1	Tropospheric Chemistry in the Integrated Forecasting						
2	System of ECMWF						
3 4 5 6	 Blechschmidt⁴, B. Josse³, M. Diamantakis¹, R. J. Engelen¹, A. Gaudel⁵, A. Inness¹, L. Jones¹, E. Katragkou⁶, V. Marecal³, VH. Peuch¹, A. Richter⁴, N 						
7	[1] European Centre for Medium-Range Weather Forecasts, Reading, UK						
8	[2] Royal Netherlands Meteorological Institute, De Belt, The Netherlands						
9	[3] Météo-France, Toulouse, France						
10	[4] Universität Bremen, Germany						
11	[5] CNRS, Laboratoire d'Aérologie, UMR 5560, Toulouse, France						
12 13	[6] Department of Meteorology and Climatology, School of Geology, Aristotle University of Thessaloniki, Greece						
14	[7] Institute for Energy and Climate Research, Forschungszentrum Jülich, Germany						
15							
16							
17 18	Correspondence to: J. Flemming (Johannes.Flemming@ecmwf.int)						
19							
20							
21							
22							
23							
24							

1

25 Abstract

26 A representation of atmospheric chemistry has been included in the Integrated Forecasting System (IFS) of the European Centre for Medium-range Weather Forecasts (ECMWF). The 27 28 new chemistry modules complement the aerosol modules of the IFS for atmospheric 29 composition, which is named C-IFS. C-IFS for chemistry supersedes a coupled system, in 30 which the Chemical Transport Model (CTM) Model for OZone and Related chemical Tracers 31 3 was two-way coupled to the IFS (IFS-MOZART). This paper contains a description of the 32 new on-line implementation, an evaluation with observations and a comparison of the 33 performance of C-IFS with MOZART and with a re-analysis of atmospheric composition produced by IFS-MOZART within the Monitoring Atmospheric Composition and Climate 34 (MACC) project. The chemical mechanism of C-IFS is an extended version of the Carbon 35 36 Bond 2005 (CB05) chemical mechanism as implemented in the CTM Transport Model 5 37 (TM5). CB05 describes tropospheric chemistry with 54 species and 126 reactions. Wet 38 deposition and lightning nitrogen monoxide (NO) emissions are modelled in C-IFS using the 39 detailed input of the IFS physics package. A one-year simulation by C-IFS, MOZART and the 40 MACC re-analysis is evaluated against ozonesondes, carbon monoxide (CO) aircraft profiles, 41 European surface observations of ozone (O_3) , CO, sulphur dioxide (SO_2) and nitrogen dioxide (NO_2) as well as satellite retrievals of CO, tropospheric NO₂ and formaldehyde. 42 43 Anthropogenic emissions from the MACC/CityZen (MACCity) inventory and biomass 44 burning emissions from the Global Fire Assimilation System (GFAS) data set were used in 45 the simulations by both C-IFS and MOZART. C-IFS (CB05) showed an improved performance with respect to MOZART for CO, upper tropospheric O₃, winter time SO₂ and 46 47 was of a similar accuracy for other evaluated species. C-IFS (CB05) is about ten times more 48 computationally efficient than IFS-MOZART.

49 **1** Introduction

50 Monitoring and forecasting of global atmospheric composition are key objectives of the 51 atmosphere service of the European Copernicus Programme. The Copernicus Atmosphere 52 Monitoring Service (CAMS) is based on combining satellite observations of atmospheric 53 composition with state-of-the-art atmospheric modelling (Flemming et al., 2013 and 54 Hollingsworth et al., 2008). For that purpose, the integrated forecasting system (IFS) of the 55 European Centre for Medium-Range Weather Forecasts (ECMWF) was extended for forecast 56 and assimilation of atmospheric composition. Modules for aerosols (Morcrette et al., 2009, 57 Benedetti et al., 2009) and greenhouse gases (Engelen et al., 2009) were integrated on-line in 58 the IFS. Because of the complexity of the chemical mechanisms for reactive gases, modules 59 for atmospheric chemistry were not initially included in the IFS. Instead a coupled system 60 (Flemming et al., 2009a) was developed, which couples the IFS to the Chemical Transport 61 Model (CTM) Model for OZone and Related chemical Tracers 3 (MOZART, Kinnison et al., 2007) or Transport Model 5 (TM5, Huijnen et al., 2010) by means of the Ocean Atmosphere 62 63 Sea Ice Soil coupling software (OASIS4) coupler software (Redler et al., 2010). Van Noije et al. (2014) coupled TM5 to IFS for climate applications in a similar approach. The coupled 64 65 system made it possible to assimilate satellite retrievals of reactive gases with the assimilation 66 algorithm of the IFS, which is also used for the assimilation of meteorological observations as 67 well as for aerosol and greenhouse gases.

68 The coupled system IFS-MOZART has been successfully used for a re-analysis of 69 atmospheric composition (Inness et al., 2013), pre-operational atmospheric composition 70 forecasts (Stein et al., 2012), forecast and assimilation of the stratospheric ozone (O_3) 71 (Flemming et al., 2011a, Lefever et al., 2014) and tropospheric carbon monoxide (CO) 72 (Eligundi et al., 2010) and O₃ (Ordonez et al., 2010). The coupled system IFS-TM5 has been 73 used in a case study on a period with intense biomass burning in Russia in 2010 (Huijnen et 74 al., 2012). Nevertheless, the coupled approach has limitations such as the need for interpolation between the IFS and CTM model grids and the duplicate simulation of transport 75 processes. Further, its computational performance is often not optimal as it can suffer from 76 77 load imbalances between the coupled components.

78 Consequently, modules for atmospheric chemistry and related physical processes have now 79 been integrated on-line in the IFS, thereby complementing the on-line integration strategy 80 already pursued for aerosol and greenhouse gases in IFS. The IFS including modules for 81 atmospheric composition is named Composition-IFS (C-IFS). C-IFS makes it possible (i) to 82 use the detailed meteorological simulation of the IFS for the simulation of the fate of 83 constituents (ii) to use the IFS data assimilation system to assimilate observations of 84 atmospheric composition and (iii) to simulate feedback processes between atmospheric composition and weather. A further advantage of C-IFS is the possibility of model runs at a 85 high horizontal and vertical resolution because of the high computational efficiency of C-IFS. 86 87 C-IFS is the global model system run in pre-operational mode as part of the Monitoring

Atmospheric Composition and Climate - Interim Implementation project (MACC II and
MACC III) in peraration of CAMS.

90 Including chemistry modules in general circulation models (GCM) to simulate interaction of 91 stratospheric O₃ (e.g. Steil et al., 1998) and aerosols (e.g. Haywood et al., 1997) in the climate system started in the mid-1990s. Later, more comprehensive schemes for tropospheric 92 93 chemistry were included in climate GCM such as ECHAM5-HAMMOZ (Pozzoli et al., 2008; 94 Rast et al., 2014) and CAM-chem (Lamarque et al., 2012) to study short-lived greenhouse 95 gases and the influence of climate change on air pollution (e.g. Fiore et al., 2010). In the UK 96 Met Office's Unified Model (UM) stratospheric chemistry (Morgenstern et al., 2009) and 97 tropospheric chemistry (O'Connor et al., 2014) can be simulated together with the GLOMAP mode aerosol scheme (Mann et al., 2010). Examples of the on-line integration of chemistry 98 99 modules in global circulation models with focus on NWP are GEM-AQ (Kaminski et al., 100 2008), GEMS-BACH (Menard et al., 2007) and GU-WRF/Chem (Zhang et al., 2012). 101 Savage et al. (2013) evaluate the performance of air quality forecast with the UM at the 102 regional scale. Baklanov et al. (2014) give a comprehensive overview of on-line coupled 103 chemistry-meteorological models for regional applications.

104 C-IFS is intended to run with several chemistry schemes for both the troposphere and the 105 stratosphere in the future. Currently, only the tropospheric chemical mechanism CB05 106 originating from the TM5 CTM (Huijnen et al., 2010) has been thoroughly tested. For 107 example, C-IFS (CB05) has been applied to study the HO₂ uptake on clouds and aerosols 108 (Huijnen et al., 2014) and pollution in the Artic (Emmons et al., 2014). The tropospheric and 109 stratospheric scheme RACMOBUS of the MOCAGE model (Bousserez et al., 2007) and the 110 MOZART 3 chemical scheme as well as an extension of the CB05 scheme with the 111 stratospheric chemical mechanism of the BASCOE model (Errera et al., 2008) have been 112 technically implemented and are being scientifically tested. Only C-IFS (CB05) is the subject 113 of this paper.

Each chemistry scheme in C-IFS consists of the specific gas phase chemical mechanism, multi-phase chemistry, the calculation of photolysis rates and upper chemical boundary conditions. Dry and wet deposition, emission injection and parameterization of lightning NO emissions as well as transport and diffusion are simulated by the same approach for all chemistry schemes. Likewise, emissions and dry deposition input data are kept the same for all configurations. 120 The purpose of this paper is to document C-IFS and to present its model performance with 121 respect to observations. Since C-IFS (CB05) replaced the current operational MACC model 122 system for reactive gases (IFS-MOZART) both in data assimilation and forecast mode, the 123 evaluation in this paper is carried out predominately with observations that are used for the 124 routine evaluation of the MACC II system. The model results are compared (i) with a MOZART stand-alone simulation, which is equivalent to a IFS-MOZART simulation and (ii) 125 126 with the MACC re-analysis (Inness et al., 2013), which is an application of IFS-MOZART in 127 data assimilation mode. All model configurations used the same emission data. The 128 comparison demonstrates that C-IFS is ready to be used operationally.

The paper is structured as follows. Section 2 is a description of the C-IFS, with focus on the newly implemented physical parameterizations and the chemical mechanism CB05. Section 3 contains the evaluation with observations of a one year simulation with C-IFS (CB05) and a comparison with the results from the MOZART run and the MACC re-analysis. The paper is concluded with a summary and an outlook in section 4.

134 2 Description of C-IFS

135 **2.1** Overview of C-IFS

The IFS consists of a spectral NWP model that applies the semi-Lagrangian (SL) semi-136 137 implicit method to solve the governing dynamical equations. The simulation of the hydrological cycle includes prognostic representations of cloud fraction, cloud liquid water, 138 139 cloud ice, rain and snow (Forbes et al., 2011). The simulations presented in this paper used 140 the IFS release CY40r1. The technical and scientific documentation of this IFS release can be 141 found at http://www.ecmwf.int/research/ifsdocs/CY40r1/index.html. Changes of the 142 operational model documented are on 143 https://software.ecmwf.int/wiki/display/IFS/Operational+changes.

At the start of the time step, the three-dimensional advection of the tracers mass mixing ratios is simulated by the SL method as described in Temperton et al. (2001) and Hortal (2002). Next, the tracers are vertically distributed by the diffusion scheme (Beljaars et al., 1998) and by convective mass fluxes (Bechtold et al., 2014). The diffusion scheme also simulates the injection of emissions and the loss by dry deposition (see section 2.4.1). The output of the convection scheme is used to calculate NO production by lightning (see section 2.4.3). Finally, the sink and source terms due to chemical conversion (see section 2.5), wet deposition (see section 2.4.2) and prescribed surface and stratospheric boundary conditionsare calculated (see section 2.5.2).

153 The chemical species and the related processes are represented only in grid-point space. The 154 horizontal grid is a reduced Gaussian grid (Hortal and Simmons, 1991). C-IFS can be run at varying vertical and horizontal resolutions. The simulations presented in this paper were 155 156 carried out at a T255 spectral resolution (i.e. truncation at wavenumber 255), which 157 corresponds to a grid box size of about 80 km. The vertical discretization uses 60 levels up to 158 the model top at 0.1 hPa (65 km) in a hybrid sigma-pressure coordinate. The vertical extent of 159 the lowest level is about 17 m; it is 100 m at about 300m above ground, 400-600 m in the middle troposphere and about 800 m at about 10 km height. 160

161 The modus operandi of C-IFS is one of a forecast model in a NWP framework. The 162 simulations of C-IFS are a sequence of daily forecasts over a period of several days. Each forecast is initialised by the ECMWF's operational analysis for the meteorological fields and 163 by the 3D chemistry fields from the previous forecast ("forecast mode"). Continuous 164 simulations over longer periods are carried out in "relaxation mode". In relaxation mode the 165 166 meteorological fields are relaxed to the fields of a meteorological re-analysis, such as ERA-167 Interim, during the run (Jung et al., 2008) to ensure realistic and consistent meteorological 168 fields.

169 **2.2 Transport**

170 The transport by advection, convection and turbulent diffusion of the chemical tracers uses 171 the same algorithms as developed for the transport of water vapour in the NWP applications of IFS. The advection is simulated with a three-dimensional semi-Lagrangian advection 172 173 scheme, which applies a quasi-montonic cubic interpolation of the departure values. Since the semi-Lagrangian advection does not formally conserve mass a global mass fixer is applied. 174 175 The effect of different global mass fixers is discussed in Diamantakis and Flemming (2014) 176 and Flemming and Huijnen (2011b). A proportional mass was used for the runs presented in this paper because of the overall best balance between the results and computational cost. 177

The vertical turbulent transport in the boundary layer is represented by a first order Kdiffusion closure. The surface emissions are injected as lower boundary flux in the diffusion scheme. The lower boundary flux condition also accounts for the dry deposition flux based on the projected surface mass mixing ratio in an implicit way. The vertical transport by 182 convection is simulated as part of the cumulus convection. It applies a bulk mass flux scheme 183 which was originally described in Tiedtke (1989). The scheme considers deep, shallow and 184 mid-level convection. Clouds are represented by a single pair of entraining/detraining plumes 185 which determine the updraught and downdraught mass fluxes. 186 (http://old.ecmwf.int/research/ifsdocs/CY40r1/ in Physical Processes, Chapter 6, pp 73-90). Highly soluble species such as nitr/ic acid (HNO₃), hydrogen peroxide (H₂O₂) and aerosol 187 188 precursors are assumed to be scavenged in the convective rain droplets and are therefore 189 excluded from the convective mass transfer.

The operator splitting between the transport and the sink and source terms follows the implementation for water vapour (Beljaars et al., 2004). Advection, diffusion and convection are simulated sequentially. The sink and source processes are simulated in parallel using an intermediate update of the mass mixing ratios with all transport tendencies. At the end of the time step tendencies from transport and sink and source terms are added together for the final update the concentration fields. Resulting negative mass mixing ratios are corrected at this point by setting the updated mass mixing ratio to a "chemical zero" of 1.0e-25 kg/kg.

197 **2.3 Emissions for 2008**

198 The anthropogenic surface emissions were given by the MACCity inventory (Granier et al., 199 2011) and aircraft NO emissions of a total of ~0.8 Tg N/yr were applied (Lamarque et al, 200 2010). Natural emissions from soils and oceans were taken from the Precursors of Ozone and 201 their Effects in the Troposphere (POET) database for 2000 (Granier et al., 2005; Olivier et al., 202 2003). The biogenic emissions were simulated off-line by the MEGAN2.1 model (Guenther et 203 al., 2006). The anthropogenic and natural emissions were used as monthly means. Daily 204 Biomass burning emissions were produced by the Global Fire Assimilation System (GFAS) 205 version 1, which is based on satellite retrievals of fire radiative power (Kaiser et al., 2012). 206 The actual emission totals used in the T255 simulation for 2008 from anthropogenic, biogenic 207 sources and biomass burning as well as lighting NO are given in Table 1.

208 **2.4 Physical parameterizations of sources and sinks**

209 2.4.1 Dry deposition

Dry deposition is an important removal mechanism at the surface in the absence of precipitation. It depends on the diffusion close to the earth surface, the properties of the constituent and on the characteristics of the surface, in particular the type and state of the vegetation and the presence of intercepted rain water. Dry deposition plays an important role in the biogeochemical cycles of nitrogen and sulphur, and it is a major loss process of tropospheric O₃. Modelling the dry deposition fluxes in C-IFS is based on a resistance model (Wesely et al., 1989), which differentiates the aerodynamic, the quasi-laminar and the canopy or surface resistance. The inverse of the total resistance is equivalent to a dry deposition velocity V_D .

The dry deposition flux F_D at the model surface is calculated based on the dry deposition velocity V_D , the mass mixing ratio X_s and air density ρ_s at the lowest model level *s*, in the following way:

$$F_D = V_D X_S \rho_S$$

The calculation of the loss by dry deposition has to account for the implicit character of the dry deposition flux since it depends on the mass mixing ratio X_s .

