GNAQPMS-Hg v1.0, a global nested atmospheric mercury transport model: Model description, evaluation and application to trans-boundary transport of Chinese anthropogenic emissions

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## 1 Abstract

2 Atmospheric mercury (Hg) is a toxic pollutant and can be transported over the whole globe due to its long lifetime in the atmosphere. For the purpose of assessing Hg 3 hemispheric transport and better characterizing regional Hg pollution, a global 4 nested atmospheric Hg transport model (GNAQPMS-Hg) has been developed. In 5 GNAQPMS-Hg, the gas and aqueous phase Hg chemistry representing the 6 7 transformation among three forms of Hg: elemental mercury (Hg(0)), divalent mercury (Hg(II)), and primary particulate mercury (Hg(P)) are calculated. A detailed 8 description of the model, including mercury emissions, gas and aqueous phase 9 10 chemistry, and dry and wet deposition is given in this study. Worldwide observations including extensive data in China have been collected for model evaluation. 11 Comparison results show that the model reasonably simulates the global mercury 12 budget and the spatial-temporal variation of surface mercury concentrations and 13 14 deposition. Overall, model predictions of annual total gaseous mercury (TGM) and wet deposition agree with observations within a factor of two, and within a factor of 15 five for oxidized mercury and dry deposition. The model performs significantly 16 better in North America and Europe than in East Asia. This can probably be 17 18 attributed to the large uncertainties in emission inventories, coarse model resolution and to the inconsistency between the simulation and observation periods in East Asia. 19 Compared to the global simulation, the nested simulation shows improved skill at 20 capturing the high spatial variability of surface Hg concentrations and deposition 21 over East Asia. In particular, the root mean square error (RMSE) of simulated Hg 22 23 wet deposition over East Asia is reduced by 24% in the nested simulation. Model sensitivity studies indicate that Chinese primary anthropogenic emissions account 24 for 30% and 62% of surface mercury concentrations and deposition over China, 25 respectively. Along the rim of the western Pacific, the contributions from Chinese 26 sources are 11% and 15.2% over the Korean Peninsula, 10.4% and 8.2% over 27 Southeast Asia, and 5.7% and 5.9% over Japan. But for North America, Europe and 28 West Asia, the contributions from China are all below 5%. 29

## 1 Introduction

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2 Since the Minamata Event in Japan in the 1960s (Harada, 1995), the toxicity of 3 mercury (Hg) on human health and the environment has caused widespread public concern. Hg is a persistent, bio-accumulated pollutant, and the only heavy metal that 4 can be transported globally in gaseous form (Schroeder and Munthe, 1998). As a 5 result, Hg has been listed as a priority pollutant by many countries and international 6 7 agencies. After a long struggle, the first global treaty (the Minamata Convention) aimed at reducing Hg emissions and releases, was adopted and signed by 92 countries 8 in 2013 (http://www.mercuryconvention.org/). This made an important advance 9 10 towards joint action to control global Hg pollution and has brought higher requirements for understanding global Hg source-receptor relationships, especially the 11 impacts of high regional emissions (e.g. from China and India) on global Hg levels. 12 However, besides the remaining uncertainties in emission estimates, poor 13 14 understanding of the chemical transformation of atmospheric mercury has made assessment of long-range transport very challenging (AMAP/UNEP, 2013). 15 Atmospheric mercury models are powerful tools to assess the fate and transport of 16 mercury in the atmosphere. A number of atmospheric mercury models have been 17 18 developed to investigate the emissions, transport, chemistry, deposition and source-receptor relationships of Hg at global and regional scales. Global models 19 include the GEOS-Chem model (Amos et al., 2012; Zhang et al., 2012), the CTM-Hg 20 model (Seigneur et al., 2004), the CAM-Chem-Hg model (Lei et al., 2013), the 21 ECHMERIT model (De Simone et al., 2014), the MSCE-Hg-Hem model (Travnikov 22 23 and Ilyin, 2009), the DEHM model (Christensen et al., 2004), and the GRAHM model (Dastoor and Durnford, 2014). Regional models include the CMAQ-Hg model (Bash, 24 2010), the STEM-Hg model (Pan et al., 2008), the CAMx-Hg model (ENVIRON, 25 2011) and the WRF-Chem-Hg model (Gencarelli et al., 2014). Application of these 26 models has greatly advanced our understanding of the global Hg cycle. However, 27 several model intercomparison studies (Ryaboshapko et al., 2007; Bullock et al., 2008; 28 29 Pirrone and Keating, 2010) have found that large uncertainties still exist in Hg models

- and there is much room for improvement, especially for simulation of reactive
- 2 gaseous mercury (RGM) and dry deposition.
- 3 Mercury is released to the atmosphere from both anthropogenic and natural sources.
- 4 Human activities have increased the amount of mercury cycling through the
- 5 atmosphere-ocean-terrestrial system by about a factor of three (Selin, 2009), although
- 6 anthropogenic sources are estimated to account for only 31% of total Hg emissions
- 7 (Pirrone et al., 2010). China has the world's largest Hg production, consumption and
- 8 emissions, and suffers the most serious Hg pollution (Jiang et al., 2006), but the
- 9 impacts of its anthropogenic emissions on global Hg levels are still unclear. Previous
- modeling studies mainly focused on long-range transport of mercury from Asia.
- Based on the GEOS-Chem model, about 7-20% of Hg deposition over the United
- 12 States (US) was found to originate from Asian anthropogenic sources, which was
- comparable to that from North American sources (Strode et al., 2008; Jaffe and Strode,
- 14 2008). Another modeling study using the CTM-Hg model with three emission
- scenarios indicated that Asian anthropogenic emissions accounted for 14–25% of Hg
- deposition over the US (Seigneur et al., 2004). Travnikov (2005) reported a
- contribution to Hg deposition from total Asian sources (including both anthropogenic
- and natural emissions) of 15% over Europe and 33% over the Artic. Corbitt et al.
- 19 (2011) further pointed out that Asian emissions are the largest contributors to
- anthropogenic deposition to all ocean basins and these contributions are expected to
- 21 further grow in the future. The above studies all treated Asian anthropogenic
- 22 emissions as a whole, and the effects of anthropogenic emissions from the world's
- largest single emitter (China) have not been explicitly assessed before. In addition,
- due to lack of observational data, little model validation has been conducted over East
- Asia (especially China) in these studies and this leads to greater uncertainty in the
- 26 conclusions. Fu et al. (2012) reviewed previous modeling studies and pointed out that
- 27 current model simulations tend to underestimate total gaseous mercury (TGM) and
- total particulate mercury (TPM) concentrations but overestimate reactive gaseous
- 29 mercury (RGM) concentrations in China. To improve Hg model skill in China, nested
- simulations with high horizontal resolution might be a good choice. Zhang et al. (2012)

- demonstrated that a nested-grid model can capture the variation of Hg wet deposition
- 2 over North America better than a global model. In this study, online nested Hg
- 3 simulation with flexible horizontal resolution was developed and evaluated.
- 4 Compared to traditional multi-scale modeling approach (using a global model to
- 5 provide initial and boundary conditions to a regional model) (Seigneur et al., 2001),
- 6 online nested method use the same physical and chemical parameterizations in the
- 7 global and nested domains which could avoid uncertainties induced by different
- 8 boundary conditions. Compared to offline nested method used in the GEOS-Chem
- 9 model (Zhang et al., 2012), online nested method can provide boundary conditions
- with higher time resolution (10 or 5 minutes) from the global domain to the nested
- domain. Hence, online nested simulation would potentially improve model
- 12 performance in regional scale.
- Therefore, a comprehensive evaluation and improvement of Hg model performance
- in China is needed to effectively reduce the uncertainties in Hg trans-boundary
- transport and a quantitative assessment of Chinese anthropogenic contribution to
- global Hg concentration and deposition levels is helpful to determine and fulfill the
- 17 Hg emission reduction tasks under the Minamata Convention.
- In this paper, we describe the development of a global nested atmospheric mercury
- transport model (GNAQPMS-Hg) incorporating the latest available physical and
- 20 chemical processes essential to the mercury life cycle. The spatial and temporal
- variability of Hg concentrations and deposition are comprehensively evaluated against
- 22 available worldwide observations, including extensive data from China. The impact of
- horizontal resolution ( $1^{\circ}x1^{\circ}$  in the global domain versus  $0.33^{\circ}x0.33^{\circ}$  in the nested
- domain) on model predictions over East Asia is examined. Finally, the trans-boundary
- 25 transport of Chinese primary anthropogenic Hg emissions is quantified using the
- 26 model.

# 2 Model description and setup

# 2.1 General description

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- 3 The atmospheric physics and chemistry component of GNAQPMS-Hg, with the
- 4 exception of the mercury module, is based on the Nested Air Quality Prediction
- 5 Modeling System (NAQPMS) (Wang et al., 2006), developed at the Institute of
- 6 Atmospheric Physics, Chinese Academy of Sciences. NAQPMS is a 3-D regional
- 7 Eulerian model which has been rigorously evaluated and widely applied to simulate
- 8 the chemical evolution and transport of ozone (Li et al., 2007; Tang et al., 2010), the
- 9 distribution and evolution of aerosol and acid rain over East Asia (Wang et al., 2002;
- Li et al., 2011; Li et al., 2012) and to provide operational air quality forecasts in mega
- cities such as Beijing, Shanghai and Guangzhou (Wang et al., 2010; Wu et al., 2012;
- Wang et al., 2009). GNAQPMS is the global version of NAQPMS and uses the same
- model framework and physical and chemical parameterization schemes.
- As a multi-scale model, GNAQPMS can simulate the transportation and formation
- of primary and secondary pollutants from urban to global scale using an online
- 16 nesting approach. It includes advection, diffusion and convection processes,
- 17 gas/aqueous/aerosol chemistry, and modules for dry and wet deposition. The
- 18 advection process is parameterized based on an accurate mass conservative,
- 19 peak-preserving algorithm provided by Walcek and Aleksic (1998). The gas phase
- 20 chemical mechanism is the CBM-Z mechanism (Zaveri and Peters, 1999), including
- 21 133 reactions for 53 species. The dry deposition module uses the parameterization of
- Wesely (1989). The wet deposition and aqueous-phase chemistry module is
- constructed based on a revised version of the RADM mechanism (Chang et al., 1987;
- Wang et al., 2002; Ge et al., 2014). A mercury module has been developed and
- coupled into the GNAQPMS model in this study, as described in Sect. 2.2 to 2.4.
- Hereafter, we call this new model GNAQPMS-Hg. Note that meteorology, emissions,
- 27 deposition and chemistry are self-consistent between the global and nested domains.

### 2.2 Mercury chemistry

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#### 2.2.1 Basic mechanism

- 3 In GNAQPMS-Hg, three forms of mercury are explicitly treated: elemental mercury
- 4 (Hg(0)), divalent mercury (Hg(II)), and primary particulate mercury (Hg(P)).
- 5 Transformations between these three forms include the gas phase oxidation of Hg(0)
- to Hg(II), the aqueous phase oxidation of Hg(0) to Hg(II), the aqueous phase
- 7 reduction of Hg(II) to Hg(0), the aqueous phase equilibria of Hg(II) species and the
- 8 aqueous phase adsorption of Hg(II) to PM. Fig. 1 depicts the mercury reaction
- 9 pathways both in the gas and aqueous phase while the detailed reactions and their rate
- constants are summarized in Table 1. In line with most global mercury models,
- 11 GNAQPMS-Hg does not include dynamic air-surface exchange during Mercury
- Depletion Events (MDEs) in Polar regions (Schroeder et al., 1998) due to lack of
- 13 fundamental data.
- In the gas phase, Hg(0) is oxidized to Hg(II) by  $O_3$ , OH, hydrogen peroxide  $(H_2O_2)$ ,
- 15 hydrogen chloride (HCl) and molecular chlorine (Cl<sub>2</sub>). The oxidized products of these
- five reactions are assumed to be in the gas phase. According to Lin et al. (2004), OH
- and O<sub>3</sub> are the dominant oxidants in the continental troposphere while Cl and Br
- dominate Hg(0) oxidation in the marine boundary layer and the upper troposphere. In
- the aqueous phase, Hg(0) is oxidized to Hg(II) by dissolved  $O_3$ , OH, and  $Cl_2$ , and
- Hg(II) can be reduced back to Hg(0) via reaction with  $HO_2$  and by the formation of
- sulfite complexes. In addition, adsorption of Hg(II) species on atmospheric particulate
- matter (PM) is simulated using an adsorption coefficient ( $K = 34 L g^{-1}$ ) recommended
- 23 by Seigneur et al. (1998).
- As shown in Table 1, the mercury chemistry requires the concentrations of several
- non-mercury species, among which O<sub>3</sub>, OH, HO<sub>2</sub>, H<sub>2</sub>O<sub>2</sub>, SO<sub>2</sub>, HCl and PM are
- simulated online with GNAQPMS-Hg. However, Cl<sub>2</sub> is not explicitly simulated, and a
- 27 typical vertical profile of Cl<sub>2</sub> concentrations is therefore prescribed. The Cl<sub>2</sub>
- concentrations are specified to be 100 ppt at the surface, 50 ppt aloft at night, 10 ppt
- during daytime over the oceans, and zero over land (Seigneur et al., 2001).

