Response to the Editor

We thank the editor for additional comments to the manuscript. We have addressed all the suggestions for changes below in a point-by-point response, and hope that the manuscript is now ready for publication.

• p.6,l.24 - p.7,l.7: The paragraph is very hard to understand. This is mostly because one never knows, which species is hosted in which media. It would have been better to follow the request of referee #2 to add identifers like SO2(g) to the species name. E.g., what is "nucleation of SO4", if -as you claim- you always mean the particulate sulfate when using SO4. To be precise, please at the phase to all species names: use "(g)" for gas and "(aq),(l) or (p)" for aqueous, liquid or particulate at your convenience.

To address this comment we have changed the paragraph from:

"The production of sulfate aerosol (SO4) in CAM4-chem and CAM5-chem is also parameterized differently. In CAM4-chem, SO4 is produced directly from sulfur dioxide (SO2) by oxidation through heterogeneous reactions on aerosols. SO2 is referred to as gas-phase species throughout the paper. In CAM5-chem, sulfate aerosols are assumed to be partially neutralized by ammonia (NH3), in the form of ammonium hydrogen sulfate (NH4HSO4). Sulfates are produced via sulfuric acid (H2SO4) condensation on existing aerosols, where H2SO4 is formed by the oxidation of SO2. Both CAM4-chem and CAM5-chem include aqueous phase production of SO4 from SO2, with more than half formed by the hydroperoxyl (HO2) uptake and subsequent hydrogen peroxide (H2O2) oxidation in clouds (Liu et al., 2012). In addition, CAM5-chem includes nucleation of SO4, which contributes less than 1% to the production of SO4 mass but is an important source of aerosol number. Also, while in CAM4-chem sulfur oxides emissions are in the form of SO2 only, in CAM5, 2.5% of SO2 is emitted in the form of SO4."

To:

"The production of sulfate aerosol (SO₄) in CAM4-chem and CAM5-chem is also parameterized differently. In this paper we always consider SO₄ in solid particle phase, SO₄ (p), and sulfur dioxide (SO₂) and sulfuric acid (H2SO₄) in CAM5, in the gas-phase, SO₂ (g) and H2SO₄ (g), if not explicitly noted differently. In CAM5-chem, sulfate aerosols are assumed to

be in the form of ammonium hydrogen sulfate (NH4HSO4 (p)), considering partial neutralization by ammonia (NH3), since NH3 and NH4 cycles are not explicitly treated in this version. In CAM4-chem, SO4 is produced directly from SO2 by oxidation through heterogeneous reactions on aerosols. In CAM5-chem, sulfates are produced via H2SO4 condensation on existing aerosols, where H2SO4 is formed by the oxidation of SO2. Both CAM4-chem and CAM5-chem include aqueous phase production of SO4 from SO2(aq) with more than half formed by the hydroperoxyl (HO2) uptake and subsequent hydrogen peroxide (H2O2) oxidation in cloud droplets (Liu et al., 2012). In addition, CAM5-chem includes homogeneous nucleation of sulfate particles from H2SO4 gas, which contributes less than 1% to the production of SO4 mass but is an important source of aerosol number. Also, while in CAM4-chem sulfur oxides emissions are in the form of SO2 only, in CAM5, 2.5% of SO2 is emitted in the form of sulfate aerosol."

- p.13, l.22: temperatures -> temperature done
- p.14, l.20: remove "macro" in macrophysics
 We changed: "The fraction of low clouds in all configurations varies between 34% and about 60% (Table1) and are caused by the different parameterizations of cloud macrophysics with some contribution from the cloud microphysics, but also by differences in the assumed minimum relative humidity values that allow clouds to form."
 To

"The fraction of low clouds in all configurations varies between 34% and about 60% (Table1) and are caused by the different parameterizations of cloud fraction and cloud condensation with some contribution from the cloud microphysics. Differences exist in the assumed minimum relative humidity values that influence where clouds form."

- p.15, l.26 p.15, l.2: I do not see how reduction in SAD produces less LNOx
 - We remove the phrase: "in additions to changes in lightning NOx (LNOx)"
- p.15, l.22: AOA was not introduced done
- p.15, l.23: space after "10 hPa" is missing done
- p.16, l.5: BDC was not introduced.

done

- p.18, l.5/6: leave out the sentence about aircraft observations here, it is more confusing than helpful.

 done
- p.18, l.15-17. Here I disagree. The model does not reproduce sulfate aerosol measurements. Only for the height of 4-5 km this might be true.
 - We agree and add "for altitudes between 4-6km"
- p.18, l.25: From Fig. 9 I would not say that BC is overestimated overall.
 - The text says that all model configurations overestimate background BC. We add change this to: "All model configurations overestimate background BC (1 μ g/m3 or less)."
- p.19,l.4: Is there any concrete proof that it is BC plumes which are underestimated?
 - This was discussed in earlier studies, as referenced at the end of the sentence.
- p.19, l.12: "An overall comparison of aerosol .." -> "A comparison of overall aerosol .." done
- p.20,l.8: "a largest" -> "the largest" done
- p. 20, l.24/25: removes first "temperatures" done
- p. 21: I think there is an issue with the references of the Figures. Fig. 13 is only referenced in the introduction (line 5), but not thereafter. Which suggests that Figure 13 could be deleted. From the content of your discussion I assume, it should be referenced in line 20ff, and in the paragraph starting line 24. If this is correct, this leaves the issue, that Figures should be numbered in the order in which they are referred to. Thus Figs. 13 and 14 should be switched.

 In the revised version of the manuscript we now refer to Figure 13 in this sentence in line 9 and therefore before Figure 14, and therefore if should not be switched: "Near surface ozone at 900 hPa is for the most part within the range of variability of ozonesonde observations in both SD and FR configurations (Figs. 12 and 13, top row)" and also later in line 24, as suggested.
- p.22, l.6: delete the second opening bracket.

- We closed the first open bracket.
- p.22, l.12: this is only true for northern hemisphere We add NH to the text.
- p.22, l.13: second and third column (first is HIPPO data) this is changed to second and third column
- p.23, l.22: up to 80%

 The range of the plot only reaches to 60%. We change the sentence to: "(up to 40% in mid and high latitudes and up to 60% in winter in the Tropics)"
- p.24, l.9-16: The adjustment of SAD shows almost no effect for most of the cases. You pick the only one, which shows a little effect. This is not convincing to me.
 - We reword: "Larger SAD values in CAM4-chem result in a faster transition of NOx to NOy and therefore reduced PAN production, as shown in the example in Fig. 21, top left panel, for SD-CAM5-chem. However, in the FR versions and for the other cases shown in Fig. 21, adjustments of the SAD between CAM4-chem and CAM5-chem configurations are less important."

"One would expect that larger SAD values result in a faster transition of NOx to NOy and therefore reduced PAN production. However, adjustments of the SAD between CAM4-chem and CAM5-chem configurations are less important in most cases, as shown in Fig. 21."

- p.25, l.10: bracket "Fig. 23, left and middle panel" done
- p.25, l.20: bracket "Fig. 23" done
- p.26: Why do you write the simulation marker differently (all lower case letters) in the subscripts and not CAM4chem as everywhere als? Changed to CAM4-chem.
- p.26, l.16: What do you mean with "The spread the annual in LNOx production"???
 - "The spread the annual in LNOx production ..." is changed to "The annual variability in LNOx production ..."
- p.27, l.23: Do you really mean model version? It seems to be more a configuration thing.
 - We change it to "in the discussed simulations"
- p.28, l.19: "a result of a coarse resolution"

done

- Appendix A: The indentation of the reactions is very strange and thus nearly unreadable. Please correct it. done
- Table 3. As you use SO4 as abbreviation for sulfate aerosol (similar to OC/BC etc.) it should be also in the abbreviation table. done
- Table 3, MOZART: Model for ... done
- Fig. 3: From the caption and the annotations it is not clear what is depicted. Why are OH, NOx and H2O in %, while Ozone, nitric acid and CO are in ppb? Additionally Ozone difference in Fig. 2 was in %. From the unit I would assume that it is Lighting NOx production and not Lighting NOx, but even this is inconsistent with ozone production in %. Please be consistent!
 - We have changed the figure to show absolute changes for all species
- Fig. 5: Why use millibars instead of hPa here? SI-units should be the standard.

done

- Fig. 6: The upper right corner of SAD SOA looks quite artificial. This may be because of a model specification for the polar region, we have not investigated SAD SOA in the stratosphere
- Figs. 8 / 21, caption: use either "indicate" or "describe" done
- Figs. 9 / 18 / 22: You never use the acronym CAM5-MAM3-chem (even not in the caption of these Figures), so please remove MAM3 in the pictures themselves.
 - We have removed MAM3 from those figures
- Fig. 10: It is strange to show the same picture in two panels of the same Figure twice. I was looking for the difference
- Fig. 10: The x-axis annotations are unreadable (less would be good). We have updated this figure and improved the annotations for the x-axis. We are just showing differences now and not the observations. The figure caption has been changed accordingly.
- Fig. 10, caption: "over only" -> "only over"? done
- Figs. 12 / 15: the numbers are often not readable in the pictures..
- Figs. 12 / 15, caption: "different numbers are correspondent .." -> "different numbers correspond"

done

- Figs. 12 / 15, caption: The meaning of the numbers is hard to read. Readablility would be greatly enhanced by adding (outlayed as a small table) a third \picture or panel for each of the regions listing the meaning of the numbers.
 - I have played around to improve this figure, but the problem is that increasing the size of the numbers will not make it more readable, since they overlay each other, while making them smaller is also not helpful. We think it is important that outliers are detected, which is possible in this plot. Also, for the figure caption, including the caption in the text makes this unreadable. We found that the figure as is, is the best presentation.
- Fig. 13 / 16: at least at current size, the coloured numbers are not easy to read. Please ensure, that the numbers are readable in the final version.
 - We have increased the font of the numbers to make them more readable
- Figs. 17 / 20: From the caption I understand that "observation model" is shown. Your discussion implies that the sign is the other way round. Please make unmistakably clear what is shown.

 We change the caption to: "Relative difference between different model configurations and aircraft observations
- Fig. 19: Somehow the longitude axis got lost ...

 We have not plotted longitude in this plot, since it does not seem to be needed.
- Fig. 23-25: At print-size, symbols are not distinguishable.
- Fig. 23: Symbol annotation is missing
 We updated both plots to make the symbols readable and added the missing annotation.

Manuscript prepared for Geosci. Model Dev. Discuss. with version 2014/09/16 7.15 Copernicus papers of the LATEX class copernicus.cls. Date: 18 April 2015

Description and evaluation of tropospheric chemistry and aerosols in the Community Earth System Model (CESM1.2)

S. Tilmes¹, J.-F. Lamarque¹, L. K. Emmons¹, D. E. Kinnison¹, P.-L. Ma², X. Liu³, S. Ghan², C. Bardeen¹, S. Arnold⁴, M. Deeter¹, F. Vitt¹, T. Ryerson⁵, J. W. Elkins⁵, F. Moore⁵, R. Spackman⁶, and M. Val Martin⁷

Correspondence to: S. Tilmes (tilmes@ucar.edu)

¹National Center for Atmospheric Research, Boulder, Colorado, USA

²Pacific Northwest National Laboratory, Richland, Washington, USA

³University of Wyoming, USA

⁴School of Earth and Environment, University of Leeds, Leeds, UK

⁵NOAA ESRL Chemical Sciences Division, Boulder, Colorado, USA

⁶NOAA ESRL Science and Technology Corporation (STC), Boulder, Colorado, USA

⁷Department of Chemical and Biological Engineering, The University of Sheffield, Sheffield, UK

Abstract

The Community Atmosphere Model (CAM), version 5, is now coupled to extensive tropospheric and stratospheric chemistry, called CAM5-chem, and is available in addition to CAM4-chem in the Community Earth System Model (CESM) version 1.2. The main focus of this paper is to compare the performance of configurations with internally derived "free running" (FR) meteorology and "specified dynamics" (SD) against observations from surface, aircraft, and satellite, as well as understand the origin of the identified differences. We focus on the representation of aerosols and chemistry. All model configurations reproduce tropospheric ozone for most regions based on in-situ and satellite observations. However, shortcomings exist in the representation of ozone precursors and aerosols. Tropospheric ozone in all model configurations agrees for the most part with ozonesondes and satellite observations in the Tropics and the Northern Hemisphere within the variability of the observations. Southern Hemispheric tropospheric ozone is consistently underestimated by up to 25%. Differences in convection and stratosphere to troposphere exchange processes are mostly responsible for differences in ozone in the different model configurations. Carbon monoxide (CO) and other volatile organic compounds are largely underestimated in Northern Hemisphere mid latitudes based on satellite and aircraft observations. Nitrogen oxides (NO_x) are biased low in the free tropical troposphere, whereas peroxyacetyl nitrate (PAN) is overestimated in particular in high northern latitudes. The present-day methane lifetime estimates are compared among the different model configurations. These range between 7.8 years in the SD configuration of CAM5-chem and 8.8 years in the FR configuration of CAM4-chem and are therefore underestimated compared to observational estimations. We find that differences in tropospheric aerosol surface area between CAM4 and CAM5 play an important role in controlling the burden of the tropical tropospheric hydroxyl radical (OH), which causes differences in tropical methane lifetime of about half a year between CAM4chem and CAM5-chem. In addition, different distributions of NO_x from lightning explain about half of the difference between SD and FR model versions in both CAM4-chem and CAM5-chem. Remaining differences in the tropical OH burden are due to enhanced tropical ozone burden in SD configurations compared to the FR versions, which are not only caused by differences in chemical production or loss, but also by transport and mixing. For future studies, we recommend the use of CAM5-chem configurations, due to improved aerosol description and inclusion of aerosol-cloud interactions. However, smaller tropospheric surface area density in the current version of CAM5-chem compared to CAM4-chem results in larger oxidizing capacity in the troposphere and therefore a shorter methane lifetime.

1 Introduction

The Community Earth System Model (CESM) is a comprehensive model that couples different independent models for atmosphere, land, ocean, sea-ice, land ice, and river runoff (e.g., Neale et al., 2013; Lamarque et al., 2012). It can be used in various configurations, depending on the use of different components and the coupling between them. The atmospheric component of CESM, the Community Atmosphere Model (CAM), has the capability of including chemistry of varying complexity. Default CESM configurations used for long-term climate model simulations usually include prescribed chemical fields in the atmosphere using monthly averages. To produce those prescribed input fields, simulations with a detailed representation of chemistry and aerosol processes are required. Furthermore, non-linear interactions between chemistry and aerosols in the atmosphere are important for chemistry-climate interactions (e.g., Lamarque et al., 2005; Isaksen et al., 2009) or for the simulation of air quality.

In CESM version 1.2, CAM version 5 (CAM5), extensive tropospheric and stratospheric chemistry, referred hereafter to as CAM5-chem, has been successfully implemented. The performance of CAM version 4 (CAM4) with interactive chemistry, referred to as CAM4-chem, has been discussed in Lamarque et al. (2012). In this study, a similar setup of both CAM4-chem and CAM5-chem allows the comparison of both versions and their performance in comparison to observations. The two atmospheric configurations CAM4-chem and CAM5-chem differ in various aspects, including the treatment of cloud, convection, turbulent mixing, and aerosol processes (e.g., Neale et al., 2013; Gent et al., 2011; Kay et al.,

2012; Liu et al., 2012), whereas the gas-phase chemistry is identical. Resulting differences in dynamics, clouds, precipitation, and radiation, will alter chemical reactions in the gas, aqueous, and aerosol phase, and removal processes, and therefore the chemical composition of the atmosphere in these configurations.

