

1 **Tropospheric Chemistry in the Integrated Forecasting**
2 **System of ECMWF**

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25 **Abstract**

26 A representation of atmospheric chemistry has been included in the Integrated Forecasting
27 System (IFS) of the European Centre for Medium-range Weather Forecasts (ECMWF). The
28 new chemistry modules complement the aerosol modules of the IFS for atmospheric
29 composition, which is named C-IFS. C-IFS for chemistry supersedes a coupled system, in
30 which the Chemical Transport Model (CTM) Model for OZone and Related chemical Tracers
31 3 was two-way coupled to the IFS (IFS-MOZART). This paper contains a description of the
32 new on-line implementation, an evaluation with observations and a comparison of the
33 performance of C-IFS with MOZART and with a re-analysis of atmospheric composition
34 produced by IFS-MOZART within the Monitoring Atmospheric Composition and Climate
35 (MACC) project. The chemical mechanism of C-IFS is an extended version of the Carbon
36 Bond 2005 (CB05) chemical mechanism as implemented in the CTM Transport Model 5
37 (TM5). CB05 describes tropospheric chemistry with 54 species and 126 reactions. Wet
38 deposition and lightning nitrogen monoxide (NO) emissions are modelled in C-IFS using the
39 detailed input of the IFS physics package. A one-year simulation by C-IFS, MOZART and the
40 MACC re-analysis is evaluated against ozonesondes, carbon monoxide (CO) aircraft profiles,
41 European surface observations of ozone (O₃), CO, sulphur dioxide (SO₂) and nitrogen dioxide
42 (NO₂) as well as satellite retrievals of CO, tropospheric NO₂ and formaldehyde.
43 Anthropogenic emissions from the MACC/CityZen (MACCcity) inventory and biomass
44 burning emissions from the Global Fire Assimilation System (GFAS) data set were used in
45 the simulations by both C-IFS and MOZART. C-IFS (CB05) showed an improved
46 performance with respect to MOZART for CO, upper tropospheric O₃, winter time SO₂ and
47 was of a similar accuracy for other evaluated species. C-IFS (CB05) is about ten times more
48 computationally efficient than IFS-MOZART.

49 **1 Introduction**

50 Monitoring and forecasting of global atmospheric composition are key objectives of the
51 atmosphere service of the European Copernicus Programme. The Copernicus Atmosphere
52 Monitoring Service (CAMS) is based on combining satellite observations of atmospheric
53 composition with state-of-the-art atmospheric modelling (Flemming et al., 2013 and
54 Hollingsworth et al., 2008). For that purpose, the integrated forecasting system (IFS) of the
55 European Centre for Medium-Range Weather Forecasts (ECMWF) was extended for forecast
56 and assimilation of atmospheric composition. Modules for aerosols (Morcrette et al., 2009,

57 Benedetti et al., 2009) and greenhouse gases (Engelen et al., 2009) were integrated on-line in
58 the IFS. Because of the complexity of the chemical mechanisms for reactive gases, modules
59 for atmospheric chemistry were not initially included in the IFS. Instead a coupled system
60 (Flemming et al., 2009a) was developed, which couples the IFS to the Chemical Transport
61 Model (CTM) Model for OZone and Related chemical Tracers 3 (MOZART, Kinnison et al.,
62 2007) or Transport Model 5 (TM5, Huijnen et al., 2010) by means of the Ocean Atmosphere
63 Sea Ice Soil coupling software (OASIS4) coupler software (Redler et al., 2010). Van Noije et
64 al. (2014) coupled TM5 to IFS for climate applications in a similar approach. The coupled
65 system made it possible to assimilate satellite retrievals of reactive gases with the assimilation
66 algorithm of the IFS, which is also used for the assimilation of meteorological observations as
67 well as for aerosol and greenhouse gases.

68 The coupled system IFS-MOZART has been successfully used for a re-analysis of
69 atmospheric composition (Inness et al., 2013), pre-operational atmospheric composition
70 forecasts (Stein et al., 2012), forecast and assimilation of the stratospheric ozone (O_3)
71 (Flemming et al., 2011a, Lefever et al., 2014) and tropospheric carbon monoxide (CO)
72 (Eligundi et al., 2010) and O_3 (Ordonez et al., 2010). The coupled system IFS-TM5 has been
73 used in a case study on a period with intense biomass burning in Russia in 2010 (Huijnen et
74 al., 2012). Nevertheless, the coupled approach has limitations such as the need for
75 interpolation between the IFS and CTM model grids and the duplicate simulation of transport
76 processes. Further, its computational performance is often not optimal as it can suffer from
77 load imbalances between the coupled components.

78 Consequently, modules for atmospheric chemistry and related physical processes have now
79 been integrated on-line in the IFS, thereby complementing the on-line integration strategy
80 already pursued for aerosol and greenhouse gases in IFS. The IFS including modules for
81 atmospheric composition is named Composition-IFS (C-IFS). C-IFS makes it possible (i) to
82 use the detailed meteorological simulation of the IFS for the simulation of the fate of
83 constituents (ii) to use the IFS data assimilation system to assimilate observations of
84 atmospheric composition and (iii) to simulate feedback processes between atmospheric
85 composition and weather. A further advantage of C-IFS is the possibility of model runs at a
86 high horizontal and vertical resolution because of the high computational efficiency of C-IFS.
87 C-IFS is the global model system run in pre-operational mode as part of the Monitoring

88 Atmospheric Composition and Climate - Interim Implementation project (MACC II and
89 MACC III) in preparation of CAMS.

90 Including chemistry modules in general circulation models (GCM) to simulate interaction of
91 stratospheric O₃ (e.g. Steil et al., 1998) and aerosols (e.g. Haywood et al., 1997) in the climate
92 system started in the mid-1990s. Later, more comprehensive schemes for tropospheric
93 chemistry were included in climate GCM such as ECHAM5-HAMMOZ (Pozzoli et al., 2008;
94 Rast et al., 2014) and CAM-chem (Lamarque et al., 2012) to study short-lived greenhouse
95 gases and the influence of climate change on air pollution (e.g. Fiore et al., 2010). In the UK
96 Met Office's Unified Model (UM) stratospheric chemistry (Morgenstern et al., 2009) and
97 tropospheric chemistry (O'Connor et al., 2014) can be simulated together with the GLOMAP
98 mode aerosol scheme (Mann et al., 2010). Examples of the on-line integration of chemistry
99 modules in global circulation models with focus on NWP are GEM-AQ (Kaminski et al.,
100 2008), GEMS-BACH (Menard et al., 2007) and GU-WRF/Chem (Zhang et al., 2012).
101 Savage et al. (2013) evaluate the performance of air quality forecast with the UM at the
102 regional scale. Baklanov et al. (2014) give a comprehensive overview of on-line coupled
103 chemistry-meteorological models for regional applications.

104 C-IFS is intended to run with several chemistry schemes for both the troposphere and the
105 stratosphere in the future. Currently, only the tropospheric chemical mechanism CB05
106 originating from the TM5 CTM (Huijnen et al., 2010) has been thoroughly tested. For
107 example, C-IFS (CB05) has been applied to study the HO₂ uptake on clouds and aerosols
108 (Huijnen et al., 2014) and pollution in the Arctic (Emmons et al., 2014). The tropospheric and
109 stratospheric scheme RACMOBUS of the MOCAGE model (Bousserez et al., 2007) and the
110 MOZART 3 chemical scheme as well as an extension of the CB05 scheme with the
111 stratospheric chemical mechanism of the BASCOE model (Errera et al., 2008) have been
112 technically implemented and are being scientifically tested. Only C-IFS (CB05) is the subject
113 of this paper.

114 Each chemistry scheme in C-IFS consists of the specific gas phase chemical mechanism,
115 multi-phase chemistry, the calculation of photolysis rates and upper chemical boundary
116 conditions. Dry and wet deposition, emission injection and parameterization of lightning NO
117 emissions as well as transport and diffusion are simulated by the same approach for all
118 chemistry schemes. Likewise, emissions and dry deposition input data are kept the same for
119 all configurations.

120 The purpose of this paper is to document C-IFS and to present its model performance with
121 respect to observations. Since C-IFS (CB05) replaced the current operational MACC model
122 system for reactive gases (IFS-MOZART) both in data assimilation and forecast mode, the
123 evaluation in this paper is carried out predominately with observations that are used for the
124 routine evaluation of the MACC II system. The model results are compared (i) with a
125 MOZART stand-alone simulation, which is equivalent to a IFS-MOZART simulation and (ii)
126 with the MACC re-analysis (Inness et al., 2013), which is an application of IFS-MOZART in
127 data assimilation mode. All model configurations used the same emission data. The
128 comparison demonstrates that C-IFS is ready to be used operationally.

129 The paper is structured as follows. Section 2 is a description of the C-IFS, with focus on the
130 newly implemented physical parameterizations and the chemical mechanism CB05. Section 3
131 contains the evaluation with observations of a one year simulation with C-IFS (CB05) and a
132 comparison with the results from the MOZART run and the MACC re-analysis. The paper is
133 concluded with a summary and an outlook in section 4.

134 **2 Description of C-IFS**

135 **2.1 Overview of C-IFS**

136 The IFS consists of a spectral NWP model that applies the semi-Lagrangian (SL) semi-
137 implicit method to solve the governing dynamical equations. The simulation of the
138 hydrological cycle includes prognostic representations of cloud fraction, cloud liquid water,
139 cloud ice, rain and snow (Forbes et al., 2011). The simulations presented in this paper used
140 the IFS release CY40r1. The technical and scientific documentation of this IFS release can be
141 found at <http://www.ecmwf.int/research/ifsdocs/CY40r1/index.html>. Changes of the
142 operational model are documented on
143 <https://software.ecmwf.int/wiki/display/IFS/Operational+changes>.

144 At the start of the time step, the three-dimensional advection of the tracers mass mixing ratios
145 is simulated by the SL method as described in Temperton et al. (2001) and Hortal (2002).
146 Next, the tracers are vertically distributed by the diffusion scheme (Beljaars et al., 1998) and
147 by convective mass fluxes (Bechtold et al., 2014). The diffusion scheme also simulates the
148 injection of emissions and the loss by dry deposition (see section 2.4.1). The output of the
149 convection scheme is used to calculate NO production by lightning (see section 2.4.3).
150 Finally, the sink and source terms due to chemical conversion (see section 2.5), wet

151 deposition (see section 2.4.2) and prescribed surface and stratospheric boundary conditions
152 are calculated (see section 2.5.2).

153 The chemical species and the related processes are represented only in grid-point space. The
154 horizontal grid is a reduced Gaussian grid (Hortal and Simmons, 1991). C-IFS can be run at
155 varying vertical and horizontal resolutions. The simulations presented in this paper were
156 carried out at a T255 spectral resolution (i.e. truncation at wavenumber 255), which
157 corresponds to a grid box size of about 80 km. The vertical discretization uses 60 levels up to
158 the model top at 0.1 hPa (65 km) in a hybrid sigma-pressure coordinate. The vertical extent of
159 the lowest level is about 17 m; it is 100 m at about 300m above ground, 400-600 m in the
160 middle troposphere and about 800 m at about 10 km height.

161 The modus operandi of C-IFS is one of a forecast model in a NWP framework. The
162 simulations of C-IFS are a sequence of daily forecasts over a period of several days. Each
163 forecast is initialised by the ECMWF's operational analysis for the meteorological fields and
164 by the 3D chemistry fields from the previous forecast ("forecast mode"). Continuous
165 simulations over longer periods are carried out in "relaxation mode". In relaxation mode the
166 meteorological fields are relaxed to the fields of a meteorological re-analysis, such as ERA-
167 Interim, during the run (Jung et al., 2008) to ensure realistic and consistent meteorological
168 fields.

169 **2.2 Transport**

170 The transport by advection, convection and turbulent diffusion of the chemical tracers uses
171 the same algorithms as developed for the transport of water vapour in the NWP applications
172 of IFS. The advection is simulated with a three-dimensional semi-Lagrangian advection
173 scheme, which applies a quasi-monotonic cubic interpolation of the departure values. Since the
174 semi-Lagrangian advection does not formally conserve mass a global mass fixer is applied.
175 The effect of different global mass fixers is discussed in Diamantakis and Flemming (2014)
176 and Flemming and Huijnen (2011b). A proportional mass was used for the runs presented in
177 this paper because of the overall best balance between the results and computational cost.

178 The vertical turbulent transport in the boundary layer is represented by a first order K-
179 diffusion closure. The surface emissions are injected as lower boundary flux in the diffusion
180 scheme. The lower boundary flux condition also accounts for the dry deposition flux based on
181 the projected surface mass mixing ratio in an implicit way. The vertical transport by

182 convection is simulated as part of the cumulus convection. It applies a bulk mass flux scheme
183 which was originally described in Tiedtke (1989). The scheme considers deep, shallow and
184 mid-level convection. Clouds are represented by a single pair of entraining/detraining plumes
185 which determine the updraught and downdraught mass fluxes.
186 (<http://old.ecmwf.int/research/ifsdocs/CY40r1/> in Physical Processes, Chapter 6, pp 73-90).
187 Highly soluble species such as nitric acid (HNO₃), hydrogen peroxide (H₂O₂) and aerosol
188 precursors are assumed to be scavenged in the convective rain droplets and are therefore
189 excluded from the convective mass transfer.

190 The operator splitting between the transport and the sink and source terms follows the
191 implementation for water vapour (Beljaars et al., 2004). Advection, diffusion and convection
192 are simulated sequentially. The sink and source processes are simulated in parallel using an
193 intermediate update of the mass mixing ratios with all transport tendencies. At the end of the
194 time step tendencies from transport and sink and source terms are added together for the final
195 update the concentration fields. Resulting negative mass mixing ratios are corrected at this
196 point by setting the updated mass mixing ratio to a “chemical zero” of 1.0e-25 kg/kg. For the
197 majority of the species the contribution of the negative fixer was below 0.1% of the
198 dominating source or sink term. The contribution was of the order of 1% for nitrogen species
199 such as NO, N₂O₅ as well as up to 3% for highly soluble species such HNO₃, HO₂, NO₃_A.
200 Large gradients of NO_x at the terminator in the stratosphere as well as intensive wet
201 deposition were the reasons for the increased occurrence of projected negative concentrations.

202 **2.3 Emissions for 2008**

203 The anthropogenic surface emissions were given by the MACCity inventory (Granier et al.,
204 2011) and aircraft NO emissions of a total of ~0.8 Tg N/yr were applied (Lamarque et al.,
205 2010). Natural emissions from soils and oceans were taken from the Precursors of Ozone and
206 their Effects in the Troposphere (POET) database for 2000 (Granier et al., 2005; Olivier et al.,
207 2003). The biogenic emissions were simulated off-line by the MEGAN2.1 model (Guenther et
208 al., 2006). The anthropogenic and natural emissions were used as monthly means without
209 accounting for the diurnal cycle. Daily Biomass burning emissions were produced by the
210 Global Fire Assimilation System (GFAS) version 1, which is based on satellite retrievals of
211 fire radiative power (Kaiser et al., 2012). The actual emission totals used in the T255
212 simulation for 2008 from anthropogenic, biogenic sources and biomass burning as well as
213 lightning NO are given in Table 1.

214 **2.4 Physical parameterizations of sources and sinks**

215 **2.4.1 Dry deposition**

216 Dry deposition is an important removal mechanism at the surface in the absence of
217 precipitation. It depends on the diffusion close to the earth surface, the properties of the
218 constituent and on the characteristics of the surface, in particular the type and state of the
219 vegetation and the presence of intercepted rain water. Dry deposition plays an important role
220 in the biogeochemical cycles of nitrogen and sulphur, and it is a major loss process of
221 tropospheric O₃. Modelling the dry deposition fluxes in C-IFS is based on a resistance model
222 (Wesely et al., 1989), which differentiates the aerodynamic, the quasi-laminar and the canopy
223 or surface resistance. The inverse of the total resistance is equivalent to a dry deposition
224 velocity V_D .

225 The dry deposition flux F_D at the model surface is calculated based on the dry deposition
226 velocity V_D , the mass mixing ratio X_s and air density ρ_s at the lowest model level s , in the
227 following way:

$$F_D = V_D X_s \rho_s$$

228 The calculation of the loss by dry deposition has to account for the implicit character of the
229 dry deposition flux since it depends on the mass mixing ratio X_s .

230 The dry deposition velocities were calculated as monthly mean values from a one-year
231 simulation using the approach described in Michou et al. (2004). It used meteorological and
232 surface input data such as wind speed, temperature, surface roughness and soil wetness from
233 the ERA-interim data set. At the surface the scheme makes a distinction between uptake
234 resistances for vegetation, bare soil, water, snow and ice. The surface and vegetation
235 resistances for the different species are calculated using the stomatal resistance of water
236 vapour. The stomatal resistance for water vapour is calculated depending on the leaf area
237 index, radiation and the soil wetness at the uppermost surface layer. Together with the
238 cuticular and mesophyllic resistances this is combined into the leaf resistance according to
239 Wesely et al. (1989) using season and surface type specific parameters as referenced in
240 Seinfeld and Pandis (1998).

241 Dry deposition velocities have higher values during the day because of lower aerodynamic
242 resistance and canopy resistance. Zhang et al. (2003) reported that averaged observed O₃ and
243 sulphur dioxide (SO₂) dry deposition velocities can be up to 4 times higher at day time than at

244 night time. As this important variation is not captured with the monthly-mean dry deposition
245 values, a +/- 50% variation is imposed on all dry deposition values based on the cosine of the
246 solar zenith angle. This modulation tends to decrease dry deposition for species with a night
247 time maximum at the lowest model level and it increases dry deposition of O₃.

248 Table A4 (supplement) contains annual total loss by dry deposition and expressed as a life-
249 time estimate by dividing by tropospheric burden for a simulation using monthly dry
250 deposition values for 2008. Dry deposition was most effective for many species in particular
251 SO₂ and ammonia (NH₃) as the respective lifetimes were one day to one week. For
252 tropospheric O₃ the respective globally averaged time scale is about 3 months. Because dry
253 deposition occurs mainly over ice-free land surfaces the corresponding time scale is at least
254 three times shorter in these areas.

255 2.4.2 Wet Deposition

256 Wet deposition is the transport and removal of soluble or scavenged constituents by
257 precipitation. It includes the following processes:

- 258 • In-cloud scavenging and removal by rain and snow (rain out)
- 259 • Release by evaporation of rain and snow
- 260 • Below cloud scavenging by precipitation falling through without formation of
261 precipitation (wash out)

262 It is important to take the sub-grid scale of cloud and precipitation-formation into account for
263 the simulation of wet deposition. The IFS cloud scheme provides information on the cloud
264 and the precipitation fraction for each grid box. It uses a random overlap assumption (Jakob
265 and Klein, 2000) to derive cloud and precipitation area fraction. The same method has been
266 used by Neu and Prather (2012), who demonstrated the importance of the overlap assumption
267 for the simulation of the wet deposition. The precipitation fluxes for the simulation of wet
268 removal in C-IFS were scaled to be valid over the precipitation fraction of the respective grid-
269 box. The loss of tracer by rain-out and wash-out was limited to the area of the grid box
270 covered by precipitation. Likewise, the cloud water and ice content is scaled to the respective
271 cloud area fraction. If the sub-grid scale distribution was not considered in this way, wet
272 deposition was lower for highly soluble species such as HNO₃ because the species is only
273 removed from the cloudy or rainy grid box fraction. For species with low solubility the wet
274 deposition loss was slightly decreased because of the decrease in effective cloud and rain
275 water.

276 Even if wet deposition removes tracer mass only in the precipitation area, the mass mixing
277 ratio representing the entire grid box is changed accordingly after each model time step. This
278 is equivalent with the assumption that there is instantaneous mixing within the grid-box at the
279 time scale of the model time step. As discussed in Huijnen et al. (2014), this assumption may
280 lead to an overestimation of the simulated tracer loss.

281 The module for wet deposition in C-IFS is based on the Harvard wet deposition scheme
282 (Jacob et al., 2000 and Liu et al., 2001). In contrast to Jacob et al. (2000), tracers scavenged in
283 wet convective updrafts are not removed as part of the convection scheme. Nevertheless, the
284 fraction of highly soluble tracers in cloud condensate is simulated to limit the amount of
285 tracers lifted upwards as only the gas phase fraction is transported by the mass flux. The
286 removal by convective precipitation is simulated in the same way as for large-scale
287 precipitation in the wet deposition routine.

288 The input fields to the wet deposition routine are the following prognostic variables,
289 calculated by the IFS cloud scheme (Forbes et al., 2011): total cloud and ice water content,
290 grid-scale rain- and snow water content and cloud and grid-scale precipitation fraction as well
291 as the derived fluxes for convective and grid-scale precipitation fluxes at the grid cell
292 interfaces. For convective precipitation a precipitation fraction of 0.05 is assumed and the
293 convective rain and snow water content is calculated assuming a droplet fall speed of 5 m/s.

294 Wash-out, evaporation and rain-out are calculated after each other for large-scale and
295 convective precipitation. The amount of trace gas dissolved in cloud droplets is calculated
296 using Henry's-law-equilibrium or assuming that 70% of aerosol precursors such as sulphate
297 (SO_4), NH_3 and nitrate (NO_3) is dissolved in the droplet. The effective Henry coefficient for
298 SO_2 , which accounts for the dissociation of SO_2 , is calculated following Seinfeld and Pandis
299 (1998, p. 350). The other Henry's law coefficients are taken from the compilation by Sander
300 (1999) (www.henrys-law.org, Table A1 in the supplement).

301 The loss by rain out is determined by the precipitation formation rate. The retention
302 coefficient R , which accounts for the retention of dissolved gas in the liquid cloud condensate
303 as it is converted to precipitation, is one for all species in warm clouds ($T > 268 \text{ K}$). For
304 mixed clouds ($T < 268 \text{ K}$) R is 0.02 for all species but 1.0 for HNO_3 and 0.6 for H_2O_2 (von
305 Blohn, 2011). In ice clouds only H_2O_2 (Lawrence and Crutzen, 1998) and HNO_3 are
306 scavenged.

307 Partial evaporation of the precipitation fluxes leads to the release of 50% of the resolved
308 tracer and 100% in the case of total evaporation (Jacobs et al., 2000). Wash-out is either
309 mass-transfer or Henry-equilibrium limited. HNO₃, aerosol precursors and other highly
310 soluble gases are washed out using a first order wash-out rate of 0.1 mm⁻¹ (Levine and
311 Schwartz, 1982) to account for the mass transfer . For less soluble gases the resolved fraction
312 in the rain water is calculated assuming Henry equilibrium in the evaporated precipitation.

