1	MESMO 3: Flexible phytoplankton stoichiometry and refractory DOM
2	
3	Katsumi Matsumoto, Tatsuro Tanioka+, and Jacob Zahn
4	
5	Department of Earth & Environmental Sciences, University of Minnesota, Minneapolis,
6	Minnesota, USA
7	⁺ Current affiliation: Department of Earth System Science, University of California Irvine,
8	Irvine, California, USA
9	
10	Contact: katsumi@umn.edu
11	
12	(revised version submitted 3/18/2021 to Geoscientific Model Development)

14 Abstract

15 We describe the third version of Minnesota Earth System Model for Ocean biogeochemistry 16 (MESMO 3), an earth system model of intermediate complexity, with a dynamical ocean, a 17 dynamic-thermodynamic sea ice, and an energy moisture balanced atmosphere. A major 18 feature of Version 3 is the flexible C:N:P ratio for the three phytoplankton functional types 19 represented in the model. The flexible stoichiometry is based on the power law formulation 20 with environmental dependence on phosphate, nitrate, temperature, and light. Other new 21 features include nitrogen fixation, water column denitrification, oxygen and temperature-22 dependent organic matter remineralization, and $CaCO_3$ production based on the concept of 23 the residual nitrate potential growth. Also, we describe the semi-labile and refractory 24 dissolved organic pools of C, N, P, and Fe that can be enabled in MEMSO 3 as an optional 25 feature. The refractory dissolved organic matter can be degraded by photodegradation at 26 the surface and hydrothermal vent degradation at the bottom. These improvements 27 provide a basis for using MESMO 3 in further investigations of the global marine carbon 28 cycle to changes in the environmental conditions of the past, present, and future.

30 1. Introduction

31 Here we document the development of the third version of the Minnesota Earth System 32 Model for Ocean biogeochemistry (MESMO 3). As described for the first two versions 33 (Matsumoto et al., 2008, 2013), MESMO is based on the non-modular version of the Grid 34 ENabled Integrated Earth (GENIE) system model (Lenton et al., 2006; Ridgwell et al., 2007). 35 The computationally efficient ocean-climate model of Edwards and Marsh (Edwards and 36 Marsh, 2005) forms the core of GENIE's physical model. MESMO is an earth system model 37 of intermediate complexity (EMIC), which occupies a midpoint in the continuum of climate 38 models that span high resolution, comprehensive coupled models on one end and box 39 models at the other end (Claussen et al., 2002). MESMO has a 3D dynamical ocean model on 40 a 36 x 36 equal-area horizontal grid with 10° increments in longitude and uniform in the 41 sine of latitude. There are 16 vertical levels. It is coupled to a 2D energy moisture balanced 42 model of the atmosphere and a 2D dynamic and thermodynamic model of sea ice. Thus, 43 MESMO retains important dynamics, which allow for simulations of transient climate 44 change, while still being computationally efficient.

45

46 Since the first version, MESMO has continued to be developed chiefly for investigations of 47 ocean biogeochemistry (Table 1). Briefly, in MESMO 1, the main improvements over the 48 predecessor GENIE focused on the biological production and remineralization as well as on 49 the uptake of natural radiocarbon $({}^{14}C)$ and anthropogenic transient tracers (Matsumoto et 50 al., 2008). The net primary production (NPP) in MESMO 1 occurred in the top two vertical 51 levels representing the surface 100 m and depended on temperature, nutrients, light, and 52 mixed layer depth (MLD). The nutrient dependence was based on the Michaelis-Menten 53 uptake kinetics of phosphate (PO_4), nitrate (NO_3), and aqueous CO_2 . The limiting nutrient 54 was determined by the Liebig's rule of the minimum relative to the fixed uptake 55 stoichiometry of C:N:P=117:16:1. A single generic phytoplankton functional type (PFT) 56 carried out NPP, which was split between particulate organic matter (POM) and dissolved 57 organic matter (DOM) in a globally constant ratio of 1:2. The semi-labile form of the 58 dissolved organic carbon (DOC) was the only form of DOM simulated in MESMO 1. The POM 59 flux across the 100 m level defined the export production. The vertical flux of POM was

driven by a fixed rate of sinking and a temperature-dependent, variable remineralizationrate.

62

63 The main aim of MESMO 2 was a credible representation of the marine silica cycle 64 (Matsumoto et al., 2013). To this end, the set of limiting nutrients (P, N, and C) in MESMO 1 65 was augmented to include iron (Fe) and silicic acid (Si(OH)₄) in MESMO 2 (Table 1). The 66 stable isotope of Si (³⁰Si) was also added as a state variable. The Fe cycle included an 67 aeolian flux of Fe, complexation with organic ligand, and particle scavenging of free Fe. The 68 scavenged Fe that reached the seafloor was removed from the model domain. This burial 69 flux of Fe balanced the aeolian flux at steady state. Also, a new PFT was added in MESMO 2 to represent chiefly diatoms. This new "large" PFT was limited by Si and characterized by a 70 71 high maximum growth rate and large half-saturation constants for the nutrient uptake 72 kinetics. It represented fast and opportunistic phytoplankton that do well under nutrient 73 replete conditions. In comparison, the "small" PFT was characterized with a lower 74 maximum growth rate and smaller half-saturation constants and outperformed the large 75 PFT in oligotrophic subtropical gyres. CaCO₃ production was associated with the "small" 76 PFT in MESMO2. The addition of Fe, Si, and the large PFT in MESMO 2 allowed it to have a 77 Fe-dependent, variable Si:N uptake ratio (Hutchins and Bruland, 1998; Takeda, 1998), 78 which is critical to simulate important features of the global ocean Si distribution. 79

80 MESMO 1 and 2 were assessed and calibrated by multi-objective tuning and extensive 81 model-data comparisons of transient tracers (anthropogenic carbon, CFCs), deep ocean 82 Δ^{14} C, and nutrients (Matsumoto et al., 2008, 2013). These versions have been employed 83 successfully in a number of studies of global distributions of carbon and carbon isotopes 84 under various conditions of the past, present, and future (Cheng et al., 2018; Lee et al., 85 2011; Matsumoto et al., 2010, 2020; Matsumoto and McNeil, 2012; Matsumoto and 86 Yokoyama, 2013; Sun and Matsumoto, 2010; Tanioka and Matsumoto, 2017; Ushie and 87 Matsumoto, 2012). Also, MESMO 1 and 2 have participated in model intercomparison 88 projects (Archer et al., 2009; Cao et al., 2009; Eby et al., 2013; Joos et al., 2013; Weaver et 89 al., 2012; Zickfeld et al., 2013).

91 In this contribution, we describe the third and latest version of MEMSO with a number of 92 substantial biogeochemical model modifications and new features that bring MESMO up to date with the evolving and accumulating knowledge of the ocean biogeochemical cvcle 93 94 (Table 1). There is no change in the physical model between MESMO 3 and MESMO 2. The 95 most significant new feature of MESMO 3 over the previous versions is the power law 96 formulation of flexible phytoplankton C:N:P ratio. Other new features include additional 97 PFT diazotrophs that carry out N-fixation, water column denitrification, the dependence of 98 organic matter remineralization on the dissolved oxygen (0_2) and temperature, and CaCO₃ 99 production based on the concept of the residual nitrate potential growth. Also, we describe 100 the semi-labile DOM for P, N, and Fe (DOP_{sl}, DON_{sl}, and DOFe_{sl}), and the refractory DOM for 101 C, P, and N (DOC_r, DOP_r, and DON_r), which can be activated as an optional feature in MESMO 102 3. Some of these features have been described separately in different publications 103 (Matsumoto et al., 2020; Matsumoto and Tanioka, 2020; Tanioka and Matsumoto, 2017, 104 2020a). This work consolidates the descriptions of all these features in a single publication. 105

106 **2. Model Description**

Here we present the full set of biogeochemical equations of MESMO 3 as well as key model
parameters (Table 2). We describe only the biogeochemical source and sink terms and omit
the physical (advective and diffusive) transport terms that are calculated by the ocean
circulation model. We discuss the production terms first, followed by remineralization
terms, followed by conservation equations that incorporate both terms.

113 **2.1 Phytoplankton Nutrient Uptake**

- 114 NPP occurs in the top two vertical levels of the ocean domain above the fixed compensation
- 115 depth (z_c) of 100 m. Key parameter values are given in Table 2a. Nutrient uptake by
- 116 phytoplankton type i (Γ_i) depends on the optimal nutrient uptake timescale (τ_i), nutrients,
- 117 temperature (T), irradiance (I), and mixed layer depth (z_{ml}) :
- 118

$$\Gamma_{i} = \frac{1}{\tau_{i}} \cdot F_{N,i} \cdot F_{T} \cdot F_{I} \cdot max \left\{ 1, \frac{z_{c}}{z_{ml}} \right\}$$

$$I$$

Subscript i refers to PFT (i =1: eukaryotes, i =2: cyanobacteria, i = 3: diazotrophs). The nutrient dependence $F_{N,i}$ is given by Liebig's law of minimum combined with Michael-Menten uptake kinetics of limiting nutrients: PO₄, NO₃, CO₂, (aq), total dissolved iron (sum of free iron and ligand-bound iron; FeT=Fe'+FeL), and silicic acid (Si(OH)₄):

$$F_{N,i} = min\left(\frac{[PO_4]}{[PO_4]+K_{PO_4,i}} \cdot [PO_4], \frac{[NO_3]}{[NO_3]+K_{NO_3,i}} \cdot [NO_3] \cdot Q_{N,i}^{-1}, \frac{[CO_2(aq)]}{[CO_2(aq)]+K_{CO_2,i}} \cdot \right)^2$$

$$[CO_2(aq)] \cdot Q_{C,i}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[Si(OH)_4]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[FeT]}{[Si(OH)_4]+K_{Si(OH)_4}} \cdot [Si(OH)_4] \cdot Q_{Si}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe,i}^{-1}, \frac{[FeT]}{[FeT]+K_{FeT,i}} \cdot [FeT] \cdot Q_{Fe$$

125

where K_{X,i} is the half-saturation concentration of nutrient X for PFT i. Only eukaryotes (i=1) are limited by Si(OH)₄. Diazotrophs (i = 3) are not limited by NO₃. Nutrient uptake Γ is based on the master nutrient variable P, and all other nutrient uptake is related to Γ by the uptake stoichiometry Q_{X,i}, where X is N, Fe, Si, or C. For example, $Q_{C,i} = \frac{1}{[P:C]_i}$ for PFT i. Thus, Q_{c,i} is numerically equivalent to C:P for PFT i, but we write the equations in terms of P:C for numerical stability and convenience. The Q_{X,i} ratios represent the flexible phytoplankton uptake stoichiometry and describe more fully in the following section 2.2.

133

134 The temperature dependence F_T of Equation 1 is given by:

135

$$F_T = \frac{T(°C) + 2}{T(°C) + 10}$$
³

136

137 which is analogous to the commonly used $Q_{10} = 2$ relationship. Light limitation F_1 of

138 Equation 1 is described by a hyperbolic function:

$$F_T = \frac{I}{I+20}$$

140 where I is the seasonally variable solar short-wave irradiance in W m⁻². Light is attenuated

141 exponentially from the ocean surface with a 20 m depth scale.

142

143Nutrient uptake in Equation 1 has a dependence on z_{ml} , which is diagnosed using the σ_t 144density gradient criterion (Levitus, 1982). Following the Sverdrup (1953) model of the

145 spring bloom, Equation 1 allows for the shoaling of z_{ml} relative to z_c to enhance nutrient 146 uptake.

147

148 **2.2 Phytoplankton uptake stoichiometry**

149As noted above, all nutrients and O_2 are related to the main model currency P by $Q_{X,i}$. We150describe three different, mutually exclusive formulations in this section. The standard

151 formulation is the power law model (Matsumoto et al., 2020; Tanioka and Matsumoto,

152 2017). The other two (Linear model and Optimality-based model of stoichiometry) are

153 alternative formulations that have been coded, and the user can activate them (one at a

154 time) in place of the power law formulation. However, the alternative formulations are not

155 calibrated. Key parameter values are given in Table 2b for the power law formulation.

156

157 **2.2.1 Power law model of stoichiometry**

158 The uptake P:C and N:C ratios are calculated using the power-law formulation as a function

159 of ambient concentrations of phosphate [PO₄], nitrate [NO₃], temperature (T), and

- 160 Irradiance (I):
- 161

$$[P:C]_{i} = [P:C]_{0,i} \cdot \left(\frac{[PO_{4}]}{[PO_{4}]_{0}}\right)^{s_{PO_{4},i}^{P:C}} \cdot \left(\frac{[NO_{3}]}{[NO_{3}]_{0}}\right)^{s_{NO_{3},i}^{P:C}} \cdot \left(\frac{T}{T_{0}}\right)^{s_{T,i}^{P:C}} \cdot \left(\frac{I}{I_{0}}\right)^{s_{I,i}^{P:C}} 5$$

$$[N:C]_{i} = [N:C]_{0,i} \cdot \left(\frac{[PO_{4}]}{[PO_{4}]_{0}}\right)^{S_{PO_{4},i}^{N:C}} \cdot \left(\frac{[NO_{3}]}{[NO_{3}]_{0}}\right)^{S_{NO_{3},i}^{N:C}} \cdot \left(\frac{T}{T_{0}}\right)^{S_{T,i}^{N:C}} \cdot \left(\frac{I}{I_{0}}\right)^{S_{T,i}^{N:C}} \tag{6}$$

163 Equations 5 and 6 are the power-law equations that calculate the change in P:C and N:C for

164 fractional changes in environmental drivers relative to the reference P:C and N:C,

respectively (Matsumoto et al., 2020; Tanioka and Matsumoto, 2017). The exponents are

166 the sensitivity factors determined by a meta-analysis (Tanioka and Matsumoto, 2020a).

167 Subscript "0" indicates the reference values (Table 2b). We have hard bounds for the

168 calculated P:C and N:C ratios to be within 26.6<C:P<546.7 and 2<C:N<30 as observed

169 (Martiny et al., 2013).