#### 2.2.2 Bromine oxidation

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2 In order to test the effect of bromine (Br) oxidation reactions on global Hg concentrations, five Br chemical reactions in the gas phase are added in addition to 3 the O<sub>3</sub>-OH oxidation mechanism. The detailed description of the Br chemical 4 reactions and their implementation in the model is shown in Section S1.1 in the 5 supplement. A model sensitivity experiment with additional Br oxidation reactions 6 was conducted and compared to the base case simulation with O<sub>3</sub>-OH oxidation 7 mechanism. Fig. S1 in the supplement shows the difference of surface TGM 8 concentrations resulting from introducing Br oxidation reactions. Decrease in TGM 9 10 concentrations is found in the whole globe. This is because additional Br chemistry transforms more Hg(0) into Hg(II), which subsequently enhances the deposition of 11 Hg(II), leading to the reduction of TGM concentrations. Larger TGM reduction is 12 found in the Northern Hemisphere than in the Southern Hemisphere. In general, the 13 change in TGM concentration is less than 0.2 ng m<sup>-3</sup> in most areas which indicates 14 that introducing Br chemistry seems to have little impact on overall TGM 15 magnitudes and patterns. These results are similar to Lei et al. (2013) which test the 16 impact of Br chemistry using the CAM-Chem-Hg model. Although adding the Br 17 18 chemistry does not significantly change the TGM pattern, but it may affect the gaseous Hg partitioning between Hg(0) and Hg(II), and hence may affect the global 19 Hg deposition patterns. More in-depth tests and analysis are needed to address these 20 impacts in the future. In the following sections, we still use the base case simulated 21 22 results without considering the possible effects of Br chemistry.

# 2.2.3 Gas-particle partitioning of Hg(II)

- Recent studies suggested that gas-particle partitioning of Hg(II) is an important
- 25 process affected global Hg concentrations and deposition (Amos et al., 2012). To test
- these effects, an empirical mechanism of gas-particle partitioning of Hg(II) was added
- to the GNAQPMS-Hg model. The detailed description of this mechanism and its
- implementation in the model is shown in Section S1.2 in the supplement. A model

- 1 sensitivity experiment with Hg(II) gas-particle partitioning module was conducted and
- 2 compared to the base case simulation with all Hg(II) existing in the gas phase. Figs.
- 3 S2-3 in the supplement show the change fraction of surface TGM concentrations and
- 4 oxidized Hg concentrations resulting from introducing the Hg(II) gas-particle
- 5 partitioning mechanism. As expected, TGM concentrations decrease while oxidized
- 6 Hg concentrations increase in the whole globe. The change fractions of TGM and
- 7 oxidized Hg concentrations are smaller than 0.1 over the middle latitude of the
- 8 Northern Hemisphere, and even smaller than 0.05 over China. Considering that the
- 9 base case simulation has overestimated oxidized Hg concentrations in most areas,
- introducing the mechanism of gas-particle partitioning of Hg(II) would further
- increase this model discrepancy. Therefore, we still use the base case simulated
- results without considering the possible effects of gas-particle partitioning of Hg(II) in
- the following sections.

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#### 2.3 Mercury deposition

- Deposition is the leading removal process of atmospheric mercury, and also a major
- cause of mercury contamination in soil and water. Studies have shown that both dry
- 17 and wet removal pathways are equally significant for the total deposition of mercury
- 18 (Pirrone and Keating, 2010; Lin et al., 2006).
- Dry deposition of Hg(0), Hg(II) and Hg(P) is accounted for in the GNAQPMS-Hg
- 20 model, and simulated with the Wesely (1989) resistance model, which considers the
- 21 effect of different land cover types and characterizes the diurnal variation of dry
- deposition velocities. The Henry's Law constant for Hg(0) is set to be 0.11 M atm<sup>-1</sup>
- 23 (Lin and Pehkonen, 1999) with a temperature factor of -4970 K (Clever et al., 1985),
- and the surface reactivity is set to zero. Hg(II) represents HgCl<sub>2</sub> and Hg(OH)<sub>2</sub>. Its
- 25 Henry's Law constant is assumed to be the same as HNO<sub>3</sub> because they have similar
- solubility (Bullock and Brehme, 2002). Like HNO<sub>3</sub>, Hg(II) has a strong tendency to
- 27 stick to surfaces and its dry deposition occurs readily, so the surface resistance for
- Hg(II) in the dry deposition scheme is set to zero. The Hg(P) dry deposition velocity
- 29 is set equal to that for sulfate, similar to that applied in the CMAQ-Hg and

1 STEM-Hg model (Bullock and Brehme, 2002; Pan et al., 2008). More detailed

2 description of the dry deposition scheme used in the model is given in Section S2.1

in the supplement. Model intercomparison studies demonstrate that there are still

4 very large uncertainties in Hg dry deposition estimates (Bullock et al., 2008), and

this can be ascribed to the wide range of treatments and physical parameters for dry

6 deposition used in different models.

The wet deposition of Hg includes in-cloud and below-cloud scavenging. In-cloud scavenging is dependent on cloud and rain water content, species solubility and chemical transformation in the liquid phase, while below-cloud scavenging depends mainly on total rainfall intensity and washout efficiency. Among the three forms of mercury, wet deposition of Hg(0) is minor compared to Hg(II) and Hg(P) due to its low solubility. Therefore, Hg(0) oxidation will enhance total Hg wet deposition. In the GNAQPMS-Hg model, wet deposition of Hg species is calculated through adapting the RADM mechanism. The physical properties (e.g. Henry's Law constant, surface reactivity, molecular diffusivity) used are the same as those in the dry deposition module. More detailed description of the wet deposition scheme used in the model is given in Section S2.2 in the supplement. Currently, the uncertainties of Hg wet deposition simulation are mainly from the assumptions made in the cloud scavenging process and the uncertainty associated with the precipitation fields (Seigneur et al., 2001; Lin et al., 2006).

#### 2.4 Mercury emissions

We include anthropogenic emissions, biomass burning emissions, geogenic emissions, land reemission and ocean emissions (including reemission) of Hg in the model. Emissions from artisanal mining and volcanoes are neglected due to lack of fundamental data. The former is estimated to be 400 Mg yr<sup>-1</sup>, and the latter 90 Mg yr<sup>-1</sup>, and they account for about 5% and 1% of global total Hg emissions (Pirrone et al., 2010). Note that biomass burning emissions, geogenic emissions, land and ocean emissions are all treated as Hg(0). Global Hg emissions in the model are compared to previous studies in Table 2, and their spatial distributions are given in Figs. S4-S6

1 in the supplement.

Anthropogenic emissions in 2000 are derived from the Arctic Monitoring and 2 Assessment Programme (AMAP) inventory (Pacyna et al., 2006; Wilson et al., 3 2006). This inventory has a horizontal resolution of 0.5°x0.5° and no seasonal 4 variation. Following Selin et al. (2008), we increase the Asian ( $0\sim60^{\circ}$ N,  $65\sim150^{\circ}$ E) 5 Hg(0) emissions in the AMAP inventory by 50% (about 300 Mg yr<sup>-1</sup>) to account for 6 the regional underestimation identified by Jaffe et al. (2005). The modified inventory 7 has a total emission of 2488 Mg yr<sup>-1</sup>, with Hg(0), Hg(II) and Hg(P) accounting for 8 63%, 29% and 8% respectively. The major source regions are Asia and Africa, 9 accounting for 59% (1480 Mg yr<sup>-1</sup>) and 16% (399 Mg yr<sup>-1</sup>), while Europe and North 10 America contribute only 7% and 6%. China has the largest emissions at country 11 level (about 785 Mg yr<sup>-1</sup>), contributing 53% and 32% to the Asian and global 12 anthropogenic Hg emissions, respectively. It is noted that the emissions over South 13 Africa in this inventory were reported to be flawed (AMAP/UNEP, 2008) and much 14 higher than reality. The effects of these flawed emissions on the simulated results 15 16 were assessed in Section S3 in the supplement. Biomass burning emissions are specified by mapping an annual mean value of 675 17 Mg yr<sup>-1</sup> (Friedli et al., 2009) to the spatial and temporal distribution of CO biomass 18 burning emissions from the IPCC-AR5 (Intergovernmental Panel on Climate Change 19 20 Fifth Assessment Report) emissions inventory (Lamarque et al., 2010). The regional and monthly emission amounts are prescribed based on Friedli et al. (2009). A 21 similar method has been used by Jung et al. (2009). 22 The geogenic emissions here represent mobilization of Hg by degassing from 23 geological reservoirs. Following Selin et al. (2007), we consider a geogenic source 24 of 500 Mg yr<sup>-1</sup> distributed according to the locations of Hg mines (Frank, 1999) as an 25 indicator of Hg deposits. No temporal variation is applied to the geogenic emissions. 26 Land and ocean emissions are not dynamically calculated in the model due to the 27 large uncertainties associated with current parameterizations. Consistent with several 28 29 previous studies (Selin et al., 2007; Seigneur et al., 2001), the global annual land reemission of Hg is assumed to be 1500 Mg yr<sup>-1</sup>. The biogenic CO emissions from 30

- the Global Emission InitiAtive (GEIA) inventory (Guenther et al., 2006) are used as
- 2 spatial and temporal surrogates to map the land reemission. Regional emission totals
- 3 from different latitude zones and land uses are prescribed based on Mason (2009).
- 4 Ocean emissions in our model are specified as 5000 Mg yr<sup>-1</sup> (including
- 5 reemission), close to the estimates of Selin et al. (2008). Similarly, ocean emissions
- 6 are mapped according to the distribution of ocean biogenic CO emissions from the
- 7 Precursors of Ozone and their Effects in the Troposphere (POET) inventory (Granier
- 8 et al., 2005). Additionally, ocean emissions are adjusted to reflect several
- 9 distribution characteristics: 1) ocean emissions are high in summer but low in winter
- 10 (Strode et al., 2007), 2) ocean emissions are largest in the Tropics and downwind of
- industrial regions (Strode et al., 2007; Soerensen et al., 2010b), and 3) ocean
- emissions are large at mid and high latitudes in the Southern Hemisphere due to high
- wind speeds (Selin et al., 2008).

### 2.5 Model setup

- 15 Two nested domains covering the whole globe and East Asia are configured in this
- study. The horizontal resolutions are 1°x1° and 0.33°x0.33°, respectively. Vertically,
- the model uses 20 terrain-following layers from the surface to 20 km a.s.l., with a
- decreasing resolution with height. Roughly, the lowest 14-18 layers are in the
- 19 troposphere and the remaining layers are in the stratosphere. The time step in the
- 20 model calculation is 600 s. The input/output frequency is 6h in the global domain but
- 3h in the nested domain. The meteorological fields are provided by the global
- version of the Weather Research and Forecasting (WRF) model. The atmospheric
- 23 lifetime of Hg(0) is 0.5-2 year (Schroeder and Munthe, 1998), and so to ensure
- 24 mixing through the global troposphere and approach steady-state, we conduct the
- simulation for a 4-year period, with the first 3 years used for initialization and the
- last year (2001) used for analyses.
- Emissions of reactive gases and aerosols used in this study are from several
- databases: 1) the IPCC-AR5 anthropogenic and biomass burning emissions for 2000
- 29 (Lamarque et al., 2010); 2) the GEIA biogenic emissions for 2000 (Guenther et al.,

- 1 2006) and lightning emissions of nitric oxide (NO<sub>x</sub>) for 1983–1990 (Price et al.,
- 2 1997); 3) the POET ocean emissions of volatile organic compounds (VOCs) for
- 3 2000 (Granier et al., 2005); 4) the soil  $NO_x$  emissions for 2001 from Yan et al. (2005).
- 4 All emissions are interpolated and remapped to match the model grids of the global
- 5 and nested domains. The initial and top boundary conditions for O<sub>3</sub>, NO<sub>x</sub>, and CO
- 6 are taken from a global chemical transport model (MOZART-V2.4) with 2.8°
- 7 resolution (Horowitz et al., 2003).
- 8 Two model simulations, with and without Chinese primary anthropogenic Hg
- 9 emissions, are carried out in this study. The differences between the two simulations
- are attributed to the influence of Chinese primary anthropogenic Hg emissions.

## 3 Model evaluation

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#### 3.1 Observational data

- 13 Compared to reactive gases and aerosols, atmospheric Hg measurements are still
- 14 quite sparse. Routine monitoring networks for atmospheric Hg concentrations and
- deposition have only been established in Europe and North America. Lack of Hg
- observational data is a great restriction against advancing our understanding of
- 17 global Hg cycling and improving our skill in modeling. There is an urgent need to
- 18 establish a coordinated global Hg monitoring network for current Hg study
- 19 (Sprovieri et al., 2010; Keeler et al., 2009).
- The observational dataset in this study is based partly on the database shared by the
- 21 GEOS-Chem Hg modeling group (public access athttps://github.com/noelleselin/
- HgBenchmark; Selin et al., 2007; Selin et al., 2008; Holmes et al., 2010). This is
- supplemented with scattered Hg observations across East Asia collected from the
- literature. The observations used in this study are summarized as follows: 1)
- 25 long-term TGM/GEM (gaseous elemental mercury) measurements at 51 land sites,
- with 49 in the Northern Hemisphere and 2 in the Southern Hemisphere; 2) long-term
- 27 RGM/TPM measurements at 26 land sites, all in the Northern Hemisphere; 3)
- short-term Hg species measurements from 6 ship cruises; 4) wet deposition

- 1 measurements from the MDN (the Mercury Deposition Network in North America,
- 2 http://nadp.sws.uiuc.edu/nadpdata/mdnalldata.asp) and EMEP (the European
- 3 Monitoring and Evaluation Programme,
- 4 http://www.nilu.no/projects/ccc/emepdata.html) monitoring networks, with 51 and 8
- 5 sites respectively; 5) dry and wet deposition measurements at 19 sites in East Asia.
- 6 Further information about the measurement sites and data sources is given in Tables
- 7 S2-S5 in the supplement. It should be noted that the time periods of the measurements
- 8 do not all match with those of the simulation, and this difference may partially explain
- 9 any model–observation discrepancies. The influence of the mismatch of time periods
- when comparing the simulated results with the observations was qualitatively
- analyzed and shown in Section S4.1 in the supplement.

