

Radiation **Optical** model for the Baltic Sea with an explicit CDOM state variable: a case study with Model ERGOM (version 1.2)

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Abstract. Colored dissolved organic matter (CDOM) in marine environments impacts primary production due to its absorption effect on the photosynthetically active radiation. In coastal seas, CDOM originates from terrestrial sources predominantly and causes spatial and temporal changing patterns of light absorption which should be considered in marine biogeochemical models. We propose a model approach in which Earth Observation (EO) products are used to define boundary conditions of CDOM concentrations in an ecosystem model of the Baltic Sea. CDOM concentrations in riverine water derived from EO products serve as forcing for the ecosystem model. For this reason, we introduced an explicit CDOM state variable in the model.

We show that the light absorption by CDOM in the model can be improved considerably compared to traditional approaches where, e.g., CDOM is estimated from salinity. **The model performance increases especially with respect to spatial CDOM patterns due to the consideration of single river properties.** A prerequisite is high quality CDOM data with sufficiently high spatial resolution which can be provided by the new generation of ESA satellite sensor systems (Sentinel 2 MSI and Sentinel 3 OLCI). **Such data are essential, especially when local differences in riverine CDOM concentrations exist.**

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1 Introduction

Colored dissolved organic matter (CDOM) is a major light absorption constituent in the marine environment and especially in coastal seas. The spectral absorption characteristic of CDOM follows an exponential function with highest absorption towards shorter wavelengths (e.g. Nelson and Siegel, 2002). By modifying the underwater light climate, CDOM has an impact on primary productivity, e.g. in clear water sufficient light intensity to enable phytoplankton growth is available down to greater depths than in turbid waters (Dutkiewicz et al., 2015). Water temperature is affected by CDOM absorption as well. In turbid

20 water, the short wave light **divergence absorption** is located in the upper water column increasing the temperature while in clear water a thicker layer is warmed but to a lesser degree (Jolliff and Smith, 2014). This process impacts especially the sea surface temperature (SST). **Model studies show an SST increase up to 2 K in coastal regions when colored organic materials are considered** (Gnanadesikan et al., 2019).

Jerlov (1976) developed a classification for different water masses based on specific optical properties. This classification is 25 widely used in numerical ocean models (e.g., Griffies, 2004). For global models, this parametrization works **reasonable** **reasonably well**. However, coastal ecosystems with substantial terrestrial runoff require more detailed parametrization of **models**. **light penetration. Especially variable light attenuation in river plumes and their environs affect the hydrodynamic and ecological response** (Cahill et al., 2008).

Marine CDOM comprises humid substances (**gelbstoff** **yellow substances**) of terrestrial origin, and **autochthonous** **autochthonously** 30 produced CDOM. Its degradation is governed by photochemical bleaching and bacterial activity. In freshwater dominated systems, like the Baltic Sea, terrestrial CDOM dominates (e.g., Stedmon et al., 2010, and references herein). **Salinity and CDOM absorption** **In such systems, salinity and light absorption due to CDOM** show a robust relationship (Neumann et al., 2015) **indicating a conservative behavior. However, the .** **The relationship is hyperbolic (instead of linear) due to indicating the** degradation processes.

For coupled physical-biogeochemical models of freshwater influenced coastal seas, a spatially resolved CDOM concentra- 35 tion is important for realistic light climate estimates. Based on the nearly conservative character of CDOM, statistical models have been developed which estimate absorption from salinity and e.g. chlorophyll (Kowalczuk et al., 2006; Neumann et al., 2015). These models usually deliver reasonable results. However, two distinct disadvantages are prominent: (i) uncertainties in model salinity propagate into the biogeochemical model and (ii) variability in CDOM riverine load (Skoog et al., 2011; Asmala et al., 2013) cannot be resolved. These disadvantages can be eliminated by introducing an independent CDOM state variable 40 into the biogeochemical model (**Dutkiewicz et al., 2015**). A necessary prerequisite are boundary data for riverine CDOM loads of sufficiently good quality **which is not commonly available for most rivers** (Pefanis et al., 2020).

In this study, we present the **generation of CDOM boundary data with the aid of satellite imagery, the** implementation of a CDOM state variable in the biogeochemical model ERGOM (Ecological ReGional Ocean Model, Leibniz Institute for Baltic Sea Research (2015)), 45 **and we discuss the the generation of CDOM boundary data with the aid of satellite imagery, and we discuss the** effect of **the new** development. We use the Baltic Sea as a study area.

the proposed model extension on the Baltic Sea ecosystem.

2 **CDOM data from Earth Observation products**

The aim of the development is to improve modeled CDOM data compared to available statistical models (Sect. 1). **In situ** data of CDOM loads are not available in sufficient spatial and temporal resolution, i.e. in the Baltic Sea, CDOM is not a parameter of the HELCOM (www.helcom.fi, last access: 22 September 2020) monitoring program. New instruments 50 and technologies in Earth Observation (EO), now in operation, are ideal tools to overcome these limitations.

2 **Methods and data**

2.1 **Characteristics of satellite data**

55 In order to estimate the amount of CDOM coming from a river, it is necessary to derive it from observations within the river or as close to the discharge point as possible. The rivers in the Baltic Sea are usually small and thus high resolution (HR) instruments such as Sentinel-2 Multi Spectral Instrument (S2-MSI) are required. The MSI has a spatial resolution of 10-60 depending on the central wavelength of the band. Water quality products are usually generated in 60 resolution in order to reduce noise. This is sufficient for estimating the CDOM absorption of most rivers.

Two Sentinel-2 satellites are currently in orbit: S2A was launched on 23 June 2015

2.1 Model development and description

60 We start presenting the optical model including an explicit CDOM state variable, its implementations as part of a biogeochemical model, and S2B on 7 March 2017. Together they provide a global revisit time of 5 days next to equator. Due to the high latitude of the Baltic Sea, the revisit time amounts to 2–3 days in this area.

Despite the frequent cloud cover, sufficient observations can be gathered to monitor river CDOM throughout the open water season (typically from March-April to October in the northern Baltic Sea).

2.2 Earth Observation processor for CDOM absorption estimation

65 EO processors are designed to convert the radiance signal acquired by the satellites into values of geophysical parameters. The estimation is based on the scattering and absorption features of the material suspended or dissolved in water. In addition to CDOM, these materials include phytoplankton cells, represented by Chlorophyll a (Chl-a), and suspended particulate matter.

One commonly used water quality processor is Case 2 Regional Coast Color (C2RCC) (Brockmann et al., 2016). It utilizes one artificial neural network (ANN) to first remove the effects of the atmosphere from the signal (atmospheric correction) and another to estimate inherent optical properties (IOPs) of water from the water leaving reflectance.

70 For estimation of CDOM absorption (a_{CDOM}), we utilized the C2RCC (version 1) output called a_{dg} (combined absorption by detritus and gelbstoff) which was calibrated to $a_{CDOM}(440)$ (absorption coefficient by CDOM at 440) values with in situ measurements from Finland with this equation:

$$a_{CDOM}(440) = 0.0654 \cdot a_{dg}^{1.45} + 0.2$$

A similar local calibration method has provided good results with other water quality parameters in earlier studies such as Koponen et al. (2007); Attila et al. (2013).

