

***Interactive comment on* “Effects of heterogeneous reactions on global tropospheric chemistry” by Phuc T. M. Ha et al.**

Anonymous Referee #1

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Comments to Ha et al.

This paper evaluated the effects of heterogeneous uptake reactions of N_2O_5 , HO_2 and RO_2 on cloud and aerosol particles by using a chemical-climate model CHASER, and the modelling results have been verified by comparing with ground-based measurements, shipboard, aircraft and satellite observations. Although the findings of this study on the changes in global abundances of NO_2 , NO_3 , O_3 , and CO , and lifetime of CH_4 are basically within the range of uncertainties of previous studies, and no new surprising finding are reported, this work provides the most comprehensive view among this kind of studies covering the lower to upper troposphere, polluted terrestrial and remote oceanic region, and seasonal to annual characteristics. Particularly the study demonstrated the heterogeneous effect in the remote areas such as oceanic region

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and the upper troposphere for the first time. The present reviewer judges this paper is acceptable for publication after considering the following comments.

1. The difference between the role of uptake of HO₂ and RO₂ should be explained more in detail. In the case of the uptake of RO₂, the reduction of the formation of PAN and organic nitrates due to the reactions, CH₃COO₂ + NO₂ → PAN, and RO₂ + NO → RONO₂, as well as the reduction of NO oxidation reaction, RO₂ + NO → RO + NO₂, RO + O₂ → HO₂ etc. are expected. How the difference in the effect of HR(HO₂) and HR(RO₂) shown in Figs. 11 and 12 can be explained by these factors? 2. Other than the well-known heterogeneous processes of N₂O₅, HO₂ and RO₂ analyzed in this study, the heterogeneous renoxification process of HNO₃ to reproduce NO_x has previously been suggested in order to explain the model overestimate of HNO₃/NO_x ratio in the free and polluted atmosphere (Hauglustaine et al., *Geophys. Res. Lett.*, 23, 2609-2612, 1996; Lary et al., *J. Geophys. Res.*, 102, 3671-3682, 1997; Li et al., *SOLA*, 11, 124-128, 2015; Akimoto et al., *Atmos. Chem. Phys.*, 19, 603-615, 2019). Although the importance of this process has not been established, the same tendency of overestimate of HNO₃ and underestimate of NO_x has been revealed in this study (Table 5). Discussion should be given for the possibility of the heterogeneous reaction of HNO₃ whether in supporting or objecting. 3. Many of the figures are rather poorly presented for readers and should be revised. (1) In most of the figures, size of inside letters and axis labels are too small (unreadable on print and difficult to read even on PC screen). (2) Fig.3: How the site for each species were selected? There is no explanation in the text. (3) Figs. 3, 4, 5: The difference between the plots for noHR_n2o5, _ho2, _ro2 and _CLD are almost indiscernible. It is suggested to show only noHR and STD in these Figures, and the difference of noHR_n2o5, _ho2, _ro2 and _CLD should be presented in some selected plots in a different Figure. (4) Figs. 9, 11, 12: The differences between the upper and lower figures are not discernible easily. It is suggested to delete the figures for HRS(N₂O₅-aerosols), HRS(HO₂-Cloud) and HRS(RO₂-Cloud) in these Figures. It would be enough to explain in the text that the uptake of N₂O₅ on aerosols, and that of HO₂ and RO₂ on cloud are major processes.

Explanation should be given in the text why the process predominate for each of the species. (5) Figs. 10, 14: Labels and units of horizontal axis should be given properly. 4. Table 2: What is the meaning of asterisk for “product*”. What do the ISO2 and MACRO2 stand for? 5. Tables 5, 6, 7, 8: Units should be given appropriately.

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