313 Table A5 (supplement) contains total loss by wet deposition and expressed as time scale in
314 days based on the tropospheric burden. For aerosol precursors nitrate, sulphate and
315 ammonium, HNO₃ and H₂O₂ wet deposition is the most important loss process with respective
316 timescales of 2–4 days.

317 2.4.3 NO emissions from lightning

318 NO emissions from lightning are a considerable contribution to the global atmospheric NO_x
319 budget. Estimates of the global annual source vary between 2–8 TgN/yr (Schumann and
320 Huntrieser, 2007). 5 TgN/yr (10.7 TgNO/yr) is the most commonly assumed value for global
321 CTMs which is about 6-7 times the value of NO emissions from aircraft (Gauss et al., 2006)
322 or 17% of the total anthropogenic emissions. NO emissions from lightning play an important
323 role in the chemistry of the atmosphere because they are released in the rather clean air of the
324 free troposphere, where they can influence the O₃ budget and hence the OH-HO₂ partitioning
325 (DeCaria et al., 2005) .

326 The parameterization of the lightning NO production in C-IFS consist of estimates of (i) the
327 flash rate density, (ii) the flash energy release and (iii) the vertical emission profile for each
328 model grid column. The estimate of the flash-rate density is based on parameters of the
329 convection scheme. The C-IFS has two options to simulate the flash-rate densities using the
330 following input parameters: (i) convective cloud height (Price and Rind, 1992) or (ii)
331 convective precipitation (Meijer et al., 2001).

332 The parameterizations distinguish between land and ocean points by assuming about 5-10
333 times higher flash rates over land. Additional checks on cloud base height, cloud extent and
334 temperature are implemented to select only clouds that are likely to generate lightning strokes.
335 The coefficients of the two parameterizations were derived from field studies and depend on
336 the model resolution. With the current implementation of C-IFS (T255L60), the global flash
337 rates were 26 and 43 flashes per seconds for the schemes by Price and Rind (1992) and Meijer

338 et al. (2001), respectively. It seemed therefore necessary to scale the coefficients to get a flash
339 rate in the range of the observed values of about 40-50 flashes per second derived from
340 observations of the Optical Transient Detector (OTD) and the Lightning Imaging Sensor
341 (LIS) (Cecil et al., 2012). Figure 1 shows the annual flash rate density simulated by the two
342 parameterisations together with observations from the LIS/OTD data set. The two approaches
343 show the main flash activity in the tropics but there were differences in the distributions over
344 land and sea. The smaller land-sea differences of Meijer et al. (2001) agreed better with the
345 observations. The observed maximum over Central Africa was well reproduced by both
346 parameterizations but the schemes produce an exaggerated maximum over tropical South
347 America. The lightning activity over the United States was underestimated by both
348 parameterisations. The parameterization by Meijer et al. (2001) has been used for the C-IFS
349 runs presented in this paper.

350 Cloud to ground (CG) and cloud to cloud (CC) flashes are assumed to release a different
351 amount of energy, which is proportional to the NO release. Price et al. (1997) suggest that the
352 energy release of CG is 10 times higher. However, more recent studies suggest a similar value
353 for CG and CC energy release based on aircraft observations and model studies (Ott et al.,
354 2010), which is followed in C-IFS. In C-IFS, CG and CC fractions are calculated using the
355 approach by Price and Rind (1993), which is based on a 4th order function of cloud height
356 above freezing level.

357 The vertical distribution of the NO release is of importance for its impact on atmospheric
358 chemistry. Many CTMs use the suggestion of Pickering et al. (1998) of a C-shape profile,
359 which peaks at the surface and in the upper troposphere. Ott et al. (2010) suggest a “backward
360 C-shape” profile which locates most of the emission in the middle of the troposphere. The
361 vertical distribution can be simulated by C-IFS (i) according to Ott et al. (2010) or (ii) as a C-
362 shape profile following Huijnen et al. (2010). The approach by Ott et al. (2010) is used in the
363 simulation presented here. As lightning NO emissions occur mostly in situations with strong
364 convective transport, differences in the injection profile had little impact.

365 As the lightning emissions depend on the convective activity they change at different
366 resolutions or after changes to the convection scheme. The C-IFS lightning emissions, using
367 the parameterization of Meijer et al. (2001) based on convective precipitation, were 4.9
368 TgN/yr at T159 resolution and 5.7 Tg N/yr at T255 resolution.

369 **2.5 CB05 chemistry scheme**

370 *2.5.1 Gas-phase chemistry*

371 The chemical mechanism is a modified version of the Carbon Bond mechanism 5 (CB05,
372 Yarwood et al., 2005), which is originally based on the work of Gery et al. (1989) with added
373 reactions from Zaveri and Peters (1999) and from Houweling et al. (1998) for isoprene. The
374 CB05 scheme adopts a lumping approach for organic species by defining a separate tracer
375 species for specific types of functional groups. The speciation of the explicit species into
376 lumped species follows the recommendations given in Yarwood et al. (2005). The CB05
377 scheme used in C-IFS has been further extended in the following way: An explicit treatment
378 of methanol (CH₃OH), ethane (C₂H₆), propane (C₃H₈), propene (C₃H₆) and acetone
379 (CH₃COCH₃) has been introduced as described in Williams et al., (2013). The isoprene
380 oxidation has been modified motivated by Archibald et al. (2010). Higher C3 peroxy-radicals
381 formed during the oxidation of C₃H₆ and C₃H₈ were included following Emmons et al.
382 (2010).

383 The CB05 scheme is supplemented with chemical reactions for the oxidation of SO₂, di-
384 methyl sulphide (DMS), methyl sulphonic acid (MSA) and NH₃, as outlined in Huijnen et al.
385 (2014). For the oxidation of DMS, the approach of Chin et al. (1996) is adopted. Table A1
386 (supplement) gives a comprehensive list of the trace gases included in the chemical scheme.

387 The reaction rates have been updated according to the recommendations given in either
388 Sander et al. (2011) or Atkinson et al. (2004, 2006). The oxidation of CO by the hydroxyl
389 radical (OH) implicitly accounts for the formation and subsequent decomposition of the
390 intermediate species HOCO as outlined in Sander et al. (2006). For lumped species, e.g.
391 ALD2, the reaction rate is determined by an average of the rates of reaction for the most
392 abundant species, e.g. C2 and C3 aldehydes, in that group. An overview of all gas-phase
393 reactions and reaction rates as applied in this version of C-IFS can be found in Table A2
394 (supplement).

395 For the loss of trace gases by heterogeneous oxidation processes, the model explicitly
396 accounts for the oxidation of SO₂ in cloud through aqueous phase reactions with H₂O₂ and O₃,
397 depending on the acidity of the solution. The pH is computed from the SO₄, MSA, HNO₃,
398 NO₃_A, NH₃ and NH₄ concentrations, as well as from a climatological CO₂ value. The pH,
399 in combination with the Henry coefficient, defines the fraction of sulphate residing in the

400 aqueous phase, compared to the gas phase concentration (Dentener and Crutzen, 1993) The
401 heterogeneous conversion of N_2O_5 into HNO_3 on cloud droplets and aerosol particles is
402 applied with a reaction probability (γ) set to 0.02 (Evans and Jacob, 2005). The surface area
403 density is computed based on a climatological aerosol size distribution function, applied to the
404 SO_4 , MSA and NO_3_A aerosol, as well as to clouds assuming a droplet size of 8 μm .

405 2.5.2 *Photolysis rates*

406 For the calculation of photo-dissociation rates an on-line parameterization for the derivation
407 of actinic fluxes is used (Williams et al., 2012, 2006). It applies a Modified Band Approach
408 (MBA) which is an updated version of the work by Landgraf and Crutzen (1998), tailored and
409 optimized for use in tropospheric CTMs. The approach uses 7 absorption bands across the
410 spectral range 202 – 695 nm. At instances of large solar zenith angles (71-85°) a different set
411 of band intervals is used. In the MBA the radiative transfer calculation using the absorption
412 and scattering components introduced by gases, aerosols and clouds is computed on-line for
413 each of 7 pre-defined band intervals based on the 2-stream solver of Zdunkowski et al.
414 (1980).

415 The optical depth of clouds is calculated based on a parameterization available in IFS (Slingo,
416 1989 and Fu et al., 1996) for the cloud optical thickness at 550 nm. For the simulation of the
417 impact of aerosols on the photolysis rates a climatological field for aerosols is used, as
418 detailed in Williams et al. (2012). There is also an option to use the MACC aerosol fields.

419 In total 20 photolysis rates are included in the scheme, as given in Table A3 (supplement).
420 The explicit nature of the MBA implies a good flexibility in terms of updating molecular
421 absorption properties (cross sections and quantum yields) and the addition of new photolysis
422 rates into the model.

423 2.5.3 *The chemical solver*

424 The chemical solver used in C-IFS (CB05) is an Euler Backward Iterative (EBI) solver
425 (Hertel et al., 1996). This solver has been originally designed for use with the CBM4
426 mechanism of Gery et al. (1989). The chemical time step is 22.5 min, which is half of the
427 dynamical model time step of 45 min at T255 resolution. Eight, four or one iterations are
428 carried out for fast-, medium- and slow-reacting chemical species to obtain a solution. The
429 number of iterations is doubled in the lowest four models levels, where the perturbations due
430 to emissions can be large.

431 2.5.4 *Stratospheric boundary conditions*

432 The modified CB05 chemical mechanism includes no halogenated species and no photolytic
433 destruction below 202 nm and is therefore not suited for the description of stratospheric
434 chemistry. Thus realistic upper boundary conditions for the longer-lived gases such as O₃,
435 methane (CH₄), and HNO₃ are needed to capture the influence of stratospheric intrusions on
436 the composition of the upper troposphere.

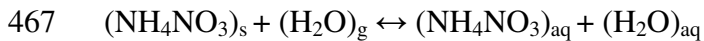
437 Stratospheric O₃ chemistry in C-IFS (CB05) is parameterized by the Cariolle scheme
438 (Cariolle and Teysse re, 2007). Chemical tendencies for stratospheric and tropospheric O₃ are
439 merged at an empirical interface of the diagnosed tropopause height in IFS. Additionally,
440 stratospheric O₃ in C-IFS can be nudged to O₃ analyses of either the MACC re-analysis
441 (Inness et al., 2013) or ERA interim (Dee et al., 2011). The tropopause height in IFS is
442 diagnosed either from the gradient in humidity or the vertical temperature gradient.

443 Stratospheric HNO₃ at 10 hPa is controlled by a climatology of HNO₃ and O₃ observations
444 from the Microwave Limb Sounder (MLS) aboard the Upper Atmosphere Research satellite
445 (UARS). HNO₃ is set to according to the observed HNO₃ - O₃ ratio and the simulated O₃
446 concentrations. Further, stratospheric CH₄ is constrained by a climatology based on
447 observations of the Halogen Occultation Experiment instrument (Gro   and Russel, 2005), at
448 45hPa and at 90 hPa in the extra-tropics, which implicitly accounts for the stratospheric
449 chemical loss of CH₄ by OH, chlorine (Cl) and oxygen (O¹D) radicals. It should be noted that
450 also the surface concentrations of CH₄ are fixed in this configuration of the model.

451 2.5.5 *Gas-aerosol partitioning*

452 Gas-aerosol partitioning is calculated using the Equilibrium Simplified Aerosol Model
453 (EQSAM, Metzger et al., 2002a, 2002b). The scheme has been simplified so that only the
454 partitioning between HNO₃ and the nitrate aerosol (NO₃⁻) and between NH₃ and the
455 ammonium aerosol (NH₄⁺) is calculated. SO₄²⁻ is assumed to remain completely in the
456 aerosol phase because of its very low vapour pressure. The assumptions of the equilibrium
457 model are that (i) aerosols are internally mixed and obey thermodynamic gas/aerosol
458 equilibrium and that (ii) the water activity of an aqueous aerosol particle is equal to the
459 ambient relative humidity (RH). Furthermore, the aerosol water mainly depends on the
460 aerosol mass and the type of the solute, so that parameterizations of single solute molalities
461 and activity coefficients can be defined, depending only on the type of the solute and RH. The
462 advantage of using such parameterizations is that the entire aerosol equilibrium composition

463 can be solved analytically. For atmospheric aerosols in thermodynamic equilibrium with the
464 ambient RH, the following reactions are considered in C-IFS. The subscripts g, s and aq
465 denote gas, solid and aqueous phase, respectively:



469 **2.6 Model budget diagnostics**

470 C-IFS computes global diagnostics for every time step to study the contribution of different
471 processes on the global budget. The basic outputs are the total and tropospheric tracer mass,
472 the global integral of the total surface emissions, integrated wet and dry deposition fluxes,
473 chemical conversion as well as elevated atmospheric emissions and the contributions of
474 prescribed upper and lower vertical boundary conditions for CH₄ and HNO₃. A time-invariant
475 pressure-based tropopause definition, which varies with latitude, is used to calculate the
476 tropospheric mass. To monitor the numerical integrity of the scheme, the contributions of the
477 corrections to ensure positiveness and global mass conservation are calculated. Optionally,
478 more detailed diagnostics can be requested that includes photolytic loss and the loss by OH
479 for the tropics and extra-tropics.

480 A detailed analysis of the global chemistry budget is beyond the scope of this paper. Only a
481 number of key terms for CO, O₃ and CH₄ is summarized here. They are compared with values
482 from the “Atmospheric Composition Change: the European Network of Excellence”
483 (ACCENT) model inter-comparisons of chemistry models by Stevenson et al. (2006) for
484 tropospheric O₃ and by Shindell et al. (2006) for CO. A more recent inter-comparison was
485 carried out within the Atmospheric Chemistry and Climate Model Intercomparison Project
486 (ACCMIP) (Lamarque et al., 2013). The ACCMIP values have been taken from Young et al.
487 (2013) for tropospheric O₃ and from Voulgarakis et al. (2013) for CH₄. It should be noted that
488 the values from these inter-comparisons are valid for present-day conditions but not
489 specifically for 2008. A further source of the differences is the height of the tropopause
490 assumed in the calculations. Overall, the comparison showed that the C-IFS (CB05) is well
491 within the range of the two multi model ensembles.

492 The annual mean of C-IFS tropospheric O₃ burden was 390 Tg. The values are at the upper
493 end of the range simulated by the ACCENT (344 ± 39 Tg) and the ACCMIP (337 ± 23 Tg)

494 models. The same holds for the loss by dry deposition, which was 1155 Tg/yr for C-IFS, 1003
495 ± 200 Tg/yr for ACCENT and in the range 687-1350 Tg/yr for ACCMIP. The tropospheric
496 chemical O_3 production of C-IFS was 4608 Tg/yr and loss 4144 Tg/yr, which is for both
497 values at the lower end of the range reported for the production (5110 ± 606 Tg/yr) and loss
498 (4668 ± 727 Tg/yr) for the ACCENT models. The comparatively simple treatment of volatile
499 organic compounds in CB05 could be an explanation for the low O_3 production and loss
500 terms. Stratospheric inflow in C-IFS, estimated as the residue from the remaining terms was
501 691 Tg and the corresponding value from the ACCENT multi-model mean is 552 ± 168 Tg.

502 The annual mean total CO burden in C-IFS was 361 Tg, which is slightly larger than the
503 ACCENT mean (345 Tg, 248-427 Tg). The total CO emissions in 2008 were 1008 Tg which
504 is in-line with the number used in ACCENT (1077 Tg/yr) but lower than the estimate (1550
505 Tg/yr) of the Third Assessment Report (Prather et al. 2001) of the Intergovernmental Panel
506 on Climate Change (IPCC), which also takes into account results from inverse modelling
507 studies. The tropospheric chemical CO production was 1434 Tg/yr, which is very close to the
508 ACCENT multi-mean of 1505 ± 236 Tg/yr. The chemical CO loss in C-IFS was 2423 Tg
509 and the loss by dry deposition 24 Tg.

510 The annual mean CH_4 total and tropospheric burdens of C-IFS (CB05) are 4874 and 4271
511 Tg/yr, respectively. The global chemical CH_4 loss by OH was 467 Tg/yr. Following
512 Stevenson et al. (2006), this leads to a global CH_4 lifetime estimate of 9.1 yr. This value is
513 within the ACCMIP range of 9.8 ± 1.6 yr but lower than an observation-based 11.2 ± 1.3 yr
514 estimate by Prather et al. (2012). CH_4 emissions were substituted by prescribed monthly
515 zonal-mean surface concentrations to avoid the long-spin up needed by a direct modelling of
516 the CH_4 surface fluxes. The CH_4 surface concentrations were derived from a latitudinal
517 interpolation of observations from the stations South Pole, Cape Grim, Mauna Loa, Mace
518 Head, Barrow and Alert as discussed in Banda et al. (2015). The resulting CH_4 flux was 488
519 Tg/yr, which is of similar size as the sum of current estimates of the total CH_4 emissions of
520 500 - 580 Tg/yr and the loss by soils of 30-40 Tg/yr (Forth Assessment Report by IPCC <
521 http://www.ipcc.ch/publications_and_data/ar4/wg1/en/ch7s7-4-1.html#ar4top).

522

523 **3 Evaluation with observations and comparison with the coupled system** 524 **IFS-MOZART**

525 The main motivation for the development of C-IFS is forecasting and assimilation of
526 atmospheric composition as part of the CAMS. Hence, the purpose of this evaluation is to
527 show how C-IFS (CB05) performs relative to the coupled CTM MOZART-3 (Kinnison et al.,
528 2007), which has been running in the coupled system IFS-MOZART in pre-operational mode
529 since 2007. C-IFS will replace the coupled system in the next update of the CAMS system.
530 The evaluation focuses on species which are relevant to global air pollution such as
531 tropospheric O₃, CO, nitrogen dioxide (NO₂), SO₂ and formaldehyde (HCHO). The MACC re-
532 analysis (Inness et al., 2013), which is an application of IFS-MOZART with assimilation of
533 observations of atmospheric composition, has been included in the evaluation as a benchmark.

534 The MACC re-analysis (REAN) and the corresponding MOZART (MOZ) stand-alone run
535 have already been evaluated with observations by Inness et al. (2013). Further, the MACC-II
536 sub-project on validation has compiled a comprehensive report assessing REAN (MACC,
537 2013). REAN has been further evaluated with surface observations in Europe and North-
538 America for O₃ by Im et al. (2014). C-IFS (CB05) has been already evaluated with a special
539 focus on hydroperoxyl (HO₂) in relation to CO in Huijnen et al. (2014). The performance of
540 an earlier version of C-IFS (CB05) in the Arctic was evaluated and inter-compared with
541 CTMs of the POLARCAT model intercomparison Project (POLMIP) by Monks et al. (2014)
542 for CO and Arnold et al. (2014) for reactive nitrogen. The POLMIP inter-comparisons show
543 that C-IFS (CB05) performs within the range of state-of-the-art CTMs.

544 **3.1 Summary of model runs setup**

545 C-IFS (CB05) was run from 1 January to 31 December 2008 with a spin up starting 1 July
546 2007 at a T255 resolution (80 km x 80 km) with 60 model levels in monthly chunks. The
547 meteorological simulation was relaxed to dynamical fields of the MACC re-analysis (see
548 section 2.1). Likewise stratospheric O₃ above the tropopause was nudged to the MACC re-
549 analysis.

550 MOZ is a run with the MOZART CTM at 1.1°x1.1° (120 x 120 km) horizontal resolution
551 using the 60 vertical levels of C-IFS. The setup of the MOZART model and the applied
552 emissions and dry deposition velocities were the same in MOZ and REAN. The most
553 important difference between MOZ and REAN is the assimilation of satellite retrieval of
554 atmospheric composition in REAN. Further, REAN was produced with the coupled system

555 IFS-MOZART whereas MOZ is a stand-alone driven by the meteorological fields of REAN.
556 The latter is equivalent to a simulation of IFS-MOZART without data assimilation of
557 atmospheric composition. The assimilated retrievals were CO and O₃ total columns,
558 stratospheric O₃ profiles and tropospheric NO₂ columns. No observations of atmospheric
559 composition have been feed in to the MOZ run. No observational information has been used
560 to improve the tropospheric simulation of the C-IFS run. Another difference between MOZ
561 and REAN is that the IFS diffusion and convection scheme, as used in C-IFS, controls the
562 vertical transport in REAN whereas MOZART's generic schemes were used in the MOZ run.

563 MOZ, REAN and C-IFS used the same anthropogenic emissions (MACCity), biogenic
564 emissions (MEGAN 2.1 Guenther et al., 2006,
565 <http://acd.ucar.edu/~guenther/MEGAN/MEGAN.htm>) and natural emissions from the POET
566 project. The biomass burning emissions for MOZ and REAN came from the Global Fire
567 Emission Data version 3 inventory which was redistributed according to Fire Radiative Power
568 observations used in GFAS. Hence, the average biomass burning emissions used by
569 MOZART (MOZ and REAN) agree well with the GFAS emissions used by C-IFS, but they
570 are not identical in temporal and spatial variability.

571 **3.2 Observations**

572 The runs (C-IFS, MOZ, REAN) were evaluated with O₃ observations from ozonesondes and
573 O₃ and CO aircraft profiles from the Measurement of Ozone, Water Vapour, Carbon
574 Monoxide and Nitrogen Oxides by Airbus in-service Aircraft (MOZAIC) program. Simulated
575 surface O₃, CO, NO₂ and SO₂ fields were compared against Global Atmospheric Watch
576 (GAW) surface observations and additionally O₃ against observations from the of the
577 European Monitoring and Evaluation Programme (EMEP) and the European air quality
578 database (AirBase). The global distributions of tropospheric NO₂ and HCHO were evaluated
579 with retrievals of tropospheric columns from Global Ozone Monitoring Experiment 2
580 (GOME-2). Measurements Of Pollution In The Troposphere (MOPITT) retrievals were used
581 for the validation of the global CO total column fields.