170

171The P:C and N:C ratios from Equations 5 and 6 can then be converted to $Q_{N,i}$ and $Q_{C,i}$ for use172in Equation 2.

173

$$Q_{C,i} = \frac{1}{\left[P:C\right]_i} \tag{7}$$

$$Q_{N,i} = \frac{1}{[P:N]_{i}} = \frac{[N:C]_{i}}{[P:C]_{i}}$$
8

174

175 **2.2.2 Linear model of stoichiometry by Galbraith & Martiny**

176 A much simpler, alternative formulation for P:C and N:C is the model of Galbraith & Martiny 177 (2015) where P:C is a linear function of $[PO_4]$ (in μ M), and N:C is a Holling type 2 functional 178 form with a frugality behavior only at very low $[NO_3]$ (in μ M). The same P:C and N:C values 179 are applied to all three PFTs.

$$[P:C] = \frac{6.9 \cdot [PO_4] + 6.0}{1000}$$

$$[N:C] = 0.125 + \frac{0.03 \cdot [NO_3]}{0.32 + [NO_3]}$$
¹⁰

182 **2.2.3 Optimality-based model of stoichiometry**

183 The optimality-based model of phytoplankton growth is based on the chain model, which 184 connects the cellular P, N, and C acquisition by a chain of limitations, where the P quota 185 limits N assimilation and the N quota drives carbon fixation (Pahlow et al., 2013; Pahlow 186 and Oschlies, 2009, 2013). Resource-allocation of cellular P, N, and C among different 187 cellular compartments are derived from balancing energy gain from gross carbon fixation 188 and energy loss due to nutrient acquisition and light-harvesting. The optimality-based 189 model by Pahlow et al. (2013) computes C:N and C:P as a function of nutrient availability 190 (PO₄ and NO₃), irradiance, and day length. Temperature dependence was added by 191 Arteaga et al. (2014) following the simple logarithmic temperature dependence on 192 maximum nutrient uptake rate following (Eppley, 1972).

193

194 Different versions of this optimality-based model have previously been successfully 195 implemented in global ocean biogeochemical models, such as the Pelagic Interactions 196 Scheme for Carbon and Ecosystem Studies (PISCES) (Kwiatkowski et al., 2018, 2019) and 197 the University of Victoria Earth System Model (UVIC) (Chien et al., 2020; Pahlow et al., 198 2020). However, as we are not describing any results in this paper, we will only mention 199 here that there is an option to calculate C:N:P using this stoichiometry model in MESMO 3. 200 The full description of the optimality-based stoichiometry model and its parameter 201 calibration are presented specifically for the UVic model elsewhere (Chien et al., 2020; 202 Pahlow et al., 2020).

203

204 **2.2.4 Stoichiometry of iron and silica**

Iron uptake stoichiometry Q_{Fe,i} is calculated as a function of FeT following the power-law
formulation of Ridgwell (2001). Key parameter values are given in Table 2c.

$$Q_{Fe,i} = [Fe:P]_i = [Fe:C]_i \cdot Q_{C,i}$$
 11

$$[Fe:C]_i = 1.0/([C:Fe]_{min,i} + [C:Fe]_{ref,i} \cdot [FeT]^{-s^{Fe:C_i}})$$
¹²

For all PFTs, the power law exponent s^{Fe:C} in Equation 12 is -0.65. The allowable Fe:C ratio is bounded at the low end by the hard-bound minimum Fe:C of 1:220,000. The scaling constant or [C:Fe]_{ref,i} is set differently for PFTs, with eukaryotes having a higher base [C:Fe]_{ref,i} than cyanobacteria and diazotrophs (115,623:1 and 31,805:1, respectively). The high end of the allowable Fe:C ratio is bounded by [C:Fe]_{min,i} (i.e., maximim Fe:C) of 15,000:1 for eukaryotes and 20,000:1 for cyanobacteria/diazotrophs. These parameters directly follow Ridgwell (2001), who fitted power-law functions to the experimental data (Sunda and Huntsman, 1995).

Silica uptake stoichiometry by eukaryotes Q_{Si} is a power law of total dissolved iron [FeT]
and increases with a decrease in [FeT] (Brzezinski, 2002). The power law exponent s^{Si:N} is
set to 0.7. The Si:N ratio is limited to a maximum of 18 and a minimum of 1.

$$Q_{Si} = [Si:P] = [Si:N] \cdot Q_{N,1}$$
 13

$$[Si:N] = min\left([Si:N]_{max}, max\left([Si:N]_{min}, \left(\frac{[FeT]}{0.5 \ nmol \ kg^{-1}}\right)^{-s^{Si:N}}\right)\right)$$
 14

 O_2 liberated by phytoplankton during photosynthesis per PO₄ consumed ($Q_{-O_2,i}$) is

224 calculated from the uptake C:P and N:P ratios (Tanioka and Matsumoto, 2020b):

$$Q_{-O_{2,i}} = 1.1Q_{C,i} + 2Q_{N,i}$$
¹⁵

2.3 Production of POM and DOM

- In the top 100 m of the model domain, where phytoplankton P uptake occurs (i.e., $\Gamma_i > 0$,
- see section 2.1), NPP is immediately routed to POM and DOM pools (Figure 1). The
- 230 production fluxes of POM, DOM_{sl}, and DOM_r from NPP are given as Jprod. Here we write the
- 231 equations in terms of P, which is the master nutrient variable:
- 232

$$Jprod_{POP_i} = (1 - fDOM) \cdot \Gamma_i$$
¹⁶

$$Jprod_{DOP_{sl}} = \sum_{i} (1 - fDOM_r) \cdot fDOM \cdot \Gamma_i$$
¹⁷

$$Jprod_{DOP_r} = \sum_{i} fDOM_r \cdot fDOM \cdot \Gamma_i$$
¹⁸

234 The term fDOM denotes the fraction of NPP that is routed to DOM as opposed to POM. 235 Likewise, fDOM_r is the fraction of DOM that is routed to DOM_r as opposed to DOM_{sl}. The 236 value of fDOM_r is not well known but estimated to be $\sim 1\%$ (Hansell, 2013), which we 237 tentatively adopt in MESMO 3. If DOM_r is not selected in the model run, $fDOM_r = 0$. In 238 previous versions of MESMO, fDOM was assigned a constant value of 0.67. In reality, a large 239 variability is observed locally for this ratio, ranging from 0.01-0.2 in temperate waters to 240 0.1-0.7 in the Southern Ocean (Dunne et al., 2005; Henson et al., 2011; Laws et al., 2000). In 241 MESMO 3, fDOM is calculated as a function of the ambient temperature following Laws et 242 al. (Laws et al., 2000):

243

$$fDOM = 1.0 - min(0.72, max(0.04, 0.62 - 0.02 \cdot T(^{\circ}C)))$$
¹⁹

244

This formulation gives low export efficiency (i.e., high fDOM) in the warmer regions
compared to the colder high latitude regions. Locally, we impose fixed fDOM upper and
lower bounds of 0.96 and 0.28, respectively, as estimated from a previous study (Dunne et
al., 2005).

In MEMSO 3, a new DOM production pathway below the production layer is available as an

- 251 option. In previous MESMO versions, sinking POM was respired in the water column with
- the loss of O₂ directly to the dissolved inorganic forms (i.e., POC-->DIC, POP-->PO₄, and
- 253 POP-->NO₃). In the new "deep POC split" pathway, sinking POM is simply broken down into
- 254 DOM without the loss of O_2 as in the production layer (Figure 1). If DOM_r is selected in the
- 255 model, the broken-down POM is further routed to both DOM_{sl} and DOM_{r} according to
- 256 fDOM_r. If not, all of the broken down POM is converted to DOM_{sl}. Thus, when the deep POC
- 257 split is activated, the presence of DOM in the deep ocean can be accounted for by *in situ*
- 258 production of DOM and DOM_r in addition to DOM transport from the surface. Thus, the
- 259 deep POC split pathway offers an alternative means to control deep ocean DOM
- 260 distribution.
- 261

262 **2.4 Production of CaCO₃ and opal by eukaryotes**

In MESMO 2, opal production was associated with the large PFT and CaCO₃ production was associated with the "small" PFT. We recognize that coccolithophorids and diatoms, which are the producers of these biogenic tests, are both eukaryotes. Therefore, in MEMSO 3, we associate both CaCO₃ and opal production with the POM production by the same eukaryote PFT (Jprod_{POM1}) :

268

$$Jprod_{CaCO3} = r^{CaCO_3:POC} \cdot Jprod_{POM1} \cdot Q_{C,1}$$
 20

$$Jprod_{opal} = Jprod_{POM1} \cdot Q_{Si}$$
 21

269

The concept of the residual nitrate potential growth (RNPG) (Balch et al., 2016) is useful in allowing competition between diatoms and non-siliceous phytoplankton within the same PFT (Matsumoto et al., 2020). Typically, in the real ocean, non-Si phytoplankton are able to grow faster and dominate the community if Si concentration is low and diatom growth is Si limited. Otherwise, diatoms are more competitive, as they have higher intrinsic growth rates. The RNPG index recasts the ambient concentrations of NO₃ and Si(OH)₄ into potential algal growth rates:

$$RNPG = \frac{[NO_3]}{[NO_3] + K_{NO_{3,1}}} - \frac{[Si(OH)_4]}{[Si(OH)_4] + K_{Si(OH)_4}}$$
22

278

If RNPG is more positive, the index indicates that nitrate-dependent growth exceeds silicadependent growth. Thus, non-Si phytoplankton are more competitive, and this leads to
higher CaCO₃ production. On the other hand, a more negative RNPG implies that silica
limitation for diatoms is relieved, leading to enhanced diatom growth and reduced CaCO₃
production. The RNPG index is incorporated in the calculation of the rain ratio r^{CaCO₃:POC}
presented in Equation 20 as:

$$r^{CaCO_3:POC} = r_0^{CaCO_3:POC} \cdot (\Omega - 1)^{\eta} \cdot min(1, max(0.1, RNPG)) \cdot k_{T, CaCO_3}$$
23

286

Equation 23 indicates the base rain ratio $r_0^{CaCO_3:POC}$ (set to 0.30) is also modified by the carbonate ion saturation state Ω by η (set to 1.28) by as well as by temperature (see Ridgwell et al. (2007) and references therein):

290

$$\Omega = \frac{[Ca^{2+}][CO_3^{2-}]}{K_{sp}}$$
24

$$k_{T,CaCO_3} = min\left(1.0, \frac{T(\circ C) + 2}{T(\circ C) + 8}\right)$$
25

291

292 K_{sp} is the solubility product of CaCO₃. The temperature dependency of CaCO₃ formation 293 $(k_{T,CaCO_3})$ is similar to that of Moore et al. (2004) where warmer temperatures favor the 294 growth of carbonate-bearing phytoplankton.

295

296 **2.5 Remineralization of POM and DOM**

297 Once produced, both POM and DOM undergo remineralization throughout the water

298 column. Key remineralization parameter values are given in Table 2d. Previously, POM

299 remineralization had a temperature dependence and decayed exponentially with depth

300 (Yamanaka et al., 2004). In MESMO 3, we incorporate an additional dependency on

301 dissolved oxygen following Laufkötter et al. (2017):

302

$$R_{POM_i} = V_{POM} \cdot e^{k_R \cdot T \cdot \frac{[O_2]}{[O_2] + K_{O_2}}} \cdot [POM_i]$$
²⁶

303

V_{POM} is the base remineralization rate, parameter k_R expresses the temperature sensitivity
of remineralization, and K₀₂ is half-saturation constant for oxygen-dependent
remineralization. When the sediment model is not coupled, any POM that reaches the
seafloor dissolves completely to its inorganic form and is returned to the overlying water.

309 In MESMO 3, all forms of semi-labile DOM remineralize at the same rate. It is represented 310 by τ_{sl} , the inverse of the time scale of DOM_{sl} decay, which has been estimated previously to 311 be ~1.5 years (Hansell, 2013):

312

$$R_{DOM_{sl}} = \tau_{sl} \cdot [DOM_{sl}]$$
²⁷

313

314 All forms of DOM_r also remineralize at the same rate in MESMO 3. In total, there are three 315 optional, additive sinks of DOM_r in the model: slow background decay, photodegradation, 316 and degradation via hydrothermal vents (Figure 1). Observations clearly indicate that the 14 C age of deep ocean DOC_r is 10³ years (e.g., Druffel et al., 1992), much older than DI¹⁴C. 317 318 Also, the deep ocean DOC_r concentration decreases modestly along the path of the deep 319 water from the deep North Atlantic to the deep North Pacific (Hansell and Carlson, 1998). 320 Thus, it is understood that there is a slow DOM_r background decay in the deep ocean. We 321 represent this ubiquitous process with τ_{bg} , which is the inverse of the background decay 322 time scale, estimated to be \sim 16,000 years (Hansell, 2013). 323

324 Observations to date indicate that photodegradation is a major sink of DOM_r (e.g., Mopper 325 et al., 1991). This process is believed to convert DOM_r that is upwelled from the ocean

- 326 interior into the euphotic zone into more labile forms of DOM. We represent
- 327 photodegradation with τ_{photo} , the inverse of the decay time scale, estimated to be ~70 years
- 328 (Yamanaka and Tajika, 1997). This occurs only in the surface.
- 329

330 Finally, observations of DOM emanating from different types of hydrothermal vents 331 indicate that they have variable impacts on the deep sea DOM_r (Lang et al., 2006). However, 332 the off-axis vents circulate far more seawater through the fractured oceanic crust than the 333 high temperature and diffuse vents and thus believed to determine the overall impact of 334 the vents on the deep sea DOM_r as a net sink (Lang et al., 2006). Here we assume simply 335 that seawater that circulates through the vents loses all DOM_r (i.e., $1/\tau_{vent} < \Delta t$, where Δt is 336 the biogeochemical model time step of 0.05 year). This means that the more seawater 337 circulates through the vents, the more DOM_r is removed: the total removal rate depends on 338 the vent flux of seawater H_{flux} . We implement the vent degradation of DOM_r in MESMO 3 by 339 first identifying the wet grid boxes located immediately above known mid-ocean ridges. We 340 then distribute the annual global H_{flux} of 4.8×10^{16} kg yr⁻¹ (Lang et al., 2006) equally among 341 those ridge-associated grid boxes. The grid cells contain a mass of seawater much greater 342 than the mass that circulates through vents in Δt (10²¹ kg vs. 10¹³ kg). Therefore, the 343 seawater mass in the vent grid cells that does not circulate through the vents in Δt is 344 subject only to background degradation in MESMO 3. 345

The three DOM_r sinks are not mutually exclusive. They can thus be combined to yield the
 total DOM_r remineralization rate:

348

$$R_{DOM_r} = (\tau_{bg} + \tau_{photo} + \tau_{vent} \cdot \frac{SW_{flux_local}}{SW_{grid}}) \cdot [DOM_r]$$
²⁸

349

350 where SW_{flux_local} is the mass of seawater that circulates through the vents in each grid box 351 in Δt , and SW_{grid} is the total mass of seawater in the same grid box.

