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1. Appendix table

No.	Species	Definition	Molecular Weight
3	NO		30
3a	$^{15}\text{NO}$	Nitric oxide	31
4	$\text{NO}_2$		46
4a	$^{15}\text{NO}_2$	Nitrogen dioxide	47
5	$\text{NO}_3$		62
5a	$^{15}\text{NO}_3$	Nitrogen trioxide	63
6	$\text{N}_2\text{O}_5$		108
6a	$^{15}\text{NNO}_5$	Dinitrogen pentoxide	109
6b	$^{15}\text{N}_2\text{O}_5$		110
7	HONO		47
7a	$\text{HO}^{15}\text{NO}$	Nitrous acid	48
8	$\text{HNO}_3$		63
8a	$\text{H}^{15}\text{NO}_3$	Nitric acid	64
9	$\text{HNO}_4$		79
9a	$\text{H}^{15}\text{NO}_4$	Pernitric acid	80
14	$\text{N}_2$		28
14a	$^{15}\text{NN}$	Nitrogen	29
14b	$^{15}\text{N}_2$		30
46	ONIT		119
46a	$^{15}\text{ONIT}$	Organic nitrate	120
47	PAN		121
47a	$^{15}\text{PAN}$	Peroxyacetyl nitrate and higher saturated PANs	122
48	TPAN		147
48a	$^{15}\text{TPAN}$	Unsaturated PANs	148
75	OLNN		136
75a	$^{15}\text{OLNN}$	$\text{NO}_3$ -alkene adduct reacting to form carbonitrates + $\text{HO}_2$	137
76	OLND		136
76a	$^{15}\text{OLND}$	$\text{NO}_3$ -alkene adduct reacting via decomposition	137

Table S1a:  $^{14}\text{N}$  and  $^{15}\text{N}$  species

No.	Species	Definition	Molecular Weight
37	HCHO	Formaldehyde	30
38	ALD	Acetaldehyde and higher aldehydes	44
40	GLY	Glyoxal	58
43	MACR	Methacrolein and other unsaturated monoaldehydes	70
41	MGLY	Methylglyxal and other $\alpha$ -carbonyl aldehydes	72
42	DCB	unsaturated dicarbonyls	87
36	CSL	cresol and other hydroxy substituted aromatics	108

Table S1b: Hydrocarbon species

Reaction No.	Reaction	Photolysis Frequency, s-1	Cross Section	Quantum Yield	$\alpha$
R1	$\text{NO}_2 \rightarrow \text{O}^3\text{P} + \text{NO}$	$7.50 \times 10^{-3}$	DeMoretal. [1994]	DeMoretal. [1994]	1
R1a	$^{15}\text{NO}_2 \rightarrow \text{O}^3\text{P} + ^{15}\text{NO}$	$7.50 \times 10^{-3}$	DeMoretal. [1994]	DeMoretal. [1994]	1.0042
R4	$\text{HONO} \rightarrow \text{HO} + \text{NO}$	$1.63 \times 10^{-3}$	DeMoretal. [1994]	DeMoretal. [1995]	1
R4a	$\text{HO}^{15}\text{NO} \rightarrow \text{HO} + ^{15}\text{NO}$	$1.63 \times 10^{-3}$	DeMoretal. [1994]	DeMoretal. [1996]	1
R5	$\text{HNO}_3 \rightarrow \text{HO} + \text{NO}_2$	$4.50 \times 10^{-7}$	DeMoretal. [1994]	assumed to be unity	1
R5a	$\text{H}^{15}\text{NO}_3 \rightarrow \text{HO} + ^{15}\text{NO}_2$	$4.50 \times 10^{-7}$	DeMoretal. [1994]	assumed to be unity	1
R6	$\text{HNO}_4 \rightarrow 0.65 \text{HO}_2 + 0.65 \text{NO}_2 + 0.35 \text{HO} + 0.35 \text{NO}_3$	$3.17 \times 10^{-6}$	DeMoretal. [1994]	assumed to be unity	1
R6a	$\text{H}^{15}\text{NO}_4 \rightarrow 0.65 \text{HO}_2 + 0.65 ^{15}\text{NO}_2 + 0.35 \text{HO} + 0.35 ^{15}\text{NO}_3$	$3.17 \times 10^{-6}$	DeMoretal. [1994]	assumed to be unity	1
R7	$\text{NO}_3 \rightarrow \text{NO} + \text{O}_2$	$2.33 \times 10^{-2}$	Waynetal. [1991]	Waynetal. [1992]	1
R7a	$^{15}\text{NO}_3 \rightarrow ^{15}\text{NO} + \text{O}_2$	$2.33 \times 10^{-2}$	Waynetal. [1991]	Waynetal. [1992]	1
R8	$\text{NO}_3 \rightarrow \text{NO}_2 + \text{O}^3\text{P}$	$1.87 \times 10^{-1}$	Waynetal. [1991]	Waynetal. [1992]	1
R8a	$^{15}\text{NO}_3 \rightarrow ^{15}\text{NO}_2 + \text{O}^3\text{P}$	$1.87 \times 10^{-1}$	Waynetal. [1991]	Waynetal. [1992]	1

Table S2a: Photolysis reactions involving N compounds

Reaction No.	Reaction	A, cm <sup>3</sup> s <sup>-1</sup>	E/R, K	k	$\alpha$
R26	O <sup>1</sup> D + N <sub>2</sub> --> O <sup>3</sup> P + N <sub>2</sub>	1.80 x 10 <sup>-11</sup>	-110	2.60 x 10 <sup>-11</sup>	1
R26a	O <sup>1</sup> D + <sup>15</sup> NN --> O <sup>3</sup> P + <sup>15</sup> NN	1.80 x 10 <sup>-11</sup>	-110	2.60 x 10 <sup>-11</sup>	1
R35	O <sup>3</sup> P + NO --> NO <sub>2</sub>	Table S2d		1.66 x 10 <sup>-12</sup>	1
R35a	O <sup>3</sup> P + <sup>15</sup> NO --> <sup>15</sup> NO <sub>2</sub>	Table S2d		1.66 x 10 <sup>-12</sup>	1
R36	O <sup>3</sup> P + NO <sub>2</sub> --> NO + O <sub>2</sub>	6.50 x 10 <sup>-12</sup>	-120	9.72 x 10 <sup>-12</sup>	1
R36a	O <sup>3</sup> P + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> NO + O <sub>2</sub>	6.50 x 10 <sup>-12</sup>	-120	9.72 x 10 <sup>-12</sup>	1
R37	O <sup>3</sup> P + NO <sub>2</sub> --> NO <sub>3</sub>	Table S2d		1.58 x 10 <sup>-12</sup>	1
R37a	O <sup>3</sup> P + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> NO <sub>3</sub>	Table S2d		1.58 x 10 <sup>-12</sup>	1
R38	HO + NO --> HONO	Table S2d		1.58 x 10 <sup>-12</sup>	1
R38a	HO + <sup>15</sup> NO --> HO <sup>15</sup> NO	Table S2d		1.58 x 10 <sup>-12</sup>	1
R39	HO + NO <sub>2</sub> --> HNO <sub>3</sub>	Table S2d		1.58 x 10 <sup>-12</sup>	1
R39a	HO + <sup>15</sup> NO <sub>2</sub> --> H <sup>15</sup> NO <sub>3</sub>	Table S2d		1.58 x 10 <sup>-12</sup>	1.04
R40	HO + NO <sub>3</sub> --> NO <sub>2</sub> + HO <sub>2</sub>	2.20 x 10 <sup>-11</sup>		2.20 x 10 <sup>-11</sup>	1
R40a	HO + <sup>15</sup> NO <sub>3</sub> --> <sup>15</sup> NO <sub>2</sub> + HO <sub>2</sub>	2.20 x 10 <sup>-11</sup>		2.20 x 10 <sup>-11</sup>	1
R41	HO <sub>2</sub> + NO --> NO <sub>2</sub> + HO	3.70 x 10 <sup>-12</sup>	-250	8.56 x 10 <sup>-12</sup>	1
R41a	HO <sub>2</sub> + <sup>15</sup> NO --> <sup>15</sup> NO <sub>2</sub> + HO	3.70 x 10 <sup>-12</sup>	-250	8.56 x 10 <sup>-12</sup>	1

R42	$\text{HO}_2 + \text{NO}_2 \rightarrow \text{HNO}_4$	Table S2d		$1.39 \times 10^{-12}$	1
R42a	$\text{HO}_2 + {}^{15}\text{NO}_2 \rightarrow \text{H}{}^{15}\text{NO}_4$	Table S2d		$1.39 \times 10^{-12}$	1
R43	$\text{HNO}_4 \rightarrow \text{HO}_2 + \text{NO}_2$	Table S2e		$8.62 \times 10^{-2}$	1
R43a	$\text{H}{}^{15}\text{NO}_4 \rightarrow \text{HO}_2 + {}^{15}\text{NO}_2$	Table S2e		$8.62 \times 10^{-2}$	1
R44	$\text{HO}_2 + \text{NO}_3 \rightarrow 0.3 \text{ HNO}_3 + 0.7 \text{ NO}_2 + 0.7 \text{ HO} + \text{O}_2$	$3.50 \times 10^{-12}$		$3.50 \times 10^{-12}$	1
R44a	$\text{HO}_2 + {}^{15}\text{NO}_3 \rightarrow 0.3 \text{ H}{}^{15}\text{NO}_3 + 0.7 {}^{15}\text{NO}_2 + 0.7 \text{ HO} + \text{O}_2$	$3.50 \times 10^{-12}$		$3.50 \times 10^{-12}$	1
R45	$\text{HO} + \text{HONO} \rightarrow \text{NO}_2 + \text{H}_2\text{O}$	$1.80 \times 10^{-11}$	390	$4.86 \times 10^{-12}$	1
R45a	$\text{HO} + \text{HO}{}^{15}\text{NO} \rightarrow {}^{15}\text{NO}_2 + \text{H}_2\text{O}$	$1.80 \times 10^{-11}$	390	$4.86 \times 10^{-12}$	1
R46	$\text{HO} + \text{HNO}_3 \rightarrow \text{NO}_3 + \text{H}_2\text{O}$	Table S2f		$1.47 \times 10^{-13}$	1
R46a	$\text{HO} + \text{H}{}^{15}\text{NO}_3 \rightarrow {}^{15}\text{NO}_3 + \text{H}_2\text{O}$	Table S2f		$1.47 \times 10^{-13}$	1
R47	$\text{HO} + \text{HNO}_4 \rightarrow \text{NO}_2 + \text{O}_2 + \text{H}_2\text{O}$	$1.30 \times 10^{-12}$	-380	$4.65 \times 10^{-12}$	1
R47a	$\text{HO} + \text{H}{}^{15}\text{NO}_4 \rightarrow {}^{15}\text{NO}_2 + \text{O}_2 + \text{H}_2\text{O}$	$1.30 \times 10^{-12}$	-380	$4.65 \times 10^{-12}$	1
R48	$\text{O}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{O}_2$	$2.00 \times 10^{-12}$	1400	$1.82 \times 10^{-14}$	1
R48a	$\text{O}_3 + {}^{15}\text{NO} \rightarrow {}^{15}\text{NO}_2 + \text{O}_2$	$2.00 \times 10^{-12}$	1400	$1.82 \times 10^{-14}$	0.9933
R49	$\text{O}_3 + \text{NO}_2 \rightarrow \text{NO}_3 + \text{O}_2$	$1.20 \times 10^{-13}$	2450	$3.23 \times 10^{-17}$	1
R49a	$\text{O}_3 + {}^{15}\text{NO}_2 \rightarrow {}^{15}\text{NO}_3 + \text{O}_2$	$1.20 \times 10^{-13}$	2450	$3.23 \times 10^{-17}$	1
R50	$\text{NO} + \text{NO} + \text{O}_2 \rightarrow \text{NO}_2 + \text{NO}_2$	$3.30 \times 10^{-39}$	-530	$1.95 \times 10^{-38}$	1

R50a	$\text{NO} + ^{15}\text{NO} + \text{O}_2 \rightarrow \text{NO}_2 + ^{15}\text{NO}_2$	$3.30 \times 10^{-39}$	-530	$1.95 \times 10^{-38}$	1
R50b	$^{15}\text{NO} + ^{15}\text{NO} + \text{O}_2 \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_2$	$3.30 \times 10^{-39}$	-530	$1.95 \times 10^{-38}$	1
R51	$\text{NO}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{NO}_2$	$1.50 \times 10^{-11}$	-170	$2.65 \times 10^{-11}$	1
R51a	$^{15}\text{NO}_3 + \text{NO} \rightarrow \text{NO}_2 + ^{15}\text{NO}_2$	$1.50 \times 10^{-11}$	-170	$2.65 \times 10^{-11}$	1
R51b	$\text{NO}_3 + ^{15}\text{NO} \rightarrow \text{NO}_2 + ^{15}\text{NO}_2$	$1.50 \times 10^{-11}$	-170	$2.65 \times 10^{-11}$	1
R51c	$^{15}\text{NO}_3 + ^{15}\text{NO} \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_2$	$1.50 \times 10^{-11}$	-170	$2.65 \times 10^{-11}$	1
R52	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{NO} + \text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	1
R52a	$\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NO} + \text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	0.5
R52b	$\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow \text{NO} + ^{15}\text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	0.5
R52c	$^{15}\text{NO}_3 + \text{NO}_2 \rightarrow ^{15}\text{NO} + \text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	0.5
R52d	$^{15}\text{NO}_3 + \text{NO}_2 \rightarrow \text{NO} + ^{15}\text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	0.5
R52e	$^{15}\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NO} + ^{15}\text{NO}_2 + \text{O}_2$	$4.50 \times 10^{-14}$	1260	$6.56 \times 10^{-16}$	1
R53	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{N}_2\text{O}_5$	Table S2d		$1.27 \times 10^{-12}$	1
R53a	$\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NNO}_5$	Table S2d		$1.27 \times 10^{-12}$	1.0266
R53b	$^{15}\text{NO}_3 + \text{NO}_2 \rightarrow ^{15}\text{NNO}_5$	Table S2d		$1.27 \times 10^{-12}$	1.0309
R53c	$^{15}\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{N}_2\text{O}_5$	Table S2d		$1.27 \times 10^{-12}$	1.057
R54	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	Table S2e		$4.36 \times 10^{-2}$	1
R54a	$^{15}\text{NNO}_5 \rightarrow ^{15}\text{NO}_2 + \text{NO}_3$	Table S2e		$4.36 \times 10^{-2}$	0.5
R54b	$^{15}\text{NNO}_5 \rightarrow \text{NO}_2 + ^{15}\text{NO}_3$	Table S2e		$4.36 \times 10^{-2}$	0.5