582 **3.2.1 In-situ observations**

583 The ozonesondes were obtained from the World Ozone and Ultraviolet Radiation Data Centre
584 (WOUDC) and from the ECWMF Meteorological Archive and Retrieval System. The
585 observation error of the sondes is about $\pm 5\%$ in the range between 200 and 10 hPa and -7 -

586 17% below 200 hPa (Beekmann et al., 1994, Komhyr et al., 1995 and Steinbrecht et al.,
587 1996). The number of soundings varied for the different stations. Typically, the sondes are
588 launched once a week but in certain periods such as during O₃ hole conditions soundings are
589 more frequent. Sonde launches were carried out mostly between 9 and 12 hours local time.
590 The global distribution of the launch sites is even enough to allow meaningful averages over
591 larger areas such North-America, Europe, the Tropics, the Arctic and Antarctica. Table 2
592 contains a list of the ozonesondes used in this study. Tilmes et al. (2012) suggest a further
593 refinement of the North-America region into Canada, Eastern and Western United States as
594 well of the Tropics into Atlantic/Africa, equatorial Americas and Eastern Indian
595 Ocean/Western Pacific based on the inter-comparison of ozone sonde observation for the
596 1994-2010 period. The results will be discussed also for the sub-regions and figures will be
597 presented in the supplement.

598 The MOZAIC program (Marengo et al., 1998 and Nédélec et al., 2003) provides profiles of
599 various trace gases taken during commercial aircraft ascents and descents at specific airports.
600 MOZAIC CO data have an accuracy of ± 5 ppbv, a precision of $\pm 5\%$, and a detection limit of
601 10 ppbv (Nédélec et al., 2003). Since the aircraft carrying the MOZAIC unit were based in
602 Frankfurt, the majority of the CO profiles (837 in 2008) were observed at this airport. A
603 further 10 of the 28 airports with observations in 2008 had a sufficient number of profiles:
604 Windhoek (323), Caracas (129), Hyderabad (125) and London-Gatwick (83) as well as the
605 North-American airports Atlanta (104), Portland (69), Philadelphia (65), Vancouver (56),
606 Toronto (46) and Dallas (43). The North-American airports were considered to be close
607 enough to make a spatial average meaningful. Because of the varying data availability the
608 North-American mean is dominated by the airports in the Eastern United States.

609 Apart from Frankfurt, typically 2 profiles (takeoff and landing) are taken within 2-3 hours or
610 with a longer gap in the case of an overnight stay. At Frankfurt there were 2-6 profiles
611 available each day, mostly in the morning and the later afternoon to the evening. At the other
612 airports the typical observation times were 6 & 18 UTC for Windhoek (+/- 0 h local time), 19
613 and 21 UTC for Hyderabad (+ 4 h local time), 20 and 22 UTC for Caracas (-6 h), 4 and 22 for
614 London (+/- 0 h) and 19 and 22 (- 5/6 h) for the North American airports. This means that
615 most of the observations were taken between the late evening and early morning hours, i.e. at
616 a time of increased stability and large CO vertical gradients close to the surface. Only the
617 observations at Caracas (afternoon) and to some extent in Frankfurt represent a more mixed

618 day-time boundary layer. The modelled column profile was obtained at the middle between
619 start and end time of the profile observation and no consideration was given to the horizontal
620 movement of the aircraft. The model columns were interpolated in time between two
621 subsequent output time steps.

622 The global atmospheric watch (GAW) program of the World Meteorological Organization is
623 a network for mainly surface based observations (WMO, 2007). The data were retrieved from
624 the World Data Centre for Greenhouse Gases [<http://ds.data.jma.go.jp/gmd/wdcgg/>]. The
625 GAW observations represent the global background away from the main polluted areas.
626 Often, the GAW observation sites are located on mountains, which makes it necessary to
627 select a model level different from the lowest model level for a sound comparison with the
628 model. In this study the procedure described in Flemming et al. (2009b) is applied to
629 determine the model level, which is based on the difference between a high resolution
630 orography and the actual station height. The data coverage for CO and O₃ was global,
631 whereas for SO₂ and NO₂ only a few observations in Europe were available at the data
632 repository.

633 The Airbase and EMEP databases host operational air quality observations from different
634 national European networks. All EMEP stations are located in rural areas, while Airbase
635 stations are designed to monitor local pollution. Many AirBase observations may therefore
636 not be representative of a global model with a horizontal resolution of 80 km. However,
637 stations of rural regime may capture the larger scale signal in particular for O₃, which is
638 spatially well correlated (Flemming et al., 2005). The EMEP observations and the rural
639 Airbase O₃ observations were used for the evaluation over Europe.

640 3.2.2 Satellite retrievals

641 Satellite retrievals of atmospheric composition are more widely used to evaluate model
642 results. Satellite data provide good horizontal coverage but have limitation with respect to the
643 vertical resolution and signal from the lowest atmospheric levels. Further, satellite
644 observations are only possible at the specific overpass time, and they can be disturbed by the
645 presence of clouds and surface properties. Depending on the instrument type global coverage
646 is achieved in several days.

647 Day-time CO total column retrievals from MOPITT, version 6 (Deeter et al., 2013b) and
648 retrievals of tropospheric columns of NO₂ (IUP-UB v0.7, Richter et al., 2005) and of HCHO

649 (IUP-UB v1.0; Wittrock et al., 2006) from GOME-2 (Callies et al., 2000) have been used for
650 the evaluation. The retrievals were averaged to monthly means values to reduce the random
651 retrieval error.

652 MOPITT is a multispectral thermal infrared (TIR) / near infrared (NIR) instrument onboard
653 the TERRA satellite with a pixel resolution of 22 km. TERRA's local equatorial crossing time
654 is approximately 10:30 a.m. The MOPITT CO level 2 pixels were binned within $1 \times 1^\circ$ within
655 each month. Deeter et al. (2013a) report a bias of about $+0.08e^{18}$ molec/cm² and a standard
656 deviation (SD) of the error of $0.19e^{18}$ molec/cm² for the TIR/NIR product version 5. This is
657 equivalent to a bias of about 4 % and a SD of 10% respectively assuming typical observations
658 of $2.0 e^{18}$ molec/cm². For the calculation of the simulated CO total column the a-priori profile
659 in combination with the averaging kernels (AK) of the retrievals were applied. They have the
660 largest values between 300 and 800 hPa. The AK have been applied to ensure that the
661 difference between retrieval and AK-weighted model column is independent of the a-priori
662 CO profiles used in the retrieval.. One should note however, that the AK-weighted column is
663 not equivalent to the modelled atmospheric CO burden anymore..

664

665 GOME-2 is a ultra violet - visible (UV-VIS) and NIR sensor designed to provide global
666 observations of atmospheric trace gases. GOME-2 flies in a sun-synchronous orbit with an
667 equator crossing time of 09:30 LT in descending mode and has a footprint of 40 x 80 km.
668 Here, tropospheric vertical columns of NO₂ and HCHO have been computed using a three
669 step approach. First, the Differential Optical Absorption Spectroscopy (DOAS; Platt, 1994)
670 method is applied to measured spectra which yields the total slant column. The DOAS
671 method is applied in a 425–497 nm wavelength window (Richter et al., 2011) for NO₂. and
672 between 337 and 353 nm for HCHO (Vrekoussis et al., 2010). Second, the reference sector
673 approach is applied to total slant columns for stratospheric correction. In a last step,
674 tropospheric slant columns are converted to tropospheric vertical columns by applying an air
675 mass factor. Only data with cloud fractions smaller than 0.2 according to the FRESCO cloud
676 data base (Wang et al., 2008) are used here. Furthermore, retrievals are limited to maximum
677 solar zenith angles of 85° for NO₂ and 60° for HCHO. Uncertainties in NO₂ satellite
678 retrievals are large and depend on the region and season. Winter values in mid and high
679 latitudes are usually associated with larger error margins. As a rough estimate, systematic
680 uncertainties in regions with significant pollution are of the order of 20% – 30%. As the

681 HCHO retrieval is performed in the UV part of the spectrum where less light is available and
682 the HCHO absorption signal is smaller than that of NO₂, the uncertainty of monthly mean
683 HCHO columns is relatively large (20% – 40%) and both noise and systematic offsets have an
684 influence on the results. However, absolute values and seasonality are retrieved more
685 accurately over HCHO hotspots.

686 For comparison to GOME-2 data, model data are vertically integrated without applying AK to
687 tropospheric vertical columns of NO₂ and HCHO, interpolated to satellite observation time
688 and then sampled to match the location of available cloud free satellite data, which has been
689 gridded to match the model resolution. The resulting daily files are then averaged over
690 months for both satellite and model data to reduce the noise.

691 **3.3 Tropospheric Ozone**

692 Figure 2 shows the monthly means of O₃ volume mixing ratios in the pressure ranges surface
693 to 700 hPa (lower troposphere, LT) 700-400 hPa (middle troposphere, MT) and 400-200 hPa
694 (upper troposphere UT) observed by sondes and averaged over Europe, North America and
695 East Asia. Figure 3 shows the same as Figure 2 for the Tropics, Arctic and Antarctica. A more
696 detailed breakdown of North America (Canada, Eastern and Western United States) and the
697 Tropics (Atlantic/Africa, equatorial Americas and Eastern Indian Ocean/Western Pacific)
698 following Times et al. (2012) is presented in the supplement. The observations have a
699 pronounced spring maximum for UT O₃ over Europe, North America and East Asia and a
700 more gradually developing maximum in late spring and summer in MT and LT. The LT
701 seasonal cycle is well re-produced in all runs for the areas of the Northern Hemisphere (NH).
702 In Europe, REAN tends to overestimate by about 5 ppb where the C-IFS and MOZ have
703 almost no bias before the annual maximum in May apart from a small negative bias in spring.
704 Later in the year, C-IFS tends to overestimate in autumn, whereas MOZ overestimates more
705 in late summer. In MT over Europe C-IFS agrees slightly better with the observations than
706 MOZ. MOZ overestimates in winter and spring and this overestimation is more prominent in
707 the UT, where MOZ is biased high throughout the year. This overestimation in UT is highest
708 in spring, where it can be 25% and more. These findings show that data assimilation in REAN
709 improved UT O₃ considerably but had only little influence in LT and MT. The overestimation
710 of MOZ in UT seems to be caused by increased stratospheric O₃ rather than a more efficient
711 transport as lower stratospheric O₃ was overestimated in MOZ. Note that stratospheric ozone
712 in C-IFS was nudged to the MACC re-analysis (see 3.1) but good agreement of C-IFS with

713 observation in UT in all three regions is also present in a run without nudging to stratospheric
714 O₃. It is therefore not only a consequence of the use of assimilated observations in C-IFS
715 (CB05).

716 Over North-America the spring time underestimation by C-IFS and MOZ is more pronounced
717 than over Europe. The underestimation occurs in all regions but was largest in early spring
718 over Canada. C-IFS also underestimates spring ozone throughout North America in MT. LT
719 summer time ozone was overestimated in North America by all models, in particular over the
720 Eastern United States. The bias of C-IFS was the smallest in LT but in contrast to MOZ and
721 REAN C-IFS underestimates summer time ozone in MT over the Eastern United States. The
722 overestimation of UT ozone by MOZ was most pronounced in Canada.

723 In East Asia all runs overestimate by 5-10 ppb in LT and MT especially in autumn and winter.
724 In the northern high latitudes (Figure 3) the negative spring bias appears in all runs in LT and
725 only for C-IFS in MT. As in the other regions, MOZ greatly overestimates UT O₃.

726 Averaged over the tropics, the annual variability is below 10 ppb with maxima in May and in
727 September caused by the dry season in South-America (May) and Africa (September). The
728 variability is well reproduced and biases are mostly below 5 ppb in the whole troposphere.
729 Note that the 400-200 hPa range (UT) in the tropics is less influenced by the stratosphere
730 because of the higher tropopause. C-IFS had smaller biases because of lower values in LT and
731 higher values in MT and UT than MOZ. A more detailed analysis for different tropical
732 regions shows that the seasonality is well captured by all models over Atlantic-Africa,
733 equatorial America and eastern Indian Ocean/Western Pacific in all three tropospheric levels.
734 However, the strong observed monthly anomalies (a observation glitch ? by one station) in
735 equatorial America in March and September were underestimated by up to 20 ppb in all
736 tropospheric levels.

737 Over the Arctic C-IFS and MOZ reproduce the seasonal cycle, which peaks in late spring, but
738 generally underestimate the observations in LT. C-IFS had a smaller bias in LT than MOZ but
739 had a larger negative bias in MT. The biggest improvement of C-IFS w.r.t to MOZ occurred
740 at the surface in Antarctica as the biases compared to the GAW surface observations were
741 greatly reduced. Notably, the assimilation (REAN) led to increased biases for LT and MT O₃,
742 in particular during polar night when UV satellite observations are not available as already
743 discussed in Flemming et al. (2011a).

744 The ability of the models to simulate O₃ near the surface is tested with rural AirBase and
745 EMEP stations (see section 3.2). Figure 4 shows monthly means and Figure 5 the average
746 diurnal cycle in different season in Europe. All runs underestimate monthly mean O₃ in spring
747 and winter and overestimate it in late summer and autumn. The overestimation in summer
748 was largest in MOZ. The recently reported (Val Martin et al. 2014) missing coupling of the
749 leaf area index to the leaf and stomatal vegetation resistance in the calculation of dry
750 deposition velocities could be an explanation of the MOZ bias. While the overestimation
751 appeared also with respect to the ozonesondes in LT (see Figure 2, left) the spring time
752 underestimation was less pronounced in LT.

753 The comparison of the diurnal cycle with observations (Figure 5) shows that C-IFS produced
754 a more realistic diurnal cycle than the MOZART model. The diurnal variability simulated by
755 the MOZART model is much less pronounced than the observations suggest. The diurnal
756 cycle of C-IFS and REAN were similar. This finding can be explained by the fact that C-IFS
757 and REAN use the IFS diffusion scheme whereas MOZART applies the diffusion scheme of
758 the MOZART CTM.

759 The negative bias of C-IFS in winter and spring seems mainly caused by an underestimation
760 of the night time values whereas the overestimation of the summer and autumn average values
761 in C-IFS were caused by an overestimation of the day time values. However, the
762 overestimation of the summer night time values by MOZART seems to be a strong
763 contribution to the average overestimation in this season.

764 **3.4 Carbon Monoxide**

765 The seasonality of CO is mainly driven by its chemical lifetime, which is lower in summer
766 because of increased photochemical activity. The seasonal cycle of the CO emissions plays
767 also an important role in the case of biomass burning and high anthropogenic emissions. The
768 global distribution of total column CO retrieved from MOPITT and from AK weighted
769 columns simulated by C-IFS, MOZ and REAN is shown for April 2008 in Figure 6 and for
770 August in Figure 7. Figures showing the corresponding biases can be found in the
771 supplement. April and August have been selected because they are the months of the NH CO
772 maximum and minimum. C-IFS reproduced well the location of the observed global maxima
773 in North-America, Europe and China as well as the biomass burning signal in Central Africa.
774 However, there was a widespread underestimation of the MOPITT values in the NH, which
775 was strongest over European Russia and Northern China. Tropical CO was slightly

776 overestimated but more strongly over Southeast Asia in April at the end of the biomass
777 burning season in this region. The lower CO columns in mid- and high latitudes in the
778 Southern Hemisphere (SH) were underestimated. The same global gradients of the bias were
779 found in MOZ and REAN. The negative NH bias in April of MOZ is however more
780 pronounced but the positive bias in the tropics is slightly reduced. The bias of MOZ seems
781 stronger over the entire land surface in NH and not predominately in the areas with high
782 emission. This is consistent with the finding of Stein et al. (2014) that dry deposition, besides
783 underestimated emissions, contributes to the large negative biases in NH in MOZ.
784 Assimilating MOPITT (V4) in REAN led to much reduced biases everywhere even though
785 the sign of bias in NH, Tropics and SH remained. In August, the NH bias is reduced but the
786 hemispheric pattern of the CO bias was similar as in April for all runs. The only regional
787 exception from the general overestimation in the tropics is the strong underestimation of CO
788 in the biomass burning maximum in Southern Africa, which points to an underestimation of
789 the GFAS biomass burning emissions in that area.

790 More insight into the seasonal cycle and the vertical CO distribution can be obtained from
791 MOZAIC aircraft profiles. CO profiles at Frankfurt (Figure 8, left) provide a continuous
792 record with about 2 - 6 observations per day. As already reported in Inness et al. (2013) and
793 Stein et al. (2014), MOZ underestimates strongly LT CO with a negative bias of 40 - 60 ppb
794 throughout the whole year. The highest underestimation occurred in April and May, i.e. at the
795 time of the observed CO maximum. C-IFS CB05 also underestimates CO but with a smaller
796 negative bias in the range of 20-40 ppb even though it used the same CO emission data as
797 MOZ. REAN has the lowest bias throughout the year but the improvement is more important
798 in winter and early spring. The comparison over London, which is representative for 4 and 22
799 UTC, leads to similar results as for Frankfurt (Figure 8, middle). The seasonal variability of
800 LT CO from MOZAIC and the model runs in North-America is very similar to the one in
801 Europe (Figure 8, right). The late winter and spring bias is slightly increased whereas the
802 summer time bias was lower for all models. The surface bias in winter and spring of MOZ, C-
803 IFS and REAN is about -50, -40 and -20 ppb respectively. In the rest of the year REAN and
804 C-IFS have a bias of about -15 ppb whereas the bias of MOZ is about twice as large.

805 MT CO was very well produced by REAN in Europe and North-America probably because
806 MOPITT has the highest sensitivity at this level. The MT bias of C-IFS is about 75% of the
807 bias of MOZ, which underestimates by about 30 ppb. In the UT the CO biases are for all

808 models mostly below 10ppb, i.e. about 10 %. C-IFS has overall the smallest CO bias whereas
809 REAN tends to overestimate and MOZ to underestimate CO over Europe and North America.

810 CO observed by MOZAIC over Windhoek (Figure 9, middle) has a pronounced maximum in
811 September because of the seasonality of biomass burning in this region. Although all runs
812 show increased CO in this period, the models without assimilation were less able to reproduce
813 the high observed CO values and are biased low up to 40 ppb in LT and MT. Biases were
814 much reduced, i.e. mostly within 10 ppb, during the rest of the year. The assimilation in
815 REAN greatly reduces the bias in the biomass burning period. In UT C-IFS had slightly
816 smaller biases of about 10 ppb than MOZ and REAN. A less complete record of the seasonal
817 variability is available for Caracas (Figure 9, left). All models tend to underestimate UT and
818 MT CO maxima in April by about 20% but in contrast to Windhoek the C-IFS and not REAN
819 has the smallest bias in LT. Hyderabad (Figure 9, right) is the only observation site where a
820 substantial overestimation of CO in LT and UT is present even though the observations are in
821 the range of 150 - 250 ppb, which is mostly higher than at any of the other airports discussed.
822 All models overestimate the seasonality because of an underestimation in JJA and an
823 overestimation during the rest of the year.

824 The outcome of the comparison with LT CO from MOZAIC is consistent with the model bias
825 with respect to the GAW surface observations in Europe (Figure 10). The winter biases were
826 larger than summer biases and MOZ showed the largest underestimation. . The GAW stations
827 measuring CO are mostly located on mountains in the Alpine region and typical annual biases
828 were about - 5, -20 and -35 ppb for REAN, C-IFS and MOZ respectively. The negative
829 biases of stations in flatter terrain such as Kollumerward tended to be larger.

830 **3.5 Nitrogen dioxide**

831 The global maxima of NO₂ are located in areas of high anthropogenic and biomass burning
832 NO emissions. The global annual distribution of annual tropospheric columns retrieved from
833 the GOME-2 instrument and simulated by the models is shown in Figure 11. C-IFS, MOZ and
834 REAN showed a very similar distribution, which can be explained by that fact that the same
835 NO emission data were used in all runs. The global patterns of the modelled fields resemble
836 the observed annual patterns to a large extent. But the models tend to underestimate the high
837 observed values in East-Asia and Europe and also simulate too little NO₂ in larger areas of
838 medium observed NO₂ levels in Asia and Central Africa as well as in the outflow areas over
839 the West-Atlantic and West Pacific Ocean. This could mean that NO emissions in the most

840 polluted areas are too low but also that the simulated lifetime of NO₂ is too short. Further, an
841 insufficient simulation of NO_x reservoir species such as PAN and the lack of alkyl nitrates in
842 CB05 might be the reason for the underestimation.

843 The validation of the seasonality of NO₂ (Figure 12) for different regions and months shows
844 that tropospheric NO₂ columns over Europe, North America, South Africa and East-Asia are
845 reasonably reproduced. The models tend to underestimate tropospheric columns over Europe
846 in summer (see Table 2 for area descriptions). However, the evaluation with GAW surface
847 stations mainly from Central and Eastern Europe (Figure 13) revealed an overestimation by
848 all models in winter and a small overestimation in summer for REAN and C-IFS. All runs
849 significantly underestimate the annual cycle of the GOME-2 NO₂ tropospheric columns over
850 East-Asia. The winter time values are only half of the observations whereas in summer
851 models agree well with observations. In Southern Africa (20°S/0°S/15°E/15°W), the models
852 overestimate the increased NO₂ values in the biomass burning season by a factor 2 but show
853 good agreement with observations in the rest of the year. The overestimation during biomass
854 burning events could be related to the assumed NO emission factor.

855 **3.6 HCHO**

856 On the global scale HCHO is mainly chemically produced by the oxidation of isoprene and
857 CH₄. Isoprene is emitted by vegetation. On the regional scale HCHO emissions from
858 anthropogenic sources, vegetation and biomass burning also contribute to the HCHO burden.

859 The annual average of tropospheric HCHO retrieved from GOME-2 and from the model runs
860 is shown in Figure 14. The observations show higher values in the tropics and the NH and
861 maxima in the rain forest regions of South America and Central Africa and in South East
862 Asia. The simulated fields of the three runs are very similar. C-IFS, MOZ and REAN
863 reproduce the observed global patterns but show a small but widespread underestimation in
864 the NH extra-Tropics and in industrialized East Asia. On the other hand HCHO is
865 overestimated in Indonesia. Figure 15 shows model time series of tropospheric HCHO against
866 corresponding GOME-2 satellite retrievals for selected regions. The models underestimated
867 satellite values over East-Asia especially in summer and overestimate HCHO columns for
868 Indonesia (5°S/5°N/100°E/120°E) throughout the year. The seasonality in Southern Africa
869 (not shown) and tropical South America (10°S/5°S/73°W/35°W) is well captured in particular
870 by C-IFS. All models also reproduced the observations rather well for the Eastern United

871 States (30°N/40°N/90°W/75°W), but tend to underestimate wintertime HCHO columns for
872 this region.

873 **3.7 Sulfur dioxide**

874 SO₂ was evaluated with available GAW surface observations from Central and Eastern
875 Europe. There were considerable differences in the performance for individual stations often
876 caused by local effects not resolved by the models. To summarize the evaluation for SO₂
877 Figure 16 shows the median of weekly observed and modelled time series. REAN and MOZ
878 greatly exaggerated the seasonal cycle since the values in winter were up to eight times larger
879 than the median of the observations. The summer values of the two runs were about 50%
880 higher than the observations. C-IFS followed better the weak seasonality of the observations
881 but suffered from a nearly constant bias of about 1 ppb (100%), which was much smaller than
882 the bias of REAN and MOZ in winter but slightly higher in summer. Overall, the on-line
883 integration of C-IFS showed lower SO₂ biases.