The amount of O₂ respired as a result of these POM and DOM remineralization processes is related to the organic carbon pools by the respiratory quotients of POC and DOC, $r_{-O_2:POC}$ and $r_{-O_2:DOC}$, respectively. These are molar ratios of O₂ consumed per unit organic carbon respired. They are variable and calculated from the ambient POM and DOM concentration (Tanioka and Matsumoto, 2020b):

358

$$r_{-O_2:POC} = 1.1 + \frac{2[PON]}{[POC]}$$
²⁹

$$r_{-O_2:DOC} = 1.1 + \frac{2[DON]}{[DOC]}$$
30

359

2.6 Remineralization of CaCO₃ and opal

361Remineralization of CaCO3 and opal particles occurs as they sink through the water column362and remains the same as in MESMO 2. Key parameter values are given in Table 2d.363Remineralization of CaCO3 is a function of temperature similar to that of particulate organic364matter remineralization but without oxygen dependency. The temperature dependence365term kR modifies the base remineralization rate V_{CaCO3} :

366

$$R_{CaCO_3} = V_{CaCO_3} \cdot e^{k_R \cdot T} \cdot [CaCO_3]$$
³¹

367

Opal remineralization in the water column follows Ridgwell et al. (2002). The rate of opal
 remineralization R_{opal} is given by the product of normalized dissolution rate (r_{opal}), base
 opal dissolution rate (k_{opal}), and opal concentration [opal]:

$$R_{opal} = r_{opal} \cdot k_{opal} \cdot [opal]$$
³²

$$r_{opal} = 0.16 \cdot \left(1 + \frac{T(\mathcal{C})}{15}\right) \cdot u_{opal} + 0.55 \cdot \left(\left(1 + \frac{T(\mathcal{C})}{400}\right)^4 \cdot u_{opal}\right)^{9.25}$$
33

$$u_{opal} = \frac{[Si(OH)_4]_{eq} - [Si(OH)_4]}{[Si(OH)_4]_{eq}}$$
34

- 373 r_{opal} is a function of temperature (T) and the degree of under-saturation (u_{opal}), which in
- turn is calculated from the ambient $[Si(OH)_4]$ and $[Si(OH)_4]$ at equilibrium. The
- 375 equilibrium concentration is a function of ambient temperature:

376

$$\log_{10} \left([Si(OH)_4]_{eq} \right) = 6.44 - \frac{968}{T(K)}$$
35

377

378 Without the sediment module of MESMO activated, both CaCO₃ and opal particles that

379 reach the seafloor are completely dissolved back to inorganic forms.

380

381 **2.7 Conservation of organic matter and biogenic tests**

The time rate of change of the biogenic organic matter and tests are given by the sum of the production terms (i.e., sources) and the remineralization terms (i.e., sinks). The circulationrelated transport terms are omitted as noted above, but the vertical transport due to

385 particle sinking is included here. The sinking speed w is the same for all particles. The sum

386 of POM_i of all the PFTs give the total POM concentrations:

$$\frac{\partial [POP]_{i}}{\partial t} = Jprod_{POP_{i}} - \frac{\partial}{\partial z} (w[POP]_{i}) - R_{POP,i}$$
³⁶

$$\frac{\partial [POC]_{i}}{\partial t} = Jprod_{POP_{i}} \cdot Q_{C,i} - \frac{\partial}{\partial z} (w[POC]_{i}) - R_{POC,i}$$
³⁷

$$\frac{\partial [PON]_{i}}{\partial t} = Jprod_{POP_{i}} \cdot Q_{N,i} - \frac{\partial}{\partial z} (w[PON]_{i}) - R_{PON,i}$$
³⁸

$$\frac{\partial [POFe]_{i}}{\partial t} = Jprod_{POP_{i}} \cdot Q_{Fe,i} - \frac{\partial}{\partial z} (w[POFe]_{i}) - R_{POFe,i}$$
³⁹

$$[POM] = \sum_{i} [POM]_{i}$$
⁴⁰

389 The time rate of change of CaCO₃ and opal is expressed in much the same way as POM:390

$$\frac{\partial [CaCO_3]}{\partial t} = Jprod_{CaCO_3} - \frac{\partial}{\partial z} (w[CaCO_3]) - R_{CaCO_3}$$
⁴¹

$$\frac{\partial [opal]}{\partial t} = Jprod_{opal} - \frac{\partial}{\partial z}(w[opal]) - R_{opal}$$
⁴²

392 The DOM pools have the production and remineralization terms without the particle393 sinking term:

$$\frac{\partial [DOP_{sl}]}{\partial t} = Jprod_{DOP_{sl}} - R_{DOP_{sl}}$$

$$43$$

$$\frac{\partial [DON_{sl}]}{\partial t} = Jprod_{DON_{sl}} - R_{DON_{sl}}$$

$$44$$

$$\frac{\partial [DOC_{sl}]}{\partial t} = Jprod_{DOC_{sl}} - R_{DOC_{sl}}$$
⁴⁵

$$\frac{\partial [DOFe_{sl}]}{\partial t} = Jprod_{DOFe_{sl}} - R_{DOFe_{sl}}$$

$$46$$

$$\frac{\partial [DOP_r]}{\partial t} = Jprod_{DOP_r} - R_{DOP_r}$$
⁴⁷

$$\frac{\partial [DON_r]}{\partial t} = Jprod_{DON_r} - R_{DON_r}$$
⁴⁸

$$\frac{\partial [DOC_r]}{\partial t} = Jprod_{DOC_r} - R_{DOC_r}$$
⁴⁹

2.8 Conservation of inorganic nutrients

The time rate of change of the inorganic nutrients have organic carbon production as sink
terms and remineralization as source terms. The production terms (J_{prod}) are zero below
the upper ocean production layer. Nutrients have a unit of mol element kg⁻¹ in the model.

$$\frac{\partial [PO_4]}{\partial t} = -\sum_i \Gamma_i + \sum_i R_{POP,i} + R_{DOP_{sl}} + R_{DOP_r}$$
50

$$\frac{\partial [NO_3]}{\partial t} = -\sum_i \Gamma_i \cdot Q_{N,i} + \sum_i R_{PON,i} + R_{DON_{sl}} + R_{DON_r} + Fix_N - Den_N$$
⁵¹

$$\frac{\partial [DIC]}{\partial t} = -\left(\sum_{i} \Gamma_{i} Q_{C,i} + J prod_{CaCO3}\right) + \sum_{i} R_{POC,i} + R_{DOC_{sl}} + R_{DOC_{r}} + R_{CaCO_{3}} + F_{gas,CO_{2}}\right)$$

$$52$$

$$\frac{\partial [ALK]}{\partial t} = -\left(2 \cdot Jprod_{caCO3} - \sum_{i} \Gamma_{i} Q_{N,i}\right) - \sum_{i} R_{PON,i} - R_{DON_{sl}} - R_{DON_{r}} - Fix_{N} + Den_{N} + 2 \cdot R_{caCO_{3}}$$
53

$$\frac{\partial [FeT]}{\partial t} = -\sum_{i} \Gamma_{i} Q_{Fe,i} + \sum_{i} R_{POFe,i} + R_{DOFe_{sl}} + R_{POM_{Fe}} + Aeolian_{Fe}$$
54

$$\frac{\partial [Si(OH)_4]}{\partial t} = -Jprod_{opal} + R_{opal}$$
⁵⁵

$$\frac{\partial [O_2]}{\partial t} = \sum_i \Gamma_i \cdot Q_{-O_2,i} - \left(r_{-O_2:DOC} \cdot \left(R_{DOC_{sl}} + R_{DOC_r} \right) + \sum_i r_{-O_2:POC,i} \cdot R_{POC,i} \right)$$

$$+ 1.25 \ Den_N + F_{gas,O_2}$$

$$56$$

In Equation 51, Fix_N is the N-fixation carried out by diazotrophs, and Den_N is the water
column denitrification. There is an air-sea gas exchange term F_{gas} in Equations 52 and 56 for
gaseous CO₂ and O₂, respectively. In Equation 53, alkalinity increases with decreasing
nitrate concentrations and increasing CaCO₃ dissolution. Equation 54 contains R_{POMFe},
which is an iron source that represents remineralization of the Fe' scavenged by sinking
particles. These terms are explained in the following sections.

2.9 Prognostic nitrogen cycle

410 Biological production by diazotrophs is stimulated when the ambient NO₃ is low. Nitrogen

411 fixed by diazotrophs during their growth is added to the marine NO₃ pool. The prognostic

- 412 nitrogen fixation model employed here is similar to that used in the HAMOCC
- 413 biogeochemical module (Paulsen et al., 2017):

414

$$Fix_N = \Gamma_3 \cdot Q_{N,3} \cdot I_{NO_3} , \qquad 57$$

$$I_{NO_3} = \left(1.0 - \frac{[NO_3]^2}{{K_{N_2}}^2 + [NO_3]^2}\right),$$
58

415

416 where Fix_N is the nitrogen fixation rate and I_{NO_3} is the nitrate dependency term in

417 quadratic Michaelis-Menten kinetics form with the half-saturation constant K_{N_2} . See Table

418 2e for the values related to the N cycle.

419

420 Water-column denitrification is formulated in an approach similar to that of the original

421 GENIE model (Ridgwell et al., 2007), in which 2 moles of NO_3 are converted to 1 mole of N_2

422 and liberating 2.5 moles of O_2 as a byproduct:

423

$$2NO_3^- + 2H^+ \to 2.5O_2 + N_2 + H_2O$$
59

424

425 Denitrification takes place in grid boxes, in which O_2 concentration is below a threshold 426 concentration $(O_{2,def})$ and is stimulated if the total global inventory of NO_3 relative to PO_4 is 427 high. In other words, denitrification can effectively act as negative feedback to nitrogen 428 fixation. The threshold O_2 concentration $(O_{2,def})$ takes the minimum of the hard-bound O_2 429 threshold concentration $(O_{2,crit})$ and the NO_3 to PO_4 ratio, scaled by a parameter k_D . The 430 parameters $O_{2,crit}$ and k_D are calibrated to give the global denitrification rate of roughly 431 100 Tg N yr⁻¹, which balances the total nitrogen fixation rate in the model. 432

$$Den_N = 0.8 [yr^{-1}] \cdot max \left(([O_2]_{def} - [O_2]), 0 \right)$$

$$60$$

$$[O_2]_{def} = min\left(O_{2,crit}, k_D \cdot \frac{[NO_3]_{inventory}}{[PO_4]_{inventory}}\right)$$

$$61$$

434 **2.10 Prognostic iron cycle**

The iron cycle in MESMO 3 remains the same as in MESMO 2. Key parameter values are
given in Table 2e. The two species of dissolved iron (Fe' and FeL) are partitioned according
to the following equilibrium relationship:

438

$$K_{ligand} = \frac{[FeL]}{[Fe] \cdot [L]}$$
⁶²

439

440 where [L] is the ligand concentration, and K_{ligand} is the conditional stability constant. The 441 sum of ligand and FeL is set at a constant value of 1 nmol kg⁻¹ everywhere. Iron is 442 introduced into the model domain by a constant fraction (3.5 weight %) of aeolian dust 443 deposition at the surface (F_{in}) following the prescribed modern flux pattern (Mahowald et 444 al., 2006) with constant solubility (β): 445

 $S_{Fe} = \beta \cdot F_{in} \tag{63}$

446

Particle-scavenged iron POM_{Fe} (note the difference from POFe) is produced below the
productive layer when sinking POM scavenges Fe['] to sinking POM:

$$J_{Fe} = -\tau_{sc} \cdot K_o \cdot [POC]^{0.58} \cdot [Fe]$$

450

451 where τ_{sc} and K_o and are empirical parameters that determine the strength of scavenging.

452 Remineralization of Fe scavenged to POM (POM_{Fe}) is identical in form to that of POM

453 remineralization:

$$R_{POM_{Fe}} = V_{POM} \cdot e^{k_R T \cdot \frac{[O_2]}{[O_2] + K_{O_2}}} \cdot [POM_{Fe}]$$

$$65$$

455

456 The conservation equation of the particle scavenged iron is thus expressed as :457

$$\frac{\partial [POM_{Fe}]}{\partial t} = J_{Fe} - \frac{\partial}{\partial z} (w[POM_{Fe}]) - R_{POM_{Fe}}$$
⁶⁶

458

Any scavenged iron that escapes remineralization in the water column reaching the
seafloor is removed from the model domain in order to keep the total Fe inventory at a
steady state.