75 The data processing and extraction are done in a Calvalus massive parallel processing system (<http://www.brockmann-consult.de/calvalus>, last access: 22 September 2020). Data extraction areas are manually defined in the vicinity of the mouths of 69 rivers that represent ERGOM input locations (Fig. 1). Location of model rivers (black squares). The green line is the coastline of the 3model. We refer to the labeled river later in the text. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008). The areas are designed so that islands, mixed pixels and shallow areas are excluded. All valid pixels (not masked as land or cloud by the pixel classification processor Idepix) within each area and image are collected and analyzed, and the 75th percentile value is chosen to represent the river a_{CDOM} . Cases in which the number of valid pixels is less than 50% of all available pixels from an area are removed from the analysis. Assumedly, these represent cases with partial cloud cover and they are discarded to keep only estimates with highest quality and low uncertainty. The arithmetic means of the 75th percentile pixel values of all valid days within each calendar month during years 2017–2019 are then computed for each extraction area.

85 We are aiming at providing the ecosystem model ERGOM with an annual cycle of CDOM loads based on monthly data. Since optical EO methods cannot provide a_{CDOM} estimates in darkness and throughout times with ice coverage, the values for the winter months have been interpolated. As a result, the dataset contains a_{CDOM} value for each month for each of the areas under investigation (69 extraction areas in total). Figure 2 shows four examples of the annual CDOM absorption cycle. The behavior of the data follows well the annual cycle: spring values are high due to the terrestrial matter brought into the coastal water by melting snow, summer values are low due to lower influx and the fall values are higher due to increasing rainfall. The aggregated monthly values of a_{CDOM} based on EO data (S2 & C2RCC V1) for four ERGOM input locations in the western coast of Finland. Location of the rivers is shown in Fig. 1.

3 Radiation model development

In this section, we describe the basics of the model development and the implementation.

we briefly introduce the circulation and biogeochemical model used for this study.

2.0.1 Fundamentals of the optical model

Starting point of the development is the radiation model proposed optical model as in the study by Neumann et al. (2015). The photosynthetic active radiation (PAR) follows an exponential decay with depth z :

$$95 \quad PAR(z) = PAR(0) \cdot \exp(-K_{PAR} \cdot z), \quad (1)$$

K_{PAR} is the underwater bulk light attenuation and is described by 5 components:

$$K_{PAR} = k_w + k_c \cdot Chl + k_{det} \cdot DET + k_{don} \cdot DON + K_{CDOM}(S), \quad (2)$$

k_c , k_{det} , and k_{don} are material specific constants and Chl , DET , and DON are concentrations of chlorophyll, detritus, and dissolved organic nitrogen, respectively. These concentrations are state variables of the ecosystem model ERGOM or, in the case of chlorophyll, can be estimated from model phytoplankton (Sect. 2.0.1). k_w is the attenuation coefficient for pure water. In Neumann et al. (2015), K_{CDOM} $K_{CDOM}(S)$ is a statistical relationship between *in situ* salinity and CDOM attenuation absorption for the Baltic Sea derived from observations. (Eq. A. For the new approach, we use the additional state variable CDOM:

$$K_{CDOM} = k_{cdom} \cdot CDOM \quad (3)$$

The PAR attenuation now reads:

$$105 \quad K_{PAR} = k_w + k_c \cdot Chl + k_{det} \cdot DET + k_{don} \cdot DON + k_{cdom} \cdot CDOM \quad (4)$$

Terrestrial CDOM behaves relatively refractory nearly conservatively in the ocean. An indication is the linear salinity–CDOM relationship in the northern Baltic (Harvey et al., 2015; Neumann et al., 2020). This is due to the fact of high freshwater supply with high CDOM concentrations. However, this relation does not apply to the central Baltic. In this region with longer residence time, the effects of CDOM degradation processes become more pronounced and observable (Skoog et al., 2011).

110 Two processes control CDOM degradation, photobleaching and biological degradation. Moran et al. (2000) study the degradation of terrestrial CDOM in the coastal ocean and find that photobleaching accounts for 80% of the degradation. Furthermore, they show that CDOM decay follows closely simple first-order kinetics. In accordance with these findings, we implement CDOM as

$$\frac{dCDOM}{dt} = -dr \cdot CDOM \quad (5)$$

115 with the degradation rate

$$dr = DRdr_0 \cdot IPAR(z). \quad (6)$$

$I(z) PAR(z)$ is the ambient PAR at depth z , $DR_0 dr_0$ is a constant, and $I(z) PAR(z)$ can be estimated as:

$$IPAR(z) = \frac{I_0}{2} r \cdot I_0 \cdot \exp(-K_{PAR} - \int_z^0 dz dz' K_{PAR}(z')) \quad (7)$$

I_0 is the solar radiation at sea surface depending on latitude, time, and sun zenith angle. We use 50% of total solar radiation for PAR (e.g., Stigebrandt and Wulff, 1987) since the invisible, long-wave part is absorbed at the water surface., which is a function of latitude and time. Factor r is the fraction of I_0 available as PAR (spectral range 400 to 700 nm). The integral consider the depth dependence of model concentrations of e.g. chlorophyll in Eq. 4.

2.1 Implementation in the biogeochemical model ERGOM

2.0.1 Implementation of the optical model

A comprehensive overview about CDOM in the ocean is given in Nelson and Siegel (2002). CDOM content usually is given as the concentration is usually given by a proxy, the light absorption for a specific wavelength, e.g. $a_{CDOM}(440)$ for 440 nm. The spectral distribution can be parameterized by an exponentially decline with wavelength.

$$a_{CDOM}(\lambda) = a_{CDOM}(\lambda_0) \cdot \exp(-s(\lambda - \lambda_0)) \quad (8)$$

s is the exponential slope parameter and varies between $0.015 - 0.025 \text{ nm}^{-1}$. For a given slope s and an absorption $a_{CDOM}(\lambda_0)$, any $a_{CDOM}(\lambda)$ can be estimated for wavelengths longer than 320 nm. We use a reference wavelength of 440 nm.

In the biogeochemical model The biogeochemical framework for implementing the CDOM state variable is the model ERGOM. In this model, state variables are given as concentrations of an element, e.g. mol carbon per m^3 , because these models primarily describe cycles of elements (carbon, phosphorus, nitrogen etc.). In order to model CDOM, a relationship between CDOM absorption and the concentration is required. Following the Lambert–Beer law, a linear relation exists. Neumann et al. (2020) derived a relationship based on optical measurements and measurements with a calibrated CDOM sensor, which we used to convert CDOM absorption into concentration and vice versa. We have to note that the accuracy of the conversion does not impact the performance of the radiation model. Both the CDOM freshwater loads optical model because both the satellite derived CDOM in freshwater and CDOM in the radiation optical model is given as $a_{CDOM}(440)$. CDOM loads are estimated from concentration times runoff which is available from Gustafsson et al. (2012).

Most model constants used in the model are provided by Neumann et al. (2015). An exception is DR_0 (We use 50% of total solar radiation for PAR (e.g., Stigebrandt and Wulff, 1987) since the invisible, long-wave part is absorbed at the water surface (factor r , Eq. 6) which. dr_0 (Eq. 6) has been estimated with a series of calibration simulations. Aim of the calibration has been was to find an optimal match of observed and simulated absorption values $a_{CDOM}(440)$.

Used constants are listed in Tab. 1. Model state variables in Eq. 4 have to be converted into appropriate units before entering the radiation optical model. We use a volume based concentration. $CDOM$ should be given as absorption because of uncertainties in the absorption–concentration relationship.

