

Lambda-PFLOTRAN 1.0: Workflow for Incorporating Organic Matter Chemistry Informed by Ultra High Resolution Mass Spectrometry into Biogeochemical Modeling

Katherine A. Muller¹, Peishi Jiang¹, Glenn Hammond¹, Tasneem Ahmadullah¹, Hyun-Seob Song², Ravi Kukkadapu¹, Nicholas Ward³, Madison Bowe³, Rosalie K. Chu¹, Qian Zhao¹, Vanessa A. Garayburu-Caruso¹, Alan Roebuck³, Xingyuan Chen¹

¹ Pacific Northwest National Laboratory, Richland, WA 99352, USA

² Department of Biological Systems Engineering, University of Nebraska—Lincoln, Lincoln, Nebraska, USA

³ Pacific Northwest National Laboratory, Sequim WA 98382, USA

Correspondence to: Katherine Muller (katherine.muller@pnnl.gov)

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Abstract. Organic matter (OM) composition plays a central role in microbial respiration of dissolved organic matter and subsequent biogeochemical reactions. Here, a direct connection of organic matter chemistry and thermodynamics to reactive transport simulators has been achieved through the newly developed Lambda-PFLOTRAN workflow tool that succinctly incorporates carbon chemistry data generated from Fourier transform ion cyclotron resonance mass spectrometry (FTICR-MS) into reaction networks to simulate organic matter degradation and the resulting biogeochemistry. Lambda-PFLOTRAN is a python-based workflow, executed through a Jupyter Notebook interface, that digests raw FTICR-MS data, develops a representative reaction network based on substrate-explicit thermodynamic modeling (also termed lambda modeling due to its key thermodynamic parameter λ used therein), and completes a biogeochemical simulation with the open source, reactive flow and transport code PFLOTRAN. The workflow consists of the following five steps: configuration, thermodynamic (lambda) analysis, sensitivity analysis, parameter estimation, and simulation output and visualization. Two test cases are provided to demonstrate the functionality of the Lambda-PFLOTRAN workflow. The first test case uses laboratory incubation data of temporal oxygen depletion to fit lambda parameters (i.e., maximum utilization rate and microbial carrying capacity). A slightly more complex second test case fits multiple lambda formulation and soil organic matter release parameters to temporal greenhouse gas generation measured during a soil incubation. Overall, the Lambda-PFLOTRAN workflow facilitates upscaling by using molecular-scale characterization to inform biogeochemical processes occurring at larger scales.

30 **1 Introduction**

31 Microbial respiration of dissolved organic carbon (DOC) is a main driver of environmental biogeochemical processes.
32 Mechanistic biogeochemical models often rely on lumping organic matter into a few distinct carbon pools (e.g.,
33 dissolved, sorbed, mineral associated or refractory, labile, etc.) (e.g., Fatichi, et al., 2019, Robertson et al., 2019, Wang
34 et al., 2013) but do not fully consider the properties of the organic matter (OM) compounds individually. Pooled
35 carbon approaches have benefits, such as assigning variable levels of bioavailability, however, this approach does not
36 capture the complex temporal dynamics of respiration driven by OM composition, as aerobic respiration rates have
37 been linked to organic carbon concentration, thermodynamics of the OM (Stegen et al., 2018, Garayburu-Caruso et
38 al., 2020), as well as the diversity of OM compounds present (Lehmann et al. 2020, Stegen et al., 2022). Such findings
39 highlight the importance of incorporating individual OM chemistry into biogeochemical modeling to capture, and
40 ultimately predict, system behavior more accurately.

41 There are many advanced instrumentation techniques capable of detecting and identifying individual OM formulae
42 that comprise a bulk OM sample (e.g., GC-MS, HPLC-MS, Fourier transform ion cyclotron resonance mass
43 spectrometry [FTICR-MS], etc.). For instance, FTICR-MS is a powerful, high-resolution, method that identifies
44 molecular formulae for individual organic compounds. In any given environmental sample, FTICR-MS (or other ultra
45 high-resolution methods) will typically resolve thousands of discrete OM molecular formulae, each with a unique
46 mass and elemental composition (Cooper et al., 2020, Bahureksa et al., 2021). However, untargeted analytical
47 techniques like FTICR-MS are only able to determine if a compound is present and cannot quantify the total
48 concentration associated with each organic matter molecule. Still, such techniques do provide immense amounts of
49 characterization data encompassing a deeper analytical window than measuring a small number of individual
50 biomarkers quantitatively (e.g., Ward et al., 2013). Utilizing such high-resolution molecular data in reactive transport
51 modeling frameworks affords new opportunity to advance carbon cycling in terrestrial, riverine and coastal systems
52 despite of various theoretical and computational challenges.

53 Substrate-explicit thermodynamic modeling (SXTM) provides an avenue for incorporating individual OM reactivity
54 based on thermodynamics (Song et al., 2020) into reactive transport models. The SXTM procedure takes the individual
55 chemical formula derived from FTICR-MS (or another high-resolution technique) and uses its thermodynamic
56 properties to generate an oxidation reaction for each molecular formula present in a sample. The corresponding
57 reaction stoichiometry is then determined by considering catabolic, anabolic, and metabolic reactions and balancing
58 energy for the overall metabolic reaction, allowing for the development of an aerobic respiration expression for each
59 OM formula.

60 Still, the sheer number of compounds identified in each sample proves difficult for model integration. Typically,
61 reactive transport simulators consider only a small number of primary species in their reaction networks, and most
62 could not support modeling each of the thousands of organic matter molecules individually. Here, the developed
63 Lambda-PFLOTRAN workflow addresses this challenge through grouping, or binning, similar compounds based on

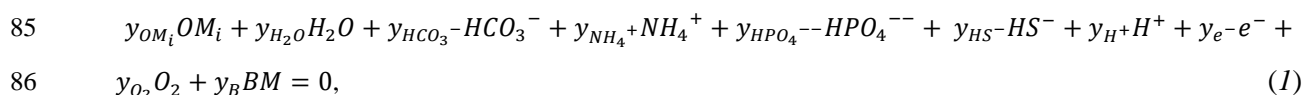
64 their thermodynamic properties, allowing for the number of species considered within the reaction network to be
65 reduced, and thus decreasing the required computational resources.

66 Lambda-PFLOTRAN is a python-based workflow that digests raw FTICR-MS data, develops a representative reaction
67 network based on substrate-explicit thermodynamic modeling (Song et al., 2020), and completes a biogeochemical
68 simulation with the open source, parallel reactive flow and transport code, PFLOTRAN (Hammond et al., 2014).
69 PFLOTRAN is developed under an open source, GNU LGPL license. The term ‘lambda’ is used here because λ is a
70 key parameter in the SXTM, which quantifies thermodynamic favorability of aerobic respiration of OM. The
71 connection between the unique reaction network developed for each FTICR-MS sample hinges on the use of
72 PFLOTRAN’s reaction sandbox capability (Hammond, 2022). The reaction sandbox gives the ability to define
73 additional custom, kinetic reactions beyond standard formulations (e.g., mineral precipitation-dissolution, Michaelis-
74 Menten, etc.). The Lambda-PFLOTRAN workflow enables upscaling by using molecular-scale information to inform
75 larger scale biogeochemical processes occurring throughout a watershed which can be simulated with PFLOTRAN.
76 Herein we describe the Lambda-PFLOTRAN workflow process including the governing expressions, workflow steps,
77 data requirements, as well as the associated assumptions and limitations. Two illustrative test cases are also included
78 to demonstrate the workflow.

79 **2 Methods**

80 **2.1 Conceptual Model**

81 Respiration modeling herein is based on thermodynamic theory by Desmond-Le Quemener and Bouchez (2014) which
82 was updated for multiple OM formulas by Song et al. (2020). The generalized form of OM molecule is assumed to
83 take the form of $C_aH_bN_cO_dP_eS^z_f$. Each molecular formula then undergoes respiration (i.e., reaction with oxygen) based
84 on the following general reaction expression:



87 This generalized expression is used to describe the oxidation of any OM molecule, i , and has been normalized to one
88 mole of biomass (BM) produced. BM is assumed to have a formula of $CH_{1.8}O_{0.5}N_{0.2}$ (Stephanopoulos et al.,
89 1998; Kleerebezem and Van Loosdrecht, 2010). OM_i represents the OM molecules as informed by FTICR-MS. Each
90 y represents the reaction stoichiometry for that reactant ($y < 0$) or product ($y > 0$). While this expression is specific for
91 cases where oxygen is the electron acceptor, such an expression could be updated for alternative electron acceptors.

92 Substrate-explicit thermodynamic modeling expressions developed from Song et al. (2020) were implemented in a
93 reaction sandbox within PFLOTRAN. The expressions were implemented in a general manner allowing for flexibility
94 in handling variations in FTICR-MS data and several user adjustable analysis configurations.