R54c	$^{15}\text{N}_2\text{O}_5 \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_3$	Table S2e		$4.36 \times 10^{-2}$	1
R55	$\text{NO}_3 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{NO}_2 + \text{O}_2$	$8.50 \times 10^{-13}$	2450	$2.29 \times 10^{-16}$	1
R55a	$\text{NO}_3 + 15\text{NO}_3 \rightarrow \text{NO}_2 + ^{15}\text{NO}_2 + \text{O}_2$	$8.50 \times 10^{-13}$	2450	$2.29 \times 10^{-16}$	1
R55b	$^{15}\text{NO}_3 + ^{15}\text{NO}_3 \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_2 + \text{O}_2$	$8.50 \times 10^{-13}$	2450	$2.29 \times 10^{-16}$	1
R88	$\text{PAN} + \text{HO} \rightarrow \text{HCHO} + \text{XO}_2 + \text{H}_2\text{O} + \text{NO}_3$	$4.00 \times 10^{-14}$		$4.00 \times 10^{-14}$	1
R88a	$^{15}\text{PAN} + \text{HO} \rightarrow \text{HCHO} + \text{XO}_2 + \text{H}_2\text{O} + ^{15}\text{NO}_3$	$4.00 \times 10^{-14}$		$4.00 \times 10^{-14}$	1
R89	$\text{TPAN} + \text{HO} \rightarrow 0.60 \text{ HKET} + 0.40 \text{ HCHO} + 0.40 \text{ HO}_2 + \text{XO}_2 + 0.40 \text{ PAN} + 0.60 \text{ NO}_3$	$3.25 \times 10^{-13}$	-500	$1.74 \times 10^{-12}$	1
R89a	$^{15}\text{TPAN} + \text{HO} \rightarrow 0.60 \text{ HKET} + 0.40 \text{ HCHO} + 0.40 \text{ HO}_2 + \text{XO}_2 + 0.40 ^{15}\text{PAN} + 0.60 ^{15}\text{NO}_3$	$3.25 \times 10^{-13}$	-500	$1.74 \times 10^{-12}$	1
R90	$\text{ONIT} + \text{HO} \rightarrow \text{HC3P} + \text{NO}_2 + \text{H}_2\text{O}$	$5.31 \times 10^{-12}$	260	$2.22 \times 10^{-12}$	1
R90a	$^{15}\text{ONIT} + \text{HO} \rightarrow \text{HC3P} + ^{15}\text{NO}_2 + \text{H}_2\text{O}$	$5.31 \times 10^{-12}$	260	$2.22 \times 10^{-12}$	1
R91	$\text{HCHO} + \text{NO}_3 \rightarrow \text{HO}_2 + \text{HNO}_3 + \text{CO}$	$3.40 \times 10^{-13}$	1900	$5.79 \times 10^{-16}$	1
R91a	$\text{HCHO} + ^{15}\text{NO}_3 \rightarrow \text{HO}_2 + \text{H}^{15}\text{NO}_3 + \text{CO}$	$3.40 \times 10^{-13}$	1900	$5.79 \times 10^{-16}$	0.9974
R92	$\text{ALD} + \text{NO}_3 \rightarrow \text{ACO}_3 + \text{HNO}_3$	$1.40 \times 10^{-12}$	1900	$2.38 \times 10^{-15}$	1
R92a	$\text{ALD} + ^{15}\text{NO}_3 \rightarrow \text{ACO}_3 + \text{H}^{15}\text{NO}_3$	$1.40 \times 10^{-12}$	1900	$2.38 \times 10^{-15}$	0.9967
R93	$\text{GLY} + \text{NO}_3 \rightarrow \text{HNO}_3 + \text{HO}_2 + 2 \text{ CO}$	$2.90 \times 10^{-12}$	1900	$4.94 \times 10^{-15}$	1
R93a	$\text{GLY} + ^{15}\text{NO}_3 \rightarrow \text{H}^{15}\text{NO}_3 + \text{HO}_2 + 2 \text{ CO}$	$2.90 \times 10^{-12}$	1900	$4.94 \times 10^{-15}$	0.9962

R94	MGLY + NO <sub>3</sub> --> HNO <sub>3</sub> + ACO <sub>3</sub> + CO	1.40 x 10 <sup>-12</sup>	1900	2.38 x 10 <sup>-15</sup>	1
R94a	MGLY + <sup>15</sup> NO <sub>3</sub> --> H <sup>15</sup> NO <sub>3</sub> + ACO <sub>3</sub> + CO	1.40 x 10 <sup>-12</sup>	1900	2.38 x 10 <sup>-15</sup>	0.9957
R95	MACR + NO <sub>3</sub> --> 0.20 TCO <sub>3</sub> + 0.20 HNO <sub>3</sub> + 0.80 OLNN + 0.80 CO	8.27 x 10 <sup>-15</sup>	150	5.00 x 10 <sup>-15</sup>	1
R95a	MACR + <sup>15</sup> NO <sub>3</sub> --> 0.20 TCO <sub>3</sub> + 0.20 H <sup>15</sup> NO <sub>3</sub> + 0.80 <sup>15</sup> OLNN + 0.80 CO	8.27 x 10 <sup>-15</sup>	150	5.00 x 10 <sup>-15</sup>	0.9958
R96	DCB + NO <sub>3</sub> --> 0.50 TCO <sub>3</sub> + 0.50 HO <sub>2</sub> + 0.50 XO <sub>2</sub> + 0.25 GLY + 0.25 ALD + 0.03 KET + 0.25 MGLY + 0.5 HNO <sub>3</sub> + 0.5 NO <sub>2</sub>	2.87 x 10 <sup>-13</sup>	1000	1.00 x 10 <sup>-14</sup>	1
R96a	DCB + <sup>15</sup> NO <sub>3</sub> --> 0.50 TCO <sub>3</sub> + 0.50 HO <sub>2</sub> + 0.50 XO <sub>2</sub> + 0.25 GLY + 0.25 ALD + 0.03 KET + 0.25 MGLY + 0.5 H <sup>15</sup> NO <sub>3</sub> + 0.5 <sup>15</sup> NO <sub>2</sub>	2.87 x 10 <sup>-13</sup>	1000	1.00 x 10 <sup>-14</sup>	0.9954
R97	CSL + NO <sub>3</sub> --> HNO <sub>3</sub> + PHO	2.20 x 10 <sup>-11</sup>		2.20 x 10 <sup>-11</sup>	1
R97a	CSL + <sup>15</sup> NO <sub>3</sub> --> H <sup>15</sup> NO <sub>3</sub> + PHO	2.20 x 10 <sup>-11</sup>		2.20 x 10 <sup>-11</sup>	0.9949
R98	ETE + NO <sub>3</sub> --> 0.80 OLNN + 0.20 OLND	Table S2c		2.05 x 10 <sup>-16</sup>	1
R98a	ETE + <sup>15</sup> NO <sub>3</sub> --> 0.80 <sup>15</sup> OLNN + 0.20 <sup>15</sup> OLND	Table S2c		2.05 x 10 <sup>-16</sup>	1
R99	OLT + NO <sub>3</sub> --> 0.43 OLNN + 0.57 OLND	1.79 x 10 <sup>-13</sup>	450	3.95 x 10 <sup>-14</sup>	1
R99a	OLT + <sup>15</sup> NO <sub>3</sub> --> <sup>15</sup> 0.43 OLNN + 0.57 <sup>15</sup> OLND	1.79 x 10 <sup>-13</sup>	450	3.95 x 10 <sup>-14</sup>	1
R100	OLI + NO <sub>3</sub> --> 0.11 OLNN + 0.89 OLND	8.64 x 10 <sup>-13</sup>	-450	3.91 x 10 <sup>-12</sup>	1

R100a	$\text{OLI} + {}^{15}\text{NO}_3 \rightarrow 0.11 {}^{15}\text{OLNN} + 0.89 {}^{15}\text{OLND}$	$8.64 \times 10^{-13}$	-450	$3.91 \times 10^{-12}$	1
R101	$\text{DIEN} + \text{NO}_3 \rightarrow 0.90 \text{ OLNN} + 0.10 \text{ OLND} + 0.90 \text{ MACR}$	$1.0 \times 10^{-13}$		$1.0 \times 10^{-13}$	1
R101a	$\text{DIEN} + {}^{15}\text{NO}_3 \rightarrow 0.90 {}^{15}\text{OLNN} + 0.10 {}^{15}\text{OLND} + 0.90 \text{ MACR}$	$1.0 \times 10^{-13}$		$1.0 \times 10^{-13}$	1
R102	$\text{ISO} + \text{NO}_3 \rightarrow 0.90 \text{ OLNN} + 0.10 \text{ OLND} + 0.90 \text{ MACR}$	$4.00 \times 10^{-12}$	446	$8.96 \times 10^{-13}$	1
R102a	$\text{ISO} + {}^{15}\text{NO}_3 \rightarrow 0.90 {}^{15}\text{OLNN} + 0.10 {}^{15}\text{OLND} + 0.90 \text{ MACR}$	$4.00 \times 10^{-12}$	446	$8.96 \times 10^{-13}$	1
R103	$\text{API} + \text{NO}_3 \rightarrow 0.10 \text{ OLNN} + 0.90 \text{ OLND}$	$1.19 \times 10^{-12}$	-490	$6.16 \times 10^{-12}$	1
R103a	$\text{API} + {}^{15}\text{NO}_3 \rightarrow 0.10 {}^{15}\text{OLNN} + 0.90 {}^{15}\text{OLND}$	$1.19 \times 10^{-12}$	-490	$6.16 \times 10^{-12}$	1
R104	$\text{LIM} + \text{NO}_3 \rightarrow 0.13 \text{ OLNN} + 0.87 \text{ OLND}$	$1.22 \times 10^{-11}$		$1.22 \times 10^{-11}$	1
R104a	$\text{LIM} + {}^{15}\text{NO}_3 \rightarrow 0.13 {}^{15}\text{OLNN} + 0.87 {}^{15}\text{OLND}$	$1.22 \times 10^{-11}$		$1.22 \times 10^{-11}$	1
R105	$\text{TPAN} + \text{NO}_3 \rightarrow 0.60 \text{ ONIT} + 0.60 \text{ NO}_3 + 0.40 \text{ PAN} + 0.40 \text{ HCHO} + 0.40 \text{ NO}_2 + \text{XO}_2$	$2.20 \times 10^{-14}$	500	$4.11 \times 10^{-15}$	1
R105a	$\text{TPAN} + {}^{15}\text{NO}_3 \rightarrow 0.30 \text{ ONIT} + 0.30 {}^{15}\text{ONIT} + 0.30 \text{ NO}_3 + 0.30 {}^{15}\text{NO}_3 + 0.20 \text{ PAN} + 0.20 {}^{15}\text{PAN} + 0.40 \text{ HCHO} + 0.20 \text{ NO}_2 + 0.20 {}^{15}\text{NO}_2 + \text{XO}_2$	$2.20 \times 10^{-14}$	500	$4.11 \times 10^{-15}$	1
R105b	${}^{15}\text{TPAN} + {}^{15}\text{NO}_3 \rightarrow 0.60 {}^{15}\text{ONIT} + 0.60 {}^{15}\text{NO}_3 + 0.40 {}^{15}\text{PAN} + 0.40 \text{ HCHO} + 0.40 {}^{15}\text{NO}_2 + \text{XO}_2$	$2.20 \times 10^{-14}$	500	$4.11 \times 10^{-15}$	1

R115	$\text{TPAN} + \text{O}_3 \rightarrow 0.70 \text{ HCHO} + 0.30 \text{ PAN} + 0.70 \text{ NO}_2 + 0.13 \text{ CO} + 0.04 \text{ H}_2 + 0.11 \text{ ORA1} + 0.08 \text{ HO}_2 + 0.036 \text{ HO} + 0.70 \text{ ACO}_3$	$2.46 \times 10^{-15}$	1700	$8.19 \times 10^{-18}$	1
R115a	$^{15}\text{TPAN} + \text{O}_3 \rightarrow 0.70 \text{ HCHO} + 0.30 \text{ }^{15}\text{PAN} + 0.70 \text{ }^{15}\text{NO}_2 + 0.13 \text{ CO} + 0.04 \text{ H}_2 + 0.11 \text{ ORA1} + 0.08 \text{ HO}_2 + 0.036 \text{ HO} + 0.70 \text{ ACO}_3$	$2.46 \times 10^{-15}$	1700	$8.19 \times 10^{-18}$	1
R116	$\text{PHO} + \text{NO}_2 \rightarrow 0.10 \text{ CSL} + \text{ONIT}$	$2.00 \times 10^{-11}$		$2.00 \times 10^{-11}$	1
R116a	$\text{PHO} + \text{ }^{15}\text{NO}_2 \rightarrow 0.10 \text{ CSL} + \text{ }^{15}\text{ONIT}$	$2.00 \times 10^{-11}$		$2.00 \times 10^{-11}$	1
R117	$\text{ADDT} + \text{NO}_2 \rightarrow \text{CSL} + \text{HONO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R117a	$\text{ADDT} + \text{ }^{15}\text{NO}_2 \rightarrow \text{CSL} + \text{HO }^{15}\text{NO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R121	$\text{ADDX} + \text{NO}_2 \rightarrow \text{CSL} + \text{HONO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R121a	$\text{ADDX} + \text{ }^{15}\text{NO}_2 \rightarrow \text{CSL} + \text{HO }^{15}\text{NO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R124	$\text{ADDC} + \text{NO}_2 \rightarrow \text{CSL} + \text{HONO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R124a	$\text{ADDC} + \text{ }^{15}\text{NO}_2 \rightarrow \text{CSL} + \text{HO }^{15}\text{NO}$	$3.60 \times 10^{-11}$		$3.60 \times 10^{-11}$	1
R127	$\text{ACO}_3 + \text{NO}_2 \rightarrow \text{PAN}$	Table S2d		$8.66 \times 10^{-12}$	1
R127a	$\text{ACO}_3 + \text{ }^{15}\text{NO}_2 \rightarrow \text{ }^{15}\text{PAN}$	Table S2d		$8.66 \times 10^{-12}$	1
R128	$\text{PAN} \rightarrow \text{ACO}_3 + \text{NO}_2$	Table S2e		$4.63 \times 10^{-4}$	1
R128a	$^{15}\text{PAN} \rightarrow \text{ACO}_3 + \text{ }^{15}\text{NO}_2$	Table S2e		$4.63 \times 10^{-4}$	1