Table 1. Constants of the [radiation optical](#) model

| Const. | Value | Unit |
|-------------|---------|-----------------------------------|
| k_w | 0.027 | m^{-1} |
| k_c | 0.029 | $\text{m}^2 (\text{mg Chl})^{-1}$ |
| k_{det} | 0.0039 | $\text{m}^2 (\text{mg N})^{-1}$ |
| k_{don} | 0.0009 | $\text{m}^2 (\text{mg N})^{-1}$ |
| k_{cdom} | 0.221 | 1 |
| $DR_0 dr_0$ | 8.75e-5 | day^{-1} |
| r | 0.5 | 1 |

The technical implementation is done by an automatic code generation. [Basis is Fundamentals are](#) a set of text files describing the biogeochemistry independently of computer language and the host system. Code templates describe physical and numerical aspects, and are specific for a certain host e.g. a circulation model. All necessary ingredients, the code generation tool, text files, and templates for several systems, can be downloaded from www.ergom.net (last access: 22 September 2020). The same technique is used e.g. in Radtke et al. (2019).

2.1 Model description

2.0.1 Circulation and biogeochemical model

For model testing, we have used a similar [setup model system](#) as in Neumann et al. (2015). The circulation model is MOM5.1 (Griffies, 2004) adapted for the Baltic Sea. The horizontal resolution is three nautical miles. Vertically, the model is resolved into 152 layers with a layer thickness of 0.5 m at the surface and gradually increasing with depth up to 2 m. The circulation model is coupled with a sea ice model (Winton, 2000) accounting for ice formation and drift.

Coupled with the circulation model is the biogeochemical model ERGOM. It describes a marine nitrogen and phosphorus cycle. Primary production, forced by PAR, is provided by three functional phytoplankton groups (large cells, small cells, and cyanobacteria). Chlorophyll concentration can be estimated from the phytoplankton groups which is used in the [radiation optical](#) model. Dead particles accumulate in the detritus state variable which is another compartment in the [radiation optical](#) model. A bulk zooplankton grazes on phytoplankton and constitutes the uppermost trophic level in the model. The metabolism of phytoplankton and zooplankton produces DON which has only little impact on light absorption. Phytoplankton and detritus can sink down in the water column and accumulate in a sediment layer. In the water column and the sediment, detritus is mineralized into dissolved inorganic nitrogen and phosphorus. Mineralization is controlled by temperature and oxygen. Oxygen is produced by primary production and consumed due to all other processes e.g. metabolism and mineralization. Coupled to the nitrogen and phosphorus cycle is a carbon cycle as described in Kuznetsov and Neumann (2013). [A schematic of the model](#)

structure is provided in Appendix B. The estimated short wave absorption (Sect. 2.0.1) feeds back into the physical part of the model and hence impacts the temperature distribution.

170 The new CDOM variable in the current model development state is not involved in the biogeochemical processes. This is justified by the fact that CDOM is relatively refractory and has a long residence time, and autochthonous CDOM produced by e.g. phytoplankton, is a small fraction. In later developments, it will be included in the carbon cycle. For this purpose, it is essential to realize CDOM as a carbon based concentration. If this will be not the case, the CDOM state variable could be implemented as an absorption and thus conversions between absorption and concentration could be prevented.

175 The model has been forced by meteorological data from the coastDat-2 data set (Geyer and Rockel, 2013). We run the model from 1948—2019. A first run was used to spin up the new CDOM tracer. In a second run, CDOM was initialized with data from the first run. In addition to the 3 nautical miles resolution, we use a 1 nautical mile resolution for the period 2017–2019. The model has been successfully used in several applications (e.g., Neumann, 2010; Neumann et al., 2015).

2.1 CDOM boundary data from Earth Observation products

180 Aim of the development is to improve the simulated light climate by a more realistic representation of CDOM concentration compared to available statistical models (Sect. 1). However, *in situ* data of CDOM loads are not available in sufficient spatial and temporal resolution, i.e. in the Baltic Sea, CDOM is not a parameter of the HELCOM (www.helcom.fi, last access: 22 September 2020) monitoring program. New instruments and technologies in Earth Observation (EO), now in operation, are ideal tools to overcome these limitations.

185 2.1.1 Characteristics of satellite data

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Earth Observation (EO) processors are a set of algorithms designed to convert the radiance signal acquired by the satellites into values of geophysical parameters. The estimation is based on the scattering and absorption features of the

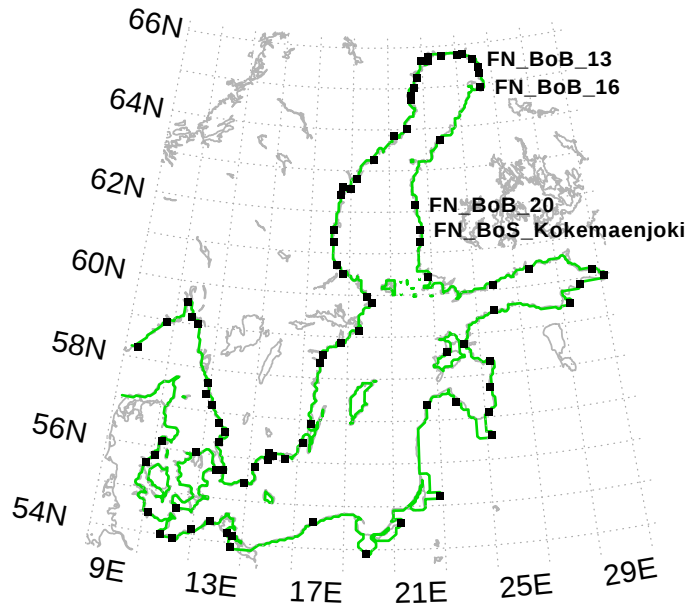


Figure 1. Location of model rivers (black squares). The green line is the coastline of the 3 nm model. We refer to the labeled river later in the text. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008)

material suspended or dissolved in water. In addition to CDOM, these materials include phytoplankton cells, represented by Chlorophyll a (Chl-a), and suspended particulate matter.

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 205 440 nm) using this equation:

$$a_{CDOM}(440) = 0.654 \cdot a_{dg}^{1.45} + 0.2 \quad (9)$$

The equation is based on in situ sampling made with a flow-through device (ac-9) (Lindfors et al., 2005; Koponen et al., 2007) during two coastal estuary measurement campaigns. These data are not yet published but a similar local calibration method has provided good results with other water quality parameters in earlier studies such as Attila et al. (2013).

210 The data processing and extraction are done in a Calvalus massive parallel processing system (<http://www.brockmann-consult.de/calvalus>, last access: 22 September 2020). Data extraction areas are manually defined in the vicinity of the mouths of 69 rivers that represent ERGOM input locations (Fig. 1). The areas are designed so that islands, mixed pixels and shallow areas are excluded. All valid pixels (not masked as land or cloud by the pixel classification processor

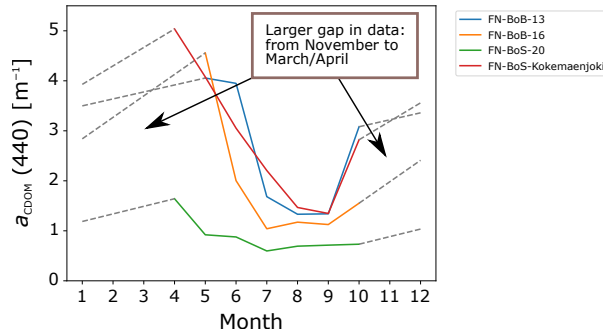


Figure 2. The aggregated monthly values of a_{CDOM} based on EO data (S2 & C2RCC V1) for four ERGOM input locations in the western coast of Finland. Location of the rivers is shown in Fig. 1.

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 225 higher due to increasing rainfall.