95 The microbial growth kinetics are described by Eq. (2):

$$96 \quad \mu_i^{kin} = \mu^{max} \exp\left(-\frac{\alpha|y_{OM,i}|}{1000V_h[OM_i]}\right) \exp\left(-\frac{\alpha|y_{O_2,i}|}{1000V_h[O_2]}\right), \quad (2)$$

97 where μ_i^{kin} is the unregulated uptake rate of reaction for OM_i [hr^{-1}], μ^{max} is the maximal microbial growth rate [hr^{-1}], $y_{OM,i}$ is the stoichiometry for OM_i [$mol-OM \cdot mol-biomass^{-1}$], V_h is microbial harvest volume [m^3]. Given the
98 physical interpretation of V_h as the microbial harvest volume, it is assumed here that the value of V_h is the same for
99 both OM_i and O_2 , $[OM_i]$ is the organic matter concentration of OM_i [$mol-OM \cdot L^{-1}$], $y_{O_2,i}$ is the stoichiometry for O_2
100 for respiration of OM_i [$mol-O_2 \cdot mol-biomass^{-1}$], $[O_2]$ is oxygen concentration [$mol-O_2 \cdot L^{-1}$], α is a microbial unit
101 conversion [$mol-biomass$] and 1000 is the conversion of m^3 to L.
102

103 Further, using a cybernetic modeling approach (after Song et al., 2018), all the unregulated uptake rates (μ_i^{kin}) are
104 normalized by the sum of unregulated uptake rates across all reactions, i following Eq. (3):

$$105 \quad u_i = \frac{\mu_i^{kin}}{\sum_{i=1}^n \mu_i^{kin}} \quad (3)$$

106 where u_i is the fraction of the unregulated rate [-]. The final regulated rate, r_i [hr^{-1}] for each reaction is then computed
107 following Eq. (4):

$$108 \quad r_i = u_i \mu_i^{kin}, \quad (4)$$

109 For implementation within PFLOTRAN, the use of inhibition terms was required to prevent negative concentrations
110 once a reactant is nearly depleted. For a reaction to proceed, all reactant species must be present above a minimum
111 concentration even if the molecules do not explicitly control the respiration rate (i.e., species other than OM and O_2 ,
112 Eq. (2). If a reactant concentration falls below a threshold concentration, the respiration rate is inhibited. Reactant
113 inhibition is computed by Eq. 5 (Kinzelbach et al., 1991) for reactant species j :

$$114 \quad I_j = 0.5 + \frac{\arctan([C_j] - C_{thj})f}{\pi}, \quad (5)$$

115 where $C_{th,i}$ is the threshold concentration [M], f is the threshold scaling factor [-]. The default C_{thj} is 10^{-20} M.

116 The reaction rates are also inhibited by the microbial carrying capacity of the system, I_{cc} , as follows in Eq. (6):

$$117 \quad I_{cc} = 1 - \frac{[BM]}{CC} \quad (6)$$

118 where [BM] is the biomass concentration [$mol-BM \cdot L^{-1}$], CC is the biomass carrying capacity [$mol-BM \cdot L^{-1}$]. I_{cc} has a
119 non-negativity constraint, so if $[BM] > CC$, then $I_{cc} = 0$.

120 These inhibition factors are applied to the overall rate expression as shown in Eq. (7).

121 $r_{i,inhibited} = r_i I_{CC} \prod I_j \quad \forall y_{i,j} < 0,$ (7)

122 The overall individual species rates, $d[C_j]/dt$, [mol-species·L⁻¹·hr⁻¹] are then computed as follows with Eq. (8):

123 $\frac{dC_j}{dt} = (\sum_{i=1}^n y_{i,j} r_{i,inhibited}) [BM],$ (8)

124 where j is the species index. The total number of species includes 7 general species (i.e., HCO₃⁻, NH₄⁺, HPO₄⁻, HS⁻,
 125 H⁺, O₂, BM (i.e., Eq (1)) and the OM species considered (i.e., typically 10). i is the reaction index, n is total number
 126 of reactions as based on the total number of OM species (typically, with this workflow $n = 10$). $y_{i,j}$ is the coefficient
 127 for species j in reaction i .

128
 129 The expression for biomass is also modified to account for biomass decay (note all biomass stoichiometries are 1 by
 130 definition):

131 $\frac{dBM}{dt} = (\sum_{i=1}^n y_{i,j} r_{i,inhibited}) [BM] - k_{deg} [BM],$ (9)

132 where k_{deg} is the biomass decay rate [hr⁻¹].

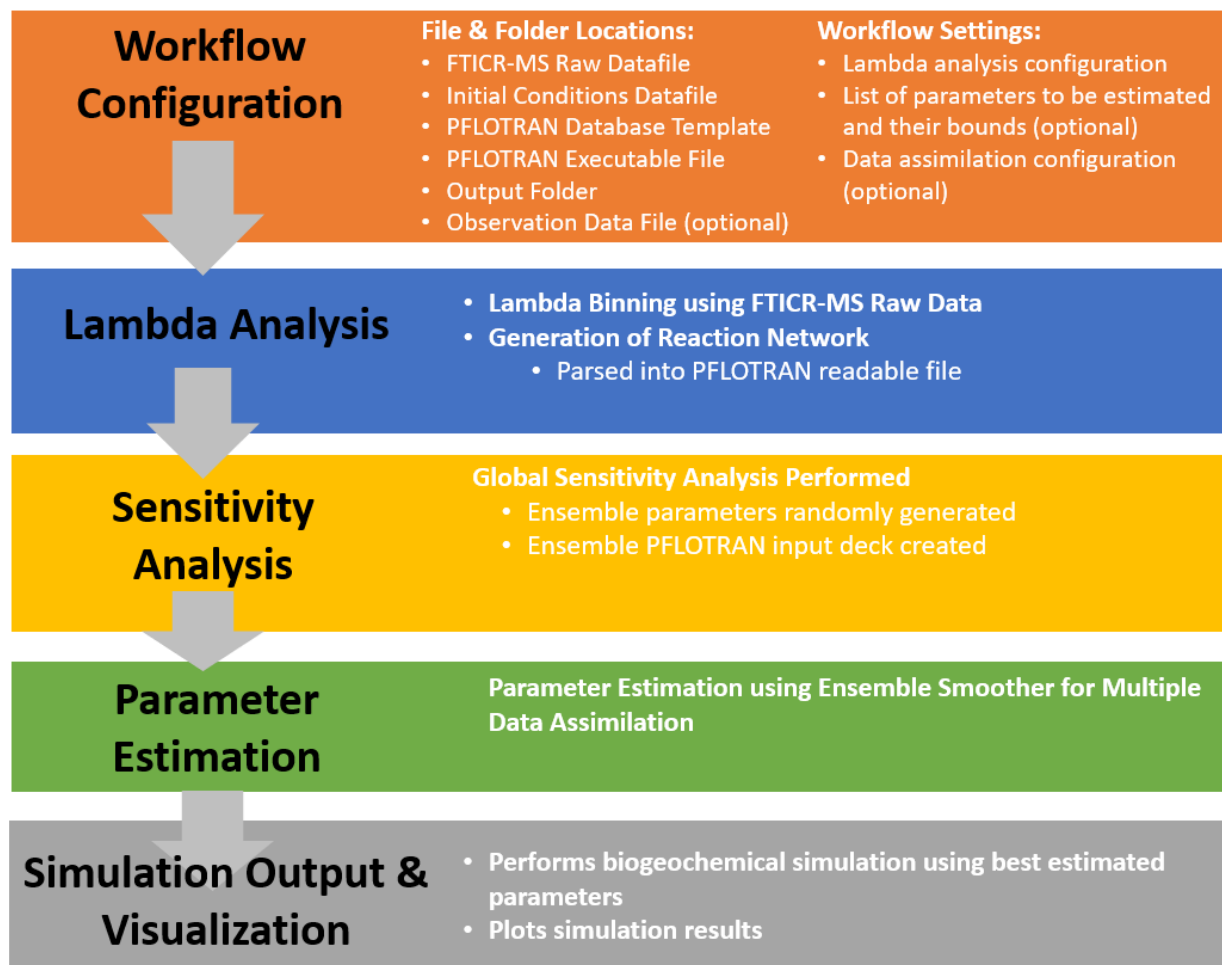
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134 2.2 Lambda Analysis and Binning

135 To reduce the number of organic compounds considered in the simulation, OM molecules are grouped, or binned,
 136 based on their λ value computed by Eq. (10):

137 $\lambda = \frac{\Delta G_{r,anabolic} + \Delta G_{r,dissipation}}{(-\Delta G_{r,catabolic})},$ (10)

138 where ΔG are the Gibbs energies for the anabolic and catabolic reactions and the associated dissipation energy,
 139 respectively. The value of λ is indicative of how many times the catabolic reaction needs to be completed to provide
 140 the energy required to synthesis one mole of biomass. Lower λ values suggest higher thermodynamic favorability of
 141 OM respiration. Using the chemical formula determined for each OM molecule, the energy balance equations are
 142 solved providing the overall reaction stoichiometry Eq. (1) and the λ is calculated. Using the λ value for each molecule,
 143 the cumulative probability distribution for the sample is produced (Figure 2).