R129	$\text{TCO}_3 + \text{NO}_2 \rightarrow \text{TPAN}$	Table S2d		$8.66 \times 10^{-12}$	1
R129a	$\text{TCO}_3 + {}^{15}\text{NO}_2 \rightarrow {}^{15}\text{TPAN}$	Table S2d		$8.66 \times 10^{-12}$	1
R130	$\text{TPAN} \rightarrow \text{TCO}_3 + \text{NO}_2$	Table S2e		$4.63 \times 10^{-4}$	1
R130a	${}^{15}\text{TPAN} \rightarrow \text{TCO}_3 + {}^{15}\text{NO}_2$	Table S2e		$4.63 \times 10^{-4}$	1
R131	$\text{MO}_2 + \text{NO} \rightarrow \text{HCHO} + \text{HO}_2 + \text{NO}_2$	$4.2 \times 10^{-12}$	-180	$7.68 \times 10^{-12}$	1
R131a	$\text{MO}_2 + {}^{15}\text{NO} \rightarrow \text{HCHO} + \text{HO}_2 + {}^{15}\text{NO}_2$	$4.2 \times 10^{-12}$	-180	$7.68 \times 10^{-12}$	1
R132	$\text{ETHP} + \text{NO} \rightarrow \text{ALD} + \text{HO}_2 + \text{NO}_2$	$8.7 \times 10^{-12}$		$8.7 \times 10^{-12}$	1
R132a	$\text{ETHP} + {}^{15}\text{NO} \rightarrow \text{ALD} + \text{HO}_2 + {}^{15}\text{NO}_2$	$8.7 \times 10^{-12}$		$8.7 \times 10^{-12}$	1
R133	$\text{HC3P} + \text{NO} \rightarrow 0.047 \text{ HCHO} + 0.233 \text{ ALD} + 0.623 \text{ KET} + 0.063 \text{ GLY} + 0.742 \text{ HO}_2 + 0.15 \text{ MO}_2 + 0.048 \text{ ETHP} + 0.048 \text{ XO}_2 + 0.059 \text{ ONIT} + 0.941 \text{ NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R133a	$\text{HC3P} + {}^{15}\text{NO} \rightarrow 0.047 \text{ HCHO} + 0.233 \text{ ALD} + 0.623 \text{ KET} + 0.063 \text{ GLY} + 0.742 \text{ HO}_2 + 0.15 \text{ MO}_2 + 0.048 \text{ ETHP} + 0.048 \text{ XO}_2 + 0.059 \text{ {}^{15}ONIT} + 0.941 \text{ {}^{15}NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R134	$\text{HC5P} + \text{NO} \rightarrow 0.021 \text{ HCHO} + 0.211 \text{ ALD} + 0.722 \text{ KET} + 0.599 \text{ HO}_2 + 0.031 \text{ MO}_2 + 0.245 \text{ ETHP} + 0.334 \text{ XO}_2 + 0.124 \text{ ONIT} + 0.876 \text{ NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1

R134a	$\text{HC5P} + ^{15}\text{NO} \rightarrow 0.021 \text{HCHO} + 0.211 \text{ALD} + 0.722 \text{KET} + 0.599 \text{HO}_2 + 0.031 \text{MO}_2 + 0.245 \text{ETHP} + 0.334 \text{XO}_2 + 0.124 ^{15}\text{ONIT} + 0.876 ^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R135	$\text{HC8P} + \text{NO} \rightarrow 0.15 \text{ALD} + 0.642 \text{KET} + 0.133 \text{ETHP} + 0.261 \text{ONIT} + 0.739 \text{NO}_2 + 0.606 \text{HO}_2 + 0.416 \text{XO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R135a	$\text{HC8P} + ^{15}\text{NO} \rightarrow 0.15 \text{ALD} + 0.642 \text{KET} + 0.133 \text{ETHP} + 0.261 ^{15}\text{ONIT} + 0.739 ^{15}\text{NO}_2 + 0.606 \text{HO}_2 + 0.416 \text{XO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R136	$\text{ETEP} + \text{NO} \rightarrow 1.6 \text{HCHO} + \text{HO}_2 + \text{NO}_2 + 0.2 \text{ALD}$	$9.0 \times 10^{-12}$		$9.0 \times 10^{-12}$	1
R136a	$\text{ETEP} + ^{15}\text{NO} \rightarrow 1.6 \text{HCHO} + \text{HO}_2 + ^{15}\text{NO}_2 + 0.2 \text{ALD}$	$9.0 \times 10^{-12}$		$9.0 \times 10^{-12}$	1
R137	$\text{OLTP} + \text{NO} \rightarrow 0.94 \text{ALD} + \text{HCHO} + \text{HO}_2 + \text{NO}_2 + 0.06 \text{KET}$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R137a	$\text{OLTP} + ^{15}\text{NO} \rightarrow 0.94 \text{ALD} + \text{HCHO} + \text{HO}_2 + ^{15}\text{NO}_2 + 0.06 \text{KET}$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R138	$\text{OLIP} + \text{NO} \rightarrow \text{HO}_2 + 1.71 \text{ALD} + 0.29 \text{KET} + \text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R138a	$\text{OLIP} + ^{15}\text{NO} \rightarrow \text{HO}_2 + 1.71 \text{ALD} + 0.29 \text{KET} + ^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R139	$\text{ISOP} + \text{NO} \rightarrow 0.446 \text{MACR} + 0.354 \text{OLT} + 0.847 \text{HO}_2 + 0.606 \text{HCHO} + 0.153 \text{ONIT} + 0.847 \text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1

R139a	ISOP + $^{15}\text{NO} \rightarrow 0.446$ MACR + 0.354 OLT + 0.847 $\text{HO}_2 + 0.606 \text{HCHO} + 0.153$ $^{15}\text{ONIT} + 0.847 \text{ }^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R140	APIP + NO $\rightarrow 0.80 \text{HO}_2 +$ 0.80 ALD + 0.80 KET + 0.20 ONIT + 0.80 NO <sub>2</sub>	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R140a	APIP + $^{15}\text{NO} \rightarrow 0.80 \text{HO}_2 +$ 0.80 ALD + 0.80 KET + 0.20 $^{15}\text{ONIT} + 0.80 \text{ }^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R141	LIMP + NO $\rightarrow 0.65 \text{HO}_2 +$ 0.40 MACR + 0.25 OLI + 0.25 HCHO + 0.35 ONIT + 0.65 NO <sub>2</sub>	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R141a	LIMP + $^{15}\text{NO} \rightarrow 0.65 \text{HO}_2 +$ 0.40 MACR + 0.25 OLI + 0.25 HCHO + 0.35 $^{15}\text{ONIT} +$ 0.65 $^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R142	TOLP + NO $\rightarrow 0.95 \text{NO}_2 +$ 0.95 HO <sub>2</sub> + 0.65 MGLY + 1.20 GLY + 0.50 DCB + 0.05 ONIT	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R142a	TOLP + $^{15}\text{NO} \rightarrow 0.95 \text{ }^{15}\text{NO}_2$ + 0.95 HO <sub>2</sub> + 0.65 MGLY + 1.20 GLY + 0.50 DCB + 0.05 $^{15}\text{ONIT}$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R143	XYLP + NO $\rightarrow 0.95 \text{NO}_2 +$ 0.95 HO <sub>2</sub> + 0.60 MGLY + 0.35 GLY + 0.95 DCB + 0.05 ONIT	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R143a	XYLP + $^{15}\text{NO} \rightarrow 0.95 \text{ }^{15}\text{NO}_2$ + 0.95 HO <sub>2</sub> + 0.60 MGLY + 0.35 GLY + 0.95 DCB + 0.05 $^{15}\text{ONIT}$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1

R144	$\text{CSLP} + \text{NO} \rightarrow \text{GLY} + \text{MGLY} + \text{HO}_2 + \text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R144a	$\text{CSLP} + {}^{15}\text{NO} \rightarrow \text{GLY} + \text{MGLY} + \text{HO}_2 + {}^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R145	$\text{ACO}_3 + \text{NO} \rightarrow \text{MO}_2 + \text{NO}_2$	$2.0 \times 10^{-11}$		$2.0 \times 10^{-11}$	1
R145a	$\text{ACO}_3 + {}^{15}\text{NO} \rightarrow \text{MO}_2 + {}^{15}\text{NO}_2$	$2.0 \times 10^{-11}$		$2.0 \times 10^{-11}$	1
R146	$\text{TCO}_3 + \text{NO} \rightarrow \text{ACO}_3 + \text{HCHO} + \text{NO}_2$	$2.0 \times 10^{-11}$		$2.0 \times 10^{-11}$	1
R146a	$\text{TCO}_3 + {}^{15}\text{NO} \rightarrow \text{ACO}_3 + \text{HCHO} + {}^{15}\text{NO}_2$	$2.0 \times 10^{-11}$		$2.0 \times 10^{-11}$	1
R147	$\text{KETP} + \text{NO} \rightarrow 0.54 \text{ MGLY} + 0.46 \text{ ALD} + 0.23 \text{ ACO}_3 + 0.77 \text{ HO}_2 + 0.16 \text{ XO}_2 + \text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R147a	$\text{KETP} + {}^{15}\text{NO} \rightarrow 0.54 \text{ MGLY} + 0.46 \text{ ALD} + 0.23 \text{ ACO}_3 + 0.77 \text{ HO}_2 + 0.16 \text{ XO}_2 + {}^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R148	$\text{OLNN} + \text{NO} \rightarrow \text{HO}_2 + \text{ONIT} + \text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R148a	${}^{15}\text{OLNN} + \text{NO} \rightarrow \text{HO}_2 + 0.5 {}^{15}\text{ONIT} + 0.5 \text{ NO}_2 + 0.5 \text{ ONIT} + 0.5 {}^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R148b	$\text{OLNN} + {}^{15}\text{NO} \rightarrow \text{HO}_2 + 0.5 \text{ ONIT} + 0.5 {}^{15}\text{NO}_2 + 0.5 {}^{15}\text{ONIT} + 0.5 \text{ NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R148c	${}^{15}\text{OLNN} + {}^{15}\text{NO} \rightarrow \text{HO}_2 + {}^{15}\text{ONIT} + {}^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R149	$\text{OLND} + \text{NO} \rightarrow 0.287 \text{ HCHO} + 1.24 \text{ ALD} + 0.464 \text{ KET} + 2 \text{ NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R149a	$\text{OLND} + {}^{15}\text{NO} \rightarrow 0.287 \text{ HCHO} + 1.24 \text{ ALD} + 0.464 \text{ KET} + \text{NO}_2 + {}^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1

R149b	$^{15}\text{OLND} + \text{NO} \rightarrow 0.287$ $\text{HCHO} + 1.24 \text{ ALD} + 0.464$ $\text{KET} + \text{NO}_2 + ^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R149c	$^{15}\text{OLND} + ^{15}\text{NO} \rightarrow 0.287$ $\text{HCHO} + 1.24 \text{ ALD} + 0.464$ $\text{KET} + 2 ^{15}\text{NO}_2$	$4.0 \times 10^{-12}$		$4.0 \times 10^{-12}$	1
R169	$\text{OLNN} + \text{HO}_2 \rightarrow \text{ONIT}$	$1.66 \times 10^{-13}$	-1300	$1.30 \times 10^{-11}$	1
R169a	$^{15}\text{OLNN} + \text{HO}_2 \rightarrow ^{15}\text{ONIT}$	$1.66 \times 10^{-13}$	-1300	$1.30 \times 10^{-11}$	1
R170	$\text{OLND} + \text{HO}_2 \rightarrow \text{ONIT}$	$1.66 \times 10^{-13}$	-1300	$1.30 \times 10^{-11}$	1
R170a	$^{15}\text{OLND} + \text{HO}_2 \rightarrow ^{15}\text{ONIT}$	$1.66 \times 10^{-13}$	-1300	$1.30 \times 10^{-11}$	1
R190	$\text{OLNN} + \text{MO}_2 \rightarrow 0.75 \text{ HCHO}$ + $\text{HO}_2 + \text{ONIT}$	$1.60 \times 10^{-13}$	-708	$1.72 \times 10^{-12}$	1
R190a	$^{15}\text{OLNN} + \text{MO}_2 \rightarrow 0.75$ $\text{HCHO} + \text{HO}_2 + ^{15}\text{ONIT}$	$1.60 \times 10^{-13}$	-708	$1.72 \times 10^{-12}$	1
R191	$\text{OLND} + \text{MO}_2 \rightarrow 0.96 \text{ HCHO}$ + $0.5 \text{ HO}_2 + 0.64 \text{ ALD} +$ $0.149 \text{ KET} + 0.5 \text{ NO}_2 + 0.5$ $\text{ONIT}$	$9.68 \times 10^{-14}$	-708	$1.04 \times 10^{-12}$	1
R191a	$^{15}\text{OLND} + \text{MO}_2 \rightarrow 0.96$ $\text{HCHO} + 0.5 \text{ HO}_2 + 0.64 \text{ ALD} +$ + $0.149 \text{ KET} + 0.5 ^{15}\text{NO}_2 +$ $0.5 ^{15}\text{ONIT}$	$9.68 \times 10^{-14}$	-708	$1.04 \times 10^{-12}$	1
R208	$\text{OLNN} + \text{ACO}_3 \rightarrow \text{ONIT} +$ 0.5 $\text{ORA2} + 0.5 \text{ MO}_2 + 0.50$ $\text{HO}_2$	$8.85 \times 10^{-13}$	-765	$1.15 \times 10^{-11}$	1
R208a	$^{15}\text{OLNN} + \text{ACO}_3 \rightarrow ^{15}\text{ONIT}$ + 0.5 $\text{ORA2} + 0.5 \text{ MO}_2 + 0.50$ $\text{HO}_2$	$8.85 \times 10^{-13}$	-765	$1.15 \times 10^{-11}$	1