3 Results and discussion

In this section, we show the improved model CDOM representation and its impact on the simulation results. Owing to the changed shortwave distribution in the water column, an effect especially on the biogeochemistry is expected. All CDOM data presented are converted into absorption at 440 nm ($a_{CDOM}(440)$). Especially for observations at different wavelengths, we
 230 use Eq. 8 with a slope s of 0.018 nm^{-1} (Kratzer and Moore, 2018). For comparison, we show $a_{CDOM}(440)$ values estimated from the CDOM state variable and from model salinity. The models differ only in the estimation of PAR which becomes evident when we show the impact on biogeochemistry.

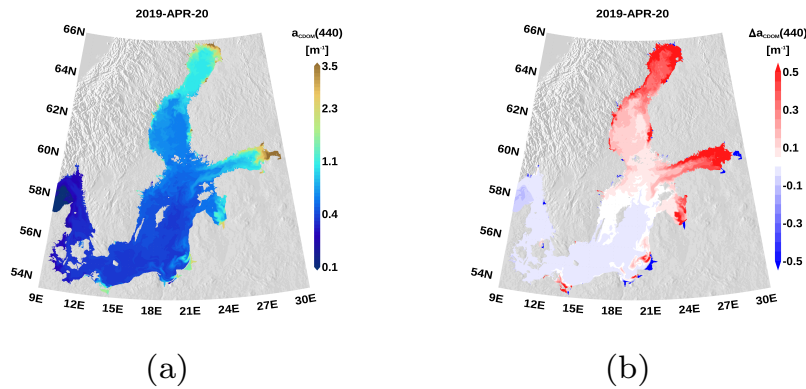


Figure 3. Snapshot of simulated surface $a_{CDOM}(440)$ at April 20th 2019 (a) and the difference to the salinity based absorption estimate (b) as seen in the 1nm resolution model.

3.1 CDOM absorption

In Fig. 3a, we show the simulated CDOM absorption at the sea surface. The snapshot clearly illustrates the spatial patterns. Strong absorption is visible in the northern Baltic and the river mouths. The difference to the salinity based estimate (Eq. 2) is depicted in the right panel. Strongest differences appear in the Gulf of Bothnia and the Gulf of Finland while in the central Baltic differences are small. Strong differences are also pronounced in river estuaries. Owing to the low salinity, the salt-CDOM relationship overestimates CDOM content in these areas. Furthermore, the new, EO based method considers individual CDOM concentrations of different rivers. Rivers of the northern catchment area carry higher CDOM loads compared to rivers of the south-eastern catchment area due to a high fraction of peat land. The range of a_{CDOM} values is demonstrated in Fig. 4.

Both datasets were compared against *in situ* data collected from monitoring stations in coastal waters of Finland and from the Northern coast of Sweden. As shown in Fig. 5, the improvement becomes obvious. With the salinity method, the correlation is low ($R^2 = 0.15$) and there are some clear overestimates (difference between $a_{CDOM}(440)$ from salinity and the one-to-one line more than 2 m^{-1}) while most data points are underestimated. With the EO CDOM method, the correlation improves significantly ($R^2 = 0.61$). There are no large overestimates and the data points move closer to the one-to-one line. Large *in situ* values ($a_{CDOM}(440) > 3 \text{ m}^{-1}$) are still underestimated with the new EO method. This underestimation is most likely caused by the following two major inaccuracies in the present version of the model input:

- The coast has many small rivers. Not all of them are yet included in this version.
- In river estuaries with low bottom depth or complex morphological structure, the shapes and formulations of the extraction areas do not sufficiently capture the incoming CDOM loading from the river. In order to avoid EO observations contaminated by bottom reflectance it was necessary to use pixels that are sufficiently far from the shoreline. Therefore, pixels of the extraction area may not represent river water as it has been already mixed with sea water. In some cases, this leads to lower concentrations especially during the low runoff season.

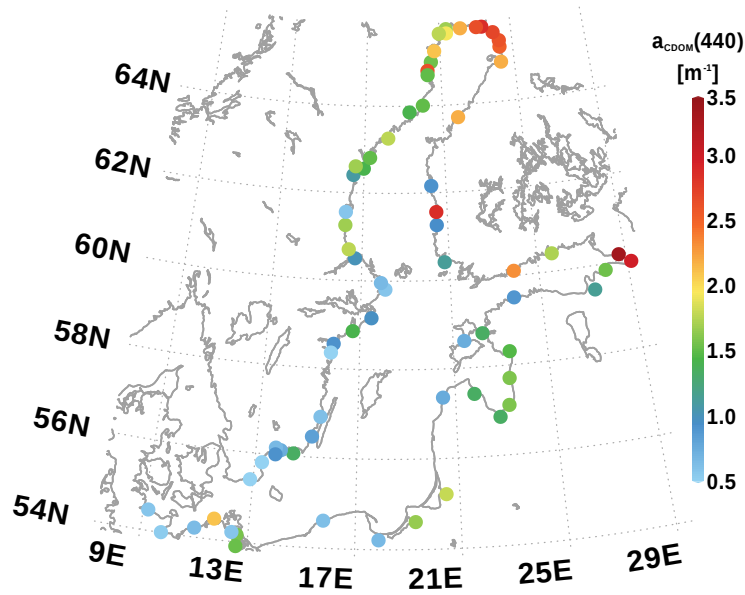


Figure 4. Mean a_{CDOM} of the individual rivers used in the model. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008).

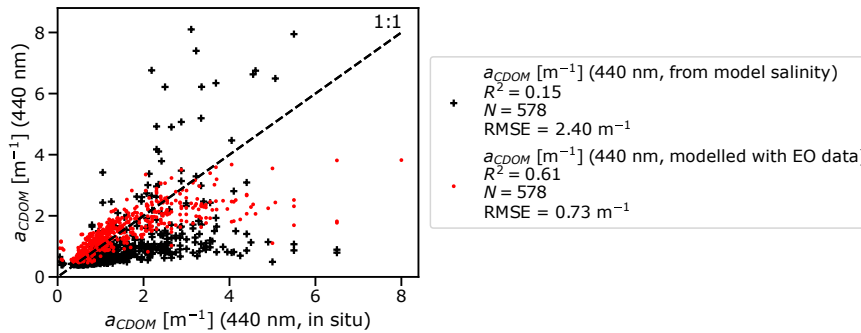


Figure 5. *In situ* a_{CDOM} (x-axis) vs. ERGOM a_{CDOM} (y-axis) estimated from salinity (in black) and ERGOM a_{CDOM} with the EO method (in red) and the location of stations (triangles in (b)).

Surface $a_{CDOM}(440)$ time series at 6 stations. Location of the stations is shown in (d). Absorption estimates based on simulated CDOM are shown in black, based on a conversion from simulated salinity in green, and red diamonds are observations. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008). Figure 6 demonstrates the different CDOM absorption estimates. Shown are time series of surface CDOM absorption at 6 stations (Fig. 6d). The green curve is the salinity based estimate and the black curve the estimate from model CDOM concentration. Red diamonds are *in situ* observations.