### 3.2 Global mercury budget

- 13 Fig. 2 gives the global mercury budget in GNAQPMS-Hg, including the cycling
- among atmosphere, ocean and land. The total atmospheric burden of Hg is 5546 Mg,
- with Hg(0), Hg(II), and Hg(P) contributing 90%, 9% and 1%, respectively. Therefore,
- mercury in the atmosphere exists mainly as Hg(0). Total emissions and deposition of
- Hg are 5163 Mg yr<sup>-1</sup> and 2866 Mg yr<sup>-1</sup> over land (a net source), and are 5000 Mg yr<sup>-1</sup>
- and 7297 Mg yr<sup>-1</sup> over ocean (a net sink), indicating that Hg is transported from land
- to ocean. For total deposition of Hg species, Hg(0) and Hg(II)/Hg(P) account for 38%
- and 62% over the earth's surface. Over land, deposition of Hg(II)/Hg(P) is more
- prominent than that of Hg(0), while they are both important over the ocean. Our
- results for total Hg deposition over ocean and Hg(II)/Hg(P) deposition over land are
- very close to that of GEOS-Chem (Selin et al., 2008). However, Hg(0) deposition
- over land derived from GNAQPMS-Hg is much smaller. This may be due to the
- lower reactivity coefficient used in the dry deposition module in GNAQPMS-Hg
- 26 (zero in GNAQPMS-Hg but 10<sup>-5</sup> in GEOS-Chem), which produces a lower dry
- 27 deposition velocity for Hg(0).
- Table 2 compares the GNAQPMS-Hg TGM budget and lifetime to those from
- 29 previous modeling studies. The TGM sources, sinks, burden and lifetime estimated

- 1 from GNAQPMS-Hg are all in the range determined by previous studies. Taking the
- 2 TGM lifetime as an example, the reported range is 0.5-1.7 years and it is 0.54 years
- 3 for GNAPQMS-Hg. In addition, similar to the results of GEOS-Chem (Selin et al.,
- 4 2007) and CAM-Chem-Hg (Lei et al., 2013), Hg dry deposition in GNAQPMS-Hg
- 5 dominates globally over wet deposition. Dry and wet deposition account for 78%
- 6 and 22%, respectively.

### 7 **3.3 Total gaseous mercury (TGM)**

As shown in Fig. 3, the main characteristics of the spatial distribution of TGM are 8 captured well by the model. High surface TGM concentrations are found in or 9 downwind of areas with intensive mercury-relative mining (e.g. Western USA) and 10 rapid industrialization (e.g. East Asia). In particular, TGM concentrations even 11 exceed 3 ng m<sup>-3</sup> in eastern China. Both model simulation and observations show a 12 significant surface interhemispheric gradient in TGM (Fig. 3 and Fig. 4). Based on 13 background observations, Lindberg et al. (2007) reported that mean Hg(0) 14 concentrations were 1.5-1.7 ng m<sup>-3</sup> in the Northern Hemisphere and 1.1-1.7 ng m<sup>-3</sup> 15 in the Southern Hemisphere. Lamborg et al. (2002) also estimated the range of 16 north-south interhemispheric TGM concentration ratios for surface air as 1.2–1.8. 17 Our model results share a general similarity with these studies. In GNAQPMS-Hg, 18 surface mean TGM concentrations in the Northern and Southern Hemisphere are 19 1.56 and 1.23 ng m<sup>-3</sup>, and the derived interhemispheric ratio is 1.27. However, it 20 21 should be noted that GNAQPMS-Hg is systematically biased low relative to cruise observations in the Northern Hemisphere, which leads to underestimation of the 22 23 TGM interhemispheric ratio compared with the range (1.49±0.12) reported by Temme et al. (2003) based on observations from several Atlantic cruises. This 24 25 disagreement was also found by several previous modeling studies (Seigneur et al., 2004; Selin et al., 2007), and can be attribute to the inability of current models to 26 reproduce the air-sea exchange of Hg reasonably (Soerensen et al., 2010a). More 27 specifically, this discrepancy is due to upwelling mercury from the sub-surface 28 ocean, possibly reflecting the legacy of past anthropogenic emissions (Holmes et al., 29

- 1 2010), and has been partially demonstrated by Soerensen et al. (2012). In general,
- 2 the simulated TGM concentrations match observations within a factor of two (Fig.
- 3 10). The correlation coefficient (R) and normalized mean bias (NMB) between
- 4 model results and observations from 51 land sites are 0.7 and -18%, respectively
- 5 (Table 3).
- Fig. 5 illustrates the mean seasonal variations of surface TGM concentrations in
- 7 North America, Europe, East Asia, the Arctic, the Antarctic (Neumayer) and South
- 8 Africa (Cape Point). In northern mid-latitudes, TGM concentrations are high in
- 9 winter but low in summer. This seasonality can be reproduced well by
- 10 GNAQPMS-Hg. The summer low is caused by high OH concentrations and frequent
- precipitation (Bergan and Rodhe, 2001). Compared with observations, the simulated
- 12 TGM monthly variations are stronger in North America but weaker in East Asia. The
- site by site comparisons in East Asia are shown in Fig. S9 in the supplement. We can
- see that nested simulation can well improve model performance in simulated TGM
- monthly variation in East Asia. At Arctic and Antarctic sites, TGM shows a spring
- minimum driven by MDEs and a summer maximum driven by reemission from the
- 17 snowpack (Steffen et al., 2005). The summer maximum is captured by
- 18 GNAQPMS-Hg because high reemission in polar summer has been taken into
- 19 account in our land reemission inventories. However, due to missing halogen
- 20 chemistry, the model fails to reproduce the spring minimum. At Cape Point, both
- observed and simulated TGM show little seasonal variation. However, simulated
- 22 monthly TGM concentrations are systematically biased high (NMB is 87%), which
- can be attributed to the flawed anthropogenic emissions in the AMAP emission
- inventories over South Africa (AMAP/UNEP, 2008). By updating the anthropogenic
- emissions over South Africa, the simulated TGM concentrations at Cape Point
- decrease from 1.77 ng m<sup>-3</sup> to 1.23 ng m<sup>-3</sup>, more close to the observed values (See
- 27 Section S3 in the supplement).
- Additional evaluation and analyses of simulated diurnal and vertical variation of
- TGM concentrations are given in Section S4.2 in the supplement.

### 3.4 Oxidized mercury

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2 Fig. 3 also shows the global distribution of oxidized mercury (defined as the sum of RGM+TPM in the observations and Hg(II)+Hg(P) in the model). Similar to TGM, a 3 pronounced north-south interhemispheric gradient is found for surface 4 concentrations of oxidized mercury, which is consistent with the global distribution 5 of emissions. Both model simulation and observations indicate that oxidized 6 mercury concentrations are much higher in East Asia than North America and 7 Europe. Compared to scarce available observations, oxidized mercury concentrations 8 are overestimated by GNAQPMS-Hg in most parts of the world (except East Asia). 9 10 This discrepancy may partially be attributed to excessive oxidation of Hg(0) by relatively high concentrations of OH and O<sub>3</sub> (especially over the ocean) and 11 uncertainties concerning Hg chemical speciation in emission inventories. The 12 simulated tropospheric mean OH concentration is 1.41x10<sup>6</sup> molec cm<sup>-3</sup>. This is at the 13 high end of the concentration range (0.65-1.56x10<sup>6</sup> molec cm<sup>-3</sup>) summarized by 14 Lawrence et al. (2001) and is about 27% higher than the ensemble mean 15 (11.1±1.8x10<sup>5</sup> molec cm<sup>-3</sup>) of the Atmospheric Chemistry and Climate Model 16 Intercomparison Project (ACCMIP) models (Voulgarakis et al., 2013). The simulated 17 18 mean surface O<sub>3</sub> in the North Pacific and North Atlantic is overestimated by 27% and 34% compared to observations from the WDCGG (World Data Centre for 19 Greenhouse Gases) network, although concentrations over land are reproduced 20 relatively well (see Figs. S10-S11 in the supplement). Besides, uncertainties of Hg 21 chemistry (e.g. gas-particle partitioning of RGM, in-plume reduction of RGM) and 22 deposition processes in the present model might also contribute to this discrepancy. 23 Overall, the simulated oxidized mercury concentrations agree with observations 24 within a factor of five (Fig. 10). The statistical indicators, R and NMB, calculated 25 from 26 land sites are 0.53 and 3% (Table 3), respectively. 26

#### 3.5 Wet deposition

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Wet deposition is mainly determined by the distribution of precipitation and Hg

- 1 concentrations. Fig. 6 and Fig. 7 evaluate the simulated annual Hg wet deposition
- 2 and accumulated precipitation over North America, Europe and East Asia. In general,
- 3 GNAQPMS-Hg reproduces the spatial patterns of Hg wet deposition relatively well.
- 4 Over North America, the maximum wet deposition occurs in the southeast,
- 5 corresponding to high OH concentrations and frequent precipitation there, while less
- 6 wet deposition occurs in the west and north, where there is much less precipitation.
- 7 GNAQPMS-Hg predicts the magnitude of mean wet deposition within 5% and
- 8 shows a good spatial correlation (R=0.76) (Table 3). These results are similar to
- 9 those of GEOS-Chem (Selin et al., 2007). However, it should be also noted that
- precipitation in the southeast is slightly overestimated by the model.
- Over Europe, model performance for wet deposition and precipitation are better
- than over North America and East Asia. High spatial correlation between the
- simulated and observed results are found for both wet deposition (R=0.78) and
- precipitation (R=0.86), and the NMBs are both less than 5% (Table 3).
- Over East Asia, Hg wet deposition is not only related to the precipitation pattern
- but also the local Hg emissions, especially in the southwest and Jilin province of
- 17 China, and in Central Japan. Model performance for wet deposition over East Asia is
- poorer than over Europe and North America. Although the spatial distribution and
- magnitude of precipitation over East Asia are seemingly well reproduced (R=0.64
- and NMB=-6%), a large underestimation (NMB=-61%) of wet deposition is found
- 21 here. Specifically, this is because the model fails to capture the high wet deposition
- 22 at certain sites. For example, the observed wet deposition over Shanghai and
- 23 Changchun are 251 and 108 μg m<sup>-2</sup> yr<sup>-1</sup> while the corresponding simulated values are
- only 25 and 13 μg m<sup>-2</sup> yr<sup>-1</sup>. This suggests that it is hard for models with coarse
- 25 horizontal resolution to characterize the high local mercury pollution in China. The
- 26 difference between the simulated and observed time periods and uncertainties in the
- 27 emission inventories may also contribute to these discrepancies.
- Fig. 8 further compares the simulated seasonal cycle of wet deposition with
- 29 measurements at MDN sites over North America and EMEP sites over Europe. No
- 30 monthly wet deposition observations are available over East Asia. Wet deposition

- and precipitation share similar monthly variations, with high values in summer and
- 2 autumn and low values in winter, as shown by both observations and simulation. In
- 3 summer and autumn, the variation in wet deposition and precipitation among sites is
- 4 larger than for other seasons, and this is evident from the greater variability in Fig. 8.
- 5 GNAQPMS-Hg tends to overestimate wet deposition and precipitation in July and
- 6 August over North America.

#### 3.6 Dry deposition

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- 8 Due to limited observations, only Hg dry deposition over East Asia is evaluated in
- 9 this study. It should be noted that data (Table S5 in the supplement) used to evaluate
- model simulation of dry deposition is not directly measured, but is inferred or
- estimated based on measurements of total Hg in through fall and rainwater, wet
- deposition and atmospheric concentrations. Associated with local Hg emissions,
- high dry deposition mainly occurs over central eastern China and central Japan (Fig.
- 14 9). The modeled dry deposition has a good spatial correlation with observations
- 15 (r=0.81), but there is a substantial negative bias (NMB=-42%, Table 3). In general,
- the simulated dry deposition agrees with observations within a factor of five (Fig.
- 17 10). Over Japan, the model results are biased high by a factor of 2-5, which may be
- 18 caused by overestimation of Hg(II) and Hg(P) emissions and missing model
- mechanism to deal with fast in-plume reduction of Hg(II) (Vijayaraghavan et al.,
- 20 2008; Amos et al., 2012; Zhang et al., 2012). Taking Tokyo as an example, observed
- Hg(P) is only 98 pg m<sup>-3</sup> while the simulated value is as high as 648 pg m<sup>-3</sup>. Modeling
- studies conducted by Pan et al. (2008) using the STEM-Hg model also found large
- 23 overestimation in dry deposition over Japan. Conversely, the model results are biased
- low by a factor of 2-5 over China, which indicates probable underestimation of
- 25 Chinese Hg emissions.

### 26 3.7 Model performance summary and comparison

- 27 In this section, we summarize the statistical performance of GNAQPMS-Hg for
- 28 TGM, oxidized mercury, and wet and dry deposition, compare the model

- 1 performance over East Asia, North America and Europe, and assess the effects of
- 2 horizontal resolution on model predictions over East Asia. As shown in Fig. 10, the
- 3 simulated TGM and wet deposition are within a factor of two of the corresponding
- 4 observations and within a factor of five for oxidized mercury and dry deposition.
- 5 The statistical performance of GNAQPMS-Hg is comparable with that of other
- 6 state-of-the-art Hg models (Bullock et al., 2008; Ryaboshapko et al., 2007; Pirrone
- 7 and Keating, 2010).