R209	OLND + ACO <sub>3</sub> --> 0.207 HCHO + 0.65 ALD + 0.167 KET + 0.484 ORA2 + 0.484 ONIT + 0.516 NO <sub>2</sub> + 0.516 MO <sub>2</sub>	5.37 x 10 <sup>-13</sup>	-765	7.00 x 10 <sup>-12</sup>	1
R209a	<sup>15</sup> OLND + ACO <sub>3</sub> --> 0.207 HCHO + 0.65 ALD + 0.167 KET + 0.484 ORA2 + 0.484 <sup>15</sup> ONIT + 0.516 <sup>15</sup> NO <sub>2</sub> + 0.516 MO <sub>2</sub>	5.37 x 10 <sup>-13</sup>	-765	7.00 x 10 <sup>-12</sup>	1
R210	OLNN + OLNN --> 2 ONIT + HO <sub>2</sub>	7.0 x 10 <sup>-14</sup>	-1000	2.00 x 10 <sup>-12</sup>	1
R210a	OLNN + <sup>15</sup> OLNN --> ONIT + <sup>15</sup> ONIT + HO <sub>2</sub>	7.0 x 10 <sup>-14</sup>	-1000	2.00 x 10 <sup>-12</sup>	1
R210b	OLNN + <sup>15</sup> OLNN --> 2 <sup>15</sup> ONIT + HO <sub>2</sub>	7.0 x 10 <sup>-14</sup>	-1000	2.00 x 10 <sup>-12</sup>	1
R211	OLNN + OLND --> 0.202 HCHO + 0.64 ALD + 0.149 KET + 0.50 HO <sub>2</sub> + 1.50 ONIT + 0.50 NO <sub>2</sub>	4.25 x 10 <sup>-14</sup>	-1000	1.22 x 10 <sup>-12</sup>	1
R211a	<sup>15</sup> OLNN + OLND --> 0.202 HCHO + 0.64 ALD + 0.149 KET + 0.50 HO <sub>2</sub> + 0.75 ONIT + 0.75 <sup>15</sup> ONIT + 0.25 NO <sub>2</sub> + 0.25 <sup>15</sup> NO <sub>2</sub>	4.25 x 10 <sup>-14</sup>	-1000	1.22 x 10 <sup>-12</sup>	1
R211b	OLNN + <sup>15</sup> OLND --> 0.202 HCHO + 0.64 ALD + 0.149 KET + 0.50 HO <sub>2</sub> + 0.75 ONIT + 0.75 <sup>15</sup> ONIT + 0.25 NO <sub>2</sub> + 0.25 <sup>15</sup> NO <sub>2</sub>	4.25 x 10 <sup>-14</sup>	-1000	1.22 x 10 <sup>-12</sup>	1
R211c	<sup>15</sup> OLNN + <sup>15</sup> OLND --> 0.202 HCHO + 0.64 ALD + 0.149 KET + 0.50 HO <sub>2</sub> + 1.50 <sup>15</sup> ONIT + 0.50 <sup>15</sup> NO <sub>2</sub>	4.25 x 10 <sup>-14</sup>	-1000	1.22 x 10 <sup>-12</sup>	1

R212	OLND + OLND --> 0.504 HCHO + 1.21 ALD + 0.285 KET + ONIT + NO <sub>2</sub>	2.96 x 10 <sup>-14</sup>	-1000	8.50 x 10 <sup>-13</sup>	1
R212a	OLND + <sup>15</sup> OLND --> 0.504 HCHO + 1.21 ALD + 0.285 KET + <sup>15</sup> ONIT + NO <sub>2</sub>	2.96 x 10 <sup>-14</sup>	-1000	8.50 x 10 <sup>-13</sup>	1
R212b	OLND + <sup>15</sup> OLND --> 0.504 HCHO + 1.21 ALD + 0.285 KET + ONIT + <sup>15</sup> NO <sub>2</sub>	2.96 x 10 <sup>-14</sup>	-1000	8.50 x 10 <sup>-13</sup>	1
R212c	<sup>15</sup> OLND + <sup>15</sup> OLND --> 0.504 HCHO + 1.21 ALD + 0.285 KET + <sup>15</sup> ONIT + <sup>15</sup> NO <sub>2</sub>	2.96 x 10 <sup>-14</sup>	-1000	8.50 x 10 <sup>-13</sup>	1
R213	MO <sub>2</sub> + NO <sub>3</sub> --> HCHO + HO <sub>2</sub> + NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R213a	MO <sub>2</sub> + <sup>15</sup> NO <sub>3</sub> --> HCHO + HO <sub>2</sub> + <sup>15</sup> NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R214	ETHP + NO <sub>3</sub> --> ALD + HO <sub>2</sub> + NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R214a	ETHP + <sup>15</sup> NO <sub>3</sub> --> ALD + HO <sub>2</sub> + <sup>15</sup> NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R215	HC3P + NO <sub>3</sub> --> 0.048 HCHO + 0.243 ALD + 0.67 KET + 0.063 GLY + 0.792 HO <sub>2</sub> + 0.155 MO <sub>2</sub> + 0.053 ETHP + 0.051 XO <sub>2</sub> + NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R215a	HC3P + <sup>15</sup> NO <sub>3</sub> --> 0.048 HCHO + 0.243 ALD + 0.67 KET + 0.063 GLY + 0.792 HO <sub>2</sub> + 0.155 MO <sub>2</sub> + 0.053 ETHP + 0.051 XO <sub>2</sub> + <sup>15</sup> NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1
R216	HC5P + NO <sub>3</sub> --> 0.021 HCHO + 0.239 ALD + 0.828 KET + 0.699 HO <sub>2</sub> + 0.04 MO <sub>2</sub> + 0.262 ETHP + 0.391 XO <sub>2</sub> + NO <sub>2</sub>	1.20 x 10 <sup>-12</sup>		1.20 x 10 <sup>-12</sup>	1

R216a	$\text{HC5P} + ^{15}\text{NO}_3 \rightarrow 0.021 \text{HCHO} + 0.239 \text{ALD} + 0.828 \text{KET} + 0.699 \text{HO}_2 + 0.04 \text{MO}_2 + 0.262 \text{ETHP} + 0.391 \text{XO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R217	$\text{HC8P} + \text{NO}_3 \rightarrow 0.187 \text{ALD} + 0.88 \text{KET} + 0.845 \text{HO}_2 + 0.155 \text{ETHP} + 0.587 \text{XO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R217a	$\text{HC8P} + ^{15}\text{NO}_3 \rightarrow 0.187 \text{ALD} + 0.88 \text{KET} + 0.845 \text{HO}_2 + 0.155 \text{ETHP} + 0.587 \text{XO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R218	$\text{ETEP} + \text{NO}_3 \rightarrow 1.6 \text{HCHO} + 0.2 \text{ALD} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R218a	$\text{ETEP} + ^{15}\text{NO}_3 \rightarrow 1.6 \text{HCHO} + 0.2 \text{ALD} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R219	$\text{OLTP} + \text{NO}_3 \rightarrow \text{HCHO} + 0.94 \text{ALD} + 0.06 \text{KET} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R219a	$\text{OLTP} + ^{15}\text{NO}_3 \rightarrow \text{HCHO} + 0.94 \text{ALD} + 0.06 \text{KET} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R220	$\text{OLIP} + \text{NO}_3 \rightarrow 1.71 \text{ALD} + 0.29 \text{KET} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R220a	$\text{OLIP} + ^{15}\text{NO}_3 \rightarrow 1.71 \text{ALD} + 0.29 \text{KET} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R221	$\text{ISOP} + \text{NO}_3 \rightarrow 0.60 \text{MACR} + 0.40 \text{OLT} + 0.686 \text{HCHO} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R221a	$\text{ISOP} + ^{15}\text{NO}_3 \rightarrow 0.60 \text{MACR} + 0.40 \text{OLT} + 0.686 \text{HCHO} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R222	$\text{APIP} + \text{NO}_3 \rightarrow \text{ALD} + \text{KET} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1

R222a	$\text{APIP} + ^{15}\text{NO}_3 \rightarrow \text{ALD} + \text{KET} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R223	$\text{LIMP} + \text{NO}_3 \rightarrow 0.60 \text{ MACR} + 0.40 \text{ OLI} + 0.40 \text{ HCHO} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R223a	$\text{LIMP} + ^{15}\text{NO}_3 \rightarrow 0.60 \text{ MACR} + 0.40 \text{ OLI} + 0.40 \text{ HCHO} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R224	$\text{TOLP} + \text{NO}_3 \rightarrow 0.70 \text{ MGLY} + 1.30\text{GLY} + 0.50 \text{ DCB} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R224a	$\text{TOLP} + ^{15}\text{NO}_3 \rightarrow 0.70 \text{ MGLY} + 1.30\text{GLY} + 0.50 \text{ DCB} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R225	$\text{XYLP} + \text{NO}_3 \rightarrow 1.26 \text{ MGLY} + 0.74 \text{ GLY} + \text{DCB} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R225a	$\text{XYLP} + ^{15}\text{NO}_3 \rightarrow 1.26 \text{ MGLY} + 0.74 \text{ GLY} + \text{DCB} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R226	$\text{CSLP} + \text{NO}_3 \rightarrow \text{GLY} + \text{MGLY} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R226a	$\text{CSLP} + ^{15}\text{NO}_3 \rightarrow \text{GLY} + \text{MGLY} + \text{HO}_2 + ^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R227	$\text{ACO}_3 + \text{NO}_3 \rightarrow \text{MO}_2 + \text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R227a	$\text{ACO}_3 + ^{15}\text{NO}_3 \rightarrow \text{MO}_2 + ^{15}\text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R228	$\text{TCO}_3 + \text{NO}_3 \rightarrow \text{HCHO} + \text{ACO}_3 + \text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R228a	$\text{TCO}_3 + ^{15}\text{NO}_3 \rightarrow \text{HCHO} + \text{ACO}_3 + ^{15}\text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R229	$\text{KETP} + \text{NO}_3 \rightarrow 0.54 \text{ MGLY} + 0.46 \text{ ALD} + 0.77 \text{ HO}_2 + 0.23 \text{ ACO}_3 + 0.16 \text{ XO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1

R229a	$\text{KETP} + {}^{15}\text{NO}_3 \rightarrow 0.54$ $\text{MGLY} + 0.46 \text{ ALD} + 0.77$ $\text{HO}_2 + 0.23 \text{ ACO}_3 + 0.16 \text{ XO}_2 + {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R230	$\text{OLNN} + \text{NO}_3 \rightarrow \text{ONIT} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R230a	$\text{OLNN} + {}^{15}\text{NO}_3 \rightarrow \text{ONIT} + \text{HO}_2 + {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R230b	${}^{15}\text{OLNN} + \text{NO}_3 \rightarrow {}^{15}\text{ONIT} + \text{HO}_2 + \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R230c	${}^{15}\text{OLNN} + {}^{15}\text{NO}_3 \rightarrow {}^{15}\text{ONIT} + \text{HO}_2 + {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R231	$\text{OLND} + \text{NO}_3 \rightarrow 0.28 \text{ HCHO} + 1.24 \text{ ALD} + 0.469 \text{ KET} + 2 \text{ NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R231a	${}^{15}\text{OLND} + \text{NO}_3 \rightarrow 0.28 \text{ HCHO} + 1.24 \text{ ALD} + 0.469 \text{ KET} + \text{NO}_2 + {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R231b	$\text{OLND} + {}^{15}\text{NO}_3 \rightarrow 0.28 \text{ HCHO} + 1.24 \text{ ALD} + 0.469 \text{ KET} + \text{NO}_2 + {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R231c	${}^{15}\text{OLND} + {}^{15}\text{NO}_3 \rightarrow 0.28 \text{ HCHO} + 1.24 \text{ ALD} + 0.469 \text{ KET} + 2 {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R236	$\text{XO}_2 + \text{NO} \rightarrow \text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R236a	$\text{XO}_2 + {}^{15}\text{NO} \rightarrow {}^{15}\text{NO}_2$	$4.00 \times 10^{-12}$		$4.00 \times 10^{-12}$	1
R237	$\text{XO}_2 + \text{NO}_3 \rightarrow \text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R237a	$\text{XO}_2 + {}^{15}\text{NO}_3 \rightarrow {}^{15}\text{NO}_2$	$1.20 \times 10^{-12}$		$1.20 \times 10^{-12}$	1
R238	$\text{NO} + {}^{15}\text{NO}_2 \rightarrow {}^{15}\text{NO} + \text{NO}_2$	$3.60 \times 10^{-14}$		$3.60 \times 10^{-14}$	1
R238a	${}^{15}\text{NO} + \text{NO}_2 \rightarrow \text{NO} + {}^{15}\text{NO}_2$	$3.60 \times 10^{-14}$	-18.467	$3.83 \times$	0.9771

				$10^{-14}$	
R239	$\text{N}_2\text{O}_5 \rightarrow \text{HNO}_3 + \text{HNO}_3$	0.1		0.1	1
R239a	$^{15}\text{NNO}_5 \rightarrow ^{15}\text{HNO}_3 + \text{HNO}_3$	0.1		0.1	0.9954
R239b	$^{15}\text{N}_2\text{O}_5 \rightarrow ^{15}\text{HNO}_3 + ^{15}\text{HNO}_3$	0.1		0.1	0.9909

