Response to reviewer #1

Thank you for your comments on our manuscript. We hope to have addressed the issues you have raised satisfactorily.

Note that full bibliographic information of the references below is to be found in the manuscript.

Instant equilibration

> The presentation of the article is much improved and the authors have provided a good supplementary document with much need details. However, I do not find the explanation they have provided, of why the instantaneous equilibrium is an appropriate approximation, convincing. In my first review I showed that the adsorption and desorption terms were not as fast as the sinking rate for large particles.
>
> After carefully reading the authors response, I returned to this issue and using a single large and single small particle class for Thorium derived 1-D vertical profile solutions. I chose the numbers based on the dominate sinking flux (CaCO3 for large) and POC for small, their Table 4 and numbers from the supplementary materials. I considered the solutions in the absence of mixing and advection. I chose a very large dissociation value of 4/yr (Note: smaller values will make the problems discussed worse, not better). The profiles obtained are attached.
>
> We see that the large particles are falling so fast, they do not reach equilibrium until at least 2000 m. This means they transport far less Th than estimated by the authors. The dissolved Th and the Th on the small particles are much larger than estimated by the authors.

We agree that it is clear that non-equilibrium scavenging is different from equilibrium scavenging. However, in your calculations there is no comparison with observations, so it does not show that modelling Th-230 and Pa-231 is impossible with an equilibrium scavenging approach.

The model you used (thank you for sending this to us via the editor) is very useful for the purpose of illustration, but the ad- and desorption values are, as far as we can see, quite arbitrary (except of course in that you fixed their ratios to be consistent with our partition coefficients).

More specifically, k_{ads} is on the low side of the values estimated in the shallow waters with ^{234}Th. ^{234}Th data suggests values ranging from $1/20$ per day to $1/200$ per day. Using these values and maintaining the $k_{\mathrm{ads}}/k_{\mathrm{des}}$ ratio constant, the modelled profile tends towards the equilibrium profile (see figure below).
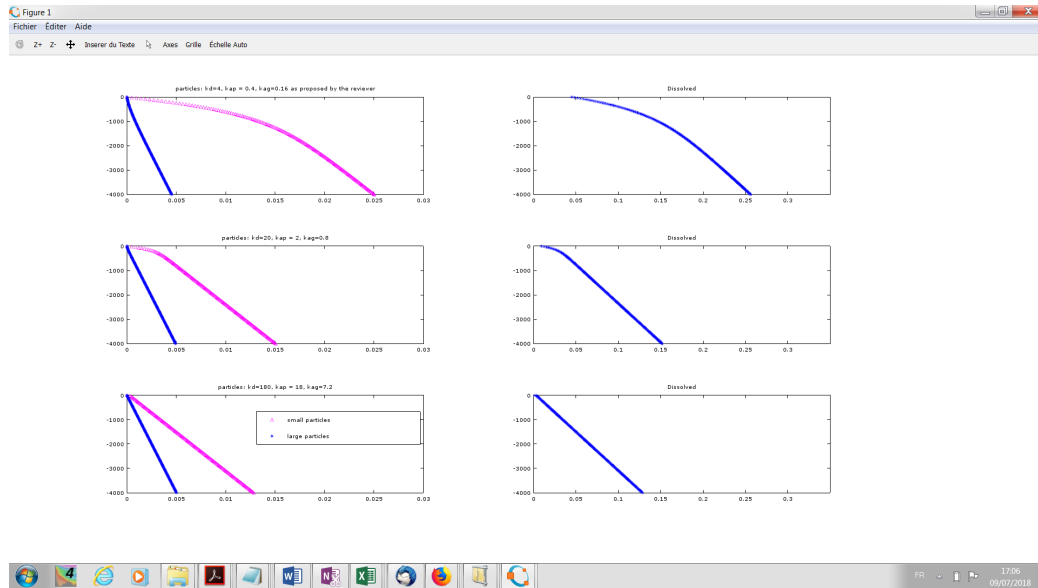

Figure 1:

The upper two panels in the figure are the graphs obtained by using the reviewer's, showing that only from 2000 m downwards the profiles are linear. The centre two panels correspond with the more realistic parameters. There the profiles are linear from 300 m downwards. The lower panels with very high k+/- in good approximation of instant equilibrium.

Furthermore, the reviewer's model (upper panels) assumes a constant particle flux and a constant particle concentration, whereas in the real ocean particle concentrations are higher in the shallow waters: they produce a higher scavenging and a lower dissolved ^{230}Th concentration.

Furthermore, equilibrium scavenging is an approach commonly used in Pa/Th modelling (Siddall et al., 2005; Dutay et al., 2009; Gu and Liu, 2017). It represent the advantage to design the model with partition coefficients that are directly constrained by observations (though admitting that consensus values are still not available, even for partition coefficients). Observations shows K values varying (decreasing) with particle flux (Roy-Barman, Chen and Wasserburg, 1996; Chase et al, 2002) that are applied in our simulation.

We have included our main arguments in the revised manuscript at the bottom of page 17.

We do realise that our approach may be imprecise, which raises the question why we do not perform simulations with slow equilibration after all. This would require more model development and simulation time, including the determination of the different k_+ and k_- values; in our opinion this belongs to a different, follow-up study. We maintain that the model as it stands is useful for the community, and, though we cannot be certain at this point, precise enough for for instance deep ocean and sediment studies.

> As expressed by the authors in their response, these ratios can be changed by changing the equilibrium constants, kappa. However, the other difference the full solution shows is that the dissolved and particulate phases are not straight lines with origins at zero depth. In particular, the dissolved phase has a distinct curvature and intercept. This curvature means, compared with the authors solution, the real difference in the upper waters (0-2000m) and the lower waters (2000-4000m) is much less than occurs for a straight line with origin at zero depth.

> On the other hand, observed profiles do not have such a large intercept. This result must mean that particles falling as fast as 50 m/d cannot play as large a role as the authors suggest.

It plays a role in the export of Th and Pa, but we don't think we overemphasise its role. We assign larger K values to small particles, which has as a consequence that much more of the radionuclides are associated with small than with big particles. Indeed, at least in our model which works quite well, the small particles play the dominant role. You may be referring to the fact that we couldn't assign too large values of K for biogenic silica and calcium carbonate. Absolutely, but this is, admittedly, because of the shortcoming

of the model that it doesn't have small versions of these particles. As a consequence, at least some of the bulk concentration of biogenic silica and calcium carbonate corresponds to (observed) small particles (in the case of calcium carbonate the modelled concentration lies in between large and total, as briefly discussed in the paper).

> It can not be expected that the results from a model that gets the 1-D profiles incorrect, would be accurate in 3-D. The interpretation of the results, due to the tuning required to correct this model deficit, is uncertain.

We agree with the last sentence: we needed to tune, within a range based on published values, to get better model results, which make interpreting the model somewhat more difficult. However, we cannot agree with the premisse of the first sentence. Your model clearly shows a difference between the instant and non-instant 1-D scavenging models, but one cannot infer from this that one or the other is more likely to be correct (or closer to reality). Moreover, we happen to be closer to the instant-equilibration 1-D model.

Other issues

> Opal is still used at least once Figure 5c: caption says uM, color bar says nM

We fixed these mistakes.

Response to reviewer #2

Thank you for your comments to our manuscript. We hope to have addressed the issues you have raised satisfactorily.

Additions are denoted in green text in the manuscript.

Scientific Quality

Key points

1. […] I question the use of the K values for lithogenic material in this study. For small particles, the values used for both isotopes are much larger than $\pm$ 1 standard deviation of the value found by Hayes et al. (2015), the range in Geibert and Usbeck (2004), or the range from other previous studies (Anderson et al., 1992; Chase et al., 2002). The only study I have seen a K value for lithogenic material greater than that reported used this study was by Luo and Ku (2004) ($K = 230 \gg 10^6$ g/g), though this value mostly reflected the variability in K with the % lithogenic composition of sediment trap material from the Southern Ocean and the Equatorial Pacific. Similarly, the source for K for small particle POC is unclear; these values for both isotopes appear to be based on those derived by Hayes et al. (2015), but why was the higher end of the range used for ^{230}Th, and the low end of the range used for ^{231}Pa? The authors should state explicitly how they arrived at the values of K for both POC and lithogenic material.

The high values that we used for the small lithogenic particles are possibly because we tuned them too high because we are lacking other small particles (e.g. Mn/Fe oxides and calcium carbonate). Hayes et al. (2015) do report higher values for Th than for Pa. (Note that they use different exponents/units for K values for Th versus Pa.) In the supplement we included a table with values that shows we tried to choose reasonable values.

2.

PISCES includes aggregation between small and big POC (Aumont et al., 2015), but in Pa-231 and Th-230 equations 9(b,c) I do not see any terms that represent aggregation of particulate from small to big particles. The authors justify the omission of adsorption and desorption in equations 9(a,b,c) by assuming an instant equilibrium between the dissolved and the two particulate phases, but I do not see a similar argument for the omission of

aggregation of POC given. I think the authors need to clarify whether and how aggregation of the particulate radionuclide pools is or is not considered. Additionally, I am curious as to how difficult it would be to include aggregation of the sPOC-bound radionuclide pool in the model, since the aggregation rate can be derived from eq. (39) of Aumont et al. (2015). For example, could an aggregation term be included in equations 9(b,c):

$dA_{i,S/B}/dt = \pm\beta A_{i,D} K_{i,POC} P^{POC}$ ...,

where the aggregation rate represents a loss for radionuclides bound to small particles (S), a source for radionuclides bound to big POC (B), and B is the aggregation rate parameter (/d). Including this term in the current setup may not make sense considering the authors already assume instant equilibriation with respect to adsorption and desorption, and POC specific aggregation rates may be on the order of adsorption and desorption rates (0.1 to 1 /yr, Murnane (1994a)) or faster (Burd, 2013), but I think it is important to at least make explicit why aggregation of radionuclides in the small POC pool is or is not considered.

We explained in the text now why there is no such aggregation in the model based on the fact that we do have instant equilibration. Here in short, because of the latter, the former would not have any effect.

3. In the conclusion, along with mentioning the potential missing effects of nepheloid layers, Mn (oxyhydr)oxides, and a stronger AMOC on simulated ^{231}Pa and ^{230}Th, I also suggest that the authors mention the sensitivities of these distributions to K for the various particle phases. Although a sensitivity analysis may not be the goal of this study, I think it is at least worth speculating on how changing K values from those used in this study could alter and potentially improve the simulated distributions. Reported K values for any given particle type may vary by up to three orders of magnitude (e.g., $10^{3.98}$ from Guo et al. (2002) vs. up to $10^{6.23}$ from Geibert and Usbeck (2004) for SiO_2), may depend on whether K is derived from field observations or laboratory experiments, and may depend on differences in types of particles used in laboratory experiments (e.g., inor-

ganic vs. biogenic SiO_2 or $CaCO_3$). Given such large ranges, it is not unreasonable to suspect that a different, yet still plausible, combination of K values could better describe the available data.

We believe that we have given enough details on the effects of different K values, for this study. Anymore would be speculation and just as good a guess for the reader as for us. Therefore, we prefer not to add on this at this point.

Minor comments

1. The authors mention that this model is an improvement of that presented in Dutay et al. (2009). I think this improvement should be shown in the paper. For example, a table showing the root mean square deviation and/or r for dissolved and particulate Th-230 and Pa-231 in the dissolved and small particulate pool (there may not be enough observations to compare the fit to big particles) for both models (the one presented and the one from Dutay et al. (2009)) could be provided. This way, the improvement in fit obtained by the new model, which is one of the key results of this study, is made explicit.

Unfortunately, their model output is not available anymore, so we cannot do this. We have put our most important model output on PANGAEA, so comparing with our study should be possible when future studies want to compare with ours.

2. The authors provide r for the fit between simulated and observed dissolved ^{230}Th at GA02 and GA03. A similar comparison should be presented for ^{231}Pa along both sections, ^{230}Th and Pa-231 in the small particle pool at GA03 (I don't think these particulate radioisotopes have been reported along GA03).

We have added a Table to the supplementary material (Table S4) that lists the r for the dissolved and small particle pools of Th-230 and Pa-231 at GA02, as well as for our sensitivity simulation (Pa–opal affinity decreased).

Presentation quality

1. We have rephrased the sentence according to your suggestion.
2. Names in the figure and caption should now be consistent.
3. We added a comment on different diameter boundaries in the text.
4. We added an extra column; this probably looks just as well as adding it to one of the columns.
5. Yes, rephrased.
6. Thanks, fixed.
7. Done (described consequences of circulation and scavenging changes).
8. Rephrased according to your suggestion.
9. Done.
10. Done.

Main changes to the manuscript

In our response to the reviewers many changes we made to the manuscript are mentioned, but for a quick overview we list here the most notable changes to the manuscript:

- page 2, lines 21–24:
- page 4, lines 5–6: clarified/made precise the diameter/velocity cut-off;
- page 5, Figure 1: made acronyms in figure and caption consisten;
- page 7, bottom: added explanation why we did not include aggregation and disaggregation of Th and Pa in the model;
- page 17, bottom: added reasoning why use instant ad/desorption instead of slow equilibration.

The rest of this PDF file is a version of the manuscript where all significant changes are highlighted in green.

A global scavenging and circulation ocean model of thorium-230 and protactinium-231 with improved particle dynamics (NEMO–ProThorP 0.1)

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Abstract. In this paper, we set forth a 3-D ocean model of the radioactive trace isotopes ^{230}Th and ^{231}Pa. The interest arises from the fact that these isotopes are extensively used for investigating particle transport in the ocean and reconstructing past
5 ocean circulation. The tracers are reversibly scavenged by biogenic and lithogenic particles.

Our simulations of ^{230}Th and ^{231}Pa are based on the NEMO–PISCES ocean biogeochemistry general circulation model, which includes biogenic particles, namely small and big particulate organic carbon, calcium carbonate and biogenic silica. Small and big lithogenic particles from dust deposition are included in our model as well. Their distributions generally compare well with the small and big lithogenic particle concentrations from recent observations from the GEOTRACES programme,
10 except for boundary nepheloid layers for which, as up to today, there are no non-trivial, prognostic models available on a global scale. Our simulations reproduce ^{230}Th and ^{231}Pa dissolved concentrations: they compare well with recent GEOTRACES observations in many parts of the ocean. Particulate ^{230}Th and ^{231}Pa concentrations are significantly improved compared to previous studies, but they are still too low because of missing particles from nepheloid layers. Our simulation reproduces the main characteristics of the ^{231}Pa/^{230}Th ratio observed in the sediments, and supports a moderate affinity of ^{231}Pa to biogenic
15 silica as suggested by recent observations, relative to ^{230}Th.

Future model development may further improve understanding, especially when this will include a more complete representation of all particles, including different size classes, manganese hydroxides and nepheloid layers. This can be done based on our model, as its source code is readily available.

1 Introduction

20 Oceanic circulation and the carbon cycle play a major role in the regulation of the past and present climate. Heat and carbon dioxide in the atmosphere tend to equilibrate with the ocean surface, and are transported down into the deep ocean through the Meridional Overturning Circulation (MOC). Biogeochemical cycling also generates organic carbon that transfers into the deep ocean through particle sinking. Because of this, the strength of the MOC and particle removal participate actively in the regulation of the climate on the Earth.

25 Trace elements are also affected by these mechanisms, and represent useful tools to provide constraints on these processes. The GEOTRACES programme has generated a unique, large data set that can now be used to better understand biogeochemical oceanic processes. Modelling quantifies and provides more information on the processes that control the oceanic distribution

of these new observations. In present day climate, it is difficult to measure the MOC strength, and for past climate there are no measurements available at all. Isotopes and trace elements from sediment cores are used as proxies to infer past ocean circulation. Several examples include carbon isotopes (Broecker et al., 1988), the cadmium/calcium ratio (Rosenthal et al., 1997), the ratio between protactinium-231 and thorium-230 (^{231}Pa/^{230}Th) (Böhm et al., 2015), and the neodymium isotope ratio ^{143}Nd/^{144}Nd (Piotrowski et al., 2005). These proxies are affected by dynamical and biogeochemical processes. Including these proxies in a climate model is a way to better understand the climatic signal they register.

We will focus on ^{230}Th and ^{231}Pa, because these isotopes are well documented by the international GEOTRACES programme, and they are particularly suitable to study the transfer of particulate matter since the isotopes' source in the ocean is perfectly known: radioactive decay of uranium isotopes. Others have modelled ^{143}Nd and ^{144}Nd (Arsouze et al., 2008, 2009; Ayache et al., 2016), ^{13}C (Tagliabue et al., 2009) and ^{14}C (Mouchet, 2013). The ratio between ^{230}Th and ^{231}Pa is used as a proxy for past ocean conditions, but this signal is potentially affected by both circulation and biogeochemical changes. Therefore, a correct understanding of the scavenging and the underlying particle dynamics is essential in order to better simulate these tracers (Dutay et al., 2015).

Protactinium-231 and thorium-230 are produced in the ocean by the α-decay of uranium-235 and uranium-234, respectively. Because the activity of uranium is approximately uniform in the ocean, ^{231}Pa and ^{230}Th are produced at a relatively constant rate ($\beta_{\mathrm{Pa}} = 2.33 \times 10^{-3}\,\mathrm{dpm\,m^{-3}\,a^{-1}}$ and $\beta_{\mathrm{Th}} = 2.52 \times 10^{-2}\,\mathrm{dpm\,m^{-3}\,a^{-1}}$, dpm = *disintegrations per minute*) (Henderson and Anderson, 2003). They are both scavenged rapidly by the many particles that reside in the ocean and settle towards the seafloor. ^{231}Pa is less sensitive to particle scavenging than ^{230}Th, which is reflected in the longer residence time of ^{231}Pa (80–200 yr) compared to that of ^{230}Th (20–40 yr) (Yu et al., 1996). ^{231}Pa and ^{230}Th are radioactive, decaying to radium isotopes and having a half-life of 32.76 kyr and 75.40 kyr, respectively. Each combination of particle–radionuclide adsorption has a different reactivity. Other factors may affect the adsorption of radionuclides onto particles. For example, smaller particles have larger surface area to volume ratios, which results in an increase in the number of these radionuclides adsorbed per particle. Additionally, while the adsorption rate is expected to increase with particle concentration, the adsorption rate may be partly limited by the coagulation of non-filtered particles (or colloids) to filtered particles (Honeyman et al., 1988; Honeyman and Santschi, 1988). The vertical distributions of natural radionuclides, such as ^{230}Th and ^{231}Pa, are hence sensitive to the distribution and mixture of particles. As a consequence of the different particle reactivities of ^{230}Th and ^{231}Pa, the dissolved concentration ratio ^{231}Pa/^{230}Th deviates from the production activity ratio of 0.093 (Anderson et al., 1983; Anderson, 2003; Rutgers van der Loeff et al., 2016).