462

463 2.11 Air-sea gas Exchange

The air-sea gas exchange formulation remains the same as in MESMO 2 and follows
Ridgwell et al. (2007). It is the function of gas transfer velocity, the ambient dissolved gas
concentration, and saturation gas concentration. The flux of CO₂ and O₂ gases across the
air-sea interface is given by:

468

$$F_{gas,CO_2} = k \cdot ([CO_2]_{sat} - [CO_2]) \cdot (1 - A)$$
⁶⁷

$$F_{gas,O_2} = k \cdot ([O_2]_{sat} - [O_2]) \cdot (1 - A)$$
68

469

where k is the gas transfer velocity, [CO₂]_{sat} and [O₂]_{sat} are saturation concentrations, and
A is the fractional ice-covered area that is calculated by the physical model. Gas transfer
velocity k is a function of wind speed (u) following Wanninkhof (1992) where Sc is the
Schmidt Number for a specific gas:

$$k = 0.31 \cdot u^2 \cdot \left(\frac{Sc}{660}\right)^{-0.5}$$
⁶⁹

476 3 Results and Discussion

477 All new results from MESMO 3 presented here are from the steady state. On a single 478 compute core at the Minnesota Supercomputing Institute, it takes approximately 1 hour to 479 complete 1,000 years of MESMO 3 simulation. The "standard" MESMO 3 has the power law 480 model of flexible stoichiometry but no DOM_r. The results from the standard model 481 (hereafter just MESMO 3) are presented in Section 3.1, and the results from the DOM_r-482 enabled model are presented in Section 3.2. In Table 3, we summarize and compare key 483 biogeochemical diagnostics from MESMO 3 against those from MESMO 2 and available 484 observational constraints. The global NPP, as well as global export production of POC and 485 opal, are comparable or somewhat lower in MESMO 3 than MESMO 2.

486

487 We relied on experience to calibrate MESMO 3 with the primary goal of reasonably 488 simulating the phytoplankton community composition and C:N:P ratio (e.g., abundant 489 cyanobacteria and high ratio for all PFTs in oligotrophic gyres). We tried to improve or at 490 least preserve the gains that we achieved in earlier versions of MESMO in terms of the global distributions of PO₄, NO₃, O₂, Si(OH)₄, and FeT (Supplemental Figures S1, S2, S3, S4, 491 492 and S5). Overall, there is a stronger nutrient depletion in the new model. For example, the 493 surface PO_4 and NO_3 in MESMO 3 are now sufficiently depleted in the subtropical gyres but 494 too low in the eastern equatorial Pacific when compared to the World Ocean Atlas (Figure 495 S1; see RMSE in Table 3). It is a challenge for MESMO and other coarse resolution models to 496 simulate narrow dynamical features such as equatorial upwelling and reproduce 497 biogeochemical features with sharp gradients. The spatial pattern of POC export that drives 498 this surface nutrient pattern is similar in the two models (Figure S2). In the 1D global 499 profile, there is a marked improvement in the subsurface distribution of O_2 in MESMO 3 500 over MESMO 2. Whereas the depth of the oxygen minimum was ~300 m in MEMOS 2, it is 501 ~1000 m in both MESMO 3 and the World Ocean Atlas (Figure S3). At 1000 m, the O₂ 502 minimum is located in the far North Pacific in MESMO 3, whereas in the World Ocean Atlas

503 it occurs in both the Northeast Pacific and the Arabian Sea. In contrast, the world ocean at 504 1000 m is too well-oxygenated in MESMO 2. We believe that the improved match in the O_2 505 minimum depth would help simulate denitrification in the correct depth range, and there is 506 a modest improvement in the data-model O₂ mismatch in terms of RMSE in MESMO 3 over 507 MESMO 2 (Table 3). The deepening of the O₂ minimum was achieved largely by increasing 508 the particle sinking speed, which tends to strengthen the biological pump and deplete the 509 surface nutrients. This also helps MESMO 3 preserves MESMO 2's surface Si(OH)₄ depletion 510 in much of the world ocean except in the North Pacific and Southern Ocean (Figure S4). 511 This is a feature captured by $Si^* < 0$ ($Si^* = [Si(OH)_4] - [NO_3]$) in observations (Sarmiento et al., 512 2004) and simulated previously by MESMO 2 and now MESMO 3. Finally, surface FeT is also 513 depleted more strongly in MESMO 3 over MESMO2, except the North Atlantic, where 514 aeolian deposition of dust from the Sahara maintains a steady Fe supply (Figure S5).

515

516 In MESMO 3, we made no effort to calibrate all the semi-labile pools of DOM: DOC_{sl} , DOP_{sl} , 517 DON_{sl} , and $DOFe_{sl}$. We note only that the surface DOC_{sl} concentration of 58 µmol kg⁻¹ and

518 DOC export production of 1.4 Pg C yr⁻¹ in MESMO 3 are higher than in MESMO 2 (24 µmol

519 kg⁻¹ and 0.4 Pg C yr⁻¹, respectively). The higher surface concentration is due to the longer τ_{sl}

520 in MESMO 3 (Table 2d). The global average of the temperature dependent fDOM in MESMO

521 3 is 0.69, which is slightly higher than the spatially uniform value of 0.67 in MESMO 2.

522

523 **3.1 Novel features of MESMO 3**

524 An important new feature of MESMO 3 is the representation of the primary producers by 525 three PFTs (Figure 2). The eukaryotes are characterized by the highest maximum growth 526 rate and high half-saturation constants. Thus, the eukaryotes are more dominant than the 527 other PFTs in the more eutrophic waters of the equatorial and polar regions (Figure 2a). 528 The cyanobacteria have smaller half-saturation constants and thus are more dominant in 529 the oligotrophic subtropical gyres (Figure 2c). The diazotrophs do not have NO₃ limitation 530 but have the lowest maximum growth rate. Thus it is much lower in abundance than the 531 other two PFTs generally, and outcompeted in transient blooms and thus excluded in 532 higher latitudes (Figure 2e).

534 Figure 2 also indicates that all three PFTs show Fe limitation in the Southern Ocean. 535 Outside the Southern Ocean, the eukaryotes are primarily limited by Si(OH)₄ (Figure 2b). 536 As far as organic carbon is concerned, we consider the eukaryotes to basically represent 537 diatoms, which are arguably the most important agent of organic carbon export. In this 538 context, the widespread silica limitation for eukaryotes would be consistent with the 539 notions that silica uptake by diatoms should be limited in $\sim 60\%$ of the world surface ocean 540 (Ragueneau et al. 2000) and that much the world ocean thermocline is filled with silica-541 depleted water (Si*<0 as noted above). On the other hand, the cyanobacteria are largely 542 limited by NO₃ outside the Southern Ocean (Figure 2d). The diazotrophs are limited by iron 543 in much of the world ocean except in the Atlantic basin (Figure 2f), where surface PO_4 is 544 strongly depleted in both observations (Mather et al., 2008) and in our model (Figure S1). 545 546 Figure 3 illustrates the influence of the RNPG index, which was implemented in MESMO 3 547 to allow for the effect of competition between diatoms and coccolithophores within the 548 same PFT (Equations 22 and 23). The eukaryote NPP (Figure 3a) is effectively split into two

549 parts: one is associated with diatoms and opal production (Figure 3b), and the other is

associated with coccolithophores and CaCO₃ production (Figure 3c). According to the RNPG

index, opal production is simulated more in the higher latitudes of the Southern Ocean and

- the North Pacific, where surface [Si(OH)₄] is abundant. Elsewhere, CaCO₃ production is
- relatively larger. The decoupling is prominent in the North Indian. Note that the spatial
- pattern of CaCO₃ production is quite different in MEMOS 3 (Figure 3c) and MESMO 2
- 555 (Figure 3d), because CaCO₃ production was associated in MESMO 2 with the "small" PFT,
- which corresponds to the cyanobacteria PFT in MESMO 3.
- 557

The global pattern of the mean C:P uptake ratio in the production layer is shown in Figure 4. Consistent with observations (Martiny et al., 2013), the simulated C:P ratio of the phytoplankton community is elevated in the oligotrophic subtropical gyres and low in the eutrophic polar waters (Figure 4a). The community C:P ratio exceeds 200 in the gyres and reaches as low as 40-50 in the Southern Ocean. The community C:P has contributions from both physiological effects (i.e., environmental drivers acting on each PFT's C:P ratio) and 564 taxonomic effects (i.e., the shift in the community composition changes the weighting of 565 each PFT's C:P ratio). Figure 4b shows that the community C:P is high in oligotrophic gyres 566 because cyanobacteria and to a lesser extent diazotrophs dominate the community, and 567 their C:P ratio is high. Conversely, the community C:P is low in the polar waters because the 568 eukaryotes dominate and their C:P ratio is low. For both eukaryotes and cyanobacteria, 569 their C:P is high in oligotrophic subtropical gyres because PO₄ is low (Figure 4c and d). This 570 physiological effect is larger in eukaryotes than cyanobacteria because the former has greater sensitivity (i.e., larger sensitivity factor $s_{PO_4}^{P:C}$, Equation 5, Table 2b). However, the 571 cyanobacteria PFT's C:P ratio has an additional sensitivity to temperature (i.e., $s_T^{P:C} \neq 0$) 572 that elevates their C:P in the lower latitudes. We do not show the C:P ratio for diazotrophs 573 574 because it is very similar to that of cyanobacteria (Figure 4b, d).

575

576 In order to gain more insights into the spatial patterns of the C:P ratio (Figure 4), we 577 examined the relationships between the C:P and C:N ratios and the four possible 578 environmental drivers for eukaryotes and cyanobacteria (Figure 5; again, diazotrophs are 579 not shown). The red plots show that there is a causal relationship between the ratios and 580 the drivers as formulated by the power law model (Equations 5 and 6). The black plots 581 show the absence of a causal relationship. For example, the C:P ratio of both eukaryotes 582 and cyanobacteria are strongly correlated with PO₄ because there is a causal relationship 583 (Figure 5a, b shown in red). Similarly, the C:N ratio of the same two PFTs have a strong 584 correlation with PO₄ (Figure 5c, d in black), but there is actually not a causal relationship (i.e., $s_{PO_4}^{N:C} = 0$, Table 2b). The C:N-PO₄ correlation exists, simply because the nutrients are 585 586 well correlated. Similarly, because temperature and photosynthetically active radiation 587 (PAR) tend to be correlated via latitude, the stoichiometry has a similar correlation to the 588 two drivers. For example, cyanobacteria C:P has a strong correlation with both 589 temperature and PAR (Figure 5j, 4n), but only the temperature is a real driver. Figure 5 590 indicates which are the dominant drivers of the C:N:P ratio in MESMO 3. For the eukaryote 591 C:P ratio, it is PO₄. For the cyanobacteria C:P ratio, the important drivers are temperature 592 and PO₄. For the C:N ratio for both eukaryotes and cyanobacteria, NO₃ is more important 593 than PAR. Figure 5 also serves to remind us that correlation does not indicate causation.

595 Figure 6 shows the community C:P and C:N ratios plotted against the four environmental 596 drivers. Unlike Figure 5, which reflected the individual PFT's physiological response, Figure 597 6 includes the effect of taxonomy as well. Still, the effects of PO₄ and temperature are 598 clearly visible on the community C:P ratio. Both low [PO₄] and warmer waters are found in 599 the lower latitudes, so the P frugality and temperature effects are additive. The effect of 600 NO₃ on the community C:N ratio is also very clear, but the effect of PAR is not as clear. Thus 601 overall, the physiological effects seen in the PFT-specific C:N:P are obvious in the 602 community C:N:P ratio.

603

604 3.2 DOM_r-enabled MESMO 3

In MESMO 2, DOC_{sl} was a standard state variable. In MESMO 3, other forms of DOM are
available as options. They are the semi-labile forms of DOM: DOP_{sl}, DON_{sl}, and DOFe_{sl}; and
the refractory forms of DOM: DOC_r, DOP_r, and DON_r. MESMO 3 is not yet calibrated with
respect to all the DOM variables, but here we demonstrate their potential use in future
biogeochemical investigations by presenting steady state DOM results from the model
experiment *LV* (experiment ID: 210310m). In this run, all three sinks of DOM_r are activated:
slow background decay, photodegradation, and degradation in hydrothermal vents.

612

613 The experiment name LV stands for "literature values." In LV, we use the literature values 614 for the key DOM remineralization model parameters (Table 2d) and $fDOM_r = 0.01$ (Hansell, 615 2013). All other model parameter values in the LV run are identical to the standard MESMO 616 3 model (Table 2). The black lines in Figure 7 show the global mean vertical profiles of the 617 total DOC ($DOC_t = DOC_{sl} + DOC_r$) in solid line and DOC_r in dashed line. Qualitatively, the 618 simulated profiles are consistent with the observations, showing a near-uniform DOC_r 619 concentration and a DOC_{sl} profile that rapidly with depth in the top few hundred meters 620 (Hansell, 2013). However, the simulated values reach 130 µmol kg⁻¹ in the surface, which is approximately twice the observations. More typically, the observed DOC_r is 30~40 µmol kg⁻ 621 ¹, and the observed DOC_{sl} attenuates with depth from $30 \sim 40 \mu$ mol kg⁻¹ near the surface. So 622

625

Figure 8 adds a lateral perspective to Figure 7. The rapid DOC_t attenuation in the vertical is strong in the lower latitudes where stratification is generally stronger. The transport of DOC_{sl} from the surface to deeper waters is evident in the high latitudes of the North Atlantic and the Southern Ocean. The DOC_t change in the deep ocean is limited. Observations of deep ocean DOC_t indicates a reduction by 29% or 14 μ mol kg⁻¹ from the deep North Atlantic to the deep North Pacific (Hansell and Carlson, 1998). Figure 8 shows that the deep ocean DOC_t gradient in *LV* is approximately 10 μ mol kg⁻¹ from 70-75 μ mol kg⁻¹

 633 1 in the North Atlantic to <65 $\mu mol~kg^{\text{-1}}$ in the North Pacific.

634

635 The horizontal DOCt distributions from the LV run can also be compared to a global
636 extrapolation based on an artificial neural network (ANN) of the available DOCt data

637 (Roshan and DeVries, 2017). At the surface, the extrapolation indicates higher DOC_t

638 concentrations in the subtropical gyres (Figure 9a), while our simulation does not clearly

639 delineate the gyres (Figure 9c). In our model, fDOM is temperature-dependent and strongly

640 controls the production of DOM. The surface DOC_t is thus more elevated in the lower

641 latitudes. Interestingly, the ANN study diagnosed higher rates of DOM production in the

642 subtropical gyres. Since the oligotrophic subtropical gyres have low NPP, the diagnosis

643 would thus suggest that somehow fDOM is higher in the gyres. At depths, both the

644 extrapolated and simulated DOC_t show a gradual decline in concentrations from the North

645 Atlantic to the North Pacific (Figure 9b, d). The highest deep DOC_t in the *LV* run is seen just

646 south of Greenland, where convection occurs in the model.