Table S2b: Chemical reactions involving N compounds

Reaction No.	Reaction	C, K <sup>-2</sup> cm <sup>3</sup> s <sup>-1</sup>	D, K	$\alpha$
R98	ETE + NO <sub>3</sub> --> 0.80 OLNN + 0.20 OLND	4.88 x 10 <sup>-18</sup>	2282	1
R98a	ETE + <sup>15</sup> NO <sub>3</sub> --> 0.80 <sup>15</sup> OLNN + 0.20 <sup>15</sup> OLND	4.88 x 10 <sup>-18</sup>	2282	0.9975

Table S2c: Reaction rate constants of the form  $k = T^2 C \exp(-D/T)$

Reaction No.	Reaction	$K_0^{300}$ , cm <sup>6</sup> s <sup>-1</sup>	n	$K_\infty^{300}$ , cm <sup>6</sup> s <sup>-1</sup>	m
R35	O <sup>3</sup> P + NO --> NO <sub>2</sub>	9.00 x 10 <sup>-32</sup>	1.5	3.00 x 10 <sup>-11</sup>	0
R35a	O <sup>3</sup> P + <sup>15</sup> NO --> <sup>15</sup> NO <sub>2</sub>	9.00 x 10 <sup>-32</sup>	1.5	3.00 x 10 <sup>-11</sup>	0
R37	O <sup>3</sup> P + NO <sub>2</sub> --> NO <sub>3</sub>	9.00 x 10 <sup>-32</sup>	2	2.20 x 10 <sup>-11</sup>	0
R37a	O <sup>3</sup> P + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> NO <sub>3</sub>	9.00 x 10 <sup>-32</sup>	2	2.20 x 10 <sup>-11</sup>	0
R38	HO + NO --> HONO	7.00 x 10 <sup>-31</sup>	2.6	1.50 x 10 <sup>-11</sup>	0.5
R38a	HO + <sup>15</sup> NO --> HO <sup>15</sup> NO	7.00 x 10 <sup>-31</sup>	2.6	1.50 x 10 <sup>-11</sup>	0.5
R39	HO + NO <sub>2</sub> --> HNO <sub>3</sub>	2.60 x 10 <sup>-30</sup>	3.2	2.40 x 10 <sup>-11</sup>	1.3
R39a	HO + <sup>15</sup> NO <sub>2</sub> --> H <sup>15</sup> NO <sub>3</sub>	2.60 x 10 <sup>-30</sup>	3.2	2.40 x 10 <sup>-11</sup>	1.3
R42	HO <sub>2</sub> + NO <sub>2</sub> --> HNO <sub>4</sub>	2.80 x 10 <sup>-31</sup>	3.2	4.70 x 10 <sup>-12</sup>	1.4
R42a	HO <sub>2</sub> + <sup>15</sup> NO <sub>2</sub> --> H <sup>15</sup> NO <sub>4</sub>	2.80 x 10 <sup>-31</sup>	3.2	4.70 x 10 <sup>-12</sup>	1.4
R53	NO <sub>3</sub> + NO <sub>2</sub> --> N <sub>2</sub> O <sub>5</sub>	2.20 x 10 <sup>-30</sup>	3.9	1.50 x 10 <sup>-12</sup>	0.7
R53a	NO <sub>3</sub> + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> NNO <sub>5</sub>	2.20 x 10 <sup>-30</sup>	3.9	1.50 x 10 <sup>-12</sup>	0.7
R53b	<sup>15</sup> NO <sub>3</sub> + NO <sub>2</sub> --> <sup>15</sup> NNO <sub>5</sub>	2.20 x 10 <sup>-30</sup>	3.9	1.50 x 10 <sup>-12</sup>	0.7
R53c	<sup>15</sup> NO <sub>3</sub> + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> N <sub>2</sub> O <sub>5</sub>	2.20 x 10 <sup>-30</sup>	3.9	1.50 x 10 <sup>-12</sup>	0.7
R127	ACO <sub>3</sub> + NO <sub>2</sub> --> PAN	9.70 x 10 <sup>-29</sup>	5.6	9.30 x 10 <sup>-12</sup>	1.5
R127a	ACO <sub>3</sub> + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> PAN	9.70 x 10 <sup>-29</sup>	5.6	9.30 x 10 <sup>-12</sup>	1.5
R129	TCO <sub>3</sub> + NO <sub>2</sub> --> TPAN	9.70 x 10 <sup>-29</sup>	5.6	9.30 x 10 <sup>-12</sup>	1.5
R129a	TCO <sub>3</sub> + <sup>15</sup> NO <sub>2</sub> --> <sup>15</sup> TPAN	9.70 x 10 <sup>-29</sup>	5.6	9.30 x 10 <sup>-12</sup>	1.5

Table S2d: Troe reactions

Reaction No.	Reaction	A	B	$K_0^{300}, \text{cm}^6 \text{s}^{-1}$	n	$K_\infty^{300}, \text{cm}^6 \text{s}^{-1}$	m
R43	$\text{HNO}_4 \rightarrow \text{HO}_2 + \text{NO}_2$	$4.76 \times 10^{+26}$	10900	$1.81 \times 10^{-31}$	3.2	$4.70 \times 10^{-12}$	1.4
R43a	$\text{H}^{15}\text{NO}_4 \rightarrow \text{HO}_2 + ^{15}\text{NO}_2$	$4.76 \times 10^{+26}$	10900	$1.81 \times 10^{-31}$	3.2	$4.70 \times 10^{-12}$	1.4
R54	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	$3.70 \times 10^{+26}$	11000	$2.20 \times 10^{-30}$	3.9	$1.50 \times 10^{-12}$	0.7
R54a	$^{15}\text{NNO}_5 \rightarrow ^{15}\text{NO}_2 + \text{NO}_3$	$3.70 \times 10^{+26}$	11000	$2.20 \times 10^{-30}$	3.9	$1.50 \times 10^{-12}$	0.7
R54b	$^{15}\text{NNO}_5 \rightarrow \text{NO}_2 + ^{15}\text{NO}_3$	$3.70 \times 10^{+26}$	11000	$2.20 \times 10^{-30}$	3.9	$1.50 \times 10^{-12}$	0.7
R54c	$^{15}\text{N}_2\text{O}_5 \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_3$	$3.70 \times 10^{+26}$	11000	$2.20 \times 10^{-30}$	3.9	$1.50 \times 10^{-12}$	0.7
R128	$\text{PAN} \rightarrow \text{ACO}_3 + \text{NO}_2$	$1.16 \times 10^{+28}$	13954	$9.70 \times 10^{-29}$	5.6	$9.30 \times 10^{-12}$	1.5
R128a	$^{15}\text{PAN} \rightarrow \text{ACO}_3 + ^{15}\text{NO}_2$	$1.16 \times 10^{+28}$	13954	$9.70 \times 10^{-29}$	5.6	$9.30 \times 10^{-12}$	1.5
R130	$\text{TPAN} \rightarrow \text{TCO}_3 + \text{NO}_2$	$1.16 \times 10^{+28}$	13954	$9.70 \times 10^{-29}$	5.6	$9.30 \times 10^{-12}$	1.5
R130a	$^{15}\text{TPAN} \rightarrow \text{TCO}_3 + ^{15}\text{NO}_2$	$1.16 \times 10^{+28}$	13954	$9.70 \times 10^{-29}$	5.6	$9.30 \times 10^{-12}$	1.5

Table S2e: Troe equilibrium reactions

Reaction No.	Reaction	Rate Constant Expression
R46	$\text{HO} + \text{HNO}_3 \rightarrow \text{NO}_3 + \text{H}_2\text{O}$	$k = k_0 + k_3/(1 + k_3/k_2)$ $k_0 = 7.2 \times 10^{-15} \times \exp(785/T)$ $k_2 = 4.1 \times 10^{-16} \times \exp(1440/T)$ $k_3 = 1.9 \times 10^{-33} \times \exp(725/T) \times [M]$ [M] is the concentration of air in molecules cm <sup>-3</sup>
R46a	$\text{HO} + \text{H}^{15}\text{NO}_3 \rightarrow {}^{15}\text{NO}_3 + \text{H}_2\text{O}$	

Table S2f: Reactions with special rate expressions

Compound	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1E+07	-
O <sub>3</sub>	10	-
NO	0.2	2.59
<sup>15</sup> NO	0.00072	9.32E-03
NO <sub>2</sub>	0.25	-
<sup>15</sup> NO <sub>2</sub>	0.0009	-
HNO <sub>3</sub>	-	-
H <sup>15</sup> NO <sub>3</sub>	-	-
CO	1000	5.6
CH <sub>4</sub>	3000	-
H <sub>2</sub>	500	-
H <sub>2</sub> O <sub>2</sub>	2	-
SO <sub>2</sub>	-	0.52
ETH	-	0.24
HC3	-	2.94
HC5	-	0.77
HC8	-	0.45
ETE	-	0.46
OLI	-	0.19
OLT	-	0.22
TOL	-	0.57
XYL	-	0.52
HCHO	1	0.14
ALD	-	0.04
KET	-	0.50
O <sub>2</sub>	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-

Table S3a: Initial concentrations and emission rates for test cases

Meteorological conditions	Values
Start Date/Time	Mar 1, 0300 LT
End Date/Time	Mar 6, 0000 LT
Latitude	33 °N
Longitude	0
Elevation, km	0
Temperature, K	298
Pressure, atm	1

Table S3b: Meteorological conditions for test cases

Compound	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1.00E+07	-
O <sub>3</sub>	10	-
NO	5	2.59
<sup>15</sup> NO	0.018	9.32E-03
NO <sub>2</sub>	10	-
<sup>15</sup> NO <sub>2</sub>	0.036	-
HNO <sub>3</sub>	-	-
H <sup>15</sup> NO <sub>3</sub>	-	-
CO	1000	5.6
CH <sub>4</sub>	3000	-
H <sub>2</sub>	500	-
H <sub>2</sub> O <sub>2</sub>	2	-
SO <sub>2</sub>	-	0.52
ETH	-	0.24
HC3	-	2.94
HC5	-	0.77
HC8	-	0.45
ETE	-	0.46
OLI	-	0.19
OLT	-	0.22
TOL	-	0.57
XYL	-	0.52
HCHO	1	0.14
ALD	-	0.04
KET	-	0.50
O <sub>2</sub>	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-

Table S3c: Initial concentrations and emission rates for the cases with low VOC emission rate

Compound	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1.00E+07	-
O <sub>3</sub>	10	-
NO	50	2.59
<sup>15</sup> NO	0.18	9.32E-03
NO <sub>2</sub>	100	-
<sup>15</sup> NO <sub>2</sub>	0.36	-
HNO <sub>3</sub>	-	-
H <sup>15</sup> NO <sub>3</sub>	-	-
CO	1000	5.6
CH <sub>4</sub>	3000	-
H <sub>2</sub>	500	-
H <sub>2</sub> O <sub>2</sub>	2	-
SO <sub>2</sub>	-	0.52
ETH	-	1.20
HC3	-	14.7
HC5	-	3.85
HC8	-	2.26
ETE	-	2.28
OLI	-	0.94
OLT	-	1.09
TOL	-	2.86
XYL	-	2.59
HCHO	1	0.69
ALD	-	0.18
KET	-	2.51
O <sub>2</sub>	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-

Table S3d: Initial concentrations and emission rates for the cases with high NO<sub>x</sub> concentration and high VOC emission rate

Compound	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1.00E+07	-
O <sub>3</sub>	10	-
NO	5	2.59
<sup>15</sup> NO	0.018	9.32E-03
NO <sub>2</sub>	10	-
<sup>15</sup> NO <sub>2</sub>	0.036	-
HNO <sub>3</sub>	-	-
H <sup>15</sup> NO <sub>3</sub>	-	-
CO	1000	5.6
CH <sub>4</sub>	3000	-
H <sub>2</sub>	500	-
H <sub>2</sub> O <sub>2</sub>	2	-
SO <sub>2</sub>	-	0.52
ETH	-	1.20
HC3	-	14.7
HC5	-	3.85
HC8	-	2.26
ETE	-	2.28
OLI	-	0.94
OLT	-	1.09
TOL	-	2.86
XYL	-	2.59
HCHO	1	0.69
ALD	-	0.18
KET	-	2.51
O <sub>2</sub>	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-

Table S3e: Initial concentrations and emission rates for the cases with intermediate NO<sub>x</sub> concentration and high VOC emission rate

Compound	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1.00E+07	-
O <sub>3</sub>	10	-
NO	0.5	2.59
<sup>15</sup> NO	0.0018	9.32E-03
NO <sub>2</sub>	1	-
<sup>15</sup> NO <sub>2</sub>	0.0036	-
HNO <sub>3</sub>	-	-
H <sup>15</sup> NO <sub>3</sub>	-	-
CO	1000	5.6
CH <sub>4</sub>	3000	-
H <sub>2</sub>	500	-
H <sub>2</sub> O <sub>2</sub>	2	-
SO <sub>2</sub>	-	0.52
ETH	-	1.20
HC3	-	14.7
HC5	-	3.85
HC8	-	2.26
ETE	-	2.28
OLI	-	0.94
OLT	-	1.09
TOL	-	2.86
XYL	-	2.59
HCHO	1	0.69
ALD	-	0.18
KET	-	2.51
O <sub>2</sub>	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-

Table S3f: Initial concentrations and emission rates for the cases with low NO<sub>x</sub> concentration and high VOC emission rate