### 8 3.7.1 East Asia versus North America and Europe

- As illustrated in Table 3, the model statistical performance for all Hg parameters in North America and Europe is better than in East Asia. For example, the RMSEs between simulated and observed TGM over North America and Europe are 0.58 and 0.17 ng m<sup>-3</sup> but up to 3.61 ng m<sup>-3</sup> over East Asia. The poor model performance over
- East Asia is probably caused by the following reasons. Firstly, there are differences
- between simulated and observed data periods. Hg measurements over East Asia
- 15 (especially China) are mainly taken from recent years, and the observed values are
- 16 higher than in year 2001, which may lead to model underestimation. Hg
- anthropogenic emissions in China had increased by 164% during 1992-2007 (Liang
- et al., 2013) is an evidence. Secondly, there is a much higher spatial variation ratio
- 19 (SVR, see Table 3) for Hg parameters in East Asia than North America and Europe.
- 20 This implies that there are very intense spatial variations in surface Hg
- 21 concentrations and deposition over East Asia which cannot be resolved at the coarse
- 22 horizontal resolution used in global models (see Section 3.7.2). Thirdly, there are
- large uncertainties in emission inventories over East Asia. Large underestimations in
- 24 Hg anthropogenic emissions over East Asia have been demonstrated in several
- previous studies (Jaffe et al., 2005; Pan et al., 2007; Friedli et al., 2004; Song et al.,
- 26 2015). This is consistent with the simulated results in this study. Except the above
- 27 factors, missing of some chemical and physical processes (e.g. gas-particle
- partitioning of Hg(II), in-plume reduction of Hg(II), dynamic land reemission) in the
- 29 present model might also contribute to the poor model performance over East Asia.

#### 3.7.2 Global versus nested simulations

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In order to assess the impact of resolution on model predictions, an online nested simulation with higher resolution (0.33°x0.33°) over East Asia was conducted and compared to the global simulation with lower resolution (1°x1°). Emissions, meteorology, deposition and chemistry are self-consistent between the global and nested domains. The nested simulation uses higher resolution model inputs (e.g. topography, meteorology, emissions) and thus has the potential to better resolve high spatial variability of Hg concentrations and deposition in regional and local scales. Fig. 7 and Fig. 9 compare the spatial distributions of simulated annual mercury wet deposition, accumulated precipitation and dry deposition over East Asia between the global and nested simulations. Although the global and nested simulations predict similar large scale patterns for Hg deposition, the nested simulation resolves many fine features which are lost in the global simulation by horizontal averaging. Firstly, in the nested domain, high deposition fluxes become more concentrated in regions with large emissions or precipitation resulting in higher spatial variability in deposition. Secondly, the nested simulation reveals elevated wet deposition in southwest China due to frequent orographic and convective precipitation. Finally, the nested simulation shows a more detailed land/ocean contrast in deposition over coastal regions. For example, over the coastal regions of southeast China and Japan, wet deposition increases due to scavenging of local emissions and enhanced precipitation (Fig. 7) while dry deposition decreases associated with the lower dry deposition velocity of Hg(0) over land than over ocean (Fig. 9). Our results are similar to those of Zhang et al. (2012) who conducted a nested simulation of Hg over North America using the GEOS-Chem model. More comparisons about the differences of dry and wet deposition and Hg budgets over East Asia between the two simulations are given in Fig. S16 and Table S6 in the supplement. Fig. 11 and Table 3 further quantitatively compare the model performance over East Asia between the global and nested domains. In the Taylor diagram (Taylor, 2001), the position of each circle (or square) quantifies how closely the simulated

- 1 results match observations. We can see that the simulated precipitation, oxidized Hg,
- 2 wet and dry deposition agree better with observations in the nested domain than in
- 3 the global domain (Fig. 11). The largest improvement is found in the simulated wet
- 4 deposition. Specifically, the statistical parameter R for simulated wet deposition
- 5 increases from 0.36 to 0.78, the NMB decreases from -61% to -28%, and the RMSE
- 6 decreases by 24% (from 60.1 to 45.5 μg m<sup>-2</sup> yr<sup>-1</sup>) (Table 3). But for TGM, oxidized
- 7 Hg and dry deposition, the statistical parameters do not change significantly. For
- 8 example, the RMSEs of simulated oxidized Hg and dry deposition decrease by 7%
- 9 and 2% respectively, but increase by 7% for simulated TGM.

#### 3.7.3 Online versus offline nested simulations

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In order to further justify the online nested method, several model sensitivity experiments were conducted and the simulated results were compared to observations over East Asia. In the base simulation, the online nested method was used and the nested domain get boundary conditions from the global domain every 10 minutes. In the sensitivity simulations, the offline method was used and the boundary conditions were applied at 3 or 6 hours. The differences between the base and the sensitivity simulations were mainly caused by different frequency of boundary conditions. As shown in Table 4, the model performances are similar between the two sensitivity simulations (with 3 and 6 hour offline boundary conditions respectively). While the model performances in the base simulation are slightly better than those in the sensitivity simulations. The largest improvement is found in the simulated oxidized mercury. Specifically, the statistical parameter R for simulated oxidized mercury increases from 0.41 to 0.45, the NMB decreases from -18% to -12%, and the RMSE decreases by 4% compared to the simulated results with 6 hour offline boundary conditions. For wet deposition, little changes are found. This is because Hg wet deposition is not only affected by air concentrations but also precipitation. These results confirmed the effectiveness of the online nested method.

# 4 Impacts of Chinese primary anthropogenic sources on global Hg

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Fig. 12 shows the contribution of Chinese primary anthropogenic sources (not 3 including reemission) to annual mercury surface concentrations and total deposition 4 in the Northern Hemisphere, and Fig. 13 gives the corresponding mean percentage 5 contributions over different world regions (defined in Fig. S17 in the supplement), as 6 7 derived from a sensitivity simulation with Chinese anthropogenic emissions shut off. In general, the largest percentage contribution is found in China itself, followed by 8 neighbouring regions like the Korean Peninsula, Southeast Asia, Mongolia and Japan, 9 but they are relatively small in other regions. Specifically, domestic anthropogenic 10 emissions contribute on average 0.6 ng m<sup>-3</sup> (ranging from below 0.1 to above 3.0) to 11 surface Hg concentrations and 18.4 µg m<sup>-2</sup> yr<sup>-1</sup> (ranging from below 2.0 to above 12 50.0) to total deposition in China. They account for about 30% and 62% on a 13 national basis, respectively. The domestic contribution to deposition consists mainly 14 of the deposition of directly emitted Hg(II) and Hg(P) near sources and deposition of 15 16 Hg(II) formed by oxidation of Chinese Hg(0). For neighboring regions, the Chinese anthropogenic contributions to surface Hg concentrations and deposition are also 17 large. For example, the percentage contributions are 11% (0.2-0.6 ng m<sup>-3</sup>) and 15.2% 18  $(8-20 \mu g m^{-2} vr^{-1})$  over the Korean Peninsula, 10.4% (0.1-0.6 ng m<sup>-3</sup>) and 8.2% (1-12) 19  $\mu g \ m^{-2} \ vr^{-1}$ ) over Southeast Asia, and 5.7% (0.1-0.4  $ng \ m^{-3}$ ) and 5.9% (2-15  $\mu g \ m^{-2}$ 20 vr<sup>-1</sup>) over Japan. For regions far away from China, the percentage contributions are 21 small. They are 4.2% (0.06-0.1 ng m<sup>-3</sup>) and 4.8% (0.5-4 µg m<sup>-2</sup> yr<sup>-1</sup>) over North 22 America, and 3.5% (below 0.08 ng  $m^{-3}$ ) and 3.0% (below 2.0  $\mu$ g  $m^{-2}$   $yr^{-1}$ ) over 23 Europe. The percentage contributions over North America determined from our 24 simulation are comparable with the modeling study of Lei et al. (2013). They 25 estimated that around 7% of TGM concentrations and 9% of total Hg deposition in 26 the United States resulted from transpacific transport of Asian anthropogenic 27 emissions. Given that about 53% of Asian anthropogenic Hg emissions are from 28

China, it is reasonable that our estimated contributions are a little smaller than those reported by Lei et al. (2013).

Finally, there are another two issues which need to be addressed. Firstly, the above 3 analysis mainly focuses on regional average contributions. However, the percentage 4 contributions vary geographically inside the region. As shown in Fig. 12, 5 contributions of domestic anthropogenic emissions to total deposition in Central 6 Eastern China can exceed 40 μg m<sup>-2</sup> yr<sup>-1</sup>, but they are below 5 μg m<sup>-2</sup> yr<sup>-1</sup> in Western 7 China. Similarly, previous studies have found that Asia emissions make a much 8 larger contribution to Hg deposition in the Western USA than in the Eastern USA 9 (Seigneur et al., 2004; Strode et al., 2008). Secondly, the contributions from 10 reemission of previously deposited anthropogenic Hg (treated as natural land or 11 ocean reemission in GNAQPMS-Hg) are not taken into account in this study. Of the 12 natural emissions, only one-third is considered not to be influenced by anthropogenic 13 activities at all (Jung et al., 2009). In addition, according to the modeling study of 14 Selin et al. (2008), 31% (including 22% primary and 9% recycled) of the deposition 15 16 over USA is from anthropogenic emissions outside of North America. When considering reemission of previously deposited anthropogenic Hg, this suggests that 17 the foreign anthropogenic contribution would increase by about 42% (from 22% to 18 31%). If we apply the same scaling factor to our attribution results, then the 19 20 estimated Chinese anthropogenic contributions to Hg deposition over North America would increase from 4.8% to 6.8%. Therefore, it is also important to consider the 21 reemission of previously deposited anthropogenic Hg. 22

## **5** Conclusions

- A global nested atmospheric mercury transport model including Hg emissions,
- 25 chemical transformation and deposition is introduced in this study. The treatment of
- Hg chemistry employs the  $O_3$ -OH oxidation and  $SO_3^{2-}$ -H $O_2$  reduction mechanisms.
- 27 The gas phase reactions of Hg are added to the CBM-Z mechanism, while the
- aqueous phase reactions and wet deposition of Hg are calculated through adapting

- the RADM mechanism. The Wesely (1989) resistance model is used to deal with Hg
- 2 dry deposition. The same meteorological fields, emissions, chemical and physical
- 3 parameterizations are used in the global and nested domains.
- The GNAQPMS-Hg model has a global mercury source of 10163 Mg yr<sup>-1</sup>,
- 5 including 2488 Mg yr<sup>-1</sup> primary anthropogenic emissions, 675 Mg yr<sup>-1</sup> biomass
- 6 burning emissions, 2000 Mg yr<sup>-1</sup> land emissions (of which 75% is reemission), and
- 7 5000 Mg yr<sup>-1</sup> from the ocean. Dynamic bidirectional air-surface exchange of Hg is
- 8 not included in the model. Instead, we simply apply static net emission fluxes to
- 9 account for natural sources (including reemission) of Hg, with total emission
- amounts determined based on published estimates.

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Based on existing routine monitoring networks (e.g. MDN, EMEP) and the published literature, global observations including surface Hg concentrations and deposition are collected for model evaluation. Compared with previous studies, many more observations over East Asia (especially China) are included in our dataset. Model evaluation shows that the spatial distribution and seasonal cycle of Hg concentrations and deposition can be reproduced reasonably well by GNAQPMS-Hg. Overall, the simulated annual TGM and wet deposition match observations within a factor of two, and within a factor of five for oxidized mercury and dry deposition. This performance is comparable with other state-of-the-art Hg models. Some model deficiencies have also been identified. GNAQPMS-Hg is systematically biased low relative to cruise observations in the Northern Hemisphere, due to poor representation of the air-sea exchange mechanism for Hg. GNAQPMS-Hg overestimates oxidized mercury concentrations in most parts of the world which may partially be caused by excessive oxidation of Hg(0) by relatively high concentrations of OH and O<sub>3</sub> and uncertainties associated with Hg chemical speciation in emission inventories. The model performs significantly better in North America and Europe than in East Asia. This can probably be attributed to the large uncertainties in emission inventories, coarse model resolution and inconsistency between the simulation and observation periods in East Asia. An online nested simulation with higher resolution (0.33°x0.33°) over East Asia was conducted to

examine the impact of horizontal resolution on model predictions. Relative to the 1 global simulation, the nested simulation can better resolve high spatial variability of 2 3 Hg concentrations and deposition over East Asia, can better capture features such as higher wet deposition due to orographic and convective precipitation, and land/ocean 4 contrast. Statistically, the RMSE of simulated wet deposition over East Asia is 5 reduced by 24% in the nested simulation. 6 To quantify the impacts of Chinese anthropogenic sources on global Hg levels, a 8 model sensitivity simulation was conducted with Chinese anthropogenic emissions shut off. The results show that these sources contribute 30% and 62% of surface 9 mercury concentrations and deposition over China. Outside of China, the largest 10 percentage contributions of 11% and 15.2% are found in the Korean Peninsula, 11 following by Southeast Asia (10.4% and 8.2%), Mongolia (6.1% and 8.6%), and 12 Japan (5.7% and 5.9%). For regions far away from China, the percentage 13 contributions are relatively small (e.g. 4.2% and 4.8% over North America; 3.5% and 14 3.0% over Europe). 15 16 To perfect the model, future improvements will be focused on the following aspects: 1) employing dynamic parameterizations for bidirectional air-surface (sea 17 and land) exchange of Hg (Selin et al., 2008; Bash, 2010; Strode et al., 2007) to 18 better reflect natural emissions (including reemission), 2) including fast in-plume 19 20 reduction of Hg(II) to better characterize Hg(II) distribution near large point sources (Amos et al., 2012), and 3) reducing uncertainties in the anthropogenic Hg emission 21 inventory, especially the Hg speciation profile. Finally, establishment of routine Hg 22

# Code availability

modeling studies in East Asia.