In addition to exploring differences between the two atmospheric model versions using internally produced meteorology, we also perform simulations in which the meteorology (temperature, winds, and surface fluxes) is nudged towards meteorological analysis (or reanalysis) fields to reduce differences in the dynamics of the two configurations. Further, two slightly different aerosol schemes of the modal aerosol model (MAM) are tested in CAM5-chem, the 3-mode version (MAM3) (Liu et al., 2012) and the 4-mode version (MAM4) (Liu et al., 2015). In addition, sensitivity studies are performed to explore differences in the oxidizing capacity of the atmosphere and therefore in tropospheric methane lifetime in the different model configurations. In this way, relationships between methane lifetime, aerosol and chemistry composition, and meteorological parameters are explored.

A comprehensive evaluation of all configurations is performed, using a set of present-day observational climatologies of different chemistry and aerosol species from ground-based, aircraft and satellite observations. Strengths and weaknesses of the various model configurations are discussed. Evaluation tools for trace gases and aerosols developed in this study are merged to the Atmospheric Model Working Group (AMWG) diagnostics package, and are available to the community on the CESM website (https://www2.cesm.ucar.edu/working-groups/amwg/amwg-diagnostics-package).

15

This paper is structured as follows. Section 2 gives details of the model configurations and experiments performed for this study. Section 3 describes present-day climatological datasets used in this study to evaluate the model. Model-to-model differences in dynamics, chemistry and aerosols, and global budgets are discussed in Sect. 4.1. A comprehensive evaluation of chemistry and aerosols, based on satellite and in-situ observations is performed in Sect. 4.2. We discuss reasons for differences in tropospheric methane lifetime of the different model configurations, an indicator of the oxidizing capacity of the atmosphere in Sect. 5. A summary and discussion of the results is given in Sect. 6.

2 Model configurations and experiments

The presented results are based on output from simulations performed System with Community Earth Model (CESM) (https://www2.cesm.ucar.edu/models/current). All model simulations are performed with prescribed sea surface temperatures and sea-ice distribution data for present-day climatological conditions, since we focus on the atmospheric component. Dry deposition of gases and aerosols are implemented in the Community Land Model (CLM) (Oleson, 2010) as described in Lamarque et al. (2012). For all experiments CLM Version 4.0 was used. CESM 1.2 can also include online calculation of biogenic emissions in CLM using the Model of Emissions of Gases and Aerosols from Nature (MEGAN) version 2.1 (Guenther et al., 2012). In this study, biogenic emissions are prescribed (see below) to ensure having the same amount of emissions in all configurations, and interactive bio-geo-chemistry was not included.

CAM4-chem uses 26 vertical levels while CAM5-chem uses 30, and they both have a model top around 40 km. The horizontal resolution of performed simulations is $1.9^{\circ} \times 2.5^{\circ}$ and we use the finite volume dynamical core. An important difference between the two atmospheric models is the cloud microphysics, which in CAM4-chem predicts only the mass concentrations of the cloud species, but in CAM5-chem predicts the number as well as mass concentrations. CAM5-chem consequently treats the microphysical effect of aerosols on clouds (Ghan et al., 2012), while in CAM4-chem aerosols impact physics and dynamics only through their interaction with radiation.

CAM4-chem and CAM5-chem further differ in the parameterization of aerosols. CAM4-chem runs with a bulk aerosol model (BAM), which considers a fixed size distribution of externally-mixed sulfate, black carbon (BC), organic carbon (OC), sea-salt and dust (Tie, 2005). Sea-salt and dust are described using four different bins. In CAM4-chem, the formation of secondary organic aerosols (SOA) is coupled to chemistry. SOA are derived using the 2-product model approach using laboratory determined yields for SOA formation from

monoterpene oxidation, isoprene and aromatic photooxidation, as described in Heald et al. (2008).

The current standard CAM5 model version, and therefore also CAM5-chem, uses the modal aerosol model with three modes (MAM3) (Liu et al., 2012). The aerosol components, including BC, primary organic matter (POM), SOA, sea-salt, dust, and sulfate, are internally mixed in each lognormal mode, and the aerosol mass and the total number in each mode are predicted. CAM5-chem is also tested with the 4-mode version, MAM4, called CAM5-MAM4-chem from here on. The main difference between these two modal versions used here is the representation of BC and OC. In MAM3 all BC and OC is assumed to be aged and hence is emitted directly into the accumulation mode with other soluble aerosol species, whereas MAM4 emits the BC and OC in the primary carbon mode and represents the aging process of BC and OC from the primary carbon mode to the accumulation mode, as done in BAM. For the SOA production in CAM5-chem, mass yields of several biogenic and anthropogenic Volatile Organic Compounds (VOCs) are prescribed. The resulting condensable secondary organic gas reversibly and kinetically partitions to the aerosol phase, as described in detail in Liu et al. (2012). The different approach in CAM5chem than CAM4-chem results in much larger burden of SOA, as shown in Tsigaridis et al. (2014). The dust emissions are calibrated so that the global dust aerosol optical depth (AOD) is between 0.025 and 0.030 (Mahowald et al., 2006). Further, sea salt emissions are calibrated to present-day conditions so that the global mean AOD (for all species) are within the reasonable range. Those values have been evaluated in Liu et al. (2012), who have shown that the difference between model simulations and observations are generally within a factor of two.

The production of sulfate aerosol (SO₄) in CAM4-chem and CAM5-chem is also parameterized differently. In CAM4-chemthis paper we always consider SO₄ in solid particle phase, SO₄ is produced directly from (p), and sulfur dioxide (SO₂) by exidation through heterogeneous reactions on aerosols. is referred to as and sulfuric acid (H₂SO₄) in CAM5, in the gas-phase-species throughout the paper, SO₂ (g) and H₂SO₄ (g), if not explicitly noted differently. In CAM5-chem, sulfate aerosols are assumed to be partially neutralized

by ammonia (), in the form of ammonium hydrogen sulfate (NH₄HSO₄). Sulfates (p)), considering partial neutralization by ammonia (NH₃), since NH₃ and ammonium NH₄⁺ cycles are not explicitly treated in this version. In CAM4-chem, SO₄ is produced directly from SO₂ by oxidation through heterogeneous reactions on aerosols. In CAM5-chem, sulfates are produced via sulfuric acid (H₂SO₄) condensation on existing aerosols, where H₂SO₄ is formed by the oxidation of SO₂. Both CAM4-chem and CAM5-chem include aqueous phase production of SO₄ from -SO₂ (aq) with more than half formed by the hydroperoxyl (HO₂) uptake and subsequent hydrogen peroxide (H₂O₂) oxidation in clouds cloud droplets (Liu et al., 2012). In addition, CAM5-chem includes nucleation of homogeneous nucleation of sulfate particles from H₂SO₄ gas, which contributes less than 1% to the production of SO₄ mass but is an important source of aerosol number. Also, while in CAM4-chem sulfur oxides emissions are in the form of SO₂ only, in CAM5, 2.5% of SO₂ is emitted in the form of sulfate aerosol.

Furthermore, the representation of removal processes is different in CAM4-chem and CAM5-chem. In CAM4-chem all of the aerosol in the cloudy fraction of the grid cell is assumed to reside within cloud droplets and is removed in proportion to the cloud water removal rate. In CAM5-chem the mass and number fraction of the cloud-borne aerosol is determined from the aerosol activation parameterization (Ghan and Easter, 2006), so that smaller particles are not removed by nucleation scavenging.

20

CAM4-chem has been run and tested with comprehensive tropospheric and stratospheric chemistry (Lamarque et al., 2012). The chemical mechanism is based on the Model for Ozone and Related chemical Tracers (MOZART), version 4 mechanism for the troposphere (Emmons et al., 2010), extended stratospheric chemistry (Kinnison et al., 2007), further updates as described in Lamarque et al. (2012), and additional reaction rate updates following JPL 2010 recommendations (Sander et al., 2011). In CESM1.2 CAM4-chem, the lumped aromatic ("TOLUENE") was replaced with the specific species benzene, xylene and toluene, along with simplified oxidation products for the two new species, to accommodate the 2-product formation of SOA (new reactions listed in Appendix A). These changes do not have an impact on the chemical performance of the model.

As in CAM4-chem, CAM5-chem couples tropospheric aerosols to chemistry through heterogeneous reactions, as listed in Lamarque et al. (2012, Table 4). Tropospheric heterogeneous reactions of chemical species are parameterized based on aerosol surface area density (SAD) and therefore depend on the overall aerosol loading. The total tropospheric SAD in both model configurations is derived using the mass and size distributions of ammonium sulfates, black carbon, and organic aerosols. The contribution of very small particles, such as the Aitken mode in MAM3 and the primary carbon mode in MAM4, to the SAD are neglected in the the model calculation of surface area density. Further, sea-salt and mineral dust aerosols do not contribute to SAD in either model version, as heterogeneous reactions are not assumed to occur on these surfaces. Since reactions on very small particles are important, this may lead to an underestimation of SAD in the model.

For all simulations, model configurations simulate wet deposition of gas species using the Neu and Prather (2012) scheme, including a bug fix to CESM1.2, where the SO_2 Henry's law coefficient has been updated, resulting in reduced washout rates. This fix resulted in an increased burden of SO_4 in CAM4-chem, which has been adjusted by increasing the inand below-cloud solubility factor of SO_4 from 0.3 to 0.4. In addition, improved calculations of dry deposition velocities for gas species, as discussed in Val Martin et al. (2014), are added to this study, which results in an improved representation of surface ozone, as discussed below.

Experiments

Two different configurations of both CAM4-chem and CAM5-chem are used in this study. In the free running (FR) version the meteorology and dynamics are internally derived. We also run CAM4-chem and CAM5-chem in a specified dynamics (SD) version of the model, called SD-CAM4-chem and SD-CAM5-chem, respectively. In this configuration, the internally derived meteorological fields are nudged every time step (30 min) by 10 % towards analysis fields (i.e., a 5 h Newtonian relaxation time scale for nudging) from the Modern-Era Retrospective Analysis For Research And Applications (MERRA) reanalysis product (http://gmao.gsfc.nasa.gov/merra/) (Rienecker et al., 2011), regridded to the model horizon-

tal resolution. The SD model version adopts the vertical levels of the analysis data up to the top of the model (around 40 km), resulting in 56 vertical levels for both CAM4-chem and CAM5-chem simulations; see Lamarque et al. (2012) and Ma et al. (2013) for details. For the SD simulations, we use meteorological analysis for the years 2000 to 2010.

Emissions and prescribed chemical fields for longer-lived substances follow the protocol defined by the Chemistry Climate Model Initiative (CCMI) hindcast simulations for the year 2000 (Eyring et al., 2013), which are repeated for all the simulated model years for both FR and SD configurations. In particular, greenhouse gases are from Meinshausen et al. (2011), surface mixing ratios of ozone depleting substances are taken from WMO (2010, Table 5-A3), anthropogenic and biofuel emissions are from the MACCity emission data set (Granier et al., 2011), and biomass burning emissions are taken from the Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP) historical emissions dataset (Lamarque et al., 2010). Biogenic emissions are prescribed in this study for all model configurations using a climatology based on MEGAN version 2.1, with the same emissions for all model experiments; carbon monoxide (CO): 1053 Tg yr⁻¹, isoprene: 525 Tg yr⁻¹, monoterpene: 97 Tg yr⁻¹, and methanol: 170 Tg yr⁻¹. All experiments use the same solar forcing, with lower boundary conditions fixed for the year 2000.

Two additional sensitivity experiments are performed to test differences between CAM4-chem and CAM5-chem that may be caused by differences in the aerosol description in the model, in particular the amount of tropospheric SAD in the different configurations. CAM5-chem simulates significantly lower SAD than CAM4 (as discussed in Sect. 4.1.2). We perform an additional CAM5-chem (CAM5-chem*) simulation where SAD is increased by a factor of 1.5 to match the averaged tropospheric SAD amount that is simulated in CAM4-chem. We also perform SD-CAM5-chem* that matches averaged tropospheric SAD of the SD-CAM4-chem simulation, requiring SAD to increase by a factor of 1.9. And finally, we perform a simulation that uses the MAM4 modal scheme, CAM5-MAM4-chem, as described above. An overview of the setup and global model diagnostics of the different model configurations is given in Table 1.

3 Present day climatological datasets

To evaluate the performance of the different model configurations, we made use of several satellite and in-situ chemical datasets. We use present-day climatological datasets with a focus on the troposphere that have been derived from observations between 1995 and 2012.

3.1 Satellite climatologies

The comparison of the model simulations to satellite observations provides a global picture on the representation of CO and ozone columns. To evaluate tropospheric and stratospheric column ozone in the model simulations, we compare the model to a present-day column ozone climatology compiled by Ziemke et al. (2011). This climatology was derived by combining retrievals from the Aura Ozone Monitoring Instrument (OMI) and Microwave Limb Sounder (MLS) observations over the period between October 2004 and December 2010. The monthly-mean thermal tropopause is used to separate between tropospheric and stratospheric ozone for the model results and satellite climatology.

For comparison with CO, a new climatology is compiled based on Measurements of Pollution in The Troposphere (MOPITT) Version 6 Level 3 data, using the multispectral (thermal-infrared plus near-infrared) total column product. This monthly mean gridded climatology on a $1^{\circ} \times 1^{\circ}$ horizontal resolution includes data between 2003 and 2012. Only daytime MOPITT data were analyzed. The Version 6 MOPITT product is similar to the validated Version 5 product (Deeter et al., 2013) with several differences (Deeter et al., 2014). The V5 products relied on a priori CO concentrations based on the MOZART chemistry transport model and National Centers for Environmental Prediction (NCEP) analysis fields. The a priori for V6 products is based on CAM4-chem simulations for the period from 2000–2009 (Lamarque et al., 2012) and the retrieval processing exploits the MERRA reanalysis product. Finally, geolocation (latitude and longitude) data are more accurate for V6 product as the result of a correction for a slight misalignment between the MOPITT instrument and the Terra spacecraft. The V6 product is described in more detail in a User's Guide available on

the MOPITT website (http://www2.acd.ucar.edu/mopitt/publications). Monthly mean Level 3 MOPITT a priori and averaging kernels are applied to monthly mean model results to account for the a priori dependence and vertical resolution of the MOPITT data. CO columns are derived for altitudes between surface and 100 hPa.

For the comparison of AOD, we use a $1^{\circ} \times 1^{\circ}$ monthly averaged climatology for present-day AOD at 550 nm, derived using various satellite data including observations from the Aerosol Robotic Network (AERONET) (Kinne, 2009).

3.2 Ozonesonde climatology

For a detailed evaluation of tropospheric ozone profiles and seasonality, a present-day ozonesonde climatology is used (Tilmes et al., 2012). This climatology covers available ozonesonde observations between 1995 and 2011 for 42 stations around the globe. Ozonesonde observations do agree reasonably well with surface and aircraft observations (Tilmes et al., 2012). Maximum summer time ozonesonde data over Eastern US is biased high by about 10 ppb compared to surface observations, but otherwise, the ozone climatology provides reliable ozone vertical profiles for different seasons and regions. In this study, monthly mean model results are interpolated to the locations of the data and aggregated over defined regions, as suggested in Tilmes et al. (2012).