647

Finally, we show that the deep ocean radiocarbon aging is larger in DIC than in DOC_t in the

649 model (Figure 10). The North Pacific-North Atlantic Δ^{14} C gradient is roughly -100‰ for

650 DIC and -70‰ for DOC_t. The oldest DOC_t Δ^{14} C is approximately -430‰ in the North Pacific.

651 If ¹⁴C decay were the only mechanism of change along the path of the deepwater

652 circulation, the Δ^{14} C gradient should be quite similar between DIC and DOC_t, which are both

- dissolved phases and transported passively by the same circulation. The one potentially
- 654 important difference is that the addition of the relatively young DI¹⁴C and DO¹⁴C to the
- deep ocean by the "deep POC split" (see Section 2.3) impacts DOC_t Δ^{14} C more than DIC Δ^{14} C,
- 656 because DOC_t is two orders of magnitude lower in concentration than DIC.
- 657

In observations, the aging of DIC and DOC_t is reportedly similar in the Antarctic Bottom 658 659 Water (below 4000 m) of the deep Pacific (Druffel et al., 2019). This may be explained by 660 the fact that there would not be much deep POC split occurring so deep in the ocean. The 661 North Pacific-North Atlantic Δ^{14} C gradient, accounting for thermonuclear bomb 14 C, may be 662 as large as -100‰ for DOCt (about -550‰ in the deep Pacific and -456‰ in the deep Atlantic) (Druffel et al., 2019). This gradient is not rigorously determined, because there is 663 664 not enough data to do an objective analysis. Therefore, the equivalent Δ^{14} C gradient for DIC cannot be determined. However, the DIC Δ^{14} C endmember values by inspection (about -665 666 250‰ in the deep Pacific and -70‰ in the deep Atlantic) (Matsumoto and Key, 2004) 667 indicate a clearly larger Δ^{14} C gradient for DIC than DOC_t as simulated by the experiment *LV*.

668

669 One lesson from the data-*LV* run mismatch in the overall DOC_t concentration (Figure 7) and 670 surface DOC_t pattern (Figure 9) is that the parameter values from the literature do not fully 671 capture the DOC cycle and/or MESMO 3 is still lacking some important DOC process. In LV, 672 the surface DOC_t is too high because DOC_r is too high, while DOC_s is not unreasonable 673 (Figure 7). DOC_r is too high because there is too much DOC_r production (e.g., $fDOM_r=1\%$ is too large), there is too little DOCr degradation (e.g., one of the DOM decay mechanisms is 674 675 too slow; Equation 28 and Table 2d), or some combination of both. For example, $fDOM_r$ is a 676 key parameter that is not well constrained by observations. Had we used 0.2% instead of 677 1% for fDOM_r, the global mean surface DOC_t drops to 76 μ mol kg⁻¹ (red line, Figure 7), 678 consistent with observations. For achieving a better surface DOC_t pattern, we may need a 679 different formulation of fDOM that is, for example, negatively related to nutrient 680 concentrations so that fDOM increases in the oligotrophic subtropical gyres (Roshan and 681 DeVries, 2017).

683 Another lesson from the DOM modeling exercise is that it is important to simulate DOP_r 684 reasonably well in order to preserve the favorable results we achieved in MESMO 3 with 685 respect to biological production and the phytoplankton C:N:P ratio. We find that in the 686 experiment LV, the global mean DOP_r concentration becomes steady at 0.45 µmol-P kg⁻¹. In 687 observations, the mean DOC_r is about 40 µmol-C kg⁻¹, and the DOC_r:DOP_r ratio is estimated to be \sim 1370:1 (Letscher and Moore, 2015), so DOP_r concentration should only be roughly 688 689 0.03 µmol-P kg⁻¹. Thus, the simulated DOP_r=0.45 µmol-P kg⁻¹ is an order of magnitude too 690 high. Because there is more P in the form of DOP_r in LV, the oceanic inventory of PO_4 691 declines, causing a nearly 10% drop in export production compared to the standard 692 MESMO 3. In LV, the decline in the surface ocean PO_4 that accompanies the change in the 693 PO_4 inventory acts on the phytoplankton physiology (i.e., P effect on C:P in Equation 5), 694 which leads to a large rise in the global mean phytoplankton community C:P export ratio 695 from 113:1 to 127:1. The implementation of preferential remineralization of DOP (and 696 DON) over DOC (Letscher and Moore, 2015) is one way to deal with the problem of too high 697 DOP_r concentrations.

698

699 **3.3 Large-scale patterns of N2 fixation and denitrification**

700 The modeled habitat of diazotrophs is concentrated in tropical and subtropical waters 701 between 40°S and 40°N and limited by iron (Figure 1e, f). Most noticeably in North Pacific 702 subtropical gyre, diazotrophs constitute $\sim 40\%$ of total NPP. The latitudinal extent of 703 diazotrophs is mainly determined by surface nitrate availability and physical factors such 704 as surface temperature and irradiance. Low nitrate availability in subtropical gyres gives 705 diazotrophs a competitive advantage over small cyanobacteria. Warm temperature and 706 high irradiance also critical physical factors that drive the growth of diazotrophs in the 707 model.

708

The modeled global depth-integrated N_2 fixation is 101 Tg N yr⁻¹ (Table 3), and this value

710 falls well within the range of observational and geochemical constraints of 80 – 200 Tg N

- 711 yr⁻¹ (Landolfi et al., 2018). In MESMO 3, N₂ fixation occurs in the North Pacific and mid-to-
- 712 low latitudes of the Atlantic basin (Supplementary Figure S6), where diazotrophs are

generally more abundant (Figure 2e). The elevated N₂ fixation rate in the North Pacific,
where nitrate limits eukaryotes and cyanobacteria (Figure 2b, d), can be explained by the
healthy growth of diazotrophs, which is not limited by N. In the subtropical and tropical
Atlantic and the Indian Ocean, high N₂ fixation is driven by elevated C:P and N:P ratio
(Figure 4), exemplified by low phosphate availability and warm surface temperature. This
spatial pattern agrees with a recent inverse model study (Wang et al., 2019), which showed
an elevated N₂ fixation rate in subtropical gyres.

720

721 Global water-column denitrification is 101 Tg N yr⁻¹ (Table 3) and is equal to the global N₂ 722 fixation because the model has reached steady state. Denitrification is restricted to the subpolar North Pacific (Figure S6), where sub-surface oxygen concentration is significantly 723 724 depleted (Figure S3d). Enhanced denitrification in this region is in qualitative agreement 725 with a previous modeling study (Bianchi et al., 2018), which showed the anaerobic niche 726 due to particle microenvironments can significantly expand the hypoxic expanses in the 727 North Pacific. However, the extent of denitrification in our model does not include the 728 eastern equatorial Pacific and northern Indian, which are important hotspots for 729 denitrification (Codispoti, 2007; Deutsch et al., 2007). This issue is typical of coarse-730 resolution global ocean biogeochemistry models that lack spatial resolution in reproducing 731 intense upwelling (Marchal et al., 1998; Najjar et al., 1992; Yamanaka and Tajika, 1997). 732

Finally, the ratio of the global inventories of NO₃ and PO₄ in MESMO 3 is just about 16 at
steady state, consistent with observations (Gruber and Sarmiento, 1997). One key model
parameter in this regard is the nitrate uptake half saturation constant of diazotrophs, K_{NO3,3}
in Equation 2. A large value of K_{NO3,3} will make it hard for diazotrophs to obtain fixed N
from NO₃, which would facilitate N₂ fixation and pushes up the global N/P ratio. With a
smaller value of K_{NO3,3}, diazotrophs will more easily uptake NO₃, thus depressing N₂
fixation, lowering the global N/P ratio.

740

741 **4. Conclusions**

742 MESMO 3, the third and latest version of MESMO, is comprehensively described here. With

a fully flexible C:N:P ratio in three PFTs, a prognostic N cycle, and more mechanistic

- 744 schemes of organic matter production and remineralization, MESMO 3 reflects the evolving 745 and accumulating knowledge of the ocean biogeochemistry. The model thus remains an 746 effective tool for investigations of the global biogeochemical cycles especially on long time 747 scale given the model's computational efficiency. In particular, MESMO 3 holds promise for 748 studying the marine DOM cycle. The optional features of MESMO 3 include the semi-labile 749 and refractory pools of C, P, N, and Fe. The fact that the literature values regarding the 750 present marine DOM cycle are unable to simulate key observations indicates an 751 opportunity for MESMO 3 to contribute to an improved understanding of the marine DOM 752 cvcle. 753 754 **Code availability** 755 The complete code of MESMO version 3.0 and results presented here are available at 756 GitHub https://github.com/gaia3intc/mesmo.git and have a DOI: 757 10.5281/zenodo.4403605. 758 759 **Author contribution** 760 KM, TT, and JZ developed the model code. KM performed the simulations, carried out 761 analyses, and archived the model code and results. KM and TT wrote the paper. 762 763 Acknowledgements
- This work was funded by the US National Science Foundation (OCE-1827948). Numerical
- 765 modeling and analysis were carried out using resources at the University of Minnesota
- 766 Supercomputing Institute.

768	Tables
769	
770	Table 1. MESMO Development
771 772 773 774 775 776 777 778 779 780	PFT = phytoplankton functional types. MESMO2 PFTs are LG = large/diatoms and SM = small. MESMO 3 PFTs are Eu = eukaryotes, Cy = cyanobacteria, and Dz = diazotrophs. OM = organic matter. RNPG = residual nitrate potential growth. T = temperature. PAR = photosynthetically active radiation. fDOM = fraction of NPP routed to dissolved organic matter (DOM). The two types of DOM are semi-labile (DOC, DOP, DON, and DOFe) and refractory (DOCr, DOPr, and DONr). Carbon isotopes (12 C, 13 C, and 14 C) are calculated separately for DOC and DOCr. The run ID is 210310m for the MESMO 3 experiment LVR and 210310o for the experiment LVR with fDOM _r =0.2%.
781	Table 2. MESMO 3 Biogeochemical Model Parameters Values
782	
783	Table 3. Key Biogeochemical Model Diagnostics
784 785 786 787 788 789 790 791	^a NPP for MESMO 2 was unavailable as a model output and therefore estimated from POC and fDOM=0.67. ^b NPP (in terms of C) is needed in the calculation of the PFT abundance. The root mean square error (RMSE) of the simulated P, N, Si, and O ₂ distributions from MESMO 2 and 3 was calculated relative to the World Ocean Atlas 2018 gridded data (Garcia et al., 2018, 2019). The model-data comparison is made in the top 100 m for nutrients and below 100 m for O ₂ . WOA18 was regridded to the MESMO 3 grid to calculate the RMSE.
 791 792 793 794 795 796 797 798 799 800 801 	References for independent constraints: (1) global NPP (Carr et al., 2006); (2) global POC export (DeVries and Weber, 2017); (3) global DOC export assumed to be 20% of total carbon export (Hansell et al., 2009; Roshan and DeVries, 2017); (4) global opal (Dunne et al., 2007); (5) global CaCO ₃ export (Berelson et al., 2007); (6) global N fixation and denitrification rates (Landolfi et al., 2018); (7) uptake C:N:P ratio is based on POM measurements (Martiny et al., 2013); (8) export C:N:P ratio is assumed to equal the subsurface remineralization ratio (Anderson and Sarmiento, 1994); (9) Deep O ₂ from WOA18 below 100 m (Garcia et al., 2019).
802	

804	Figures
805	
806	Figure 1. Schematic diagram of DOM cycling in MESMO 2 versus MESMO 3. In the new
807	model, DOM_{r} can be activated. DOM_{r} is produced from POM breakdown, which can occur in
808	the production layer or throughout the water column in the "deep POC split." Possible
809	$\mathrm{DOM}_{\mathrm{r}}$ remineralization mechanisms are the slow background degradation that occurs
810	everywhere, thermal degradation in hydrothermal vents, and photodegradation in the
811	surface. See text for details.
812	
813	Figure 2. NPP-based surface phytoplankton functional type (PFT) abundance and nutrient
814	limitation in MESMO 3. Fractional abundance and nutrient limitation for eukaryotes (a, b),
815	cyanobacteria (c, d), and diazotrophs (e, f).
816	
817	Figure 3. Eukaryote production in MESMO 3 and $CaCO_3$ export in MESMO 2. In MESMO 3,
818	eukaryote NPP (a) is linked to both opal export (b) and $CaCO_3$ export (c) but the two export
819	productions are differentiated by the residual nitrate potential growth (RNPG). Compare
820	CaCO ₃ export in MESMO 3 (c) to MESMO 2 (d). Unit = mol m^{-2} year ⁻¹ .
821	
822	Figure 4. Uptake C:P ratio in the top 100 m in MESMO 3: (a) phytoplankton community C:P,
823	(b) zonal mean C:P of all three PFTs and phytoplankton community, (c) eukaryote C:P, and
824	(d) cyanobacteria C:P. The colors in (b) indicate: black = community C:P, red = eukaryote
825	C:P, green = cyanobacteria C:P, and blue = diazotroph C:P. Also, (b) shows the range of
826	observed C:P ratio binned by latitude (Martiny et al., 2013).
827	
828	Figure 5. Scatter plots of surface ocean eukaryote and cyanobacteria C:P and C:N vs.
829	environmental drivers in MESMO 3. Columns: 1 = eukaryote C:P, 2 = cyanobacteria C:P, 3 =
830	eukaryote C:N, and 4 = cyanobacteria C:N. Rows: 1 = PO ₄ , 2 = NO ₃ , 3 = temperature, and 4 =
831	PAR. Red indicates causal relationship according to the power law formulation of flexible
832	C:N:P ratio. PAR = photosynthetically active radiation in W m^{-2} .
833	

Figure 6. Scatter plots of surface ocean community C:P and C:N vs environmental drivers inMESMO 3.