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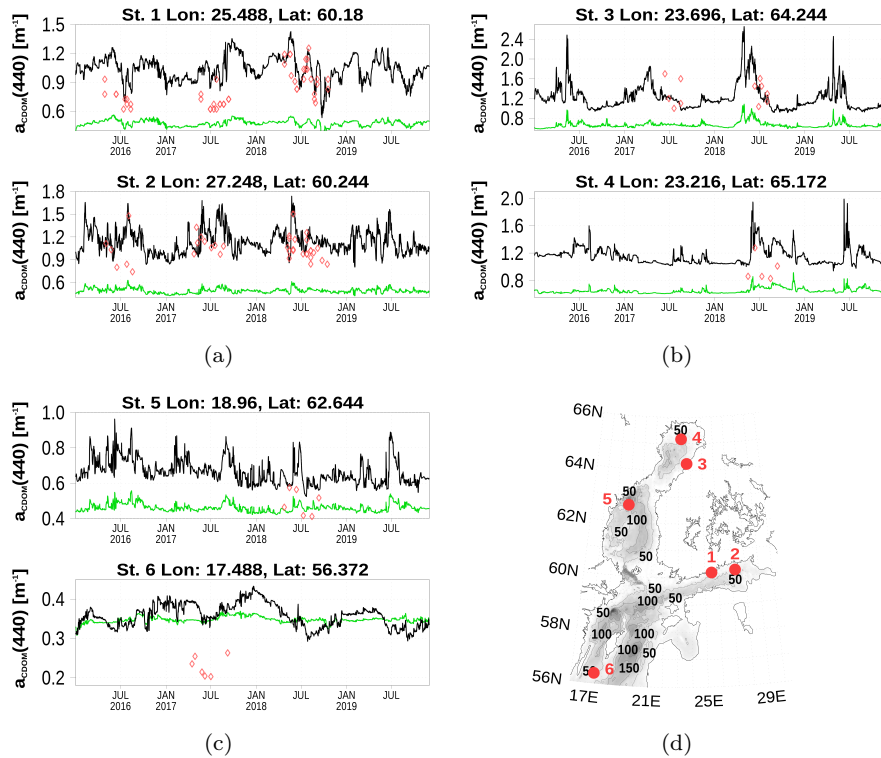


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At stations 1–4, absorption values from simulated CDOM are much closer to observations compared with salt based estimates. The absorption difference at stations 5 and 6 is less pronounced which is also evident from Fig. 3. The seasonal variability is stronger for absorption derived from the model CDOM variable (compared to salt based absorption) and reflects the observed variability. The reason is the annual cycle of riverine CDOM concentration in addition to the runoff cycle. In the central Baltic Sea, both methods overestimate CDOM absorption.

3.2 Impact on biogeochemistry

As an example of the changed light absorption impact on the biogeochemistry, we show annual mean profiles of selected variables in Fig. 7. We have chosen station 4 from Fig. 3 since at this station the CDOM absorption is considerably increased due to the new, EO based, radiation optical model and it is located in the center of the Bothnian Bay. Owing to the increased CDOM absorption, PAR is reduced in the EO model approach (Fig. 7a) as expected. Consequently, primary production (PP) is reduced (Fig. 7b). However, in the uppermost layer, PP is increased. As a result of reduced PP, phytoplankton concentration

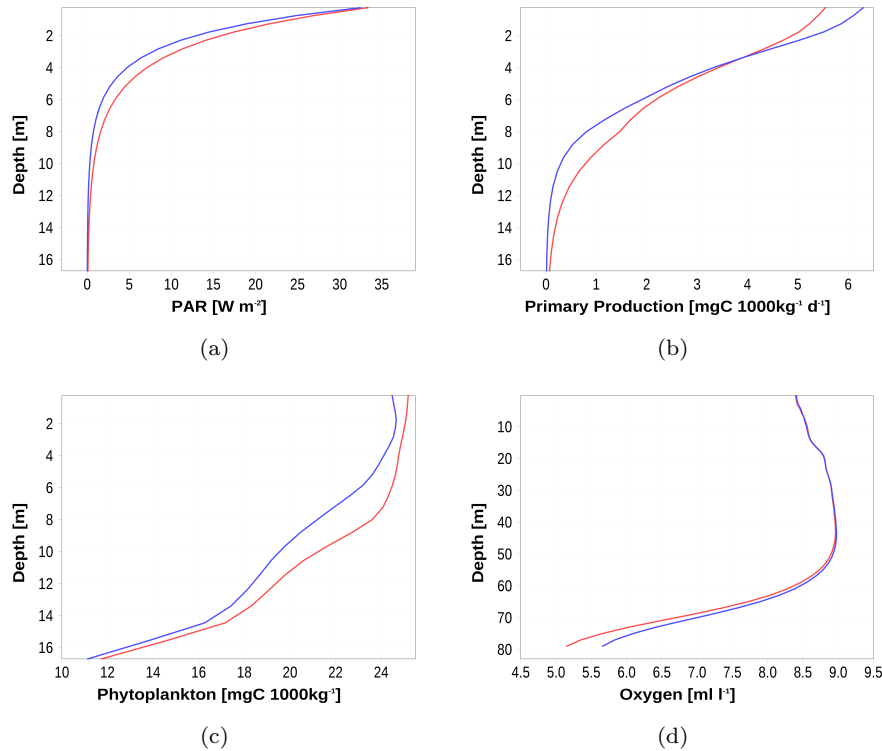


Figure 7. 2018 annual mean profiles at station 4 (Fig. 6d). Blue curve shows data from the model with an explicit CDOM state variable and the red curve is from the model with CDOM absorption estimates based on [salinity relation to salinity](#). For oxygen (d), we show the whole water column.

270 shows lower values (Fig. 7c). An integrated response is the increased bottom oxygen concentration (Fig. 7d). Less net PP results in less accumulation of organic matter in the deep water of the basin and subsequently reduced oxygen consumption. The impact on water temperature is small (order of 0.01 K, not shown). The effect is a temperature increase in the surface layer and a lower temperature below.

We demonstrate changes in the biogeochemistry with a climatology of surface nutrient concentrations at three stations in
 275 Fig. 8. For this analysis, we use data from the 3 nautical miles model version because of the longer simulation period. Shown are data from the EO model (black) and the previous (salinity based) model (green) together with observations (red). In the Gulf of Finland at station KAS-11, the spring bloom related nutrient depletion is delayed by 2 weeks (Fig. 8a and b). Sufficient PAR intensity, initiating a bloom, is available later in the season. The winter nutrient concentrations are elevated compared to the salinity model version. At the Bothnian Bay station (Fig. 8c and d), the spring bloom delay is less pronounced. In this area,
 280 the longer sea ice coverage dominates the PAR in spring. At station BY15 in the Baltic Proper (Fig. 8e and f), the difference between both model versions is small due to similar CDOM absorption (Fig. 6).

4 Summary and conclusions

In this study, we propose an approach for **considering** light absorption due to terrestrial CDOM in a marine ecosystem model for the Baltic Sea. An explicit consideration is necessary if large amounts of terrestrial CDOM enter the marine system and strong coastal-sea gradients develop. In such cases, a uniform light absorption due to CDOM cannot account for the *in situ* light climate in a sufficient way. A common **An often applied** approach uses CDOM-salinity relationships for CDOM **absorption** estimates (Kowalczyk et al., 2006, 2010; Neumann et al., 2015) but with distinct disadvantages (see Sect. 1).

Our approach uses an explicit CDOM state variable as part of the biogeochemical model. In order to improve the simulated absorption compared to the salt approximation, a high quality data set of riverine CDOM loads is necessary. This has been accomplished by using earth observation data from **Sentinel 2 Sentinel-2** MSI. The high spatial resolution (10 m–60 m) allows to observe the river mouths directly. A difficulty in the regions of higher latitude like the Baltic Sea area is the insolation, the occurrence of sea ice, and the frequent cloud cover in winter. Continuous observations are not possible during this time. We have used a linear interpolation to bridge the winter data gap. This could be validated by ground truth measurements in winter possibly guiding for another than linear interpolation.