3.3 Aircraft climatologies

For the evaluation of various chemical species, averaged profiles from various aircraft campaigns between 1995 and 2010 were derived for different regions and seasons around the globe. Details of aircraft campaigns included between 1995 and 2010 are given in Table 2. More details, including information of earlier aircraft campaigns, are provided on https://www2.acd.ucar.edu/gcm/aircraft-climatology. As discussed in Emmons et al. (2000), for each aircraft campaign, regions with high frequency occurrence of vertical profiles from the aircraft are identified. Mean and median profiles of available species are compiled over these regions, as well as percentiles of the distribution with a 1 km vertical resolution. Pro-

files that are outliers of the distribution were removed. Following this approach, we extended the existing climatology as described in Emmons et al. (2000), to include additional aircraft campaigns up to 2010.

The largest sampling frequency of aircraft observations included in this study is over Europe and the US during spring and summer. For each observed regional profile, monthlymean model results are averaged over the location and months of the observations. It is assumed that these regional profiles represent typical background conditions. However, one has to keep in mind that aircraft campaigns often target specific atmospheric conditions that may not be captured in multi-year average model results. Nevertheless, the combination of the numerous aircraft campaigns provides a general overview on the behavior of the chemistry in the model. In this way, aircraft data provide a very powerful evaluation tool, because various species were observed at the same time during the flight and can be evaluated side by side. A comparison is performed for ozone (O₃), CO, nitrogen oxides (NO_x), and peroxyacetyl nitrate (CH₃COO₂NO₂ or PAN), selected hydrocarbons, SO₂ and sulfate aerosol for selected aircraft campaigns. In addition, we averaged profiles over certain altitude intervals and grouped them into four regions and four seasons, to identify systematic differences between models and observations.

A data set derived during the HIAPER (High-Performance Instrumented Airborne Platform for Environmental Research) Pole-to-Pole Observations (HIPPO) campaigns (Wofsy et al., 2011) is available for model evaluation purposes (Wofsy et al., 2012). During the campaigns, profiles from 85° N–65° S over the Pacific Ocean and North America were sampled in January and November 2009, March/April 2010, June/July 2011, and August/September 2011. Each of the campaigns sampled very similar flight tracks over the Pacific and North America, which provides information for comparing similar regions in different seasons (Wofsy et al., 2011). For this paper, we use O₃, BC, and PAN data (Schwarz et al., 2013; Wofsy et al., 2011). The aircraft profiles sampled during different HIPPO campaigns were averaged over 5° latitude intervals along the flight path over the Pacific Ocean to produce a gridded dataset that can be easily compared to model output. Likewise, model results are binned over the same latitude regions as done for the aircraft

observations. Here, we compare the observations to monthly mean model data that are aligned with the months of the corresponding campaign. It has to be kept in mind that the HIPPO dataset, even though observing the background atmosphere over the Pacific, is influenced by the specific situation for the particular year. This climatological comparison has shortcomings, in particular because the emissions of the particular year were not considered.

3.4 Surface observations

We use two sets of surface observations in this study. Surface observations from the United States Interagency Monitoring of Protected Visual Environments (IMPROVE) dataset (http://vista.cira.colostate.edu/improve/), (Malm, 2004), is used for years 1998–2009, to compare sulfur dioxide and sulfate aerosol with the model results. The IMPROVE network includes 165 sites in the US. Major fine particles (with diameter < 2.5 micro meter) are monitored including aerosol species, sulfates, nitrates, organics, light-absorbing carbon, and wind-blown dust. IMPROVE sites are located in rural environments and therefore will not describe the conditions found in large urban areas.

Ozone surface observations are used to evaluate daily ozone concentration in our model configurations. Daily averages from available hourly surface ozone data were derived from the Clean Air Status and Trends Network (CASTNET) (http://java.epa.gov/castnet/) and the European Monitoring and Evaluation Programme (EMEP) network in Europe (http://www.emep.int/) for years 1995-2010, as shown in Tilmes et al. (2012).

4 Performance for different model configurations

4.1 Model-to-model comparison

Differences in the physics, including cloud and aerosol schemes between CAM4-chem and CAM5-chem (as described above), result in large differences in tropospheric surface area density, temperatures temperature, relative humidity and cloud fraction, with implications for

chemistry, in particular ozone. Additional differences in the vertical resolution of different model configurations influence convection and dynamics in troposphere and stratosphere and therefore atmospheric composition. The comparison of zonal and annual mean meteorological as well as chemical constituencies between different model versions helps to explain differences in ozone and other chemical tracers.

4.1.1 Dynamics and chemistry

CAM5-chem simulates more ozone in the stratosphere than CAM4-chem, most pronounced in high latitudes in the lower stratosphere. This is aligned with lower temperatures in the stratosphere in the tropics and mid-latitudes in CAM5-chem compared to CAM4-chem, resulting in reduced ozone destroying gas-phase chemistry. Further, lower ozone mixing ratios and a cold bias are present in CAM5-chem right around the tropical tropopause in comparison to CAM4-chem. Reduced ozone around the tropical tropopause can affect temperatures at the cold point and above (Bardeen et al., 2013).

Differences in zonal winds point to a weaker polar vortex in CAM5-chem compared to CAM4-chem, whereby zonal winds in CAM5-chem are more aligned with analysis fields than in CAM4-chem (not shown). Corresponding higher temperatures in the polar lower-most stratosphere are consistent with higher ozone mixing ratios in high latitudes due to a reduction in halogen activation.

Differences in the microphysics between CAM4-chem and CAM5-chem result in significantly larger relative humidity in the troposphere in mid and high latitudes in CAM5-chem compared to CAM4-chem (Fig. 1, as discussed in Bardeen et al. (2013)). The fraction of low clouds in all configurations varies between 34 % and about 60 % (Table 1) and are caused by the different parameterizations of cloud macrophysics fraction and cloud condensation with some contribution from the cloud microphysics, but also by differences. Differences exist in the assumed minimum relative humidity values that allow clouds to influence where clouds form. Differences in cloud fraction between different configurations impact photolysis rates in the lower troposphere and therefore ozone photochemistry (discussed below), and also precipitation and removal processes.

Large differences between CAM5-chem and CAM4-chem configurations are present in the tropospheric surface area density (SAD), as further discussed below. Those differences impact tropospheric chemistry, whereby less SAD in CAM5-chem results in the reduction of NO_x , OH, and therefore changes in CO and ozone production, in additions to changes in lightning (L), see further discussion in Sect. 5.

However, differences in dynamics between CAM5-chem and CAM4-chem have a stronger impact on ozone than differences in clouds and SAD, as shown in comparing SD-CAM5-chem and SD-CAM4-chem (Fig. 1, bottom row). In these two configurations, winds and temperatures are nudged to meteorological analyzed fields. Similarities in the meteorological fields lead to much smaller differences in ozone than between the FR versions, despite the large differences in relative humidity, clouds fraction, and SAD, which are similar to the differences between two free running model versions.

The impact of differences in dynamics for tropospheric chemistry is further supported in comparing CAM5-chem and SD-CAM5-chem (Fig. 2 and 3). In these two model simulations, differences in clouds and SAD are much smaller than between CAM4-chem and CAM5-chem. However, the FR version produces a significantly stronger polar vortex and lower temperatures in high latitudes than the SD version. SD simulations driven by MERRA temperatures are higher than the FR model versions. As shown in Bardeen et al. (2013), differences of the microphysics between different model versions determine the relative humidity in the model, and therefore the relationship between water and temperature. Warmer temperatures in SD-CAM5-chem compared to CAM5-chem therefore cause an increase in water vapor in the stratosphere.

Dynamical differences in the tropics and the stratosphere are investigated for the different model configurations in analysing the H_2O tape recorder (Mote et al., 1996) (Fig. 4) and stratospheric AOAage of air (AOA), as described in Garcia et al. (2011), (Fig. 5). The tropical vertical transport between 23° S and 23° N and 100 and 10 hPa is analyzed for different model configurations based on the magnitude and slope of the H_2O tape recorder (Fig. 4). The slope and magnitude of the tape recorder, as derived from MLS observations between 2005 and 2011 (Fig. 4, bottom row), is best reproduced by the SD configurations, even

though H₂O mixing ratios are too large in SD-CAM5-chem. CAM5-chem reproduces the magnitude of the tape recorder, while minimum H₂O mixing ratios are too low, and shows a reduced slope compared to SD-CAM5-chem. This points to a faster updraft of air masses above the TTL. CAM4-chem poorly simulates the slope compared to other model configurations, whereas SD-CAM4-chem shows a reasonable magnitude of the tape recorder in comparison to MLS observations. Consistent with the poor representation of the slope of the tape recorder compared to observations, CAM4-chem and CAM5-chem produce much shorter stratospheric AOA compared to the SD configurations (Fig. 5). This is consistent with a stronger BDC Brewer Dobson Circulation (BDC) in both free running model configurations and stronger stratosphere to troposphere exchange (STE) (Table 1). Slightly larger AOA values in the tropics and high latitudes are simulated in CAM5-chem compared to CAM4-chem configurations.

The comparison of chemical constituents in the two model configurations further supports a stronger tropical vertical transport in CAM5-chem compared to SD-CAM5-chem and stronger STE in high latitudes (Fig. 3). Stronger tropical vertical transport (mostly in deep convection) in CAM5-chem is evident due to higher mixing ratios in CO and lower mixing ratios of nitric acid in the upper tropical troposphere. The resulting higher CO mixing ratios in the upper troposphere together with increased lightning NO_x (LNO_x) production in mid-latitudes lead to greater ozone production, while reduced LNO_x production in the tropical belt reduces ozone production. Furthermore, increased nitric acid in addition to higher ozone mixing ratios in high northern latitudes point to more STE. Additionally, lower NO_x and CO values in the boundary layer in CAM5-chem indicate that increased STE rather than chemical processing results in larger ozone mixing ratios in CAM5-chem than SD-CAM5chem. Differences in low clouds between CAM5-chem and SD-CAM5-chem also impact chemistry and result in reduced ozone production in the boundary layer in CAM5-chem. Similar differences are present between CAM4-chem to SD-CAM4-chem, however, with smaller differences in STE in high latitudes compared to the CAM5-chem configurations (not shown).

4.1.2 Aerosol burden and Surface Area Density (SAD)

15

Optical depth and aerosol loading from the different model configurations are listed in Table 1. Total optical depth is somewhat smaller in CAM4-chem than in the CAM5-chem configuration, which is due to different amounts of internally derived sea-salt and dust emissions, but also differences in the sulfate burden in comparison to observations, as discussed in Sect. 4.2.1. The largest differences in aerosol burden between the configurations occur in the burden of SOA, with about 50 % larger values in CAM5-chem compared to CAM4-chem (as discussed above). The burden of organic matter and black carbon is slightly larger in CAM4-chem compared to CAM5-chem using MAM3, due to the different handling of these aerosols in the two configurations. More similar values of BC and OC in CAM4-chem are simulated in CAM5-MAM4-chem. Running 2 modes for BC in CAM5-MAM4-chem compared to CAM5-chem increases the BC burden by 37 % (see Table 1). SO₄ burdens in CAM4-chem are slightly larger than in CAM5-chem. This is because of the different way SO₄ formation and washout is parameterized, as described in Sect. 2.

Heterogeneous reactions on aerosol particles in the model do not directly relate to the aerosol burden, but rather depend on the amount of tropospheric SAD. SAD depends not only on aerosol burden or mass, but also on their size distribution. For the same aerosol burden, smaller particles provide a larger SAD than larger particles. Both the SD and FR version CAM5-chem simulate much smaller SAD than CAM4-chem. This has implications for chemistry and climate (see Sect. 5). The total tropospheric SAD in the model includes SAD from SO₄, nitrates, POM, SOA, and BC modes.

We compare the burden and SAD between SD-CAM5-chem and SD-CAM4-chem for SO₄, BC, and SOA (Fig. 6). Both magnitude and sign of the differences in burden do not agree with differences in SAD, which is caused by different description of the size distribution of aerosols in the two model versions. In CAM4-chem, BAM assumes a fixed mean radius of 69.5 nm (Emmons et al., 2010; Lamarque et al., 2012), while in MAM3, the size distribution of aerosols is represented in three different modes. For instance, most of SO₄ in the middle and upper troposphere is in the accumulation mode, with a dry diameter size

range of 58–270 nm (Liu et al., 2012). On average, SO₄ particles are larger in CAM5-chem compared to CAM4-chem. Larger particles in CAM5-chem in the upper troposphere result in smaller SAD despite the slightly larger SO₄ burden compared to CAM4-chem. The increase of BC burden in CAM5-MAM4-chem does not result in an increase of SAD in the model, because only the aged mode of BC is considered in the calculation of SAD. Instead SAD in MAM4 is slightly reduced compared to MAM3 (see Sect. 5).

4.2 Evaluation of model results

4.2.1 Aerosols and Aerosol Optical Depth (AOD)

For the evaluation of aerosols, we compare simulated SO_2 and SO_4 at the surface with observations over the US from the IMPROVE network (see Sect. 3.4), shown in Fig. 7 for SD-CAM4-chem and SD-CAM5-chem, only. Aircraft observations are considered over the US and high latitudes to evaluate the tropospheric distributions (Fig. 8). All model configurations overestimate SO_2 at the surface, as shown here for the SD configurations (Fig. 7) with larger values in CAM5-chem then in CAM4-chem. Annual SO_4 concentrations for all model configurations are about twice as large as observations in rural areas over the US suggest, in particular in summer. In winter, median SO_4 values in SD-CAM4-chem are biased low compared to observation while SD-CAM5-chem is biased high, whereas CAM4-chem values are biased high and CAM5-chem are biased low (not shown).

Comparisons to aircraft observations over the US (Fig. 8), show very good agreement for SO_2 that are very close to the observed values for two of the campaigns, while simulated values are slightly larger for ARCTAS-CARB. Further, the model configurations reproduce observed sulfate aerosol for some campaigns for altitudes between 4–6 km, with the exception of CAM4-chem, while boundary layer values are more than double to those observed. In high latitudes, all model configurations underestimate SO_2 and SO_4 compared to observations from aircraft campaigns ARCTAS and ARCPAC in spring. Those campaigns in particular sampled highly concentrated fire plumes that are not captured by climatological simulations. In comparison to aircraft observations over Central Canada in July 2008, sim-

ulated SO₄ values in the free troposphere are in the range of variability of observations (Fig. 8, bottom right panels).

The evaluation of simulated BC for CAM4-chem, CAM5-chem, and CAM5-MAM4-chem, is performed by comparing to HIPPO aircraft campaigns over the Pacific Ocean (Sect. 3.3), as shown in Fig. 9. All model configurations overestimate background BC , as (about 1 µg/m³ or less), as the case for other climate models (Schwarz et al., 2010; Wang et al., 2014; Samset et al., 2014). The most realistic representation of background BC is in CAM5-chem, where primary BC is assumed to be immediately transitioned into the aged mode and therefore directly emitted in the aged mode. On the other hand, all configurations largely underestimate BC plumes, especially in NH mid and high latitudes in winter and spring, and in August in the Southern Hemisphere (SH). Shortcomings in the simulation of BC plumes are likely caused by a potential underestimate of BC emissions, as well as shortcomings in transport and wet removal by convection (Ma et al., 2013; Wang et al., 2013), while the overestimation of background values may be in part caused by a too long lifetime of BC in the model configurations (Samset et al., 2014).

