

1 **Evaluation of the global aerosol microphysical ModelE2-TOMAS model**
2 **against satellite and ground-based observations**

3
4 Y. H. Lee¹, P. J. Adams², and D. T. Shindell¹

5 ¹ Earth and Ocean Sciences, Nicholas School of the Environment, Duke University,
6 Durham, NC 27708

7 ² Department of Civil and Environmental Engineering and Department of Engineering
8 Public Policy, Carnegie Mellon University, Pittsburgh, PA USA

9
10 Corresponding author: Y. H. Lee (yunha.lee.00@gmail.com)

11
12 **Abstract**

13
14 The Two-Moment Aerosol Sectional microphysics model (TOMAS) has been integrated
15 into the state-of-the-art general circulation model, GISS ModelE2. This paper provides a
16 detailed description of the ModelE2-TOMAS model and evaluates the model against
17 various observations including aerosol precursor gas concentrations, aerosol mass and
18 number concentrations, and aerosol optical depths. Additionally, global budgets in
19 ModelE2-TOMAS are compared with those of other global aerosol models, and the
20 ModelE2-TOMAS model is compared to the default aerosol model in ModelE2, which is
21 a One-Moment Aerosol (OMA) model (i.e., no aerosol microphysics). Overall, the
22 ModelE2-TOMAS predictions are within the range of other global aerosol model
23 predictions, and the model has a reasonable agreement (mostly within a factor of two)
24 with observations of sulphur species and other aerosol components as well as aerosol
25 optical depth. However, ModelE2-TOMAS (as well as ModelE2-OMA) cannot capture
26 the observed vertical distribution of sulphur dioxide over the Pacific Ocean possibly due
27 to overly strong convective transport and overpredicted precipitation. The ModelE2-
28 TOMAS model simulates observed aerosol number concentrations and cloud
29 condensation nuclei concentrations roughly within a factor of two. Anthropogenic aerosol
30 burdens in ModelE2-OMA differ from ModelE2-TOMAS by a few percent to a factor of
31 2 regionally, mainly due to differences in aerosol processes including deposition, cloud
32 processing, and emission parameterizations. We observed larger differences for naturally
33 emitted aerosols such as sea salt and mineral dust, as those emission rates are quite
34 different due to different upper size cutoff assumptions.

35
36 **1. Introduction**

37
38 Aerosols perturb the energy balance of the Earth-atmosphere system by scattering
39 and absorbing solar and terrestrial radiation, known as the aerosol direct effect, and by
40 modifying cloud properties by acting as cloud condensation nuclei (CCN), known as

1 aerosol indirect effects (e.g. Lohmann and Feichter, 2005; Forster and Ramaswamy,
2 2007). The recently published IPCC AR5 (Intergovernmental Panel on Climate Change
3 Fifth Assessment Report) refers to these as aerosol-radiation interactions and aerosol-
4 cloud interactions, respectively (Boucher et al., 2013). For light-absorbing aerosols such
5 as black carbon and mineral dust, the ambient air can be heated as a result of their direct
6 effect, affecting relative humidity and atmospheric stability, which is known as the semi-
7 direct effect. The largest uncertainty in estimating anthropogenic climate forcing is from
8 the aerosol indirect effects (Myhre et al., 2013b). Since it is not easily estimated from
9 observations due to natural variability in cloud properties and the lack of observations of
10 the pre-industrial atmosphere, estimates of aerosol indirect forcing have been mainly
11 based on general circulation models (GCMs). Thus, there have been growing efforts to
12 develop and improve aerosol microphysics models for a more physically based
13 representation of atmospheric aerosol number and CCN concentrations (e.g. Adams and
14 Seinfeld, 2002; Easter et al., 2004; Vignati et al., 2004; Lauer et al., 2005; Liu et al.,
15 2005; Spracklen et al., 2005; Stier et al., 2005; Bauer et al., 2008; Trivitayanurak et al.,
16 2008; Yu and Luo, 2009; Mann et al., 2010; Lee and Adams, 2012).

17 Aerosol microphysics models can be broadly categorized into modal and sectional
18 methods, depending on how they represent the aerosol size distribution. In general,
19 modal-based methods use an analytical function (e.g. a lognormal distribution) to
20 represent a subset of the particle population. Sectional methods represent a size
21 distribution by predicting aerosols in several size sections or “bins”. Additionally,
22 sectional and modal methods may differ from each other in numerous ways, including the
23 number of moments of the size distributions that are tracked in each section or mode.

24 Sectional methods can be divided into single-moment sectional methods that
25 typically track either aerosol number or mass in each bin and two-moment sectional
26 methods that explicitly track both aerosol number (i.e. 0th moment) and mass (i.e. 1st
27 mass moment or 3rd radial moment) in each size section. Unlike single-moment sectional
28 approaches, two-moment sectional methods can conserve both number and mass very
29 accurately (Tzivion et al., 1987; Feingold et al., 1988; Harrington and Kreidenweis, 1998;
30 Tzivion et al., 2001; Adams and Seinfeld, 2002; Jung et al., 2006) but have a high
31 computational burden. The modal approaches are generally more computationally

1 efficient but may not represent abrupt transitions in a size distribution well, which can
2 occur during cloud processing (Zhang et al., 1999).

3 The TOMAS aerosol microphysics model (Adams and Seinfeld, 2002; Tzivion et
4 al., 1987; Tzivion et al., 1989) has been developed to study tropospheric aerosol
5 microphysics and predict cloud condensation nuclei (CCN) concentrations. The TOMAS
6 model has been previously implemented into the climate model of Goddard Institute for
7 Space Studies General Circulation Model II-prime (GISS GCM II-prime), referred to as
8 “GISS-TOMAS” (Lee and Adams, 2010). It has also been incorporated into GEOS-
9 CHEM (Trivitayanurak et al. 2008), the regional model PMCAMx-UF (Jung et al.,
10 2010), and the Large-Eddy Simulation model (Stevens et al., 2012; Singh et al., 2014).
11 The GISS GCM II-prime has horizontal grid dimensions of 4° latitude and 5° longitude,
12 with nine vertical sigma layers between the surface and the 10 hPa level (Hansen et al.,
13 1983). Modules for each of the major aerosol species have been developed for the GISS
14 GCM II-prime, and the GISS-TOMAS model has been evaluated with ground-level
15 measurements such as number and mass concentrations, deposition fluxes, and remote
16 sensing observations (Adams and Seinfeld, 2002; Pierce and Adams, 2006; Pierce et al.,
17 2007; Lee et al., 2009; Lee and Adams, 2010). Despite the accuracy in predicting aerosol
18 microphysical process in TOMAS, the original version of TOMAS has a heavy
19 computational burden. Lee and Adams (2012) developed less computationally expensive
20 configurations of the TOMAS model (Fast TOMAS), which are 2-3 times faster than the
21 original TOMAS model with only a few percent increases in microphysical errors.
22 However, a remaining weakness for the GISS-TOMAS model is the outdated host model,
23 the GISS GCM II-prime.

24 Here, we incorporate the TOMAS model into the new version of GISS GCM (i.e.,
25 ModelE2), referred to as “ModelE2-TOMAS”. ModelE2 now has three different aerosol
26 models available: TOMAS, the One-Moment Aerosol model (hereafter, referred to as
27 OMA) (e.g. Koch et al., 2006) that has no microphysics, and the modal-based aerosol
28 microphysics model, MATRIX (Multiconfiguration Aerosol TRacker of mIXing state)
29 (Bauer et al., 2008). The combination of several aerosol models allows ModelE2 to
30 explore the uncertainties in predicting aerosol characteristics and their climate effects that
31 are associated with aerosol modelling (e.g. different numerical approaches) in the same

1 host model. We also note that it was important to implement the TOMAS aerosol model
2 into the ModelE2 host model because uncertainties in the estimates of aerosol forcing
3 come not only from aerosol modelling itself but also other parts of the host GCM (e.g.,
4 cloud physics, planetary boundary layer, and advection).

5 ModelE2-TOMAS has been used in several recent studies under the Atmospheric
6 Chemistry and Climate Model Intercomparison Project (ACCMIP) which aims to
7 understand composition changes and the associated radiative forcing between 1850 and
8 2100 (Bowman et al., 2013; Lamarque et al., 2013a; Lamarque et al., 2013b; Lee et al.,
9 2013a; Nabat et al., 2013; Naik et al., 2013; Shindell et al., 2013; Stevenson et al., 2013;
10 Young et al., 2013). Here we give a detailed description of ModelE2-TOMAS and
11 evaluate against ModelE2-OMA (e.g., Schmidt et al., 2014) and observations of aerosol
12 mass and number as well as aerosol optical depth. Section 2 and 3 provide descriptions of
13 ModelE2-OMA and ModelE2-TOMAS, respectively. Section 4 explains the emissions
14 and design of the simulations. Section 5 presents global budgets of the simulated aerosols
15 and the evaluation of the ModelE2-TOMAS and ModelE2-OMA against observations of
16 aerosol mass concentrations and aerosol optical depth and the evaluation of the TOMAS
17 number predictions against observations. Conclusions follow in Section 6. We note that
18 aerosol direct and indirect forcings using ModelE2-TOMAS will be discussed in a
19 separate paper.

20

21 **2. GISS GCM ModelE2**

22 In this section, we briefly describe ModelE2 (Schmidt et al. (2014)), the GISS climate
23 model used to perform Coupled Model Intercomparison Phase 5 (CMIP5; Taylor et al,
24 2012). The model physics are mostly similar to GISS ModelE (CMIP3 version: Schmidt
25 et al., 2006). The model has 2° latitude by 2.5° longitude resolution, with 40 vertical
26 hybrid sigma layers from the surface to 0.1 hPa (80 km). Tracers, heat, and humidity are
27 advected using the highly nondiffusive Quadratic Upstream Scheme (Prather, 1986). The
28 radiation scheme accounts for size-dependent scattering properties of clouds and aerosols
29 based on Mie scattering (Hansen et al., 1983) and non-spherical light scattering of cirrus
30 and dust particles based on T-matrix theory (Mishchenko et al., 1996). It also includes the
31 impact of water uptake by hygroscopic species on their radiative properties. In the model,
32

1 clouds are distinguished into convective and large-scale stratiform clouds. The clouds
2 parameterizations are similar to Del Genio (1993) and Del Genio et al. (1996) but have
3 been improved in several respects (see details in Schmidt et al., 2006; Schmidt et al.,
4 2014). The physics timestep is 30 minutes, and the radiation is calculated every 2.5 hours.

5 **2.1 ModelE2-OMA description**

6

7 ModelE2 includes a default aerosol module, OMA (One-Moment Aerosol), which has
8 no microphysics. ModelE2-OMA has sulphate (Koch et al., 2006; Koch et al., 2007;
9 Koch et al., 2011), carbonaceous aerosols (Koch et al., 2007), secondary organic aerosols
10 (Tsigaridis and Kanakidou, 2007), sea salt (Koch et al., 2006; Tsigaridis et al., 2013),
11 dust (Miller et al., 2006), and nitrate (Bauer et al., 2007). Along with sulphate, the model
12 also predicts sulphur dioxide, dimethyl sulfide (DMS) and methanesulfonic acid (MSA)
13 (Koch et al., 2006). The secondary organic aerosol formation is computed using a two-
14 product model with isoprene, monoterpenes, and sesquiterpenes as SOA precursors
15 (described in Tsigaridis and Kanakidou, 2007). Sea salt particles have two size classes
16 with a fine mode (0.1 μm to 1 μm in dry radii) and a coarse mode (1 μm to 4 μm in dry
17 radii). Dust particles have four size classes with radii between 0.1-1 μm (clay), 1-2 μm
18 (silt1), 2-4 μm (silt2), and 4-8 μm (silt3). The model accounts for heterogeneous
19 chemistry on mineral dust particle surfaces to form nitrate and sulphate (Bauer and Koch,
20 2005).

21 In ModelE2, the surface boundary conditions are defined using dry deposition and
22 interactive surface sources (Koch et al., 2006). The dry deposition scheme is tightly
23 coupled to the model's boundary layer scheme and is based on a resistance-in-series
24 scheme derived from the Harvard GISS-CTM, which is applied between the surface layer
25 (10 m) and the ground (Koch et al., 2006). Wet deposition is determined by several
26 processes including rainout within clouds, washout below precipitating regions,
27 scavenging within and below cloud updrafts, evaporation of falling precipitation,
28 transport along with convective plumes, and detrainment and evaporation from
29 convective plumes (Koch et al., 2006; Shindell et al., 2006). ModelE2 includes a
30 dissolved species budget scheme for stratiform clouds, which has an impact on sulphate

1 formation via aqueous oxidation, since some sulphate formed in clouds undergoes wet
2 scavenging instead of being added back to the sulphate in air (Koch et al., 2006).

3 Tropospheric/stratospheric chemistry in ModelE2 includes 156 chemical reactions
4 among 51 gas species (Shindell et al., 2013). In ModelE2, chemistry and aerosols are
5 fully interactive, so that the oxidation fields used for sulphate formation are from the
6 chemistry model (not prescribed) and the photolysis rates are affected by light attenuation
7 by aerosols (Shindell et al., 2013). Photolysis rates are computed using the Fast-J2
8 scheme (Bian and Prather, 2002). Aerosol indirect effects are based on an empirical
9 parameterization that compute cloud droplet number concentrations as a function of
10 aerosol mass (Menon et al., 2002; Menon et al., 2008).

11

12 3. ModelE2-TOMAS description

13

14 The TOMAS aerosol microphysics model uses a sectional approach that represents
15 the aerosol size distribution by predicting the amount of aerosol in several size categories
16 or “bins”. TOMAS tracks two moments of the aerosol size distribution in each size bin:
17 total aerosol number (i.e., 0th moment) and mass (i.e., 1st mass moment). Total mass is
18 decomposed into several aerosol species, allowing prediction of the size-resolved aerosol
19 composition. In total, ten quantities are tracked for each size bin: sulphate mass, sea-salt
20 mass, mass of pure (hydrophobic) elemental carbon (EC), mass of mixed (aged) EC,
21 mass of hydrophobic organic matter (OM), mass of hydrophilic OM, mass of mineral
22 dust, mass of ammonium, mass of water and the number of aerosol particles in that bin.
23 In TOMAS, all ammonia becomes ammonium aerosol until sulfate is neutralized to form
24 ammonium sulfate; the excess ammonia after neutralization remains as free gas-phase
25 ammonia. The ammonium aerosol is partitioned into each size bin in proportion to the
26 sulfate mass. However, ammonium is not size-resolved (i.e., bulk tracer) for the purpose
27 of model processes outside of TOMAS such as advection and deposition. In addition, the
28 model tracks four bulk gas-phase species: sulphur dioxide (SO_2), dimethylsulfide (DMS),
29 sulphuric acid (H_2SO_4), and a lumped gas-phase tracer that represents oxidized organic
30 vapours forming secondary organic aerosol (SOA). The gas-phase H_2SO_4 is assumed to
31 be in pseudo-steady state equilibrium between its chemical production and
32 condensational/nucleation losses (Pierce and Adams, 2009a). Water uptake by sulphate

1 and sea salt is based on a polynomial fit based on ISORROPIA, a thermodynamic
2 equilibrium model for inorganic aerosols (Nenes et al., 1998). For water uptake by
3 hydrophilic OM, it is based on the observations of Dick et al. (2000). The size section
4 boundary is defined by dry particle mass, such that addition or removal of aerosol water
5 mass does not move particles between sections. In general, TOMAS treats all aerosols as
6 internally mixed during microphysics such as calculating condensation and coagulation
7 rates. However, a portion of EC is treated as externally mixed for the purpose of wet
8 deposition. A detailed description of the TOMAS microphysics scheme can be found in
9 Adams and Seinfeld (2002), Lee and Adams (2012), and Lee et al. (2013b).

10 Several alternative nucleation schemes are available in TOMAS, including binary
11 nucleation (Vehkamaki et al., 2002), ternary nucleation (Napari et al., 2002), ion-induced
12 nucleation (Modgil et al., 2005), and activation nucleation with an A factor of $2 \times 10^{-6} \text{ s}^{-1}$
13 (Sihto et al., 2006) for the boundary layer (~up to 900 hPa). For the simulations used in
14 this paper, only binary nucleation is used. The boundary-layer nucleation is off in all
15 simulations because it tends to overpredict aerosol number concentrations in our model.
16 Also we do not show any run with the ternary nucleation (Napari et al., 2002) because it
17 overpredicts aerosol number concentration severely (not shown).

18 With Fast TOMAS models, the TOMAS microphysics module became more flexible
19 in term of varying particle size resolution, i.e. the number of size bins (Lee and Adams,
20 2012). For the size range of 10 nm to 10 μm , the original TOMAS uses 30 bins (size
21 boundary is defined with mass doubling), and the Fast TOMAS uses 15 bins or 12 bins
22 (size boundary is defined with mass quadrupling). As discussed in Lee and Adams
23 (2012), the Fast TOMAS reduces the computational burden by 2-3 times while generally
24 predicting CCN concentrations within a few percent of the original TOMAS. The lower
25 size cutoff in TOMAS can also vary from 10 nm to 3 nm or from 10 nm to 1 nm (Lee et
26 al., 2013b). Among several possible configurations, ModelE2-TOMAS currently uses
27 either 12 bins covering 10 nm to 10 μm or 15 bins covering 3 nm to 10 μm , which is the
28 most computationally efficient version of TOMAS for the given size range. In this paper,
29 we used TOMAS with 15 bins covering 3 nm to 10 μm (TOMAS15; see Table S1 in the
30 supplementary materials): 3 bins cover from 3 nm to 10 nm, 10 bins from 10 nm to 1 μm
31 and the last two bins from 1 μm to 10 μm . The TOMAS15 version has become the

1 default model configuration for ModelE2-TOMAS, so we will refer to it as ModelE2-
2 TOMAS throughout this paper. More configurations will be available in the near future.

3 The wet deposition scheme in ModelE2-TOMAS is identical to the one used in
4 ModelE2-OMA except for the following. First, ModelE2-TOMAS adds sulphate mass
5 produced in the aqueous phase directly to the bin-resolved sulphate mass in ambient air
6 rather than maintaining a separate tracer for dissolved sulphate. Compared to ModelE2-
7 OMA, this is a simplification because the sulphate formed in the aqueous phase will be
8 only released as interstitial aerosol when the cloud water evaporates. It is adopted here
9 for simplicity but will be improved in the future. The other difference is that the wet/dry
10 deposition in ModelE2-TOMAS accounts for particle size dependence. For in-cloud
11 scavenging, modified Köhler theory is used to obtain the critical supersaturation for
12 activation of each size section and to determine which particles activate and are subject to
13 in-cloud (nucleation) scavenging (Pierce et al., 2007). To determine activation, we
14 assume kappa values of 0.7 for sulfate, 1.3 for sea-salt, and 0.15 for hydrophilic OM. The
15 fraction of activated aerosols removed by wet deposition is proportional to the fraction of
16 cloud water that precipitates, which is computed in each model layer. Wet deposition
17 accounts for re-evaporation of precipitation. For in-cloud scavenging, the large-scale and
18 convective clouds in the model are assumed to have a supersaturation of 0.2%; unlike
19 GISS-TOMAS that used a supersaturation of 1.0% for convective clouds, a
20 supersaturation of 0.2% is assumed in ModelE2-TOMAS in order to capture the observed
21 Hoppel gap (\sim 100 nm) in the marine boundary layer. Note that the activation described
22 here to determine in-cloud scavenging is not used for computing cloud droplet number
23 concentrations (see below). For below-cloud scavenging, a first-order removal scheme
24 implemented for bulk aerosols by Koch et al. (1999) is modified for size-resolved
25 aerosols (Adams and Seinfeld, 2002). Dry deposition is identical to the existing
26 resistance-in-series scheme in ModelE2, but ModelE2-TOMAS considers size-dependent
27 gravitational settling of particles and size-dependent resistance in the quasi-laminar
28 sublayer (Seinfeld and Pandis, 1998; Adams and Seinfeld, 2002).

