A single-column ocean-biogeochemistry model (GOTM-TOPAZ) 2 version 1.0

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8 Abstract. Recently, Earth System Models (ESMs) have begun to consider the marine ecosystem to reduce errors in climate 9 simulations. However, many models are unable to fully represent the ocean biology-induced climate feedback, which is due 10 in part to significant bias in the simulated biogeochemical properties. Therefore, we developed the Generic Ocean 11 Turbulence Model-Tracers of Phytoplankton with Allometric Zooplankton (GOTM-TOPAZ), a single-column ocean 12 biogeochemistry model that can be used to improve ocean biogeochemical processes in ESMs. This model was developed by 13 combining the GOTM, a single-column model that can simulate the physical environment of the ocean, and TOPAZ, a 14 biogeochemical module. Here, the original form of TOPAZ has been modified and modularized to allow easy coupling with 15 other physical ocean models. To demonstrate interactions between ocean physics and biogeochemical processes, the model 16 was designed to allow ocean temperature to change due to absorption of visible light by chlorophyll in phytoplankton. We 17 also added a module to reproduce upwelling and the air-sea gas transfer process for oxygen and carbon dioxide, which are of 18 particular importance for marine ecosystems. The simulated variables (e.g., chlorophyll, oxygen, nitrogen, phosphorus, 19 silicon) of GOTM-TOPAZ were evaluated by comparison against observations. The temporal variability of the observed 20 upper-ocean (0–20 m) chlorophyll is well captured by the GOTM-TOPAZ model with a correlation coefficient of 0.51. The 21 surface correlation coefficients between the GOTM-TOPAZ oxygen, nitrogen, phosphorus, and silicon are 0.47, 0.30, 0.16, 22 and 0.19, respectively. We compared the GOTM-TOPAZ simulations with those from MOM-TOPAZ and found that 23 GOTM-TOPAZ showed relatively lower correlations, which is most likely due to the limitations of the single-column model. 24 Results also indicate that source/sink terms may contribute to the biases in the surface layer (< 60 m), while initial values are 25 important for realistic simulations in the deep sea (> 250 m). Despite this limitation, we argue that our GOTM-TOPAZ 26 model is a good starting point for further investigation of key biogeochemical processes and is also useful to couple complex 27 biogeochemical processes with various oceanic global circulation models.

29 1 Introduction

Over several decades, climate researchers have accumulated significant knowledge on atmosphere-land-ocean feedback 30 31 processes through various studies related to climate systems (Friedlingstein et al., 2006; Soden and Held, 2006; Dirmeyer et 32 al., 2012; Randerson et al., 2015). With the advancement of coupled modeling techniques and an exponential increase in the 33 number of computer resources available, climate research institutions worldwide began competing to develop earth system 34 models (ESMs) (Dunne et al., 2012a; Dunne et al., 2012b; Jones and Sellar, 2015; Sokolov et al., 2018). ESMs are often 35 coupled with biogeochemistry models that consider the atmosphere-ocean carbon cycle and ocean ecosystem cycles (Dunne 36 et al. 2012b; Yool et al., 2013; Azhar et al., 2014; Stock et al., 2014; Aumont et al., 2015). Recently, reproductions of ocean 37 ecosystems in ESMs have become very precise with the addition of physiological details, such as light or nutrient 38 acclimation, and the division of various phytoplankton and zooplankton into functional groups (Hense et al., 2017).

39 The following processes are generally considered the most important in ocean biogeochemistry models: the ocean 40 ecosystem cycle, including phytoplankton and zooplankton; the biogeochemical carbon cycle; and the biogeochemical cycle 41 of key nutrients (P, N, Fe, and Si) (Dunne et al., 2012b; Aumont et al., 2015). These three cycles are not independent and 42 include mutual material exchange through chemical mechanisms. There are still no accurate methodologies with which to 43 differentiate biogeochemical variables and to represent biogeochemical processes as formulas (Sauerland et al., 2018). In 44 other words, biogeochemical processes are reproduced in the model via parameterization that adjusts the parameters of a 45 formula based on observations and some general parameters (e.g., maximum phytoplankton growth rate) that are adjusted 46 until the model produces reasonable results (Sauerland et al., 2018).

47 Researchers have been using single-column models (SCMs) to control the parameterizations and increase their 48 understanding of the physical processes in models. Betts and Miller (1986) suggested that SCMs were an effective tool with 49 which to develop and control the convective scheme of an atmospheric model, while Price et al. (1986) used an ocean SCM 50 to study the daily cycle of the mixed layer in the Pacific Ocean. An SCM allows for control of physics parameters, alongside 51 large-scale forcing influences, and, unlike 3D models, it has a low calculation cost. Accordingly, SCMs have been viewed as 52 essential tools with which to develop and improve numerical models (Lebassi-Habtezion and Caldwell, 2015; Hartung et al., 53 2018). SCM-based studies are essential for improving ocean-biogeochemical processes, which are reproduced in climate 54 models based on column physics (Evans and Garcon, 1997; Burchard et al., 2006; Bruggenman and Bolding, 2014). Even 55 the latest analyses of the ESMs included in the Coupled Model Intercomparison Project Phase 5 (CMIP5) show high biases 56 and inter-model diversity in ocean biogeochemical variables (Lim et al., 2017). Therefore, a single-column form of a 57 biogeochemistry model might be a useful tool to meet the ongoing demand for improvements in biogeochemistry models in 58 ESMs.