More work is also needed to improve the representation of POM and SOA, which are not further discussed in this study but were evaluated in Tsigaridis et al. (2014). Large uncertainties exist in the amount of global SOA distribution from observations, and the representation of these aerosols in models, and more future work is needed for both understanding observational yields in comparison to model results.

An overall comparison of A comparison of overall aerosol can be given by comparing Aerosol Optical Depth (AOD) from satellite and AERONET observations (see Sect. 3.1) with model results, as shown for CAM4-chem and CAM5-chem (Fig. 10). AOD derived using CAM5-MAM4-chem (not shown) is very similar to CAM5-chem. The global AOD average in CAM4-chem is slightly lower compared to the observations dataset, while it is higher in CAM5-chem. An overestimation of AOD compared to the climatology occurs in CAM5-chem in Northern Africa and the Middle East, and around 30° N and 30° S over the ocean in CAM5-chem. The AOD bias in the subtropical ocean (mostly from coarse mode sea salt) can be due to the model deficiency representing the sea salt emission or sedi-

mentation (scavenging) process that requires further investigation. Using reanalysis winds do not reduce this bias (not shown). Furthermore, AOD values are underestimated over polluted regions like India, South-East Asia in both model configurations. CAM5-chem has a tendency towards lower AOD in northern mid and high latitudes, which could be a result of the significant underestimation of high BC plumes in these regions. Larger values than observed in CAM4-chem over Eastern US and Europe may be in part a result of the larger simulated SO₄ burden.

4.2.2 Ozone

The zonal mean seasonal cycle of tropospheric and stratospheric O_3 column is evaluated in comparison to a monthly-mean OMI/MLS climatology (Sect. 3.1), Figure 11 (middle and right columns). The tropospheric ozone column in CAM4-chem and CAM5-chem is overestimated between fall and spring in the NH mid-latitudes, while it is slightly underestimated in the tropics. On the other hand, SD configurations overestimate column ozone in the tropics in summer. All configurations underestimate tropospheric O_3 column in the SH, with a the largest deviations to the observations between September and December. Differences between the FR and SD configurations in NH mid to high latitudes are aligned with a stronger STE and stronger BDC between fall and spring in the FR versions, as discussed in Sect. 4.1.1. The reason for differences of the different model configurations in tropical tropospheric ozone column are further discussed in Sect. 5. The underestimation of tropospheric ozone in the SH, especially in October in the tropics and mid-latitudes may be caused by an underestimation of biomass burning at this time of the year, which is consistent with the underestimation of CO column at the same season in the SH (see below).

Stratospheric ozone column is reasonably well reproduced for the tropics and midlatitudes, showing slightly more ozone in the SD versions compared to the FR versions. In high latitudes, the ozone column is largely overestimated in winter and spring in each hemisphere compared to the climatology, which points to shortcomings in stratospheric transport most pronounced in the FR simulations. On the other hand, the underestimation of column O_3 in the SH in October and December point to the well known cold bias of polar vortex temperatures in the FR model versions (Eyring et al., 2010). SD configurations do not show the low bias in ozone column during the ozone hole season in both hemispheres, but instead slightly overestimate column ozone at that time. The reason for this is that temperatures in the SD configurations temperatures are slightly higher than for the FR versions especially the lower stratosphere in high latitudes.

Ozonesonde observations (Sect. 3.2), aircraft data (Sect. 3.3), and surface observations (Sect. 3.4) are used to evaluate the simulated tropospheric chemical composition in more detail. We use a Taylor-like diagram to illustrate relative differences between model configurations and ozonesonde observations, and correlations of the seasonal cycle for different regions, seasons, and different pressure levels, see Figs. 12 and 15. In addition, seasonal cycle comparisons between model results and observations for specific regions are illustrated in Figs. 13 and 16. A comparison of surface ozone is performed, showing probability distribution functions between model results and observations for Western and Eastern North America and Western Europe in Figure 14.

15

Near surface ozone at 900 hPa is for the most part within the range of variability of ozonesonde observations in both SD and FR configurations (Fig.Figs. 12 and 13, top row). The high bias in summer over Eastern US and Western Europe, as reported in earlier studies (e.g., Lamarque et al., 2012), has been significantly reduced, due to an improved calculation of dry deposition velocities (Val Martin et al., 2014). In comparison to surface observations (Fig. 14), in winter, FR model configurations slightly overestimate maximum ozone values for North America and Western Europe. SD configurations show a low bias for Eastern North America and Western Europe. In summer, all model configurations show a high bias of about 10-15 ppb. However, maximum ozone mixing ratios do agree with observations, whereas low ozone mixing ratios are overestimated. A high bias of about 10 ppb can be attributed to the coarse model resolution, which leads to an overestimate of ozone production, because of diluted emissions of ozone precursors and therefore an increase in the lower ozone mixing ratios of its distribution (e.g. Pfister et al., 2014). Ozonesondes are not compared to the model configurations at the surface. Those agree well to surface ob-

servations, besides they bias high over Eastern US in summer, as discussed in Tilmes et al. (2012).

In the mid-troposphere, model results agree well with ozonesonde observations at 500 hPa (Fig. 12, bottom row). The seasonal cycle is well produced, in particular for the FR configurations in mid- and high latitudes, with correlations around 0.95 compared to the observations—(Fig. 13, bottom row). The somewhat higher bias in winter and spring over Western Europe and high latitudes in CAM5-chem in 500 hPa contributes to the high bias in 900 hPa, as more ozone is transported downward, discussed in Sect. 4.1. The low bias in ozone in the West Pacific / East Indian Ocean is due to the stronger convection in the FR model configurations compared to SD. This bias is also shown in the comparisons at 250 hPa, (Figs. 15 and 16). At 50 hPa, all configurations show a high ozone bias by at least 20 % in the tropics during winter and spring. Mid- and high latitude ozone in the stratosphere is reproduced well for all configurations within the range of variability.

Comparisons to the aircraft climatology in the free troposphere (2–7 km), (Fig. 17, top row) confirm the high bias of ozone in CAM5-chem and the low bias in the SD configuration at NH high latitudes, as well as the low bias in the Tropics in fall. Deviations from the aircraft climatology are much larger (up to 40% in mid and high latitudes and up to 60% in winter in the Tropics) compared to the ozonesonde observations (up to 25%).

In comparison to HIPPO aircraft observations over the Pacific, ozone mixing ratios are biased high in mid and high latitudes in both CAM4-chem and CAM5-chem configurations, mainly in fall and winter (Fig. 18 first and second second and third column). In addition, in spring CAM5-chem simulates larger ozone in the NH mid and high latitudes than the other model configurations. The high ozone bias in both CAM4-chem and CAM5-chem in the remote region of the Pacific further points to a too strong STE in the FR versions. In the tropical troposphere, CAM5-chem reproduces observed mean ozone mixing ratios very well, while there is also the low biased summer and fall. However, SD configurations simulate larger ozone mixing ratios in winter and spring compared to ozonesondes and HIPPO observations.

The better representation of tropical ozone in the SD configurations in summer and fall may therefore be the result of more realistic convection, or due to a larger production of LNO_x in this region. The observations further confirm that STE in winter and spring in midand high latitudes is slightly too strong in CAM5-chem compared to the other configurations.

4.2.3 CO and hydrocarbons

In comparison to MOPITT satellite observations (Fig. 11, left column) all model configurations show a significant low bias in column CO with a maximum in spring and fall in the NH and a smaller bias in October in the SH. The tropical CO column agrees within 5% with the observations. Regional differences in column CO between CAM5-chem and MO-PITT (Fig. 19) occur over polluted regions, especially in April and July for the NH and over South America and southern Africa in October. This points to a significant underestimation of CO biomass burning emissions over those regions. Further, CO is largely overestimated in January over Central Africa, which points to an overestimation of fire emissions.

CO and other hydrocarbons are strongly controlled by emissions, but also directly impacted by the amount of OH in the atmosphere. The comparison of CO between aircraft measurement and CAM5-chem model results, averaged over 2–7 km (Fig. 20), confirms the pronounced underestimation of CO mixing ratios in the NH troposphere for seasons where data are available. Inter-model differences can be explained by differences in the oxidizing capacity of the atmosphere, showing largest values for CAM4-chem, consistent with the longest methane lifetime with that configuration (Table, 1, and further discussed in Sect. 5). Furthermore, in the tropics, in spring, aircraft campaigns show in some regions larger propane (C_3H_8) , and to some degree large acetylene (C_2H_2) and CO values (Fig. 17). Too strong convection in the tropics may lead to enhanced mixing ratios of short-lived species, like C_3H_8 (with an approximately 10 day lifetime) in this region, while longer-lived species are still underestimated by the models for the same campaigns.

4.2.4 NO_x and PAN

10

Differences in the simulation of NO_x and PAN between the configurations will have implications for simulated distributions of tropospheric ozone. As for ozone, in the FR version, especially CAM5-chem, both PAN and NO_x mixing ratios in the NH mid and high latitudes are slightly larger compared to the SD versions (Fig. 17). Model comparisons to aircraft observations, show in general an underestimation of NO_x and PAN up 80%. Some aircraft campaigns observed much higher NO_x and PAN values than simulated, for instance ARC-PAC in 2008 and SOS in 1999. Both of these campaigns targeted regions with a significant contribution of biomass burning pollution and local pollution.

In the tropics, ozone deviations from specific aircraft observations often occur along with biases in ozone precursors, NO_x , PAN, and CO, and C_3H_8 , see Figs. 17 and 20. Variations in biases between observations and model results are expected in comparing to aircraft campaigns that targeted specific conditions. We investigate aircraft profiles from those campaigns, where the models reproduced ozone and CO mixing ratios reasonably well in the troposphere (Fig. 21). In this way, shortcomings in NO_x and PAN can be identified. In general, PAN is overestimated in the free tropical troposphere, which can be an indicator of too much convection in the model compared to observations (e.g., Fischer et al., 2014). In comparison to HIPPO observations of PAN (Fig. 22), all model configurations strongly overestimate PAN in the upper troposphere, and in the NH troposphere especially in winter. Values in the lower troposphere in tropics and the SH are reasonably well reproduced.

Sensitivity studies, CAM5-chem* and SD-CAM5-chem* (Sect. 2), where SAD is increased in CAM5-chem configurations to the amount simulated in CAM4-chem simulations (see Table 1), show that only a small fraction of the differences in PAN mixing ratios between the different configurations can be attributed to differences in SAD (Fig. 21). Larger SAD values in CAM4-chem One would expect that larger SAD values result in a faster transition of NO_x to NO_y and therefore reduced PAN production, as shown in the example in Fig. 21, top left panel, for SD CAM5 chem. However, in the FR versions and for the other cases shown in Fig. 21,—However, adjustments of the SAD between CAM4-chem and

CAM5-chem configurations is less important are less important in most cases, as shown in Fig. 21.

5 Methane lifetime and OH differences in CAM4-chem and CAM5-chem

Tropospheric chemistry is strongly controlled by the oxidizing capacity of the atmosphere. The most abundant oxidants in the troposphere are OH, ozone, and nitrate radical (NO₃). These control the atmospheric lifetimes of trace gases, including methane. Methane lifetime can therefore be considered as an indicator for the performance of the model. Model configurations differ largely in tropospheric methane lifetime and often underestimate recent observational estimates of 10.2 years (Prinn, 2005) and 11.3 years (Prather et al., 2012). The reason for differences cannot be easily ascribed to specific processes in models that contributed to the intercomparison projects such as the Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP) (Voulgarakis et al., 2013; Naik et al., 2013).

In this study, all simulations are based on the same framework and run with the same emissions, the same gas-phase chemistry, and in the case of the SD versions, nudged with the same dynamics. Differences in the oxidizing capacity of the atmosphere can be therefore attributed to model physics, aerosol description, and differences in dynamics between SD and FR versions, caused by differences in vertical resolution and transport processes.

Tropospheric methane lifetime in all model configurations in this study varies between 7.6 to 8.8 years (Table 1), which is significantly lower than observational estimates. Tropospheric methane lifetime and CO burden in the tropics (between 30° S– 30° N) are both correlated to the tropical OH burden (e.g., Wang and Jacob, 1998; Murray et al., 2014), with slightly different correlations for different model configurations, Fig. 23 —(left and middle panel). Since CO and methane are both controlled by OH, all model configurations show a very similar CH₄/CO correlation (see Fig. 23, right panel).

To understand the processes that lead to the spread of tropical OH in different model configurations in this study, we explore relationships between annual averages of tropical OH burden and other variables averaged over 30° S-30° N over the troposphere, including

25

tropospheric SAD, H_2O_2 , LNO_x , HNO_3 , tropospheric and stratospheric column ozone, and ozone production (Figs. 24 and 25).

A consistent difference in OH burden exists between CAM5-chem and CAM4-chem in both FR and SD versions, whereby the CH₄ lifetime of CAM4-chem is about half a year longer than in CAM5-chem , see (Fig. 23). Based on the sensitivity simulations CAM5-chem* and SD-CAM5-chem*, most of the difference in OH burden can be attributed to the differences in SAD between CAM4-chem and CAM5-chem (Fig. 24, left top panel). The increased SAD results in increased heterogeneous reaction and therefore increased H₂O₂ (Fig. 24, right top), and further reductions in NO_x burden in comparison to LNO_x production (Fig. 25, left panel). This is due to the fact that enhanced tropospheric heterogeneous reactions increase both the uptake of dinitrogen pentoxide (N₂O₅) as well as the uptake of HO₂ on aerosols, which is the major aqueous-phase source of H₂O₂. The hydrolysis of N₂O₅ on aerosols results in a reduction of NO_x. Increased H₂O₂ further results in increased production of sulfate, since the reaction of H₂O₂ with SO₂ in cloud drops is the most significant contributor to sulfate formation (Seinfeld and Pandis, 2012). For the gas-phase chemistry, the decrease of NO_x leads to a reduction of ozone and, together with the reduction in HO_x, this leads to reduced OH and therefore to an increase in methane lifetime.

However, SAD differences do not explain all the differences in the OH burden, especially between FR and SD configurations. To further analyze factors that control OH burden, we scale OH to a fixed SAD value for all configurations and use the mean tropical tropospheric SAD derived using CAM4-chem results (SAD cam4-chem CAM4-chem) as a reference. For this, we use the slope of the line that describes the OH/SAD change between CAM5-chem and CAM5-chem* configurations, $S_{\rm SAD}$, see blue and cyan line in Fig. 24, left top panel, to adjust the OH burden for all configurations to the SAD reference for SD and FR configurations:

$$OH (adjusted) = OH + S_{SAD} \cdot (SAD_{\underline{cam4chem}CAM4*\underline{chem}} - SAD_{model}).$$
 (1)

As discussed in Murray et al. (2014), OH is strongly correlated to NO_x and CO emissions, as well as to the stratospheric ozone column. Since all the simulations were performed with the same CO and NO_x emissions, differences in NO_x emissions are due to variations in

 LNO_x . Indeed, Fig. 24, middle top panel, shows a strong dependency of the OH burden to LNO_x . The spread the annual annual variability in LNO_x production is much larger in the SD simulations compared to the FR configurations, which is likely introduced by the use of climatological SSTs in the FR configurations. However, the same LNO_x in FR and SD does not result in the same OH burden, which shows inter-model differences are only in part (about half) a result of differences in LNO_x (Fig. 25, top, middle panel).

On the other hand, variations in OH cannot be explained by differences in stratospheric column ozone between the different model simulations. Stratospheric column ozone in the model increases between FR and SD configurations. One would expect a decrease in OH as a result of reduced photolysis rates with increasing stratospheric ozone.