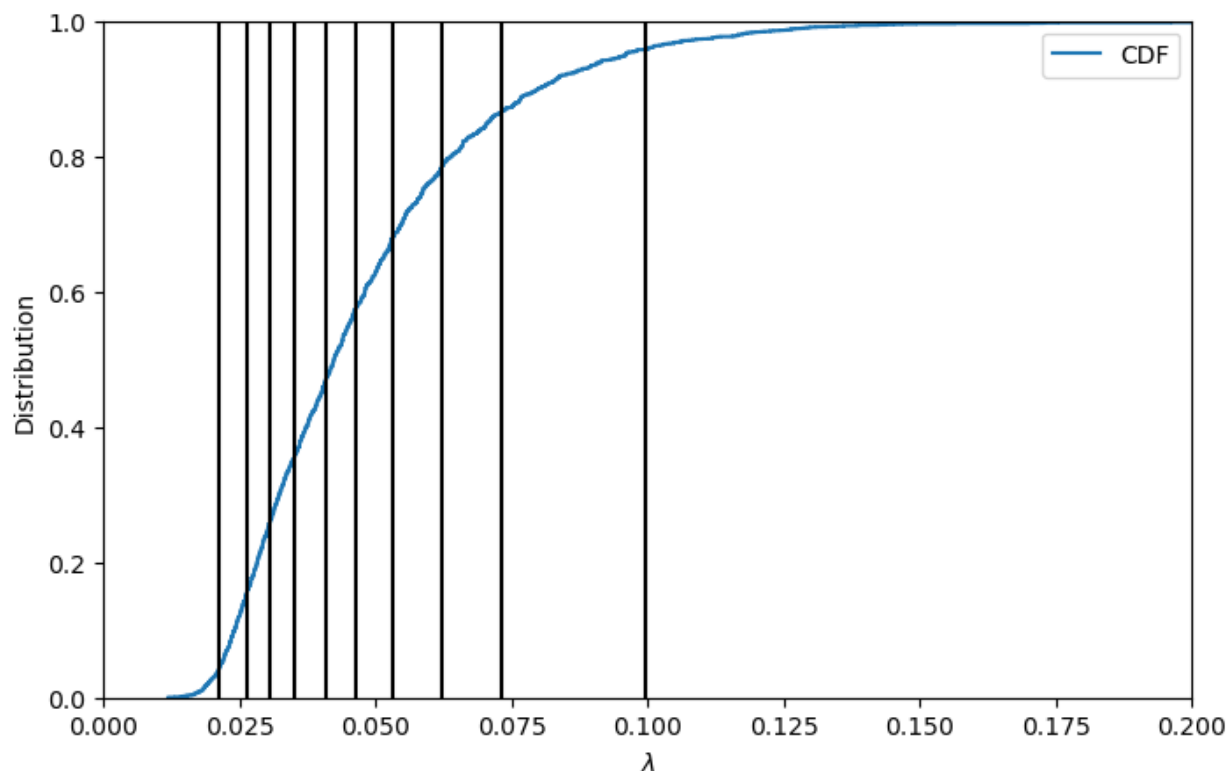


144

145 **Figure 1:** Flow Chart of the Lambda-PFLOTRAN Workflow.

146

147 It is this conversion from individual compounds to a distribution that is critical for reducing the entire sample down
 148 to a representative set of expressions. The λ bins are then formed by splitting the cumulative probability distribution
 149 into equally weighted sections as which to define the overall sample by. The illustrative example shown in Fig. 2
 150 demonstrates the sample distribution being divided into 10 sections (i.e., in this case each section contains 10% of the
 151 overall sample distribution).



152
 153 **Figure 2:** Lambda binning to convert raw FTICR-MS into a representative reaction network using the cumulative probability
 154 distribution function (CDF) for Test Case 1a. Vertical lines display the average λ value for each of the 10 bins (left to right, λ bin
 155 1 to 10).

156
 157 Each section is used to determine a representative organic matter formula and the associated reaction and
 158 stoichiometry of that λ bin. The group of representative reactions (one per bin) is called the reaction network. A
 159 demonstrative reaction network defined by λ analysis and binning is shown in Table 1.

160
 161 **Table 1:** Reaction Network Developed from Lambda Theory for Test Case 1a

Bin Number	Representative Organic Matter Species Formula	λ	y_{OM}	$y_{HCO_3^-}$	$y_{NH_4^+}$	$y_{HPO_4^{2-}}$	y_{HS^-}	y_{H^+}	y_{O_2}
1	$C_{31}H_{44}N_{0.33}O_{4.8}P_{0.6}S_{0.3}$	0.021	-0.05	0.64	-0.17	-0.18	0.03	0.02	-1.07
2	$C_{26}H_{39}N_{0.20}O_{7.0}P_{0.6}S_{0.1}$	0.026	-0.07	0.68	-0.10	-0.19	0.04	0.01	-1.06
3	$C_{22}H_{36}N_{0.24}O_{7.5}P_{0.5}S_{0.1}$	0.031	-0.08	0.69	-0.02	-0.18	0.04	0.01	-1.06
4	$C_{20}H_{32}N_{0.28}O_{7.3}P_{0.4}S_{0.1}$	0.035	-0.08	0.72	-0.08	-0.18	0.04	0.01	-1.05
5	$C_{19}H_{29}N_{0.48}O_{7.9}P_{0.3}S_{0.2}$	0.041	-0.09	0.79	-0.17	-0.16	0.03	0.02	-1.04
6	$C_{18}H_{26}N_{0.68}O_{8.1}P_{0.2}S_{0.2}$	0.046	-0.10	0.85	-0.27	-0.13	0.02	0.02	-1.03
7	$C_{17}H_{24}N_{0.69}O_{8.1}P_{0.2}S_{0.2}$	0.053	-0.11	0.90	-0.32	-0.12	0.02	0.02	-1.02
8	$C_{15}H_{20}N_{0.67}O_{7.6}P_{0.2}S_{0.2}$	0.062	-0.13	0.94	-0.42	-0.11	0.02	0.03	-1.00

9	$C_{13}H_{19}N_{1.13}O_{87.4}P_{0.1}S_{0.2}$	0.073	-0.15	1.01	-0.48	-0.03	0.01	0.03	-1.00
10	$C_{10}H_{15}N_{1.56}O_{6.5}P_{0.1}S_{0.2}$	0.100	-0.21	1.17	-0.75	0.12	0.01	0.04	-0.97

162
163 Currently, the representative OM molecule that defines each bin is computed as the average chemical formula of all
164 the molecules present in that λ section. The disadvantage of this approach is that unrealistic compounds are defined
165 as representative molecules instead of realistic molecules. The issue with selecting a single, but real compound, from
166 within each λ section resides in chemical complexity and variation - for instance some molecules may contain low
167 levels of phosphorous or sulfur and others may not contain either element in the chemical formula. Thus, requiring
168 the representative chemical formula to be a real compound present in the sample would create basis which would
169 propagate through the reaction network and into the resulting biogeochemical simulation results.

170 **2.3 Lambda-PFLOTRAN Workflow**

171 The Lambda-PFLOTRAN workflow digests raw FTICR-MS data, calculates the λ distribution for the sample,
172 generates the λ bins and corresponding reaction network, and completes a biogeochemical simulation using
173 PFLOTRAN. Further, we incorporated sensitivity analysis and ensemble data assimilation to enable an in-depth
174 exploration of the impact of reaction parameters on respiration as well as a straightforward parameter estimation
175 method to fit model parameters to experimental data.

176 The workflow is implemented through a user-friendly Jupyter notebook interface (Kluyver et al., 2016) where a user
177 can configure the simulation parameters by adjusting initial concentrations, λ binning configuration, parameter values
178 and/or ranges, and data assimilation options. Based on the user's data file and the associated parameters, scripts within
179 the Jupyter notebook write the corresponding PFLOTRAN input files, including OM molecules and aqueous
180 chemistry. The PFLOTRAN simulations are completed locally through a Docker container making this capability
181 much more user-friendly and accessible. The progress of the data assimilation tool used for parameter fitting is
182 illustrated within the Jupyter notebook. The resulting best fit final biogeochemical simulation is output visually with
183 plots and as a text file (when applicable).

184 The Lambda-PFLOTRAN workflow steps are shown in Figure 1 and described in detail in the following subsections:

185 **2.3.1 Step 1 – Workflow configuration**

186 The first step is to set up the workflow configuration for a Lambda-PFLTORAN application. This includes specifying
187 the file and folder locations of the following information: 1) FTICR-MS raw data file (.csv), 2) initial species
188 concentrations file (.csv) that includes starting molar concentrations for HCO_3^- , NH_4^+ , HPO_4^{2-} , HS^- , H^+ , O_2 (aq), BM
189 and total organic carbon (TOC), 3) PFLOTRAN database template file, 4) PFLOTRAN executable file, 5) workflow
190 output folder, and if completing parameter estimation, (6) the data observation file (.csv), if applicable.

