

1 **Description and evaluation of the UKCA stratosphere-troposphere chemistry scheme**
2 **(StratTrop vn 1.0) implemented in UKESM1.**

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4 Archibald, Alexander T. ^{1,2,*}, O'Connor, Fiona M. ³, Abraham, N. Luke^{1,2}, Archer-Nicholls,
5 Scott¹, Chipperfield, Martyn P.^{4,5}, Dalvi, Mohit³, Folberth, Gerd A. ³, Dennison, Fraser ^{6,+},
6 Dhomse, Sandip S.^{4,5}, Griffiths, Paul T. ^{1,2}, Hardacre, Catherine³, Hewitt, Alan J. ³, Hill,
7 Richard³, Johnson, Colin E. ³, Keeble, James^{1,2}, Köhler, Marcus O.^{1,7,†}, Morgenstern, Olaf⁶,
8 Mulcahy, Jane P. ³, Ordóñez, Carlos^{3,‡}, Pope, Richard J.^{4,5}, Rumbold, Steven T. ⁸, Russo,
9 Maria R. ^{1,2}, Savage, Nicholas³, Sellar, Alistair³, Stringer, Marc⁸, Turnock, Steven³, Wild,
10 Oliver⁹ and Zeng, Guang⁶.

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12 1) Department of Chemistry, University of Cambridge, Cambridge, UK, CB2 1EW
13 2) NCAS-Climate, University of Cambridge, UK, CB2 1EW
14 3) Met Office Hadley Centre, FitzRoy Road, Exeter, UK, EX1 3PB
15 4) School of Earth and Environment, University of Leeds, Leeds, UK, LS2 9JT.
16 5) National Centre for Earth Observation (NCEO), University of Leeds, Leeds, U.K.
17 6) National Institute of Water & Atmospheric Research Ltd (NIWA), 301 Evans Bay
18 Parade, Greta Point, Wellington, New Zealand.
19 7) Centre for Ocean and Atmospheric Sciences, School of Environmental Sciences,
20 University of East Anglia, Norwich, U.K.
21 8) NCAS-Climate, Department of Meteorology, University of Reading, Reading, UK,
22 RG6 6BB.
23 9) Lancaster Environment Centre, Lancaster University, Lancaster, UK, LA1 4YQ

24
25 (†) Now at ECMWF, Reading, UK.

26 (‡) Now at Departamento de Física de la Tierra y Astrofísica, Facultad de Ciencias
27 Físicas, Universidad Complutense de Madrid, Madrid 28040, Spain

28 (*) Now at now at CSIRO Oceans and Atmosphere, Aspendale, Australia
29

30 *Email: ata27@cam.ac.uk
31
32

33 **Abstract**

34 Here we present a description of the UKCA StratTrop chemical mechanism which is used in
35 the UKESM1 Earth System Model for CMIP6. The StratTrop chemical mechanism is a merger
36 of previously well evaluated tropospheric and stratospheric mechanisms and we provide
37 results from a series of bespoke integrations to assess the overall performance of the model.
38

39 We find that the StratTrop scheme performs well when compared to a wide array of
40 observations. The analysis we present here focuses on key components of atmospheric
41 composition, namely the performance of the model to simulate ozone in the stratosphere and
42 troposphere and constituents that are important for ozone in these regions. We find that the
43 results obtained for tropospheric ozone and its budget terms from the use of the StratTrop
44 mechanism are sensitive to the host model; simulations with the same chemical mechanism
45 run in an earlier version of the MetUM host model show a range of sensitivity to emissions that
46 the current model does not fall within.
47

1 Whilst the general model performance is suitable for use in the UKESM1 CMIP6 integrations,
2 we note some shortcomings in the scheme that future targeted studies will address.
3
4

5 **1.0 Introduction**

6 The ability to model the composition of the atmosphere is vital for a wide range of applications
7 relevant to society at large. Atmospheric composition modelling can broadly be subdivided
8 into two sub disciplines: (1) aerosol processes and microphysics and (2) atmospheric
9 chemistry. Coupling these processes in climate models is paramount for being able to simulate
10 atmospheric composition at the global scale. The most societally important questions revolve
11 around understanding how the composition of the atmosphere has changed over the past,
12 attributing this change, understanding how this system is likely to change into the future and
13 what the impacts of these changes are on the Earth system and on human health. It is these
14 pressing issues that have led to the development of the new UK Earth System Model,
15 UKESM1 (Sellar et al., 2019a), which uses the UK Chemistry and Aerosol model (UKCA)
16 (O'Connor et al., 2014; Morgenstern et al., 2009; Mulcahy et al., 2018) as its key component
17 to simulate atmospheric composition in the Earth system. The key challenge UKCA is applied
18 to is understanding and predicting how the concentrations of a range of trace gases, especially
19 the greenhouse gases methane (CH₄), ozone (O₃) and nitrous oxide (N₂O), and aerosol
20 species will evolve in the Earth system under a range of different forcings. UKCA simulates
21 the processes that control the formation and destruction of these species. Here we describe
22 and document the performance of the version of UKCA used in UKESM1, which includes a
23 representation of combined stratospheric and tropospheric chemistry that enhances the
24 capability of UKCA beyond the version used in the Atmospheric Chemistry and Climate Model
25 Intercomparison Project (ACCMIP; Young et al., 2013; O'Connor et al., 2014) and the recent
26 Chemistry-Climate Model Initiative (CCMI) intercomparison (Bednarz et al., 2018; Hardiman
27 et al., 2017; Morgenstern et al., 2017). There have been a number of previous versions of
28 UKCA with defined scopes but we denote the version used in UKESM1 and described here
29 as UKCA StratTrop, to signify its purpose of holistic treatment of composition processes in the
30 troposphere and stratosphere.
31

32 As a result of the Chemistry-Climate Model Validation Activity (CCMVal), it was recommended
33 that models which are aimed at simulating the coupled ozone-climate problem should include
34 processes to enable interactive ozone in the troposphere and stratosphere (Morgenstern et
35 al., 2010). Chemistry-climate models (CCMs) use schemes to describe the reactions that
36 chemical compounds undergo. These chemistry schemes can be constructed to explicitly
37 model a specific chemical reaction system (e.g. Aumont et al., 2005) but in most applications
38 the chemistry schemes are heavily simplified. Until recently, models of atmospheric chemistry
39 tended to focus on chemistry schemes formulated for limited regions of the atmosphere;
40 detailed schemes have been constructed to examine phenomena such as stratospheric ozone
41 depletion or tropospheric air pollution. Examples of this using the UKCA model framework are
42 two studies of the effects of the eruption of Mt. Pinatubo, where Telford et al. (2009) used the
43 stratospheric scheme of Morgenstern et al. (2009) to study the effects of the eruption on
44 stratospheric ozone, whereas Telford et al. (2010) used the tropospheric scheme of O'Connor

1 et al. (2014) to examine the effects on tropospheric oxidising capacity. Whilst the chemical
2 schemes described in O'Connor et al. (2014) (hereafter OC14) and Morgenstern et al. (2009)
3 (hereafter MO09) have some overlap (for example the use of some common reactions) the
4 schemes were developed with specific applications in scope. The reason for partitioning
5 chemical complexity up like this is to reduce the computational resources required. Moreover,
6 simulations with these process limitations were found to be able to capture the phenomena of
7 interest.

8 However, increases in computational power and a drive to answer a greater number of
9 questions from model simulations have allowed models that simulate both the stratosphere
10 and troposphere to be developed and which are now widely used (e.g. Pitari et al., 2002;
11 Jockel et al., 2006; Lamarque et al., 2008; Morgenstern et al., 2012). The removal of the need
12 for prescribed upper boundary conditions (for the stratosphere), and a more comprehensive
13 chemistry scheme, make their increased cost worth bearing. In this work, we describe the
14 implementation of a combined chemistry scheme suitable for simulating the stratosphere and
15 the troposphere within the UKCA model as used in UKESM1 (Sellar et al., 2019a). This
16 scheme, UKCA StratTrop, builds on and combines the existing stratospheric (MO09) and
17 tropospheric schemes (OC14). In various configurations of UKCA (under the names
18 HadGEM3-ES, UМУKCA-UCAM, NIWA-UKCA, ACCESS), this combined chemical scheme
19 has already been used to study: stratospheric ozone and its sensitivity to changes in bromine
20 (Yang et al., 2014), subsequent circulation changes (Braesicke et al., 2013), and how it may
21 be impacted by certain forms of geoengineering (Tang et al., 2014); the role of ozone radiative
22 feedback on temperature and humidity biases at the tropical tropopause layer (TTL)
23 (Hardiman et al., 2015); the effects on tropospheric and stratospheric ozone changes under
24 climate and emissions changes following the Representative Concentration Pathways (RCPs)
25 (Banerjee et al., 2015; Dhomse et al., 2018); climate-induced changes in lightning (Banerjee
26 et al., 2014) and changes in methane chemistry between the present day and the last
27 interglacial (Quiquet et al., 2015). The scheme has been included in model simulations as part
28 of the CCMI project (Eyring et al., 2013; Hardiman et al., 2017; Morgenstern et al., 2017;
29 Dhomse et al., 2018) as well as all future Earth System modelling studies using the UKESM1
30 model (Sellar et al., 2019a).

31
32 This paper is organised in the following sections: In Section 2, we present a thorough
33 description of UKCA StratTrop, including the physical model and details of the chemistry
34 scheme, followed by a detailed description of the emissions used and some notes on the
35 historical development of the scheme. In Section 3, we describe two 15-year simulations we
36 have performed with UKCA StratTrop in an atmosphere-only configuration of UKESM1. In
37 Section 4, we use these simulations to review the performance of UKCA StratTrop, focusing
38 on the model's ability to simulate key features of tropospheric and stratospheric chemistry as
39 simulated by other models or observed using in situ and remote sensing measurements.
40 Finally, in Section 5, we discuss the performance of the model and make some
41 recommendations for further targeted studies.

42

43

2.0 Model Description

In this section, we present a thorough description of UKCA StratTrop, from the host physical model to the detailed process representation of the StratTrop chemistry scheme.

2.1 Physical Model

The physical model to which the UKCA StratTrop chemistry scheme has been coupled is the Global Atmosphere 7.1/Global Land 7.0 (GA7.1/GL7.0; Walters et al., 2019) configuration of the Hadley Centre Global Environment Model version 3 (HadGEM3; Hewitt et al., 2011).

The coupling between the UKCA StratTrop chemistry scheme and the GA7.1/GL7.0 configuration of HadGEM3 is based on the Met Office's Unified Model (MetUM; Brown et al., 2012). As a result, UKCA uses aspects of MetUM for the large-scale advection, convective transport, and boundary layer mixing of its tracers. The large-scale advection makes use of the semi-implicit semi-Lagrangian formulation of the ENDGame dynamical core (Wood et al., 2014) to solve the non-hydrostatic, fully compressible deep-atmosphere equations of motion. These are discretized on to a regular latitude-longitude grid, with Arakawa C-grid staggering (Arakawa and Lamb, 1977). The discretization in the vertical uses Charney–Phillips staggering (Charney and Phillips, 1953) with terrain-following hybrid height coordinates. Although GA7.1/GL7.0 can be run at a variety of resolutions, as detailed in Walters et al. (2019), the resolution here is N96L85 ($1.875^\circ \times 1.25^\circ$ longitude-latitude) i.e. approximately 135 km resolution in the horizontal and with 85 terrain-following levels spanning the altitude range from the surface to 85 km. Of the 85 model levels, 50 lie below 18 km and 35 levels are above 18 km (Walters et al., 2019). Mass conservation of UKCA tracers is achieved with the optimised conservative filter (OCF) scheme (Zerroukat and Allen, 2015); use of this scheme for virtual dry potential temperature resulted in reducing the warm bias at the TTL (Hardiman et al., 2015; Walters et al., 2019). This conservation scheme is also used for moist prognostics (e.g. water vapour mass mixing ratio and prognostic cloud fields). Although this makes the conservation scheme for moist prognostics consistent with the treatment of UKCA tracers and virtual dry potential temperature, Walters et al. (2019) found that it had little impact on moisture biases in the lower stratosphere.

The convective transport of UKCA tracers is treated within the MetUM convection scheme. It is essentially the mass flux scheme of Gregory and Rowntree (1990) but with updates for downdrafts (Gregory and Allen, 1991), convective momentum transport (Gregory et al., 1997), and Convective Available Potential Energy closure. The scheme involves diagnosis of possible convection from the boundary layer, followed by a call to shallow or deep convection on selected grid points based on the diagnosis from step one, and then a call to the mid-level convection scheme at all points. One key difference between the convective treatment of UKCA chemical and aerosol tracers is that convective scavenging of aerosols (simulated with GLOMAP-mode) is coupled with the convective transport following Kipling et al. (2013), whereas for chemical tracers, convective transport and scavenging are treated independently. Further details on the convection scheme in GA7.1 can be found in Walters et al. (2019). Finally, mixing over the full depth of the troposphere is carried out by the so-called “boundary-layer” scheme in GA7.1; this scheme is that of Lock et al. (2000), but with updates from Lock (2001) and Brown et al. (2008).

1 The GA7.1/GL7.0 configuration described in Walters et al. (2019) already includes the two-
2 moment GLOMAP-mode aerosol scheme from UKCA (Mann et al., 2010; Mulcahy et al., 2018;
3 Mulcahy et al., 2019), in which sulphate and secondary organic aerosol (SOA) formation is
4 driven by prescribed oxidant fields. In the UKCA-StratTrop configuration described here, the
5 oxidants driving secondary aerosol formation are fully interactive; this coupling between UKCA
6 chemistry and GLOMAP-mode is fully described in Mulcahy et al. (2019). Together with
7 dynamic vegetation and a terrestrial carbon/nitrogen scheme (Sellar et al., 2019a),
8 GA7.1/GL7.0 and UKCA StratTrop make up the atmospheric and land components of the UK
9 Earth System Model, UKESM1 (Sellar et al., 2019a) which will be used as part of the UK
10 contribution to the 6th Coupled Model Intercomparison Project (CMIP6; Eyring et al., 2016).

11

12 **2.2 Chemistry scheme**

13 The UKCA StratTrop scheme is based on a merger between the stratospheric scheme of
14 MO09 and the tropospheric “TropIsop” scheme of OC14. StratTrop simulates the O_x , HO_x and
15 NO_x chemical cycles and the oxidation of carbon monoxide, ethane, propane, and isoprene in
16 addition to chlorine and bromine chemistry, including heterogeneous processes on polar
17 stratospheric clouds (PSCs) and liquid sulphate aerosols (SAs). The level of detail of the VOC
18 oxidation is far from the complexity of explicit representations (e.g. Aumont et al., 2005) but
19 the VOCs simulated are treated as discrete species.

20

21 Wet deposition is parameterised using the approach of Giannakopoulos et al. (1999). Dry
22 deposition is parameterised employing a resistance type model (Wesely, 1989) using the
23 implementation described in OC14, updated to account for advancements in the Joint UK Land
24 Environment Simulator (JULES; Best et al., 2011), in particular a significant increase in land
25 surface types (an increase from 9 to 27; see below for more details). Interactive photolysis is
26 represented with the Fast-JX scheme (Neu et al., 2007), as implemented in Telford et al.
27 (2013). Fast-JX covers the wavelength range of 177 to 750 nm. For shorter wavelengths,
28 effective above 60 km of altitude, a correction is applied to the photolysis rates following the
29 formulation of Lary and Pyle (1991).

30

31 The StratTrop scheme includes emissions of 12 chemical species: nitrogen oxide (NO),
32 carbon monoxide (CO), formaldehyde (HCHO), ethane (C_2H_6), propane (C_3H_8), acetaldehyde
33 (CH_3CHO), acetone ($(CH_3)_2CO$), methanol (CH_3OH) and isoprene (C_5H_8) in addition to trace
34 gas aerosol-precursor emissions (dimethyl sulphide (DMS), sulphur dioxide (SO_2), and
35 monoterpenes). For the implementation used in UKESM1, emissions may be prescribed or
36 interactive and are described in more detail in Sections 2.6.1 to 2.6.3. A further 7 long-lived
37 species (N_2O , CF_2Cl_2 , $CFCl_3$, CH_3Br , COS, H_2 , and CH_4) are constrained by lower boundary
38 conditions; for more details see Section 2.6.4.

39

40 UKCA StratTrop was developed by starting with the stratospheric chemistry scheme (MO09)
41 and adding aspects of chemistry unique to the tropospheric scheme (OC14). In most cases
42 the formulation and reaction coefficients are taken from reference evaluations (JPL and
43 IUPAC) or the Master Chemical Mechanism, as detailed in OC14. Table 1 provides a list of
44 the chemical tracers included in the StratTrop configuration used in UKESM1. In total the
45 model employs 84 species and represents the chemistry of 81 of these. O_2 , N_2 and CO_2 are

1 not treated as chemically active species. Note that the scheme has a simplified treatment of
 2 stratospheric halocarbons and lumps all chlorine and bromine source gases into CFC-11,
 3 CFC-12 and CH₃Br. This chemistry scheme accounts for 199 bimolecular reactions (Table
 4 S1), 25 uni- and termolecular reactions (Table S2), 59 photolytic reactions (Table S3), 5
 5 heterogeneous reactions (Table S4) and 3 aqueous phase reactions for the sulfur cycle (Table
 6 S5). Hence, UKCA-StratTrop describes the oxidation of organic compounds – e.g. methane,
 7 ethane, propane and isoprene and their oxidation products – coupled to the inorganic
 8 chemistry of O_x, NO_x, HO_x, ClO_x and BrO_x, using a continuous set of equations with no artificial
 9 boundaries imposed on where to stop performing chemistry. Except for water vapour, at the
 10 top two levels, the mixing ratios of all species are held identical to those at the third highest
 11 level. The time-dependent chemical reactions are integrated forward in time using an implicit
 12 backward Euler solver with Newton Raphson iteration (Wild and Prather, 2000). This solver
 13 has a relative convergence criterion of 10⁻⁴ with a time step of 60 minutes throughout the
 14 atmosphere. An extensive discussion of the solver used here is presented in Esentürk et al.
 15 (2018).

16

17 **Table 1.** List of chemical species in UKCA StratTrop. Species in italics are not advected tracers but
 18 are calculated using a steady state approximation. Species in bold are set as constant mixing ratios
 19 throughout the atmosphere. †The molecular mass of Sec_Org is set to 150 g/mol.

Name	Formula	Dry Deposit ed	Wet Deposited	Emitted or LBC
O(3P)	O(³ P)	No	No	No
O(1D)	O(¹ D)	No	No	No
O3	O ₃	Yes	No	No
N	N	No	No	No
NO	NO	Yes	No	Emitted
NO3	NO ₃	Yes	Yes	No
NO2	NO ₂	Yes	No	No
N2O5	N ₂ O ₅	Yes	Yes	No
HO2NO2	HO ₂ NO ₂	Yes	Yes	No
HONO2	HONO ₂	Yes	Yes	No
H2O2	H ₂ O ₂	Yes	Yes	No
CH4	CH ₄	No	No	LBC

CO	CO	Yes	No	Emitted
HCHO	HCHO	Yes	Yes	Emitted
MeOO	CH ₃ OO	No	Yes	No
MeOOH	CH ₃ OOH	Yes	Yes	No
H	H	No	No	No
H ₂ O	H ₂ O	No	No	No
OH	OH	No	No	No
HO ₂	HO ₂	No	Yes	No
Cl	Cl	No	No	No
Cl ₂ O ₂	Cl ₂ O ₂	No	No	No
ClO	ClO	No	No	No
OCIO	OCIO	No	No	No
Br	Br	No	No	No
BrO	BrO	No	No	No
BrCl	BrCl	No	No	No
BrONO ₂	BrONO ₂	No	Yes	No
N ₂ O	N ₂ O	No	No	LBC
HCl	HCl	Yes	Yes	No
HOCl	HOCl	Yes	Yes	No
HBr	HBr	Yes	Yes	No
HOBr	HOBr	Yes	Yes	No
ClONO ₂	ClONO ₂	No	Yes	No
CFCI ₃	CFCI ₃	No	No	LBC
CF ₂ Cl ₂	CF ₂ Cl ₂	No	No	LBC
MeBr	CH ₃ Br	No	No	LBC
HONO	HONO	Yes	Yes	No
C ₂ H ₆	C ₂ H ₆	No	No	Emitted

EtOO	C ₂ H ₅ OO	No	No	No
EtOOH	C ₂ H ₅ OOH	Yes	Yes	No
MeCHO	CH ₃ CHO	Yes	No	Emitted
MeCO ₃	CH ₃ C(O)OO	No	No	No
PAN	PAN	Yes	No	No
C ₃ H ₈	C ₃ H ₈	No	No	Emitted
n-PrOO	C ₃ H ₇ OO	No	No	No
i-PrOO	CH ₃ CH(OO)CH ₃	No	No	No
n-PrOOH	C ₃ H ₇ OOH	Yes	Yes	No
i-PrOOH	CH ₃ CH(OOH)CH ₃	Yes	Yes	No
EtCHO	C ₂ H ₅ CHO	Yes	No	No
EtCO ₃	C ₂ H ₅ C(O)OO	No	No	No
Me ₂ CO	CH ₃ C(O)CH ₃	No	No	Emitted
MeCOCH ₂ O O	CH ₃ C(O)CH ₂ OO	No	No	No
MeCOCH ₂ O OH	CH ₃ C(O)CH ₂ OOH	Yes	Yes	No
PPAN	PPAN	Yes	No	No
MeONO ₂	MeONO ₂	No	No	No
C ₅ H ₈	C ₅ H ₈	No	No	Emitted
ISO ₂	HOC ₅ H ₈ OO	No	No	No
ISOOH	HOC ₅ H ₈ OOH	Yes	Yes	No
ISON	ISON	Yes	Yes	No
MACR	C ₄ H ₆ O	Yes	No	No
MACRO ₂	C ₄ H ₆ O(OO)	No	No	No
MACROOH	C ₄ H ₆ O(OOH)	Yes	Yes	No
MPAN	MPAN	Yes	No	No
HACET	CH ₃ C(O)CH ₂ OH	Yes	Yes	No
MGLY	CH ₃ COCHHO	Yes	Yes	No

NALD	NALD	Yes	No	No
HCOOH	HC(O)OH	Yes	Yes	No
MeCO3H	CH ₃ C(O)OOH	Yes	Yes	No
MeCO2H	CH ₃ C(O)OH	Yes	Yes	No
H ₂	H ₂	No	No	LBC
MeOH	CH ₃ OH	Yes	Yes	Emitted
CO₂	CO₂	No	No	No
O₂	O₂	No	No	No
N₂	N₂	No	No	No
DMS	CH ₃ SCH ₃	No	No	Emitted
SO ₂	SO ₂	Yes	Yes	Emitted
H ₂ SO ₄	H ₂ SO ₄	Yes	No	No
MSA	MSA	No	No	No
DMSO	DMSO	Yes	Yes	No
COS	COS	No	No	Emitted
SO ₃	SO ₃	No	No	No
Monoterp	C ₁₀ H ₁₆	Yes	No	Emitted
Sec_Org	[†] Sec_Org	Yes	Yes	No

1

2 The treatment of Polar Stratospheric Cloud (PSC) has been recently expanded in UKCA
3 (Dennison et al. 2019), but these improvements did not make it into the UKESM1 version of
4 UKCA discussed here, which remains unmodified from the original Morgenstern et al. (2009)
5 scheme. The abundance of nitric acid trihydrate (NAT) and mixed NAT/ice polar stratospheric
6 clouds is calculated following Chipperfield (1999) assuming thermodynamic equilibrium with
7 gas-phase HNO₃ and water vapour; the treatment of reactions on liquid sulfate aerosol also
8 follows Chipperfield (1999). Sedimentation of PSCs is included in the model whilst dehydration
9 is handled as part of the model's hydrological cycle. Denitrification is prescribed in the same
10 way as in Chipperfield (1999) with two different sedimentation velocities. We refer the reader
11 to Morgenstern et al. (2009) and Dennison et al. (2019) for further details.

12

13 The stratospheric sulfate aerosol optical depth, used in the radiation scheme of MetUM, is
14 modified to be consistent with the aerosols used in the heterogeneous chemistry which, by
15 default, are taken from a surface area density climatology prepared for the CMIP6 model
16 intercomparison (Luo, personal communication). The surface aerosol density is converted to

1 mass mixing ratio, using a climatology of particle size (Thomason and Peter, 2006) and
2 assuming a density of 1700 kg/m³.

3 4 **2.3 Photolysis**

5 The most significant new development relative to MO09 and OC14 in the UKCA-StratTrop
6 scheme used in UKESM1 is the interactive Fast-JX photolysis scheme which is applied to
7 derive photolysis rates between 177 and 750 nm (Neu et al., 2007) as described in Telford et
8 al. (2013). This is an important new addition as it enables interactive treatment of photolysis
9 rates (key drivers for the photochemistry of the atmosphere) under changing climate and
10 atmospheric composition. For shorter wavelengths, relevant above 60 km, a correction is
11 added, to account for photolysis occurring between 112 and 177 nm, following Lary and Pyle
12 (1991).

13
14 In older versions of UKCA (i.e. MO09 and OC14) pre-calculated photolysis frequencies were
15 applied in the model. Sellar et al (2019) shows a comparison of these and we note here that
16 the switch from pre-calculated to on-line interactive photolysis calculations has had a
17 significant effect on shortening the model simulated methane lifetime and increasing the
18 tropospheric mean [OH] (Telford et al., 2013; O'Connor et al., 2014; Voulgarakis et al., 2009),
19 as shown in Figure 4.