Tropospheric ozone is an important driver for the OH burden in all the different model configurations. More tropospheric ozone results in higher OH burden. The question remains why tropospheric ozone is larger in the SD than the FR version. Considering ozone production, increased SAD between CAM5-chem and CAM5-chem* reduces ozone production as a result of the reduced NO_x burden. However, the same amount of ozone production in FR and SD versions does not result in the same OH burden (see Fig. 25, bottom, right panel). Therefore, enhanced ozone in the SD versions is not only due to differences in chemical production of ozone, but must be also due to differences in transport processes between SD and FR version. This is further supported by the OH to HNO₃ correlations (Fig. 25, middle panel). Larger HNO₃ burden is simulated in the SD configurations than in the FR versions, which is pointing at less stratospheric contribution in the FR configurations. Another source of HNO₃ in the troposphere is LNO₄. The correlation between HNO₃ and LNO₄ clearly supports the conclusion that larger HNO₃ mixing ratios in the SD configuration compared to the FR simulations are not due to differences in HNO₃ production (Fig. 25, right panel). Furthermore, smaller tropical tropospheric ozone burden in CAM5-chem compared to CAM4-chem is not aligned with the larger ozone production in CAM5-chem due to larger LNO_x. Differences are therefore likely a result of differences in transport and mixing processes in the tropics.

6 Conclusions

10

15

20

25

The evaluation of the different model configurations using various observations of aerosol and chemical species shows a realistic performance of the model versions especially in simulating tropospheric ozone. Agreements and shortcomings of each model version against observations are summarized in the following:

- Surface values of SO₂ and SO₄ over rural areas of the US are largely overestimated in most model configurations, whereas median values of SO₂ are overestimated by at least a factor of four and sulfate aerosol (SO₄) is overestimated by about 100% compared to IMPROVE observations. In this model version the discussed simulations, anthropogenic emissions of SO₂ and SO₄ are emitted at the surface, which can lead to an underestimated transport into the free troposphere. Comparisons to aircraft observations in the troposphere show a reasonable agreement between models and observations in SO₂ and SO₄, besides a high bias in SO₄ in CAM4-chem over the US. Profiles of SO₂ and SO₄ in high latitudes are for the most part underestimated in the model.
- The different representation of BC in CAM4-chem and CAM5-chem results in a larger burden of BC in CAM4-chem, which is due to its consideration of primary and aged BC. A similar description in CAM5-MAM4-chem leads to enhanced BC burden compared to CAM5-chem. BC plumes are in general underestimated in all model configurations while background values over the Pacific Ocean are overestimated, whereby CAM5-chem agrees best with observations.
- AOD points to a significant underestimation of biomass burning emissions in the model, and some overestimation in CAM4-chem over West Europe and Eastern US that may be due to the overestimation of SO₄. An overestimation of AOD over the Pacific points to too large background values in aerosols, potentially also from sea-salt, which is more pronounced in CAM5-chem than in CAM4-chem.

Tropospheric ozone in the tropics and the Northern Hemisphere is very well represented in all model configurations and agrees within the variability of ozonesonde observations of about 25 %. Surface observations are well reproduced in winter. The summer high bias of all models over Western Europe and Northern America can be for the most part contributed to a high bias in low and medium ozone mixing ratios as a result the a coarse resolution of the model configurations. In the free troposphere, FR configurations slightly overestimate ozone in mid and high latitudes and underestimate ozone in the tropical free troposphere in summer and fall, while SD configurations slightly overestimate ozone in the upper tropical troposphere and in part underestimate ozone in high latitudes. Southern Hemisphere tropospheric ozone is underestimated by 10–25 % in all model configurations. The comparison to aircraft observations confirms the differences based on ozonesonde observations, but models show a large bias up to 40 % compared to observations.

5

10

15

20

25

- CO is largely underestimated in the Northern Hemisphere, especially in spring, and in the SH in October, pointing to the underestimation of emissions. Other hydrocarbons that are most frequently observed during aircraft campaigns are also significantly underestimated for all seasons. The lowest values of CO and hydrocarbons occur in SD-CAM5-Chem in the tropics. CO is in reasonable agreement with the observations in the tropics.
- PAN is in general overestimated in the upper troposphere in comparison to aircraft observations for all model configurations, while NO_x is underestimated in comparison to aircraft observations, in particular in high latitudes. The largest bias of simulated PAN in comparison to HIPPO observations occurs in mid and high northern latitudes throughout the troposphere in winter months.

Differences in CAM4-chem and CAM5-chem, and FR and SD configurations are to a large part driven by differences in dynamics, including temperature, transport, and mixing processes. Differences in the H_2O tape recorder and in AOA indicate that the Brewer—Dobson circulation is too strong in the FR model configurations, while both diagnostics are

reasonably reproduced in the SD configurations. This is consistent with the overestimation of ozone in high latitudes in FR, particularly in winter and spring for CAM5-chem. Further, shortcomings in transport and mixing are likely responsible for slightly larger ozone mixing ratios in the tropical troposphere in SD compared to FR versions of the model.

Differences in the oxidizing capacity of the atmosphere, which impacts methane and CO lifetime between different model configurations, are largely controlled by tropospheric surface area density, lightning NO_x , and differences in tropospheric ozone. Smaller SAD values in CAM5-chem are responsible for the smaller methane lifetime compared to CAM4-chem. Smaller values in surface area density in CAM5-chem compared to CAM4-chem are a result of different aerosol descriptions in the two model configurations. An underestimation of SAD in the model is possible, because BC plumes are significantly underestimated over source regions. Since background aerosols are in general overestimated, shortcomings may exist in the calculation of SAD. For example, sea-salt and dust provide surfaces for heterogeneous reactions that have not been taken into account in any of the simulations (Evans and Jacob, 2005).

Besides SAD, tropospheric ozone impacts the oxidizing capacity of the model. For the SD configuration, larger ozone mixing ratios in the tropics compared to FR result in reduced methane lifetime. Therefore, variations in transport and mixing is an important driver for differences in ozone and therefore methane lifetime, which is critical for climate simulations.

20

Methane lifetime is in general underestimated in all model configurations compared to observational estimates, with a difference of about one year between the different configurations. The main reason for the underestimation compared to observations is likely due to shortcomings in CO and other hydrocarbon emissions, as also found in other model studies (Stein et al., 2014; Monks et al., 2014; Emmons et al., 2014). This is supported by the underestimation of CO over source regions, but also by the underestimation of AOD over source regions, pointing to a general underestimation of biomass burning emissions. Also, the underestimation of isoprene emissions can result in a significant underestimation of methane lifetime (Pike and Young, 2006).

In summary, both CAM4-chem and CAM5-chem configurations are well suited tools for atmospheric-chemistry modeling studies, considering the shortcomings discussed in this study. We recommend the use of CAM5-chem in future studies, due to the improved description of aerosol processes and cloud interactions. Ongoing work is contributing to further improving CAM5-chem configurations.

Appendix A: Additional reactions in CAM4-chem

Reaction					Rate
BENZENE	+	ОН	\rightarrow	BENO ₂	; $2.3*10^{-12}*exp(-193./T)$
$BENO_2$	+	HO_2	\rightarrow	BENOOH	; $1.4*10^{-12}*exp(700./T)$
$BENO_2$	+	NO	\rightarrow	0.9*GLYOXAL + 0.9*BIGALD +	; $2.6*10^{-12}*exp(350./T)$
				$0.9*NO_2 + 0.9*HO_2$	
XYLENE	+	ОН	\rightarrow	$XYLO_2$; $2.3 * 10^{-11}$
$XYLO_2$	+	HO_2	\rightarrow	XYLOOH	; $1.4*10^{-12}*exp(700./T)$
$XYLO_2$	+	NO	\rightarrow	0.62*BIGALD + 0.34*GLYOXAL +	; $2.6*10^{-12}*exp(350./T)$
				0.54*CH ₃ COCHO 0.9*NO ₂ + 0.9*HO ₂	

Acknowledgements. We thank the HIPPO team for performing reliable aircraft observations used in this study, in particular Steven Wofsy for leading the campaigns, and Joshua Schwarz and Anne Perring or providing black carbon observations, and Ru-Shan Gao for providing ozone observations. We further thank Kenneth Aikin for providing airborne observations in an unified and user-friendly format. MERRA data used in this study have been provided by the Global Modeling and Assimilation Office (GMAO) at NASA Goddard Space Flight Center through the NASA GES DISC online archive. The CESM project is supported by the National Science Foundation and the Office of Science (BER) of the US Department of Energy. The National Center for Atmospheric Research is funded by the National Science Foundation. S. Ghan and P.-L. Ma were supported by the US Department of Energy, Office of Science, Basic Energy Research as part of the Scientific Discoveries through Advanced Computing program. The Pacific Northwest National Laboratory is operated for DOE by Battelle Memorial Institute under contract DE-AC05-76RL01830.

References

15

20

25

- Bardeen, C. G., Gettelman, A., Jensen, E. J., Heymsfield, A., Conley, A. J., Delanoë, J., Deng, M., and Toon, O. B.: Improved cirrus simulations in a general circulation model using CARMA sectional microphysics, J. Geophys. Res.-Atmos., 118, 11679–11697, doi:10.1002/2013JD020193, 2013.
- Deeter, M. N., Martínez-Alonso, S., Edwards, D. P., Emmons, L. K., Gille, J. C., Worden, H. M., Pittman, J. V., Daube, B. C., and Wofsy, S. C.: Validation of MOPITT Version 5 thermal-infrared, near-infrared, and multispectral carbon monoxide profile retrievals for 2000–2011, J. Geophys. Res.-Atmos., 118, 6710–6725, doi:10.1002/jgrd.50272, 2013.
- Deeter, M. N., Martínez-Alonso, S., Edwards, D. P., Emmons, L. K., Gille, J. C., Worden, H. M., Sweeney, C., Pittman, J. V., Daube, B. C., and Wofsy, S. C.: The MOPITT Version 6 product: algorithm enhancements and validation, Atmos. Meas. Tech., 7, 3623–3632, doi:10.5194/amt-7-3623-2014, 2014.
- Emmons, L. K., Hauglustaine, D. A., Müller, J.-F., Carroll, M. A., Brasseur, G. P., Brunner, D., Staehelin, J., Thouret, V., and Marenco, A.: Data composites of airborne observations of tropospheric ozone and its precursors, J. Geophys. Res., 105, 20497, 2000,
- Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D., Granier, C., Guenther, A., Kinnison, D., Laepple, T., Orlando, J., Tie, X., Tyndall, G., Wiedinmyer, C., Baugh-

- cum, S. L., and Kloster, S.: Description and evaluation of the Model for Ozone and Related chemical Tracers, version 4 (MOZART-4), Geosci. Model Dev., 3, 43–67, doi:10.5194/gmd-3-43-2010, 2010.
- Emmons, L. K., Arnold, S. R., Monks, S. A., Huijnen, V., Tilmes, S., Law, K. S., Thomas, J. L., Raut, J.-C., Bouarar, I., Turquety, S., Long, Y., Duncan, B., Steenrod, S., Strode, S., Flemming, J., Mao, J., Langner, J., Thompson, A. M., Tarasick, D., Apel, E. C., Blake, D. R., Cohen, R. C., Dibb, J., Diskin, G. S., Fried, A., Hall, S. R., Huey, L. G., Weinheimer, A. J., Wisthaler, A., Mikoviny, T., Nowak, J., Peischl, J., Roberts, J. M., Ryerson, T., Warneke, C., and Helmig, D.: The POLARCAT Model Intercomparison Project (POLMIP): overview and evaluation with observations, Atmos. Chem. Phys. Discuss., 14, 29331–29393, doi:10.5194/acpd-14-29331-2014, 2014.

10

15

20

- Evans, M. J. and Jacob, D. J.: Impact of new laboratory studies of N₂O₅ hydrolysis on global model budgets of tropospheric nitrogen oxides, ozone, and OH, Geophys. Res. Lett., 32, L09813, doi:10.1029/2005GL022469, 2005.
- Eyring, V., Shepherd, T. G., and Waugh, D. W. (Eds.): SPARC Report on the Evaluation of Chemistry-Climate Models, {SPARC} Report No. 5, WCRP-132, WMO/TD-No.1526, 2010.
- Eyring, V., Lamarque, J.-F., and Hess, P.: Overview of IGAC/SPARC Chemistry-Climate Model Initiative (CCMI) Community Simulations in Support of Upcoming Ozone and Climate Assessments, Tech. rep., 2013.
- Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F., Singh, H. B., Roiger, A., Ries, L., Talbot, R.W., Dzepina, K., and Pandey Deolal, S.: Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, Atmos. Chem. Phys., 14, 2679–2698, doi:10.5194/acp-14-2679-2014, 2014.
- Garcia, R. R., Randel, W. J., and Kinnison, D. E.: On the determination of age of air trends from atmospheric trace species, J. Atmos. Sci., 68, 139–154, doi:10.1175/2010JAS3527.1, 2011.
- Gent, P. R., Danabasoglu, G., Donner, L. J., Holland, M. M., Hunke, E. C., Jayne, S. R., Lawrence, D. M., Neale, R. B., Rasch, P. J., Vertenstein, M., Worley, P. H., Yang, Z.-L., and Zhang, M.: The Community Climate System Model Version 4, J. Climate, 24, 4973–4991, doi:10.1175/2011JCLI4083.1, 2011.
 - Ghan, S. J. and Easter, R. C.: Impact of cloud-borne aerosol representation on aerosol direct and indirect effects, Atmos. Chem. Phys., 6, 4163–4174, doi:10.5194/acp-6-4163-2006, 2006.
 - Ghan, S. J., Liu, X., Easter, R. C., Zaveri, R., Rasch, P. J., Yoon, J.-H., and Eaton, B.: Toward a minimal representation of aerosols in climate models: comparative decomposi-

- tion of aerosol direct, semidirect, and indirect radiative forcing, J. Climate, 25, 6461–6476, doi:10.1175/JCLI-D-11-00650.1, 2012.
- Granier, C., Bessagnet, B., Bond, T., D'Angiola, A., Denier van der Gon, H., Frost, G. J., Heil, A., Kaiser, J. W., Kinne, S., Klimont, Z., Kloster, S., Lamarque, J.-F., Liousse, C., Masui, T., Meleux, F., Mieville, A., Ohara, T., Raut, J.-C., Riahi, K., Schultz, M. G., Smith, S. J., Thompson, A., van Aardenne, J., van der Werf, G. R., and van Vuuren, D. P.: Evolution of anthropogenic and biomass burning emissions of air pollutants at global and regional scales during the 1980–2010 period, Climatic Change, 109, 163–190, doi:10.1007/s10584-011-0154-1, 2011.