191 The user is also asked to configure workflow settings related to: (1) the lambda analysis configuration, including
192 number of λ bins and method to define the λ bins (i.e., cumulative vs uniform); (2) the respiration modeling parameter

193 setup, including the list of the parameters to be estimated and their associated upper and lower bounds and (3) the data
194 assimilation configuration (see below).

195 **2.3.2 Step 2 – Organic Matter Chemistry using Lambda Analysis**

196 With only an input of FTICR-MS data, the workflow first performs the lambda analysis (Section 2.2) to group OM
197 molecules into various λ bins based on each compound's thermodynamics (Figure 2) and produce the corresponding
198 reaction network for respiration (Table 1). The default number of λ bins is 10, although this can be adjusted in the
199 workflow configuration by the user, if desired. The generated reaction network is then automatically parsed by the
200 workflow into a text file that can be read by PFLOTRAN.

201 **2.3.3 Step 3 – Sensitivity Analysis using Mutual Information**

202 This step performs the global sensitivity analysis on the parameters to be estimated. Ensemble parameters are first
203 generated by randomly sampling from their predefined ranges in the configuration step and saved into an HDF5 file.
204 Then, the workflow generates a PFLOTRAN input deck to conduct ensemble simulations using the ensemble
205 parameters. The generated ensemble model states enables a global sensitivity analysis using mutual information
206 (Cover and Thomas, 2006; Jiang et al, 2022) as follows:

$$207 \quad I(X;Y) = H(Y) - H(Y|X) = \sum_{x=x} \sum_{y=y} p(x,y) \log \left(\frac{p(x,y)}{p(x)p(y)} \right), \quad (11)$$

208 where x and y are the specific values of X and Y , respectively; $H(Y)$ is the Shannon's entropy of Y ; $H(Y|X)$ is the
209 conditional entropy of Y given X ; p is the probability density function. Higher I indicate stronger sensitivity between
210 X and Y . Besides sensitivity analysis, the ensemble parameter/states also serve as the prior information for parameter
211 estimation at the next step.

212 **2.3.4 Step 4 – Parameter Estimation using Ensemble Smoother for Multiple Data Assimilation**

213 The workflow adopts Ensemble Smoother for Multiple Data Assimilation (Emerick and Reynolds, 2013; Jiang et al,
214 2021), abbreviated as ESMDA, for data assimilation in this step. Rooted in ensemble Kalman filter, ESMDA is an
215 iterative data assimilation approach that assimilates the observations on the entire time period for multiple times to
216 reduce the uncertainty of the estimated or posterior parameters. During each iteration of ESMDA, the model
217 parameters are updated based on the following equation:

$$218 \quad m_{k,l}^u = m_{k,l}^f + C_{MD,l}^f (C_{DD,l}^f + \alpha_l C_D)^{-1} \left(d_{obs} + \sqrt{\alpha_l} C_D^{\frac{1}{2}} z_k - d_{k,l}^f \right), \quad k = 1, \dots, N_e \text{ and } l = 1, \dots, L, \quad (12)$$

219 where the subscripts k and l are the indices of the ensemble member and the iteration, respectively; the superscripts u
220 and f are the updated and forecast parameters or states, respectively; N_e is the number of ensemble members; L is the
221 number of iterations; $m_{k,l}^f$ and $m_{k,l}^u$ are the k th ensemble member of the forecast/prior and updated/posterior
222 parameters, respectively, at the l th iteration; d_{obs} is the observation; z_k is the observation noise sampled from

223 independent standard normal distributions for the k th ensemble member; $d_{k,l}^f$ is the k th ensemble member of the
224 predicted observation states by the model using $m_{k,l}^f$; $C_{MD,l}^f$ is the cross-covariance matrix between the prior parameters
225 m_l^f and the predicted observation states d_l^f ; $C_{DD,l}^f$ is the auto-covariance matrix of the predicted observation states d_l^f ;
226 C_D is the auto-covariance matrix of the observation error; and α_l is the inflation coefficient at the l th iteration with the
227 sum of all α_l equal to one.

228
229 Here, the assimilation starts with taking the ensemble model parameters/states in Step 3 and the provided observations,
230 and calculates the posterior parameters using ensemble Kalman filter, updates the prior parameters with the current
231 posterior for the next iteration, and then repeats the whole process for multiple times (typically 3 to 5 iterations, as
232 defined by the user). The final estimated parameters are obtained from the posterior parameter at the last iteration and
233 are updated in the parameter HDF5 file. The parameter estimation is implemented in a way that allows assimilating
234 either a single (e.g., Test Case 1) or multiple observed species simultaneously through a simple change of the inputs.
235 For example, if temporal experimental or field data is available for oxygen, pH, and total carbon, all these data sources
236 could be simultaneously fit to, with only minor adjustments to Jupyter notebook.

237 **2.3.5 Step 5 – Simulation Output and Visualization**

238 The last step performs the ensemble simulation of the biogeochemical modeling a final time using the estimated
239 parameters in Step 4. Optionally, users can further pick the realization with the best performance. The user has the
240 option to select their preferred goodness of fit metric from the following options as a means for selecting the best
241 performing simulation: R-squared (R^2), Root Mean Squared Error (RMSE), Modified Kling-Gupta Efficiency
242 (mKGE), Nash-Sutcliffe Model Efficiency Coefficient (NSE), or Correlation Coefficient (CorC). Based on the
243 selection, the final time series of aqueous chemistry, oxygen consumption, CO_2 production, and lambda binned, and
244 total organic carbon concentrations will be computed and plotted.

245 **3 Test Cases**

246 **3.1 Test Case 1 - Oxygen Depletion Incubation Experiments.**

247 In the first illustrative example, the workflow was used to fit μ_{max} to laboratory incubation experiments where oxygen
248 levels were measured over two hours in a closed reactor. The incubation experiments were completed as part of the
249 Worldwide Hydrobiogeochemistry Observation Network for Dynamic River Systems (WHONDRS) program
250 (Goldman et al, 2020). For these incubations, sediment was taken from three locations within a stream (i.e., upstream
251 [Test Case 1a], midstream [Test Case 1b], and downstream [Test Case 1c]) in the Yakima River Basin in Washington,
252 USA for subsequent laboratory respiration experiments. FTICR-MS was used to determine the OM chemistry from
253 each sediment sample, resulting in variable formulae being identified in each sample. Formula assignments for all the
254 samples included herein were completed using formultitude (Tolic et al., 2017). Total dissolved organic carbon
255 concentration paired with the FTICR-MS sample and biomass measurements taken at the start of each experiment
256 were used as the initial concentrations for each of the simulations. Due to the absence of quantitative data related to

257 how the total carbon mass is distributed between various the OM compounds, the total carbon concentration (on a per-
258 C basis) was assumed to be split equally between each of the λ bins. The total organic carbon concentration was
259 distributed into each λ bin using Eq. (13). While this assumption results in equal distribution of carbon between the
260 bins, consequently, it assigns different initial species concentrations due to varying carbon concentrations between the
261 molecules.

$$262 \quad [C_{\lambda bin}]_0 = \frac{[TOC]}{n_{\lambda bin} n_{C_{\lambda bin}}} \quad (13)$$

263 Where: $[C_{\lambda bin}]_0$ is the initial species concentration in each λ bin [$\text{mol} \cdot \text{L}^{-1}$]; TOC is the total organic carbon measured
264 [$\text{mol-carbon} \cdot \text{L}^{-1}$]; $n_{\lambda bin}$ is the number of λ bins [-]; and $n_{C_{\lambda bin}}$ is the number of carbon molecules in the assumed
265 formula for the λ bins [$\text{mol-carbon} \cdot \text{mol-molecule}^{-1}$].