- 836
- 837 **Figure 7.** Global mean vertical profiles of DOC from the DOM_R-enabled MESMO 3. DOC_t
- 838 (DOC_{sl}+DOC_r, black line) and DOC_r (black dashed line) from the *LV* run. Red line is DOC_t
- after reducing fDOM_r from 1% in LV to 0.2% (Experiment 2103100). Unit = μ mol kg⁻¹.
- 840
- Figure 8. Global depth-latitude transect of DOC_t from the DOM_R -enabled MESMO 3 *LV* run.
- 842 Transects are N-S along 25°W in the Atlantic, E-W along 60°S in the Southern Ocean, and N-
- 843 S along 165°E in the Pacific. Unit = μ mol kg⁻¹.
- 844
- **Figure 9.** Assessment of surface and deep ocean DOC_t from the DOM_R-enabled MESMO 3 *LV*
- run. Data-derived DOCt distributions in the top 100 m (a) and 2500-4000 m (b). Model-
- simulated DOC_t distributions in the top 100 m (c) and 2500-4000 m (d). Date-derived DOC_t
- are from Roshan and DeVries (Roshan and DeVries, 2017). Unit = μ mol kg⁻¹.
- 849
- Figure 10. Δ^{14} C of deep ocean DIC (a) and DOC_t (b) from the DOM_R-enabled MESMO 3 *LV*
- run. Vertical average over 2500-4000 m water depth. Unit = ‰.
- 852

853 **References:**

Anderson, L. A. and Sarmiento, J. L.: Redfield ratios of remineralization determined by
nutrient data analysis, Global Biogeochem. Cycles, 8(1), 65–80,

856 doi:10.1029/93GB03318, 1994.

- Archer, D., Eby, M., Brovkin, V., Ridgwell, A., Cao, L., Mikolajewicz, U., Caldeira, K.,
- 858 Matsumoto, K., Munhoven, G., Montenegro, A. and Tokos, K.: Atmospheric Lifetime of
- Fossil Fuel Carbon Dioxide, Annu. Rev. Earth Planet. Sci., 37(1), 117–134,
- 860 doi:10.1146/annurev.earth.031208.100206, 2009.
- 861 Arteaga, L., Pahlow, M. and Oschlies, A.: Global patterns of phytoplankton nutrient and light
- 862 colimitation inferred from an optimality-based model, Global Biogeochem. Cycles,
- 863 28(7), 648–661, doi:10.1002/2013GB004668, 2014.
- Balch, W. M., Bates, N. R., Lam, P. J., Twining, B. S., Rosengard, S. Z., Bowler, B. C., Drapeau, D.
- T., Garley, R., Lubelczyk, L. C., Mitchell, C. and Rauschenberg, S.: Factors regulating the
 Great Calcite Belt in the Southern Ocean and its biogeochemical significance, Global
 Biogeochem. Cycles, 30(8), 1124–1144, doi:10.1002/2016GB005414, 2016.
- 868 Berelson, W. M., Balch, W. M., Najjar, R., Feely, R. A., Sabine, C. and Lee, K.: Relating
- 869 estimates of CaCO3 production, export, and dissolution in the water column to
- 870 measurements of CaCO3 rain into sediment traps and dissolution on the sea floor:A
- 871 revised global carbonate budget, Global Biogeochem. Cycles, 21(1), 1–15,
- 872 doi:10.1029/2006GB002803, 2007.
- 873 Bianchi, D., Weber, T. S., Kiko, R. and Deutsch, C.: Global niche of marine anaerobic
- 874 metabolisms expanded by particle microenvironments, Nat. Geosci., 11(4), 263–268,
- 875 doi:10.1038/s41561-018-0081-0, 2018.
- Brzezinski, M. A.: A switch from Si(OH) 4 to NO 3 depletion in the glacial Southern Ocean,
 Geophys. Res. Lett., 29(12), doi:10.1029/2001gl014349, 2002.
- 878 Cao, L., Eby, M., Ridgwell, A., Caldeira, K., Archer, D., Ishida, A., Joos, F., Matsumoto, K.,
- 879 Mikolajewicz, U., Mouchet, A., Orr, J. C., Plattner, G.-K., Schlitzer, R., Tokos, K.,
- 880 Totterdell, I., Tschumi, T., Yamanaka, Y. and Yool, A.: The role of ocean transport in the
- uptake of anthropogenic CO₂, Biogeosciences, 6(3), 375–390, doi:10.5194/bg-6-375-
- 882 2009, 2009.
- 883 Carr, M. E., Friedrichs, M. A. M., Schmeltz, M., Noguchi Aita, M., Antoine, D., Arrigo, K. R.,

- Asanuma, I., Aumont, O., Barber, R., Behrenfeld, M., Bidigare, R., Buitenhuis, E. T.,
- 885 Campbell, J., Ciotti, A., Dierssen, H., Dowell, M., Dunne, J., Esaias, W., Gentili, B., Gregg,
- 886 W., Groom, S., Hoepffner, N., Ishizaka, J., Kameda, T., Le Quéré, C., Lohrenz, S., Marra, J.,
- 887 Mélin, F., Moore, K., Morel, A., Reddy, T. E., Ryan, J., Scardi, M., Smyth, T., Turpie, K.,
- 888 Tilstone, G., Waters, K. and Yamanaka, Y.: A comparison of global estimates of marine
- primary production from ocean color, Deep. Res. Part II Top. Stud. Oceanogr., 53(5–7),
- 890 741–770, doi:10.1016/j.dsr2.2006.01.028, 2006.
- 891 Cheng, H., Edwards, R. L., Southon, J., Matsumoto, K., Feinberg, J. M., Sinha, A., Zhou, W., Li,
- H., Li, X., Xu, Y., Chen, S., Tan, M., Wang, Q., Wang, Y., Ning, Y., Lawrence Edwards, R.,
- 893 Southon, J., Matsumoto, K., Feinberg, J. M., Sinha, A., Zhou, W., Li, H., Li, X., Xu, Y., Chen,
- 894 S., Tan, M., Wang, Q., Wang, Y. and Ning, Y.: Atmospheric 14C/12C changes during the
- last glacial period from hulu cave, Science (80-.)., 362(6420), 1293–1297,
- 896 doi:10.1126/science.aau0747, 2018.
- 897 Chien, C. Te, Pahlow, M., Schartau, M. and Oschlies, A.: Optimality-based non-Redfield
- plankton-ecosystem model (OPEM v1.1) in UVic-ESCM 2.9 Part 2: Sensitivity analysis
 and model calibration, Geosci. Model Dev., 13(10), 4691–4712, doi:10.5194/gmd-134691-2020, 2020.
- 901 Claussen, M., Mysak, L., Weaver, A., Crucifix, M., Fichefet, T., Loutre, M. F., Weber, S., Alcamo,
- 902 J., Alexeev, V., Berger, A., Calov, R., Ganopolski, A., Goosse, H., Lohmann, G., Lunkeit, F.,
- 903 Mokhov, I., Petoukhov, V., Stone, P. and Wang, Z.: Earth system models of intermediate
- 904 complexity: Closing the gap in the spectrum of climate system models, Clim. Dyn.,
- 905 18(7), 579–586, doi:10.1007/s00382-001-0200-1, 2002.
- 906 Codispoti, L. A.: An oceanic fixed nitrogen sink exceeding 400 Tg N a-1 vs the concept of
- 907 homeostasis in the fixed-nitrogen inventory, Biogeosciences, 4(2), 233–253,
- 908 doi:10.5194/bg-4-233-2007, 2007.
- 909 Deutsch, C., Sarmiento, J. L., Sigman, D. M., Gruber, N. and Dunne, J. P.: Spatial coupling of
- 910 nitrogen inputs and losses in the ocean, Nature, 445(7124), 163–167,
- 911 doi:10.1038/nature05392, 2007.
- 912 DeVries, T. and Weber, T.: The export and fate of organic matter in the ocean: New
- 913 constraints from combining satellite and oceanographic tracer observations, Global
- 914 Biogeochem. Cycles, 31(3), 535–555, doi:10.1002/2016GB005551, 2017.

- Druffel, E. R. M., Williams, P. M., Bauer, J. E. and Ertel, J. R.: Cycling of dissolved and
 particulate organic matter in the open ocean, J. Geophys. Res., 97(C10), 15639,
 doi:10.1029/92JC01511, 1992.
- 918 Druffel, E. R. M., Griffin, S., Wang, N., Garcia, N. G., McNichol, A. P., Key, R. M. and Walker, B.
- D.: Dissolved Organic Radiocarbon in the Central Pacific Ocean, Geophys. Res. Lett.,
 46(10), 5396–5403, doi:10.1029/2019GL083149, 2019.
- 921 Dunne, J. P., Armstrong, R. A., Gnanadesikan, A. and Sarmiento, J. L.: Empirical and
- mechanistic models for the particle export ratio, Global Biogeochem. Cycles, 19(4),
 n/a-n/a, doi:10.1029/2004GB002390, 2005.
- Dunne, J. P., Sarmiento, J. L. and Gnanadesikan, A.: A synthesis of global particle export from
 the surface ocean and cycling through the ocean interior and on the seafloor, Global
 Biogeochem. Cycles, 21(4), 1–16, doi:10.1029/2006GB002907, 2007.
- 927 Eby, M., Weaver, A. J., Alexander, K., Zickfeld, K., Abe-Ouchi, A., Cimatoribus, A. A., Crespin,
- 928 E., Drijfhout, S. S., Edwards, N. R., Eliseev, A. V., Feulner, G., Fichefet, T., Forest, C. E.,
- 929 Goosse, H., Holden, P. B., Joos, F., Kawamiya, M., Kicklighter, D., Kienert, H., Matsumoto,
- 930 K., Mokhov, I. I., Monier, E., Olsen, S. M., Pedersen, J. O. P., Perrette, M., Philippon-
- 931 Berthier, G., Ridgwell, A., Schlosser, A., Schneider von Deimling, T., Shaffer, G., Smith, R.
- 932 S., Spahni, R., Sokolov, A. P., Steinacher, M., Tachiiri, K., Tokos, K., Yoshimori, M., Zeng,
- 933 N. and Zhao, F.: Historical and idealized climate model experiments: an
- 934 intercomparison of Earth system models of intermediate complexity, Clim. Past, 9(3),
- 935 1111–1140, doi:10.5194/cp-9-1111-2013, 2013.
- 936 Edwards, N. R. and Marsh, R.: Uncertainties due to transport-parameter sensitivity in an
- 937 efficient 3-D ocean-climate model, Clim. Dyn., 24(4), 415–433, doi:10.1007/s00382938 004-0508-8, 2005.
- Eppley, R. W.: Temperature and phytoplankton growth in the sea, Fish. Bull., 70(4), 1063–
 85, 1972.
- Galbraith, E. D. and Martiny, A. C.: A simple nutrient-dependence mechanism for predicting
 the stoichiometry of marine ecosystems, Proc. Natl. Acad. Sci., 112(27), 8199–8204,
 doi:10.1073/pnas.1423917112, 2015.
- Garcia, H., Weathers, K. W., Paver, C. R., Smolyar, I., Boyer, T. P., Locarnini, R. A., Zweng, M.
- 945 M., Mishonov, A. V., Baranova, O. K., Seidov, D. and Reagan, J. R.: World Ocean Atlas

- 946 2018. Volume 4: Dissolved Inorganic Nutrients (phosphate, nitrate and nitrate+nitrite,
 947 silicate), NOAA Atlas NESDIS 84, 84(July), 35, 2018.
- Garcia, H. E., Weathers, K., Paver, C. R., Smolyar, I., Boyer, T. P., Locarnini, R. A., Zweng, M.
- 949 M., Mishonov, A. V., Baranova, O. K., Seidov, D. and Reagan, J. R.: World Ocean Atlas
- 950 2018, Volume 3: Dissolved Oxygen, Apparent Oxygen Utilization, and Oxygen
- 951 Saturation, NOAA Atlas NESDIS, 3(83), 38 pp., 2019.
- 952 Gruber, N. and Sarmiento, J. L.: Global patterns of marine nitrogen fixation and
- 953 denitrification, Global Biogeochem. Cycles, 11(2), 235–266, doi:10.1029/97GB00077,
 954 1997.
- Hansell, D. A.: Recalcitrant Dissolved Organic Carbon Fractions, Ann. Rev. Mar. Sci., 5, 421–
 445, doi:10.1146/annurev-marine-120710-100757, 2013.
- Hansell, D. A. and Carlson, C. A.: Deep-ocean gradients in the concentration of dissolved
 organic carbon, Nature, 395(6699), 263–266, doi:10.1038/26200, 1998.
- Hansell, D. A., Carlson, C. A., Repeta, D. J. and Schlitzer, R.: Dissolved Organic Matter in the
 Ocean: A Controversy stimulates new insights, Oceanography, 22(4), 202–211,
- 961 doi:https://doi.org/10.5670/oceanog.2009.109, 2009.
- 962 Henson, S. A., Sanders, R., Madsen, E., Morris, P. J., Le Moigne, F. and Quartly, G. D.: A
- 963 reduced estimate of the strength of the ocean's biological carbon pump, Geophys. Res.
 964 Lett., 38(4), n/a-n/a, doi:10.1029/2011GL046735, 2011.
- Hutchins, D. A. and Bruland, K. W.: Iron-limited growth and Si:N ratios in a costal upwelling
 regime, Nature, 393(June), 561–564, 1998.
- 967 Joos, F., Roth, R., Fuglestvedt, J. S., Peters, G. P., Enting, I. G., Von Bloh, W., Brovkin, V., Burke,
- 968 E. J., Eby, M., Edwards, N. R., Friedrich, T., Frölicher, T. L., Halloran, P. R., Holden, P. B.,
- 969 Jones, C., Kleinen, T., Mackenzie, F. T., Matsumoto, K., Meinshausen, M., Plattner, G. K.,
- 970 Reisinger, A., Segschneider, J., Shaffer, G., Steinacher, M., Strassmann, K., Tanaka, K.,
- 971 Timmermann, A. and Weaver, A. J.: Carbon dioxide and climate impulse response
- 972 functions for the computation of greenhouse gas metrics: A multi-model analysis,
- 973 Atmos. Chem. Phys., 13(5), 2793–2825, doi:10.5194/acp-13-2793-2013, 2013.
- 974 Kwiatkowski, L., Aumont, O., Bopp, L. and Ciais, P.: The Impact of Variable Phytoplankton
- 975 Stoichiometry on Projections of Primary Production, Food Quality, and Carbon Uptake
- 976 in the Global Ocean, Global Biogeochem. Cycles, 516–528,

- 977 doi:10.1002/2017GB005799, 2018.
- Kwiatkowski, L., Aumont, O. and Bopp, L.: Consistent trophic amplification of marine
 biomass declines under climate change, Glob. Chang. Biol., 25(1), 218–229,

980 doi:10.1111/gcb.14468, 2019.