The oceanic biogeochemical cycle affects not only the physical environment of the upper ocean but also that of the entire climate system, and such changes produce feedback that, in turn, alters the ocean ecosystem (Hense et al., 2017; Lim et al., 2017; Park et al., 2018). Hense et al. (2017) presented the CO₂ cycle, gas and particle cycle, and changes in the physical environment of the upper ocean by chlorophyll as important climate-ocean biogeochemistry feedback loops reproduced in ESMs that are currently available. An ESM that reproduces all three of these biological mechanisms does not exist today; however, all of these mechanisms need to be properly reproduced in the ESMs to reduce the uncertainty in predicting future climate change. This would allow ESMs to change in a fundamentally different way. Furthermore, there are generally time constraints in repeated experiments using ocean general circulation models (OGCMs) and biogeochemistry models due to their complexity and the heavy calculation required. Consequently, SCMs are crucial for applying and testing new climateocean-biogeochemistry feedbacks in existing ESMs.

69 In this study, we developed the Generic Ocean Turbulence Model–Tracers of Phytoplankton with Allometric Zooplankton 70 (GOTM-TOPAZ), which is a single-column ocean-biogeochemistry model. GOTM is a one-dimensional ocean model that 71 focuses on reproducing statistical turbulence closures (see http://www.gotm.net); TOPAZ is an ocean-biogeochemistry 72 model developed by the Geophysical Fluid Dynamics Laboratory (GFDL) and coupled with the ESM2M and ESM2G 73 models (Dunne et al., 2012a; Dunne et al., 2012b). We modularized TOPAZ to apply external physical environmental data while modifying it as an SCM. It was then combined with a GOTM utilizing an air-sea gas exchange for CO₂ and O₂ and 74 75 optical feedback from photosynthesis by chlorophyll. A w-advection prescription module that can reproduce upwelling was also added to this model. To verify GOTM-TOPAZ, we selected points in the East/Japan Sea off the coast of the Korean 76 77 Peninsula upon which to conduct simulations. The results produced by the model were compared to observed data and 78 results from OGCMs to verify its reliability.

79

80 2 The Physical Ocean Model: General Ocean Turbulence Model (GOTM)

In GOTM-TOPAZ, the GOTM version 4.0 is applied to ocean physics. The physical bases of the GOTM are Reynoldsaveraged Navier–Stokes equations in a rotational coordinate system (Eqs. 1 and 2). Moreover, the temperature and salinity equations derived using these methods are given in Eqs. 3 and 4, respectively. GOTM uses one-dimensional potential temperature, salinity, and horizontal velocity based on these four equations, as shown below:

85

86
$$\partial_t u - v \partial_{zz} u + \partial_z \langle u'w' \rangle = -\frac{1}{\rho_0} \partial_x p + fv$$
 (1)

87
$$\partial_t v - v \partial_{zz} v + \partial_z \langle v' w' \rangle = -\frac{1}{\rho_0} \partial_y p - f u$$
 (2)

88
$$\partial_t T - \nu' \partial_{zz} T + \partial_z \langle w' T' \rangle = \frac{\partial_z I}{c_p \rho_0}$$
 (3)

89
$$\partial_t S - \nu'' \partial_{zz} S + \partial_z \langle w' S' \rangle = \tau_R^{-1} (S_R - S).$$
 (4)

90

91 In Eqs. (1) and (2), u, v, and w represent the mean velocities in the spatial directions x (eastward), y (northward), and z92 (upward), respectively; v represents the molecular diffusivity of momentum; ρ_0 represents a constant reference density; p

93 represents pressure; and f represents the Coriolis parameter. In Eq. (3), the temperature (T) equation, v' represents the molecular diffusivity due to heat; c_n represents the heat capacity; and I represents the vertical divergence of short-wave 94 95 radiation. The effect of solar radiation absorbed by seawater is included in this equation; thus, Eq. (3) is closely associated 96 with the radiation parameterization method. Moreover, a coupled ocean biogeochemistry model must contain an additional 97 short-wave absorption process associated with chlorophyll synthesis distributed throughout the upper-ocean layer (Morel and 98 Antoine, 1994; Cloern et al., 1995; Manizza et al., 2005; Litchman et al., 2015; Hense et al., 2017). Based on the 99 methodology of Manizza et al. (2005), we applied a visible light absorption process due to chlorophyll synthesis, explained 100 in detail in Sect 4.4, to the coupled model. Equation (4) explains the vertical distribution of salinity (S). In this equation, v''101 represents the molecular diffusivity of salinity; τ_R represents the relaxation time scale; and S_R represents the observed 102 salinity distribution. In other words, the terms on the right side of this equation express the "relaxation" process based on 103 observations. Unlike 3D models, SCMs cannot reproduce horizontal advection. Therefore, as salinity is greatly affected by horizontal advection, it is necessary to prescribe and supplement the observed value to the simulated value with the terms on 104 105 the right side of Eq. (4) (Burchard et al., 2006). Please see Umlauf and Burchard (2003, 2005), Umlauf et al. (2005), and 106 Burchard et al. (2006) for further detailed information on the GOTM.

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108 3 The Ocean Biogeochemistry Model: Tracers of Phytoplankton with Allometric Zooplankton (TOPAZ)

We chose TOPAZ version 2.0 to couple with the GOTM. TOPAZ simulates the nitrogen, phosphorus, iron, dissolved oxygen, and lithogenic material cycles as well as the ocean carbon cycle while also considering zooplankton and phytoplankton growth cycles. It divides phytoplankton into small and large groups based on size, including the group of nitrogen-fixing diazotrophs. Consequently, TOPAZ handles a total of 30 prognostic and 11 diagnostic tracers. The local changes in the tracers simulated in TOPAZ can be explained by the following equation:

114

115
$$\partial_t C = -\nabla \cdot \vec{v}C + \nabla K \nabla C + S_C.$$
 (5)

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Equation (5) is an advection-diffusion equation for each state variable *C* simulated in TOPAZ. In this equation, \vec{v} represents the velocity vector calculated in the ocean model, K represents diffusivity, and S_c represents the sources minus the sinks of *C* calculated at each point in the model. TOPAZ is received data from the ocean model in terms of the transport tendency of the tracers associated with advection and horizontal diffusion, and it calculates vertical diffusion and source/sink terms internally. The biological processes of TOPAZ were reproduced with a focus on phytoplankton growth, nutrient and light limitations, the grazing process, and empirical formulas derived from observations. These are followed by the Redfield ratio (Redfield et al., 1963), Liebig's law of the minimum (de Baar, 1994), and size considerations (large organisms feed on smaller ones), which were used to establish the ocean ecosystem model (Dunne et al., 2012b). Please see Dunne et al. (2012b)

125 for further detailed information on TOPAZ.