20 21 **2.4 Dry deposition**

22 In UKCA the representation of dry deposition follows the resistance-in-series model as
23 described by Wesely (1989) in which the removal of material at the surface is described by
24 three resistances, r_a , r_b , and r_c . The deposition velocity v_d (m s⁻¹) is then a function of these
25 three resistance terms according to:

$$26 \quad v_d = \frac{1}{r_a + r_b + r_c}, \quad \text{Eq. 1}$$

27
28 where r_a denotes the aerodynamic resistance to dry deposition, r_b is the quasi-laminar
29 resistance term, and r_c represents the resistance to uptake at the surface. Of these three terms
30 r_c tends to be the most complex because it encompasses a variety of exchange fluxes, such
31 as stomatal and cuticular uptake, assimilation by soil microbes, etc. The uptake at the surface
32 also depends strongly on the presence of dew, rain, or snow which can interrupt the deposition
33 process altogether.
34

35 36 **2.4.1 Dry deposition of gas-phase species**

37 Surface dry deposition is calculated interactive at every time step for a number of atmospheric
38 gas-phase species (c.f., Table 1 for a list of deposited species). The aerodynamic resistance
39 r_a is given by:

$$40 \quad r_a = \frac{\ln\left(\frac{z}{z_0}\right) - \psi}{k \times u^*}, \quad \text{Eq. 2}$$

41
42 where z_0 is the roughness length, ψ denotes the Businger dimensionless stability function, k
43 is the von Karman constant, and u^* is the friction velocity. r_a represents the resistance to
44 turbulent mixing in the boundary layer and therefore depends crucially on the stability of the
45 boundary layer. It is independent of the chemical species that is deposited.
46

1
2 The quasi-laminar resistance r_b , on the other hand, depends on the chemical and physical
3 properties of the deposited species. It describes the transport through the thin, laminar layer
4 of air closest to the surface. Transport through this layer is diffusive due to the absence of
5 turbulent mixing.
6
7 The third resistance term r_c depends on both the physico-chemical properties of the deposited
8 species and the properties and condition of the respective surface to which deposition occurs.
9 The surface can be anything from bare soil or rock to vegetation and even urban environments.
10 Surface uptake varies with season, time of day and current meteorological conditions. The
11 largest individual surface type is water in the form of the world's oceans. In this latter case
12 solubility clearly plays the key role (Hardacre et al., 2015; Luhar et al., 2017).
13
14 A particularly important surface uptake process is the deposition flux to the terrestrial
15 vegetation. In this case a number of pathways exist which are commonly integrated into the
16 so-called "Big-leaf" model (Smith et al., 2000; Seinfeld and Pandis, 2006). Of all the deposition
17 pathways manifesting in vegetated regions, for most species the most important is uptake
18 through the stomata. Through these tiny pores in the leaf surface plants take up carbon dioxide
19 from the atmosphere and exchange water vapour and oxygen with it. This exchange also
20 includes all other species that make up the ambient air, including pollutants such as for
21 instance ozone. For this, the specific type of vegetation is crucial. Ozone deposition fluxes, for
22 instance, vary widely between forests and grasslands.
23
24 The calculation of the surface resistance term and land surface type information provided by
25 the dynamic vegetation model JULES (Best et al., 2011; Clark et al., 2011) is used in UKCA.
26 JULES forms part of UKESM1 and is thus coupled with UKCA. Within JULES, various land
27 surface type configurations may be selected. In the most simple configuration, which was also
28 used in the UKESM1 predecessor model HadGEM2-ES, any land-based grid box at the
29 surface can be subdivided into variable-sized fractions assigned to any of 9 different surface
30 types: broadleaf trees, needleleaf trees, C3 grasses, C4 grasses, shrubs, bare soil, rivers
31 and lakes, urban environments and ice. Non-land grid boxes are treated separately.
32
33 Since then, the number of land surface types in JULES has increased substantially (c.f. Harper
34 et al., 2018). Apart from the original 9-tile version (5 vegetation and 4 non-vegetation types),
35 13, 17, and also 27-tile configurations are now included. The upgrade to the 13-tile
36 configuration increases the number of vegetation types by introducing 3 broadleaf plant
37 functional types (PFTs), 2 needleleaf PFTs, and 2 shrub PFTs; the number of grass-related
38 PFTs as well as the number of non-vegetation type remains the same. In this configuration.
39 The 17-tile configuration further extends the number of PFTs by introducing 4 cropland types,
40 two C3-grass related and two C4-grass related PFTs; again, the number of non-vegetation
41 types remains the same. Finally, the 27-tile land surface configuration, corresponding to the
42 UKESM1 release configurations and the configurations used for this manuscript, introduces a
43 substantial number of additional land ice tiles. Each of these land surface and PFT tiles offers
44 a specific resistance to dry deposition of atmospheric gas-phase species.
45
46 For dry deposition of aerosols a slightly different treatment is taken to that described above
47 and we direct the reader to Mulchay et al. (2019) and references therein for more details.
48

2.5 Wet deposition

The wet deposition scheme employed in UKCA for the removal of tropospheric gas-phase species through convective and stratiform precipitation is the same as that described in O'Connor et al., 2014. The original scheme was implemented from the TOMCAT chemistry transport model (CTM) where it previously had been validated by Giannakopoulos (1998) and Giannakopoulos et al. (1999). In this paper we provide a brief description of the scheme but will not present an evaluation because there have been no changes since the last published version. For an in-depth performance evaluation in UKCA we refer to section 3.4 in O'Connor et al. (2014).

Following a scheme originally developed by Walton et al. (1988) wet deposition is parameterized as a first-order loss process which is calculated as a function of the three-dimensional convective and stratiform precipitation. The climate model provides the required precipitation activity to UKCA. The wet scavenging rate r is calculated at every grid box and time step according to:

$$r = S_j \times p_j(l) \quad \text{Eq. 3}$$

where S_j is the wet scavenging coefficient for precipitation type j (cm^{-1}) and $p_j(l)$ is the precipitation rate for type j (convective or stratiform), provided at model level l (cm h^{-1}).

Scavenging coefficients for nitric acid (HNO_3) of 2.4 cm^{-1} and 4.7 cm^{-1} for stratiform and convective precipitation, respectively, are applied (c.f., Penner et al., 1991). These parameters are scaled down for individual species using the fraction of each species in the aqueous phase, f_{aq} , calculated by:

$$f_{aq} = \frac{L \times H_{eff} \times R \times T}{1 + L \times H_{eff} \times R \times T} \quad \text{Eq. 4}$$

where L represents the liquid water content, R the universal gas constant, T denotes ambient temperature, and H_{eff} is the effective Henry's Law constant for each species. H_{eff} includes the effects of solubility, dissociation, and complex formation. Tables S6, S7, and S8 (in the supplement) summarise the parameters used in the UKCA wet deposition scheme for each soluble species included in the StratTrop chemical mechanism.

Furthermore, in the scheme precipitation only occurs over a fraction of the grid box. This fraction is assumed to be 1.0 and 0.3 for stratiform and convective precipitation, respectively. These fractions are applied in the calculation of the grid-box-mean wet scavenging rate for both precipitation types after which point the two rates are added together.

2.6 Emissions

This section describes the implementation of tropospheric ozone precursor emissions used in the UKCA StratTrop scheme in detail. The scheme includes the emissions of nine chemical species: nitric oxide (NO), carbon monoxide (CO), formaldehyde (HCHO), ethane (C_2H_6), propane (C_3H_8), acetaldehyde (MeCHO), acetone (Me_2CO), isoprene (C_5H_8), and methanol (MeOH). Emissions to UKCA can be broadly classified into two categories: *offline*, where pre-

1 computed fluxes are read from input files; and *online*, where fluxes are computed in real-time
2 during the simulation making use of online meteorological variables from the MetUM. The
3 implementation of *offline* emissions will be described in Section 2.6.1. Examples of *online*
4 emissions currently in UKCA StratTrop are biogenic volatile organic compound (BVOC)
5 emissions (Section 2.6.2) and lightning NO_x (Section 2.6.3). All emissions, including offline
6 emissions, have interannual variability over the time period of the model simulations.

7
8 When UKCA StratTrop is coupled to the UKCA aerosol scheme, GLOMAP-mode (Mann et al.,
9 2010) as here, there are additional trace gas aerosol-precursor emissions for dimethyl
10 sulphide (DMS), sulphur dioxide (SO₂), and monoterpenes (C₁₀H₁₆). These emissions will be
11 discussed in the context of the UKESM1 aerosol performance in Mulcahy et al. (2019); the
12 focus here will solely be on the tropospheric ozone precursor emissions. Table S10 and
13 Figures S7-S8 summarise the mean global annual emissions totals for the time period
14 considered here (2005-2014) and their global and seasonal distributions.

17 **2.6.1 Offline Anthropogenic and Natural Emissions**

18 *Offline* tropospheric ozone precursor emissions are either injected into the model's lowest
19 layer or, in the case of aircraft emissions and some biomass burning emissions, injected into
20 a number of model levels. The emissions are added to the appropriate UKCA tracers (see
21 Table 1) and mixed simultaneously by the boundary-layer mixing scheme (Section 2.1). While
22 boreal and temperate forest and deforestation emissions (van Marle et al., 2017) of black
23 carbon (BC) and organic carbon (OC) are considered 'high-level' (Mulcahy et al., 2019) and
24 are spread uniformly up to level 20 (~3 km in L85), all gas-phase biomass burning emissions
25 are added to the surface layer.

26
27 For anthropogenic emissions, we make use of historical (1750–2014) annual emissions of
28 reactive gases from the Community Emissions Data System (CEDS; Hoesly et al., 2018) that
29 were prepared for use in CMIP6. The CEDS emissions are generally greater than those of
30 other emission datasets (e.g. Lamarque et al., 2010) for the years that are used in the
31 simulations evaluated here (i.e. 2005-2014). Biomass burning emissions are taken from van
32 Marle et al. (2017). They combined satellite observations from 1997 with various proxies and
33 output from six fire models participating in the Fire Model Intercomparison Project (FireMIP;
34 Rabin et al., 2017) to provide a complete dataset of biomass burning emissions from 1750 to
35 2014 for use in CMIP6. As was the case for anthropogenic emissions, emissions from the
36 years 2005-2014 are used here. For both anthropogenic and biomass burning, the emissions
37 were re-gridded from their native resolution to N96L85 while conserving global annual totals
38 and seasonal cycles. Emissions of all C₂ and C₃ VOCs are included as ethane and propane,
39 respectively.

40
41 For natural emissions which are not simulated, *offline* emissions are prescribed through the
42 provision of pre-computed fluxes. For example, oceanic emissions of CO, ethane (including
43 ethene (C₂H₄)), propane (including propene (C₃H₆)) are taken from the POET (Granier et al.,
44 2005) inventory for the year 1990 which contains one annual cycle with 12 monthly fluxes.
45 These fluxes are applied perpetually to all years of the time series. Biogenic emissions of
46 acetaldehyde (MeCHO) make use of combined emissions of MeCHO and other aldehydes
47 from the MACCity-MEGAN emissions inventory (Sindelarova et al., 2014); biogenic emissions
48 of CO, HCHO, MeOH, and propane (including C₃H₆) are also taken from this inventory. For

1 biogenic acetone emissions, emissions of acetone and *other* ketones from the MACCity-
2 MEGAN emissions inventory (Sindelarova et al., 2014) are combined. Based on the years
3 2001-2010, a monthly mean climatology is derived and applied to all years (see Section 3 for
4 the implementation of the emission in the model). Finally, soil emissions of NO_x are distributed
5 according to Yienger and Levy (1995) and scaled to give a global annual total of 12.0 Tg NO/yr
6 and again perpetually applied to all years.

7 8 **2.6.2 Biogenic VOC emissions**

9 In the standard configuration of UKCA StratTrop in UKESM1, emissions of organic compounds
10 from the natural environment (BVOC) are added to UKCA interactively (Sellar et al., 2019a).
11 Specifically, emissions of isoprene (C₅H₈) and (mono-)terpenes are *online*, the latter
12 represented by a lumped compound in UKCA with the formula C₁₀H₁₆ and a corresponding
13 molecular weight of 136 g mol⁻¹, are calculated by the interactive biogenic VOC (iBVOC)
14 emission model (Pacifico et al., 2011). Emission fluxes are passed to UKCA at every model
15 time step.

16
17 In iBVOC the emissions of isoprene are coupled to the gross primary productivity of the
18 terrestrial vegetation (Arneth et al., 2007; Pacifico et al., 2011). The biogenic emission of all
19 other organic compounds included in the iBVOC model, i.e., (mono-)terpenes, methanol, and
20 acetone, follow the original model described in Guenther et al. (1995). Note that the current
21 configuration of UKCA used in UKESM1 does not make use of the interactive emissions of
22 methanol or acetone; these are *offline* as discussed in Section 2.6.1. To the best of our
23 knowledge, in the case of the non-isoprene biogenic VOCs there exists no equivalent process-
24 based formulation for an interactive BVOC emission model applicable to Earth System Models
25 (ESMs).

26
27 For present-day conditions total global annual emissions of isoprene amount to 495.9 (±13.6)
28 Tg(C) yr⁻¹. This number represents the 10-year average annual total emission strength and
29 the uncertainty quantified by the standard deviation over the 10-year period between 2005
30 and 2014 taken from a historic run with UKESM1 (Sellar et al., 2019a). This is in good
31 agreement with estimates reported for other emission models (e.g. Arneth et al., 2008;
32 Guenther et al., 2012; Messina et al., 2016; Müller et al., 2008; Sindelarova et al., 2014;
33 Stavrakou et al., 2009; Young et al., 2009). For the global annual total (mono-)terpene
34 emissions, iBVOC calculates 115.1 (±1.6) Tg(C) yr⁻¹ over the same period of model simulation.
35 This model estimate is in reasonable good agreement with the literature (e.g., Folberth et al.,
36 2006; Lathièrè et al., 2006; Arneth et al., 2007, 2011; Acosta Navarro et al., 2014; Sindelarova
37 et al., 2014; Bauwens et al., 2016; Messina et al., 2016).

38
39 In the configuration of UKCA StratTrop used in UKESM1, isoprene is included in the gas-
40 phase chemistry but does not contribute to the formation of secondary organic aerosol (SOA).
41 Emissions of (mono-)terpenes are oxidised using a fixed yield approach (e.g. Kelly et al., 2018)
42 to form SOA in the GLOMAP-mode aerosol scheme - see Table S1 and Mulcahy et al. (2019)
43 for a detailed description and evaluation.

44 45 **2.6.3 Emissions of NO_x from lightning**

46 The lightning NO_x emissions scheme in UKCA StratTrop is based on the cloud top
47 parameterisation proposed by Price and Rind (1992). Based on satellite data and storm
48 measurements, the lightning flash density is parameterised as:

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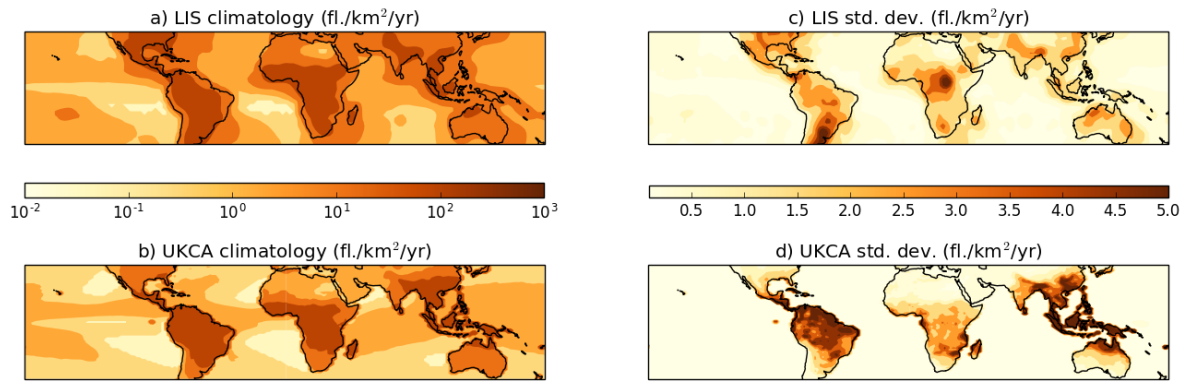
$$F_l = 3.44 \times 10^{-5} H^{4.9} \quad \text{Eq. 5}$$

$$F_o = 6.2 \times 10^{-4} H^{1.3} \quad \text{Eq. 6}$$

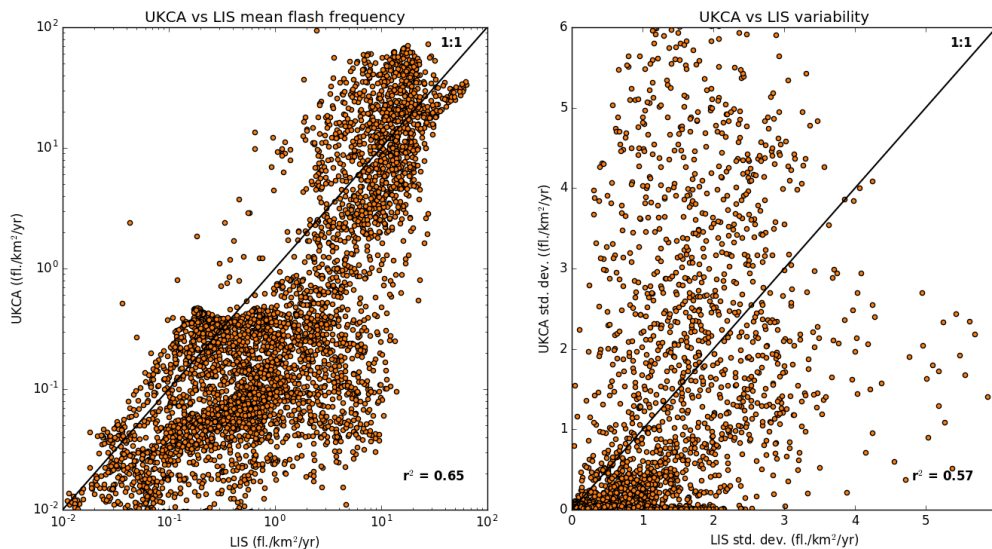
where F is the flash density (flash min^{-1}), H is the cloud top height (km), and the l and o subscripts are used to represent the land and ocean, respectively, and to distinguish between the updraft velocities experienced over the two surfaces. The scheme also differentiates between cloud-to-cloud and cloud-to-ground flashes based on the grid cell latitude (Price and Rind, 1993) and is resolution-independent by the implementation of a spatial calibration factor (Prince and Rind, 1994). A minimum cloud depth of 5 km is required for NO_x emissions to be activated and is diagnosed on a timestep basis from the physical model's convection scheme. For NO_x production, the parameterisation assumes that the production efficiency per unit of energy discharged is $25 \times 10^{16} \text{ molec (NO) J}^{-1}$, with the energy discharged from cloud-to-ground flashes ($3.0 \times 10^9 \text{ J flash}^{-1}$) being approximately 3 times greater than that for cloud-to-cloud ($0.9 \times 10^9 \text{ J flash}^{-1}$) flashes (Schumann and Huntrieser, 2007).

This implementation is identical to that implemented in HadGEM2-ES (Collins et al., 2011) by O'Connor et al. (2014) except that NO_x emissions are now distributed linearly in altitude in $\log(\text{pressure})$ rather than linearly in pressure. Whereas global annual lightning emissions in HadGEM2-ES were inadvertently too low (O'Connor et al., 2014; Young et al., 2013), here, the emissions have been scaled to give an average global annual emission rate of 5.93 and 5.98 Tg N yr^{-1} over the period 2005 to 2014 in the free-running and nudged simulations, respectively. When compared with anthropogenic, biomass burning and natural emissions, lightning contributes approximately 10 % to the global annual NO_x emission rate, consistent with estimates from Schumann and Huntrieser (2007).

Figure 1 shows tropical distributions of decadal mean annual flash density as observed by the Lightning Imaging Sensor (LIS) on board the Tropical Rainfall Measuring Mission (TRMM) satellite (Mack et al., 2007) in comparison with the free-running simulation being evaluated here (see Section 3 for details). It demonstrates that UKCA is capable of capturing the broad features of the observed climatology, with peak densities over S. America, Africa, and East Asia; the spatial coefficient of determination (R^2) between the modelled and observed climatology is 0.65 and 0.69 in the free-running and nudged (not shown) simulations, respectively. However, the model tends to be biased low in regions of low flash density (e.g. over the oceans and towards the extra tropics) compared to the observations (Figure 2), consistent with the assessment of Finney et al. (2014). In considering the variability, the spatial R^2 between the modelled and observed standard deviation is 0.57 and 0.59 in the free-running and nudged simulations, respectively. The variability from UKCA is comparable in magnitude to that observed over Africa, albeit displaced geographically. Over the Maritime continent and S. America, for example, UKCA overestimates the variability relative to the LIS observations.



1
 2 **Figure 1.** Tropical distribution of the LIS-observed climatological annual mean lightning flash density
 3 over the period 1999-2013 in a) in comparison with the modelled annual mean climatology from the
 4 period 2005-2014 in b). The corresponding standard deviation of the observed and modelled
 5 climatologies are shown in c) and d), respectively.
 6



7
 8 **Figure 2.** Scatter plot of the modelled versus the LIS-observed multi-annual annual mean lightning
 9 flash density (left) and the standard deviation (right).
 10

11 Whilst the skill of the cloud top parameterisation is good relative to other parameterisations,
 12 (Finney et al., 2014) and the performance here in the free-running and nudged model
 13 simulations is consistent with that assessment, raising the diagnosed cloud top height over
 14 land to the power of 4.9 makes the cloud top parameterisation susceptible to model biases in
 15 cloud top height, as noted by Allen and Pickering (2002) and Tost et al. (2007). Lightning is
 16 potentially a key chemistry-climate interaction in Earth System Models but the sensitivity to
 17 how it is represented (i.e. using cloud top height (Banerjee et al., 2014) or ice-flux based
 18 parameterisations (Finney et al., 2018)) warrants further investigation. Indeed, Hakim et al.
 19 (2019) recently identified uncertainty in modelled lightning NO_x in the Indian subcontinent as
 20 being an important source of uncertainty in model simulations of tropospheric ozone in that
 21 region.
 22

2.6.4 Lower boundary conditions

Lower boundary conditions are provided at the surface for the chemical species CH₄, N₂O, CFC-11 (CFCl₃), CFC-12 (CF₂Cl₂), CH₃Br, H₂, and COS. Values for H₂ and COS are fixed at 500 ppb and 482.8 ppt, respectively (invariant with time). Values for the remaining species are specified using time series data provided for the 5th Coupled Model Intercomparison Project (CMIP5) for the greenhouse gas concentrations (see *RCP webpage in references*). The values provided are valid on the 1st July for each year specified, and are linearly interpolated in time to give daily values if data for more than one time-point is defined. CFC-11, CFC-12, and CH₃Br also contain contributions from other Cl and Br containing source gases which are not explicitly treated in the model to ensure that there is the correct stratospheric chlorine and bromine loading, with these contributing species given in Table 2. These values are converted into a two-dimensional “effective emission” field at each timestep that is used to fix the surface concentrations of these species.

Table 2. List of halocarbons (not explicitly treated in the model) contributing to the lower boundary conditions of CFC-11, CFC-12, and CH₃Br. Note that H-1211 contributes to both CFC-11 and CH₃Br as it contains both Cl and Br. Contributions are included by moles of Cl or Br.

CFC-11	CFC-12	CH ₃ Br
CCl ₄	CFC-113	H-1211
CH ₃ CCl ₃	CFC-114	H-1202
HCFC-141b	CFC-115	H-1301
HCFC-142b	HCFC-22	H-2402
H-1211		
CH ₃ Cl		

2.7 Coupling with other Earth System components

Secondary aerosol formation of sulphate and organic carbon in UKESM1 (Sellar et al., 2019a) is determined by oxidants (OH, O₃, H₂O₂, NO₃) modelled interactively by the UKCA StratTrop chemistry scheme. For further details on the oxidation of sulphate and SOA precursors, chemistry-aerosol coupling, and the scientific performance of the aerosol scheme (GLOMAP-mode; Mann et al., 2010) in UKCA and UKESM1, the reader is referred to Mulcahy et al. (2019).

In the HadGEM2-ES model (Collins et al., 2011) used for CMIP5, radiative feedbacks between UKCA modelled methane and tropospheric ozone concentrations were active (OC14); stratospheric ozone was prescribed and combined with the modelled interactive tropospheric concentrations. In UKESM1 (Sellar et al., 2019a), however, the coupling between the UKCA modelled radiatively active trace gases and the radiation scheme has been extended to include N₂O and stratospheric ozone (in addition to methane and tropospheric ozone). Although chlorofluorocarbons (CFCs) and hydrochlorofluorocarbons (HCFCs) are modelled in UKCA StratTrop, the radiation scheme cannot handle the speciation. Therefore, separate

1 lumped species (CFC12-eq and HFC134a-eq) are prescribed in the radiation scheme (see
2 Section 2.6.4 on how the lumping/mapping is done).

3 4 **2.7.1 Heterogeneous chemistry couplings**

5 In UKCA StratTrop as implemented in UKESM1, 5 different heterogeneous reactions are
6 included (see Table S4). These reactions occur on the modelled soluble aerosol surface area,
7 which in the troposphere is calculated interactively using GLOMAP-mode by summing over all
8 soluble aerosol modes. In the stratosphere (defined here as being 12 km above the surface)
9 the aerosol surface area comes from the stratospheric sulfate surface area density input
10 climatology, discussed in Sellar et al. (2019b). The combining of the stratospheric aerosol
11 surface area density from the climatology and the interactive components of GLOMAP-mode
12 is calculated at each UKCA time step and only the soluble aerosol modes simulated by
13 GLOMAP are included in the calculation.