As both particle dynamics and circulation of the ocean affect ^{230}Th and ^{231}Pa, numerical biogeochemical general circulation models are used to study the relative contribution of these mechanisms. The isotopes have been simulated in models of intermediate complexity, for instance by Henderson et al. (1999) (LSG-OGCM), Marchal et al. (2000) (EMIC 2.5D), Heinze et al. (2006) (HAMOCC) and Luo et al. (2010). More complex ocean general circulation models models have also been used to simulate these tracers Dutay et al. (2009) (NEMO–PISCES), Siddall et al. (2007); Rempfer et al. (2017) (Bern3D) and Gu and Liu (2017) (CESM). Dutay et al. (2009) demonstrated that the particle concentration simulated by the PISCES model in the deep ocean was a factor of 2 in the mesopelagic zone to 50 in the deepest ocean too low. This lead to overestimated radionuclide concentrations in the deep ocean. Therefore, it is crucial to improve the representation of the particles (Dutay et al., 2009). Rempfer et al. (2017) showed that taking into account additional sinks at the seafloor and at the ocean margins yields an improved agreement with observations, especially for the dissolved phases of ^{230}Th and ^{231}Pa. Particulate ratios improved to a lesser extent; the authors have not presented evaluation of their simulated particulate concentrations. Gu and Liu (2017) had similar goals as this study.

In this study, we try to improve on previous studies that simulate the distribution of ^{230}Th and ^{231}Pa. Our approach is to improve the mechanistic description of the particle and radionuclide cycling, as well as on the side of system design and reusability of the model. We evaluate how well the model fits with observations, but tuning is not one of our main goals. Just like Dutay et al. (2009), we use the NEMO–OPA (Nucleus for European Modelling of the Ocean – Océan PAralelisé) ocean general circulation model (Madec and the NEMO team, 2016) and the PISCES biogeochemical model (Aumont et al., 2015).

5   Aumont et al. (2017) showed that dissolution rates of Particulate Organic Carbon (POC) in PISCES were overestimated, and they improved on this by introducing a spectrum of different labilities. This improved the simulation of both small and big POC significantly, so we use this same model for our simulations. There are several improvements since Dutay et al. (2009), namely the new model

- includes lithogenic particles from dust deposition;

10   – improved the biogeochemistry, affecting the biogenic particle distributions (Aumont et al., 2015, 2017);

- includes three different phases per nuclide (dissolved, and adsorbed onto big (and small) particles), whereas Dutay et al. (2009) include only a single compartment (total concentration) for each nuclide from which they calculated the respective phases based on chemical equilibrium;

- is more precise/explicit on the mathematical formalism;

15   – is written in Fortran 95 instead of Fortran 77, making the extension of its use to other modern Fortran models easier;

- is part of a modern model framework, NEMO–TOP, facilitating its use with other models in this framework.

The main objective of this paper is to improve on the simulation of ^{230}Th and ^{231}Pa (the dissolved and two particulate size classes of particles for both nuclides), based on an improved modelling of small and big particles.

New observations are available from the GEOTRACES programme. Especially the North Atlantic GA03 transects will be 20   used for model validation; because on this transect not only dissolved ^{230}Th and ^{231}Pa have been measured (Hayes et al., 2015b), but also their adsorbed forms (Hayes et al., 2015b) and biogenic and lithogenic particle concentrations in two size classes (Lam et al., 2015).

2   Model description

In order to simulate the biogenic particle dynamics and its interaction with the ^{230}Th and ^{231}Pa trace isotopes, we use the bio-25   geochemical circulation model NEMO–PISCES (Madec, 2008; Aumont et al., 2015). This model has been employed for many other studies concerning trace metals, as well as large-scale ocean biogeochemistry (e.g. Gehlen et al., 2007; Arsouze et al., 2009; Dutay et al., 2009; Tagliabue et al., 2010; Van Hulten et al., 2013, 2014, 2017b). We force PISCES by a climatological year of circulation fields (including turbulent diffusion) that was obtained from the dynamical component of NEMO. Table 1 gives an overview of this and other components of the model and their relevant properties.

5   All model fields are defined on the ORCA2 discrete coordinate system, an irregular grid covering the whole world ocean with a nominal resolution of $2° \times 2°$, with an increased resolution in the meridional direction near the equator and Antarctica, and in both horizontal directions in the Mediterranean, Red, Black and Caspian Seas. On the Northern Hemisphere, it has two coordinate singularities, one in Canada and the other in Russia, such that both singularities fall outside the computational domain. The vertical resolution of the ORCA2 grid is 10 m in the upper 100 m, increasing downwards to 500 m, such that there 10   are 30 layers in total and the ocean has a maximum depth of 5000 m (Madec and Imbard, 1996; Murray, 1996). The timestep

model component	improvements or relevant properties	timestep
circulation (OPA)	prescribed forcing (off-line physics) (e.g. used by Arsouze et al., 2009)	6.0 h
biogenic particles	prognostically integrated by PISCES (Aumont et al., 2015)	1.5 h
– POC	two size classes settling with $w_s = 2\,\mathrm{m\,d^{-1}}$ and $w_b = 50\,\mathrm{m\,d^{-1}}$; variable reactivity (Aumont et al., 2017)	
– $CaCO_3$	one size class settling with $w_b = 50\,\mathrm{m\,d^{-1}}$; higher order dissolution (e.g. Subhas et al., 2015)	
– $bSiO_2$	one size class settling with $w_b = 50\,\mathrm{m\,d^{-1}}$; no changes	
lithogenic particles	based on forcing of dust deposition; two size classes settling with $w_s = 2\,\mathrm{m\,d^{-1}}$ and $w_b = 50\,\mathrm{m\,d^{-1}}$	1.5 h
radionuclide model	six prognostic tracers (dissolved, small and big adsorbed; ^{231}Pa and ^{230}Th)	6.0 h

Table 1. Model components and properties essential for the radionuclide model. OPA stands for Océan PAralalisé.

of the model is 6 h for the dynamics and the radionuclides, and 1.5 h for the biogeochemistry (PISCES) and the lithogenic particles. When necessary, sub-timestepping is done for all sinking components in the model.

2.1 Circulation

The circulation was obtained by forcing NEMO in the ORCA2 configuration with climatological air-sea boundary conditions, consisting of heat, fresh water and momentum fluxes that were derived from bulk formulae. They are functions of wind, sea surface temperature, air temperature, air humidity and evaporation minus precipitation. For the tropics, daily wind stress was used, which was based on European Remote Sensing (ERS) satellite data, and for the polar regions, NCEP/NCAR re-analysis data (Kalnay et al., 1996; Kistler et al., 2001). Surface salinity was restored with a timescale of 60 days towards the seasonal Polar Science Center Hydrographic Climatology (PHC) dataset to avoid model drift (Timmermann et al., 2005). The last year of this 200-year simulation is used as our one-year climatology with a resolution of five days of the dynamics.

2.2 Particle dynamics

This version of PISCES includes two size classes of POC, both with differential remineralisation rates (Aumont et al., 2017), one size class of biogenic silica ($bSiO_2$) and one class of calcium carbonate ($CaCO_3$) (Fig. 1). In the model, particles sink down with two velocities: $w_s = 2\,\mathrm{m\,d^{-1}}$ and $w_b = 50\,\mathrm{m\,d^{-1}}$. Aumont et al. (2015) marks the diameter boundary at $100\,\mu m$, but as we need to compare with observations, the size classes will be taken to correspond with "small" particles ($\varnothing \leq 51\,\mu m$) and "big" particles ($\varnothing > 51\,\mu m$).

The dissolution equation for calcium carbonate is given by

$$\frac{\mathrm{d}[CaCO_3]}{\mathrm{d}t} = -R[CaCO_3],\qquad(1)$$

where the rate variable (constant in t) for calcite dissolution is

$$R = k \cdot (1 - \Omega)^n,\qquad(2)$$

where k and n are two somewhat restricted, though tunable, parameters that signify a dissolution rate constant and a reaction rate order, respectively. The calcite saturation state is given by

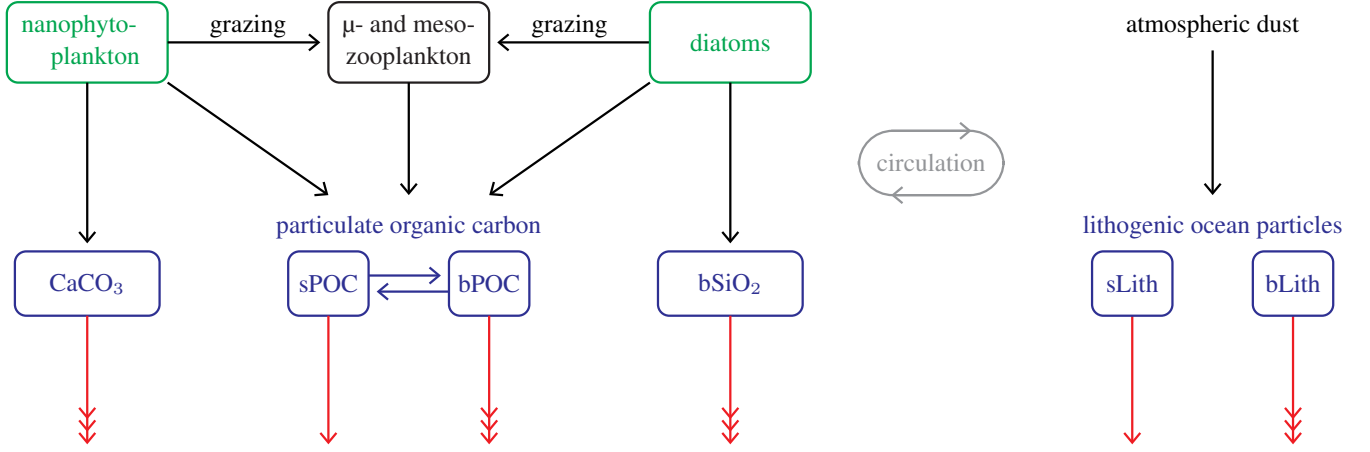$$\Omega = \frac{[Ca^{2+}][CO_3^{2-}]}{K'_{sp}},\qquad(3)$$

4

Figure 1. Conceptual model of particle dynamics, on which the numerical model is based. Nanophytoplankton and diatoms (in green) take up nutrients and CO_2, which are released again from respiration and remineralisation of $bSiO_2$, POC, dissolved organic matter and lithogenic particles. The nutrients, CO_2 and dissolved organic matter are not represented in the figure, because only particles impact ^{231}Pa and ^{230}Th. Zooplankton are denoted by the black box. All sinking particles are denoted by blue boxes. Effectively, this figure comprises the internal cycling of PISCES, minus details that are not of interest here, plus the lithogenic dust model. Small and big particulate Organic Matter are denoted by sPOM and bPOM, both subject to the differential lability scheme (Aumont et al., 2017), and $bSiO_2$ stands for biogenic silica. Similarly, sLith and bLith stand for small and big lithogenic particles. Sinking is denoted by the red arrows; triple arrows means fast, normal arrows slow.

where K'_{sp} is the product of the saturation concentrations of calcium and carbonate. This can be approximated by

$$\Omega \approx \frac{[CO_3^{2-}]}{[CO_3^{2-}]_{sat}}, \tag{4}$$

because relative variations in $[Ca^{2+}]$ are small. For $CaCO_3$ we changed the standard first-order dissolution kinetics parameterisation to a fourth-order dissolution, based on evidence of Keir (1980) and subsequent studies. For our simulation we used a calcite dissolution rate constant of $k = 2.5 \, mo^{-1}$ and a dissolution order of $n = 3.9$ (Subhas et al., 2015).

In addition to biogenic particles, we introduced lithogenic dust particles in the model. The yearly average dust flux is derived from Hauglustaine et al. (2004) and is presented in Fig. 2. It is used by the model as the input of lithogenic ocean particles as well as for nutrient supply in PISCES. This dust deposition field has been tested in biogeochemical studies with this configuration of NEMO (e.g. Van Hulten et al., 2013).

Small and big lithogenic particles are added to the upper layer of the ocean according to:

$$\left.\frac{\partial P^{Litho}}{\partial t}\right|_{surface} = \frac{f}{\Delta z_1} \cdot \Phi_{dust}, \tag{5}$$

where P^{Litho} is the small (big) lithogenic particle concentration, f is the fraction of the dust that gets partitioned into the small (big) lithogenic particles in the ocean, $\Delta z_1 = 10 \, m$ is the thickness of the upper model layer, and Φ_{dust} is the dust flux.

Our model has two size classes for lithogenic particles, so this equation is applied for two different concentrations P^{Litho} and respective fractions f. We set the small lithogenic dust flux fraction to 20 % and the big one to 80 %. Once partitioned in the ocean, the lithogenic particles sink down, changing their concentrations throughout the ocean according to

$$\frac{dP^{Litho}}{dt} = -w \cdot \frac{\partial P^{Litho}}{\partial z} + \left(\mathcal{A}\nabla_h^2 + \mathcal{B}\frac{\partial^2}{\partial z^2}\right)P^{Litho}, \tag{6}$$

where w is the settling velocity, set to the constant $2 \, m \, d^{-1}$ for the small lithogenic particles and to $50 \, m \, d^{-1}$ for the big particles. The depth, z, is positive upwards, ∇_h is the horizontal divergence, and $\mathcal{A}$ and $\mathcal{B}$ are respectively the horizontal and
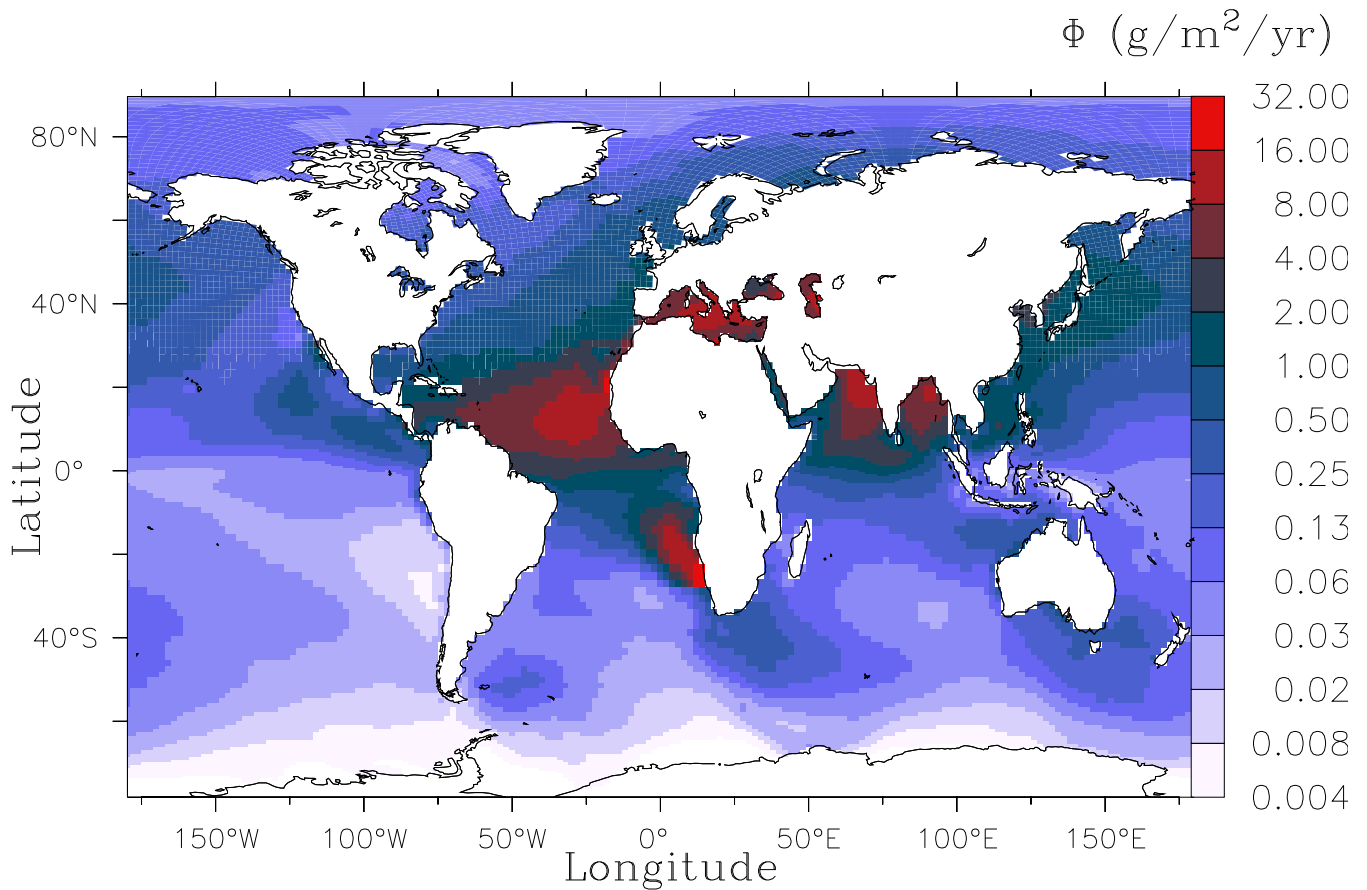
Figure 2. Dust deposition on a logarithmic scale $(\mathrm{g\,m^{-2}\,a^{-1}})$. It is the integrated flux over the 12-months climatology based on Hauglustaine et al. (2004).

vertical eddy diffusivity coefficients. The material derivative includes a term for eddy-induced velocity (Gent and McWilliams, 1990; Gent et al., 1995). Equation (6) is identical to the settling of small and big POC (Aumont et al., 2015). Of course, there are also biological and chemical sources and sinks for POC. However, for lithogenic particles there are no such sources or sinks because the only source in our model is dust deposition and we assume the lithogenic particles are refractory. The lithogenic particles are removed from the model domain when arriving at the seafloor, which means that they are buried in the sediment.