126

127 4 The Ocean Biogeochemistry Coupled Model: GOTM-TOPAZ

TOPAZ was initially coupled with Modular Ocean Model 5 (MOM5), an OGCM developed by the GFDL. We separated TOPAZ from MOM5 and constructed two modules by separating the initialization and main calculation subroutines. This model was then modified into an SCM while adding interfaces associated with surface flux prescriptions (boundary conditions) and initial data input.

In our new coupled model, the GOTM provided ocean physics calculations for TOPAZ, and TOPAZ relayed optical feedback from the chlorophyll simulated according to these data to GOTM. A subroutine that calculates the optical feedback from chlorophyll and another that prescribes the w-advection were added to GOTM-TOPAZ (see Fig. 1 for the flow diagram). Upwelling that usually occurs along coastal areas due to wind plays a major role in changing the vertical distribution of zooplankton and phytoplankton by supplying the surface layer with nutrient-rich intermediate water (Krezel et al., 2005; Lips and Lips, 2010; Shin et al., 2017). We connected the w-advection module in GOTM to TOPAZ so that the upwelling was reproduced in TOPAZ.

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140 4.1 Initial Conditions

The initial data needed to run GOTM-TOPAZ can be divided into the data needed to operate the GOTM and TOPAZ models individually. To run the GOTM, it is necessary to have the initial ocean data (temperature and salinity) and the salinity data for the duration of the model run time. The latter are needed to relax the GOTM. For TOPAZ, initial data are needed for the 30 prognostic and 11 diagnostic tracers.

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146 **4.2 Boundary Conditions**

Atmospheric forcing data must be prescribed in GOTM-TOPAZ because it is not coupled with an atmospheric model. The atmospheric forcing variables needed to run the model are: 10 m u-wind; v-wind [m s⁻¹]; surface (2 m) air pressure [hPa]; surface (2 m) air temperature [°C]; relative humidity [%], wet bulb temperature [°C], or dew point temperature [°C]; and cloud cover [1/10].

Values for surface or bottom fluxes for a few types of tracers must be provided to accurately simulate ocean biogeochemical variables. TOPAZ includes processes for variables including sediment calcite cycling and the external bottom fluxes of O₂, NH₄, PO₄, and alkalinity (Dunne et al., 2012b). However, it does not include a process for calculating 154 the atmosphere-ocean surface flux. Therefore, we added processes for calculating the surface fluxes of O_2 , NO_3 , NH_4 , 155 alkalinity, lithogenic aluminosilicate, dissolved iron, and dissolved inorganic carbon. Of the subroutines shown in Fig. 1., the 156 calculation of the surface fluxes is implemented using generic topaz column physics. The surface flux of NO₃, NH₄, 157 lithogenic aluminosilicate, and dissolved iron is prescribed using monthly average climate values, while alkalinity is 158 calculated from prescribed NO₃ dry/wet deposition values. These surface flux data are provided by the Australian Research 159 Council's Centre of Excellence for Climate System Science (ARCCSS; http://climate-cms.unsw.wikispaces.net/Data). The 160 following equation was used to calculate the air-sea gas transfer for O_2 and CO_2 (dissolved inorganic carbon):

161

162
$$F = k_w \rho([A] - [A]_{sat})$$
 (6)

163

164 Here, F is the upward flux of gas A, and k_w is its gas transfer velocity, which can be calculated as a function of the 165 Schmidt number and wind speed at 10 m (Wanninkhof, 1992). ρ is the density of surface seawater, [A] is the concentration $[\mu mol kg^{-1}]$ of gas A at the surface of the ocean, and $[A]_{sat}$ is the corresponding saturation concentration of gas A in 166 equilibrium with a water vapor-saturated atmosphere at total atmospheric pressure (Najjar and Orr, 1998). [A] is predicted by 167 the model. Please see Najjar and Orr (1998) for further detailed information related to Eq. (6). 168

169

170 **4.3 Ocean Physics**

The GOTM simulates the physics of oceanic environments based on Eqs. (1)–(4). In the coupled model, the GOTM relays 171 the following simulated one-dimensional ocean physical variables to the TOPAZ module at each time step: potential 172 173 temperature [°C]; salinity [psu]; thermal diffusion coefficient $[m^2 sec^{-1}]$; density [kg m^{-3}]; thickness [m]; mixed layer thickness [m]; and radiation [w m^{-2}]. 174

175

4.4 Optical Feedback 176

177 As explained in Sect. 2, the photosynthesis of chlorophyll distributed throughout the upper ocean is known to have physical 178 effects. Manizza et al. (2005) used satellite observation data and OGCMs to conduct a study of changes in ocean irradiance 179 due to the absorption of visible light by chlorophyll. We used their methodology to apply the optical feedback from 180 chlorophyll on GOTM-TOPAZ in the following manner:

182
$$k_{\lambda} = k_{sw(\lambda)} + \chi_{(\lambda)} \cdot [chl]^{e_{(\lambda)}}$$
(7)