29 To compute the cloud microphysics properties as a function of aerosols (i.e., the
30 aerosol-cloud interactions), ModelE2-TOMAS uses a physical-based activation
31 parameterization from Nenes and Seinfeld (2002). A critical supersaturation is computed

1 in the parameterization using a model updraft velocity that is computed based on a large-
2 scale vertical velocity and sub-grid velocity.

3 In ModelE2-TOMAS, Mie theory is used to compute size-resolved AOD. For each
4 grid cell, particle compositions (including aerosol-water) in each individual size bin are
5 used to compute the volume-averaged refractive index and optical properties based on
6 Mie theory. The optical properties are used to compute aerosol optical depth taking into
7 account the aerosol concentration.

8 ModelE2-TOMAS is coupled to the same gas chemistry model (Shindell et al., 2013)
9 as ModelE2-OMA. So the oxidation fields used for sulphate formation are from the
10 chemistry model. However, unlike ModelE2-OMA, the photolysis rates are not affected
11 by aerosols.

12

13 **4. Description of the simulations**

14 **4.1 Emissions**

15

16 The emissions used in this study are summarized in Table 1. The simulations used year-
17 2000 emissions from the anthropogenic emissions inventory created for CMIP5
18 (Lamarque et al., 2012) and climatologically averaged biomass burning emissions from
19 GFED3 for 1997 to 2009 (van der Werf et al., 2010). For SO₂, in addition to the
20 anthropogenic emissions, continuous volcanic emissions from GEIA (Global Emissions
21 Initiative; Andres and Kasgnoc, 1998) are used but increased by a factor of 1.5 as in the
22 AEROCOM intercomparison emissions in Dentener et al. (2006). Sea-salt emissions are
23 based on Gong (2003), which extends the lower size limit of the Monahan et al. (1986)
24 emission from 0.4 μm to 0.02 μm. Dust emissions are based on the source distribution
25 from Ginoux et al. (2001) and are proportional to the third power of the wind speed (at 10
26 m in height) above a threshold that is a function of soil moisture. Subgrid-scale variation
27 of the wind speed in a GCM grid box, which is created by boundary-layer turbulence and
28 dry/wet convection, is accounted for in the modelled dust emissions (Cakmur et al., 2006;
29 Miller et al., 2006). DMS emissions are based on the seawater DMS concentrations of
30 Kettle et al. (1999). For the sea-to-air transfer function used in the DMS emissions, the

1 ModelE2-TOMAS runs are based on Liss and Merlivat (1986), and the ModelE2-OMA
2 model run is based on Nightingale et al. (2000).

3 Nightingale et al. (2000) provides a revised parameterization based on observations of
4 the sea-to-air transfer rate scatter between two classical parameterizations (i.e., Liss and
5 Merlivat, 1986; Wanninkhof, 1992), and it has been more favoured in many global DMS
6 models than the two classical parameterizations. However, DMS emissions are quite
7 uncertain. Estimates of the global DMS emissions range from 16 to 54 Tg S yr⁻¹ (Kettle
8 and Andreae, 2000), depending on the choice of DMS sea surface climatology, sea-to-air
9 transfer rate parameterization, and wind speed data. DMS emission rates from ModelE2-
10 TOMAS (16.1 Tg S yr⁻¹; see Table 3) and ModelE2-OMA (28.7 Tg S yr⁻¹; see Table 3)
11 are within this range. The Liss and Merlivat (1986) parameterization is used in ModelE2-
12 TOMAS, because a ModelE2-TOMAS run based on Nightingale et al. (2000)
13 overpredicts the SO₂ concentrations over remote oceanic regions especially in the
14 southern hemisphere. Koch et al. (2006) showed that the sea-to-air transfer function from
15 Nightingale et al. (2000) increased annual DMS emissions by roughly a factor of two
16 compared to the emission based on Liss and Merlivat (1986). This was desirable in
17 ModelE2-OMA because of the underprediction of sulphate in remote oceanic regions in
18 that model, although the model DMS and MSA (oxidized from DMS) tended to be
19 excessive in SH oceanic regions especially near Antarctica. However, despite the higher
20 DMS emissions, it turned out that the sulphate was still underpredicted because sulphate
21 formed by aqueous oxidation was subject to wet scavenging before releasing to the
22 ambient air as a result of the updated dissolved species budget scheme (Koch et al.,
23 2006).

24 **4.2 ModelE2-TOMAS run setup**

25

26 We performed the simulations nudged with winds from the MERRA (Modern Era
27 Retrospective-analysis for Research and Applications; Rienecker et al., 2011) reanalysis
28 meteorological fields from 2000 to 2003 with 3 years spin-up (i.e., 1997-1999). Primary
29 emissions of particulate sulphate are assumed to be 1.0 percent of total sulphur emissions.
30 Emissions size distributions assumed for ModelE2-TOMAS are summarized in Table 2.
31 Primary sulphate emissions are assumed to have a bi-modal lognormal distribution that

1 assigns 5% of the primary sulphate emissions as a nucleation mode with a geometric
2 number mean diameter (GMD) of 10 nm and a geometric standard deviation (GSD) of
3 1.6 and the rest as an Aitken mode with GMD of 70 nm and GSD of 2. For fossil fuel and
4 biofuel emissions, the size of primary carbonaceous aerosol emissions are assumed to
5 follow a lognormal size distribution with a GMD of 60 nm and a GSD of 1.59 for both
6 EC and OM (Stier et al., 2005). For carbonaceous aerosols of biomass burning emissions,
7 a lognormal size distribution is assumed to have a GMD of 150 nm and a GSD of 1.59.
8 Note that although the emission size distribution for biofuel emissions is generally
9 assumed to be the same as that for biomass burning emissions (e.g. Dentener et al., 2006),
10 in ModelE2-TOMAS run, we assumes the biofuel emission size distributions follow the
11 finer fossil fuel size settings because the CMIP5 emissions does not provide a separate
12 category for biofuel emissions (e.g. biofuel used for cooking and heating are assigned as
13 the residential sector, which also includes fossil fuel usage). The OC (organic carbon):
14 OM (organic matter) ratio is assumed to be 1:1.4.

15 ModelE2-TOMAS assumes larger particles for primary sulphate and carbonaceous
16 aerosols than GISS-TOMAS (e.g., Lee et al., 2013b) to capture the observed aerosol
17 number concentrations better. This is very likely due to the following: 1) GISS-TOMAS
18 attributed the primary sulphate emissions only to anthropogenic sulphur emissions
19 (excluding biomass burning emissions), while ModelE2-TOMAS attributes these to all
20 sulphur emissions; 2) GISS-TOMAS applied the biomass-burning emission size
21 distributions of carbonaceous aerosols to the biofuel emissions, which is coarser than the
22 fossil fuel emission size distribution. Note also that the emission size distributions used
23 for biomass burning and volcanic emissions are finer than the AEROCOM
24 recommendations in Dentener et al. (2006). However, the model number concentrations
25 and size distributions are changed little when applying the AEROCOM recommended
26 emission distributions (not shown). Note that the biomass burning and volcanic emissions
27 for sulphur are 1.4 Tg S yr^{-1} and $12.5 \text{ Tg S yr}^{-1}$, respectively.

28 Following the soil size assumptions used in GISS-TOMAS (Lee et al., 2009), the clay
29 distribution is assumed to have a GMD of $0.14 \mu\text{m}$ and a GSD of 2, and the silt
30 distribution, a GMD of $1.15 \mu\text{m}$ and a GSD of 2. Using this distribution, fifteen percent

1 of the silt emissions flux falls out of the upper size cutoff (i.e., 10 μm), and is therefore
2 not received by any of the TOMAS size bins.

3 Compared to the run setup described above (hereafter, referred to as the “BASE”
4 run), we additionally ran two other sensitivity runs with the ModelE2-TOMAS model by
5 perturbing the nucleation process to evaluate changes in number concentrations (Table 1).
6 The first sensitivity run is called “NoNUC”, in which we turned off nucleation to
7 estimate the contribution of primary emissions to aerosol number concentrations. The
8 other run is called “LowNUC”, in which we reduced the nucleation rate by using 5 times
9 lower sulphuric acid concentrations to compute nucleation rates. Note that sulphuric acid
10 concentrations are not perturbed in other processes, and the model sulphuric acid budget
11 is little influenced by this treatment.

12 4.3 ModelE2-OMA run setup

13 To compare to the ModelE2-TOMAS run, we also ran the ModelE2-OMA model
14 nudged to the same MERRA reanalysis meteorology with 3 years spin-up. However, the
15 natural emissions and associated settings are not always the same between the two
16 models because we chose to maintain the natural emissions/setup used in ModelE2-
17 OMA, which has been chosen carefully in previous studies. To assist the interpretation
18 of the results, we briefly summarize the differences between ModelE2-OMA and
19 ModelE2-TOMAS. First, as mentioned in Section 4.1, the ModelE2-OMA model uses
20 the sea-to-air transfer function of Nightingale et al. (2000) instead of Liss and Merlivat
21 (1986), because Koch et al. (2006) argues that the newer DMS emissions improve
22 sulphate predictions in the remote marine locations. Second, the same sea salt and dust
23 emission schemes are applied in both aerosol models, but different assumptions for the
24 upper limit of particle size are used: 8 μm in diameter for sea salt and 16 μm in diameter
25 for dust in the ModelE2-OMA model; 10 μm in diameter for all species in ModelE2-
26 TOMAS. Third, the ModelE2-OMA model assumes 2.5% of the total sulphur as primary
27 sulphate as followed by the AEROCOM study (Dentener et al., 2006) whereas ModelE2-
28 TOMAS assumes only 1%. Aerosol number predictions are sensitive to the primary
29 sulphate assumption, but sulphate mass concentrations are not. When using the 2.5%
30 assumption in ModelE2-TOMAS, we found that the simulated aerosol number
31

1 concentrations were biased high, and the model size distribution predictions were also
2 poor. Note that Pierce and Adams (2009b) shows that GISS-TOMAS also overpredicts
3 aerosol number concentration with the 2.5% assumptions.

4

5 **5. Model results and evaluation**

6

7 In this section, we present global-annual budgets, spatial distributions, and
8 evaluations of the model aerosol precursor gases (in Sect. 5.1), aerosol mass (in Sect. 5.2
9 and 5.3), aerosol optical depths (AODs; in Sect. 5.4) and aerosol number (in Sect. 5.5 and
10 5.6). The observations used for model evaluations are from surface-based, aircraft-based
11 and remote-sensing measurements. More details of the observations are provided in each
12 subsection. To compare with the ModelE2-TOMAS results, we included the ModelE2-
13 OMA results in global-annual budgets and model evaluations. Only the BASE run results
14 are used in Section 5.1 to 5.4 because the predicted aerosol precursor gases
15 concentrations, aerosol mass concentrations, and AODs from the nucleation sensitivity
16 runs are quite similar to the BASE run.

17 Model skill is quantified in terms of log-mean normalized bias (LMNB) and log-
18 mean normalized error (LMNE) when evaluating with annual-mean concentrations
19 measurements and normalized mean bias (NMB) and correlation coefficient (R) when
20 evaluating with an observed annual cycle and aerosol optical depth (both monthly and
21 annually averaged AODs).

22 **5.1 Aerosol precursor gases**

23

24 Global budgets of DMS and SO₂ in ModelE2-TOMAS are presented in Table 3 with
25 a range obtained from several global models including Wang et al. (2011), Liu et al.
26 (2005), and those listed in Liu et al. (2005). The DMS and SO₂ budgets in ModelE2-
27 TOMAS are within the ranges of the other global models. In the case of ModelE2-OMA
28 (in Table 3), the global burden of DMS is about a factor of two higher than the ModelE2-
29 TOMAS model because the DMS emission rate is ~78% higher by using the sea-to-air
30 transfer functions by Nightingale et al. (2000). Despite the different DMS emissions and
31 SO₂ emissions (due to the primary sulphate emission assumption, 1% versus 2.5%), the

1 global burden of SO₂ is quite similar to that in the ModelE2-TOMAS model. The
2 dominant SO₂ removal processes are aqueous oxidation and dry deposition in both
3 simulations. Boucher et al. (2003) simulates atmospheric DMS in the LMD-ZT model
4 using the same DMS emission schemes as ModelE2-OMA (i.e., EXP1 in their study) and
5 ModelE2-TOMAS (i.e., EXP4 in their study). The global DMS budgets from ModelE2-
6 OMA and ModelE2-TOMAS agree quite well with the EXP1 (within 25 %) and EXP4
7 (within 15%).

8 When using the same DMS emissions in ModelE2-TOMAS as in ModelE2-OMA, the
9 DMS global budgets are almost identical, but the SO₂ budgets vary substantially due to
10 the differences in SO₂ modelling, i.e., heterogeneous SO₂ oxidation and photolysis (see
11 Section 2). The heterogeneous sulphur dioxide oxidation on dust aerosol surfaces, which
12 is only included in ModelE2-OMA, accounts for 25% of the total gas-phase oxidation
13 loss. Based on Bauer and Koch (2005), including the heterogeneous chemistry, global
14 SO₂ burden can decrease by 32%, and the global sulphate burden can increase by 3%.
15 The simulated photolysis rates in ModelE2-OMA are affected by aerosol optical depth,
16 affecting hydroxyl (OH) and other gas tracer concentrations - Naik et al. (2013) show a
17 higher OH concentration in ModelE2-TOMAS than ModelE2-OMA. Overall, using the
18 same DMS emissions in ModelE2-TOMAS results in a higher SO₂ burden and worse
19 agreement for SO₂ and sulphate concentrations over remote oceanic regions (not shown).

20 The global budgets of H₂SO₄ and SOA precursor gas in the ModelE2-TOMAS model
21 are not included in Table 3 but are summarized here. The simulated H₂SO₄ has a total
22 production rate of 12.3 Tg S yr⁻¹, matching the SO₂ gas-phase oxidation, and is used in
23 aerosol microphysics (i.e. 12 Tg yr⁻¹ for condensation and 0.3 Tg yr⁻¹ for nucleation).
24 The model SOA precursor gas has a total production rate of 17.1 Tg yr⁻¹, assumed to be
25 10% of the terpene emission, and is condensed as hydrophilic OM. For ModelE2-OMA,
26 the total production rate of SOA is 14.6 Tg yr⁻¹. This is quite comparable to ModelE2-
27 TOMAS, which treats SOA much more simply and has a production rate of 17.1 Tg yr⁻¹.
28 The global burden of SOA in ModelE2-OMA is 0.6 Tg yr⁻¹.

29 Figure 1 shows global maps of annual-mean DMS and SO₂ column mass
30 concentrations. The spatial distribution of DMS concentrations shown in Fig. 1a is driven
31 by its emission and interactive OH and NO₃ concentrations, which oxidize DMS to form

1 MSA and SO₂. The model DMS concentrations are most pronounced in the Southern
2 Ocean and the Northern Atlantic oceans due to high seawater DMS concentrations during
3 summer. The simulated SO₂ concentration shown in Fig. 1b is very high over industrial
4 regions due to the anthropogenic emissions and is also high over the Southern Ocean due
5 to DMS oxidation. Several local hotspots of SO₂ shown in Fig. 1b are due to volcanic
6 emissions.

7 Annually averaged surface-layer SO₂ concentrations from both ModelE2-TOMAS
8 and ModelE2-OMA are evaluated against observations from the EMEP (European
9 Monitoring and Evaluation Programme, <http://www.emep.int>) and CASTNET (Clean Air
10 Status and Trends Network, <http://epa.gov/castnet/javaweb/index.html>) networks (see
11 Fig. 2). We used 2000-2004 mean SO₂ measurements for the EMEP network and 1995-
12 2005 mean SO₂ data for the CASTNET network. Performance of ModelE2-TOMAS and
13 ModelE2-OMA for predicted SO₂ concentrations in these locations is almost the same
14 (i.e. LMNB=0.25-0.26 and LMNE=0.34 for the EMEP network; LMNB=0.09 and
15 LMNE=0.29 for the CASTNET network), and the continental SO₂ predictions agree with
16 the observation on average roughly within a factor of two. The two aerosol models are
17 almost the same because the anthropogenic emissions, which are identical in both
18 models, are dominant at these locations.

19 Figure 3 compares surface-layer SO₂ and DMS mixing ratios from the two aerosol
20 models against observations at three Southern Hemisphere remote sites: Amsterdam
21 Island (DMS from Sciare et al., 2000; SO₂ from Nguyen et al., 1992), Cape Grim (Ayers
22 et al., 1995), and Dumont (Jourdain and Legrand, 2001). Note that Dumont has only
23 DMS measurements. For DMS, both models capture the observed seasonal cycle (i.e.
24 R>0.8) generally well but are less successful over Dumont. The DMS concentrations
25 seem to agree well against the observations when using the sea-air transfer function of
26 Liss and Merlivat (1986), i.e., the case for ModelE2-TOMAS, but this run underpredicts
27 during winter season at the Amsterdam Island site (in Fig. 3a) and during all season at
28 Dumont site (in Fig. 3c). Earlier, we mentioned that the global DMS budgets from
29 Boucher et al. (2003) agree well with those from ModelE2 when using the same DMS
30 emission parameterization. However, Boucher et al. (2003) shows the better agreement to
31 the same DMS measurements when using the sea-air transfer function of Nightingale et

1 al. (2000) at Amsterdam Island and Cape Grim (i.e., the case for ModelE2-OMA),
2 because their DMS mixing ratios from that simulations are actually more close to
3 ModelE2-TOMAS. This may suggest that, at least over SH high latitude regions, the
4 surface wind speed in ModelE2 is much stronger than that in LMD-ZT, resulting in
5 higher DMS emissions and burden. We need to investigate further to find out a source for
6 the difference though.

7 The ModelE2-TOMAS SO₂ shows very good agreement at Amsterdam Island but a
8 high bias at Cape Grim (see Figs. 3d and e). We considered whether the overpredicted
9 SO₂ in the model might be influenced by the emissions in the adjacent grids, which is
10 mentioned in Mann et al. (2010), because the measured SO₂ at Cape Grim is filtered to
11 include the marine sector only (Ayers et al., 1991). Sampling the model SO₂ from
12 adjacent grids toward marine areas, the overprediction is reduced significantly (LMB is
13 reduced from 10 to 3) but is still severe. The most plausible reason for the overprediction
14 of SO₂ at Cape Grim might be the lack of SO₂ oxidation by ozone on sea salt particles,
15 which is missing in our model. Korhonen et al. (2008) shows a reduction of SO₂
16 concentrations by a factor of 5 in January and a factor of 20 in July at Cape Grim when
17 including SO₂ oxidation on sea spray particles, although their treatment of the reactions
18 might overestimate the SO₂ oxidation rates.

19 Simulated DMS and SO₂ vertical profiles over the Pacific Ocean are compared
20 against two sets of aircraft observations in Figures 4 and 5: PEM-Tropics-A performed
21 during August-October, 1996 (Hoell et al., 1999) and PEM-Tropics-B during March-
22 April, 1999 (Raper et al., 2001). Note that PEM-Tropics-A DC8 aircraft data is used and
23 most of them were during September 1996, but model evaluation is little changed by
24 comparing with model outputs in September or August-October average. The observed
25 vertical profile data used here are binned into altitude ranges (Emmons et al., 2000).
26 Model outputs are averaged over the observational time period and domain. Simulated
27 DMS vertical profiles are very similar between the two aerosol models, although the
28 surface DMS is different as their emissions are not the same. Both models show good
29 agreement with the observations (mostly within 25th and 75th percentile of observed
30 values), especially capturing a strong concentration decrease from the surface to the free
31 troposphere.