2.3 Radionuclides

Thorium-230 and protactinium-231 are produced throughout the ocean from the decay of ^{234}U and ^{235}U, respectively. Because of long residence times of over 200 kyr, these uranium isotopes are approximately homogeneously distributed throughout the ocean, and do not change much over time (Ku et al., 1977). The residence time of ^{234}U is 3.2–5.6×10^5 a, which is much longer than the full mixing time of the world ocean of about 10^3 a (Dunk et al., 2002). The ^{234}U concentration can vary about $10\,\%$, depending mostly on the salinity (Owens et al., 2011), but that is smaller than uncertainties arising from other assumptions in our model. Therefore, we assume that the production rates of ^{230}Th and ^{231}Pa are constant, both in space and time.

The ^{230}Th and ^{231}Pa radionuclides are reversibly scavenged by biogenic and lithogenic particles (Fig. 3). We will assume, as in previous studies (e.g. Siddall et al., 2005; Dutay et al., 2009), that the adsorption and desorption reaction rates are sufficiently fast compared to radionuclide production, decay, advection, mixing, change in particle distribution and settling of the adsorbed
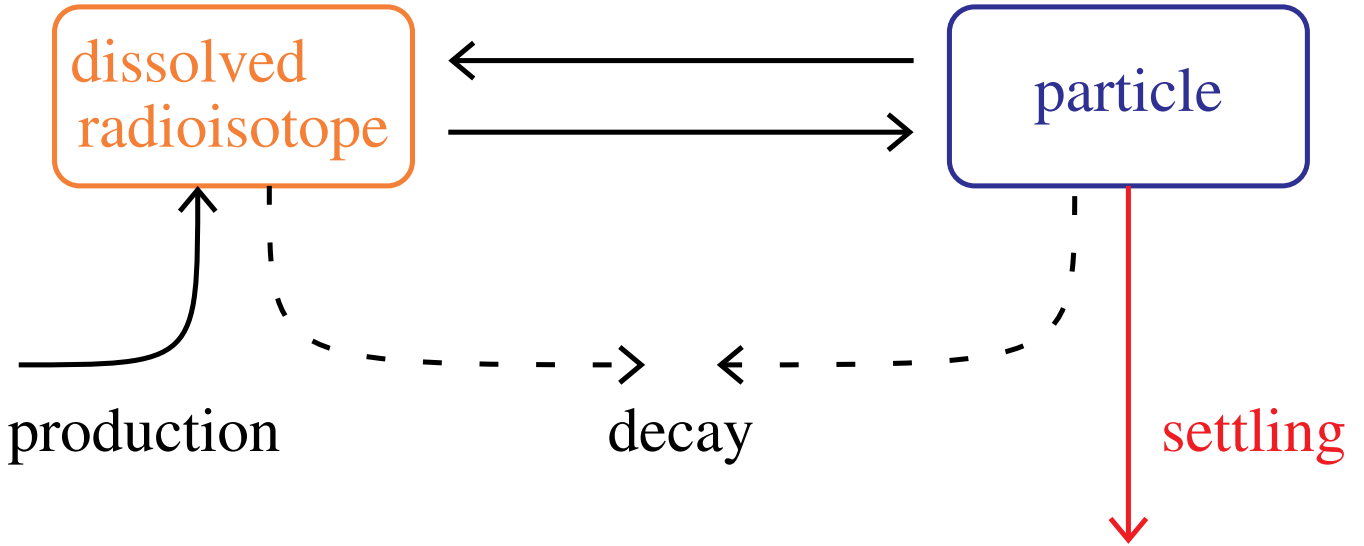
6

Figure 3. The conceptual reversible scavenging model for the radionuclides. The radioisotopes, ^{230}Th and ^{231}Pa, are depicted in orange when in the dissolved phase. Just like with the other PISCES tracers, the dissolved and particulate radionuclides are transported by circulation and eddy diffusion.

phases (see discussion in section 6). Because of that, we equilibrate between the dissolved and adsorbed phases instantly at each time step. This means, only considering ad- and desorption processes at this point, that we must solve this set of equations for every nuclide i:

$$0 = \frac{\partial}{\partial t}(A_{i,D} + A_{i,S} + A_{i,B}),\tag{7a}$$

$$A_{i,S} = A_{i,D}\sum_{j \in S} K_{ij}P^j,\tag{7b}$$

$$A_{i,B} = A_{i,D}\sum_{j \in B} K_{ij}P^j,\tag{7c}$$

where A_i stands for the activity of nuclide $i \in \{^{230}\text{Th}, ^{231}\text{Pa}\}$,[1] and where P^j stands for the concentration of particle $j \in \{\text{sPOC}, \text{bPOC}, \text{CaCO}_3, \text{bSiO}_2, \text{sLitho}, \text{bLitho}\}$ (in gram per gram of seawater, so strictly this is a *mass fraction*). D is the set of non-sinking phases (here only dissolved), S of small particles that sink with $2\,\text{m}\,\text{d}^{-1}$ and B of big particles that sink with $50\,\text{m}\,\text{d}^{-1}$ (in the same way as Eq. (6) for lithogenic particles). For any particle size class $J \in \{S, B\}$ and any radionuclide i, we define $A_{i,J} = \sum_{j \in J} A_{ij}$. Finally, K_{ij} is the equilibrium partition coefficient of nuclide i for particle j. **(addressing report reviewer 2–Q2:)** As with small and big POC, adsorbed ^{230}Th and ^{231}Pa should aggregate and disaggregate. However, any transference of radionuclides between two particles types or size classes, in addition to the instant equilibration, would not have any effect as there would quickly be a new equilibrium of the radionuclides between the particles and the dissolved phase at the next instant reequilibration (which happens each timestep). Therefore there is no equation for the radionuclides addressing aggregation.

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[1]Concentration of radionuclides (amount per unit of volume) is proportional to its (radio)activity (disintegrations per unit of time). These terms are used interchangeably throughout this paper. Moreover, they are considered identical: the activity or concentration $[^{230}\text{Th}]_D \equiv A_{230\,\text{Th}}$ is expressed in $\text{mBq}\,\text{m}^{-3}$, and thus we report activity ratios of ^{231}Pa/^{230}Th.

Since we cannot solve this analytically, this will be done numerically by first assigning a new value to the dissolved nuclide activity:[2]

$$A_{i,D} := \frac{A_{i,D} + A_{i,S} + A_{i,B}}{1 + \sum_j K_{ij} P^j}. \tag{8}$$

Then, we calculate the activity of i that is adsorbed onto small and big particles by applying Eqs (7b) and (7c). With this approach, the small and big adsorbed concentrations equilibrate instantly. Assuming that the change in adsorption strength is much smaller than the relative change in tracer activity, $\forall_{i,J}: \partial_t \sum_{j \in J} K_{ij} P^j \ll \partial_t A_{i,D}/A_{i,D}$, the total activity of every i is conserved, i.e. Eq. (7a) holds. Then,

Proof. Let the total adsorption strength for any isotope i be $Q_i = \sum_j K_{ij} P^j$ and the total amount of the same isotope $T_i = A_{i,D} + A_{i,S} + A_{i,B}$, and let primes denote the updated concentrations. Assume that the adsorption strength for every isotope i is constant ($Q_i' \equiv Q_i$).

$$\begin{aligned}
T_i' &= \frac{T_i}{1 + Q_i'} + A_{i,D}' Q_i' \\
&= \frac{T_i}{1 + Q_i} + \frac{T_i}{1 + Q_i} Q_i \\
&= \frac{T_i}{1 + Q_i}(1 + Q_i) = T_i
\end{aligned}$$

$$\Rightarrow \partial_t T_i' = \partial_t T_i = 0. \qquad \square$$

The ^{230}Th and ^{231}Pa that are adsorbed onto the particles follow the same law as the small and big (lithogenic) particles (Eq. 6). Of course, by definition, the adsorbed radioisotope and the particle settle with the same speed, and thus we have implemented it.

The decay terms of ^{230}Th and ^{231}Pa are much smaller than the other sources and sinks, but they are included in the model:

$$\frac{\mathrm{d}A_{i,D}}{\mathrm{d}t} = \beta_i - \lambda_i A_{i,D} + \left(\mathcal{A}\nabla_h^2 + \mathcal{B}\frac{\partial^2}{\partial z^2}\right)A_{i,D}, \tag{9a}$$

$$\frac{\mathrm{d}A_{i,S}}{\mathrm{d}t} = -\lambda_i A_{i,S} - w_s \frac{\partial A_{i,S}}{\partial z} + \left(\mathcal{A}\nabla_h^2 + \mathcal{B}\frac{\partial^2}{\partial z^2}\right)A_{i,S}, \tag{9b}$$

$$\frac{\mathrm{d}A_{i,B}}{\mathrm{d}t} = -\lambda_i A_{i,B} - w_b \frac{\partial A_{i,B}}{\partial z} + \left(\mathcal{A}\nabla_h^2 + \mathcal{B}\frac{\partial^2}{\partial z^2}\right)A_{i,B}, \tag{9c}$$

where β_i and λ_i are respectively the production rate and radioactive decay of isotope i.

3 Simulations

For the numerical simulations, we forced the just described model, an adjusted version of PISCES with lithogenic particles and radiotracers, with the off-line circulation fields. The model was spun up for 500 yr, after which it was in an approximate steady state (decadal drift of $-0.002\,\%$ for total ^{230}Th and $+0.058\,\%$ for total ^{231}Pa). Protactinium-231 has a larger drift than thorium-230, because ^{230}Th is everywhere in the ocean more quickly removed because of its high particle reactivity. The lithogenic particles are in a steady state, and the PISCES variables are in an approximate steady state (e.g. phosphate shows a drift of $-0.005\,\%$ per decade.

The partition coefficients K_{ij}, with $i \in \{\mathrm{Th}, \mathrm{Pa}\}$, depend on the type of particle j and are given in Table 2. These coefficients have large uncertainties still, but their values can be constrained by reported values from different experimental studies (e.g. Chase et al., 2002; Geibert and Usbeck, 2004; Hayes et al., 2015b). Therefore, we had quite some freedom in prescribing

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[2]When summation bounds are not specified, the union of all phases, $D \cup S \cup B$, is assumed.

Particle	K_{Pa} $(Mg\,g^{-1})$	K_{Th} $(Mg\,g^{-1})$	Settling speed
small POC	2.0	5.0	$2\,m\,d^{-1}$
big POC	0.4	1.0	$50\,m\,d^{-1}$
biogenic silica	0.5 (0.4)	0.5 (1.0)	$50\,m\,d^{-1}$
$CaCO_3$	0.12	5.0	$50\,m\,d^{-1}$
small lithogen.	10.0	50.0	$2\,m\,d^{-1}$
big lithogen.	1.0	5.0	$50\,m\,d^{-1}$

Table 2. Partition coefficients for the different modelled particles in 10^6 gram of seawater per gram of particles. Between braces is the value for the sensitivity simulation discussed in Section 6.1.

values. The adsorption onto calcium carbonate is a factor two decreased from Chase et al. (2002). This brings the K value of Pa closer to that in Hayes et al. (2015b) based on field measurements, namely $0.9\pm0.4\ Mg\,g^{-1}$ (10^6 gram of seawater per gram of particles), but since we maintained the ratio of K values, the partition coefficient of Th is much smaller than found by Hayes et al. (2015b). It is consistent with the laboratory results of Geibert and Usbeck (2004), but they cleaned the biogenic silica aggresively, bought carbonate that should be organic-free and clay (smectite) was purchased as a commercial standard and not cleaned (Geibert and Usbeck, 2004, as well as pers. comm. January 2018). In the real ocean, solid minerals are not clean but have organic matter around them, giving rise to increased scavenging (Chuang et al., 2014, for diatoms). Therefore, our K's for at least biogenic silica and calcium carbonate may be lower than what is typically considered realistic. The actual reason for the relatively low K values is that we only have fast sinking biogenic silica and calcium carbonate particles in the model that represent also smaller particles that sink slower. Thus using higher K values would result in too much export of Pa and Th. For small lithogenic particles, K is set about a factor five larger than literature (Geibert and Usbeck, 2004; Hayes et al., 2015b), whereas big lithogenic particles have a smaller value than reported in the literature. The supplementary document contains a more exhaustive table of information, including an estimated literature range (Table S3). The consequences of chosen values of the partition coefficients is discussed further in Section 6.

4 Observations

In this study, we will focus on the GEOTRACES GA03 transect in the North Atlantic Ocean. Recently, a large number of measurements on both radionuclides and their carrier particles have been collected on this transect. This unique combination makes it especially useful to evaluate a radionuclide scavenging model. We will also compare our global ocean model with several observational datasets throughout the ocean (Table 3). Observations obtained from the GEOTRACES programme are denoted with the respective GEOTRACES transect number (Mawji et al., 2015).

For the carrier particles most of our data come from Lam et al. (2015), which is a recent GEOTRACES dataset at the GA03 transect in the North Atlantic Ocean. An older compilation of particles is taken from Lam et al. (2011). We use some older data as well (Table 3). Particle concentrations were determined by filtering seawater in situ.

Concentrations of both dissolved and particulate phases of ^{230}Th and ^{231}Pa were taken from Hayes et al. (2015a) (GA03) and Hayes et al. (2014) (Pacific Ocean). Other data are listed in Table 3.

To evaluate the sediment ^{231}Pa/^{230}Th flux of the model, we will compare with compilations of the Holocene (i.e. top core particulate concentrations) (Table 3).

Transect	Year	Expedition	Ocean basin	Tracers used	Citation
Carrier particles POC, $CaCO_3$, $bSiO_2$ and lithogenic					
–	1973–74	GEOSECS st.235,239,306	Central Pacific	$CaCO_3$	Brun-Cottan et al. (1991)
–	1985–91	Alcyone-5, Eve-1, Hydros-6	North Pacific	s/bPOC	Druffel et al. (1992)
–	1982–97	WCR, Line P, SOFeX, K2, JGOFS	Pacific and Atlantic	s/bPOC, $bSiO_2$, $CaCO_3$	Lam et al. (2011)
GA03	2011	US GT10 and GT11	North Atlantic	POC, $bSiO_2$, $CaCO_3$; s and b	Lam et al. (2015)
Radionuclides ^{230}Th and ^{231}Pa					
–	1979	R/V *Knorr* cruise 73/leg 16	Central-east Pacific	^{230}Th$_{D,S \cup B}$	Bacon and Anderson (1982)
–	1994	R/V *Moana Wave*, HOT-57	Central Pacific	^{230}Th$_{D,D \cup S \cup B}$	Roy-Barman et al. (1996)
–	1976–98	(multiple)	global	^{230}Th$_{D,S,B}$	Henderson et al. (1999)
–	1983–02	(multiple)	global	^{230}Th$_{D,S \cup B}$	Marchal et al. (2007)
GIPY5_w	2008	ANT XXIV/3	Southern Oc. / Drake	^{230}Th$_{D,S,B}$	Venchiarutti et al. (2011)
GIPY5_e	2008	ANT XXIV/3	Southern Oc. / 0° E	^{230}Th$_{D,S,B}$	Venchiarutti et al. (2011)
–	2009	SO202-INOPEX, ALOHA, SAFe	North Pacific	^{230}Th$_D$	Hayes et al. (2013)
GA02	2011	JC 057	Southwest Atlantic	^{230}Th$_D$, ^{231}Pa$_D$	Deng et al. (2014)
GA03	2010	US GT10 and GT11	North Atlantic	^{230}Th$_{D,S,B}$, ^{231}Pa$_{D,S,B}$	Hayes et al. (2015b)
Sediment top core ^{231}Pa/^{230}Th					
–	1966–97	(multiple)	global	total particulate	Holocene compilation[a]
–	2009	R/V *M. Dufresne*, MD173/RETRO3	global	total particulate	Burckel et al. (2015, 2016)

Table 3. Observations used for comparison with the model simulations.

[a] http://climotope.earth.ox.ac.uk/data_compilations

5   Results

5.1   Circulation

As mentioned before, our model is part of a global general circulation model, but instead of solving the Navier-Stokes equa-
tions, our tracers are advected by the circulation fields of a previous simulation of the dynamical ocean model. Since the
overturning circulation is of importance for the redistribution of the tracers, we here present basic results.