14
15 Heterogeneous reactions are extremely important for simulating composition change in the
16 stratosphere (Keeble et al., 2014) and there is increasing attention to the simulation of these
17 processes in the troposphere too (e.g. Jacob et al., 2000; Lowe et al., 2015). One of the most
18 important tropospheric heterogeneous reactions is that of N_2O_5 on aerosol surfaces (Jacob et
19 al., 2000). This reaction is complicated because of the dependence of the uptake parameter
20 (γ) on the composition of the aerosol as well as on temperature and relative humidity (Bertram
21 and Thornton 2009). Macintyre and Evans (2010) suggest that models that use high values of
22 $\gamma_{\text{N}_2\text{O}_5}$ (~ 0.1) overestimate the impact of changing aerosol loadings on tropospheric
23 composition through heterogeneous uptake. In UKCA StratTrop, $\gamma_{\text{N}_2\text{O}_5}$ is set at this higher
24 value, 0.1, throughout the atmosphere. In part this compensates for the fact that there is an
25 important missing aerosol surface in UKESM1 in the troposphere, in the form of nitrate
26 aerosol. The lack of nitrate aerosol is an issue for UKESM1 simulations of particulate matter,
27 particularly in regions with high levels of ammonia emissions. An improved understanding of
28 $\gamma_{\text{N}_2\text{O}_5}$ is needed to both understand current composition but also the combined impact of
29 changing gas and aerosol-phase composition. Whilst more sophisticated treatments of $\gamma_{\text{N}_2\text{O}_5}$
30 are available (e.g. Bertram and Thornton 2009) and have been included in versions of UKCA,
31 further work is required to improve this aspect of the mechanism for UKCA in UKESM1.

32 33 **2.7.2 Chemical production of H_2O**

34 There are many chemical reactions which consume or produce water vapour in the
35 troposphere and stratosphere. For example, reactions between the hydroxyl radical (OH) and
36 VOCs usually result in the production of a water molecule:

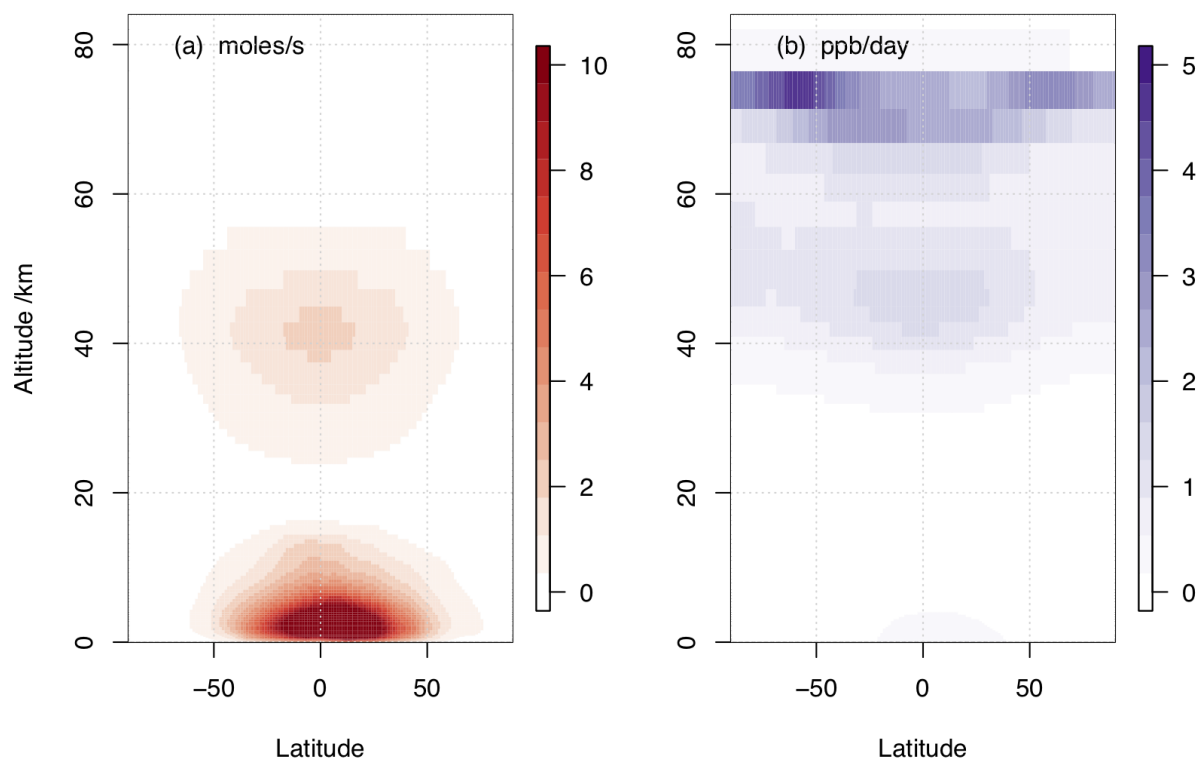


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39
40 In the troposphere the chemical source of water vapour is negligible compared with that from
41 the oceans and evapotranspiration from the Earth's land surface, but given the low
42 temperatures around the tropopause, chemically produced water is very important in the lower
43 stratosphere. Furthermore, the main source of chemical water in the middle to upper
44 stratosphere comes from the oxidation of CH_4 . Complete oxidation of CH_4 to CO_2 can result
45 in the net production of two water molecules.

1 In previous versions of UKCA, such as that used in HadGEM2-ES, the oxidation of CH₄ to
2 produce chemical water was neglected. Instead stratospheric water vapour was simulated
3 using the following simple relationship:

$$2 \times [\text{CH}_4] + [\text{H}_2\text{O}] = 3.75 \text{ (ppm)} \quad \text{Eq. 8}$$

4
5
6
7 where UKCA was used to calculate [CH₄]. In UKCA StratTrop as implemented in UKESM1 we
8 now include interactive H₂O production from all chemical reactions in the mechanism. In this
9 way UKCA now passes the water vapour field after the chemistry step back to the main climate
10 model where it is used in other routines. The annual mean zonal mean chemical production
11 of H₂O as simulated by UKESM1 is shown in Figure 3. There are two clear regions which
12 dominate where H₂O chemical production takes place, in the tropical lower troposphere and
13 the tropical upper stratosphere. In both regions the primary source of chemical water is the
14 oxidation of CH₄. Figure 3 compares the absolute production of chemical water (panel a) and
15 the production of chemical water as expressed in mixing ratio units (panel b). In this sense,
16 panel (b) shows the relative production of chemical water is greatest in the upper stratosphere.
17 The contribution of this source of stratospheric H₂O to the present day forcing of climate
18 relative to the pre-industrial period will be assessed in O'Connor et al. (2019).



20
21 **Figure 3.** Multiannual mean zonal mean production of H₂O from the UKCA StratTrop mechanism in
22 UKESM1. Panel (a) shows the production in moles/s and panel (b) in ppb/day, highlighting the larger
23 relative source of water from chemical processes in the upper atmosphere.

24 2.7.3 Future couplings

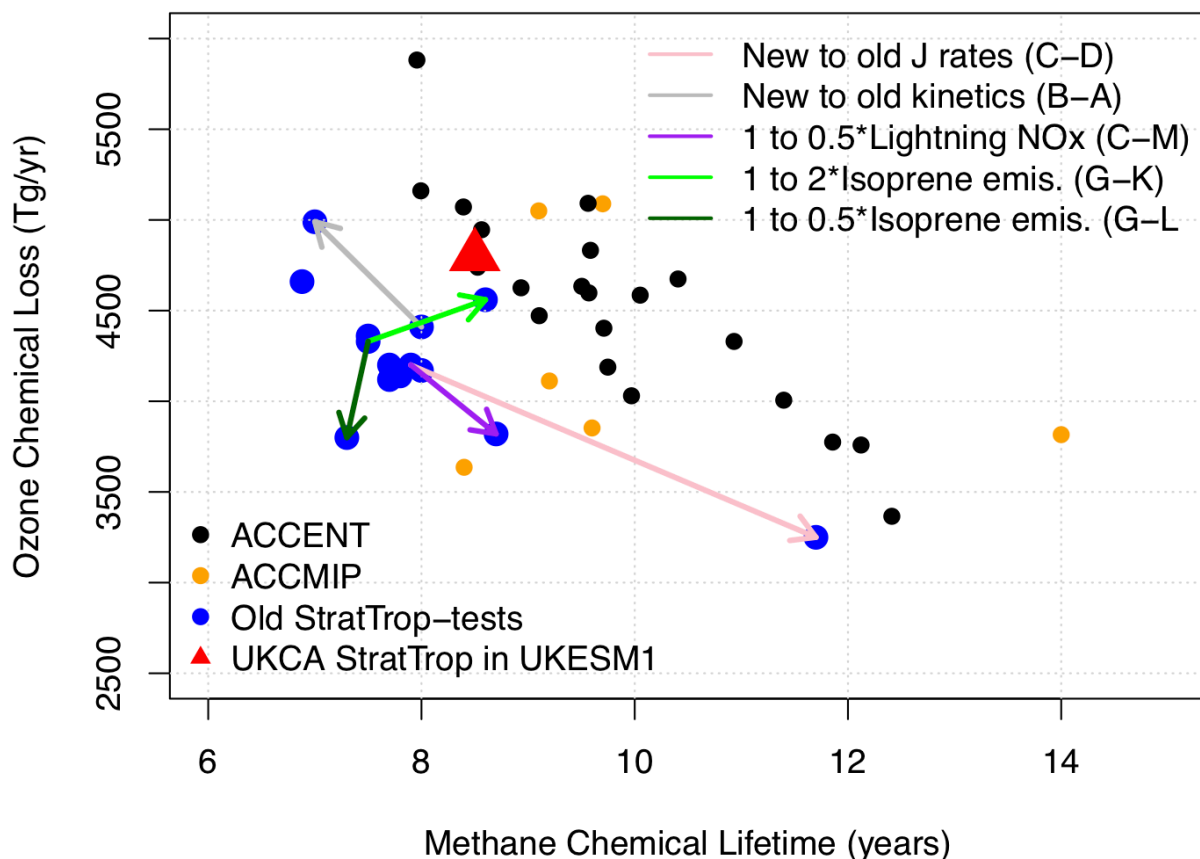
25 Although UKESM1 (Sellar et al., 2019a) represents a significant enhancement in the
26 representation of atmospheric chemistry and Earth System interactions, a number of key
27 interactions are not included. For example, the coupling of aerosols with Fast-JX is omitted
28 despite the impact of aerosols on tropospheric photochemical production of ozone (e.g. Xing
29

1 et al., 2017; Wang et al., 2019). This development is currently underway and will be included
2 in future versions of UKCA and UKESM. Ozone damage to natural and managed ecosystems
3 (e.g. Ashmore, 2005) has an important impact on the strength of carbon uptake by vegetation
4 (Sitch et al., 2007; Oliver et al., 2018) and has yet to be implemented. In addition, although
5 the terrestrial carbon cycle considers nitrogen availability/limitation, nitrogen deposition rates
6 are prescribed in UKESM1; future work will include implementing a nitrate aerosol scheme in
7 GLOMAP-mode and coupling deposition of both oxidised and reduced nitrogen from the
8 atmosphere to the terrestrial biosphere.

11 **2.8 Historic development of the chemistry scheme**

12 During the development of the StratTrop chemistry scheme, several simulations were run to
13 test the scheme and its sensitivity to different (a) rate coefficients (updating the JPL and IUPAC
14 recommendations), (b) reactions (by looking at the sensitivity to specific reactions associated
15 with isoprene oxidation (Archibald et al., 2011) and the reaction between HO₂ and NO
16 (Butkovzkaya et al., 2005; 2007; 2009)), (c) treatment of photolysis, (d) emissions and (e)
17 deposition parameters. These one-at-a-time simulations are outlined in Table S9 in the
18 Supplement. It should be noted that these simulations provide an ensemble of opportunity;
19 they were not designed to probe model sensitivity in a targeted way. However, they result in
20 some useful information which helped the development of the StratTrop mechanism. These
21 simulations made use of an older version of the MetUM and earlier atmosphere-only version
22 of UKCA, which is now deprecated. That version of UKCA ran at a lower resolution than the
23 version discussed in this paper and used in UKESM1 (about half the resolution). Results from
24 these simulations are shown in Figure 4 where they are compared against results from model
25 intercomparison studies (further analysis of the model sensitivity tests is presented in the
26 Supplement Figures S1 to S6). Figure 4 focuses on a subset of the full range of experiments
27 performed but contextualises these by comparing to results from the ACCENT simulations
28 discussed in Stevenson et al. (2006) (black dots) and the ACCMIP simulations discussed in
29 Young et al. (2013) (orange dots). In addition to the early sensitivity tests (the blue dots in
30 Figure 4), we also show the results from the simulations presented here, labelled UKESM1
31 (red triangle in Figure 4). The figure focuses on the relationship between methane lifetime and
32 ozone chemical loss, important metrics for representing key sources and sinks of tropospheric
33 OH (Wild 2007). Both metrics are calculated by masking out the stratosphere. The methane
34 lifetime is calculated by dividing the burden of methane in the model by the reaction flux
35 between methane and OH in the troposphere and so represents the lifetime with respect to
36 OH in the troposphere. The ozone loss is calculated by summing the reaction fluxes which are
37 key for O₃ loss in the troposphere (reactions of O₃ with HO_x species and the reaction between
38 O(¹D) and H₂O). The experiments outlined in Table S9 and shown in Figure 4 emphasise that
39 the range in O₃ loss and CH₄ lifetime spanned by changing aspects of the UKCA model span
40 a range as wide as that covered by the ACCMIP models (Young et al., 2013). In other words,
41 the *ensemble of opportunity* from the early tests of the UKCA StratTrop scheme span as wide
42 a range in the metrics presented as the structurally different ACCMIP and ACCENT models.
43 Interestingly, the UKESM1 simulations discussed in this paper in detail lie close to the
44 ACCENT ensemble (black dots), yet the early test simulations using the same chemical
45 mechanism but an earlier version of the MetUM model do not (the blue cluster of dots). This

1 highlights that structural changes in the underlying meteorological model can substantially
 2 influence key metrics of atmospheric composition through changes in the distribution of
 3 clouds, water vapour and other key variables.
 4



5
 6 **Figure 4.** Comparison of early tests of the StratTrop scheme running in an older version of UKCA (blue
 7 dots) with the scheme applied in UKESM1 (red triangle; free running simulation) and other CCMs which
 8 took part in the ACCMIP intercomparison (orange dots) and CTMs which took part in the ACCENT
 9 intercomparison (black dots). The letters in the legend (i.e. B-A) refer to the experiments outlined in
 10 Table S9.

11
 12 These sensitivity studies highlight some important points. Simulations using kinetic data
 13 recommendations from IUPAC and JPL updated from 2005 to 2011 led to a decrease in model
 14 methane lifetime and an increase in ozone chemical loss flux (grey arrow), indicating
 15 increased photochemical activity. The attribution of which rate coefficients were dominant in
 16 this behaviour is outside of the scope of this work. Similarly, we note that the metrics analysed
 17 are sensitive to lightning NO_x (Banerjee et al., 2014); decreasing the lightning NO_x emissions
 18 by 50 % (to ~3 Tg/yr) results in an increased methane lifetime of ~1 year (purple arrow). Figure
 19 4 also highlights a non-linear response in the simulations to changes in isoprene emissions;
 20 scaling them by a factor of two (100% increase and 50% decrease, green arrows) leads to a
 21 highly non-linear response in the metrics analysed. Finally we note that the change which had
 22 the biggest impact on the metrics was switching to the FAST-JX photolysis scheme (Telford
 23 et al., 2013) from pre-calculated photolysis rates and a look-up table (pink arrow). The main
 24 reason for this is that the pre-calculated photolysis rates had underestimated rates for the

1 photolysis of O₃ to O(¹D). This behaviour has been documented previously (Voulgarakis et al.,
2 2009; Telford et al., 2013).

3
4 In addition to the tests described above we found during the testing of the StratTrop scheme
5 that inclusion of the termolecular reaction:



7 which has been shown to exhibit both pressure and water vapour dependence (Butkovzkaya
8 et al., 2005; 2007; 2009), led to large changes in the metrics analysed in Figure 4 (see Section
9 1.2 of the Supplement for further details). Previous modelling work highlighted that this could
10 have an important impact on the simulation of ozone (Cariolle et al., 2008). However, owing
11 to uncertainty in its recommendation between the recent evaluations by JPL and IUPAC we
12 have omitted it from the StratTrop scheme used in UKESM1.

13 14 **3.0 Model simulations to evaluate UKCA StratTrop in UKESM1.**

15 In this section, we discuss a series of simulations that have been performed to evaluate the
16 performance of the UKCA StratTrop scheme in UKESM1. These simulations link closely to
17 the UKESM1 historical and AMIP simulations by using similar inputs, e.g. emissions, and
18 crucially the version of UKCA StratTrop is identical to that used in UKESM1 (Sellar et al.,
19 2019a).

20
21 Simulations analysed in this paper have been carried out with an atmosphere-only
22 configuration of UKESM1 (Sellar et al., 2019a). The sea surface temperatures and sea ice
23 cover used to drive the model are those specified for the historical period by the 6th Coupled
24 Model Intercomparison Project (CMIP6 project; Durack et al., 2016). Land cover fraction,
25 vegetation canopy height and leaf area index (LAI) have been provided as multi-annual
26 monthly mean climatologies, derived from a historical simulation of UKESM1 which includes
27 the dynamic vegetation model TRIFFID (Cox, 2001). Anthropogenic and biomass burning
28 emissions of ozone precursors are prescribed on a monthly basis using a 2005-2014
29 timeseries from Hoesly et al. (2018) (see section 2.6) and van Marle et al. (2017), respectively.
30 Land-based biogenic emissions not simulated within the JULES model (e.g. CO) are provided
31 as monthly climatologies for the period 2001-2010 from the MEGAN-MACC dataset
32 (Sindelarova et al 2014), supplemented by soil NO_x emissions based on Yienger and Levy
33 (1995) and oceanic emissions from POET. Greenhouse gas concentrations for CFC-12, CH₄,
34 CO₂, HFC-134 and N₂O are derived from the dataset generated by Meinshausen et al (2017)
35 for CMIP6. Concentrations of other CFCs seen only by UKCA are derived from the same
36 dataset but described in more detail under Lower Boundary Conditions (Section 2.6.4). The
37 model is initialised using output after nearly 150 years of the UKESM1 coupled historical
38 simulation. The land surface setup used in this paper is based on a 27 sub-grid tile
39 configuration including 13 plant functional types (three broadleaf tree tiles, two needleleaf tree
40 tiles, three C3-grass tiles including crops, three C4-grass tiles including crops, and two tiles
41 representing shrubs), one water tile (to represent lakes), one tile for bare soil, one urban tile
42 and 11 land ice tiles.

43
44 Two simulations have been carried out using the atmosphere-only configuration, covering
45 January 1999 to December 2014. The first is a *free-running* (FR) simulation where the
46 meteorology is allowed to evolve independently based on the influence of the aforementioned
47 forcing agents. The second is a *Nudged* (ND) simulation where the meteorology, though under

1 the same forcings as the FR simulation, is in addition relaxed toward the ECMWF's ERA-
2 Interim reanalysis (Dee et al., 2011) using the nudging functionality in the MetUM (Telford et
3 al., 2008). Nudging is applied to model temperature and winds from about 1.2 km (to be
4 generally free of the boundary layer) to 65 km (maximum height of ERA data), using an e-
5 folding relaxation timescale of 6 hours. In the following section, output from the ND simulation
6 will mainly be used for the comparison of modelled fields with observations, unless otherwise
7 stated, in order to reduce biases. On the other hand, the FR simulation will be useful to
8 document some key performance indicators such as the tropospheric oxidising capacity (OH
9 concentrations and methane lifetime) or the middle atmosphere age of air.

10
11 For both simulations, output from the first 6 years is considered as spin-up and analysis from
12 the years 2005-2014 inclusive is presented in this paper. Model fields used in the analysis
13 have been output mainly as monthly means. In addition, some aerosol-related fields were
14 produced at daily and 6-hourly intervals, while ozone, nitric acid and nitrogen dioxide at the
15 surface were produced at hourly intervals.

16
17 Table S10 provides a summary of the sectors contributing to the emissions of the nine
18 tropospheric ozone precursor species treated in UKCA StratTrop and their corresponding
19 global annual totals, averaged (mean) over the 2005-2014 time period covered by the two
20 simulations. Figures S7 and S8 show the multi-annual global annual mean distributions and
21 the seasonal cycle for different emission sectors and regions for NO and CO, respectively.

22 23 **4.0 Evaluation of model fields**

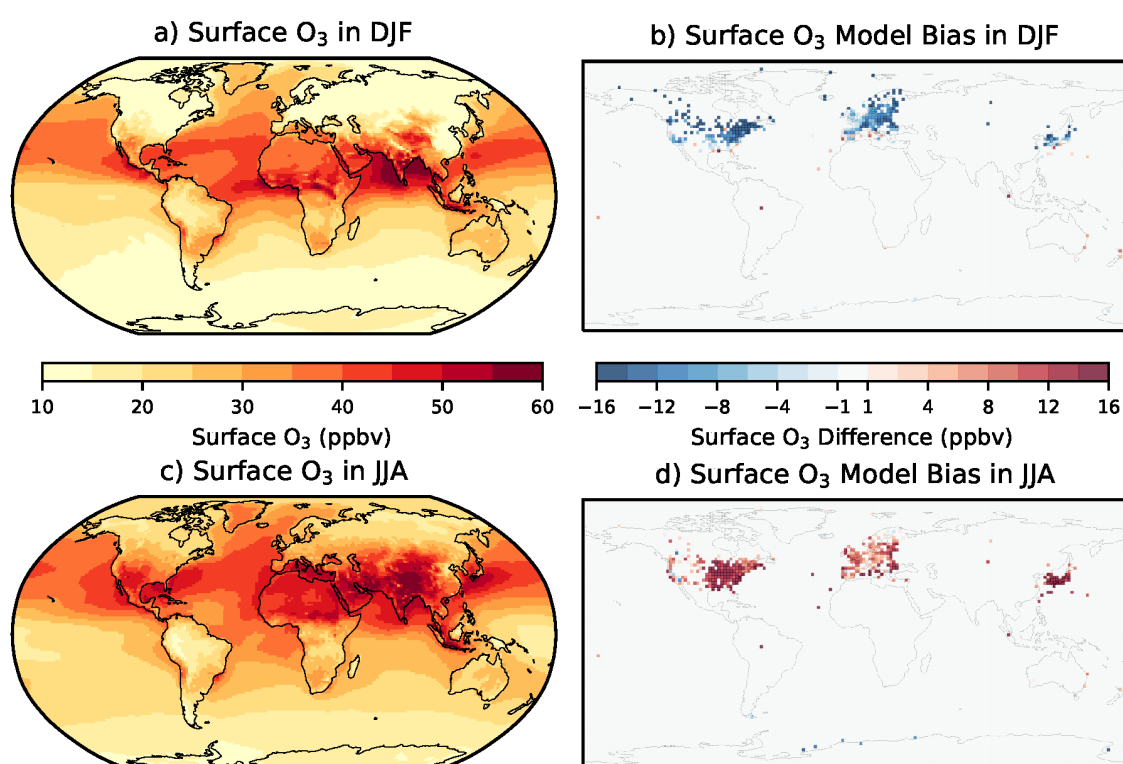
24 We start our evaluation of UKCA StratTrop in UKESM1 by assessing the performance of the
25 model in the troposphere, against surface observations, and build up the evaluation to focus
26 on tropospheric integrated quantities and stratospheric quantities before concluding with an
27 analysis of transport in the model. The evaluation presented here is mainly targeted at model
28 fields which are relevant to document the model's ability to reproduce tropospheric and
29 stratospheric ozone. Some additional evaluation of H₂O₂, important for the oxidation of SO₂
30 in the aqueous phase, is presented in Section 2.2 of the supplement.

31 32 **4.1 Evaluation of surface ozone against TOAR observations**

33 The surface O₃ concentrations in the ND simulation with UKCA StratTrop in UKESM1 for
34 December-January-February (DJF) and June-July-August (JJA) (seasonal means, calculated
35 from monthly means, over the 2005-2014 period) show elevated values across the tropics in
36 both seasons as well as in the northern mid-latitudes in JJA (Figure 5a and c). Maximum
37 surface O₃ concentrations of more than 60 ppb are simulated across the Middle East, Northern
38 Africa and South Asia in JJA due to large anthropogenic and biogenic sources of O₃
39 precursors. In DJF, surface O₃ concentrations are lower over the continental northern mid-
40 latitudes due to slow O₃ production and an enhanced O₃ removal from elevated NO_x
41 emissions. Meanwhile, surface O₃ concentrations are slightly higher over oceanic areas (North
42 Atlantic and North West Pacific) than over land in DJF, probably due to transport from the
43 stratosphere and a reduced chemical sink from weaker photolysis of O₃ (Banerjee et al., 2016).
44 Surface O₃ concentrations are slightly higher over some oceanic areas in JJA, indicating long
45 range transport from polluted continental areas.