	Rural									
Case	1	2	3	4	5	6	7	8	9	
H <sub>2</sub> O, %	1	1	1	1	1	1	1	1	1	
NO, ppb	0.1	0.1	0.1	0.334	0.334	0.334	1	1	1	
<sup>15</sup> NO, ppb	3.600E-04	3.600E-04	3.600E-04	1.202E-03	1.202E-03	1.202E-03	3.600E-03	3.600E-03	3.600E-03	
NO <sub>2</sub> , ppb	0.4	0.4	0.4	1.336	1.336	1.336	4	4	4	
<sup>15</sup> NO <sub>2</sub> , ppb	0.00144	0.00144	0.00144	0.00481	0.00481	0.00481	0.0144	0.0144	0.0144	
HNO <sub>3</sub> , ppb	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	
H <sup>15</sup> NO <sub>3</sub> , ppb	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	
CH <sub>4</sub> , ppb	1800	1800	1800	1800	1800	1800	1800	1800	1800	
CO, ppb	200	200	200	200	200	200	200	200	200	
SO <sub>2</sub> , ppb	5	5	5	5	5	5	5	5	5	
O <sub>3</sub> , ppb	10	10	10	10	10	10	10	10	10	
H <sub>2</sub> , ppb	500	500	500	500	500	500	500	500	500	
H <sub>2</sub> O <sub>2</sub> , ppb	2	2	2	2	2	2	2	2	2	
ETH, ppb	0.79	2.607	7.9	0.79	2.607	7.9	0.79	2.607	7.9	
HC3, ppb	0.6575	2.16975	6.575	0.6575	2.16975	6.575	0.6575	2.16975	6.575	
HC5, ppb	0.295	0.9735	2.95	0.295	0.9735	2.95	0.295	0.9735	2.95	
HC8, ppb	0.13	0.429	1.3	0.13	0.429	1.3	0.13	0.429	1.3	
ETE, ppb	0.3	0.99	3	0.3	0.99	3	0.3	0.99	3	
OLT, ppb	0.1	0.33	1	0.1	0.33	1	0.1	0.33	1	
OLI, ppb	0	0	0	0	0	0	0	0	0	
ISO, ppb	0.12	0.396	1.2	0.12	0.396	1.2	0.12	0.396	1.2	
TOL,	0.2667	0.88011	2.667	0.2667	0.88011	2.667	0.2667	0.88011	2.667	

ppb										
XYL, ppb	0.0375	0.12375	0.375	0.0375	0.12375	0.375	0.0375	0.12375	0.375	0.375
CSL, ppb	0	0	0	0	0	0	0	0	0	0
HCHO, ppb	0.1	0.33	1	0.1	0.33	1	0.1	0.33	1	1
ALD, ppb	0.05	0.165	0.5	0.05	0.165	0.5	0.05	0.165	0.5	0.5
O <sub>2</sub> , %	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9
N <sub>2</sub> , %	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394
<sup>15</sup> NN, %	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596
<sup>15</sup> N <sub>2</sub> , %	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010
Urban										
Case	10	11	12	13	14	15	16	17	18	
H <sub>2</sub> O, %	1	1	1	1	1	1	1	1	1	1
NO, ppb	8	8	8	26.4	26.4	26.4	80	80	80	
<sup>15</sup> NO, ppb	2.880E-02	2.880E-02	2.880E-02	9.504E-02	9.504E-02	9.504E-02	2.880E-01	2.880E-01	2.880E-01	
NO <sub>2</sub> , ppb	2	2	2	6.6	6.6	6.6	20	20	20	
<sup>15</sup> NO <sub>2</sub> , ppb	0.0072	0.0072	0.0072	0.02376	0.02376	0.02376	0.072	0.072	0.072	
HNO <sub>3</sub> , ppb	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1
H <sup>15</sup> NO <sub>3</sub> , ppb	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036	0.00036
CH <sub>4</sub> , ppb	3000	3000	3000	3000	3000	3000	3000	3000	3000	3000
CO, ppb	1000	1000	1000	1000	1000	1000	1000	1000	1000	1000
SO <sub>2</sub> , ppb	30	30	30	30	30	30	30	30	30	30
O <sub>3</sub> , ppb	10	10	10	10	10	10	10	10	10	10
H <sub>2</sub> , ppb	500	500	500	500	500	500	500	500	500	500
H <sub>2</sub> O <sub>2</sub> , ppb	2	2	2	2	2	2	2	2	2	2
ETH, ppb	2.75	9.075	27.5	2.75	9.075	27.5	2.75	9.075	27.5	27.5
HC3,	4.842	15.9786	48.42	4.842	15.9786	48.42	4.842	15.9786	48.42	

ppb									
HC5, ppb	3.043	10.0419	30.43	3.043	10.0419	30.43	3.043	10.0419	30.43
HC8, ppb	2.223	7.3359	22.23	2.223	7.3359	22.23	2.223	7.3359	22.23
ETE, ppb	3.5	11.55	35	3.5	11.55	35	3.5	11.55	35
OLT, ppb	2.667	8.8011	26.67	2.667	8.8011	26.67	2.667	8.8011	26.67
OLI, ppb	1.25	4.125	12.5	1.25	4.125	12.5	1.25	4.125	12.5
ISO, ppb	0	0	0	0	0	0	0	0	0
TOL, ppb	1.81	5.973	18.1	1.81	5.973	18.1	1.81	5.973	18.1
XYL, ppb	0.625	2.0625	6.25	0.625	2.0625	6.25	0.625	2.0625	6.25
CSL, ppb	0.333	1.0989	3.33	0.333	1.0989	3.33	0.333	1.0989	3.33
HCHO, ppb	2.5	8.25	25	2.5	8.25	25	2.5	8.25	25
ALD, ppb	1.25	4.125	12.5	1.25	4.125	12.5	1.25	4.125	12.5
O <sub>2</sub> , %	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9
N <sub>2</sub> , %	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394	77.4394
<sup>15</sup> NN, %	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596	0.5596
<sup>15</sup> N <sub>2</sub> , %	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010	0.0010

Table S4a: Initial concentrations for inter-comparison cases 1-18

Meteorological conditions	Values
Start Date/Time	June 21, 0600 LT
End Date/Time	June 23, 0600 LT
Latitude	40 °N
Longitude	0
Elevation, km	0
Temperature, K	298
Pressure, atm	1

Table S4b: Meteorological conditions for inter-comparison cases 1-18

	Case 19		Case 20	
Compound	Initial Concentrations, ppb	Emissions, ppt/min	Initial Concentrations, ppb	Emissions, ppt/min
H <sub>2</sub> O	1 x 10 <sup>7</sup>	-	1 x 10 <sup>7</sup>	-
O <sub>3</sub>	50	-	30	-
NO	0.2	2.6	0.02	0.26
<sup>15</sup> NO	7.20E-04	9.36E-03	7.20E-05	9.36E-04
NO <sub>2</sub>	0.5	-	0.05	-
<sup>15</sup> NO <sub>2</sub>	0.0018	-	0.00018	-
HNO <sub>3</sub>	0.1	-	0.01	-
H <sup>15</sup> NO <sub>3</sub>	0.00036	-	0.000036	-
CO	200	5.7	104	0.57
CH <sub>4</sub>	1700	-	1700	-
H <sub>2</sub>	500	-	500	-
H <sub>2</sub> O <sub>2</sub>	2	-	0.2	-
SO <sub>2</sub>	-	0.52	-	0.052
ETH	-	0.24	-	0.024
HC3	-	2.6	-	0.26
HC5	-	0.76	-	0.076
HC8	-	0.45	-	0.045
ETE	-	0.46	-	0.046
OLI	-	0.19	-	0.019
OLT	-	0.22	-	0.022
TOL	-	0.57	-	0.057
XYL	-	0.52	-	0.052
HCHO	1	0.14	0.1	0.014
ALD	-	0.036	-	0.0036
KET	-	0.32	-	0.032
O <sub>2</sub>	2.09E+08	-	2.09E+08	-
N <sub>2</sub>	7.74394109E+08	-	7.74394109E+08	-
<sup>15</sup> NN	5.59578240E+06	-	5.59578240E+06	-
<sup>15</sup> N <sub>2</sub>	1.01088000E+04	-	1.01088000E+04	-

Table S5a: Initial concentrations and emission rates for inter-comparison cases 19-20

Meteorological conditions	Values
Start Date/Time	June 21, 0600 LT
End Date/Time	June 26, 0600 LT
Latitude	40 °N
Longitude	0
Elevation, km	0
Temperature, K	298
Pressure, atm	1

Table S5b: Meteorological conditions for inter-comparison cases 19-20

Reaction No.	Reaction	$\alpha$
R1	$\text{NO}_2 \rightarrow \text{O}^3\text{P} + \text{NO}$	1
R1a	$^{15}\text{NO}_2 \rightarrow \text{O}^3\text{P} + ^{15}\text{NO}$	1.0042
R39	$\text{HO} + \text{NO}_2 \rightarrow \text{HNO}_3$	1
R39a	$\text{HO} + ^{15}\text{NO}_2 \rightarrow \text{H}^{15}\text{NO}_3$	1.04
R48	$\text{O}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{O}_2$	1
R48a	$\text{O}_3 + ^{15}\text{NO} \rightarrow ^{15}\text{NO}_2 + \text{O}_2$	0.9933
R91	$\text{HCHO} + \text{NO}_3 \rightarrow \text{HO}_2 + \text{HNO}_3 + \text{CO}$	1
R91a	$\text{HCHO} + ^{15}\text{NO}_3 \rightarrow \text{HO}_2 + \text{H}^{15}\text{NO}_3 + \text{CO}$	0.9974
R92	$\text{ALD} + \text{NO}_3 \rightarrow \text{ACO}_3 + \text{HNO}_3$	1
R92a	$\text{ALD} + ^{15}\text{NO}_3 \rightarrow \text{ACO}_3 + \text{H}^{15}\text{NO}_3$	0.9967
R93	$\text{GLY} + \text{NO}_3 \rightarrow \text{HNO}_3 + \text{HO}_2 + 2 \text{CO}$	1
R93a	$\text{GLY} + ^{15}\text{NO}_3 \rightarrow \text{H}^{15}\text{NO}_3 + \text{HO}_2 + 2 \text{CO}$	0.9962
R94	$\text{MGLY} + \text{NO}_3 \rightarrow \text{HNO}_3 + \text{ACO}_3 + \text{CO}$	1
R94a	$\text{MGLY} + ^{15}\text{NO}_3 \rightarrow \text{H}^{15}\text{NO}_3 + \text{ACO}_3 + \text{CO}$	0.9957
R95	$\text{MACR} + \text{NO}_3 \rightarrow 0.20 \text{TCO}_3 + 0.20 \text{HNO}_3 + 0.80 \text{OLNN} + 0.80 \text{CO}$	1
R95a	$\text{MACR} + ^{15}\text{NO}_3 \rightarrow 0.20 \text{TCO}_3 + 0.20 \text{H}^{15}\text{NO}_3 + 0.80 ^{15}\text{OLNN} + 0.80 \text{CO}$	0.9958
R96	$\text{DCB} + \text{NO}_3 \rightarrow 0.50 \text{TCO}_3 + 0.50 \text{HO}_2 + 0.50 \text{XO}_2 + 0.25 \text{GLY} + 0.25 \text{ALD} + 0.03 \text{KET} + 0.25 \text{MGLY} + 0.5 \text{HNO}_3 + 0.5 \text{NO}_2$	1
R96a	$\text{DCB} + ^{15}\text{NO}_3 \rightarrow 0.50 \text{TCO}_3 + 0.50 \text{HO}_2 + 0.50 \text{XO}_2 + 0.25 \text{GLY} + 0.25 \text{ALD} + 0.03 \text{KET} + 0.25 \text{MGLY} + 0.5 \text{H}^{15}\text{NO}_3 + 0.5 ^{15}\text{NO}_2$	0.9954
R97	$\text{CSL} + \text{NO}_3 \rightarrow \text{HNO}_3 + \text{PHO}$	1
R97a	$\text{CSL} + ^{15}\text{NO}_3 \rightarrow \text{H}^{15}\text{NO}_3 + \text{PHO}$	0.9949
R238	$\text{NO} + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NO} + \text{NO}_2$	1
R238a	$^{15}\text{NO} + \text{NO}_2 \rightarrow \text{NO} + ^{15}\text{NO}_2$	0.9771
Table S6: Fractionation factors of Leighton cycle, $\text{NO}_x$ isotope exchange, OH production of $\text{HNO}_3$ , and KIE effects of $\text{NO}_3$ reacting with hydrocarbons.		

Reaction No.	Reaction	$\alpha$
R1	$\text{NO}_2 \rightarrow \text{O}^3\text{P} + \text{NO}$	1
R1a	$^{15}\text{NO}_2 \rightarrow \text{O}^3\text{P} + ^{15}\text{NO}$	1.0042
R39	$\text{HO} + \text{NO}_2 \rightarrow \text{HNO}_3$	1
R39a	$\text{HO} + ^{15}\text{NO}_2 \rightarrow \text{H}^{15}\text{NO}_3$	1.04
R48	$\text{O}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{O}_2$	1
R48a	$\text{O}_3 + ^{15}\text{NO} \rightarrow ^{15}\text{NO}_2 + \text{O}_2$	0.9933
R53	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{N}_2\text{O}_5$	1
R53a	$\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NNO}_5$	1.0266
R53b	$^{15}\text{NO}_3 + \text{NO}_2 \rightarrow ^{15}\text{NNO}_5$	1.0309
R53c	$^{15}\text{NO}_3 + ^{15}\text{NO}_2 \rightarrow ^{15}\text{N}_2\text{O}_5$	1.057
R54	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	1
R54a	$^{15}\text{NNO}_5 \rightarrow ^{15}\text{NO}_2 + \text{NO}_3$	0.5
R54b	$^{15}\text{NNO}_5 \rightarrow \text{NO}_2 + ^{15}\text{NO}_3$	0.5
R54c	$^{15}\text{N}_2\text{O}_5 \rightarrow ^{15}\text{NO}_2 + ^{15}\text{NO}_3$	1
R238	$\text{NO} + ^{15}\text{NO}_2 \rightarrow ^{15}\text{NO} + \text{NO}_2$	1
R238a	$^{15}\text{NO} + \text{NO}_2 \rightarrow \text{NO} + ^{15}\text{NO}_2$	0.9771
R239	$\text{N}_2\text{O}_5 \rightarrow \text{HNO}_3 + \text{HNO}_3$	1
R239a	$^{15}\text{NNO}_5 \rightarrow ^{15}\text{HNO}_3 + \text{HNO}_3$	0.9954
R239b	$^{15}\text{N}_2\text{O}_5 \rightarrow ^{15}\text{HNO}_3 + ^{15}\text{HNO}_3$	0.9909
Table S7: Fractionation factors of Leighton cycle, $\text{NO}_x$ isotope exchange, OH production of $\text{HNO}_3$ , and $\text{N}_2\text{O}_5$ heterogeneous reactions.		