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Please contact Huansheng Chen (E-mail: chenhuansheng@mail.iap.ac.cn) to obtain the source code of GNAQPMS-Hg.

monitoring networks would be also very helpful for enhancing and improving

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# 8 References

- 9 AMAP/UNEP: Technical Background Report to the Global Atmospheric Mercury Assessment,
- Tech. rep., Arctic Monitoring and Assessment Programme / UNEP Chemicals Branch, http:
- 11 //www.unep.org/hazardoussubstances/, 2008.
- 12 AMAP/UNEP: Technical Background Report for the Global Mercury Assessment 2013, Tech.
- rep., Arctic Monitoring and Assessment Programme AMAP and United Nations Environment
- Programme (UNEP) Chemicals Branch, http://www.unep.org/hazardoussubstances/
- 15 Mercury/Informationmaterials/ReportsandPublications/tabid/3593/Default.aspx, 2013.
- Amos, H. M., Jacob, D. J., Holmes, C. D., Fisher, J. A., Wang, Q., Yantosca, R. M., Corbitt, E. S.,
- Galarneau, E., Rutter, A. P., Gustin, M. S., Steffen, A., Schauer, J. J., Graydon, J. A., Louis, V.
- L. St., Talbot, R. W., Edgerton, E. S., Zhang, Y., and Sunderland, E. M.: Gas-particle
- 19 partitioning of atmospheric Hg(II) and its effect on global mercury deposition, Atmos. Chem.
- 20 Phys., 12, 591-603, doi:10.5194/acp-12-591-2012, 2012.
- 21 Ariya, P. A., Khalizov, A., and Gidas, A.: Reactions of gaseous mercury with atomic and
- 22 molecular halogens: Kinetics, product studies, and atmospheric implications, J. Phys. Chem. A,
- 23 106, 7310-7320, doi:10.1021/jp0207190, 2002.
- 24 Bash, J. O.: Description and initial simulation of a dynamic bidirectional air-surface exchange
- 25 model for mercury in Community Multiscale Air Quality (CMAQ) model, J. Geophys.
- 26 Res.-Atmos., 115, D06305, doi:10.1029/2009jd012834, 2010.
- 27 Bergan, T., and Rodhe, H.: Oxidation of elemental mercury in the atmosphere: constraints
- 28 imposed by global scale modelling, J. Atmos. Chem., 40, 191-212,
- 29 doi:10.1023/a:1011929927896, 2001.
- 30 Bullock, O. R., and Brehme, K. A.: Atmospheric mercury simulation using the CMAQ model:
- 31 formulation description and analysis of wet deposition results, Atmos. Environ., 36,
- 32 2135-2146, doi:10.1016/s1352-2310(02)00220-0, 2002.
- 33 Bullock, O. R., Jr., Atkinson, D., Braverman, T., Civerolo, K., Dastoor, A., Davignon, D., Ku,
- 34 J.-Y., Lohman, K., Myers, T. C., Park, R. J., Seigneur, C., Selin, N. E., Sistla, G., and
- Vijayaraghavan, K.: The North American Mercury Model Intercomparison Study (NAMMIS):
- 36 Study description and model-to-model comparisons, J. Geophys. Res.-Atmos., 113, D17310,
- 37 doi:10.1029/2008jd009803, 2008.

- 1 Chang, J. S., Brost, R. A., Isaksen, I. S. A., Madronich, S., Middleton, P., Stockwell, W. R., and
- 2 Walcek, C. J.: A three-dimensional Eulerian acid deposition model: Physical concepts and
- 3 formulation, J. Geophys. Res.-Atmos., 92, 14681-14700, doi:10.1029/JD092iD12p14681,
- 4 1987.
- 5 Christensen, J. H., Brandt, J., Frohn, L. M., and Skov, H.: Modelling of mercury in the Arctic
- 6 with the Danish Eulerian Hemispheric Model, Atmos. Chem. Phys., 4, 2251-2257, 2004.
- 7 Clever, H. L., Johnson, S. A., and Derrick, M. E.: The solubility of mercury and some sparingly
- 8 soluble mercury salts in water and aqueous-electrolyte solutions, J. Phys. Chem. Ref. Data, 14,
- 9 631-681, 1985.
- 10 Corbitt, E. S., Jacob, D. J., Holmes, C. D., Streets, D. G., and Sunderland, E. M.: Global
- source-receptor relationships for mercury deposition under present-day and 2050 emissions
- scenarios, Environ. Sci. Technol., 45, 10477-10484, doi:10.1021/es202496y, 2011.
- 13 Dastoor, A. P., and Durnford, D. A.: Arctic Ocean: Is It a Sink or a Source of Atmospheric
- 14 Mercury, Environ. Sci. Tehnol., 48(3), 1707-1717, 2014.
- De Simone, F., Gencarelli, C. N., Hedgecok, I. M., Pirrone, N.: Global atmospheric cycle of
- mercury: a model study on the impact of oxidation mechanisms, Environ. Sci. Pollut. Res., 21,
- 17 4110-4123, 2014.
- 18 ENVIRON: User's guide for Comprehensive Air Quality Model with Extensions Version 5.40,
- 19 ENVIRON International Corporation, Novato, California, 2011.
- 20 Frank, D. G.: Mineral Resource Data System (MRDS) data in Arc-View shape file format, for
- spatial data delivery project, U.S. Geol. Surv., Spokane, Wash., 1999.
- Friedli, H. R., Radke, L. F., Prescott, R., Li, P., Woo, J. H., and Carmichael, G. R.: Mercury in
- the atmosphere around Japan, Korea, and China as observed during the 2001 ACE-Asia field
- campaign: Measurements, distributions, sources, and implications, J. Geophys. Res.-Atmos.,
- 25 109, D19s25, doi:10.1029/2003jd004244, 2004.
- 26 Friedli, H. R., Arellano, A. F., Cinnirella, S., and Pirrone, N.: Initial estimates of mercury
- emissions to the atmosphere from global biomass burning, Environ. Sci. Technol., 43,
- 28 3507-3513, doi:10.1021/es802703g, 2009.
- 29 Fu, X., Feng, X., Sommar, J., and Wang, S.: A review of studies on atmospheric mercury in
- 30 China, Sci. Total Environ., 421, 73-81, doi:10.1016/j.scitotenv.2011.09.089, 2012.
- 31 Ge, B., Wang, Z., Xu, X., Wu, J., Yu, X., and Li, J.: Wet deposition of acidifying substances in
- 32 different regions of China and the rest of East Asia: Modeling with updated NAQPMS,
- 33 Environ. Pollut., 187, 10-21, doi:http://dx.doi.org/10.1016/j.envpol.2013.12.014, 2014.
- 34 Gencarelli, C. N., De Simone, F., Hedgecok, I. M., Sprovieri, F., Pirrone, N.: Development and
- application of a regional-scale atmospheric mercury model based on WRF/Chem: a
- Mediterranean area investigation, Environ. Sci. Pollut. Res., 21, 4095-4109, 2014.
- 37 Granier, C., Lamarque, J. F., Mieville, A., Muller, J. F., and Olivier, J.: POET, a database of
- 38 surface emissions of ozone precursors , tech. report, available at:
- 39 http://www.aero.jussieu.fr/projet/ACCENT/POET.php (last access: 10 June 2013), 2005.
- 40 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of
- 41 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
- 42 Aerosols from Nature), Atmos. Chem. Phys., 6, 3181-3210, 2006.
- 43 Hall, B., and Bloom., N.: Report to EPRI, Palo Alto, CA., USA,1993.
- 44 Hall, B.: The gas phase oxidation of elemental mercury by ozone, Water Air Soil Pollut., 80,

- 1 301-315, doi:10.1007/bf01189680, 1995.
- 2 Harada, M.: Minamata Disease Methylmercury poisoning in Japan caused by environmental
- 3 pollution, Crit. Rev. Toxicol., 25, 1-24, doi:10.3109/10408449509089885, 1995.
- 4 Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R., and Slemr, F.: Global
- 5 atmospheric model for mercury including oxidation by bromine atoms, Atmos. Chem. Phys.,
- 6 10, 12037-12057, doi:10.5194/acp-10-12037-2010, 2010.
- 7 Horowitz, L. W., Walters, S., Mauzerall, D. L., Emmons, L. K., Rasch, P. J., Granier, C., Tie, X.
- 8 X., Lamarque, J. F., Schultz, M. G., Tyndall, G. S., Orlando, J. J., and Brasseur, G. P.: A global
- 9 simulation of tropospheric ozone and related tracers: Description and evaluation of MOZART,
- version 2, J. Geophys. Res.-Atmos., 108, 4784, doi:10.1029/2002jd002853, 2003.
- 11 Jaffe, D., Prestbo, E., Swartzendruber, P., Weiss-Penzias, P., Kato, S., Takami, A., Hatakeyama,
- 12 S., and Kajii, Y.: Export of atmospheric mercury from Asia, Atmos. Environ., 39, 3029-3038,
- doi:10.1016/j.atmosenv.2005.01.030, 2005.
- 14 Jaffe, D., and Strode, S.: Sources, fate and transport of atmospheric mercury from Asia, Environ.
- 15 Chem., 5, 121-126, doi:10.1071/en08010, 2008.
- Jiang, G., Shi, J., and Feng, X.: Mercury pollution in China: An overview of the past and current
- sources of the toxic metal, Environ. Sci. Technol., 40, 3673-3678, 2006.
- 18 Jung, G., Hedgecock, I. M., and Pirrone, N.: ECHMERIT V1.0-a new global fully coupled
- mercury-chemistry and transport model, Geosci. Model Dev., 2, 175-195, 2009.
- 20 Keeler, G. J., Pirrone, N., Bullock, R., and Sillman, S.: The need for a coordinated global Hg
- 21 monitoring network for global and regional models validation, in: Mercury Fate and Transport
- in the Global Atmosphere, edited by: Mason, R., and Pirrone, N., Springer, USA, 391-424,
- 23 2009.
- Lamarque, J. F., Bond, T. C., Eyring, V., Granier, C., Heil, A., Klimont, Z., Lee, D., Liousse, C.,
- 25 Mieville, A., Owen, B., Schultz, M. G., Shindell, D., Smith, S. J., Stehfest, E., Van Aardenne,
- J., Cooper, O. R., Kainuma, M., Mahowald, N., McConnell, J. R., Naik, V., Riahi, K., and van
- Vuuren, D. P.: Historical (1850-2000) gridded anthropogenic and biomass burning emissions
- of reactive gases and aerosols: methodology and application, Atmos. Chem. Phys., 10,
- 29 7017-7039, doi:10.5194/acp-10-7017-2010, 2010.
- 30 Lamborg, C. H., Fitzgerald, W. F., O'Donnell, J., and Torgersen, T.: A non-steady-state
- 31 compartmental model of global-scale mercury biogeochemistry with interhemispheric
- 32 atmospheric gradients, Geochim. Cosmochim. Ac., 66, 1105-1118,
- 33 doi:10.1016/s0016-7037(01)00841-9, 2002.
- 34 Lawrence, M. G., Jöckel, P., and von Kuhlmann, R.: What does the global mean OH
- 35 concentration tell us?, Atmos. Chem. Phys., 1, 37-49, doi:10.5194/acp-1-37-2001, 2001.
- Lei, H., Liang, X., Wuebbles, D. J., and Tao, Z.: Model analyses of atmospheric mercury: present
- air quality and effects of transpacific transport on the United States, Atmos. Chem. Phys., 13,
- 38 10807-10825, doi:10.5194/acp-13-10807-2013, 2013.
- 39 Li, J., Wang, Z., Akimoto, H., Gao, C., Pochanart, P., and Wang, X.: Modeling study of ozone
- seasonal cycle in lower troposphere over east Asia, J. Geophys. Res.-Atmos., 112, D22s25,
- 41 doi:10.1029/2006jd008209, 2007.
- 42 Li, J., Wang, Z., Wang, X., Yamaji, K., Takigawa, M., Kanaya, Y., Pochanart, P., Liu, Y., Irie, H.,
- 43 Hu, B., Tanimoto, H., and Akimoto, H.: Impacts of aerosols on summertime tropospheric
- 44 photolysis frequencies and photochemistry over Central Eastern China, Atmos. Environ., 45,