224 The dry deposition velocities were calculated as monthly mean values from a one-year 225 simulation using the approach described in Michou et al. (2004). It used meteorological and 226 surface input data such as wind speed, temperature, surface roughness and soil wetness from 227 the ERA-interim data set. At the surface the scheme makes a distinction between uptake 228 resistances for vegetation, bare soil, water, snow and ice. The surface and vegetation 229 resistances for the different species are calculated using the stomatal resistance of water 230 vapour. The stomatal resistance for water vapour is calculated depending on the leaf area 231 index, radiation and the soil wetness at the uppermost surface layer. Together with the 232 cuticlular and mesophyllic resistances this is combined into the leaf resistance according to 233 Wesely et al. (1989) using season and surface type specific parameters as referenced in 234 Seinfeld and Pandis (1998).

Dry deposition velocities have higher values during the day because of lower aerodynamic resistance and canopy resistance. Zhang et al. (2003) reported that averaged observed O_3 and sulphur dioxide (SO₂) dry deposition velocities can be up to 4 times higher at day time than at night time. As this important variation is not captured with the monthly-mean dry deposition values, a +/- 50% variation is imposed on all dry deposition values based on the cosine of the solar zenith angle. This modulation tends to decrease dry deposition for species with a night time maximum at the lowest model level and it increases dry deposition of O_3 . Table A4 (supplement) contains annual total loss by dry deposition and expressed as a lifetime estimate by dividing by tropospheric burden for a simulation using monthly dry deposition values for 2008. Dry deposition was most effective for many species in particular SO₂ and ammonia (NH₃) as the respective lifetimes were one day to one week. For tropospheric O₃ the respective globally averaged time scale is about 3 months. Because dry deposition occurs mainly over ice-free land surfaces the corresponding time scale is at least three times shorter in these areas.

249 2.4.2 Wet Deposition

- Wet deposition is the transport and removal of soluble or scavenged constituents by precipitation. It includes the following processes:
- In-cloud scavenging and removal by rain and snow (rain out)
- Release by evaporation of rain and snow
- Below cloud scavenging by precipitation falling through without formation of
 precipitation (wash out)

256 It is important to take the sub-grid scale of cloud and precipitation-formation into account for 257 the simulation of wet deposition. The IFS cloud scheme provides information on the cloud 258 and the precipitation fraction for each grid box. It uses a random overlap assumption (Jakob 259 and Klein, 2000) to derive cloud and precipitation area fraction. The same method has been 260 used by Neu and Prather (2012), who demonstrated the importance of the overlap assumption 261 for the simulation of the wet deposition. The precipitation fluxes for the simulation of wet 262 removal in C-IFS were scaled to be valid over the precipitation fraction of the respective gridbox. The loss of tracer by rain-out and wash-out was limited to the area of the grid box 263 264 covered by precipitation. Likewise, the cloud water and ice content is scaled to the respective 265 cloud area fraction. If the sub-grid scale distribution was not considered in this way, wet 266 deposition was lower for highly soluble species such as HNO₃ because the species is only 267 removed from the cloudy or rainy grid box fraction. For species with low solubility the wet 268 deposition loss was slightly decreased because of the decrease in effective cloud and rain 269 water.

Even if wet deposition removes tracer mass only in the precipitation area, the mass mixing ratio representing the entire grid box is changed accordingly after each model time step. This is equivalent with the assumption that there is instantaneous mixing within the grid-box at the time scale of the model time step. As discussed in Huijnen et al. (2014), this assumption maylead to an overestimation of the simulated tracer loss.

The module for wet deposition in C-IFS is based on the Harvard wet deposition scheme (Jacob et al., 2000 and Liu et al., 2001). In contrast to Jacob et al. (2000), tracers scavenged in wet convective updrafts are not removed as part of the convection scheme. Nevertheless, the fraction of highly soluble tracers in cloud condensate is simulated to limit the amount of tracers lifted upwards as only the gas phase fraction is transported by the mass flux. The removal by convective precipitation is simulated in the same way as for large-scale precipitation in the wet deposition routine.

The input fields to the wet deposition routine are the following prognostic variables, calculated by the IFS cloud scheme (Forbes et al., 2011): total cloud and ice water content, grid-scale rain- and snow water content and cloud and grid-scale precipitation fraction as well as the derived fluxes for convective and grid-scale precipitation fluxes at the grid cell interfaces. For convective precipitation a precipitation fraction of 0.05 is assumed and the convective rain and snow water content is calculated assuming a droplet fall speed of 5 m/s.

Wash-out, evaporation and rain-out are calculated after each other for large-scale and convective precipitation. The amount of trace gas dissolved in cloud droplets is calculated using Henrys-law-equilibrium or assuming that 70% of aerosol precursors such as sulphate (SO₄), NH₃ and nitrate (NO₃) is dissolved in the droplet. The effective Henry coefficient for SO₂, which accounts for the dissociation of SO₂, is calculated following Seinfeld and Pandis (1998, p. 350). The other Henry's law coefficients are taken from the compilation by Sander (1999) (www.henrys-law.org, Table A1 in the supplement).

The loss by rain out is determined by the precipitation formation rate. The retention coefficient R, which accounts for the retention of dissolved gas in the liquid cloud condensate as it is converted to precipitation, is one for all species in warm clouds (T > 268 K). For mixed clouds (T < 268 K) R is 0.02 for all species but 1.0 for HNO₃ and 0.6 for H₂O₂ (von Blohn, 2011). In ice clouds only H₂O₂ (Lawrence and Crutzen, 1998) and HNO₃ are scavenged.

Partial evaporation of the precipitation fluxes leads to the release of 50% of the resolved tracer and 100% in the case of total evaporation (Jacobs et al., 2000). Wash-out is either mass-transfer or Henry-equilibrium limited. HNO₃, aerosol precursors and other highly soluble gases are washed out using a first order wash-out rate of 0.1 mm⁻¹ (Levine and

- Schwartz, 1982) to account for the mass transfer . For less soluble gases the resolved fraction
 in the rain water is calculated assuming Henry equilibrium in the evaporated precipitation.
- 307 Table A5 (supplement) contains total loss by wet deposition and expressed as time scale in 308 days based on the tropospheric burden. For aerosol precursors nitrate, sulphate and 309 ammonium, HNO_3 and H_2O_2 wet deposition is the most important loss process with respective 310 timescales of 2–4 days.

311 2.4.3 NO emissions from lightning

312 NO emissions from lightning are a considerable contribution to the global atmospheric NO_x 313 budget. Estimates of the global annual source vary between 2-8 TgN/yr (Schumann and 314 Huntrieser, 2007). 5 TgN/yr (10.7 TgNO/yr) is the most commonly assumed value for global 315 CTMs which is about 6-7 times the value of NO emissions from aircraft (Gauss et al., 2006) 316 or 17% of the total anthropogenic emissions. NO emissions from lightning play an important 317 role in the chemistry of the atmosphere because they are released in the rather clean air of the 318 free troposphere, where they can influence the O_3 budget and hence the OH-HO₂ partitioning 319 (DeCaria et al., 2005).

The parameterization of the lightning NO production in C-IFS consist of estimates of (i) the flash rate density, (ii) the flash energy release and (iii) the vertical emission profile for each model grid column. The estimate of the flash-rate density is based on parameters of the convection scheme. The C-IFS has two options to simulate the flash-rate densities using the following input parameters: (i) convective cloud height (Price and Rind, 1992) or (ii) convective precipitation (Meijer et al., 2001).

326 The parameterizations distinguish between land and ocean points by assuming about 5-10 327 times higher flash rates over land. Additional checks on cloud base height, cloud extent and temperature are implemented to select only clouds that are likely to generate lightning strokes. 328 The coefficients of the two parameterizations were derived from field studies and depend on 329 330 the model resolution. With the current implementation of C-IFS (T255L60), the global flash 331 rates were 26 and 43 flashes per seconds for the schemes by Price and Rind (1992) and Meijer 332 et al. (2001), respectively. It seemed therefore necessary to scale the coefficients to get a flash 333 rate in the range of the observed values of about 40-50 flashes per second derived from 334 observations of the Optical Transient Detector (OTD) and the Lightning Imaging Sensor





337 Figure 1 shows the annual flash rate density simulated by the two parameterisations together 338 with observations from the LIS/OTD data set. The two approaches show the main flash 339 activity in the tropics but there were differences in the distributions over land and sea. The 340 smaller land-sea differences of Meijer et al. (2001) agreed better with the observations. The 341 observed maximum over Central Africa was well reproduced by both parameterizations but 342 the schemes produce an exaggerated maximum over tropical South America. The lightning 343 activity over the United States was underestimated by both parameterisations. The 344 parameterization by Meijer et al. (2001) has been used for the C-IFS runs presented in this 345 paper.

Cloud to ground (CG) and cloud to cloud (CC) flashes are assumed to release a different amount of energy, which is proportional to the NO release. Price et al. (1997) suggest that the energy release of CG is 10 times higher. However, more recent studies suggest a similar value for CG and CC energy release based on aircraft observations and model studies (Ott et al., 2010), which is followed in C-IFS. In C-IFS, CG and CC fractions are calculated using the approach by Price and Rind (1993), which is based on a 4th order function of cloud height above freezing level.

The vertical distribution of the NO release is of importance for its impact on atmospheric chemistry. Many CTMs use the suggestion of Pickering et al. (1998) of a C-shape profile, which peaks at the surface and in the upper troposphere. Ott et al. (2010) suggest a "backward C-shape" profile which locates most of the emission in the middle of the troposphere. The vertical distribution can be simulated by C-IFS (i) according to Ott et al. (2010) or (ii) as a Cshape profile following Huijnen et al. (2010). The approach by Ott et al. (2010) is used in the

- simulation presented here. As lightning NO emissions occur mostly in situations with strongconvective transport, differences in the injection profile had little impact.
- As the lightning emissions depend on the convective activity they change at different
 resolutions or after changes to the convection scheme. The C-IFS lightning emissions were
 4.9 TgN/yr at T159 resolution and 5.7 Tg N/yr at T255 resolution.
- 364 **2.5 CB05 chemistry scheme**

365 2.5.1 Gas-phase chemistry

366 The chemical mechanism is a modified version of the Carbon Bond mechanism 5 (CB05, 367 Yarwood et al., 2005), which is originally based on the work of Gery et al. (1989) with added 368 reactions from Zaveri and Peters (1999) and from Houweling et al. (1998) for isoprene. The 369 CB05 scheme adopts a lumping approach for organic species by defining a separate tracer 370 species for specific types of functional groups. The speciation of the explicit species into 371 lumped species follows the recommendations given in Yarwood et al. (2005). The CB05 372 scheme used in C-IFS has been further extended in the following way: An explicit treatment 373 of methanol (CH₃OH), ethane (C₂H₆), propane (C₃H₈), propene (C₃H₆) and acetone 374 (CH₃COCH₃) has been introduced as described in Williams et al., (2013). The isoprene 375 oxidation has been modified motivated by Archibald et al. (2010). Higher C3 peroxy-radicals 376 formed during the oxidation of C_3H_6 and C_3H_8 were included following Emmons et al. 377 (2010).

The CB05 scheme is supplemented with chemical reactions for the oxidation of SO_2 , dimethyl sulphide (DMS), methyl sulphonic acid (MSA) and NH₃, as outlined in Huijnen et al. (2014). For the oxidation of DMS, the approach of Chin et al. (1996) is adopted. Table A1 (supplement) gives a comprehensive list of the trace gases included in the chemical scheme.

382 The reaction rates have been updated according to the recommendations given in either 383 Sander et al. (2011) or Atkinson et al. (2004, 2006). The oxidation of CO by the hydroxyl 384 radical (OH) implicitly accounts for the formation and subsequent decomposition of the 385 intermediate species HOCO as outlined in Sander et al. (2006). For lumped species, e.g. 386 ALD2, the reaction rate is determined by an average of the rates of reaction for the most 387 abundant species, e.g. C2 and C3 aldehydes, in that group. An overview of all gas-phase 388 reactions and reaction rates as applied in this version of C-IFS can be found in Table A2 389 (supplement).

390 For the loss of trace gases by heterogeneous oxidation processes, the model explicitly 391 accounts for the oxidation of SO₂ in cloud through aqueous phase reactions with H₂O₂ and O₃, 392 depending on the acidity of the solution. The pH is computed from the SO4, MSA, HNO3, 393 NO3 A, NH3 and NH4 concentrations, as well as from a climatological CO2 value. The pH, 394 in combination with the Henry coefficient, defines the factor of sulphate residing in the 395 aqueous phase, compared to the gas phase concentration (Dentener and Crutzen, 1993) The 396 heterogeneous conversion of N₂O₅ into HNO₃ on cloud droplets and aerosol particles is 397 applied with a reaction probability (γ) set to 0.02 (Evans and Jacob, 2005). The surface area 398 density is computed based on a climatological aerosol size distribution function, applied to the 399 SO₄, MSA and NO₃_A aerosol, as well as to clouds assuming a droplet size of 8 µm.

400 2.5.2 Photolysis rates

401 For the calculation of photo-dissociation rates an on-line parameterization for the derivation 402 of actinic fluxes is used (Williams et al., 2012, 2006). It applies a Modified Band Approach 403 (MBA) which is an updated version of the work by Landgraf and Crutzen (1998), tailored and 404 optimized for use in tropospheric CTMs. The approach uses 7 absorption bands across the 405 spectral range 202 – 695 nm. At instances of large solar zenith angles (71-85°) a different set 406 of band intervals is used. In the MBA the radiative transfer calculation using the absorption 407 and scattering components introduced by gases, aerosols and clouds is computed on-line for 408 each of 7 pre-defined band intervals based on the 2-stream solver of Zdunkowski et al. 409 (1980).

The optical depth of clouds is calculated based on a parameterization available in IFS (Slingo, 1989 and Fu et al., 1996) for the cloud optical thickness at 550 nm. For the simulation of the impact of aerosols on the photolysis rates a climatological field for aerosols is used, as detailed in Williams et al. (2012). There is also an option to use the MACC aerosol fields.

In total 20 photolysis rates are included in the scheme, as given in Table A3 (supplement).
The explicit nature of the MBA implies a good flexibility in terms of updating molecular
absorption properties (cross sections and quantum yields) and the addition of new photolysis
rates into the model.

418 2.5.3 The chemical solver

419 The chemical solver used in C-IFS (CB05) is an Euler Backward Iterative (EBI) solver 420 (Hertel et al., 1996). This solver has been originally designed for use with the CBM4 421 mechanism of Gery et al. (1989). The chemical time step is 22.5 min, which is half of the 422 dynamical model time step of 45 min at T255 resolution. Eight, four or one iterations are 423 carried out for fast-, medium- and slow-reacting chemical species to obtain a solution. The 424 number of iterations is doubled in the lowest four models levels, where the perturbations due 425 to emissions can be large.

426 2.5.4 Stratospheric boundary conditions

The modified CB05 chemical mechanism includes no halogenated species and no photolytic destruction below 202 nm and is therefore not suited for the description of stratospheric chemistry. Thus realistic upper boundary conditions for the longer-lived gases such as O_3 , methane (CH₄), and HNO₃ are needed to capture the influence of stratospheric intrusions on the composition of the upper troposphere.

432 Stratospheric O_3 chemistry in C-IFS (CB05) is parameterized by the Cariolle scheme 433 (Cariolle and Teyssèdre, 2007). Chemical tendencies for stratospheric and tropospheric O_3 are 434 merged at an empirical interface of the diagnosed tropopause height in IFS. Additionally, 435 stratospheric O_3 in C-IFS can be nudged to O_3 analyses of either the MACC re-analysis 436 (Inness et al., 2013) or ERA interim (Dee et al., 2011). The tropopause height in IFS is 437 diagnosed either from the gradient in humidity or the vertical temperature gradient.

