

Authors' response to reviewers

- Italic font is quoted from reviewer.
- Author responses begin with [CW].
- Line numbers mentioned in our responses below are in reference to the marked-up version of the manuscript here.

Anonymous Referee #1

Received and published: 26 March 2018

General comments:

This discussion paper addresses the high-spatial resolution modelling of PAHs, which is relevant within the scope of GMD and important for the estimation of air quality. The presented model improves upon existing PAH prediction capabilities, making novel improvements to an existing tool's performance. The methods and assumptions are generally clearly outlined and valid with the conclusions supported by the results, with the exception of some specifics discussed below. The specific model improvements discussed are precisely and clearly presented, and therefore should be reproducible. To the best of my knowledge, the authors give proper credit to related work and clearly indicate their own contribution. The title clearly reflects the contents of the paper and the abstract provides a complete and relatively concise summary. Overall, the paper is very well structured and clear, and the language is completely fluent and precise. The number and quality of the references is appropriate, as is the supplementary material.

The authors provide a link to model source code, but do not include a user manual or compilation/run instructions and dependencies. This paper represents an advance in PAH modelling, where the model appears to be limited by the availability of inputs, particularly emissions; the inability of such a highresolution model to capture daily variability appears to be likely due to unresolved variability in emissions. However, the ability to make even seasonal-scale PAH estimates comparable to highly local measurements given emissions scenarios is an important asset for air-quality science. While the abstract, conclusion, and body of the paper include ambiguous use of "statistically indistinguishable" which overstates the performance of model, the actual performance of the model represents a sufficient advance for PAH modelling.

[CW] Thank you for your thorough review of our paper. We have taken all of your comments into account in our revised manuscript. Please see below for more details.

Regarding the overstatement of the model's performance: please our responses to the specific comments below.

Regarding the source code: while the code is in the public domain ([the Zenodo site](#)), unfortunately ECCC does not have the resources to support a community model, and therefore does not have instructions available for the compilation of the code and its underlying subroutine libraries on various platforms. However, I had added the following information to our Zenodo record:

"GEM-MACH is an extension of the standard GEM model which is available from <https://github.com/mfvalin?tab=repositories>. The executable for GEM-MACH is obtained by providing this chemistry library to GEM when generating its executable."

In addition, I've added a figure of the calling sequence for the model to the revised supplemental material (Section E). The "Data and code availability" section of the revised paper has been updated with this additional information. I hope this is sufficient.

Specific comments:

Adjusting Ksw to measurements taken in 2002 is a good way to navigate their high uncertainty, but some of the adjustments are incredibly large. E.g. PHEN and PYR Ksw increases by almost 2 orders of magnitude. The authors should discuss the justification of such a large change in the context of the prior uncertainty and/or possible missing mechanisms.

[CW] Firstly, your comment helped us to notice that the Ksw values we had in Table 1 of the manuscript were actually values that we initially calculated before we followed through with the

method described in the supplemental material, starting at the bottom of p9, continuing top of p.10 (related to selecting samples that contained BaP rather than all of the samples, many of which were missing BaP measurements). Table 1 in the revised manuscript is now updated to the correct values used, but these are not much different from those in the original manuscript – thus your comment still applies. However, the K_{sw} values we calculated in this study – while they are different from the Galarneau et al (2014) values – are still in line with K_{sw} values found in published sources. See Table R.1. below, and note that the Dachs & Eisenreich (2000) values for phenanthrene are of the same order of magnitude as our empirically-derived K_{sw}. There is also a range from Jonker and Koelmans (2002), which for anthracene reach close to our value. Our value for pyrene is not too much greater than the top end found in Bucheli & Gustafsson (2000, ES&T), and so on. We have added the quoted literature sources of K_{sw} to the Table 1 in the revised manuscript.

Table R.1: K_{sw} from the literature, and this study.

	PHEN	ANTH	FLRT	PYR	BaA	CHRY	BaP
Original from Galarneau et al (2014)	4.34 x10 ⁵	1.55 x10 ⁶	2.24 x10 ⁶	1.70 x10 ⁶	3.74x10 ⁷	2.82 x10 ⁷	9.59 x10 ⁷
From Dachs Eisenreich (2000, ES&T)	1.26 x10 ⁷	NA	6.31 x10 ⁷	5.01 x10 ⁷	NA	3.16 x10 ⁸	NA
From Jonker and Koelmans (2002, ES&T)	1.86 x10 ⁵ - 3.72 x10 ⁶	3.80x10 ⁵ - 1.26 x10 ⁷	5.01 x10 ⁵ - 9.12 x10 ⁶	5.13 x10 ⁵ - 8.91 x10 ⁶	4.07 x10 ⁶ - 1.82 x10 ⁸	3.24 x10 ⁶ - 3.39 x10 ⁸	2.57 x10 ⁷ - -1.17 x10 ⁹
From Xu et al. (2012, E&ES)	4.79 x10 ⁵ - 5.13 x10 ⁵	1.29 x10 ⁶ - 1.91 x10 ⁶	1.86 x10 ⁵ - 3.72 x10 ⁶	3.09 x10 ⁶ - 4.07 x10 ⁶	6.31 x10 ⁷ - 7.94 x10 ⁷	3.89 x10 ⁷ - 5.75 x10 ⁷	2.00 x10 ⁸ - -3.39 x10 ⁸
From Bucheli and Gustafsson (2000, ES&T)	2.57 x10 ⁵ - 5.89 x10 ⁶	NA	2.82 x10 ⁵ - 1.62x10 ⁶	2.75 x10 ⁶ - 1.55 x10 ⁷	NA	NA	NA
this study	2.95x10⁷	3.75x10⁷	7.71x10⁷	8.98x10⁷	1.39 x10⁸	2.09 x10⁷	1.04 x10⁸

Putting in context of the prior uncertainty and/or possible missing mechanisms: We use K_{sw} combined with K_{aw} to estimate K_{sa}. This estimation may be inaccurate if equilibrium is not achieved. There is also no temperature dependence in published K_{sw} (but there is in K_{aw}) which can make a large difference at some temperatures and will be noticeable for compounds whose partitioning varies a lot across the gas-to-particle spectrum depending on temperature. The large correction and large range of published K_{sw} values lends further support for the need to measure K_{sa} directly on soots of atmospheric relevance. We have added this extra information into the revised manuscript (lines 191-194).

Equation 6 holds only if m_{i,gas} is equal to the total mass of PAH in the parcel of air considered; i.e. prior to partitioning, all of the PAH is gas-phase. Does this mean that once the PAH partitions to water it is considered lost? Is partitioning to cloud-water irreversible? A clarification of the fate of PAH that undergoes water uptake but not precipitative loss from the atmosphere is warranted.

[CW] In GEM-MACH we have separate tracers for particulate PAHs and gas-phase PAHs which are passed into the scavenging subroutines. There could be particle-phase PAHs in the same parcel, which would undergo scavenging via the particle-scavenging mechanism. However, if we understand the reviewer's comment correctly, we think they are talking about the gas-water

partitioning of Equation 6, which applies only to the gas-phase mass of PAHs in the parcel. Yes, at the start of each chemistry time step, all of the gaseous PAHs are considered to be in the gas phase (none start out in the aqueous phase, and the within-cloud aqueous fraction is calculated at every time step when cloud is present). To further clarify the cloud-water scavenging process: Once the gaseous PAHs partition to cloud-water in GEM-MACH (which is re-calculated at each time step) they are subject to rain-out (cloud-to-rain conversion) process; the relative amount of PAH mass within the cloud water which is transferred to precipitation is thus lost from the amount within the cloud. At the end of each chemistry step, the fraction of the dissolved PAH tracers contained within cloud water which are *not* removed by this rain-out process will be returned to gas phase. For the fraction that go into rain water, there is also a parameterization in the rain scavenging code for possible evaporation of rain before it reaches the ground. That fraction of PAHs released via evaporation of precipitation would also return to the atmosphere, albeit in the column below the cloud. The remaining fraction that isn't evaporated is counted towards the wet deposition in the model, which is treated as irreversible (i.e., no re-emission of PAHs from the ground after they are deposited). The revised manuscript has been updated with these further clarifications (lines 282-289), and we hope they satisfy the reviewer's question.

L460 Biases of BaA and BaP indicate NO₃ reactions that are noted in the literature. Here a few more lines of discussion of this point would be helpful. A back-of-the-envelope estimate of the effect of these reactions would increase confidence that these are the reason (or not) for the remaining bias.

[CW] According to Keyte et al (2013), the PAH-NO₃ reaction is actually more important for gas-phase PAHs. Since BaA and BaP have a relatively small fraction in the gas-phase, a gaseous NO₃ loss mechanism would have a relatively small impact on total (gas+particle) BaA and BaP. That said, Liu et al (2012, ES&T) determined k_{NO_3} second order rate coefficients for heterogeneous (on-particle) reactions with pyrene, chrysene, and benz(a)anthracene. The BaA reaction rate was $k_{\text{NO}_3} = 4.3\text{E-}12 \text{ cm}^3/\text{molec}/\text{s}$. When we do a back-of-the-envelope calculation of BaA lifetime (the time required for BaA concentrations to drop to 1/e of its initial concentration) with respect to heterogeneous reaction with NO₃, using typical night-time concentrations of NO₃ from the model (~3.2pptv or 6.67E7 molecules/cm³), we get a lifetime of about 1 hour for particulate BaA. Thus, at moderate to high concentrations of NO₃, gas-phase reactions of PAHs with NO₃ may be a significant loss, and thus a probable cause for the remaining bias. However, this calculation assumes that the one Liu et al (2012) study is correct and applies to all atmospheric conditions (temperatures, etc). More research should be done to determine whether this reaction should go into PAH CTMs. We have added this discussion to the revised manuscript at the end of that paragraph (lines 476-488).

L560 The R-values in these site-by-site comparisons are fairly low even after the anomalous sites are removed. Discussion of the causes of this low correlation is warranted. At the considered time and spatial scales, is unresolved time-variability in emissions too large? The conclusions should highlight further the reliance on accurate emissions and their time and spatial distributions for daily and small-scale predictions. The inability of the model to reproduce short-timescale PAH concentrations is a weakness considering its high resolution.

[CW] We agree that accurate emissions at finer time and spatial resolutions would greatly improve model results. Ideally, increasing model resolution would increase R at shorter timescales, however, given the large dynamic range and sharp spatial gradients that PAH concentrations have and the difficulty in modelling plume locations accurately, in reality we end up with *lower* R values at higher resolution than those reported in other studies*. Thus, the errors in modelled transport

are likely another cause for the low correlation coefficients at high time and spatial resolution (in addition to the unresolved time-variability in emissions). We have added this to the text of the revised manuscript (lines 593-597).

It was not our intention to dwell on the R values in the model-measurement comparison because of the high uncertainties in PAH modelling and emissions. For example, we did not report on the R values we obtained for the NAPS network analysis (which were much better than those for the NATTS network, at about $R=0.6-0.7$ depending on the PAH species). However, we mentioned them in this part of the text to emphasize the improvement when grid points with known point-source emission errors are removed.

*Note many other PAH modelling studies do not report correlation coefficients (e.g., Aulinger et al, 2011 reported index of agreement instead of R; Zhang et al, 2016, reported mean fractional bias and mean fraction error), and those that do report R values have model-measurement pairs that have been averaged over larger spatial scales or larger timescales (or both). For example, Aulinger et al (2007) achieved better R values of 0.75 for weekly averages, and 0.58 for 2-day averages, however, they only had 6 measurement sites in their study, and coarser grid (18-km) resolution, which may have smoothed over some of the variability we see at higher resolution. Also Friedman et al (2012) achieved good R values of 0.64 or higher, however, these were for their 5-year mean data over 15-16 sites globally (most sites in Europe, NE-North America, and China), at $4^{\circ}\times 5^{\circ}$ resolution – both spatial and time scales that smooth out variability. Matthias et al (2009) report good correlation coefficients of 0.3 to 0.8, but these were for the times series at each site, rather than for all sites together. Also their model was at coarser ($1^{\circ}\times 1^{\circ}$) resolution than ours.

The number of aspects of the simulation that are compared to observations is a major asset for this work. The comparison to K_p , particulate fraction, wet deposition, and concentrations across many sites and PAHs allows a very detailed and transparent assessment of the model.

[CW] Thank you!

The winter/summer differences in wet deposition show that snow-initiated wet deposition is a definite weakness of the model. In the conclusions, while the authors mention that snow scavenging is new to the model, it should be acknowledged that it requires improvements going forward, along with possibilities of what these improvements might be.

[CW] We added this text to the conclusion (at the end of the first paragraph, line 732-734): “However, the modelled wet deposition was biased high - particularly in the wintertime - thus further improvements to these parametrizations are required if the model is to be used for deposition studies.” and we added some additional text to the first paragraph of section 4.4 (lines 667-669): “...removing PAHs in the model. This may be due to our simplification of a constant snow surface area, which may be set too high, or due to inaccuracies in the modelled or measured precipitation.”

The simplification of a constant snow surface area may have introduced a high bias in snow scavenging if actual surface area of snow was smaller. Ranges of $0.01 \text{ m}^2/\text{g}$ to $15 \text{ m}^2/\text{g}$ have been reported in the literature for snow surface area (e.g., Hoff et al, 1998; Hanot et al, 1999; Legagneux et al, 2002; Domine et al, 2007; and Hachikubu et al, 2014), but these are based usually on measurements from fallen snow (rather than falling snow). Fresh snow surface area was usually on the higher end, which is why we chose a value of $1 \text{ m}^2/\text{g}$, which was at the high end of the reported values. The interfacial adsorption coefficient was dependent on modelled saturation

vapour pressure for PAHs, which also may have introduced errors. The modelled gas-particle partitioning is likely *not* to blame, because the modelled particulate fraction was underestimated compared to IADN measurements (Fig.10) and snow scavenging of particles is more efficient than that of the gas-phase.

L675 *“but it is at least promising to see that there are no particular sites where the model is consistently too high or too low, rather the errors in spatial distribution are haphazard and may be due to propagation of error, rather than any major error with the PAH scavenging scheme itself” I think that this paragraph should be re-written. The difference in the model compared to the observations is definitely due to the propagation of error, in all cases. The fact that these errors are different for different PAHs and sites is a result of the complexity of the processes involved, as the authors write, but I do not think that this makes them more or less promising.*

[CW] We have rephrased the second half of that paragraph (lines 709-718) as follows:

“... highest in Burlington). The lack of spatial or temporal pattern in the sign and/or magnitude of wet deposition biases indicates that there is no major error with the PAH scavenging scheme itself. Given that wet deposition of PAHs relies on the correct simulation of many model factors (meteorology, scavenging parameters, atmospheric concentrations, etc), our work suggests that more process studies aimed at quantifying wet deposition are needed. In fact, other PAH models also overestimate PAH deposition (Matthias et al, 2009; Friedman et al, 2012).”

L705 *“we have determined from our sensitivity test that the GEM-MACH-PAH model has a linear response to a 50% variation in mobile emission factors, simulating concentrations that vary up to 30%.” In the results section, the authors present a /non-linear/ response of concentrations to variations in mobile emission factors. Increasing emission from 50% to 100% yields _10% difference in concentrations, while increasing from 100% to 150% yields _30% difference in concentrations. Figure 8) Further displays non-linearity of mobile source factor. What is the reason that it is not linear? What change in total emissions results from the mobile source change?*

[CW] Our apologies, this is an important error in the manuscript that you caught. What we actually did was scale the mobile emissions by 0.5 (halved) and 2.0 (doubled, *not* 1.5). And what we get when we sum over the domain in total PAH area emissions (doesn't include major points) for each scaled scenario is in Table R.2 below – which was calculated for 14 UTC on a Monday in October as an example. This amounts to -10% for the 0.5x change in mobile emissions and +20% for the 2.0x change in mobile emissions. So the results are linear, as the latter is double the former. The manuscript has been corrected and clarified and all mention of 1.5x has been replaced with 2.0x.