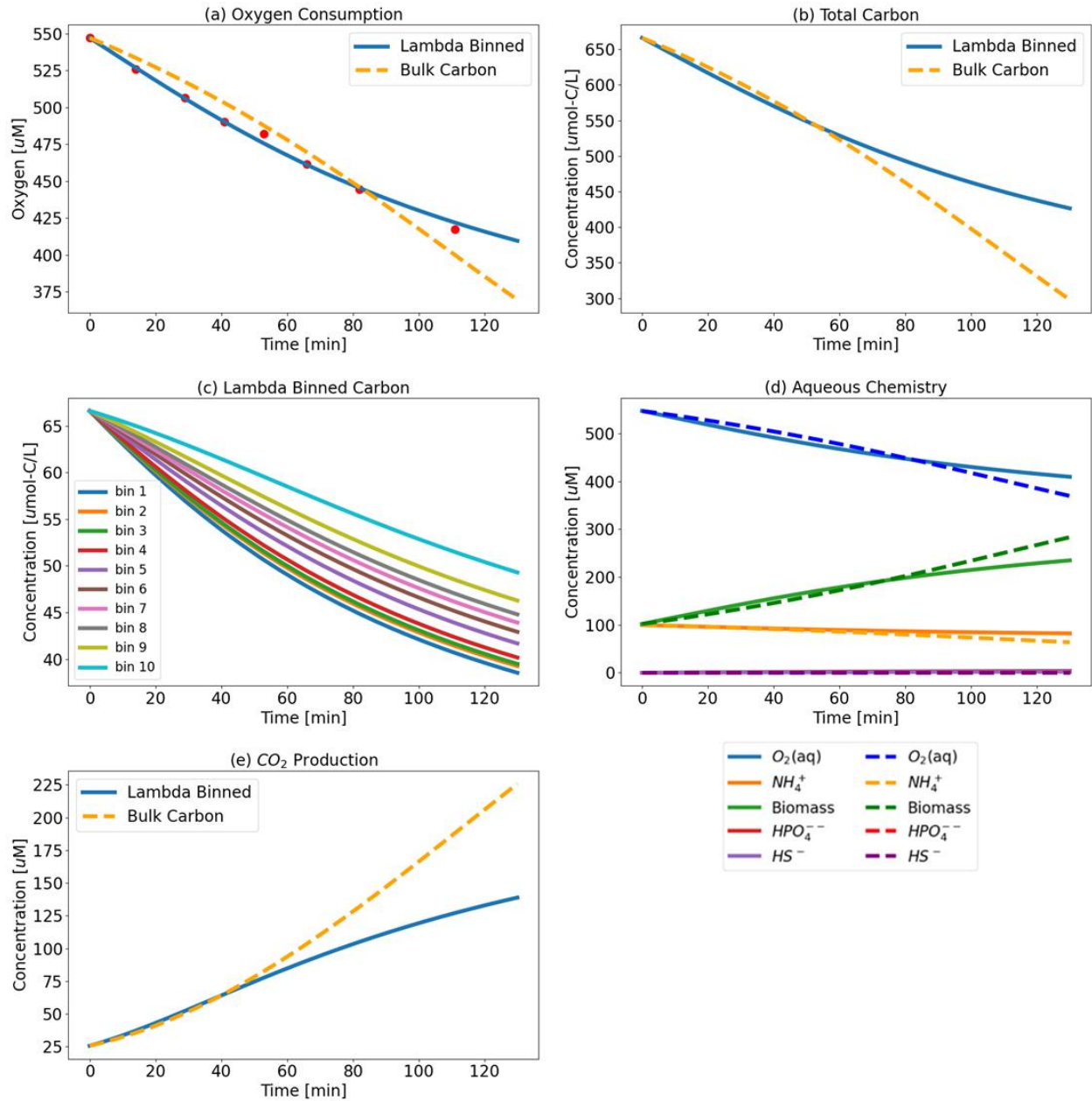
266 Using the Lambda-PFLOTRAN workflow, the FTICR-MS data from each laboratory experiment was digested into
267 the corresponding λ bins to create the individual reaction network. The Jupyter Notebook for this example is
268 “Test_Case1-WHONDERS.ipynb” and is available at <https://doi.org/10.15485/2281403>.

269
270 μ_{\max} was fit to the provided experimental oxygen data. The final lambda binned fit, along with corresponding carbon
271 consumption (individual and total) and aqueous chemistry is displayed in Figure 3 (and in the supporting information
272 Fig. S1 and S2 for Test Cases 1b [midstream] and 1c [downstream], respectively). To evaluate the use of lambda
273 binned OM obtained from FTICR-MS (Figure 3), the workflow was also run for a baseline case where μ_{\max} was fit
274 again, but this time assuming a generic bulk OM form of CH_2O for comparison. The reaction network developed for
275 a generic OM molecule of CH_2O is shown in Eq. 14.



277 This reaction network is used in the Lambda-PFLOTRAN workflow for bulk OM simulations.

278 Fitted μ_{\max} values for the lambda binned model is 0.25 min^{-1} ($R^2 = 0.99$) and fitted μ_{\max} to the bulk OM CH_2O model
279 is 0.032 min^{-1} ($R^2 = 0.96$). V_h and CC are fixed at assumed values of 10 m^3 and 1 M , respectively in both simulations.



280
 281 **Figure 3:** Test Case 1a Results – (a) Oxygen Consumption where Lambda-PFLOTRAN workflow was used to fit (blue line) to
 282 experimental respiration data (red dots) and (b) Total Carbon Consumption; (c) Individual Organic Matter Consumption by λ bin;
 283 and (d) biogeochemistry including O_2 (aq) (blue); Biomass (green); NH_4^+ (orange); HS^- (purple); and HPO_4^- (red); and (e) CO_2
 284 production for the upstream incubation. The dashed orange lines (in a, b and e) show simulation results assuming a generic OM
 285 species of CH_2O for comparison.

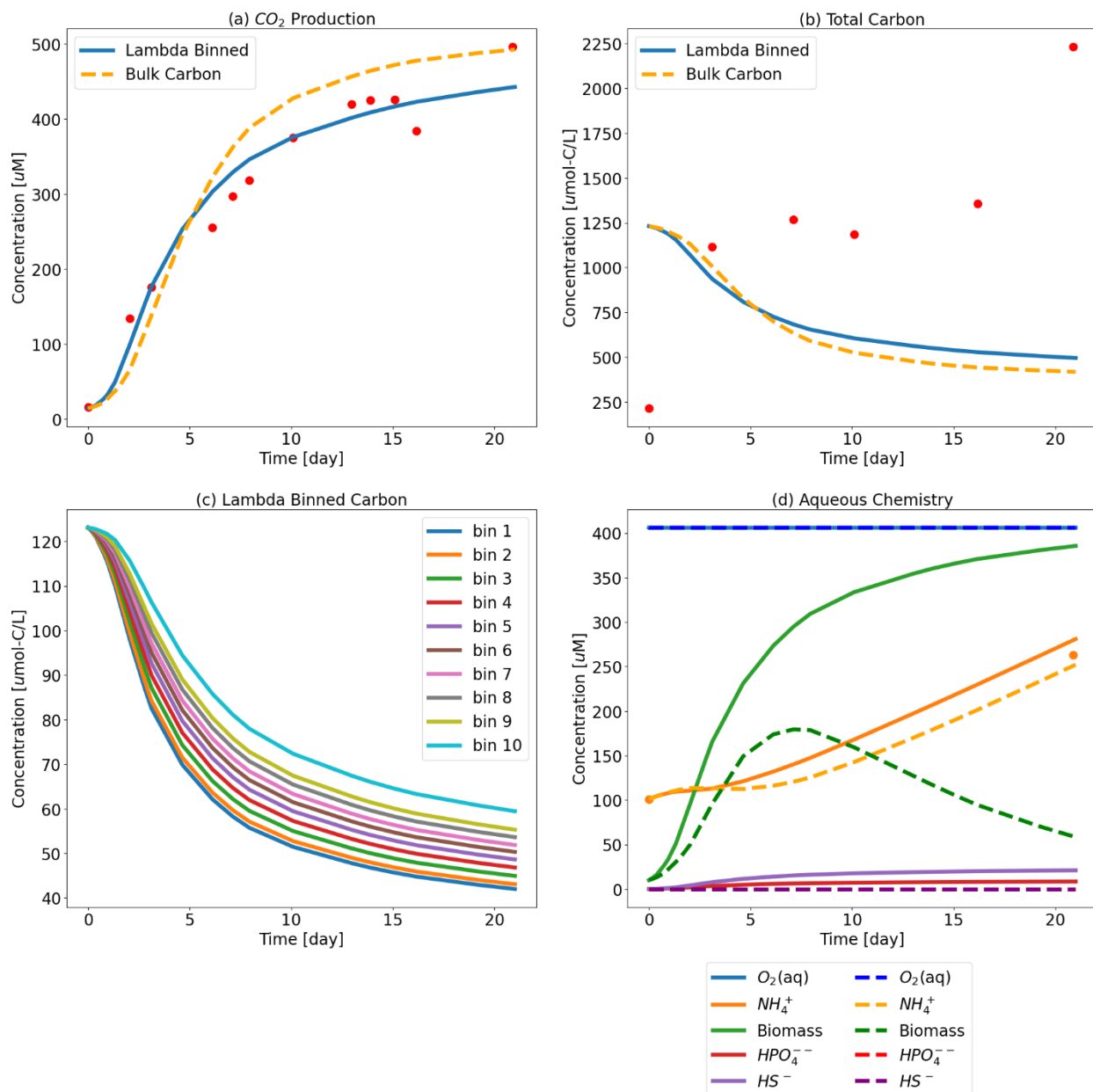
286 However, even over the short time frame of this simulation (i.e., only 120 minutes), the difference between assuming
 287 the generic CH_2O and using the more detailed organic matter chemistry resulted in different predictions of total carbon
 288 and CO_2 generation. The bulk OM model predicts more carbon consumption and greater CO_2 production than the
 289 lambda binned model. The bulk OM model estimates that 50% of the initial total carbon is consumed over the first

290 120 mins, whereas the lambda binned model predicts 34% consumption. Similarly, the bulk OM model estimates
291 approximately 35% more CO₂ generation as compared the lambda binned model. The effects on aqueous chemistry
292 over this short duration are more muted, albeit still present.