	Urban		Rural		Forest		Marine	
Compound	Initial Concentrations ppb	Emissions ppt/min						
H <sub>2</sub> O	1 x 10 <sup>7</sup>	-						
O <sub>3</sub>	300	-	50	-	10	-	10	-
NO	90	156	4.9	3.9	0.054	0.156	0.027	0.208
<sup>15</sup> NO	0.324	0.5616	0.01764	0.01404	1.944E-04	0.0005616	9.720E-05	0.0007488
NO <sub>2</sub>	90	-	2.1	-	0.006	-	0.003	-
<sup>15</sup> NO <sub>2</sub>	0.324	-	0.00756	-	2.160E-05	-	1.080E-05	-
HNO <sub>3</sub>	20	-	6	-	0.29	-	0.145	-
H <sup>15</sup> NO <sub>3</sub>	0.072	-	0.0216	-	1.044E-03	-	5.22E-04	-
PAN	20	-	3	-	0.05	-	0.025	-
<sup>15</sup> PAN	0.072	-	0.0108	-	1.800E-04	-	9.00E-05	-
CO	1000	5.7	200	5.7	207	5.7	-	5.7
CH <sub>4</sub>	3000	-	1800	-	1657	-	-	-
H <sub>2</sub>	500	-	500	-	500	-	500	-
H <sub>2</sub> O <sub>2</sub>	2	-	2	-	2	-	2	-
SO <sub>2</sub>	30	0.52	5	0.52	-	0.52	-	0.52
ETH	30.730	4.8	2.557	0.36	0.98	0.216	1.304	0.144
HC3	53.777	52	1.147	3.9	0.84	2.34	1.306	1.56
HC5	33.803	15.2	1.368	1.14	0.344	0.684	0.199	0.456
HC8	24.694	9	0.259	0.675	0.1	0.405	0.025	0.27
ETE	17.986	9.2	0.100	0.69	0.97	0.414	1.469	0.276
OLI	3.063	3.8	-	0.285	-	0.171	-	0.114
OLT	6.535	4.4	0.066	0.33	0.39	0.198	0.404	0.132
ISO	-	-	0.082	-	2.040	-	-	-
TOL	17.667	11.4	0.372	0.855	0.93	0.513	0.545	0.342
XYL	15.802	10.4	0.195	0.78	0.090	0.468	0.140	0.312
HCHO	22.610	2.8	1.748	0.21	-	0.126	-	0.084
ALD	25.694	0.72	0.368	0.054	-	0.0324	-	0.0216
KET	-	6.4	-	0.48	-	0.288	-	0.192
API	-	-	-	-	0.130	-	-	-
O <sub>2</sub>	2.09E+08	-	2.09E+08	-	2.09E+08	-	2.09E+08	-

$\text{N}_2$	7.7439E+08	-	7.7439E+08	-	7.7439E+08	-	7.7439E+08	-
$^{15}\text{NN}$	5.5958E+06	-	5.5958E+06	-	5.5958E+06	-	5.5958E+06	-
$^{15}\text{N}_2$	1.0109E+04	-	1.0109E+04	-	1.0109E+04	-	1.0109E+04	-

Table S8: Initial concentrations and emission rates for test cases under different environments

## 2. Detail information for case studies inputs

The completed *i*<sub>N</sub>RACM was tested with different cases with various initial concentrations and emission rates, which stands for urban, rural, forest, and marine conditions. Stockwell (1997) provided 18 test cases without emission (9 for urban condition, 9 for rural condition), and 2 test cases with emission (1 for polluted atmosphere, 1 for clean atmosphere), while the concentration of chemicals in these test cases does not agree with atmospheric observations [Altshuller, 1989; Baugues, 1986; Greenberg & Zimmerman, 1984; Logan, 1989; National Research Council, 1992; Torres & Buchan, 1988; Zimmerman et al., 1988].

First, Stockwell used the exact same initial ozone concentration of 10 ppb all of the 18 test cases without emission, which is severely underestimated for both urban (100-400 ppb) and rural (50-120 ppb) conditions [Cleveland et al., 1977; Gregory et al., 1988; Janach, 1989; Kirchhoff, 1988; LeFohn & Pinkerton, 1988; Logan, 1989]. Even for the so-called polluted atmosphere case, the initial ozone concentration was set to 50 ppb, which is hard to recognize as “polluted”. The constant ozone concentration among urban and rural in Stockwell’s cases also causes the NO/NO<sub>x</sub> ratio being not reasonable. According to Toores and Buchan (1988), with higher ozone concentration during the day, NO/NO<sub>x</sub> becomes lower. This agrees with R48 (NO + O<sub>3</sub> → NO<sub>2</sub> + O<sub>2</sub>), with higher ozone concentration, more NO is consumed, thus more NO<sub>2</sub> is produced. Thus, NO/NO<sub>x</sub> ratio in rural condition will be higher than in urban condition, which is opposite to Stockwell’s cases. Second, Stockwell’s cases considered NO<sub>x</sub> to be the only compounds of NO<sub>y</sub> in all the test cases. In fact, as the distance between sampling site and major anthropogenic sources increases, the conversion of NO<sub>x</sub> to HNO<sub>3</sub> and PAN increases. As a result, NO<sub>x</sub> account for 30-60% of NO<sub>y</sub> at rural sites, and 15% at forest sites and marine sites [Carroll et al., 1992; Fahey et al., 1986; National Research Council, 1992; Parrish et al., 1986; Williams et al., 1987]. Third, for Stockwell’s 9 urban cases and 9 rural cases, the emission rates of all chemical compounds are set to zero. Without the emission of NO and VOC, the simulated NO<sub>x</sub> and VOC deplete quickly, which decay the NO<sub>x</sub> cycle and the conversion to NO<sub>y</sub>, as well as ozone concentration (Fig. S1). Fourth, the NO<sub>x</sub> concentration in Stockwell’s urban cases is too low, with the maximum of 100 ppb at the initial state, but quickly deplete to be less than 10 ppb in 6 hours. Baugues (1986) measured NO<sub>x</sub> concentration among 30 sites among major cities in the eastern and southern United States in 1984 and 1985. Baugues’s measurement shows 17 out of 21 sites in 1984 and 15 out of 19 sites in 1985 with maximum concentration of NO<sub>x</sub> higher than 100 ppb, 10 out of 21 sites in 1984 and 7 out of 19 sites in 1985 with maximum concentration of NO<sub>x</sub> higher than 200 ppb, which indicate the obvious higher NO<sub>x</sub>

concentration than Stockwell's cases. For the so-called polluted atmosphere, the initial concentration of  $\text{NO}_x$  is only 0.7 ppb, with emission rate of 2.6 ppt/min, the maximum concentration only reaches 2.4 ppb around 6 am at day 2 to day 6, which again cannot stand for "polluted".

Thus, instead of replicating Stockwell's cases, we set up four conditions that stands for urban, rural, forest, and marine, with the initial concentrations based on various measurements from previous studies. The emission rates of NO and total VOCs was tuned until the simulation results satisfied with these following features: a). The concentration of  $\text{NO}_x$  changes diurnally and stabilized through time; b). The concentration of  $\text{O}_3$  changes diurnally and stabilized through time; c).  $\text{HNO}_3$  is produced primarily during daytime; d). VOCs are produced primarily during nighttime (Fig. S2-5). The molar fraction of each VOC species with respect to total VOC emission rate was obtained from Stockwell's (1997) emission cases.

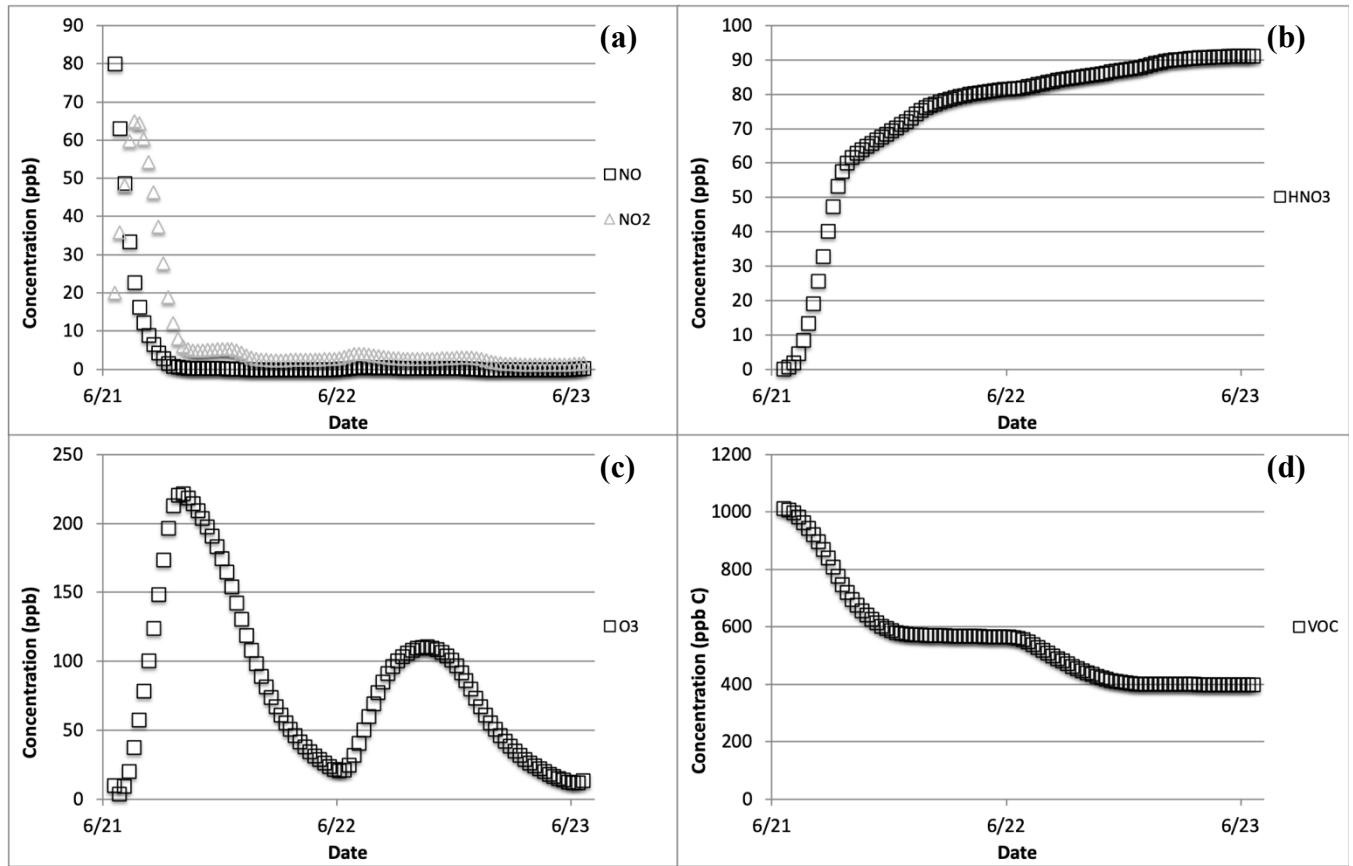


Figure S1. The concentrations of NO<sub>x</sub> (top left, a), HNO<sub>3</sub> (top right, b), O<sub>3</sub> (bottom left, c), and total VOC (bottom right, d) for Stockwell's urban case with initial concentration of NO<sub>x</sub> at 100 ppb, VOC at 1000 ppb C, simulation starts from Jun 21.