The results (Sect. 3) show that the proposed approach clearly improves the ability of the model to estimate CDOM and thus light absorption especially in the northern parts of the Baltic Sea where the impacts of terrestrial CDOM are large. This underlines the performance of the combined approach to increase the predictive capability of ecosystem models. The method can be further improved by adding more rivers to the model and improving the quality of CDOM data from Sentinel 2 MSI.

For the model CDOM, we have applied a light sensitive degradation. Although this is the dominating degradation process for terrestrial CDOM (Moran et al., 2000), bacterial breakdown contributes to the degradation as well. Technically, such a process can easily be implemented. However, to our knowledge comprehensive process studies in the Baltic Sea are not done yet. Therefore, we have decided that bacterial breakdown is subject to later developments.

We consider only terrestrial CDOM in our model. In regions with high runoff, like the Baltic Sea, terrestrial CDOM is the dominating fraction (Harvey et al., 2015; Stedmon and Markager, 2003). However, a further step toward a more sophisticated model could be the inclusion of autochthonous CDOM .

as e.g. in Dutkiewicz et al. (2015).

Code and data availability. *In situ* absorption observations are available from http://eo.ymparisto.fi/data/water/Baltic_SeaLaBio/. Monthly CDOM absorption data are available from http://eo.ymparisto.fi/data/water/Baltic_SeaLaBio/CDOM_input_to_ERGOM/. Model data can be accessed via https://thredds-iow.io-warnemuende.de/thredds/catalogs/projects/SeaLaBio/catalog_sealabio.html.

The code of the biogeochemical model is available at www.ergom.net (last access: 22 September 2020). The ocean model "Modular Ocean Model MOM 5-1", used in this study, is available from the developers repository <https://github.com/mom-ocean/MOM5> (last access: 1 December 2020). The meteorological forcing is archived at https://cera-www.dkrz.de/WDCC/ui/cersearch/entry?acronym=coastDat-2_COSMO-CLM (last access: 1 December 2020). The version of the model code used to produce the results in this study is archived on Zenodo at <https://doi.org/10.5281/zenodo.4299873> (last access: 1 December 2020). In addition to the source code, the archive includes initial fields and boundary conditions except the meteorological forcing.

Nitrate and phosphate data used for model comparison are available from the ICES database <https://ocean.ices.dk/Helcom/Helcom.aspx?Mode=1> (last access: 30 April 2021).

Sample availability. Simulated CDOM data: https://www.i4.ymparisto.fi/i4/eng/tarkka_beta/index.html?type=ERGOM_CDOM&date=2019-12-01&lang=en&zoom=5&lat=61.46508&lon=32.98851

320 **Appendix A: Impact of photobleaching**

Photobleaching accounts for slow decomposition of CDOM. Although, the CDOM decomposition is slow compared to decomposition of e.g. *in situ* detritus, in water bodies with longer residence time it becomes important. For example, in the salt – CDOM absorption relationship, decomposition is considered by a hyperbolic function (Neumann et al., 2015, eq. 10):

$$325 \quad a_{\text{CDOM}} = 1.26 S^{-0.627} [\text{m}^{-1}] \quad (\text{A1})$$

We demonstrate the effect of the implemented photobleaching on the model CDOM concentration by comparing a simulation without photobleaching with the control run. The 3 nautical miles setup was used for this study. The photobleaching was switched on in 1980 and then the simulation was continued until 2019. Figure A1 shows the differences developing due to lacking a degradation process. Absorption values are far away from observations and even after 40 simulation
330 years, a new steady state is not achieved (Fig. A1a). Spatial patterns after 39 simulation years show that largest differences occur in the central basins. Differences are smaller in freshwater dominated regions (Fig. A1b) like river estuaries. Small differences are also in the vicinity of the open boundary toward the North Sea.

Appendix B: Simplified schematic of the biogeochemical model ERGOM

In Fig. B1, we show the structure of the biogeochemical model ERGOM. Ellipses are state variables and rectangles are
335 processes describing the transfer from one to another state variable. The meaning of the state variable symbols are given in Table B1.

Author contributions. TN developed the ecosystem model components. SK led the project. SV performed the processing of EO data. PY performed CDOM absorption measurements in the central Baltic Sea. All authors designed the project, and contributed to data analysis and writing the manuscript.

340 *Competing interests.* The authors declare that they have no conflict of interest.

Table B1. State variables of the biogeochemical model ERGOM

| Symbol | State Variable |
|---------------------------|--|
| O ₂ | dissolved oxygen |
| N ₂ | dissolved nitrogen |
| CDOM | colored dissolved organic matter |
| DIC | dissolved inorganic carbon |
| TA | total alkalinity |
| NH ₄ | ammonium |
| NO ₃ | nitrate |
| PO ₄ | phosphate |
| SO ₄ | sulfate |
| S | sulfur |
| H ₂ S | hydrogen sulfide |
| large cells | large cell phytoplankton |
| small cells | small cell phytoplankton |
| cyanobacteria | cyanobacteria |
| zooplankton | bulk zooplankton |
| detritus | detritus |
| DOC | dissolved organic carbon |
| DOC – N | DOC with additional nitrogen |
| DOC – P | DOC with additional phosphorus |
| POC | particulate organic carbon |
| POC – N | POC with additional nitrogen |
| POC – P | PC with additional phosphorus |
| sediment detritus | detritus accumulated in the sediment layer |
| Fe(III) – PO ₄ | iron-3-phosphate |