183
$$I_{IR} = I_0 \cdot 0.58$$
 (8)
184 $I_{VIS} = I_0 \cdot 0.42$ (9)

$$184 \quad I_{VIS} = I_0 \cdot 0.42$$

185
$$I_{RED} = I_{BLUE} = \frac{I_{VIS}}{2}$$
 (10)
186 $I_{(z)} = I_{IR} \cdot e^{-k_{IR}z} + I_{RED(z-1)} \cdot e^{-k_{(r)}\Delta z} + I_{BLUE(z-1)} \cdot e^{-k_{(b)}\Delta z}$. (11)

In these equations, visible light was divided into red and blue/green bands in accordance with Manizza et al. (2005). In Eq. (7), λ represents the wavelength of these bands and $k_{sw(\lambda)}$ represents the light attenuation coefficient of optically pure seawater, which has values of 0.225 m⁻¹ and 0.0232 m⁻¹, respectively, in red and blue/green bands. In these bands, the values of the pigment adsorption $\chi_{(\lambda)}$ are 0.037 and 0.074 m⁻² mg Chl m⁻³, respectively; $e_{(\lambda)}$, the power law for absorption, has values of 0.629 and 0.674 [no units], respectively. Moreover, [chl] represents the concentration of chlorophyll in mg Chl m⁻³.

Infrared light (I_{IR}) and visible light (I_{VIS}) that reach mean open ocean conditions are set in Eqs. (8) and (9), respectively, by default. However, GOTM-TOPAZ can change the light extinction method by modifying the namelist in the GOTM (see http://www.gotm.net) and this can also be used to change the coefficients of I_{IR} and I_{VIS} . The total irradiance of the red and blue/green bands that reach the ocean surface is represented in Eq. (10). Ultimately, the irradiance of visible light transmitted at each vertical level (z) can be calculated in GOTM-TOPAZ using Eq. (11). Moreover, the sum of the second and third terms on the right side of Eq. (11) represents photosynthetically active radiation (PAR) and is used in TOPAZ to calculate the growth rate of phytoplankton groups.

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202 **4.5 w-advection**

As mentioned at the beginning of Sect. 4, the upwelling phenomenon generated by coastal winds is known to affect phytoplankton growth by supplying nutrient-rich intermediate water to the upper ocean. The GOTM is already designed to allow users to prescribe w-advection to experiments. Therefore, we linked the subroutines of the GOTM that are related to w-advection to TOPAZ, so GOTM-TOPAZ users can study the impact of upwelling on the biogeochemical environment of the ocean. Users can prescribe vertical advection as a constant or input the velocities by time and depth in ASCII format to reproduce the desired form of vertical motions. Please refer to the GOTM homepage (http://www.gotm.net) and Burchard et al. (2006) for further technical details and numerical analysis of the w-advection in GOTM.

210

211 5 Experimental Setup

The East/Japan Sea is unique, with its steep topography and three large, deep, and semi-enclosed basins. Moreover, it is somewhat isolated from other major oceans, connects to the Pacific Ocean through a narrow strait, and is sometimes referred to as a miniature ocean since it contains a double gyre and experiences various oceanic phenomena (Ichiye, 1984). The high215 temperature, high-salinity Tsushima Warm Current (TWC) introduced through the Korea Strait is divided into two main 216 branches: the nearshore branch, which flows northeastward along the Japanese coast, and the East Korea Warm Current 217 (EKWC), which flows northward along the Korean coast (Uda, 1934; Tanioka, 1968; Moriyasu, 1972) (Fig. 2). Apart from 218 these two main branches, there is another that exists offshore of the first branch, but it is not present all year (Shimomura and 219 Miyata, 1957; Kawabe, 1982). To the north, the North Korea Cold Current (NKCC) flows southward along the Korean coast. 220 Furthermore, the 200-400 m East Sea Intermediate Water (ESIW) is known for its high concentration of dissolved oxygen 221 and the appearance of a salinity-minimum layer (Kim and Chung, 1984; Kim and Kim, 1999). The East/Japan Sea is divided 222 into warm and cold regions relative to the 40° N parallel, and, since the current pattern and characteristics of the East/Japan 223 Sea vary spatially and seasonally, this region is very important to oceanographic studies. This region is also considered 224 important for biogeochemical research (Joo et al., 2014; Kim et al., 2016; Shin et al., 2017) for the following reasons: the 225 nutrient-rich seawater that flows along the southern coast of the Korean Peninsula due to inflow from the Nakdong River, 226 which is located at its southeastern end; the influence of a strong southerly wind during the summer, which causes upwelling 227 off the coast of the East/Japan Sea; and the transport of this nutrient- and chlorophyll-rich seawater near Ulleungdo Island by 228 the EKWC. We selected three points that have features typical of the East/Japan Sea and for which observation data suitable 229 to use for verification exist (Fig. 2): point 107, where the EKWC and NKCC meet (130.0° E, 38.0° N); point 104, which is 230 an important location along the EKWC (131.3° E, 37.1° N); and point 102, which is in the middle of a warm eddy created as 231 the EKWC moves north (130.6° E, 36.1° N). As noted previously, these points are in regions with strong advection and thus 232 may not be suitable for testing GOTM-TOPAZ, which is an SCM. However, since the results obtained using GOTM-233 TOPAZ were significant when compared to the observations, we think that this shows that it is possible to perform 234 sensitivity experiments using GOTM-TOPAZ at several kinds of locations.

235 The observed data, such as seawater temperature and salinity, were used to initialize and relax vertical structures in the 236 GOTM throughout the simulation. These data were provided by the National Institute of Fisheries Science (NIFS; 237 http://www.nifs.go.kr/kodc). The water temperature and salinity data from the NIFS were measured at 15-m intervals at 238 depths of 0 m to 500 m. They were measured once in February, April, June, August, October, and December every year 239 beginning in 1961. For the initial data on prognostic/diagnostic tracers in TOPAZ, we used the data provided by ARCCSS 240 for use with MOM5 (http://climate-cms.unsw.wikispaces.net/Data). These initial tracer data were interpolated for each 241 location, and a spin-up was applied over 14 years for use in the experiments. For atmospheric forcing data, we input 0.75° 242 ERA-Interim reanalysis data provided by the European Centre for Medium Range Weather Forecasts (Dee et al., 2011). We 243 applied global data to our model by interpolating the latitude and longitude values of the test points.