- 1 1817-1829, doi:10.1016/j.atmosenv.2011.01.016, 2011.
- 2 Li, J., Wang, Z., Zhuang, G., Luo, G., Sun, Y., and Wang, Q.: Mixing of Asian mineral dust with
- anthropogenic pollutants over East Asia: a model case study of a super-duststorm in March
- 4 2010, Atmos. Chem. Phys., 12, 7591-7607, doi:10.5194/acp-12-7591-2012, 2012.
- 5 Liang, S., Xu, M., Liu, Z., Suh, S. and Zhang T.: Socioeconomic Drivers of Mercury Emissions
- 6 in China from 1992 to 2007, Environ. Sci. Technol., 47(7): 3234-3240, 2013.
- 7 Lin, C. J., and Pehkonen, S. O.: Aqueous free radical chemistry of mercury in the presence of
- 8 iron oxides and ambient aerosol, Atmos. Environ., 31, 4125-4137,
- 9 doi:http://dx.doi.org/10.1016/S1352-2310(97)00269-0, 1997.
- 10 Lin, C. J., and Pehkonen, S. O.: Oxidation of elemental mercury by aqueous chlorine
- 11 (HOCl/OCl<sup>-</sup>): Implications for tropospheric mercury chemistry, J. Geophys. Res.-Atmos., 103,
- 28093-28102, doi:10.1029/98jd02304, 1998.
- Lin, C. J., Pongprueksa, P., Lindberg, S. E., Pehkonen, S. O., Byun, D., and Jang, C.: Scientific
- uncertainties in atmospheric mercury models I: Model science evaluation, Atmos. Environ., 40,
- doi:2911-2928, 10.1016/j.atmosenv.2006.01.009, 2006.
- 16 Lin, C. J., and Pehkonen, S. O.: The chemistry of atmospheric mercury: a review, Atmos.
- 17 Environ., 33, 2067-2079, doi:10.1016/s1352-2310(98)00387-2, 1999.
- 18 Lin, C. J., Pongprueks, P., Ho, T. C., and Jang, C.: Development of mercury modeling schemes
- 19 within CMAQ framework: Science and model implementation issues, In: Proceedings of the
- 20 2004 CMAS Models-3 Conference, Research Triangle Park, NC, October 18-20 (CD-ROM).
- 21 2004.
- Lindberg, S., Bullock, R., Ebinghaus, R., Engstrom, D., Feng, X., Fitzgerald, W., Pirrone, N.,
- Prestbo, E., and Seigneur, C.: A synthesis of progress and uncertainties in attributing the
- sources of mercury in deposition, Ambio, 36, 19-32, 2007.
- Lindqvist, O., and Rodhe, H.: Atmospheric mercury A review, Tellus B, 37, 136-159, 1985.
- 26 Mason, R.: Mercury emissions from natural processes and their importance in the global mercury
- 27 cycle, in: Mercury Fate and Transport in the Global Atmosphere, edited by: Mason, R., and
- 28 Pirrone, N., Springer, USA, 173-191, 2009.
- 29 Munthe, J.: The aqueous oxidation of elemental mercury by ozone, Atmos. Environ., 26,
- 30 1461-1468, doi:10.1016/0960-1686(92)90131-4, 1992.
- 31 Pacyna, E. G., Pacyna, J. M., Steenhuisen, F., and Wilson, S.: Global anthropogenic mercury
- 32 emission inventory for 2000, Atmos. Environ., 40, 4048-4063,
- 33 doi:http://dx.doi.org/10.1016/j.atmosenv.2006.03.041, 2006.
- Pan, L., Chai, T., Carmichael, G. R., Tang, Y., Streets, D., Woo, J.-H., Friedli, H. R., and Radke,
- L. F.: Top-down estimate of mercury emissions in China using four-dimensional variational
- 36 data assimilation, Atmos. Environ., 41, 2804-2819,
- 37 doi:http://dx.doi.org/10.1016/j.atmosenv.2006.11.048, 2007.
- 38 Pan, L., Carmichael, G. R., Adhikary, B., Tang, Y., Streets, D., Woo, J.-H., Friedli, H. R., and
- Radke, L. F.: A regional analysis of the fate and transport of mercury in East Asia and an
- 40 assessment of major uncertainties, Atmos. Environ., 42, 1144-1159,
- 41 doi:10.1016/j.atmosenv.2007.10.045, 2008.
- 42 Pehkonen, S. O., and Lin, C. J.: Aqueous photochemistry of mercury with organic acids, J. Air
- 43 Waste Manage. Assoc., 48, 144-150, doi:10.1080/10473289.1998.10463661, 1998.
- 44 Pirrone, N., Cinnirella, S., Feng, X., Finkelman, R. B., Friedli, H. R., Leaner, J., Mason, R.,

- 1 Mukherjee, A. B., Stracher, G. B., Streets, D. G., and Telmer, K.: Global mercury emissions to
- the atmosphere from anthropogenic and natural sources, Atmos. Chem. Phys., 10, 5951-5964,
- 3 doi:10.5194/acp-10-5951-2010, 2010.
- 4 Pirrone, N., and Keating, T.: Hemispheric Transport of Air Pollution 2010 Part B: Mercury,
- 5 United Nations, New York and Geneva, 210 pp., 2010.
- 6 Price, C., Penner, J., and Prather, M.: NO<sub>x</sub> from lightning: 1. Global distribution based on
- 7 lightning physics, J. Geophys. Res.-Atmos., 102, 5929-5941, doi:10.1029/96jd03504, 1997.
- 8 Ryaboshapko, A., Bullock, O. R., Jr., Christensen, J., Cohen, M., Dastoor, A., Ilyin, I., Petersen,
- 9 G., Syrakov, D., Travnikov, O., Artz, R. S., Davignon, D., Draxler, R. R., Munthe, J., and
- 10 Pacyna, J.: Intercomparison study of atmospheric mercury models: 2. Modelling results vs.
- long-term observations and comparison of country deposition budgets, Sci. Total Environ.,
- 12 377, 319-333, doi:10.1016/j.scitotenv.2007.01.071, 2007.
- 13 Sanemasa, I.: The solubility of elemental mercury vapor in water, Bull. Chem. Soc. Jpn., 48,
- 14 1795-1798, 1975.
- 15 Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D. R., and
- Berg, T.: Arctic springtime depletion of mercury, Nature, 394, 331-332, doi:10.1038/28530,
- 17 1998.
- 18 Schroeder, W. H., and Munthe, J.: Atmospheric mercury An overview, Atmos. Environ., 32,
- 19 809-822, doi:10.1016/s1352-2310(97)00293-8, 1998.
- 20 Seigneur, C., Abeck, H., Chia, G., Reinhard, M., Bloom, N. S., Prestbo, E., and Saxena, P.:
- 21 Mercury adsorption to elemental carbon (soot) particles and atmospheric particulate matter,
- 22 Atmos. Environ., 32, 2649-2657, doi:http://dx.doi.org/10.1016/S1352-2310(97)00415-9, 1998.
- 23 Seigneur, C., Karamchandani, P., Lohman, K., Vijayaraghavan, K., and Shia, R. L.: Multiscale
- 24 modeling of the atmospheric fate and transport of mercury, J. Geophys. Res.-Atmos., 106,
- 25 27795-27809, doi:10.1029/2000jd000273, 2001.
- 26 Seigneur, C., Vijayaraghavan, K., Lohman, K., Karamchandani, P., and Scott, C.: Global source
- attribution for mercury deposition in the United States, Environ. Sci. Technol., 38, 555-569,
- doi:10.1021/es034109t, 2004.
- 29 Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Strode, S., Jaegle, L., and Jaffe, D.:
- 30 Chemical cycling and deposition of atmospheric mercury: Global constraints from
- 31 observations, J. Geophys. Res.-Atmos., 112, D02308, doi:10.1029/2006jd007450, 2007.
- 32 Selin, N. E., Jacob, D. J., Yantosca, R. M., Strode, S., Jaegle, L., and Sunderland, E. M.: Global
- 33 3-D land-ocean-atmosphere model for mercury: Present-day versus preindustrial cycles and
- 34 anthropogenic enrichment factors for deposition, Global Biogeochem. Cy., 22, Gb3099,
- 35 doi:10.1029/2008gb003282, 2008.
- 36 Selin, N. E.: Global biogeochemical cycling of mercury: A review, Annu. Rev. Environ. Resour.,
- 34, 43, 2009.
- 38 Sillen, L. G., Martell, A. E., and Bjerrum, J.: Stability constants of metal-ion complexes, Chem.
- 39 Soc., 17, 754,1964.
- 40 Soerensen, A. L., Skov, H., Jacob, D. J., Soerensen, B. T., and Johnson, M. S.: Global
- 41 concentrations of gaseous elemental mercury and reactive gaseous mercury in the marine
- 42 boundary layer, Environ. Sci. Technol., 44, 7425-7430, doi:10.1021/es903839n, 2010a.
- 43 Soerensen, A. L., Sunderland, E. M., Holmes, C. D., Jacob, D. J., Yantosca, R. M., Skov, H.,
- Christensen, J. H., Strode, S. A., and Mason, R. P.: An improved global model for air-sea

- 1 exchange of mercury: High concentrations over the North Atlantic, Environ. Sci. Technol., 44,
- 2 8574-8580, doi:10.1021/es102032g, 2010b.
- 3 Soerensen, A. L., Jacob, D. J., Streets, D., Witt, M., Ebinghaus, R., Mason, R. P., Andersson, M.
- and Sunderland E. M.,: Multi-decadal decline of mercury in the North Atlantic atmosphere
- 5 explained by changing subsurface seawater concentrations, Geophys. Res. Lett. 39: Art
- 6 #L21810, 2012.
- 7 Sommar, J., Gardfeldt, K., Stromberg, D., and Feng, X. B.: A kinetic study of the gas-phase
- 8 reaction between the hydroxyl radical and atomic mercury, Atmos. Environ., 35, 3049-3054,
- 9 doi:10.1016/s1352-2310(01)00108-x, 2001.
- 10 Song, S., Selin, N. E., Soerensen, A. L., Angot, H., Artz, R., Brooks, S., Brunke, E.-G., Conley,
- G., Dommergue, A., Ebinghaus, R., Holsen, T. M., Jaffe, D. A., Kang, S., Kelley, P., Luke, W.
- T., Magand, O., Marumoto, K., Pfaffhuber, K. A., Ren, X., Sheu, G.-R., Slemr, F., Warneke, T.,
- Weigelt, A., Weiss-Penzias, P., Wip, D. C., and Zhang, Q.: Top-down constraints on
- 14 atmospheric mercury emissions and implications for global biogeochemical cycling, Atmos.
- 15 Chem. Phys., 15, 7103-7125, doi:10.5194/acp-15-7103-2015, 2015.
- Sprovieri, F., Pirrone, N., Ebinghaus, R., Kock, H., and Dommergue, A.: A review of worldwide
- 17 atmospheric mercury measurements, Atmos. Chem. Phys., 10, 8245-8265,
- doi:10.5194/acp-10-8245-2010, 2010.
- 19 Steffen, A., Schroeder, W., Macdonald, R., Poissant, L., and Konopley, A.: Mercury in the Arctic
- atmosphere: An analysis of eight years of measurements of GEM at Alert (Canada) and a
- comparison with observations at Amderma (Russia) and Kuujjuarapik (Canada), Sci. Total
- 22 Environ., 342, 185-198, doi:10.1016/j.scitotenv.2004.12.048, 2005.
- Strode, S. A., Jaegle, L., Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Mason, R. P., and
- Slemr, F.: Air-sea exchange in the global mercury cycle, Global Biogeochem. Cy., 21, Gb1017,
- 25 doi:10.1029/2006gb002766, 2007.
- 26 Strode, S. A., Jaegle, L., Jaffe, D. A., Swartzendruber, P. C., Selin, N. E., Holmes, C., and
- 27 Yantosca, R. M.: Trans-Pacific transport of mercury, J. Geophys. Res.-Atmos., 113, D15305,
- 28 doi:10.1029/2007jd009428, 2008.
- 29 Tang, X., Wang, Z., Zhu, J., Gbaguidi, A. E., Wu, Q., Li, J., and Zhu, T.: Sensitivity of ozone to
- precursor emissions in urban Beijing with a Monte Carlo scheme, Atmos. Environ., 44,
- 31 3833-3842, doi:http://dx.doi.org/10.1016/j.atmosenv.2010.06.026, 2010.
- 32 Taylor, K. E.: Summarizing multiple aspects of model performance in a single diagram, J.
- 33 Geophys. Res.-Atmos., 106, 7183-7192, doi:10.1029/2000jd900719, 2001.
- 34 Temme, C., Slemr, F., Ebinghaus, R., and Einax, J. W.: Distribution of mercury over the Atlantic
- 35 Ocean in 1996 and 1999-2001, Atmos. Environ., 37, 1889-1897,
- 36 doi:10.1016/s1352-2310(03)00069-4, 2003.
- 37 Tokos, J. J. S., Hall, B., Calhoun, J. A., and Prestbo, E. M.: Homogeneous gas-phase reaction of
- Hg0 with  $H_2O_2$ ,  $O_3$ , CH3I, and (CH3)(2)S: Implications for atmospheric Hg cycling, Atmos.
- 39 Environ., 32, 823-827, doi:10.1016/s1352-2310(97)00171-4, 1998.
- 40 Travnikov, O.: Contribution of the intercontinental atmospheric transport to mercury pollution in
- 41 the Northern Hemisphere, Atmos. Environ., 39, 7541-7548,
- 42 doi:10.1016/j.atmosenv.2005.07.066, 2005.
- 43 Travnikov, O., and Ilyin, I.: The EMEP/MSC-E mercury modeling system, in: Mercury Fate and
- Transport in the Global Atmosphere, edited by: Mason, R., and Pirrone, N., Springer, USA,