Table R.2: Total PAH emissions (g/s) in the Pan Am domain for 14 UTC on Monday in October

	0.5x mobile emiss	1.0x mobile emiss	2.0x mobile emiss
benzene	2530.3	3030.8	4031.7
phen	25.3	29.0	36.5
anth	3.5	3.8	4.6
flrt	11.5	12.17	13.6
pyr	11.15	11.9	13.5
baa	4.3	4.7	5.4
chry	4.13	4.5	5.11
bap	5.0	5.8	7.4
Total of 7 PAHs	64.8	71.9	86.1

Abstract, L694 and ca. L570 “with concentrations statistically indistinguishable from observations, at 2.5-km resolution”. If I understand the analysis, this phrasing highly overstates the performance of the model. Firstly, the 2.5 km resolution is not a significant part of the model-measurement comparison. The sites are a few dozen distributed all across the Northeastern U.S. and Southern Ontario, and are aggregated, and therefore

the comparison is not testing the high spatial resolution. Secondly, by grouping all of the measurement-model pairs for the whole domain and season, a more accurate statement of the agreement would be “Over the domain as a whole and on the seasonal time-scale, the model is unbiased with respect to measurements.” The above phrasing is misleading and ambiguous, and must be changed to at least clearly state the statistical test performed.

[CW] We have rephrased the text in the revised manuscript to better indicate the model’s performance as you’ve suggested.

E.g., the conclusion (line 735-741) now reads as follows: “Over the model domain, at seasonal time-scales, the GEM-MACH-PAH simulation of benzene and six semi-volatile PAHs (PHEN, ANTH, FLRT, PYR, BaA, CHRY) is statistically unbiased with respect to measurements ($t < 1$ and $p > 0.05$). For the seventh PAH species, BaP; its summertime average is simulated to a similar level of accuracy. However, it appears the model’s OH, O₃ and PM biases were additive, resulting in a wintertime average that is biased significantly high for BaP.”

The abstract was similarly changed (line 18). Finally, near line 570 of original manuscript, when $t < 1$ and $p > 0.05$, this by definition means that the two datasets have no significant difference. Therefore, our expression that the model was indistinguishable from the measurements is technically true. However, to not overstate the performance of the model, we have added the caveat in the revised version (line 598) that it is when taking all of the data *across the domain* (rather than at each specific site).

That said, our analysis of the Hamilton, ON measurement campaign has specifically assessed the model’s spatial performance at high resolution.

Technical corrections:

Equation 1: “b” in equation, but “B” in text

[CW] Corrected in revised manuscript, line 166.

L165 “amoung” should be “among”

[CW] Corrected in revised manuscript, line 168.

L222 “In order to investigate whether these U.S. values would be representative of conditions in Canada and whether only have those two fuel-type categories are adequate, . . .” I believe should read: “In order to investigate whether these U.S. values would be representative of conditions in Canada and whether having only these two fuel-type categories is adequate, . . .”

[CW] Thank you. The phrase has been revised, line 228.

Equation 4: the “reduced” and “i” labeling is confusing; indicating a forward timestepping would be clearer.

[CW] Thank you. We’ve revised such that $[BaP]_i$ is now $[BaP]_t$ and $[BaP]_{reduced}$ is now $[BaP]_{t+dt}$.

L269 “. . .for the seven PAHs is a linear relationship with inverse temperature.” Should read “. . .for the seven PAHs are proportional to inverse temperature.”

[CW] Yes, thank you. This has been revised, line 275.

L380 “NATTS is an a U.S.” should read “NATTS is a U.S.”

[CW] Corrected in revised manuscript, line 394.

L444 “. . . slope of the best-fit line is very close to 1.” Here it is preferable to simply quantify the slope and remove the qualitative phrase “very close”.

[CW] We’ve changed this to “the slope of the best-fit line is 0.96.”, line 458-459.

L648 “gages” should read “gauges”

[CW] Thank you. This has been corrected in the revised manuscript, line 682.

Figure 4: The white circles are difficult to see on difference maps. A more visible color should be used. In all 4 map panels, the grey color of some of the dots is not on the color scale.

[CW] We have added text in the caption to explain that the grey dots are sites that are missing measurements in the summer (but not in the winter). We also changed the background to a darker color in the two right-hand maps so that the white dots are more visible (see Fig. R.1 below):

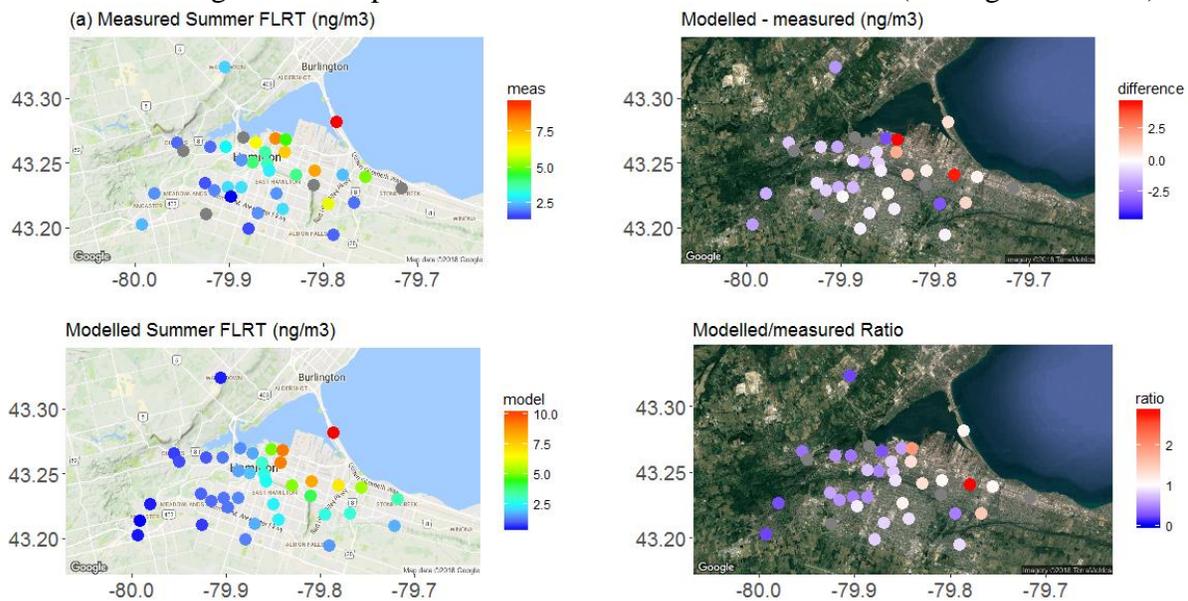


Fig R.1: replacement for Fig 4a in the paper.

Figure 5 a) text too small Figure 5 b) great figure, but purple overlay hard to discern.

[CW] Fig 5.a was made larger, and the font is now more readable. The corrected-BaP box has been moved to be beside the uncorrected BaP box – see Fig R.2 below for the revised figure.

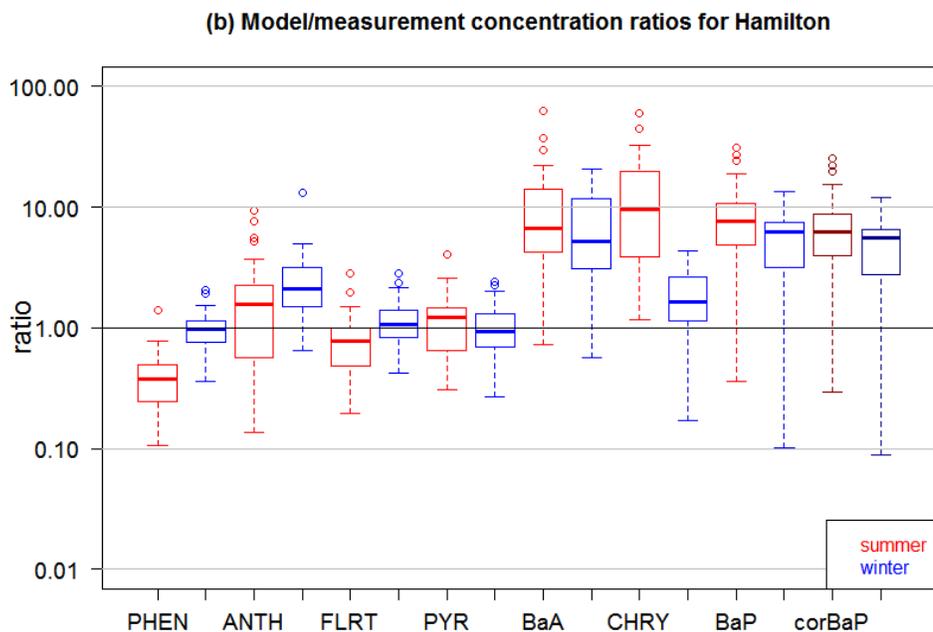


Fig R.2: replacement for Fig 5b in the paper.

Figure 6 a) Should remove irrelevant labels on each y-axis ($\leq 1e4$ on left, $> 1e3$ on right)

[CW] Thank you for this suggestion. These have been removed.

Figure 10 a) remove meaningless negative particle fraction axis labels Figure 10: b) does not exist but is mentioned in Figure 10 caption.

[CW] The plots have been split up and each has an appropriate y-axis (no negative particulate fractions, and no ratios greater than 100) – see Fig. R3 below. We removed the reference to (a) and (b) in the caption. Thank you!

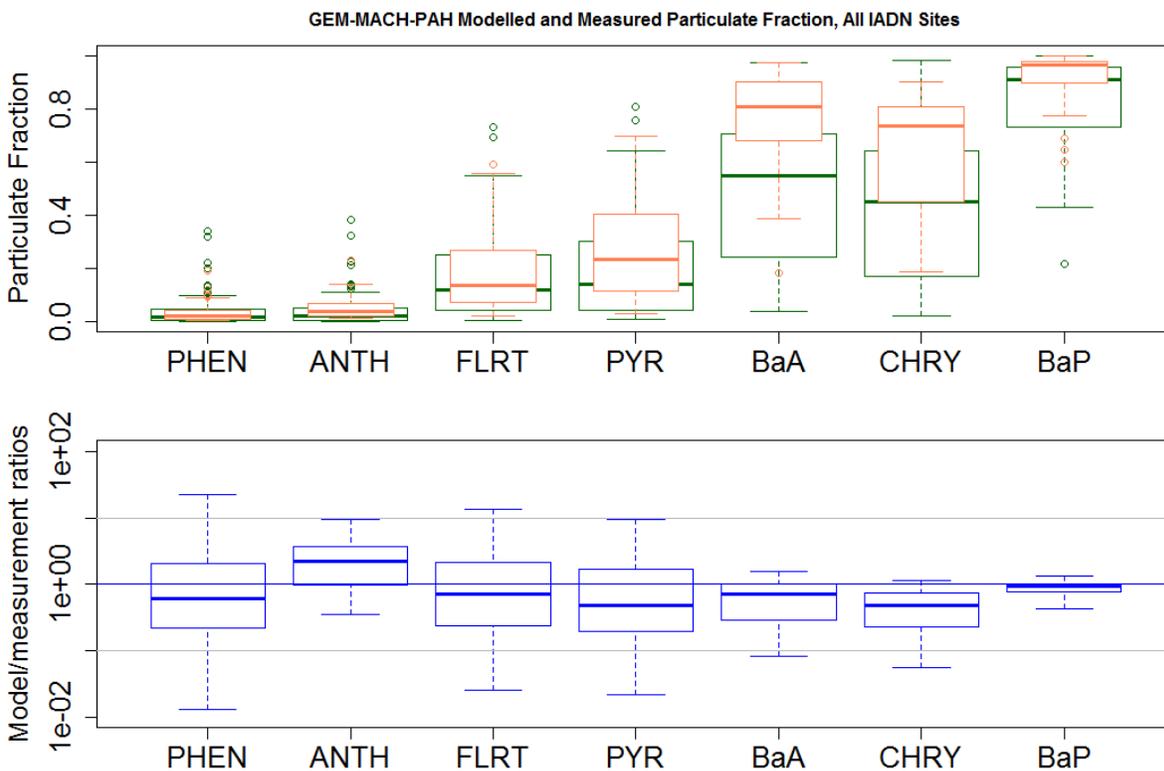


Fig R.3: Replacement for Fig.10 in the manuscript. New caption is: “GEM-MACH-PAH modelled (green) and measured (orange) particulate fraction (f) of all PAHs at all IADN sites, and their model/measurement ratios (blue). The blue line indicates the 1-to-1 line, and the grey lines are for ratios of 10 and 0.1.”

Anonymous Referee #2

Received and published: 18 June 2018

General comments:

The present manuscript describes the improvements in PAH modelling in North America. The results with an analysis and discussion of the biases are lengthy and clearly presented. The strength and remaining limitations of the modelling system are put in evidence.

[CW] Thank you for your review of our paper. Below we will address your comments.

Major comment: Possible reasons like the missing reaction with NO₃ radicals are given for the high BaP model bias. I would like to see a discussion of what the recent results of Mu et al. 2018(DOI:10.1126/sciadv.aap7314), if implemented in GEM-MACH-PAH, would change the predictions for BaP. The reduced OA diffusivity would increase BaP lifetime especially in winter. On the other hand the ROI temperature-dependent reaction of BaP is predicted to be the major cause of changes compared to the Kwamena’s parameterization.

[CW] Thank you for pointing us to this recent paper by Mu et al. As noted in the original manuscript (p8, line 257), we chose the Kwamena approach since the BaP-O₃ scheme from that work were mid-range of the three available in the literature (including P_{schl} and Kahan as well), while that described in Mu et al presents a fourth option, potentially worth considering for low temperature conditions (e.g., in Arctic or global simulations). We have added the following discussion to our revised paper (line 480-488):

“Additionally, Mu et al (2018) suggest that the heterogeneous BaP-O₃ reaction should be temperature-, humidity-, and organic aerosol phase state-dependent (none of which are taken into account in the Kwamena scheme used in our work). However, it has been shown that the Kwamena scheme and the Mu scheme produce similar results in mid-latitudes (where our study is located) (Mu et al, 2018). Spring/summertime BaP would be minimally affected, as outdoor temperatures at that time of year resemble the room temperature laboratory conditions that the Kwamena scheme was based on. Additionally, our positive model bias would likely increase in the fall-wintertime, when low temperatures and humidity would increase BaP lifetime in the Mu scheme.”

Furthermore, we expect the relative change in results to be small with respect to the impact of emissions uncertainties.

Minor comments: In a few instances references to figure panels a), b) and c) are given although no trace of it can be found on figure 6 and 7, for example.

[CW] Thank you for catching those errors. In our revised manuscript, we have clarified all figures so that (a), (b), (c), etc are included and consistent with the text and captions (Figs, 2, 6, 7, and 11 updated).

GEM-MACH-PAH (rev2488): a new high-resolution chemical transport model for North American PAHs and benzene

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Abstract. Environment and Climate Change Canada’s online air quality forecasting model, GEM-MACH, was extended to simulate atmospheric concentrations of benzene and seven polycyclic aromatic hydrocarbons (PAHs): phenanthrene, anthracene, fluoranthene, pyrene, benz(a)anthracene, chrysene, and benzo(a)pyrene. In the expanded model, benzene and PAHs are emitted from major
5 point, area, and mobile sources, with emissions based on recent emission factors. Modelled PAHs undergo gas-particle partitioning (whereas benzene is only in the gas phase), atmospheric transport, oxidation, cloud processing, and dry and wet deposition. To represent PAH gas-particle partitioning, the Dachs-Eisenreich scheme was used, and we have improved gas-particle partitioning parameters based on an empirical analysis to get significantly better gas-particle partitioning results than
10 the previous North American PAH model, AURAMS-PAH. Other added process ~~parameterizations~~ parametrizations include the particle phase benzo(a)pyrene reaction with ozone via the Kwamena scheme and gas-phase scavenging of PAHs by snow via ~~vapor vapour~~ sorption to the snow surface.

The resulting GEM-MACH-PAH model was used to generate the first online model simulations of PAH emissions, transport, chemical transformation and deposition for a high resolution domain
15 (2.5-km grid cell spacing) in North America, ~~centered~~ centred on the PAH-data-rich region of southern Ontario, Canada and the north-eastern United States. Model output for two seasons was compared to measurements from three monitoring networks spanning Canada and the U.S. Average summertime model results were found to be statistically ~~indistinguishable~~ unbiased from measurements of benzene and all seven PAHs. The same was true for the winter seasonal mean, except for
20 benzo(a)pyrene (BaP), which had a statistically significant positive bias. We present evidence that the benzo(a)pyrene results may be ameliorated via further improvements to PM and oxidant processes and transport. Our analysis focused on four key components to the prediction of atmospheric PAH levels: spatial variability; sensitivity to mobile emissions; gas-particle partitioning; and wet

deposition. Spatial variability of PAHs/PM_{2.5} at 2.5-km resolution was found to be comparable to
25 measurements. Predicted ambient surface concentrations of benzene and the PAHs were found to be
critically dependent on mobile emission factors, indicating the mobile emissions sector has a signif-
icant influence on ambient PAH levels in the study region. PAH wet deposition was overestimated
due to additive precipitation biases in the model and the measurements. Our overall performance
evaluation suggests that GEM-MACH-PAH can provide seasonal estimates for benzene and PAHs
30 and be suitable for emissions scenario simulations.

