

1 **Source apportionment and sensitivity**
2 **analysis: two methodologies with two**
3 **different purposes.**

4
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10

11 **Abstract**

12 In this work are reviewed the existing methodologies for source
13 apportionment and sensitivity analysis to identify key differences
14 and stress their implicit limitations. The emphasis is laid on the
15 differences between source “impacts” (sensitivity analysis) and
16 “contributions” (source apportionment) obtained by using four
17 different methodologies: brute force top-down, brute force bottom-
18 up, tagged species and decoupled direct method (DDM). A simple
19 theoretical example to compare these approaches is used
20 highlighting differences and potential implications for policy. When
21 the relationships between concentration and emissions are linear,
22 impacts and contributions are equivalent concepts. In this case,
23 source apportionment and sensitivity analysis may be used
24 indifferently for both air quality planning purposes and quantifying
25 sources contributions.

26 However, this study demonstrates that when the relationship
27 between emissions and concentrations is non-linear, sensitivity
28 approaches are not suitable to retrieve source contributions and
29 source apportionment methods are not appropriate to evaluate the
30 impact of abatement strategies. A quantification of the potential non-
31 linearities should therefore be the first step prior to source
32 apportionment or planning applications, to prevent any limitations in
33 their use. When non-linearity is mild, these limitations may however
34 be acceptable in the context of the other uncertainties inherent to
35 complex models.

36
37 Moreover, when using sensitivity analysis for planning, it is
38 important to note that, under non-linear circumstances, the
39 calculated impacts will only provide information for the exact
40 conditions (e.g. emission reduction share) that are simulated.

41

42

43 **Keywords:** source apportionment, sensitivity analysis, abatement strategies,
44 air quality planning

45

46 **1. Introduction**

47

48 When pollutant concentrations exceed the thresholds set in the
49 legislation, competent authorities must take actions to abate
50 pollution. Those abatement strategies consist in reducing the
51 precursor's emission of the different activity sector to reduce
52 pollutant concentrations but they are challenging to design because
53 of the complex relationships that link emissions and pollutants.
54 Indeed, the concentration of a pollutant at a given location generally
55 results from direct emissions and from interactions in the
56 atmosphere among different emission precursors, emitted by a
57 variety of sources. For example, particulate matter (denoted here as
58 PM) results from the interaction and combination of five different
59 precursors (PPM, NO_x, SO₂, NH₃, and VOC) which can be emitted
60 by different activity macro-sectors (e.g. residential, transport,
61 industrial and agriculture [Seinfeld and Pandis, 2016]).

62 Two different approaches are currently used to support air quality
63 decision makers: source apportionment and sensitivity analysis.

64

65 • Source apportionment quantifies the **contribution** of an emission
66 source (or precursor) to the concentration of one pollutant at one
67 given location.

68 • Sensitivity analysis estimates the **impact** on pollutant
69 concentration that results from a change of one or more emission
70 sources.

71

72 In practice, source apportionment is often used for planning
 73 purposes. It is indeed intuitive to use source apportionment to detect
 74 the activity sectors that need to be tackled in priority in an air quality
 75 plan. On the other hand, sensitivity analysis is often used as an
 76 approach to derive source contributions, e.g. SHERPA (Thunis et al.
 77 2017), FASST (Crippa et al. 2017), GAINS (Kieseewetter et al.
 78 2015)...

79 The main objective of this work is to review the existing
 80 methodologies, identify key differences and stress their implicit
 81 limitations. We will particularly focus on the differences between
 82 concentration “impacts” (sensitivity) and “contributions” (source
 83 apportionment) obtained with different methodologies. We make use
 84 of a simple theoretical example to compare the approaches, highlight
 85 differences and potential implications in terms of policy. In the
 86 following sections, we analyze first how these methodologies work
 87 in a simple linear case before generalizing it to more complex non-
 88 linear situations.

89

90 **2. Linear simplification and implications**

91

92 Let's consider C a pollutant concentration at one location that is a
 93 function of three variables (E_1 , E_2 and E_3), i.e. the emissions of three
 94 precursors or sources within a given domain: $C = C(E_1, E_2, E_3)$. For
 95 a linear relationship between the function C and the three variables
 96 E_1 , E_2 and E_3 , we can write:

97

$$C(E_1, E_2, E_3) = C(0, 0, 0) + P_1 E_1 + P_2 E_2 + P_3 E_3 \quad (1)$$

98

99 where P_1 , P_2 and P_3 are three constant coefficients.

100

101 On the other hand, the sensitivity of the concentration to a change of
 102 a given emission source can be quantified via partial derivatives. For
 103 expression (1) this gives:

104

$$\frac{\partial C}{\partial E_1} = P_1 ; \quad \frac{\partial C}{\partial E_2} = P_2 ; \quad \frac{\partial C}{\partial E_3} = P_3$$

105

106 In Clappier et al. (2017) the coefficients (P_1 , P_2 and P_3) are referred
 107 to “potencies”. The authors used this concept of “potencies” to
 108 analyze the model response to emission changes in different European
 109 countries.

110

111 The consequences of a linear relationship between concentration and
 112 emission sources are twofold:

113 1) all higher order derivatives (order 2 and beyond) are null, including
 114 those involving two or more emission sources (crossed derivatives) as
 115 the impact of a change in one emission source is independent from all
 116 others.

117 2) the first-order partial derivatives are constant and can therefore be
 118 calculated with finite differencing, between any couple of emission
 119 levels, for example a base case (denoted BC) and a background
 120 (denoted as 0).

121 The potency equations then read as:

122

$$123 \quad P_1 = \frac{\Delta C_0^{BC_1}}{E_1^{BC}} ; P_2 = \frac{\Delta C_0^{BC_2}}{E_2^{BC}} ; P_3 = \frac{\Delta C_0^{BC_3}}{E_3^{BC}}$$

124

$$125 \quad \text{with } \Delta C_0^{BC_1} = C(E_1^{BC}, 0, 0) - C(0, 0, 0),$$

$$126 \quad \Delta C_0^{BC_2} = C(0, E_2^{BC}, 0) - C(0, 0, 0)$$

$$127 \quad \Delta C_0^{BC_3} = C(0, 0, E_3^{BC}) - C(0, 0, 0)$$

128

129 Together with ”potencies”, Clappier et al. (2017) also introduce the
 130 concept of “potential”, defined as the concentration change resulting
 131 from a total reduction of the emissions (from BC to 0). The “potential”
 132 can be calculated via relation (1) applied between the base-case and
 133 background levels as:

134

$$\Delta C_0^{BC} = \Delta C_0^{BC_1} + \Delta C_0^{BC_2} + \Delta C_0^{BC_3} \quad (2)$$

135

$$136 \quad \text{where } \Delta C_0^{BC} = C(E_1^{BC}, E_2^{BC}, E_3^{BC}) - C(0, 0, 0)$$

137

138 Equation (2) can directly be used for source apportionment purpose,
139 with $\Delta C_0^{BC_1}$ the concentration change resulting from a total reduction
140 of the emission source (or precursor) E_1 , reflecting the contribution
141 of E_1 to the base case concentration. Similarly $\Delta C_0^{BC_2}$ and $\Delta C_0^{BC_3}$ are
142 the contributions of E_2 and E_3 . Equation (2) shows that, in the
143 linear case, the concentration change resulting from a simultaneous
144 reduction of all emission sources (ΔC_0^{BC}) is equal to the sum of the
145 emission sources contributions.

146

147 In the next sections, we will explore how this simple conclusion
148 changes when non-linear relationships are considered. In particular,
149 we will assess which implications (and limitations) these non-
150 linearities have in terms of source apportionment and sensitivity
151 analysis.