884 As no SO₂ observations were assimilated in REAN and identical SO₂ emission were used, the
885 differences between the runs were caused by differences in the simulation of vertical mixing,
886 sulphur chemistry and wet and dry deposition in C-IFS and MOZART. The winter time bias
887 of REAN and MOZ could be introduced by the diffusion scheme in MOZART.

888 **3.8 Computational cost**

889 The computational cost is an important factor for the operational applications in CAMS. The
890 computational cost of different configurations of IFS, C-IFS and IFS-MOZART are given in
891 Table 3. Computational cost is expressed in billing units (BU) of the ECMWF IBM Power 7
892 super-computer. BUs are proportional to the number of used Central Processing Unit (CPU)
893 times the simulation time.

894 The increase of cost because of the simulation of the CB05 chemistry with respect to an NWP
895 run is about a factor 4 at the resolutions T159 (110km), T255 (80 km) and T511 (40 km). C-
896 IFS (CB05) is about 8 times more efficient than the coupled system IFS-MOZART at a T159
897 resolution and about 15 times more at a T255 resolution. This strong relative increase in cost
898 of IFS-MOZART is caused by the increasing memory requirements of the IFS at higher
899 resolution, or also in data assimilation mode. However, there is insufficient parallelism in
900 MOZART to exploit the larger number of CPU for speeding up the simulation of the coupled
901 system.

902 C-IFS with the MOZART chemical mechanism, i.e. the same chemistry scheme as in IFS-
903 MOZART, is about 2 times and C-IFS with RACMOBUS 7 times more costly than C-IFS
904 (CB05) at a T159 resolution. Both the MOZART and the RACMOBUS schemes encompass a
905 larger number of species and reactions and include a full stratospheric chemistry scheme,
906 which is missing in CB05. The overhead because of the doubled number of advected species
907 in C-IFS RACMOBUS and MOZART is however small because of the efficiency of the SL
908 advection scheme.

909 **4 Summary and outlook**

910 Modules for the simulation of atmospheric chemistry have been implemented on-line in the
911 Integrated Forecasting System (IFS) of ECMWF. The chemistry scheme complements the
912 already integrated modules for aerosol and greenhouse gases as part of the IFS for
913 atmospheric composition (C-IFS). C-IFS for chemistry replaces the coupled system IFS-
914 MOZART for forecast and assimilation of reactive gases within the pre-operational
915 Copernicus Atmosphere Monitoring Service.

916 C-IFS applies the chemical mechanism CB05, which describes tropospheric chemistry with
917 55 species and 126 reactions. C-IFS benefits from the detailed cloud and precipitation physics
918 of the IFS for the calculation of wet deposition and lightning NO emission. Wet deposition
919 modelling is based on Jacob (2000) and accounts for the sub-grid scale distribution of clouds
920 and precipitation. Dry deposition is modelled using pre-calculated monthly-mean dry
921 deposition velocities following (Wesely, 1989) with a superimposed diurnal cycle. Surface
922 emissions and dry deposition fluxes are applied as surface boundary conditions of the diffusion
923 scheme. Lightning emissions of NO can be calculated either by cloud height (Price and Rind,
924 1993) or by convective precipitation (Meijer et al., 2001). The latter parameterization was
925 used in this study. The anthropogenic emissions were taken from the MACCity inventory and
926 biomass burning emissions from the GFAS data set for 2008.

927 An evaluation for the troposphere of a simulation in 2008 with C-IFS (CB05) and the
928 MOZART CTM (MOZ) as well as with the MACC re-analysis (REAN) was carried out. The
929 model results were compared against ozonesondes, MOZAIC CO aircraft profiles, European
930 surface observations of O₃, CO, SO₂ and NO₂ and global satellite retrievals of CO, NO₂ and
931 HCHO. The evaluation showed that C-IFS performs better or with similar accuracy as
932 MOZART and mostly of similar quality as the MACC re-analysis. It should be noted that

933 satellite retrievals of CO, O₃ and NO₂ were assimilated in the MACC re-analysis to improve
934 the realism of the fields simulated by IFS-MOZART.

935 In comparison to MOZ, C-IFS (CB05) had smaller biases (i) for CO in the Northern
936 Hemisphere, (ii) for O₃ in the upper troposphere and (ii) for winter-time SO₂ at the surface in
937 Europe. Further, the diurnal cycle of surface O₃, tested with rural European Air quality
938 observations, showed greater realism in the C-IFS simulation. As both models used the same
939 emission data, the improvements can be explained by the differences in the chemical
940 mechanism and the simulation of wet and dry deposition. However, the improvements in SO₂
941 and the diurnal cycle of O₃ are most probably caused by the more consistent interplay of
942 diffusion and sink and sources processes in the on-line integrated C-IFS.

943 There is still room for improvement of C-IFS (CB05). It underestimated surface O₃ over
944 Europe and North America in spring and overestimated it in late summer and autumn. CO
945 was still underestimated by C-IFS in particular in Europe and North America throughout the
946 year but more in spring and winter, and in the biomass burning season in Africa. Winter time
947 tropospheric NO₂ over China as retrieved from the GOME-2 instrument was two times higher
948 than the fields modelled by C-IFS, MOZART and the MACC re-analysis.

949 Although only one chemical mechanism is described in the paper, C-IFS is a model that can
950 apply multiple chemistry schemes. The implementation of the chemistry schemes of the
951 CTMs MOCAGE and MOZART has technically been completed but further optimisation and
952 evaluation is required. Both schemes offer a description of stratospheric chemistry, which is
953 not included in the tropospheric scheme CB05. For this reason it is intended to combine the
954 CB05 mechanism with the BASCOE stratospheric mechanism. An inter-comparison of the
955 performance of the different chemical mechanism is planned.

956 It is foreseen to further improve the link between the physics and chemistry packages in IFS.
957 For example, the detailed information from the IFS surface scheme will be utilised for the
958 calculation of dry deposition and biogenic emissions. A first important step is to replace the
959 climatological dry deposition velocities with-online calculated values. Further, the impact of
960 the simulated O₃ fields, once the stratospheric chemistry is fully implemented, on the IFS
961 radiation scheme and the corresponding feedback on the temperature fields will be
962 investigated.

963 Another ongoing development is to link more closely the greenhouse gas, aerosol and gas-
964 phase chemistry modules of C-IFS. Relevant chemical conversion terms can already be fed to

965 the GLOMAP aerosol (Mann et al, 2010) module for the simulation of secondary aerosols.
966 The calculation of photolysis rates can account for the presence of aerosols, and HO₂ uptake
967 on aerosols can be simulated (Huijnen et al., 2014).

968 In summary, C-IFS is a new global chemistry weather model for forecast and assimilation of
969 atmospheric composition. C-IFS (CB05) has already been successfully applied in data
970 assimilation mode (Inness et al., 2015). C-IFS offers improvements over the coupled system
971 IFS-MOZART because (i) it simulates several trace gas C-IFS (CB05)es with better accuracy,
972 (ii) it is computational several times more efficient in particular at high resolution and (iii) it
973 better facilitates the implementation of feedback processes between gas-phase and aerosol
974 processes as well as between atmospheric composition and meteorology.

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986

987 **Code Availability**

988 The C-IFS source code is integrated in ECWMF's IFS code, which is only available subject to
989 a licence agreement with ECMWF. ECMWF member-state weather services and their
990 approved partners will get access granted. The IFS code without modules for assimilation and
991 chemistry can be obtained for educational and academic purposes as part of the openIFS
992 release (<https://software.ecmwf.int/wiki/display/OIFS/OpenIFS+Home>). A detailed
993 documentation of the IFS code is available from
994 <https://software.ecmwf.int/wiki/display/IFS/CY40R1+Official+IFS+Documentation>. The
995 CB05 chemistry module of C-IFS was originally developed in the TM5 chemistry-transport

996 model. Readers interested in the TM5 code can contact the TM5 developers
997 (<http://tm5.sourceforge.net>) or can go directly to the TM5 wiki page,
998 http://tm.knmi.nl/index.php/Main_Page.
999

1000 **References**

- 1001 Archibald, A. T., Cooke, M. C., Utembe, S. R., Shallcross, D. E., Derwent, R. G., and Jenkin,
1002 M. E.: Impacts of mechanistic changes on HO_x formation and recycling in the oxidation of
1003 isoprene, *Atmos. Chem. Phys.*, 10, 8097-8118, doi:10.5194/acp-10-8097-2010, 2010.
- 1004 Arnold, S. R., Emmons, L. K., Monks, S. A., Law, K. S., Ridley, D. A., Turquety, S., Tilmes,
1005 S., Thomas, J. L., Bouarar, I., Flemming, J., Huijnen, V., Mao, J., Duncan, B. N., Steenrod,
1006 S., Yoshida, Y., Langner, J., and Long, Y.: Biomass burning influence on high latitude
1007 tropospheric ozone and reactive nitrogen in summer 2008: a multi-model analysis based on
1008 POLMIP simulations, *Atmos. Chem. Phys. Discuss.*, 14, 24573-24621, doi:10.5194/acpd-14-
1009 24573-2014, 2014.
- 1010 Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G.,
1011 Jenkin, M. E., Rossi, M. J. and Troe, J.: Evaluated kinetic and photochemical data for
1012 atmospheric chemistry: Volume I – gas phase reactions of Ox, HO_x, NO_x and SO_x species,
1013 *Atmos. Chem. Phys.*, 4, 1461–1738, doi:10.5194/acp-4-1461-2004, 2004.
- 1014 Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G.,
1015 Jenkin, M. E., Rossi, M. J., Troe, J., and IUPAC Subcommittee: Evaluated kinetic and
1016 photochemical data for atmospheric chemistry: Volume II – gas phase reactions of organic
1017 species, *Atmos. Chem. Phys.*, 6, 3625–4055, doi:10.5194/acp-6-3625-2006, 2006.
- 1018 Baklanov, A., Schlünzen, K., Suppan, P., Baldasano, J., Brunner, D., Aksoyoglu, S.,
1019 Carmichael, G., Douros, J., Flemming, J., Forkel, R., Galmarini, S., Gauss, M., Grell, G.,
1020 Hirtl, M., Joffre, S., Jorba, O., Kaas, E., Kaasik, M., Kallos, G., Kong, X., Korsholm, U.,
1021 Kurganskiy, A., Kushta, J., Lohmann, U., Mahura, A., Manders-Groot, A., Maurizi, A.,
1022 Moussiopoulos, N., Rao, S. T., Savage, N., Seigneur, C., Sokhi, R. S., Solazzo, E., Solomos,
1023 S., Sørensen, B., Tsegas, G., Vignati, E., Vogel, B., and Zhang, Y.: Online coupled regional
1024 meteorology chemistry models in Europe: current status and prospects, *Atmos. Chem. Phys.*,
1025 14, 317-398, doi:10.5194/acp-14-317-2014, 2014.
- 1026 Banda, N., Krol, M, van Noije, T. , van Weele, M. , Williams, J. E., Le Sager, P. , Niemeier,
1027 U., Thomason, L. and Röckmann, T. : The effect of stratospheric sulfur from Mount Pinatubo
1028 on tropospheric oxidizing capacity and methane, *J. Geophys. Res. Atmos.*, 119,
1029 doi:10.1002/2014JD022137, 2014.

1030

1031

1032 Barkley, M., Description of MEGAN biogenic VOC emissions in GEOS-Chem, 2010.
1033 http://acmg.seas.harvard.edu/geos/wiki_docs/emissions/megan.pdf

1034 Bechtold, P., Semane, N., Lopez, P., Chaboureau, J-P, Beljaars, A., Bormann, N: 2014:
1035 Representing Equilibrium and Nonequilibrium Convection in Large-Scale Models. *J. Atmos.*
1036 *Sci.*, 71, 734–753. doi: <http://dx.doi.org/10.1175/JAS-D-13-0163.1>. 2014.

1037 Beekmann M., Ancellet G., Megie G., Smit H. G. J., and Kley D.: Intercomparison campaign
1038 for vertical ozone profiles including electrochemical sondes of ECC and Brewer-Mast type
1039 and aground based UV-differential absorption radar, *J. Atmos. Chem.*, 10, 259–288, 1994.

1040 Beljaars, A. and Viterbo, P.: The role of the boundary layer in a numerical weather prediction
1041 model, in: *Clear and cloudy boundary layers*, A.A.M. Holtslag and P. Duynkerke (eds.),
1042 Royal Netherlands Academy of Arts and Sciences, p. 287-304, Amsterdam, North Holland
1043 Publishers, 1998.

1044 Beljaars, A., Bechtold, P., Kohler, M., Morcrette, J-J., Tompkins, A., Viterbo, P. and Wedi,
1045 N.: The numerics of physical parameterization, Seminar on Recent developments in
1046 numerical methods for atmospheric and ocean modelling, 6-10 September,
1047 <http://www.ecmwf.int/publications/library/do/references/>, 2004.

1048 Benedetti, A., Morcrette, J.-J., Boucher, O., Dethof, A., Engelen, R. J., Fisher, M., Flentje, H.,
1049 Huneeus, N., Jones, L., Kaiser, J. W., Kinne, S., Mangold, A., Razinger, M., Simmons, A. J.,
1050 Suttie, M., and the GEMS-AER team: Aerosol analysis and forecast in the European Centre
1051 for Medium-Range Weather Forecasts Integrated Forecast System: 2. Data assimilation, *J.*
1052 *Geophys. Res.*, 114, D13205, doi:10.1029/2008JD011115, 2009.

1053 Bousserez, N., Attié, J.-L., Peuch, V.-H., Michou, M., and Pfister, G.: Evaluation of the
1054 MOCAGE chemistry and transport model during the ICARTT/ITOP experiment, *J. Geophys.*
1055 *Res.*, 112, D10S42, doi:10.1029/2006JD007595, 2007.

1056 Callies, J., Corpaciacoli, E., Eisinger, M., Hahne, A., and Lefebvre, A.: GOME-2 Metop's
1057 Second Generation Sensor for Operational Ozone Monitoring, *ESA Bulletin*, 102, 2000.

1058 Cariolle, D. and Deque, M.: Southern hemisphere medium-scale waves and total ozone
1059 disturbances in a spectral general circulation model, *J. Geophys. Res.*, 91D, 10825–10846,
1060 1986.

1061 Cariolle, D. and Teyssèdre, H.: A revised linear ozone photochemistry parameterization for
1062 use in transport and general circulation models: multi-annual simulations, *Atmos. Chem.*
1063 *Phys.*, 7, 2183-2196, doi:10.5194/acp-7-2183-2007, 2007.

1064 Carslaw, K. S., Luo, B., Peter, T., and Clegg, S. L.: Vapour pressures of
1065 H₂SO₄/HNO₃/HBr/H₂O solutions to low stratospheric temperatures, *Geophys. Res. Lett.*, 22,
1066 247-250, 1995.

1067 Cecil, D.J., Buechler, D. E., Blakeslee, R. J. : Gridded lightning climatology from TRMM-
1068 LIS and OTD: Dataset description, *Atmospheric Research*, 135–136, 404-414,
1069 doi:10.1016/j.atmosres.2012.06.028, 2012.

1070 Chin, M., D. J. Jacob, G. M. Gardner, M. S. Foreman-Fowler, P. A. Spiro, and D. L. Savoie:
1071 A global three-dimensional model of tropospheric sulfate, *J. Geophys. Res.*, 101, (D13),
1072 18,667–18,690, 1996

1073 DeCaria, A. J., Pickering, K. E. , Stenchikov, G. L. and Ott, L. E.: Lightning-generated NO_x
1074 and its impact on tropospheric ozone production: A three-dimensional modeling study of a
1075 Stratosphere-Troposphere Experiment: Radiation, Aerosols and Ozone (STERA0-A)
1076 thunderstorm, *J. Geophys. Res.*, 110, D14303, doi:10.1029/2004JD005556, 2005.

1077 Dee, D.P., Uppala, S.M., Simmons, A.J., Berrisford, P., Poli, P., Kobayashi, S., Andrae, U.,
1078 Balmaseda, M.A., Balsamo, G., Bauer, P., Bechtold, P., Beljaars, A.C.M., van de Berg, L.,
1079 Bidlot, J., Bormann, N., Delsol, C., Dragani, R., Fuentes, M., Geer, A.J., Haimberger, L.,
1080 Healy, S.B., Hersbach, H., Hólm, E.V., Isaksen, L., Kållberg, P., Köhler, M., Matricardi, M.,
1081 McNally, A.P., Monge-Sanz, B.M., Morcrette, J.-J., Park, B.-K., Peubey, C., de Rosnay, P.,
1082 Tavolato, C., Thépaut, J.-N., Vitart, F.: The ERA-Interim reanalysis: Configuration and
1083 performance of the data assimilation system, *Quarterly Journal of the Royal Meteorological*
1084 *Society*, 2011.

1085 Deeter, M. N., S. Martínez-Alonso, D. P. Edwards, L. K. Emmons, J. C. Gille, H. M. Worden,
1086 J. V. Pittman, B. C. Daube, and S. C. Wofsy: Validation of MOPITT Version 5 thermal-
1087 infrared, near-infrared, and multispectral carbon monoxide profile retrievals for 2000–2011, *J.*
1088 *Geophys. Res. Atmos.*, 118, 6710–6725, doi:10.1002/jgrd.50272, 2013a.

1089 Deeter, M.N., MOPITT Version 6 Product User's Guide, Technical Report, NCAR, Boulder,
1090 USA, 2013.137 (656), pp. 553-597, 2013b.

1091 Dentener, F. J. and Crutzen, P. J.: Reaction of N₂O₅ on tropospheric aerosols: Impact on the
1092 global distributions of NO_x, O₃ and OH, *J. Geophys. Res.*, 98(D4), 7149–7163, 1993.

1093 Diamantakis, M. and Flemming, J.: Global mass fixer algorithms for conservative tracer
1094 transport in the ECMWF model, *Geosci. Model Dev.*, 7, 965-979, doi:10.5194/gmd-7-965-
1095 2014, 2014.

1096 Elguindi, N., Clark, H., Ordóñez, C., Thouret, V., Flemming, J., Stein, O., Huijnen, V.,
1097 Moinat, P., Inness, A., Peuch, V.-H., Stohl, A., Turquety, S., Athier, G., Cammas, J.-P., and
1098 Schultz, M.: Current status of the ability of the GEMS/MACC models to reproduce the
1099 tropospheric CO vertical distribution as measured by MOZAIC, *Geosci. Model Dev.*, 3, 501-
1100 518, doi:10.5194/gmd-3-501-2010, 2010.

1101 Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D.,
1102 Granier, C., Guenther, A., Kinnison, D., Laepple, T., Orlando, J., Tie, X., Tyndall, G.,
1103 Wiedinmyer, C., Baughcum, S. L., and Kloster, S.: Description and evaluation of the Model
1104 for Ozone and Related chemical Tracers, version 4 (MOZART-4), *Geosci. Model Dev.*, 3, 43-
1105 67, doi:10.5194/gmd-3-43-2010, 2010.

1106 Emmons, L.K., Arnold, S., Monks, S., Huijnen, V., Tilmes, S., Law, K., Thomas, J.L., Raut,
1107 J.-C., Bouarar, I., Turquety, S., Long, Y., Duncan, B., Steenrod, S., Strode, S., Flemming, J.
1108 Mao, J., Langner, J., Thompson, A., Tarasick, D., Apel, E., Blake, D., Brune, W., Cohen, R.,
1109 Dibb, J., Diskin, G. S., Fried, A., Hall, S., Huey, G., Weinheimer, ennberg, P., Wisthaler, A.,
1110 de Gouw, J., Holloway, J., Montzka, S., Nowak, J., Roberts, J. and Ryerson, J.: The
1111 POLARCAT Model Intercomparison Project (POLMIP): Overview and evaluation with
1112 observations, *Atmospheric Chemistry and Physics*, to be submitted, 2014.

1113 Engelen, R. J., Serrar, S., and Chevallier, F.: Four-dimensional data assimilation of
1114 atmospheric CO₂ using AIRS observations, *J. Geophys. Res.*, 114, D03303,
1115 doi:10.1029/2008JD010739, 2009.

1116 Errera, Q., Daerden, F., Chabrilat, S., Lambert, J. C., Lahoz, W. A., Viscardy, S., Bonjean,
1117 S., and Fonteyn, D.: 4D-Var assimilation of MIPAS chemical observations: ozone and
1118 nitrogen dioxide analyses, *Atmos. Chem. Phys.*, 8, 6169-6187, doi:10.5194/acp-8-6169-2008,
1119 2008.

1120 Evans, M. J., and Jacob, D. J.: Impact of new laboratory studies of N₂O₅ hydrolysis on global
1121 model budgets of tropospheric nitrogen oxides, ozone, and OH, *Geophys. Res. Lett.*, 32,
1122 L09813, doi:10.1029/2005GL022469, 2005.

1123 Fiore, A.M., Naik, V., Spracklen, D.V., Steiner, A., Unger, N., Prather, M., Bergmann, D.,
1124 Cameron-Smith, P.J., Cionni, I., Collins, W.J., Dalsoren, S., Eyring, V., Folberth, G.A.,
1125 Ginoux, P., Horowitz, L.W., Josse, B., Lamarque, J.-F., MacKenzie, I.A., Nagashima, T.,
1126 O'Connor, F.M., Righi, M., Rumbold, S.T., Shindell, D.T., Skeie, R.B., Sudo, K., Szopa, S.,
1127 Takemura, T., Zeng, G., Global air quality and climate, *Chemical Society Reviews*, 41 (19),
1128 pp. 6663-6683, 2012.