Figure S1 of the supplementary document presents the Overturning Stream Function (OSF) of the Atlantic Ocean. The OSF
is defined as the zonally (through the basin) and vertically (from the surface downwards) integrated meridional component
of the current velocity. We use this as a measure for the Atlantic Meridional Overturning Circulation (AMOC). The upper
overturning cell transports 14 Sv ($1 \, \text{Sv} = 10^6 \, \text{m}^3 \, \text{s}^{-1}$) on average. Observations suggest higher values of about 19 ± 5 Sv
(Cunningham et al., 2007; Rayner et al., 2011; Sinha et al., 2018). The lower cell has an overturning strength of about 6 Sv,
which is stronger than the about 2 Sv from estimates (e.g. Talley et al., 2003). This overestimation is mostly due to the fact
that the AMOC is shallow. The AMOC reaches about $2500 \, \text{m}$ depth, whereas observational studies show the deep water sinks
down to about $4000 \, \text{m}$. At around 30° N the AMOC has weakened notably, which is consistent with some studies (e.g. Johnson,
2008), but the Antarctic Bottom Water (AABW) does not go as far north as other studies suggest (e.g. 45° N in Van Aken,
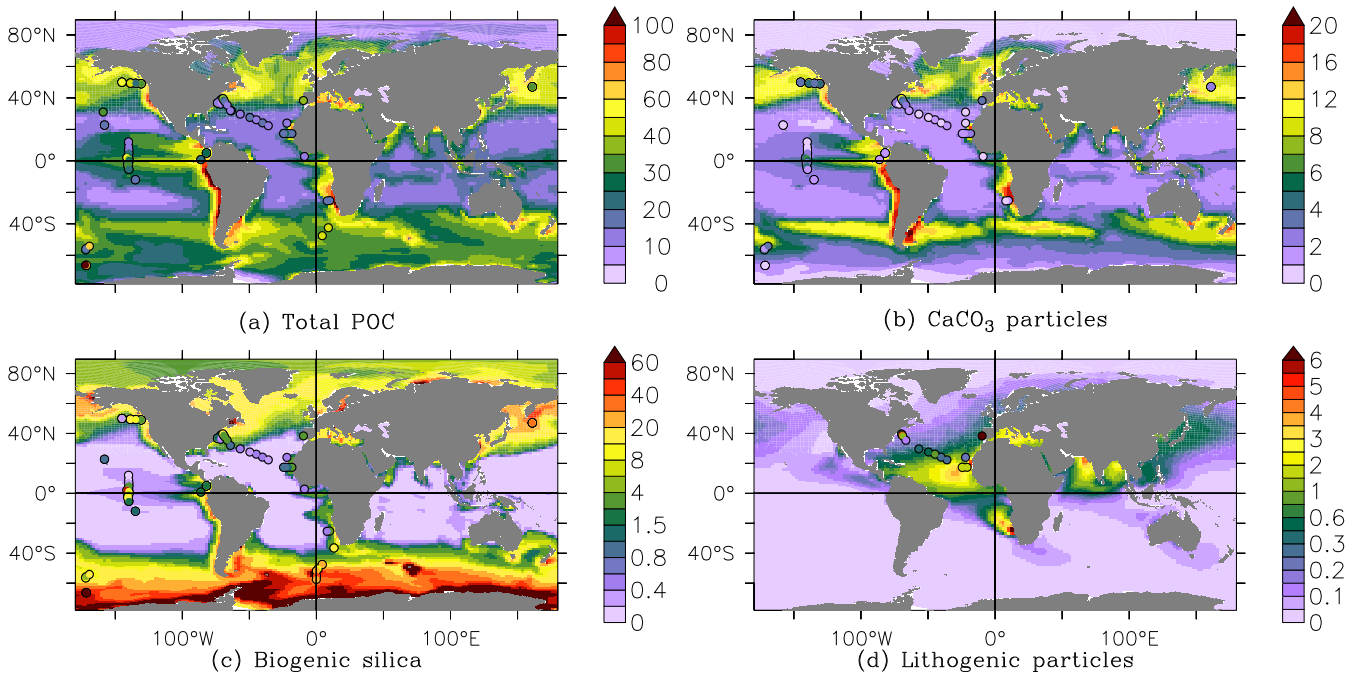2011).

Figure 4. Particle concentrations from the PISCES model at the surface ocean. Observations are represented as coloured discs (Brun-Cottan et al., 1991; Druffel et al., 1992; Lam et al., 2011, 2015). The unit of every modelled and measured particle concentration is $\mathrm{mg\,m^{-3}}$.

The Antarctic Circumpolar Current is also an important feature for large-scale ocean circulation. The through-flow at Drake Passage is a good measure for that. In our model the flux through Drake Passage is 200 Sv, which slightly overestimates recently reported values (e.g. 173 ± 11 Sv in Donohue et al., 2016).

5.2 Particles

Figure 4 shows the surface concentrations of total POC, big calcium carbonate, big biogenic silica and total lithogenic particles. Observations are plotted as coloured discs on top of the modelled concentrations. The modelled biogenic particles include living matter. For the model, we assume POC includes all phytoplankton and microzooplankton (not mesozooplankton since it may swim away), calcium carbonate contains the assumed fraction of $CaCO_3$ in the phytoplankton, and biogenic silica includes living diatoms. The modelled concentrations and patterns of total POC compare well with the observations (Fig. 4a). For instance, coastal and equatorial Pacific POC concentrations are both in the model and the observations elevated compared to other regions, albeit that the spatial extent of the oligotrophic regions appears to be too small in the model. We compare our modelled $CaCO_3$ (big, fast-sinking, calcium carbonate particles), with the big calcium carbonate particles from observations. As the model tries to represent $CaCO_3$ with only one size class, neither comparing with only big particles nor comparing with total $CaCO_3$ is completely fair. Incidentally, we have data of big $CaCO_3$ for both the Atlantic and Pacific Ocean, whereas only for the Atlantic Ocean we have small particle data. Even though the meridional patterns of big $CaCO_3$ particles are reproduced, its concentrations are generally overestimated, especially in the Gulf of Alaska (Fig. 4b). Contrarily, *total* observed $[CaCO_3]$, which is only provided along section GA03 where small particulate $CaCO_3$ measurements are available, is higher than the prediction of our modelled (only big) $CaCO_3$ particle concentration. The model produces a reasonable spatial distribution of biogenic silica: there are more elevated values in the high latitudes, but concentrations are underestimated at the surface at low latitudes (Fig. 4c).
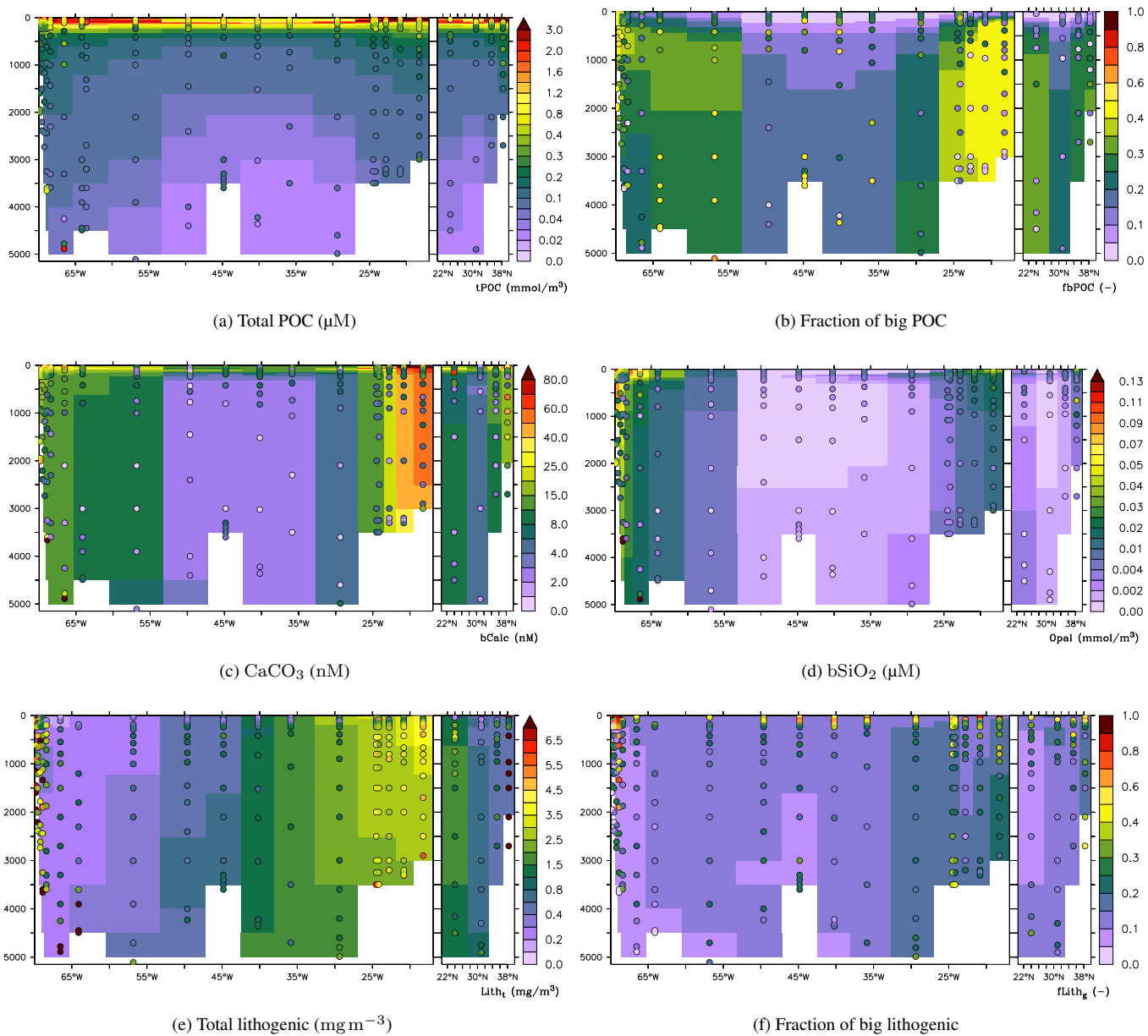
(a) Total POC (μM)

(b) Fraction of big POC

(c) CaCO₃ (nM)

(d) bSiO₂ (μM)

(e) Total lithogenic ($mg\,m^{-3}$)

(f) Fraction of big lithogenic

Figure 5. Particle fields at the GA03 North Atlantic GEOTRACES transect; observations (Lam et al., 2015) as coloured discs.

12

The recent large dataset obtained from the full-ocean depth GEOTRACES GA03 transect (Lam et al., 2015) provides an opportunity to analyse the performance of PISCES in more detail. PISCES generally produces the right order of magnitude for all four biogenic particles (small and big POC, calcium carbonate and biogenic silica), but there still remain some shortcomings in their distributions (Fig. 5a–d).

– The total POC concentration in Fig. 5a is up to a factor of 4 underestimated in the deep oligotrophic ocean. In the upper 200 m of the ocean the model overestimates the observations, though at some points at the surface the POC concentration is underestimated (also Fig. 4a). In both the observations and the model the fraction of big POC varies from zero to 0.6 (Fig. 5b), but the spatial distributions are very different. More detailed results and discussion on the simulation of POC are to be found in Aumont et al. (2017).

– The model underestimates the $CaCO_3$ concentrations from $70°$ W to $25°$ W by a factor of 2 to 10, but east of that up to Africa the prediction lies close to or overestimates the observations. In the Canary Basin and up to Portugal (meridional transect at the right), the model reproduces the right order of magnitude, but it does not reproduce the correct profiles everywhere (Fig. 5c).

– Biogenic silica concentrations are generally reproduced but the model overestimates the higher concentrations observed along the western margin (Fig. 5d).

Finally, the modelled total lithogenic particle concentration at the surface shows, as expected, a close resemblance with dust deposition patterns (Fig. 2), and mostly compares well with observations (Fig. 4d). Only near the coasts of the USA and of Portugal the model strongly underestimates lithogenic particle concentrations. Concerning the deep ocean, our model captures quite nicely the general distribution at the GA03 transect (Figs. 5e and 5f). However, the concentrations near the western boundary are strongly underestimated, especially those of big lithogenic particles. Our model reproduces the observed fraction of big lithogenic particles of ∼0.1 to ∼0.3 in most of the deep ocean, but it underestimates the much larger fraction of big particles observed in the upper 200 m of the ocean (0.1–0.2 in the model versus 0.3–0.9 in the observations).

5.3 Thorium-230 and protactinium-231

$[^{230}Th]_D$ stands for the concentration of ^{230}Th that is in the dissolved phases. Similarly, $[^{230}Th]_S$ is the concentration of ^{230}Th that is adsorbed onto the small particles S, $[^{230}Th]_B$ the concentration adsorbed onto the big particles B; and similarly for ^{231}Pa. The modelled dissolved distributions of ^{230}Th and ^{231}Pa are underestimated at the surface (Figs 6 and 7). Below the surface, $[^{230}Th]_D$ is much better captured by the model (also Fig. 8a for the GA03 transect). However, observations show lower values near the bottom in the western part of the GA03 section and at $45°$ W, associated with more intense scavenging related to respectively nepheloid layers and manganese oxides from hydrothermal vents, which are not produced in the simulation (Figs 8a and 8b).

In the intermediate and deep waters of all the oceans, modelled $[^{231}Pa]_D$ is generally of the correct order of magnitude but is strongly overestimated below 2500 m depth (Fig. 8b), especially in the Pacific Ocean (Fig. 7c,d).

The concentrations of the adsorbed phases of ^{230}Th at GA03 are presented in Figs 8c and 8e. In the deep ocean, modelled $[^{230}Th]_S$ and $[^{230}Th]_B$ have lower values than $[^{230}Th]_D$, which is consistent with observations. Furthermore, compared with Dutay et al. (2009), we have notably improved $[^{230}Th]_S$ and $[^{230}Th]_B$. Other global modelling studies have not reported adsorbed concentrations of $[^{230}Th]_D$ and $[^{231}Pa]_D$. However, the model still underestimates the observed $[^{230}Th]_S$ and $[^{230}Th]_B$.

Contrarily, small ^{231}Pa particles are overestimated in our model below 1500 m at the GA03 transect (Fig. 8d). Big ^{231}Pa particles are simulated more realistically (Fig. 8f).
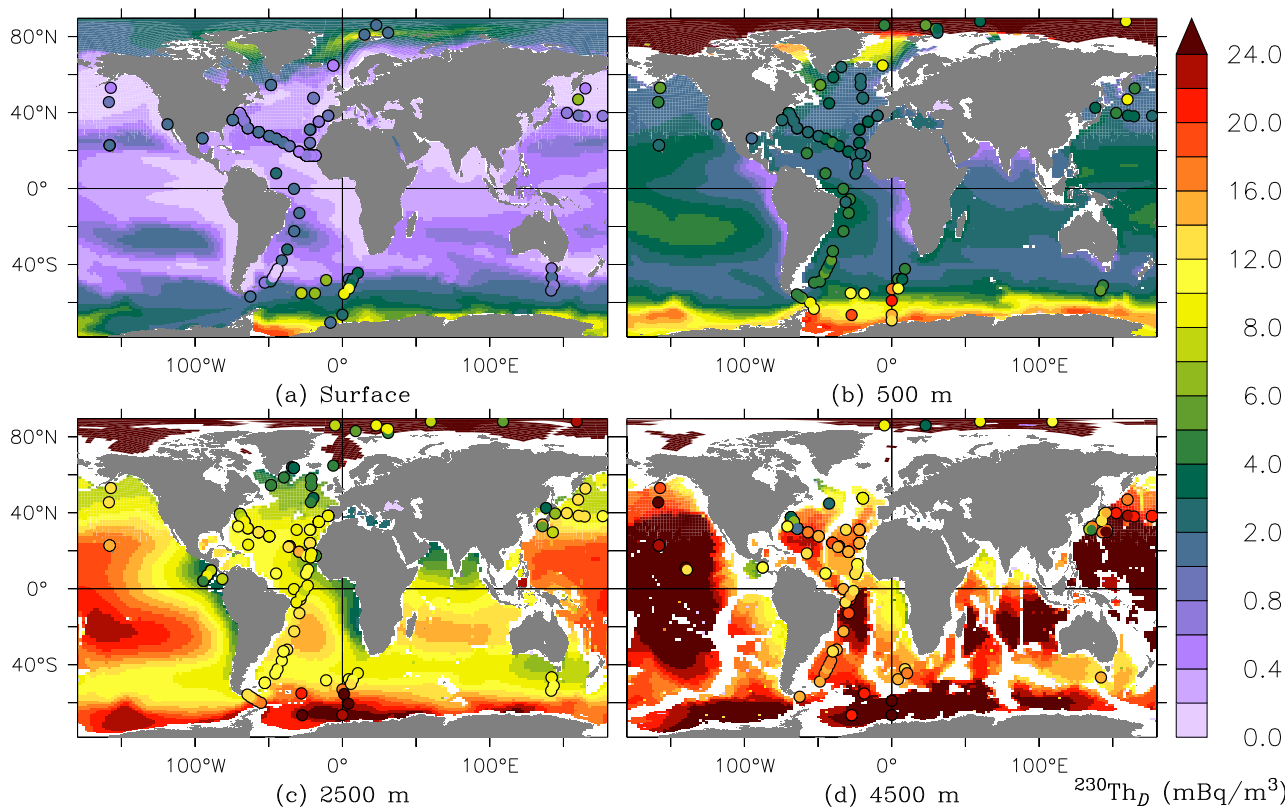
13