152

153 **3. Brute Force method**

154

155 The “brute force” method consists in estimating the concentration
156 change by performing and subtracting two simulations, one with and
157 the second without a specific emission source to be analysed
158 (Blanchard, 1999; Yarwood et al., 2004).

159 In non-linear situations, the concentration change resulting from a
160 set of emission sources is not anymore equivalent to the sum of the
161 concentration changes resulting from emission sources changed
162 individually. In the following, we will refer to the work of Stein and
163 Alpert (1993) who proposed an approach to decompose an overall
164 impact into single (one emission source only) and combined
165 (multiple emission sources) impacts.

166

167

168 3.1. Bottom-up formulation

169

170 We consider here three precursor’s emission E_1 , E_2 and E_3 that are
171 changing from a low (denoted as “L”) to a high level (denoted as
172 “H”). In a bottom-up approach, the low emission level is chosen as
173 the reference. With these definitions and notations, the impact on

174 concentration resulting from a change of one only of the three
 175 precursor's emissions can be written as follows:
 176

$$\begin{aligned}\Delta C_{\bar{L}}^{H_1} &= C(E_1^H, E_2^L, E_3^L) - C(E_1^L, E_2^L, E_3^L) \\ \Delta C_{\bar{L}}^{H_2} &= C(E_1^L, E_2^H, E_3^L) - C(E_1^L, E_2^L, E_3^L) \\ \Delta C_{\bar{L}}^{H_3} &= C(E_1^L, E_2^L, E_3^H) - C(E_1^L, E_2^L, E_3^L)\end{aligned}$$

177
 178 While the impact on concentration resulting from the simultaneous
 179 changes of two or three precursor's emissions would be written as:
 180

$$\begin{aligned}\Delta C_{\bar{L}}^{H_1, H_2} &= C(E_1^H, E_2^H, E_3^L) - C(E_1^L, E_2^L, E_3^L) \\ \Delta C_{\bar{L}}^{H_1, H_3} &= C(E_1^H, E_2^L, E_3^H) - C(E_1^L, E_2^L, E_3^L) \\ \Delta C_{\bar{L}}^{H_2, H_3} &= C(E_1^L, E_2^H, E_3^H) - C(E_1^L, E_2^L, E_3^L) \\ \Delta C_{\bar{L}}^{\bar{H}} &= C(E_1^H, E_2^H, E_3^H) - C(E_1^L, E_2^L, E_3^L)\end{aligned}$$

181 Using a similar notation, the decomposition of Stein and Alpert
 182 (1993) applied to 2 variables (E_1 and E_2) would read as:
 183

$$\Delta C_{\bar{L}}^{H_1, H_2} = \Delta C_{\bar{L}}^{H_1} + \Delta C_{\bar{L}}^{H_2} + \hat{C}^{int} \quad (3)$$

184
 185 where $\Delta C_{\bar{L}}^{H_1}$ and $\Delta C_{\bar{L}}^{H_2}$ are the impacts induced by the change in
 186 emission sources E_1 and E_2 taken independently and $\Delta C_{\bar{L}}^{H_1, H_2}$ the
 187 impact induced from E_1 and E_2 taken simultaneously.
 188
 189

190 It is clear from (3) that the impact of a simultaneous change of two
 191 emission sources is not equivalent to the sum of the individual
 192 impacts, as highlighted by the additional term \hat{C}^{int} . This term that
 193 quantifies the interaction between the two emission sources can be
 194 calculated using (3) as:
 195

$$\hat{C}^{int} = \hat{C}_{\bar{L}}^{H_1, H_2} = \Delta C_{\bar{L}}^{H_1, H_2} - \Delta C_{\bar{L}}^{H_1} - \Delta C_{\bar{L}}^{H_2} \quad (4)$$

196
 197 The Stein and Alpert formulation can similarly be applied with 3
 198 emission sources:

199

$$\Delta C_{\bar{L}}^{\bar{H}} = \Delta C_{\bar{L}}^{H_1} + \Delta C_{\bar{L}}^{H_2} + \Delta C_{\bar{L}}^{H_3} + \hat{C}^{int} \quad (5)$$

200

201 Where $\Delta C_{\bar{L}}^{H_1}$, $\Delta C_{\bar{L}}^{H_2}$ and $\Delta C_{\bar{L}}^{H_3}$ are the impact on concentration
202 resulting from single emission changes in the sources and
203

$$\hat{C}^{int} = \hat{C}_{\bar{L}}^{H_1,H_2} + \hat{C}_{\bar{L}}^{H_1,H_3} + \hat{C}_{\bar{L}}^{H_2,H_3} + \hat{C}_{\bar{L}}^{H_1,H_2,H_3} \quad (6)$$

204

205 where $\hat{C}_{\bar{L}}^{H_1,H_2}$, $\hat{C}_{\bar{L}}^{H_1,H_3}$ and $\hat{C}_{\bar{L}}^{H_2,H_3}$ are the double interaction terms that
206 can be further decomposed via equation (4). $\hat{C}_{\bar{L}}^{H_1,H_2,H_3}$ is the triple
207 interaction term (between E_1 , E_2 , E_3) which can be decomposed by
208 combining (5) and (6) as:
209

$$\begin{aligned} \hat{C}_{\bar{L}}^{H_1,H_2,H_3} &= \Delta C_{\bar{L}}^{\bar{H}} - \Delta C_{\bar{L}}^{H_1} - \Delta C_{\bar{L}}^{H_2} - \Delta C_{\bar{L}}^{H_3} \\ &\quad - \Delta C_{\bar{L}}^{H_1,H_2} - \Delta C_{\bar{L}}^{H_1,H_3} - \Delta C_{\bar{L}}^{H_2,H_3} \end{aligned}$$

210

211 3.2. Top-down formulation

212

213 In a top-down formulation, the highest emission level is chosen as
214 reference. The Stein Alpert formulation for three precursors can then
215 be expressed similarly to the bottom-up formulation as:
216

$$\Delta C_{\bar{L}}^{\bar{H}} = \Delta C_{L_1}^{\bar{H}} + \Delta C_{L_2}^{\bar{H}} + \Delta C_{L_3}^{\bar{H}} + \hat{C}^{int} \quad (7)$$

217

218 where $\Delta C_{L_1}^{\bar{H}}$, $\Delta C_{L_2}^{\bar{H}}$ and $\Delta C_{L_3}^{\bar{H}}$ are the impacts on concentration induced
219 by reducing E_1 , E_2 and E_3 independently whereas \hat{C}^{int} is the
220 interaction term which itself can be decomposed into a series of
221 double interactions and a triple interaction terms:
222

$$\hat{C}^{int} = \hat{C}_{L_1,L_2}^{\bar{H}} + \hat{C}_{L_1,L_3}^{\bar{H}} + \hat{C}_{L_2,L_3}^{\bar{H}} + \hat{C}_{L_1,L_2,L_3}^{\bar{H}} \quad (8)$$

223

224 It is important to stress that the top-down single impacts are not
225 equivalent to their bottom-up counterparts. The relation between

226 these bottom-up and top-down impacts can be expressed as (here for
227 the case of E_3):

228

$$229 \quad \Delta C_{L_3}^{\bar{H}} = C(E_1^H, E_2^H, E_3^H) - C(E_1^H, E_2^H, E_3^L)$$

230

231

$$232 \quad \Delta C_{L_3}^{\bar{H}} = C(E_1^H, E_2^H, E_3^H) - C(E_1^L, E_2^L, E_3^L) \\ 233 \quad \quad \quad - [C(E_1^H, E_2^H, E_3^L) - C(E_1^L, E_2^L, E_3^L)]$$

234

$$235 \quad \Delta C_{L_3}^{\bar{H}} = \Delta C_L^{\bar{H}} - \Delta C_L^{H_1 H_2} \quad (9)$$

235

236 Using equations (3), (4), (5) and (6), equation (9) can be re-
237 expressed as:

238

$$\Delta C_{L_3}^{\bar{H}} = \Delta C_L^{H_3} + \hat{C}_L^{H_1, H_3} + \hat{C}_L^{H_2, H_3} + \hat{C}_L^{H_1, H_2, H_3} \quad (10)$$

239

240 In other words, the top-down impact on concentration of an emission
241 source (obtained by switching off the emission source while all
242 others remain unchanged) is not equivalent to its bottom-up
243 counterpart, (obtained by switching on the emission source while all
244 others are switched off). Relation (10) indeed clearly shows that
245 additional interaction terms need to be considered. The implications
246 resulting from these differences are highlighted in Section 5 in
247 which some theoretical examples are described.