438 Stratospheric HNO₃ at 10 hPa is controlled by a climatology of HNO₃ and O₃ observations 439 from the Microwave Limb Sounder (MLS) aboard the Upper Atmosphere Research satellite (UARS). HNO₃ is set to according to the observed HNO₃ - O_3 ratio and the simulated O_3 440 441 concentrations. Further, stratospheric CH₄ is constrained by a climatology based on 442 observations of the Halogen Occultation Experiment instrument (Grooß and Russel, 2005), at 443 45hPa and at 90 hPa in the extra-tropics, which implicitly accounts for the stratospheric chemical loss of CH₄ by OH, chlorine (Cl) and oxygen (O¹D) radicals. It should be noted that 444 also the surface concentrations of CH₄ are fixed in this configuration of the model. 445

446 2.5.5 Gas-aerosol partitioning

447 Gas-aerosol partitioning is calculated using the Equilibrium Simplified Aerosol Model 448 (EQSAM, Metzger et al., 2002a, 2002b). The scheme has been simplified so that only the 449 partitioning between HNO₃ and the nitrate aerosol (NO⁻₃) and between NH₃ and the 450 ammonium aerosol (NH⁺₄) is calculated. SO²⁻₄ is assumed to remain completely in the 451 aerosol phase because of its very low vapour pressure. The assumptions of the equilibrium 452 model are that (i) aerosols are internally mixed and obey thermodynamic gas/aerosol 453 equilibrium and that (ii) the water activity of an aqueous aerosol particle is equal to the 454 ambient relative humidity (RH). Furthermore, the aerosol water mainly depends on the 455 aerosol mass and the type of the solute, so that parameterizations of single solute molalities 456 and activity coefficients can be defined, depending only on the type of the solute and RH. The advantage of using such parameterizations is that the entire aerosol equilibrium composition 457 458 can be solved analytically. For atmospheric aerosols in thermodynamic equilibrium with the 459 ambient RH, the following reactions are considered in C-IFS. The subscripts g, s and aq 460 denote gas, solid and aqueous phase, respectively:

- 461 $(NH_3)_g + (HNO_3)_g \leftrightarrow (NH_4NO_3)_s$
- 462 $(NH_4NO_3)_s + (H_2O)_g \leftrightarrow (NH_4NO_3)_{aq} + (H_2O)_{aq}$
- 463 $(NH_4NO_3)_{aq} + (H_2O)_g \leftrightarrow (NH_4^+)_{aq} + (NO_3^-)_{aq} + (H_2O)_{aq}$

464 **2.6**

6 Model budget diagnostics

465 C-IFS computes global diagnostics for every time step to study the contribution of different processes on the global budget. The basic outputs are the total and tropospheric tracer mass, 466 467 the global integral of the total surface emissions, integrated wet and dry deposition fluxes, 468 chemical conversion as well as elevated atmospheric emissions and the contributions of 469 prescribed upper and lower vertical boundary conditions for CH₄ and HNO₃. A time-invariant 470 pressure-based tropopause definition, which varies with latitude, is used to calculate the 471 tropospheric mass. To monitor the numerical integrity of the scheme, the contributions of the 472 corrections to ensure positiveness and global mass conservation are calculated. Optionally, 473 more detailed diagnostics can be requested that includes photolytic loss and the loss by OH 474 for the tropics and extra-tropics.

475 A detailed analysis of the global chemistry budget is beyond the scope of this paper. Only a 476 number of key terms for CO, O₃ and CH₄ is summarized here. They are compared with values 477 from the "Atmospheric Composition Change: the European Network of Excellence" 478 (ACCENT) model inter-comparisons of chemistry models by Stevenson et al. (2006) for 479 tropospheric O₃ and by Shindell et al. (2006) for CO. A more recent inter-comparison was 480 carried out within the Atmospheric Chemistry and Climate Model Intercomparison Project 481 (ACCMIP) (Lamarque et al., 2013). The ACCMIP values have been taken from Young et al. 482 (2013) for tropospheric O₃ and from Voulgarakis et al. (2013) for CH₄. It should be noted that

483 the values from these inter-comparison are valid for present-day conditions but not 484 specifically for 2008. A further source of the differences is the height of the tropopause 485 assumed in the calculations. Overall, the comparison showed that the C-IFS (CB05) is well 486 within the range of the multi model ensemble.

- 487 The annual mean of C-IFS tropospheric O_3 burden was 390 Tg. The values are at the upper
- 488 end of the range simulated by the ACCENT (344 \pm 39 Tg) and the ACCMIP (337 \pm 23 Tg)
- 489 models. The same holds for the loss by dry deposition, which was 1155 Tg/yr for C-IFS, 1003 490 ± 200 Tg/yr for ACCENT and in the range 687-1350 Tg/yr for ACCMIP. The tropospheric 491 chemical O₃ production of C-IFS was 4608 Tg/yr and loss 4144 Tg/yr, which is for both 492 values at the lower end of the range reported for the production $(5110 \pm 606 \text{ Tg/yr})$ and loss 493 $(4668 \pm 727 \text{ Tg/yr})$ for the ACCENT models. The comparatively simple treatment of volatile 494 organic compounds in CB05 could be an explanation for the low O_3 production and loss 495 terms. Stratospheric inflow in C-IFS, estimated as the residue from the remaining terms was 496 691 Tg and the corresponding value from the ACCENT multi-model mean is 552 ± 168 Tg.
- 497 The annual mean total CO burden in C-IFS was 361 Tg, which is slightly larger than the 498 ACCENT mean (345 Tg, 248-427 Tg). The total CO emissions in 2008 were 1008 Tg which 499 is in-line with the number used in ACCENT (1077 Tg/yr) but lower than the estimate (1550 500 Tg/yr) of the Third Assessment Report (Prather et al. 2001) of the Intergovernmental Panel 501 on Climate Change (IPCC), which also takes into account results from inverse modelling 502 studies. The tropospheric chemical CO production was 1434 Tg/yr, which is very close to the 503 ACCENT multi-mean of 1505 +/- 236 Tg/yr. The chemical CO loss in C-IFS was 2423 Tg 504 and the loss by dry deposition 24 Tg.
- 505 The annual mean CH₄ total and tropospheric burdens of C-IFS (CB05) are 4874 and 4271 506 Tg/yr, respectively. The global chemical CH₄ loss by OH was 467 Tg/yr. Following 507 Stevenson et al. (2006), this leads to a global CH₄ lifetime estimate of 9.1 yr. This value is within the ACCMIP range of 9.8±1.6 yr but lower than an observation-based 11.2±1.3 yr 508 509 estimate by Prather et al., 2012. CH₄ emissions were substituted by prescribed monthly zonal-510 mean surface concentrations to avoid the long-spin up needed by a direct modeling of the CH₄ 511 surface fluxes. The CH₄ surface concentrations were derived from a latitudinal interpolation 512 of observations from the stations South Pole, Cape Grim, Mauna Loa, Mace Head, Barrow 513 and Alert. The resulting CH₄ flux was 488 Tg/yr, which is of similar size as the sum of current estimates of the total CH₄ emissions of 500 - 580 Tg/yr and the loss by soils of 30-40 514

515	Tg/yr	(Forth	Assessment	Report	by	IPCC<

516 http://www.ipcc.ch/publications_and_data/ar4/wg1/en/ch7s7-4-1.html#ar4top).

517

5183Evaluation with observations and comparison with the coupled system519IFS-MOZART

520 The main motivation for the development of C-IFS is forecasting and assimilation of 521 atmospheric composition as part of the CAMS. Hence, the purpose of this evaluation is to 522 show how C-IFS (CB05) performs relative to the coupled CTM MOZART-3 (Kinnison et al., 523 2007), which has been running in the coupled system IFS-MOZART in pre-operational mode 524 since 2007. C-IFS will replace the coupled system in the next update of the CAMS system. 525 The evaluation focuses on species which are relevant to global air pollution such as 526 tropospheric O₃, CO, nitrogen doxide (NO₂), SO₂ and formaldehyde (HCHO). The MACC re-527 analysis (Inness et al., 2013), which is an application of IFS-MOZART with assimilation of 528 observations of atmospheric composition, has been included in the evaluation as a benchmark.

529 The MACC re-analysis (REAN) and the corresponding MOZART (MOZ) stand-alone run 530 have already been evaluated with observations by Inness et al. (2013). Further, the MACC-II 531 sub-project on validation has compiled a comprehensive report assessing REAN (MACC, 532 2013). REAN has been further evaluated with surface observations in Europe and North-533 America for O₃ by Im et al. (2014). C-IFS (CB05) has been already evaluated with a special 534 focus on hydroperoxyl (HO₂) in relation to CO in Huijnen et al. (2014). The performance of 535 an earlier version of C-IFS (CB05) in the Arctic was evaluated and inter-compared with 536 CTMs of the POLARCAT model intercomparison Project (POLMIP) by Monks et al. (2014) 537 for CO and Arnold et al. (2014) for reactive nitrogen. The POLMIP inter-comparisons show 538 that C-IFS (CB05) performs within the range of state-of-the-art CTMs.

539 **3.1** Summary of model runs setup

540 C-IFS (CB05) was run from 1 January to 31 December 2008 with a spin up starting 1 July 541 2007 at a T255 resolution (80 km x 80 km) with 60 model levels in monthly chunks. The 542 meteorological simulation was relaxed to dynamical fields of the MACC re-analysis (see 543 section 2.1). Likewise stratospheric O_3 above the tropopause was nudged to the MACC re-544 analysis.

MOZ is a run with the MOZART CTM at 1.1°×1.1° (120 x 120 km) horizontal resolution 545 using the 60 vertical levels of C-IFS. The setup of the MOZART model and the applied 546 547 emissions and dry deposition velocities were the same in MOZ and REAN. The most 548 important difference between MOZ and REAN is the assimilation of satellite retrieval of 549 atmospheric composition in REAN. Further, REAN was produced with the coupled system 550 IFS-MOZART whereas MOZ is a stand-alone driven by the meteorological fields of REAN. 551 The latter is equivalent to a simulation of IFS-MOZART without data assimilation of 552 atmospheric composition. The assimilated retrievals were CO and O₃ total columns, 553 stratospheric O₃ profiles and tropospheric NO₂ columns. No observations of atmospheric 554 composition have been feed in to the MOZ run. No observational information has been used 555 to improve the tropospheric simulation of the C-IFS run. Another difference between MOZ 556 and REAN is that the IFS diffusion and convection scheme, as used in C-IFS, controls the 557 vertical transport in REAN whereas MOZART's generic schemes were used in the MOZ run.

558 MOZ, REAN and C-IFS used the same anthropogenic emissions (MACCity), biogenic 559 (MEGAN 2.1 Guenther al., 2006, emissions et http://acd.ucar.edu/~guenther/MEGAN/MEGAN.htm) and natural emissions from the POET 560 561 project. The biomass burning emissions for MOZ and REAN came from the Global Fire 562 Emission Data version 3 inventory which was redistributed according to Fire Radiative Power 563 observations used in GFAS. Hence, the average biomass burning emissions used by 564 MOZART (MOZ and REAN) agree well with the GFAS emissions used by C-IFS, but they 565 are not identical in temporal and spatial variability.

566 **3.2 Observations**

567 The runs (C-IFS, MOZ, REAN) were evaluated with O₃ observations from ozonesondes and 568 O₃ and CO aircraft profiles from the Measurement of Ozone, Water Vapour, Carbon 569 Monoxide and Nitrogen Oxides by Airbus in-service Aircraft (MOZAIC) program. Simulated 570 surface O₃, CO, NO₂ and SO₂ fields were compared against Global Atmospheric Watch (GAW) surface observations and additionally O₃ against observations from the of the 571 572 European Monitoring and Evaluation Programme (EMEP) and the European air quality 573 database (AirBase). The global distributions of tropospheric NO₂ and HCHO were evaluated 574 with retrievals of tropospheric columns from Global Ozone Monitoring Experiment 2 575 (GOME-2). Measurements Of Pollution In The Troposphere (MOPITT) retrievals were used 576 for the validation of the global CO total column fields.

577 3.2.1 In-situ observations

578 The ozonesondes were obtained from the World Ozone and Ultraviolet Radiation Data Centre 579 (WOUDC) and from the ECWMF Meteorological Archive and Retrieval System. The observation error of the sondes is about $\pm 5\%$ in the range between 200 and 10 hPa and -7 -580 581 17% below 200 hPa (Beekmann et al., 1994, Komhyr et al., 1995 and Steinbrecht et al., 582 1996). The number of soundings varied for the different stations. Typically, the sondes are 583 launched once a week but in certain periods such as during O₃ hole conditions soundings are more frequent. Sonde launches were carried out mostly between 9 and 12 hours local time. 584 585 The global distribution of the launch sites is even enough to allow meaningful averages over larger areas such North-America, Europe, the Tropics, the Artic and Antarctica. Table 2 586 587 contains a list of the ozonesondes used in this study. Tilmes et al. (2012) suggest a further 588 refinement of the North-America region into Canada, Eastern and Western United States as 589 well of the Tropics into Atlantic/Africa, equatorial Americas and Eastern Indian 590 Ocean/Western Pacific based on the inter-comparison of ozone sonde observation for the 591 1994-2010 period. The results will be discussed also for the sub-regions and figures will be 592 presented in the supplement.

593 The MOZAIC program (Marenco et al., 1998 and Nédélec et al., 2003) provides profiles of 594 various trace gases taken during commercial aircraft ascents and descents at specific airports. 595 MOZAIC CO data have an accuracy of ± 5 ppby, a precision of $\pm 5\%$, and a detection limit of 596 10 ppbv (Nédélec et al., 2003). Since the aircraft carrying the MOZAIC unit were based in 597 Frankfurt, the majority of the CO profiles (837 in 2008) were observed at this airport. A 598 further 10 of the 28 airports with observations in 2008 had a sufficient number of profiles: 599 Windhoek (323), Caracas (129), Hyderabad (125) and London-Gatwick (83) as well as the 600 North-American airports Atlanta (104), Portland (69), Philadelphia (65), Vancouver (56), 601 Toronto (46) and Dallas (43). The North-American airports were considered to be close 602 enough to make a spatial average meaningful. Because of the varying data availability the 603 North-American mean is dominated by the airports in the Eastern United States.

Apart from Frankfurt, typically 2 profiles (takeoff and landing) are taken within 2-3 hours or with a longer gap in the case of an overnight stay. At Frankfurt there were 2-6 profiles available each day, mostly in the morning and the later afternoon to the evening. At the other airports the typical observation times were 6 & 18 UTC for Windhoek (+/- 0 h local time), 19 and 21 UTC for Hyderabad (+ 4 h local time), 20 and 22 UTC for Caracas (-6 h), 4 and 22 for London (+/- 0 h) and 19 and 22 (- 5/6 h) for the North American airports. This means that most of the observations were taken between the late evening and early morning hours, i.e. at a time of increased stability and large CO vertical gradients close to the surface. Only the observations at Caracas (afternoon) and to some extent in Frankfurt represent a more mixed day-time boundary layer. The modelled column profile was obtained at the middle between start and end time of the profile observation. The model columns were interpolated in time between two subsequent output time steps.

The global atmospheric watch (GAW) program of the World Meteorological Organization is 616 617 a network for mainly surface based observations (WMO, 2007). The data were retrieved from 618 the World Data Centre for Greenhouse Gases [http://ds.data.jma.go.jp/gmd/wdcgg/]. The 619 GAW observations represent the global background away from the main polluted areas. Often, the GAW observation sites are located on mountains, which makes it necessary to 620 621 select a model level different from the lowest model level for a sound comparison with the 622 model. In this study the procedure described in Flemming et al. (2009b) is applied to 623 determine the model level, which is based on the difference between a high resolution orography and the actual station height. The data coverage for CO and O₃ was global, 624 625 whereas for SO₂ and NO₂ only a few observations in Europe were available at the data 626 repository.

The Airbase and EMEP databases host operational air quality observations from different national European networks. All EMEP stations are located in rural areas, while Airbase stations are designed to monitor local pollution. Many AirBase observations may therefore not be representative of a global model with a horizontal resolution of 80 km. However, stations of rural regime may capture the larger scale signal in particular for O_3 , which is spatially well correlated (Flemming et al., 2005). The EMEP observations and the rural Airbase O_3 observations were used for the evaluation over Europe.

634 3.2.2 Satellite retrievals

635 Satellite retrievals of atmospheric composition are more widely used to evaluate model 636 results. Satellite data provide good horizontal coverage but have limitation with respect to the 637 vertical resolution and signal from the lowest atmospheric levels. Further, satellite 638 observations are only possible at the specific overpass time, and they can be disturbed by the 639 presence of clouds and surface properties. Depending on the instrument type global coverage 640 is achieved in several days. 641 Day-time CO total column retrievals from MOPITT, version 6 (Deeter et al., 2013b) and 642 retrievals of tropospheric columns of NO₂ (IUP-UB v0.7, Richter et al., 2005) and of HCHO 643 (IUP-UB v1.0; Wittrock et al., 2006) from GOME-2 (Callies et al., 2000) have been used for 644 the evaluation. The retrievals were averaged to monthly means values to reduce the random 645 retrieval error.

646 MOPITT is a multispectral thermal infrared (TIR) / near infrared (NIR) instrument onboard 647 the TERRA satellite with a pixel resolution of 22 km. TERRA's local equatorial crossing time is approximately 10:30 a.m. The MOPITT CO level 2 pixels were binned within $1x1^{\circ}$ within 648 each month. Deeter et al. (2013a) report a bias of about $+0.08e^{18}$ molec/cm² and a standard 649 deviation (SD) of the error of $0.19e^{18}$ molec/cm² for the TIR/NIR product version 5. This is 650 equivalent to a bias of about 4 % and a SD of 10% respectively assuming typical observations 651 of 2.0 e^{18} molec/cm². For the calculation of the simulated CO total column the a-priori profile 652 in combination with the averaging kernels (AK) of the retrievals were applied. They have the 653 654 largest values between 300 and 800 hPa. The AK have been applied to ensure that the 655 difference between retrieval and AK-weighted model column is independent of the a-priori CO profiles used in the retrieval.. One should note however, that the AK-weighted column is 656 657 not equivalent to the modelled atmospheric CO burden anymore..