- 1 571-587, 2009.
- 2 Van Loon, L., Mader, E., and Scott, S. L.: Reduction of the aqueous mercuric ion by sulfite: UV
- 3 spectrum of HgSO<sub>3</sub> and its intramolecular redox reaction, J. Phys. Chem. A, 104, 1621-1626,
- 4 doi:10.1021/jp994268s, 2000.
- 5 Van Loon, L. L., Mader, E. A., and Scott, S. L.: Sulfite stabilization and reduction of the aqueous
- 6 mercuric ion: Kinetic determination of sequential formation constants, J. Phys. Chem. A, 105,
- 7 3190-3195, doi:10.1021/jp003803h, 2001.
- 8 Vijayaraghavan, K., P. Karamchandani, C. Seigneur, R. Balmori, and S.-Y. Chen: Plume-in-grid
- 9 modeling of atmospheric mercury, J. Geophys. Res., 113, D24305,
- doi:10.1029/2008JD010580, 2008.
- Voulgarakis, A., Naik, V., Lamarque, J. F., Shindell, D. T., Young, P. J., Prather, M. J., Wild, O.,
- 12 Field, R. D., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins, W. J., Dalsøren, S. B.,
- Doherty, R. M., Eyring, V., Faluvegi, G., Folberth, G. A., Horowitz, L. W., Josse, B.,
- MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T., Stevenson, D. S.,
- 15 Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Analysis of present day and future OH and
- methane lifetime in the ACCMIP simulations, Atmos. Chem. Phys., 13, 2563-2587,
- doi:10.5194/acp-13-2563-2013, 2013.
- 18 Walcek, C. J., and Aleksic, N. M.: A simple but accurate mass conservative, peak-preserving,
- mixing ratio bounded advection algorithm with Fortran code, Atmos. Environ., 32, 3863-3880,
- 20 doi:10.1016/s1352-2310(98)00099-5, 1998.
- Wang, Q., Fu, Q., Wang, Z., Wang, T., Liu, P., Lu, T., Duan, Y., and Huang, Y.: Application of
- ensemble numerical model system on the air quality forecast in Shanghai (in Chinese),
- Environmental Monitoring and Forewarning, 2(4), 1-6+11, 2010.
- 24 Wang, Z., Akimoto, H., and Uno, I.: Neutralization of soil aerosol and its impact on the
- distribution of acid rain over east Asia: Observations and model results, J. Geophys.
- 26 Res.-Atmos., 107, 4389, doi:10.1029/2001jd001040, 2002.
- Wang, Z., Xie, F., Wang, X., An, J., and Zhu, J.: Development and application of Nested Air
- 28 Quality Prediction Modeling System (in Chinese), Chinese Journal of Atmospheric Sciences,
- 29 30(5), 778-790, 2006.
- Wang, Z., Wu, Q., Gbaguidi, A., Yan, P., Zhang, W., Wang, W., and Tang, X.: Ensemble air
- 31 quality multi-model forecast system for Beijing (EMS-Beijing): Model description and
- 32 preliminary application (in Chinese), Journal of Nanjing University of Information Science &
- Technology (Natural Science Edition), 1(1), 19-26, 2009.
- 34 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in
- 35 regional-scale numerical models, Atmos. Environ., 23, 1293-1304,
- 36 doi:http://dx.doi.org/10.1016/0004-6981(89)90153-4, 1989.
- Wilson, S. J., Steenhuisen, F., Pacyna, J. M., and Pacyna, E. G.: Mapping the spatial distribution
- 38 of global anthropogenic mercury atmospheric emission inventories, Atmos. Environ., 40,
- 39 4621-4632, doi:http://dx.doi.org/10.1016/j.atmosenv.2006.03.042, 2006.
- 40 Wu, Q., Wang, Z., Chen, H., Zhou, W., and Wenig, M.: An evaluation of air quality modeling
- 41 over the Pearl River Delta during November 2006, Meteorol. Atmos. Phys., 116, 113-132,
- 42 doi:10.1007/s00703-011-0179-z, 2012.
- 43 Yan, X., Ohara, T., and Akimoto, H.: Statistical modeling of global soil NOx emissions, Global
- 44 Biogeochem. Cy., 19, 2005.

- 1 Zhang, Y., Jaegle, L., van Donkelaar, A., Martin, R. V., Holmes, C. D., Amos, H. M., Wang, Q.,
- Talbot, R., Artz, R., Brooks, S., Luke, W., Holsen, T. M., Felton, D., Miller, E. K., Perry, K. D.,
- 3 Schmeltz, D., Steffen, A., Tordon, R., Weiss-Penzias, P., and Zsolway, R.: Nested-grid
- 4 simulation of mercury over North America, Atmos. Chem. Phys., 12, 6095-6111,
- 5 doi:10.5194/acp-12-6095-2012, 2012.

- 6 Zaveri, R., and Peters, L.: A new lumped structure photochemical mechanism for large-scale
- 7 applications, J. Geophys. Res.-Atmos., 104, 30387-30415,1999.

Table 1. Reactions and rate constants used in the GNAQPMS-Hg model.

NO.	Reaction	Rates (k or K) <sup>a</sup>	References							
Gas-phase reactions										
RG1	$Hg(0)(g)+O_3(g)\rightarrow Hg(II)(g)$	$3x10^{-20} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$	Hall (1995)							
RG2	$Hg(0)(g)+HCl(g)\rightarrow HgCl_2(g)$	$1 \times 10^{-19} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$	Hall and Bloom (1993)							
RG3	$Hg(0)(g)+H_2O_2(g)\rightarrow Hg(OH)_2(g)$	$8.5 \times 10^{-19} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$	Tokos et al. (1998)							
RG4	$Hg(0)(g)+Cl_2(g)\rightarrow HgCl_2(g)$	$2.6 \times 10^{-18} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$	Ariya et al. (2002)							
RG5	$Hg(0)(g)+OH(g)\rightarrow Hg(OH)_2(g)$	$8x10^{-14} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$	Sommar et al. (2001)							
Gas-liquid equilibria										
GL1	$Hg(0)(g) \leftrightarrow Hg(0)(aq)$	0.11 M atm <sup>-1</sup>	Sanemasa (1975)							
GL2	$HgCl_2(g) \leftrightarrow HgCl_2(aq)$	$1.4 \times 10^6  \text{M atm}^{-1}$	Lindqvist and Rodhe (1985)							
GL3	$Hg(OH)_2(g) \leftrightarrow Hg(OH)_2(aq)$	$1.2 \times 10^4  \text{M atm}^{-1}$	Lindqvist and Rodhe (1985)							
Aqueous-phase equilibria										
AE1	$HgCl_2(aq) \leftrightarrow Hg^{2+} + 2Cl^{-}$	$1 \times 10^{-14} \mathrm{M}^2$	Sillen et al. (1964)							
AE2	$Hg(OH)_2(aq) \leftrightarrow Hg^{2+} + 2OH^{-}$	$1 \times 10^{-22} \text{ M}^2$	Sillen et al. (1964)							
AE3	$Hg^{2+} + SO_3^{2-} \leftrightarrow HgSO_3$	$2.1 \times 10^{13} \mathrm{M}^{-1}$	Van Loon et al. (2001)							
AE4	$HgSO_3+SO_3^{2-} \leftrightarrow Hg(SO_3)_2^{2-}$	$1 \times 10^{10} \text{ M}^{-1}$	Van Loon et al. (2001)							
Aqueous-phase reaction										
RA1	$Hg(0)(aq)+O_3(aq)\rightarrow Hg^{2+}$	$4.7 \times 10^7 \mathrm{M}^{-1} \mathrm{s}^{-1}$	Munthe (1992)							
RA2	$Hg(0)(aq)+OH(aq)\rightarrow Hg^{2+}$	$2x10^9 \text{ M}^{-1} \text{ s}^{-1}$	Lin and Pehkonen (1997)							
RA3	$HgSO_3(aq) \rightarrow Hg(0)(aq)$	$0.0106 \text{ s}^{-1}$	Van Loon et al. (2000)							
RA4	$Hg(II)(aq)+HO_2(aq)\rightarrow Hg(0)(aq)$	$1.7 \times 10^4 \mathrm{M}^{-1} \mathrm{s}^{-1}$	Pehkonen and Lin (1998)							
RA5	$Hg(0)(aq)+HOCl(aq)\rightarrow Hg^{2+}$	$2.09 \times 10^6 \mathrm{M}^{-1} \mathrm{s}^{-1}$	Lin and Pehkonen (1998)							
RA6	$Hg(0)(aq)+OCl \rightarrow Hg^{2+}$	$1.99 \times 10^6  M^{-1}  s^{-1}$	Lin and Pehkonen (1998)							
Adsorption of Hg(II) on PM in the aqueous-phase										
AD1	$Hg(II)(aq) \leftrightarrow Hg(II)(p)$	$34 L g^{-1}$	Seigneur et al. (1998)							

<sup>&</sup>lt;sup>a</sup> The reaction rate constants are for temperatures in the range of 20 to 25<sup>o</sup>C. No temperature dependence information is available.

**Table 2.** Global budgets of TGM in the literature (Unit: Mg yr<sup>-1</sup>).

	Bergan et	Shia et al.	Lamborg et	Mason et	Seigneur et	Selin et al.	Selin et al.	This
	al. (1999)	(1999)	al. (2002)	al. (2002)	al. (2004)	(2007)	(2008)	work
<b>Total Sources</b>	6050	6100	4400	6600	6411	7000	11200	10163
anthropogenic	2150	2100	2600	2400	2143	2200	3400	2488
land	2500	2000	1000	1600	2290	2000	2800	2675
ocean	1400	2000	800	2600	1978	2800	5000	5000
<b>Total Sinks</b>	6050	6100	4200	6600	6411	7000	11200	10163
Wet deposition		2800		3920		2100		2283
Dry deposition		3300		2680		4700		7880
TGM Burden	6050	10400	5220	5000	7690	5360	5600	5507
TGM lifetime(yr)	1	1.7	1.3	0.76	1.2	0.79	0.5	0.54

**Table 3.** Statistical summary of comparisons of the model results with observations<sup>a</sup>.

Parameter	Region	R	NMB	RMSE	SVR <sup>b</sup>
TGM	East Asia Nested	0.51	-39%	3.87	2.56
	East Asia	0.54	-32%	3.61	2.56
	North America	0.69	18%	0.58	0.48
	Europe	0.57	-8%	0.17	0.35
	Global	0.70	-18%	2.22	_
Oxidized mercury	East Asia Nested	0.45	-12%	242	3.66
	East Asia	0.31	-10%	259	3.66
	North America	0.53	148%	28	1.61
	Europe	0.91	155%	48	1.00
	Global	0.53	3%	185	-
Wet deposition	East Asia Nested	0.78	-28%	45.5	6.69
	East Asia	0.36	-61%	60.1	6.69
	North America	0.76	-4%	4.3	1.89
	Europe	0.78	4%	1.5	1.40
	Global	0.38	-36%	29.3	-
Dry deposition	East Asia Nested	0.88	-42%	87.0	-
	East Asia	0.81	-42%	88.5	-

 $<sup>^{</sup>a}$  R, NMB, RMSE, SVR represent correlation coefficient, normalized mean bias, root mean square error, spatial variation ratio. Units of TGM, oxidized mercury, wet and dry deposition are ng m<sup>-3</sup>, pg m<sup>-2</sup> yr<sup>-1</sup>, μg m<sup>-2</sup> yr<sup>-1</sup> respectively.

<sup>&</sup>lt;sup>b</sup> SVR defines as (max-min)/mean observations over all sites.

**Table 4.** Statistical comparisons of the online and offline nested simulation results with observations over East Asia<sup>a</sup>.

Parameter	Case <sup>b</sup>	R	NMB	RMSE
TGM	Base simulation	0.51	-39%	3.87
	3 hour boundary conditions	0.50	-42%	3.94
	6 hour boundary conditions	0.50	-42%	3.95
Oxidized mercury	Base simulation	0.45	-12%	242.15
	3 hour boundary conditions	0.42	-17%	251.52
	6 hour boundary conditions	0.41	-18%	252.78
Wet deposition	Base simulation	0.78	-28%	45.47
	3 hour boundary conditions	0.78	-29%	45.75
	6 hour boundary conditions	0.78	-29%	45.78
Dry deposition	Base simulation	0.88	-42%	87.02
	3 hour boundary conditions	0.88	-44%	88.55
	6 hour boundary conditions	0.87	-45%	88.96

<sup>&</sup>lt;sup>a</sup> *R*, NMB, RMSE represent correlation coefficient, normalized mean bias, root mean square error. Units of TGM, oxidized mercury, wet and dry deposition are ng m<sup>-3</sup>, pg m<sup>-3</sup>, μg m<sup>-2</sup> yr<sup>-1</sup> respectively.

b In the base simulation, the nested domain gets boundary conditions from the global domain every 10 minutes.

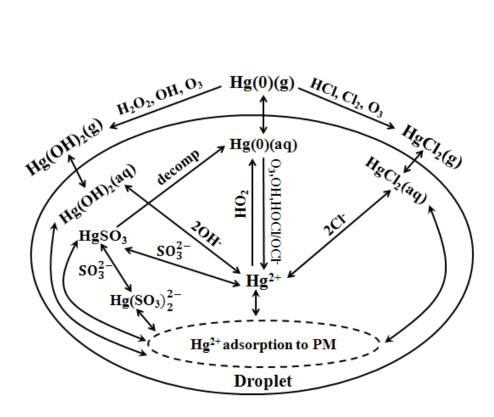


Fig. 1. Schematic of different mercury reactions utilized in the GNAQPMS-Hg model.