248

249 **4. Source apportionment and sensitivity analysis**

250

251 4.1. Tagged species techniques

252

253 Equation (2) shows that, when the relationship between concentration
254 and several emission sources is linear, the contribution of a specific
255 source (source apportionment) can be computed as the impact on
256 concentration obtained by a full reduction of this source (sensitivity).
257 Moreover, the sum of the impacts on concentration obtained by
258 reduction of the single sources ($\Delta C_0^{BC_1} + \Delta C_0^{BC_2} + \Delta C_0^{BC_3}$) is equivalent to

259 the impact on concentration resulting from a simultaneous abatement
 260 of all sources (ΔC_0^{BC}). In such a case, the concentration impacts are
 261 equal to source contributions and source apportionment and
 262 sensitivity analysis lead to similar results. This is not the case,
 263 however, when the relationship between concentrations and
 264 emissions is nonlinear. In their approach, Stein and Alpert express the
 265 difference between the impact caused by a simultaneous abatement
 266 and the sum of the impacts caused by individual abatement as
 267 interactions among the different sources. The Stein and Alpert
 268 formulation applied between the base-case and background levels is
 269 very close to equation (2) but with an additional term that accounts
 270 for interactions:

271

$$272 \quad \Delta C_0^{BC} = \Delta C_0^{BC_1} + \Delta C_0^{BC_2} + \Delta C_0^{BC_3} + \hat{C}^{int}$$

273

274 Because the interaction terms cannot not be attributed to a single
 275 emission source as they represent the interaction between two and
 276 more emission sources, the Stein and Alpert methodology does not
 277 allow identifying the full contribution of each individual source. It
 278 cannot therefore be used for source apportionment purpose, unless the
 279 interaction terms are negligible as in the linear case.

280 Unlike the Stein and Alpert methodology, the tagged species
 281 methodology is designed for source apportionment purposes. This
 282 methodology tags each precursor and quantifies its contribution (in
 283 terms of mass) to the pollutant concentration.

284 Tagged algorithms are implemented in several chemical transport
 285 model systems (Yarwood et al., 2004; Wagstrom et al., 2008,
 286 ENVIRON, 2014; Bhave et al., 2007; Wang et al., 2009; Kranenburg
 287 et al., 2013).

288

289 In tagging approaches, the effect of the full reduction of all sources is
 290 directly expressed as the sum of the source contributions:

291

$$292 \quad \Delta C_0^{BC} = \delta C_1 + \delta C_2 + \delta C_3$$

293

294 where δC_1 , δC_2 , δC_3 are the contributions of sources E_1 , E_2 and E_3
 295 resulting from the tagged species approach resolution.

296

297 Tagging methodologies split the interaction terms into fractions and
 298 attribute these fractions to the source contributions, on the basis of
 299 mass weighting factors.

300

301

$$\delta C_1 = \Delta C_0^{BC_1} + \alpha \hat{C}^{int}$$

302

303

304 Because the tagged species approach mixes interaction terms and
 305 single concentration impacts into sources contributions, it is not
 306 suitable to estimate the effect of emission reduction when non-
 307 linearities are present (Burr and Zhang 2011a, 2011b). Indeed these
 308 two types of terms may react in very different ways to emission
 309 reductions. This fact is detailed in the examples provided below.

310 On the other hand, the strength of this method is that it allows for a
 311 direct comparison of the source contributions with measurements (or
 312 measurement based methods like receptor models).

313 Note that similar tagging methods are also used in the frame of
 314 climate-chemical studies at global scale (e.g. Horowitz and Jacob,
 315 1999; Lelieveld and Dentener, 2000; Meijer et al., 2000; Grewe, 2004;
 316 Gromov et al., 2010; Butler et al., 2011; Emmons et al., 2012; Grewe
 317 et al., 2012; Grewe et al., 2017).

318

319 4.2. DDM

320

321 The decoupled direct method (DDM) is designed to calculate
 322 directly sensitivities to emission changes (Dunker et al., 1984;
 323 Dunker et al., 2002). It aims to compute the first order derivatives
 324 (which correspond to the potencies mentioned in section 2).

325

326

$$\frac{\partial C}{\partial E_1} ; \frac{\partial C}{\partial E_2} ; \frac{\partial C}{\partial E_3}$$

327

328 The Taylor formula is applied at first order to calculate the
 329 concentration change between two emission levels (denoted H and
 330 L).

331

332
$$\Delta C_L^{\bar{H}} = \Delta E_1 \left. \frac{\partial C}{\partial E_1} \right|_H + \Delta E_2 \left. \frac{\partial C}{\partial E_2} \right|_H + \Delta E_3 \left. \frac{\partial C}{\partial E_3} \right|_H$$

333

334 with
$$\Delta E_1 = E_1^H - E_1^L, \Delta E_2 = E_2^H - E_2^L, \Delta E_3 = E_3^H - E_3^L$$

335

336 In the linear case, the first order derivatives are constant and the first
 337 order approximation of the Taylor formula gives the exact
 338 expression of the impact on concentration of an emission change
 339 between H and L. When the emission-concentration relationship is
 340 nonlinear, the first derivatives are not constants. The first order
 341 Taylor formula cannot take into account the nonlinear effects. It is a
 342 linear approximation based on derivatives computed at a given
 343 emission reference level (level H in our example). The estimation of
 344 the impact on concentration of an emission change between H and L
 345 is accurate enough if level L is close enough to level H.

346

347 HDDM is another method (Hakami et al., 2003) which aims to
 348 increase the accuracy of the DDM method by computing second
 349 order derivatives.

350 DDM (and HDDM) gives similar information to the Stein Alpert
 351 formulation applied with the Brute force top-down approach
 352 (because the reference level is H). For the same reason than for the
 353 Stein Alpert approach, these two methods are suitable for source
 354 apportionment purpose only if the relation between concentration
 355 and emission is close to linearity.

356

357 DDM (and HDDM) approximates the impact on concentration from
 358 an emission change between the two level H and L, using
 359 derivatives computed at level H. This impact is accurate enough if
 360 the level L is close enough to the reference level H.

361 Dunker (2015) showed how to use first-order sensitivity to
 362 determine source contributions between two model cases, e.g., to
 363 apportion the difference between the current atmosphere (and
 364 natural conditions) to specific human activities. Along the same
 365 lines, Simon et al. (2013) used first order sensitivity to construct
 366 emission response surfaces. To cope with potential non-linearities
 367 and the need to compute higher-order derivatives, a powerful

368 alternative is to compute first-order sensitivities at several emission
 369 levels.

370

371

372 **5. Example**

373

374 In this section, examples are designed to illustrate the differences in
 375 terms of contribution and impact estimates when the approaches
 376 discussed previously are used. In these examples, we focus on the
 377 formation of particulate matter (PM) in the atmosphere and only
 378 consider three formation processes: direct emissions (primary PM
 379 denoted as PPM); formation through reactions with nitrogen oxides
 380 (NO_2) and ammonia (NH_3) and formation through reactions with
 381 sulphur oxide (SO_2) and NH_3 .