658

659 GOME-2 is a ultra violet - visibile (UV-VIS) and NIR sensor designed to provide global observations of atmospheric trace gases. GOME-2 flies in a sun-synchronous orbit with an 660 661 equator crossing time of 09:30 LT in descending mode and has a footprint of 40 x 80 km. Here, tropospheric vertical columns of NO₂ and HCHO have been computed using a three 662 663 step approach. First, the Differential Optical Absorption Spectroscopy (DOAS; Platt, 1994) 664 method is applied to measured spectra which yields the total slant column. The DOAS 665 method is applied in a 425–497 nm wavelength window (Richter et al., 2011) for NO₂, and between 337 and 353 nm for HCHO (Vrekoussis et al., 2010). Second, the reference sector 666 667 approach is applied to total slant columns for stratospheric correction. In a last step, tropospheric slant columns are converted to tropospheric vertical columns by applying an air 668 669 mass factor. Only data with cloud fractions smaller than 0.2 according to the FRESCO cloud 670 data base (Wang et al., 2008) are used here. Furthermore, retrievals are limited to maximum solar zenith angles of 85° for NO2 and 60° for HCHO. Uncertainties in NO₂ satellite 671 retrievals are large and depend on the region and season. Winter values in mid and high 672

673 latitudes are usually associated with larger error margins. As a rough estimate, systematic 674 uncertainties in regions with significant pollution are of the order of 20% - 30%. As the 675 HCHO retrieval is performed in the UV part of the spectrum where less light is available and 676 the HCHO absorption signal is smaller than that of NO2, the uncertainty of monthly mean 677 HCHO columns is relatively large (20% - 40%) and both noise and systematic offsets have an 678 influence on the results. However, absolute values and seasonality are retrieved more 679 accurately over HCHO hotspots.

For comparison to GOME-2 data, model data are vertically integrated without applying AK to tropospheric vertical columns of NO_2 and HCHO, interpolated to satellite observation time and then sampled to match the location of available cloud free satellite data, which has been gridded to match the model resolution. The resulting daily files are then averaged over months for both satellite and model data to reduce the noise.

685 **3.3 Tropospheric Ozone**

690

Figure 2 shows the monthly means of O₃ volume mixing ratios in the pressure ranges surface
to 700 hPa (lower troposphere, LT) 700-400 hPa (middle troposphere, MT) and 400-200 hPa
(upper troposphere UT) observed by sondes and averaged over Europe, North America and
East



23

691 Figure 3 shows the same as Figure 2 for the Tropics, Arctic and Antarctica. A more detailed 692 breakdown of North America (Canada, Eastern and Western United States) and the Tropics 693 (Atlantic/Africa, equatorial Americas and Eastern Indian Ocean/Western Pacific) following 694 Times et al. (2012) is presented in the supplement. The observations have a pronounced 695 spring maximum for UT O₃ over Europe, North America and East Asia and a more gradually developing maximum in late spring and summer in MT and LT. The LT seasonal cycle is well 696 697 re-produced in all runs for the areas of the Northern Hemisphere (NH). In Europe, REAN 698 tends to overestimate by about 5 ppb where the C-IFS and MOZ have almost no bias before 699 the annual maximum in May apart from a small negative bias in spring. Later in the year, C-700 IFS tends to overestimate in autumn, whereas MOZ overestimates more in late summer. In 701 MT over Europe C-IFS agrees slightly better with the observations than MOZ. MOZ 702 overestimates in winter and spring and this overestimation is more prominent in the UT, 703 where MOZ is biased high throughout the year. This overestimation in UT is highest in 704 spring, where it can be 25% and more. These findings show that data assimilation in REAN 705 improved UT O₃ considerably but had only little influence in LT and MT. The overestimation 706 of MOZ in UT seems to be caused by increased stratospheric O₃ rather than a more efficient 707 transport as lower stratospheric O₃ was overestimated in MOZ. The good agreement of C-IFS 708 with observation in UT in all three regions is also present in a run without nudging to 709 stratospheric O₃. It is therefore not a consequence of the use of assimilated observations in C-710 IFS (CB05).

711 Over North-America the spring time underestimation by C-IFS and MOZ is more pronounced 712 than over Europe. The underestimation occurs in all regions but was largest in early spring 713 over Canada. C-IFS also underestimates spring ozone throughout North America in MT. LT 714 summer time ozone was overestimated in North America by all models, in particular over the 715 Eastern United States. The bias of C-IFS was the smallest in LT but in contrast to MOZ and 716 REAN C-IFS underestimates summer time ozone in MT over the Eastern United States. The 717 overestimation of UT ozone by MOZ was most pronounced in Canada.

718 In East Asia all runs overestimate by 5-10 ppb in LT and MT especially in autumn and winter.





719

Figure 3) the negative spring bias appears in all runs in LT and only for C-IFS in MT. As in
the other regions, MOZ greatly overestimates UT O₃.

723 Averaged over the tropics, the annual variability is below 10 ppb with maxima in May and in 724 September caused by the dry season in South-America (May) and Africa (September). The variability is well reproduced and biases are mostly below 5 ppb in the whole troposphere. 725 726 Note that the 400-200 hPa range (UT) in the tropics is less influenced by the stratosphere 727 because of the higher tropopause. C-IFS had smaller biases because of lower values in LT and higher values in MT and UT than MOZ. A more detailed analysis for different tropical 728 729 regions shows that the seasonality is well captured by all models over Atlantic-Africa, 730 equatorial America and eastern Indian Ocean/Western Pacific in all three tropospheric levels. 731 However, the strong observed monthly anomalies (a observation glitch ? by one station) in 732 equatorial America in March and September were underestimated by up to 20 ppb in all 733 tropospheric levels.

Over the Arctic C-IFS and MOZ reproduce the seasonal cycle, which peaks in late spring, but
 generally underestimate the observations in LT. C-IFS had a smaller bias in LT than MOZ but

- had a larger negative bias in MT. The biggest improvement of C-IFS w.r.t to MOZ occurred
- at the surface in Antarctica as the biases compared to the GAW surface observations were
- 738 greatly reduced. Notably, the assimilation (REAN) led to increased biases for LT and MT O₃,
- in particular during polar night when UV satellite observations are not available as already
- 740 discussed in Flemming et al. (2011a).
- The ability of the models to simulate O_3 near the surface is tested with rural AirBase and
- T42 EMEP stations (see section 3.2).



743

744 Figure 4 shows monthly means and Figure 5 the average diurnal cycle in different season in Europe. All runs underestimate monthly mean O₃ in spring and winter and overestimate it in 745 746 late summer and autumn. The overestimation in summer was largest in MOZ. The recently 747 reported (Val Martin at al. 2014) missing coupling of the leaf area index to the leaf and 748 stomatal vegetation resistance in the calculation of dry deposition velocities could be an 749 explanation of the MOZ bias. While the overestimation appeared also with respect to the 750 ozonesondes in LT (see Figure 2, left) the spring time underestimation was less pronounced in 751 LT.

- 752 The comparison of the diurnal cycle with observations (Figure 5) shows that C-IFS produced
- a more realistic diurnal cycle than the MOZART model. The diurnal variability simulated by
- the MOZART model is much less pronounced than the observations suggest. The diurnal
- 755 cycle of C-IFS and REAN were similar. This finding can be explained by the fact that C-IFS
- and REAN use the IFS diffusion scheme whereas MOZART applies the diffusion scheme of
- the MOZART CTM.
- 758 The negative bias of C-IFS in winter and spring seems mainly caused by an underestimation
- of the night time values whereas the overestimation of the summer and autumn average values
- 760 in C-IFS were caused by an overestimation of the day time values. However, the
- 761 overestimation of the summer night time values by MOZART seems to be a strong
- contribution to the average overestimation in this season.

763 **3.4 Carbon Monoxide**

764 The seasonality of CO is mainly driven by its chemical lifetime, which is lower in summer 765 because of increased photochemical activity. The seasonal cycle of the CO emissions plays 766 also an important role in the case of biomass burning and high anthropogenic emissions. The 767 global distribution of total column CO retrieved from MOPITT and from AK weighted columns simulated by C-IFS, MOZ and REAN is shown for April 2008 in Figure 6 and for 768 769 August in Figure 7. April and August have been selected because they are the months of the NH CO maximum and minimum. C-IFS reproduced well the location of the observed global 770 771 maxima in North-America, Europe and China as well as the biomass burning signal in Central 772 Africa. However, there was a widespread underestimation of the MOPITT values in the NH, 773 which was strongest over European Russia and Northern China. Tropical CO was slightly 774 overestimated but more strongly over Southeast Asia in April at the end of the biomass 775 burning season in this region. The lower CO columns in mid- and high latitudes in the 776 Southern Hemisphere (SH) were underestimated. The same global gradients of the bias were 777 found in MOZ and REAN. The negative NH bias in April of MOZ is however more pronounced but the positive bias in the tropics is slightly reduced. The bias of MOZ seems 778 779 stronger over the entire land surface in NH and not predominately in the areas with high emission. This is consistent with the finding of Stein et al. (2014) that dry deposition, besides 780 781 underestimated emissions, contributes to the large negative biases in NH in MOZ. 782 Assimilating MOPITT (V4) in REAN led to much reduced biases everywhere even though 783 the sign of bias in NH, Tropics and SH remained. In August, the NH bias is reduced but the

hemispheric pattern of the CO bias was similar as in April for all runs. The only regional exception from the general overestimation in the tropics is the strong underestimation of CO in the biomass burning maximum in Southern Africa, which points to an underestimation of the GFAS biomass burning emissions in that area.

788 More insight into the seasonal cycle and the vertical CO distribution can be obtained from 789 MOZAIC aircraft profiles. CO profiles at Frankfurt (Figure 8, left) provide a continuous 790 record with about 2 - 6 observations per day. As already reported in Inness et al. (2013) and 791 Stein et al. (2014), MOZ underestimates strongly LT CO with a negative bias of 40 - 60 ppb 792 throughout the whole year. The highest underestimation occurred in April and May, i.e. at the 793 time of the observed CO maximum. C-IFS CB05 also underestimates CO but with a smaller 794 negative bias in the range of 20-40 ppb even though it used the same CO emission data as 795 MOZ. REAN has the lowest bias throughout the year but the improvement is more important in winter and early spring. The comparison over London, which is representative for 4 and 22 796 797 UTC, leads to similar results as for Frankfurt (Figure 8, middle). The seasonal variability of 798 LT CO from MOZAIC and the model runs in North-America is very similar to the one in 799 Europe (Figure 8, right). The late winter and spring bias is slightly increased whereas the 800 summer time bias was lower for all models. The surface bias in winter and spring of MOZ, C-801 IFS and REAN is about -50, -40 and -20 ppb respectively. In the rest of the year REAN and 802 C-IFS have a bias of about -15 ppb whereas the bias of MOZ is about twice as large.

MT CO was very well produced by REAN in Europe and North-America probably because MOPITT has the highest sensitivity at this level. The MT bias of C-IFS is about 75% of the bias of MOZ, which underestimates by about 30 ppb. In the UT the CO biases are for all models mostly below 10ppb, i.e. about 10 %. C-IFS has overall the smallest CO bias whereas REAN tends to overestimate and MOZ to underestimate CO over Europe and North America.

808 CO observed by MOZAIC over Windhoek (Figure 9, middle) has a pronounced maximum in 809 September because of the seasonality of biomass burning in this region. Although all runs 810 show increased CO in this period, the models without assimilation were less able to reproduce the high observed CO values and are biased low up to 40 ppb in LT and MT. Biases were 811 812 much reduced, i.e. mostly within 10 ppb, during the rest of the year. The assimilation in 813 REAN greatly reduces the bias in the biomass burning period. In UT C-IFS had slightly 814 smaller biases of about 10 ppb than MOZ and REAN. A less complete record of the seasonal variability is available for Caracas (Figure 9, left). All models tend to underestimate UT and 815

MT CO maxima in April by about 20% but in contrast to Windhoek the C-IFS and not REAN has the smallest bias in LT. Hyderabad (Figure 9, right) is the only observation site were a substantial overestimation of CO in LT and UT is present even though the observations are in the range of 150 - 250 ppb, which is mostly higher than at any of the other airports discussed. All models overestimate the seasonality because of an underestimation in JJA and an overestimation during the rest of the year.

The outcome of the comparison with LT CO from MOZAIC is consistent with the model bias with respect to the GAW surface observations in Europe (Figure 10). The winter biases were larger than summer biases and MOZ showed the largest understimation. ,. The GAW stations measuring CO are mostly located on mountains in the Alpine region and typical annual biases were about - 5, -20 and -35 ppb for REAN, C-IFS and MOZ respectively. The negative biases of stations in flatter terrain such as Kollumerward tended to be larger.

828 **3.5** Nitrogen dioxide

829 The global maxima of NO₂ are located in areas of high anthropogenic and biomass burning 830 NO emissions. The global annual distribution of annual tropospheric columns retrieved from 831 the GOME-2 instrument and simulated by the models is shown in



832

833

834 Figure 11. C-IFS, MOZ and REAN showed a very similar distribution, which can be 835 explained by that fact that the same NO emission data were used in all runs. The global patterns of the modelled fields resemble the observed annual patterns to a large extent. But the 836 837 models tend to underestimate the high observed values in East-Asia and Europe and also 838 simulate too little NO₂ in larger areas of medium observed NO₂ levels in Asia and Central 839 Africa as well as in the outflow areas over the West-Atlantic and West Pacific Ocean. This 840 could mean that NO emissions in the most polluted areas are too low but also that the 841 simulated lifetime of NO₂ is too short. Further, an insufficient simulation of NO_x reservoir 842 species such as PAN and the lack of alkyl nitrates in CB05 might be the reason for the 843 underestimation.



Figure 12) for different regions and months shows that tropospheric NO₂ columns over Europe, North America, South Africa and East-Asia are reasonably reproduced. The models tend to underestimate tropospheric columns over Europe in summer (see Table 2 for area descriptions). However, the evaluation with GAW surface stations mainly from Central and Eastern Europe (



851

852 Figure 13) revealed an overestimation by all models in winter and a small overestimation in 853 summer for REAN and C-IFS. All runs significantly underestimate the annual cycle of the 854 GOME-2 NO₂ tropospheric columns over East-Asia. The winter time values are only half of 855 the observations whereas in summer models agree well with observations. In Southern Africa 856 (20°S/0°S/15°E/15°W), the models overestimate the increased NO₂ values in the biomass 857 burning season by a factor 2 but show good agreement with observations in the rest of the 858 year. The overestimation during biomass burning events could be related to the assumed NO 859 emission factor.

860 **3.6 HCHO**

861 On the global scale HCHO is mainly chemically produced by the oxidation of isoprene and 862 CH₄. Isoprene is emitted by vegetation. On the regional scale HCHO emissions from 863 anthropogenic sources, vegetation and biomass burning also contribute to the HCHO burden.

864 The annual average of tropospheric HCHO retrieved from GOME-2 and from the model runs

shown

865 is





Figure 14. The observations show higher values in the tropics and the NH and maxima in the rain forest regions of South America and Central Africa and in South East Asia. The simulated fields of the three runs are very similar. C-IFS, MOZ and REAN reproduce the observed global patterns but show a small but widespread underestimation in the NH extra-Tropics and in industrialized East Asia. On the other hand HCHO is overestimated in Indonesia.

873

in





874

Figure 15 shows model time series of tropospheric HCHO against corresponding GOME-2 876 877 satellite retrievals for selected regions. The models underestimated satellite values over East-878 summer and overestimate HCHO Asia especially in columns for Indonesia 879 (5°S/5°N/100°E/120°E) throughout the year. The seasonality in Southern Africa (not shown) 880 and tropical South America (10°S/5°S/73°W/35°W) is well captured in particular by C-IFS. 881 All models also reproduced the observations rather well for the Eastern United States 882 (30°N/40°N/90°W/75°W), but tend to underestimate wintertime HCHO columns for this 883 region.

884 **3.7 Sulfur dioxide**

885 SO₂ was evaluated with available GAW surface observations from Central and Eastern

886 Europe. There were considerable differences in the performance for individual stations often



888

Figure 16 shows the median of weekly observed and modelled time series. REAN and MOZ greatly exaggerated the seasonal cycle since the values in winter were up to eight times larger than the median of the observations. The summer values of the two runs were about 50% higher than the observations. C-IFS followed better the weak seasonality of the observations but suffered from a nearly constant bias of about 1 ppb (100%), which was much smaller than the bias of REAN and MOZ in winter but slightly higher in summer. Overall, the on-line integration of C-IFS showed lower SO₂ biases.

As no SO₂ observations were assimilated in REAN and identical SO₂ emission were used, the
differences between the runs were caused by differences in the simulation of vertical mixing,
sulphur chemistry and wet and dry deposition in C-IFS and MOZART. The winter time bias
of REAN and MOZ could be introduced by the diffusion scheme in MOZART.