1 Introduction

Polycyclic aromatic hydrocarbons (PAHs) are semi-volatile atmospheric pollutants that have numer-
ous negative health effects (some are carcinogenic, mutagenic, and teratogenic) (Kim et al., 2013).
Measurements of PAHs in North America are sparse in both time (typically 24-hour averages, every 6
35 days) and space (limited surface measurement networks), yet show ambient concentrations that reg-
ularly exceed the Ontario provincial government’s health-based threshold (Galarneau et al., 2016).
Similarly, benzene is a gas-phase single-ring aromatic hydrocarbon, is a known carcinogen, and
also exceeds atmospheric health-based guidelines (Galarneau et al., 2016). Accurate, 3-dimensional
modelling of PAHs and benzene can fill in the space-time gaps of the measurements, identify atmo-
40 spheric processes that are responsible for the threshold exceedances, and simulate effects of emis-
sions scenarios.

AURAMS (A Unified Regional Air quality Modelling System) was an offline (meteorology from
a weather forecast model used as an input), Eulerian 3-D chemical transport model (CTM) de-
veloped by Environment and Climate Change Canada (ECCC). In Galarneau et al. (2014), AU-
45 RAMS was modified to include seven PAH species (phenanthrene, anthracene, fluoranthene, pyrene,
benz(a)anthracene, chrysene, and benzo(a)pyrene – hereafter abbreviated to PHEN, ANTH, FLRT,
PYR BaA, CHRY, BaP, respectively). AURAMS-PAH included emissions, transport, gas-particle
partitioning, oxidation of the gas-phase PAHs with OH, dry deposition, and wet deposition of the
particle-phase PAHs. This model was able to accurately simulate the 2002 annual average PAH
50 concentrations in North America when compared to 45 measurement sites, located in Ontario,
the north-eastern U.S., and California. However, the AURAMS-PAH gas-particle partitioning over-
predicted the gas phase for the lighter PAH species, and was employed at relatively poor time and
spatial resolutions. It was also missing two known PAH loss processes: the surface reaction of O₃
on particulate BaP (Kwamena et al., 2004, 2007; Ringuet et al., 2012; Keyte et al., 2013; Liu et al.,
55 2014), and snow scavenging of gas-phase PAHs (Franz and Eisenreich, 1998; Daly and Wania, 2004;
Lei and Wania, 2004; Skrdlíková et al., 2011). These missing processes, along with the coarse (42-
km) spatial resolution, may have contributed to the differences between model results and measure-

ments. Also, this model used PAH emission factors for mobile emissions which are now out-of-date for representing the modern vehicle fleet.

60 Other PAH CTMs include GEOS-Chem (Friedman and Selin, 2012; Thackray et al., 2015), which is a global model, CMAQ, which was run on the Europe continental domain in Aulinger et al. (2007), and in North America in Zhang et al. (2016, 2017), FARM (Flexible Air quality Regional Model) (Gariazzo et al., 2007), which is a regional model, applied for the region of Rome, Italy, ~~and WRF-Chem-PAH (Mu et al., 2017), which modelled East Asia.~~ The most relevant of these model
65 studies to our own is the one by Zhang et al. (2016, 2017), whereby they ran CMAQ with 16 PAH species added, at 36-km resolution in a mainly U.S. domain (that included parts of Canada and Mexico), evaluated their model results against NATTS measurements, and used their results to determine cancer risk to U.S. human populations from various sources.

Therefore, the goal of this study is to update and improve ECCC's PAH modelling capabilities by
70 using a more advanced model framework, updating emission inventories, and utilizing better process representation of PAHs than were used in AURAMS-PAH to allow better exploration of PAH processes and scenarios. To achieve this goal, GEM-MACH (Global Environment Multiscale model – Modelling Air quality and CHEmistry), ECCC's next generation, online air quality forecasting model (meteorology and air-quality are predicted in the same code) was modified to include the same seven
75 PAH species, as well as benzene. PAH processes parameterization were improved in the following ways: 1. On-road mobile PAH emissions were updated with more recent data, to better represent the modern vehicle fleet; 2. gas-particle partitioning parameters were improved based on empirical results and analysis of AURAMS model output; 3. process representation for the on-particle O₃ - particulate BaP reaction was added to the model; and 4. process representation for in- and below-
80 cloud wet scavenging (including scavenging by snow) were added for gas-phase PAHs and benzene. Simulations using GEM-MACH-PAH were then carried out at high (2.5-km) spatial resolution in a small, but densely populated North American domain including southern Ontario, and most of the northeastern U.S. (Fig. 1) for summer and winter of 2009. We refer to this region as the "Pan Am" domain because it was created for high-resolution air quality modelling during the 2015 Pan American Games in Ontario (Joe et al., 2017). This domain contains approximately 109 million people, including about 38% of the Canadian population and 30% of the U.S. population. The model results were evaluated using measurements from a high-spatial-resolution campaign in Hamilton, Ontario (Anastasopoulos et al., 2012), as well as the binational Integrated Atmospheric Deposition Network (IADN), the Canadian National Air Pollution Surveillance network (NAPS), and the U.S. National
90 Air Toxics Trends Stations network (NATTS). We focus our model evaluation on spatial variations at high resolution, estimating the level of model sensitivity to uncertainties in the PAH emission factors, gas-particle partitioning, and wet deposition, which are all related to novel aspects of the GEM-MACH-PAH model.

The following sections will further describe the GEM-MACH-PAH model (Section 2), the measurements used for evaluation (Section 3), the results of the model evaluation (Section 4), and conclusions (Section 5).

2 Model Description

In the present study, we have modified ECCC's high-resolution air quality forecasting model, GEM-MACH (hereafter called "GEM-MACH-PAH"), to include benzene and seven PAHs (in both gas and particle-phases) and have carried out 6 months of simulations in 2009 at the highest resolution (2.5-km grid cell size) in a North American domain yet reported for PAH simulations, to our knowledge. We have also tracked PAH wet deposition, and gas-particle partitioning, and have attempted to qualify model sensitivity to uncertainty in mobile emission factors, which has not been reported in other model studies.

2.1 GEM-MACH overview

GEM-MACH is an on-line, 3D chemical transport model, which is embedded in GEM, ECCC's operational numerical weather prediction model (Côté et al., 1998b, a; Moran et al., 2010) ([available online here: https://github.com/mfvalin?tab=repositories](https://github.com/mfvalin?tab=repositories)). On-line models such as GEM-MACH improve air-quality chemical prediction performance by reducing interpolation errors between different model coordinate systems and removing the input/output time and disk storage required for the transfer of meteorological input files to their off-line CTM counterparts (e.g., Baklanov et al., 2014). The coupling to meteorology is a one-way process in this version, whereby chemistry does not influence the meteorology. More detailed description of the gas-, aqueous-, and particle-phase process representations of GEM-MACH, and an evaluation of its performance for common pollutants such as ozone, particulate matter (PM), and ammonia appears in Moran et al. (2013); Makar et al. (2015b, a); Gong et al. (2015), and Whaley et al. (2018). Here we will focus on the model changes made to include PAH species and processes.

GEM-MACH is used to provide ECCC's twice-daily, 48-hour operational public forecasts of criteria air pollutants (ozone, nitrogen oxides, PM), as well as the Air Quality Health Index [<https://ec.gc.ca/cas-aqhi/>]. To reduce the computational burden for forecasting, the PM size distribution is represented using a simplified sectional treatment consisting of two size bins, a fine-fraction bin for particles with Stokes diameter from 0 to 2.5 μm and a coarse-fraction bin for particles with Stokes diameter from 2.5 to 10 μm (Moran et al., 2010), with sub-binning used for those particle processes requiring a finer particulate size distribution. Here, we utilize the research version of GEM-MACH version 2, revision 2476, with this two-size-bin representation as our starting point for PAH modifications. The model grid used corresponds to a rotated latitude-longitude map projection with 2.5-km horizontal grid spacing

and a hybrid vertical coordinate with 80-level vertical discretization spanning the atmosphere from the surface to 0.1 hPa.

2.2 Model modifications for benzene and PAH species

130 Our modifications to GEM-MACH include adding benzene and seven gas-phase and 14 particle-phase (7 species \times 2 size bins) PAHs to the species arrays, and adding the gas-particle partitioning subroutine described in Galarneau et al. (2014), but with updated partitioning parameters (see Section 2.2.1). Since PAHs have very small concentrations relative to criteria air contaminants, as in Galarneau et al. (2014), we assume they do not have a significant effect on oxidant concentrations
135 (O_3 and OH). Thus, the PAHs in GEM-MACH-PAH make use of the outcomes of the model's gas and aqueous-phase chemistry in a diagnostic fashion for PAH oxidation. Processes in which the PAHs participate directly include advection, vertical diffusion, plume rise of major point source emissions, aerosol particle microphysics, in- and below-cloud scavenging, and dry and wet deposition of both gas and particle phases. Some of these processes and/or their controlling parameters
140 were updated relative to Galarneau et al. (2014) and are described in the subsections below. Like AURAMS-PAH, the total (gas+particle) PAH emissions were treated as gas-phase emissions in GEM-MACH-PAH, since these quickly repartition between particles and gas phases following emission. The non-PAH and PAH emissions are described further below. [The MACH part of the model code is available here: https://zenodo.org/record/1162252#.Wt39IMZII-J](https://zenodo.org/record/1162252#.Wt39IMZII-J), and the call sequence of
145 [the code is included in Appendix E of the supplemental information.](#)

2.2.1 Gas-particle partitioning

As PAHs are semi-volatile organic compounds that partition between the particulate and gaseous phases of the atmosphere, their partitioning is a major determinant of their atmospheric fate (Bidleman, 1988). Despite decades of study (Junge, 1977; Yamasaki et al., 1982; Bidleman and Foreman, 1987;
150 Pankow, 1987; Smith and Harrison, 1996; Dachs and Eisenreich, 2000; Lohmann and Lammel, 2004; Keyte et al., 2013), the mechanisms responsible for PAH partitioning and its spatiotemporal variability are not well-understood. AURAMS-PAH included two parametrizations to calculate gas/particle partitioning: Junge-Pankow, JP (Junge, 1977; Pankow, 1987) and Dachs-Eisenreich, DE (Dachs and Eisenreich, 2000), both of which, when applied for partitioning in AURAMS-PAH, assigned too much PAH
155 mass to the gas phase. The two schemes resulted in surprisingly similar gas-particle partitioning (Galarneau et al., 2014). We carried out post-processing and analysis on the AURAMS-PAH model output from both schemes as well as the observations of gas and particle PAHs from the Galarneau et al. (2014) study, in order to determine which scheme to proceed with in GEM-MACH-PAH, and how it could be improved.

160 Measured PAH partitioning typically takes the linear form of:

$$\log K_{p,k} = m_K \log p_{L,k}^{\circ} + b_K, \quad (1)$$

where $K_{p,k}$ is the partitioning coefficient for each PAH species, k:

$$\log K_{p,k} = \log \left[\frac{C_p / C_{TSP}}{C_g} \right], \quad (2)$$

and C_p , C_{TSP} , and C_g are the concentrations of the particulate PAH, the total suspended particles, and the gas-phase PAH, respectively. $p_{L,k}^{\circ}$ is the sub-cooled liquid vapour pressure of the k'th gas, and m_k and B_k are empirically derived coefficients. This linear relationship (where the $\log K_p$ of all PAH species in any given measurement sample fall on a line relative to their $\log p_L$) is common among homologous compound groups such as PAHs, polychlorinated biphenyls (PCBs), and polychlorinated dioxins and furans. However, the JP formulation only allows for $m_K = -1$ (see Section B in the supplemental material for more detailed information and a derivation). Conversely, observation-based estimates show a wide variety of $|m_K|$ values that are usually less than 1 (e.g., Fig. B.1a in the Supplemental Material), and this could be the reason why the AURAMS-PAH JP model results under-predicted the particulate fraction.

Therefore, we proceeded with the Dachs-Eisenreich formulation in GEM-MACH, but with improved parameters. The Dachs-Eisenreich (DE) partitioning formulation was adapted from work examining water-sediment partitioning (Dachs and Eisenreich, 2000). The DE expression for K_p (Eq. (B.2.1) is related to the octanol-air and soot-air ($K_{SA,k}$) partitioning coefficients, the latter depending on the soot-water ($K_{SW,k}$) and air-water partitioning coefficients. The soot-water partitioning coefficients are highly uncertain. Their values in the literature span two orders of magnitude for the same compound (Dachs and Eisenreich, 2000; Bucheli and Gustafsson, 2000; Jonker and Koelmans, 2002; Xu et al., 2012). $K_{SW,k}$ from Jonker and Koelmans (2002), was used in AURAMS-PAH. However, using the 2002 measurement data and their average m_K , we have determined new $K_{SW,k}$ values based on ambient observations that improve the DE particulate fraction representation (see Section B.2 in the supplemental material for this process, and Table 1 for the original and new values). The purple boxes in Fig. 2 represent the results of the AURAMS-PAH partitioning module, making use of our new $K_{SW,k}$ values instead of the originals.

While the adjusted K_{SW} values in Table 1 are significantly different from those in the original model (based on Jonker and Koelmans (2002) as adjusted by the relative contributions of PM mass to the domain total in the inventory of Galarneau et al. (2007)), particularly for lower molecular-weight species, they fall within the range of values found in the literature (e.g., Dachs and Eisenreich, 2000; Bucheli and Gustafsson, 2000; Jonker and Koelmans, 2002; Xu et al., 2012-). The large correction and large range of published $K_{SW,k}$ values (which lack a temperature dependence) lends further support for the need to measure $K_{SA,k}$ directly on soots of atmospheric relevance so that they need not be estimated based on the highly uncertain/variable $K_{SW,k}$.

195 **2.2.2 Emissions**

Chemical (non-PAH) emissions in GEM-MACH make use of data from the U.S. Environmental Protection Agency (EPA)'s 2011 National Emissions Inventory (NEI), and Canada's 2010 Air Pollutant Emission Inventory (APEI), these being the closest available inventory years to the year in which our simulations takes place (2009). PAH model emissions were created with the SMOKE emissions
200 processing system (Sparse Matrix Operator Kernel Emissions, <https://www.cmascenter.org/smoke/>), which utilized PAH-to-TOG (total organic gases) emission factors, that were originally compiled for AURAMS-PAH by Galarneau et al. (2007, 2014). Below we outline the further modifications and updates that we made to this existing emissions database, in order to generate updated PAH emissions for modeling.

205 *PAH stationary emissions*

Most of the PAH emission factors (EFs) used for the 2002 AURAMS-PAH model were compiled from the U.S. EPA's Locating and Estimating Series (U.S. EPA, 1998), AP-42 document (US EPA, 1995), and the 1999 National Emissions Inventory (NEI99), (Galarneau et al., 2007). PAH EFs for stationary sources that were published between 1999 and the present are not substantially different
210 from those already being used in SMOKE. For example, recent literature on agricultural burning (e.g., Dhammapala et al., 2007; Hall et al., 2012) reported EFs that were close (within a factor of two) to those already in the inventory. Only emissions from iron and steel production were updated to those in Odabasi et al. (2009) for electric arc furnaces, as the values used in Galarneau et al. (2007) were derived from literature published before 1990, and were 1-2 orders of magnitude larger (hence
215 likely represented outdated (or absent) pollution control equipment).

PAH mobile emissions

On-road mobile PAH emission factors in AURAMS-PAH were taken from NEI99 (Galarneau et al., 2007). The mobile emissions in this inventory may no longer be relevant as the values compiled were from an older (1990s) vehicle fleet. Therefore, we employed updated EFs for more current on-road
220 mobile emissions in Canada and the U.S. for 2009 modelling. Also, some off-road emissions, such as emissions from helicopter and marine (large ships) were not considered before, and were added to the inventory (from Chen et al. (2006) and Agrawal et al. (2008), respectively) in this study.