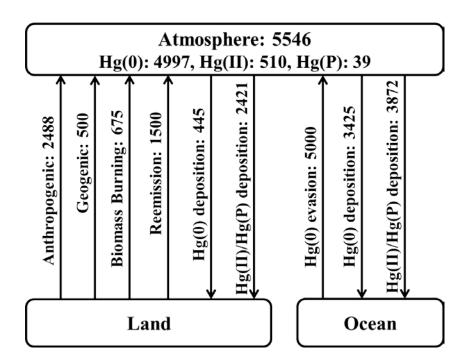
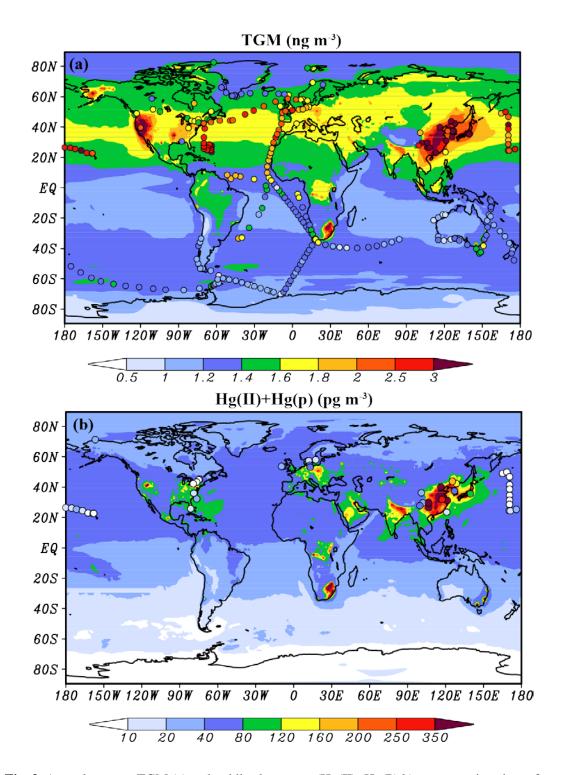
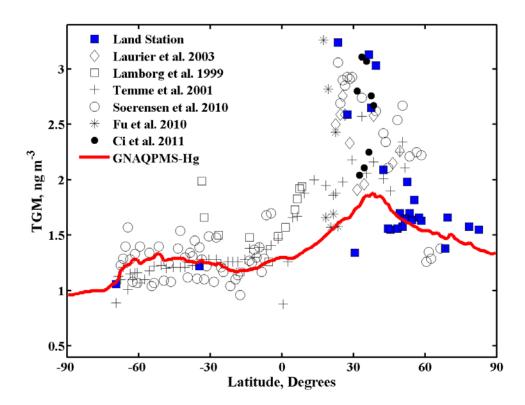


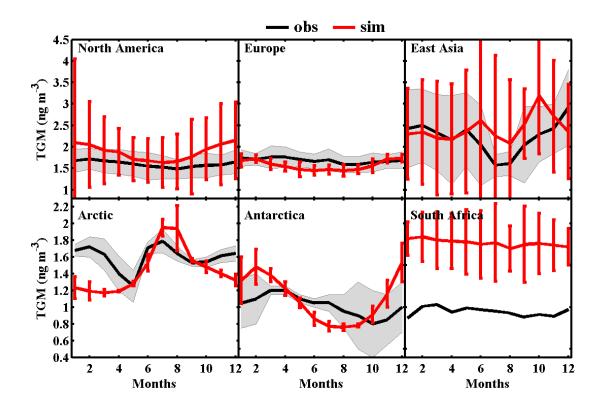
Fig. 2. Global atmospheric mercury budget in GNAQPMS-Hg. Units are Mg yr<sup>-1</sup>.



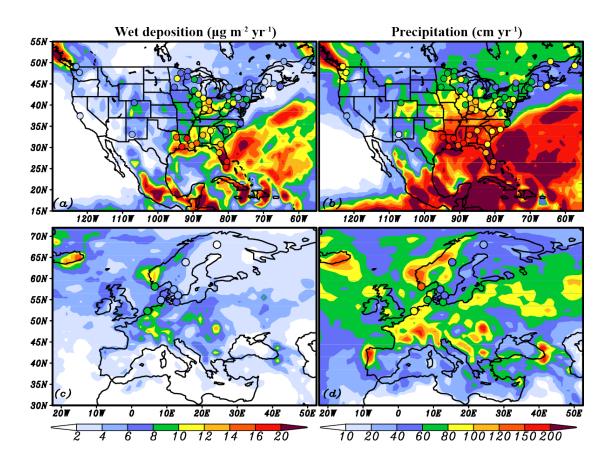
**Fig. 3.** Annual average TGM (a) and oxidized mercury (Hg(II)+Hg(P),b) concentrations in surface air. Model results (background, for year 2001) are compared to observations (circles) from long-term surface sites and short-term ship cruises. Units of TGM and oxidized mercury are ng  $m^{-3}$  and pg  $m^{-3}$ , respectively.



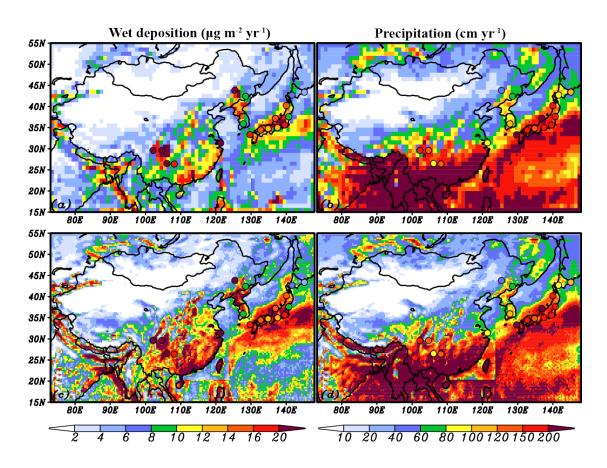
**Fig. 4.** Variation of surface TGM concentrations (ng m<sup>-3</sup>) with latitude. Zonally averaged, annual mean model results (line) are compared to observations (symbols).



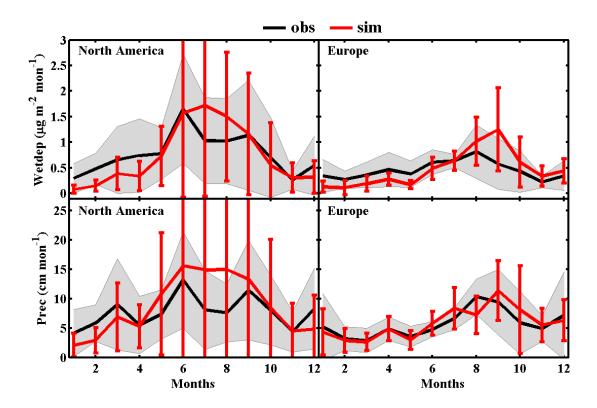
**Fig. 5.** Mean seasonal variation of TGM (ng m<sup>-3</sup>) at North America, Europe, East Asia, Arctic, Antarctica and South Africa sites. Gray shaded areas and red vertical bars show one standard deviation over the sites for observations and for model results.



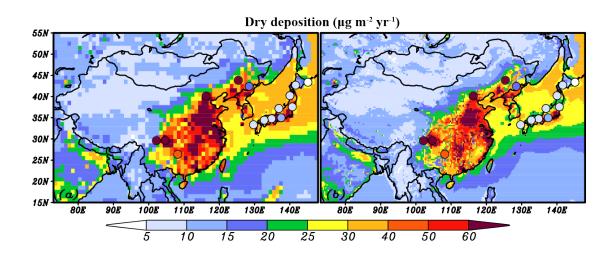
**Fig. 6.** Simulated annual mercury wet deposition ( $\mu g \ m^{-2} \ yr^{-1}$ ) and accumulated precipitation (cm yr<sup>-1</sup>) over North America (a, b) and Europe (c, d) in 2001. Overlaid points show observations for the same year from the Mercury Deposition Network (MDN) over North America, and the European Monitoring and Evaluation Programme (EMEP) over Europe.



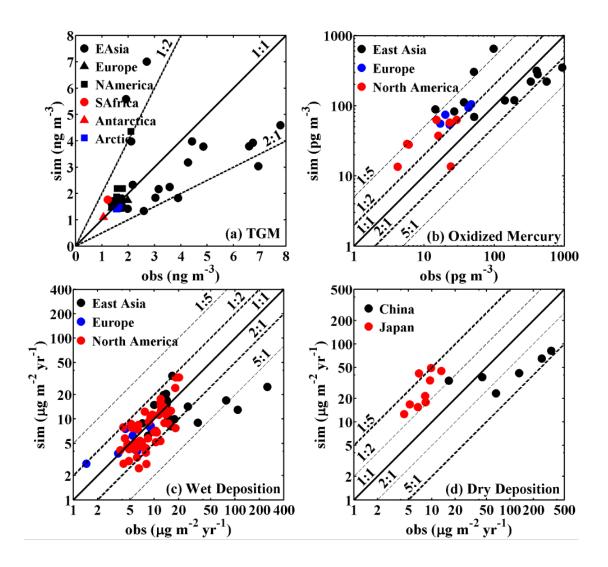
**Fig. 7.** Simulated annual mercury wet deposition ( $\mu g \ m^{-2} \ yr^{-1}$ ) and accumulated precipitation (cm  $yr^{-1}$ ) over East Asia in the global (a, b) and nested (c, d) domains in 2001. Overlaid points show observations collected from the literature. Note that observations and simulated results are not in the same year.



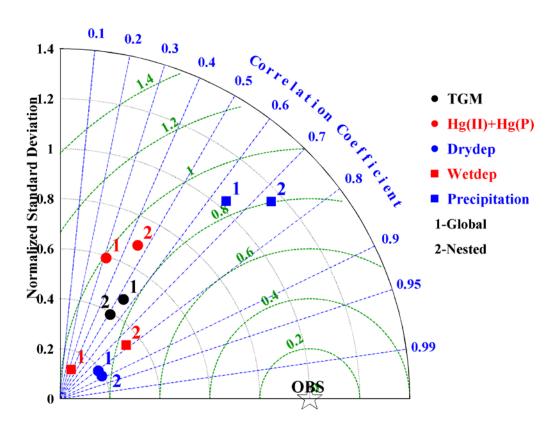
**Fig. 8.** Mean seasonal variation of mercury wet deposition (μg m<sup>-2</sup> mon<sup>-1</sup>) and accumulated precipitation (cm mon<sup>-1</sup>) at North America (51 sites averaged) and Europe (8 sites averaged) sites in 2001. Gray shaded areas and red vertical bars show one standard deviation over the sites for observations and for model results.



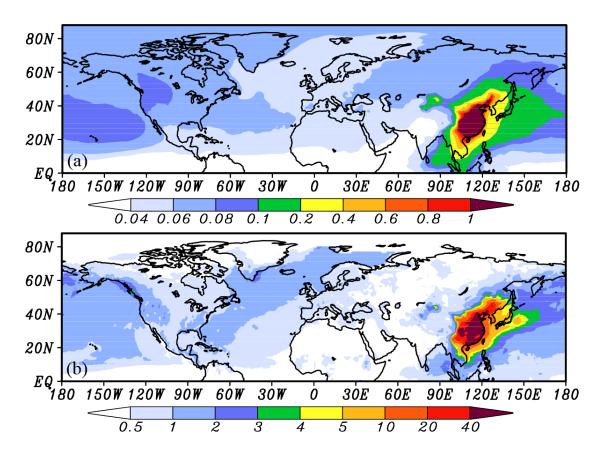
**Fig. 9.** Simulated annual mercury dry deposition ( $\mu$ g m<sup>-2</sup> yr<sup>-1</sup>) over East Asia in the global (a) and nested (b) domains in 2001. Overlaid points show observations collected from the literature. Note that observations and simulated results are not in the same year.



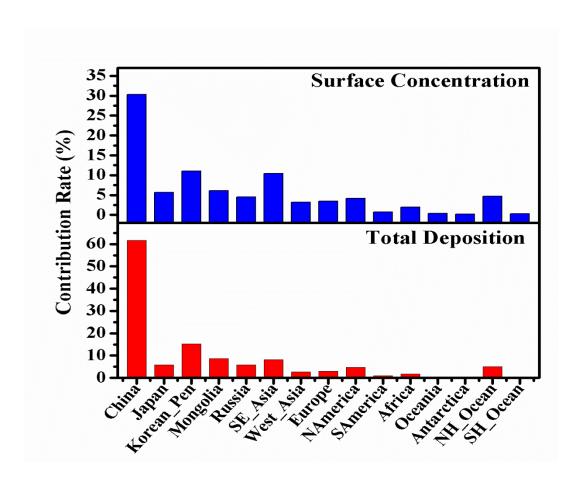
**Fig. 10.** Simulated vs. observed TGM (a), oxidized mercury (b), wet deposition (c), dry deposition (d) in different regions. Note that coordinates are different in different panels. Units of TGM, oxidized mercury, wet deposition and dry deposition are ng m<sup>-3</sup>, pg m<sup>-3</sup>,  $\mu$ g m<sup>-2</sup> yr<sup>-1</sup>,  $\mu$ g m<sup>-2</sup> yr<sup>-1</sup>, respectively.



**Fig. 11.** Taylor Diagram of simulated annual TGM, Hg(II)+Hg(P), dry deposition, wet deposition and precipitation over East Asia in the global and nested domains (denoted by 1 and 2).



**Fig. 12.** Contributions of Chinese primary anthropogenic sources to (a) annual mercury surface concentrations and (b) total (wet plus dry) deposition in the Northern Hemisphere. The units of mercury concentrations and deposition are ng m<sup>-3</sup> and  $\mu$ g m<sup>-2</sup> yr<sup>-1</sup>, respectively.



**Fig. 13.** Mean percentage contributions (%) from Chinese primary anthropogenic sources to annual mercury surface concentrations and total (wet plus dry) deposition over different world regions.