We used the monthly average of observed seawater temperature and salinity data from the analysis fields in the EN.4.2.1, provided by the Hadley Centre at the Met Office (Good et al., 2013), to verify the results from GOTM-TOPAZ following the adjusted method in Gouretski and Reseghetti (2010). With respect to chlorophyll, we compared the results simulated by the model using observational data with a resolution of 9 km gathered by the NASA Goddard Space Flight Center's Sea-Viewing Wide Field-of-View Sensor (SeaWiFS) from October 1997 to December 2007 (McClain et al., 1998). The results of 249 simulations of dissolved oxygen and nutrients such as nitrogen, phosphorus, and silicon were tested using observational data 250 from the NIFS; these data were measured once every year, in February, April, June, August, October, and December, at 251 depths of 0, 20, 50, and 100 m. Specific measurement dates and times were not fixed, so we viewed the measurement data as 252 values that represented each month and used them to verify the model. Data from a model that operated MOM5, the Sea-Ice 253 Simulator, and TOPAZ together (MOM) were used for comparative analysis. MOM was operated using CORE-II forcing 254 data (Large and Yeager, 2009) from 1950 to 2008. We also used data from the Surface Ocean CO₂ Atlas (SOCAT) (Bakker 255 et al., 2016) from the analysis period to verify the CO₂ air-sea gas flux in TOPAZ. The time periods for which SOCAT 256 observational data exist for point 102 are April 2001, January 2005, November 2008, and December 2008. For points 104 257 and 107, the time period is April 2001. Finally, we performed a spin-up for 14 years on the initial data at each point and 258 analyzed the results of operating GOTM-TOPAZ from 1999 to 2008.

259

260 6 Results

261 Figure 3 shows the results of the GOTM-TOPAZ simulation and observational data (EN.4.2.1) as vertical distributions of the 262 water column over time. The vertical distributions of salinity at all points are well simulated and are comparable to the 263 observations, although this could also be because relaxation was applied. The water temperature at point 107, as simulated 264 by GOTM-TOPAZ, showed a cold bias in the upper layer at a depth of around 120 m (Fig. 3a). This appears to be the effect of large-scale forcing (from the EKWC) that GOTM-TOPAZ could not resolve. Similar differences in water temperature 265 266 also appeared at points 104 and 102 (Fig. 3b and 3c). Observational results showed that the water temperature was 267 particularly affected by the ESIW, a finding that did not appear in the GOTM-TOPAZ results. It was determined that since 268 GOTM-TOPAZ could not reproduce advection from the ESIW, there were differences (warm bias) in the vertical water 269 temperature distributions near depths of 200 m compared to the observational results at all points (Fig. 3).

270 We used SeaWiFS data to measure chlorophyll concentrations using light reflected from the ocean surface and thus 271 verified the results simulated by GOTM-TOPAZ. However, part of the reflected light reaches the satellite from the mixed 272 layer below the ocean surface due to a backscattering effect (Jochum et al., 2009; Park et al., 2013). Therefore, we compared 273 chlorophyll anomalies averaged up to 20 m in the data from each model and chlorophyll from SeaWiFS. The mean chlorophyll concentration at depths of 0-20 m, as simulated by GOTM-TOPAZ and MOM, had similar seasonal variabilities 274 275 at point 107; their correlation coefficients versus the observational data were 0.53 and 0.60, respectively, which is 276 statistically significant (p < 0.001) (Fig. 4a). At points 104 and 102, these correlation coefficients of GOTM-TOPAZ versus 277 the observational data were 0.25 (p < 0.01) (Fig. 4c) and 0.32 (p < 0.001) (Fig. 4e), respectively. In the case in which the 278 maximum concentration of chlorophyll at all points occurred annually on the surface layer, GOTM-TOPAZ showed smaller 279 errors against the observational results than did MOM (Fig. 4a, 4c, and 4e).

280 Phytoplankton in the East/Japan Sea are generally present in the highest concentrations at depths of around 10-60 m (Rho et al., 2012). Therefore, we averaged chlorophyll concentrations from 20-80 m to verify the model results (Fig. 4b, 4d and 281 282 4f). However, since observational data for chlorophyll in the subsurface layer (~20-80 m) were unavailable, the MOM and 283 GOTM-TOPAZ results were compared instead. There were slight differences in the scale of the minimum and maximum 284 concentrations of chlorophyll in the subsurface layer at point 107, but the two models had a correlation coefficient of 0.59 (p 285 < 0.01) and a similar seasonal variability (Fig. 4b). At points 104 and 102, the GOTM-TOPAZ chlorophyll results had a 286 slightly lower correlation coefficient against the observational data than MOM did, but its seasonal variability was similar to 287 that of the observation data and the results from MOM (Fig. 4d and 4f). However, when compared to the results from MOM, the time series of the chlorophyll anomaly in the ocean surface and subsurface layers simulated by GOTM-TOPAZ appear to 288 289 show a time shift (Fig. 4). In the TOPAZ module in MOM, the transport tendencies of each tracer were calculated in the 290 ocean model; however, this process was not carried out in GOTM-TOPAZ. In addition, MOM and GOTM-TOPAZ are not 291 only just different models of the marine physical environment; the atmospheric forcing data they each use are also different. 292 Therefore, there are complex reasons for the differences in the results of the two models, and further detailed experiments 293 and analysis are required.