382

383 $\text{PPM} \longrightarrow \text{PM}$

384 $2\text{NO}_2 + \text{H}_2\text{O} + \frac{1}{2}\text{O}_2 \longrightarrow 2\text{HNO}_3$

385 $2\text{NH}_3 + 2\text{HNO}_3 \longrightarrow 2\text{NH}_4\text{NO}_3$

386 $\text{SO}_2 + \text{H}_2\text{O} + \frac{1}{2}\text{O}_2 \longrightarrow \text{H}_2\text{SO}_4$

387 $2\text{NH}_3 + \text{H}_2\text{SO}_4 \longrightarrow (\text{NH}_4)_2\text{SO}_4$

388

389 These reactions pathways are summarised by the following system
 390 of reactions:

391

392 $\text{PPM} \rightarrow \text{PM}[\text{PPM}]$

393 $\text{NO}_2 + \text{NH}_3 \rightarrow \text{PM}[\text{NH}_4\text{NO}_3]$

394 $\text{SO}_2 + 2\text{NH}_3 \rightarrow \text{PM}[(\text{NH}_4)_2\text{SO}_4]$

395

396 This system is further simplified by assuming that all reactions have
 397 comparable kinetics (reaction speed) and have reached their
 398 equilibrium. From these three reactions, 1 PM mole can be produced
 399 by 1 PPM mole, by the combination of 1 NH_3 and 1 NO_2 moles or
 400 by the combination of 1 SO_2 and 2 NH_3 moles.

401
 402 We also limit our examples to emissions from three activity sectors.
 403 The residential sector (R) only emits PPM and NO₂, the agricultural
 404 sector (A) only emits NH₃ and the industrial sector (I) only emits
 405 PPM and SO₂ (Figure 1). We assume for convenience that no
 406 background pollution is present (i.e. there is no PM when all
 407 emissions are zero). Two situations are considered: a “non-limited
 408 regime” where the NH₃ quantity is sufficient to react with all moles
 409 of NO₂ and SO₂ and a “limited regime” where the NH₃ quantity of is
 410 not sufficient to react with all moles of NO₂ and SO₂.

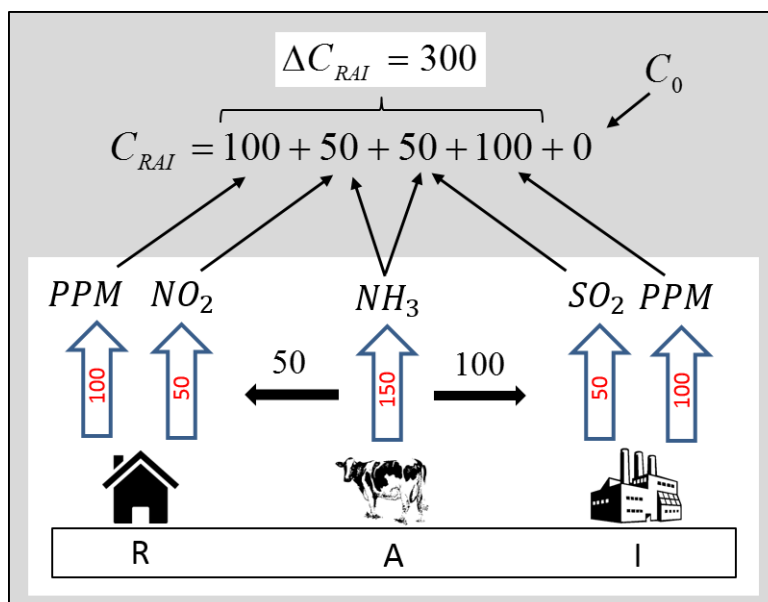
411

412 5.1. Non-limited regime

413

414 In this first example, the quantity of precursors (in terms of mass) is
 415 large enough to feed all reactions. The agricultural sector emits 150
 416 NH₃ moles which can react with 50 NO₂ moles emitted by the
 417 residential sector and 50 SO₂ moles emitted by industrial sector. 100
 418 PPM moles are emitted by the residential sector as well by the
 419 industrial sector (Figure 1).

420



421

422 Figure 1: Example of PPM, NO₂, SO₂ and NH₃ emissions released
 423 by three activity sectors: residential (R), agricultural (A) and
 424 industrial (I). For convenience, no units are associated to emissions
 425 and concentrations

426

427 Let's first calculate the PM concentration produced with and without
 428 each of the sources:

429

430 No source: C_0 is the PM concentration obtained when all emissions
 431 are set to zero. Since we assumed a zero background pollution, $C_0 =$
 432 0.

433

434 One source only: C_R (resp. C_A and C_I) is the PM concentration
 435 reached when only the residential (resp. agricultural and industrial)
 436 sector releases emissions:

437 - $C_R = 100$ produced by PPM emissions (NO₂ emissions do not
 438 produce PM as no NH₃ is available).

439 - $C_A = 0$ because NO₂ and SO₂ are not available to react with NH₃.

440 - $C_I = 100$ produced by the PPM emissions (SO₂ emissions do not
 441 produce PM as no NH₃ is available).

442

443 Two sources: C_{RA} , C_{RI} , and C_{AI} are the concentrations obtained when
 444 two (out of three) activity sectors release their emissions
 445 simultaneously (the *RA* subscripts correspond to residential and
 446 agriculture, *RI* to residential and industrial, *AI* to agriculture and
 447 industrial):

448 - $C_{RA} = 150$: 100 produced by PPM emissions from the residential
 449 sector and 50 produced by the 50 NO₂ released by the residential
 450 sector reacting with the 50 NH₃ emitted by agriculture (100 NH₃
 451 moles remain unused).

452 - $C_{RI} = 200$: 100 produced by PPM emissions from the residential
 453 sector and 100 produced by PPM emissions from the industrial
 454 sector.

455 - $C_{AI} = 150$: 100 produced by PPM industrial emissions and 50
 456 from the combination of 50 SO₂ (industry) and 100 NH₃
 457 (agriculture).

458

459 Three sources: C_{RAI} is the concentrations obtained when all
 460 emissions are released simultaneously.
 461 - $C_{RAI} = 300$: 200 from PPM (residential and industry), 50 from
 462 reaction between NO_2 and NH_3 and 50 from reaction between SO_2
 463 and NH_3 .

464

465 **Brute-force Bottom-up (BF-BU) method**

466

467 The contribution of each activity sector is calculated as the
 468 concentration change resulting from a 100% emission increase from
 469 the lowest emission level (previously denoted “L” or background) to
 470 the highest level (denoted as “H”, the base case C_{RAI} obtained with
 471 all emissions)

472

473 In a bottom-up approach, the concentration associated with the
 474 lowest emission level is considered as the reference. Concentration
 475 impacts are then computed by the difference between any situation
 476 (e.g. one, two or three sources present) and this reference.