Figure 6. The dissolved thorium-230 activity at four depth levels $(\mathrm{mBq\,m^{-3}})$, observations as discs at the same colour scale.
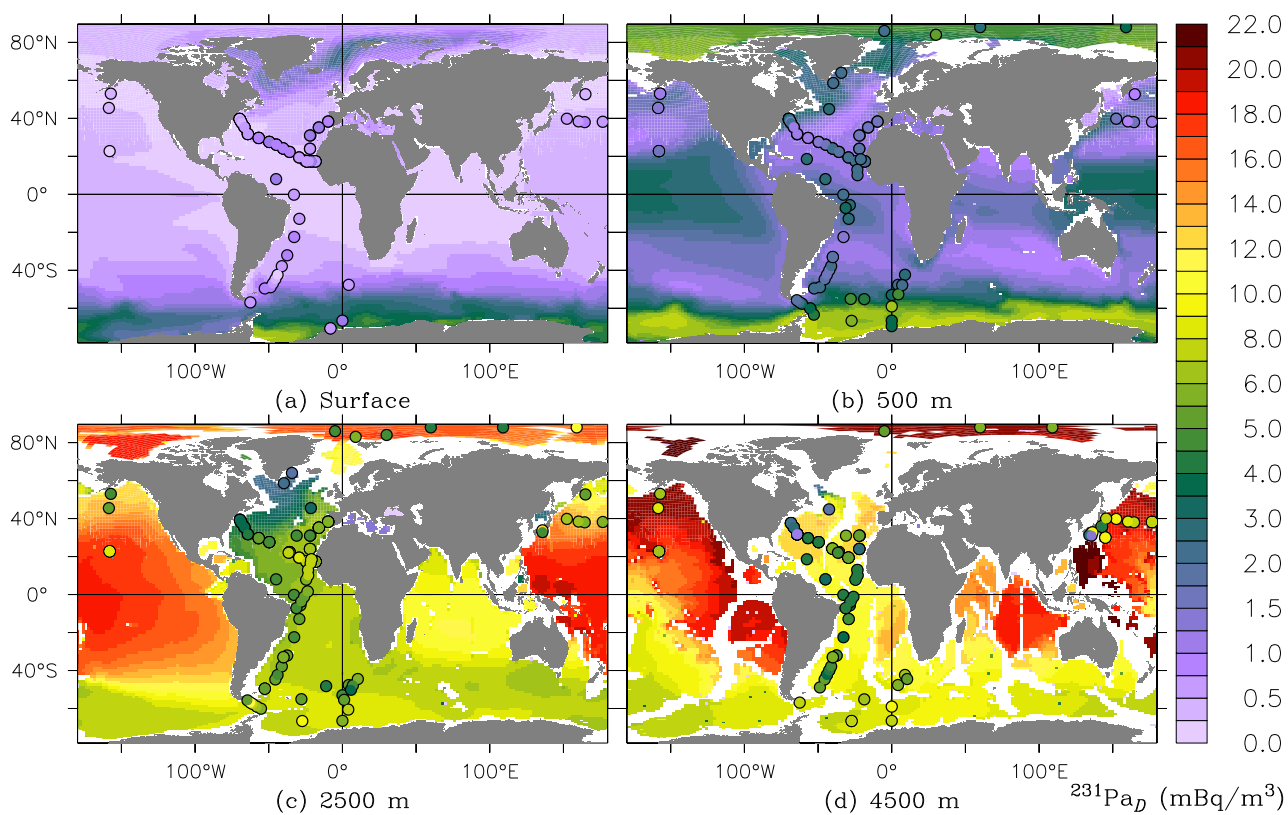
Figure 7. The dissolved protactinium-231 activity at four depth levels $(\mathrm{mBq\,m^{-3}})$, observations as coloured discs.
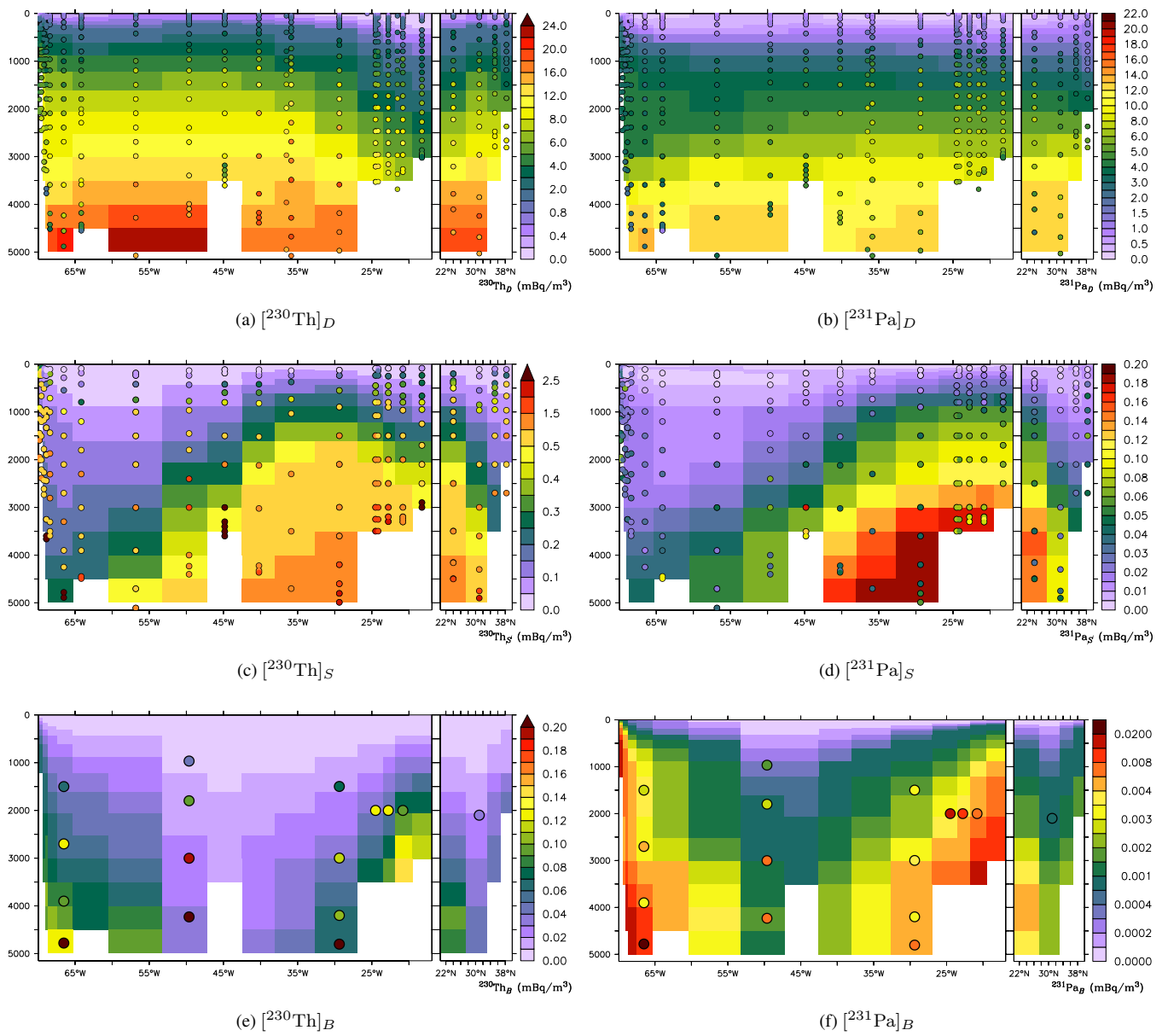
14

Figure 8. Radionuclide concentrations at the GA03 North Atlantic GEOTRACES transect. (a) and (b) display dissolved concentrations, (c) and (d) show the amounts on small particles, and (e) and (f) those on big particles. Everything is in $(\mathrm{mBq\,m^{-3}})$. Observations (Hayes et al., 2015a) are shown as coloured discs.
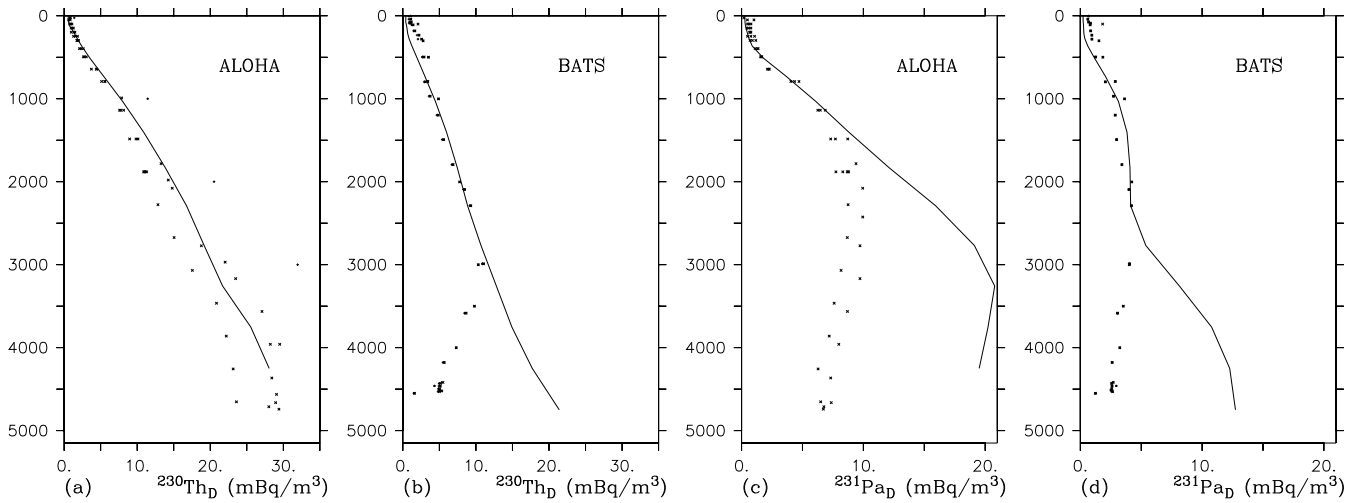
Figure 9. Radionuclide concentrations in $\mathrm{mBq\,m^{-3}}$ at stations in the North Pacific (around ALOHA station, $158°$ W, $23°$ N) and North Atlantic Oceans (around the BATS station, $64°$ W, $29°$ N). The left two panels (a,b) display $[^{230}\mathrm{Th}]_D$, and the right two (c,d) $[^{231}\mathrm{Pa}]_D$. The line is the model, and the speckles are the observations.

%		POC	$bSiO_2$	$CaCO_3$	Litho.
	global stock	35.5	10.5	3.1	50.8
$^{231}\mathrm{Pa}$	global particle flx	16.5	53.5	15.8	14.2
	North Atlantic flx	16.5	25.7	15.6	42.1
	Southern Ocean flx	5.6	93.7	0.6	0.1
	global stock	21.3	3.2	24.3	51.2
$^{230}\mathrm{Th}$	global particle flx	5.9	9.9	75.8	8.7
	North Atlantic flx	4.4	2.8	69.9	22.9
	Southern Ocean flx	9.7	71.1	18.9	0.3

Table 4. Relative global budget of $^{231}\mathrm{Pa}$ and $^{230}\mathrm{Th}$ in the different particulate phases ("global stock") in percentages. Globally averaged fluxes in the different phases ("global particle flx"), and the same but for the North Atlantic Ocean ($44–24°$ W, $25–56°$ N) and the Weddell Sea ($44–24°$ W, $76–63°$ S), all in percentages associated with each particle type.

To show clearly the change of $[^{230}\mathrm{Th}]_D$ and $[^{231}\mathrm{Pa}]_D$ with respect to depth, profiles are plotted in Fig. 9. $[^{230}\mathrm{Th}]_D$ is linearly increasing downwards, which matches with the observations, but below 3000 m near Bermuda, $[^{230}\mathrm{Th}]_D$ increases where it should decrease according to the observations. The shape of $[^{231}\mathrm{Pa}]_D$ matches that of the observations for the upper 2 km, below which our model overestimates the observations up to a factor four at the bottom of the ocean.

Globally in the model, only 0.3 % of the $^{231}\mathrm{Pa}$ sits in the particulate pool, and for $^{230}\mathrm{Th}$ this is 1.3 %. The rest is in the dissolved phase. Observational data show that more than 10 % of $^{230}\mathrm{Th}$ sits in the particulate pool, typically adsorbed onto small calcium carbonate particles. The underestimation by our model can be explained by the fact that we only have big, fast-sinking $CaCO_3$ particles in the model. Since the big particles are removed quickly, these cannot act as a realistic $CaCO_3$-associated $^{230}\mathrm{Th}$ pool without depleting $^{230}\mathrm{Th}$, and thus we underestimate the particulate radionuclide concentrations.

Table 4 gives two types of information on the adsorbed phases. One is the amount fraction of radionuclide adsorbed onto each particle ("global stock"). The other is the fraction of the flux carried by each type of particle (the other rows) for the global

16

ocean and two regions of the ocean. No distinction is made between big and small particles (they are summed). The stock and the flux are not necessarily the same, because the big particles have a different settling velocity from the small particles. For example, only for POC and lithogenic particles we have slowly sinking particles on which much of the radionuclides are adsorbed, whereas the big affinity of Pa to $bSiO_2$ mostly results in a strong export (and similarly for Th on $CaCO_3$).

Only a small amount of ^{230}Th is adsorbed onto biogenic silica (3 %), though the amount is larger for ^{231}Pa (11 %). However, the global settling flux of ^{231}Pa is primarily (54 %) determined by $bSiO_2$, particularly due to the high $bSiO_2$ fluxes in the Southern Ocean. Most of the modelled particulate ^{231}Pa and ^{230}Th is on the lithogenic particles, but lithogenic particles account for only 14 % of the ^{231}Pa flux and 9 % of the ^{230}Th flux. If lithogenic particles from nepheloid layers were included, this fraction would be higher. The flux contribution is also different from what is in the pools when we look at $CaCO_3$: only 24 % of the ^{230}Th is on calcium carbonate, but $CaCO_3$ is for 76 % responsible for the ^{230}Th export. These discrepancies between the contribution of each particle type to the flux and to the stock of radionuclides arise from the different settling velocities for each type of particle.

The relative contributions of the different particulate phases also vary for different regions of the ocean. Most notably, biogenic silica is for 94 % responsible for ^{231}Pa export in the Southern Ocean.

Compared to Dutay et al. (2009), the improvement of the ^{230}Th profile shape is improved, but only to a small extent due to the large improvement in the POC representation (Aumont et al., 2017), because the POC accounts for only 6 % of the vertical transport of ^{230}Th (Table 4). The improvement is mostly due to the addition of lithogenic dust particles and the adjusted $CaCO_3$ dissolution. Lithogenic particles and $CaCO_3$ hardly dissolve compared to POC and hence the downward particle flux remains constant, which is in agreement with 1-D models which show linear profiles of ^{230}Th$_D$ (e.g. Roy-Barman, 2009).

5.4 Sedimentation flux ratio of ^{231}Pa/^{230}Th

The sedimentation flux of the total adsorbed ^{231}Pa/^{230}Th ratio captures the general patterns of the upper sediment core ^{231}Pa/^{230}Th concentration ratio (Fig. 10). Consistent with observations, the model produces more elevated ^{231}Pa/^{230}Th ratio values in the Southern Ocean, the northern part of the Indian Ocean, the Pacific Ocean, and along the coastal upwelling regions. However, at some places, like much of the Southern Ocean, our model tends to overestimate the ratio derived from the sediment core observations.

6 Discussion

We succeed to generate the global patterns of the dissolved nuclide concentrations. However, we underestimate radioactivities in the surface waters (Figs 7a and 6a). This shortcoming may be due to the instant equilibration between the dissolved and the adsorbed phases (Henderson et al., 1999). After decay of U into ^{231}Pa and ^{230}Th, the tracers instantly adsorb onto the particles in the surface ocean and are exported from the mixed layer. **(addressing report reviewer 1:)** The equilibrium scavenging approach is commonly used in Pa/Th modelling (Siddall et al., 2005; Dutay et al., 2009; Gu and Liu, 2017). It allows the model to use partition coefficients that can be directly constrained by observations (although consensus values for these coefficients are still not available). A non-equilibrium approach to scavenging could be considered, but it would necessitate the use of adsorption and desorption coefficients (k_+, k_-) that are currently very poorly constrained by observations. Observations shows equilibrium partition coefficient K values varying (decreasing) with particle flux (Roy-Barman et al., 1996; Chase et al., 2002), and we apply these in our simulation (i.e. lower K values for large, fast sinking particles). This parameterisation produces effects that are similar to those of a non-equilibrium system for fast sinking particles.
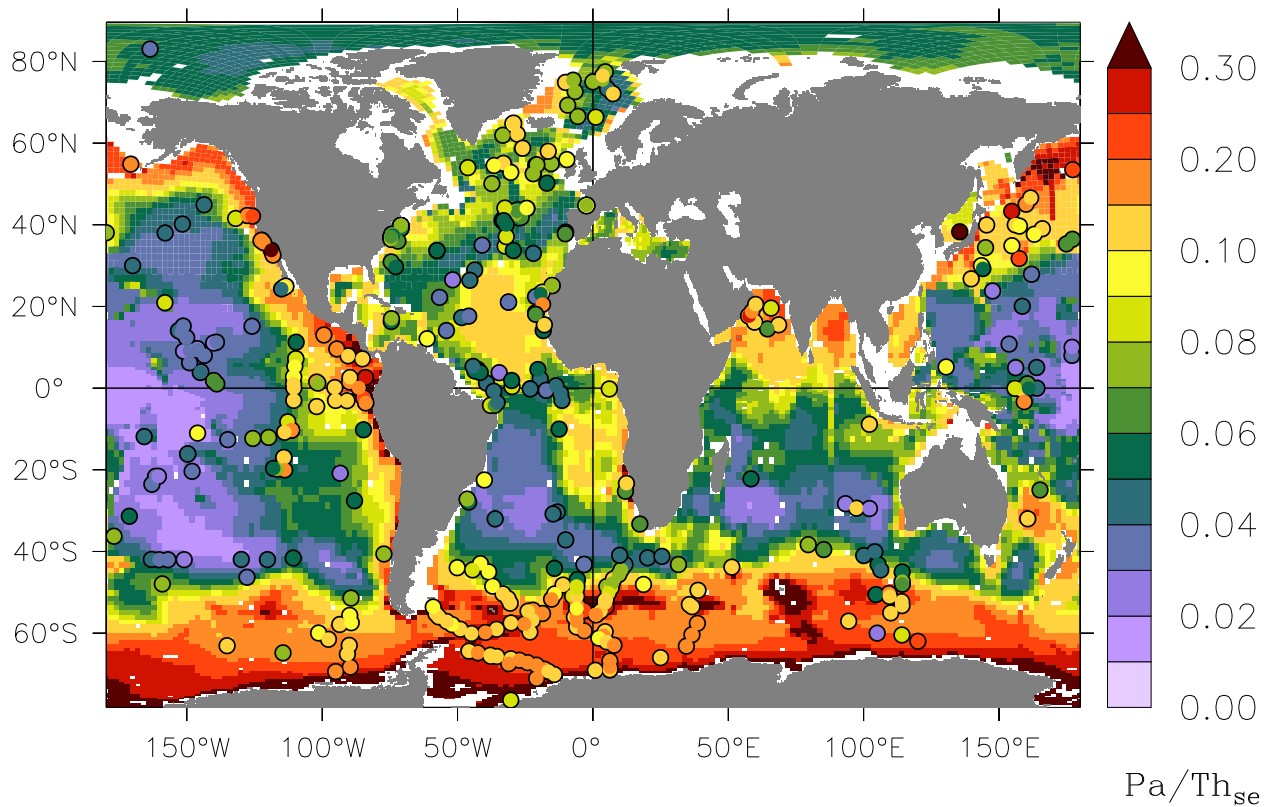
Figure 10. Sedimented ^{231}Pa/^{230}Th ratio. Top core measurements are presented as coloured discs. This is the standard simulation with $K_{\mathrm{Pa,bSiO_2}}/K_{\mathrm{Th,bSiO_2}} = 1$.