900 **3.8 Computational cost**

901 The computational cost is an important factor for the operational applications in CAMS. The
902 computational cost of different configurations of IFS, C-IFS and IFS-MOZART are given in
903 Table 3. Computational cost is expressed in billing units (BU) of the ECMWF IBM Power 7

904 super-computer. BUs are proportional to the number of used Central Processing Unit (CPU)905 times the simulation time.

906 The increase of cost because of the simulation of the CB05 chemistry with respect to an NWP 907 run is about a factor 4 at the resolutions T159 (110km), T255 (80 km) and T511 (40 km). C-908 IFS (CB05) is about 8 times more efficient than the coupled system IFS-MOZART at a T159 909 resolution and about 15 times more at a T255 resolution. This strong relative increase in cost 910 of IFS-MOZART is caused by the increasing memory requirements of the IFS at higher 911 resolution, or also in data assimilation mode. However, there is insufficient parallelism in 912 MOZART to exploit the larger number of CPU for speeding up the simulation of the coupled 913 system.

C-IFS with the MOZART chemical mechanism, i.e. the same chemistry scheme as in IFSMOZART, is about 2 times and C-IFS with RACMOBUS 7 times more costly than C-IFS
(CB05) at a T159 resolution. Both the MOZART and the RACMOBUS schemes encompass a
larger number of species and reactions and include a full stratospheric chemistry scheme,
which is missing in CB05. The overhead because of the doubled number of advected species
in C-IFS RACMOBUS and MOZART is however small because of the efficency of the SL
advection scheme.

921 4 Summary and outlook

Modules for the simulation of atmospheric chemistry have been implemented on-line in the Integrated Forecasting System (IFS) of ECMWF. The chemistry scheme complements the already integrated modules for aerosol and greenhouse gases as part of the IFS for atmospheric composition (C-IFS). C-IFS for chemistry replaces the coupled system IFS-MOZART for forecast and assimilation of reactive gases within the pre-operational Copernicus Atmosphere Monitoring Service.

928 C-IFS applies the chemical mechanism CB05, which describes tropospheric chemistry with 929 55 species and 126 reactions. C-IFS benefits from the detailed cloud and precipitation physics 930 of the IFS for the calculation of wet deposition and lightning NO emission. Wet deposition 931 modelling is based on Jacob (2000) and accounts for the sub-grid scale distribution of clouds 932 and precipitation. Dry deposition is modelled using pre-calculated monthly-mean dry 933 deposition velocities following (Wesely, 1989) with a superimposed diurnal cycle. Surface 934 emissions and dry deposition fluxes are applied as surface boundary conditions of the diffusion

- scheme. Lightning emissions of NO can be calculated either by cloud height (Price and Rind,
 1993) or by convective precipitation (Meijer et al., 2001). The latter parameterization was
 used in this study. The anthropogenic emissions were taken from the MACCity inventory and
 biomass burning emissions from the GFAS data set for 2008.
- 939 An evaluation for the troposphere of a simulation in 2008 with C-IFS (CB05) and the 940 MOZART CTM (MOZ) as well as with the MACC re-analysis (REAN) was carried out. The 941 model results were compared against ozonesondes, MOZAIC CO aircraft profiles, European 942 surface observations of O₃, CO, SO₂ and NO₂ and global satellite retrievals of CO, NO₂ and 943 HCHO. The evaluation showed that C-IFS preforms better or with similar accuracy as 944 MOZART and mostly of similar quality as the MACC re-analysis. It should be noted that 945 satellite retrievals of CO, O₃ and NO₂ were assimilated in the MACC re-analysis to improve 946 the realism of the fields simulated by IFS-MOZART.
- 947 In comparison to MOZ, C-IFS (CB05) had smaller biases (i) for CO in the Northern 948 Hemisphere, (ii) for O_3 in the upper troposphere and (ii) for winter-time SO_2 at the surface in 949 Europe. Further, the diurnal cycle of surface O₃, tested with rural European Air quality 950 observations, showed greater realism in the C-IFS simulation. As both models used the same 951 emission data, the improvements can be explained by the differences in the chemical 952 mechanism and the simulation of wet and dry deposition. However, the improvements in SO₂ 953 and the diurnal cycle of O_3 are most probably caused by the more consistent interplay of 954 diffusion and sink and sources processes in the on-line integrated C-IFS.
- There is still room for improvement of C-IFS (CB05). It underestimated surface O_3 over Europe and North America in spring and overestimated it in late summer and autumn. CO was still underestimated by C-IFS in particular in Europe and North America throughout the year but more in spring and winter, and in the biomass burning season in Africa. Winter time tropospheric NO₂ over China as retrieved from the GOME-2 instrument was two times higher than the fields modelled by C-IFS, MOZART and the MACC re-analysis.
- Although only one chemical mechanism is described in the paper, C-IFS is a model that can apply multiple chemistry schemes. The implementation of the chemistry schemes of the CTMs MOCAGE and MOZART has technically been completed but further optimisation and evaluation is required. Both schemes offer a description of stratospheric chemistry, which is not included in the tropospheric scheme CB05. For this reason it is intended to combine the
966 CB05 mechanism with the BASCOE stratospheric mechanism. An inter-comparison of the967 performance of the different chemical mechanism is planned.

It is foreseen to further improve the link between the physics and chemistry packages in IFS. For example, the detailed information from the IFS surface scheme will be utilised for the calculation of dry deposition and biogenic emissions. A first important step is to replace the climatological dry deposition velocities with-online calculated values. Further, the impact of the simulated O_3 fields, once the stratospheric chemistry is fully implemented, on the IFS radiation scheme and the corresponding feedback on the temperature fields will be investigated.

Another ongoing development is to link more closely the greenhouse gas, aerosol and gasphase chemistry modules of C-IFS. Relevant chemical conversion terms can already be fed to
the GLOMAP aerosol (Mann et al, 2010) module for the simulation of secondary aerosols.
The calculation of photolysis rates can account for the presence of aerosols, and HO₂ uptake
on aerosols can be simulated (Huijnen et al., 2014).

980 In summary, C-IFS is a new global chemistry weather model for forecast and assimilation of 981 atmospheric composition. C-IFS (CB05) has already been successfully applied in data 982 assimilation mode and a paper on the subject is in preparation (Inness et al., 2014). C-IFS 983 offers improvements over the coupled system IFS-MOZART because (i) it simulates several 984 trace gas C-IFS (CB05)es with better accuracy, (ii) it is computational several times more 985 efficient in particular at high resolution and (iii) it better facilitates the implementation of 986 feedback processes between gas-phase and aerosol processes as well as between atmospheric 987 composition and meteorology.

988 Acknowledgments

989 MACC II is funded by the European Union's Seventh Framework Programme (FP7) under 990 Grant Agreement no. 283576. The MOPITT data were obtained from the NASA Langley 991 Research Atmospheric Science Data Center. We are grateful to the World Ozone and 992 Ultraviolet Radiation Data Centre (WOUDC) for providing ozonesonde obervations. We 993 thank the Global Atmospheric Watch programme for the provision of NO₂, CO and SO₂ 994 surface observations. We thank the European Environmental Agency for providing access to 995 European O₃ observations in the AirBase data base. We also thank the MOZAIC 996 (Measurements of OZone, water vapour, carbon monoxide and nitrogen oxides by in-service

- AIrbus aircraft) and IAGOS (In-Service Aircraft for a Global Observing System) programmes
- 998 for providing CO profile observations.

1000 **References**

- Archibald, A. T., Cooke, M. C., Utembe, S. R., Shallcross, D. E., Derwent, R. G., and Jenkin,
 M. E.: Impacts of mechanistic changes on HOx formation and recycling in the oxidation of
 isoprene, Atmos. Chem. Phys., 10, 8097-8118, doi:10.5194/acp-10-8097-2010, 2010.
- 1004 Arnold, S. R., Emmons, L. K., Monks, S. A., Law, K. S., Ridley, D. A., Turquety, S., Tilmes,
- 1005 S., Thomas, J. L., Bouarar, I., Flemming, J., Huijnen, V., Mao, J., Duncan, B. N., Steenrod,
- 1006 S., Yoshida, Y., Langner, J., and Long, Y.: Biomass burning influence on high latitude
- tropospheric ozone and reactive nitrogen in summer 2008: a multi-model analysis based on
 POLMIP simulations, Atmos. Chem. Phys. Discuss., 14, 24573-24621, doi:10.5194/acpd-1424573-2014, 2014.
- 1010 Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G.,
- 1011 Jenkin, M. E., Rossi, M. J. and Troe, J.: Evaluated kinetic and photochemical data for
- 1012 atmospheric chemistry: Volume I gas phase reactions of Ox, HOx, NOx and SOx, species,
- 1013 Atmos. Chem. Phys., 4, 1461–1738, doi:10.5194/acp-4-1461-2004, 2004.
- Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G.,
 Jenkin, M. E., Rossi, M. J., Troe, J., and IUPAC Subcommittee: Evaluated kinetic and
 photochemical data for atmospheric chemistry: Volume II gas phase reactions of organic
 species, Atmos. Chem. Phys., 6, 3625–4055, doi:10.5194/acp-6-3625-2006, 2006.
- 1018 Baklanov, A., Schlünzen, K., Suppan, P., Baldasano, J., Brunner, D., Aksoyoglu, S.,
- 1019 Carmichael, G., Douros, J., Flemming, J., Forkel, R., Galmarini, S., Gauss, M., Grell, G.,
- 1020 Hirtl, M., Joffre, S., Jorba, O., Kaas, E., Kaasik, M., Kallos, G., Kong, X., Korsholm, U.,
- 1021 Kurganskiy, A., Kushta, J., Lohmann, U., Mahura, A., Manders-Groot, A., Maurizi, A.,
- 1022 Moussiopoulos, N., Rao, S. T., Savage, N., Seigneur, C., Sokhi, R. S., Solazzo, E., Solomos,
- 1023 S., Sørensen, B., Tsegas, G., Vignati, E., Vogel, B., and Zhang, Y.: Online coupled regional
- 1024 meteorology chemistry models in Europe: current status and prospects, Atmos. Chem. Phys.,
 1025 14, 317-398, doi:10.5194/acp-14-317-2014, 2014.
- 1026
- Banda, N., Krol, M, van Noije, T., van Weele, M., Williams, J. E., Le Sager, P., Niemeier,
 U., Thomason, L. and Röckmann, T.: The effect of stratospheric sulfur from Mount Pinatubo
 on tropospheric oxidizing capacity and methane, J. Geophys. Res. Atmos., 119,
 doi:10.1002/2014JD022137, 2014.

- Barkley, M., Description of MEGAN biogenic VOC emissions in GEOS-Chem, 2010.
 http://acmg.seas.harvard.edu/geos/wiki_docs/emissions/megan.pdf
- 1035 Bechtold, P., Semane, N., Lopez, P., Chaboureau, J-P, Beljaars, A., Bormann, N: 2014:
- 1036 Representing Equilibrium and Nonequilibrium Convection in Large-Scale Models. J. Atmos.
- 1037 Sci., 71, 734–753. doi: http://dx.doi.org/10.1175/JAS-D-13-0163.1. 2014.
- Beekmann M., Ancellet G., Megie G., Smit H. G. J., and Kley D.: Intercomparison campaign
 for vertical ozone profiles including electrochemical sondes of ECC and Brewer-Mast type
 and aground based UV-differential absorption radar, J. Atmos. Chem., 10, 259–288, 1994.
- Beljaars, A. and Viterbo, P.: The role of the boundary layer in a numerical weather prediction
 model, in: Clear and cloudy boundary layers, A.A.M. Holtslag and P. Duynkerke (eds.),
 Royal Netherlands Academy of Arts and Sciences, p. 287-304, Amsterdam, North Holland
 Publishers, 1998.
- 1045 Beljaars, A., Bechtold, P., Kohler, M., Morcrette, J-J., Tompkins, A., Viterbo, P. and Wedi, 1046 N.: The numerics of physical parameterization, Seminar on Recent developments in methods modelling, 1047 numerical for atmospheric and ocean 6-10 September, 1048 http://www.ecmwf.int/publications/library/do/references/, 2004.
- 1049 Benedetti, A., Morcrette, J.-J., Boucher, O., Dethof, A., Engelen, R. J., Fisher, M., Flentje, H.,
- 1050 Huneeus, N., Jones, L., Kaiser, J. W., Kinne, S., Mangold, A., Razinger, M., Simmons, A. J.,
- 1051 Suttie, M., and the GEMS-AER team: Aerosol analysis and forecast in the European Centre
- 1052 for Medium-Range Weather Forecasts Integrated Forecast System: 2. Data assimilation, J.
- 1053 Geophys. Res., 114, D13205, doi:10.1029/2008JD011115, 2009.
- 1054 Bousserez, N., Attié, J.-L., Peuch, V.-H., Michou, M., and Pfister, G.: Evaluation of the
- 1055 MOCAGE chemistry and transport model during the ICARTT/ITOP experiment, J. Geophys.
- 1056 Res., 112, D10S42, doi:10.1029/2006JD007595, 2007.
- 1057 Callies, J., Corpacicioli, E., Eisinger, M., Hahne, A., and Lefebvre, A.: GOME-2 Metops's
- 1058 Second Generation Sensor for Operational Ozone Monitoring, ESA Bulletin, 102, 2000.

- 1059 Cariolle, D. and Deque, M.:. Southern hemisphere medium-scale waves and total ozone
 1060 disturbances in a spectral general circulation model, J. Geophys. Res., 91D, 10825–10846,
 1061 1986.
- Cariolle, D. and Teyssèdre, H.: A revised linear ozone photochemistry parameterization for
 use in transport and general circulation models: multi-annual simulations, Atmos. Chem.
 Phys., 7, 2183-2196, doi:10.5194/acp-7-2183-2007, 2007.
- 1065 Carslaw, K. S., Luo, B., Peter, T., and Clegg, S. L.: Vapour pressures of
 1066 H2SO4/HNO3/HBr/H2O solutions to low stratospheric temperatures, Geophys. Res. Lett., 22,
 1067 247-250, 1995.
- 1068 Cecil, D.J., Buechler, D. E., Blakeslee, R. J. : Gridded lightning climatology from TRMM1069 LIS and OTD: Dataset description, Atmospheric Research, 135–136, 404-414,
 1070 doi:10.1016/j.atmosres.2012.06.028, 2012.
- 1071 Chin, M., D. J. Jacob, G. M. Gardner, M. S. Foreman-Fowler, P. A. Spiro, and D. L. Savoie:
 1072 A global three-dimensional model of tropospheric sulfate, J. Geophys. Res., 101, (D13),
 1073 18,667–18,690, 1996
- 1074 DeCaria, A. J., Pickering, K. E., Stenchikov, G. L. and Ott, L. E.: Lightning-generated NO_X 1075 and its impact on tropospheric ozone production: A three-dimensional modeling study of a 1076 Stratosphere-Troposphere Experiment: Radiation, Aerosols and Ozone (STERAO-A) 1077 thunderstorm, J. Geophys. Res., 110, D14303, doi:10.1029/2004JD005556, 2005.
- 1078 Dee, D.P., Uppala, S.M., Simmons, A.J., Berrisford, P., Poli, P., Kobayashi, S., Andrae, U.,
- 1079 Balmaseda, M.A., Balsamo, G., Bauer, P., Bechtold, P., Beljaars, A.C.M., van de Berg, L.,
- 1080 Bidlot, J., Bormann, N., Delsol, C., Dragani, R., Fuentes, M., Geer, A.J., Haimberger, L.,
- 1081 Healy, S.B., Hersbach, H., Hólm, E.V., Isaksen, L., Kållberg, P., Köhler, M., Matricardi, M.,
- 1082 Mcnally, A.P., Monge-Sanz, B.M., Morcrette, J.-J., Park, B.-K., Peubey, C., de Rosnay, P.,
- Tavolato, C., Thépaut, J.-N., Vitart, F.: The ERA-Interim reanalysis: Configuration and
 performance of the data assimilation system, Quarterly Journal of the Royal Meteorological
 Society, 2011.
- 1086 Deeter, M. N., S. Martínez-Alonso, D. P. Edwards, L. K. Emmons, J. C. Gille, H. M. Worden,
- 1087 J. V. Pittman, B. C. Daube, and S. C. Wofsy: Validation of MOPITT Version 5 thermal-
- 1088 infrared, near-infrared, and multispectral carbon monoxide profile retrievals for 2000–2011, J.
- 1089 Geophys. Res. Atmos., 118, 6710–6725, doi:10.1002/jgrd.50272, 2013a.