477

478 - with one source: $\Delta C_R^{BU} = C_R - C_0 = 100$

479 $\Delta C_A^{BU} = C_A - C_0 = 0$

480 $\Delta C_I^{BU} = C_I - C_0 = 100$

481

482 - with two sources: $\Delta C_{RA}^{BU} = C_{RA} - C_0 = 150$

483 $\Delta C_{RI}^{BU} = C_{RI} - C_0 = 200$

484 $\Delta C_{AI}^{BU} = C_{AI} - C_0 = 150$

485

486 - with three sources: $\Delta C_{RAI} = C_{RAI} - C_0 = 300$

487

488 To calculate the interaction terms, we use the Stein-Alpert
 489 formulation using Eq. (5) and Eq. (6):

490

491 $\Delta C_{RAI} = \Delta C_R^{BU} + \Delta C_A^{BU} + \Delta C_I^{BU} + \hat{C}_{RA}^{BU} + \hat{C}_{RI}^{BU} + \hat{C}_{AI}^{BU} + \hat{C}_{RAI}^{BU}$

492

493 from which the interaction terms are obtained by application of (4)

494 and (6):

495

496 $\hat{C}_{RA}^{BU} = \Delta C_{RA}^{BU} - \Delta C_R^{BU} - \Delta C_A^{BU} = 50$

497 $\hat{C}_{RI}^{BU} = \Delta C_{RI}^{BU} - \Delta C_R^{BU} - \Delta C_I^{BU} = 0$

$$\begin{aligned} 498 \quad \hat{C}_{AI}^{BU} &= \Delta C_{AI}^{BU} - \Delta C_A^{BU} - \Delta C_I^{BU} = 50 \\ 499 \quad \hat{C}_{RAI}^{BU} &= \Delta C_{RAI}^{BU} - \Delta C_R^{BU} - \Delta C_A^{BU} - \Delta C_I^{BU} - \hat{C}_{RA}^{BU} - \hat{C}_{RI}^{BU} - \hat{C}_{AI}^{BU} = 0 \\ 500 \end{aligned}$$

501 As can be seen from this example, the system behaves non-linearly
502 and the interaction terms (e.g. \hat{C}_{RA}^{BU}) are not-zero. Moreover the sum
503 of the individual impacts ($\Delta C_R^{BU} + \Delta C_A^{BU} + \Delta C_I^{BU} = 200$)
504 underestimates the overall impact ($\Delta C_{RAI}^{BU} = 300$). These results are
505 graphically represented in Figure 2 (third column).
506

507 **Brute-force Top-down (BF-TD) method**

508
509 In a BF-TD approach, the higher emission level (base case, C_{RAI}) is
510 the reference and the impact of each activity sector is calculated as
511 the concentration change resulting from a 100% emission decrease
512 (of one, two or three sources) from this reference to the background
513 level.
514

515
516 - with one source: When all emissions from one sector are reduced
517 (e.g. residential), the other two sector remain active (agricultural
518 and industry). In this case, the Top-down impact is the difference
519 between the base case concentration and the concentration
520 resulting from the agricultural and industrial emissions only. A
521 similar reasoning can be made for all sectors:
522

$$\begin{aligned} 523 \quad \Delta C_R^{TD} &= C_{RAI} - C_{AI} = 150 \\ 524 \quad \Delta C_A^{TD} &= C_{RAI} - C_{RI} = 100 \\ 525 \quad \Delta C_I^{TD} &= C_{RAI} - C_{RA} = 150 \\ 526 \end{aligned}$$

527 - with two sources: The Top-down impact due to a full reduction
528 of two sectors (e.g. residential and agriculture) is similarly
529 computed as the difference between the base case concentration
530 and the concentration resulting from the remaining sector
531 (industry).
532

$$\begin{aligned} 533 \quad \Delta C_{RA}^{TD} &= C_{RAI} - C_I = 200 \\ 534 \quad \Delta C_{RI}^{TD} &= C_{RAI} - C_A = 300 \\ 535 \quad \Delta C_{AI}^{TD} &= C_{RAI} - C_R = 200 \end{aligned}$$

536

537

538 - with three sources: The impact resulting from the simultaneous
 539 reduction of all three sources is similar in the Top-down and
 540 Bottom-up approaches:

541

542

$$\Delta C_{RAI} = C_{RAI} - C_0 = 300$$

543

544 The interaction terms can be obtained in a similar way to the
 545 bottom-up approach by using the Stein and Alpert formulation for
 546 ΔC_{RAI} :

547

$$\Delta C_{RAI} = \Delta C_R^{TD} + \Delta C_A^{TD} + \Delta C_I^{TD} + \hat{C}_{RA}^{TD} + \hat{C}_{RI}^{TD} + \hat{C}_{AI}^{TD} + \hat{C}_{RAI}^{TD}$$

549

550 The interaction terms are given by:

551

$$\hat{C}_{RA}^{TD} = \Delta C_{RA}^{TD} - \Delta C_R^{TD} - \Delta C_A^{TD} = -50$$

$$\hat{C}_{RI}^{TD} = \Delta C_{RI}^{TD} - \Delta C_R^{TD} - \Delta C_I^{TD} = 0$$

$$\hat{C}_{AI}^{TD} = \Delta C_{AI}^{TD} - \Delta C_A^{TD} - \Delta C_I^{TD} = -50$$

$$\hat{C}_{RAI}^{TD} = \Delta C_{RAI}^{TD} - \Delta C_R^{TD} - \Delta C_A^{TD} - \Delta C_I^{TD} - \hat{C}_{RA}^{TD} - \hat{C}_{RI}^{TD} - \hat{C}_{AI}^{TD} = 0$$

556

557 With this approach, a non-linear behavior is also observed and
 558 interaction terms are not zero. It is also interesting to note that the
 559 triple interaction term (\hat{C}_{RAI}^{TD}) is null. The sum of the individual
 560 impacts ($\Delta C_R^{TD} + \Delta C_A^{TD} + \Delta C_I^{TD} = 400$) overestimates the overall
 561 impact ($\Delta C_{RAI} = 300$). We further discuss these aspects at the end of
 562 this section. These results are graphically represented in Figure 2
 563 (fourth and fifth columns).

564

565

566 **Tagged species approach**

567

568 Compared to Brute-force, the tagged species approach calculates the
 569 share of each source to the overall concentration change. These
 570 shares are referred to as contributions and have the main property
 571 that the sum of the individual contributions is equal to the overall
 572 concentration impact, by definition, i.e.:

573

$$\Delta C_{RAI} = \delta C_R^{TAG} + \delta C_A^{TAG} + \delta C_I^{TAG}$$

574

575

576

577 The sector contributions are computed by tracking the mass of their
578 emitted species contributing to PM formation (in our example:

579 PM[PPM] , PM[NH₄NO₃] and PM[(NH₄)₂SO₄])

580

581 - PM[PPM] is formed from PPM. The 100 moles from the residential
582 sector lead to 100 moles of PM. The same applies to the 100 moles
583 from industry.

584

585 - PM[NH₄NO₃] is formed by combination of NH₃ and NO₂. The
586 share between these two contributions is obtained by application
587 of stoichiometric molar mass ratios:

588

$$589 \quad a_1 = \frac{[NO_3]_m}{[NO_3]_m + [NH_4]_m} = 0.78$$

590

591 In our example, 50 moles of PM[NH₄NO₃] are formed by
592 combination of NO₂ (50 moles) from the residential sector and
593 NH₃ (50 moles) from agriculture. The contribution attributed to
594 NO₂ is $50 \times a_1$ whereas the contribution attributed to NH₃ is
595 $50 \times (1 - a_1)$.

596

597

598 - PM[(NH₄)₂SO₄] is formed by combination of NH₃ and SO₂. The
599 following stoichiometric mass ratio is used:

600

$$601 \quad a_2 = \frac{[SO_4]_m}{[SO_4]_m + 2[NH_4]_m} = 0.73$$

602

603 The contribution attributed to SO₂ is $50 \times a_2$ whereas the
604 contribution attributed to NH₃ is $50 \times (1 - a_2)$.

605

606 The contribution of each sector is then obtained as the sum of their
607 precursor contribution shares as follows:

608

$$609 \quad \delta C_R^{TAG} = 100 + 50 \times a_1 = 138.7$$

$$610 \quad \delta C_A^{TAG} = 50 \times (1 - a_1) + 50 \times (1 - a_2) = 24.9$$

611 $\delta C_I^{TAG} = 100 + 50 \times a_2 = 136.4$

612

613 By definition the sum of the contributions ($\delta C_R^{TAG} + \delta C_A^{TAG} +$
 614 $\delta C_I^{TAG} = 300$) is exactly equal to the overall concentration impact
 615 ($\Delta C_{RAI} = 300$).