The reversible scavenging model uses partition coefficients K_{ij} for the different isotopes i and particles j. Many studies provide constraints on these coefficients but estimations vary strongly between different studies. Small particles are the most important scavengers and largely control the shape of the vertical profile of the tracers (Dutay et al., 2009). This also shows from sedimentary isotope records (Kretschmer et al., 2011). This is explained by the specific surface of small particles, which is larger than that of big particles. Therefore higher adsorption values are set for small particles compared to the bigger particles of the same type (namely, 5 times higher for POC, and 10 times higher for lithogenic particles). Compared to Dutay et al. (2009), we succeed to simulate the tracer concentrations using reasonable K values. This is due to improvement in (especially small) particle concentrations, which were generated with the lability parametrisation of POC (Aumont et al., 2017), the change of the $CaCO_3$ dissolution parameterisation, and the addition of lithogenic dust in our ocean model. This shows that at least two particle size classes are needed to simulate dissolved and particulate thorium and protactinium. Our model does not include small particles for all types (there are neither small $CaCO_3$ nor small $bSiO_2$ particles), even though we expect that better results can be reached upon adding those particles akin to small POC and small lithogenic particles.

On the GA02 transect, the correlation coefficient r of the data–model comparison of $[^{230}\mathrm{Th}]_D$ is 0.80, which is close to what was reported by Rempfer et al. (2017) in their simulations without bottom-boundary scavenging. On the GA03 section, $r = 0.78$. The correlation between model and observation are presented as scatter plots (supplementary document, Fig. S2).

There are a couple of interesting features to note on the comparisons. Firstly, the hydrothermal plume over the ridge results in ^{230}Th-poor seawater that is not reproduced by the model (red triangle in Fig. S2, GA03). Secondly, in the western boundary currents, very low dissolved ^{230}Th concentrations are observed due to strong nepheloids and/or possibly strong deep ventilation from the north. Neither of these processes are represented in the model. The lack of nepheloids has an especially large impact

18

on the GA02 transect, which has a relatively large number of measured profiles near the western boundary. Finally, between 32 and 20° W, the model tends to underestimate ^{230}Th. The value of the correlation index is not that important, because the goal of this study is not to give a perfect model–data comparison.

The provided dust flux does not distinguish between different dust particle sizes or types. Furthermore, dust deposition is uncertain; therefore we chose a dust flux that was available to the ORCA2 configuration of NEMO, and its biogeochemistry has been tested with that dust flux. However, given the significant impact of dust particles to thorium and protactinium concentrations, it would be useful to look into the effect of different dust deposition fields, and of using different dust particles size classes. A good model of ^{230}Th (and other isotopes of thorium) could even constrain the sporadic and uncertain dust deposition, but all this falls outside the scope of the present study.

Our model adds 80 % of the dust deposition into the fast-sinking lithogenic particle compartment (corresponding to $\varnothing >$ 51 μm), and only 20 % goes into the slowly-sinking compartment (corresponding to $\varnothing \leq 51$ μm). This is necessary to reproduce the distributions of lithogenic particles. However, airborne dust particles typically have a diameter of smaller than 20 μm (Knippertz and Stuut, 2014) and thus fall clearly within the small size class of less than 50 μm. Recently, there have been observations of aerosols larger than 20 μm (Van Does et al., 2016), but they do not reach far enough into the atmosphere above the open ocean to explain the high concentration of big lithogenic ocean particles. The explanation for the apparently required high fraction of big lithogenic particles is that smaller aerosols aggregate in the upper layers of the ocean. Moreover, though more hypothetically, there may be strong aggregation with biogenic particles just below the surface, below which the aggregates partly disaggregate again to result in the ~0.2 big lithogenic fraction in the deep ocean (Fig. 5f).

The underestimation of lithogenic particle concentrations at the western boundary is expected. The reason is that we only have dust deposition as a source of lithogenic particles, whereas nepheloid layers are not included. Nepheloid layers are at least locally an important source of lithogenic (and biogenic) particles (Lam et al., 2015). With the transport of ^{230}Th and ^{231}Pa through the western boundary current, a significant portion may be scavenged. Moreover, the periodic transport through the North Atlantic Gyre would result in lower $[^{230}\text{Th}]_D$ and $[^{231}\text{Pa}]_D$ throughout a large part of the North Atlantic Ocean, possibly improving the concentrations in the deep ocean. Therefore it will be useful to include nepheloid layers in the future. We have not done this so far, because we do not know how to model nepheloid layers. Except for trivial models, like the one of Rempfer et al. (2017) who forced an additional constant scavenging rate in the bottom box of the ocean model, no large-scale, prognostic nepheloid models have been developed.

We overestimate the radionuclide activity in the deep ocean, which is partly because we underestimate particle concentrations. Especially thorium adsorbs well onto lithogenic particles, which are underestimated in the West Atlantic Ocean (in the subsurface, west of about 43° W, Fig. 5e). Below 2 km depth, also POC is underestimated (Fig. 5a), even though the lability parameterisation improved the distribution by over an order of magnitude compared with previous versions of PISCES (Aumont et al., 2017). Small POC and the small lithogenic particles are the only small particles. Their underestimation in the deep ocean results in an overestimation of the radionuclides in the deep ocean. In our simulation, $[^{230}\text{Th}]_D$ is not that much overestimated (outside the Arctic Ocean). However, $[^{231}\text{Pa}]_D$ is strongly overestimated in the deep Pacific and Atlantic Oceans.

Clearly the model does not remove ^{231}Pa efficiently enough from the deep ocean (Figs 7c,d, 8b and Fig. 12–upper panel). Two likely reasons may be that the waters are not well ventilated in the model (older than in reality), or that there is not enough scavenging. The Atlantic OSF (Fig. S1) shows that the upper overturning cell is too shallow compared to observational studies, so there is too little ventilation. This means that ^{231}Pa can build up in the deep water due to sluggish ventilation of the physical model (Dutay et al., 2002; Biastoch et al., 2008). Moreover, the AABW has weakened in the deep North Atlantic Ocean near the high-$[^{231}\text{Pa}]_D$ region (Fig. 12–upper panel), which may also contribute to the high dissolved ^{231}Pa concentrations. The volume transport of the lower cell is about 6 Sv near 20° N and still 2 Sv near 40° N, which is in the order of magnitude of

what is reported by literature. Moreover, the relationship between sediment ^{231}Pa/^{230}Th at any given site and the overturning circulation is very complex. In general, any increased opportunity for protactinium to scavenge would increase the sedimentary ^{231}Pa/^{230}Th ratio. Specifically, a slower overturning would mean more Pa-scavenging and a higher ratio "just downstream of the sites of deep water formation" (Luo et al., 2010). Similarly, relatively stronger scavenging of Pa means such a higher sedimentary ^{231}Pa/^{230}Th, which we can see in the Southern Ocean and some coastal areas where the concentration of bSiO$_2$ is high. For these reasons, it is not obvious whether the weak overturning is large enough to explain the discrepancy between the modelled and the observed $[^{231}$Pa$]_D$.

In reality, there are many regions in the North Atlantic Ocean where there is sediment resuspension and where there are nepheloid layers (Gardner and Sullivan, 1981; Lam et al., 2015; Gardner et al., 2017). This is consistent with the fact that our model underestimates lithogenic particle concentrations in the West Atlantic Ocean (Fig. 5e). The additional lithogenic particles would scavenge more ^{231}Pa, resulting in lower $[^{231}$Pa$]_D$. Even though the enforced scavenging occurs near the floor and the western boundary of the ocean, a strong flux of water passes through the latter region, and is transported through the North Atlantic Gyre and through the rest of the Atlantic, diluting high tracer concentrations. The small POC and lithogenic particles are now underestimated in much of the deep ocean. They would contribute to lower radio tracer concentrations as well.

Rempfer et al. (2017) confirmed that an additional bottom sink affects ^{231}Pa and ^{230}Th significantly. They used a homogeneous extra scavenger in their model grid's bottom grid cell. Therefore, it would be worth testing if a realistic distribution of nepheloids would result in the right amount of scavenging.

Other particles may include resuspended (nepheloidal) biogenic particles but also manganese and iron hydroxides that are not included in our model, and a smaller class of calcium carbonate and biogenic silica particles in addition to the big-particle classes that are already in the model. Manganese oxides are available especially near hydrothermal vents, but also throughout the Pacific Ocean in low concentrations but much higher than in the Atlantic Ocean. Indeed, it has been argued that hydrothermal inputs may provide additional removal of ^{231}Pa and ^{230}Th (Lopez et al., 2015; Rutgers van der Loeff et al., 2016; German et al., 2016), and recently this is confirmed based on an analysis of the $\sim 10°$ S zonal GP16 transect in the Pacific Ocean (Pavia et al., 2017).

Finally, the particle concentration effect (Honeyman et al., 1988) should be taken into account in future models, particularly when regions of high particle concentration (e.g., ocean margins and nepheloid layers) are included.

6.1 Effect of scavenging by bSiO$_2$ on the ^{231}Pa/^{230}Th sedimentation flux ratio

Previously, it was suggested that Pa has a stronger affinity for bSiO$_2$ than Th has, i.e. $F_{\mathrm{Pa/Th,bSiO_2}} > 1$ (Henderson et al., 1999; Chase et al., 2002; Dutay et al., 2009). This "standard view" has weakened over time, so it is often accepted that $F_{\mathrm{Pa/Th,bSiO_2}} \gtrsim 1$ (Venchiarutti et al., 2011; Rutgers van der Loeff et al., 2016; Rempfer et al., 2017). We argue here that it is Th which has the stronger affinity to bSiO$_2$ ($F_{\mathrm{Pa/Th,bSiO_2}} < 1$), similarly (though not quite) like other particles.

Our standard simulation (with $F_{\mathrm{Pa/Th,bSiO_2}} = 1$) yields a too high sedimentation ^{231}Pa/^{230}Th ratio compared to top-core sediment observations below diatom-rich areas such as the Southern Ocean. In that region, bSiO$_2$ largely controls the particle scavenging of the two tracers, especially that of ^{231}Pa (Table 4). In order to estimate how strongly the model depends on the value of the bSiO$_2$ partition coefficient we performed a sensitivity experiment where we set $K_{\mathrm{Pa,bSiO_2}} = 0.4\,\mathrm{Mg\,g^{-1}}$ and $K_{\mathrm{Th,bSiO_2}} = 1.0\,\mathrm{Mg\,g^{-1}}$ such that now $F_{\mathrm{Pa/Th,bSiO_2}} = 0.4$ (Table 2).

As a result of this reduction, ^{231}Pa/^{230}Th flux ratio is not that much overestimated anymore in diatom-rich regions (Fig. 11). This result suggests that protactinium has a weaker affinity to biogenic silica than thorium has, though the fractionation is closer to 1 than that of other particle types (Table 2). This result is consistent with Geibert and Usbeck (2004) whose laboratory study
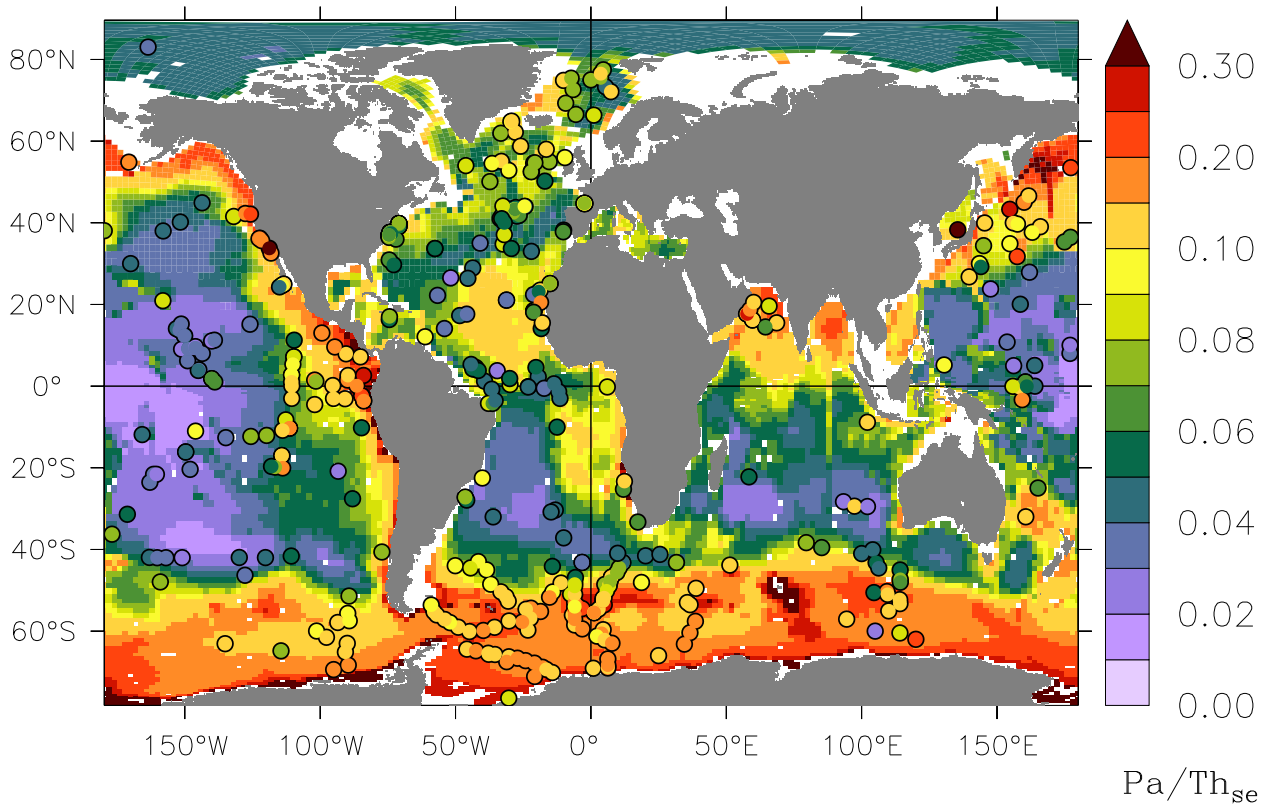
Figure 11. Sedimented ^{231}Pa/^{230}Th ratio when we set $K_{\mathrm{Pa,bSiO_2}}/K_{\mathrm{Th,bSiO_2}} = 0.4$. Top core measurements are presented as discs.

shows no definitive affinity of either Th or Pa to biogenic silica. Their average K values actually suggest that thorium has a higher affinity to biogenic silica.

We included lithogenic particles in the model from deposited dust. Since lithogenic particles have a strong relative affinity
25 with ^{230}Th ($K_{\mathrm{Pa,Lith}}/K_{\mathrm{Th,Lith}} = 0.2$) and since they are mostly prevalent in the (northern) Atlantic Ocean, much of the ^{230}Th is scavenged in the Atlantic Ocean. Calcium carbonate $K_{\mathrm{Pa}}/K_{\mathrm{Th}}$ is even smaller (0.024) and is present througout much of the Atlantic but not in the Southern Ocean. This leaves protactinium to be scavenged by biogenic silica in the Southern Ocean when it arrives there through the AMOC. Therefore, even with the smaller-than-conventional K ratio of 0.4 for biogenic silica, the familiar pattern of higher values of ^{231}Pa/^{230}Th in the Southern Ocean compared to most of the rest of the ocean is still
30 reproduced.

7   Conclusion

The purpose of this study is two-fold. Firstly, we set out a model of ^{231}Pa and ^{230}Th, complete with all the necessary particle improvements and additions. This includes an improved underlying particle dynamics from NEMO–PISCES as well as the addition of lithogenic particles. We have succeeded in this, and the model has been presented and implemented such that it can
35 be used for future studies, including for the modelling and validation of ocean circulation in the past.