MOVES 2014, the latest version of the U.S. EPA's motor vehicle emissions simulator (www.epa.gov/moves) contains a more recent standard set of mobile EFs, separated into one set of factors for gasoline vehicles, based on one large, 2008 study of vehicles in the U.S. (Kishan et al., 2008), and one set of
225 factors for diesel vehicles, based on another large study in the U.S. (Khalek et al., 2009). In order to investigate whether these U.S. values would be representative of conditions in Canada and whether ~~only have those~~ having only these two fuel-type categories ~~are is~~ adequate, when this would neglect studies that have reported different EFs for several different vehicle/fuel categories (e.g., cars, trucks, buses, motorcycles; light- or heavy-duty; gasoline or diesel), we compiled and researched
230 PAH-to-TOG emission factors for these classes of mobile sources from over 30 recent (1999 to

present) publications, as well as from the U.S. EPA's SPECIATE v4.4 database (containing data from 1990 to 2012: <https://www.epa.gov/air-emissions-modeling/speciate-version-45-through-40>). Please refer to Section C in the supplemental material for this mobile emission factor analysis. In this analysis, we found that the MOVES2014 EFs provided the best results in the model, thus they were selected for use in our simulations.

2.2.3 On-particle BaP-O₃ reaction

The only PAH oxidation reactions included in AURAMS-PAH were temperature-independent OH reactions with each gas-phase PAH species (Galarneau et al., 2014), which were also added to the GEM-MACH-PAH model. Temperature-dependent OH reaction rates were not pursued because Brubaker and Hites (1998) determined that only k_{OH} for fluoranthene has a slight temperature dependence, but dependence was smaller than their error levels. Also, Friedman and Selin (2012) performed a phenanthrene sensitivity study with their model, and determined that including temperature dependence in k_{OH} did not affect their mean non-urban mid-latitude concentrations.

The AURAMS-PAH model overestimated BaP concentrations compared to measurements (Galarneau et al., 2014). This could be due to two O₃-related factors: (1) Particulate BaP measurements are known to be affected by on-filter O₃ degradation, causing measured particulate BaP measurements to be biased low (Menichini, 2009); (2) Heterogeneous BaP degradation by O₃ in ambient air (Keyte et al., 2013) was not simulated in AURAMS-PAH, thereby biasing modelled concentrations high. We therefore added a particle-phase BaP-O₃ reaction in GEM-MACH-PAH to account for the latter atmospheric process as described next. For the former, on-filter O₃ reaction, we have attempted to correct the measurements as described in Section 3.

In GEM-MACH-PAH we used the Kwamena scheme (Kwamena et al., 2004) for the atmospheric on-particle BaP-O₃ reaction, as this scheme produced the best results in Friedman and Selin (2012)'s global model, and according to our sensitivity calculations, other schemes either overestimate (e.g., Pöschl et al., 2001) or underestimate (e.g., Kahan et al., 2006) the amount of BaP destroyed by this reaction. The Kwamena scheme was used because it produced BaP loss consistent with measurement studies (Ringuet et al., 2012; Jariyasopit et al., 2014; Liu et al., 2014). The reaction rate, k , is expressed as follows:

$$k = \frac{k_{max} K_{O_3} [O_3]}{1 + K_{O_3} [O_3]}, \quad (3)$$

where, $k_{max} = 0.060 \pm 0.018 \text{ s}^{-1}$ and $K_{O_3} = (2.8 \pm 1.4) \times 10^{-15} \text{ cm}^3$. The Kwamena scheme is expressed in the model as:

$$[BaP]_{\text{reduced } t+\delta t} = [BaP]_{it} e^{-k\delta t}, \quad (4)$$

where δt is the model time step in seconds. This formulation does not depend on the particle size, but rather on the overall bulk particulate concentration, and the concentration of O₃.

2.2.4 Dry and wet removal of PAHs and benzene

Gas-phase dry deposition follows a multiple resistance approach and single-layer “big leaf” approach (Wesely, 1989; Zhang et al., 2002; Makar et al., 2018) with a temperature dependency for Henry’s Law constants and for water solubility (Sander, 1999; Ma et al., 2010). Dry deposition of benzene and PAHs is also output by the model; however, measurements of the deposition flux of these species were unavailable during the study period.

PAH particle-phase dry deposition is treated following (Zhang et al., 2001), resulting in size-dependent particle deposition velocities.

Gas-phase benzene and PAHs undergo cloud and rain scavenging via Henry’s law. Henry’s law partition coefficients (K_{AW}) for the seven PAHs ~~is a linear relationship with~~ are proportional to inverse temperature. The mass of benzene and PAHs in the gas-phase (as opposed to the aqueous phase in cloud droplets and raindrops) is derived from:

$$K_{AW,k} = \frac{m_{gas}/V_{air}}{m_{aq}/V_{h2o}} = m_{AW}/T + b_{AW}, \quad (5)$$

and solving for m_{gas} :

$$m_{gas} = \frac{\frac{V_{air}}{V_{h2o}} K_{AW,k} m_i^{gas}}{\left(1 + \frac{V_{air}}{V_{h2o}} K_{AW,k}\right)}, \quad (6)$$

where, m_i^{gas} is the initial mass of the gas-phase PAH before the Henry’s law partitioning, and the remaining PAH mass is scavenged to the liquid rain or cloud phases (m_{aq}). Once the gaseous PAHs are scavenged to cloud-water (which is re-calculated at each time step), they are subject to rain-out (cloud-to-rain conversion) process. At the end of each chemistry step, the fraction of the dissolved tracers not removed by the rain-out process will be returned to gas phase. For the fraction that do go into rain water, there is a parametrization in the rain scavenging code for re-evaporation of the rain before it reaches the ground. That fraction of PAHs would also return to the atmosphere. The remaining fraction that is not re-evaporated is counted towards the wet deposition in the model, which is irreversible (i.e., no re-emission of PAHs from the ground after they are deposited).

Note that Okochi et al. (2004) reported that assuming Henry’s Law equilibrium for benzene ~~underpredicts~~ under-predicts the extent of wet-deposition. In the absence of a suitable alternative parameterization, we used Henry’s Law partitioning and therefore obtained a conservative estimate of wet deposition for benzene. Note that benzene wet deposition is not evaluated in this paper as no measurements are available.

Where temperatures are $<0^{\circ}\text{C}$ below-cloud, or $<-15^{\circ}\text{C}$ in-cloud, scavenging of gas-phase benzene and PAHs by snow and cloud-ice is done via surface adsorption following the formulation used in Franz and Eisenreich (1998), which was also used by Wania et al. (1999); Lei and Wania (2004) and Friedman and Selin (2012):

$$W_g = K_{ia,k}(SA)\rho, \quad (7)$$

300 where W_g is the gas scavenging ratio (equal to the concentration of PAH in snow over the concentration of PAH in air - both in moles/m³), $K_{ia,k}$ is the interfacial adsorption coefficient (equal to the mass adsorbed per surface area of snow to the atmospheric vapor phase concentration - both in ng/m³), SA is the specific surface area of the snow crystal, for which we use a constant 1 m²/g based on literature values for fresh snow precipitation, which are highly variable, and for which no clear
305 relationship with temperature or wind speed has been found (Hoff et al., 1998; Hanot and Dominé, 1999; Domine et al., 2007; Hachikubo et al., 2014), ρ is the density of ice (0.917 g/cm³), and $K_{ia,k}$ is calculated from the following (Franz and Eisenreich, 1998):

$$\log(K_{ia,k}) = -1.2 \log p_{L,k}^{\circ} - 5.82, \quad (8)$$

Eq. (7) is used to determine the fraction of PAH mass in the gas and snow/cloud-ice phases.

310 Particle-phase PAHs are treated as passive tracers that undergo wet removal along with the modelled aerosol particles (Gong et al., 2006; Wang et al., 2010). The cloud and precipitation processes above are applied sequentially in the model using operator splitting, and the amount of PAH deposited from wet deposition is output by the model. Table D.1 in the supplemental material provides all of the constants used in the model.

315 2.3 Model setup for two 3-month simulations

GEM-MACH-PAH, rev2488, was run from 8 May to 13 August 2009 and from 18 October 2009 to 5 January 2010, where the first week from each period is treated as a spin-up period (for chemical concentrations to stabilize and for the initial condition effects to be negligible: e.g., Samaali et al., 2009), and were not used in our evaluation. The time periods were chosen to coincide with as many
320 PAH concentration and deposition measurements as possible, while limiting simulation duration to reduce computational expenses.

The chemical initial and boundary conditions for the outer nest North American domain were taken from a one-year MOZART simulation for all pollutants (Emmons et al., 2010; Pendlebury et al., 2017), except for benzene and PAHs. Initial and boundary conditions for PAHs and benzene were set
325 to zero for the North American domain as its boundaries are generally away from PAH and benzene sources (e.g., over the ocean), and are also very distant from the Pan Am domain. The simulations for the nested Pan Am region were run using the chemical initial and boundary conditions from the 10-km North American model run.

The model simulation was carried out in sequence of 27-hour staggered simulations starting at 00
330 UTC, in order to reinitialize meteorology with the analysis at 10-km resolution. The first three hours in the 2.5-km domain were discarded as spin-up to reduce the dependency on the 10-km resolution meteorological initial conditions. Each 27-hour simulation used the chemical concentrations from the end of the previous simulation as initial conditions for the next 27 hours, and this sequence continued until each 3-month period was complete.

335 GEM-MACH-PAH was run in the 2-size-bin mode to represent the PM size distribution, which
means that particles fall in either fine mode ($PM_{2.5}$ - diameter $2.5 \mu\text{m}$ or less) or coarse mode
(PM_{10} - $PM_{2.5}$ - diameter 2.5 - $10 \mu\text{m}$).

3 Measurement Description

We compare the GEM-MACH-PAH predictions to all of the benzene and PAH measurements avail-
340 able in the Pan Am domain during the two time periods in 2009. These include a high-spatial-
resolution urban measurement campaign in the Hamilton, Ontario region, as well as network mon-
itoring stations from NAPS, NATTS, and IADN. Locations of PAH and benzene measurement sta-
tions are plotted in Fig. 3 and 4a. Note that all measurement stations were *not* equipped with oxidant
removal technology; therefore, all measured PAHs, especially benzo(a)pyrene (which has the high-
345 est particulate fraction, and is the most reactive with O_3), would have had losses due to reaction
with ozone on the filters (Menichini, 2009; Liu et al., 2014), and thus would be biased low com-
pared to concentrations in ambient air. Accordingly, we have applied an O_3 correction to the BaP
measurements in this study, as the literature suggests that the BaP sampling artifact is substantial,
with around 20-72% lost on average during sampling (Menichini, 2009; Liu et al., 2014). Note that
350 our correction follows the linear method recommended by Schauer et al. (2003), which is dependent
only on O_3 concentrations. However, other studies state that the O_3 degradation of BaP is more
complex, with additional dependencies on the resident atmospheric lifetime of BaP (Goriaux et al.,
2006), and relative humidity (Pitts Jr. et al., 1986; Umwelterhebungen and Gerätesicherheit, 2002;
Menichini, 2009). However, those studies did not provide an alternative correction equation. There-
355 fore, in our results, we will present both the Schauer-corrected BaP measurements (for sites that had
 O_3 monitors nearby), as well as the original reported BaP from the measurements, given the lack of
a better correction for the sampling artifact.

3.1 Hamilton measurement campaign

Ambient measurements of $PM_{2.5}$ and 16 PAH species were collected from a dense network of mea-
360 surement sites in Hamilton, Ontario during June/July 2009 and December 2009. These measure-
ments are described in Anastasopoulos et al. (2012), where they found a high level of intra-urban
variability for the PAHs; 3-4 times more variable than $PM_{2.5}$ concentrations.

There were 43 measurement sites operating during the summer period (see Fig 4a), and 46 sites
during the winter period. All measurements are from 2-week integrated time frames (24 June to
365 8 July and 2 to 16 December) taken with URG personal pesticide samplers, which collected gas
and particle-phase PAHs less than $2.5 \mu\text{m}$ in diameter in 40 m^3 of sample air. The $PM_{2.5}$ measure-
ments were made at the same sites using a three-stage Harvard Cascade Impactor. The particle-phase
PAHs (up to $2.5 \mu\text{m}$ in diameter) were collected on a Teflon filter, gas-phase PAHs were collected

in polyurethane foam (PUF), and total (gas + particle) PAHs concentrations were reported in ng/m³
370 as determined by gas chromatography/mass selective detection. PM sample filter masses were deter-
mined by gravimetric analysis. (Anastasopoulos et al., 2012)

O₃ measurements that were used to correct the Hamilton BaP measurements came from three
monitoring sites in the Hamilton region ("Downtown", "Mountain", and "West") from the Ontario
Ministry of Environment and Climate Change (MOECC) website for historical air quality data
375 (<http://www.airqualityontario.com/history/>).

3.2 National Air Pollution Surveillance Program

NAPS is a Canadian program to provide accurate and long-term air quality data of a uniform stan-
dard across the country. NAPS is managed under a cooperative agreement between ECCC and the
provinces, territories, and some municipal governments. There are currently 286 NAPS measure-
380 ment sites in 203 communities located in every province and territory (www.ec.gc.ca/rnsps-naps/).

Under this program, PAH samples were collected over 24 hours, beginning and ending at midnight
(local), typically every 6 days, with a sample volume range of 600-800 m³ (Environment Canada,
2013). Benzene samples were collected in 6-L stainless steel canisters over 24 hours, starting at
midnight, every 3 days (Galarneau et al., 2016).

385 Within the Pan Am domain, total PAHs (gas+particle-phases combined) and benzene were mea-
sured at eight NAPS sites (listed in Table A.1 in the supplemental material; Fig. 3), and their 2009
data were downloaded from the following url: <http://maps-cartes.ec.gc.ca/rnsps-naps/data.aspx>.

For BaP measurement corrections, the NAPS network also measures hourly O₃ at four of these
eight PAH/benzene sites (Windsor, Hamilton, Simcoe, and Egbert). Two of the missing sites (Toronto
390 and Etobicoke) had nearby O₃ measurements from MOECC, but the last two (Burnt Island, and
Point Petre), which are rural sites, had no O₃ measurements nearby. Therefore, BaP could only be
corrected at six of the eight NAPS sites in the Pan Am domain.

3.3 National Air Toxics Trends Stations network

NATTS is ~~an~~ a U.S. program to monitor toxic air pollutants in accordance with the U.S. Government
395 Performance Results Act, which requires the U.S. EPA to reduce the risk of cancer and other serious
health effects associated with hazardous air pollutants (HAPS) by achieving a 75% reduction in air
toxics emissions chemicals, based on 1993 levels (U.S. EPA, 2009). Regulated under the Clean Air
Act are 188 HAPS species including benzene and the seven PAHs in this study.

Every six days, 24-hour ambient air samples are collected starting at midnight LT. Analysis of the
400 samples is done by high resolution gas chromatography/mass spectrometry (GCMS) Selective Ion
Monitoring (SIM) mode to get total (gas + particle) PAH concentrations, and benzene concentrations
(Eastern Research Group, Inc., 2009).

There are 115 NATTS sites within the model domain (Table A.1 in the supplemental material, Fig. 3), but only 21 sites measured PAHs, while 113 sites measured benzene. Those data were downloaded from the following url: www.epa.gov/ttnamti1/toxdat.html#data. Measurement methods in NATTS are very similar to those of NAPS.

Since O_3 was not measured at the NATTS sites, NATTS BaP was corrected with the nearest O_3 monitor data found at the U.S. EPA and CASTNET websites: www.epa.gov/outdoor-air-quality-data/download-daily-data and <https://java.epa.gov/castnet/reportPage.do>, respectively.

410 3.4 Integrated Atmospheric Deposition Network

IADN was mandated by the 1987 Canada-U.S. Water Quality Agreement, and was initiated in 1990 to measure atmospheric concentrations of persistent toxic pollutants in the Great Lakes basin. There are nine IADN sites total within our Pan Am model domain, and they are listed in Table A.1 of the supplemental material (see also Fig. 3). Six of the nine IADN sites report gas- and particle-phase PAH atmospheric concentrations separately (labelled "PAHs" in Table A.1), and a different set of six sites report wet deposition of PAHs (labelled "PAH wet dep" in Table A.1) using sampled precipitation concentrations (Blanchard et al., 2005). Thus, these data can be used to evaluate the model's gas-particle partitioning and deposition, respectively. Benzene was not measured by IADN, nor was C_{TSP} or PM_{10} in 2009, with the unfortunate result that K_p can not be calculated directly from the IADN measurements. O_3 was also not measured by IADN, necessitating the use of the nearest O_3 monitors in order to carry out the BaP oxidation correction. This latter step was possible only for observation stations at Cleveland and Chicago. The other four IADN air sites were rural/background locations, and did not have any O_3 measurements nearby.