616

617 Note that a decomposition of the non-linear interaction terms
 618 obtained in the top-down or bottom-up approach, using the above
 619 stoichiometric ratios would lead to similar results than the tagged
 620 approach. These results are graphically represented in Figure 2
 621 (second column).

622

623

624 DDM

625

626 In this methodology, delta concentrations and interaction terms are
 627 estimated with first order partial derivatives computed from the
 628 highest emission level (base case in our example). Being a
 629 sensitivity approach using level H as reference, DDM shows clear
 630 analogies with the BF-TD.

631

632
$$\left. \frac{\partial C}{\partial \alpha_R} \right|_{TD} = 150 \quad \left. \frac{\partial C}{\partial \alpha_A} \right|_{TD} = 100 \quad \left. \frac{\partial C}{\partial \alpha_I} \right|_{TD} = 150$$

633

634

635 where α_R , α_A and α_I are percentage emission changes from the base
 636 case for the residential, agricultural and industrial sectors.

637 The first-order derivatives are evaluated using finite differencing
 638 between the base case and a level characterised by emissions that are
 639 10% lower for each activity sector.

640 The concentration changes resulting from a 100% emission
 641 reduction (i.e. between the base case and the zero emission case)
 642 could be estimated by setting α_R , α_A and α_I to unity.

643

644
$$\Delta C_R^{HDDM} = \left. \frac{\partial C}{\partial \alpha_R} \right|_{TD} = 150$$

645
$$\Delta C_A^{HDDM} = \left. \frac{\partial C}{\partial \alpha_A} \right|_{TD} = 100$$

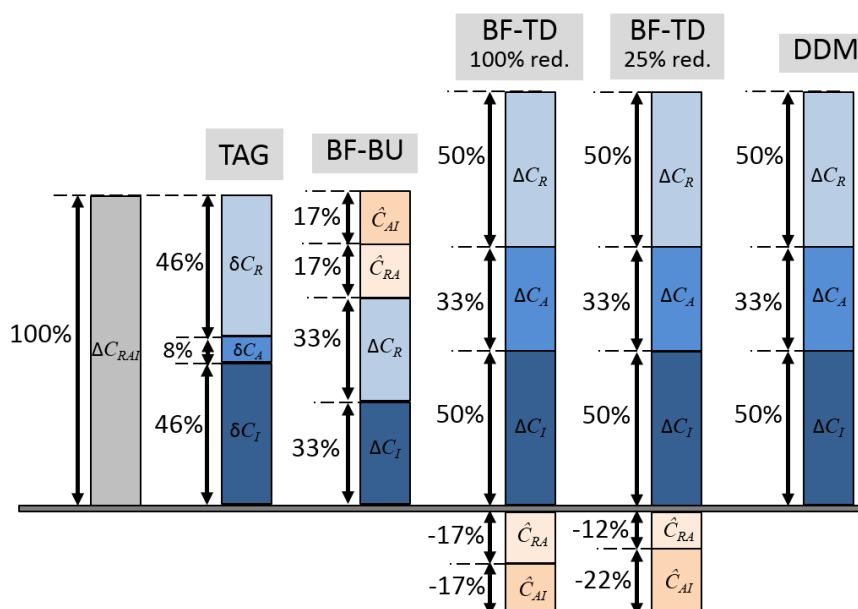
$$646 \quad \Delta C_I^{HDDM} = \left. \frac{\partial C}{\partial \alpha_I} \right|_{TD} = 150$$

647

648

649 We see from this last example that both the total PM and the
 650 contribution of the sources are then comparable with those obtained
 651 with the BF-TD method. Their interpretation is similar (Figure 2,
 652 sixth column). In their work, Koo et al. (2009) present a detailed
 653 comparison between a DDM and a tagged species approach in a 3-D
 654 PM model and show which sensitivities are similar to
 655 apportionment, and which are not.

656



657

658 Figure 2: Schematic representation of the allocation of PM to its
 659 sources in the non-limited example. The expected total PM is
 660 displayed in the grey bar on the left.

661

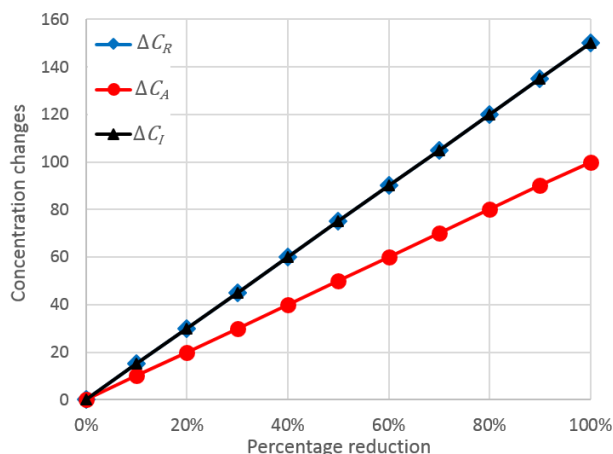
662 Comparative overview

663

664 In the linear case (second paragraph) we have seen that a single
 665 source contribution can be computed as the impact resulting from a
 666 full reduction of this source. However, source contributions and

667 concentration impacts should not be confused as they are different in
 668 most situations. The example presented in this paragraph illustrates
 669 this clearly for a non-linear system. Indeed the contributions of the
 670 single sources computed by the tagged species approach ($\delta C_R^{TAG} =$
 671 138 ; $\delta C_A^{TAG} = 24$; $\delta C_I^{TAG} = 136$) differ from the concentration
 672 impacts resulting from a total abatement of these single sources
 673 computed by the BF-TD ($\Delta C_R^{TD} = 150$; $\Delta C_A^{TD} = 100$; $\Delta C_I^{TD} = 150$)
 674 method. Moreover, the sum of the concentration impacts obtained
 675 with either the BF-TD approach for single sources does not equal the
 676 total concentration impact ($\Delta C_{RAI} = 300$). This is also valid for any
 677 selection of two sectors ($\Delta C_R^{TD} + \Delta C_A^{TD} = 250 \neq \Delta C_{RA}^{TD} = 200$). Note
 678 that similarly to BF-TD, the concentration impacts computed as
 679 increases from the background (BF-BU) show the same behaviour
 680 ($\Delta C_R^{BU} + \Delta C_A^{BU} = 100 \neq \Delta C_{RA}^{BU} = 150$).

681
 682 Figure 3 shows that the impact on concentration is proportional to
 683 the emission reduction indicating that the relationship between
 684 emission and concentration changes is linear. However, this example
 685 also illustrates the fact that linearity encompasses two aspects: 1)
 686 the interaction terms are zero ($\hat{C}^{int} = 0$); 2) the ratios between
 687 concentration change and emission changes ($\Delta C/\Delta E$) remain
 688 constant, regardless of the calculation bounds (denoted “H” and “L”
 689 in Section 4). In the current example the ratios $\Delta C/\Delta E$ are constant
 690 (linear trend of ΔC on Figure 3) but the relationship between
 691 concentration and emission is not linear because of the non-zero
 692 interaction terms (not shown) ($\hat{C}_{RA}^{TD} = -50$ and $\hat{C}_{AI}^{TD} = -50$).
 693 However, even with zero interaction terms, we can still observe a
 694 non-linear behaviour with the emission reduction percentage. The
 695 evaluation of linearity therefore requires two tests: one to quantify
 696 the interaction terms and the second to assess the deviation from a
 697 linear trend with respect to the emission reduction percentage.
 698



699
 700 Figure 3: Evolution of the concentration changes resulting from
 701 different percentage of source abatement (Top down approach) for
 702 the three sectors (Residential, Agricultural and Industrial).