1129 Flemming, J., Stern, R., and Yamartino, R. J.: A new air quality regime classification scheme
1130 for O₃, NO₂, SO₂ and PM₁₀ observations sites, *Atmos. Environ.*, 39, 6121–6129, 2005

1131 Flemming, J., Inness, A., Flentje, H., Huijnen, V., Moinat, P., Schultz, M. G., and Stein, O.:
1132 Coupling global chemistry transport models to ECMWF's integrated forecast system, *Geosci.*
1133 *Model Dev.*, 2, 253-265, doi:10.5194/gmd-2-253-2009, 2009a.

1134 Flemming, J., Inness, A., Flentje, H., Huijnen, V., Moinat, P., Schultz, M. G., and Stein, O.:
1135 Coupling global chemistry transport models to ECMWF's integrated forecast system
1136 ECMWF technical memorandum 590
1137 [http://old.ecmwf.int/publications/library/ecpublications/_pdf/tm/501-600/tm590.pdf]
1138 tm590.pdf, 2009 b.

1139 Flemming, J., Inness, A., Jones, L., Eskes, H. J., Huijnen, V., Schultz, M. G., Stein, O.,
1140 Cariolle, D., Kinnison, D., and Brasseur, G.: Forecasts and assimilation experiments of the
1141 Antarctic ozone hole 2008, *Atmos. Chem. Phys.*, 11, 1961–1977, doi:10.5194/acp-11-1961-
1142 2011, 2011 a.

1143 Flemming, J. and Huijnen, V.: IFS Tracer Transport Study, MACC Deliverable G-
1144 RG 4.2, Tech. rep., ECMWF, [http://www.gmes-atmosphere.eu/documents/deliverables/g-
1145 rg/ifs_transport_study.pdf](http://www.gmes-atmosphere.eu/documents/deliverables/g-rg/ifs_transport_study.pdf), 2011 b.

1146 Flemming, J.; Peuch, V.-H.; Engelen, R.; Kaiser, J.W. A European Global-to-Regional Air
1147 Pollution Forecasting System that Combines Modeling with Satellite Observations; *EM*
1148 *Magazine of A&WMA*, November 2013, pp. 6-10.
1149 [https://www.researchgate.net/publication/259535688_A_European_Global-to-
1150 Regional_Air_Pollution_Forecasting_System_that_Combines_Modeling_with_Satellite_Obse
1151 rvations](https://www.researchgate.net/publication/259535688_A_European_Global-to-Regional_Air_Pollution_Forecasting_System_that_Combines_Modeling_with_Satellite_Observations), 2013

1152 Forbes, R.M., A.M. Tompkins & A. Untch, A new prognostic bulk-microphysics scheme for
1153 the IFS. ECMWF Tech. Memo. No. 649, 2011.

1154 Fu, Q., Yang, P. and Sun, W. B.: An accurate parametrization of the infrared radiative
1155 properties of cyrrus clouds of climate models. *J. Climate*, 11, 2223–2237, 1998

1156 Gauss, M., Isaksen, I. S. A., Lee, D. S., and Søvde, O. A.: Impact of aircraft NO_x emissions
1157 on the atmosphere – tradeoffs to reduce the impact, *Atmos. Chem. Phys.*, 6, 1529-1548,
1158 doi:10.5194/acp-6-1529-2006, 2006.

1159 Gery, M., Whitten, G. Z., Killus, J. P., and Dodge, M. C.: A photochemical kinetics
1160 mechanism for urban and regional scale computer modelling, *J. Geophys. Res.*, 94, 18925–
1161 18956, 1989.

1162 Granier, C., J.F. Lamarque, A. Mieville, J.F. Muller, J. Olivier, J. Orlando, J. Peters, G.
1163 Petron, G. Tyndall, S. Wallens, POET, a database of surface emissions of ozone precursors,
1164 available on internet at <http://www.aero.jussieu.fr/projet/ACCENT/POET.php> , 2005.

1165 Granier, C., B. Bessagnet, T. Bond, A. D'Angiola, H.D.v.d. Gon, G.J. Frost, A. Heil, J.W.
1166 Kaiser, S. Kinne, Z. Klimont, S. Kloster, J.-F. Lamarque, C. Lioussse, T. Masui, F. Meleux, A.
1167 Mieville, T. Ohara, J.-C. Raut, K. Riahi, M.G. Schultz, S.J. Smith, A. Thomson, J.v.
1168 Aardenne, G.R.v.d. Werf, and D.P.v. Vuuren, Evolution of anthropogenic and biomass
1169 burning emissions of air pollutants at global and regional scales during the 1980-2010 period,
1170 *Climatic Change*, 109(1-2), 163-190, doi:10.1007/s10584-011-0154-1, 2011.

1171 Grell, G. A., Peckham, S. E. , Schmitz, R , McKeen, S. A.,. Frost, G. J, Skamarock, W. and
1172 Eder B.: Fully coupled online chemistry within the WRF model, *Atmospheric Environment* ,
1173 39, 37, 6957-6975, 2005.

1174 Grooß, J.-U. and Russell III, J. M.: Technical note: A stratospheric climatology for O₃, H₂O,
1175 CH₄, NO_x, HCl and HF derived from HALOE measurements, *Atmos. Chem. Phys.*, 5, 2797–
1176 2807, doi:10.5194/acp-5-2797-2005, 2005

1177 Guenther, A. B., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates
1178 of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
1179 Aerosols from Nature), *Atmos. Chem. Phys.*, 6, 3181–3210, doi:10.5194/acp-6-3181-2006,
1180 2006.

1181 Haywood, J. M., Roberts, D. L., Slingo, A., Edwards, J. M., and Shine, K. P.: General
1182 circulation model calculations of the direct radiative forcing by anthropogenic sulfate and
1183 fossil-fuel soot aerosol, *Journal of Climate*, 10, 1562–1577, 1997.

1184 Hertel, O., Berkowicz, R., Christensen, J. and Hov Ø: Test of two numerical schemes for use
1185 in atmospheric transport-chemistry models *Atmos. Environ.*, 27A(16), 2591–2611, 1993.

1186 Hollingsworth, A., Engelen, R.J., Textor, C., Benedetti, A., Boucher, O. , Chevallier, F.,
1187 Dethof, A., Elbern, H., Eskes, H., Flemming, J., Granier, C., Kaiser, J.W. , Morcrette, J.-J.,
1188 Rayner, P., Peuch, V.H., Rouil, L., Schultz, M.G., Simmons, A.J and The GEMS
1189 Consortium: Toward a Monitoring and Forecasting System For Atmospheric Composition:
1190 The GEMS Project. *Bull. Amer. Meteor. Soc.*, 89, 1147-1164, 2008.

1191 Holtslag, A.A. and B. Boville: Local versus nonlocal boundary-layer diffusion in a global
1192 climate model, *J. Clim.*, 6, 1825-1842, 1993.

1193 Horowitz, L. W., Walters, S., Mauzerall, D. L., Emmons, L. K., Rasch, P. J., Granier, C., Tie,
1194 X., Lamarque, J.-F., Schultz, M. G., Tyndall, G. S., Orlando, J. J., and Brasseur, G. P.: A
1195 global simulation of tropospheric ozone and related tracers, Description and Evaluation of
1196 MOZART version 2, *J. Geophys. Res.*, 108, 4784, doi:10.1029/2002JD002853, 2003.

1197 Hortal, M. and Simmons, A. J.: Use of reduced Gaussian grids in spectral models, *Mon.*
1198 *Weather Rev.*, 119, 1057 1074, 1991.

1199 Hortal, M.: The development and testing of a new two-time-level semi-Lagrangian scheme
1200 (SETTLS) in the ECMWF forecast model, 128, 1671–1687, DOI: 10.1002/qj.200212858314,
1201 2002.

1202 Houweling, S., Dentener, F. J., and Lelieveld, J.: The impact of nonmethane hydrocarbon
1203 compounds on tropospheric photochemistry, *J. Geophys. Res.*, 103(D9), 10673–10696, 1998.

1204 Huijnen, V., Williams, J., van Weele, M., van Noije, T., Krol, M., Dentener, F., Segers, A.,
1205 Houweling, S., Peters, W., de Laat, J., Boersma, F., Bergamaschi, P., van Velthoven, P., Le
1206 Sager, P., Eskes, H., Alkemade, F., Scheele, R., Nédélec, P., and Pätz, H.-W.: The global
1207 chemistry transport model TM5: description and evaluation of the tropospheric chemistry
1208 version 3.0, *Geosci. Model Dev.*, 3, 445-473, doi:10.5194/gmd-3-445-2010.

1209 Huijnen, V., Flemming, J., Kaiser, J. W., Inness, A., Leitão, J., Heil, A., Eskes, H. J., Schultz,
1210 M. G., Benedetti, A., Hadji-Lazaro, J., Dufour, G., and Eremenko, M.: Hindcast experiments
1211 of tropospheric composition during the summer 2010 fires over western Russia, *Atmos.*
1212 *Chem. Phys.*, 12, 4341-4364, doi:10.5194/acp-12-4341-2012, 2012.

1213 Huijnen, V., Williams, J. E., and Flemming, J.: Modeling global impacts of heterogeneous
1214 loss of HO₂ on cloud droplets, ice particles and aerosols, *Atmos. Chem. Phys. Discuss.*, 14,
1215 8575-8632, doi:10.5194/acpd-14-8575-2014, 2014.

1216 Im, U., Bianconi, R., Solazzo, E., Kioutsioukis, I., Badia, A., Balzarini, A., Baró, R., Bellasio,
1217 R., Brunner, D., Chemel, C., Curci, G., Flemming, J., Forkel, R., Giordano, L., Jiménez-
1218 Guerrero, P., Hirtl, M., Hodzic, A., Honzak, L., Jorba, O., Knote, C., Kuenen, J.J.P., Makar,
1219 P.A., Manders-Groot, A., Neal, L., Pérez, J.L., Pirovano, G., Pouliot, G., San Jose, R.,
1220 Savage, N., Schroder, W., Sokhi, R.S., Syrakov, D., Torian, A., Tuccella, P., Werhahn, J.,
1221 Wolke, R., Yahya, K., Zabkar, R., Zhang, Y., Zhang, J., Hogrefe, C., Galmarini, S.:
1222 Evaluation of operational on-line-coupled regional air quality models over Europe and North
1223 America in the context of AQMEII phase 2. Part I: Ozone, *Atmospheric Environment*, doi:
1224 10.1016/j.atmosenv.2014.09.042, 2014.

1225 Inness, A., Baier, F., Benedetti, A., Bouarar, I., Chabrillat, S., Clark, H., Clerbaux, C.,
1226 Coheur, P., Engelen, R. J., Errera, Q., Flemming, J., George, M., Granier, C., Hadji-Lazaro,
1227 J., Huijnen, V., Hurtmans, D., Jones, L., Kaiser, J. W., Kapsomenakis, J., Lefever, K., Leitão,
1228 J., Razinger, M., Richter, A., Schultz, M. G., Simmons, A. J., Suttie, M., Stein, O., Thépaut,
1229 J.-N., Thouret, V., Vrekoussis, M., Zerefos, C., and the MACC team: The MACC reanalysis:
1230 an 8 yr data set of atmospheric composition, *Atmos. Chem. Phys.*, 13, 4073-4109,
1231 doi:10.5194/acp-13-4073-2013, 2013.

1232

1233 Inness, A., Blechschmidt, A.-M., Bouarar, I., Chabrillat, S., Crepulja, M., Engelen, R. J.,
1234 Eskes, H., Flemming, J., Gaudel, A., Hendrick, F., Huijnen, V., Jones, L., Kapsomenakis, J.,
1235 Katragkou, E., Keppens, A., Langerock, B., de Mazière, M., Melas, D., Parrington, M.,
1236 Peuch, V. H., Razinger, M., Richter, A., Schultz, M. G., Suttie, M., Thouret, V., Vrekoussis,
1237 M., Wagner, A., and Zerefos, C.: Data assimilation of satellite retrieved ozone, carbon
1238 monoxide and nitrogen dioxide with ECMWF's Composition-IFS, *Atmos. Chem. Phys.*
1239 *Discuss.*, 15, 4265-4331, doi:10.5194/acpd-15-4265-2015, 2015.

1240 Jacob, D.J. H. Liu, C.Mari, and R.M. Yantosca, Harvard wet deposition scheme for GMI,
1241 Harvard University Atmospheric Chemistry Modeling Group, revised March 2000.
1242 http://acmg.seas.harvard.edu/geos/wiki_docs/deposition/wetdep.jacob_etal_2000.pdf

1243 Jakob, C. and Klein, S.: A parameterization of the effects of cloud and precipitation overlap
1244 for use in general-circulation models, *Q. J. Roy. Meteor. Soc.*, 126, 2525–2544, 2000.

1245 Jung, T., T. N. Palmer, M. J. Rodwell, and S. Serrar, 2008: Diagnosing forecast error using
1246 relaxation experiments. ECMWF Newsletter 82, ECMWF, Shinfield Park, Reading, Berkshire
1247 RG2 9AX, UK.

1248 Kaiser, J. W., Heil, A., Andreae, M. O., Benedetti, A., Chubarova, N., Jones, L., Morcrette,
1249 J.-J., Razinger, M., Schultz, M. G., Suttie, M., and van der Werf, G. R.: Biomass burning
1250 emissions estimated with a global fire assimilation system based on observed fire radiative
1251 power, *Biogeosciences*, 9, 527-554, doi:10.5194/bg-9-527-2012, 2012.

1252 Kaminski, J. W., Neary, L., Struzewska, J., McConnell, J. C., Lupu, A., Jarosz, J., Toyota, K.,
1253 Gong, S. L., Côté, J., Liu, X., Chance, K., and Richter, A.: GEM-AQ, an on-line global
1254 multiscale chemical weather modelling system: model description and evaluation of gas phase
1255 chemistry processes, *Atmos. Chem. Phys.*, 8, 3255-3281, 2008.

1256 Kinnison, D. E., Brasseur, G. P., Walters, S. , Garcia, R. R., Marsh, D. R , Sassi, F., Harvey,
1257 V. L., Randall, C. E., Emmons, L., Lamarque, J. F., Hess, P. , Orlando, J. J., Tie, X. X. ,
1258 Randel, W. , Pan, L. L., Gettelman, A. , Granier, C., Diehl, T., Niemeier, U. and Simmons, A.
1259 J.: Sensitivity of Chemical Tracers to Meteorological Parameters in the MOZART-3
1260 Chemical Transport Model. *J. Geophys. Res.*, 112, D03303, doi:10.1029/2008JD010739,2007.

1261 Komhyr, W. D., Barnes, R. A., Borthers, G. B., Lathrop, J. A., Kerr, J. B., and Opperman, D.
1262 P.: Electrochemical concentration cell ozonesonde performance evaluation during STOIC
1263 1989, *J. Geophys. Res.*, 100, 9231–9244, 1995

1264 Lamarque, J.-F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C.
1265 L., Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P. J., and Tyndall, G. K.:
1266 CAM-chem: description and evaluation of interactive atmospheric chemistry in the
1267 Community Earth System Model, *Geosci. Model Dev.*, 5, 369-411, doi:10.5194/gmd-5-369-
1268 2012, 2012.

1269 Lamarque, J.-F., Shindell, D. T., Josse, B., Young, P. J., Cionni, I., Eyring, V., Bergmann, D.,
1270 Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi, G., Folberth, G.,
1271 Ghan, S. J., Horowitz, L. W., Lee, Y. H., MacKenzie, I. A., Nagashima, T., Naik, V.,
1272 Plummer, D., Righi, M., Rumbold, S. T., Schulz, M., Skeie, R. B., Stevenson, D. S., Strode,
1273 S., Sudo, K., Szopa, S., Voulgarakis, A., and Zeng, G.: The Atmospheric Chemistry and
1274 Climate Model Intercomparison Project (ACCMIP): overview and description of models,

1275 simulations and climate diagnostics, *Geosci. Model Dev.*, 6, 179-206, doi:10.5194/gmd-6-
1276 179-2013, 2013.

1277 Landgraf, J. and Crutzen, P. J.: An efficient method for online calculations of photolysis and
1278 heating rates, *J. Atmos. Sci.*, 55, 863–878, 1998.

1279 Lawrence, M.G. and Crutzen, P. J: The impact of cloud particle gravitational settling on
1280 soluble trace gas distributions. *Tellus B*, 50: 263–289. doi: 10.1034/j.1600-0889.1998.

1281 Lefever, K., van der A, R., Baier, F., Christophe, Y., Errera, Q., Eskes, H., Flemming, J.,
1282 Inness, A., Jones, L., Lambert, J.-C., Langerock, B., Schultz, M. G., Stein, O., Wagner, A.,
1283 and Chabrillat, S.: Copernicus atmospheric service for stratospheric ozone: validation and
1284 intercomparison of four near real-time analyses, 2009–2012, *Atmos. Chem. Phys. Discuss.*,
1285 14, 12461-12523, doi:10.5194/acpd-14-12461-2014, 2014.

1286 Levine, S. Z. and Schwartz, S. E.: In-cloud and below-cloud scavenging of nitric acid vapor.
1287 *Atmos. Environ.* doi:10.1016/0004-6981(82)90266-9, 1982.

1288 Lin, J.-T., McElroy, M. B., and Boersma, K. F.: Constraint of anthropogenic NO_x emissions
1289 in China from different sectors: a new methodology using multiple satellite retrievals, *Atmos.*
1290 *Chem. Phys.*, 10, 63-78, doi:10.5194/acp-10-63-2010, 2010.

1291 Liu, H., Jacob, D.J., Bey, I., Yantosca, R.M., 2001. Constraints from ²¹⁰Pb and ⁷Be on wet
1292 deposition and transport in a global three-dimensional chemical tracer model driven by
1293 assimilated meteorological fields. *Journal of Geophysical Research* 106, 12109e12128.

1294 McGregor, J. L.: C-CAM Geometric Aspects and Dynamical Formulation, Tech. Rep. 70,
1295 CSIRO Atmospheric Research, Aspendale, Victoria, 2005.

1296 MACC VAL report, Validation report of the MACC reanalysis of global atmospheric
1297 composition Period, 2003-
1298 2011.[http://www.copernicusatmosphere.eu/documents/maccii/deliverables/val/MACCII_VAL](http://www.copernicusatmosphere.eu/documents/maccii/deliverables/val/MACCII_VAL_DEL_D_83.4_REAreport02_20130207.pdf)
1299 [L_DEL_D_83.4_REAreport02_20130207.pdf](http://www.copernicusatmosphere.eu/documents/maccii/deliverables/val/MACCII_VAL_DEL_D_83.4_REAreport02_20130207.pdf), 2013

1300 Mann, G. W., Carslaw, K. S., Spracklen, D. V., Ridley, D. A., Manktelow, P. T.,
1301 Chipperfield, M. P., Pickering, S. J., and Johnson, C. E.: Description and evaluation of
1302 GLOMAP-mode: a modal global aerosol microphysics model for the UKCA composition-
1303 climate model, *Geosci. Model Dev.*, 3, 519-551, doi:10.5194/gmd-3-519-2010, 20103-651-
1304 2010, 2010.

1305 Marenco, A., Thouret, V., Nédelec, P., Smit, H. G., Helten, M., Kley, D., Karcher, F., Simon,
1306 P., Law, K., Pyle, J., Poschmann, G., Von Wrede, R., Hume, C., and Cook, T.: Measurement
1307 of ozone and water vapour by Airbus in-service air-craft: The MOZAIC airborne programme,
1308 an overview, *J. Geophys. Res.*, 103, 25631–25642, 1998.

1309 Mari, C., Jacob, D. J., and Bechtold, P.: Transport and scavenging of soluble gases in a deep
1310 convective cloud, *J. Geophys. Res.*, 105, 22 255-22 267, 2000.

1311 Matsumi, Y., Comes, F. J., Hancock, G., Hofzumahus, A., Hynes, A. J., Kawasaki, M., and
1312 Ravishankara, A. R.: Quantum yields for the production of O(1D) in the ultraviolet photolysis
1313 of ozone: recommendation based on evaluation of laboratory data, *J. Geophys. Res.*, 107,
1314 4024, doi:10.1029/2001JD000510, 2002.

1315 Meijer, E.W., P. F. J. van Velthoven, D. W. Brunner, H. Huntrieser and H. Kelder:
1316 Improvement and evaluation of the parameterization of nitrogen oxide production by
1317 lightning, *Physics and Chemistry of the Earth, Part C, Volume 26, Issue 8, Pages 577-583,*
1318 2001.

1319 Ménard , R. et al. ,Coupled chemical-dynamical data assimilation, Final Report, ESA/ESTEC.
1320 2007.

1321 Metzger, S., F. Dentener, S. Pandis, and J. Lelieveld, Gas/aerosol partitioning, 1, A
1322 computationally efficient model, *J. Geophys. Res.*, 107(D16), doi:10.1029/2001JD001102,
1323 2002a.

1324 Metzger, S., Dentener, F., Krol, M. C., Jeuken, A., and Lelieveld, J.: Gas/aerosol partitioning
1325 2. Global modeling results, *J. Geophys. Res.*, 107(D16), 4313, doi:10.1029/2001JD001103,
1326 2002b.

1327 Michou M., P. Laville, D. Serça, A. Fotiadi, P. Bouchou and V.-H. Peuch, Measured and
1328 modeled dry deposition velocities over the ESCOMPTE area, *Atmos. Res.*, 74 (1-4), 89-116,
1329 2004.

1330 Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N.,
1331 Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L.,
1332 Steenrod, S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Schlager, H., and Ancellet, G.:
1333 Multi-model study of chemical and physical controls on transport of anthropogenic and
1334 biomass burning pollution to the Arctic, *Atmos. Chem. Phys. Discuss.*, 14, 25281-25350,
1335 doi:10.5194/acpd-14-25281-2014, 2014.

1336 Morcrette, J.-J., Boucher, O., Jones, L., Salmond, D. , Bechtold, P., Beljaars, A., Benedetti,
1337 A., Bonet, A., Kaiser, J. W., Razinger, M., Schulz, M. , Serrar, S. , Simmons, A. J., Sofiev,
1338 M., Suttie, M., Tompkins, A. M. and Untch, A.: Aerosol analysis and forecast in the ECMWF
1339 Integrated Forecast System. Part I: Forward modelling, *J. Geophys. Res.*, 2009.