PAHs were collected by high-volume sampler for periods of 24 hours beginning at 08:00 Eastern Standard Time, every 12 days. At Canadian IADN sites, glass fiber filters and PUF sorbent collected the particulate and gaseous fractions, whereas the U.S. stations collected PAHs with quartz fiber filters and XAD resin (Blanchard et al., 2005). Sample volume for the U.S. method is about 800 m^3 , but is 400 m^3 for the Canadian method to minimize breakthrough of volatile species during warm summer months (Blanchard et al., 2005).

Wet deposition of PAHs are measured with MIC-B precipitation collectors. The U.S stations used XAD-2 resin column cartridges for accumulating the organics on a 28-day cumulative basis, while the Canadian stations use a dichloromethane solvent extraction system, also on a 28-day cumulative basis. Both countries collect samples on a monthly basis. Note that one of the six wet deposition sites, St Clair, Ontario (STC), only had valid measurements during February 2009, which was not a time period simulated here. Therefore, only five IADN sites appear in our wet deposition analysis in Section 4.4 below.

Note that Point Petre and Burnt Island are NAPS stations co-located with IADN. IADN data were downloaded from the following url: <http://open.canada.ca/data/en/dataset/531d6054-4179-4883-8022-1175cdfb6911>.

4 Model Evaluation

440 In this section we evaluate GEM-MACH-PAH's performance for benzene and PAH surface concentrations, their spatial variation, gas-particle partitioning, and wet deposition. We also assess the sensitivity of the model output to PAH emission factors for mobile sources.

4.1 PAH concentrations in the Hamilton region

GEM-MACH-PAH output for gas + fine-PM PAH were compared to measurements of same from
445 the 2009 Hamilton campaign (Anastasopoulos et al., 2012). Fig 4a shows a map of measured and modelled fluoranthene concentrations (14-day average) in the summer time period, as well as their differences and ratios. Here we see that GEM-MACH-PAH has captured intra-city variability, and that the differences between observations and simulated values are, at a maximum, $2.8\times$ too high. The model is biased low in the upwind/background areas of the city, and a high in the eastern areas of
450 the city (Fig. 4a), and this pattern is seen across all seven PAHs. The spatial pattern in the PAH bias is less apparent when PAH/PM_{2.5} ratios are plotted (in ng/ μ g – shown in Fig. EF.1 in the supplemental material) – removing the dependency on modelling PM correctly (since fractions of the PAH are particulate). Therefore, the spatial pattern in the PAH bias is mainly due to the pattern in the model PM bias, which is shown in Fig. EF.2 in the supplemental material.

455 When the spatial variability is represented by the standard deviation over the mean (σ/mean), the model achieves very similar spatial variability to the measurements (Fig. 4b). The scatter plot of model vs measurements for summertime fluoranthene concentrations (Fig. 5a, FLRT selected as a good example) has correlation coefficient R^2 of 0.57, and the slope of the best-fit line is ~~very close to 1.~~ 0.96. The other PAH species had similar results, where, except for FLRT, the slopes and R^2
460 values were better in the winter than in the summer.

The model bias (given as a model/measurement ratio) for all PAHs is shown as box and whiskers in Fig. 5b. Here we see that wintertime biases are smaller than those in the summertime for all PAHs except for ANTH. The four lightest PAHs (left side of Fig. 5b) have model/measurement ratios near 1 (except summertime PHEN), but the three heaviest PAHs, are biased high (except for wintertime
465 CHRY). We will see this same pattern for the model bias (small for lighter PAHs, high for BaA and BaP) in the next sections as well.

BaA and BaP are the most reactive of the heavier species, thus the lack of O₃ correction to the BaA measurements may be partially responsible for the model-measurement differences. However, mean O₃ for the three measurement stations in the Hamilton region was only 20-26 ppbv/day, thus
470 the O₃-corrected BaP was approximately 20% greater than the reported BaP concentrations. The median BaP bias was brought down to 6.2 from 7.6 in the summer, and 5.5 from 6.3 in the winter - these are shown as the purple boxes in Fig. 5b. Additional reactions with BaA and BaP, such as with NO₃, are noted in the literature (~~Keyte et al., 2013; Mu et al., 2017~~)(Keyte et al., 2013), but were not

included in GEM-MACH-PAH at this time, given larger uncertainties in those reactions. However,
475 our model biases appear to indicate that those missing reactions may need to be considered for fur-
ther model improvement. Indeed, a back-of-the-envelope calculation using a reaction rate reported
in Liu et al. (2012) implied that the lifetime of particulate BaA with respect to a heterogeneous
reaction with NO₃ (using mean night-time concentrations of NO₃ from the model) was about 1
hour. More research should be done to determine the range of uncertainty on that reaction and
480 whether it should go into PAH models. Additionally, Mu et al. (2018) suggest that the heterogeneous
BaP-O₃ reaction should be temperature-, humidity-, and organic aerosol phase state-dependent (none
of which were taken into account in the Kwamena scheme used in our work). However, it has been
shown that the Kwamena scheme and the Mu scheme produce similar results in mid-latitudes (where
our study is located) (Mu et al., 2018). Spring/summertime BaP would be minimally affected, as
485 outdoor temperatures at that time of year resemble the room temperature laboratory conditions that
the Kwamena scheme was based on. Additionally, our positive model bias would likely increase in
the fall-wintertime, when low temperatures and humidity would increase BaP lifetime in the Mu
scheme.

In order to remove the impact of the model's PM predictions on the PAH comparison, we also
490 plotted the PAH/PM_{2.5} model-over-measurement ratios (shown in Fig. EF.3). There we see all of
the ratios reduced – which improves results for the heavier PAHs, but increases the low bias for the
lighter PAHs. The reason the bias decreases for all seven PAHs is that the model PM_{2.5} is overesti-
mated by a factor of 2 in the summertime, and a factor of 1.4 in the wintertime (average across all
sites in the Hamilton region).

495 4.2 PAH and benzene concentrations from the NAPS, NATTS, and IADN networks

Modelled 24-hour-average total (gas+particle) PAH and benzene concentrations can be evaluated
with every-6th-day measurements from the NATTS, NAPS, and IADN surface measurement net-
works, which sample much of the model domain well (Fig. 3). As with the Hamilton evaluation, the
model has very good agreement for seasonal averages at the monitoring network sites for benzene,
500 phenanthrene, anthracene, fluoranthene, and pyrene, which all have model/measurement ratios (red
and blue boxes) close to 1, and their concentrations (green and orange boxes) overlapping in Fig. 6a.
The 24-hour average model (daily) and measurements (every 6th day) have been averaged over each
of the 3-month time periods. BaA and CHRY are within a factor of 5 of the measurements in the
summertime, but worse in the wintertime. BaP is overestimated by the model in both summer and
505 winter by about a factor of 10, although the measurement-corrected BaP has a slightly reduced bias.
We have not shown the O₃-corrected BaP measurements in the plots because the changes are small,
similar to the Hamilton plot (Fig. 5a).

When the model biases are examined more closely we find a few patterns to determine the cause(s). The following list summarizes some observations from our evaluation of each model-measurement pair (24-hour averages, not seasonal averages):

510 – **By site - overestimations:** all PAHs are significantly overestimated at the Kennedy Township, Pennsylvania (NW of Pittsburgh) NATTS site (Fig 6b). There appears to be a major emissions point source near that station that is emitting too much PAH in our model compared to reality. There are in fact hundreds of point sources in the emissions inventory that are within 20 km
515 of Kennedy Township, but one in particular emits a relatively large amount of VOCs, and is associated with the “Secondary Metal Production; Aluminum; Raw Material Charging” source category, which has very large PAH-to-TOG EFs in Galarneau et al. (2007) because aluminum smelter emissions are largely particulate, so expressing EFs as a large fraction of TOG was somewhat artificial. However, our results indicate that the PAH EFs for that PAH speciation
520 profile (1036b) should be reduced substantially compared to Galarneau et al. (2007). In order to ensure that this facility did not begin operation *after* 2009 - which is a risk when using a 2011 inventory to model the year 2009 and would result in a large overestimation as well, we have further verified that the facility existed and was emitting similar VOC amounts in the NEI2008 inventory as well.

525 PHEN (Fig. 7c) and ANTH (not shown) are also greatly overestimated in New York City, however, none of the other PAHs are biased particularly high there. However, we note that the measurements for New York City appear erroneously low, as the reported PHEN concentrations there are around the same magnitude as those in Underhill, VT (Fig. 7c), which is a background site, near a national park.

530 Most PAHs are also overestimated at the Gary, Indiana site (Fig 6b)), which may also have a nearby major point emissions source that is too high compared to reality. The heavier PAHs (BaA, CHRY, and BaP) are also overestimated at the Toronto Gage Institute NAPS site, but are only slightly higher there than the average model/measurement ratio for those species (not shown).

535 – **By site - underestimates:** all PAHs are markedly underestimated at the Liberty, Pennsylvania site (e.g., Figs. 6b, and 7c), implying that there may be industry emissions of PAHs here that are missing, mis-allocated, or misplaced in the NEI2011 inventory, or an improper PAH speciation profile applied. Similarly, PAHs are underestimated in Buffalo, New York and at Franklin Furnace, Ohio. As these are not large cities, there may be industrial emissions that
540 are not reported in the NEI2011 emissions inventory (or are reported at too low levels) - perhaps because those facilities shut down in 2010 (or installed emission control technology), which would mean the problem simply lies in using a 2011 inventory to model 2009.

545 However, when we further investigated stacks near Buffalo, NY, we found that the facility with the largest CO and VOC emissions had zero PAH emissions. This facility is associated with the generic process of "Primary metal production; By-product Coke Manufacturing", which did not have an associated PAH-to-TOG profile in Galarneau et al. (2007), because the source category codes that follow it (such as flushing liquor circulation tank, excess-ammonia liquor tank, tar dehydrator, tar interceding sump, tar storage, etc) are not expected to emit PAHs to air. However, our results imply that the PAH speciation profile for "By Product Coke Oven Stack Gas" (0011b) would have been more appropriate for this facility and its use might eliminate the model bias near Buffalo in future studies.

555 – **By month:** All PAH species have lower mod/meas ratios in the summer than in the winter (shown in Fig. 6a by season for all PAHs and in Fig. 6c for FLRT by month) – implying that modelled hydroxyl radical (OH) and/or PM biases (which have strong seasonal cycles) are impacting modelled PAHs. For example, if model OH is too high in the summer, or too low in the winter, this would cause the U-shaped pattern that we see when plotting model/measurement ratio vs. month (Fig. 6c) and it would be particularly pronounced for the lighter, gas-phase PAHs, which it is. Another possibility is seasonal bias in the representation of atmospheric vertical stability: if the modeled stability is too low in the summer and too high in the winter, then winter emissions will tend to be trapped in inversions more than observed, and summer emissions will be diluted by excessive vertical mixing. However, evidence in Makar et al. (2010) and Stroud et al. (2012) suggest that model stability is too high (not too low) for the summer time period in those studies.

565 – **By season:** For the four lightest PAHs, the model/measurement ratios are <1 in the summer, and >1 in the winter (Fig. 6a). As mentioned above, this is likely due to modelled OH being too high in the summer and too low in the winter. BaA and CHRY follow a similar seasonal difference but do not straddle the ratio=1 mark.

570 BaP, on the other hand has a model/measurement ratio that is slightly higher in the summer than in the winter (Fig. 6a). For BaP, the OH bias could be offset by an opposite O₃ bias in the model. Indeed, it has been shown (Makar et al., 2010; Stroud et al., 2012) that the processes in GEM-MACH cause urban, surface O₃ to be too low in the summertime, due to insufficient vertical mixing and excessive titration from NO_x, and surface PM tends to be too high in the wintertime due to overestimation of wintertime atmospheric stability (e.g., lack of an urban heat island parameterization in the driving meteorology). These factors, together with the BaP measurement bias due to on-filter reaction with O₃, may explain the high model BaP bias.

Thus, the generally high bias of modelled BaP may to be due to additive OH, O₃ and PM model biases (plus the missing O₃ denuder technology in the measurements), impacting BaP more than the

other species because BaP has the highest O₃ reactivity, and the highest particulate fraction of the seven PAHs examined here.

580 When the five measurement sites mentioned in “By Site”, above (Kennedy Township, PA; Gary, IN; Liberty PA; Buffalo, NY; Franklin Furnace, OH) are removed from the NATTS analysis (because errors in their nearby emissions were identified), model-measurement correlation (R) and slopes improve. For example, the model vs. measurement best-fit-line slope for PHEN doubles from 0.3 to 0.6 when those sites are removed, and its R increases from 0.16 to 0.32. The slopes and R values
585 of the four heaviest PAHs all move from *negative* to positive. PYR has the largest improvement, going from slope=-0.049 and R=-0.028 to slope=0.26 and R=0.35. To the extent that the model prediction errors at the other sites may reflect emissions inaccuracy, having an accurate major point emission inventory for the time period modelled, along with proper PAH speciation profiles are extremely important requirements for modelling PAHs well at high resolution. The cases with large
590 discrepancies mentioned above highlight the need to be as specific as possible when assigning source category codes to facility processes (which is difficult given that there are tens of thousands of point sources in the inventories).

In addition to emissions errors (for which fine temporal information is lacking), transport errors can cause poorer agreement between model and measurements at high spatial- and time-scales. For
595 example, PAH concentrations have sharp gradients and large dynamic range. Thus, if plume transport is off by a few kilometers, the result will be a very large difference between modelled and measured concentrations downwind of point sources.

That said, when **using doing** a paired t-test on all data in the model domain to examine whether the summertime and wintertime modelled averages are the same as the measured averages, we found that
600 the model was indistinguishable from the measurements for all PAH species ($t < 1$ and $p > 0.05$), *except* for wintertime BaP (which has $t > 1$, and $p < 0.05$, however, even with the O₃-corrected measurements). At finer time scales (e.g., *daily* model-measurement pairs) only modelled ANTH was statistically indistinguishable from measurements. Therefore, GEM-MACH-PAH can accurately model benzene and PAHs seasonally, but not daily.

605 4.2.1 Sensitivity of model to mobile emission factors

As discussed in the previous section, ensuring the accuracy of major point source emissions is important for model-measurement agreement near industrial locations. However, those major point source emissions tend to be located far from large population centres where human exposure is concentrated. In our inventory, mobile emissions make up 44%, 45%, 19%, 32%, 14%, 21%, and
610 30% of total PAH emissions, for PHEN, ANTH, FLRT, PYR, BaA, CHRY, and BaP, respectively, in our continental model domain, and studies have shown that the bulk of emissions within population centers-centres is likely to originate from on-road mobile sector emissions (Dunbar et al., 2001; Pachón et al., 2013; Kuoppamäki et al., 2014; Miao et al., 2015). Thus, in order to accurately model

ambient PAHs in urban centres, the uncertainty in emission factors from on-road vehicles may play
615 a more significant role than major point sources.

We have thus carried out 2-week sensitivity simulations (9-24 May and 18 Oct-2 Nov 2009)
wherein the mobile emissions of PAHs were scaled by factors of 0.5 ~~and 1.5~~ (halved) and 2.0 (doubled).
This is approximately equivalent to the 25th and 75th percentiles in the range of emission factors
found in the recent literature.

620 In Fig. 7, we show the surface PHEN time series from the measurements, base model run, and
0.5 and ~~1.5~~ 2.0 scaled model runs at the IADN, NAPS, and NATTS sites. It is clear that – while
a relatively small PAH source overall – changes to mobile emissions makes a large change in am-
bient PAH concentrations at certain urban locations, such as Philadelphia, PA, New York, NY, and
Burlington, Etobicoke, and Windsor, ON.

625 On average, there is about a 20-30% increase in PAH concentrations when mobile emissions are
~~increased by 50%~~ doubled, and a 5-10% decrease in PAH concentrations when mobile emissions are
~~decreased by 50%~~ halved (Fig. 8, PHEN and BaP shown as examples) – with a larger sensitivity
in the summer than the winter, and slightly larger sensitivity at NATTS (U.S.) sites than at NAPS
(Canada) sites. The predicted ambient concentrations generally follow the increase or decrease in
630 on-road mobile emissions monotonically.