- 1090 Deeter, M.N., MOPITT Version 6 Product User's Guide, Technical Report, NCAR, Boulder,
 1091 USA, 2013.137 (656), pp. 553-597, 2013b.
- 1092 Dentener, F. J. and Crutzen, P. J.: Reaction of N2O5 on tropospheric aerosols: Impact on the 1093 global distributions of NOx, O3 and OH, J. Geophys. Res., 98(D4), 7149–7163, 1993.
- Diamantakis, M. and Flemming, J.: Global mass fixer algorithms for conservative tracer
 transport in the ECMWF model, Geosci. Model Dev., 7, 965-979, doi:10.5194/gmd-7-9652014, 2014.
- 1097 Elguindi, N., Clark, H., Ordóñez, C., Thouret, V., Flemming, J., Stein, O., Huijnen, V.,
- 1098 Moinat, P., Inness, A., Peuch, V.-H., Stohl, A., Turquety, S., Athier, G., Cammas, J.-P., and
- 1099 Schultz, M.: Current status of the ability of the GEMS/MACC models to reproduce the
- 1100 tropospheric CO vertical distribution as measured by MOZAIC, Geosci. Model Dev., 3, 501-
- 1101 518, doi:10.5194/gmd-3-501-2010, 2010.
- Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D.,
 Granier, C., Guenther, A., Kinnison, D., Laepple, T., Orlando, J., Tie, X., Tyndall, G.,
- Wiedinmyer, C., Baughcum, S. L., and Kloster, S.: Description and evaluation of the Model
 for Ozone and Related chemical Tracers, version 4 (MOZART-4), Geosci. Model Dev., 3, 43-
- 1106 67, doi:10.5194/gmd-3-43-2010, 2010.
- 1107 Emmons, L.K., Arnold, S., Monks, S., Huijnen, V., Tilmes, S., Law, K., Thomas, J.L., Raut,
- 1108 J-C., Bouarar, I., Turquety, S., Long, Y., Duncan, B., Steenrod, S., Strode, S., Flemming, J.
- 1109 Mao, J., Langner, J., Thompson, A., Tarasick, D., Apel, E., Blake, D., Brune, W., Cohen, R.,
- 1110 Dibb, J., Diskin, G. S., Fried, A., Hall, S., Huey, G., Weinheimer, ennberg, P., Wisthaler, A.,
- 1111 de Gouw, J., Holloway, J., Montzka, S., Nowak, J., Roberts, J. and Ryerson, J.: The
- 1112 POLARCAT Model Intercomparison Project (POLMIP): Overview and evaluation with
- 1113 observations, Atmospheric Chemistry and Physics, to be submitted, 2014.
- Engelen, R. J., Serrar, S., and Chevallier, F.: Four-dimensional data assimilation of
 atmospheric CO2 using AIRS observations, J. Geophys. Res., 114, D03303,
 doi:10.1029/2008JD010739, 2009.
- 1117 Errera, Q., Daerden, F., Chabrillat, S., Lambert, J. C., Lahoz, W. A., Viscardy, S., Bonjean,
- 1118 S., and Fonteyn, D.: 4D-Var assimilation of MIPAS chemical observations: ozone and
- 1119 nitrogen dioxide analyses, Atmos. Chem. Phys., 8, 6169-6187, doi:10.5194/acp-8-6169-2008,
- 1120 2008.

- 1121 Evans, M. J., and Jacob, D. J.: Impact of new laboratory studies of N2O5 hydrolysis on global
- 1122 model budgets of tropospheric nitrogen oxides, ozone, and OH, Geophys. Res. Lett., 32,
- 1123 L09813, doi:10.1029/2005GL022469, 2005.

Fiore, A.M., Naik, V., Spracklen, D.V., Steiner, A., Unger, N., Prather, M., Bergmann, D.,
Cameron-Smith, P.J., Cionni, I., Collins, W.J., Dalsoren, S., Eyring, V., Folberth, G.A.,
Ginoux, P., Horowitz, L.W., Josse, B., Lamarque, J.-F., MacKenzie, I.A., Nagashima, T.,
O'Connor, F.M., Righi, M., Rumbold, S.T., Shindell, D.T., Skeie, R.B., Sudo, K., Szopa, S.,
Takemura, T., Zeng, G., Global air quality and climate, Chemical Society Reviews, 41 (19),
pp. 6663-6683, 2012.

- Flemming, J., Stern, R., and Yamartino, R. J.: A new air quality regime classification scheme
 for O₃, NO₂, SO₂ and PM10 observations sites, Atmos. Environ., 39, 6121–6129, 2005
- 1132 Flemming, J., Inness, A., Flentje, H., Huijnen, V., Moinat, P., Schultz, M. G., and Stein, O.:
- 1133 Coupling global chemistry transport models to ECMWF's integrated forecast system, Geosci.
- 1134 Model Dev., 2, 253-265, doi:10.5194/gmd-2-253-2009, 2009a.
- 1135 Flemming, J., Inness, A., Flentje, H., Huijnen, V., Moinat, P., Schultz, M. G., and Stein, O.:
- 1136 Coupling global chemistry transport models to ECMWF's integrated forecast system
- 1137 ECMWF technical memorandum 590
- 1138 [http://old.ecmwf.int/publications/library/ecpublications/_pdf/tm/501-600/tm590.pdf]
- 1139 tm590.pdf, 2009 b.
- 1140 Flemming, J., Inness, A., Jones, L., Eskes, H. J., Huijnen, V., Schultz, M. G., Stein, O.,
- 1141 Cariolle, D., Kinnison, D., and Brasseur, G.: Forecasts and assimilation experiments of the
- 1142 Antarctic ozone hole 2008, Atmos. Chem. Phys., 11, 1961–1977, doi:10.5194/acp-11-1961-
- 1143 2011, 2011 a.
- 1144 Flemming, J. and Huijnen, V.: IFS Tracer Transport Study, MACC Deliverable G-
- 1145 RG 4.2, Tech. rep., ECMWF, http://www.gmes-atmosphere.eu/documents/deliverables/g-
- 1146 rg/ifs transport study.pdf, 2011 b.
- 1147 Flemming, J.; Peuch, V.-H.; Engelen, R.; Kaiser, J.W. A European Global-to-Regional Air
- 1148 Pollution Forecasting System that Combines Modeling with Satellite Observations; EM
- 1149 Magazine of A&WMA, November 2013, pp. 6-10.
- 1150 https://www.researchgate.net/publication/259535688_A_European_Global-to-
- 1151 Regional_Air_Pollution_Forecasting_System_that_Combines_Modeling_with_Satellite_Obse
- 1152 rvations, 2013

- 1153 Forbes, R.M., A.M. Tompkins & A. Untch, A new prognostic bulk-microphysics scheme for
- 1154 the IFS. ECMWF Tech. Memo. No. 649, 2011.
- Fu, Q., Yang, P. and Sun, W. B.:. An accurate parametrization of the infrared radiative
 properties of cyrrus clouds of climate models. J. Climate, 11, 2223–2237, 1998
- 1157 Gauss, M., Isaksen, I. S. A., Lee, D. S., and Søvde, O. A.: Impact of aircraft NOx emissions
- 1158 on the atmosphere tradeoffs to reduce the impact, Atmos. Chem. Phys., 6, 1529-1548,
- 1159 doi:10.5194/acp-6-1529-2006, 2006.
- Gery, M., Whitten, G. Z., Killus, J. P., and Dodge, M. C.: A photochemical kinetics
 mechanism for urban and regional scale computer modelling, J. Geophys. Res., 94, 18925–
 18956, 1989.
- 1163 Granier, C., J.F. Lamarque, A. Mieville, J.F. Muller, J. Olivier, J. Orlando, J. Peters, G.
- 1164 Petron, G. Tyndall, S. Wallens, POET, a database of surface emissions of ozone precursors,
- 1165 available on internet at http://www.aero.jussieu.fr/projet/ACCENT/POET.php, 2005.
- 1166 Granier, C., B. Bessagnet, T. Bond, A. D'Angiola, H.D.v.d. Gon, G.J. Frost, A. Heil, J.W.
- 1167 Kaiser, S. Kinne, Z. Klimont, S. Kloster, J.-F. Lamarque, C. Liousse, T. Masui, F. Meleux, A.
- 1168 Mieville, T. Ohara, J.-C. Raut, K. Riahi, M.G. Schultz, S.J. Smith, A. Thomson, J.v.
- 1169 Aardenne, G.R.v.d. Werf, and D.P.v. Vuuren, Evolution of anthropogenic and biomass
- 1170 burning emissions of air pollutants at global and regional scales during the 1980-2010 period,
- 1171 Climatic Change, 109(1-2), 163-190, doi:10.1007/s10584-011-0154-1, 2011.
- Grell, G. A., Peckham, S. E., Schmitz, R, McKeen, S. A., Frost, G. J, Skamarock, W. and
 Eder B.: Fully coupled online chemistry within the WRF model, Atmospheric Environment,
 39, 37, 6957-6975, 2005.
- 1175 Grooß, J.-U. and Russell III, J. M.: Technical note: A stratospheric climatology for O3, H2O,
- 1176 CH4, NOx, HCl and HF derived from HALOE measurements, Atmos. Chem. Phys., 5, 2797–
- 1177 2807, doi:10.5194/acp-5-2797-2005, 2005
- 1178 Guenther, A. B., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates
- of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
 Aerosols from Nature), Atmos. Chem. Phys., 6, 3181–3210, doi:10.5194/acp-6-3181-2006,
- 1181 2006.

- Haywood, J. M., Roberts, D. L., Slingo, A., Edwards, J. M., and Shine, K. P.: General
 circulation model calculations of the direct radiative forcing by anthropogenic sulfate and
 fossil-fuel soot aerosol, Journal of Climate, 10, 1562–1577, 1997.
- Hertel, O., Berkowicz, R., Christensen, J. and Hov Ø: Test of two numerical schemes for use
 in atmospheric transport-chemistry models Atmos. Environ., 27A(16), 2591–2611, 1993.
- 1187 Hollingsworth, A., Engelen, R.J., Textor, C., Benedetti, A., Boucher, O., Chevallier, F.,
- 1188 Dethof, A., Elbern, H., Eskes, H., Flemming, J., Granier, C., Kaiser, J.W., Morcrette, J.-J.,
- 1189 Rayner, P., Peuch, V.H., Rouil, L., Schultz, M.G., Simmons, A.J and The GEMS
- 1190 Consortium: Toward a Monitoring and Forecasting System For Atmospheric Composition:
- 1191 The GEMS Project. Bull. Amer. Meteor. Soc., 89, 1147-1164, 2008.
- 1192 Holtslag, A.A. and B. Boville: Local versus nonlocal boundary-layer diffusion in a global
- 1193 climate model, J. Clim., 6, 1825-1842, 1993.
- 1194 Horowitz, L. W., Walters, S., Mauzerall, D. L., Emmons, L. K., Rasch, P. J., Granier, C., Tie,
- 1195 X., Lamarque, J.-F., Schultz, M. G., Tyndall, G. S., Orlando, J. J., and Brasseur, G. P.: A
- 1196 global simulation of tropospheric ozone and related tracers, Description and Evaluation of
- 1197 MOZART version 2, J. Geophys. Res., 108, 4784, doi:10.1029/2002JD002853, 2003.
- Hortal, M. and Simmons, A. J.: Use of reduced Gaussian grids in spectral models, Mon.
 Weather Rev., 119, 1057 1074, 1991.
- Hortal, M.: The development and testing of a new two-time-level semi-Lagrangian scheme
 (SETTLS) in the ECMWF forecast model, 128, 1671–1687, DOI: 10.1002/qj.200212858314,
 2002.
- Houweling, S., Dentener, F. J., and Lelieveld, J.: The impact of nonmethane hydrocarbon
 compounds on tropospheric photochemistry, J. Geophys. Res., 103(D9), 10673–10696, 1998.
- 1205 Huijnen, V., Williams, J., van Weele, M., van Noije, T., Krol, M., Dentener, F., Segers, A.,
- 1206 Houweling, S., Peters, W., de Laat, J., Boersma, F., Bergamaschi, P., van Velthoven, P., Le
- 1207 Sager, P., Eskes, H., Alkemade, F., Scheele, R., Nédélec, P., and Pätz, H.-W.: The global
- 1208 chemistry transport model TM5: description and evaluation of the tropospheric chemistry
- 1209 version 3.0, Geosci. Model Dev., 3, 445-473, doi:10.5194/gmd-3-445-2010.
- 1210 Huijnen, V., Flemming, J., Kaiser, J. W., Inness, A., Leitão, J., Heil, A., Eskes, H. J., Schultz,
- 1211 M. G., Benedetti, A., Hadji-Lazaro, J., Dufour, G., and Eremenko, M.: Hindcast experiments

- 1212 of tropospheric composition during the summer 2010 fires over western Russia, Atmos.
- 1213 Chem. Phys., 12, 4341-4364, doi:10.5194/acp-12-4341-2012, 2012.
- Huijnen, V., Williams, J. E., and Flemming, J.: Modeling global impacts of heterogeneous
 loss of HO2 on cloud droplets, ice particles and aerosols, Atmos. Chem. Phys. Discuss., 14,
 8575-8632, doi:10.5194/acpd-14-8575-2014, 2014.
- 1217 Im, U., Bianconi, R., Solazzo, E., Kioutsioukis, I., Badia, A., Balzarini, A., Baró, R., Bellasio,
- 1218 R., Brunner, D., Chemel, C., Curci, G., Flemming, J., Forkel, R., Giordano, L., Jiménez-
- 1219 Guerrero, P., Hirtl, M., Hodzic, A., Honzak, L., Jorba, O., Knote, C., Kuenen, J.J.P., Makar,
- 1220 P.A., Manders-Groot, A., Neal, L., Pérez, J.L., Pirovano, G., Pouliot, G., San Jose, R.,
- 1221 Savage, N., Schroder, W., Sokhi, R.S., Syrakov, D., Torian, A., Tuccella, P., Werhahn, J.,
- 1222 Wolke, R., Yahya, K., Zabkar, R., Zhang, Y., Zhang, J., Hogrefe, C., Galmarini, S.:
- 1223 Evaluation of operational on-line-coupled regional air quality models over Europe and North
- 1224 America in the context of AQMEII phase 2. Part I: Ozone, Atmospheric Environment , doi:
- 1225 10.1016/j.atmosenv.2014.09.042, 2014.
- 1226 Inness, A., Baier, F., Benedetti, A., Bouarar, I., Chabrillat, S., Clark, H., Clerbaux, C.,
- 1227 Coheur, P., Engelen, R. J., Errera, Q., Flemming, J., George, M., Granier, C., Hadji-Lazaro,
- 1228 J., Huijnen, V., Hurtmans, D., Jones, L., Kaiser, J. W., Kapsomenakis, J., Lefever, K., Leitão,
- 1229 J., Razinger, M., Richter, A., Schultz, M. G., Simmons, A. J., Suttie, M., Stein, O., Thépaut,
- 1230 J.-N., Thouret, V., Vrekoussis, M., Zerefos, C., and the MACC team: The MACC reanalysis:
- 1231 an 8 yr data set of atmospheric composition, Atmos. Chem. Phys., 13, 4073-4109,
- 1232 doi:10.5194/acp-13-4073-2013, 2013.
- 1233 Inness, A., Blechschmidt, A., Bouarar, I., Chabrillat, S., Crepulja, M., Engelen, R. J., Errera,
- 1234 Q., Flemming, J., Gaudel, A., Huijnen, V., Jones, L., Kapsomenakis, J., Keppens A.,
- 1235 Lambert, J.-C., Langerock, B., Peuch, V.H., Razinger, M., Richter, A., Schultz, M. G., Suttie,
- 1236 M., Thouret, V., Vrekoussis, M., Wagner, A. and Zerefos, C. : Data assimilation experiments
- 1237 of satellite retrievals of O3, CO and NO2 with Composition IFS, in preparation for Geosci.
- 1238 Model Dev, 2014.
- 1239 Jacob, D.J. H. Liu, C.Mari, and R.M. Yantosca, Harvard wet deposition scheme for GMI,
- 1240 Harvard University Atmospheric Chemistry Modeling Group, revised March 2000.
- 1241 http://acmg.seas.harvard.edu/geos/wiki_docs/deposition/wetdep.jacob_etal_2000.pdf
- 1242 Jakob, C. and Klein, S.: A parameterization of the effects of cloud and precipitation overlap
- 1243 for use in general-circulation models, Q. J. Roy. Meteor. Soc., 126, 2525–2544, 2000.