293 **3.2 Test Case 2 - Respiration Incubation Experiments.**

294 Test Case 2 uses soil respiration incubation data from Ward et al. (2023) aimed at investigating the influence of soil
295 type, oxygen condition (aerobic vs. anaerobic), and seawater exposure (fresh vs. saline) on respiration extent and rate.
296 For these experiments, temporal measurements were collected for CO₂ generation, dissolved organic carbon (DOC),
297 organic matter formulas via FTICR-MS and other bulk aqueous chemistry (i.e., pH, NH₄⁺, and other metals and ions)
298 creating a rich dataset for calibration of system specific lambda model parameters. These incubations were setup by
299 adding dry soil to the reactor and then adding water (resulting in a soil:water ratio ranging from 1:11 to 1:16). The soil
300 and water were shaken vigorously for five minutes, and then sampled for the initial time point prior to officially
301 starting the incubation. For the aerobic experiments, the reactor headspace was cycled every 24 hours to measure CO₂
302 generated but also to ensure the system was kept aerobic; this was only performed five days per week, with no
303 measurements taken on the weekend due to logistical constraints. Upon experiment completion, the increase in DOC
304 concentrations indicated organic carbon was being kinetically released from the soil into the aqueous phase over the
305 course of the 21-day experiment. Similarly, measured NH₄⁺ concentrations also increased during the experiment. To
306 address this within our reactive transport model, a source of nitrogen was assumed to be released from the soil as well
307 ($N_{release}$). Both carbon and nitrogen release are included in this example and are assumed to follow a zero-order constant
308 release rate. Any organic carbon released from the soil was fractionated into each λ bin on the same per-carbon basis
309 assumed for the initial total organic carbon. This was implemented through a dependent function that calculated the
310 release of carbon into each λ bin based on a fitted single bulk $k_{release}$ rate. Mathematically in PFLOTRAN the constant
311 oxygen conditions were implemented through a gas-liquid partitioning expression with a fast exchange term. These
312 three additional processes were added to describe the experimental conditions of Test Case 2 more accurately (i.e.,
313 release of carbon, nitrogen and sustained aerobic conditions); however, a PFLOTRAN input deck can be expanded
314 and customized to include a host of additional processes and full geochemistry for a specific system of interest. For
315 instance, aqueous complexation, mineral dissolution and precipitation, sorption, and redox reactions can be added, all
316 of which can influence the resultant pH and carbon, nitrogen, and other nutrient dynamics.

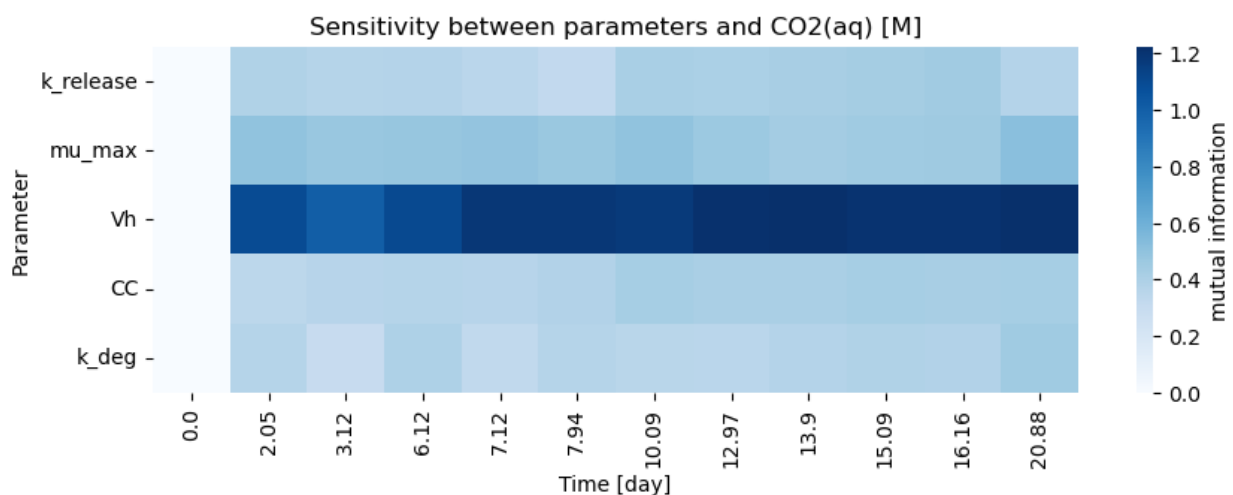
317 The workflow was used to fit μ_{max} , V_h , CC , k_{deg} , as well as $k_{release}$, to the temporal CO₂ generation for a single aerobic
318 soil incubation (Figure 5). The Jupyter Notebook for this example is “Test_Case2-Colloids.ipynb”.



320
 321 **Figure 4.** Test Case 2 Results – (a) CO₂ production where Lambda-PFLOTTRAN workflow was used to fit (blue line) to
 322 experimental respiration data (red dots) and (b) the corresponding Total Organic Carbon; (c) Individual Organic Matter
 323 Consumption by λ bin, and (d) the corresponding biogeochemistry including O₂ (aq) (blue); Biomass (green); NH₄⁺ (orange); HS⁻
 324 (purple); and HPO₄⁻ (red). Dots indicate experimental data. The dashed orange lines in the top two figures show simulation results
 325 assuming a generic OM species of CH₂O for comparison. Fitted parameters for lambda binned model are $k_{\text{release}} = 5.5 \times 10^{-12} \text{ day}^{-1}$;
 326 $\mu_{\text{max}} = 37.6 \text{ day}^{-1}$, $V_h = 5.0 \text{ m}^3$, $CC = 0.12 \text{ M}$, and $k_{\text{deg}} = 1 \times 10^{-3} \text{ day}^{-1}$ ($R^2 = 0.953$) and fitted bulk OM CH₂O model values are k_{release}
 327 $= 2.0 \times 10^{-12} \text{ day}^{-1}$; $\mu_{\text{max}} = 47 \text{ day}^{-1}$, $V_h = 1.0 \text{ m}^3$, $CC = 0.77 \text{ M}$, and $k_{\text{deg}} = 0.15 \text{ day}^{-1}$ ($R^2 = 0.909$).

328
 329 For the purposes for showcasing the workflow, five parameters were estimated in this test case example, and as a
 330 result the models are over parametrized given the amount of data available. Parameter sensitivity over the course of

331 simulation time is shown in Figure 5 and suggests this system is highly sensitive to V_h . It should be noted that both
 332 these model fits are also highly sensitive to the allowable parameter space as user defined by the lower and upper
 333 parameter bounds. In general, parameterization efforts are inherently challenging. For Lambda-PFLOTRAN, which
 334 models microbially mediated processes, it is recommended to initially focus on constraining biomass parameters (i.e.,
 335 CC, k_{deg} , and V_h) by measuring temporal changes in biomass concentrations. Further, V_h and μ_{max} are typically highly
 336 sensitive and often correlated. However, since V_h represents the theoretical volume accessible to microbes and cannot
 337 be directly measured, it is suggested to fix V_h within a range of 1-10 m^3 . If these microbial parameters can be
 338 adequately constrained, focus can shift to μ_{max} , the maximum microbial growth rate, which significantly influences
 339 overall respiration and is expected to exhibit the highest variability across different locations and conditions.



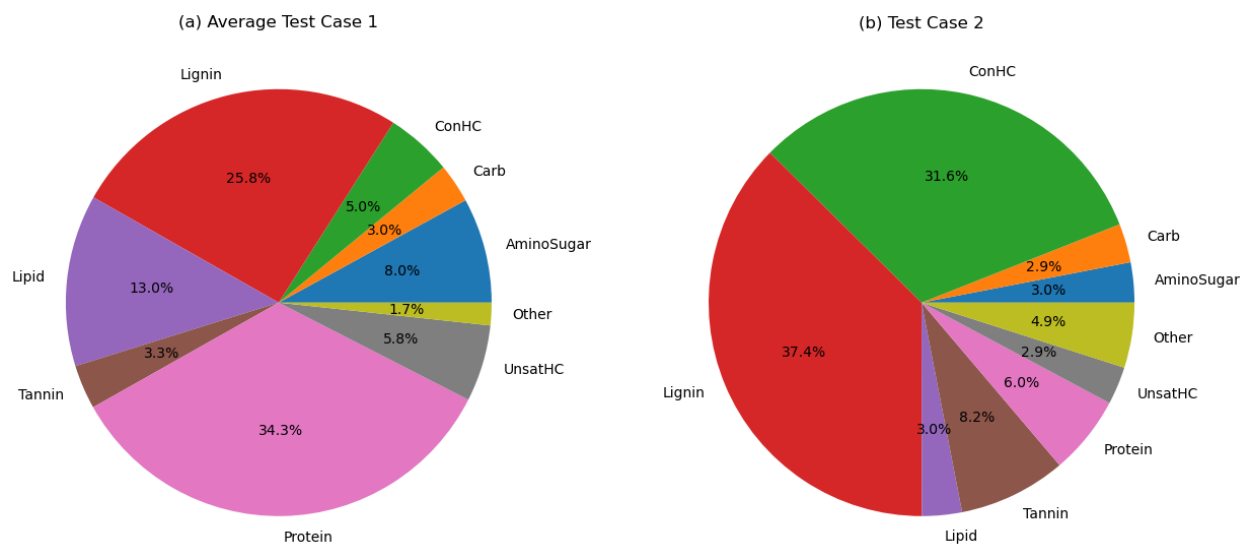
340
 341 **Figure 5.** Test Case 2 - Sensitivity Analysis Output during Parameter Estimation. The sensitivity of five fitted parameters ($k_{release}$,
 342 μ_{max} , V_h , CC, and k_{deg}) on temporal aqueous CO₂ concentrations as a function of time.