281 Landolfi, A., Kähler, P., Koeve, W. and Oschlies, A.: Global marine N2 fixation estimates:

982 From observations to models, Front. Microbiol., 9(SEP), 1–8,

983 doi:10.3389/fmicb.2018.02112, 2018.

984 Lang, S. Q., Butterfield, D. A., Lilley, M. D., Paul Johnson, H. and Hedges, J. I.: Dissolved

985 organic carbon in ridge-axis and ridge-flank hydrothermal systems, Geochim.

```
986 Cosmochim. Acta, 70(15), 3830–3842, doi:10.1016/j.gca.2006.04.031, 2006.
```

287 Laufkötter, C., John, J. G., Stock, C. A. and Dunne, J. P.: Temperature and oxygen dependence

- 988 of the remineralization of organic matter, Global Biogeochem. Cycles, 31(7), 1038–
 989 1050, doi:10.1002/2017GB005643, 2017.
- Laws, E. A., Falkowski, P. G., Smith, W. O., Ducklow, H. W. and McCarthy, J. J.: Temperature
 effects on export production in the open ocean, Global Biogeochem. Cycles, 14(4),

992 1231–1246, doi:10.1029/1999GB001229, 2000.

- 993 Lee, S. Y., Chiang, J. C. H., Matsumoto, K. and Tokos, K. S.: Southern Ocean wind response to
- North Atlantic cooling and the rise in atmospheric CO 2 : Modeling perspective and
- 995 paleoceanographic implications, Paleoceanography, 26, 1–16,
- 996 doi:10.1029/2010PA002004, 2011.
- 997 Lenton, T. M., Williamson, M. S., Edwards, N. R., Marsh, R., Price, A. R., Ridgwell, A. J.,
- Shepherd, J. G. and Cox, S. J.: Millennial timescale carbon cycle and climate change in an
 efficient Earth system model, Clim. Dyn., 26(7–8), 687–711, doi:10.1007/s00382-0060109-9, 2006.
- 1001 Letscher, R. T. and Moore, J. K.: Preferential remineralization of dissolved organic
- 1002 phosphorus and non-Redfield DOM dynamics in the global ocean: Impacts on marine
- 1003 productivity, nitrogen fixation, and carbon export, Global Biogeochem. Cycles, 29(3),
- 1004 325-340, doi:10.1002/2014GB004904, 2015.
- Levitus, S.: Climatological atlas of the world ocean, US Department of Commerce, National
 Oceanic and Atmospheric Administration., 1982.
- 1007 Mahowald, N. M., Muhs, D. R., Levis, S., Rasch, P. J., Yoshioka, M., Zender, C. S. and Luo, C.:

- 1008 Change in atmospheric mineral aerosols in response to climate: Last glacial period,
- 1009 preindustrial, modern, and doubled carbon dioxide climates, J. Geophys. Res. Atmos.,

1010 111(D10), n/a-n/a, doi:10.1029/2005JD006653, 2006.

- 1011 Marchal, O., Stocker, T. F. and Joos, F.: A latitude-depth, circulation-biogeochemical ocean
- 1012 model for palaeoclimate studies. Development and sensitivities, Tellus, Ser. B Chem.
- 1013 Phys. Meteorol., 50B(3), 290–316, doi:10.1034/j.1600-0889.1998.t01-2-00006.x,
- 1014 1998.
- Martiny, A. C., Pham, C. T. A. A., Primeau, F. W., Vrugt, J. A., Moore, J. K., Levin, S. A. and
 Lomas, M. W.: Strong latitudinal patterns in the elemental ratios of marine plankton
 and organic matter, Nat. Geosci., 6(4), 279–283, doi:10.1038/ngeo1757, 2013.
- 1018 Mather, R. L., Reynolds, S. E., Wolff, G. A., Williams, R. G., Torres-Valdes, S., Woodward, E. M.
- 1019 S., Landolfi, A., Pan, X., Sanders, R. and Achterberg, E. P.: Phosphorus cycling in the
- 1020 North and South Atlantic Ocean subtropical gyres, Nat. Geosci., 1(7), 439–443,
- 1021 doi:10.1038/ngeo232, 2008.
- 1022 Matsumoto, K. and Key, R. M.: Natural radiocarbon distribution in the deep ocean, in Global
- 1023 Environmental Change in the Ocean and on Land, edited by M. Shiyomi, H. Kawahata,
- 1024 H. Koizumi, A. Tsuda, and Y. Awaya, pp. 45–58, Terrapub, Tokyo. [online] Available
- 1025 from: http://svr4.terrapub.co.jp/e-library/kawahata/pdf/045.pdf, 2004.
- 1026 Matsumoto, K. and McNeil, B.: Decoupled response of ocean acidification to variations in 1027 climate sensitivity, J. Clim., 26, 1764–1771, doi:10.1175/JCLI-D-12-00290.1, 2012.
- 1028 Matsumoto, K. and Tanioka, T.: Shifts in regional production as a driver of future global
- 1029 ocean production stoichiometry, Environ. Res. Lett., 15(124027), doi:10.1088/17481030 9326/abc4b0, 2020.
- Matsumoto, K. and Yokoyama, Y.: Atmospheric Δ 14 C reduction in simulations of Atlantic
 overturning circulation shutdown, Global Biogeochem. Cycles, 27(2), 296–304,
 doi:10.1002/gbc.20035, 2013.
- 1034 Matsumoto, K., Tokos, K. S., Price, A. R. and Cox, S. J.: First description of the Minnesota
- 1035 Earth System Model for Ocean biogeochemistry (MESMO 1.0), Geosci. Model Dev., 1(1),
- 1036 1–15, doi:10.5194/gmd-1-1-2008, 2008.
- 1037 Matsumoto, K., Tokos, K., Chikamoto, M. and Ridgwell, A.: Characterizing post-industrial
- 1038 changes in the ocean carbon cycle in an Earth system model, Tellus B Chem. Phys.

- 1039 Meteorol., 62(4), 296–313, doi:10.1111/j.1600-0889.2010.00461.x, 2010.
- 1040 Matsumoto, K., Tokos, K., Huston, A. and Joy-Warren, H.: MESMO 2: a mechanistic marine
- 1041 silica cycle and coupling to a simple terrestrial scheme, Geosci. Model Dev., 6(2), 477–
- 1042 494, doi:10.5194/gmd-6-477-2013, 2013.
- Matsumoto, K., Rickaby, R. and Tanioka, T.: Carbon Export Buffering and CO2 Drawdown by
 Flexible Phytoplankton C:N:P Under Glacial Conditions, Paleoceanogr.
- 1045 Paleoclimatology, 35(7), 1–22, doi:10.1029/2019PA003823, 2020.
- 1046 Moore, J. K., Doney, S. C. and Lindsay, K.: Upper ocean ecosystem dynamics and iron cycling
- 1047 in a global three-dimensional model, Global Biogeochem. Cycles, 18(4), 1–21,
- 1048 doi:10.1029/2004GB002220, 2004.
- 1049 Mopper, K., Zhou, X., Kieber, R. J., Kieber, D. J., Sikorski, R. J. and Jones, R. D.: Photochemical
- degradation of dissolved organic carbon and its impact on the oceanic carbon cycle,
 Nature, 353(6339), 60–62, doi:10.1038/353060a0, 1991.
- Najjar, R. G., Sarmiento, J. L. and Toggweiler, J. R.: Downward transport and fate of organic
 matter in the ocean: Simulations with a general circulation model, Global Biogeochem.
 Cycles, 6(1), 45–76, doi:10.1029/91GB02718, 1992.
- Pahlow, M. and Oschlies, A.: Chain model of phytoplankton P, N and light colimitation, Mar.
 Ecol. Prog. Ser., 376(2), 69–83, doi:10.3354/meps07748, 2009.
- Pahlow, M. and Oschlies, A.: Optimal allocation backs droop's cell-quota model, Mar. Ecol.
 Prog. Ser., 473(2010), 1–5, doi:10.3354/meps10181, 2013.
- Pahlow, M., Dietze, H. and Oschlies, A.: Optimality-based model of phytoplankton growth
 and diazotrophy, Mar. Ecol. Prog. Ser., 489, 1–16, doi:10.3354/meps10449, 2013.
- 1061 Pahlow, M., Chien, C. Te, Arteaga, L. A. and Oschlies, A.: Optimality-based non-Redfield
- 1062 plankton-ecosystem model (OPEM v1.1) in UVic-ESCM 2.9 Part 1: Implementation
- and model behaviour, Geosci. Model Dev., 13(10), 4663–4690, doi:10.5194/gmd-134663-2020, 2020.
- Paulsen, H., Ilyina, T., Six, K. D. and Stemmler, I.: Incorporating a prognostic representation
 of marine nitrogen fixers into the global ocean biogeochemical model HAMOCC, J. Adv.
- 1067Model. Earth Syst., 9(1), 438–464, doi:10.1002/2016MS000737, 2017.
- Ridgwell, A.: Glacial-interglacial perturbations in the global carbon cycle, Ph, D. thesis, 134
 pp., Univ. East Anglia, Norwich, U. K. Ridgwell, A. J., U. Edwards, 2001.

1070	Ridgwell, A., Hargreaves, J. C., Edwards, N. R., Annan, J. D., Lenton, T. M., Marsh, R., Yool, A.
1071	and Watson, A.: Marine geochemical data assimilation in an efficient Earth System
1072	Model of global biogeochemical cycling, Biogeosciences, 4(4), 87–104, doi:10.5194/bg-
1073	4-87-2007, 2007.
1074	Ridgwell, A. J., Watson, A. J. and Archer, D. E.: Modeling the response of the oceanic Si
1075	inventory to perturbation, and consequences for atmospheric CO ₂ , Global
1076	Biogeochem. Cycles, 16(4), 19-1-19–25, doi:10.1029/2002GB001877, 2002.
1077	Roshan, S. and DeVries, T.: Efficient dissolved organic carbon production and export in the
1078	oligotrophic ocean, Nat. Commun., 8(1), 2036, doi:10.1038/s41467-017-02227-3,
1079	2017.
1080	Sarmiento, J. L., Gruber, N., Brzezinski, M. A. and Dunne, J. P.: High-latitude controls of
1081	thermocline nutrients and low latitude biological productivity, Nature, 427(6969), 56–
1082	60, doi:10.1038/nature02127, 2004.
1083	Sun, X. and Matsumoto, K.: Effects of sea ice on atmospheric p CO 2 : A revised view and
1084	implications for glacial and future climates, J. Geophys. Res., 115(G02015),
1085	doi:10.1029/2009JG001023, 2010.
1086	Sunda, W. G. and Huntsman, S. A.: Iron uptake and growth limitation in oceanic and coastal
1087	phytoplankton, Mar. Chem., 50(1–4), 189–206, doi:10.1016/0304-4203(95)00035-P,
1088	1995.
1089	Sverdrup, H. U.: On the conditions for the vernal blooming of phytoplankton, J. Cons. Perm.
1090	Int. Pour l'Exploration La Mer, 18, 287–195, 1953.
1091	Takeda, S.: Influence of iron availability on nutrient consumption ratio, Nature, 393(JUNE),
1092	774–777, 1998.
1093	Tanioka, T. and Matsumoto, K.: Buffering of Ocean Export Production by Flexible Elemental
1094	Stoichiometry of Particulate Organic Matter, Global Biogeochem. Cycles, 31(10), 1528–
1095	1542, doi:10.1002/2017GB005670, 2017.
1096	Tanioka, T. and Matsumoto, K.: A meta-analysis on environmental drivers of marine
1097	phytoplankton C : N : P, Biogeosciences, 17(11), 2939–2954, doi:10.5194/bg-17-2939-
1098	2020, 2020a.
1099	Tanioka, T. and Matsumoto, K.: Stability of Marine Organic Matter Respiration
1100	Stoichiometry, Geophys. Res. Lett., 47(1), 1–10, doi:10.1029/2019GL085564, 2020b.

- Ushie, H. and Matsumoto, K.: The role of shelf nutrients on glacial-interglacial CO 2 : A
 negative feedback, Global Biogeochem. Cycles, 26, 1–10, doi:10.1029/2011GB004147,
 2012.
- Wang, W.-L., Moore, J. K., Martiny, A. C. and Primeau, F. W.: Convergent estimates of marine
 nitrogen fixation, Nature, 566(7743), 205–211, doi:10.1038/s41586-019-0911-2,
 2019.
- Wanninkhof, R.: Relationship between wind speed and gas exchange over the ocean, J.
 Geophys. Res., 97(C5), 7373–7382, doi:10.1029/92JC00188, 1992.

1109 Weaver, A. J., Sedlá, J., Eby, M., Alexander, K., Crespin, E., Fichefet, T., Philippon-berthier, G.,

1110 Joos, F., Kawamiya, M., Matsumoto, K., Steinacher, M., Tachiiri, K., Tokos, K., Yoshimori,

1111 M. and Zickfeld, K.: Stability of the Atlantic meridional overturning circulation : A

1112 model intercomparison, Geophys. Res. Lett., 39, 1–7, doi:10.1029/2012GL053763,

1113 2012.

Yamanaka, Y. and Tajika, E.: Role of dissolved organic matter in the marine biogeochemical
cycle: Studies using an ocean biogeochemical general circulation model, Global
Biogeochem. Cycles, 11(4), 599–612, doi:10.1029/97GB02301, 1997.