1 In the case of SO₂, even though the agreement is not as good as that seen for DMS,
2 both aerosol models seem to capture the observed magnitude approximately within a
3 factor of two (see Fig. 5). The overall vertical patterns shown in the models are
4 frequently not in agreement with the observations. Except at Hawaii, our model do not
5 capture the enhanced SO₂ concentrations in the boundary layer shown in the observation,
6 even though the model DMS is quite well captured. The poor prediction of SO₂ vertical
7 profile might be due to 1) too much precipitation near tropics in ModelE2 (see Fig. 9 in
8 Schmidt et al., 2014) and 2) too strong vertical transport (e.g., via deep convection over
9 the tropical Pacific Ocean) in the model. The latter can be supported by that the small
10 DMS peak at 8 km and the elevated SO₂ in the upper/free troposphere in the models (see
11 Figs. 4 and 5),. Although the elevated SO₂ mixing ratios might be due to wet scavenging
12 (including aqueous chemistry) of SO₂ being too weak, we did not see any noticeable
13 improvement when increasing SO₂ Henry's law constant by a factor of two in the model
14 (not shown). A large peak in the mid-troposphere at Hawaii in the models results from
15 volcanic SO₂ emissions, while the observations show a similar peak only during March-
16 April 1999, which is heavily influenced by volcanic emissions (Thornton et al., 1999).
17 During August-October 1996, the observations at Tahiti and Easter Island show transport
18 of volcanic SO₂ emissions in the middle and upper troposphere (Thorton et al., 1999),
19 which is not captured in the model. Since our model includes only continuous volcanic
20 emission with a yearly resolution, our model fails to simulate variability in volcanic SO₂
21 emissions at higher time resolution.

22 **5.2 Aerosol mass budgets and distributions**

23 Globally and annually averaged budgets of aerosols in the ModelE2-TOMAS model
24 are shown in Table 4. For the sulphate and EC budgets, we use the ACCMIP multi-model
25 mean from Shindell et al. (2013) and Lee et al. (2013a), which is based on 8 ACCMIP
26 models using the same AR5 emission scenario. Note that the biomass burning emission in
27 this study is GFEDv3 inventory averaged from 1997 to 2009, while the ACCMIP models
28 use GFEDv2 inventory averaged from 1997 to 2006. We do not compare with the
29 AEROCOM phase 2 multi-model mean presented in Myhre et al. (2013a) because the
30 aerosol budgets in Myhre et al. (2013) are for anthropogenic aerosols, which is defined as

1 the difference between the present-day run and pre-industrial run. For the lifetime and
2 deposition rate coefficient budgets, we compare with the AEROCOM Phase 1 multi-
3 model mean presented in Textor et al. (2006) –hereafter, referred to as AEROCOM Phase
4 1.

5 For sulphate, the ModelE2-TOMAS total source rate is lower than the ACCMIP
6 mean (43.7 Tg yr^{-1} vs. $51.7 \text{ Tg S yr}^{-1}$), and the global burden is the same as the ACCMIP
7 mean burden (0.67 Tg S) due to the slightly longer lifetime in the ModelE2-TOMAS
8 model (5.7 days vs. 5.0 days). Note that the GISS-E2-R-TOMAS model used for
9 ACCMIP is almost identical to ModelE2-TOMAS evaluated here except for the sulfate
10 modelling. The sulphate and DMS emissions used in GISS-E2-R-TOMAS are identical to
11 those used in ModelE2-OMA in this paper. For dry deposition coefficient (the inverse of
12 the lifetime), ModelE2-TOMAS has a particularly small value. However, the longer
13 overall sulphate lifetime is contributed by both dry and wet deposition, rather than dry
14 deposition. When increasing the dry deposition coefficient to the AEROCOM Phase 1
15 mean alone, the overall lifetime is decreased from 5.6 days to 4.8 days. Doing the same
16 for wet deposition with no change in the dry deposition, the overall lifetime decreases
17 from 5.6 days to 4.5 days.. Wet deposition accounts for 98% of the total deposition in
18 ModelE2-TOMAS, which is much higher than AEROCOM Phase 1, and convective
19 clouds contribute 27% of the wet deposition.

20 Global-annual EC in ModelE2-TOMAS is 0.19 Tg, which is very comparable to the
21 ACCMIP mean (0.16 Tg). Similar to sulphate, wet deposition contributes >95% of total
22 deposition of EC, which is higher than other ACCMIP models (see Table 3 in Lee et al.,
23 2013a), and 24% of wet deposition is by convective clouds. For EC and OM, their
24 lifetimes are longer than the ACCMIP mean as well as AEROCOM Phase 1 means but
25 still within the standard deviation. Despite smaller dry deposition coefficients for EC and
26 OM compared to the AEROCOM mean, their wet deposition coefficients are quite
27 comparable to the AEROCOM mean.

28 The global annual burden of sea salt and dust in ModelE2-TOMAS are 3.6 Tg and 9.1
29 Tg, respectively. For sea salt and dust, dry deposition is as important as wet deposition
30 due to their large particle sizes, accounting for 68% and 52% of total deposition,
31 respectively. Since the size coverage of sea salt and dust in our model do not necessarily

1 match with those in the AEROCOM Phase 1 models, we do not compare the lifetime and
2 removal rate coefficients, which is strongly influenced by the upper size cutoff used in
3 their emissions.

4 Despite the same host model and the same anthropogenic emission scenarios as the
5 ModelE2-TOMAS model, the ModelE2-OMA model shows significantly different
6 aerosol mass budgets (in Table 5), which must arise from using different deposition
7 assumptions and other aerosol modelling treatments (see Section 2 for the details).
8 Sulphate burden and lifetime in ModelE2-OMA is roughly half of that in ModelE2-
9 TOMAS. The total source rate of SO₄ is about 20% higher than ModelE2-TOMAS and is
10 close to the ACCMIP mean value. The ModelE2-OMA model has a shorter lifetime for
11 EC, leading to ~40% lower burden compared to ModelE2-TOMAS. The OM burden is
12 quite similar between two models. SOA formation rate is slightly different (14.6 Tg yr⁻¹
13 for ModelE2-OMA and 17.1 Tg yr⁻¹ for ModelE2-TOMAS), but the difference is only a
14 small portion (about 3-4%) of the total OM source rate. The sea-salt emission rate is
15 lower than that in ModelE2-TOMAS due to the maximum size cutoff of 8 μm assumed in
16 sea-salt emission, but its burden is more than a factor of two higher. For dust particles,
17 the emission rate is higher than that in ModelE2-TOMAS due to the coarser size cutoff in
18 their emissions (i.e. up to 16 μm), and the burden is higher. For sea salt, the contribution
19 of wet deposition to total deposition is more than 2 times higher than that in ModelE2-
20 TOMAS. Unlike ModelE2-TOMAS, ModelE2-OMA has nitrate aerosol, which has a
21 global burden of 1.6 Tg with a lifetime of 6.4 days.

22 The removal rate coefficient of dry deposition is about 50-60 times higher for
23 sulphate, EC and OM in ModelE2-OMA, making it more comparable to the AEROCOM
24 mean values. The lower dry deposition rates with the ModelE2-TOMAS model is likely
25 due to the size-dependent dry deposition parameterization. The dry deposition velocity is
26 not saved in the ModelE2 output currently, so alternatively we refer to the Figure 1 in
27 Adams and Seinfeld (2001) that presents the global and annual-average of size-resolved
28 dry deposition velocities in GISS-TOMAS (sulphate alone) compared to the size-
29 independent one. Although the dry deposition velocities from ModelE2-TOMAS might
30 not be exactly the same as the ones in the model version used in Adams and Seinfeld
31 (2001) due to the updates made in ModelE2 (e.g. the boundary layer module), this point

1 should be valid because the dry deposition parameterizations in both models have been
2 little changed. Despite the large differences in dry deposition rates for accumulation
3 mode particles, dry deposition is a fairly minor removal pathway in both models.

4 Figure 6 shows simulated global distributions of annual-mean concentrations of
5 ModelE2-TOMAS sulphate, EC, OM, sea salt and dust in the lowermost layer. The
6 sulphate concentrations are high over industrial regions, driven by the SO₂ emissions and
7 OH/H₂O₂ oxidant concentrations. Simulated EC and OM concentrations are high over the
8 biomass burning regions and the industrial regions, especially East Asia and South Asia,
9 but the OM concentrations are particularly pronounced over biomass burning regions due
10 to their higher emissions. Due to the SOA formation, the OM concentrations over
11 Midwest US and Central Siberia are also noticeably high. The sea salt concentrations are
12 distributed fairly uniformly over the oceans but are higher over the Southern Ocean and
13 lower over the oceans near the tropics as expected due to wind speed variations. The dust
14 concentrations are pronounced over the source regions such as Northern Africa, Arabia,
15 Northern/Northwestern China, and Australia.

16 Figure 7 shows zonal distributions of annual-mean aerosol concentrations in
17 ModelE2-TOMAS. The sulphate concentrations are highest between 0° N and 50° N due
18 to the high anthropogenic emissions in the NH. Over the SH, the sulphate concentrations
19 are mostly a result of DMS oxidation. The zonal-mean EC and OM concentrations are
20 high from the tropics to ~50° N. Similar to Fig. 6, the high EC and OM concentrations
21 are around 30° N to 50° N due to fossil fuel emissions, but the OM concentrations are
22 also large around the tropics due to biomass burning emissions. The two small spikes
23 shown the EC and OM concentrations between 10° S and 10° N are a result of the
24 injection height used in the biomass burning emissions. Since a significant amount of
25 sulfate and OM are also formed through chemical reactions in the atmosphere, their
26 vertical gradients are relatively small. The sea salt concentrations are high from 60° S to
27 50° N with a peak around 30° S to 60° S due to the large open ocean in the SH. A strong
28 dust plume is shown at around 0°-30° N due to the large Northern African and Middle
29 East dust emissions, and a small dust plume at around 30° S due to Australian emissions.
30 In the model, the PM10 concentrations in upper troposphere are dominated by sulfate and
31 dust particles.

1 **5.3 Aerosol mass concentrations evaluation**
2

3 The simulated surface-layer mass concentrations of aerosols are evaluated against
4 various observations: 1) sulphate/sea salt/dust concentrations at 23 long-term observation
5 sites operated by University of Miami (e.g. Prospero and Bonatti, 1969; Savoie and
6 Prospero, 1989; Arimoto et al., 1990); 2) speciated PM_{2.5} concentrations from the
7 Interagency Monitoring of Protected Visual Environment (IMPROVE) sites in the United
8 States that are annual-averages from 2000 to 2008 (Debell et al., 2006; Hand et al.,
9 2011); 3) speciated PM_{2.5} concentrations from various European observations (Putaud et
10 al., 2010) (hereafter, referred to as European sites); 4) a large set of PM_{2.5} observations
11 assembled in support of the Global Burden of Disease Study (GBD Study 2010,
12 <http://www.globalburden.org>); 5) deposition flux measurements obtained from Ginoux et
13 al. (2001), Tegen et al. (2002), and Mahowald et al. (2009). The PM_{2.5} dataset from the
14 GBD Study consists of a worldwide set of annual-average PM_{2.5} largely drawn from
15 official monitoring networks for 2005 (in some cases from 2004-2006). The GBD PM_{2.5}
16 dataset includes the IMPROVE network and the European site measurements, so we only
17 present the PM_{2.5} evaluation with the GBD dataset. For details of the GBD PM_{2.5}
18 dataset, the reader is referred to the description in the supplementary material in Shindell
19 et al. (2011).

20 Figure 8 compares the model annual-mean surface-layer sulphate mass concentrations
21 to the observations from the IMPROVE network, the European sites, and the University
22 of Miami network. Simulated sulphate agrees well with observations, mostly within a
23 factor of two. Compared to ModelE2-OMA, the ModelE2-TOMAS model shows better
24 agreement in the European sites (i.e. LMNB=-0.06 and LMNE=0.13 in Fig. 8b) but
25 worse in the IMPROVE sites (i.e. LMNB=0.06 and LMNE=0.16 in Fig. 8a) and the
26 remote oceanic sites (i.e. LMNB=0.04 and LMNE=0.22 in Fig. 8c). Over the US, both
27 models overpredict systematically at lower observed concentrations (i.e. below 1 $\mu\text{g m}^{-3}$
28 of measured SO₄ concentrations), which are mostly located over the western US.

29 Monthly mean surface-layer sulphate concentrations are evaluated using observations
30 from the University of Miami in Figure 9. The simulated sulphate mass concentrations
31 from both models usually falls within the standard deviation of the observed values. The

1 sulphate predictions in both ModelE2-OMA and ModelE2-TOMAS become quite similar
2 over the SH because about a factor of two lower DMS emissions are used in ModelE2-
3 TOMAS. Using the same DMS emissions as ModelE2-OMA, ModelE2-TOMAS tends to
4 overpredict sulphate concentrations noticeably over the SH (not shown), and the higher
5 SO₄ concentration with ModelE2-TOMAS could be explained by 1) a longer lifetime due
6 to different deposition parameterizations, 2) letting all SO₄ formed from aqueous
7 oxidation to evaporate without accounting for cloud evaporation, 3) a stronger oxidation
8 state resulted from un-degraded photolysis rates by aerosol optical depth (more SO₄ can
9 be formed from DMS oxidation). Despite the fact that ModelE2-OMA accounts for the
10 heterogeneous SO₄ formation on dust particles, SO₄ concentrations in near dust source
11 regions are still higher in the ModelE2-TOMAS model due to the shorter lifetime in the
12 ModelE2-OMA model.

13 Simulated annual-mean surface-layer sea salt concentrations are evaluated with the
14 IMPROVE network, the European sites, and the University of Miami network (Fig. 10).
15 Both aerosol models are biased strongly high over the US and Europe but biased low near
16 the tropics. Unlike ModelE2-OMA, ModelE2-TOMAS underpredicts sea salt
17 concentrations at several remote sites (see Fig. 10c). Compared to other aerosol
18 components, the agreement between model sea salt and the observations is worse over the
19 United States and Europe. For the evaluation of monthly mean surface-layer sea salt
20 concentrations against the University of Miami dataset in Figure 11, the model
21 predictions fall within the observed standard deviation at about a half of the 26 sites, but
22 these sites are not necessarily the same between the two aerosol models. The
23 overprediction of sea salt in continents may suggest that ModelE2 has a stronger
24 transport from ocean to in-land, as there is no obvious overprediction over adjacent
25 oceanic sites. The ModelE2-OMA model exhibits a particularly large overprediction over
26 most SH sites. Both models tend to be biased significantly low at some of the sites near
27 the tropics where the observed sea salt concentrations are high. Similar underprediction is
28 also shown in mineral dust (see Fig. 13). This might be due to fast wet scavenging due to
29 overpredicted precipitation in that area (see Fig. 9 in Schmidt et al., 2014).

30 For ModelE2-TOMAS, simulated annual-mean surface-layer dust concentrations are
31 mostly within a factor of two of the measurements (in Fig. 12): 6 sites are excluded in

1 Fig. 12 due to an incomplete annual cycle. ModelE2-TOMAS shows good agreement at
2 the IMPROVE and European sites, while ModelE2-OMA tends to underpredict. This is
3 due to the emission size assumptions (resulting in more clay emissions than ModelE2-
4 OMA) and the differences in the deposition parameterizations (resulting in slightly longer
5 lifetime for clay particles; ~9 days in ModelE2-TOMAS and 6.5 days in ModelE2-
6 OMA). Both models are biased low compared to the University of Miami dataset, and
7 only 7-8 sites among 20 sites fall within a factor of two agreement. A few sites show a
8 severe underprediction for both models, mostly located near the tropics and in SH high
9 latitudes.

10 Monthly mean surface-layer dust concentrations are evaluated using the observations
11 from the University of Miami in Figure 13: no measurement data is available at 3 sites (Reunion
12 Island, Invercargill, and Marion Island), but we still include them to compare the
13 two aerosol model predictions. The model captures the observed annual cycle of dust
14 very well at most NH sites but not in the SH sites except Norfolk and Mawson. The
15 model captures the observed magnitude well at sites located relatively near the source
16 regions (e.g., Sal Island, Barbados, Bermuda for African dust; Jeju Island and Hedo for
17 Asian dust; Cape Grim and Norfolk Island for Australian dust). Both models underpredict
18 dust concentrations in the NH/SH high latitude (45°) regions except Mace Head, but the
19 simulated dust seems to be within the observed standard deviation as the observations
20 have a large standard deviation for dust. The observed peak concentration at Heimaey
21 Iceland is the second highest after Sal island. Our models underpredict this site severely
22 probably because our dust emission parameterization is not designed to simulate a dust
23 event in humid areas such as Iceland. Prospero et al. (2012) points out that dust emissions
24 at high latitudes (e.g., Alaska and Iceland) are mostly due to individual dust events or
25 single seasons and link large dust events at Heimaey Iceland during 1997 to 2002 with
26 glacial outburst floods.

27 Figure 14 compares simulated annual-mean dust deposition fluxes against
28 observations obtained from Ginoux et al. (2001), Tegen et al. (2002), and Mahowald et
29 al. (2009). Data is classified by the influencing source region, presented in different
30 colours in Fig. 14. Except for the minor source category, the model dust deposition fluxes
31 tend to be underestimated at most locations and agree with observations only within a

1 factor of 5-8 on average. This may indicate that dust emissions are too low in ModelE2,
2 but deposition fluxes measurements could contain large particles especially close to
3 source regions (e.g., Duce, 1995) or local emissions (Uematsu et al., 1985) that are not
4 simulated in the model. For ModelE2-TOMAS, the upper size limit being 10 μm , which
5 is too small for the dust particles near the sources, can explain some of the severe
6 underprediction, and Lee et al. (2009) also shows similar disagreement using the GISS-
7 TOMAS model.

8 Figure 15 shows simulated annual-mean surface-layer EC and OM concentrations
9 compared against the observations from the IMPROVE network and the European sites
10 from Putaud et al. (2010). The simulated EC and OM in both models agree very well at
11 the IMPROVE sites (for EC, LMNB=-0.08 to 0.05; for OM, LMNB=-0.17 to -0.08).
12 Note that we applied an OM/OC ratio of 1.4 to the IMPROVE network to make it
13 consistent with our model assumption, but the IMPROVE OM data provided to us was
14 based on the ratio of 1.8. Over Europe, the model predictions are still reasonable (within
15 a factor of 2-3), but the agreement is slightly worse than the IMPROVE sites. Since these
16 sites are mostly adjacent to the source/emissions, the good agreement suggests that the
17 emission inventory (used in this study) is well represented for these regions.

18 Figure 16 compares simulated annual-mean PM2.5 concentrations against the GBD
19 dataset. Note that ModelE2-OMA includes nitrate mass into its PM2.5, and the nitrate
20 contributes to PM2.5 rather significantly. The GBD data is classified/presented by a
21 region listed in Figs. 16a and b. The aerosol models capture the observation quite well in
22 most locations (overall LMNB is -0.08 to -0.03 and LMNE is 0.2 to 0.25), but both
23 models show the worst agreement for Oceania regions (LMNB/LMNE=0.4-0.58) and
24 Latin America (LMNB/LMNE =-0.33 to -0.48). The PM2.5 overprediction in Oceania is
25 mainly due to too much fine mode sea-salt particles (the overall agreement in Oceania is
26 little influenced by sulfate or dust particles). Note that the sea salt comparison to the
27 Miami dataset (in Figures 11) shows severe underpredictions in several sites in Oceania
28 because their concentrations are likely dominantly by coarse mode sea salt particles. The
29 underprediction of PM2.5 in Latin America might be related to the biomass burning
30 emissions. It is consistent with the model AOD being biased low over biomass burning
31 source regions that is shown in Section 5.4. Model evaluation with the observed PM10

1 concentrations using the IMPROVE and European sites was also performed and is similar
2 to the PM_{2.5} evaluation results (not shown): LMNB and LMNE are -0.01 and 0.17 for
3 ModelE2-TOMAS and 0.0 and 0.29 for ModelE2-OMA, respectively.