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References

- Asmala, E., Autio, R., Kaartokallio, H., Pitkänen, L., Stedmon, C. A., and Thomas, D. N.: Bioavailability of riverine dissolved organic matter
345 in three Baltic Sea estuaries and the effect of catchment land use, *Biogeosciences*, 10, 6969–6986, <https://doi.org/10.5194/bg-10-6969-2013>, 2013.
- Attila, J., Koponen, S., Kallio, K., Lindfors, A., Kaitala, S., and Ylöstalo, P.: MERIS Case II water processor comparison on coastal sites of the northern Baltic Sea, *Remote Sensing of Environment*, 128, 138 – 149, <https://doi.org/10.1016/j.rse.2012.07.009>, 2013.
- Brockmann, C., Doerffer, R., Peters, M., Stelzer, K., Embacher, S., and Ruescas, A.: Evolution of the C2RCC neural network for Sentinel
350 2 and 3 for the retrieval of ocean colour products in normal and extreme optically complex waters, website, http://step.esa.int/docs/extra/Evolution%20of%20the%20C2RCC_LPS16.pdf, last access: 7 July 2020, 2016.
- Cahill, B., Schofield, O., Chant, R., Wilkin, J., Hunter, E., Glenn, S., and Bissett, P.: Dynamics of turbid buoyant plumes and the feedbacks on near-shore biogeochemistry and physics, *Geophysical Research Letters*, 35, <https://doi.org/10.1029/2008GL033595>, 2008.
- Dutkiewicz, S., Hickman, A. E., Jahn, O., Gregg, W. W., Mouw, C. B., and Follows, M. J.: Capturing optically important constituents and
355 properties in a marine biogeochemical and ecosystem model, *Biogeosciences*, 12, 4447–4481, <https://doi.org/10.5194/bg-12-4447-2015>, 2015.
- Geyer, B. and Rockel, B.: coastDat-2 COSMO-CLM Atmospheric Reconstruction, https://doi.org/10.1594/WDCC/coastDat-2_COSMO-CLM, 2013.
- Gnanadesikan, A., Kim, G. E., and Pradal, M.-A. S.: Impact of Colored Dissolved Materials on the Annual Cycle of Sea
360 Surface Temperature: Potential Implications for Extreme Ocean Temperatures, *Geophysical Research Letters*, 46, 861–869, <https://doi.org/10.1029/2018GL080695>, 2019.
- Griffies, S. M.: *Fundamentals of Ocean Climate Models*, Princeton University Press, Princeton, NJ, 2004.
- Gustafsson, B. G., Schenk, F., Blenckner, T., Eilola, K., Meier, H. E. M., Müller-Karulis, B., Neumann, T., Ruoho-Airola, T., Savchuk, O. P.,
and Zorita, E.: Reconstructing the Development of Baltic Sea Eutrophication 1850–2006, *AMBIO*, 41, <https://doi.org/10.1007/s13280-012-0318-x>, G02023, 2012.
365
- Harvey, E. T., Kratzer, S., and Andersson, A.: Relationships between colored dissolved organic matter and dissolved organic carbon in different coastal gradients of the Baltic Sea, *AMBIO*, 44, 392–401, <https://doi.org/10.1007/s13280-015-0658-4>, 2015.
- Jerlov, N., ed.: *Marine Optics*, vol. 14 of *Elsevier Oceanography Series*, Elsevier, [https://doi.org/10.1016/S0422-9894\(08\)70789-X](https://doi.org/10.1016/S0422-9894(08)70789-X), 1976.
- Jolliff, J. K. and Smith, T. A.: Biological modulation of upper ocean physics: Simulating the biothermal feedback effect in Monterey Bay,
370 California, *Journal of Geophysical Research: Biogeosciences*, 119, 703–721, <https://doi.org/10.1002/2013JG002522>, 2014.
- Koponen, S., Attila, J., Pulliainen, J., Kallio, K., Pyhälähti, T., Lindfors, A., Rasmus, K., and Hallikainen, M.: A case study of air-borne and satellite remote sensing of a spring bloom event in the Gulf of Finland, *Continental Shelf Research*, 27, 228 – 244, <https://doi.org/10.1016/j.csr.2006.10.006>, 2007.
- Kowalczyk, P., A. Stedmon, C., and Markager, S.: Modeling absorption by CDOM in the Baltic Sea from season, salinity and chlorophyll,
375 *Marine Chemistry*, 101, 1 – 11, <https://doi.org/10.1016/j.marchem.2005.12.005>, 2006.
- Kowalczyk, P., Zabłocka, M., Sagan, S., and Kuliński, K.: Fluorescence measured in situ as a proxy of CDOM absorption and DOC concentration in the Baltic Sea, *Oceanologia*, 52, 431–471, 2010.
- Kratzer, S. and Moore, G.: Inherent Optical Properties of the Baltic Sea in Comparison to Other Seas and Oceans, *Remote Sensing*, 10, 418, <https://doi.org/10.3390/rs10030418>, 2018.

- 380 Kuznetsov, I. and Neumann, T.: Simulation of carbon dynamics in the Baltic Sea with a 3D model, *Journal of Marine Systems*, 111–112, 167–174, <https://doi.org/10.1016/j.jmarsys.2012.10.011>, 2013.
- Leibniz Institute for Baltic Sea Research: ERGOM: Ecological ReGional Ocean Model, <http://www.ergom.net>, last access: 27 July 2020, 2015.
- Lindfors, A., Rasmus, K., and Strömbeck, N.: Point or pointless – quality of ground data, *International Journal of Remote Sensing*, 26, 385 415–423, <https://doi.org/10.1080/01431160410001720261>, 2005.
- Moran, M. A., Sheldon Jr., W. M., and Zepp, R. G.: Carbon loss and optical property changes during long-term photochemical and biological degradation of estuarine dissolved organic matter, *Limnology and Oceanography*, 45, 1254–1264, <https://doi.org/10.4319/lo.2000.45.6.1254>, 2000.
- Nelson, N. B. and Siegel, D. A.: Chapter 11 - Chromophoric DOM in the Open Ocean, in: *Biogeochemistry of Marine Dissolved Organic Matter*, edited by Hansell, D. A. and Carlson, C. A., pp. 547 – 578, Academic Press, San Diego, <https://doi.org/10.1016/B978-012323841-2/50013-0>, 2002.
- 390 Neumann, T.: Climate-change effects on the Baltic Sea ecosystem: A model study, *Journal of Marine Systems*, 81, 213–224, <https://doi.org/10.1016/j.jmarsys.2009.12.001>, 2010.
- Neumann, T., Siegel, H., and Gerth, M.: A new radiation model for Baltic Sea ecosystem modelling, *Journal of Marine Systems*, 152, 83–91, 395 <https://doi.org/10.1016/j.jmarsys.2015.08.001>, 2015.
- Neumann, T., Siegel, H., Moros, M., Gerth, M., Kniebusch, M., and Heydebreck, D.: Ventilation of the northern Baltic Sea, *Ocean Science*, 16, 767–780, <https://doi.org/10.5194/os-16-767-2020>, 2020.
- Pefanis, V., Losa, S. N., Losch, M., Janout, M. A., and Bracher, A.: Amplified Arctic Surface Warming and Sea Ice Loss Due to Phytoplankton and Colored Dissolved Material, *Geophysical Research Letters*, 47, e2020GL088795, <https://doi.org/10.1029/2020GL088795>, 400 e2020GL088795 10.1029/2020GL088795, 2020.
- Radtke, H., Lipka, M., Bunke, D., Morys, C., Woelfel, J., Cahill, B., Böttcher, M. E., Forster, S., Leipe, T., Rehder, G., and Neumann, T.: Ecological ReGional Ocean Model with vertically resolved sediments (ERGOM SED 1.0): coupling benthic and pelagic biogeochemistry of the south-western Baltic Sea, *Geoscientific Model Development*, 12, 275–320, <https://doi.org/10.5194/gmd-12-275-2019>, 2019.
- Seifert, T., Tauber, F., and Kayser, B.: Digital topography of the Baltic Sea, <https://www.io-warnemuende.de/topography-of-the-baltic-sea.html>, last access: 27 July 2020, 2008.
- 405 Skoog, A., Wedborg, M., and Fogelqvist, E.: Decoupling of total organic carbon concentrations and humic substance fluorescence in an extended temperate estuary, *Marine Chemistry*, 124, 68 – 77, <https://doi.org/10.1016/j.marchem.2010.12.003>, 2011.
- Stedmon, C. and Markager, S.: Behaviour of the optical properties of coloured dissolved organic matter under conservative mixing, *Estuarine, Coastal and Shelf Science*, 57, 973 – 979, [https://doi.org/10.1016/S0272-7714\(03\)00003-9](https://doi.org/10.1016/S0272-7714(03)00003-9), 2003.
- 410 Stedmon, C. A., Osburn, C. L., and Kragh, T.: Tracing water mass mixing in the Baltic–North Sea transition zone using the optical properties of coloured dissolved organic matter, *Estuarine, Coastal and Shelf Science*, 87, 156 – 162, <https://doi.org/10.1016/j.ecss.2009.12.022>, 2010.
- Stigebrandt, A. and Wulff, F.: A model for the dynamics of nutrients and oxygen in the Baltic proper, *Journal of Marine Research*, 45, 729–759, 1987.
- 415 Winton, M.: A Reformulated Three-Layer Sea Ice Model, *Journal of Atmospheric and Oceanic Technology*, 17, 525–531, [https://doi.org/10.1175/1520-0426\(2000\)017<0525:ARTLSI>2.0.CO;2](https://doi.org/10.1175/1520-0426(2000)017<0525:ARTLSI>2.0.CO;2), 2000.