4.3 Gas-particle partitioning of PAHs

The IADN network also allows us to assess model predictions of gas-particle partitioning of PAHs
at six sites (24-hour averages, every six days). Fig. 9 shows a time series of pyrene particulate
fraction (ϕ_k). Both model and measurements show higher ϕ_k in the wintertime, when there are
635 higher PM concentrations for PAH adsorption, and lower temperatures. Generally, the model seems
to underestimate ϕ_k at background sites (e.g., Burnt Island), and overestimate ϕ_k at urban sites
(e.g., Chicago), and this is true for all PAH species. Thus, in Fig. 10, which shows the results for
all PAHs at all sites, the model (green) has a larger range of ϕ_k than the measurements (orange).
This is caused by the model over- and underestimating PM concentrations at urban and rural sites,
640 respectively. For example, wind-blown dust is not included in the model; however, it is known to
be a potentially significant contributor to total PM in rural areas. Also, due to an underestimate of
vertical mixing in the model, PM tends to be biased high in urban areas, near emissions, due to a
lack of a parameterization for urban heat islands (Stroud et al., 2012).

Comparing Fig. 10 (blue boxes) to Fig. 2b (green boxes), we see significant improvement over
645 the original AURAMS-PAH partitioning due to the improved K_{SW} parameters described in Section
2.2.1.

Generally, the gas-particle partitioning scheme in the model results in model/measurement ratios
well within an order of magnitude, given by the gray lines in Fig. 10. ϕ_k for BaA and CHRY are
still underestimated, but this may be related to modelled PM errors as noted earlier. We note that

650 the addition of C_{TSP} or even PM_{10} measurements at IADN sites (which existed in 2002, but not in 2009) would allow for the calculation of measured partitioning coefficients ($\log K_p$, Eq. (2)), which could be used to validate the modelled $\log K_p$ in future work. Since K_p takes total suspended particle into account, it removes the dependency on modelled PM, thus would increase confidence that the modelled partitioning is working properly, despite model errors in PM.

655 The fact that the GEM-MACH-PAH model partitioning of BaA and CHRY (and BaP to a lesser extent) puts too much concentration in the gas phase, may help explain why these species in particular are overestimated in the model. While in the gas phase, these species are less likely to be removed from the atmosphere, so their concentrations would erroneously build up in the model.

4.4 Wet deposition of PAHs

660 When compared to the IADN one-month wet deposition measurements, the model generally overestimates wet deposition for all PAHs, as is shown in Table 2, and Fig. 11. In Fig. 11, the blue lines shows the ideal 1:1 model:measurement ratio, and most of the data lie well above these lines. By site (Fig. 11a), the modelled wet deposition was slightly better at urban locations (Toronto and Cleveland) than suburban and background sites (Burlington, Sturgeon Point, and Point Petre). By 665 month (Fig. 11b), the wet deposition from the model is best represented in June and July, whereas, wet deposition is greatly overestimated in the winter, implying that the current snow adsorption ~~parameterization~~ parametrization may be too effective at removing PAHs in the model. This may be due to our simplification of a constant snow surface area, which may be set too high, or due to inaccuracies in the modelled or measured precipitation.

670 The IADN measurements, which report the concentration of PAHs in the collected ~~rainwater~~ precipitation (in pg/L), were converted to pgPAH/m² in order to compare to the wet deposition output of the model. However, this conversion assumes that the volume of ~~rainwater-precipitation~~ reported by IADN was the *total* ~~rain-fall-precipitation~~ in the container's cross-sectional area, and in fact, it is not. The IADN wet deposition collectors are actually known to *not* sample all of the 675 ~~rainfall-precipitation~~ because the samplers aren't in the correct configuration to get an accurate ~~rainfall-precipitation~~ measurement (Dryfhout-Clark, personal communication). When we compared the actual ~~rainfall-precipitation~~ amounts (from separate meteorological rain gauge data) to IADN ~~rain~~ precipitation volumes at the Point Petre location in January 2009, we found that only 68% of the total ~~rainfall-precipitation~~ was captured by the wet deposition sampler. Therefore, if that correction factor 680 were applied to all IADN wet deposition measurements, they would increase by a factor of approximately 1.5, which would improve our comparison, but not eliminate the total bias. If IADN sites added separate, accurate rain ~~gagesgauges~~, then we could apply a "~~rainfall-precipitation~~ correction" to the IADN wet deposition measurements in a thorough, consistent way in future work.

685 Aside from the measurement bias, the modelled PAH wet deposition bias will also be dependent on the model's overall ability to predict accurate ~~rainfall~~ precipitation. We compared the modelled

daily accumulated precipitation to the precipitation measured with the accurate ~~gages~~-~~gauges~~ at Burnt Island and Point Petre, and found that, while the model's median precipitation bias was only about 0.2 mm, there was a large standard deviation, and there were some incidences where the model greatly over-predicted high ~~rain~~-~~precipitation~~ events. Those incidences would result in greater modelled wet deposition of PAHs than was measured, and because we sum over a month, there is a significant likelihood of an ~~overprediction~~-~~over-prediction~~ occurring in that long time frame. Indeed, the median ratios in Table 2, which are less sensitive to high outliers than the mean is, are substantially lower than the mean for most species.

Therefore, the model bias in wet deposition would appear to be caused by three additive factors: (1) measurements themselves having a negative bias relative to reality, due to insufficient capture of the net fluxes of precipitation, (2) modelled precipitation being biased high, and (3) a positive model bias in atmospheric PAH concentrations (which was highest for BaA and BaP in particular). Other PAH models have reported similar overestimates of PAH deposition (Matthias et al., 2009; Friedman and Selin, 2012)

The reverse reasoning can be applied, whereby we can see if high atmospheric concentrations of BaA and BaP were caused by too little wet deposition. In this case, since both of these species have correspondingly high wet deposition in the model (Fig. 11), it would appear that underestimation of wet deposition is probably not one of the causes.

Fig. 12 shows results from a sample month (June 2009) for pyrene (PYR). The spatial distribution of wet deposition was not captured, with the model predicting lower PYR deposition in Toronto and Sturgeon Point than the measurements, higher at Point Petre, and about equal at Burlington. This spatial pattern is *not* the same for all PAHs, and even differs by month for the same PAH (e.g., PYR deposition in the next month, July, is low at Point Petre, high at Toronto, and highest in Burlington). The lack of spatial or temporal pattern in the sign and/or magnitude of wet deposition biases indicates that there is no major error with the PAH scavenging scheme itself. Given that wet deposition of PAHs relies on ~~getting the correct simulation of~~ many model factors ~~correct~~(meteorology, scavenging parameters, atmospheric concentrations, etc), ~~it not surprising that the model error for PAH wet deposition is large, but it is at least promising to see that there are no particular sites where the model is consistently too high or too low, rather the errors in spatial distribution are haphazard and may be due to propagation of error, rather than any major error with the PAH scavenging scheme itself~~our work suggests that more process studies aimed at quantifying wet deposition are needed. In fact, other PAH models also overestimate PAH deposition (Matthias et al., 2009; Friedman and Selin, 2012).

5 Conclusions

720 Through this work, a high resolution chemical transport model for North American air toxics was created that allows us to see variations within a densely populated area. GEM-MACH-PAH was developed and run at 2.5-km resolution for air quality forecasting and for simulating the impacts of emissions scenarios. Relative to AURAMS-PAH, on-road mobile emissions, gas-particle partitioning, and scavenging were all improved in this study. Mobile PAH emission factors from different sources were evaluated and the MOVES 2014 factors achieved the best model results compared to those in the recent literature and in the SPECIATE database. Parameters used in the gas-particle partitioning scheme (particularly $K_{SW,k}$) were improved based on the observed relationship between $\log K_p$ and $\log p$, resulting in much better agreement between model and observations than was achieved with AURAMS-PAH. This is an important improvement because the particle/gas partitioning determines deposition and inhalation - both pathways of exposure in humans and ecosystems. Finally, we added snow scavenging, which was not a process included in AURAMS-PAH, and updated wet scavenging parameters. However, the modelled wet deposition was biased high - particularly in the wintertime - thus further improvements to these parametrizations are required if the model is to be used for deposition studies.

735 ~~Overall, Over the model domain, at seasonal time-scales, the GEM-MACH-PAH simulates simulation of benzene and six semi-volatile PAHs (PHEN, ANTH, FLRT, PYR, BaA, CHRY) at seasonal time-scales with concentrations statistically indistinguishable from observations, at 2.5-km resolution is statistically unbiased with respect to measurements ($t < 1$ and $p > 0.05$).~~ For the seventh PAH species, BaP; its summertime average is simulated to a similar level of accuracy. However, it appears the model's OH, O₃, and PM biases were additive, resulting in a wintertime average that is biased significantly high for BaP. Lack of removal of BaP via wet deposition was ruled out as a cause, but the lack of an O₃ denuder system in the measurements contributes a small amount to the model-measurement differences as well. When we corrected BaP measurements using the Schauer et al. (2003) O₃ relationship, we found reductions of about 20% in the model/measurement ratios, improving the model performance.

Our results have shown that the major point source emissions play a large role in producing accurate model results near and downwind of industrial facilities, but also that the uncertainty associated with on-road mobile emission factors plays a large role in the accuracy of simulations near and within cities. In fact, we have determined from our sensitivity test that the GEM-MACH-PAH model has a linear response to a ~~50-50 to +100%~~ variation in mobile emission factors, simulating concentrations that vary up to 30%. The spatial variability at high (2.5-km) resolution is modelled to within 50% of Hamilton, Ontario measurements, although the model places higher concentrations in polluted areas, and lower concentrations in background areas than the measurements suggest, which is correlated to the spatial distribution of the model's PM bias. With this information, we can use the high-resolution GEM-MACH-PAH model for studying vehicle emissions scenarios in order to determine intra- and

inter-city variations due to motor vehicles, with an understanding of the range of uncertainty that such a study would have.

Additional improvements to PAH modelling efforts could be achieved with general model improvements to its treatment of particulate matter (e.g., better ~~parameterizations~~parametrizations for wind-blown dust in rural areas, and better ~~parameterizations~~parametrizations for urban heat islands in urban areas). Also, additional reactions with particulate BaA and BaP in the model (e.g., with NO₃) may reduce their bias further. It would also be beneficial to any future model/measurement studies, if the PAH measurement networks utilized ozone denuder technology so that particle-phase PAHs are not underestimated in the reported observations, as well as improved and consistent ~~rain~~precipitation collection at wet deposition measurement sites. Partitioning could be better assessed if sites that measure PAH gas and particle phases separately (like IADN in this study) also measured C_{TSP} or PM₁₀. Finally, accurate emissions of PAHs at finer time and spatial resolutions could greatly improve model results.

6 Data and code availability

770 Data availability: Please refer to Section 3 for the websites where the observations can be freely downloaded.

Model code availability: GEM-MACH - Atmospheric chemistry library for the GEM numerical atmospheric model Copyright (C) 2007-2013 - Air Quality Research Division and National Prediction Operations division, Environment and Climate Change Canada. This library is free software which can be redistributed and/or modified under the terms of the GNU Lesser General Public License as published by the Free Software Foundation; either version 2.1 of the License, or any later version. The ~~CHEM-MACH-PAH (chemistry)~~ code can be downloaded from this Zenodo site: <https://zenodo.org/record/1162252#.Wm9DtK1IJZQ>, DOI:10.5281/zenodo.1162252-, and the GEM (meteorology) code is available here: <https://github.com/mfvalin?tab=repositories>. The executable for GEM-MACH-PAH is obtained by providing the chemistry library (MACH-PAH) to GEM when generating its executable.

Acknowledgements. The authors gratefully acknowledge funding from ECCC's Climate Change and Air Pollution program (CCAP, formerly called the Clean Air Regulatory Agenda). Thank you to Angelos Anastasopoulos and Amanda Wheeler for the Hamilton measurement data, to Helena Dryfhout-Clark for consultation on IADN data products, and Armaan Ladak for his help with the measurement data. Figures were made using the R-based OpenAir package (Carslaw and Ropkins, 2012; Carslaw, 2015).

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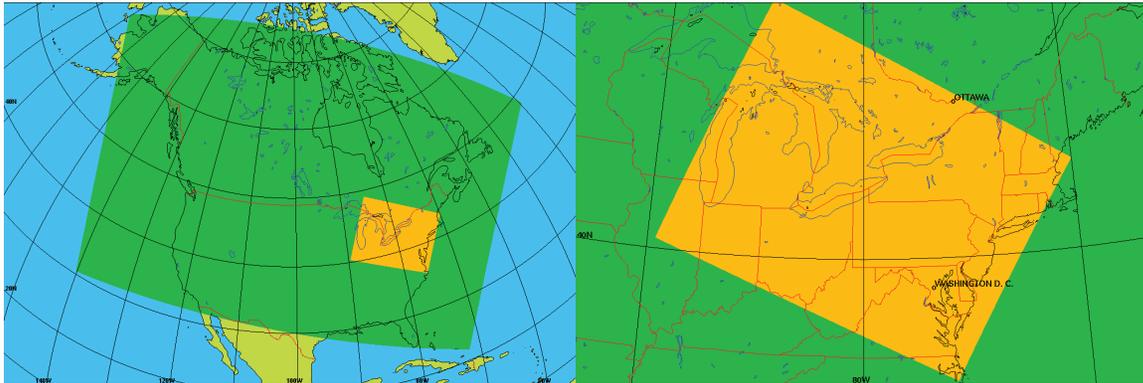


Figure 1. North American model domain with 10-km horizontal grid spacing (green), and the nested “Pan Am model” domain with 2.5-km horizontal grid spacing (orange).

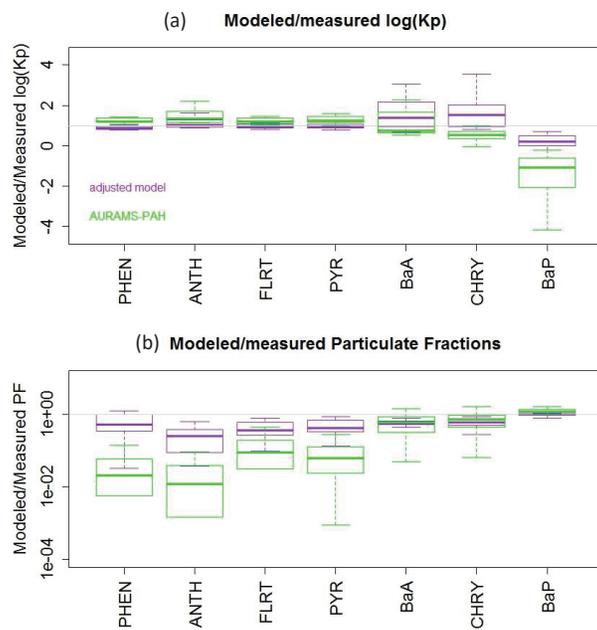


Figure 2. All-site ensemble of modeled/measured ratios of (top) log Kp, and (bottom) particulate fraction, for each PAH. Shown is the “adjusted model” (purple, from Eq. (B.2.1)), and the original AURAMS-PAH model (green). Box and Whiskers: thick line is the median, boxes extend to the 25th and 75th percentiles, and whiskers extend to the minimum and maximum.

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All PAH and benzene measurement sites

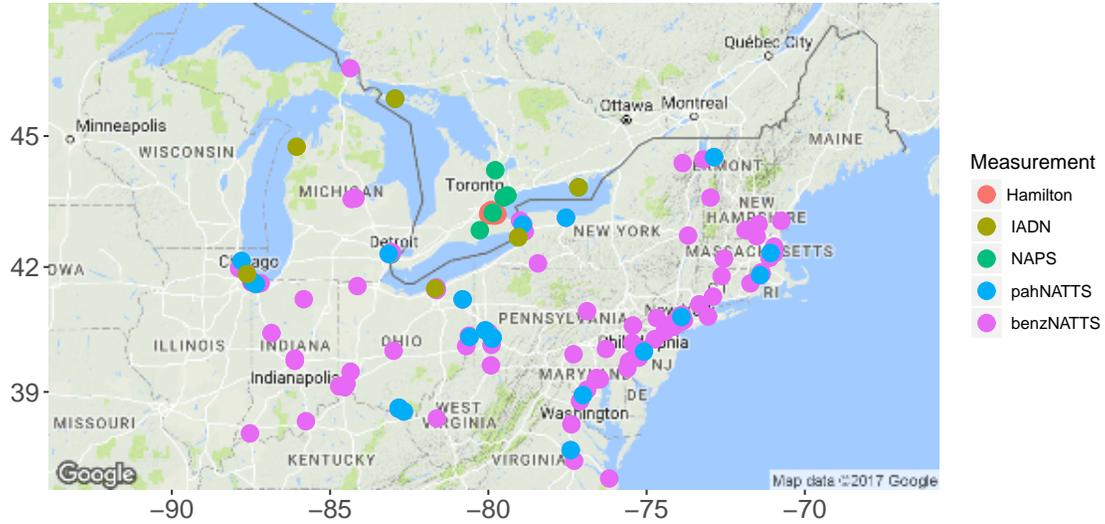


Figure 3. NAPS, NATTS, IADN and Hamilton measurement sites in the model domain.