4 **5.4 Aerosol optical depth evaluation**

5 Simulated annual-mean clear-sky aerosol optical depths (AODs) at 550 nm are
6 compared with observations from the Terra MODIS (MODerate resolution Image
7 Spectroradiometer; e.g. Abdou et al., 2005; Remer et al., 2008) and MISR (Multiangle
8 Image SpectroRadiometer; e.g. Diner et al., 1998; Kahn et al., 2005) satellite instruments
9 averaged over 2004-2006 (Fig. 17). Specifically, we use Terra MODIS Level 3
10 (MOD08_M3.051), Aqua MODIS Level 3 (MYD08_M3.051), and Terra MISR Level 3
11 (MIL3MAE4), which are monthly products with 1x1 degree resolution. We also use the
12 Deep Blue AOD (e.g., Hsu et al., 2006) from Terra MODIS to increase its spatial
13 coverage, and all the satellite data was obtained from
14 <http://disc.sci.gsfc.nasa.gov/giovanni>. When the Deep Blue AOD and the “regular” AOD
15 from Terra MODIS are both available, the former is used. However, we notice that the
16 Deep Blue AOD shows some local hotspots (e.g. over South America) that are not shown
17 in the “regular” MODIS AOD as well as the MISR AOD. Our model calculates clear-sky
18 AOD by including only AOD values calculated in model locations where clouds are not
19 present (i.e., cloud-free grid-box only). The ModelE2-TOMAS and ModelE2-OMA
20 models capture the broad spatial features seen in the satellite measurements: 1) very high
21 AODs over desert regions in and near Northern Africa and the Arabian Peninsula
22 associated with mineral dust, 2) the band of locally enhanced AOD over the Southern
23 Ocean associated with sea salt, 3) high AODs over East Asia and India due to high
24 anthropogenic emissions. However, the models underestimate the AOD over East Asia
25 and the Indo-Gangetic plain, especially compared to MODIS. Including nitrate in
26 ModelE2-OMA, the AOD predictions increase particularly over Europe and East Asia.
27 Without nitrate, they become quite similar to the ModelE2-TOMAS AOD values in the
28 regions where the model nitrate predictions are significant. The models show an
29 enhanced AOD over the biomass burning regions such as tropical South America, Africa
30 and Indonesia but it is clearly underestimated. The simulated AOD in North America and

1 high latitude regions appears also to be lower than the satellite observations. Comparing
2 with ModelE2-OMA AOD, ModelE2-TOMAS shows a stronger AOD over Africa due to
3 its higher dust burden and a lower AOD over the marine areas especially the Southern
4 Ocean associated with sea salt.

5 We present spatial correlations and biases between the models and the satellite data
6 (Table 6). For these analyses, the annual-mean satellite AOD fields are regridded to
7 $2^\circ \times 2.5^\circ$ horizontal resolution, and the models are sampled only where the satellite AOD
8 is available. Correlation coefficients between the model and satellite AODs are around
9 0.6-0.7 for ModelE2-TOMAS and around 0.4-0.5 for ModelE2-OMA. Given that the
10 correlation coefficient between MODIS and MISR is 0.79, the ModelE2-TOMAS model
11 shows a good correlation with these satellite data. Both models show better correlations
12 with MISR AOD. Compared to ModelE2-OMA, ModelE2-TOMAS shows a strong
13 negative bias (about -29 to -34%) for both satellite datasets because of noticeably low
14 AODs over the oceanic regions (where sea salt is dominant) shown in Fig. 16 and
15 possibly because of missing component such as nitrate aerosols, which contributes to
16 AOD significantly over Europe and China in the ModelE2-OMA model. Over Europe,
17 the ModelE2-OMA model overpredicts AOD due to nitrate though. Without nitrate in
18 ModelE2-OMA, the normalized mean bias (NMB) falls from 8-16% to -16 to -21%. Both
19 models show lower AOD over China, India, and biomass burning regions and a similar
20 underprediction is shown in by the ACCMIP models (Shindell et al., 2013), indicating a
21 possibility of aerosol emissions being underestimated in these regions.

22 Figure 18 presents simulated monthly mean AODs compared against AERONET
23 (AErosol RObotic NETwork; Holben et al., 1998; Holben et al., 2001) measurements at
24 28 sites that represent the following characteristic regions: polluted continental, marine,
25 biomass-burning, and dusty regions (see Table 7 for individual site information). Both
26 ModelE2-TOMAS and ModelE2-OMAunderestimate the maximum AOD during
27 summer by a factor of 2-3 in the biomass burning sites (1 to 6) but capture the observed
28 annual cycle quite well ($R > \sim 0.9$). Unlike other biomass burning sites, simulated AODs
29 at Ilorin and Banizoumbou are comparable to the observations possibly due to the
30 influence of mineral dust. The agreement between the models and the AERONET AOD
31 is generally good in the dusty regions (9 to 16). The ModelE2-TOMAS model shows a

1 slight overprediction of AOD during spring at Capo Verde and Bidi Bahn which are
2 located near the African dust sources, although it shows a good agreement at Barbados
3 where is also influenced by African dust.

4 For ModelE2-TOMAS, all polluted continental sites (17 to 24) show large
5 underpredictions, while the model tends to capture the observed annual cycle well.
6 However, the model surface aerosol mass concentrations agree well with the observations
7 from IMPROVE network and several European sites (see Figs. 8, 10, 12, 15, and 16).
8 This might be due to the fact that column AOD depends on many additional factors (e.g.,
9 optical properties and vertical distribution of aerosols) and provides a measure of total
10 radiatively active aerosols in the atmosphere. Misrepresentation of these factors or
11 missing a chemical component could introduce a bias in the model AODs. For ModelE2-
12 OMA, the model AOD without nitrate is quite similar to the ModelE2-TOMAS AOD.
13 Including nitrate in ModelE2-OMA, it simulates the annual-mean AERONET AOD
14 relatively well but the observed annual cycle worse. Although the inclusion of nitrate is
15 helpful for the underpredicted AOD in polluted regions, the overprediction of wintertime
16 AOD suggests that the model nitrate is too large.

17 With the ModelE2-TOMAS model, the oceanic sites (25 to 30) are generally
18 underpredicted roughly by a factor of two except for Bermuda (25) and Lanai (26). By
19 contrast, the ModelE2-OMA model captures the observed magnitude relatively well but
20 overpredicts at Bermuda and Lanai. Note that Bermuda (23) and Rottnest Island (27) are
21 influenced by long-range transported mineral dust. The ModelE2-OMA model predicts
22 AOD that is more comparable to observations in these oceanic sites than ModelE2-
23 TOMAS. Compared to ModelE2-OMA, the underprediction of sea salt concentrations at
24 the remote sites (shown in Figs. 10 and 11 in Sect. 5.3) and the underpredictions of
25 AODs in the remote oceanic sites in ModelE2-TOMAS may be due to a faster sea salt
26 removal rate (see global budgets in Table 4 for the details). It is worth mentioning that
27 the inclusion of marine organic aerosols, which are not included in this paper, may not
28 increase AOD noticeably in remote oceanic sites. Using the same host model, Tsigaridis
29 et al. (2013) shows that Southern Ocean AOD is quite insensitive to the inclusion of
30 marine organic particles but is strongly sensitive to the sea-salt emissions
31 parameterization (see Figure 9 of Tsigaridis et al. (2013)).

1 **5.5 Aerosol number budgets and its distributions**
2

3 This section includes only ModelE2-TOMAS results, as ModelE2-OMA does not
4 predict aerosol number concentrations. Global mean number budgets for all three
5 simulations, including two sensitivity runs for nucleation rates, are presented in Table 8
6 (see Section 4.2 and Table 1 for the details of run descriptions). Compared to the BASE
7 run, global mean CN3 (particles with dry diameters larger than 3 nm), CN10 (particles
8 with dry diameters larger than 10 nm), and CN100 (particles with dry diameters larger
9 than 100 nm) concentrations in the LowNUC run are decreased by 74%, 33%, and 7% in
10 the troposphere and 29%, 17%, and 3% in the lowermost layer, respectively. The smaller
11 impact on CN in the lowermost layer is due to the presence of the primary emissions near
12 the surface. It is also because the binary nucleation parameterization used in this model
13 produces few particles in the boundary layer; however, nucleated particles in the
14 upper/free troposphere may be carried down to the surface and influence CN there. Since
15 the LowNUC run produces fewer nucleated particles than the BASE run, each nucleated
16 particle is more likely to grow more efficiently to form CCN-sized particles, as there is
17 less competition for sulphuric acid and condensable organics. We can see this using the
18 number budgets in Table 8. The increase in CN70 with the BASE case nucleation (i.e.,
19 BASE CN70 – NONUC CN70) is 51 cm^{-3} for the nucleation rate (J_3) of $0.131 \text{ cm}^{-3} \text{ s}^{-1}$
20 and that with the LowNUC case nucleation, 37 cm^{-3} for J_3 of $0.013 \text{ cm}^{-3} \text{ s}^{-1}$. In the BASE
21 run, J_3 is 10 times higher but the CN70 increase by nucleation is only ~ 1.4 times higher
22 than those in the LowNUC run.

23 Aerosol number burdens normalized by tropospheric volume are dramatically
24 reduced when nucleation is turned off in the NoNUC run. For example, global mean CN3
25 and CN10 are decreased by 95% and 76% in the troposphere and by 42% and 31% in the
26 surface layer. We found that 24% of tropospheric CCN-sized particles (i.e., CN100) and
27 $\sim 10\%$ of surface-layer CCN-sized particles result from binary nucleation in our model;
28 the contribution of the nucleated particles to the CCN concentrations is larger as the
29 cutoff size (e.g. 100 nm in CN100) gets lower.

30 Figure 19 shows global distributions of annually averaged CN3, CN10, and CN100 in
31 the lowermost layer for the BASE, LowNUC, and NoNUC runs, and Fig. 20 presents
32 their zonal distributions. For the CN3 distributions, the BASE run shows a high

concentration (over 5000 cm⁻³) in the upper/free troposphere and the entire Antarctica troposphere due to nucleation and over polluted continental regions due to the primary emissions (see Figs. 19a and 20a). Note that the CN3 near the surface in Antarctica comes from nucleated particles from the upper/free troposphere, as the binary parameterization does not predict nucleation in the boundary layer. CN3 in the LowNUC run (Figs. 19d and 20d) is decreased significantly in the free troposphere and in the surface layer over Eurasia and western Antarctica (fewer nucleated particles formed above subside to the surface). When nucleation is switched off, CN3 is very close to CN10 near to the surface (Figs. 19g and h) because nucleation contributes most CN between 3 nm to 10nm. For CN10, its spatial pattern over the continents is quite similar among the runs as it is mostly driven by primary emissions except for some locations heavily influenced by the nucleated particles formed in the upper/free troposphere (Figs. 19b, e and h and Figs. 20b, e, h). A pronounced difference in CN10 is shown over oceans, indicating a larger contribution of nucleated particles to CN10 in these regions. This is consistent with Merikanto et al. (2009), which shows higher contribution of nucleation to CN over oceanic regions. CN100 differs little among runs except for SH high latitudes where binary nucleation plays an important role (Figs. 19c, f, and i). This is consistent with Pierce and Adams (2009) showing that the exhibited high SH latitude region for the most positive changes in CCN(0.2%) by turning on binary nucleation and, again, with Merikanto et al. (2009) showing 65% of CCN(0.2%) in Antarctica resulted from upper tropospheric nucleation. CN100 shows a maximum at 10° N to 40° N because most anthropogenic primary emissions are located in mid-NH latitudes. Rather surprisingly, dust particles in our model contribute to CN100 quite significantly over the source regions. This is opposite to the results from GISS-TOMAS (Lee et al., 2009), which shows a 10-20% reduction in CCN(0.2%) when introducing mineral dust emissions. Despite the direct source of CCN-sized particles from dust emissions, CCN and ultrafine particles that grow to become CCN are scavenged via coagulation with coarse dust particles, and dust particles compete for condensable sulfuric acid, leading to a slower growth rate of ultrafine particles.

30 5.6 Aerosol number evaluation

31

1 The CN measurement dataset compiled by Spracklen et al. (2010) is used to evaluate
2 simulated annual-mean CN concentrations (in Fig. 21) and monthly mean CN
3 concentrations (in Fig. 22). Details of the measurement procedures for each site,
4 including the instrument type and minimum cutoff diameter (varying from 3nm to 14
5 nm), can be found in Table 2 of Spracklen et al. (2010). Due to an incomplete annual
6 cycle, we excluded Mt. Waliguan, Finokalia, Listvyanka, and Weybourne. The
7 measurement sites are classified into three categories: FT (free troposphere; 1 to 8 in Fig.
8 22), MBL (marine boundary layer; 9 to 15 in Fig. 22), and CBL (continental boundary
9 layer; 16 to 32 in Fig. 22). We sampled the model values to match the altitude of each
10 measurement site except for the free troposphere sites that use an altitude 30% lower
11 because it improved the annual cycles prediction significantly. Previous studies pointed
12 out that free tropospheric sites can be influenced by upslope winds that carry the
13 planetary boundary layer air, so it cannot be assumed to be in the free troposphere all the
14 time (e.g., Baltensperger, 1997; Collaud Coen et al., 2011). Our model does not seem to
15 simulate this well, so sampling the model predictions at lower altitude (i.e., 30% lower)
16 helps to increase the influence of PBL air. In Fig. 21, a whisker plot is used to present the
17 three run results; the maximum of the whisker line for the BASE run; the circle symbol in
18 the middle of the whisker line for the LowNUC run; the minimum for the NoNUC run.

19 On average, the annual-mean CN concentrations in the model agree with the
20 observations well for the all three categories (LMNB= -0.26 to 0.16; LMNE=0.13 to
21 0.22; see Fig. 21), although the LowNUC simulation shows the best agreement to
22 observation. Simulated annual cycles at individual sites also show that the model captures
23 the measured magnitude reasonably but overpredicts the CN during winter (November to
24 March) that is worse at the CBL sites and results in poor seasonality (especially in the
25 BASE run). The poor seasonality in ModelE2-TOMAS in all three runs suggests that
26 other factors may play a role such as missing seasonal variation in primary emissions or
27 scavenging that causes the poor seasonality rather than nucleation. The evaluation of CN
28 at South Pole shows that the model predicts too strong nucleation throughout the year.

29 Figure 23 compares the observed size distributions at six European sites during winter
30 (DJF: December to February; Figs. 23a to f) and summer (JJA: June to August; Figs. 23d
31 to l) to the model. The observed size distributions are averaged during morning,

1 afternoon, and night, while the model results are not broken into the three periods. This
2 data is obtained from Putaud et al. (2003) and, for the same sites shown in Figure 21, the
3 temporal coverage used in Putaud et al. (2003) is not necessarily matched with them.
4 Most sites are reasonably close to the dataset used in Figure 21. However, for
5 Jungfraujoch, the total CN concentrations summed from the size distribution data, which
6 covers from June 1997 to May 1998, is about a factor of two lower than the CN shown in
7 Figure 21, which covers from 1995 to 1999 and 2003-2007. As expected, the simulated
8 size distributions of the model simulations differ according to what nucleation scheme is
9 used. The impact of nucleation is more notable for particles smaller than ~50 nm, as
10 expected, and also during winter. In general, the higher nucleation rates tend to
11 overpredict Aitken model particles at most sites. For Harwell (Figs. 23c and i), particles
12 below 50 nm are overpredicted even without nucleation. For the summer season, the
13 model shows less sensitivity to nucleation rates and has better skill at capturing the
14 observed size distributions.

15 The observed CN5, CN15, and CN120 concentrations from the LACE campaign
16 (Petzold et al., 2002) are compared with three model runs (Figs. 24a to c). Below 700-
17 800 mbar, all three model runs predict concentrations roughly within the observed CN5,
18 CN15, and CN120 ranges. Above 700-800 mbar, the BASE simulation overpredicts CN5
19 and CN15 by approximately an order of magnitude and 2-3 times, respectively. For the
20 NoNUC simulation, it captures the lower side of the observed CN15 but fails to capture
21 the increasing CN5 concentrations with height above 600 mbar. All model runs basically
22 show almost identical CN120 concentrations and fall on the lower edge of the observed
23 range.

24 We compare CN3 vertical profile measurements averaged into the 3 latitude bands
25 over the Pacific Ocean (Clarke and Kapustin, 2002) with the model (Figs. 24 d to f). The
26 simulated CN3 profiles in LowNUC agree well with the observation, capturing the
27 increasing CN3 with height. Although the BASE run shows the increasing pattern
28 correctly, it overpredicts CN3 severely above approximately 6 km. The NoNUC run
29 disagrees with the observations for all latitudes and altitudes and clearly fails to reproduce
30 the high number concentrations in the upper troposphere that result from nucleation.

1 The observed Aitken mode and accumulation mode concentrations and size
2 distributions in the marine boundary layer (Heintzenberg et al., 2000) are compared with
3 the model in Figs. 25 and 26. The measurements are aggregated into 15-degree latitude
4 ranges. For the Aitken mode (Fig. 25a), the LowNUC run shows the closest agreement to
5 observations. However, whereas the observations show higher concentrations in the SH
6 than in the NH, all model simulations show the opposite tendency. Similarly, other
7 global models with binary nucleation show underpredicted CN concentrations in the SH
8 and either well-simulated or overpredicted CN in the NH (e.g. Easter et al., 2004;
9 Spracklen et al., 2005; Pierce and Adams, 2006; Pierce et al., 2007; Trivitayanurak et al.,
10 2008; Mann et al., 2010; Wang et al., 2011). For the accumulation mode (Fig. 25b), the
11 three model runs are quite similar to each other and are within the observed range.

12 In Fig. 26, ModelE2-TOMAS captures the bimodal size distribution shown in the
13 observations reasonably, which is mainly determined by the activation diameter assumed
14 in wet deposition (and cloud processing): ~80 nm as an activation diameter
15 (supersaturation of 0.2%) for both large-scale clouds and convective clouds in the model.
16 Note that ModelE2-TOMAS cannot capture the observed bimodal distribution when a
17 supersaturation of 1.0% is assumed for convective clouds, unlike the other TOMAS
18 models (e.g. Pierce et al., 2007; Trivitayanurak et al., 2008): a peak at around 20-30 nm
19 appears when the supersaturation of 1.0% is assumed for convective clouds. This
20 suggests that, compared to GISS GCM II', convective clouds are more frequent in
21 ModelE2. Although the model captures the observations successfully using fixed
22 supersaturation assumptions, future work is needed to link the in-cloud supersaturation to
23 cloud and aerosol properties.

24 Simulated CCN concentrations are compared against a dataset of CCN measurements
25 compiled by Spracklen et al. (2011) in Figs. 27 and 28: see Table 1 in Spracklen et al.
26 (2011) for the details regarding each site. The CCN dataset includes a total of 277
27 measurements at 80 locations using various instruments from 1971 to 2009.
28 Approximately 70% of the observations were taken after 1990. Most have sampling
29 periods of days to weeks except the observations at Cape Grim and Mace Head. For Cape
30 Grim and Mace Head, an annual cycle is available, so we present them separately in Fig.
31 28. In Fig. 27, the CCN data is divided into two groups: CCN in the MBL (marine

1 boundary layer) and CCN in the CBL (continental boundary layer). Note that all CCN
2 measurements used here are in the boundary layer. For CCN in the CBL, the model CCN
3 shows good agreement with the observation in all three simulations ($LMNB=0.11\text{-}0.2$
4 and $LMNE=0.31\text{-}0.34$). For CCN in the MBL, the model predictions are, on average,
5 within a factor of 1.5-2 of the observations for all three runs but, relative to several
6 measurements of CCN concentration between ~ 100 and $\sim 300 \text{ cm}^{-3}$, are biased high, by
7 roughly a factor of two.

8 For the annual cycle of CCN concentrations at Cape Grim and Mace Head (Fig. 28),
9 the model overpredicts in all months even without nucleation ($NMB>1.0$). The CCN
10 overpredictions at Cape Grim might be influenced by overpredicted SO_2 (shown in Fig.
11 3), which could lead to overpredicted condensational growth of Aitken mode particles.
12 As with the SO_2 evaluation in Fig. 3, the CCN overprediction decreases by sampling
13 adjacent grids toward the ocean (not shown), but the model CCN is still higher than the
14 measurements. Given the fact that most CCN observations have very short duration (days
15 to weeks) in a single year and, according to Spracklen et al. (2011), the relative
16 uncertainties in the measurement data range from about 5-40%, mostly in 10-20% very
17 roughly, the overall model-to-observation agreement is satisfactory.

18

19 6. Conclusions

20

21 We have implemented the Two-Moment Aerosol Sectional (TOMAS) microphysics
22 model into the new version of the GISS GCM (i.e. ModelE2), called “ModelE2-
23 TOMAS”. This paper has compared the global budgets of ModelE2-TOMAS to other
24 global aerosol models and evaluates the model with various observations such as aerosol
25 precursor gas concentrations, aerosol mass and number concentrations, and aerosol
26 optical depth.