343 Any additional experimental data, either collected during incubations or through independent experiments (e.g.,
 344 carbon release from the soil in an abiotic system), would be expected to help constraint the model and improve
 345 parameterization. Additionally, it is unclear why the model is unable to capture the total organic carbon behavior in
 346 Test Case 2. One potential explanation is that some of the released organic carbon may not be fully bioavailable and
 347 thus the model may be compensating for this by artificially reducing the concentration of OM available for respiration.

348 4 Variability and Impact of Organic Matter Speciation

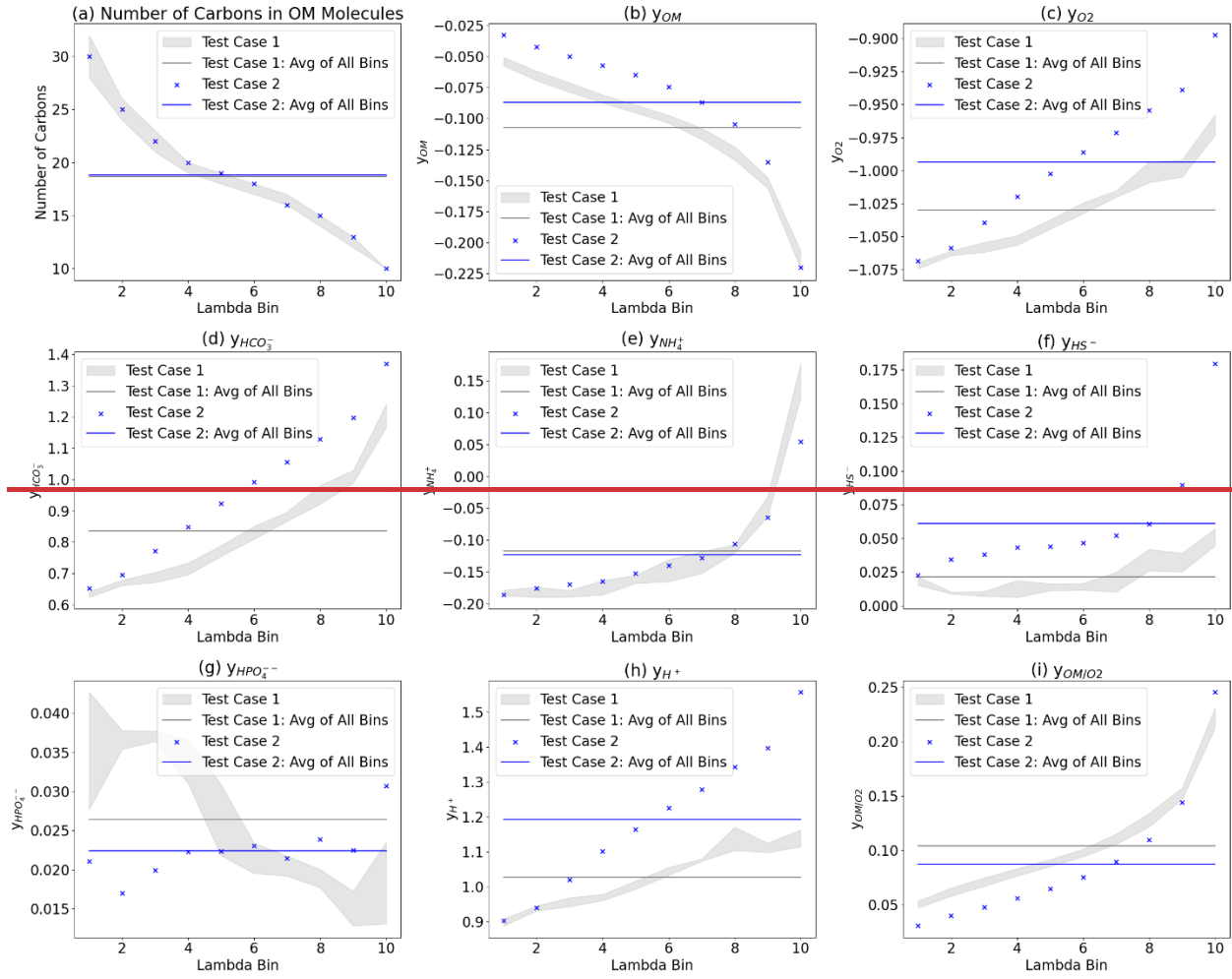
349 The variability in OM speciation was briefly assessed by comparing FTICR-MS data from Test Cases 1 and 2. Each
 350 identified OM species was classified into one of nine compound classes. For Test Case 1, the average of the three Test
 351 Case 1 samples (1a - upstream, 1b - midstream, and 1c - downstream) was computed. The predominant classes were
 352 proteins ($34 \pm 1\%$), lignin ($26 \pm 1\%$), and lipids ($13 \pm 2\%$), with the errors representing the standard deviation among
 353 the Test Case 1a-c samples. The low standard deviation suggests consistent reproducibility in OM speciation for

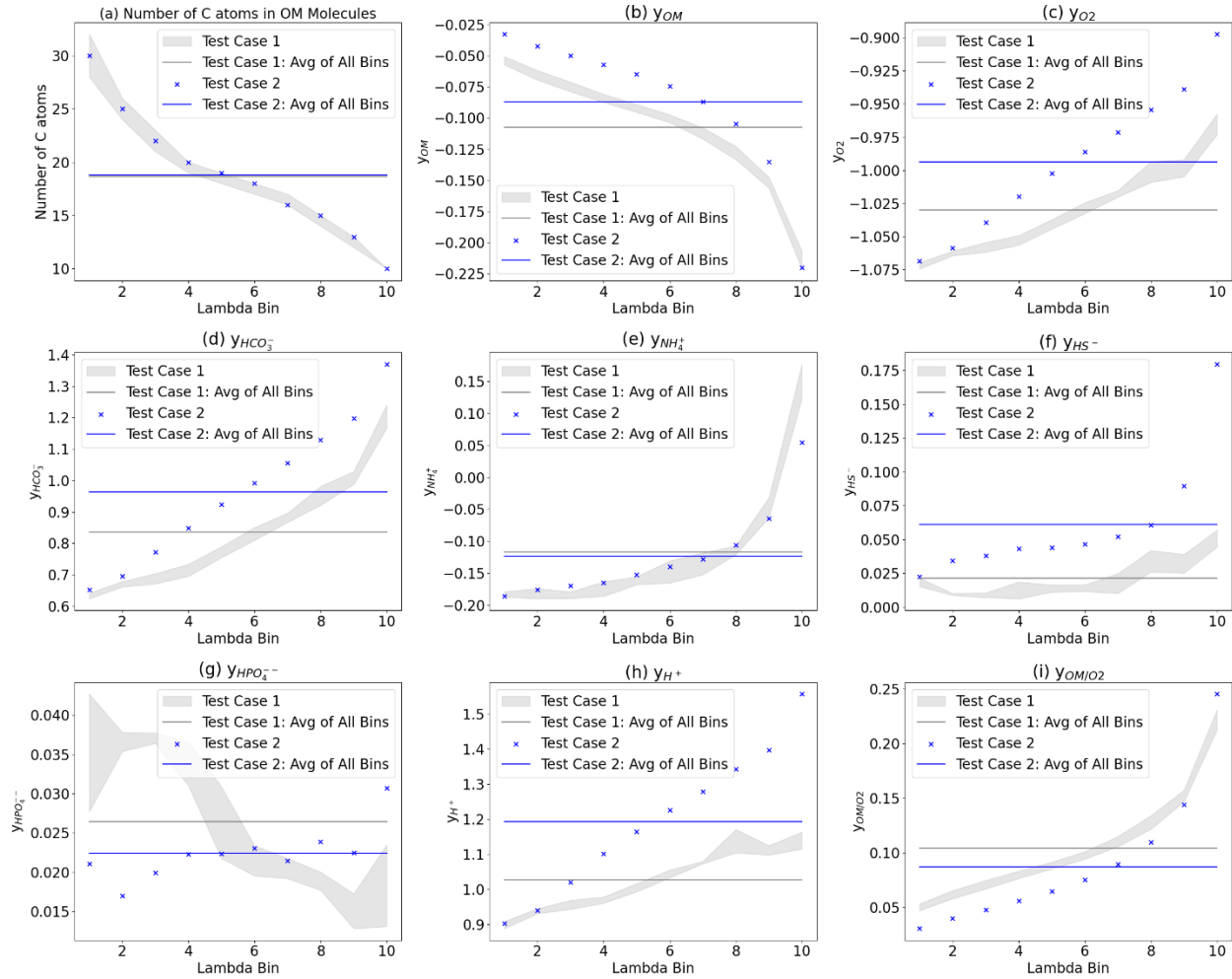
354 samples taken from nearby locations. In contrast, OM in Test Case 2 was primarily composed of lignin (37.4%) and
355 concentrated hydrocarbons (32%). The full distribution of compound classes is presented in Figure 5.