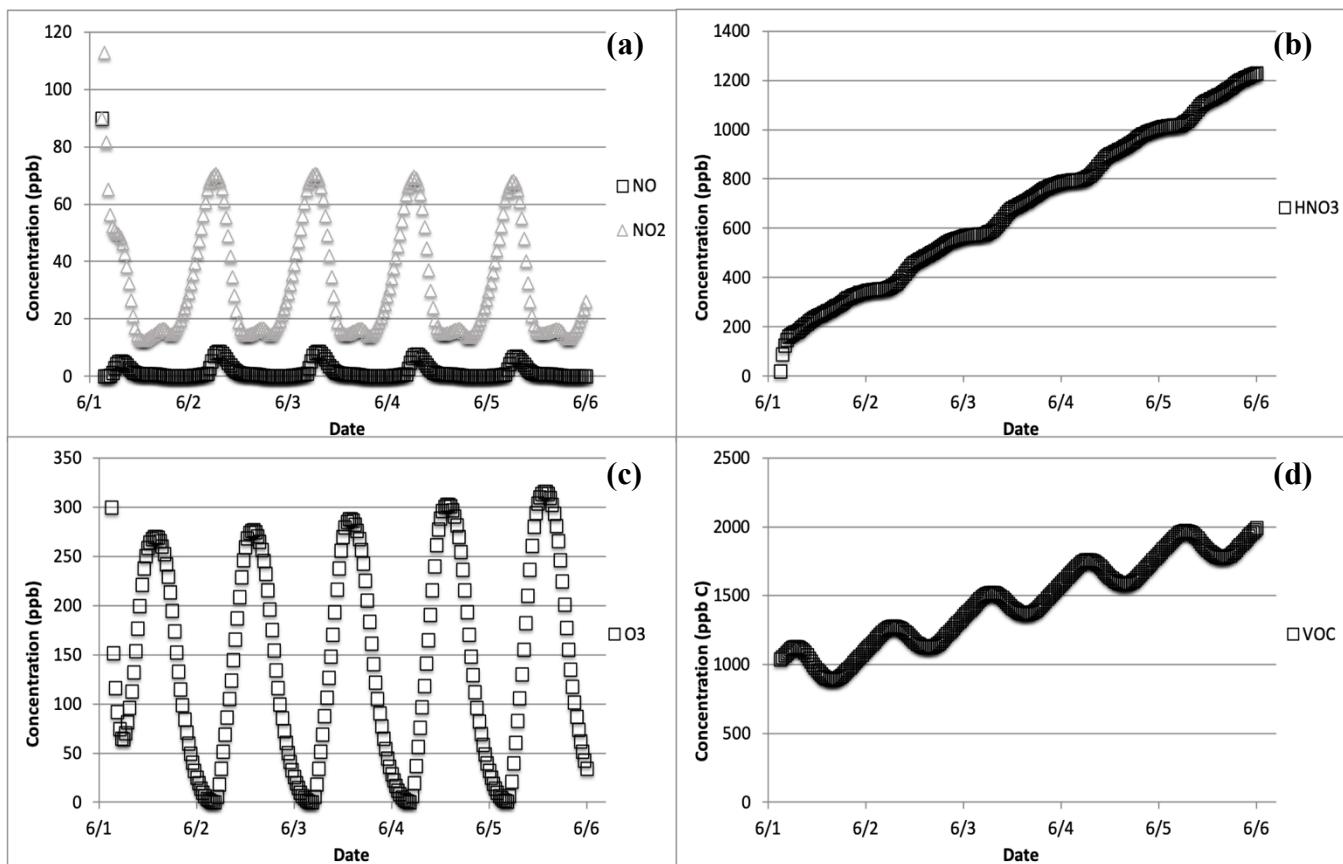


Figure S2. The concentrations of NO<sub>x</sub> (top left, a), HNO<sub>3</sub> (top right, b), O<sub>3</sub> (bottom left, c), and total VOC (bottom right, d) for urban condition, with initial concentrations and emission rate listed in Table S8, simulation starts from Jun 1.

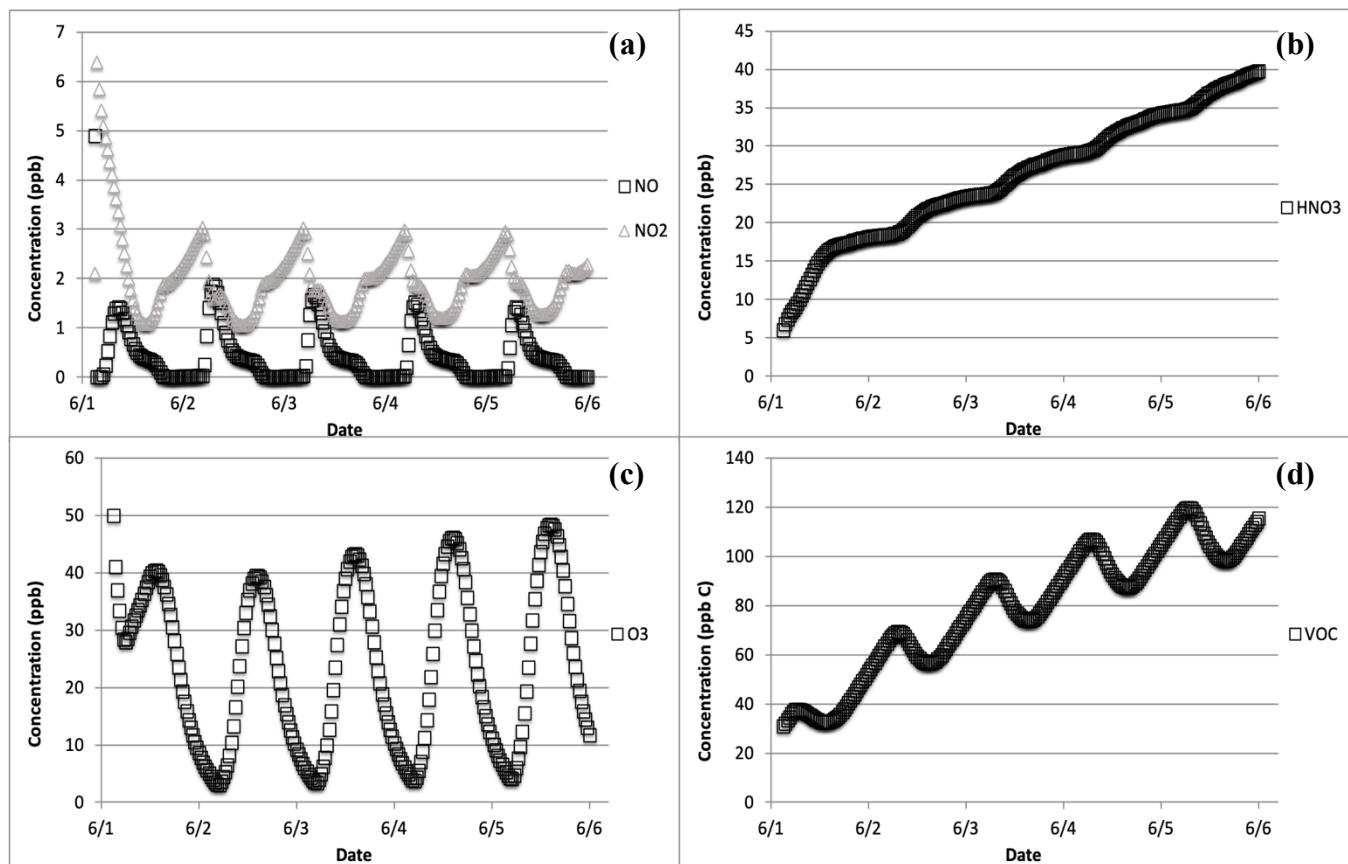


Figure S3. The concentrations of NO<sub>x</sub> (top left, a), HNO<sub>3</sub> (top right, b), O<sub>3</sub> (bottom left, c), and total VOC (bottom right, d) for rural condition, with initial concentrations and emission rate listed in Table S8, simulation starts from Jun 1.

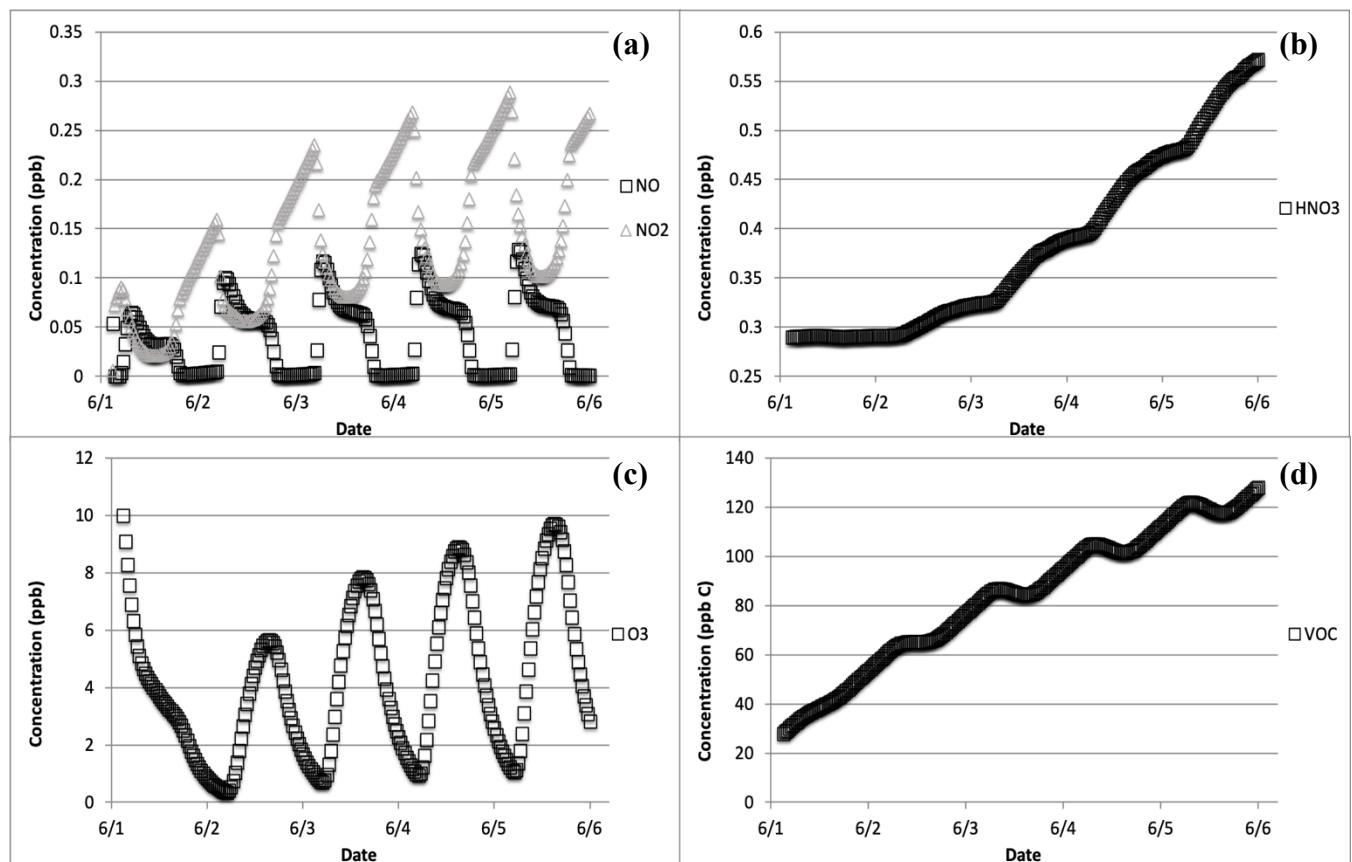


Figure S4. The concentrations of NO<sub>x</sub> (top left, a), HNO<sub>3</sub> (top right, b), O<sub>3</sub> (bottom left, c), and total VOC (bottom right, d) for forest condition, with initial concentrations and emission rate listed in Table S8, simulation starts from Jun 1.

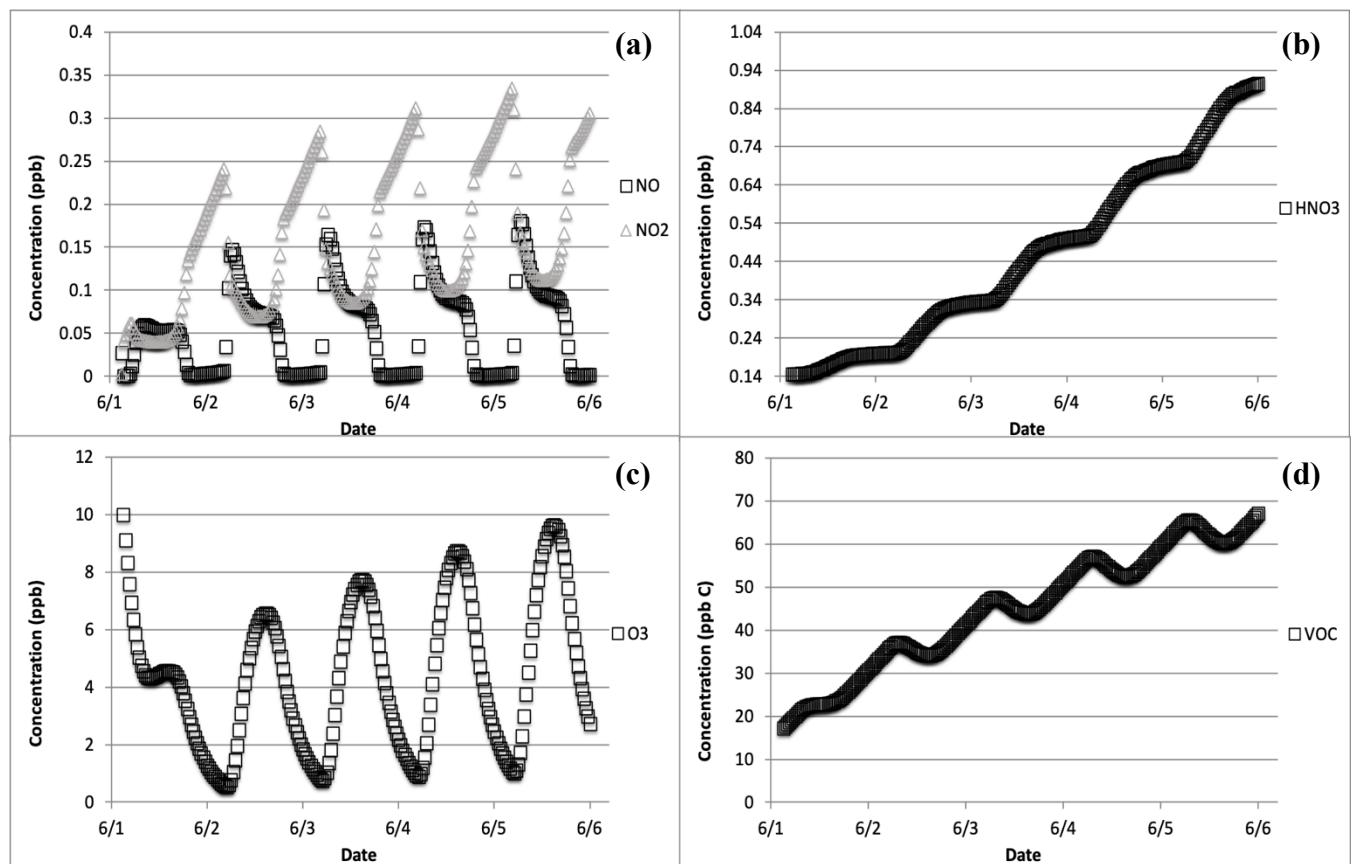


Figure S5. The concentrations of NO<sub>x</sub> (top left), HNO<sub>3</sub> (top right), O<sub>3</sub> (bottom left), and total VOC (bottom right) for marine condition, with initial concentrations and emission rate listed in Table S8, simulation starts from Jun 1.

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