27 Global budgets of aerosols and aerosol precursor gases in ModelE2-TOMAS are
28 similar to those in other global aerosol models, and the ModelE2-TOMAS model agrees
29 reasonably (mostly within a factor of two) with long-term observed aerosol precursor gas
30 and aerosol mass concentrations. The model captures the broad spatial features shown in
31 the MODIS and MISR annual-mean AOD distributions as well as the observed seasonal

1 trends of AOD at several AERONET sites. The model predicts the observed annual-mean
2 CN (the minimum cutoff varying from 3nm to 14 nm) concentrations very well and the
3 observed vertical profiles of aerosol number over Germany (i.e., LACE campaign) and in
4 the marine boundary layer. For CCN, the model shows good skill in capturing the
5 observations. We conclude that the model is realistic enough to be useful for many types
6 of scientific study.

7 However, the evaluation has also highlighted some weaknesses in ModelE2-TOMAS
8 to be revisited in the future. First, ModelE2-TOMAS predicts too much SO₂ lifted into
9 the upper/free troposphere over the Pacific Ocean possibly due to overly strong
10 convective transport. This is also seen in ModelE2-OMA (and in gaseous tracers),
11 suggesting this might be a host model problem. Second, the model AOD is
12 underpredicted over polluted continents, even though mass concentrations of each aerosol
13 component at the surface are well simulated (or at least not underpredicted) in the model.
14 Missing nitrate in ModelE2-TOMAS may not be the main contributor, as the inclusion of
15 nitrate in ModelE2-OMA decreases its ability to capture the observed seasonality in
16 AERONET polluted continents sites. Third, the model tends to underpredict aerosol
17 loading (and AOD) over biomass burning emission regions. This is a common issue in
18 global aerosol models, and this might be due to underestimation of biomass burning
19 emissions. Fourth, the ModelE2-TOMAS AOD prediction is biased low over the SH
20 high-latitude oceans, which suggests underpredicted sea salt burden in this area. Fifth, the
21 simulated CN seasonality is poor at some CBL sites due to overpredicted CN during the
22 wintertime. Similarly, ModelE2-TOMAS predicts the observed number size distributions
23 over European sites during the summer season reasonably but not for the winter season.
24 The model overpredicts Aitken mode particles during the winter season, which happens
25 even without nucleation, possibly due to a problem in primary emissions representation
26 or a bias in the model scavenging that causes the poor seasonality. Finally, ModelE2-
27 TOMAS seems to predict faster nucleation rates using binary nucleation (Vehkamaki et
28 al., 2002) than other global microphysics models using the same binary nucleation
29 (including GISS-TOMAS). For instance, the observed CN vertical profiles are captured
30 the best when nucleation rates are reduced (our LowNUC run), while Lauer et al. (2005)
31 presents significantly underpredicted CN profiles with the same binary nucleation scheme

1 using ECHAM/MADE. To investigate this issue further, we need to perform a model
2 evaluation against observation-derived nucleation-relevant metrics, which has been done
3 in Westervelt et al. (2013). Because of the fine size assumptions for the primary
4 emissions (see Section 4.2 for the details) and/or the fast nucleation rates using binary
5 nucleation in ModelE2-TOMAS, we do not include boundary nucleation as it
6 overpredicted CN near the surface. We conclude that further work is necessary to
7 improve the realism of some aspects of ModelE2-TOMAS and to better understand the
8 size-resolved physical processes (e.g. microphysics, emissions, and depositions).

9 Comparing the ModelE2-TOMAS model with the ModelE2-OMA model in
10 ModelE2, some aerosol species burdens and concentrations differ significantly, even with
11 using the same host model and identical anthropogenic emissions, mainly because the
12 deposition parameterizations and some of the emission-related assumptions are different.
13 In the case of sea salt and dust, the size ranges assumed by each model are different,
14 resulting in different emission rates and burdens.

15 Analysis of multiple aerosol model results help to identify where a model bias might
16 originate from: aerosol modelling or the host GCM or elsewhere such as emissions. We
17 found that some of the large differences in aerosol predictions between the two aerosol
18 models are due to aerosol modelling. This is valuable information, as this is not easy to
19 constrain using observations. Having more than one aerosol physics representation in the
20 NASA GISS ModelE2 will serve as a useful tool to study the uncertainty in aerosol
21 modelling and to guide our efforts to improve the models.

22

23 **Code Availability**

24 Currently, we do not have a publicly available version of either NASA GISS ModelE2-
25 TOMAS or ModelE2 itself. The ModelE2-TOMAS code may be provided upon the
26 request. However, it may be quite challenging to compile and run ModelE2-TOMAS in a
27 new computer environment as the model has been developed in a NASA NCCS
28 supercomputer (http://www.nccs.nasa.gov/discover_front.html) and a user manual has
29 not been developed.
30 It is worth noting that the public version of ModelE2 (the TOMAS model is not a part of
31 standard version of ModelE2) will be available in the next few years, and the NASA
32 GISS ModelE (which has been used for IPCC AR4 simulations) is publicly available: see
33 details in <http://www.giss.nasa.gov/tools/modelE/>.

1

2 **Acknowledgements**

3 Funding for this research was supported by NASA ACMAP and MAP program.
4 Resources supporting the simulations were provided by the NASA High-End Computing
5 (HEC) Program through the NASA Center for Climate Simulation (NCCS) at Goddard
6 Space Flight Center. We gratefully acknowledge the mission scientists and Principal
7 Investigators who provided the data used in this research effort (EMEP, CASTNET,
8 IMPROVE, and University of Miami networks, AERONET, NASA MODIS and MISR).
9 We also acknowledge the PIs for the CN and CCN surface measurements at the WMO-
10 GAW sites (J. Ogren, Earth System Research Laboratory, NOAA, US; J. Gras, CSIRO
11 Marine and Atmospheric Research, Australia; U. Baltensperger, Paul Scherrer Institute,
12 Switzerland; U. Kaminski, Deutscher Wetterdienst (DWD), Germany; S.G. Jennings,
13 National University of Ireland Galway, Ireland; R. Weller, Alfred Wegener Institute for
14 Polar and Marine Research, Germany; Y. Viisanen, Finnish Meteorological Institute,
15 Finland). We further acknowledge that Cape Grim data are provided by the Australian
16 Bureau of Meteorology (Cape Grim Baseline Air Pollution Station) and CSIRO Ocean
17 and Atmosphere Flagship. We thank Graham Mann (University of Leeds, UK) and
18 Dominick Spracklen (University of Leeds, UK) for providing aerosol number
19 measurements data collection. We also thank Greg Faluvegi (NASA GISS) and Maxwell
20 Kelley (NASA GISS) for providing technical support during the model development.
21 Finally, we thank two anonymous reviewers for constructive comments and suggestions
22 that have improved the paper considerably.

23

24

25 **References**

- 26
- 27 Abdou, W. A., Diner, D. J., Martonchik, J. V., Bruegge, C. J., Kahn, R. A., Gaitley, B. J.,
28 Crean, K. A., Remer, L. A., and Holben, B.: Comparison of coincident Multiangle
29 Imaging Spectroradiometer and Moderate Resolution Imaging Spectroradiometer aerosol
30 optical depths over land and ocean scenes containing Aerosol Robotic Network sites, *J.*
31 *Geophys. Res.-Atmos.*, 110, D10S07, doi: 10.1029/2004jd004693, 2005.
32 Adams, P. J., and Seinfeld, J. H.: Predicting global aerosol size distributions in general
33 circulation models, *J. Geophys. Res.-Atmos.*, 107, 4370, doi:10.1029/2001JD001010,
34 2002.
35 Andres, R. J., and Kasgnoc, A. D.: A time-averaged inventory of subaerial volcanic
36 sulfur emissions, *Journal of Geophysical Research: Atmospheres*, 103, 25251-25261,
37 10.1029/98JD02091, 1998.
38 Arimoto, R., Ray, B. J., Duce, R. A., Hewitt, A. D., Boldi, R., and Hudson, A.:
39 CONCENTRATIONS, SOURCES, AND FLUXES OF TRACE-ELEMENTS IN THE
40 REMOTE MARINE ATMOSPHERE OF NEW-ZEALAND, *J. Geophys. Res.-Atmos.*,
41 95, 22389-22405, 10.1029/JD095iD13p22389, 1990.
42 Ayers, G. P., Ivey, J. P., and Gillett, R. W.: COHERENCE BETWEEN SEASONAL
43 CYCLES OF DIMETHYL SULFIDE, METHANESULFONATE AND SULFATE IN
44 MARINE AIR, *Nature*, 349, 404-406, 10.1038/349404a0, 1991.

- 1 Baltensperger, U.: Aerosol climatology at the high, Älpine site Jungffaujoch,
 2 Switzerland, Journal of Geophysical Research, 102, 19707-19715, 1997.
- 3 Bauer, S. E., and Koch, D.: Impact of heterogeneous sulfate formation at mineral dust
 4 surfaces on aerosol loads and radiative forcing in the Goddard Institute for Space Studies
 5 general circulation model, J. Geophys. Res.-Atmos., 110, D17, D17202, doi:
 6 10.1029/2005jd005870, 2005.
- 7 Bauer, S. E., Koch, D., Unger, N., Metzger, S. M., Shindell, D. T., and Streets, D. G.:
 8 Nitrate aerosols today and in 2030: a global simulation including aerosols and
 9 tropospheric ozone, Atmospheric Chemistry and Physics, 7, 5043-5059, 2007.
- 10 Bauer, S. E., Wright, D. L., Koch, D., Lewis, E. R., McGraw, R., Chang, L. S., Schwartz,
 11 S. E., and Ruedy, R.: MATRIX (Multiconfiguration Aerosol TRacker of mIXing state):
 12 an aerosol microphysical module for global atmospheric models, Atmospheric Chemistry
 13 and Physics, 8, 6003-6035, 2008.
- 14 Bian, H. S., and Prather, M. J.: Fast-J2: Accurate simulation of stratospheric photolysis in
 15 global chemical models, Journal of Atmospheric Chemistry, 41, 281-296,
 16 10.1023/a:1014980619462, 2002.
- 17 Boucher, O., Moulin, C., Belviso, S., Aumont, O., Bopp, L., Cosme, E., von Kuhlmann,
 18 R., Lawrence, M. G., Pham, M., Reddy, M. S., Sciare, J., and Venkataraman, C.: DMS
 19 atmospheric concentrations and sulphate aerosol indirect radiative forcing: a sensitivity
 20 study to the DMS source representation and oxidation, Atmos. Chem. Phys., 3, 49-65,
 21 10.5194/acp-3-49-2003, 2003.
- 22 Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen,
 23 V.-M., Kondo, Y., Liao, H., Lohmann, U., Rasch, P., Satheesh, S. K., Sherwood, S.,
 24 Stevens, B., and Zhang, X. Y.: Clouds and Aerosols. In: Climate Change 2013: The
 25 Physical Science Basis. , Contribution of Working Group I to the Fifth Assessment
 26 Report of the Intergovernmental Panel on Climate Change, Cambridge University Press.
 27 Cambridge, United Kingdom and New York, NY, USA, 571-657, 2013.
- 28 Bowman, K. W., Shindell, D. T., Worden, H. M., Lamarque, J. F., Young, P. J.,
 29 Stevenson, D. S., Qu, Z., de la Torre, M., Bergmann, D., Cameron-Smith, P. J., Collins,
 30 W. J., Doherty, R., Dalsoren, S. B., Faluvegi, G., Folberth, G., Horowitz, L. W., Josse, B.
 31 M., Lee, Y. H., MacKenzie, I. A., Myhre, G., Nagashima, T., Naik, V., Plummer, D. A.,
 32 Rumbold, S. T., Skeie, R. B., Strode, S. A., Sudo, K., Szopa, S., Voulgarakis, A., Zeng,
 33 G., Kulawik, S. S., Aghedo, A. M., and Worden, J. R.: Evaluation of ACCMIP outgoing
 34 longwave radiation from tropospheric ozone using TES satellite observations,
 35 Atmospheric Chemistry and Physics, 13, 4057-4072, 10.5194/acp-13-4057-2013, 2013.
- 36 Cakmur, R. V., Miller, R. L., Perlitz, J., Geogdzhayev, I. V., Ginoux, P., Koch, D.,
 37 Kohfeld, K. E., Tegen, I., and Zender, C. S.: Constraining the magnitude of the global
 38 dust cycle by minimizing the difference between a model and observations, J. Geophys.
 39 Res.-Atmos., 111, D06207, doi:10.1029/2005jd005791, 2006.
- 40 Clarke, A. D., and Kapustin, V. N.: A pacific aerosol survey. Part I: A decade of data on
 41 particle production, transport, evolution, and mixing in the troposphere, Journal of the
 42 Atmospheric Sciences, 59, 363-382, 10.1175/1520-
 43 0469(2002)059<0363:apaspi>2.0.co;2, 2002.
- 44 Collaud Coen, M., Weingartner, E., Furger, M., Nyeki, S., Prévôt, A. S. H., Steinbacher,
 45 M., and Baltensperger, U.: Aerosol climatology and planetary boundary influence at the

- 1 Jungfraujoch analyzed by synoptic weather types, *Atmos. Chem. Phys.*, 11, 5931-5944,
2 10.5194/acp-11-5931-2011, 2011.
- 3 Debell, L. J., Gebhart, K. A., Malm, W. C., Pitchford, M. L., Schichtel, B. A., and White,
4 W. H.: Spatial and seasonal patterns and temporal variability of haze
5 and its constituents in the United States: report IV, 2006.
- 6 Del Genio, A. D., Yao, M. S., Kovari, W., and Lo, K. K. W.: A prognostic cloud water
7 parameterization for global climate models, *J Climate*, 9, 270-304, 1996.
- 8 Del Genio, A. D., and Yao, M-S: Efficient cumulus parameterization for long-term
9 climate studies: The GISS scheme, *The Representation of Cumulus Convection in*
10 *Numerical Models*, American Meteorological Society, Boston, Mass., 181- 184 pp.,
11 1993.
- 12 Dentener, F., Kinne, S., Bond, T., Boucher, O., Cofala, J., Generoso, S., Ginoux, P.,
13 Gong, S., Hoelzemann, J. J., Ito, A., Marelli, L., Penner, J. E., Putaud, J. P., Textor, C.,
14 Schulz, M., van der Werf, G. R., and Wilson, J.: Emissions of primary aerosol and
15 precursor gases in the years 2000 and 1750 prescribed data-sets for AeroCom,
16 *Atmospheric Chemistry and Physics*, 6, 4321-4344, 2006.
- 17 Dick, W. D., Saxena, P., and McMurry, P. H.: Estimation of water uptake by organic
18 compounds in submicron aerosols measured during the Southeastern Aerosol and
19 Visibility Study, *J. Geophys. Res.-Atmos.*, 105, 1471-1479, 10.1029/1999jd901001,
20 2000.
- 21 Diner, D. J., Beckert, J. C., Reilly, T. H., Bruegge, C. J., Conel, J. E., Kahn, R. A.,
22 Martonchik, J. V., Ackerman, T. P., Davies, R., Gerstl, S. A. W., Gordon, H. R., Muller,
23 J. P., Myneni, R. B., Sellers, P. J., Pinty, B., and Verstraete, M. M.: Multi-angle Imaging
24 SpectroRadiometer (MISR) - Instrument description and experiment overview, *Ieee*
25 *Transactions on Geoscience and Remote Sensing*, 36, 1072-1087, 10.1109/36.700992,
26 1998.
- 27 Duce, R. A.: Sources, distributions, and fluxes of mineral aerosols and their relationship
28 to climate, *Aerosol Forcing of Climate*, edited by: Charlson, R. J., and Heintzenberg, J.,
29 43-72 pp., 1995.
- 30 Easter, R. C., Ghan, S. J., Zhang, Y., Saylor, R. D., Chapman, E. G., Laulainen, N. S.,
31 Abdul-Razzak, H., Leung, L. R., Bian, X. D., and Zaveri, R. A.: MIRAGE: Model
32 description and evaluation of aerosols and trace gases, *J. Geophys. Res.-Atmos.*, 109,
33 D20210, doi:10.1029/2004JD004571, 2004.
- 34 Emmons, L. K., Hauglustaine, D. A., Muller, J. F., Carroll, M. A., Brasseur, G. P.,
35 Brunner, D., Staehelin, J., Thouret, V., and Marenco, A.: Data composites of airborne
36 observations of tropospheric ozone and its precursors, *J. Geophys. Res.-Atmos.*, 105,
37 20497-20538, 10.1029/2000jd900232, 2000.
- 38 Feingold, G., Tzivion, S., and Levin, Z.: EVOLUTION OF RAINDROP SPECTRA .1.
39 SOLUTION TO THE STOCHASTIC COLLECTION BREAKUP EQUATION USING
40 THE METHOD OF MOMENTS, *Journal of the Atmospheric Sciences*, 45, 3387-3399,
41 10.1175/1520-0469(1988)045<3387:eurspi>2.0.co;2, 1988.
- 42 Forster, P., and Ramaswamy, V.: Changes in Atmospheric Constituents and in Radiative
43 Forcing, *Climate Change 2007: The Physical Science Basis*, edited by: Solomon, S., Qin,
44 D., Manning, M., Marquis, M., Averyt, K., Tignor, M. M. B., Miller, H. L., and Chen, Z.
45 L., 129-234 pp., 2007.

- 1 Ginoux, P., Chin, M., Tegen, I., Prospero, J. M., Holben, B., Dubovik, O., and Lin, S. J.:
2 Sources and distributions of dust aerosols simulated with the GOCART model, *J.*
3 *Geophys. Res.-Atmos.*, 106, 20255-20273, 2001.
- 4 Gong, S. L.: A parameterization of sea-salt aerosol source function for sub- and super-
5 micron particles, *Global Biogeochemical Cycles*, 17, 1097, doi:10.1029/2003gb002079,
6 2003.
- 7 Hand, J. L., Copeland, S. A., Day, D. E., Dillner, A. M., Indresand, H., Malm, W. C.,
8 McDade, C. E., Moore, C. T., Pitchford, M. L., Schichtel, B. A., and Watson, J. G.:
9 Spatial and Seasonal Patterns and Temporal Variability of Haze and its Constituents in
10 the United States: Report V, Colo. State Univ., Fort Collins, 2011.
- 11 Hansen, J., Russell, G., Rind, D., Stone, P., Lacis, A., Lebedeff, S., Ruedy, R., and
12 Travis, L.: Efficient 3-Dimensional Global-Models for Climate Studies - Model-I and
13 Model-II, *Mon Weather Rev*, 111, 609-662, 1983.
- 14 Harrington, D. Y., and Kreidenweis, S. M.: Simulations of sulfate aerosol dynamics - Part
15 II. Model intercomparison, *Atmospheric Environment*, 32, 1701-1709, 10.1016/S1352-
16 2310(97)00453-6, 1998.
- 17 Heintzenberg, J., Covert, D. C., and Van Dingenen, R.: Size distribution and chemical
18 composition of marine aerosols: a compilation and review, *Tellus Series B-Chemical and*
19 *Physical Meteorology*, 52, 1104-1122, 10.1034/j.1600-0889.2000.00136.x, 2000.
- 20 Hoell, J. M., Davis, D. D., Jacob, D. J., Rodgers, M. O., Newell, R. E., Fuelberg, H. E.,
21 McNeal, R. J., Raper, J. L., and Bendura, R. J.: Pacific Exploratory Mission in the
22 tropical Pacific: PEM-Tropics A, August-September 1996, *J. Geophys. Res.-Atmos.*, 104,
23 5567-5583, 10.1029/1998jd100074, 1999.
- 24 Holben, B. N., Eck, T. F., Slutsker, I., Tanre, D., Buis, J. P., Setzer, A., Vermote, E.,
25 Reagan, J. A., Kaufman, Y. J., Nakajima, T., Lavenu, F., Jankowiak, I., and Smirnov, A.:
26 AERONET - A federated instrument network and data archive for aerosol
27 characterization, *Remote Sensing of Environment*, 66, 1-16, 10.1016/s0034-
28 4257(98)00031-5, 1998.
- 29 Holben, B. N., Tanre, D., Smirnov, A., Eck, T. F., Slutsker, I., Abu Hassan, N., Newcomb,
30 W. W., Schafer, J. S., Chatenet, B., Lavenu, F., Kaufman, Y. J., Castle, J. V., Setzer, A.,
31 Markham, B., Clark, D., Frouin, R., Halthore, R., Karneli, A., O'Neill, N. T., Pietras, C.,
32 Pinker, R. T., Voss, K., and Zibordi, G.: An emerging ground-based aerosol climatology:
33 Aerosol optical depth from AERONET, *J. Geophys. Res.-Atmos.*, 106, 12067-12097,
34 10.1029/2001jd900014, 2001.
- 35 Hsu, N. C., Tsay, S. C., King, M. D., and Herman, J. R.: Deep blue retrievals of Asian
36 aerosol properties during ACE-Asia, *Ieee Transactions on Geoscience and Remote*
37 *Sensing*, 44, 3180-3195, 10.1109/tgrs.2006.879540, 2006.
- 38 Jourdain, B., and Legrand, M.: Seasonal variations of atmospheric dimethylsulfide,
39 dimethylsulfoxide, sulfur dioxide, methanesulfonate, and non-sea-salt sulfate aerosols at
40 Dumont d'Urville (coastal Antarctica) (December 1998 to July 1999), *J. Geophys. Res.-*
41 *Atmos.*, 106, 14391-14408, 10.1029/2000jd900841, 2001.
- 42 Jung, J. G., Adams, P. J., and Pandis, S. N.: Simulating the size distribution and chemical
43 composition of ultrafine particles during nucleation events, *Atmospheric Environment*,
44 40, 2248-2259, 2006.