Secondly, this model helps us to study the interplay between particle and water transport that control the ^{231}Pa and ^{230}Th profiles. We have shown that by improving on the particles, we improved the radionuclide distributions and fluxes, and we have shown to what extent the different particle phases drive the scavenging of ^{231}Pa and ^{230}Th.
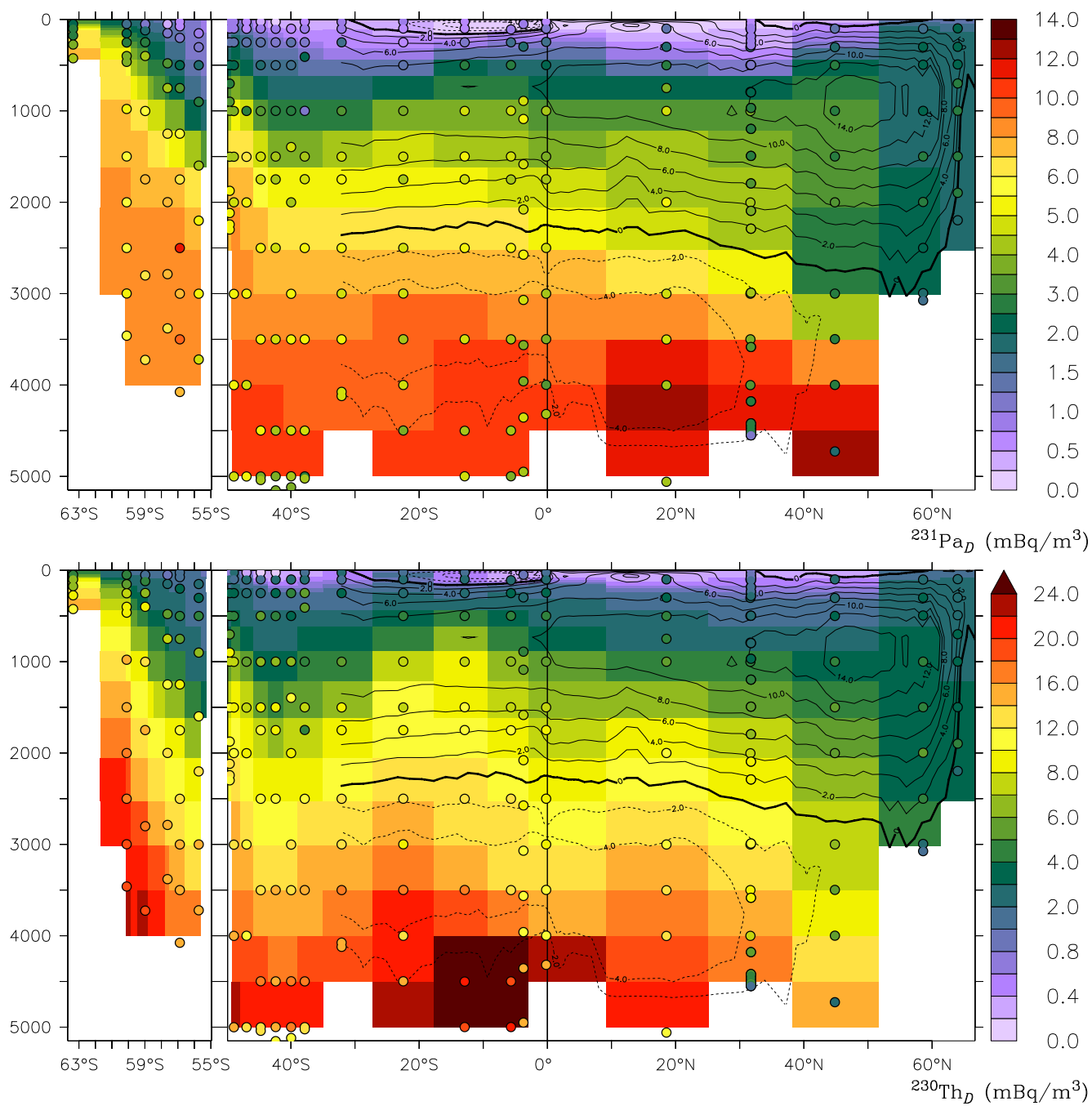
Figure 12. Dissolved protactinium-231 (upper panel), and thorium-230 (lower panel) at the Drake Passage (GIPY5) and West Atlantic (GA02) transects ($\mathrm{mBq\,m^{-3}}$); observations as discs. The Atlantic OSF, in Sv, is superimposed as a contour.

Dissolved ^{230}Th and ^{231}Pa concentrations are realistic in the intermediate-depth ocean (big improvement compared to Dutay et al. (2009)) but not in the upper part of the ocean. A finite equilibration time between the different radionuclide phases may help. The model can be extended to include this in future versions. It would double the number of adsorption/desorption parameters whose values should be determined through a careful literature and model sensitivity study.

In the deep ocean, the overestimation of ^{230}Th and ^{231}Pa in several regions is likely to be caused, at least partly, by missing nepheloid particles, but also manganese oxides (from hydrothermal vents) may improve the distributions. **(addressing report reviewer 2–Q3:)** Of course, reported adsorption/desorption parameters vary a lot between different studies, meaning that different, yet still plausible, combination of partition coefficients could better describe the available data. Additionally, the circulation may be too weak, not freshening AABW fast enough, and hence $[^{230}\mathrm{Th}]_D$ and $[^{231}\mathrm{Pa}]_D$ become too high near the ocean floor.

8 Code availability

The mathematical variable names correspond to the Fortran variable as described in Table S1 of the supplementary document. We encourage the reader to use our model, NEMO–ProThorP, to build extended and improved models. We have implemented two particle size classes of adsorbed nuclides, which is the minimum that is needed to produce good results. Instead of introducing a new tracer for every particle type, we only distinguish between "adsorbed onto small particles" and "adsorbed onto big particles", meaning that one is restricted, in this set-up, to include only particles that sink with the two respective settling speeds for scavenging. Whereas this is somewhat restricting, it is computationally efficient, and it restricts the number of degrees of freedom (which is usually a good thing in complex models). NEMO–ProThorP can also be used to set up a non-instant equilibration model (with a k_+ and k_-) that may yield better concentrations in the upper ocean.

The underlying model, NEMO 3.6 (svn revision 5283), can be downloaded from http://www.nemo-ocean.eu/ (Madec and the NEMO team, 2016) after creating a login. NEMO includes the biogeochemical PISCES model (Aumont et al., 2015). For reproducibility purposes, version 0.1.0 of the radionuclide and lithogenic particles model code can be obtained at https://dx.doi.org/10.5281/zenodo.1009065 (Van Hulten et al., 2017a). However, we plan to make this code available as part of the NEMO model.

The NEMO model is available under the CeCILL free software licence, modelled after the GNU GPL. The lithogenic particles and ^{230}Th/^{231}Pa-specific code is licensed under the same terms, or, at your option, under the GNU General Public License version 3 or higher. The authors do appreciate if, besides the legal adherence to copyleft and attribution, they are informed about the use of the code.

Finally, the most important raw model output files are available at PANGAEA (van Hulten et al., 2018). The Julia scripts used for the statistics are included in the supplementary document. This may be useful to reproduce the results or further inspect the model output.

9 Author contributions

The model and the simulations were designed by MvH, JCD and MRB. MvH implemented the code and performed the simulations. The manuscript was prepared by MvH with close collaboration and major contributions from MRB and JCD.

The authors declare that they have no conflict of interest.

25 References

Aken, H.M. van: GEOTRACES, the hydrography of the Western Atlantic Ocean, 2011.

Anderson, R.: Chemical tracers of particle transport, Treatise on Geochemistry, 6, 247–273, doi:10.1016/B0-08-043751-6/06111-9, 2003.

Anderson, R., Bacon, M., and Brewer, P.: Removal of ^{230}Th and ^{231}Pa from the open ocean, Earth Planet. Sc. Lett., 62, 7–23, doi:10.1016/0012-821X(83)90067-5, 1983.

Arsouze, T., Dutay, J.-C., Kageyama, M., Lacan, F., Alkama, R., Marti, O., and Jeandel, C.: A modeling sensitivity study of the influence of the Atlantic meridional overturning circulation on neodymium isotopic composition at the Last Glacial Maximum, Clim. of the Past, 4, 191–203, doi:10.5194/cp-4-191-2008, 2008.

Arsouze, T., Dutay, J.-C., Lacan, F., and Jeandel, C.: Reconstructing the Nd oceanic cycle using a coupled dynamical–biogeochemical model, Biogeosci., 6, 2829–2846, doi:10.5194/bg-6-2829-2009, 2009.

Aumont, O., Ethé, C., Tagliabue, A., Bopp, L., and Gehlen, M.: PISCES-v2: an ocean biogeochemical model for carbon and ecosystem studies, Geosci. Model Dev., 8, 2465–2513, doi:10.5194/gmd-8-2465-2015, 2015.

Aumont, O., Hulten, M.M.P. van, Roy-Barman, M., Dutay, J.-C., Éthé, C., and Gehlen, M.: Variable reactivity of particulate organic matter in a global ocean biogeochemical model, Biogeosci., 14, 2321–2341, doi:10.5194/bg-14-2321-2017, 2017.

Ayache, M., Dutay, J.-C., Arsouze, T., Révillon, S., Beuvier, J., and Jeandel, C.: High-resolution neodymium characterization along the Mediterranean margins and modelling of ε_{Nd} distribution in the Mediterranean basins, Biogeosci., 13, 5259–5276, doi:10.5194/bg-13-5259-2016, 2016.

Bacon, M. and Anderson, R.: Distribution of Thorium Isotopes Between Dissolved and Particulate Forms in The Deep Sea, J. Geophys. Res., 87, 2045–2056, doi:10.1029/JC087iC03p02045, 1982.

Biastoch, A., Böning, C. W., Getzlaff, J., Molines, J.-M., and Madec, G.: Causes of Interannual–Decadal Variability in the Meridional Overturning Circulation of the Midlatitude North Atlantic Ocean, J. Climate, 21, 6599–6615, doi:10.1175/2008JCLI2404.1, 2008.

Böhm, E., Lippold, J., Gutjahr, M., Frank, M., Blaser, P., Antz, B., Fohlmeister, J., Frank, N., Andersen, M. B., and Deininger, M.: Strong and deep Atlantic meridional overturning circulation during the last glacial cycle, Nature, 517, 73–76, doi:10.1038/nature14059, 2015.

Broecker, W. S., Andree, M., Bonani, G., Wolfli, W., Oeschger, H., Klas, M., Mix, A., and Curry, W.: Preliminary estimates for the radiocarbon age of deep water in the glacial ocean, Paleoceanography, 3, 659–669, doi:10.1029/PA003i006p00659, 1988.

Brun-Cottan, J. C., Auger, R., Lambere, C. E., and Chesselet, R.: Sources and transport of suspended calcites in Pacific Deep Water, J. Mar. Res., 49, 543–564, doi:10.1357/002224091784995837, 1991.

Burckel, P., Waelbroeck, C., Gherardi, J. M., Pichat, S., Arz, H., Lippold, J., Dokken, T., and Thil, F.: Atlantic Ocean circulation changes preceded millennial tropical South America rainfall events during the last glacial, Geophys. Res. Lett., 42, 411–418, doi:10.1002/2014GL062512, 2014GL062512, 2015.

Burckel, P., Waelbroeck, C., Luo, Y., Roche, D. M., Pichat, S., Jaccard, S. L., Gherardi, J., Govin, A., Lippold, J., and Thil, F.: Changes in the geometry and strength of the Atlantic meridional overturning circulation during the last glacial (20–50 ka), Clim. of the Past, 12, 2061–2075, doi:10.5194/cp-12-2061-2016, 2016.

Chase, Z., Anderson, R., Fleisher, M., and Kubik, P.: The influence of particle composition and particle flux on scavenging of Th, Pa and Be in the ocean, Earth Planet. Sc. Lett., 204, 215–229, 2002.

Chuang, C.-Y., Santschi, P. H., Jiang, Y., Ho, Y.-F., Quigg, A., Guo, L., Ayranov, M., and Schumann, D.: Important role of biomolecules from diatoms in the scavenging of particle-reactive radionuclides of thorium, protactinium, lead, polonium, and beryllium in the ocean: A case study with Phaeodactylum tricornutum, Limnol. Oceanogr., 59, 1256–1266, doi:10.4319/lo.2014.59.4.1256, 2014.

Cunningham, S., Kanzow, T., Rayner, D., Baringer, M., Johns, W., Marotzke, J., Longworth, H., Grant, E., Hirschi, J.-M., Beal, L., et al.: Temporal variability of the Atlantic meridional overturning circulation at 26.5° N, Science, 317, 935–938, doi:10.1126/science.1141304, 2007.

Deng, F., Thomas, A. L., Rijkenberg, M. J., and Henderson, G. M.: Controls on seawater ^{231}Pa, ^{230}Th and ^{232}Th concentrations along the flow paths of deep waters in the Southwest Atlantic, Earth Planet. Sc. Lett., 390, 93–102, doi:10.1016/j.epsl.2013.12.038, 2014.

25   Does, M. van, Korte, L. F., Munday, C. I., Brummer, G.-J. A., and Stuut, J.-B. W.: Particle size traces modern Saharan dust transport and deposition across the equatorial North Atlantic, Atmospheric Chem. Phys. Discuss., 16, 13 697–13 710, doi:10.5194/acp-16-13697-2016, 2016.

Donohue, K. A., Tracey, K. L., Watts, D. R., Chidichimo, M. P., and Chereskin, T. K.: Mean Antarctic Circumpolar Current transport measured in Drake Passage, Geophys. Res. Lett., 43, 11,760–11,767, doi:10.1002/2016GL070319, 2016.

30   Druffel, E., Williams, P., Bauer, J., and Ertel, J.: Cycling of dissolved and particulate organic matter in the open ocean, J. Geophys. Res., 97, 15 639–15 659, doi:10.1029/92JC01511, http://escholarship.org/uc/item/5pt6w45d, 1992.

Dunk, R., Mills, R., and Jenkins, W.: A reevaluation of the oceanic uranium budget for the Holocene, Chemical Geology, 190, 45–67, doi:10.1016/S0009-2541(02)00110-9, geochemistry of Crustal Fluids-Fluids in the Crust and Chemical Fluxes at the Earth's Surface, 2002.

35   Dutay, J., Bullister, J., Doney, S., Orr, J., Najjar, R., Caldeira, K., Campin, J., Drange, H., Follows, M., Gao, Y., et al.: Evaluation of ocean model ventilation with CFC-11, Ocean Model., 4, 89–120, doi:10.1016/S1463-5003(01)00013-0, 2002.

Dutay, J., Lacan, F., Roy-Barman, M., and Bopp, L.: Influence of particle size and type on ^{231}Pa and ^{230}Th simulation with a global coupled biogeochemical-ocean general circulation model: A first approach, Geochem. Geophy. Geosy., 10, Q01 011, doi:10.1029/2008GC002291, 2009.

40   Dutay, J., Tagliabue, A., Kriest, I., and Hulten, M.M.P. van: Modelling the role of marine particle on large scale ^{231}Pa, ^{230}Th, Iron and Aluminium distributions, Prog. Oceanogr., 133, 66–72, doi:10.1016/j.pocean.2015.01.010, 2015.

Gardner, W. D. and Sullivan, L. G.: Benthic storms, Science, 213, 329–331, 1981.

Gardner, W. D., Tucholke, B. E., Richardson, M. J., and Biscaye, P. E.: Benthic storms, nepheloid layers, and linkage with upper ocean dynamics in the western North Atlantic, Mar. Geol., 385, 304–327, doi:10.1016/j.margeo.2016.12.012, 2017.

Gehlen, M., Gangstø, R., Schneider, B., Bopp, L., Aumont, O., and Ethé, C.: The fate of pelagic CaCO$_3$ production in a high CO$_2$ ocean: a model study, Biogeosci., 4, 505–519, doi:10.5194/bg-4-505-2007, 2007.

Geibert, W. and Usbeck, R.: Adsorption of thorium and protactinium onto different particle types: experimental findings, Geochim. Cos-
5   mochim. Ac., 68, 1489–1501, doi:10.1016/j.gca.2003.10.011, 2004.

Gent, P. and McWilliams, J.: Isopycnal mixing in ocean circulation models, J. Phys. Oceanogr., 20, 150–155, doi:10.1175/1520-0485(1990)020<0150:IMIOCM>2.0.CO;2, 1990.

Gent, P., Willebrand, J., McDougall, T., and McWilliams, J.: Parameterizing eddy-induced tracer transports in ocean circulation models, J. Phys. Oceanogr., 25, 463–474, doi:10.1175/1520-0485(1995)025<0463:PEITTI>2.0.CO;2, 1995.

10   German, C. R., Casciotti, K. A., Dutay, J.-C., Heimbürger, L. E., Jenkins, W. J., Measures, C. I., Mills, R. A., Obata, H., Schlitzer, R., Tagliabue, A., Turner, D. R., and Whitby, H.: Hydrothermal impacts on trace element and isotope ocean biogeochemistry, Phil. Trans. R. Soc. A, 374, doi:10.1098/rsta.2016.0035, part of the special issue *Biological and climatic impacts of ocean trace element chemistry*, 2016.

Gu, S. and Liu, Z.: ^{231}Pa and ^{230}Th in the ocean model of the Community Earth System Model (CESM1.3), Geosci. Model Dev., 10, 4723–4742, doi:10.5194/gmd-10-4723-2017, 2017.

15   Hauglustaine, D., Hourdin, F., Jourdain, L., Filiberti, M.-A., Walters, S., Lamarque, J.-F., and Holland, E.: Interactive chemistry in the Labora-toire de Météorologie Dynamique general circulation model, J. Geophys. Res.: Atmospheres, 109, D04 314, doi:10.1029/2003JD003957, 2004.

Hayes, C. T., Anderson, R. F., Fleisher, M. Q., Serno, S., Winckler, G., and Gersonde, R.: Quantifying lithogenic inputs to the North Pacific Ocean using the long-lived thorium isotopes, Earth Planet. Sc. Lett., 383, 16–25, doi:10.1016/j.epsl.2013.09.025, 2013.

20   Hayes, C. T., Anderson, R. F., Fleisher, M. Q., Serno, S., Winckler, G., and Gersonde, R.: Biogeography in ^{231}Pa/^{230}Th ratios and a balanced ^{231}Pa budget for the Pacific Ocean, Earth Planet. Sc. Lett., 391, 307–318, doi:10.1016/j.epsl.2014.02.001, 2014.

Hayes, C. T., Anderson, R. F., Fleisher, M. Q., Huang, K.-F., Robinson, L. F., Lu, Y., Cheng, H., Edwards, R. L., and Moran, S. B.: ^{230}Th and ^{231}Pa on GEOTRACES GA03, the US GEOTRACES North Atlantic transect, and implications for modern and paleoceanographic chemical fluxes, Deep-Sea Res. Pt II, 116, 29–41, doi:10.1016/j.dsr2.2014.07.007, 2015a.