46
47 Surface O₃ concentrations simulated in the nudged configuration of UKESM1 have been
48 evaluated over the period 2005-2014 by comparing to the gridded monthly mean rural

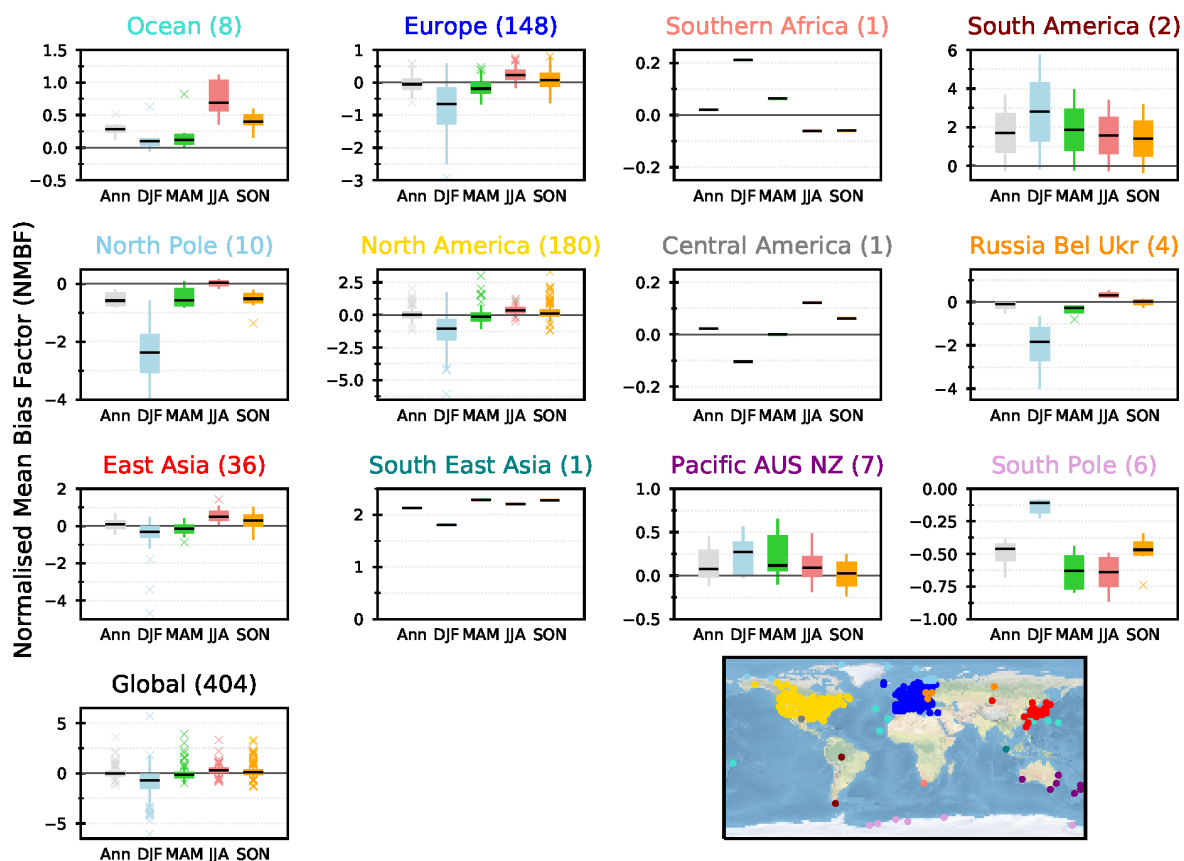
1 observations in the TOAR database over the same time period (Schultz et al., 2017). These
 2 data provide a global perspective on surface O₃ and is by far the most comprehensive surface
 3 O₃ database for use in evaluation of global models. However, the TOAR database does not
 4 provide globally uniform coverage and as such the evaluation of the model performance for
 5 surface O₃ over key regions, such as South Asia (Hakim et al., 2019), will be analysed in more
 6 specific follow up studies making use of bespoke datasets. Figure 5b and d shows that the
 7 model underpredicts surface O₃ concentrations in DJF and overpredicts O₃ in JJA across the
 8 northern midlatitudes, in a similar way to other global models (Young et al., 2018). Potential
 9 reasons for these discrepancies could be the coarse model resolution, associated errors in
 10 the emissions inventories, errors in the vertical injection of the emissions (for example we
 11 inject most of the NO_x near the surface which will titrate O₃), representation of VOCs in the
 12 chemistry scheme and uncertainties in O₃ loss processes (dry deposition).
 13



14
 15 **Figure 5.** Simulated (ND) seasonal mean surface O₃ concentrations in a) December-January-February
 16 (DJF) and c) June-July-August (JJA) over the 2005-2014 period. Difference between simulated and
 17 observed surface O₃ from the gridded TOAR database in b) DJF and d) JJA.
 18

19 Each grid point containing observations has been evaluated against the corresponding model
 20 values by calculating a normalised mean bias factor (NMBF, Yu et al., 2006). Figure 6 shows
 21 the distribution of NMBFs within a particular region for different seasons. Over northern
 22 midlatitudes (Europe, North America and East Asia) the model clearly underrepresents
 23 surface O₃ in DJF (by a factor of 1.5 to 2), suggesting excessive O₃ titration by NO_x. The model
 24 agrees better with observations in other seasons across these regions, with a slight
 25 overprediction in JJA. The limited available observations in other regions (<10 grid points)

1 makes it difficult to draw firm conclusions but suggests that UKCA StratTrop in UKESM1 tends
 2 to overpredict surface O₃ across the oceanic and southern hemisphere sites. The model
 3 consistently underpredicts observed surface O₃ at sites located in Antarctica, implying a lack
 4 of transport and a too low modelled O₃ lifetime in this region, particularly in March-April-May
 5 (MAM) and JJA.
 6

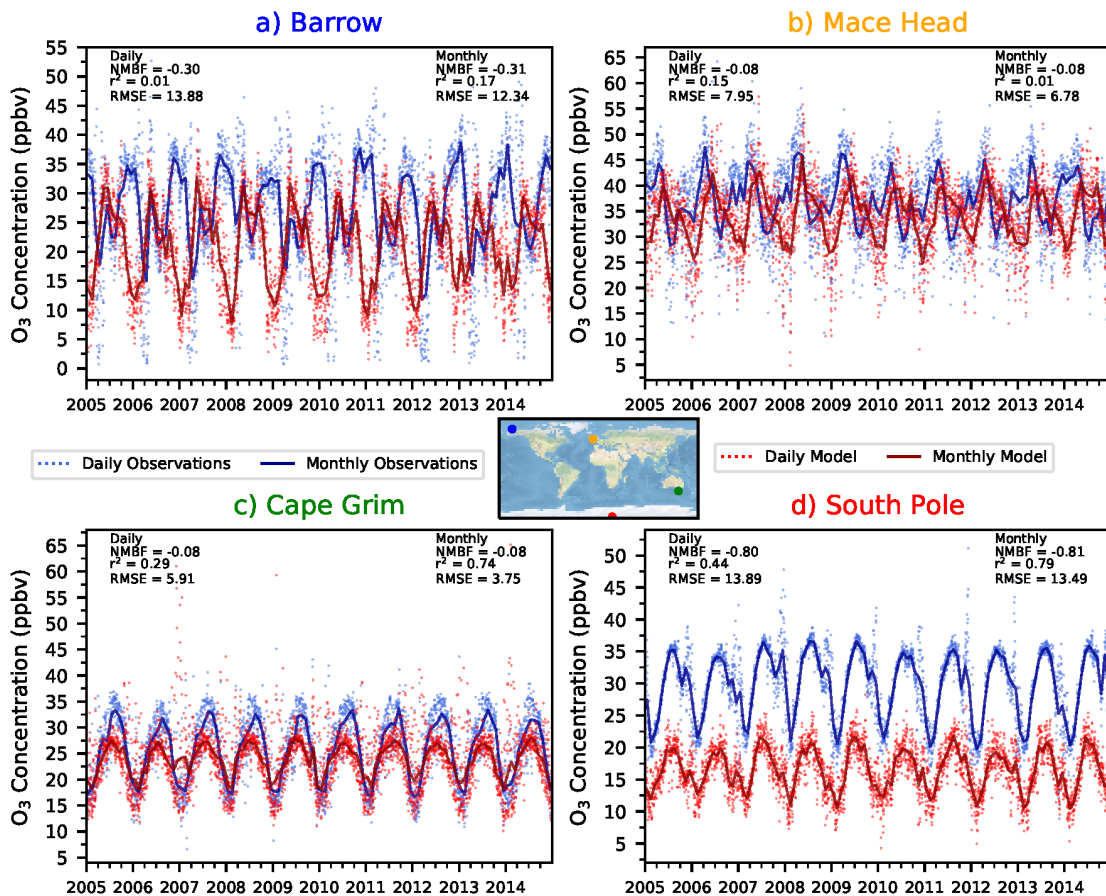


7
 8
 9 **Figure 6.** Normalised mean bias factors (NMBF) calculated for annual and seasonal means by
 10 comparing modelled concentrations of surface O₃ in the UKESM1 ND simulations to gridded
 11 observations from the TOAR database across each region over the 2005-2014 period. The solid line
 12 shows the median value for the region, the boxes show the 25th and 75th percentile values (Q1 and
 13 Q3), with the error bars extending from the boxes to cover the main body of the data, from (Q1-1.5 IQR)
 14 to (Q3+1.5 IQR) (IQR = interquartile range), and crosses representing outliers (values >1.5 x
 15 interquartile range). The total number of sites used for each region is shown in parenthesis.
 16 Comparisons on annual (grey), DJF (blue), MAM (green), JJA (red) and SON (orange) timescales are
 17 shown.
 18

19 Simulated daily and monthly mean surface O₃ concentrations over the period 2005-2014 from
 20 UKESM1 have been interpolated and compared to four individual measurement locations from
 21 the TOAR database with daily and monthly mean observations (Figure 7). UKESM1 is able to
 22 reproduce the seasonal cycle of surface O₃ observed at Cape Grimm ($r^2 = 0.74$, NMBF = -0.08)
 23 and South Pole ($r^2 = 0.79$, NMBF = -0.81), although it underestimates the magnitude in JJA at
 24 Cape Grimm and in all seasons at the South Pole. There is reasonably good model
 25 observational agreement in JJA at the two northern hemisphere sites (Barrow and Mace Head)
 26 (albeit with some disagreement in the phase), although in DJF the model underpredicts

1 surface O₃ at both sites. The surface model evaluation of UKESM1 at selected individual
 2 measurement locations exhibits a similar performance to that of HadGEM2-ES in O'Connor et
 3 al. (2014).

4
 5
 6



7
 8 **Figure 7.** Simulated (ND) and observed daily and monthly mean surface O₃ over the period 2005-2014
 9 at four individual monitoring locations of a) Barrow, b) Mace Head, c) Cape Grimm and d) South Pole

10

11 4.2 Dry deposition of ozone – comparison with HTAP models and observations

12 1030 Tg (O₃), around 20 % to 25 % of the gross chemical ozone production in the troposphere,
 13 is removed from the atmosphere in the ND simulation through dry deposition at the surface
 14 (Stevenson et al., 2006; Wild, 2007; Young et al., 2013; Hardacre et al., 2015). Uptake by
 15 terrestrial vegetation plays a crucial role, however, Hardacre et al. (2015) demonstrated that
 16 the oceans represent a very important sink, too. Much uncertainty still remains about the exact
 17 magnitude and many of the processes around ozone removal at the surface (e.g., Hardacre
 18 et al. 2015; Luhar et al., 2017). A thorough evaluation and, if necessary, re-calibration of ozone
 19 dry deposition models is, thus, critical in developing robust models of atmospheric
 20 composition.

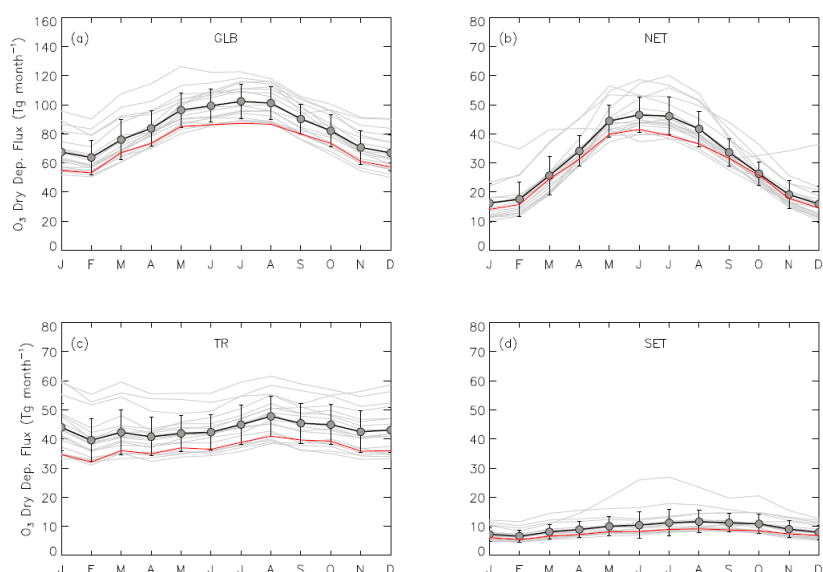
21

22 4.2.1 Comparison with the HTAP multi-model ensemble ozone deposition fluxes

23 Figure 8 shows a comparison of multi-annual average monthly mean ozone deposition
 24 modelled by UKCA StratTrop in UKESM1 with a multi-model ensemble of 15 HTAP
 25 atmospheric composition models (Hardacre et al., 2015). The StratTrop model data here are

1 taken from the ND simulation. Monthly mean ozone deposition is depicted for the entire global
 2 domain (Figure 8 a) and split into the northern extra-tropics, the tropics, and the southern
 3 extra-tropics, respectively, each representing a distinctly different deposition regime (Figure 8
 4 b-d). The solid black line and filled circles represent ensemble average monthly mean ozone
 5 deposition with the error bars indicating $\pm 1\sigma$ in the single model monthly mean ensemble; the
 6 solid grey lines represent single model monthly means from the HTAP models indicating the
 7 spread in the multi-model ensemble. The multi-annual average (10 years) monthly mean
 8 ozone dry deposition flux modelled by UKESM1-UKCA is shown as the red solid line.

9
 10 In general, ozone dry deposition from UKCA StratTrop in UKESM1 compares favourably with
 11 the HTAP multi-model ensemble falling nearly always within the 1σ -range of the HTAP multi-
 12 model average. UKCA StratTrop also correlates well with the multi-model average seasonal
 13 cycle for each of the depicted regions; however, a systematic low-bias is evident, particularly
 14 in the global and tropical domains (panels a and c in Figure 8). Most of the low bias occurs in
 15 the tropical region. Since the tropics are dominated by both a large ocean surface area and
 16 the most productive portion of the Earth's terrestrial vegetation in the form of the tropical rain
 17 forests of South America, equatorial Africa and the maritime continent, the tropical low bias in
 18 the model could be due to an underestimation of O_3 concentration, the stomatal ozone uptake
 19 by tropical rain forests or a similar underestimation of O_3 removal at the ocean's air-sea
 20 interface. The latter seems less likely in view of the relatively good performance in the southern
 21 extratropics which are also dominated by a large ocean surface.



24
 25 **Figure 8.** Multi-annual average monthly mean O_3 dry deposition fluxes for the global domain (a) and
 26 three latitudinal sections (b-d): northern extra-tropics (NET; 90N-30N; b), tropics (TR; 30N-30S; c), and
 27 southern extra-tropics (SET; 30S-90S; d) for 15 models participating in the HTAP model
 28 intercomparison. Multi-model ensemble average (solid black line and filled circles) and single model
 29 monthly means (grey solid lines) were provided by Hardacre et al. (2015). Error bars indicate $\pm 1\sigma$ in the
 30 single model monthly means. Solid red line shows UKCA ND StratTrop multi-annual average (2005-
 31 2014) monthly mean O_3 dry deposition fluxes. (Figure based on Hardacre et al. (2015)).

32 33 4.2.2 Comparison with observations of ozone deposition fluxes

1 Measurements of ozone dry deposition fluxes collected over extended periods of time are still
 2 very sparse, however, a number of long-term datasets exist. Hardacre et al. (2015) compiled
 3 a comprehensive dataset from available long-term and short-term observations. This
 4 comprehensive dataset has been adopted for our evaluation of O₃ dry deposition in UKCA
 5 StratTrop in UKESM1. Table 3 summarises the locations of all the measurement sites included
 6 in this comparison. A comparison of the dry deposition fluxes of ozone with observations at
 7 these 16 sites is presented in Figure 9. Some sites cover the seasonal cycle over several
 8 years (e.g., Castel Porziano, Harvard Forest, or Ulborg) and others only offer data spanning
 9 less than one month (e.g., Klippeneck, Le Dezert, or Viols en Levant).

10
 11

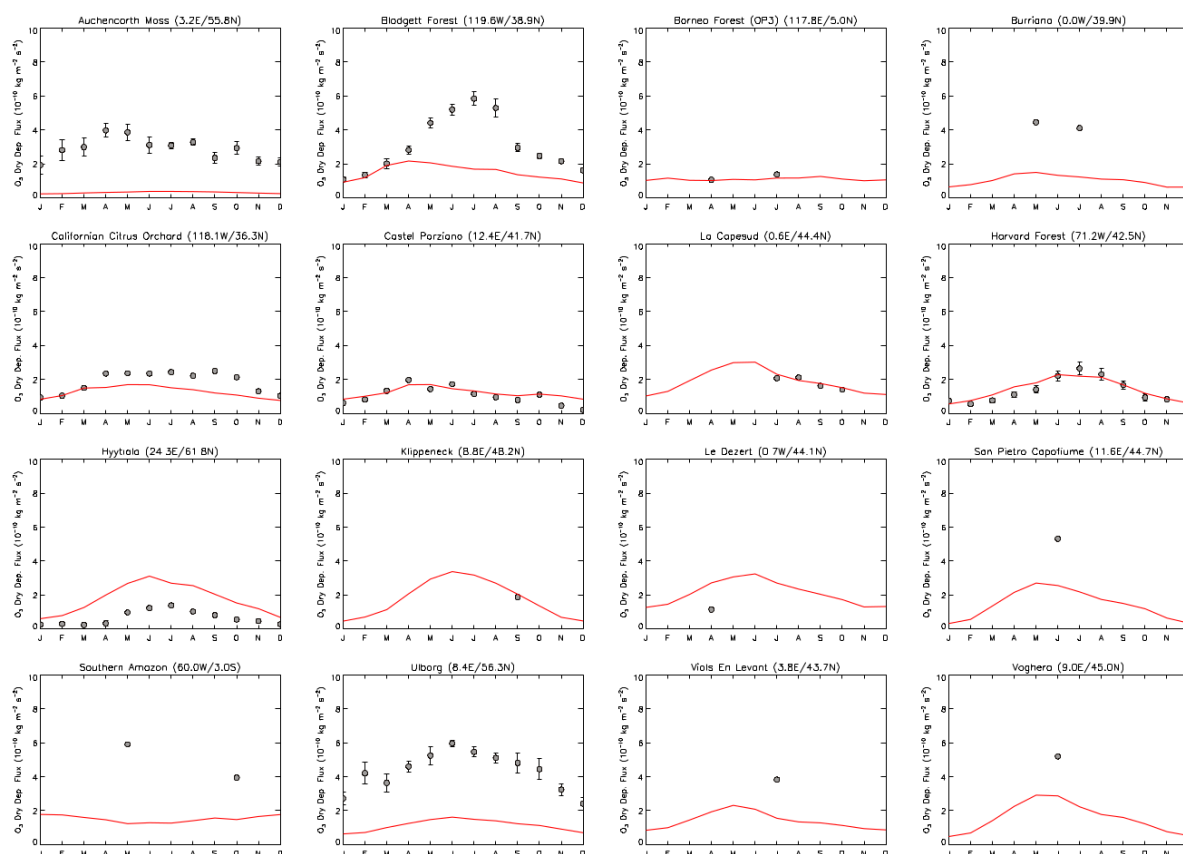
Table 3. Ozone surface dry deposition measurement sites (reproduced from Hardacre et al., 2015).

Site Name	Grid reference	Land cover	Sampling height (m)	Sampling Period	Ref.
Long-term sites					
Auchencorth Moss	55°47'N 3°14'E	Moorland	0.3-3.0	Oct 1995–Dec 2000	1
Blodgett Forest	38°53'N 120°37'W	Pine plantation	12.5	Jan 2001–Dec 2007	2
Citrus Orchard	36°21'N 119°5'W	Citrus Orchard	1.0–9.2	Oct 2009–Nov 2010	4
Castel Porziano	41°44'N 12°24'E	Holm Oak	35	Jan–Dec 2013	5
Harvard Forest	42°32'N 72°11'W	Mixed deciduous forest	30	Jan 1992–Dec 2001	7
Hyytiala	61°51'N 24°17'E	Scots Pine forest	23	Jan 2002–Dec 2003	8
Ulborg	56°17'N 8°25'E	Mixed coniferous	18, 36	Oct 1995–Dec 2000	10
Short-term sites					
Borneo OP3	45°8'N 117°51'E	Tropical forest	75	Apr, Jul 2008	1
Burriana	39°55'N 0°03'W	Citrus Orchard	10	16–29 Jul 1995 28 Apr–3 May 1996	3
La Cape Sud	44°24'N 0°38'E	Maize Crop	3.4, 3.7, 6.4	Jul–Oct 2007	6
Klippeneck	48°10'N 8°45'E	Grass	2, 8	1–22 Sep 1992	3
Le Dezert	44°05'N 0°43'E	Pine forest	37	16–18 Apr 1997	3
San Pietro Capofiume	44°39'N 11°37'E	Beet crop	8	15–22 Jun 1993	3
South-western Amazon	3°00'S 60°00'W	Tropical forest	53	May 1999 Sep–Oct 1999	9
Viols en Levant	43°41'N 3°47'E	Mediterranean shrub	37	16–24 Jul 1998	3
Voghera	45°01'N 9°00'E	Onion field	2.5	May–Jul 2003	11

12 References: 1) Fowler et al. (2001); 2) Fares et al. (2010); 3) Cieslik (2004); 4) Fares et al. (2012);
 13 5) Fares et al. (2014); 6) Stella et al. (2011); 7) Munger et al. (1996); 8) Rannik et al. (2012);
 14 9) Fan et al. (1990); 10) Mikkelsen et al. (2004, 2000); 11) Gerosa et al. (2007)
 15

1 Due to its removal via stomatal exchange and relative insolubility in water, O₃ dry deposition
 2 depends strongly on the underlying land surface type. Therefore, a reliable representation of
 3 ozone dry deposition in models requires not only the composition model to perform well. A
 4 robust model of the land surface including dynamic vegetation is also indispensable. The land
 5 surface representation in UKCA StratTrop in UKESM1 relies on JULES (Best et al., 2011;
 6 Clark et al., 2011). Thus, a comparison of ozone dry deposition (or any dry deposition process
 7 for that matter) reflects on the broader Earth system framework than just the atmospheric
 8 composition component alone.

9
 10



11
 12 **Figure 9.** Comparison of observed and modelled monthly mean ozone dry deposition fluxes. Grey
 13 circles indicate monthly mean ozone deposition fluxes at measurement sites (c.f. Table 3 for site
 14 details); error bars denote standard errors. Solid red lines represent modelled multi-annual average
 15 monthly mean O₃ deposition fluxes extracted from UKCA StratTrop ND in UKESM1 at the site locations
 16 by interpolation of the nearest grid boxes (averaged over all surface tiles in these grid boxes). Ozone
 17 dry deposition fluxes are given in 10⁻¹⁰ kg m⁻² s⁻¹, measurement data from Hardacre et al. (2015) and
 18 references therein.

19
 20 Overall, Figure 9 shows that the UKCA(StratTrop)/JULES/UKESM1 framework shows a
 21 reasonably good performance, albeit with some substantial model-to-observation deviations
 22 evident from Figure 9. At the Castel Porziano, La Cape Sud, and Harvard Forest sites the
 23 model reproduces well both magnitude and seasonal cycle of ozone dry deposition. To a
 24 somewhat lesser degree model performance is also good at the California Citrus Orchard and
 25 Hyytiälä sites. At both locations the model captures most of the seasonal cycle well but fails

1 to reproduce the magnitude of the flux fully. Interestingly, there is no systematic bias in the
2 model-to-obs deviations with respect to magnitude and land cover type.

3
4 Further locations with good model-to-obs agreement include the densely forested OP3 site in
5 Borneo and Klippeneck site in Germany. However, these sites only provide campaign data for
6 a limited period of time. The model shows very low skill in reproducing either magnitude or
7 seasonal cycle at three sites with long-term observational records; namely Auchencorth Moss
8 (Scotland, UK), Blodgett Forest (California, U.S.A), and Ulborg (Denmark). In all three cases
9 the model severely underestimates O₃ dry deposition fluxes. The model also shows a fairly
10 low skill in reproducing the seasonal cycle at these three sites. Potential reasons for the low
11 model skill at these long-term observation sites include modelled surface ozone levels,
12 deposition velocities and the appropriateness of the vegetation type, but more detailed
13 analysis is required to explore these further. However, by and large, the model performance
14 appears reasonable when compared to both observations and other models, although with an
15 overall negative bias.

16 17 18 **4.3 Model simulated methane and OH**

19 Here we discuss the performance of UKCA StratTrop modelled methane and OH distributions
20 in the troposphere. OH is the primary oxidising agent in the troposphere and is the key
21 determinant on the burden of methane in the troposphere (Monks et al., 2015).

22
23 A commonly cited indicator of tropospheric oxidising capacity, the tropospheric lifetime of
24 methane with respect to OH has been calculated for the FR simulation, averaged over the
25 entire length of the run. The modelled average tropospheric mean methane lifetime with
26 respect to OH oxidation is calculated to be 8.5 years (with a standard deviation of 0.1 years).
27 This value is in good agreement with the ACCMIP ensemble average of 9.7 ± 1.5 years (Naik
28 et al., 2013) (i.e. falling within one standard deviation of the ACCMIP ensemble). We note that
29 the methane lifetime for UKESM1 is much shorter than the methane lifetime for HadGEM2-
30 ES. Figure 4 shows this is largely down to improvement in the treatment of photolysis since
31 HadGEM2-ES (Telford et al., 2013).

32
33 We further focus our analysis on comparing the climatological distribution of OH as a function
34 of latitude and altitude (Figure 10).