294 We evaluated the performance of GOTM-TOPAZ in terms of simulations of dissolved oxygen, nitrogen, phosphorus, and 295 silicon. The sea surface dissolved oxygen at point 107 simulated by GOTM-TOPAZ and MOM had correlation coefficients 296 of 0.47 (p < 0.001) and 0.50 (p < 0.001), respectively, versus the observed data (Fig. 5a). The GOTM-TOPAZ correlation coefficient versus the observed data was 0.31 (p < 0.001) for nitrogen, 0.16 (p < 0.10) for phosphorus, and 0.19 (p < 0.05) 297 298 for silicon; these were lower than the correlation coefficients between MOM and the observed data (0.36, 0.24, and 0.33, 299 respectively; p < 0.001). However, GOTM-TOPAZ seemed to depict the seasonal variability of nutrients at the sea surface 300 well (Fig. 5b–d). At point 104, the GOTM-TOPAZ correlation coefficient was 0.37 (p < 0.001) for dissolved oxygen, 0.54 (p301 < 0.001) for nitrogen, 0.2 (p < 0.05) for phosphorus, and 0.1 (statistically non-significant) for silicon (Fig. 6). For point 102, 302 the GOTM-TOPAZ correlation coefficient was 0.59 (p < 0.001) for dissolved oxygen, 0.24 (p < 0.01) for nitrogen, 0.09303 (statistically non-significant) for phosphorus, and 0.2 (p < 0.01) for silicon (Fig. 7). In these two points, GOTM-TOPAZ 304 showed values for surface dissolved oxygen and nutrients with seasonal variabilities that were similar to those of the 305 observed data and the data from MOM (Figs. 6-7).

306 Figures 8–10 show a comparison of the vertical profiles of dissolved oxygen, nitrogen, phosphorus, and silicon averaged 307 for February, August, and the entire period from 1999 to 2008 at points 107, 104, and 102. Mixing in the upper ocean occurs 308 actively during winter due to strong winds, and GOTM-TOPAZ simulated dissolved oxygen (surface to 250 m) and nitrogen 309 (surface to 100 m) concentrations well during that season (Figs. 8–10a). However, for phosphorus and silicon at the same 310 depths, there was a difference between the GOTM-TOPAZ results and the observational data. In the case of all points, the 311 concentrations of nitrogen, phosphorus, and silicon simulated by GOTM-TOPAZ from the surface to 60 m decreased during 312 August, and these concentrations were clearly distinguishable from each depth due to strong stratification in the summer 313 (Figs. 8–10b). These stratifications appeared in the observational data. During this season, the oxygen concentration 314 simulated by GOTM-TOPAZ, increased sharply from depths of 20–60 m at points 107, 104, and 102 (Figs. 8–10b). This 315 seems to have been caused by the creation of oxygen from photosynthesis by phytoplankton. However, a highly concentrated 316 dissolved oxygen concentration is not apparent in the observational data, because the warm water, which is characterized by 317 low dissolved oxygen, is transported by the EKWC during the summer season (Rho et al., 2012). The concentrations of 318 dissolved oxygen from 80-250 m at point 107 were similar in both the results from GOTM-TOPAZ and in the 10-year 319 observational data (Fig. 8c). However, the differences increased beyond depths of 250 m. Nonetheless, the results 320 demonstrated that dissolved oxygen at 80–250 m, nitrogen, and phosphorus (but not silicon) are well simulated over 10 years 321 using GOTM-TOPAZ (Fig. 8c). The vertical distributions of dissolved oxygen and nutrients at points 104 and 102 as 322 simulated by GOTM-TOPAZ over the same time period also showed similar patterns as those at point 107 (Figs. 9–10).

323 In addition, the magnitudes of the source and sink terms of GOTM-TOPAZ were analyzed. When TOPAZ was 324 implemented three-dimensionally by being coupled with MOM, the concentration of tracers was calculated through 325 advection-diffusion processes as well as source/sink processes. On the other hand, in the case of GOTM-TOPAZ, which is 326 an SCM, it determined the tendency of state variables through vertical diffusion and source and sink terms without 327 considering advection and horizontal diffusion. At every point, the bias of dissolved oxygen seemed to be larger in summer 328 than in winter, where the vertical diffusion is stronger. Since there was a bias also in the deep sea (< 250 m), we focused on 329 source and sink terms rather than on vertical diffusion. Figures 11-13 show 10-year (1999-2008) average source and sink 330 terms of nutrients (nitrate, phosphate, silicate) and dissolved oxygen. The production of dissolved oxygen is attributable to 331 nitrate, ammonia, and nitrogen fixation, while its loss occurs in the production of NH_4 from non-sinking particles, sinking 332 particles, and dissolved organic matter and nitrification. The production of nitrate is caused by nitrification, and its loss is 333 determined by denitrification and uptake by phytoplankton. In the phosphate and silicate, the production is attributable to 334 dissolved organic matter and particles, and the loss is determined by uptake due to phytoplankton (Dunne et al., 2012b).

As shown in Figures 11–13, the source and sink of dissolved oxygen and nutrients occurred mainly in the surface layer (< 60 m), and their influence seemed to be negligible at deeper depths. The source of dissolved oxygen was remarkable in the surface layer during summer, because phytoplankton flourishes in summer. This pattern was commonly observed at all three points. The surface layer of point 102, which is the southernmost point, showed more production (consumption) of dissolved oxygen (nutrients) than did the other points in winter. Being located at the southernmost location, point 102 was greatly affected by the warm current (EKWC), which resulted in flourishing phytoplankton. However, even at this point, the source and sink of both the dissolved oxygen and nutrients made few contributions at 250 m or deeper.