25   Hayes, C. T., Anderson, R. F., Fleisher, M. Q., Lam, P. J., Ohnemus, D. C., Huang, K.-F., Robinson, L. F., Lu, Y., Cheng, H., Edwards, R. L., and Moran, S. B.: Intensity of Th and Pa scavenging partitioned by particle chemistry in the North Atlantic Ocean, Mar. Chem., 170, 49–60, doi:10.1016/j.marchem.2015.01.006, 2015b.

Heinze, C., Gehlen, M., and Land, C.: On the potential of [230]Th, [231]Pa, and [10]Be for marine rain ratio determinations: A modeling study, Global Biogeochem. Cy., 20, doi:10.1029/2005GB002595, gB2018, 2006.

30   Henderson, G. M. and Anderson, R. F.: The U-series Toolbox for Paleoceanography, Reviews in Mineralogy and Geochemistry, 52, 493–531, doi:10.2113/0520493, http://rimg.geoscienceworld.org/content/52/1/493.full.pdf+html, 2003.

Henderson, G. M., Heinze, C., Anderson, R. F., and Winguth, A. M.: Global distribution of the [230]Th flux to ocean sediments constrained by GCM modelling, Deep-Sea Res. Pt I, 46, 1861–1893, doi:10.1016/S0967-0637(99)00030-8, observational data: http://climotope.earth.ox.ac.uk/data_compilations/holocene_231pa230th_dataset_notes_and_references, 1999.

35   Honeyman, B. D. and Santschi, P. H.: Metals in aquatic systems, Environmental science & technology, 22, 862–871, doi:10.1021/es00173a002, 1988.

Honeyman, B. D., Balistrieri, L. S., and Murray, J. W.: Oceanic trace metal scavenging, Deep-Sea Res. Pt A, 35, 227–246, doi:10.1016/0198-0149(88)90038-6, 1988.

Hulten, M.M.P. van: ComPlot: Comparison Plotter to visually evaluate ocean model simulations, The Journal of Open Source Software, 2, doi:10.21105/joss.00368, 2017.

40   Hulten, M.M.P. van, Sterl, A., Dutay, J.-C., Tagliabue, A., Gehlen, M., Baar, H.J.W. de, and Middag, R.: Aluminium in an ocean general circulation model compared with the West Atlantic GEOTRACES cruises, J. Mar. Syst., 126, 3–23, doi:10.1016/j.jmarsys.2012.05.005, traces and Tracers: Selected papers from the Joint Liège Colloquium on Ocean Dynamics – Bonus-GoodHope – GEOTRACES meeting, 2013.

45   Hulten, M.M.P. van, Sterl, A., Middag, R., de Baar, H., Gehlen, M., Dutay, J.-C., and Tagliabue, A.: On the effects of circulation, sediment resuspension and biological incorporation by diatoms in an ocean model of aluminium, Biogeosci., 11, 3757–3779, doi:10.5194/bg-11-3757-2014, 2014.

Hulten, M.M.P. van, Dutay, J., and Roy-Barman, M.: Ocean model of Protactinium, Thorium and Particles (ProThorP), doi:10.5281/zenodo.1009064, software package, 2017a.

5   Hulten, M.M.P. van, Middag, R., Dutay, J.-C., de Baar, H., Roy-Barman, M., Gehlen, M., Tagliabue, A., and Sterl, A.: Manganese in the West Atlantic Ocean in the context of the first global ocean circulation model of manganese, Biogeosci., pp. 1123–1152, doi:10.5194/bg-14-1123-2017, 2017b.

Johnson, G. C.: Quantifying Antarctic Bottom Water and North Atlantic Deep Water volumes, J. Geophys. Res.: Oceans, 113, doi:10.1029/2007JC004477, c05027, 2008.

10   Kalnay, E., Kanamitsu, M., Kistler, R., Collins, W., Deaven, D., Gandin, L., Iredell, M., Saha, S., White, G., Woollen, J., Zhu, Y., Leetmaa, A., Reynolds, R., Chelliah, M., Ebisuzaki, W., Higgins, W., Janowiak, J., Mo, K. C., Ropelewski, C., Wang, J., Jenne, R., and Joseph, D.: The NCEP/NCAR 40-Year Reanalysis Project, Bull. Amer. Meteor. Soc., 77, 437–471, doi:10.1175/1520-0477(1996)077<0437:TNYRP>2.0.CO;2, 1996.

Keir, R. S.: The dissolution kinetics of biogenic calcium carbonates in seawater, Geochim. Cosmochim. Ac., 44, 241–252, doi:10.1016/0016-7037(80)90135-0, 1980.

15   Kistler, R., Collins, W., Saha, S., White, G., Woollen, J., Kalnay, E., Chelliah, M., Ebisuzaki, W., Kanamitsu, M., Kousky, V., et al.: The NCEP-NCAR 50-year reanalysis, Bull. Am. Meteorol. Soc., 82, 247–267, doi:10.1175/1520-0477(2001)082<0247:TNNYRM>2.3.CO;2, 2001.

Knippertz, P. and Stuut, J.-B. W.: Mineral Dust, Springer Dordrecht Heidelberg New York London, doi:10.1007/978-94-017-8978-3, 2014.

20   Kretschmer, S., Geibert, W., van der Loeff, M. M. R., Schnabel, C., Xu, S., and Mollenhauer, G.: Fractionation of [230]Th, [231]Pa, and [10]Be induced by particle size and composition within an opal-rich sediment of the Atlantic Southern Ocean, Geochim. Cosmochim. Ac., 75, 6971–6987, doi:10.1016/j.gca.2011.09.012, 2011.

Ku, T.-L., Knauss, K. G., and Mathieu, G. G.: Uranium in open ocean: concentration and isotopic composition, Deep-Sea Res., 24, 1005–1017, doi:10.1016/0146-6291(77)90571-9, 1977.

25 Lam, P. J., Doney, S. C., and Bishop, J. K. B.: The dynamic ocean biological pump, Global Biogeochem. Cy., 25, GB3009, doi:10.1029/2010GB003868, 2011.

Lam, P. J., Ohnemus, D. C., and Auro, M. E.: Size-fractionated major particle composition and concentrations from the US GEOTRACES North Atlantic zonal transect, Deep-Sea Res. Pt II, 116, 303–320, doi:10.1016/j.dsr2.2014.11.020, 2015.

Lopez, G. I., Marcantonio, F., Lyle, M., and Lynch-Stieglitz, J.: Dissolved and particulate ^{230}Th–^{232}Th in the Central Equatorial Pa-
30 cific Ocean: Evidence for far-field transport of the East Pacific Rise hydrothermal plume, Earth Planet. Sc. Lett., 431, 87–95, doi:10.1016/j.epsl.2015.09.019, 2015.

Luo, Y., Francois, R., and Allen, S. E.: Sediment ^{231}Pa/^{230}Th as a recorder of the rate of the Atlantic meridional overturning circulation: insights from a 2-D model, Ocean Sci., 6, 381–400, doi:10.5194/os-6-381-2010, 2010.

Madec, G.: NEMO ocean engine, Note du Pole de Modélisation, Institut Pierre-Simon Laplace, 2008.

35 Madec, G. and Imbard, M.: A global ocean mesh to overcome the North Pole singularity, Clim. Dyn., 12, 381–388, doi:10.1007/BF00211684, 1996.

Madec, G. and the NEMO team: NEMO ocean engine, Note du Pole de Modélisation, Institut Pierre-Simon Laplace, 2016.

Marchal, O., François, R., Stocker, T., and Joos, F.: Ocean thermohaline circulation and sedimentary ^{231}Pa/^{230}Th ratio, Paleoceanography, 15, 625–641, doi:10.1029/2000PA000496, 2000.

40 Marchal, O., François, R., and Scholten, J.: Contribution of ^{230}Th measurements to the estimation of the abyssal circulation, Deep-Sea Res. Pt I, 54, 557–585, doi:10.1016/j.dsr.2007.01.002, there is an erratum (for figure labels), 2007.

Mawji, E., Schlitzer, R., Dodas, E. M., Abadie, C., Abouchami, W., Anderson, R. F., Baars, O., Bakker, K., Baskaran, M., Bates, N. R., Bluhm, K., Bowie, A., Bown, J., Boye, M., Boyle, E. A., Branellec, P., Bruland, K. W., Brzezinski, M. A., Bucciarelli, E., Buesseler, K., Butler, E., Cai, P., Cardinal, D., Casciotti, K., Chaves, J., Cheng, H., Chever, F., Church, T. M., Colman, A. S., Conway, T. M., Croot,
45 P. L., Cutter, G. A., de Baar, H. J., de Souza, G. F., Dehairs, F., Deng, F., Dieu, H. T., Dulaquais, G., Echegoyen-Sanz, Y., Edwards, R. L., Fahrbach, E., Fitzsimmons, J., Fleisher, M., Frank, M., Friedrich, J., Fripiat, F., Galer, S. J., Gamo, T., Solsona, E. G., Gerringa, L. J., Godoy, J. M., Gonzalez, S., Grossteffan, E., Hatta, M., Hayes, C. T., Heller, M. I., Henderson, G., Huang, K.-F., Jeandel, C., Jenkins, W. J., John, S., Kenna, T. C., Klunder, M., Kretschmer, S., Kumamoto, Y., Laan, P., Labatut, M., Lacan, F., Lam, P. J., Lannuzel, D., le Moigne, F., Lechtenfeld, O. J., Lohan, M. C., Lu, Y., Masqué, P., McClain, C. R., Measures, C., Middag, R., Moffett, J., Navidad, A.,
5   Nishioka, J., Noble, A., Obata, H., Ohnemus, D. C., Owens, S., Planchon, F., Pradoux, C., Puigcorbé, V., Quay, P., Radic, A., Rehkämper, M., Remenyi, T., Rijkenberg, M. J., Rintoul, S., Robinson, L. F., Roeske, T., Rosenberg, M., Rutgers van der Loeff, M., Ryabenko, E., Saito, M. A., Roshan, S., Salt, L., Sarthou, G., Schauer, U., Scott, P., Sedwick, P. N., Sha, L., Shiller, A. M., Sigman, D. M., Smethie, W., Smith, G. J., Sohrin, Y., Speich, S., Stichel, T., Stutsman, J., Swift, J. H., Tagliabue, A., Thomas, A., Tsunogai, U., Twining, B. S., van Aken, H. M., van Heuven, S., van Ooijen, J., van Weerlee, E., Venchiarutti, C., Voelker, A. H., Wake, B., Warner, M. J., Woodward,
10 E. M. S., Wu, J., Wyatt, N., Yoshikawa, H., Zheng, X.-Y., Xue, Z., Zieringer, M., and Zimmer, L. A.: The GEOTRACES Intermediate Data Product 2014, Mar. Chem., 177, 1–8, doi:10.1016/j.marchem.2015.04.005, the complete dataset can be downloaded from http://www.geotraces.org/dp/idp2014., 2015.

Mouchet, A.: The Ocean Bomb Radiocarbon Inventory Revisited, Radiocarbon, 55, 1580–1594, doi:10.1017/S0033822200048505, 2013.

Murray, R.: Explicit generation of orthogonal grids for ocean models, J. Comput. Phys., 126, 251–273, doi:10.1006/jcph.1996.0136, 1996.

15 Owens, S., Buesseler, K., and Sims, K.: Re-evaluating the ^{238}U-salinity relationship in seawater: Implications for the ^{238}U–^{234}Th disequilibrium method, Mar. Chem., 127, 31–39, doi:10.1016/j.marchem.2011.07.005, 2011.

Pavia, F., Anderson, R., Vivancos, S., Fleisher, M., Lam, P., Lu, Y., Cheng, H., Zhang, P., and Edwards, R. L.: Intense hydrothermal scavenging of ^{230}Th and ^{231}Pa in the deep Southeast Pacific, Mar. Chem., doi:10.1016/j.marchem.2017.08.003, 2017.

Piotrowski, A. M., Goldstein, S. L., Hemming, S. R., and Fairbanks, R. G.: Temporal Relationships of Carbon Cycling and Ocean Circulation
20 at Glacial Boundaries, Science, 307, 1933–1938, doi:10.1126/science.1104883, 2005.

Rayner, D., Hirschi, J.-M., Kanzow, T., Johns, W., Wright, P., Frajka-Williams, E., Bryden, H., Meinen, C., Baringer, M., Marotzke, J., et al.: Monitoring the Atlantic meridional overturning circulation, Deep-Sea Res. Pt II, 58, 1744–1753, doi:10.1016/j.dsr2.2010.10.056, 2011.

Rempfer, J., Stocker, T. F., Joos, F., Lippold, J., and Jaccard, S. L.: New insights into cycling of ^{231}Pa and ^{230}Th in the Atlantic Ocean, Earth Planet. Sc. Lett., 468, 27–37, doi:10.1016/j.epsl.2017.03.027, units of ^{231}Pa and ^{230}Th activities/concentrations are dpm/m^3, 2017.

28

25 Rosenthal, Y., Boyle, E. A., and Labeyrie, L.: Last Glacial Maximum paleochemistry and deepwater circulation in the Southern Ocean: Evidence from foraminiferal cadmium, Paleoceanography, 12, 787–796, doi:10.1029/97PA02508, 1997.

Roy-Barman, M.: Modelling the effect of boundary scavenging on Thorium and Protactinium profiles in the ocean, Biogeosci., 6, 3091–3107, doi:10.5194/bg-6-3091-2009, 2009.

Roy-Barman, M., Chen, J., and Wasserburg, G.: ^{230}Th–^{232}Th systematics in the central Pacific Ocean: The sources and the fates of thorium,
30 Earth Planet. Sc. Lett., 139, 351–363, doi:10.1016/0012-821X(96)00017-9, 1996.

Rutgers van der Loeff, M., Venchiarutti, C., Stimac, I., Ooijen, Jan van, Huhn, O., Rohardt, G., and Strass, V.: Meridional circulation across the Antarctic Circumpolar Current serves as a double ^{231}Pa and ^{230}Th trap, Earth Planet. Sc. Lett., 455, 73–84, doi:10.1016/j.epsl.2016.07.027, 2016.

Siddall, M., Henderson, G., Edwards, N., Frank, M., Müller, S., Stocker, T., and Joos, F.: ^{231}Pa/^{230}Th fractionation by ocean transport,
35 biogenic particle flux and particle type, Earth Planet. Sc. Lett., 237, 135–155, doi:10.1016/j.epsl.2005.05.031, 2005.

Siddall, M., Stocker, T. F., Henderson, G. M., Joos, F., Frank, M., Edwards, N. R., Ritz, S. P., and Müller, S. A.: Modeling the relationship between ^{231}Pa/^{230}Th distribution in North Atlantic sediment and Atlantic meridional overturning circulation, Paleoceanography, 22, doi:10.1029/2006PA001358, 2007.

Sinha, B., Smeed, D., McCarthy, G., Moat, B., Josey, S., Hirschi, J.-M., Frajka-Williams, E., Blaker, A., Rayner, D., and Madec, G.: The accuracy of estimates of the overturning circulation from basin-wide mooring arrays, Prog. Oceanogr., 160, 101–123, doi:10.1016/j.pocean.2017.12.001, 2018.

715 Subhas, A. V., Rollins, N. E., Berelson, W. M., Dong, S., Erez, J., and Adkins, J. F.: A novel determination of calcite dissolution kinetics in seawater, Geochim. Cosmochim. Ac., 170, 51–68, doi:10.1016/j.gca.2015.08.011, 2015.

Tagliabue, A., Bopp, L., Roche, D., Bouttes, N., Dutay, J., Alkama, R., Kageyama, M., Michel, E., Paillard, D., et al.: Quantifying the roles of ocean circulation and biogeochemistry in governing ocean carbon-13 and atmospheric carbon dioxide at the last glacial maximum, Clim. of the Past, 5, 695–706, 2009.

720 Tagliabue, A., Bopp, L., Dutay, J.-C., Bowie, A., Chever, F., Jean-Baptiste, P., Bucciarelli, E., Lannuzel, D., Remenyi, T., Sarthou, G., Aumont, O., Gehlen, M., and Jeandel, C.: Hydrothermal contribution to the oceanic dissolved iron inventory, Nat. Geosci., 3, 252–256, doi:10.1038/NGEO818, 2010.

Talley, L., Reid, J., and Robbins, P.: Data-based meridional overturning streamfunctions for the global ocean, J. Climate, 16, 3213–3226, doi:10.1175/1520-0442(2003)016<3213:DMOSFT>2.0.CO;2, 2003.

725 Timmermann, R., Goosse, H., Madec, G., Fichefet, T., Ethé, C., and Dulière, V.: On the representation of high latitude processes in the ORCA-LIM global coupled sea ice–ocean model, Ocean Model., 8, 175–201, doi:10.1016/j.ocemod.2003.12.009, 2005.

van Hulten, M. M., Dutay, J.-C., and Roy-Barman, M.: Output from a global scavenging and circulation ocean model of thorium-230 and protactinium-231 (ProThorP 0.1), doi:10.1594/PANGAEA.887261, https://doi.pangaea.de/10.1594/PANGAEA.887261, supplement to: van Hulten, M.M.P. et al. (in review): A global scavenging and circulation ocean model of thorium-230 and protactinium-231 with
730 realistic particle dynamics (NEMO-ProThorP 0.1). Geoscientific Model Development Discussions, 1-32, 10.5194/gmd-2017-274, 2018.

Venchiarutti, C., Rutgers van der Loeff, M., and Stimac, I.: Scavenging of ^{231}Pa and thorium isotopes based on dissolved and size-fractionated particulate distributions at Drake Passage (ANTXXIV-3), Deep-Sea Res. Pt II, 58, 2767–2784, doi:10.1016/j.dsr2.2010.10.040, physics, Carbon Dioxide, Trace Elements and Isotopes in the Southern Ocean: The Polarstern Expeditions ANT XXIV-3 (2008) and ANT XXIII/3 (2006), 2011.

735 Yu, E.-F., Francois, R., and Bacon, M. P.: Similar rates of modern and last-glacial ocean thermohaline circulation inferred, Nature, 379, 22, doi:10.1038/379689a0, 1996.