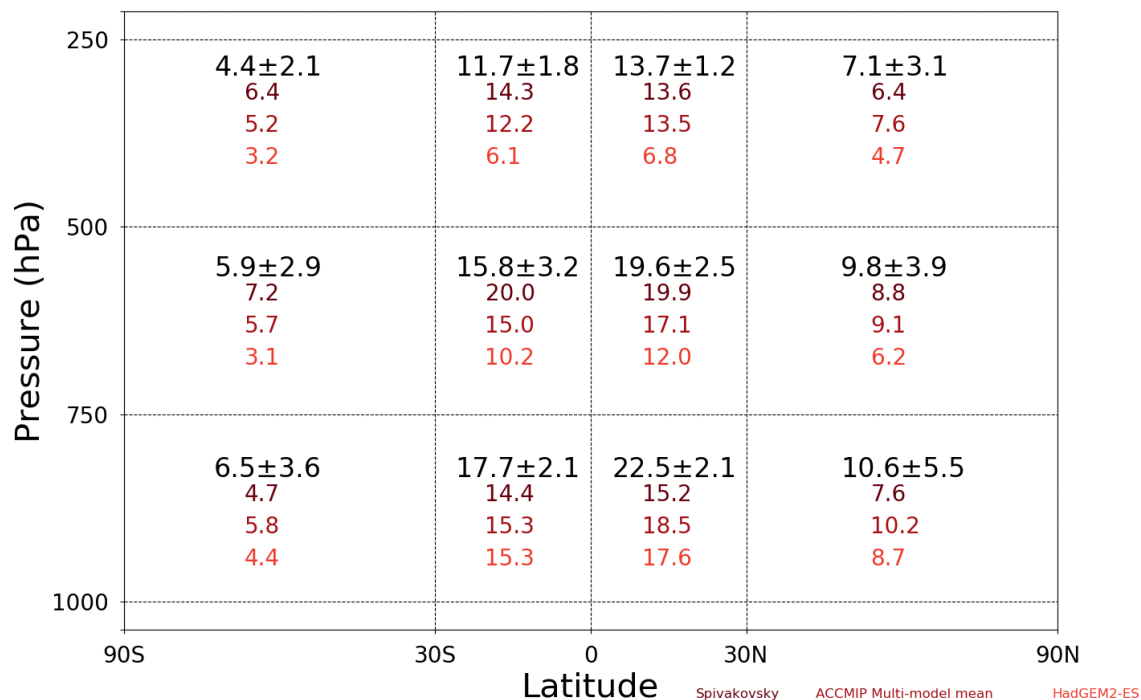
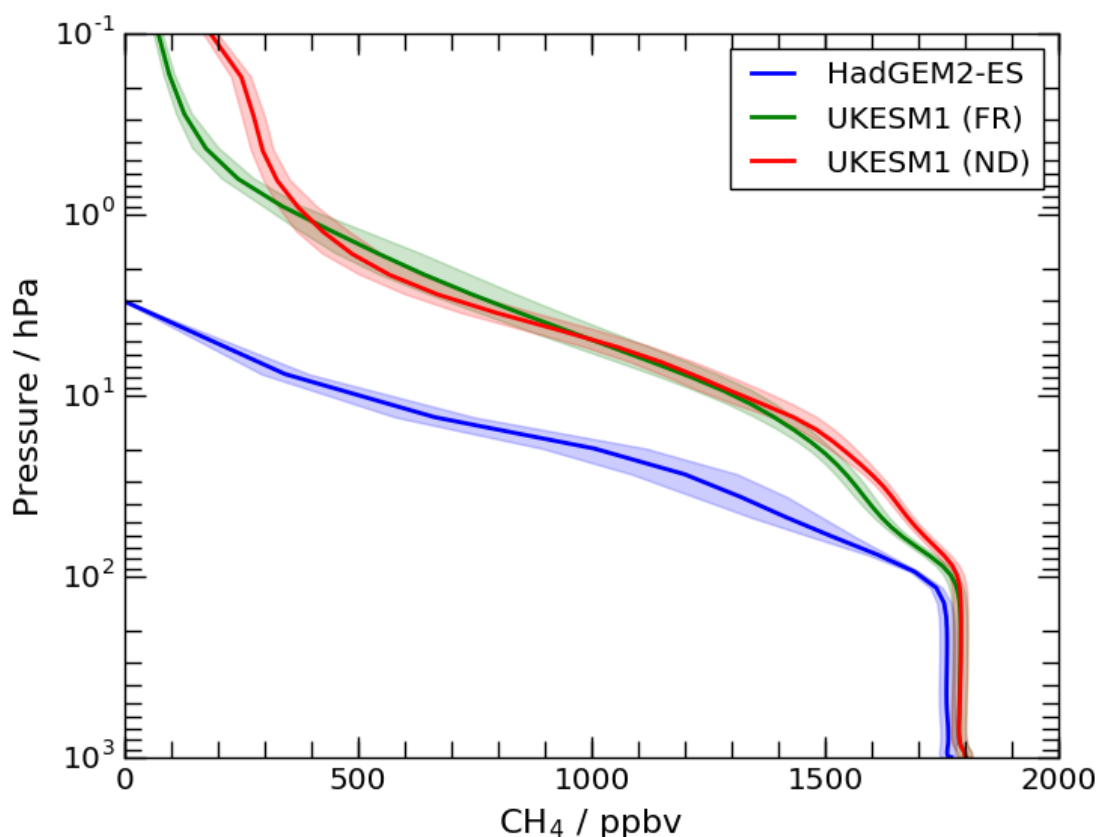


Figure 10. Evaluation of the UKCA StratTrop zonal distribution of tropospheric [OH] ($\times 10^5$ molecules cm^{-3}) in the FR simulation. Values plotted in black refer to the UKCA StratTrop multi-annual mean [OH] in each region of latitude and pressure range, with the values following being ± 1 standard deviation around the mean.

The FR UKCA StratTrop simulation result in a global mean tropospheric [OH] of 1.22×10^6 molecules cm^{-3} , averaged over the period 2005-2014. As with the methane lifetime, this value is slightly higher than the ACCMIP ensemble mean ($11.1 \pm 1.6 \times 10^5$ molecules cm^{-3}) but sits within the standard deviation of the ACCMIP ensemble mean (Naik et al., 2013). Figure 10 shows how the distribution of [OH] varies throughout the troposphere relative to the ACCMIP multi model mean, the HadGEM2-ES model and the data from Spivakovsky et al. (2000) who pioneered the development of [OH] climatologies in the troposphere. Compared against these data, UKCA StratTrop in UKESM1 performs well: The global tropospheric mean [OH] is within 10% of the ACCMIP ensemble mean. The model captures the latitudinal and vertical profiles found in the other data sets and agrees on the magnitude of [OH] in 10 of the 12 regions analysed (when considering the model uncertainty).

The [OH] is higher in UKCA StratTrop than in HadGEM2-ES, partly because of different emissions used in the HadGEM2-ES study, but also in part owing to the change in photolysis scheme (as discussed previously). UKCA StratTrop agrees better with the ACCMIP multi model mean than Spivakovsky or HadGEM2-ES, but the tropics from 1000-750 hPa are regions where the model consistently disagrees with the other datasets, simulating higher levels of OH in these regions. These regions of the troposphere are the regions where most CH_4 is oxidised and so high biases in the model here will tend to lead to lower CH_4 lifetimes than in observation-derived estimates.

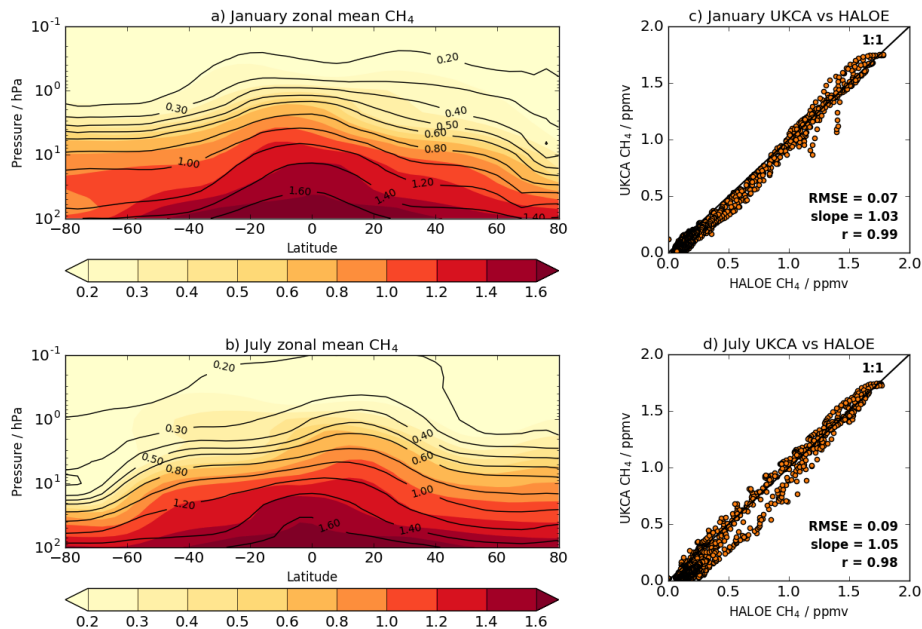
1 In the previous configurations of UKCA (MO09 and OC14), methane concentrations fell off too
 2 quickly with height above the tropopause; this was attributed to the stratospheric transport
 3 timescale being too long in the respective physical model. Comparisons of methane columns
 4 from the HadGEM2-UKCA coupled model with SCIAMACHY, for example, were too low and
 5 required modelled methane above 300 hPa to be overwritten with Halogen Occultation
 6 Experiment (HALOE, Russell et al., 1993) and Atmospheric Chemistry Experiment (ACE,
 7 Bernath et al., 2005) assimilated output from TOMCAT (Hayman et al., 2014). Figure 11 shows
 8 that the fall-off of methane with height in both the FR and ND simulations of UKESM1 is less
 9 rapid than in HadGEM2 and is consistent with the age of air in the model being comparable to
 10 that inferred from observations (Section 4.6). As comparisons with surface observations and
 11 SCIAMACHY (with its strong sensitivity to surface concentrations) are not appropriate here,
 12 because surface methane concentrations are relaxed to LBCs (Section 2.6.4), only
 13 comparisons with stratospheric observations are shown here.
 14



15
 16 **Figure 11.** Vertical profiles of the mean tropical ($\pm 10^\circ \text{N}$) modelled methane from multi-annual annual
 17 mean output from an atmosphere-only free-running simulation of HadGEM2-ES (OC14; blue) and
 18 atmosphere-only free running FR; green) and nudged (ND; red) simulations of UKCA StratTrop in
 19 UKESM1 (this study). The shading represents ± 1 standard deviation about the multi-annual mean.

20
 21 Figure 12 shows multi-annual zonal mean comparisons for January and July of modelled
 22 methane from the free-running (FR) simulation against the HALOE/Cyrogenic Limb Array
 23 Etalon Spectrometer (CLAES) climatology (Kumer et al., 1993). It indicates that UKCA
 24 StratTrop in UKESM1 is capable of simulating the absolute concentrations as well as the
 25 morphology of the observed distribution. The only exception to this is the tongue of methane-
 26 depleted air descending from the mesosphere over the SH high latitudes in July, which was

1 also evident in MO09. Nevertheless, UKESM1 is able to capture the observed vertical fall-off
 2 with height. There is an excellent 1:1 correspondence between the model and observations:
 3 the slope of the least squares fits for January and July are within 0.05 of unity, the correlation
 4 coefficients are greater than 0.98 and the root mean square errors between UKESM1 and the
 5 HALOE/CLAES climatology are less than 0.1 ppm for the free-running (Figure 12) and nudged
 6 (not shown) simulations.



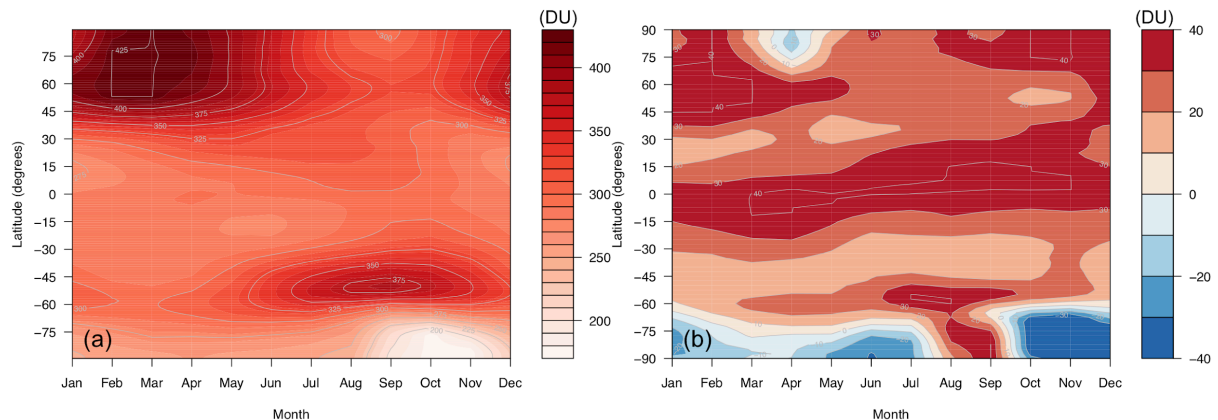
7
 8 **Figure 12.** Multi-annual (2005-2014) monthly zonal mean methane in ppm from the UKESM1 FR
 9 simulation in a) January and b) July, with scatter plots of modelled versus observed concentrations for
 10 January and July in panels c) and d), respectively. The coloured contours in a) and b) are from UKCA
 11 and the black contours are the HALOE/CLAES climatology. The scatter plots also include a 1:1 line and
 12 the root mean square error (RMSE), the slope of a least squares linear fit, and the correlation coefficient
 13 (r).

14 4.4 Comparison of total ozone column

15 Here we discuss the modelled total ozone column through analysis of the data from the FR
 16 simulation averaged over the 2005-2014 period. We note here that there is little difference
 17 between the ND and FR total column so for simplicity we focus on the FR data.
 18

19
 20 Figure 13 (panel a) shows the multi annual average total ozone column in Dobson Units as a
 21 function of latitude and time. As with most chemistry-climate models (Dhomse et al., 2018),
 22 UKCA simulates the main features of the total column well, with a minimum in the tropics and
 23 maxima at high latitudes during the hemispheric spring seasons. When compared with the
 24 total column ozone in older versions of UKCA (M09 Figure 9) the current model configuration
 25 simulates similar biases at high latitudes but a pronounced positive bias in the tropics. Figure
 26 13 (panel b) highlights that the tropical column is biased high by 30-40 DU when compared to
 27 the Bodeker climatology (Hassler et al., 2008) and the Antarctic ozone hole extends for too
 28 long in the model, leading to low biases in the austral summer. The high biases in total column
 29 ozone in the tropics are very likely in part to be driven by high biases of around 15 DU in the
 30 tropical tropospheric ozone column (see Section 4.5 below). The extra tropical biases may

1 well be related to this bias through the transport of ozone-rich air in the UTLS into this region
2 but further work is needed to resolve the causes of the bias in the total ozone column.



3
4 **Figure 13.** Evaluation of the FR UKCA total ozone climatology. Panel (a) shows the FR simulation
5 multi annual mean total column ozone climatology (in Dobson Units - DU). Panel (b) shows the
6 difference between the FR ozone climatology and the Bodeker ozone climatology v2.8.
7
8

9 **4.5 Comparisons with satellite retrievals of tropospheric columns of O₃, CO and NO₂**

10 Here we compare the results from the UKCA StratTrop runs against satellite data with a focus
11 on assessing performance in the troposphere. In all cases, the runs analysed are the nudged
12 dynamics runs (ND) discussed in Section 3. Nudging enables a more robust comparison
13 against the satellite observations as it reduces biases caused by circulation errors in the free
14 running model, although we note that it does not completely remove these biases (Orbe et al.,
15 2018; Chrysanthou et al., 2019). As well as nudging, the model output is sampled
16 instantaneously every three hours to allow for time and space sampling to the satellite data
17 locations. The comparison between the model and the observations is made using OMI-MLS
18 for the tropospheric column of O₃, MOPITT for the tropospheric column of CO and OMI for
19 tropospheric column of NO₂.

20
21 In the following analysis, the stratosphere has been removed by screening out regions where
22 the monthly mean ozone exceeded 125 ppb, the ozonopause; columns are calculated by
23 summing variables from the surface to the height at which the ozonopause starts. The model
24 ozone data presented here has not been corrected to account for optically thick clouds in the
25 troposphere which may affect retrieved ozone profiles (Ziemke et al. 2006) since averaging
26 kernel (AK) information is not available for the OMI-MLS dataset. As satellite measurement
27 errors were not available, we have used 2 × standard deviation of the retrievals to estimate
28 when the differences between modelled and observed ozone are significant. This implies that
29 the stippling area in the plots, corresponding to grid cells where |model bias| > satellite error,
30 could be reduced (i.e. better agreement with the observations) if the satellite error is added to
31 the 2 × stdev. The plots therefore show a ‘worst case scenario’.

32
33 The model fields have been co-located in time and space with the observations to reduce
34 representation errors. For each satellite retrieval, the nearest model grid box is sub-sampled
35 within 3 hours of the observation and the model profile interpolated onto the satellite pressure
36 grid. The satellite AKs (where available) are then applied to the model profile, to account for
37 the vertical sensitivity of the instrument. Then the model sub-columns are calculated and
38 summed between the surface and the tropopause to determine the co-located model

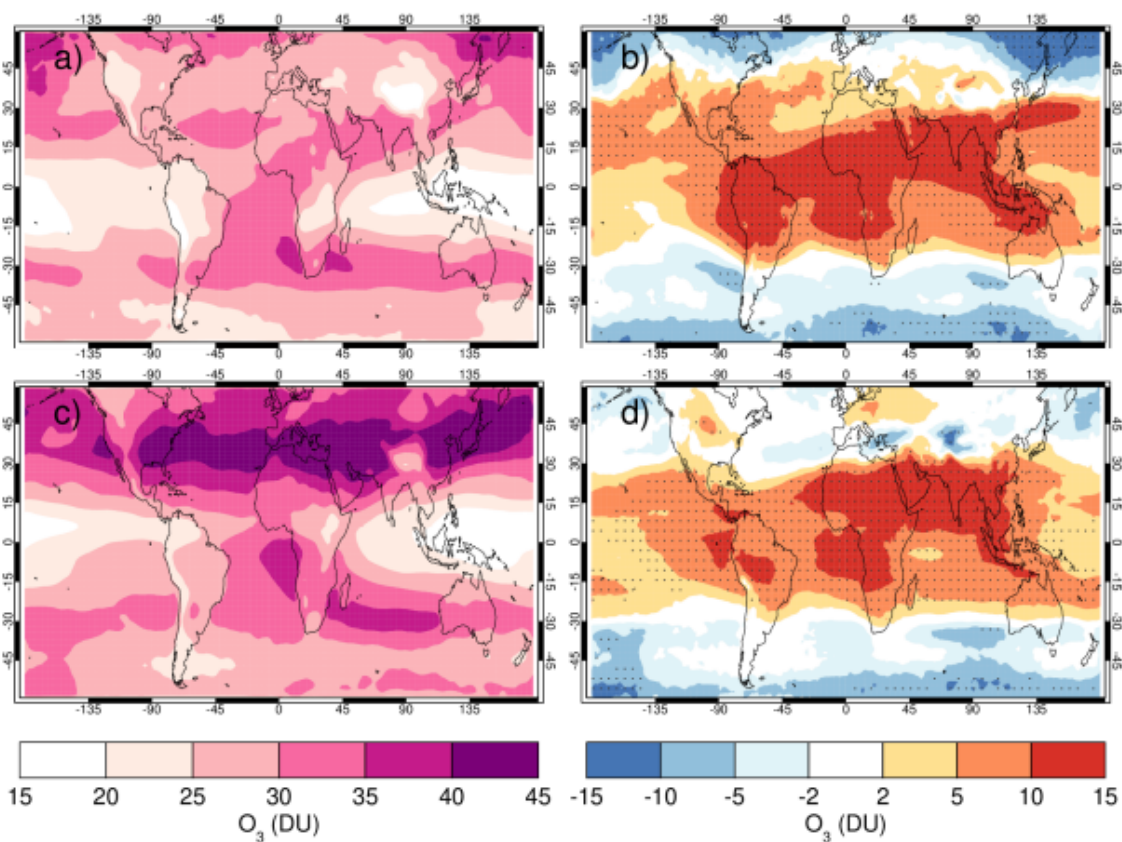
1 tropospheric column. The equations used to apply the OMI NO₂ and MOPITT CO AKs to the
 2 model profiles are:

3
 4 $y = A \cdot x$ Eq. 10

5 $y = 10^{(A(\log_{10}(x) - \log_{10}(x_a)) + \log_{10}(x_a))}$ Eq. 11

6
 7 where x is the co-located model profile interpolated onto the satellite pressure grid, A is the
 8 satellite averaging kernel, x_a is the satellite a priori and y is the modified model profile. Here x
 9 for NO₂ is in sub-columns with units of (10¹⁵ molecules cm⁻²), while x for CO has units of vmr
 10 before conversion into sub-columns/tropospheric column. Tropopause height information was
 11 provided by the OMI NO₂ files, but for MOPITT derived tropospheric column CO we use the
 12 climatological tropopause, described by Monks et al. (2017).

13
 14

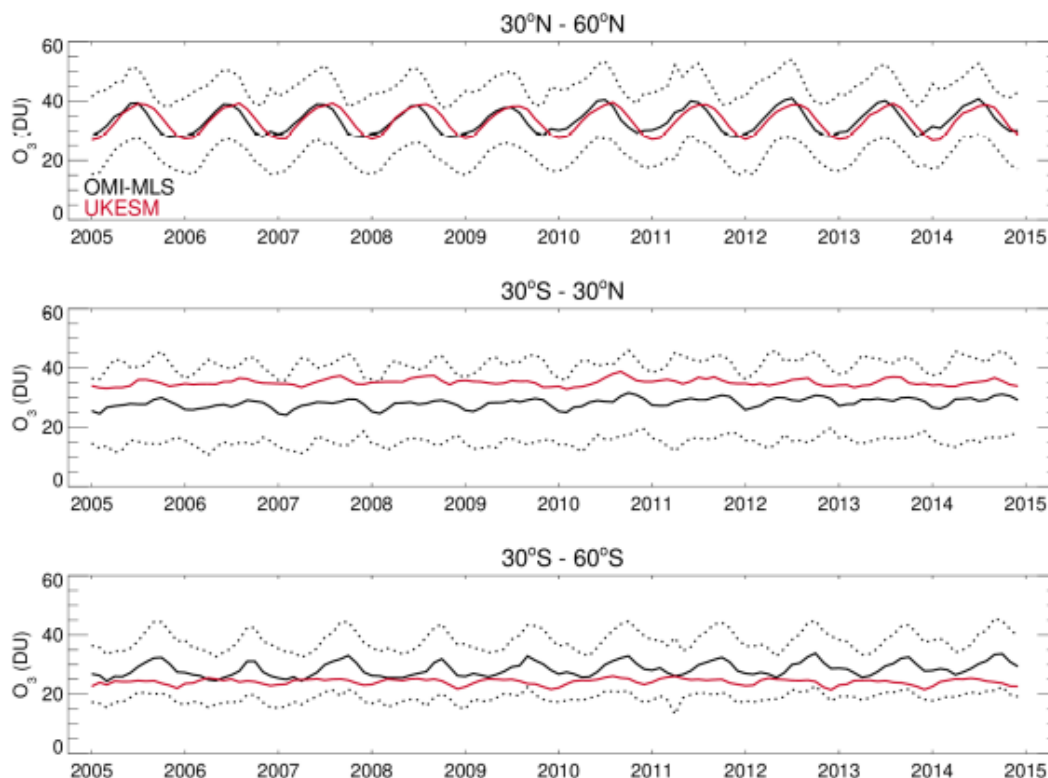


15
 16 **Figure 14.** Comparison of observed and ND modelled tropospheric ozone columns (DU) from the ND
 17 simulations. Plots show seasonal means and differences for the period 2005-2014. a) OMI-MLS
 18 tropospheric column (DJF); b) difference between the model nudged simulations and OMI-MLS
 19 tropospheric column (DJF); c) OMI-MLS (JJA) tropospheric column; d) difference between model and
 20 OMI-MLS tropospheric column (JJA). Stippling indicates gridpoints where $|bias| > 2 \times stdev$ of obs.