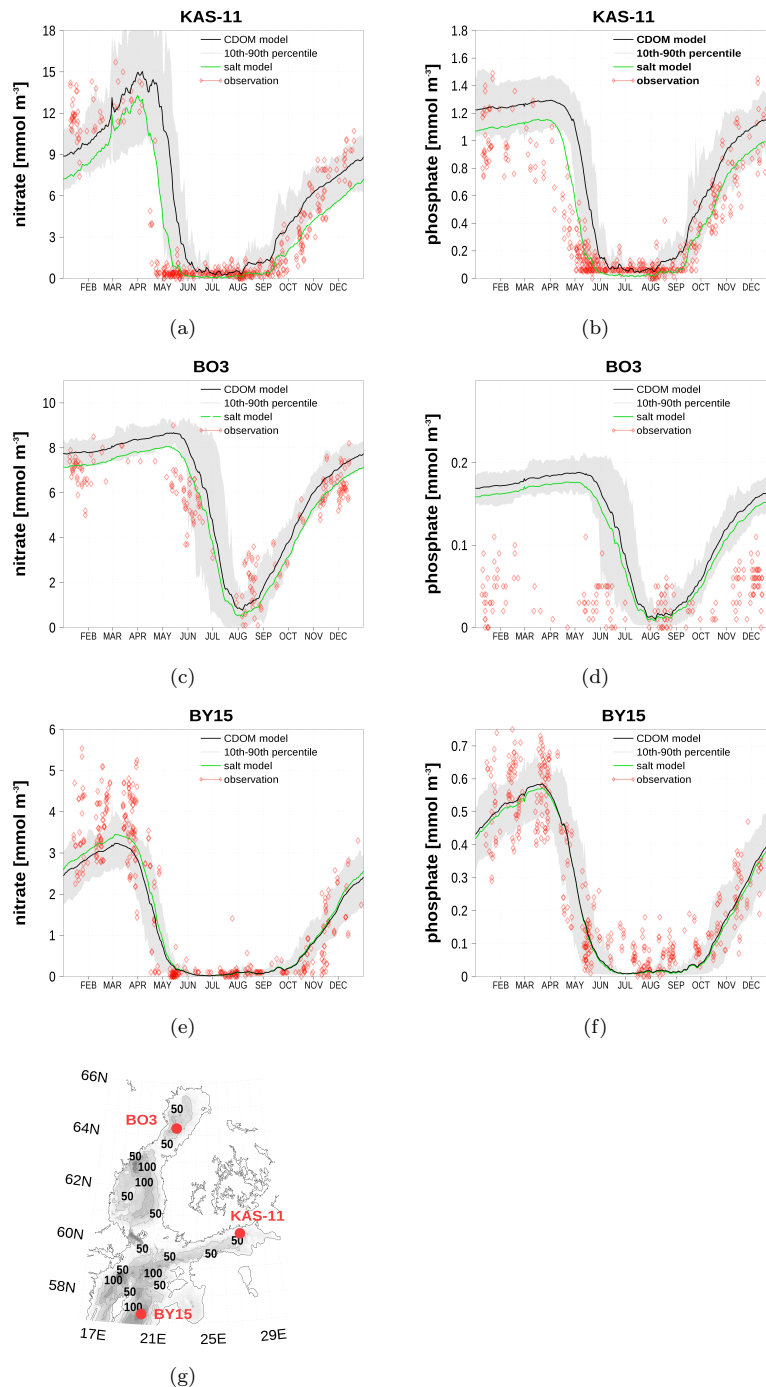


Figure 8. Climatology (1990–2019) of surface nitrate and surface phosphate at 3 stations. Black lines show the new CDOM based model, green line the salt based model, and red diamonds are observation (<http://www.ices.dk>, last access: 19 June 2020). The shaded area is the range between 10th and 90th percentile of the black line. Simulated data are from the 3 n.m. model version. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008).

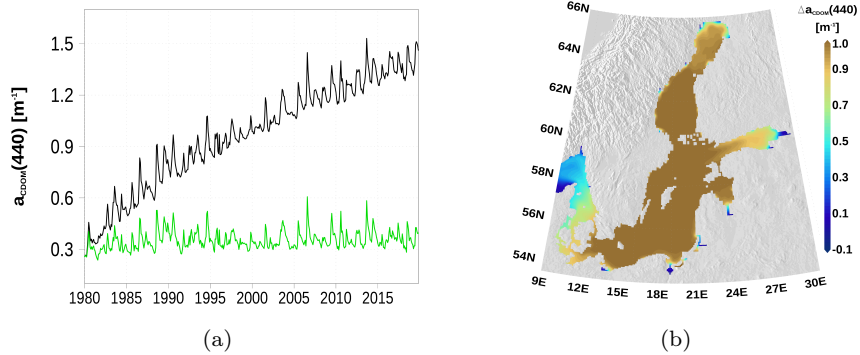


Figure A1. The effect of photobleaching on CDOM absorption. The black line in (a) is from the model without photobleaching at station BY15 (Fig. 8). In (b) the difference in 2018 of the surface CDOM absorption is shown (without photobleaching minus control run). Simulated CDOM concentration have been converted into absorption. The map was created using the software package GrADS 2.1.1.b0 (<http://cola.gmu.edu/grads/>), using published bathymetry data (Seifert et al., 2008).

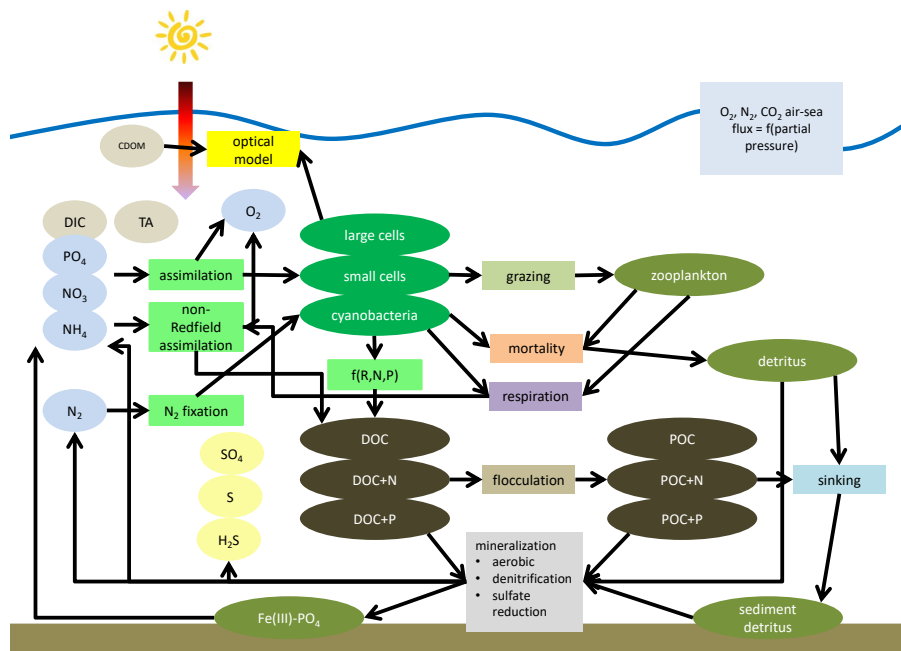


Figure B1. Simplified schematic of the biogeochemical model ERGOM. State variables are shown as ellipses and processes as rectangles.