703

704

705

706 5.2. Limited regime

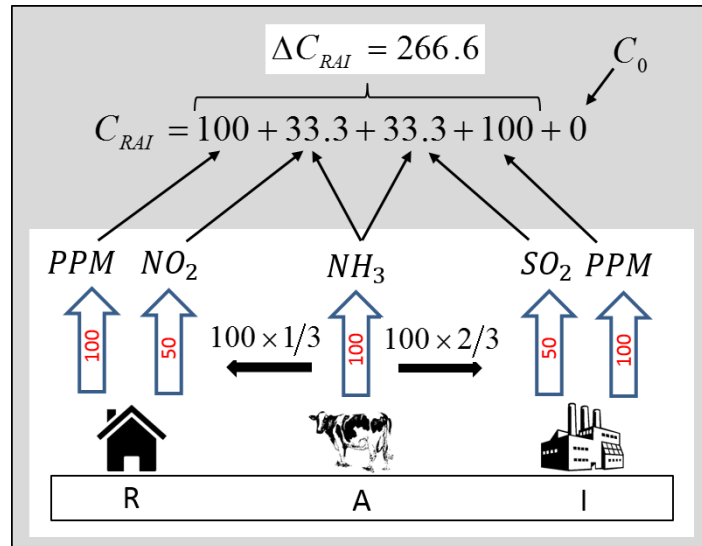
707

708 This example is similar to the previous, except that the emissions of
 709 NH_3 are reduced from 150 to 100 moles.

710 When all sources release emissions, the 100 moles of NH_3 are
 711 shared into $100/3=33.3$ moles which react with NO_2 to form 33.3
 712 moles of $\text{PM}[\text{NH}_4\text{NO}_3]$ and $100 \times 2/3=66.6$ moles which react with
 713 SO_2 to give 33.3 moles of $\text{PM}[(\text{NH}_4)_2\text{SO}_4]$. Because the mass of
 714 NH_3 is not enough to react with all the NO_2 and SO_2 mass, 16.7
 715 moles of NO_2 and 16.7 moles of SO_2 remain unused (Figure 4).

716 Note that when the agricultural source is active with one only of the
 717 two other sources (residential or industrial) the NH_3 100 moles are
 718 then sufficient to consume all the NO_2 or SO_2 and lead to 50 moles
 719 of PM in either case.

720



721
722 Figure 4: Example with three sources in an ammonia-limited regime.
723 The mass emitted by each source is expressed in moles.

724
725 The PM concentrations obtained when one or two sources are active
726 are similar to the previous example:

727
728 $C_0 = 0 ; C_R = 100 ; C_A = 0 ; C_I = 100$
729 $C_{RA} = 150 ; C_{RI} = 200 ; C_{AI} = 150$

730
731 But the result differ when all sources are active: $C_{RAI} = 266.6$ (200
732 from PPM (residential industry), 33.3 from reaction between NO_2
733 and NH_3 and 33.3 from reaction between SO_2 and NH_3).

734
735 **Bottom-up brute-force method (BF-BU)**
736

737 The BF-BU approach computes all concentration impacts from the
738 background concentration (C_0). The Stein-Alpert terms are similar to
739 the non-limited case, excepted for ΔC_{RAI} and \hat{C}_{RAI} :

740
741 $\Delta C_R^{BU} = 100$ $\hat{C}_{RA}^{BU} = 50$
742 $\Delta C_A^{BU} = 0$ $\hat{C}_{RI}^{BU} = 0$
743 $\Delta C_I^{BU} = 100$ $\hat{C}_{AI}^{BU} = 50$
744 $\Delta C_{RAI} = 266.6$ $\hat{C}_{RAI} = -33.3$

745
 746 The limiting effect of NH_3 appears only in the negative triple
 747 interaction term (\hat{C}_{RAI}). These results are graphically represented in
 748 Figure 5 (third column).

749

750 **Top-down brute-force method (BF-TD)**

751

752 The Top-down approach uses the base case (C_{RAI}) concentration as
 753 reference to compute the concentration impact. In this case, all
 754 Stein-Alpert terms are different from the non-limited regime:

755

$$\begin{array}{ll}
 756 & \Delta C_R^{TD} = 116.6 & \hat{C}_{RA}^{TD} = -16.6 \\
 757 & \Delta C_A^{TD} = 66.6 & \hat{C}_{RI}^{TD} = 33.3 \\
 758 & \Delta C_I^{TD} = 116.6 & \hat{C}_{AI}^{TD} = -16.6 \\
 759 & \Delta C_{RAI} = 266.6 & \hat{C}_{RAI} = -33.3
 \end{array}$$

760

761 These results are graphically represented in Figure 5 (fourth and
 762 fifth columns).

763

764 **Tagged approach**

765

766 The contribution of each source is computed similarly to the non-
 767 limited regime. The production of 33.3 moles of $\text{PM}[\text{NH}_4\text{NO}_3]$ and
 768 33.3 moles of $\text{PM}[(\text{NH}_4)_2\text{SO}_4]$ are split among the different sectors
 769 using the stoichiometric coefficients a_1 and a_2 :

770

$$\begin{array}{l}
 771 \delta C_R^{TAG} = 100 + 33.3 * a_1 = 125.8 \\
 772 \delta C_A^{TAG} = 33.3 * (1 - a_1) + 33.3 * (1 - a_2) = 16.6 \\
 773 \delta C_I^{TAG} = 100 + 33.3 * a_2 = 124.2
 \end{array}$$

774

775 These results are graphically represented in Figure 5 (second
 776 column).

777

778

779 **DDM**

780

781 As shown below, DDM only considers first derivatives, which are
 782 not suitable to estimate higher order interaction terms. The
 783 calculation of the first order derivatives in this example gives:
 784

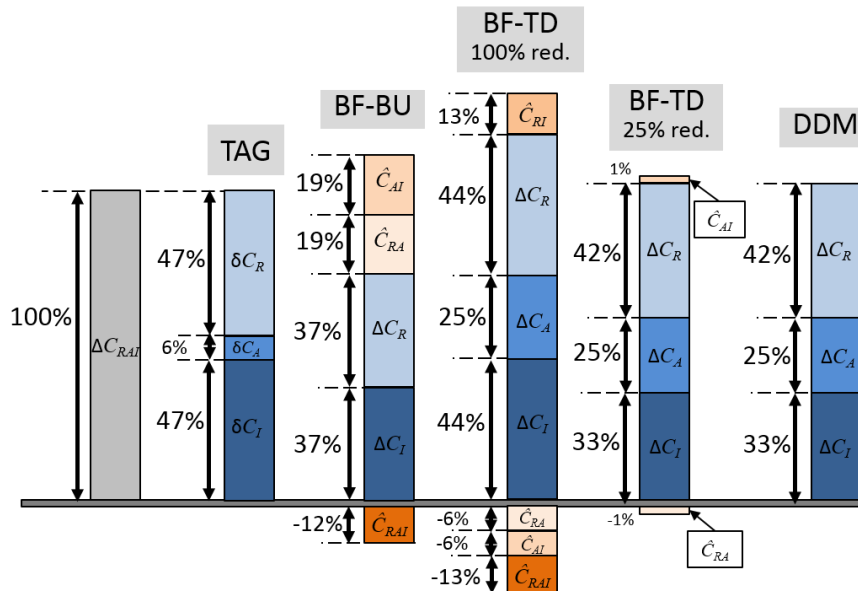
$$785 \quad \Delta C_R^{HDDM} = \left. \frac{\partial C}{\partial \alpha_R} \right|_{TD} = 111.5$$

$$786 \quad \Delta C_A^{HDDM} = \left. \frac{\partial C}{\partial \alpha_A} \right|_{TD} = 66.7$$

$$787 \quad \Delta C_I^{HDDM} = \left. \frac{\partial C}{\partial \alpha_I} \right|_{TD} = 88.1$$

788

789 These results are graphically represented in Figure 5 (sixth column).



790

791 Figure 5: Schematic representation of the allocation of PM to its
 792 sources in the ammonia-limited example. The expected total PM is
 793 displayed in the grey bar on the left.