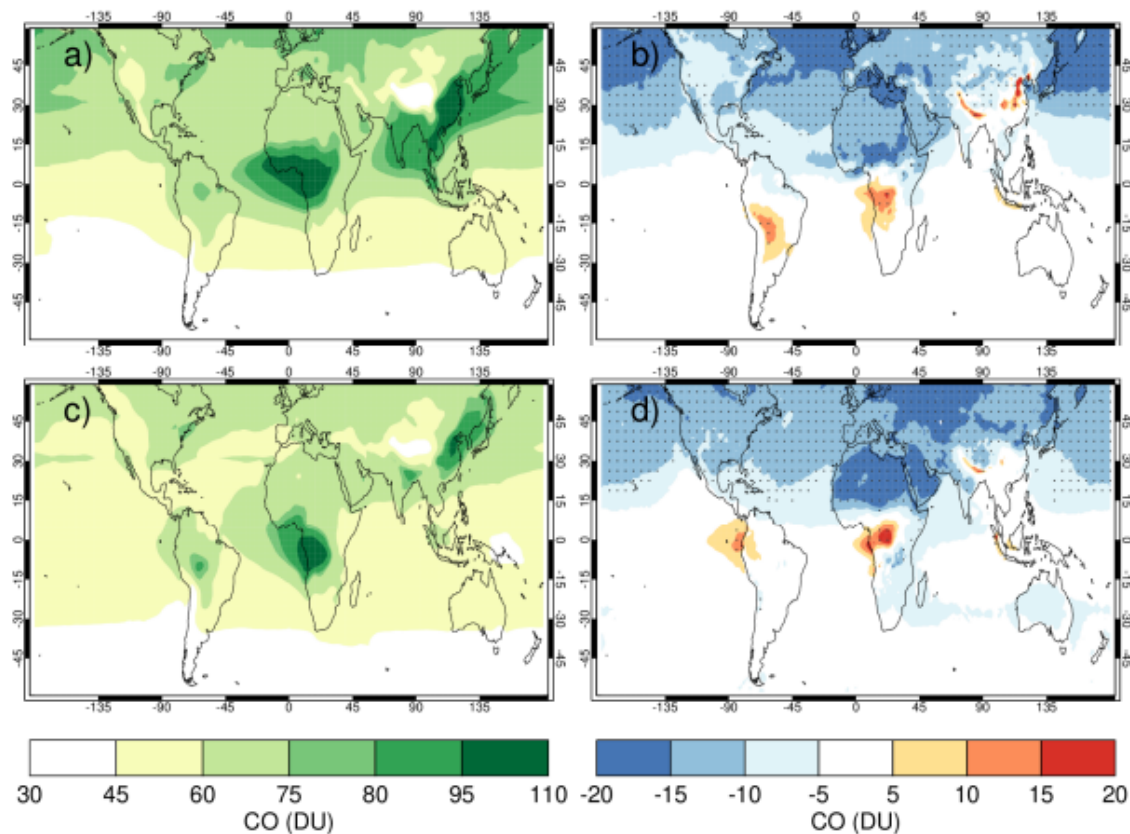
21

22 The modelled tropospheric ozone column (TC_O3) is evaluated against the OMI-MLS
 23 tropospheric ozone column (Ziemke et al., 2006). The general agreement between UKCA
 24 StratTrop and OMI-MLS is good and in line with many other CCMs (Young et al., 2013). A
 25 general feature of the model is a small underestimation in the tropospheric ozone column in
 26 the southern hemisphere extratropics, generally good agreement in the northern hemisphere

1 extratropics but significant positive biases of 15 DU in the tropics (Figure 14). The
 2 underestimation in tropospheric ozone in the Southern mid-latitudes is worse in the late
 3 summer and early autumn when OMI-MLS shows a seasonal maximum in the Southern
 4 Hemisphere that the model fails to reproduce (Figure 15 bottom panel).
 5
 6 For the Northern mid latitudes, Figure 14 panel b, shows that in DJF the model overestimates
 7 tropospheric ozone over large parts of the North Atlantic Ocean while underestimating it over
 8 Northern Russia and large parts of the North Pacific Ocean. These two biases counteract each
 9 other in the timeseries plot (Figure 15, top panel) to give good net agreement overall. It is
 10 worth noting that the timeseries plots show that there are very small, if any, trends in
 11 tropospheric column ozone when averaging across these large domains. Fig 14 panel d and
 12 Figure 15 top panel show that in JJA the model biases in the Northern mid latitudes are
 13 generally very small and the amplitude and phase of the modelled seasonal cycle is in good
 14 agreement with the OMI-MLS data. In the Tropics the differences shown in Fig 14 panel b and
 15 d are around 25-50%. There are potentially several causes for this including (a) the
 16 representation of chemistry in this region, (b) the underlying emission inventories (c) the
 17 deposition rates (which are on the low end of compared with other models) and (d) the
 18 emissions of ozone precursors. The pattern of the bias strongly resembles patterns in the
 19 emissions of NO_x from lightning. It has been noted before that the modelled tropospheric
 20 ozone is extremely sensitive to the average global NO_x emitted by lightning, which is mainly
 21 centred around the tropics. The model bias in the tropics might be a result of the simplified
 22 parameterisation of lightning NO_x emissions and further work will focus on reducing this bias.



23
 24 **Figure 15.** Tropospheric column O₃ (DU) zonal time-series (30-60°N - top panel, 30°N-30°S - middle
 25 panel, 30-60°S - bottom panel) for the ND simulation (red) and OMI-MLS (black). Dashed lines
 26 represent the satellite uncertainty range ($\pm 2 \times \text{stdev}$).



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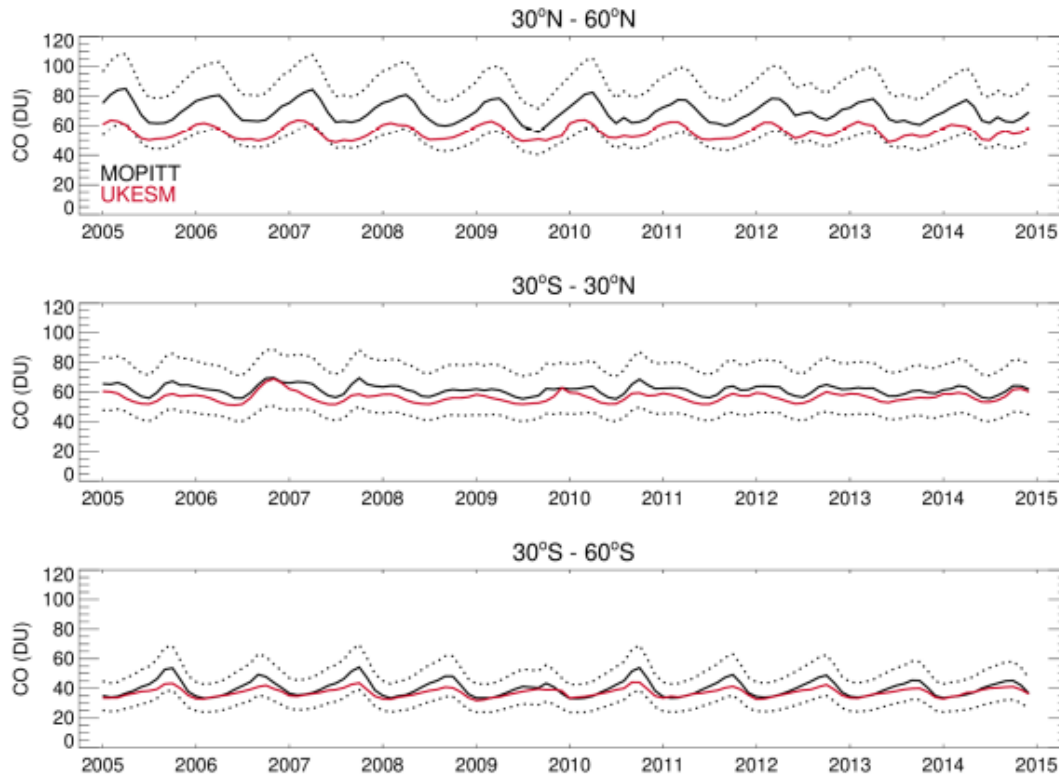
26

Figure 16. Comparison of observed and modelled tropospheric CO columns (DU) from the ND simulations. Plots show seasonal means and differences for the period 2005-2014. a) MOPITT tropospheric column (DJF); b) difference between model and MOPITT tropospheric column (DJF); c) MOPITT (JJA) tropospheric column; d) difference between model and MOPITT tropospheric column (JJA). Stippling indicates gridpoints where $|\text{bias}| > \text{satellite error}$.

Figure 16 shows a comparison of tropospheric column of CO in the UKCA StratTrop nudged dynamics runs with retrievals from the MOPITT instrument on board Terra (Emmons et al., 2004). The MOPITT data reveal that the tropospheric column CO (TC_CO) is highest over anthropogenic and biomass burning emission regions, and lowest over the remote oceans. There is a strong north south gradient which is set up from the short lifetime of CO (~ 30 days) and the time scales for interhemispheric mixing. (NB Figure 16 a highlights strong emissions of CO in DJF in the Northern midlatitudes). The general feature evident from Figure 16 is that the model significantly underestimates TC_CO in the northern hemisphere (NH), in both winter and summer. The negative bias in TC_CO is especially large at high northern latitudes, consistent with surface CO biases in this region (e.g. Shindell et al., 2006). Whilst the NH shows a negative bias, there is a strong positive bias in CO in regions associated with agricultural (Indo-Gangetic plains) and forest burning (central Africa and northern South America).

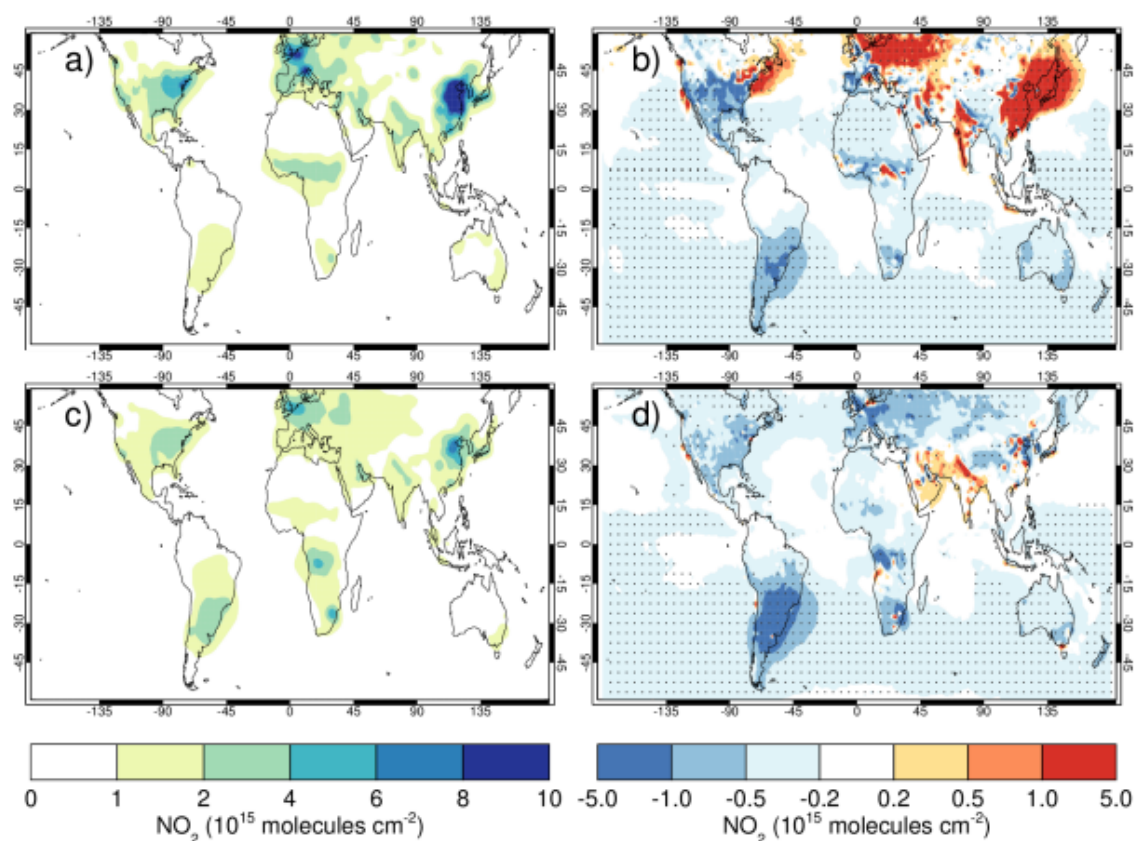
There are a number of reasons for the model-satellite biases in TC_CO, including 1) CO emissions in the NH are underestimated (Miyazaki et al., 2015), 2) insufficient secondary production of CO from non-methane VOC oxidation (e.g. Grant et al., 2010), 3) excess biomass burning emissions in the southern hemisphere (SH) during DJF (potentially the same

1 cause in central Africa in JJA), 4) strong loss through OH in the NH in both seasons. We note
2 that these types of biases are not unique to UKCA StratTrop and that further work is required
3 to ameliorate them (Shindell et al., 2006).
4



5
6 **Figure 17.** Tropospheric column CO (DU) zonal time-series (30-60°N - top panel, 30°N-30°S - middle
7 panel, 30-60°S - bottom panel) for the ND simulation (red) and MOPITT (black). Dashed lines
8 represent the satellite uncertainty range.
9

10 As shown in Figure 17, there is no clear trend in modelled and observed TC_CO over time.
11 However, both data sets show seasonal cycles in TC_CO in the NH and SH with a very muted
12 seasonal cycle in the tropics. The model simulations again underestimate (~10-20 DU)
13 TC_CO in the NH mid-latitudes but successfully capture the amplitude and phase of the
14 seasonal cycle (albeit with a slightly smaller amplitude) and the magnitude of interannual
15 variability well. In the Southern Hemisphere, the model is doing very well capturing absolute
16 concentration, seasonal cycle and interannual variability, although it underestimates the peaks
17 during the Austral winter. There is also an underestimation of CO in the tropics despite the
18 positive bias over biomass burning areas.
19
20

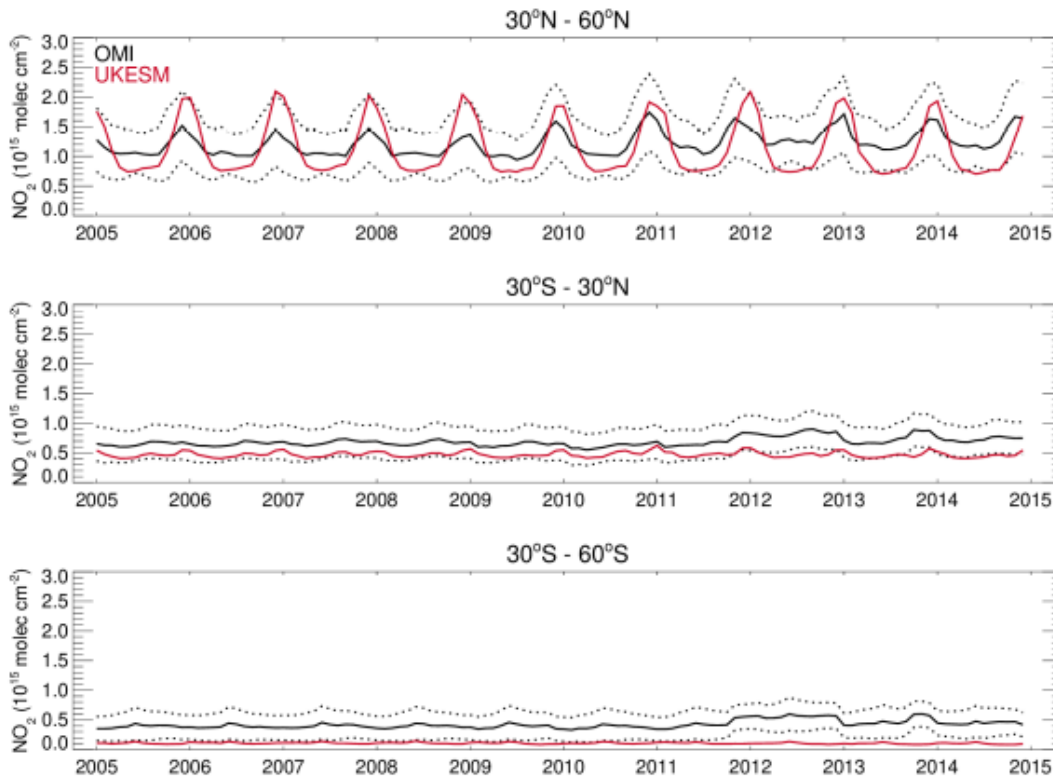


1
 2 **Figure 18.** Comparison of observed and ND modelled UKCA StratTrop in UKESM1 tropospheric NO₂
 3 columns (molecules cm⁻²). Plots show seasonal means and differences for the period 2005-2014. a)
 4 OMI tropospheric column (DJF); b) difference between model and OMI tropospheric column (DJF); c)
 5 OMI (JJA) tropospheric column; d) difference between model and OMI tropospheric column (JJA).
 6 Stippling indicates gridpoints where |bias| > satellite error.

7
 8 Finally we focus on the comparison of modelled and observed tropospheric NO₂ columns. The
 9 observed tropospheric NO₂ column (TC_NO2) data come from the OMI instrument on board
 10 AURA (Boersma et al., 2007). The observed NO₂ column is highly heterogeneous and
 11 localised to the major industrialised regions, where anthropogenic emissions are highest, and
 12 major biomass burning zones (Figure 18). The figure highlights strong seasonal differences in
 13 the observations, with TC_NO2 being larger in winter (panel a) than in summer (panel c), most
 14 likely as a result of higher emissions and longer NO₂ lifetime than in the former season.
 15 Averaged across the whole troposphere, the model compares well with OMI TC_NO2 spatially
 16 (Figure 18 b, d). However, there are very significant positive biases over the main
 17 anthropogenic emission regions (i.e. South Asia, Eastern Europe, East Asia and outflow from
 18 the US eastern seaboard), particularly in the boreal winter. These biases in TC_NO2 are only
 19 weakly correlated with the biases in TC_O3 in these regions, suggesting different causes and
 20 they are dominant in different regions of the atmosphere (boundary layer vs free troposphere).
 21 A high bias in TC_NO2 extends out from the North China plains region, across the Sea of
 22 Japan and into the Pacific Ocean suggesting either errors in the underlying emission inventory
 23 or in the modelled NO₂ lifetime.

24
 25 Over biomass burning regions, there is evidence for low biases over central Africa and South
 26 America (mainly in JJA). This may well be a vertical sensitivity issue in the comparison of the

1 data sets. As OMI has peak sensitivity in the mid-upper troposphere, OMI detects enhanced
 2 NO₂ values over biomass burning regions due to the buoyant fire plumes. In UKESM1, the
 3 gas-phase anthropogenic and biomass burning emissions are added to the surface level, so
 4 most of the NO_x will be trapped in the boundary layer where OMI is less sensitive. Therefore,
 5 the satellite AKs will give this sub-column less weighting and a negative bias occurs.
 6



7
 8 **Figure 19.** Tropospheric column NO₂ (10^{15} molecules cm^{-2}) zonal time-series (30-60°N - top panel,
 9 30°N-30°S - middle panel, 30-60°S - bottom panel) for the ND simulation (red) and MOPITT (black).
 10 Dashed lines represent the satellite uncertainty range.

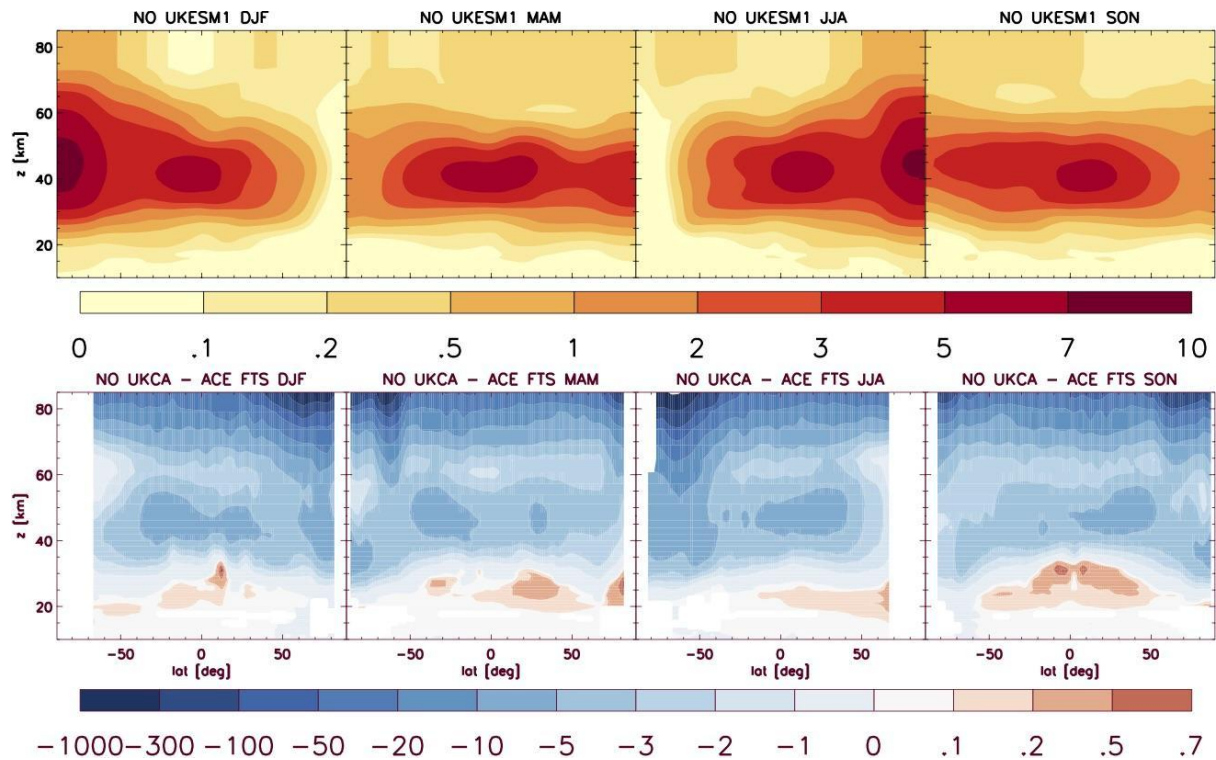
11
 12 Figure 19 highlights that in both the model simulation and satellite data, the average Southern
 13 Hemisphere extra tropical TC_NO2 is lower than in the Northern Hemisphere, due to fewer
 14 emission sources. However, in the model there is a significant low bias in this region, ~ 50%.
 15 This bias is largest over the oceans and may be connected with biases in the representation
 16 of NO_y species (i.e. PAN) which are large contributors to NO_x in this region.
 17 In the Northern extratropics, the model simulated TC_NO2 is within the observational
 18 uncertainty but with too large a seasonal cycle, the simulated mean annual minima/maxima
 19 being much lower/higher than the observed mean annual minima/maxima.

20
 21 **4.6 Evaluation of zonal mean stratospheric composition**

22 Sellar et al. (2019a) provide an overview of the simulation of total column ozone. Their results
 23 and ours (see Figure 13) indicate that UKESM1 produces relatively realistic ozone fields albeit
 24 with some remaining issues. Among these is a tendency for the Antarctic ozone hole to be too
 25 persistent, insufficiently variable, and on average too deep. This is linked to a stratospheric
 26 cold bias noted before (Dennison et al., 2019).

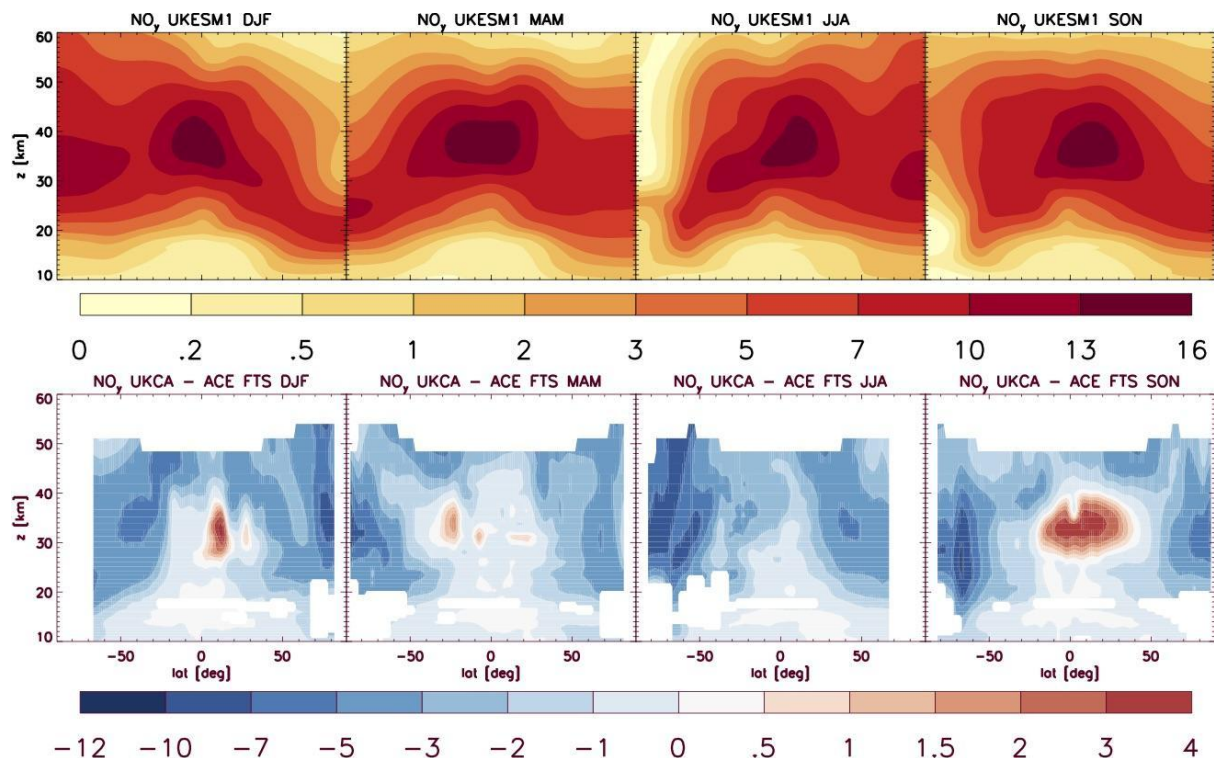
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In the analyses below, UKCA StratTrop seasonal- and zonal-mean composition fields from the FR simulation are compared to selected species from the Atmospheric Chemistry Experiment - Fourier Transform Spectrometer (ACE-FTS) climatology version 3.5. ACE-FTS is an ongoing satellite mission sponsored by the Canadian Space Agency; it uses solar occultation to observe a substantial number of species with a coverage extending in some cases into the mesosphere. The climatologies used here cover the period of February 2004 to February 2013 (http://www.ace.uwaterloo.ca/climatology_3.5.php.) Here we focus on NO, NO_y (defined here as NO + NO₂ + HNO₃), CO, H₂O, and O₃. Climatologies of N₂O₅ and ClONO₂ measurements by ACE-FTS are also available but are not included in the NO_y calculation presented here because of their more restricted coverages than the NO, NO₂, and HNO₃ climatologies. Both would contribute relatively minor amounts to NO_y compared to the large biases discussed below.



16 -1000 -300 -100 -50 -20 -10 -5 -3 -2 -1 0 .1 .2 .5 .7
17 **Figure 20.** Zonal-, seasonal-, and multiannual mean nitric oxide (NO) volume mixing ratio, in ppb.
18 Top: UKCA StratTrop, February 2004 to February 2013. Bottom: Bias versus the ACE-FTS v3.5
19 climatology, same units and period. The climatology represents the average of AM (sunrise) and PM
20 (sunset) measurements, while the model values are averaged over all local times. The model data
21 analysed are from the FR simulation.