1340 Morgenstern, O., Braesicke, P., O'Connor, F. M., Bushell, A. C., Johnson, C. E., Osprey, S.
1341 M., and Pyle, J. A.: Evaluation of the new UKCA climate-composition model – Part 1: The
1342 stratosphere, *Geosci. Model Dev.*, 2, 43–57, doi:10.5194/gmd-2-43-2009, 2009.

1343 Nedelec, P., Cammas, J.-P., Thouret, V., Athier, G., Cousin, J.-M., Legrand, C., Abonnel, C.,
1344 Lecoœur, F., Cayez, G., and Marizy, C.: An improved infrared carbon monoxide analyser for
1345 routine measurements aboard commercial Airbus aircraft: technical validation and first
1346 scientific results of the MOZAIC III programme, *Atmos. Chem. Phys.*, 3, 1551–1564,
1347 doi:10.5194/acp-3-1551-2003, 2003.

1348 Neu, J. L. and Prather, M. J.: Toward a more physical representation of precipitation
1349 scavenging in global chemistry models: cloud overlap and ice physics and their impact on
1350 tropospheric ozone, *Atmos. Chem. Phys.*, 12, 3289-3310, doi:10.5194/acp-12-3289-2012,
1351 2012.

1352 O'Connor, F. M., Johnson, C. E., Morgenstern, O., Abraham, N. L., Braesicke, P., Dalvi, M.,
1353 Folberth, G. A., Sanderson, M. G., Telford, P. J., Voulgarakis, A., Young, P. J., Zeng, G.,
1354 Collins, W. J., and Pyle, J. A.: Evaluation of the new UKCA climate-composition model –
1355 Part 2: The Troposphere, *Geosci. Model Dev.*, 7, 41-91, doi:10.5194/gmd-7-41-2014, 2014.

1356 Olivier J., J. Peters, C. Granier, G. Petron, J.F. Muller, and S. Wallens: Present and future
1357 surface emissions of atmospheric compounds, POET report #2, EU project EVK2-1999-
1358 00011, 2003

1359 Ordóñez, C., Elguindi, N., Stein, O., Huijnen, V., Flemming, J., Inness, A., Flentje, H.,
1360 Katragkou, E., Moinat, P., Peuch, V.-H., Segers, A., Thouret, V., Athier, G., van Weele, M.,
1361 Zerefos, C. S., Cammas, J.-P., and Schultz, M. G.: Global model simulations of air pollution
1362 during the 2003 European heat wave, *Atmos. Chem. Phys.*, 10, 789-815, doi:10.5194/acp-10-
1363 789-2010, 2010.

1364 Ott, L. E., K. E. Pickering, G. L. Stenchikov, D. J. Allen, A. J. DeCaria, B. Ridley, R.-F. Lin,
1365 S. Lang, and W.-K. Tao (2010), Production of lightning NO_x and its vertical distribution

1366 calculated from three-dimensional cloud-scale chemical transport model simulations, J.
1367 Geophys. Res., 115, D04301, doi:10.1029/2009JD011880.

1368 Pickering, K. E., Y. Wang, W.-K. Tao, C. Price, and J.-F. Müller: Vertical distributions of
1369 lightning NO_x for use in regional and global chemical transport models, J. Geophys. Res.,
1370 103, 31,203 – 31,216, doi:10.1029/98JD0265. 1998.

1371 Pozzoli L., Bey, I., Rast, J. S., Schultz, M. G., Stier, P., and Feichter, J.: Trace gas and aerosol
1372 interactions in the fully coupled model of aerosol-chemistry-climate ECHAM5-HAMMOZ,
1373 PART I: Model description and insights from the spring 2001 TRACE-P experiment, J.
1374 Geophys. Res., 113, 2008.

1375 Prather, M., Ehhalt, D., et al.: Atmospheric chemistry and greenhouse gases, in: Climate
1376 Change 2001: The Scientific Basis, edited by: Houghton, J. T., Ding, Y., Griggs, D. J., et al.,
1377 239–287, Cambridge University Press, Cambridge, UK, 2001.

1378 Prather, M. J., Holmes, C. D., and Hsu, J.: Reactive greenhouse gas scenarios: Systematic
1379 exploration of uncertainties and the role of atmospheric chemistry, Geophys. Res. Lett., 39,
1380 L09803,doi:10.1029/2012GL051440, 2012.

1381 Price, C., and Rind, D.: A simple lightning parameterization for calculating global lightning
1382 distributions, J. Geophys. Res., 97, 9919-9933, 1992.

1383 Price, C., and Rind, D.: What determinest he cloud-to-ground fraction in thunderstorms?
1384 Geophys Res. Lett., 20, 463-466, 1993.

1385 Price, C., J. Penner, and M. Prather: NO_x from lightning 1. Global distributions based on
1386 lightning physics, J. Geophys. Res., 102, 5929–5941, doi:10.1029/96JD03504, 1997.

1387 Rast, S., Schultz, M.G. , Bey, I., van Noije, T. , Aghedo, A.M. , Brasseur, G.P., Diehl, T.,
1388 Esch, M., Ganzeveld, L., Kirchner, I., Kornblueh, L., Rhodin, A. , Röckner, E. , Schmidt, H. ,
1389 Schröder, S., Schulzweida, U., Stier, P., Thomas, K., Walters, S.: Evaluation of the
1390 tropospheric chemistry general circulation model ECHAM5–MOZ and its application to the
1391 analysis of the chemical composition of the troposphere with an emphasis on the late RETRO
1392 period 1990–2000, Reports on Earth-System Science, 114, Max-Planck Institut fuer
1393 Meteorologie, Hamburg, 2014.

1394 Redler, R., Valcke, S. and Ritzdorf, H.: OASIS4 - A Coupling Software for Next Generation
1395 Earth System Modelling, *Geoscience Model Development*, 3, 87 - 104, DOI:10.5194/gmd-3-
1396 87-2010.

1397 Richter, A., Burrows, J. P., Nüß, H., Granier, C, Niemeier, U., Increase in tropospheric
1398 nitrogen dioxide over China observed from space, *Nature*, 437, 129-132, doi:
1399 10.1038/nature04092, 2005.

1400 Sander, R., Compilation of Henry's Law Constants for Inorganic and Organic Species of
1401 Potential Importance in Environmental Chemistry, MPI for Chemistry Mainz, Germany,
1402 1999, <http://www.henrys-law.org/henry.pdf>

1403 Sander, S. P., Friedl, R. R., Golden, D. M., Kurylo, M. J., Moortgat, G. K., Keller-Rudek, H.,
1404 Wine, P. H., Ravishankara, A. R., Kolb, C. E., Molina, M. J., Finlayson-Pitts, B. J., Huie, R.
1405 E., and Orkin, V. L.: Chemical Kinetics and Photochemical Data for Use in Atmospheric
1406 Studies, Evaluation Number 15, JPL Publication 06-02, Jet Propulsion Laboratory, Pasadena,
1407 Calif., 2006.

1408 Sander, S. P., Abbatt, J. R., Burkholder, J. B., Friedl, R. R., Golden, D. M., Huie, R. E., Kolb,
1409 C. E., Kurylo, G., Moortgat, K., Orkin, V. L. and Wine, P. H.: Chemical kinetics and
1410 Photochemical Data for Use in Atmospheric studies, Evaluation No.17, JPL Publication 10-6,
1411 Jet Propulsion Laboratory, Pasadena, 2011.

1412 Savage, N. H., Agnew, P., Davis, L. S., Ordóñez, C., Thorpe, R., Johnson, C. E., O'Connor, F.
1413 M., and Dalvi, M.: Air quality modelling using the Met Office Unified Model (AQUUM OS24-
1414 26): model description and initial evaluation, *Geosci. Model Dev.*, 6, 353-372,
1415 doi:10.5194/gmd-6-353-2013, 2013.

1416 Schumann, U., and H. Huntrieser: The global lightning-induced nitrogen oxides source,
1417 *Atmos. Chem. Phys.*, 7, 3823–3907, 2007.

1418 Seinfeld J. H. and Pandis S. N., *Atmospheric Chemistry and Physics: From Air Pollution to*
1419 *Climate Change*, 1st edition, J. Wiley, New York, 1998.

1420 Shindell, D.T., G. Faluvegi, D.S. Stevenson, M.C. Krol, L.K. Emmons, J.-F. Lamarque, G.
1421 Pétron, F.J. Dentener, K. Ellingsen, M.G. Schultz, O. Wild, M. Amann, C.S. Atherton, D.J.
1422 Bergmann, I. Bey, T. Butler, J. Cofala, W.J. Collins, R.G. Derwent, R.M. Doherty, J. Drevet,
1423 H.J. Eskes, A.M. Fiore, M. Gauss, D.A. Hauglustaine, L.W. Horowitz, I.S.A. Isaksen, M.G.
1424 Lawrence, V. Montanaro, J.-F. Müller, G. Pitari, M.J. Prather, J.A. Pyle, S. Rast, J.M.

1425 Rodriguez, M.G. Sanderson, N.H. Savage, S.E. Strahan, K. Sudo, S. Szopa, N. Unger, T.P.C.
1426 van Noije, and G. Zeng: Multi-model simulations of carbon monoxide: Comparison with
1427 observations and projected near-future changes. *J. Geophys. Res.*, 111, D19306,
1428 doi:10.1029/2006JD007100, 2006.

1429 Slingo, A.: A GCM parameterization for the shortwave radiative properties of water clouds. *J.*
1430 *Atmos. Sci.*, 46, 1419–1427, 1989.

1431 Smithson, P. A., IPCC, 2001: climate change 2001: the scientific basis. Contribution of
1432 Working Group 1 to the Third Assessment Report of the Intergovernmental Panel on Climate
1433 Change, edited by J. T. Houghton, Y. Ding, D. J. Griggs, M. Noguer, P. J. van der Linden, X.
1434 Dai, K. Maskell and C. A. Johnson (eds). Cambridge University Press, Cambridge, UK, and
1435 New York, USA, *Int. J. Climatol.*, 22: 1144. doi: 10.1002/joc.763, 2002.

1436 Steil, B., Dameris, M., Brühl, C., Crutzen, P. J., Grewe, V., Ponater, M., and Sausen, R.:
1437 Development of a chemistry module for GCMs: first results of a multiannual integration,
1438 *Ann. Geophys.*, 16, 205-228, doi:10.1007/s00585-998-0205-8, 1998.

1439 Stein, O., Flemming, J., Inness, A., Kaiser, J.W., Schultz, M.G. , Global reactive gases
1440 forecasts and reanalysis in the MACC project, *Journal of Integrative Environmental Sciences*,
1441 9, Iss. sup1, 57-70, doi:10.1080/1943815X.2012.696545, 2012.

1442 Stein, O., Schultz, M. G., Bouarar, I., Clark, H., Huijnen, V., Gaudel, A., George, M., and
1443 Clerbaux, C.: On the wintertime low bias of Northern Hemisphere carbon monoxide in global
1444 model studies, *Atmos. Chem. Phys. Discuss.*, 14, 245-301, doi:10.5194/acpd-14-245-2014,
1445 2014.

1446 Steinbrecht, W., Shwartz, R., and Claude, H.: New pump correction for the Brewer-Mast
1447 ozonesonde: Determination from experiment and instrument intercomparisons, *J. Atmos.*
1448 *Ocean. Tech.*15, 144–156, 1998.

1449 Stevenson, D. S., et al. (2006), Multimodel ensemble simulations of present-day and near-
1450 future tropospheric ozone, *J. Geophys. Res.*, 111, D08301, doi:10.1029/2005JD006338.

1451 Temperton, C., Hortal, M. and Simmons, A.: A two-time-level semi-Lagrangian global
1452 spectral model, *QJR*, 127, 111-127, 2001.

1453 Tiedtke, M. A: comprehensive mass flux scheme for cumulus parameterization in large-scale
1454 models. *Mon. Weather. Rev.*, 117(8):1779-1800, 1989.

1455 Tilmes, S., Lamarque, J.-F., Emmons, L. K., Conley, A., Schultz, M. G., Saunio, M.,
1456 Thouret, V., Thompson, A. M., Oltmans, S. J., Johnson, B., and Tarasick, D.: Technical Note:
1457 Ozonesonde climatology between 1995 and 2011: description, evaluation and applications,
1458 *Atmos. Chem. Phys.*, 12, 7475-7497, doi:10.5194/acp-12-7475-2012, 2012.

1459 Wesely, M.L.: Parameterization of Surface Resistances to Gaseous Dry Deposition in
1460 Regional-Scale Numerical Models. *Atmos. Environ.*, 23, 1293-1304, 1989.

1461 Val Martin, M., Heald, C. L. and Arnold, S. R.: Coupling dry deposition to vegetation
1462 phenology in the Community Earth System Model: Implications for the simulation of surface
1463 O₃, *Geophys. Res. Lett.*, 41, 2988–2996, doi:10.1002/2014GL059651., 2014
1464 van Noije, T. P. C., Le Sager, P., Segers, A. J., van Velthoven, P. F. J., Krol, M. C., Hazeleger, W., Williams,
1465 A. G., and Chambers, S. D.: Simulation of tropospheric chemistry and aerosols with the
1466 climate model EC-Earth, *Geosci. Model Dev.*, 7, 2435-2475, doi:10.5194/gmd-7-2435-2014,
1467 2014.

1468 von Blohn, N., Diehl, K., Mitra, S. K., and Borrmann, S.: Wind tunnel experiments on the
1469 retention of trace gases during riming: nitric acid, hydrochloric acid, and hydrogen peroxide,
1470 *Atmos. Chem. Phys.*, 11, 11569-11579, doi:10.5194/acp-11-11569-2011, 2011.

1471 Voulgarakis, A., Naik, V., Lamarque, J.-F., Shindell, D. T., Young, P. J., Prather, M. J., Wild,
1472 O., Field, R. D., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins, W. J., Dalsøren, S. B.,
1473 Doherty, R. M., Eyring, V., Faluvegi, G., Folberth, G. A., Horowitz, L. W., Josse, B.,
1474 MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T., Stevenson, D.
1475 S., Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Analysis of present day and future OH
1476 and methane lifetime in the ACCMIP simulations, *Atmos. Chem. Phys.*, 13, 2563-2587,
1477 doi:10.5194/acp-13-2563-2013, 2013.

1478 Vrekoussis, M., Wittrock, F., Richter, A., and Burrows, J. P.: GOME-2 observations of
1479 oxygenated VOCs: what can we learn from the ratio glyoxal to formaldehyde on a global
1480 scale?, *Atmos. Chem. Phys.*, 10, 10145-10160, doi:10.5194/acp-10-10145-2010, 2010.

1481 Williams, J. E., Strunk, A., Huijnen, V., and van Weele, M.: The application of the Modified
1482 Band Approach for the calculation of on-line photodissociation rate constants in TM5:
1483 implications for oxidative capacity, *Geosci. Model Dev.*, 5, 15-35, doi:10.5194/gmd-5-15-
1484 2012, 2012.

1485 Williams, J. E., van Velthoven, P. F. J., and Brenninkmeijer, C. A. M.: Quantifying the
1486 uncertainty in simulating global tropospheric composition due to the variability in global
1487 emission estimates of Biogenic Volatile Organic 2857-2013, 2013.

- 1488 Wittrock, F., A. Richter, H. Oetjen, J. P. Burrows, M. Kanakidou, S. Myriokefalitakis, R.
1489 Volkamer, S. Beirle, U. Platt, and T. Wagner, Simultaneous global observations of glyoxal
1490 and formaldehyde from space, *Geophys. Res. Lett.*, 33, L16804, doi:10.1029/2006GL026310,
1491 2006Compounds, *Atmos. Chem. Phys.*, 13, 2857-2891, doi:10.5194/acp-13-2857-2013, 2013.
- 1492 Yarwood, G., Rao, S., Yocke, M., and Whitten, G.: Updates to the carbon bond chemical
1493 mechanism: CB05. Final report to the US EPA, EPA Report Number: RT-0400675, available
1494 at: www.camx.com, last access: 1 July 2014, 2005.
- 1495 WMO (2007), WMO Global Atmosphere Watch (GAW) Strategic Plan: 2008 – 2015. World
1496 Meteorological Organization, Geneva, Switzerland, 2007.
- 1497 Young, P. J., Archibald, A. T., Bowman, K. W., Lamarque, J.-F., Naik, V., Stevenson, D. S.,
1498 Tilmes, S., Voulgarakis, A., Wild, O., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins,
1499 W. J., Dalsøren, S. B., Doherty, R. M., Eyring, V., Faluvegi, G., Horowitz, L. W., Josse, B.,
1500 Lee, Y. H., MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T.,
1501 Skeie, R. B., Shindell, D. T., Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Pre-industrial
1502 to end 21st century projections of tropospheric ozone from the Atmospheric Chemistry and
1503 Climate Model Intercomparison Project (ACCMIP), *Atmos. Chem. Phys.*, 13, 2063-2090,
1504 doi:10.5194/acp-13-2063-2013, 2013.
- 1505 Zaveri, R. A. and Peters, L. K.: A new lumped structure photochemical mechanism for large-
1506 scale applications, *J. Geophys. Res.*, 104, 30387–30415, doi:10.1029/1999JD900876, 1999.
- 1507 Zdunkowski, W. G., Welsch, R. M., and Kord, G. J.: An investigation of the structure of
1508 typical 2-stream methods for the calculation of solar fluxes and heating rates in clouds,
1509 *Contrib. Atmos. Phys.*, 53, 215–238, 1980.
- 1510 Zhang, L., Brook, J. R., and Vet, R.: A revised parameterization for gaseous dry deposition in
1511 air-quality models, *Atmos. Chem. Phys.*, 3, 2067-2082, doi:10.5194/acp-3-2067-2003, 2003.
- 1512 Zhang, Y.: On-line coupled meteorology and chemistry models: history, current status, and
1513 outlook, *Atmos. Chem. Phys.*, 8, 2895-2032, 2008.
- 1514 Zhang, Y., Karamchandani, P., Glotfelty, T., Streets, D. G., Grell, G., Nenes, A., Yu, F. and
1515 Bennartz, R.: Development and initial application of the global-through-urban weather
1516 research and forecasting model with chemistry (GU-WRF/Chem), *J. Geophys. Res.*, 117,
1517 D20206, doi:10.1029/2012JD017966, 2012.

1518 Table 1 Annual emissions from anthropogenic, biogenic and natural sources and biomass
 1519 burning for 2008 in Tg for a C-IFS (CB05) run at T255 resolution. Anthropogenic NO
 1520 emissions contain a contribution of 1.8 Tg aircraft emissions and 12.3 Tg (5.7 Tg N) lightning
 1521 emissions (LiNO) is added in the biomass burning columns.

Species	Anthropogenic	Biogenic natural	and Biomass burning
CO	584	96	328
NO	70 + 1.8	10	9.2 + 12.3 (LiNO)
HCHO	3.4	4.0	4.9
CH ₃ OH	2.2	159	8.5
C ₂ H ₆	3.4	1.1	2.3
C ₂ H ₅ OH	3.1	0	0
C ₂ H ₄	7.7	18	4.3
C ₃ H ₈	4.0	1.3	1.2
C ₃ H ₆	3.5	7.6	2.5
Parafins (Tg C)	31	18	1.7
Olefines (Tg C)	2.4	0	0.7
Aldehydes (Tg C)	1.1	6.1	2.1
CH ₃ COCH ₃	1.3	28	2.4
Isoprene	0	523	0
Terpenes	0	97	0
SO ₂	98	9	2.2
DMS	0	38	0.2
NH ₃	40	11	6.2

1522

1523

1524 Table 2 Ozone sondes sites used in the evaluation for different regions

Region	Area S/W/N/E	Stations (Number of observations)
Europe	35°N/20°W/60°N/40°E	Barajas (52), DeBilt (57), Hohenpeissenberg (126), Legionowo (48), Lindenberg (52), Observatoire de Haute-Provence (46), Payerne (158), Prague (49), Uccle (142) and Valentia Observatory (49)
North America:	30°N/135°W/60°N/60°W	Boulder (65), Bratts Lake (61), Churchill (61), Egbert (29), Goose Bay (47), Kelowna (72), Stony Plain (77), Wallops (51), Yarmouth (60), Narragansett (7) and Trinidad Head (35)
Arctic:	60°N/180°W/90°N/180°E	Alert (52), Eureka (83), Keflavik (8), Lerwick (49), Ny-Aalesund (77), Resolute (63), Scoresbysund (54), Sodankyla (63), Summit (81) and Thule(15)
Tropics	20°S/180°W/20°N/180°E	Alajuela (47), Ascension Island (32), Hilo (47), Kuala Lumpur (24), Nairobi (39), Natal (48), Paramaribo (35), Poona (13), Samoa (33), San Cristobal (28), Suva (28), Thiruvananthapuram (12) and Watukosek (19)
East Asia	15°N/100°E/45°N/142°E	Hong Kong Observatory (49), Naha (37), Sapporo (42) and Tateno Tsukuba (49)
Antarctic	90°S/180°W/60°S/180°E	Davis (24), Dumont d'Urville (38), Maitri (9), Marambio (66), Neumayer (72), South Pole (63), Syowa(41) and McMurdo (18)

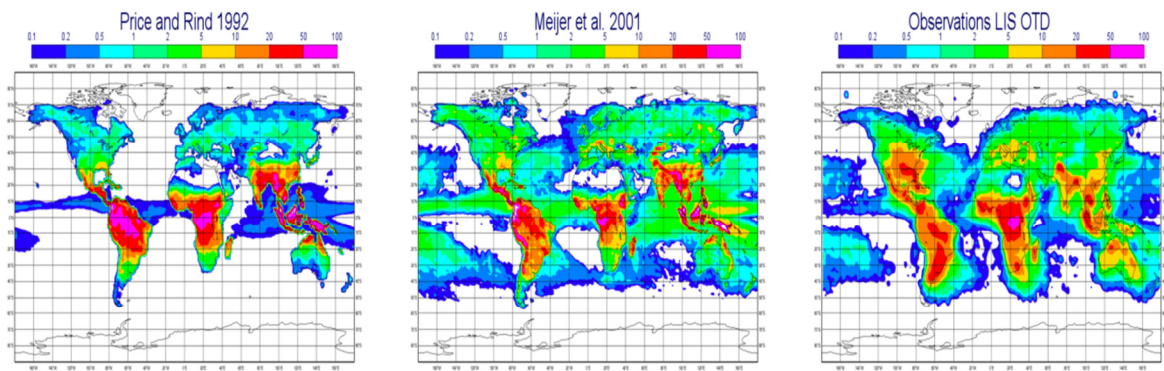
1525

1526 Table 3 Computational cost (BU) of a 24 h forecasts of different horizontal model resolutions
 1527 (60 levels) and chemistry schemes of C-IFS, IFS-MOZART and IFS, *not fully optimised.