- 1244 Jung, T., T. N. Palmer, M. J. Rodwell, and S. Serrar, 2008: Diagnosing forecast error using
- relaxation experiments. ECMWF Newsletter 82, ECMWF, Shinfield Park, Reading, BerkshireRG2 9AX, UK.
- 1247 Kaiser, J. W., Heil, A., Andreae, M. O., Benedetti, A., Chubarova, N., Jones, L., Morcrette,
- 1248 J.-J., Razinger, M., Schultz, M. G., Suttie, M., and van der Werf, G. R.: Biomass burning
- 1249 emissions estimated with a global fire assimilation system based on observed fire radiative
- 1250 power, Biogeosciences, 9, 527-554, doi:10.5194/bg-9-527-2012, 2012.
- 1251 Kaminski, J. W., Neary, L., Struzewska, J., McConnell, J. C., Lupu, A., Jarosz, J., Toyota, K.,
- 1252 Gong, S. L., Côté, J., Liu, X., Chance, K., and Richter, A.: GEM-AQ, an on-line global
- 1253 multiscale chemical weather modelling system: model description and evaluation of gas phase
- 1254 chemistry processes, Atmos. Chem. Phys., 8, 3255-3281, 2008.
- 1255 Kinnison, D. E., Brasseur, G. P., Walters, S., Garcia, R. R., Marsh, D. R, Sassi, F., Harvey,
- 1256 V. L., Randall, C. E., Emmons, L., Lamarque, J. F., Hess, P., Orlando, J. J., Tie, X. X.,
- 1257 Randel, W., Pan, L. L., Gettelman, A., Granier, C., Diehl, T., Niemeier, U. and Simmons, A.
- 1258 J.: Sensitivity of Chemical Tracers to Meteorological Parameters in the MOZART-3
- 1259 Chemical Transport Model. J. Geophys. Res, 112, D03303, doi:10.1029/2008JD010739,2007.
- Komhyr, W. D., Barnes, R. A., Borthers, G. B., Lathrop, J. A., Kerr, J. B., and Opperman, D.
 P.: Electrochemical concentration cell ozonesonde performance evaluation during STOIC
- 1262 1989, J. Geophys. Res., 100, 9231–9244, 1995
- 1263 Lamarque, J.-F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C.
- L., Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P. J., and Tyndall, G. K.:
 CAM-chem: description and evaluation of interactive atmospheric chemistry in the
 Community Earth System Model, Geosci. Model Dev., 5, 369-411, doi:10.5194/gmd-5-3692012, 2012.
- 1268 Lamarque, J.-F., Shindell, D. T., Josse, B., Young, P. J., Cionni, I., Eyring, V., Bergmann, D.,
- 1269 Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi, G., Folberth, G.,
- 1270 Ghan, S. J., Horowitz, L. W., Lee, Y. H., MacKenzie, I. A., Nagashima, T., Naik, V.,
- 1271 Plummer, D., Righi, M., Rumbold, S. T., Schulz, M., Skeie, R. B., Stevenson, D. S., Strode,
- 1272 S., Sudo, K., Szopa, S., Voulgarakis, A., and Zeng, G.: The Atmospheric Chemistry and
- 1273 Climate Model Intercomparison Project (ACCMIP): overview and description of models,

- simulations and climate diagnostics, Geosci. Model Dev., 6, 179-206, doi:10.5194/gmd-6-179-2013, 2013.
- Landgraf, J. and Crutzen, P. J.: An efficient method for online calculations of photolysis andheating trates, J. Atmos. Sci., 55, 863–878, 1998.
- Lawrence, M.G. and Crutzen, P. J: The impact of cloud particle gravitational settling on soluble trace gas distributions. Tellus B, 50: 263–289. doi: 10.1034/j.1600-0889.1998.
- 1280 Lefever, K., van der A, R., Baier, F., Christophe, Y., Errera, Q., Eskes, H., Flemming, J.,
- 1281 Inness, A., Jones, L., Lambert, J.-C., Langerock, B., Schultz, M. G., Stein, O., Wagner, A.,
- 1282 and Chabrillat, S.: Copernicus atmospheric service for stratospheric ozone: validation and
- 1283 intercomparison of four near real-time analyses, 2009–2012, Atmos. Chem. Phys. Discuss.,
- 1284 14, 12461-12523, doi:10.5194/acpd-14-12461-2014, 2014.
- 1285 Levine, S. Z. and Schwartz, S. E.: In-cloud and below-cloud scavenging of nitric acid vapor.
- 1286 Atmos. Environ. doi:10.1016/0004-6981(82)90266-9, 1982.
- 1287 Lin, J.-T., McElroy, M. B., and Boersma, K. F.: Constraint of anthropogenic NOx emissions
- in China from different sectors: a new methodology using multiple satellite retrievals, Atmos.
 Chem. Phys., 10, 63-78, doi:10.5194/acp-10-63-2010, 2010.
- 1290 Liu, H., Jacob, D.J., Bey, I., Yantosca, R.M., 2001. Constraints from 210 Pb and 7 Be on wet
- 1291 deposition and transport in a global three-dimensional chemical tracer model driven by
- assimilated meteorological fields. Journal of Geophysical Research 106, 12109e12128.
- 1293 McGregor, J. L.: C-CAM Geometric Aspects and Dynamical Formulation, Tech. Rep. 70,
- 1294 CSIRO Atmospheric Research, Aspendale, Victoria, 2005.
- 1295MACC VAL report, Validation report of the MACC reanalysis of global atmospheric1296compositionPeriod,2003-
- 1297 2011.http://www.copernicusatmosphere.eu/documents/maccii/deliverables/val/MACCII_VA
- 1298 L_DEL_D_83.4_REAreport02_20130207.pdf, 2013
- Mann, G. W., Carslaw, K. S., Spracklen, D. V., Ridley, D. A., Manktelow, P. T., Chipperfield, M. P., Pickering, S. J., and Johnson, C. E.: Description and evaluation of GLOMAP-mode: a modal global aerosol microphysics model for the UKCA compositionclimate model, Geosci. Model Dev., 3, 519-551, doi:10.5194/gmd-3-519-2010, 20103-651-2010, 2010.

- 1304 Marenco, A., Thouret, V., Nedelec, P., Smit, H. G., Helten, M., Kley, D., Karcher, F., Simon,
- 1305 P., Law, K., Pyle, J., Poschmann, G., Von Wrede, R., Hume, C., and Cook, T.: Measurement
- 1306 of ozone and water vapour by Airbus in-service air-craft: The MOZAIC airborne programme,
- 1307 an overview, J. Geophys. Res., 103, 25631–25642, 1998.
- Mari, C., Jacob, D. J., and Bechtold, P.: Transport and scavenging of soluble gases in a deepconvective cloud, J. Geophys. Res., 105, 22 255-22 267, 2000.
- 1310 Matsumi, Y., Comes, F. J., Hancock, G., Hofzumahus, A., Hynes, A. J., Kawasaki, M., and
- 1311 Ravishankara, A. R.: Quantum yields for the production of O(1D) in the ultraviolet photolysis
- 1312 of ozone: recommendation based on evaluation of laboratory data, J. Geophys. Res., 107,
- 1313 4024, doi:10.1029/2001JD000510, 2002.
- Meijer, E.W., P. F. J. van Velthoven, D. W. Brunner, H. Huntrieser and H. Kelder:
 Improvement and evaluation of the parameterization of nitrogen oxide production by
 lightning, Physics and Chemistry of the Earth, Part C, Volume 26, Issue 8, Pages 577-583,
 2001.
- 1318 Ménard , R. et al. ,Coupled chemical-dynamical data assimilation, Final Report, ESA/ESTEC.1319 2007.
- Metzger, S., F. Dentener, S. Pandis, and J. Lelieveld, Gas/aerosol partitioning, 1, A
 computationally efficient model, J. Geophys. Res., 107(D16), doi:10.1029/2001JD001102,
 2002a.
- Metzger, S., Dentener, F., Krol, M. C., Jeuken, A., and Lelieveld, J.: Gas/aerosol partitioning
 2. Global modeling results, J. Geophys. Res., 107(D16), 4313, doi:10.1029/2001JD001103,
 2002b.
- Michou M., P. Laville, D. Serça, A. Fotiadi, P. Bouchou and V.-H. Peuch, Measured and
 modeled dry deposition velocities over the ESCOMPTE area, Atmos. Res., 74 (1-4), 89-116,
 2004.
- Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N.,
 Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L.,
 Steenrod, S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Schlager, H., and Ancellet, G.:
 Multi-model study of chemical and physical controls on transport of anthropogenic and
 biomass burning pollution to the Arctic, Atmos. Chem. Phys. Discuss., 14, 25281-25350,
 doi:10.5194/acpd-14-25281-2014, 2014.

- 1335 Morcrette, J.-J., Boucher, O., Jones, L., Salmond, D., Bechtold, P., Beljaars, A., Benedetti,
- 1336 A., Bonet, A., Kaiser, J. W., Razinger, M., Schulz, M., Serrar, S., Simmons, A. J., Sofiev,
- 1337 M., Suttie, M., Tompkins, A. M. and Untch, A.: Aerosol analysis and forecast in the ECMWF
- 1338 Integrated Forecast System. Part I: Forward modelling, J. Geophys. Res., 2009.
- 1339 Morgenstern, O., Braesicke, P., O'Connor, F. M., Bushell, A. C., Johnson, C. E., Osprey, S.
- 1340 M., and Pyle, J. A.: Evaluation of the new UKCA climate-composition model Part 1: The
- 1341 stratosphere, Geosci. Model Dev., 2, 43–57, doi:10.5194/gmd-2-43-2009, 2009.
- 1342 Nedelec, P., Cammas, J.-P., Thouret, V., Athier, G., Cousin, J.-M., Legrand, C., Abonnel, C.,
- 1343 Lecoeur, F., Cayez, G., and Marizy, C.: An improved infrared carbon monoxide analyser for
- routine measurements aboard commercial Airbus aircraft: technical validation and first
 scientific results of the MOZAIC III programme, Atmos. Chem. Phys., 3, 1551–1564,
 doi:10.5194/acp-3-1551-2003, 2003.
- Neu, J. L. and Prather, M. J.: Toward a more physical representation of precipitation
 scavenging in global chemistry models: cloud overlap and ice physics and their impact on
 tropospheric ozone, Atmos. Chem. Phys., 12, 3289-3310, doi:10.5194/acp-12-3289-2012,
 2012.
- 1351 O'Connor, F. M., Johnson, C. E., Morgenstern, O., Abraham, N. L., Braesicke, P., Dalvi, M.,
- 1352 Folberth, G. A., Sanderson, M. G., Telford, P. J., Voulgarakis, A., Young, P. J., Zeng, G.,
- 1353 Collins, W. J., and Pyle, J. A.: Evaluation of the new UKCA climate-composition model -
- 1354 Part 2: The Troposphere, Geosci. Model Dev., 7, 41-91, doi:10.5194/gmd-7-41-2014, 2014.
- Olivier J., J. Peters, C. Granier, G. Petron, J.F. Muller, and S. Wallens: Present and future
 surface emissions of atmospheric compounds, POET report #2, EU project EVK2-199900011, 2003
- 1358 Ordóñez, C., Elguindi, N., Stein, O., Huijnen, V., Flemming, J., Inness, A., Flentje, H.,
- 1359 Katragkou, E., Moinat, P., Peuch, V.-H., Segers, A., Thouret, V., Athier, G., van Weele, M.,
- 1360 Zerefos, C. S., Cammas, J.-P., and Schultz, M. G.: Global model simulations of air pollution
- during the 2003 European heat wave, Atmos. Chem. Phys., 10, 789-815, doi:10.5194/acp-10-
- 1362 789-2010, 2010.
- Ott, L. E., K. E. Pickering, G. L. Stenchikov, D. J. Allen, A. J. DeCaria, B. Ridley, R.-F. Lin,
 S. Lang, and W.-K. Tao (2010), Production of lightning NOx and its vertical distribution

- calculated from three-dimensional cloud-scale chemical transport model simulations, J.
 Geophys. Res., 115, D04301, doi:10.1029/2009JD011880.
- 1367 Pickering, K. E., Y. Wang, W.-K. Tao, C. Price, and J.-F. Müller: Vertical distributions of
- lightning NOx for use in regional and global chemical transport models, J. Geophys. Res.,
 103, 31,203 31,216, doi:10.1029/98JD0265. 1998.
- 1370 Pozzoli L., Bey, I., Rast, J. S., Schultz, M. G., Stier, P., and Feichter, J.: Trace gas and aerosol
- 1371 interactions in the fully coupled model of aerosol-chemistry-climate ECHAM5-HAMMOZ,
- 1372 PART I: Model description and insights from the spring 2001 TRACE-P experiment, J.
- 1373 Geophys. Res., 113, 2008.
- 1374 Prather, M., Ehhalt, D., et al.: Atmospheric chemistry and greenhouse gases, in: Climate
- 1375 Change 2001: The Scientific Basis, edited by: Houghton, J. T., Ding, Y., Griggs, D. J., et al.,
- 1376 239–287, Cambridge University Press, Cambridge, UK, 2001.
- 1377 Prather, M. J., Holmes, C. D., and Hsu, J.: Reactive greenhouse gas scenarios: Systematic
- 1378 exploration of uncertainties and the role of atmospheric chemistry, Geophys. Res. Lett., 39,
- 1379 L09803,doi:10.1029/2012GL051440, 2012.
- Price, C., and Rind, D.: A simple lightning parameterization for calculating global lightning
 distributions, J. Geophys. Res., 9 7, 9919-9933, 1992.
- Price, C., and Rind, D.: What determinest he cloud-to-ground fraction in thunderstorms?Geophys Res. Lett., 20, 463-466, 1993.
- Price, C., J. Penner, and M. Prather: NOx from lightning 1. Global distributions based on
 lightning physics, J. Geophys. Res., 102, 5929–5941, doi:10.1029/96JD03504, 1997.
- Rast, S., Schultz, M.G., Bey, I., van Noije, T., Aghedo, A.M., Brasseur, G.P., Diehl, T.,
 Esch, M., Ganzeveld, L., Kirchner, I., Kornblueh, L., Rhodin, A., Röckner, E., Schmidt, H.,
 Schröder, S., Schulzweida, U., Stier, P., Thomas, K., Walters, S.: Evaluation of the
 tropospheric chemistry general circulation model ECHAM5–MOZ and its application to the
 analysis of the chemical composition of the troposphere with an emphasis on the late RETRO
- 1391 period 1990-2000, Reports on Earth-System Science, 114, Max-Planck Institut fuer
- 1392 Meteorologie, Hamburg, 2014.

- 1393 Redler, R., Valcke, S. and Ritzdorf, H.: OASIS4 A Coupling Software for Next Generation
- Earth System Modelling, Geoscience Model Development, 3, 87 104, DOI:10.5194/gmd-3-87-2010.
- Richter, A., Burrows, J. P., Nüß, H., Granier, C, Niemeier, U., Increase in tropospheric
 nitrogen dioxide over China observed from space, Nature, 437, 129-132, doi:
 10.1038/nature04092, 2005.
- 1399 Sander, R., Compilation of Henry's Law Constants for Inorganic and Organic Species of
- 1400 Potential Importance in Environmental Chemistry, MPI for Chemistry Mainz, Germany,
- 1401 1999, http://www.henrys-law.org/henry.pdf
- 1402 Sander, S. P., Friedl, R. R., Golden, D. M., Kurylo, M. J., Moortgat, G. K., Keller-Rudek, H.,
- 1403 Wine, P. H., Ravishankara, A. R., Kolb, C. E., Molina, M. J., Finlayson-Pitts, B. J., Huie, R.
- 1404 E., and Orkin, V. L: Chemical Kinetics and Photochemical Data for Use in Atmospheric
- Studies, Evaluation Number 15, JPL Publication 06-02, Jet Propulsion Laboratory, Pasadena,Calif., 2006.
- Sander, S. P., Abbatt, J. R., Burkholder, J. B., Friedl, R. R., Golden, D. M., Huie, R. E., Kolb,
 C. E., Kurylo, G., Moortgat, K., Orkin, V. L. and Wine, P. H.: Chemical kinetics and
- Photochemical Data for Use in Atmospheric studies, Evaluation No.17, JPL Publication 10-6,Jet Propulsion Laboratory, Pasadena, 2011.
- 1411 Savage, N. H., Agnew, P., Davis, L. S., Ordóñez, C., Thorpe, R., Johnson, C. E., O'Connor, F.
- 1412 M., and Dalvi, M.: Air quality modelling using the Met Office Unified Model (AQUM OS24-
- 1413 26): model description and initial evaluation, Geosci. Model Dev., 6, 353-372,
- 1414 doi:10.5194/gmd-6-353-2013, 2013.
- 1415 Schumann, U., and H. Huntrieser: The global lightning-induced nitrogen oxides source,
- 1416 Atmos. Chem. Phys., 7, 3823–3907, 2007.
- 1417 Seinfeld J. H. and Pandis S. N., Atmospheric Chemistry and Physics: From Air Pollution to
- 1418 Climate Change, 1st edition, J. Wiley, New York, 1998.
- 1419 Shindell, D.T., G. Faluvegi, D.S. Stevenson, M.C. Krol, L.K. Emmons, J.-F. Lamarque, G.
- 1420 Pétron, F.J. Dentener, K. Ellingsen, M.G. Schultz, O. Wild, M. Amann, C.S. Atherton, D.J.
- 1421 Bergmann, I. Bey, T. Butler, J. Cofala, W.J. Collins, R.G. Derwent, R.M. Doherty, J. Drevet,
- 1422 H.J. Eskes, A.M. Fiore, M. Gauss, D.A. Hauglustaine, L.W. Horowitz, I.S.A. Isaksen, M.G.
- 1423 Lawrence, V. Montanaro, J.-F. Müller, G. Pitari, M.J. Prather, J.A. Pyle, S. Rast, J.M.