- 1 Jung, J. G., Fountoukis, C., Adams, P. J., and Pandis, S. N.: Simulation of in situ ultrafine
 2 particle formation in the eastern United States using PMCAMx-UF, *J. Geophys. Res.-*
 3 *Atmos.*, 115, D03203, doi: 10.1029/2009jd012313, 2010.
- 4 Kahn, R. A., Gaitley, B. J., Martonchik, J. V., Diner, D. J., Crean, K. A., and Holben, B.:
 5 Multiangle Imaging Spectroradiometer (MISR) global aerosol optical depth validation
 6 based on 2 years of coincident Aerosol Robotic Network (AERONET) observations, *J.*
 7 *Geophys. Res.-Atmos.*, 110, D10s04
 8 10.1029/2004jd004706, 2005.
- 9 Kettle, A. J., Andreae, M. O., Amouroux, D., Andreae, T. W., Bates, T. S., Berresheim,
 10 H., Bingemer, H., Boniforti, R., Curran, M. A. J., DiTullio, G. R., Helas, G., Jones, G. B.,
 11 Keller, M. D., Kiene, R. P., Leck, C., Levasseur, M., Malin, G., Maspero, M., Matrai, P.,
 12 McTaggart, A. R., Mihalopoulos, N., Nguyen, B. C., Novo, A., Putaud, J. P.,
 13 Rapsomanikis, S., Roberts, G., Schebeske, G., Sharma, S., Simo, R., Staubes, R., Turner,
 14 S., and Uher, G.: A global database of sea surface dimethylsulfide (DMS) measurements
 15 and a procedure to predict sea surface DMS as a function of latitude, longitude, and
 16 month, *Global Biogeochemical Cycles*, 13, 399-444, 10.1029/1999gb900004, 1999.
- 17 Kettle, A. J., and Andreae, M. O.: Flux of dimethylsulfide from the oceans: A
 18 comparison of updated data sets and flux models, *Journal of Geophysical Research:*
 19 *Atmospheres*, 105, 26793-26808, 10.1029/2000JD900252, 2000.
- 20 Koch, D., Jacob, D., Tegen, I., Rind, D., and Chin, M.: Tropospheric sulfur simulation
 21 and sulfate direct radiative forcing in the Goddard Institute for Space Studies general
 22 circulation model, *J. Geophys. Res.-Atmos.*, 104, 23799-23822, 10.1029/1999jd900248,
 23 1999.
- 24 Koch, D., Schmidt, G. A., and Field, C. V.: Sulfur, sea salt, and radionuclide aerosols in
 25 GISS ModelE, *J. Geophys. Res.-Atmos.*, 111, D06206, doi:10.1029/2004jd005550, 2006.
- 26 Koch, D., Bond, T. C., Streets, D., Unger, N., and van der Werf, G. R.: Global impacts of
 27 aerosols from particular source regions and sectors, *J. Geophys. Res.-Atmos.*, 112,
 28 D02205, doi: 10.1029/2005jd007024, 2007.
- 29 Koch, D., Bauer, S. E., Del Genio, A., Faluvegi, G., McConnell, J. R., Menon, S., Miller,
 30 R. L., Rind, D., Ruedy, R., Schmidt, G. A., and Shindell, D.: Coupled Aerosol-
 31 Chemistry-Climate Twentieth-Century Transient Model Investigation: Trends in Short-
 32 Lived Species and Climate Responses, *J Climate*, 24, 2693-2714,
 33 10.1175/2011jcli3582.1, 2011.
- 34 Korhonen, H., Carslaw, K. S., Spracklen, D. V., Mann, G. W., and Woodhouse, M. T.:
 35 Influence of oceanic dimethyl sulfide emissions on cloud condensation nuclei
 36 concentrations and seasonality over the remote Southern Hemisphere oceans: A global
 37 model study, *J. Geophys. Res.-Atmos.*, 113, D15204, doi: 10.1029/2007jd009718, 2008.
- 38 Lamarque, J.-F., Shindell, D. T., Josse, B., Eyring, V., Young, P. J., Cionni, I.,
 39 Bergmann, D., Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi,
 40 G., Folberth, G., Ghan, S. J., Horowitz, L. W., Lee, Y. H., McKenzie, I., Nagashima, T.,
 41 Naik, V., Plummer, D., Rumbold, S., Skeie, R., Stevenson, D. S., Strode, S., Sudo, K.,
 42 Szopa, S., Voulgarakis, A., and Zeng, G.: The Atmospheric Chemistry and Climate
 43 Model Intercomparison Project (ACCMIP): Overview and description of models,
 44 simulations and climate diagnostics, *Geoscientific Model Development*, accepted, 2012.
- 45 Lamarque, J. F., Dentener, F., McConnell, J., Ro, C. U., Shaw, M., Vet, R., Bergmann,
 46 D., Cameron-Smith, P., Dalsoren, S., Doherty, R., Faluvegi, G., Ghan, S. J., Josse, B.,

- 1 Lee, Y. H., MacKenzie, I. A., Plummer, D., Shindell, D. T., Skeie, R. B., Stevenson, D.
 2 Strode, S., Zeng, G., Curran, M., Dahl-Jensen, D., Das, S., Fritzsche, D., and Nolan,
 3 M.: Multi-model mean nitrogen and sulfur deposition from the Atmospheric Chemistry
 4 and Climate Model Intercomparison Project (ACCMIP): evaluation of historical and
 5 projected future changes, *Atmospheric Chemistry and Physics*, 13, 7997-8018, 2013a.
 6 Lamarque, J. F., Shindell, D. T., Josse, B., Young, P. J., Cionni, I., Eyring, V.,
 7 Bergmann, D., Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi,
 8 G., Folberth, G., Ghan, S. J., Horowitz, L. W., Lee, Y. H., MacKenzie, I. A., Nagashima,
 9 T., Naik, V., Plummer, D., Righi, M., Rumbold, S. T., Schulz, M., Skeie, R. B.,
 10 Stevenson, D. S., Strode, S., Sudo, K., Szopa, S., Voulgarakis, A., and Zeng, G.: The
 11 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP):
 12 overview and description of models, simulations and climate diagnostics, *Geoscientific
 13 Model Development*, 6, 179-206, 10.5194/gmd-6-179-2013, 2013b.
 14 Lauer, A., Hendricks, J., Ackermann, I., Schell, B., Hass, H., and Metzger, S.: Simulating
 15 aerosol microphysics with the ECHAM/MADE GCM - Part I: Model description and
 16 comparison with observations, *Atmospheric Chemistry and Physics*, 5, 3251-3276, 2005.
 17 Lee, Y. H., Chen, K., and Adams, P. J.: Development of a global model of mineral dust
 18 aerosol microphysics, *Atmospheric Chemistry and Physics*, 9, 2441-2458, 2009.
 19 Lee, Y. H., and Adams, P. J.: Evaluation of aerosol distributions in the GISS-TOMAS
 20 global aerosol microphysics model with remote sensing observations, *Atmospheric
 21 Chemistry and Physics*, 10, 2129-2144, 2010.
 22 Lee, Y. H., and Adams, P. J.: A Fast and Efficient Version of the Two-Moment Aerosol
 23 Sectional (TOMAS) Global Aerosol Microphysics Model, *Aerosol Science and
 24 Technology*, 46, 678-689, 10.1080/02786826.2011.643259, 2012.
 25 Lee, Y. H., Lamarque, J. F., Flanner, M. G., Jiao, C., Shindell, D. T., Berntsen, T.,
 26 Bisiaux, M. M., Cao, J., Collins, W. J., Curran, M., Edwards, R., Faluvegi, G., Ghan, S.,
 27 Horowitz, L. W., McConnell, J. R., Ming, J., Myhre, G., Nagashima, T., Naik, V.,
 28 Rumbold, S. T., Skeie, R. B., Sudo, K., Takemura, T., Thevenon, F., Xu, B., and Yoon, J.
 29 H.: Evaluation of preindustrial to present-day black carbon and its albedo forcing from
 30 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP),
 31 *Atmospheric Chemistry and Physics*, 13, 2607-2634, 10.5194/acp-13-2607-2013, 2013a.
 32 Lee, Y. H., Pierce, J. R., and Adams, P. J.: Representation of nucleation mode
 33 microphysics in a global aerosol model with sectional microphysics, *Geoscientific Model
 34 Development*, 6, 1221-1232, 2013b.
 35 Liss, P. S., and Merlivat, L.: Air-sea gas exchange rates: Introduction and synthesis, *The
 36 Role of Air-Sea Exchange in Geochemical Cycling*, Springer, New York, 1986.
 37 Liu, X. H., Penner, J. E., and Herzog, M.: Global modeling of aerosol dynamics: Model
 38 description, evaluation, and interactions between sulfate and nonsulfate aerosols, *J.
 39 Geophys. Res.-Atmos.*, 110, D18206, doi: 10.1029/2004jd005674, 2005.
 40 Lohmann, U., and Feichter, J.: Global indirect aerosol effects: a review, *Atmospheric
 41 Chemistry and Physics*, 5, 715-737, 2005.
 42 Mahowald, N. M., Engelstaedter, S., Luo, C., Sealy, A., Artaxo, P., Benitez-Nelson, C.,
 43 Bonnet, S., Chen, Y., Chuang, P. Y., Cohen, D. D., Dulac, F., Herut, B., Johansen, A. M.,
 44 Kubilay, N., Losno, R., Maenhaut, W., Paytan, A., Prospero, J. A., Shank, L. M., and
 45 Siefert, R. L.: Atmospheric Iron Deposition: Global Distribution, Variability, and Human

- 1 Perturbations, in: Annual Review of Marine Science, Annual Review of Marine Science,
2 245-278, 2009.
- 3 Mann, G. W., Carslaw, K. S., Spracklen, D. V., Ridley, D. A., Manktelow, P. T.,
4 Chipperfield, M. P., Pickering, S. J., and Johnson, C. E.: Description and evaluation of
5 GLOMAP-mode: a modal global aerosol microphysics model for the UKCA
6 composition-climate model, Geoscientific Model Development, 3, 519-551,
7 10.5194/gmd-3-519-2010, 2010.
- 8 Menon, S., Del Genio, A. D., Koch, D., and Tselioudis, G.: GCM Simulations of the
9 aerosol indirect effect: Sensitivity to cloud parameterization and aerosol burden, Journal
10 of the Atmospheric Sciences, 59, 692-713, 10.1175/1520-
11 0469(2002)059<0692:gsotai>2.0.co;2, 2002.
- 12 Menon, S., Del Genio, A. D., Kaufman, Y., Bennartz, R., Koch, D., Loeb, N., and
13 Orlikowski, D.: Analyzing signatures of aerosol-cloud interactions from satellite
14 retrievals and the GISS GCM to constrain the aerosol indirect effect, J. Geophys. Res.-
15 Atmos., 113, D14s22
16 10.1029/2007jd009442, 2008.
- 17 Merikanto, J., Spracklen, D. V., Mann, G. W., Pickering, S. J., and Carslaw, K. S.:
18 Impact of nucleation on global CCN, Atmospheric Chemistry and Physics, 9, 8601-8616,
19 2009.
- 20 Miller, R. L., Cakmur, R. V., Perlitz, J., Geogdzhayev, I. V., Ginoux, P., Koch, D.,
21 Kohfeld, K. E., Prigent, C., Ruedy, R., Schmidt, G. A., and Tegen, I.: Mineral dust
22 aerosols in the NASA goddard institute for Space Sciences ModelE atmospheric general
23 circulation model, J. Geophys. Res.-Atmos., 111, D06208, doi: 10.1029/2005jd005796,
24 2006.
- 25 Mishchenko, M. I., Travis, L. D., and Mackowski, D. W.: T-matrix computations of light
26 scattering by nonspherical particles: A review, Journal of Quantitative Spectroscopy &
27 Radiative Transfer, 55, 535-575, 10.1016/0022-4073(96)00002-7, 1996.
- 28 Monahan, E. C., Spiel, D. E., and David, K. L.: A model of marine aerosol generation via
29 whitecaps and wave disruption, in Oceanic Whitecaps, edited by: Monahan, E. C., and
30 Mac Niocaill, G., Oceanographic Sciences Library, 167-174 pp., 1986.
- 31 Myhre, G., Samset, B. H., Schulz, M., Balkanski, Y., Bauer, S., Berntsen, T. K., Bian, H.,
32 Bellouin, N., Chin, M., Diehl, T., Easter, R. C., Feichter, J., Ghan, S. J., Hauglustaine, D.,
33 Iversen, T., Kinne, S., Kirkev^g, A., Lamarque, J. F., Lin, G., Liu, X., Lund, M. T., Luo,
34 G., Ma, X., van Noije, T., Penner, J. E., Rasch, P. J., Ruiz, A., Seland, Ø., Skeie, R. B.,
35 Stier, P., Takemura, T., Tsigaridis, K., Wang, P., Wang, Z., Xu, L., Yu, H., Yu, F., Yoon,
36 J. H., Zhang, K., Zhang, H., and Zhou, C.: Radiative forcing of the direct aerosol effect
37 from AeroCom Phase II simulations, Atmos. Chem. Phys., 13, 1853-1877, 10.5194/acp-
38 13-1853-2013a, 2013a.
- 39 Myhre, G., Shindell, D., Brèon, F.-M., Collins, W., Fuglestvedt, J., Huang, J., Koch, D.,
40 Lamarque, J. F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G.,
41 Takemura, T., and Zhang, H.: Anthropogenic and Natural Radiative Forcing. In: Climate
42 Change 2013: The Physical Science Basis. , Contribution of Working Group I to the Fifth
43 Assessment
- 44 Report of the Intergovernmental Panel on Climate Change, Cambridge University Press.
45 Cambridge, United Kingdom and New York, NY, USA, 2013b.

- 1 Nabat, P., Somot, S., Mallet, M., Chiapello, I., Morcrette, J. J., Solmon, F., Szopa, S.,
2 Dulac, F., Collins, W., Ghan, S., Horowitz, L. W., Lamarque, J. F., Lee, Y. H., Naik, V.,
3 Nagashima, T., Shindell, D., and Skeie, R.: A 4-D climatology (1979-2009) of the
4 monthly tropospheric aerosol optical depth distribution over the Mediterranean region
5 from a comparative evaluation and blending of remote sensing and model products,
6 Atmospheric Measurement Techniques, 6, 1287-1314, 10.5194/amt-6-1287-2013, 2013.
7 Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L. W., Lamarque, J. F., Lin, M.,
8 Prather, M. J., Young, P. J., Bergmann, D., Cameron-Smith, P. J., Cionni, I., Collins, W.
9 J., Dalsoren, S. B., Doherty, R., Eyring, V., Faluvegi, G., Folberth, G. A., Josse, B., Lee,
10 Y. H., MacKenzie, I. A., Nagashima, T., van Noije, T. P. C., Plummer, D. A., Righi, M.,
11 Rumbold, S. T., Skeie, R., Shindell, D. T., Stevenson, D. S., Strode, S., Sudo, K., Szopa,
12 S., and Zeng, G.: Preindustrial to present-day changes in tropospheric hydroxyl radical
13 and methane lifetime from the Atmospheric Chemistry and Climate Model
14 Intercomparison Project (ACCMIP), Atmospheric Chemistry and Physics, 13, 5277-
15 5298, 10.5194/acp-13-5277-2013, 2013.
16 Napari, I., Noppel, M., Vehkamaki, H., and Kulmala, M.: Parametrization of ternary
17 nucleation rates for H₂SO₄-NH₃-H₂O vapors, J. Geophys. Res.-Atmos., 107, 4381, doi:
18 10.1029/2002JD002132, 2002.
19 Nenes, A., Pandis, S. N., and Pilinis, C.: ISORROPIA: A New Thermodynamic
20 Equilibrium Model for Multiphase Multicomponent Inorganic Aerosols, Aquatic
21 Geochemistry, 4, 123-152, 10.1023/A:1009604003981, 1998.
22 Nguyen, B. C., Mihalopoulos, N., Putaud, J. P., Gaudry, A., Gallet, L., Keene, W. C., and
23 Galloway, J. N.: COVARIATIONS IN OCEANIC DIMETHYL SULFIDE, ITS
24 OXIDATION-PRODUCTS AND RAIN ACIDITY AT AMSTERDAM ISLAND IN
25 THE SOUTHERN INDIAN-OCEAN, Journal of Atmospheric Chemistry, 15, 39-53,
26 10.1007/bf00053608, 1992.
27 Nightingale, P. D., Malin, G., Law, C. S., Watson, A. J., Liss, P. S., Liddicoat, M. I.,
28 Boutin, J., and Upstill-Goddard, R. C.: In situ evaluation of air-sea gas exchange
29 parameterizations using novel conservative and volatile tracers, Global Biogeochemical
30 Cycles, 14, 373-387, 10.1029/1999gb900091, 2000.
31 Petzold, A., Fiebig, M., Flentje, H., Keil, A., Leiterer, U., Schroder, F., Stifter, A.,
32 Wendisch, M., and Wendling, P.: Vertical variability of aerosol properties observed at a
33 continental site during the Lindenberg Aerosol Characterization Experiment (LACE 98),
34 J. Geophys. Res.-Atmos., 107, 8128, 8128, doi: 10.1029/2001jd001043, 2002.
35 Pierce, J. R., and Adams, P. J.: Global evaluation of CCN formation by direct emission of
36 sea salt and growth of ultrafine sea salt, J. Geophys. Res.-Atmos., 111, D06203,
37 doi:10.1029/2005JD006186, 2006.
38 Pierce, J. R., Chen, K., and Adams, P. J.: Contribution of primary carbonaceous aerosol
39 to cloud condensation nuclei: processes and uncertainties evaluated with a global aerosol
40 microphysics model, Atmospheric Chemistry and Physics, 7, 5447-5466, 2007.
41 Pierce, J. R., and Adams, P. J.: A Computationally Efficient Aerosol
42 Nucleation/Condensation Method: Pseudo-Steady-State Sulfuric Acid, Aerosol Science
43 and Technology, 43, 216-226, 2009a.
44 Pierce, J. R., and Adams, P. J.: Uncertainty in global CCN concentrations from uncertain
45 aerosol nucleation and primary emission rates, Atmospheric Chemistry and Physics, 9,
46 1339-1356, 2009b.