Resolution	IFS-MOZART	C-IFS (MOZART)*	C-IFS (MOCAGE)*	C-IFS (CB05)	IFS
T159	205	56	147	20	6
T255	1200	-	-	55	12
T511	-	-	-	700	125

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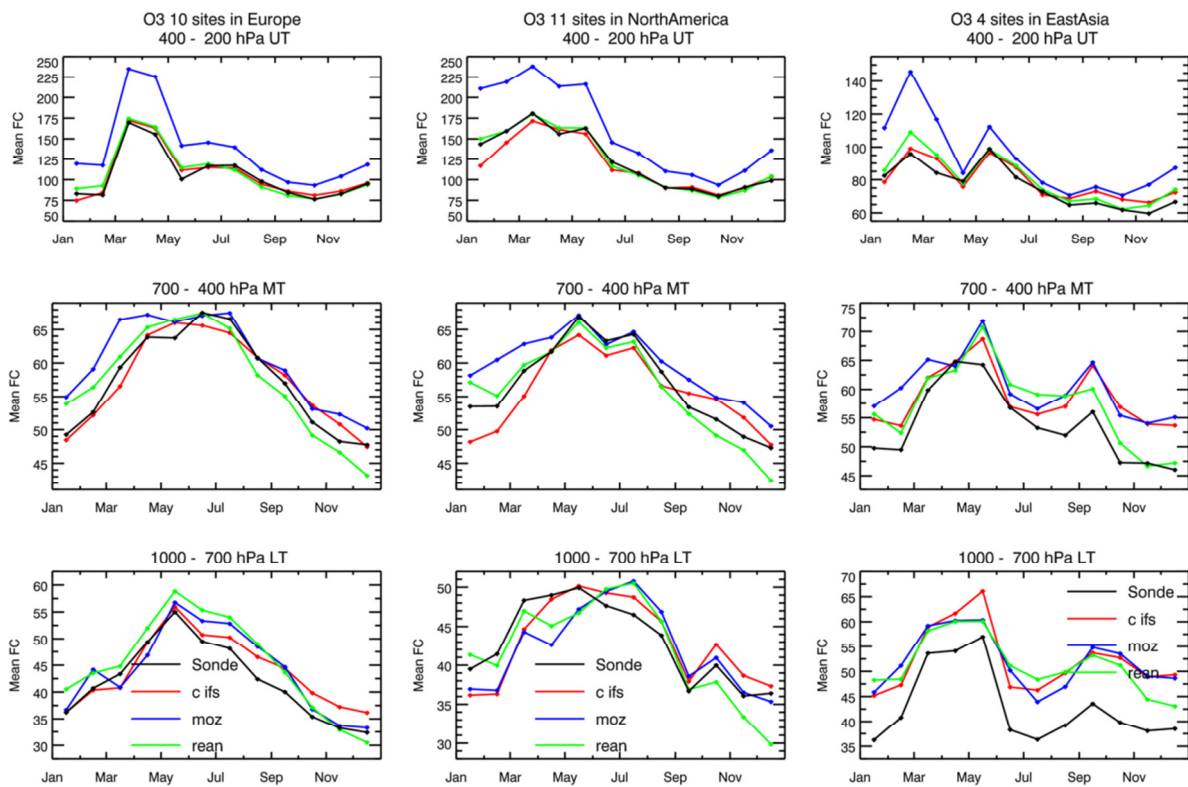
1531

1532 Figure 1 Flash density in flashes/(km² yr) from the IFS input data using the parameterization

1533 by Price and Rind (1992) (left), Meijer et al. (2001) (middle) and observations from the LIS

1534 OTD data base (right). All fields were scaled to an annual flash density of 46 fl/s.

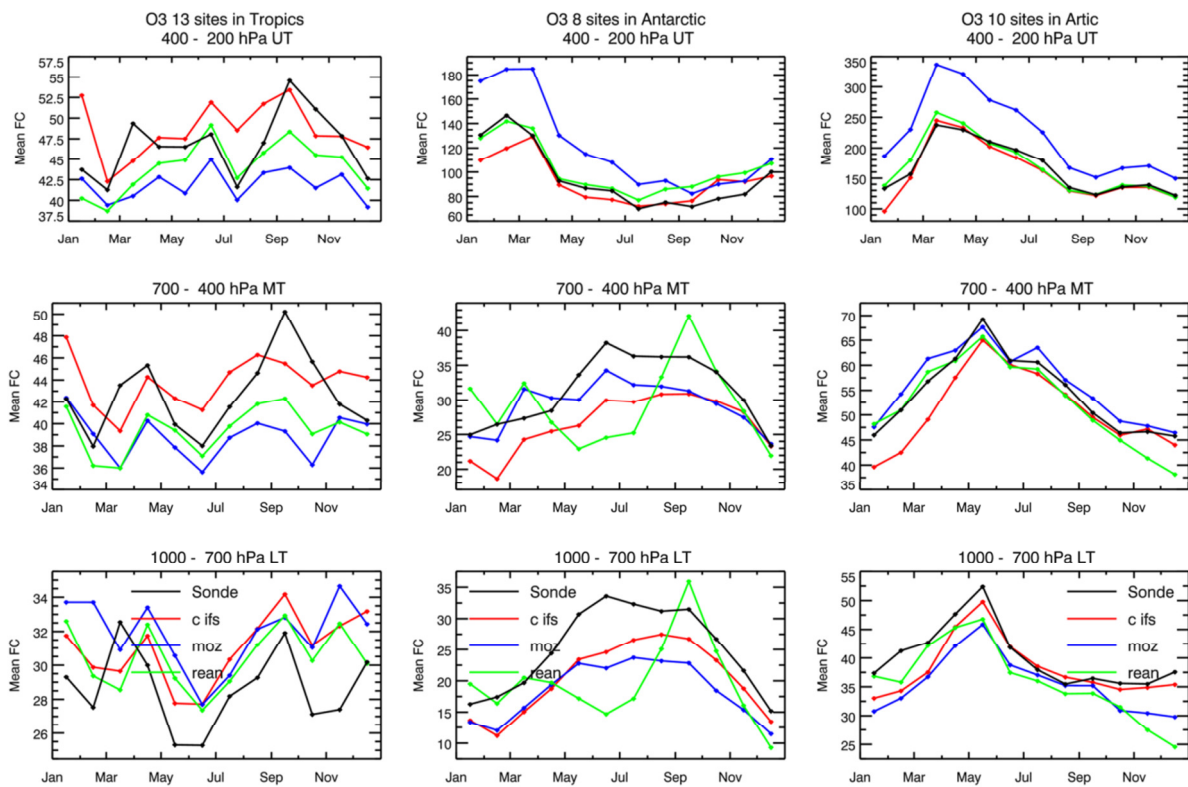
1535



1536

1537 Figure 2 Tropospheric ozone volume mixing ratios (ppb) over Europe (left) and North-
 1538 America (middle) and East Asia (right) averaged in the pressure range 1000-700 hPa
 1539 (bottom), 700-400 hPa (middle) and 400-200 hPa (top) observed by ozonesondes (black) and
 1540 simulated by C-IFS (red), MOZ (blue) and REAN (green) in 2008.

1541

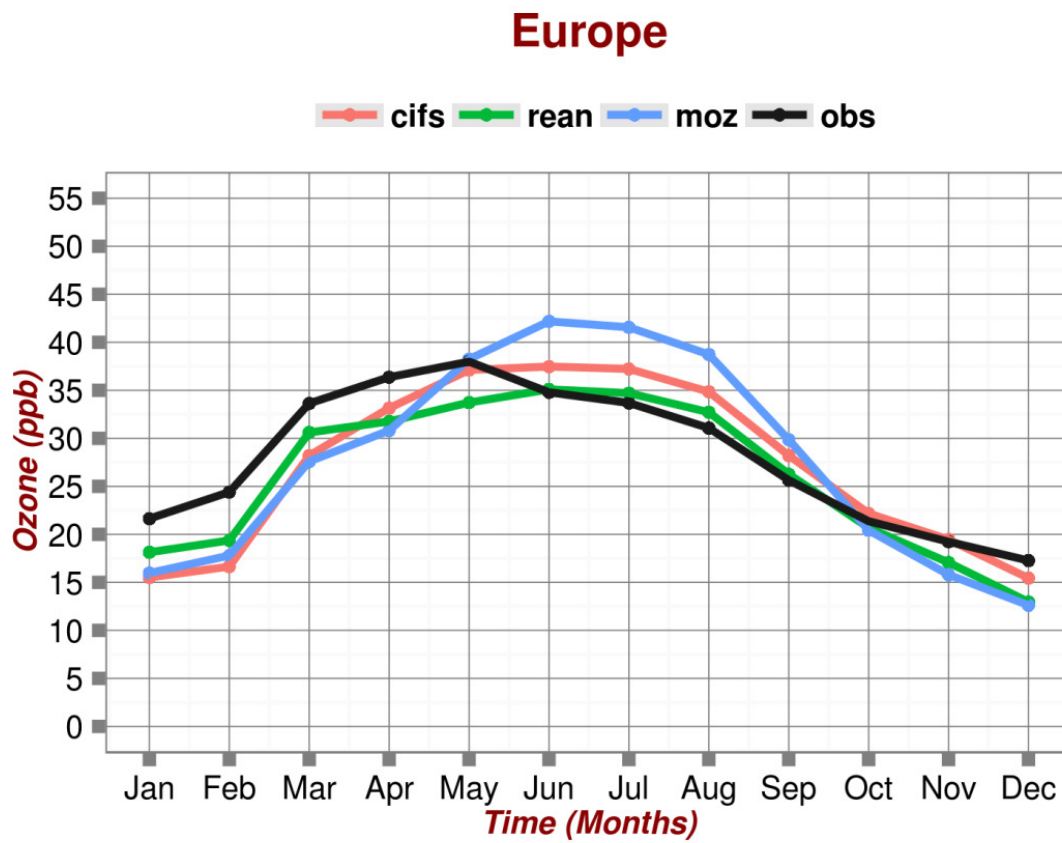


1542

1543 Figure 3 Tropospheric ozone volume mixing ratios (ppb) over the Tropics (left) Antarctica
 1544 (middle) and Arctic (right) averaged in the pressure bands 1000-700 hPa (bottom), 700-400
 1545 hPa (middle) and 400-200 hPa (top) observed by ozonesondes and simulated by C-IFS (red),
 1546 MOZ (blue) and REAN (green) in 2008.

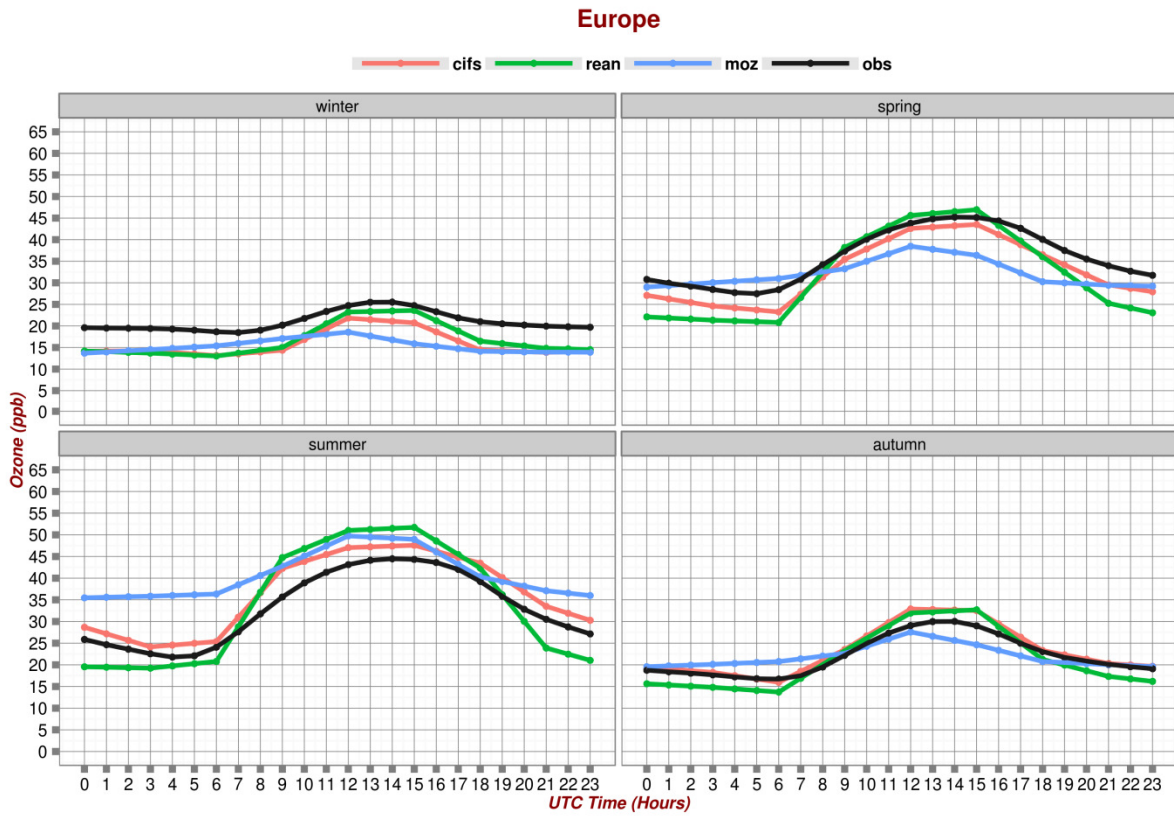
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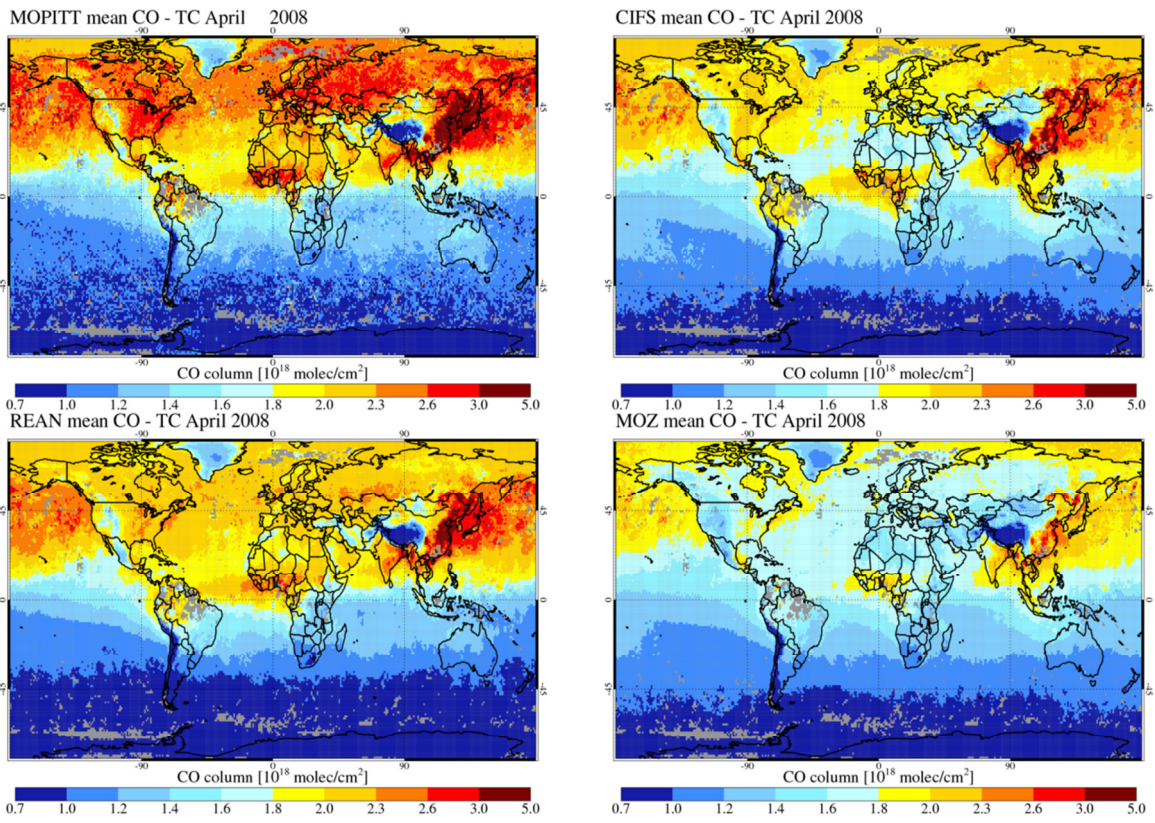
1550 Figure 4 Annual cycle of the mean ozone volume mixing ratios (ppb) at rural sites of the
1551 EMEP and AirBase data base and simulated by C-IFS (red), MOZ (blue) and REAN (green).



1552

1553 Figure 5 Diurnal cycle of surface ozone volume mixing ratios (ppb) over Europe in winter
 1554 (top, left), spring (top, right), summer (bottom, left) and autumn (bottom, right) at rural site of
 1555 the EMEP and AirBase data base and simulated by C-IFS (red), MOZ (blue) and REAN
 1556 (green).

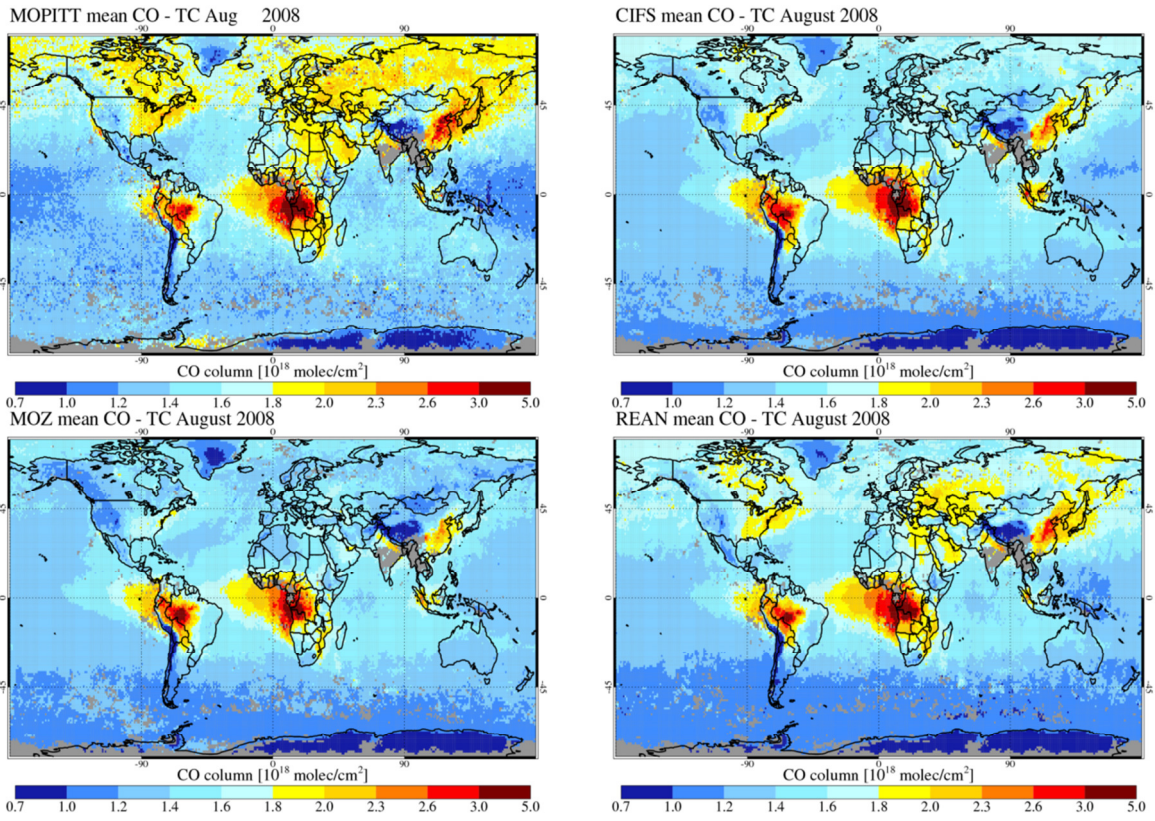
1557



1558

1559 Figure 6 CO total column retrieval (MOPITT V6) for April 2008 (top left) and simulated by
 1560 C-IFS (top right), MOZ (bottom left) and REAN (bottom right), AK are applied.

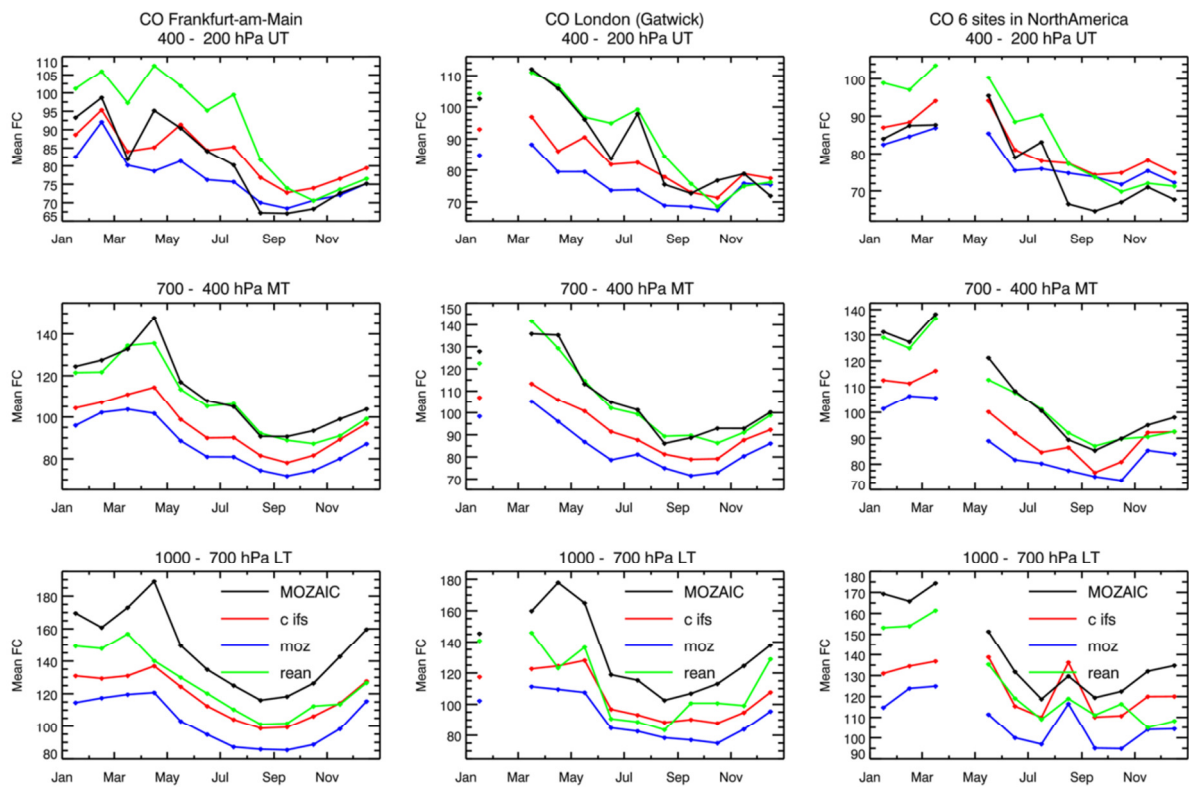
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1563 Figure 7 CO total column retrieval (MOPITT V6) for August 2008 (top left) and simulated by
 1564 C-IFS (top right), MOZ (bottom right) and REAN (bottom left), AK are applied.