1117 Yamanaka, Y., Yoshie, N., Fujii, M., Aita, M. N. and Kishi, M. J.: An Ecosystem Model Coupled

1118 with Nitrogen-Silicon-Carbon Cycles Applied to Station A7 in the Northwestern Pacific,

1119 J. Oceanogr., 60(2), 227–241, doi:10.1023/B:JOCE.0000038329.91976.7d, 2004.

1120 Zickfeld, K., Eby, M., Weaver, A. J., Alexander, K., Crespin, E., Edwards, N. R., Eliseev, A. V.,

1121 Feulner, G., Fichefet, T., Forest, C. E., Friedlingstein, P., Goosse, H., Holden, P. B., Joos, F.,

1122 Kawamiya, M., Kicklighter, D., Kienert, H., Matsumoto, K., Mokhov, I. I., Monier, E.,

1123 Olsen, S. M., Pedersen, J. O. P., Perrette, M., Philippon-Berthier, G., Ridgwell, A.,

1124 Schlosser, A., Von Deimling, T. S., Shaffer, G., Sokolov, A., Spahni, R., Steinacher, M.,

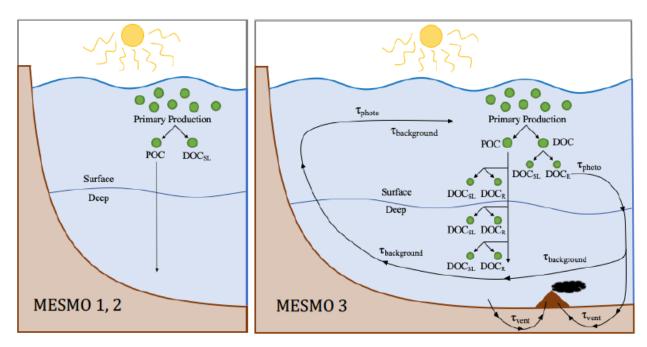
1125 Tachiiri, K., Tokos, K. S., Yoshimori, M., Zeng, N. and Zhao, F.: Long-Term climate

1126 change commitment and reversibility: An EMIC intercomparison, J. Clim., 26(16),

1127 5782–5809, doi:10.1175/JCLI-D-12-00584.1, 2013.

- 1128
- 1129







1132 **Figure 1.** Schematic diagram of DOM cycling in MESMO 2 versus MESMO 3. In the new

1133 model, DOM_r can be activated. DOM_r is produced from POM breakdown, which can occur in

1134 the production layer or throughout the water column in the "deep POC split." Possible

1135 DOM_r remineralization mechanisms are the slow background degradation that occurs

1136 everywhere, thermal degradation in hydrothermal vents, and photodegradation in the

- 1137 surface. See text for details.
- 1138
- 1139



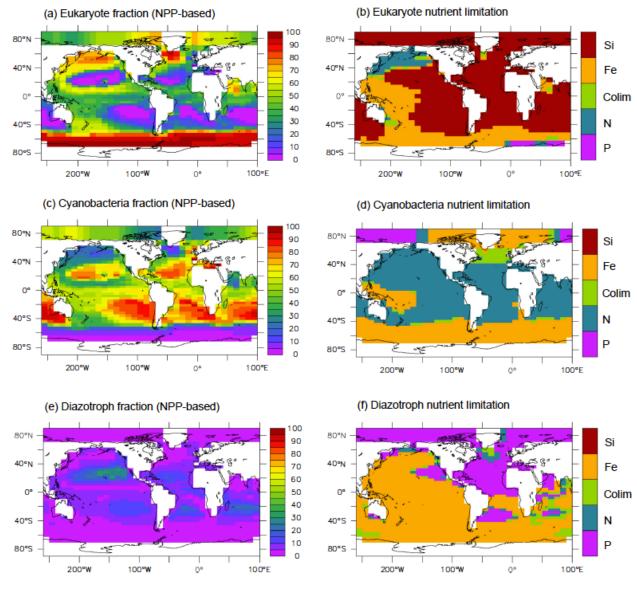
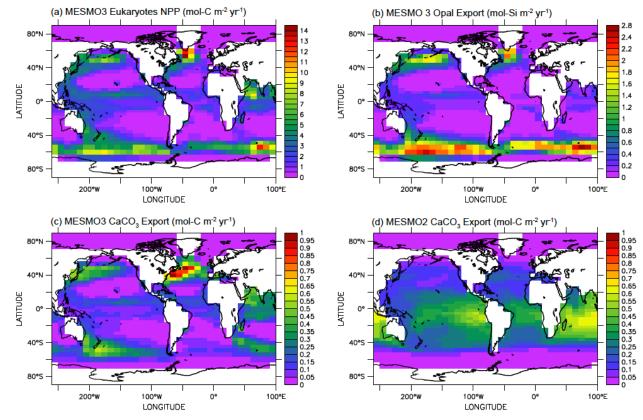


Figure 2. NPP-based surface phytoplankton functional type (PFT) abundance and nutrient
limitation in MESMO 3. Fractional abundance and nutrient limitation for eukaryotes (a, b),

- 1144 cyanobacteria (c, d), and diazotrophs (e, f).



1147

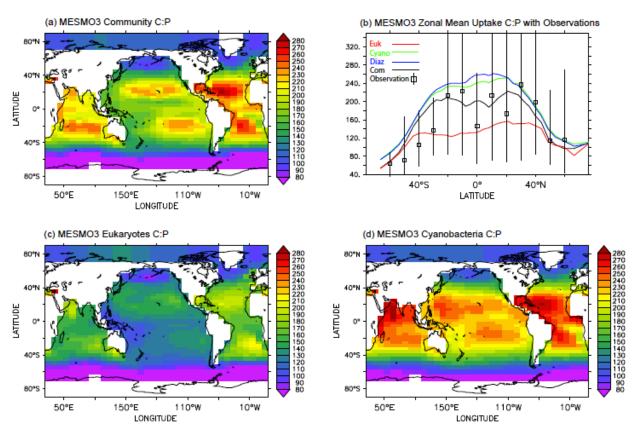
1148 **Figure 3.** Eukaryote production in MESMO 3 and CaCO₃ export in MESMO 2. In MESMO 3,

eukaryote NPP (a) is linked to both opal export (b) and $CaCO_3$ export (c) but the two export

1150 productions are differentiated by the residual nitrate potential growth (RNPG). Compare

1151 CaCO₃ export in MESMO 3 (c) to MESMO 2 (d). Unit = mol m^{-2} year⁻¹.





1154

Figure 4. Uptake C:P ratio in the top 100 m in MESMO 3: (a) phytoplankton community C:P,
(b) zonal mean C:P of all three PFTs and phytoplankton community, (c) eukaryote C:P, and

- 1157 (d) cyanobacteria C:P. The colors in (b) indicate: black = community C:P, red = eukaryote
- 1158 C:P, green = cyanobacteria C:P, and blue = diazotroph C:P. Also, (b) shows the range of
- 1159 observed C:P ratio binned by latitude (Martiny et al., 2013).
- 1160
- 1161

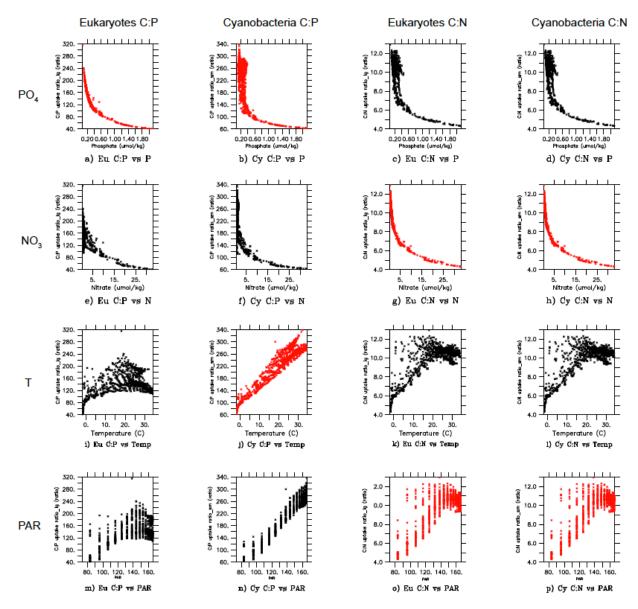




Figure 5. Scatter plots of surface ocean eukaryote and cyanobacteria C:P and C:N vs.

environmental drivers in MESMO 3. Columns: 1 = eukaryote C:P, 2 = cyanobacteria C:P, 3 =

eukaryote C:N, and 4 = cyanobacteria C:N. Rows: 1 = PO₄, 2 = NO₃, 3 = temperature, and 4 =

1166 PAR. Red indicates causal relationship according to the power law formulation of flexible

1167 C:N:P ratio. PAR = photosynthetically active radiation in W m⁻².

- 1168
- 1169

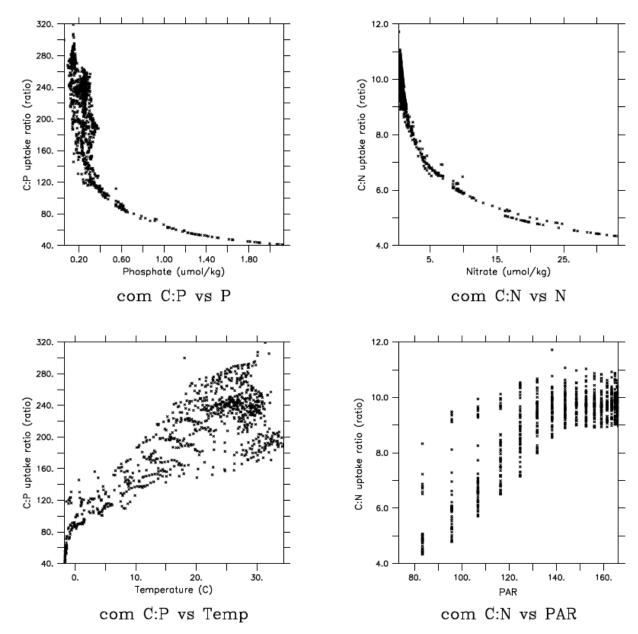
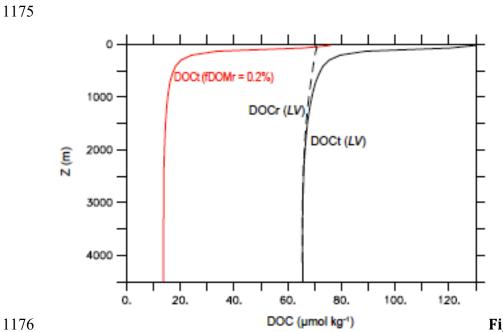




Figure 6. Scatter plots of surface ocean community C:P and C:N vs environmental drivers inMESMO 3.





1177 vertical profiles of DOC from the DOM_R-enabled MESMO 3. DOC_t (DOC_{sl}+DOC_r, black line)

and DOC_r (black dashed line) from the *LV* run (Experiment 210310m). Red line is DOC_t

1179 after reducing fDOM_r from 1% in *LV* to 0.2% (Experiment 2103100). Unit = μ mol kg⁻¹.

- 1180
- 1181

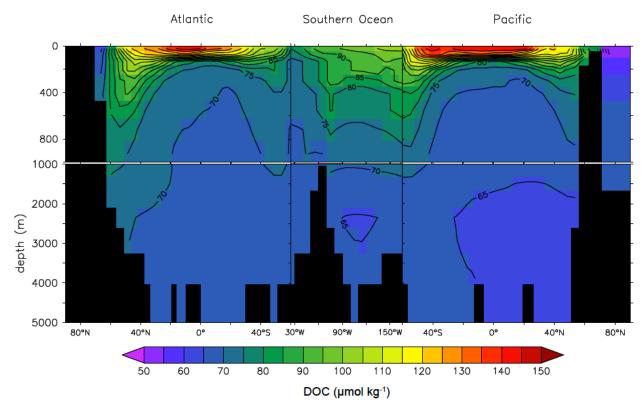
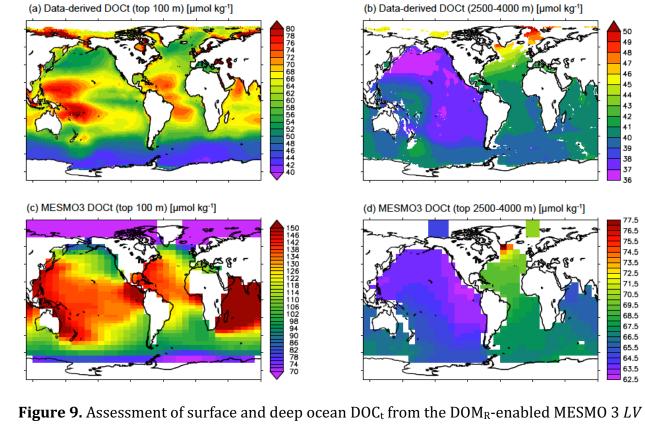


Figure 8. Global depth-latitude transect of DOC_t from the DOM_R -enabled MESMO 3 *LV* run.

1184 Transects are N-S along 25°W in the Atlantic, E-W along 60°S in the Southern Ocean, and N-

1185 S along 165°E in the Pacific. Unit = μ mol kg⁻¹.



- run. Data-derived DOCt distributions in the top 100 m (a) and 2500-4000 m (b). Model-
- 1191 simulated DOC_t distributions in the top 100 m (c) and 2500-4000 m (d). Date-derived DOC_t
- 1192 are from Roshan and DeVries (Roshan and DeVries, 2017). Unit = μ mol kg⁻¹.

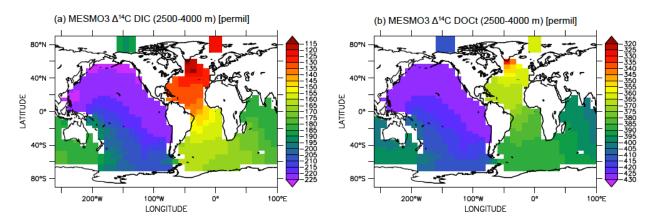


Figure 10. Δ^{14} C of deep ocean DIC (a) and DOC_t (b) from the DOM_R-enabled MESMO 3 *LV*

- run. Vertical average over 2500-4000 m water depth. Unit = ‰.