356
357 **Figure 5.** Distribution of Organic Matter Compound Classes: (a) Test Case 1 and (b) Test Case 2.
358 Note: Test Case 1 is the average of Test Case samples 1a-c. ConHC = Condensed Hydrocarbon; UnsathC = Unsaturated
359 Hydrocarbon

360
361 The influence of the sample OM speciation on the λ binned reaction networks was also assessed. Figure 6 illustrates
362 the impact of OM speciation on the corresponding λ binned reaction networks, with three key observations. First, the
363 variability in OM speciation between different samples is evident when comparing Test Case 1 and Test Case 2. To
364 enhance visual clarity, the range of Test Case 1 samples (1a-c) is depicted as a grey shaded region, showing the spread
365 between the minimum and maximum values of the three samples. For Test Case 2, data from the single FTICR-MS
366 sample is represented by blue dots. Test Case 1 and 2 have distinct λ derived reaction networks as indicated by the
367 little overlap between the grey region and the blue dots in Figures 6b-i.





369

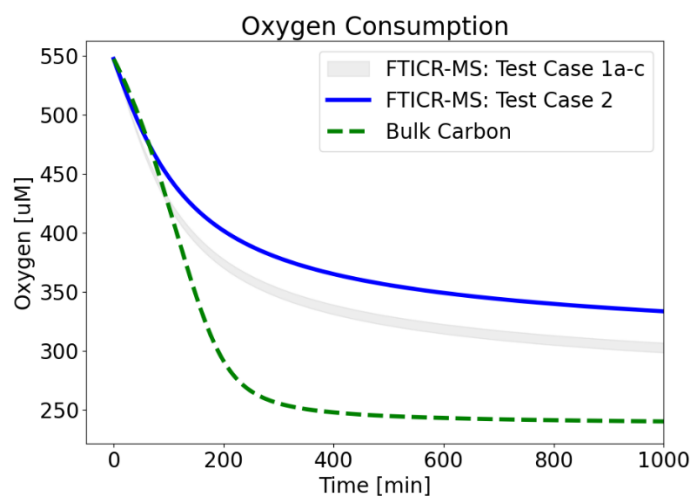
370 **Figure 6.** Comparison of Lambda-Binned Reaction Network Parameters: (a) number of carbons in the OM; stoichiometric
 371 coefficient, y , for (b) OM, (c) O_2 , (d) HCO_3^- , (e) NH_4^+ , (f) HS^- , (g) HPO_4^{2-} , (h) H^+ ; and (i) ratio of OM/ O_2 coefficients for Test
 372 Case 1a-c (grey dots); the average of all λ bins for Test Case 1 (grey line); Test Case 2 (blue x); and the average of all λ bins for
 373 Test Case 2 (blue line). The grey shaded area highlights the range of values for Test Case 1a-c for better visual comparison.

374

375 Second, the λ binning process captures the OM speciation variation within a sample. To illustrate this intrasample
 376 variability, a line representing the average of all λ bins is shown on Figure 6 (grey line for Test Case 1, blue line for
 377 Test Case 2). The difference between the reaction network coefficients (vertical axis) for the λ binning (grey shaded
 378 area and blue dots) and the Test Case average lines highlights the extent of this variability. Finally, although the λ
 379 binning process resulted in a similar number of carbon atoms to OM molecules within each λ bin for both test cases
 380 (Figure 6a), the resulting stoichiometric coefficients in the reaction networks differ significantly (Figures 6b-h). These
 381 stoichiometric differences lead to variations in biogeochemical outcomes, such as OM-to-oxygen utilization ratios
 382 during aerobic respiration (Figure 6i). These differences are due to the additional elements beyond carbon in the OM
 383 molecules (i.e., nitrogen, oxygen, sulfur, hydrogen, and phosphorus).

384

385 To further assess and isolate the effect of OM speciation, extended forward simulations were performed by only
386 varying FTICR-MS input data (Figure 7). FTICR-MS samples from Test Cases 1a-c and Test Case 2 were tested.
387 These simulations replicate Figure 3 (i.e., Test Case 1a conditions and fitted μ_{\max} values) with the expectation of OM
388 speciation, and demonstrate the significant impact of OM chemistry and speciation on overall predicted behavior,
389 especially over longer time periods.
390



391
392 **Figure 7.** Influence of OM Speciation on Oxygen Consumption. FTICR-MS data from Test Cases 1a-c (grey shaded area), and
393 Test Case 2 (blue line) were used as inputs. Bulk CH_2O OM (green line) was also plotted for reference. Best fit μ_{\max} values to Test
394 Case 1a were used (i.e., lambda binned $\mu_{\max} = 0.25 \text{ min}^{-1}$; bulk OM $\mu_{\max} = 0.032 \text{ min}^{-1}$).
395

396 The clear variability in OM speciation, differences between a generic OM reaction network and one informed by
397 FTICR-MS, and the impact of OM chemistry on biogeochemical predictive simulations underscore the importance of
398 incorporating site-specific OM chemistry informed by ultra high resolution characterization into biogeochemical
399 models.

400 5 Conclusions

401 Overall, Lambda-PFLOTRAN workflow provides an important linkage between molecular scale organic matter
402 characterization and reactive transport simulations. This workflow allows for the influence of organic matter
403 composition to be utilized within simulators to provide a more comprehensive understanding of the system chemistry
404 and behavior, moving beyond the standard assumption of bulk organic matter chemistry and composition. While there
405 are current limitations due to how composition is characterized and quantified, this workflow connecting
406 characterization information to simulations is an important advancement that can be refined as these laboratory
407 techniques improve over time.

408 One of the major limitations surrounding this method, is the lack of understanding of organic matter compound
409 bioavailability, resulting in a large conceptual gap as to how various organic carbon compounds may be utilized by

410 microbes. In the absence of such information, all identified organic matter molecules are assumed to have equal
411 bioavailability within this modeling framework when, in reality, compounds will exhibit varying degrees of
412 bioavailability depending on factors such as associated size fraction, carbon pool, and environmental factors (Schmidt
413 et al., 2011; Ahamed et al., 2023). Until improved understanding is established to discern individual compound
414 bioavailability, this will remain as a limitation.

415 Another limitation of this method resides around the analytical limitations of organic carbon characterization and
416 quantification. For instance, FTICR-MS focuses on water soluble organic matter which may provide a bias-basis in
417 the types of carbon identified by this technique (Tfaily et al., 2017). Additionally, as mentioned previously, FTICR-
418 MS is qualitative, it does not provide structural information and will not differentiate between different isomers that
419 have the same molecular formulas, it is only able to identify molecular formula is present or absent and not the
420 concentration associated with each peak. Here, this has been addressed by assuming equal distribution of total carbon
421 between the formulas within each λ bin on a per-carbon basis. This caveat can be easily updated in the workflow if
422 new analytical advances are made that provide more quantitative information. Some existing approaches could be
423 suitable for this type of modeling such as using quantitative biomarkers that cover major compound classes (Kim and
424 Blair, 2023); but further advances in obtaining both high resolution and quantitative OM characterization would
425 greatly aid in how we understand and model ecosystems.

426

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439

440 **Code Availability:**

441 The source code, installation requirements, example test case notebooks, and associated data are available in ESS
442 DIVE at <https://doi.org/10.15485/2281403>

443

444 **Author Contribution:**

445 KM: conceptualization, formal analysis, methodology, software, writing- original draft preparation; PJ: methodology,
446 software, writing- original draft preparation; GH: methodology, software, writing-review & editing; TA: data curation,
447 software, writing-review & editing; HS: methodology, writing-review & editing; RK: supervision; NW: supervision,
448 writing-review & editing; MB: investigation; RC: investigation; QZ: investigation; VG: investigation, data curation;
449 AR: investigation; XC: conceptualization, investigation, writing-review & editing

450

451 **Competing Interests:** The authors declare that they have no conflict of interest.

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