10

15

20

25

- Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K., and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN2.1): an extended and updated framework for modeling biogenic emissions, Geosci. Model Dev., 5, 1471–1492, doi:10.5194/gmd-5-1471-2012, 2012.
- Heald, C. L., Henze, D. K., Horowitz, L. W., Feddema, J., Lamarque, J.-F., Guenther, A., Hess, P. G., Vitt, F., Seinfeld, J. H., Goldstein, A. H., and Fung, I.: Predicted change in global secondary organic aerosol concentrations in response to future climate, emissions, and land use change, J. Geophys. Res.-Atmos., 113, D05211, doi:10.1029/2007JD009092, 2008.
- Isaksen, I. S. a., Granier, C., Myhre, G., Berntsen, T. K., Dalsø ren, S. B., Gauss, M., Laj, P., Lohmann, U., Maione, M., Monks, P., Prevot, a. S. H., Raes, F., Richter, a., Jo, P., Rognerud, B., Schulz, M., Shindell, D., Stevenson, D. S., Storelvmo, T., Wang, W., Weele, M. V., Wild, M., and Wuebbles, D., Atmospheric composition change: Climate Chemistry interactions, *Atmospheric Environment*, 43, 5138–5192, 2009.
- Kay, J. E., Holland, M. M., Bitz, C. M., Blanchard-Wrigglesworth, E., Gettelman, A., Conley, A., and Bailey, D.: The influence of local feedbacks and northward heat transport on the equilibrium arctic climate response to increased greenhouse gas forcing, J. Climate, 25, 5433–5450, doi:10.1175/JCLI-D-11-00622.1, 2012.
- Kinne, S.: Remote sensing data combinations-superior global maps for aerosol optical depth, in: Satellite Aerosol Remote Sensing over Land, Springer, Berlin Heidelberg, 361–381, 2009.
- Kinnison, D. E., Brasseur, G. P., Walters, S., Garcia, R. R., Marsch, D. A., Sassi, F., Boville, B. A., Harvey, V. L., Randall, C. E., Emmons, L., Lamarque, J. F., Hess, P., Orlando, J. J., Tie, X. X., Randel, W., Pan, L. L., Gettelman, A., Granier, C., Diehl, T., Niemaier, U., and Simmons, A. J.: Sensitivity of chemical tracers to meteorological parameters in the MOZART-3 chemical transport model, J. Geophys. Res., 112, D20302, 2007.

- Lamarque, J.-F., Hess, P., Emmons, L., Buja, L., Washington, W., and Granier, C.: Tropospheric ozone evolution between 1890 and 1990, J. Geophys. Res., 110, D08304, doi:10.1029/2004JD005537, 2005.
- Lamarque, J.-F., Bond, T. C., Eyring, V., Granier, C., Heil, A., Klimont, Z., Lee, D., Liousse, C., Mieville, A., Owen, B., Schultz, M. G., Shindell, D., Smith, S. J., Stehfest, E., Van Aardenne, J., Cooper, O. R., Kainuma, M., Mahowald, N., McConnell, J. R., Naik, V., Riahi, K., and van Vuuren, D. P.: Historical (1850–2000) gridded anthropogenic and biomass burning emissions of reactive gases and aerosols: methodology and application, Atmos. Chem. Phys., 10, 7017–7039, doi:10.5194/acp-10-7017-2010, 2010.

15

25

- Lamarque, J.-F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C. L., Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P. J., and Tyndall, G. K.: CAM-chem: description and evaluation of interactive atmospheric chemistry in the Community Earth System Model, Geosci. Model Dev., 5, 369–411, doi:10.5194/gmd-5-369-2012, 2012.
- Liu, X., Easter, R. C., Ghan, S. J., Zaveri, R., Rasch, P., Shi, X., Lamarque, J.-F., Gettelman, A., Morrison, H., Vitt, F., Conley, A., Park, S., Neale, R., Hannay, C., Ekman, A. M. L., Hess, P., Mahowald, N., Collins, W., Iacono, M. J., Bretherton, C. S., Flanner, M. G., and Mitchell, D.: Toward a minimal representation of aerosols in climate models: description and evaluation in the Community Atmosphere Model CAM5, Geosci. Model Dev., 5, 709–739, doi:10.5194/gmd-5-709-2012, 2012.
- Liu, X., Ma, P.-L., Wang, H., Tilmes, S., Balwinder, S, Easter, R. C., Ghan, S. J., Rasch, P.: Development of a 4-Mode Version of Modal Aerosol Module for the Community Atmospheric Model, Geosci. Model Dev. Disc., to be submitted, 2015.
 - Ma, P.-L., Rasch, P. J., Wang, H., Zhang, K., Easter, R. C., Tilmes, S., Fast, J. D., Liu, X., Yoon, J.-H., and Lamarque, J.-F.: The role of circulation features on black carbon transport into the Arctic in the Community Atmosphere Model version 5 (CAM5), J. Geophys. Res.-Atmos., 118, 4657–4669, doi:10.1002/jgrd.50411, 2013.
 - Mahowald, N. M., Yoshioka, M., Collins, W. D., Conley, A. J., Fillmore, D. W., and Coleman, D. B., Climate response and radiative forcing from mineral aerosols during the last glacial maximum, pre-industrial, current and doubled-carbon dioxide climates, *Geophysical Research Letters*, *33*, 2–5, 2006.
 - Malm, W. C.: Spatial and monthly trends in speciated fine particle concentration in the United States, J. Geophys. Res., 109, D03306, doi:10.1029/2003JD003739, 2004.

- Meinshausen, M., Smith, S. J., Calvin, K., Daniel, J. S., Kainuma, M. L. T., Lamarque, J.-F., Matsumoto, K., Montzka, S. A., Raper, S. C. B., Riahi, K., Thomson, A., Velders, G. J. M., and Vuuren, D. P.: The RCP greenhouse gas concentrations and their extensions from 1765 to 2300, Climatic Change, 109, 213–241, http://link.springer.com/10.1007/s10584-011-0156-z, 2011.
- Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N., Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L., Steenrod, S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Schlager, H., and Ancellet, G.: Multi-model study of chemical and physical controls on transport of anthropogenic and biomass burning pollution to the Arctic, Atmos. Chem. Phys. Discuss., 14, 25281–25350, 2014.
- Montzka, S. a., Krol, M., Dlugokencky, E., Hall, B., Jöckel, P., and Lelieveld, J.: Small interannual variability of global atmospheric hydroxyl, Science (New York, NY), 331, 67–69, 2011.
 - Mote, P. W., Rosenlof, K. H., McIntyre, M. E., Carr, E. S., Gille, J. G., Holton, J. R., Kinnersley, J. S., Pumphrey, H. C., Russell III, J. M., and Waters, J. W.: An atmospheric tape recorder: the imprint of tropical tropopause temperatures on stratospheric water vapor, J. Geophys. Res., 101, 3989–4006, 1996.

20

- Murray, L. T., Mickley, L. J., Kaplan, J. O., Sofen, E. D., Pfeiffer, M., and Alexander, B.: Factors controlling variability in the oxidative capacity of the troposphere since the Last Glacial Maximum, Atmos. Chem. Phys., 14, 3589–3622, doi:10.5194/acp-14-3589-2014, 2014.
- Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L. W., Lamarque, J.-F., Lin, M., Prather, M. J., Young, P. J., Bergmann, D., Cameron-Smith, P. J., Cionni, I., Collins, W. J., Dalsøren, S. B., Doherty, R., Eyring, V., Faluvegi, G., Folberth, G. A., Josse, B., Lee, Y. H., MacKenzie, I. A., Nagashima, T., van Noije, T. P. C., Plummer, D. A., Righi, M., Rumbold, S. T., Skeie, R., Shindell, D. T., Stevenson, D. S., Strode, S., Sudo, K., Szopa, S., and Zeng, G.: Preindustrial to present-day changes in tropospheric hydroxyl radical and methane lifetime from the Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP), Atmos. Chem. Phys., 13, 5277–5298, doi:10.5194/acp-13-5277-2013, 2013.
 - Neale, R. B., Richter, J., Park, S., Lauritzen, P. H., Vavrus, S. J., Rasch, P. J., and Zhang, M.: The Mean Climate of the Community Atmosphere Model (CAM4) in forced SST and fully coupled experiments, J. Climate, 26, 5150–5168, doi:10.1175/JCLI-D-12-00236.1, 2013.
- Neu, J. L. and Prather, M. J.: Toward a more physical representation of precipitation scavenging in global chemistry models: cloud overlap and ice physics and their impact on tropospheric ozone, Atmos. Chem. Phys., 12, 3289–3310, doi:10.5194/acp-12-3289-2012, 2012.

- Oleson, K. W.: Technical Description of version 4.0 of the Community Land Model (CLM), NCAR Technical Note, NCAR/TN-47, 257, 2010.
- Pfister, G. G., Walters, S., Lamarque, J. F., Fast, J., Barth, M. C., Wong, J., Done, J., Holland, G., and Bruyère, C. L., Projections of future summertime ozone over the U.S., *Journal of Geophysical Research: Atmospheres*, *119*, 5559–5582, 2014.
- Pike, R. C. and Young, P. J.: How plants can influence tropospheric chemistry: the role of isoprene emissions from the biosphere, Weather, 64, 2006.
- Prather, M. J., Holmes, C. D., and Hsu, J.: Reactive greenhouse gas scenarios: systematic exploration of uncertainties and the role of atmospheric chemistry, Geophys. Res. Lett., 39, L09803, doi:10.1029/2012GL051440, 2012.

15

20

25

- Prinn, R. G.: Evidence for variability of atmospheric hydroxyl radicals over the past quarter century, Geophys. Res. Lett., 32, L07809, doi:10.1029/2004GL022228, 2005.
- Rienecker, M. M., Suarez, M. J., Gelaro, R., Todling, R., Bacmeister, J., Liu, E., Bosilovich, M. G., Schubert, S. D., Takacs, L., Kim, G.-K., Bloom, S., Chen, J., Collins, D., Conaty, A., and et al. da Silva, MERRA NASA's Modern-Era Retrospective Analysis for Research and Applications, *J. Climate*, *24*, 3624–3648, 2011.
- Samset, B. H., Myhre, G., Herber, A., Kondo, Y., Li, S.-M., Moteki, N., Koike, M., Oshima, N., Schwarz, J. P., Balkanski, Y., Bauer, S. E., Bellouin, N., Berntsen, T. K., Bian, H., Chin, M., Diehl, T., Easter, R. C., Ghan, S. J., Iversen, T., Kirkevåg, A., Lamarque, J.-F., Lin, G., Liu, X., Penner, J. E., Schulz, M., Seland, Ø., Skeie, R. B., Stier, P., Takemura, T., Tsigaridis, K., and Zhang, K.: Modelled black carbon radiative forcing and atmospheric lifetime in Aero-Com Phase II constrained by aircraft observations, Atmos. Chem. Phys., 14, 12465–12477, doi:10.5194/acp-14-12465-2014, 2014.
- Sander, S. P., Friedl, R. R., Barker, J. R., Golden, D. M., Kurylo, M. J., Sciences, G. E., Wine, P. H., Abbatt, J. P. D., Burkholder, J. B., Kolb, C. E., Moortgat, G. K., Huie, R. E., and Orkin, V. L.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies Evaluation Number 17 NASA Panel for Data Evaluation, JLP Publ., 10–6, 2011.
- Schwarz, J. P., Spackman, J. R., Gao, R. S., Watts, L. a., Stier, P., Schulz, M., Davis, S. M., Wofsy, S. C., and Fahey, D. W.: Global-scale black carbon profiles observed in the remote atmosphere and compared to models, Geophys. Res. Lett., 37, L18812, doi:10.1029/2010GL044372, 2010.

- Schwarz, J. P., Samset, B. H., Perring, A. E., Spackman, J. R., Gao, R. S., Stier, P., Schulz, M., Moore, F. L., Ray, E. A., and Fahey, D. W.: Global-scale seasonally resolved black carbon vertical profiles over the Pacific, Geophys. Res. Lett., 40, 5542–5547, doi:10.1002/2013GL057775, 2013.
- Seinfeld, J. H. and Pandis, S. N.: Atmospheric Chemistry and Physics: from Air Pollution to Climate Change, John Wiley and Sons, New York, 2012.
- Stein, O., Schultz, M. G., Bouarar, I., Clark, H., Huijnen, V., Gaudel, A., George, M., and Clerbaux, C.: On the wintertime low bias of Northern Hemisphere carbon monoxide found in global model simulations, Atmos. Chem. Phys., 14, 9295–9316, doi:10.5194/acp-14-9295-2014, 2014.
- Tie, X.: Assessment of the global impact of aerosols on tropospheric oxidants, J. Geophys. Res., 110, D03204, doi:10.1029/2004JD005359, 2005.

- Tilmes, S., Lamarque, J.-F., Emmons, L. K., Conley, A., Schultz, M. G., Saunois, M., Thouret, V., Thompson, A. M., Oltmans, S. J., Johnson, B., and Tarasick, D.: Technical Note: Ozonesonde climatology between 1995 and 2011: description, evaluation and applications, Atmos. Chem. Phys., 12, 7475–7497, doi:10.5194/acp-12-7475-2012, 2012.
- Tsigaridis, K., Daskalakis, N., Kanakidou, M., Adams, P. J., Artaxo, P., Bahadur, R., Balkanski, Y., Bauer, S. E., Bellouin, N., Benedetti, A., Bergman, T., Berntsen, T. K., Beukes, J. P., Bian, H., Carslaw, K. S., Chin, M., Curci, G., Diehl, T., Easter, R. C., Ghan, S. J., Gong, S. L., Hodzic, A., Hoyle, C. R., Iversen, T., Jathar, S., Jimenez, J. L., Kaiser, J. W., Kirkevåg, A., Koch, D., Kokkola, H., Lee, Y. H, Lin, G., Liu, X., Luo, G., Ma, X., Mann, G. W., Mihalopoulos, N., Morcrette, J.-J., Müller, J.-F., Myhre, G., Myriokefalitakis, S., Ng, N. L., O'Donnell, D., Penner, J. E., Pozzoli, L., Pringle, K. J., Russell, L. M., Schulz, M., Sciare, J., Seland, Ø., Shindell, D. T., Sillman, S., Skeie, R. B., Spracklen, D., Stavrakou, T., Steenrod, S. D., Takemura, T., Tiitta, P., Tilmes, S., Tost, H., van Noije, T., van Zyl, P. G., von Salzen, K., Yu, F., Wang, Z., Wang, Z., Zaveri, R. A., Zhang, H., Zhang, K., Zhang, Q., and Zhang, X.: The AeroCom evaluation and intercomparison of organic aerosol in global models, Atmos. Chem. Phys., 14, 10845–10895, doi:10.5194/acp-14-10845-2014, 2014.
 - Val Martin, M., Heald, C. L., and Arnold, S. R.: Coupling dry deposition to vegetation phenology in the Community Earth System Model: implications for the simulation of surface O₃, Geophys. Res. Lett., 41, 2988–2996, doi:10.1002/2014GL059651, 2014.
- Voulgarakis, Α.. **Analysis** of present day and future OH and methane lifetime in the **ACCMIP** simulations. Atmos. Chem. Phvs. http://www.atmos-chem-phys.net/13/2563/2013/acp-13-2563-2013.html, 2013.

- Wang, H., Easter, R. C., Rasch, P. J., Wang, M., Liu, X., Ghan, S. J., Qian, Y., Yoon, J.-H., Ma, P.-L., and Vinoj, V.: Sensitivity of remote aerosol distributions to representation of cloud-aerosol interactions in a global climate model, Geosci. Model Dev., 6, 765-782, doi:10.5194/gmd-6-765-2013, 2013.
- Wang, Q., Jacob, D. J., Spackman, J. R., Perring, A. E., Schwarz, J. P., Moteki, N., Marais, E. A., Ge, C., Wang, J., and Barrett, S. R. H.: Global budget and radiative forcing of black carbon aerosol: constraints from pole-to-pole (HIPPO) observations across the Pacific, J. Geophys. Res.-Atmos., 119, 195-206, doi:10.1002/2013JD020824, 2014.
 - Wang, Y. and Jacob, J.: Anthropogenic forcing on tropospheric ozone and OH since preindustrial

15

- times sphere, J. Geophys. Res., 103, 1998.