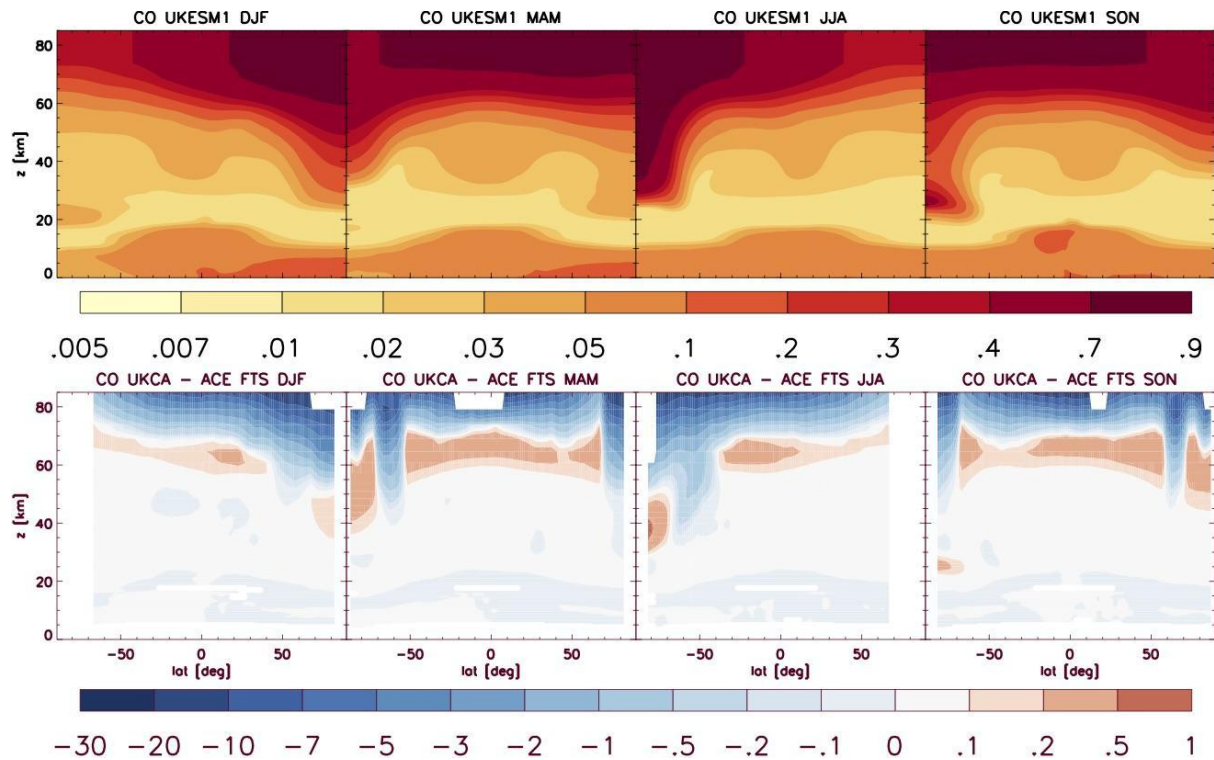
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1 -12 -10 -7 -5 -3 -2 -1 0 .5 1 1.5 2 3 4
 2 **Figure 21.** Same as Figure 22 but for odd nitrogen (NO_y), defined here as $\text{NO}_y = \text{NO} + \text{NO}_2 + \text{HNO}_3$,
 3 in ppb. For ACE-FTS, NO and NO_2 fields are the average of AM (sunrise) and PM (sunset)
 4 measurements. The model values are averaged over all local times.

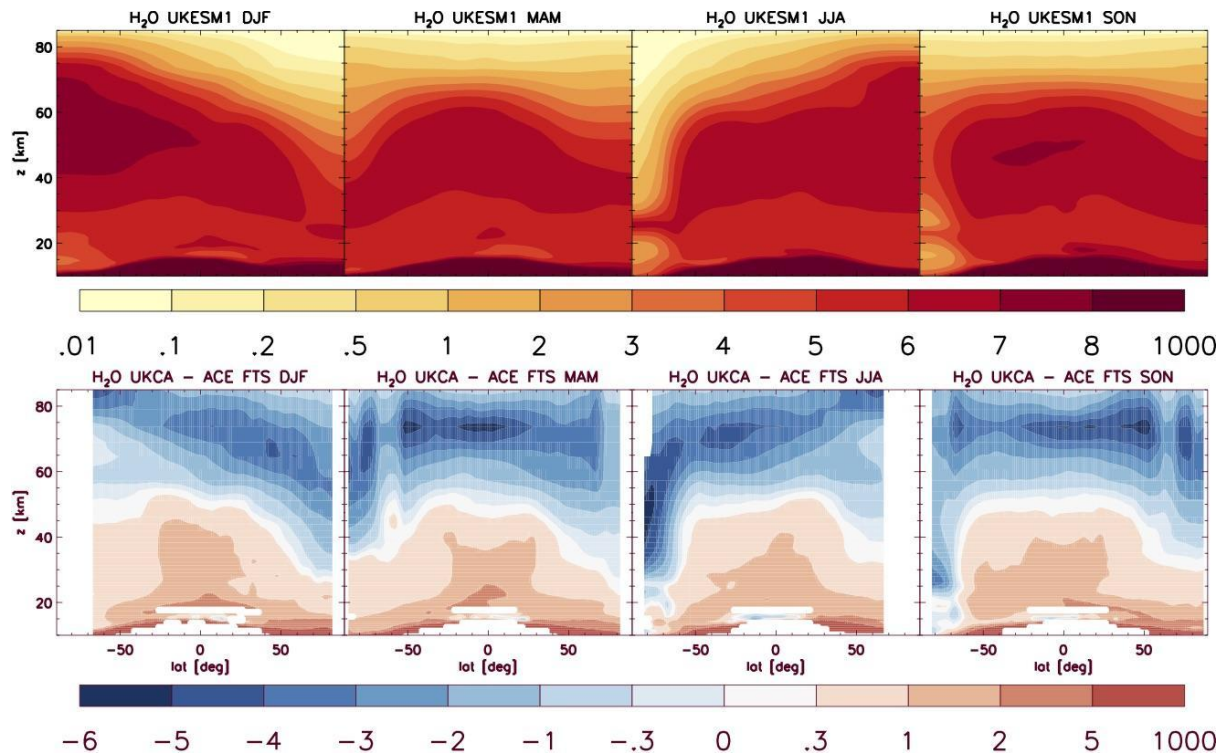
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NO is underestimated throughout the model domain (Figure 20). In the troposphere and much of the stratosphere, NO is subject to a large diurnal cycle. When exposed to sunlight it is maintained by photolysis but converts to NO_2 at night by reacting with O_3 . However, near and above the top of the region covered by NO_2 measurements, at ~ 50 km, this conversion becomes slow and NO is the dominant form of nitrogen in the ACE-FTS measurements also at night. This implies that the large underestimation of NO seen above 50 km, which reaches about 1 ppm, is not a sampling problem associated with imperfect spatiotemporal matching of satellite and model data. Rather it reflects a model shortcoming. To illustrate the consequences of this issue for stratospheric composition, we compare NO_y (Figure 21). This diagnostic reveals tongues of nitrogen-depleted air descending in the polar vortices of both hemispheres in the model which in the ACE-FTS measurements are however relatively nitrogen-rich. This discrepancy lasts into southern spring when NO_y is underestimated by up to 12 ppb at around 70°S . The depletion of HNO_3 due to denitrification in the lower Antarctic polar vortex appears to be well reproduced in winter but is perhaps overestimated in spring, in line with the generally excessively long lifetime of the polar vortex in the model (Sellar et al., 2019a; not shown).



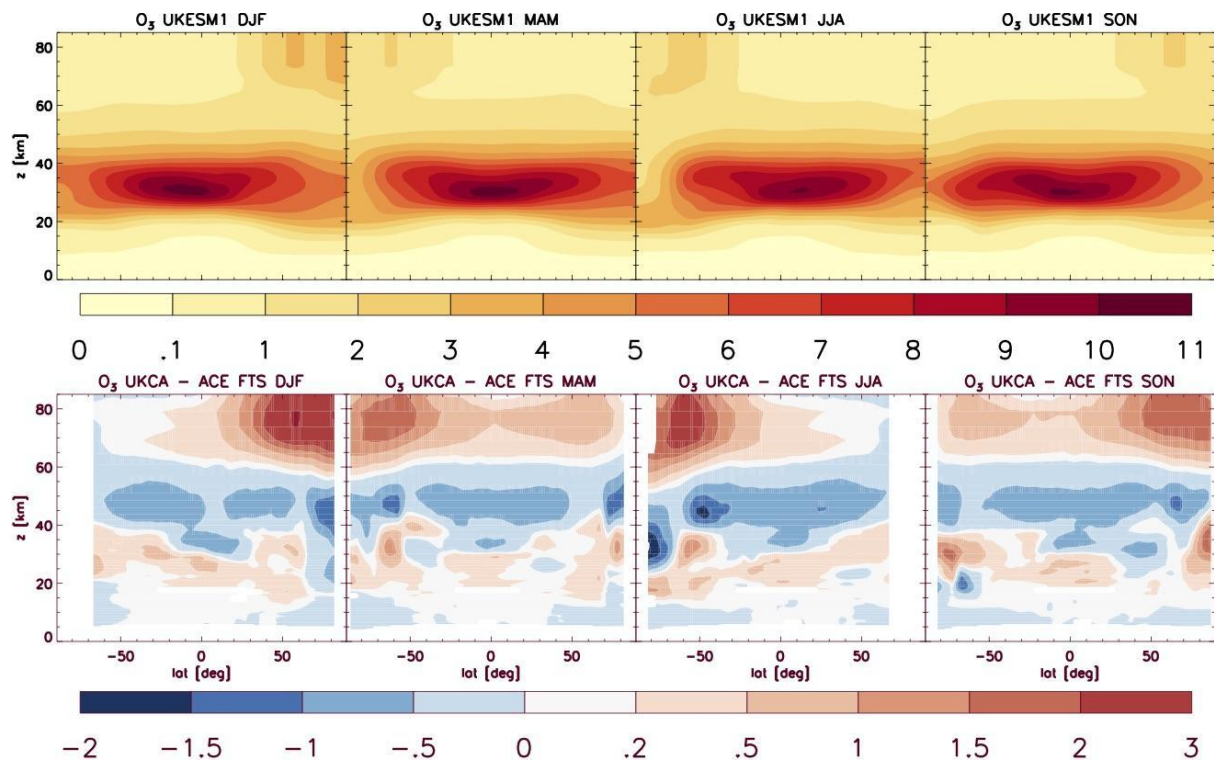
1 -30 -20 -10 -7 -5 -3 -2 -1 -0.5 -0.2 -0.1 0 .1 .2 .5 1
 2 **Figure 22.** Same as Figure 20 but for carbon monoxide (CO), in ppm.
 3

4 The model gets the shape of the distribution of CO about right, but substantially
 5 underestimates the amount of CO in the mesosphere (Figure 22). A variant simulation with a
 6 modified top-boundary condition (TBC), whereby the top two levels are not overwritten with
 7 the third-highest level, reveals that with this variant TBC CO would now be overestimated.
 8 Essentially, CO production is due to CO₂ photolysis which is extremely height sensitive. The
 9 simulation shows that mesospheric air reaches the lower polar vortex in Antarctic spring; this
 10 process is relatively well simulated in the model.
 11



1
2 **Figure 23.** Same as Figure 20 but for water vapour (H₂O), in ppm.

3
4 In much of the stratosphere, H₂O is overestimated by 0.3 to 2 ppm, suggesting that perhaps
5 the tropical-tropopause cold point is still slightly too warm (Figure 23). This has been a
6 persistent problem in the MetUM coupled to UKCA (Morgenstern et al., 2009) and a significant
7 amount of work identified remedies to this issue in earlier versions of UKCA StratTrop
8 (Hardiman et al., 2015). One cause highlighted by Hardiman et al. (2015) was the role of ozone
9 in the upper troposphere / lower stratosphere (UTLS) region. Biases in ozone here are
10 important to this issue of stratospheric moistening. In addition a new development in UKCA
11 StratTrop has been the interactive simulation of H₂O from CH₄ oxidation in the stratosphere
12 and so biases in CH₄ or the transport of CH₄ into the stratosphere may also play a role. Further
13 work will focus on understanding the causes of this H₂O bias. In the mesosphere and in the
14 polar vortices, however, H₂O is underestimated by several ppm in many locations. Unlike all
15 other gas phase chemical species, H₂O is not subject to the overwriting of the top two levels.
16 It photolyses at similarly short wavelengths as CO₂ (see above); an overestimation of its
17 photolysis may explain a large amount of the mesospheric bias.
18



1 **Figure 24.** Same as Figure 20 but for ozone (O_3), in ppm.

2
3
4 Figure 24 highlights generally a good simulation of stratospheric ozone in UKCA StratTrop. In
5 the lower stratosphere, ozone is mostly overestimated (by around 0.2 to 1 ppm), whereas in
6 the upper stratosphere it is underestimated by similar amounts. Larger underestimations exist
7 in Antarctic winter. In the mesosphere, ozone is generally overestimated.

8
9 Taken together, these disagreements indicate some progress with the simulation of odd
10 nitrogen compounds albeit with substantial remaining problems. HNO_3 is now in better
11 agreement with observations than documented by Morgenstern et al. (2009). However this
12 appears to be mostly the case because ACE-FTS finds considerably more HNO_3 in the
13 stratosphere than the older Upper Atmosphere Research Satellite (UARS) data used there
14 (Randel et al., 1998). The substantial deficit of NO in the mesosphere is most likely the result
15 of missing model physics: Energetic particle precipitation (EPP) is well documented to cause
16 the break-up of nitrogen molecules and the formation of NO_x (for a review see e.g. Sinnhuber
17 et al., 2012), but this process is not represented in UKCA StratTrop. This model deficiency
18 results in a misrepresentation of odd nitrogen descending in the polar vortices towards the
19 ozone layer. This might explain the NO_y deficit in winter/spring over both poles, although
20 further studies are needed to confirm this. This problem is receiving much more attention here
21 than e.g. in the earlier investigation by Morgenstern et al. (2009) because the newer ACE-FTS
22 satellite data offer much better coverage of high latitudes and altitudes than the observational
23 references used by Morgenstern et al. (2009).

24
25 Morgenstern et al. (2009) had to artificially reduce water vapour at the tropical tropopause; the
26 reasonable agreement found here is achieved without such an intervention. H_2O loss and CO
27 production are both the result of photolysis of molecules (CO_2 , H_2O) in the mesosphere where
28 the photolysis rate increases sharply with height and may be sensitive to assumptions about
29 the residual ozone column above the model top. In combination, these findings suggest that

1 this residual ozone column (which is a parameter in the photolysis scheme) may be too small,
 2 or that making this a simple universal constant in the model may be inadequate.

3

4 **4.7 Analysis of zonal asymmetry of ozone**

5 Stratospheric ozone is often validated against zonal-mean satellite data (e.g. see above). As
 6 the simulation of ozone improves in models, attention turns to higher-order diagnostics. A
 7 recent analysis by Dennison et al. (2017) revealed that zonal asymmetries of the stratospheric
 8 polar vortex, in simulations by a model closely related to UKESM1, were strongly
 9 underestimated; the vortex was generally too circular and its centre too close to the South
 10 Pole, when in reality the southern polar vortex is often distorted and displaced towards the
 11 Indian-Ocean sector. Dennison et al. found a westward progression of this displacement,
 12 which their model failed to reproduce. The climate impacts of ozone depletion are also often
 13 thought of in zonal-mean terms (e.g. Kang et al., 2011); any effort to attribute regional climate
 14 change beyond the zonal-mean to ozone depletion might well be impeded by such model
 15 behaviour. Hence here we briefly assess how UKCA StratTrop handles zonal asymmetries of
 16 the Antarctic polar vortex. Here we focus on the Historic UKESM1 simulations (Sellar et al.,
 17 2019a), which use the same version of UKCA StratTrop documented here, rather than the
 18 experiments discussed in Section 3.

19

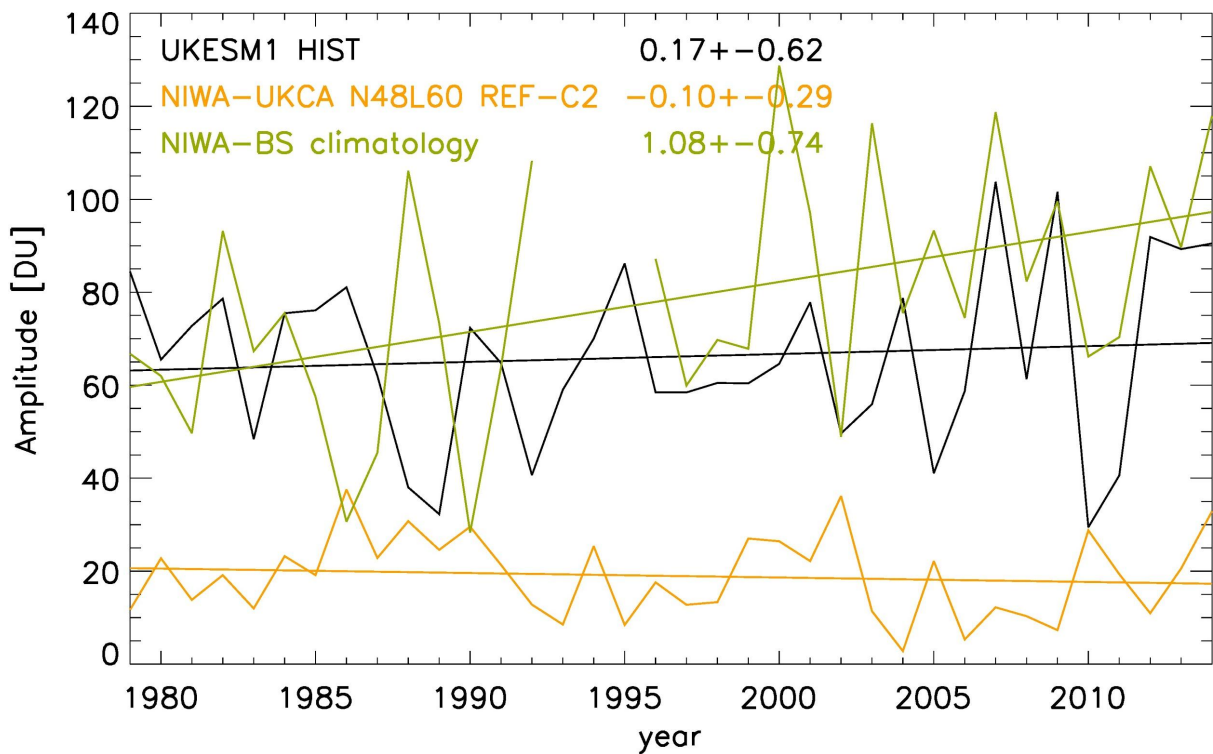
20 The analysis consists of expanding TCO in a Fourier series:

21 $O_3 = ZMO3 + A \cos(\lambda + b) + \text{higher order terms (ignored here)}$ Eq. 12

22

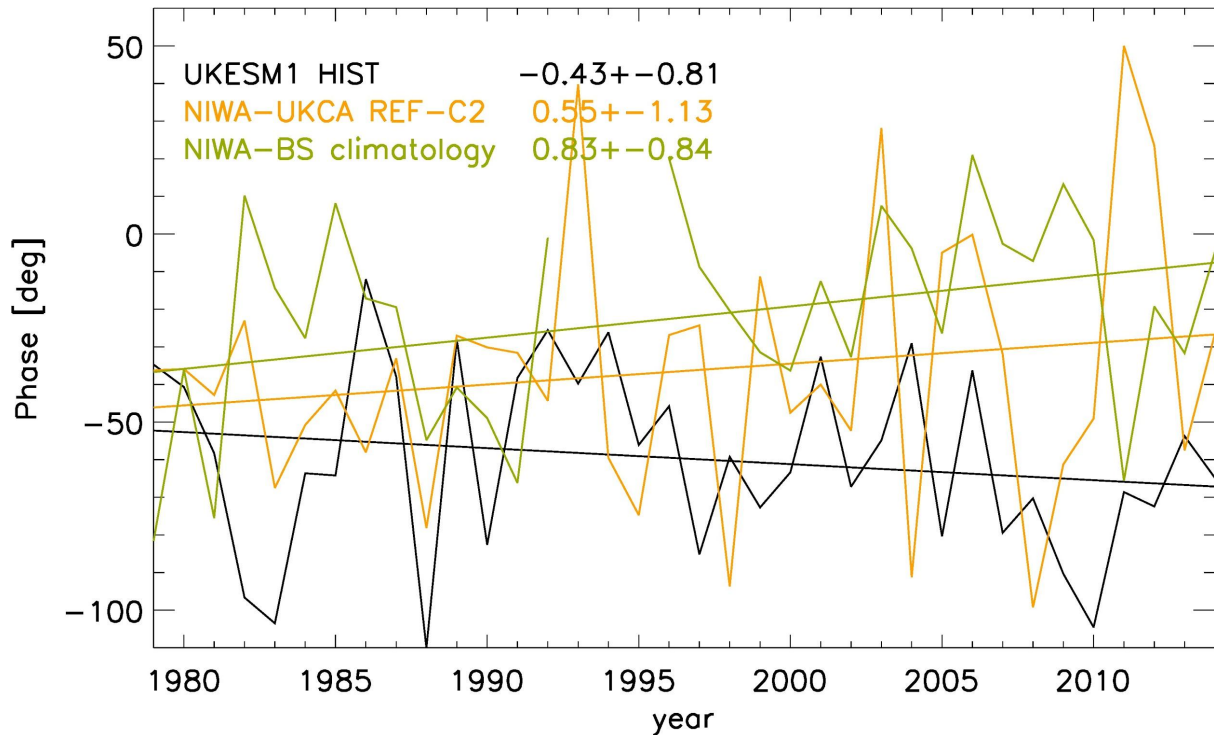
23 Here O_3 is monthly-mean total-column ozone, meridionally averaged over 60S to 70S, ZMO3
 24 is its zonal mean, $A \geq 0$ is the amplitude of the zonal asymmetry, λ is longitude, and b is the
 25 phase shift. $b=0$ would correspond to an ozone maximum occurring at the Greenwich Meridian
 26 and a minimum occurring at the Date Line. Positive values for b correspond to a westward
 27 displacement of these features.

28



29

1 **Figure 25.** Amplitude A [DU] of the zonal asymmetry in total-column ozone at 60-70S in October.
 2 Green: NIWA-Bodeker Scientific total-column ozone climatology, vn. 3.4. Orange: The model used by
 3 Dennison et al. (2017), NIWA-UKCA. The data represent the average of 5 CCMI REF-C2 simulations
 4 by their model. Black: UKESM1. The data represent the average of two CMIP6 “historical” simulations
 5 (Sellar et al., 2019a). Straight lines are linear regression fits. The numbers represent mean trends and
 6 associated 95% confidence intervals in DU/year.
 7
 8



9
 10 **Figure 26.** Same as Figure 25 but for the phase b , in degrees.
 11

12 Figure 25 displays A for the months of October (when the ozone hole typically is deepest).
 13 The NIWA-Bodeker Scientific total-column ozone climatology
 14 (<http://www.bodekerscientific.com/data/total-column-ozone>, green colour) indicates that the
 15 zonal asymmetry is typically about 40 to 120 DU in size, and on average there is a positive
 16 trend, with the ozone asymmetry increasing significantly by nearly 40 DU between 1979 and
 17 2014. UKESM1 (black) reproduces the magnitude and variability of the ozone asymmetry, a
 18 big advance over the model used by Dennison et al. (2017) (orange). The difference in the
 19 trend is not statistically significant at the 95% confidence level. For the phase b (Figure 26) we
 20 find that the model produces an ozone peak usually around 60-70E (i.e. in the Indian Ocean
 21 sector) whereas in the NIWA-Bodeker Scientific climatology this maximum occurs further west,
 22 on average around 20-30E. The mean eastward trend simulated by UKESM1 is outside the
 23 range of possibilities for the observations (which indicate a westward trend), but the
 24 uncertainty intervals overlap.
 25
 26

27 **4.8 Evaluation of transport and long-lived tracer-tracer correlation**

28 Our final aspect of model evaluation focuses on the comparison of the large-scale transport
 29 in the modelled middle atmosphere, analysed through comparison of the modelled age of air
 30 profiles against age of air determined using observations of SF₆ made by the MIPAS

1 instrument (Stiller et al., 2008) and through comparison of observed (ACE-FTS) and modelled
2 tracer-tracer correlations. The model data analysed here are from the FR simulation.

3
4 A simple but powerful way to test the representation of stratospheric chemistry in a model is
5 to analyse the correlations between long-lived trace gases (e.g. Chapter 6, SPARC 2006).
6 Long-lived tracers are known to exhibit compact correlations with each other (Plumb and Ko,
7 1992) and comparison of modelled and observed correlations can test aspects of the model
8 chemistry independent of dynamics. This is particularly useful when comparing complex 3-D
9 climate models such as UKESM1 with observations made by a range of platforms at different
10 spatial resolution and coverage, and under different meteorological conditions.