- 1 Prather, M. J.: Numerical Advection by Conservation of 2nd-Order Moments, *J.
 2 Geophys. Res.-Atmos.*, 91, 6671-6681, 1986.
 3 Prospero, J. M., and Bonatti, E.: CONTINENTAL DUST IN ATMOSPHERE OF
 4 EASTERN EQUATORIAL PACIFIC, *Journal of Geophysical Research*, 74, 3362-3371,
 5 doi: 10.1029/JC074i013p03362, 1969.
 6 Prospero, J. M., Bullard, J. E., and Hodgkins, R.: High-Latitude Dust Over the North
 7 Atlantic: Inputs from Icelandic Proglacial Dust Storms, *Science*, 335, 1078-1082,
 8 10.1126/science.1217447, 2012.
 9 Putaud, J. P.: A European aerosol phenomenology; physical and chemical characteristics
 10 of particulate matter at kerbside, urban, rural and background sites in Europe, *European
 11 Commission, EUR 20411 EN*, 2003.
 12 Putaud, J. P., Van Dingenen, R., Alastuey, A., Bauer, H., Birmili, W., Cyrys, J., Flentje,
 13 H., Fuzzi, S., Gehrig, R., Hansson, H. C., Harrison, R. M., Herrmann, H., Hitzenberger,
 14 R., Hueglin, C., Jones, A. M., Kasper-Giebl, A., Kiss, G., Kousa, A., Kuhlbusch, T. A. J.,
 15 Loeschau, G., Maenhaut, W., Molnar, A., Moreno, T., Pekkanen, J., Perrino, C., Pitz, M.,
 16 Puxbaum, H., Querol, X., Rodriguez, S., Salma, I., Schwarz, J., Smolik, J., Schneider, J.,
 17 Spindler, G., ten Brink, H., Tursic, J., Viana, M., Wiedensohler, A., and Raes, F.: A
 18 European aerosol phenomenology-3: Physical and chemical characteristics of particulate
 19 matter from 60 rural, urban, and kerbside sites across Europe, *Atmospheric Environment*,
 20 44, 1308-1320, 10.1016/j.atmosenv.2009.12.011, 2010.
 21 Raper, J. L., Kleb, M. M., Jacob, D. J., Davis, D. D., Newell, R. E., Fuelberg, H. E.,
 22 Bendura, R. J., Hoell, J. M., and McNeal, R. J.: Pacific Exploratory Mission in the
 23 tropical Pacific: PEM-Tropics B, March-April 1999, *J. Geophys. Res.-Atmos.*, 106,
 24 32401-32425, 10.1029/2000jd900833, 2001.
 25 Remer, L. A., Kleidman, R. G., Levy, R. C., Kaufman, Y. J., Tanre, D., Mattoo, S.,
 26 Martins, J. V., Ichoku, C., Koren, I., Yu, H., and Holben, B. N.: Global aerosol
 27 climatology from the MODIS satellite sensors, *J. Geophys. Res.-Atmos.*, 113, D14s07
 28 10.1029/2007jd009661, 2008.
 29 Rienecker, M. M., Suarez, M. J., Gelaro, R., Todling, R., Bacmeister, J., Liu, E.,
 30 Bosilovich, M. G., Schubert, S. D., Takacs, L., Kim, G.-K., Bloom, S., Chen, J., Collins,
 31 D., Conaty, A., da Silva, A., Gu, W., Joiner, J., Koster, R. D., Lucchesi, R., Molod, A.,
 32 Owens, T., Pawson, S., Pegion, P., Redder, C. R., Reichle, R., Robertson, F. R., Ruddick,
 33 A. G., Sienkiewicz, M., and Woollen, J.: MERRA: NASA's Modern-Era Retrospective
 34 Analysis for Research and Applications, *J Climate*, 24, 3624-3648, 10.1175/JCLI-D-11-
 35 00015.1, 2011.
 36 Savoie, D. L., and Prospero, J. M.: COMPARISON OF OCEANIC AND
 37 CONTINENTAL SOURCES OF NON-SEA-SALT SULFATE OVER THE PACIFIC-
 38 OCEAN, *Nature*, 339, 685-687, 10.1038/339685a0, 1989.
 39 Schmidt, G. A., Ruedy, R., Hansen, J. E., Aleinov, I., Bell, N., Bauer, M., Bauer, S.,
 40 Cairns, B., Canuto, V., Cheng, Y., Del Genio, A., Faluvegi, G., Friend, A. D., Hall, T.
 41 M., Hu, Y. Y., Kelley, M., Kiang, N. Y., Koch, D., Lacis, A. A., Lerner, J., Lo, K. K.,
 42 Miller, R. L., Nazarenko, L., Oinas, V., Perlitz, J., Perlitz, J., Rind, D., Romanou, A.,
 43 Russell, G. L., Sato, M., Shindell, D. T., Stone, P. H., Sun, S., Tausnev, N., Thresher, D.,
 44 and Yao, M. S.: Present-day atmospheric simulations using GISS ModelE: Comparison
 45 to in situ, satellite, and reanalysis data, *J Climate*, 19, 153-192, 10.1175/jcli3612.1, 2006.

- 1 Schmidt, G. A., Kelley, M., Nazarenko, L., Ruedy, R., Russell, G. L., Aleinov, I., Bauer,
2 M., Bauer, S. E., Bhat, M. K., Bleck, R., Canuto, V., Chen, Y.-H., Cheng, Y., Clune, T.
3 L., Del Genio, A., de Fainchtein, R., Faluvegi, G., Hansen, J. E., Healy, R. J., Kiang, N.
4 Y., Koch, D., Lacis, A. A., LeGrande, A. N., Lerner, J., Lo, K. K., Matthews, E. E.,
5 Menon, S., Miller, R. L., Oinas, V., Oloso, A. O., Perlitz, J. P., Puma, M. J., Putman,
6 W. M., Rind, D., Romanou, A., Sato, M., Shindell, D. T., Sun, S., Syed, R. A., Tausnev,
7 N., Tsigaridis, K., Unger, N., Voulgarakis, A., Yao, M.-S., and Zhang, J.: Configuration
8 and assessment of the GISS ModelE2 contributions to the CMIP5 archive, *Journal of*
9 *Advances in Modeling Earth Systems*, 6, 141-184, 10.1002/2013ms000265, 2014.
- 10 Seinfeld, J. H., and Pandis, S. N.: *Atmospheric Chemistry and Physics*, John Wiley and
11 Sons, New York, 1998.
- 12 Shindell, D., Faluvegi, G., Walsh, M., Anenberg, S. C., Van Dingenen, R., Muller, N. Z.,
13 Austin, J., Koch, D., and Milly, G.: Climate, health, agricultural and economic impacts of
14 tighter vehicle-emission standards, *Nature Climate Change*, 1, 59-66,
15 10.1038/nclimate1066, 2011.
- 16 Shindell, D. T., Faluvegi, G., Unger, N., Aguilar, E., Schmidt, G. A., Koch, D. M., Bauer,
17 S. E., and Miller, R. L.: Simulations of preindustrial, present-day, and 2100 conditions in
18 the NASA GISS composition and climate model G-PUCCINI, *Atmospheric Chemistry*
19 and *Physics*, 6, 4427-4459, 2006.
- 20 Shindell, D. T., Lamarque, J. F., Schulz, M., Flanner, M., Jiao, C., Chin, M., Young, P. J.,
21 Lee, Y. H., Rotstain, L., Mahowald, N., Milly, G., Faluvegi, G., Balkanski, Y., Collins,
22 W. J., Conley, A. J., Dalsoren, S., Easter, R., Ghan, S., Horowitz, L., Liu, X., Myhre, G.,
23 Nagashima, T., Naik, V., Rumbold, S. T., Skeie, R., Sudo, K., Szopa, S., Takemura, T.,
24 Voulgarakis, A., Yoon, J. H., and Lo, F.: Radiative forcing in the ACCMIP historical and
25 future climate simulations, *Atmospheric Chemistry and Physics*, 13, 2939-2974,
26 10.5194/acp-13-2939-2013, 2013.
- 27 Sihto, S. L., Kulmala, M., Kerminen, V. M., Dal Maso, M., Petaja, T., Riipinen, I.,
28 Korhonen, H., Arnold, F., Janson, R., Boy, M., Laaksonen, A., and Lehtinen, K. E. J.:
29 Atmospheric sulphuric acid and aerosol formation: implications from atmospheric
30 measurements for nucleation and early growth mechanisms, *Atmospheric Chemistry and*
31 *Physics*, 6, 4079-4091, 2006.
- 32 Singh, S., Adams, P. J., Misquitta, A., Lee, K. J., Lipsky, E. M., and Robinson, A. L.:
33 Computational Analysis of Particle Nucleation in Dilution Tunnels: Effects of Flow
34 Configuration and Tunnel Geometry, *Aerosol Science and Technology*, 48, 638-648,
35 10.1080/02786826.2014.910291, 2014.
- 36 Spracklen, D. V., Pringle, K. J., Carslaw, K. S., Chipperfield, M. P., and Mann, G. W.: A
37 global off-line model of size-resolved aerosol microphysics: I. Model development and
38 prediction of aerosol properties, *Atmospheric Chemistry and Physics*, 5, 2227-2252,
39 2005.
- 40 Spracklen, D. V., Carslaw, K. S., Merikanto, J., Mann, G. W., Reddington, C. L.,
41 Pickering, S., Ogren, J. A., Andrews, E., Baltensperger, U., Weingartner, E., Boy, M.,
42 Kulmala, M., Laakso, L., Lihavainen, H., Kivekas, N., Komppula, M., Mihalopoulos, N.,
43 Kouvarakis, G., Jennings, S. G., O'Dowd, C., Birmili, W., Wiedensohler, A., Weller, R.,
44 Gras, J., Laj, P., Sellegri, K., Bonn, B., Krejci, R., Laaksonen, A., Hamed, A., Minikin,
45 A., Harrison, R. M., Talbot, R., and Sun, J.: Explaining global surface aerosol number

- 1 concentrations in terms of primary emissions and particle formation, Atmospheric
2 Chemistry and Physics, 10, 4775-4793, 10.5194/acp-10-4775-2010, 2010.
- 3 Spracklen, D. V., Carslaw, K. S., Poeschl, U., Rap, A., and Forster, P. M.: Global cloud
4 condensation nuclei influenced by carbonaceous combustion aerosol, Atmospheric
5 Chemistry and Physics, 11, 9067-9087, 10.5194/acp-11-9067-2011, 2011.
- 6 Stevens, R. G., Pierce, J. R., Brock, C. A., Reed, M. K., Crawford, J. H., Holloway, J. S.,
7 Ryerson, T. B., Huey, L. G., and Nowak, J. B.: Nucleation and growth of sulfate aerosol
8 in coal-fired power plant plumes: sensitivity to background aerosol and meteorology,
9 Atmos. Chem. Phys., 12, 189-206, 10.5194/acp-12-189-2012, 2012.
- 10 Stevenson, D. S., Young, P. J., Naik, V., Lamarque, J. F., Shindell, D. T., Voulgarakis,
11 A., Skeie, R. B., Dalsoren, S. B., Myhre, G., Berntsen, T. K., Folberth, G. A., Rumbold,
12 S. T., Collins, W. J., MacKenzie, I. A., Doherty, R. M., Zeng, G., van Noije, T. P. C.,
13 Strunk, A., Bergmann, D., Cameron-Smith, P., Plummer, D. A., Strode, S. A., Horowitz,
14 L., Lee, Y. H., Szopa, S., Sudo, K., Nagashima, T., Josse, B., Cionni, I., Righi, M.,
15 Eyring, V., Conley, A., Bowman, K. W., Wild, O., and Archibald, A.: Tropospheric
16 ozone changes, radiative forcing and attribution to emissions in the Atmospheric
17 Chemistry and Climate Model Intercomparison Project (ACCMIP), Atmospheric
18 Chemistry and Physics, 13, 3063-3085, 10.5194/acp-13-3063-2013, 2013.
- 19 Stier, P., Feichter, J., Kinne, S., Kloster, S., Vignati, E., Wilson, J., Ganzeveld, L., Tegen,
20 I., Werner, M., Balkanski, Y., Schulz, M., Boucher, O., Minikin, A., and Petzold, A.: The
21 aerosol-climate model ECHAM5-HAM, Atmospheric Chemistry and Physics, 5, 1125-
22 1156, 2005.
- 23 Tegen, I., Harrison, S. P., Kohfeld, K., Prentice, I. C., Coe, M., and Heimann, M.: Impact
24 of vegetation and preferential source areas on global dust aerosol: Results from a model
25 study, J. Geophys. Res.-Atmos., 107, 4576, 4576, doi: 10.1029/2001jd000963, 2002.
- 26 Textor, C., Schulz, M., Guibert, S., Kinne, S., Balkanski, Y., Bauer, S., Berntsen, T.,
27 Berglen, T., Boucher, O., Chin, M., Dentener, F., Diehl, T., Easter, R., Feichter, H.,
28 Fillmore, D., Ghan, S., Ginoux, P., Gong, S., Kristjansson, J. E., Krol, M., Lauer, A.,
29 Lamarque, J. F., Liu, X., Montanaro, V., Myhre, G., Penner, J., Pitari, G., Reddy, S.,
30 Seland, O., Stier, P., Takemura, T., and Tie, X.: Analysis and quantification of the
31 diversities of aerosol life cycles within AeroCom, Atmospheric Chemistry and Physics, 6,
32 1777-1813, 2006.
- 33 Thornton, D. C., Bandy, A. R., Blomquist, B. W., Driedger, A. R., and Wade, T. P.:
34 Sulfur dioxide distribution over the Pacific Ocean 1991–1996, Journal of Geophysical
35 Research: Atmospheres, 104, 5845-5854, 10.1029/1998JD100048, 1999.
- 36 Trivitayanurak, W., Adams, P. J., Spracklen, D. V., and Carslaw, K. S.: Tropospheric
37 aerosol microphysics simulation with assimilated meteorology: model description and
38 intermodel comparison, Atmospheric Chemistry and Physics, 8, 3149-3168, 2008.
- 39 Tsigaridis, K., and Kanakidou, M.: Secondary organic aerosol importance in the future
40 atmosphere, Atmospheric Environment, 41, 4682-4692, 10.1016/j.atmosenv.2007.03.045,
41 2007.
- 42 Tsigaridis, K., Koch, D., and Menon, S.: Uncertainties and importance of sea spray
43 composition on aerosol direct and indirect effects, J. Geophys. Res.-Atmos., 118, 220-
44 235, 10.1029/2012jd018165, 2013.

- 1 Tzivion, S., Feingold, G., and Levin, Z.: An Efficient Numerical Solution to the
2 Stochastic Collection Equation, Journal of the Atmospheric Sciences, 44, 3139-3149,
3 1987.
- 4 Tzivion, S., Feingold, G., and Levin, Z.: THE EVOLUTION OF RAINDROP SPECTRA
5 .2. COLLISIONAL COLLECTION BREAKUP AND EVAPORATION IN A
6 RAINSHAFT, Journal of the Atmospheric Sciences, 46, 3312-3327, 10.1175/1520-
7 0469(1989)046<3312:teorsp>2.0.co;2, 1989.
- 8 Tzivion, S., Reisin, T. G., and Levin, Z.: A new formulation of the spectral multi-moment
9 method for calculating the kinetic collection equation: More accuracy with fewer bins,
10 Journal of Computational Physics, 171, 418-422, 10.1006/jcph.2001.6776, 2001.
- 11 Uematsu, M., Duce, R. A., and Prospero, J. M.: DEPOSITION OF ATMOSPHERIC
12 MINERAL PARTICLES IN THE NORTH PACIFIC-OCEAN, Journal of Atmospheric
13 Chemistry, 3, 123-138, 10.1007/bf00049372, 1985.
- 14 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P.
15 S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire emissions
16 and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997-
17 2009), Atmospheric Chemistry and Physics, 10, 11707-11735, 10.5194/acp-10-11707-
18 2010, 2010.
- 19 Vehkamaki, H., Kulmala, M., Napari, I., Lehtinen, K. E. J., Timmreck, C., Noppel, M.,
20 and Laaksonen, A.: An improved parameterization for sulfuric acid-water nucleation
21 rates for tropospheric and stratospheric conditions, J. Geophys. Res.-Atmos., 107, 4622,
22 DOI: 10.1029/2002JD002184, 2002.
- 23 Vignati, E., Wilson, J., and Stier, P.: M7: An efficient size-resolved aerosol microphysics
24 module for large-scale aerosol transport models, J. Geophys. Res.-Atmos., 109, D22202,
25 doi:10.1029/2003jd004485, 2004.
- 26 Wang, M., Ghan, S., Ovchinnikov, M., Liu, X., Easter, R., Kassianov, E., Qian, Y., and
27 Morrison, H.: Aerosol indirect effects in a multi-scale aerosol-climate model PNNL-
28 MMF, Atmospheric Chemistry and Physics, 11, 5431-5455, 10.5194/acp-11-5431-2011,
29 2011.
- 30 Wanninkhof, R.: Relationship between wind speed and gas exchange over the ocean,
31 Journal of Geophysical Research: Oceans, 97, 7373-7382, 10.1029/92jc00188, 1992.
- 32 Westervelt, D. M., Pierce, J. R., Riipinen, I., Trivitayanurak, W., Hamed, A., Kulmala,
33 M., Laaksonen, A., Decesari, S., and Adams, P. J.: Formation and growth of nucleated
34 particles into cloud condensation nuclei: model-measurement comparison, Atmos. Chem.
35 Phys., 13, 7645-7663, 10.5194/acp-13-7645-2013, 2013.
- 36 Young, P. J., Archibald, A. T., Bowman, K. W., Lamarque, J. F., Naik, V., Stevenson, D.
37 S., Tilmes, S., Voulgarakis, A., Wild, O., Bergmann, D., Cameron-Smith, P., Cionni, I.,
38 Collins, W. J., Dalsoren, S. B., Doherty, R. M., Eyring, V., Faluvegi, G., Horowitz, L.
39 W., Josse, B., Lee, Y. H., MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M.,
40 Rumbold, S. T., Skeie, R. B., Shindell, D. T., Strode, S. A., Sudo, K., Szopa, S., and
41 Zeng, G.: Pre-industrial to end 21st century projections of tropospheric ozone from the
42 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP),
43 Atmospheric Chemistry and Physics, 13, 2063-2090, 10.5194/acp-13-2063-2013, 2013.
- 44 Yu, F., and Luo, G.: Simulation of particle size distribution with a global aerosol model:
45 contribution of nucleation to aerosol and CCN number concentrations, Atmospheric
46 Chemistry and Physics, 9, 7691-7710, 2009.

1 Zhang, Y., Seigneur, C., Seinfeld, J. H., Jacobson, M. Z., and Binkowski, F. S.:
2 Simulation of aerosol dynamics: A comparative review of algorithms used in air quality
3 models, *Aerosol Science and Technology*, 31, 487-514, 1999.

4
5

1 Table 1. Aerosol and precursor gas emissions used in ModelE2-TOMAS and
 2 ModelE2-OMA and the nucleation scheme used in the ModelE2-TOMAS simulations
 3

Emission/Process	TOMAS model	Bulk model
Anthropogenic emissions	CMIP5 2000 emissions (Lamarque et al., 2012)	CMIP5 2000 emissions (Lamarque et al., 2012)
Biomass burning emissions	Climatological-average GFED3 emissions from 1997 to 2009 (van der Werf et al., 2010)	Climatological-average GFED3 emissions from 1997 to 2009 (van der Werf et al., 2010)
Primary sulfate emission assumption	1.0% of total sulfur emissions	2.5% of total sulfur emissions
DMS emission	Seawater DMS concentrations from Kettle et al. (1999) Sea-to-air transfer function from Liss and Merlivat (1986)	Seawater DMS concentrations from Kettle et al. (1999) Sea-to-air transfer function from Nightingale et al. (2000)
Sea-salt emission	Gong et al. (2002); the upper diameter limit of 10 um	Gong et al. (2002); the upper diameter limit of 8 um
Dust emission	See text for the details; the upper size diameter of 10 um	See text for the details; the upper size diameter of 16 um
Nucleation	Three nucleation cases 1. BASE - Binary nucleation 2. LowNUC - Binary nucleation with 5 times lower sulfuric acid concentrations 3. NoNUC – no nucleation	N/A

4
 5

6 Table 2. Emission size distributions assumed in ModelE2-TOMAS. *This is the soil
 7 size assumption used in ModelE2-TOMAS, and the dust emission size distribution is
 8 additionally influenced by meteorological variables. GMD stands for geometric mean
 9 diameter, and GSD for geometric standard deviation.