 MO: Scientific Assessment of Ozone Depletion, available at:

 www.unep.ch/ozone/AssessmentPanels/SAP/ScientificAssessment2010/00-SAP-2010-Assement-rego WMO: 2010.
- Wofsy, S. C. and the HIPPO team: HIAPER Pole-to-Pole Observations (HIPPO): fine-grained, globalscale measurements of climatically important atmospheric gases and aerosols, Philos. T. Ser. A, 369, 2073-86, http://www.ncbi.nlm.nih.gov/pubmed/21502177, 2011.
- Wofsy, S., Daube, B. C., Jimenez, J., et al.: HIPPO Merged 10-second Meteorology, Atmospheric Chemistry, Aerosol Data (R 20121129), Tech. rep., Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA, doi:10.3334/CDIAC/hippo 010, 2012.
- Ziemke, J. R., Chandra, S., Labow, G. J., Bhartia, P. K., Froidevaux, L., and Witte, J. C.: A global climatology of tropospheric and stratospheric ozone derived from Aura OMI and MLS measurements, Atmos. Chem. Phys., 11, 9237-9251, doi:10.5194/acp-11-9237-2011, 2011.

Table 1. Overview of model experiments, setup between different simulations, and global model diagnostics. Lifetimes and burdens are calculated for the troposphere defined for regions where ozone is below 150 ppb.

			CAM5-Chem	CAM5-Chem*	SD CAM5-Chem	SD-CAM5-Chem*	CAM5-Chem MAM4
Sim. Years	20 years	2000-2009	20 years	10 years	2000–2009	2000-2009	20 years
Meteorology	CAM4	MERRA (10%)	CAM5	CAM5	MERRA (10%)	MERRA (10 %)	CAM5
Aerosol	BAM	BAM	MAM3	MAM3, 1.5*SAD	MAM3	MAM3, 1.9*SAD	MAM4
Vert. Res.	26L	56L	30L	30L	56L	56L	30L
CH ₄ Burden (Tg)	4153	4074	4102	4098	4064	4067	4103
CH ₄ Lifet. (yr)	8.82	8.40	8.24	8.4	7.83	8.13	8.18
CO Burden (Tg)	308	301	289	294	283	291	287
CO Lifet. (yr)	0.135	0.129	0.132	0.129	0.120	0.125	0.131
O ₃ Burden (Tg)	310	309	314	310	313	306	315
O ₃ Lifet. (days)	24	24	23	23	24	24	23
O ₃ Net. chem. ^a (Tg yr ⁻¹)	515	471	507	480	480	454	518
O_3 STE (Tg yr ⁻¹)	344	356	386	401	362	362	377
LNO _× (Tg N yr ⁻¹)	4.3	4.2	4.8	4.6	4.7	4.3	4.9
Total Optical Depth	0.126	0.108	0.142	0.142	0.153	0.153	0.143
SAD trop	0.35	0.43	0.23	0.35	0.24	0.44	0.22
POM Burden (Tg C)	0.72	0.76	0.58	0.59	0.66	0.66	0.84
SOA Burden (Tg C)	0.97	1.00	1.63	1.63	1.92	1.92	1.63
BC Burden (Tg C)	0.119	0.121	0.082	0.082	0.093	0.093	0.110
SO ₄ Burden (Tg S)	0.54	0.51	0.48	0.47	0.51	0.50	0.48
SO ₄ Aqu. Prod. (Tg S yr ⁻¹)	42.8	46.8	30.2	30.8	30.2	31.2	30.0
SO ₄ Chem. Prod. (Tg S yr ⁻¹)	11.2	10.3	13.7	12.2	14.4	13.7	13.8
SO ₄ Lifet. (days)	3.6	3.3	3.8	3.7	3.9	3.5	3.8
TOA residual ^b	2.88		1.35	1.33			1.36
FSDSc (W m ⁻²)	183.4	153.6	181.0	181.0	176.0	176.0	180.9
FSDSC ^d (W m ⁻²)	246.5	247.6	244.2	244.2	243.4	243.4	243.9
high clouds (%)	31.9	29.3	38.5	38.6	38.5	40.8	38.3
med. clouds (%)	19.0	21.3	27.3	27.3	27.3	27.3	27.2
low clouds (%)	34.3	59.1	43.8	43.8	49.7	49.7	43.8
total clouds (%)	53.9	69.9	64.4	64.5	68.3	68.3	64.3

a Net chemical tendency of O3.

^b Top of the atmosphere (TOA) residual.

^c Downwelling solar flux at surface.

d Clearsky downwelling solar flux at surface.

Table 2. Measurements form aircraft campaigns used in this study

Campaign	Year	Months	Platform	O ₃	CO	NO	NO_{\times}	NO_y	PAN	HNO ₃	ОН	H_2O_2	C_2H_6	C ₃ H ₈	C_2H_4	C_2H_2	SO_2	SO ₄
TOTE	1995	12	DC-8	×	×	×		×										
VOTE	1996	01	DC-8	\times	\times	×		×										
STRAT	1995/96	01-12	ER-2	×		×		×										
PEM-Trop-A	1996	08-10	P3/DC-8	×	×	×	×		×	×	×	×	×	×	×	×	×	
SONEX	1997	10-11	DC-8	×	×	×		×	×		×	×	×	×	×	×	×	
POLARIS	1997	04-06, 09	ER-2	×	×								×	×		×		
POLINAT-2	1997	09-10	Falkon	×	×	×	×						×	×		×		
PEM-Trop-B	1999	03-04	P3/DC-8	×	×	×	×		×	×	×	×	×	×	×	×	×	
ACCENT	1999	04, 09-10	WB57	×	×													
SOS	1999	06, 07	NOAA WP-3D	×	×	×	×	×		×							×	
SOLVE	99/00	12, 03	DC-8	×	×	×		×										
SOLVE	99/00	12-03	ER-2	×	×													
TOPSE	2000	02-05	C130	×	×	×	×	×	×	×	×	×	×	×	×	×	×	
TRACE-P	2000	02-04	P3/DC8	×	×	×	×		×		×	×	×	×	×	×	×	
TexAQS	2000	08, 09	NOAA WP-3D	×	×	×	×	×	×	×			×	×	×	×	×	
ITCT	2002	04, 05	NOAA WP-3D	×	×	×	×	×	×	×			×	×	×	×	×	
Crystal Face	2002	06-07	WB57	\times	\times					×								
INTEX-A	2004	03-08	DC8	\times	\times	×	(NO_2)		×	×	×	×	×	×	×	×	×	×
NEAQS-ITCT	2004	07, 08	NOAA WP-3D	\times	\times	×	×	×	×	×			×	×	×	×	×	
Ave Fall	2004	10, 11	WB57	\times	\times					×								
Ave Houston	2005	06	WB57	\times	\times					×		×						
Polar Ave	2005	01, 02	WB57	×	×		(NO_2)			×								
Cr-Ave	2006	01, 02	WB57	\times	\times					×								
INTEX-B	2006	03-08	DC8	\times	\times	×	(NO_2)		×	×	×	×	×	×	×	×	×	×
TexAQS	2006	09, 10	NOAA WP-3D	\times	\times	×	×	×	×	×			×	×	×	×	×	×
TC4	2007	07	WB57	×	×								×					
ARCPAC	2008	03, 04	NOAA WP-3D	×	×	×	×	×	×	×							×	×
ARCTAS	2008	04-06	DC-8	×	×	×	(NO_2)	×	×	×	×	×	×	×	×	×	×	×
START08	2008	04-06	G5	×	×	×		×					×	×		×		
CalNex	2010	05, 06	NOAA WP-3D	×	×	×	×	×	×	×							×	×

Table 3. Summary of abbreviations used in this article.

Discussion Paper

Abbreviation	Definition
AERONET	AErosol RObotic NETwork
ACCMIP	Atmospheric Chemistry and Climate Model Intercomparison Project
AMWG	Atmospheric Model Working Group
AOA	age of air
AOD	aerosol optical depth
BAM	bulk aerosol model
BC	black carbon
BDC	Brewer Dobson Circulation
CAM	Community Atmosphere Model
CCMI	Chemistry Climate Model Initiative
CESM	Community Earth System Model
FR	free running
HIAPER	High-Performance Instrumented Airborne Platform for Environmental Research
HIPPO	HIAPER Pole-to-Pole Observations
IMPROVE	Interagency Monitoring of Protected Visual Environments
MACCity	MACC / CityZEN EU projects
MAM	modal aerosol model
MEGAN	Model of Emissions of Gases and Aerosols from Nature
MERRA	Modern-Era Retrospective Analysis For Research And Applications
MLS	Microwave Limb Sounder
MOPITT	Measurements of Pollution in The Troposphere
MOZART	odel Model for Ozone and Related chemical Tracers
NCEP	National Centers for Environmental Prediction
NH	Northern Hemisphere
OC	organic carbon
OMI	Aura Ozone Monitoring Instrument
POM	primary organic matter
SAD	surface area density
SD	specified dynamics
SH	Southern Hemisphere
SOA	secondary organic aerosols
SO ₄	sulfate aerosol
STE	stratosphere to troposphere ex 4A ange
TTL	tropical tropopause layer

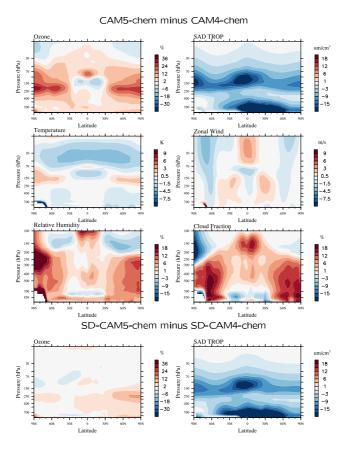


Figure 1. Comparison of ozone, tropospheric surface area density (SAD TROP), temperature, zonal wind, relative humidity (RH), and cloud fraction, between CAM5-chem and CAM4-chem (row 1–3), and between SD-CAM5-chem and SD-CAM4-chem (row 4).

CAM5-chem minus SD-CAM5-chem

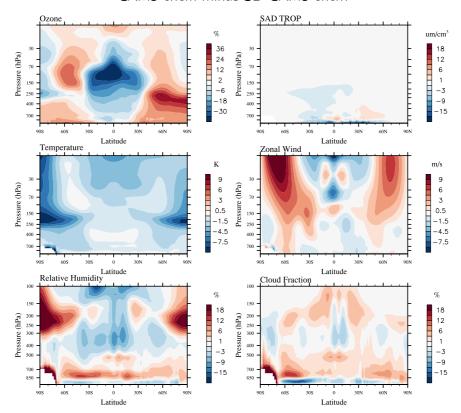


Figure 2. Comparison of ozone, tropospheric surface area density (SAD TROP), temperature, zonal wind, relative humidity (RH), and cloud fraction, between CAM5-chem and SD-CAM5-chem.

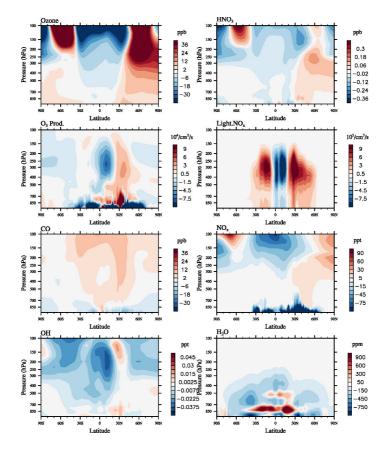


Figure 3. Comparison of ozone, nitric acid, ozone production, lightning NO_x , carbon monoxide, NO_x , hydroxyl radical, and water vapor, between CAM5-chem and SD-CAM5-chem.

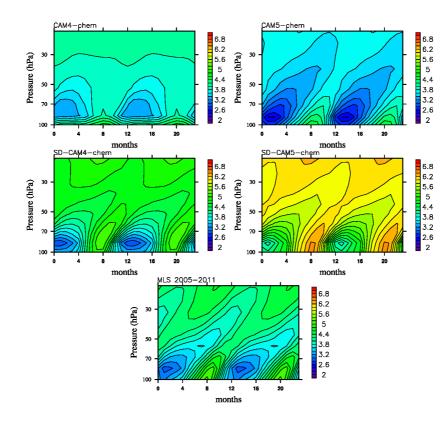


Figure 4. Zonal average water vapor tape recorder (in ppm) of different model configurations, CAM4-chem (top left), CAM5-chem (top right), SD-CAM4-chem (middle left), SD-CAM5-chem (middle right) and MLS satellite observations averaged over year 2005–2011 (bottom panel), composited over 12 months for all simulated years, and repeated over 24 months.

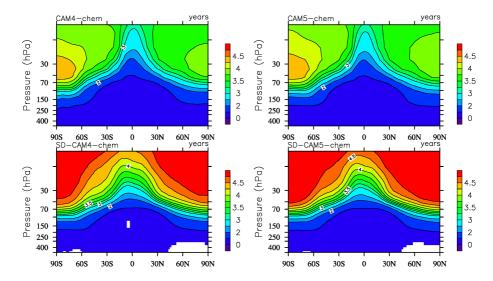


Figure 5. Age of air of different model configurations and simulated years for CAM4-chem (top left), CAM5-chem (top right), SD-CAM4-chem (bottom left), SD-CAM5-chem (bottom right).

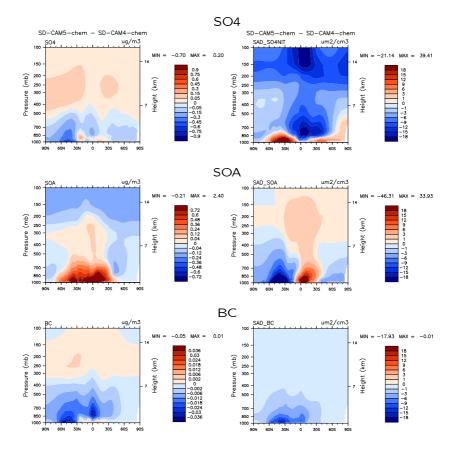


Figure 6. Comparison of aerosol burden (left) and surface area density (right) between SD-CAM5-chem and SD-CAM4-chem of sulfate aerosol (SO₄), SOA, and BC.

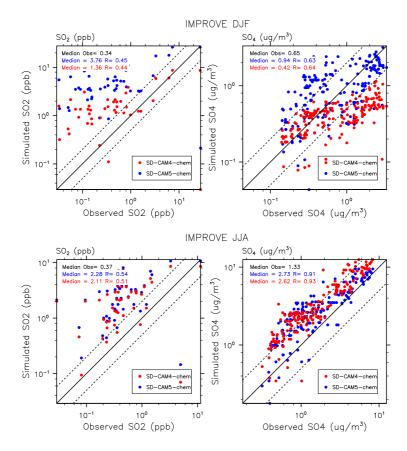


Figure 7. Comparison between IMPROVE network observations over the US in winter (December/January/February) in comparison to SD-CAM5-chem (blue) and SD-CAM4-chem (red) for SO_2 (left) and sulfate aerosol (SO_4) (right) and different seasons, DJF (top) and JJA (right). The median and correlation coefficient (R) between observations and model results are given on the top left of each panel.