1565

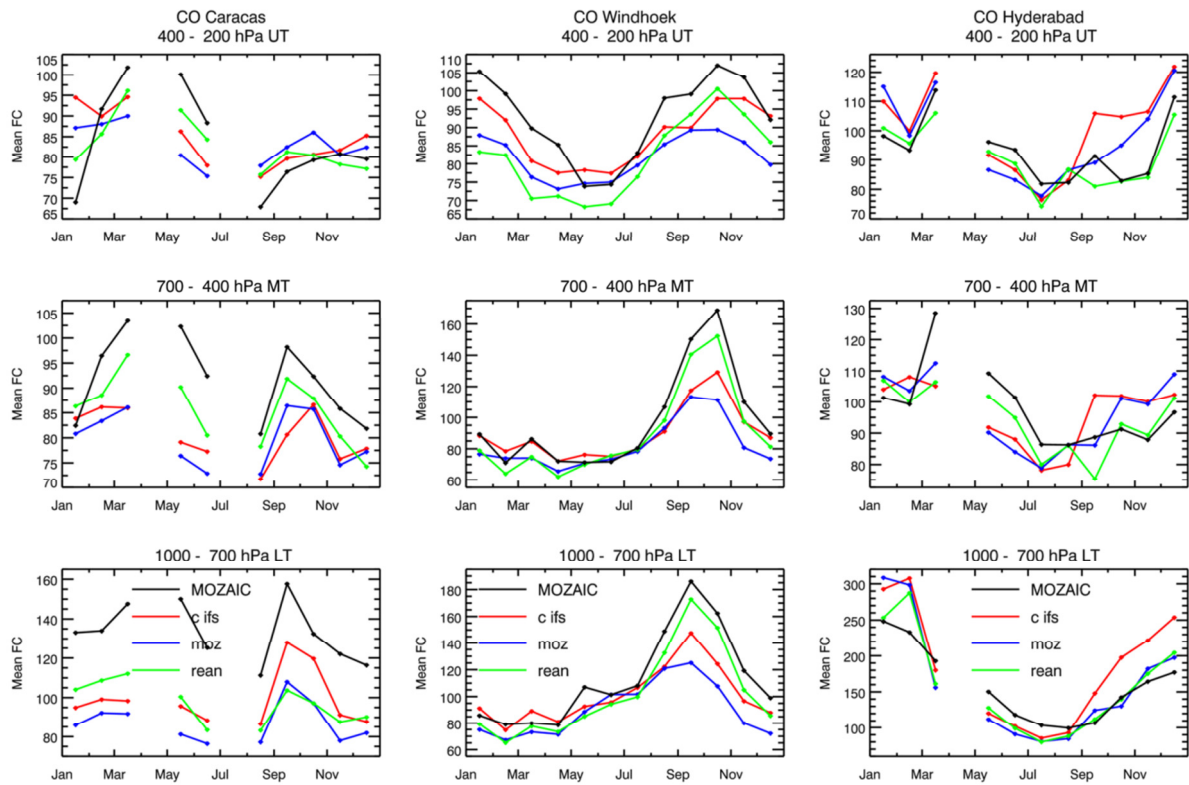


1566

1567 Figure 8 CO volume mixing ratios (ppb) over Frankfurt (left), London (middle) and North
 1568 America (left, averaged over 6 airports) averaged in the pressure bands 1000-700 hPa
 1569 (bottom), 700-400 hPa (middle) and 400-200 hPa (top) observed by MOZAIC and simulated
 1570 by C-IFS (red), MOZ (blue) and REAN (green) in 2008.

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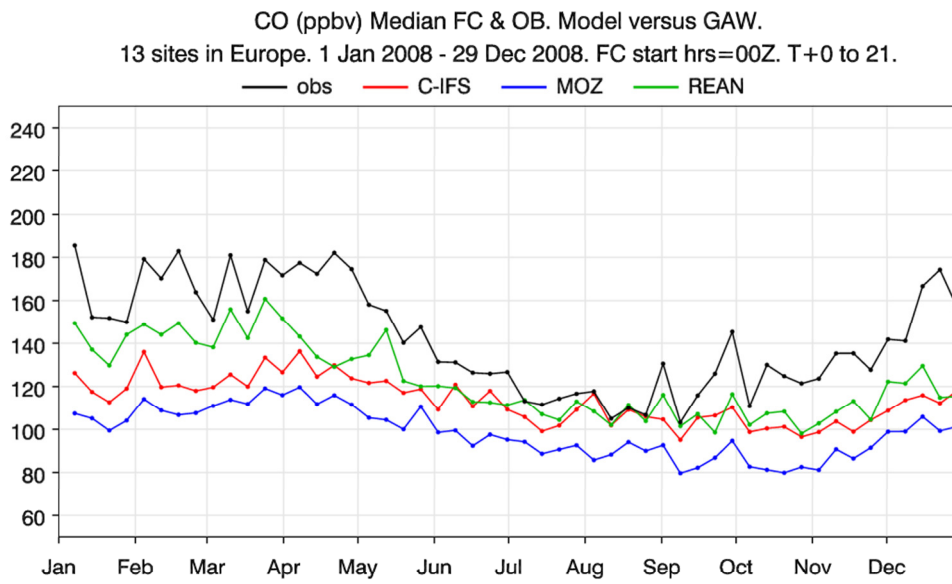


1573

1574 Figure 9 CO volume mixing ratios (ppb) over Caracas (left) Windhoek (middle) and
1575 Hyderabad (right) averaged in the pressure bands 1000-700 hPa (bottom), 700-400 hPa
1576 (middle) and 400-200 hPa (top) observed by MOZAIC, and simulated by C-IFS (red), MOZ
1577 (blue) and REAN (green) in 2008.

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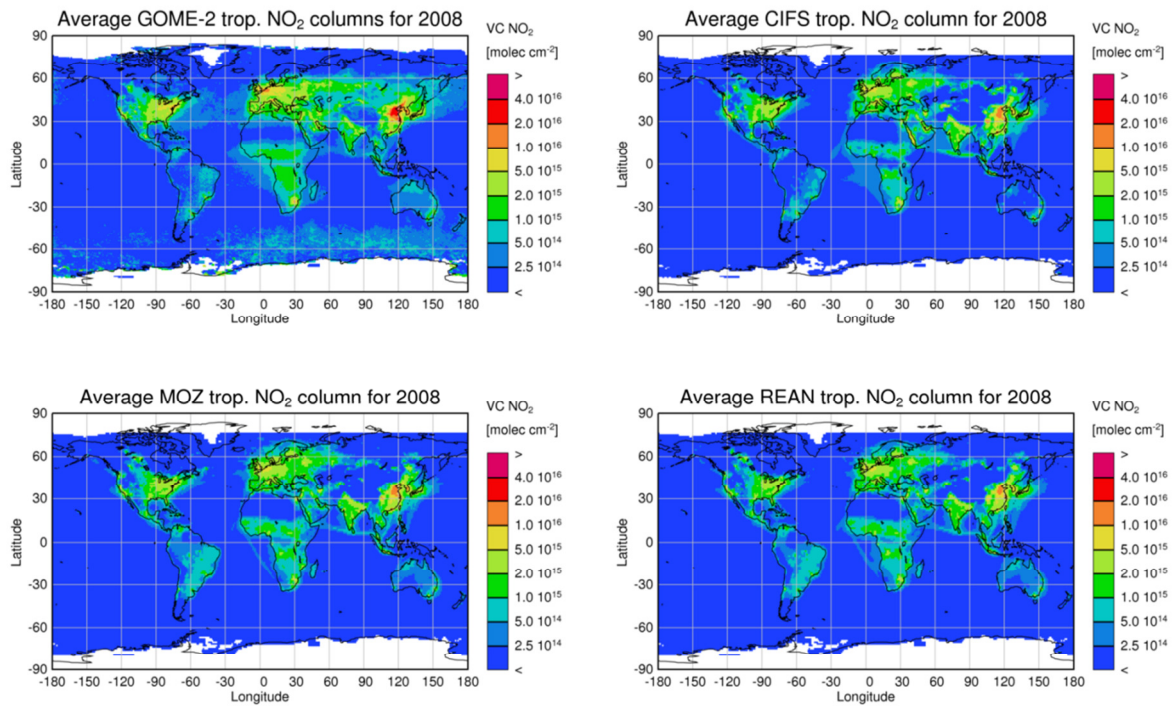
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1581 Figure 10 Time series of median of weekly CO surface volume mixing ratios (ppb) in Europe
1582 (13 GAW sites) and model results of C-IFS, MOZ and REAN.

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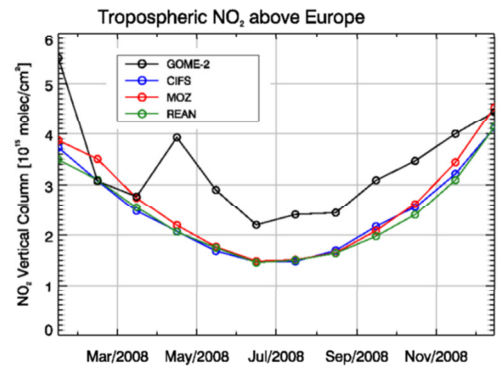
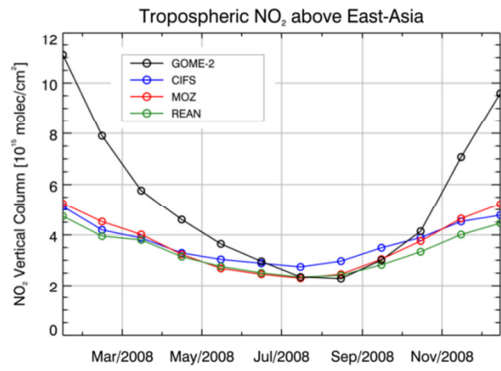
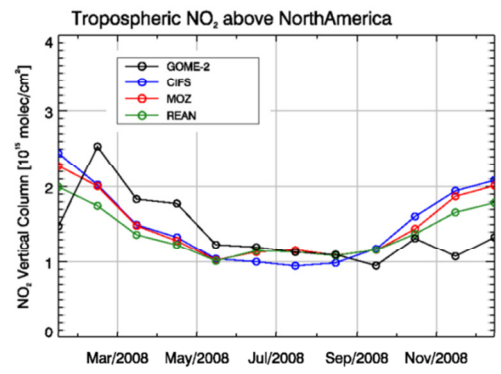
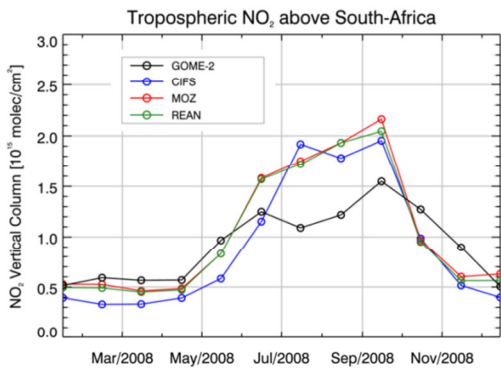


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1586 Figure 11 NO₂ tropospheric column retrieval (GOME-2) for 2008 (top left) and by C-IFS (top
 1587 right), REAN (bottom right) and MOZ (bottom left)

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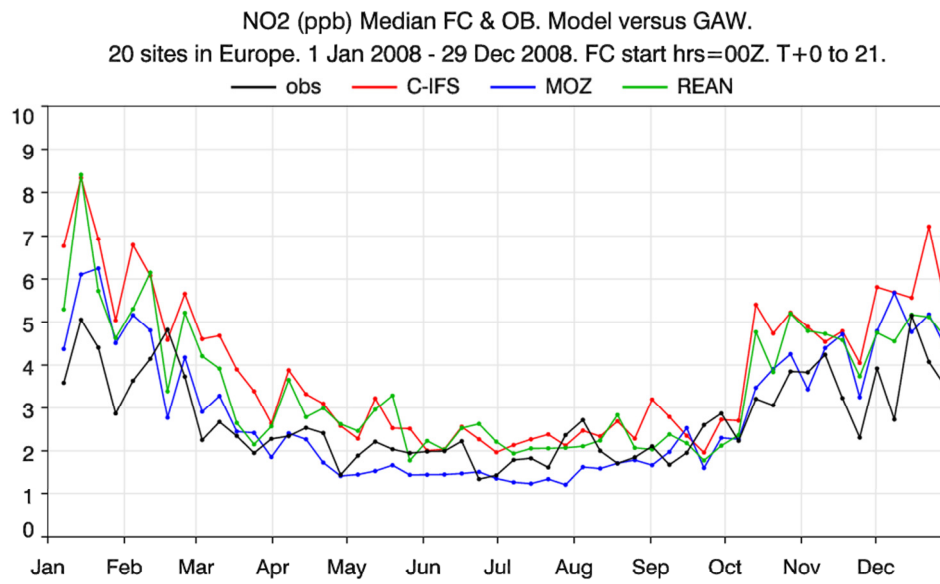
1589

1590 Figure 12 Time series of area-averaged tropospheric NO₂ columns [10¹⁵ molec cm⁻²] from
 1591 GOME-2 compared to model results of C-IFS (CB05) (blue), MOZ (red) and REAN (green)
 1592 for different regions.

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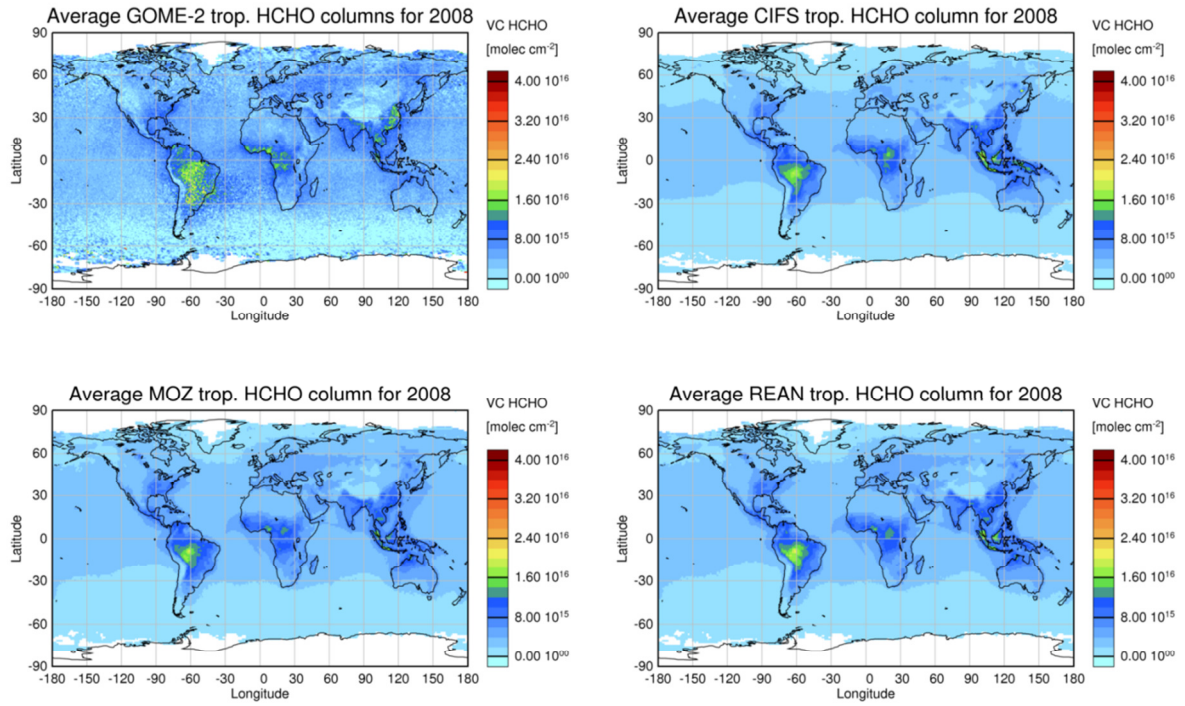
1596

1597 Figure 13 Time series of median of weekly surface NO₂ volume mixing ratios (ppb) in
1598 Europe (20 GAW sites) and model results of C-IFS, MOZ and REAN.

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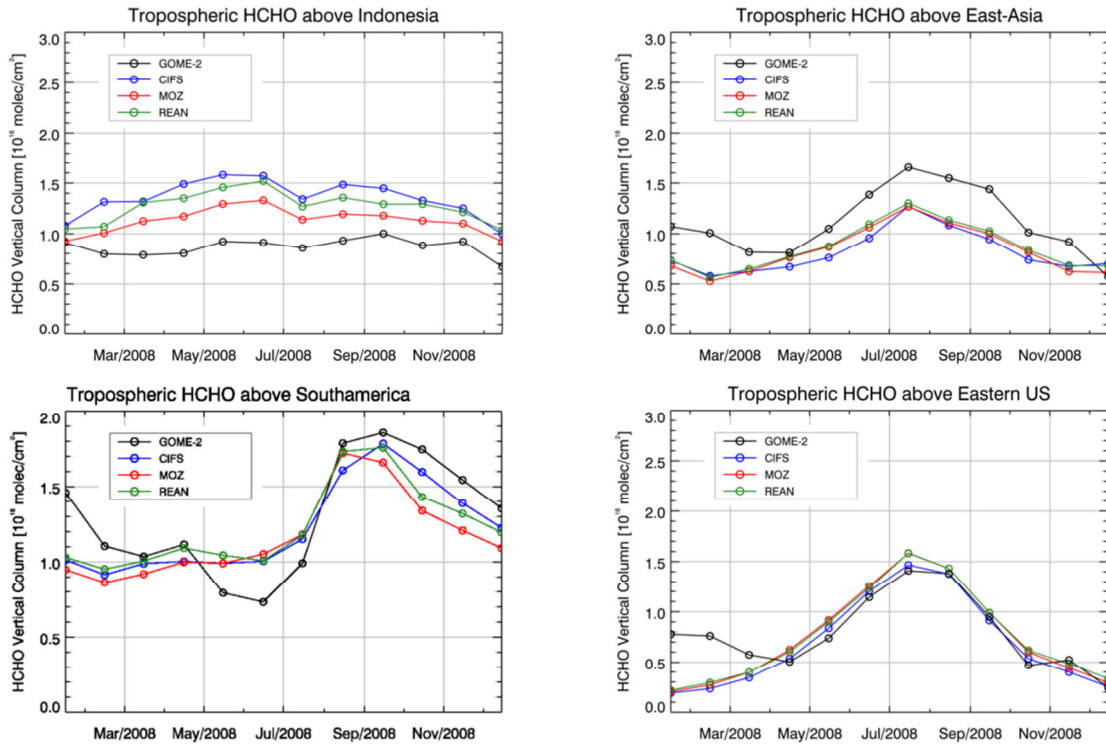
1602

1603 Figure 14 HCHO tropospheric column retrieval (GOME-2) for 2008 (top left) and by C-IFS

1604 (top right), REAN (bottom right) and MOZ (bottom left)

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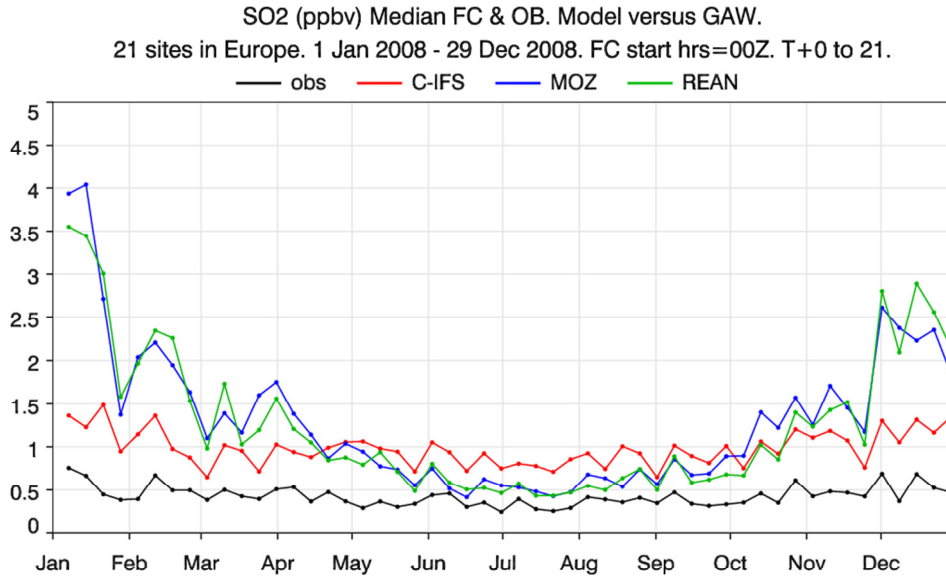
1609 Figure 15 Time series of area-averaged tropospheric HCHO columns [10^{16} molec cm⁻²] from
1610 GOME-2 compared to model results of C-IFS, MOZ and REAN for different regions.

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Figure 16 Time series of median of weekly surface SO₂ volume mixing ratios (ppb) in Europe (21 GAW sites) and model results of C-IFS, MOZ and REAN