- 1424 Rodriguez, M.G. Sanderson, N.H. Savage, S.E. Strahan, K. Sudo, S. Szopa, N. Unger, T.P.C.
- van Noije, and G. Zeng: Multi-model simulations of carbon monoxide: Comparison with
 observations and projected near-future changes. J. Geophys. Res., 111, D19306,
 doi:10.1029/2006JD007100, 2006.
- 1428 Slingo, A.: A GCM parameterization for the shortwave radiative properties of water clouds. J.
- 1429 Atmos. Sci., 46, 1419–1427, 1989.
- 1430 Smithson, P. A., IPCC, 2001: climate change 2001: the scientific basis. Contribution of
- 1431 Working Group 1 to the Third Assessment Report of the Intergovernmental Panel on Climate
- 1432 Change, edited by J. T. Houghton, Y. Ding, D. J. Griggs, M. Noguer, P. J. van der Linden, X.
- 1433 Dai, K. Maskell and C. A. Johnson (eds). Cambridge University Press, Cambridge, UK, and
- 1434 New York, USA, Int. J. Climatol., 22: 1144. doi: 10.1002/joc.763, 2002.
- 1435 Steil, B., Dameris, M., Brühl, C., Crutzen, P. J., Grewe, V., Ponater, M., and Sausen, R.:
- 1436 Development of a chemistry module for GCMs: first results of a multiannual integration,
- 1437 Ann. Geophys., 16, 205-228, doi:10.1007/s00585-998-0205-8, 1998.
- Stein, O., Flemming, J., Inness, A., Kaiser, J.W., Schultz, M.G., Global reactive gases
 forecasts and reanalysis in the MACC project, Journal of Integrative Environmental Sciences,
 9, Iss. sup1, 57-70, doi:10.1080/1943815X.2012.696545, 2012.
- Stein, O., Schultz, M. G., Bouarar, I., Clark, H., Huijnen, V., Gaudel, A., George, M., and
 Clerbaux, C.: On the wintertime low bias of Northern Hemisphere carbon monoxide in global
 model studies, Atmos. Chem. Phys. Discuss., 14, 245-301, doi:10.5194/acpd-14-245-2014,
 2014.
- Steinbrecht, W., Shwartz, R., and Claude, H.: New pump correction for the Brewer-Mast
 ozonesonde: Determination from experiment and instrument intercomparisons, J. Atmos.
 Ocean. Tech.15, 144–156, 1998.
- Stevenson, D. S., et al. (2006), Multimodel ensemble simulations of present-day and nearfuture tropospheric ozone, J. Geophys. Res., 111, D08301, doi:10.1029/2005JD006338.
- Temperton, C., Hortal, M. and Simmons, A.: A two-time-level semi-Lagrangian globalspectral model, QJR, 127, 111-127, 2001.
- 1452 Tiedtke, M. A: comprehensive mass flux scheme for cumulus parameterization in large-scale
- 1453 models. Mon. Weather. Rev., 117(8):1779-1800, 1989.

- 1454 Tilmes, S., Lamarque, J.-F., Emmons, L. K., Conley, A., Schultz, M. G., Saunois, M.,
- 1455 Thouret, V., Thompson, A. M., Oltmans, S. J., Johnson, B., and Tarasick, D.: Technical Note:
- 1456 Ozonesonde climatology between 1995 and 2011: description, evaluation and applications,
- 1457 Atmos. Chem. Phys., 12, 7475-7497, doi:10.5194/acp-12-7475-2012, 2012.
- Wesely, M.L.: Parameterization of Surface Resistances to Gaseous Dry Deposition in
 Regional-Scale Numerical Models. Atmos. Environ., 23, 1293-1304, 1989.
- 1460 Val Martin, M., Heald, C. L. and Arnold, S. R.: Coupling dry deposition to vegetation
- 1461 phenology in the Community Earth System Model: Implications for the simulation of surface
- 1462 O3, Geophys. Res. Lett., 41, 2988–2996, doi:10.1002/2014GL059651., 2014van Noije, T. P.
- 1463 C., Le Sager, P., Segers, A. J., van Velthoven, P. F. J., Krol, M. C., Hazeleger, W., Williams,
- 1464 A. G., and Chambers, S. D.: Simulation of tropospheric chemistry and aerosols with the
- 1465 climate model EC-Earth, Geosci. Model Dev., 7, 2435-2475, doi:10.5194/gmd-7-2435-2014,
- 1466 2014.
- von Blohn, N., Diehl, K., Mitra, S. K., and Borrmann, S.: Wind tunnel experiments on the
 retention of trace gases during riming: nitric acid, hydrochloric acid, and hydrogen peroxide,
 Atmos. Chem. Phys., 11, 11569-11579, doi:10.5194/acp-11-11569-2011, 2011.
- 1470 Voulgarakis, A., Naik, V., Lamarque, J.-F., Shindell, D. T., Young, P. J., Prather, M. J., Wild,
- 1471 O., Field, R. D., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins, W. J., Dalsøren, S. B.,
- 1472 Doherty, R. M., Eyring, V., Faluvegi, G., Folberth, G. A., Horowitz, L. W., Josse, B.,
- 1473 MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T., Stevenson, D.
- S., Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Analysis of present day and future OH
 and methane lifetime in the ACCMIP simulations, Atmos. Chem. Phys., 13, 2563-2587,
 doi:10.5194/acp-13-2563-2013, 2013.
- 1477 Vrekoussis, M., Wittrock, F., Richter, A., and Burrows, J. P.: GOME-2 observations of
 1478 oxygenated VOCs: what can we learn from the ratio glyoxal to formaldehyde on a global
 1479 scale?, Atmos. Chem. Phys., 10, 10145-10160, doi:10.5194/acp-10-10145-2010, 2010.
- Williams, J. E., Strunk, A., Huijnen, V., and van Weele, M.: The application of the Modified
 Band Approach for the calculation of on-line photodissociation rate constants in TM5:
 implications for oxidative capacity, Geosci. Model Dev., 5, 15-35, doi:10.5194/gmd-5-152012, 2012.
- Williams, J. E., van Velthoven, P. F. J., and Brenninkmeijer, C. A. M.: Quantifying the
 uncertainty in simulating global tropospheric composition due to the variability in global
 emission estimates of Biogenic Volatile Organic 2857-2013, 2013.

- Wittrock, F., A. Richter, H. Oetjen, J. P. Burrows, M. Kanakidou, S. Myriokefalitakis, R.
 Volkamer, S. Beirle, U. Platt, and T. Wagner, Simultaneous global observations of glyoxal
 and formaldehyde from space, Geophys. Res. Lett., 33, L16804, doi:10.1029/2006GL026310,
 2006Compounds, Atmos. Chem. Phys., 13, 2857-2891, doi:10.5194/acp-13-2857-2013, 2013.
- Yarwood, G., Rao, S., Yocke, M., and Whitten, G.: Updates to the carbon bond chemical
 mechanism: CB05. Final report to the US EPA, EPA Report Number: RT-0400675, available
 at: www.camx.com, last access: 1 July 2014, 2005.
- 1494 WMO (2007), WMO Global Atmosphere Watch (GAW) Strategic Plan: 2008 2015. World
 1495 Meteorological Organization, Geneva, Switzerland, 2007.
- 1496 Young, P. J., Archibald, A. T., Bowman, K. W., Lamarque, J.-F., Naik, V., Stevenson, D. S.,
- 1497 Tilmes, S., Voulgarakis, A., Wild, O., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins,
- 1498 W. J., Dalsøren, S. B., Doherty, R. M., Eyring, V., Faluvegi, G., Horowitz, L. W., Josse, B.,
- 1499 Lee, Y. H., MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T.,
- 1500 Skeie, R. B., Shindell, D. T., Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Pre-industrial
- 1501 to end 21st century projections of tropospheric ozone from the Atmospheric Chemistry and
- 1502 Climate Model Intercomparison Project (ACCMIP), Atmos. Chem. Phys., 13, 2063-2090,
- 1503 doi:10.5194/acp-13-2063-2013, 2013.
- Zaveri, R. A. and Peters, L. K.: A new lumped structure photochemical mechanism for largescale applications, J. Geophys. Res., 104, 30387–30415, doi:10.1029/1999JD900876, 1999.
- Zdunkowski, W. G., Welsch, R. M., and Kord, G. J.: An investigation of the structure of
 typical 2-stream methods for the calculation of solar fluxes and heating rates in clouds,
 Contrib. Atmos. Phys., 53, 215–238, 1980.
- Zhang, L., Brook, J. R., and Vet, R.: A revised parameterization for gaseous dry deposition in
 air-quality models, Atmos. Chem. Phys., 3, 2067-2082, doi:10.5194/acp-3-2067-2003, 2003.
- 1511 Zhang, Y.: On-line coupled meteorology and chemistry models: history, current status, and
- 1512 outlook, Atmos. Chem. Phys, 8, 2895-2032, 2008.
- 1513 Zhang, Y., Karamchandani, P., Glotfelty, T., Streets, D. G., Grell, G., Nenes, A., Yu, F. and
- 1514 Bennartz, R.: Development and initial application of the global-through-urban weather
- 1515 research and forecasting model with chemistry (GU-WRF/Chem), J. Geophys. Res., 117,
- 1516 D20206, doi:10.1029/2012JD017966, 2012.

1517 Table 1 Annual emissions from anthropogenic, biogenic and natural sources and biomass

1518 burning for 2008 in Tg for a C-IFS (CB05) run at T255 resolution. Anthropogenic NO

1519 emissions contain a contribution of 1.8 Tg aircraft emissions and 12.3 Tg (5.7 Tg N) lightning

1520	emissions	(LiNO) is	added in	the biomass	burning columns.
------	-----------	-----------	----------	-------------	------------------

Species	Anthropogenic	Biogenic natural	and	Biomass burning
СО	584	96		328
NO	70 + 1.8	10		9.2 + 12.3 (LiNO)
НСНО	3.4	4.0		4.9
CH ₃ OH	2.2	159		8.5
C_2H_6	3.4	1.1		2.3
C ₂ H ₅ OH	3.1	0		0
C_2H_4	7.7	18		4.3
C_3H_8	4.0	1.3		1.2
C_3H_6	3.5	7.6		2.5
Parafins (Tg C)	31	18		1.7
Olefines (Tg C)	2.4	0		0.7
Aldehydes (Tg C)	1.1	6.1		2.1
CH ₃ COCH ₃	1.3	28		2.4
Isoprene	0	523		0
Terpenes	0	97		0
SO ₂	98	9		2.2
DMS	0	38		0.2
NH ₃	40	11		6.2

1521

Region	Area S/W/N/E	Stations (Number of observations)
Europe	35°N/20°W/60°N/40°E	Barajas (52), DeBilt (57), Hohenpeissenberg (126), Legionowo (48), Lindenberg (52), Observatoire de Haute-Provence (46), Payerne (158), Prague (49), Uccle (142) and Valentia Observatory (49)
North America:	30°N/135°W/60°N/60°W	Boulder (65), Bratts Lake (61), Churchill (61), Egbert (29), Goose Bay (47), Kelowna (72), Stony Plain (77), Wallops (51), Yarmouth (60), Narragansett (7) and Trinidad Head (35)
Arctic:	60°N/180°W/90°N/180°E	Alert (52), Eureka (83), Keflavik (8), Lerwick (49), Ny-Aalesund (77), Resolute (63), Scoresbysund (54), Sodankyla (63), Summit (81) and Thule(15)
Tropics	20°S/180°W/20°N/180°E	Alajuela (47), Ascension Island (32), Hilo (47), Kuala Lumpur (24), Nairobi (39), Natal (48), Paramaribo (35), Poona (13), Samoa (33), San Cristobal (28), Suva (28), Thiruvananthapuram (12) and Watukosek (19)
East Asia	15°N/100°E/45°N/142°E	Hong Kong Observatory (49), Naha (37), Sapporo (42) and Tateno Tsukuba (49)
Antarctic	90°S/180°W/60°S/180°E	Davis (24), Dumont d'Urville (38), Maitri (9), Marambio (66), Neumayer (72), South Pole (63), Syowa(41) and McMurdo (18)

1523 Table 2 Ozone sondes sites used in the evaluation for different regions

1525 Table 3 Computational cost (BU) of a 24 h forecasts of different horizontal model resolutions

Resolution	IFS-MOZART	C-IFS (MOZART)*	C-IFS (MOCAGE)*	C-IFS (CB05)	IFS
T159	205	56	147	20	6
T255	1200	-	-	55	12
T511	-	-	-	700	125

1526 (60 levels) and chemistry schemes of C-IFS, IFS-MOZART and IFS, *not fully optimised.

1527



1529

1530 Figure 1 Flash density in flashes/(km² yr) from the IFS input data using the parameterization

- 1531 by Price and Rind (1992) (left), Meijer et al. (2001) (middle) and observations from the LIS
- 1532 OTD data base (right). All fields were scaled to an annual flash density of 46 fl/s.
- 1533



Figure 2 Tropospheric ozone volume mixing ratios (ppb) over Europe (left) and North-America (middle) and East Asia (right) averaged in the pressure range 1000-700 hPa (bottom), 700-400 hPa (middle) and 400-200 hPa (top) observed by ozonesondes (black) and simulated by C-IFS (red), MOZ (blue) and REAN (green) in 2008.



Figure 3 Tropospheric ozone volume mixing ratios (ppb) over the Tropics (left) Antarctica
(middle) and Arctic (right) averaged in the pressure bands 1000-700 hPa (bottom), 700-400
hPa (middle) and 400-200 hPa (top) observed by ozonesondes and simulated by C-IFS (red),
MOZ (blue) and REAN (green) in 2008.



1548 Figure 4 Annual cycle of the mean ozone volume mixing ratios (ppb) at rural sites of the1549 EMEP and AirBase data base and simulated by C-IFS (red), MOZ (blue) and REAN (green).



Figure 5 Diurnal cycle of surface ozone volume mixing ratios (ppb) over Europe in winter (top, left), spring (top, right), summer (bottom, left) and autumn (bottom, right) at rural site of the EMEP and AirBase data base and simulated by C-IFS (red), MOZ (blue) and REAN (green).



- 1557 Figure 6 CO total column retrieval (MOPITT V6) for April 2008 (top left) and simulated by
- 1558 C-IFS (top right), MOZ (bottom left) and REAN (bottom right), AK are applied.



1561 Figure 7 CO total column retrieval (MOPITT V6) for August 2008 (top left) and simulated by

1562 C-IFS (top right), MOZ (bottom right) and REAN (bottom left), AK are applied.

1563



Figure 8 CO volume mixing ratios (ppb) over Frankfurt (left), London (middle) and North America (left, averaged over 6 airports) averaged in the pressure bands 1000-700 hPa (bottom), 700-400 hPa (middle) and 400-200 hPa (top) observed by MOZAIC and simulated by C-IFS (red), MOZ (blue) and REAN (green) in 2008.





1572 Figure 9 CO volume mixing ratios (ppb) over Caracas (left) Windhoek (middle) and
1573 Hyderabad (right) averaged in the pressure bands 1000-700 hPa (bottom), 700-400 hPa
1574 (middle) and 400-200 hPa (top) observed by MOZAIC, and simulated by C-IFS (red), MOZ
1575 (blue) and REAN (green) in 2008.



1579 Figure 10 Time series of median of weekly CO surface volume mixing ratios (ppb) in Europe1580 (13 GAW sites) and model results of C-IFS, MOZ and REAN.



1584 Figure 11 NO₂ tropospheric column retrieval (GOME-2) for 2008 (top left) and by C-IFS (top

- 1585 right), REAN (bottom right) and MOZ (bottom left)



Figure 12 Time series of area-averaged tropospheric NO₂ columns [10¹⁵ molec cm-2] from
GOME-2 compared to model results of C-IFS (CB05) (blue), MOZ (red) and REAN (green)
for different regions.



Figure 13 Time series of median of weekly surface NO₂ volume mixing ratios (ppb) in
Europe (20 GAW sites) and model results of C-IFS, MOZ and REAN.





1601 Figure 14 HCHO tropospheric column retrieval (GOME-2) for 2008 (top left) and by C-IFS

- 1602 (top right), REAN (bottom right) and MOZ (bottom left)
- 1603



Figure 15 Time series of area-averaged tropospheric HCHO columns [10¹⁶ molec cm-2] from
GOME-2 compared to model results of C-IFS, MOZ and REAN for different regions.





Figure 16 Time series of median of weekly surface SO₂ volume mixing ratios (ppb) in Europe (21 GAW sites) and model resuls of C-IFS, MOZ and REAN