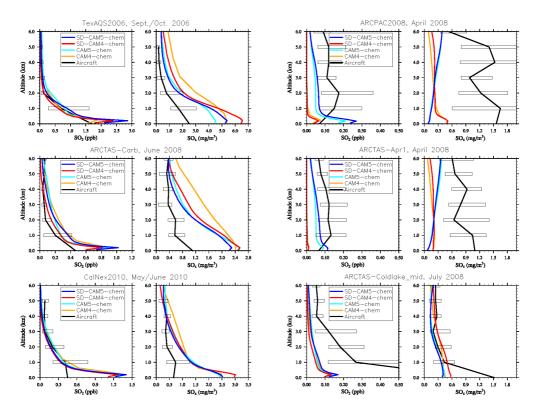


Figure 8. Comparison of SO_2 (left) and sulfate aerosol (SO_4) (right) between different model configurations and aircraft observations over the US (two left columns) and at high latitudes (2 right columns). Black lines show the median of aircraft profiles and error bars indicate describe the range between the 25th and 75th percentile of the distribution. Model results are averaged over the region and months of each campaign.

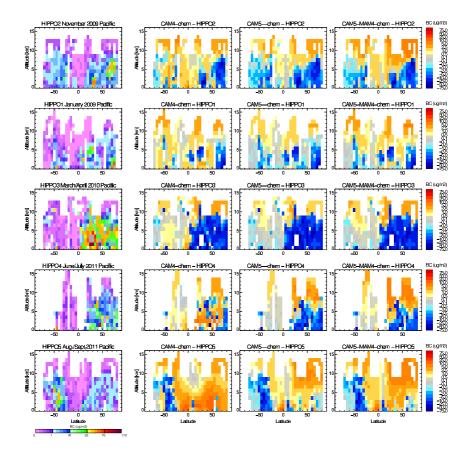


Figure 9. HIPPO BC observations for different HIPPO aircraft campaigns taken over the Pacific (left column) and differences between the different model configurations and observations, CAM4-chem (second column), CAM5-chem (third column) and CAM5-MAM4-chem (fourth column).

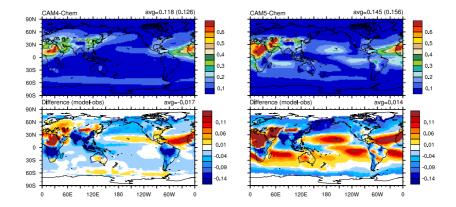


Figure 10. Aerosol Top row: aerosol optical depth at 550 nm for CAM4-chem (left) and CAM5-chem (right)in comparison to the satellite and AERONET composite (middle). Differences Bottom row: differences between model and observations are show in the bottom rowfrom satellite and AERONET composite (Kinne, 2009). Numbers in the parenthesis are the global average AOD over only over areas where the satellite composite has a valid value.

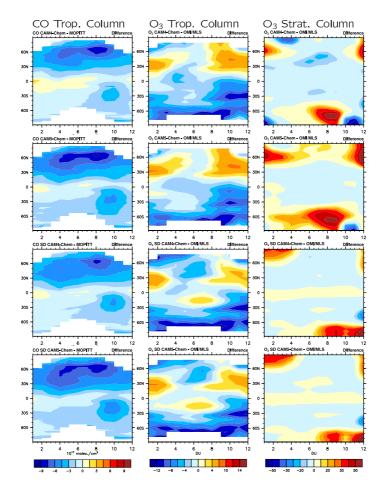


Figure 11. Differences between model results and observations of zonally averaged CO column below 100hPa from the present-day MOPITT climatology (left), and OMI/MLS tropospheric and stratospheric column climatology (right).

Comparison to Ozonesondes

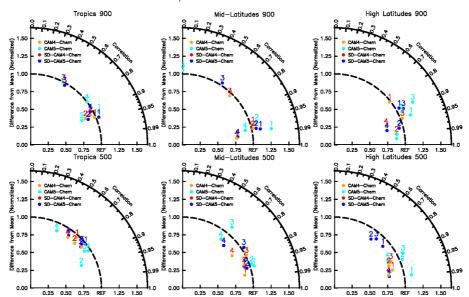


Figure 12. Taylor-like diagram comparing the mean and correlation of the seasonal cycle between observations using a present-day ozonesonde climatology between 1995–2011 and model results, interpolated to the same locations as sampled by the observations and for different pressure levels, 900 hPa (top panel) and 500 hPa (bottom panel). Different numbers are correspondent correspond to a specific region, as defined in Tilmes et al. (2012). Left panels: 1 – NH-Subtropics; 2 – W-Pacific/East Indian Ocean; 3 – equat. Americas; 4 – Atlantic/Africa. Middle panels: 1 – Western Europe; 2 – Eastern US; 3 – Japan; 4 – SH Mid-Latitudes. Right panels: 1 – NH Polar West; 2 – NH Polar East; 3 – Canada; 4 – SH Polar.

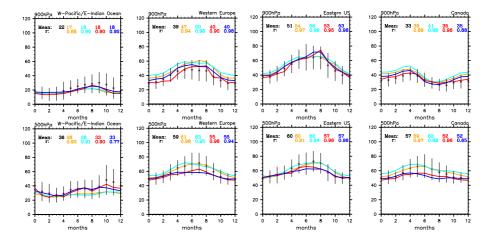


Figure 13. Seasonal cycle comparison between observations using a present-day ozonesonde climatology between 1995–2011 (black) and model results: CAM5-chem (cyan) and CAM4-chem (orange), SD-CAM5-chem (blue) and SD-CAM4-chem (red). Model results are interpolated to the same locations as sampled by the observations and for different pressure levels, 900 hPa (top panel) and 500 hPa (bottom panel) for selected regions. The standard deviation of ozonesonde observations is shown as error bars and the mean and correlation of the seasonal cycle between observations and model results are printed on the top of each figure.

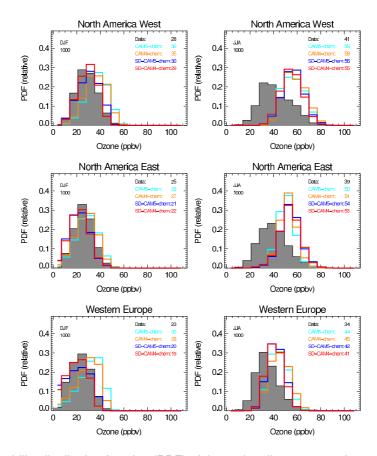


Figure 14. Probability distribution function (PDF) of the regionally-aggregated ozone distribution for Western North America, Eastern North America, and Western Europe from surface ozone observations (grey shaded area) in comparison to regionally-aggregated ozone distributions from model results interpolated to the location of the ozone stations (different colors), for winter (left) and summer (right).

Comparison to Ozonesondes

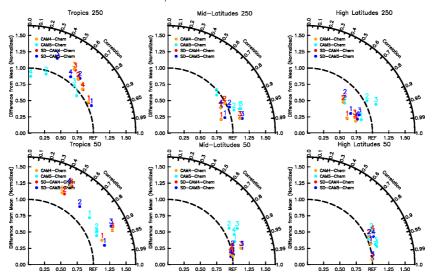


Figure 15. As Fig. 12, but for different pressure levels, 250 hPa (top panel) and 50 hPa (bottom panel). Different numbers are correspondent correspond to a specific region, as defined in Tilmes et al. (2012). Left panels: 1 – NH-Subtropics; 2 – W-Pacific/East Indian Ocean; 3 – equat. Americas; 4 – Atlantic/Africa. Middle panels: 1 – Western Europe; 2 – Eastern US; 3 – Japan; 4 – SH Mid-Latitudes. Right panels: 1 – NH Polar West; 2 – NH Polar East; 3 – Canada; 4 – SH Polar.

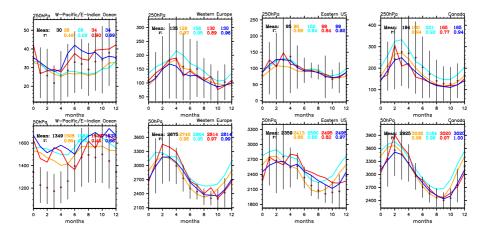


Figure 16. As Fig. 13, but for different pressure levels, 250 hPa (top panel) and 50 hPa (bottom panel).

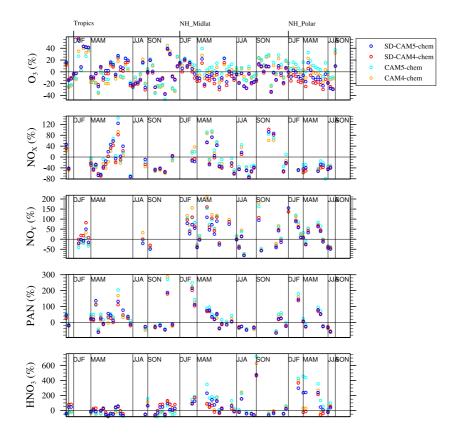


Figure 17. Relative differences between aircraft observations and different model configurations and aircraft observations (different colors) over different regions and seasons as listed in Table 1 and sorted with regard to season and location (see text for more details), averaged over 2–7 km, for O₃, NO_x, NO_y, PAN, and HNO₃.

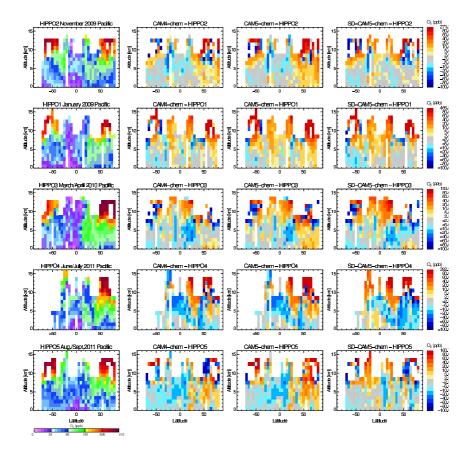


Figure 18. HIPPO O_3 observations for different HIPPO aircraft campaigns taken over the Pacific, left column, and differences between the different model configurations and observations, CAM4-chem (second column), CAM5-chem (third column) and SD-CAM5-chem (fourth column).

CAM5-Chem minus MOPITT CO Climatology (2003-2012)

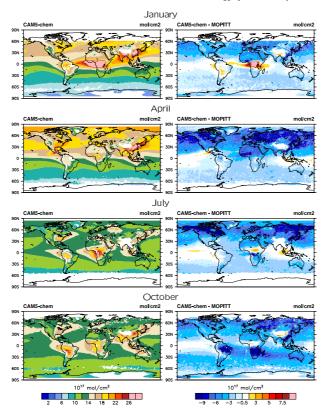


Figure 19. Regional comparison of CO column for different months, between CAM5-chem model results and MOPITT observations. Model results are shown on the left, and differences between CAM5-chem and MOPITT on the right. The MOPITT averaging kernels and a priori are applied to the model results to account for the a priori dependence and vertical resolution of the MOPITT data.

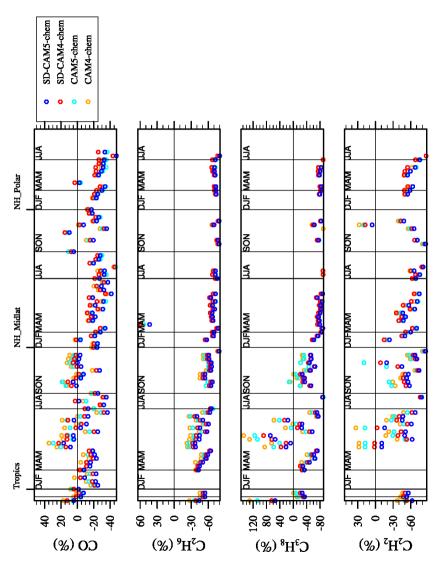


Figure 20. As Fig. 17, but instead for CO, C_2H_6 , C_3H_8 , and C_2H_2 .

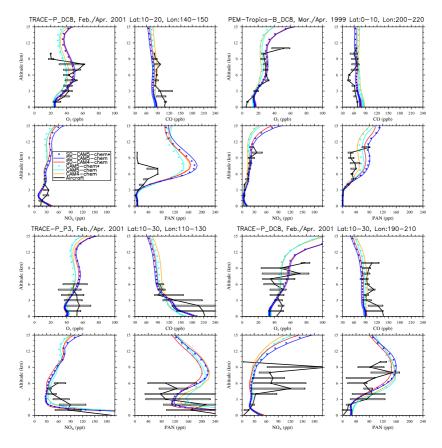


Figure 21. Comparisons of vertical profiles of ozone, CO, NO_x and PAN, from different tropical aircraft campaigns and different model configurations. Black lines show the median of aircraft profiles and error bars indicate describe the range between the 25th and 75th percentile of the distribution. Model results are averaged over the region and months of each campaign.

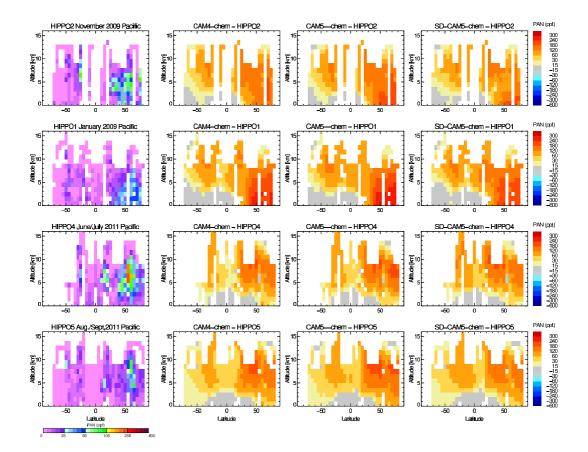


Figure 22. HIPPO PAN observations for different HIPPO aircraft campaigns taken over the Pacific, left column, and differences between the different model configurations and observations, CAM4-chem (second column), CAM5-chem (third column) and SD-CAM5-chem (fourth column).

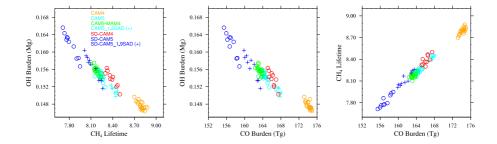


Figure 23. Correlations between tropospheric OH burden, methane lifetime, and CO, for different simulations. OH and CO burden are column integrated tropical averages (30° S–30° N). Each symbol of each configuration (see legend) represents an annual average value.

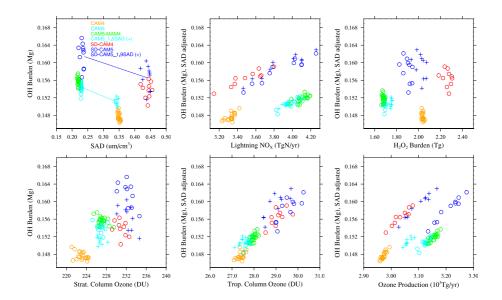


Figure 24. Column integrated tropospheric and tropical OH burden in (30° S–30° N), left top panel, and OH burden, adjusted to a reference SAD value (see text) for the other panels, in correlation to different variables that are integrated over the same region. Each symbol of each configuration (see legend) represents an annual average value.

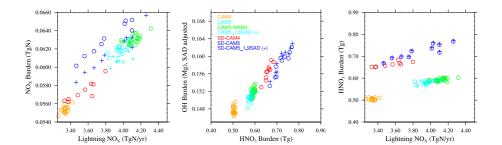


Figure 25. Correlations of tropospheric column integrated NO_x to column integrated lightning NO_x over the tropics (left panel); correlation of OH burden, adjusted to a reference SAD value (see text) to column integrated HNO_3 over the tropics (middle panel); correlations of column integrated HNO_3 to column integrated lightning NO_x over the tropics (right panel).