Accordingly, it could be inferred that the simulation of biogeochemical variables in the deep sea (< 250 m) would be more affected by initial values than by source/sink. In order to verify this assumption, the model was simulated by setting the initial data as the observations. The results indicated that the bias of dissolved oxygen was significantly reduced in the deep sea (Fig. 14). This result indicates that tracers simulated by GOTM-TOPAZ greatly depend on source/sink processes in the surface layer (< 60 m) and are sensitive to initial values in the deep sea. Finally, to verify the air-sea gas exchange simulated by GOTM-TOPAZ, we compared the monthly average sea surface CO₂ concentrations in the model and in SOCAT. The correlation coefficient between the sea surface CO₂ concentration simulated by GOTM-TOPAZ and the observational data was 0.94 (Fig. 15). However, there were no more than six months for which the observational values existed at all points; therefore, this is a statistically insignificant value.

351

352 7 Discussion

353 In this paper, we explain the major models that comprise GOTM-TOPAZ and the biological-physical feedback loop that they 354 reproduce. In addition, we compiled data from three points of scientific importance in the East/Japan Sea, near the Korean 355 Peninsula and analyzed the results of operating GOTM-TOPAZ for a decade (~1999-2008). We compared ocean water 356 temperatures, salinity, and biogeochemical variables such as chlorophyll, dissolved oxygen, nitrogen, phosphorus, and 357 silicon concentrations against the observational data and output from the OGCM to evaluate the performance of GOTM-358 TOPAZ. The results showed that GOTM-TOPAZ had lower correlation coefficients than did OGCM but that it simulated 359 seasonal variability in a similar manner overall. In addition, we analyzed the magnitudes of the source/sink terms for 360 dissolved oxygen and nutrients, which were simulated by GOTM-TOPAZ. This analysis revealed the characteristics of the 361 model and the cause of the bias, which was shown in the vertical profile of dissolved oxygen. Consequently, GOTM-TOPAZ is mainly affected by source/sink terms in the surface layer (< 60 m) and is sensitive to initial values in the deep sea (> 250362 363 m). Future users of GOTM-TOPAZ need to consider such characteristics when designing an experiment.

The SCM (1D model) includes important physical processes and has a much lower computation cost than do the 3D models; this means that a variety of experiments can be performed repeatedly. With this advantage, 1D models can be useful to track mechanisms that are difficult to understand using 3D models. We believe that TOPAZ, in particular, can be used to obtain insights on the interactions between the chemical makeup and organisms in the ocean because it accounts for complex biogeochemical mechanisms. In addition, the key processes which are studied via TOPAZ can later be implemented into 3D models.

A variety of single-column ocean biogeochemical models have already been developed. However, GOTM-TOPAZ includes complex biogeochemical processes and models over 30 kinds of tracers; the other models, which have only simple structures, do not (Dunne et al., 2012b). Furthermore, GOTM-TOPAZ considers the gas transfer caused by changes in the atmosphere and the physical environment of the ocean, depicting the deposition of dissolved iron, lithogenic aluminosilicate, NH₄, and NO₃ due to aerosols. We believe that the sophistication of TOPAZ provides researchers with the opportunity to perform a variety of experiments.

For example, aerosol concentrations are continuously increasing over the East Asia region and are known to affect precipitation and atmospheric circulation. Thus, there is a possibility that aerosols affect oceanic biogeochemical processes as deposition occurs into the ocean, and this cannot be ignored. A variety of numerical experiments are necessary to understand this process, but they are difficult to perform using 3D models due to limitations in computing resources. However, as previously noted, GOTM-TOPAZ is fast; as such, it is useful for understanding the biogeochemical changes that occur in the ocean when the concentration of aerosols or CO_2 in the atmosphere changes. In addition, recent studies have reported that the distribution of fisheries is changing due to changes in phytoplankton size structure caused by the upwelling intensity on the coast of the East/Japan Sea (Shin et al., 2017). The TOPAZ phytoplankton are divided into two types depending on their size, which should prove to be useful in this type of future research.

385 In addition, GOTM-TOPAZ can be used in studies on feedback mechanisms in the biogeochemical and physical 386 environment of the ocean. Sonntag and Hense (2011) used a simple biogeochemistry model linked to GOTM (GOTM-BIO) to analyze the effects of phytoplankton on the physical environment of the upper ocean. The feedback from cyanobacteria, 387 388 particularly during surface blooms that cause changes in ocean surface albedo, the solar light absorption rate, and the 389 momentum relayed to the ocean by wind were applied to the model during the experiment. Sonntag and Hense (2011) 390 provided us a better understanding of the needs and direction to focus on with GOTM-TOPAZ, and we plan to apply various 391 climate-ocean biogeochemistry feedback mechanisms to it in future research. We also plan to evolve GOTM-TOPAZ into a 392 single ESM by coupling an atmospheric SCM and a model that reproduces atmospheric chemical mechanisms with GOTM-393 TOPAZ.

We separated TOPAZ from MOM and constructed a model with separate initiation and column physics modules, thus introducing the possibility of more easily coupling it with various other ocean models in the future. We are currently conducting a study on coupling TOPAZ with the Nucleus for European Modelling of the Ocean (NEMO), another OGCM that is already coupled with other biogeochemistry models, such as the MEDUSA (Yool et al., 2013) and the PISCES (Aumont et al., 2015). If NEMO and TOPAZ can be coupled successfully, a comparative analysis of the simulation results from the each biogeochemistry model might provide the driving force for improving the modelling of physical processes associated with ocean-biogeochemistry.

401

402 **Code and data availability:**

403 The GOTM-TOPAZ software is based on GOTM version 4 and MOM version 5, both available for download from their 404 respective distribution sites (https://gotm.net, https://www.gfdl.noaa.gov/). GOTM-TOPAZ is freely available at 405 https://doi.org/10.5281/zenodo.1405270.

407 Author contribution:

H.C.J. and B.K.M. drafted the paper, performed the experiments, and were primarily responsible for developing GOTMTOPAZ. J.W., H.S.P., J.L., and Y.H.B. contributed to code debugging and writing the paper.