11
12 Figure 27 shows the correlations of CH₄ vs N₂O, CH₄ vs H₂O and NO_y vs N₂O from a present
13 day UKESM1 simulation (2005-2010) as well as from ACE and MIPAS satellite data. The ACE
14 V4 (2004-2018) data was obtained from <http://www.ace.uwaterloo.ca/data.php> and monthly
15 mean zonal mean values at 5° latitude bins were created by averaging all profiles with retrieval
16 errors less than 100%. The Michelson Interferometer for Passive Atmospheric Sounding
17 (MIPAS) V1.4 data used here is an update of that used in CCMVal-2010 report (SPARC, 2010)
18 (see <http://eodg.atm.ox.ac.uk/MIPAS/>). Co-located profiles of H₂O, CH₄, N₂O, NO₂, and HNO₃
19 are retrieved simultaneously for both day and night time profiles and are available for the
20 mission period (2002-2012). MIPAS data was obtained via [ftp://ftp.ceda.ac.uk/neodc/mipas-](ftp://ftp.ceda.ac.uk/neodc/mipas-oxford/data/)
21 [oxford/data/](ftp://ftp.ceda.ac.uk/neodc/mipas-oxford/data/).

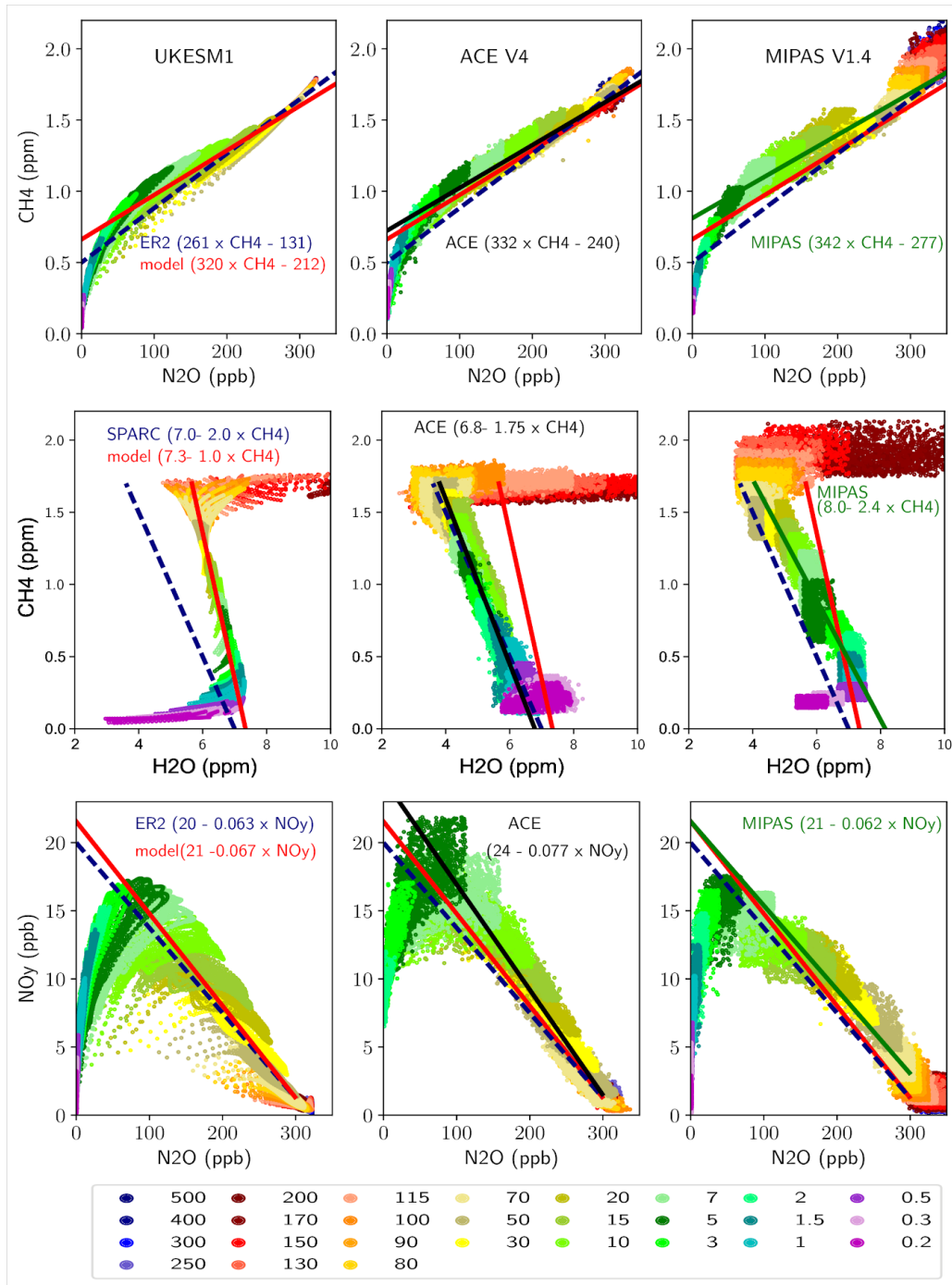
22
23 CH₄ and N₂O are two chemically independent, but long-lived, tracers with significant
24 stratospheric sinks. Accordingly, they are expected to show compact correlations in the
25 stratosphere (Plumb and Ko, 1992). Overall, UKESM1 seems to show very good agreement
26 with the recent satellite-observed relationships, suggesting that the relative loss of CH₄ and
27 N₂O in the stratosphere is well represented. However, the model and the satellite observations
28 differ slightly from the older ER-2 in-situ lower stratospheric observations, possibly due to
29 different relative changes in CH₄ and N₂O in recent years. Note also that the model simulation
30 covers the period 2000-2004 while ACE data covers 2004-2018, hence even after applying
31 the quality flag ACE CH₄ and N₂O values in the troposphere are larger than model values.

32
33 More noticeable model-observation differences are found the CH₄:H₂O correlation. These two
34 long-lived tracers are chemically linked in the stratosphere: CH₄ oxidation leads to the
35 production of nearly 2 molecules of H₂O (with a small yield of H₂). As the maximum observed
36 upper stratosphere H₂O mixing ratio is typically around 7 ppm, and CH₄ is the primary source
37 of stratospheric H₂O, the H₂O vs CH₄ relationship is expected to be close to H₂O + 2×CH₄=7
38 ppm, which is included in the plots as a reference. The ACE observations show a slightly
39 weaker relationship (H₂O + 1.75×CH₄ = 6.8) while MIPAS data shows a stronger slope, which
40 is larger than 2 (H₂O + 2.4×CH₄ = 8.0). There will be some uncertainty in the satellite data but
41 it is clear that UKESM1 has a significantly different relationship. The upper stratospheric H₂O
42 values are reasonable but the lower stratosphere seems to be much wetter compared to
43 observations (see Section 4.6). For example, near 90 hPa most of the ACE profiles show H₂O
44 values close to 3 ppm, whereas modelled values hardly go below 5 ppm, suggesting water
45 vapour entry mixing ratios near the tropical tropopause layer are not well constrained in the
46 model. However, in UKESM1 CH₄ oxidation appears to yield only 1 H₂O per CH₄ oxidised,
47 which allows the model to achieve realistic upper stratospheric H₂O values. Further detailed
48 studies are required to verify the cause of this model discrepancy. We have noted that there

1 is a missing H₂O product in the reaction HO₂ + MeOO (listed in Table S1). However, we
2 calculate that this reaction only accounts for 2.3% of the fate of MeOO in the stratosphere
3 (which is dominated by reaction with NO), so it appears unlikely that this is the source of the
4 bias.

5
6 Finally, we compare the the NO_y vs N₂O tracers, which are also chemically linked. N₂O is main
7 source of stratospheric NO_y with a yield of about 6% via reaction of O(¹D) (see equation 6.2b
8 in SPARC, 2010). ACE NO_y values are calculated simply by adding the observations of HNO₃,
9 NO, NO₂, 2N₂O₅ and ClONO₂. For MIPAS, zonal mean (5° latitude bin) monthly mean profiles
10 were calculated by averaging all the measurements with standard errors less than 100%. For
11 NO_y:N₂O plots, only nighttime profiles are selected (SZA >95) and NO_y is calculated as
12 HNO₃+ NO₂+ 2N₂O₅ + ClONO₂. For large values of N₂O, the UKESM1 correlation is less
13 compact than the observations, although the modelled slope indicates a realistic 6.7% yield of
14 NO_y. The model also produces a reasonable peak NO_y mixing ratio of around 17 ppb,
15 although this is slightly smaller than observations, in particular from ACE. The model also
16 tends to simulate larger occurrences of low NO_y values for a given N₂O, which may be an
17 indication of strong polar denitrification.

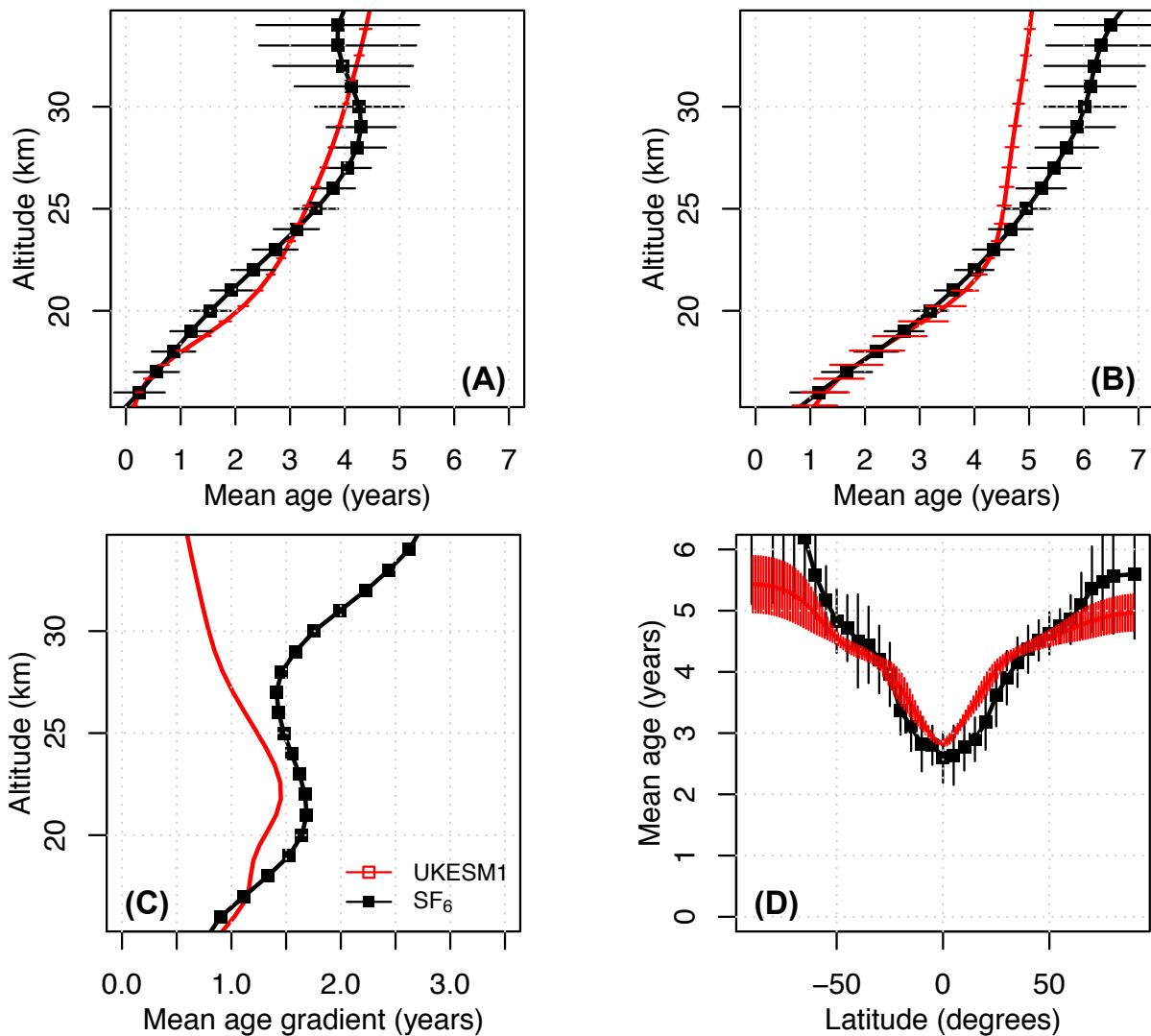
18
19



1
2 **Figure 27.** Correlations between selected long-lived chemical species (monthly mean zonal mean
3 values for 60°S-60°N) from FR UKESM1 (left column), ACE V4 data (centre) and MIPAS data (right).
4 The coloured legend shows the corresponding pressure level (hPa) of the data points. The linear
5 regression fits to the model, ACE and MIPAS data are shown in the respective panels along with the
6 equations of the lines. The MIPAS data is the same as that used in Figures 6.12, 6.13 and 6.14 in the
7 CCMVal-2 report (SPARC 2010). ACE NO_y values are calculated as NO_y = NO + NO₂ + HNO₃ +
8 2N₂O₅ + ClONO₂. (Top row): CH₄ vs N₂O. The linear fit is calculated for N₂O values ranging from 100
9 to 300 ppb. The dashed line shows estimated fit from ER-2 data (N₂O (ppb)) = 261.8×CH₄ (ppm) -131,

1 see Kawa et al., 1993). (Middle row): CH₄ vs H₂O. The linear fit is calculated for CH₄ values ranging
 2 from 0.5 to 1.5 ppm. The dashed line represents H₂O + 2CH₄ = 7 ppm. (Bottom row): NO_y vs N₂O.
 3 The linear fit is calculated for N₂O values ranging from 100 to 300 ppb and the dashed line shows the
 4 equation NO_y (ppb) = 20.0 - 0.0625×N₂O (ppb), based on mid-latitude balloon profiles and ER-2 data
 5 (see Kondo et al 1996).
 6

7 Figure 28 compares data from the FR simulation and observations. The FR run is shown here
 8 as this allows for a more robust comparison of the model data where it is not constrained by
 9 the re-analysis meteorology. Figure 28 shows the modelled multiannual mean age of air profile
 10 in the stratosphere against observations of SF₆ from 2002-2010 used to calculate the age of air
 11 from the MIPAS instrument (Stiller et al., 2008). The model includes a diagnostic to quantify
 12 the age of air. This is effectively a “species” in the model that is emitted at the model surface
 13 continually and undergoes full tracer advection and diffusion. Whilst below the modelled
 14 tropopause (based on a merger of the 380 K and 2 PVU surfaces) the tracer is set to have an
 15 age of zero; above the tropopause the tracer has its age increased every model time step that
 16 it stays above the tropopause.
 17



18

19

20 **Figure 28.** Comparison of multi-annual mean FR modelled (red) age-of-air with values derived from
 21 MIPAS observations of SF₆ (black) (Stiller et al., 2008). Panel A: Tropical profile. Panel B: NH
 22 Midlatitude profile. Panel C: Midlatitude-Tropics profile. Panel D: Mean age at 23 km (~50hPa).

1
2 Figure 28a shows the modelled mean tropical ($\pm 10^\circ$) age profile as a function of altitude and
3 that there is very good agreement between the model and the values derived from MIPAS
4 observations, with an increase in the age of air as both profiles increase in altitude and a
5 maximum age of around 5 years. The modelled northern hemisphere midlatitude ($35^\circ - 45^\circ\text{N}$)
6 age profile (panel b) agrees very well with the observations from 16 km to about 24 km, but
7 the model tends to simulate an age of air which is younger than the observations above 24
8 km (up to a year difference younger). Panel (c) shows the difference between the mid-latitude
9 and tropical profiles and further emphasises good agreement of the model with the
10 observations below 23 km but divergence above this altitude. However, the zonal cross
11 section at 23 km (~ 50 hPa) (panel d) shows that the model generally falls within the
12 observational uncertainty (1 standard deviation of the multiannual observations) at all
13 latitudes.

14 15 16 **5.0 Discussion and Conclusions**

17 In this paper we have documented the species and reactions that make up the UKCA
18 StratTrop mechanism for the first time and performed an evaluation of the model output for
19 the recent past. UKCA is the module for simulating chemical and aerosol processes in the
20 UKESM1 Earth System model (Sellar et al., 2019a) and UKCA StratTrop enables a holistic
21 representation of gas-phase chemistry in the troposphere and stratosphere; important for
22 understanding short-lived climate forcers.

23
24 Our focus here has been to document the performance of the chemical fields simulated by
25 UKCA StratTrop as implemented in UKESM1; the aerosol schemes, processes and
26 performance are discussed in detail in Mulchay et al. (2019). Further studies are planned
27 which will assess the role of composition-climate Earth system couplings in the UKESM1
28 framework. Hence, we present simulations which have enabled a more focused assessment
29 of key performance indicators of the UCKA StratTrop scheme. We have analysed data from
30 two model runs; the first was a *free-running* (FR) simulation where the meteorology was
31 allowed to evolve independently based on the influence of the prescribed forcing agents
32 (SSTs, GHGs and sea ice) and the second was a *Nudged* (ND) simulation where the
33 meteorology was relaxed toward ERA-Interim reanalysis.

34
35 In general, and focusing on the gas phase as we have here, we find that the performance of
36 UKCA StratTrop in UKESM1 is in line with the range of models that are applied to simulating
37 the coupled chemistry-climate system (Young et al., 2013; 2018).

38
39 Our key performance indicators have included:

- 40 • An assessment of the magnitude and spatial distribution of lightning NO_x :

41 We note here that whilst the model simulates a global annual total lightning NO_x emissions
42 magnitude that is in the middle of the range quoted in the literature based on observational
43 constraints (~ 6 Tg/yr), and the spatial distribution in lightning flash frequency matches well
44 with observations from satellites, the variability in lightning flash frequency is not in good
45 agreement with the observations (Figure 2). The UKESM1 model predicts too much lightning
46 activity in the tropics at the expense of the extra tropics, something which could be resolved
47 by moving to an ice-flux based scheme (Finney et al., 2018). Moreover, the vertical profile of
48 lightning NO_x may have a significant impact on modelled O_3 . Hakim et al. (2019) have shown

1 that across India the vertical profile in simulated lightning NO_x is very model dependent. We
2 suggest further work is performed to better understand the impacts of both the spatial
3 distribution of lightning NO_x and the impacts of lightning NO_x on the tropospheric column
4 biases in O_3 in the model.

- 5
- 6 • Surface ozone correlations and mean bias against TOAR observations:

7 TOAR (Shultz et al., 2018) provides the chemistry modelling community with an
8 unprecedented dataset to evaluate surface O_3 . In our analysis of the FR and ND runs
9 presented here, we show that the annual mean bias is very low, but this hides biases in
10 summer and wintertime (Young et al., 2018). However, we suggest that further work be
11 performed to understand the cause of the low and high biases in surface O_3 , especially with
12 regards to how these may impact studies that use UKESM1 surface O_3 in health assessment
13 studies.

- 14
- 15 • The tropospheric oxidising capacity:

16 A key component to determine the lifetime of emitted reactive gases in the troposphere is the
17 oxidising capacity. Whilst this has to be inferred from observations (i.e. through the inferred
18 lifetime of methane) it is an important metric to evaluate the model against. In this study we
19 found that the methane lifetime in the troposphere with respect to OH was 8.5 years, within
20 the ACCMIP multi-model range but slightly low compared to observational analyses (Naik et
21 al., 2013). When compared against other model estimates of the zonal mean distribution of
22 OH, UKESM1 performs well in 10 out of 12 regions analysed; with a significant high bias in
23 the tropical boundary layer. This is a region where the majority of methane oxidation takes
24 place and may explain the slightly low modelled methane lifetime. With the recent
25 development of aircraft OH datasets appropriate for global model evaluation (Prather et al.,
26 2017) we intend to extend this analysis further and interrogate the model with these data to
27 confirm if the bias is indeed large compared with direct observations.

- 28
- 29 • Tropospheric columns of reactive gases (CO , NO_2 and O_3):

30 The analysis of the model ND runs highlighted some success and failure in the model's
31 representation of tropospheric columns of CO , NO_2 and O_3 . The best performance was found
32 for O_3 (Figures 14-15), although we note that there is a significant positive bias in the tropics
33 (which has been shown to have an effect on modelled tropospheric photolysis rates (Hall et
34 al., 2018)). In part we believe this bias is connected with the vertical profile and magnitude of
35 lightning NO_x and further work will focus specifically on this area. The modelled tropospheric
36 column of CO shows significant negative biases in the northern hemisphere (Figure 16). In
37 part this is believed to relate to biases in the representation of higher hydrocarbons that could
38 contribute significantly to secondary CO production (Grant et al., 2010) but high OH could also
39 be a contributing factor. The performance of modelled NO_2 tropospheric columns was found
40 to be generally acceptable in northern midlatitudes (Figure 19) but there are large biases in
41 regions of high emissions (such as the North China plains (Figure 18)). One hypothesis is that
42 the model simulates too little OH in the regions of high NO_2 emissions, owing to lack of reactive
43 VOC emissions and titration of O_3 , which extends the lifetime of NO_2 in these regions. Further
44 studies are required to evaluate the modelled NO_2 lifetime and its response to changes in
45 emissions of NO_x .

- 46
- 47 • Biases in stratospheric composition

1 By examining selected climatologies of observations from satellite (Figures 20-24) we have
2 been able to show here that the simulation of stratospheric composition has improved
3 significantly in StratTrop compared with the older “stratosphere”-focused scheme of MO09. In
4 part this is largely due to improvements in the dynamical model (MetUM) and reductions in
5 biases in modelled water vapour (Hardiman et al., 2017). Key questions remain about the
6 fidelity of the upper stratospheric/mesospheric photolysis rates and the upper boundary
7 conditions. Given the generally poorer performance of NO and NO_y it would be useful to
8 investigate the implementation of parametrised EPP to see if this ameliorates the problems.
9 Further work is also required to understand the cause of the disagreement between the
10 CH₄:H₂O correlation in the stratosphere, which suggests that too little H₂O is produced from
11 methane oxidation in the model.

12

- 13 ● Middle atmosphere age of air:

14 The modelled middle atmosphere circulation has been evaluated against age of air derived
15 from observations of SF₆ and through the use of tracer-tracer correlations. These tracer-tracer
16 correlations further motivate the need for a more detailed investigation of the modelled
17 stratospheric NO_y and its budget (production and loss). The comparison of the age of air in
18 the model generally looks acceptable in the middle stratosphere but tends to deviate at higher
19 altitudes. In part there is more uncertainty in observations at higher altitudes (owing to loss
20 processes of SF₆) but further studies are required to understand if these biases are dependent
21 on the resolution of the model. To understand this a high top, > 120 km, version of the model
22 is in preparation as are simulations of UKESM1 at much higher horizontal resolution (~ 25
23 km).

24

25 In summary, UKCA StratTrop represents a substantial step forward compared to previous
26 versions of UKCA. We have shown here that it is well suited to the challenges of representing
27 interactions in a coupled Earth System Model (key for CMIP6 and beyond) and we have
28 identified key areas and components for future development that will further improve the model
29 in the future.

30

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33 efforts in developing and applying the model. In particular we would like to acknowledge Prof.
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36 datasets used in this paper to help evaluate the model. This work used JASMIN, the UK
37 collaborative data analysis facility. We would like to thank Bodeker Scientific, funded by the
38 New Zealand Deep South National Science Challenge, for providing the combined NIWA-BS
39 total column ozone database.

40

41 **Data and Model Code Availability**

42 Due to intellectual property rights restrictions, we cannot provide either the source code or
43 documentation papers for the UM (including UKCA) or JULES.

44 *Obtaining the UM (including UKCA).* The Met Office Unified Model (MetUM) is available for
45 use under licence. A number of research organisations and national meteorological services
46 use the UM in collaboration with the Met Office to undertake basic atmospheric process
47 research, produce forecasts, develop the UM code, and build and evaluate Earth system
48 models. For further information on how to apply for a licence, see

1 <http://www.metoffice.gov.uk/research/modelling-systems/unified-model> (last access: 14
2 August 2019).

3 *Obtaining JULES.* JULES is available under licence, free of charge. For further information on
4 how to gain permission to use JULES for research purposes see [http://jules-](http://jules-lsm.github.io/access_req/JULES_access.html)
5 [lsm.github.io/access_req/JULES_access.html](http://jules-lsm.github.io/access_req/JULES_access.html) (last access: 14 August 2019).

6 *Details of the simulations performed.* UM and JULES simulations are compiled and run in
7 suites developed using the Rose suite engine
8 (<http://metomi.github.io/rose/doc/html/index.html>, last access: 14 August 2019) and scheduled
9 using the Cylc workflow engine (<https://cylc.github.io/cylc/>, last access: 14 August 2019). Both
10 Rose and Cylc are available under version 3 of the GNU General Public License (GPL). In this
11 framework, the suite contains the information required to extract and build the code as well as
12 configure and run the simulations. Each suite is labelled with a unique identifier and is held in
13 the same revision-controlled repository service in which we hold and develop the model's
14 code. This means that these suites are available to any licensed user of both the UM and
15 JULES.

16 All code related to the offline emissions is freely available on Github: [https://github.com/acsis-](https://github.com/acsis-project/emissions)
17 [project/emissions](https://github.com/acsis-project/emissions) and the data for biogenic emissions are available for free download
18 from <http://eccad.sedoo.fr/>. The model-satellite evaluation codes are available on request.
19 We acknowledge the use of the TEMIS OMI NO₂ (DOMINO vn 2.0;
20 <http://www.temis.nl/airpollution/no2.html>) data and NASA's MOPITT CO (vn7.0;
21 <https://search.earthdata.nasa.gov/>) data. The observations used to evaluate age of air were
22 the IMK/IAA generated MIPAS-ENVISAT datasets developed at KIT and available from:
23 <http://www.imk-asf.kit.edu/english/308.php>.

24
25

26 **Author contributions.** ATA and NLA lead the initial development of StratTrop and ATA and
27 FMO'C lead the writing of the manuscript. NLA, FMO'C, JPM, AJH, GAF, MD, ST, contributed
28 during model development, data analysis and paper preparation. Simulation design, setup,
29 and execution was performed by FMO'C and MD. CH and OW provided the ozone data for
30 the O₃ dry deposition evaluation. All co-authors contributed to writing sections of the
31 manuscript, performing evaluation and reviewing drafts of the manuscript.

32

33 **Competing interests.** The authors declare that they have no conflict of interest.

34

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6
7

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