794

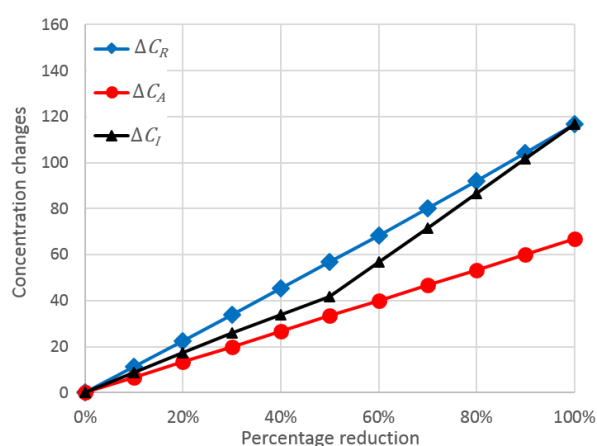
795

796 Comparative overview

797

798 The main difference with respect to the non-limited regime is the
 799 appearance of a triple interaction term that will also lead to

800 differences between the BF-TD and the DDM approaches, given the
 801 fact that the latter only accounts for first order terms.
 802 In comparison to the non-limited regime, the calculation of the
 803 concentration impacts resulting from different percentage of
 804 emission reduction shows non-linear trends (Figure 6). A
 805 discontinuity appears at 50% reduction for the abatement of
 806 industrial emissions. This discontinuity corresponds to a change of
 807 chemical regime. Below the 50% reduction level, the quantity of
 808 NH_3 is not sufficient to feed the reactions with NO_2 and SO_2 (with
 809 no SO_2 reduction, 50 moles of NO_2 and 50 moles of SO_2 would
 810 require 150 moles of NH_3 but only 100 are available) while beyond
 811 this 50% reduction level the quantity of NH_3 is then enough to feed
 812 the reactions with NO_2 and SO_2 (with 50% SO_2 reduction, 50 moles
 813 of NO_2 and 25 moles of SO_2 requires 100 moles of NH_3).
 814
 815



816
 817 Figure 6: Evolution of the concentration changes resulting from
 818 different percentage of source abatement (Top down approach) for
 819 the three sectors (Residential, Agricultural and Industrial).
 820

821
 822 The methodologies presented in this section aim at decomposing the
 823 impact of an ensemble of sources into different terms attributed to
 824 each individual sources. The terms computed by methodologies
 825 designed for source apportionment (like TAG) are named source

826 contributions. Their sum is always equal to the combined impact of
827 all sources. On the other hand, the terms computed by sensitivity
828 analysis represent the emission change of each individual source and
829 their sum is equal to the combined impact of all sources only if the
830 relationship between emissions and concentrations is linear (see
831 Section 2). Grewe et al. (2010) and Grewe (2013) who used simple
832 differential equations to reproduce the ozone tropospheric chemistry,
833 also highlighted this point in their work. In non-linear situations, the
834 source contributions computed for source apportionment and the
835 source impacts computed for sensitivity analysis are different (see
836 Figure 5 where column 2 shows different results than column 3 or
837 4). Non-linearity also implies that the calculation of the source
838 impacts depends from the bounds used to estimate the concentration
839 changes (denoted “H” and “L” in Section 4). The BF-BU and BF-
840 TD (columns 3 and 4 in Figure 5) give different results because they
841 are not using the same reference level (“L” for the BU and “H” for
842 the TD as defined in Section 4). Moreover, the results depend from
843 the percentage of emission changes applied to calculate the source
844 impacts as demonstrated by the different source impacts computed
845 with the BF-TD for 100% and 25% emission reductions (columns 4
846 and 5 in Figure 5). We expect that lower percentage emission
847 reductions generate less non-linearity and lead to a better agreement
848 between the BF-TD and the DDM method (columns 5 and 6 in
849 Figure 5).

850 In synthesis, the second example illustrates that all the
851 methodologies tested to find source contributions and source
852 impacts give different results when the relationship emissions-
853 concentrations is non-linear. A quantification of the potential non-
854 linearities should therefore be the first step prior to source
855 apportionment or planning applications, to prevent any limitations in
856 their use. When non-linearity is mild, these limitations may however
857 be acceptable in the context of the other uncertainties inherent to
858 complex models.

859

860 **6. Conclusions**

861

862 In this work, we compared source apportionment and sensitivity
863 approaches and investigated their domain of application. While
864 sensitivity analysis refers to impacts to characterize the
865 concentration change resulting from a given emission change, source
866 apportionment aims to quantify contributions, by attributing a
867 fraction of the pollutant concentration to each emission source. In
868 the case of linear (or close to linear) relationships between
869 concentration and emissions, impacts and contributions are
870 equivalent (or close to) concepts. Source apportionment may then be
871 used for air quality planning purposes and vice versa, sensitivity
872 analysis may be used for quantifying sources contributions.

873

874 In many cases, however, linearity is not a valid assumption. In such
875 cases, sensitivity approaches cannot be used to retrieve source
876 apportionment information, unless non-linear interaction terms are
877 explicitly accounted for. On the other hand, source apportionment
878 approaches (e.g. tagged species approach) intrinsically account for
879 these non-linear interactions into their source contributions. But
880 because it mix interaction terms and impacts, which may react in
881 opposite directions, source apportionment should not be used to
882 evaluate the impact of abatement strategies

883 Even when using sensitivity analysis for planning, it is important to
884 note that, under non-linear conditions, the calculated impacts will
885 only provide information for the exact conditions that are
886 considered. Impacts for an emission reduction of 50% are only valid
887 for exactly that percentage of reduction and extrapolation to air
888 quality planning with any other emission reduction levels would be
889 inappropriate, unless additional scenarios are tested. Along the same
890 line of reasoning, the importance of the non-linear interaction terms
891 (among precursors) should be quantified as well when assessing the
892 impact of more sources or precursors. Finally, these non-linear
893 interaction terms are in most cases not constant with the emission
894 reduction intensities, which would imposes the further need to
895 quantify them for different levels of emission reduction. Calculating
896 sensitivities and interactions at various level of emission reductions
897 seems the only alternative when non-linearities are important. In this
898 respect, new approaches like the path-integral methodology

899 proposed by Grewe et al. (2012) might represent a powerful
900 approach.

901

902 Fortunately, not all cases are so complex to require the full
903 quantification of all non-linear interaction terms. Thunis et al. (2015)
904 showed that for yearly average relationships between emission and
905 concentration changes, linearity is a realistic assumption, implying
906 the possible use of source apportionment and sensitivity analysis for
907 both purposes. Some integrated assessment tools (e.g. GAINS,
908 SHERPA) take advantage of this assumption to retrieve source
909 apportionment information from calculated CTM sensitivities.
910 Although non-linearities are important for short-term time averages
911 (e.g. daily means, episodes), they are likely not associated to every
912 process. For instance, non-linear interactions are expected to be
913 more relevant for secondary pollutants, especially under limited
914 regimes. The challenge consists, therefore, in screening the system
915 for significant non-linearities and account for them by calculating
916 explicitly the relevant non-linear interaction terms.

917

918 One main strength of source apportionment approaches is to provide
919 contribution estimates that can be cross validated with source
920 apportionment derived from measurements (i.e. receptor modelling,
921 for a detailed description see e.g. Belis et al., 2013). This step is
922 crucial for the evaluation of chemistry transport models.

923

924 **7. Code availability**

925 No specific code is attached to this work as all presented examples
926 can easily be replicated.

927

928

929 **References**

930

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