Table 1. Original (from [Galarneau et al. \(2014\)](#), based on Jonker and Koelmans (2002)) and adjusted (based on AURAMS-PAH model-measurement analysis for North America) K_{SW} values, along with other K_{SW} cited in the literature. All values are $\times 10^5$, and unitless.

	PHEN	ANTH	FLRT	PYR	BaA	CHRY	BaP
Original Ksw	4.34 E5	1.55E6-15.5	2.24E6-22.4	1.70E6-17.0	3.74E7-374	2.82E7-282	9.59E7-959
Adjusted Ksw	3.32E7-295	4.21E7-375	8.24E7-771	9.84E7-898	1.75E8-1390	3.23E7-209	1.10E8-1040
Dachs and Eisenreich (2000)	126		631	501		3160	
Jonker and Koelmans (2002)	1.86-37.2	3.80-12.6	5.01-91.2	5.13-89.7	40.7-1820	32.4-3390	257-11700
Xu et al. (2012)	4.79-5.13	12.9-19.1	1.86-37.2	30.9-40.7	631-794	389-575	2000-3390
Bucheli and Gustafsson (2000)	2.57-58.9		2.82-16.2	27.5-155			

Table 2. Mean and median GEM-MACH-PAH model/measurement ratios for PAH wet deposition

	PHEN	ANTH	FLRT	PYR	BaA	CHRY	BaP
mean ratio	17.5	47.4	11.5	7.4	22.2	6.1	15.0
median ratio	8.1	10.5	8.5	5.4	17.9	6.4	10.3

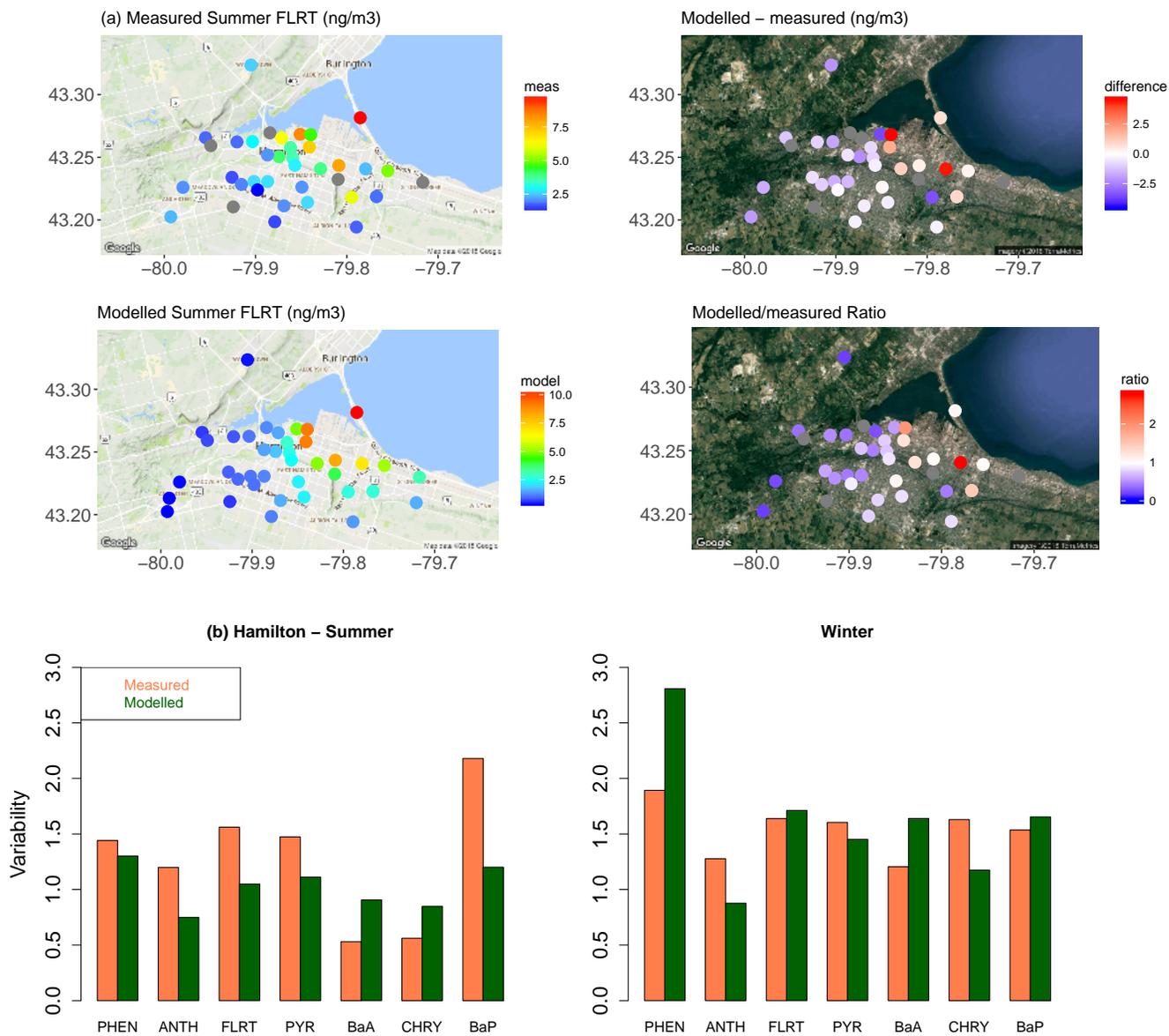
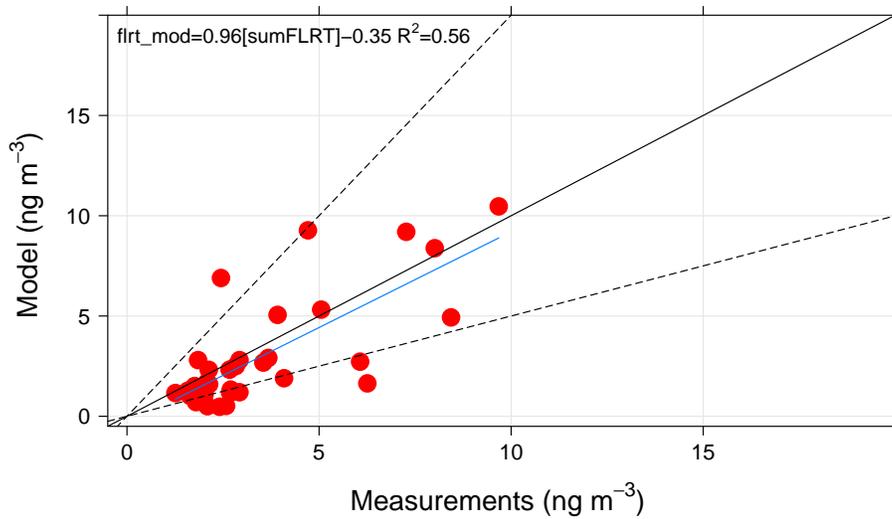


Figure 4. (a) Map of 2-week summertime average fluoranthene concentrations in Hamilton, Ontario: (left) from measurements and GEM-MACH-PAH model, and (right) their differences and ratios. [Note the grey dots are measurement sites that had missing data during this time period.](#) (b) Spatial variability in the Hamilton data in (left) summer, and (right) winter.

(a) Hamilton Summer Model vs Measurements – FLRT



(b) Model/measurement concentration ratios for Hamilton

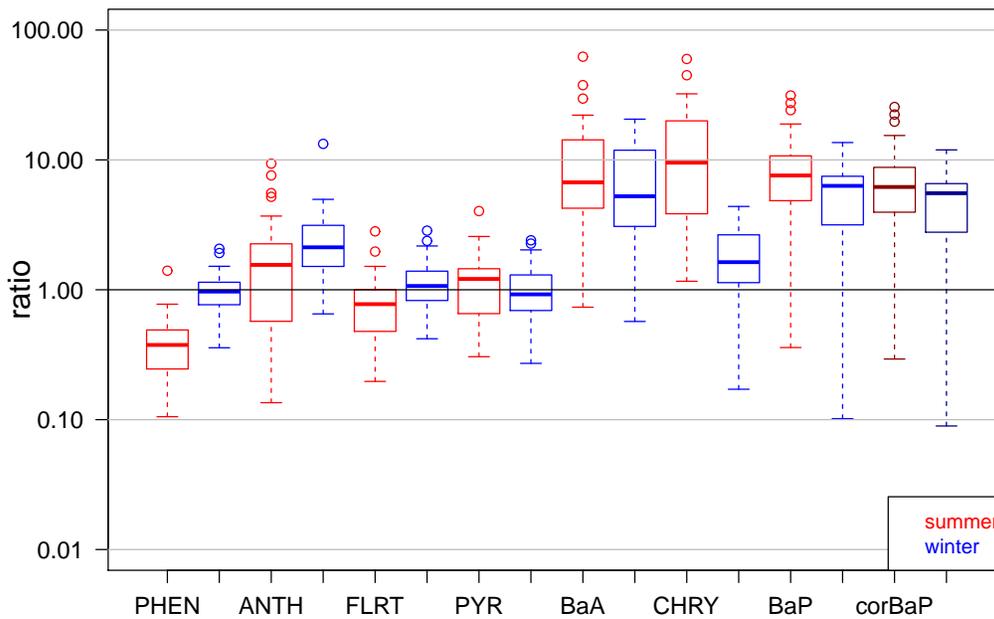


Figure 5. (a) GEM-MACH-PAH model vs. measurement scatter-plot of 2-week summertime fluoranthene concentrations at 40+ sites in Hamilton. (b) Frequency distributions of GEM-MACH-PAH model/measurement ratios of PAH concentrations for the Hamilton measurement-model pairs for all sites from both summer and winter. Purple-Darker-coloured boxes are results from O_3 -corrected BaP measurements.

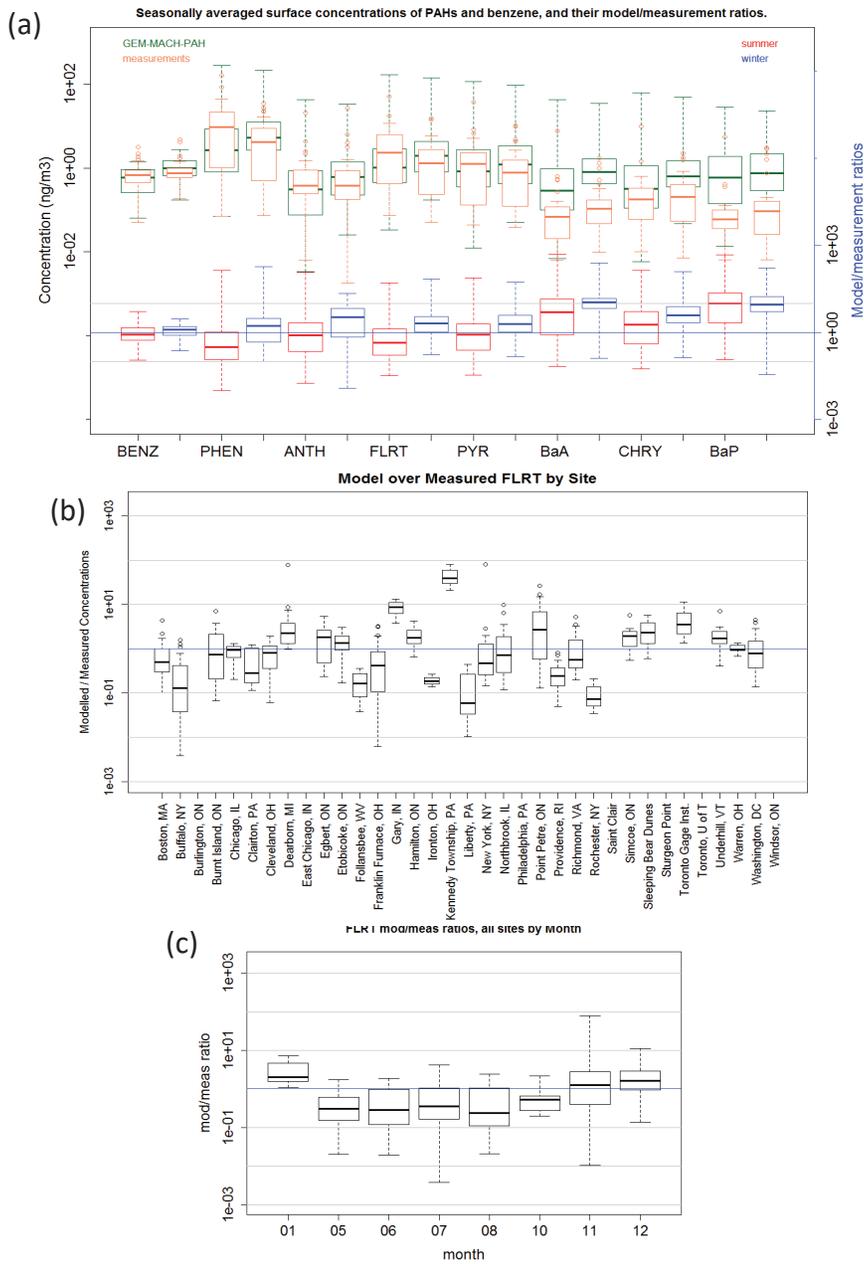


Figure 6. (a) Frequency distributions of GEM-MACH-PAH (green) and measured (orange) benzene (gas) and PAH (gas+particle) seasonal average concentrations at all IADN, NAPS, and NATTS sites. Modelled/measured concentration ratios also shown for summer (red) and winter (blue), with grey lines indicating agreement within an order of magnitude. (b) Modelled/measured concentrations for each daily model-measurement pair, separated by site (FLRT given as example), (c) same as (b) but separated by month.

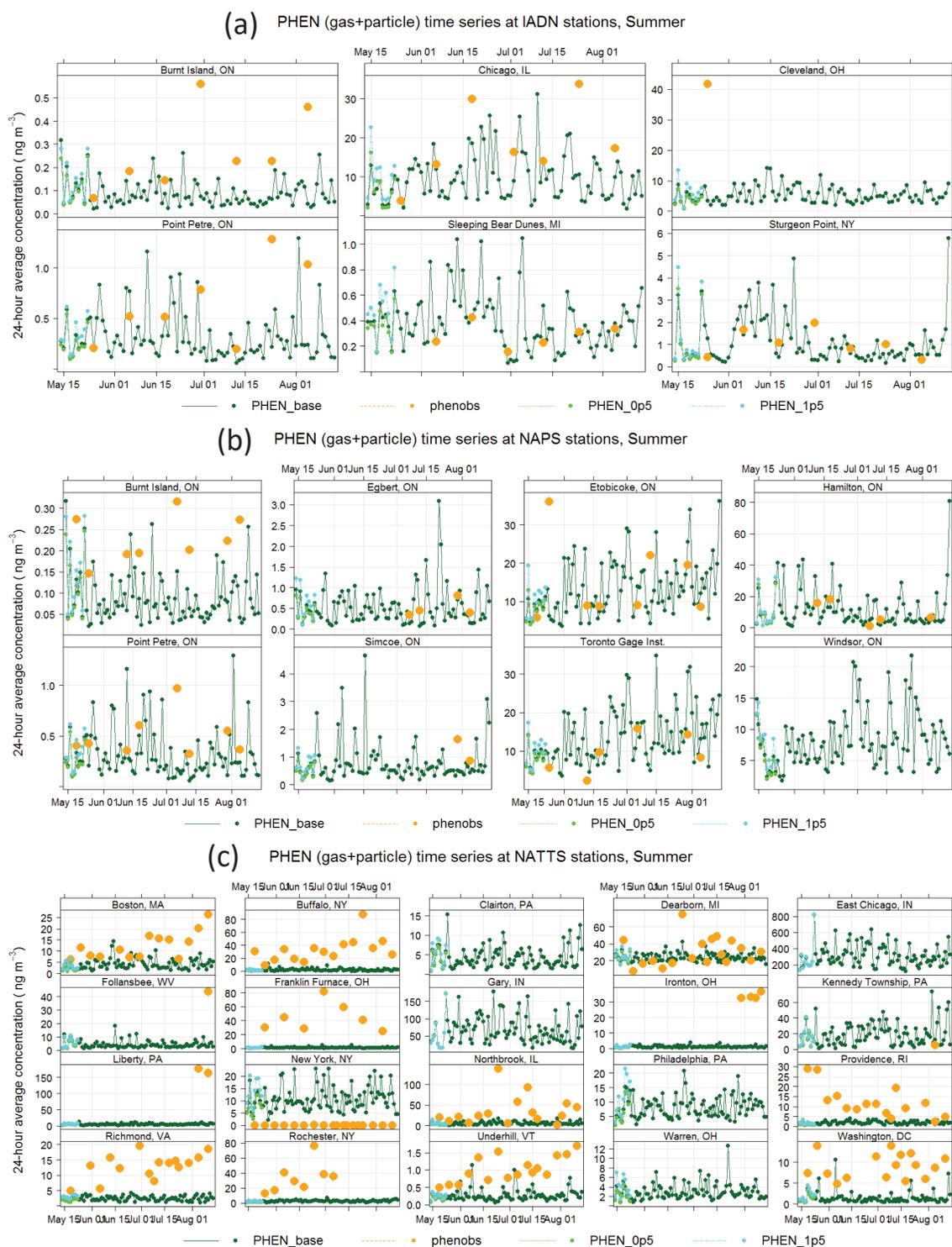


Figure 7. Phenanthrene time series for the summer 2009 period for the IADN (binational), NAPS (Canada) and NATTS (U.S.) networks. Orange=measurements, dark green=base GEM-MACH-PAH model, light green=GEM-MACH-PAH with $0.5\times$ mobile emissions, cyan=GEM-MACH-PAH with $1.52.0\times$ mobile emissions.