411 **Competing interests:**

412 The authors declare that they have no conflicts of interest.

413

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Table 1: List of abbreviations.

Abbreviation	Full form
ESM	Earth System Model
SCM	Single Column Model
OGCM	Ocean Global Circulation Models
CMIP5	Coupled Model Intercomparison Project 5 (fifth phase)
GFDL	Geophysical Fluid Dynamics Laboratory
ARCCSS	Australian Research Council Centre of Excellence for Climate System Science
NIFS	National Institute of Fisheries Science
ESM2M	Earth System Model version 2, with Modular Ocean Model Version 4.1
ESM2G	Earth System Model version 2, with General Ocean Layer Dynamics
ECMWF	European Centre for Medium-Range Weather Forecasts
GOTM	General Ocean Turbulence Model
TOPAZ	Tracers of Phytoplankton with Allometric Zooplankton
MOM5	Modular Ocean Model version 5
NEMO	Nucleus for European Modelling of the Ocean
MEDUSA	Model of Ecosystem Dynamics, Nutrients Utilization, Sequestration and Acidification
PISCES	Pelagic Interactions Scheme for Carbon and Ecosystem Studies
SOCAT	Surface Ocean CO ₂ Atlas
SeaWiFS	Sea-viewing Wide Field-of-view Sensor
CORE-II	Coordinated Ocean-ice Reference Experiments II
PAR	Photosynthetically Active Radiation
Т₩С	Tsushima Warm Current
EKWC	East Korea Warm Current
NKCC	North Korea Cold Current
NB	Nearshore Branch
OB	Offshore Branch
ESIW	East Sea Intermediate Water

GOTM-TOPAZ



Figure 1. Flow diagram of the Fortran subroutines comprising the Generic Ocean Turbulence Model–Tracers of Phytoplankton
 with Allometric Zooplankton.



Figure 2. Location of points (107, 104, 102) in the East/Japan Sea and flow of the nearby North Korea Cold Current (NKCC), East
 Korea Warm Current (EKWC), Offshore Branch (OB) of the Tsushima Warm Current, and the Nearshore Branch (NB) of the
 Tsushima Warm Current.





Figure 3. Comparison of the vertical distribution for water temperature [°C], salinity [psu], and the difference (GOTM-TOPAZ minus the observations) at points (a) 107, (b) 104, and (c) 102 for the 10-year period (1999–2008).



618 Figure 4. Chlorophyll anomaly time series and correlation values for observational data (black lines), MOM5_SIS_TOPAZ results

- 619 (blue lines), and GOTM-TOPAZ results (red lines) for the 10-year period 1999–2008. (a), (c), and (e) are the mean values at depths 620 \geq 20 m and the correlations between the observations and each model at points 107, 104, and 102, respectively. (b), (d), and (f) are
- 621 the mean values at depths of 20–80 m and the correlation between the two models at points 107, 104, and 102, respectively.



Figure 5. Anomaly time series and correlation values from observational data (black lines), MOM results (blue lines), and GOTM-TOPAZ results (red lines) for concentrations of (a) dissolved oxygen, (b) nitrogen, (c) phosphorus, and (d) silicon at point 107 for the 10-year period 1999–2008; in this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.



Figure 6. Anomaly time series and correlation values from observational data (black lines), MOM results (blue lines), and GOTM-

- TOPAZ results (red lines) for concentrations of (a) dissolved oxygen, (b) nitrogen, (c) phosphorus, and (d) silicon at point 104 for the 10-year period 1999-2008; in this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.





Figure 7. Anomaly time series and correlation values from observational data (black lines), MOM results (blue lines), and GOTM TOPAZ results (red lines) for concentrations of (a) dissolved oxygen, (b) nitrogen, (c) phosphorus, and (d) silicon at point 102 for
 the 10-year period 1999–2008; in this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.





Figure 8. Vertical profiles from observational data (black dots) and GOTM-TOPAZ results (red dots) at point 107 for concentrations of dissolved oxygen, nitrogen, phosphorus, and silicon averaged from 1999–2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma. In this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.



Figure 9. Vertical profiles from observational data (black dots) and GOTM-TOPAZ results (red dots) at point 104 for concentrations of dissolved oxygen, nitrogen, phosphorus, and silicon averaged from 1999–2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma. In this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.

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Figure 10. Vertical profiles from observational data (black dots) and GOTM-TOPAZ results (red dots) at point 102 for concentrations of dissolved oxygen, nitrogen, phosphorus, and silicon averaged from 1999–2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma. In this figure, nitrogen, phosphorus, and silicon include NO₃, PO₄, and SIO₄, respectively.

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Figure 11. Vertical profiles of the tendencies of source and sink terms in GOTM-TOPAZ at point 107 for the 10-year period 1999–
 2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma.

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Figure 12. Vertical profiles of the tendencies of source and sink terms in GOTM-TOPAZ at point 104 for the 10-year period 1999–
 2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma.

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Figure 13. Vertical profiles of the tendencies of source and sink terms in GOTM-TOPAZ at point 102 for the 10-year period 1999–
 2008; (a) for February; (b) for August; and (c) annually. The shaded areas represent 1 sigma.



Figure 14. Vertical profiles from observations (black dots) and GOTM-TOPAZ results (red dots) for concentrations of dissolved oxygen averaged from 1999–2008; (a) for point 107; (b) for point 104; and (c) for point 102. GOTM-TOPAZ is simulated by prescribing observations for the initial data. The shaded areas represent 1 sigma.



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715 Figure 15. Scatterplot of mean monthly sea surface CO₂ concentrations as observed by the Surface Ocean CO₂ Atlas and 716 simulated by GOTM-TOPAZ. The thin dotted lines around the 1-to-1 line represent ± 1 and 2 µmol kg⁻¹.