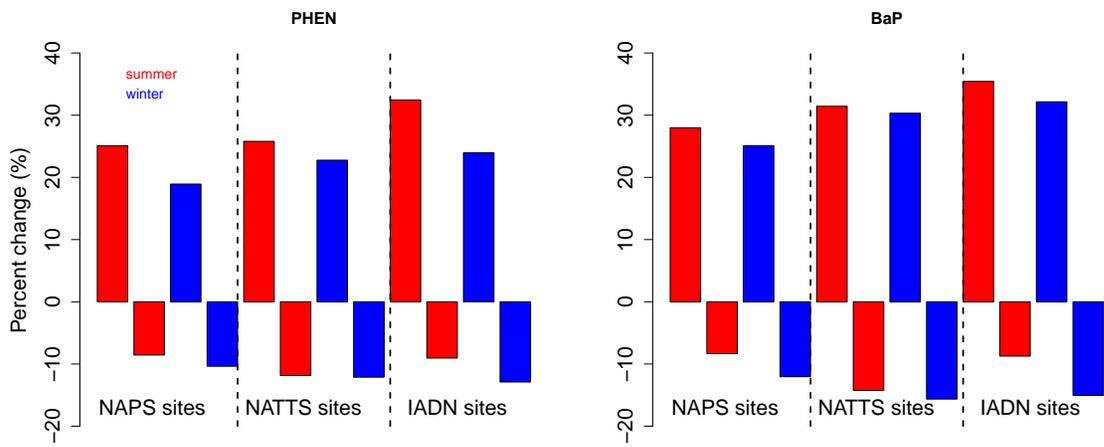


Figure 8. Average percent change in surface PHEN and BaP concentrations by season when PAH on-road mobile emissions are scaled up or down by factors of ± 5 , ± 2 , and 0.5, respectively.

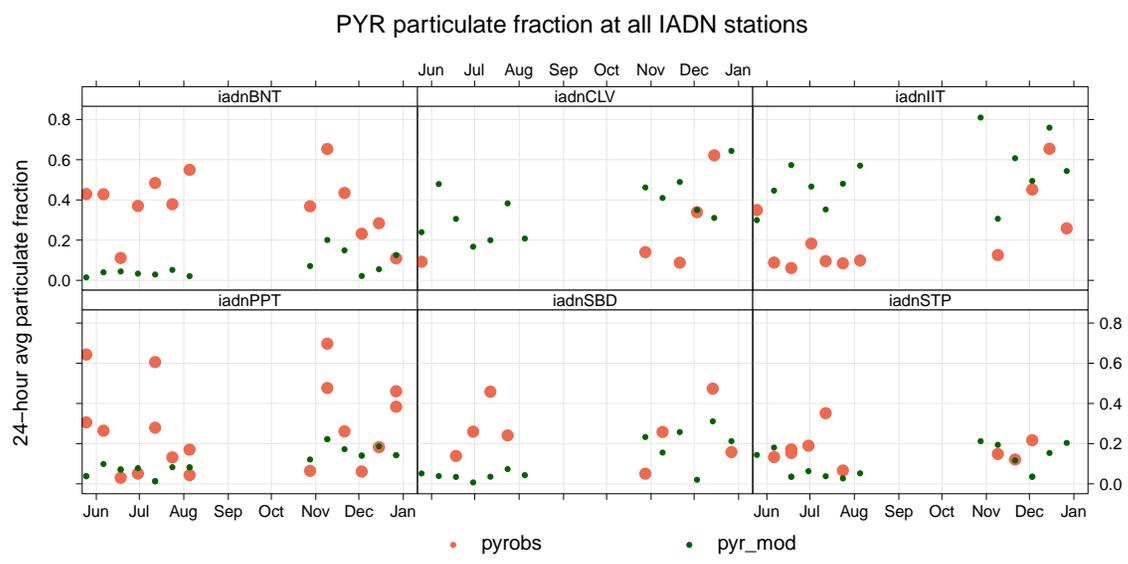


Figure 9. Time series of pyrene particulate fraction at six IADN network sites (BNT=Burnt Island, CLV=Cleveland, IIT=Chicago, PPT=Point Petre, SBD=Sleeping Bear Dunes, and STP=Sturgeon Point). GEM-MACH-PAH values are denoted by (green dots) and IADN measurements by orange dots.

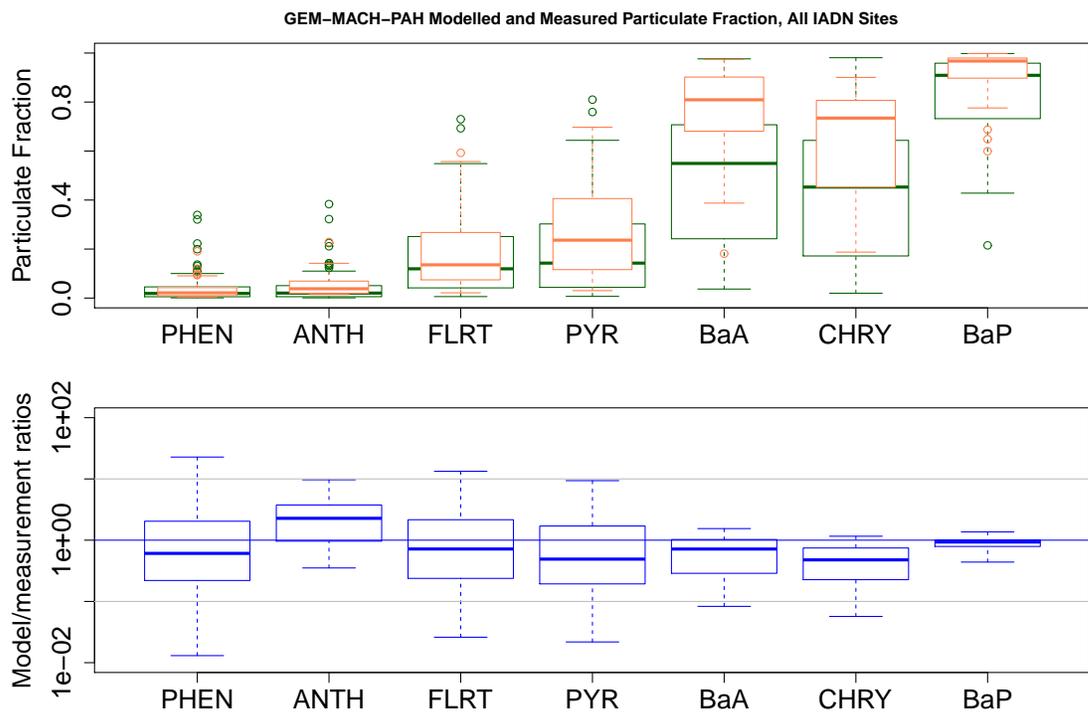


Figure 10. (a) GEM-MACH-PAH modelled (green) and measured (orange) particulate fraction (ϕ) of all PAHs at all IADN sites, and their model/measurement ratios (b) Same as (a), but for partitioning coefficient (K_p). The blue line indicates the 1-to-1 line, and the gray lines are for ratios of 10 and 0.1.

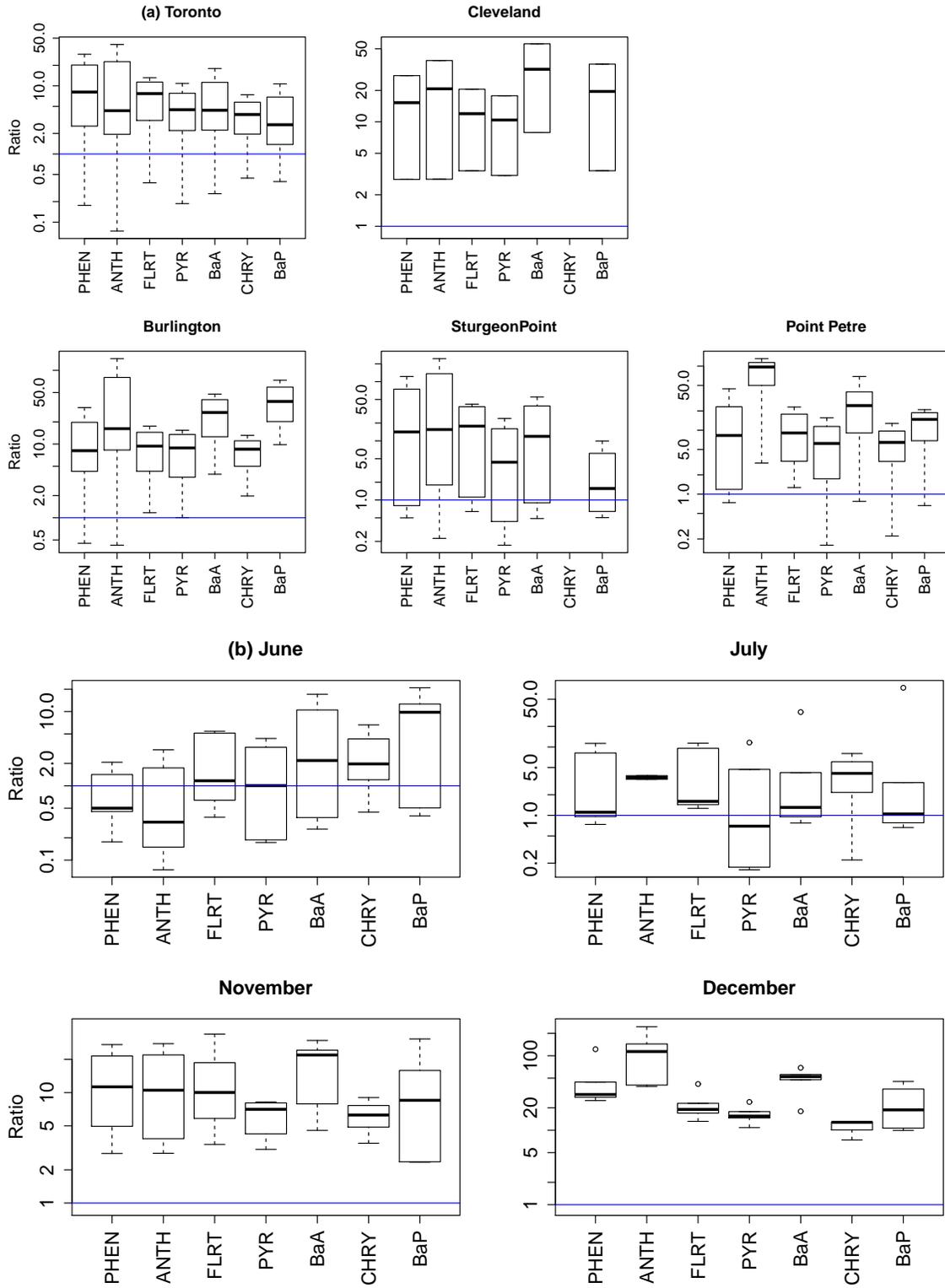


Figure 11. GEM-MACH-PAH model/measurement wet deposition ratios for all PAHs a) for five sites (all months) and b) for four months (all IADN sites).

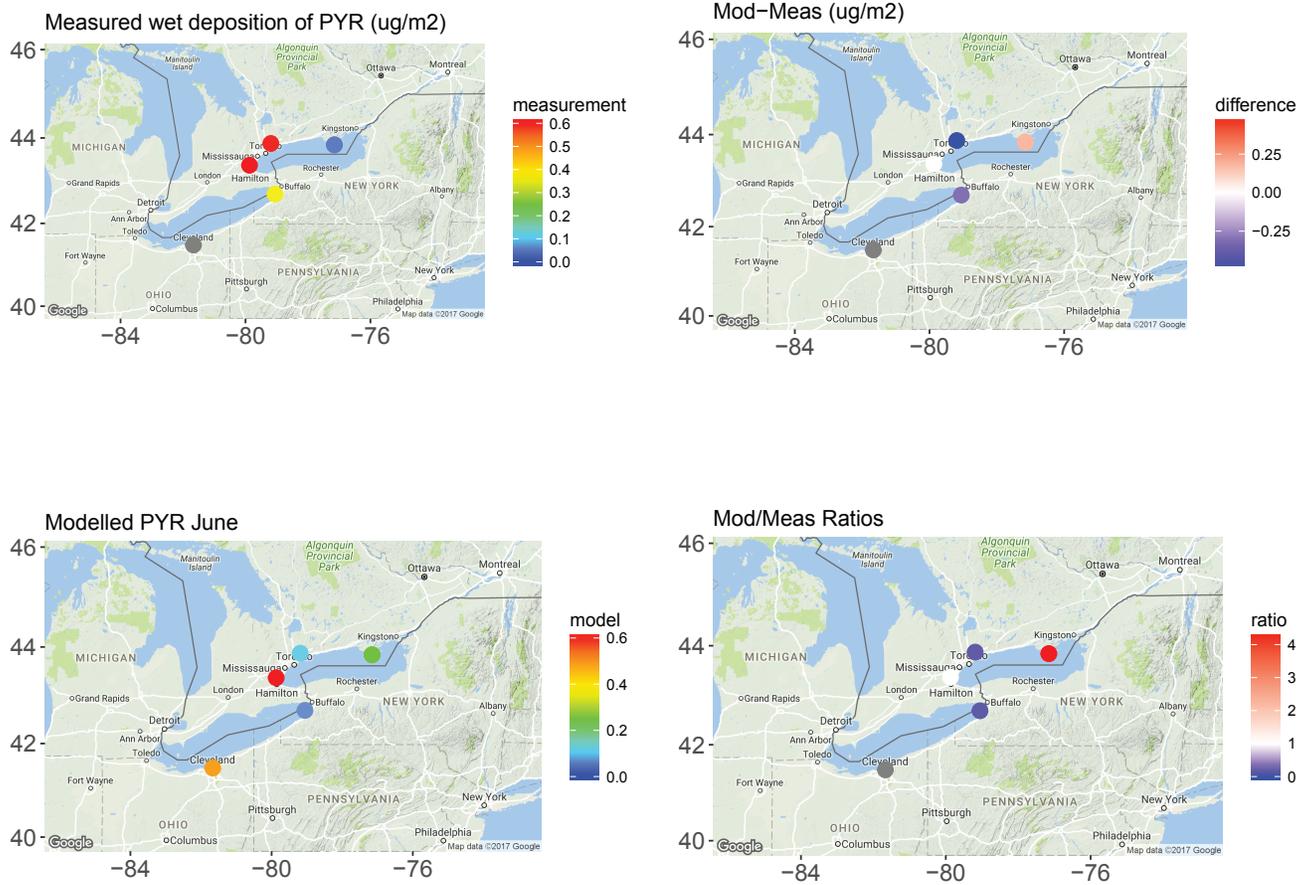


Figure 12. One-month (June 2009) wet deposition of pyrene from the (left) measurements and GEM-MACH-PAH model and (right) their differences and ratios.