10

11

Species	Emissions	Size assumptions
Sulfate	All emissions	Bimodal distribution GMD=10nm, GSD=1.6 (5% of total mass) GMD=70 nm, GSD=2.0 (95% of total mass)
EC and OC	Fossil fuel and Biofuel Biomass burning	GMD=60 nm, GSD=1.59 GMD=150 nm, GSD=1.59
Dust*	Clay Silt	GMD=140 nm, GSD=2.0 GMD=1.15 μm , GSD=2.0

1 Table 3. Global budgets for DMS and SO₂ from the BASE run in ModelE2-TOMAS and
 2 ModelE2-OMA. The ModelE2-TOMAS values are presented before slash and the
 3 ModelE2-OMA values are after slash. Values in parentheses are ranges from other
 4 global models including Wang et al. (2011), Liu et al. (2005), and those listed in Liu
 5 et al. (2005).

6

	DMS	SO2
Burden [Tg S]	0.05 / 0.11 (0.02-0.15)	0.36 / 0.38 (0.2-0.69)
Total source [Tg S yr ⁻¹]	16.1 / 28.7	80 / 90
Emission	16.1 / 28.7 (10.7-23.7)	65.6 / 64.7 (61.2-92.0)
Chemistry	-	14.4 / 25.3
Sink [Tg S yr ⁻¹]	16.1 / 28.7	80 / 89 12.3 / 14.6
Gas-phase oxidation	16.1 / 28.7	(6.1-22.0) 30.8 / 35.8
Aqueous-phase oxidation	-	(24.5-57.8) 0.36 / 0.4
Wet deposition	-	(0-19.9) 37 / 38.8
Dry deposition	-	(15.78-55)
Lifetime [days]	1.2 / 1.5 (0.5-3.0)	1.9 / 1.5 (0.6-2.6)

7
 9

1 Table 4. Global aerosol budgets of the BASE run in ModelE2-TOMAS. Values in the
 2 parentheses are the mean and normalized standard deviations obtained from Table
 3 10 in Textor et al. (2006). The mass budgets for sulfate are presented as Tg S.
 4

	Sulfate	Elemental carbon (EC)	Organic matter (OM)	Sea-salt	Dust
Burden [Tg]	0.67	0.19	1.2	3.6	9.1
Total source [Tg yr ⁻¹]	43.7	7.4	60.8	3231.9	705.8
Emission	0.66	7.4	43.7	3231.9	705.8
Wet deposition [Tg yr ⁻¹]	42.9	7.1	59.1	1046.9	336.8
By convective clouds [%]	27	24	24	54	29
Dry deposition [Tg yr ⁻¹]	0.8	0.3	1.6	2184.9	369.9
Lifetime [days]	5.6 (4.1, 18%)	9.6 (7.1, 33%)	7.2 (6.5, 27%)	0.4 (0.5, 58%)	4.7 (4.1, 43%)
Removal rate coefficient [day ⁻¹]	0.18 (0.25, 18%)	0.1 (0.15, 21%)	0.14 (0.16, 24%)	2.4 (5.1, 188%)	0.21 (0.31, 62%)
Wet deposition	0.18 (0.22, 22%)	0.1 (0.12, 31%)	0.14 (0.14, 32%)	0.79 (0.79, 77%)	0.1 (0.08, 42%)
Dry deposition	0.0032 (0.03, 55%)	0.004 (0.03, 55%)	0.0037 (0.03, 49%)	1.6 (4.3, 219%)	0.11 (0.23, 84%)
Wet/(Wet+Dry) [%]	98 (89, 8%)	96 (79, 17%)	97 (80, 16%)	32 (31, 65%)	48 (33, 54%)

5
 6 Table 5. Global aerosol budgets in ModelE2-OMA. Note that the sulfate and nitrate
 7 budgets are presented as Tg S and Tg N, respectively.

	Sulfate	EC	OM	Sea-salt	Dust	Nitrate
Burden [Tg]	0.38	0.12	1.1	9.3	11.4	0.37
Total source [Tg yr-1]	52.0	7.4	58.5	2866.7	1071.8	21.0
Emission	1.7	7.4	43.7	2866.7	1071.8	
Wet deposition [Tg yr-1]	46.6	5.4	44.9	2059.1	407.7	17.7
By convective clouds [%]	21	37	32	39	52	29
Dry deposition [Tg yr-1]	5.4	2.0	13.7	806.9	664.1	3.3
Lifetime [days]	2.6	5.8	7.1	1.2	3.9	6.4
Removal rate coefficient [day-1]	0.39	0.17	0.14	0.84	0.26	0.16
Wet deposition	0.35	0.13	0.11	0.61	0.10	0.13
Dry deposition	0.04	0.046	0.033	0.24	0.16	0.025
Wet/(Wet+Dry) [%]	90	73	77	72	38	84

1 Table 6. Statistical measures of model predictions compared to satellites. For
2 ModelE2-OMA, the model prediction without nitrate aerosols is also presented in
3 parentheses.

4

		Model vs. MODIS	Model vs. MISR	MODIS vs. MISR
ModelE2-TOMAS	Correlation	0.63	0.73	0.79
	NMB [%]	-29%	-34%	8%
ModelE2-OMA	Correlation	0.45 (0.45)	0.52 (0.55)	0.79
	NMB [%]	16% (-16%)	8% (-21%)	8%

5

1 Table 7. Locations of AERONET sites and corresponding measurement time periods.
 2

	Sites	Longitude	Latitude	Years
1	Alta Floresta	56.0° W	9.9° S	1999-2005
2	Abracos Hill	62.0° W	11° S	1999-2005
3	Cuiaba-Miranda	56.0° W	15.7° S	2001-2005
4	Mongu	23.2° E	15.2° S	1995-2005
5	Ilorin	4.3° E	8.3° N	1998-2005
6	Banizombou	2.0° E	13.0° N	1995-2005
7	Capo Verde	22.9° W	16.7° N	1994-2004
8	Bidi Bahn	2.5° W	14.1° N	1996-1997
9	Barbados	59.5° W	13.2° N	1996-2000
10	Sede Boker	34.8° E	30.9° N	1998-2005
11	Bahrain	50.6° E	26.2° N	2004-2005
12	Solar Village	46.4° E	24.9° N	1999-2005
13	Dalanzadgad	104.4° E	43.6° N	1997-2005
14	Yulin	109.7° E	38.3° N	2001-2002
15	Sevilleta	106.9° W	34.4° N	1994-2005
16	Cart site	97.5° W	36.6° N	1996-2005
17	Bondville	88.4° W	40.1° N	1996-2005
18	GSFC	76.8° W	39.0° N	1995-2005
19	Mexico city	99.2° W	19.3° N	1999-2005
20	Ispra	8.6° E	45.8° N	2001-2005
21	Kanpur	80.3° E	26.5° N	2001-2005
22	Shirahama	135.4° E	33.7° N	2000-2005
23	Bermuda	64.7° W	32.4° N	1996-2005
24	Lanai	156.9° W	20.7° N	1996-2004
25	Dry Tortugas	82.9° W	24.6° N	1996-2003
26	Tahiti	149.6° W	17.6° S	1999-2005
27	Rottnest Island	115.5° E	32.0° N	2001-2004
28	Nauru	166.9° E	0.5° S	1999-2005

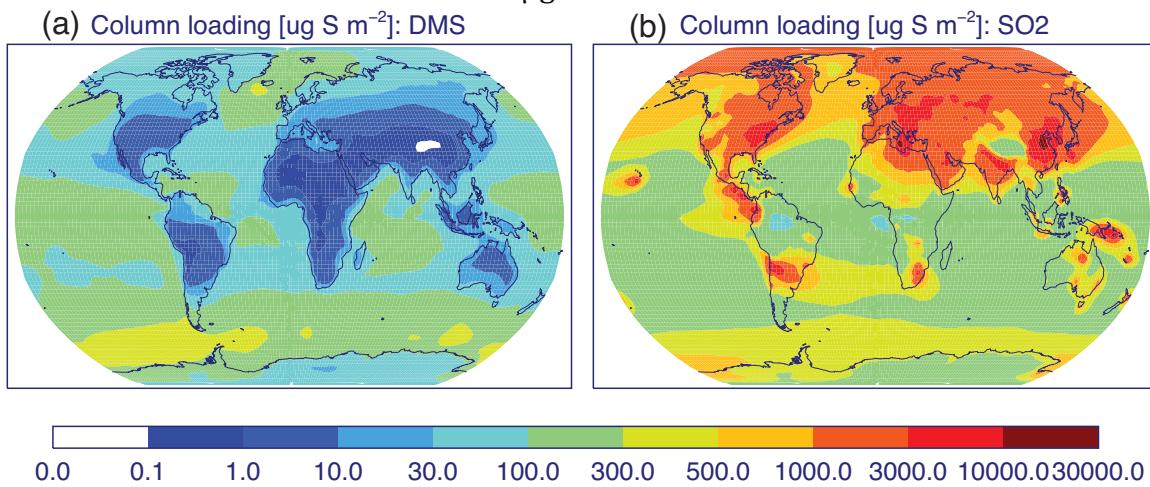
3

1 Table 8. Summary of global annual-average of tropospheric and surface-layer
 2 aerosol number budgets including J3 (new particle formation rates at 3 nm), CN3
 3 (number concentration of particles with $D_p \geq 3$ nm), CN10($D_p \geq 10$ nm), CN100(D_p
 4 ≥ 100 nm) in ModelE2-TOMAS. Values normalized by tropospheric volume at 273 K
 5 and 1 atm.
 6

	Emission rate [cm ⁻³ s ⁻¹]	Tropospheric J3 [cm ⁻³ s ⁻¹]	Tropospheric CN3 [cm ⁻³]	Tropospheric CN10 [cm ⁻³]	Tropospheric CN100 [cm ⁻³]	Surface-layer CN3 [cm ⁻³]	Surface-layer CN10 [cm ⁻³]	Surface-layer CN100 [cm ⁻³]
Base	5.47x10 ⁻⁴	0.131	4852	939	211	1622	1331	416
LowNUC	5.47 x10 ⁻⁴	0.013	1277	628	197	1152	1111	405
NoNUC	5.47 x10 ⁻⁴	0.000	222	221	159	935	919	374

7
 8

1 Figure 1. Annual-average column mass concentrations of (a) DMS and (b) SO₂ in the
 2 ModelE2-TOMAS BASE run. Units are $\mu\text{g S m}^{-2}$.

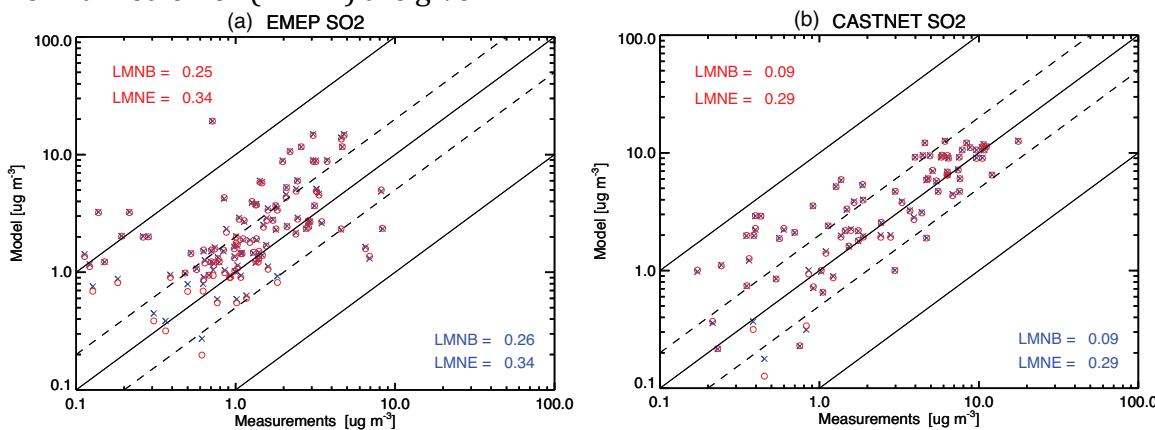


3

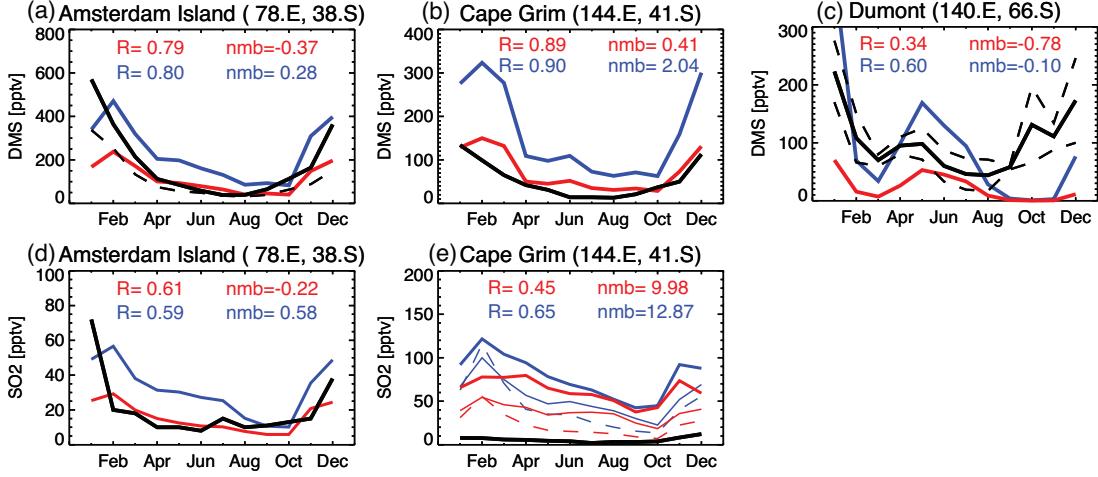
4

5

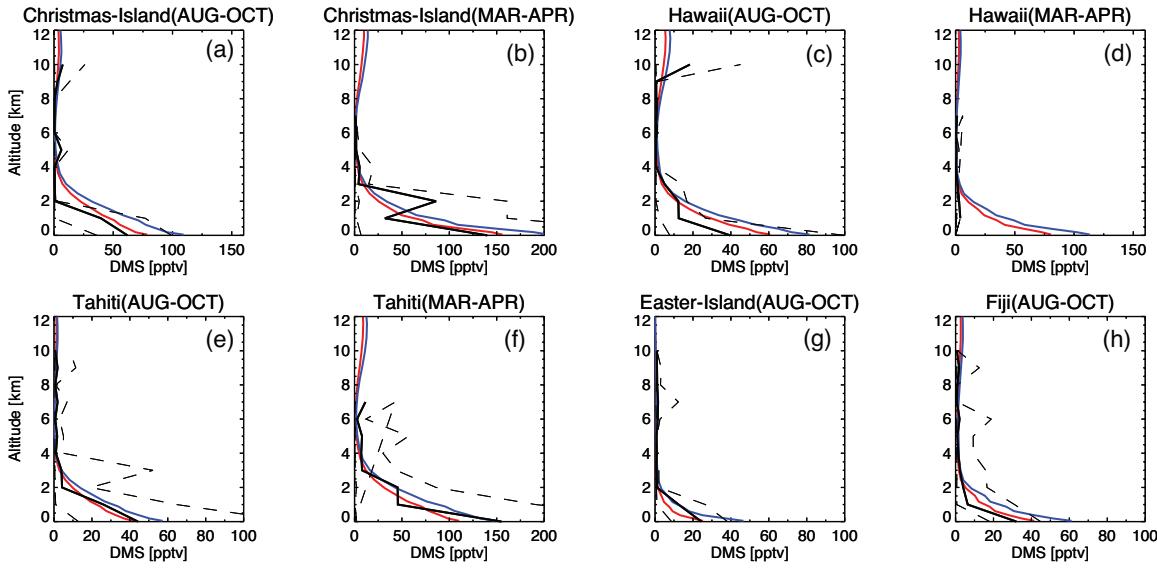
6 Figure 2. Scatter plot of annual-mean surface SO₂ concentrations [$\mu\text{g m}^{-3}$] for the
 7 model (red for ModelE2-TOMAS and blue for ModelE2-OMA) compared to the
 8 observations in the EMEP (European Monitoring and Evaluation Programme,
 9 Loevblad et al., 2004; a) and CASTNET (Clean Air Status and Trends Network, Malm
 10 et al., 2002; b) networks. Log-mean normalized bias (LMNB) and log-mean
 11 normalized error (LMNE) are given.



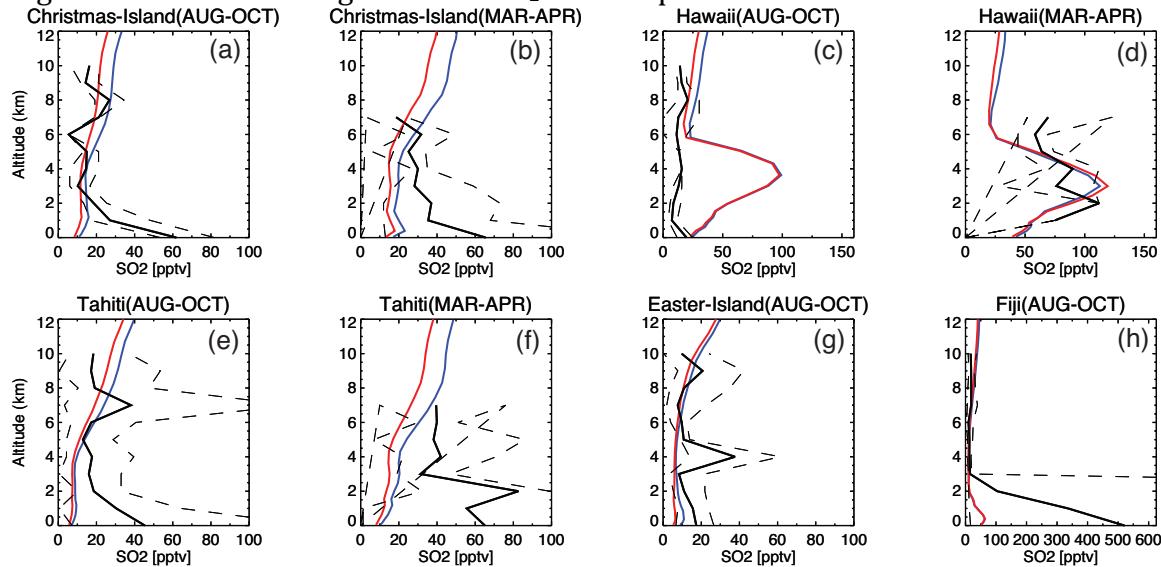
1 Figure 3. Comparisons of monthly averaged surface DMS (a to c) and SO₂ (d to e)
 2 mixing ratios [pptv] simulated (red for ModelE2-TOMAS and blue for ModelE2-
 3 OMA) and measured (black) at Amsterdam Island [Nguyen et al., 1992] and Cape
 4 Grim [Ayers et al., 1991]. Only DMS at Dumont D'Urville [Jourdain and Legrand,
 5 2001]. Correlation (R) and normalized mean bias (NMB) are given.



6
 7
 8
 9 Figure 4. Comparison of DMS vertical profiles over the Pacific Ocean simulated (red
 10 for ModelE2-TOMAS and blue for ModelE2-OMA) and observed (black solid line).
 11 Observations are from PEM-Tropic-A (August-October 1996 in the tropical Pacific;
 12 Christmas Island in a; Hawaii in c; Tahiti in e; Easter Island in g; Fiji in h) and PEM-
 13 Tropic-B (March-April 1999 in the tropical Pacific; Christmas Island in b; Hawaii in
 14 d; Tahiti in f). The dashed lines represent 25th and 75th percentiles of the observed
 15 values.
 16



1 Figure 5. The same as Fig. 4 but for SO₂ vertical profiles.



2

3

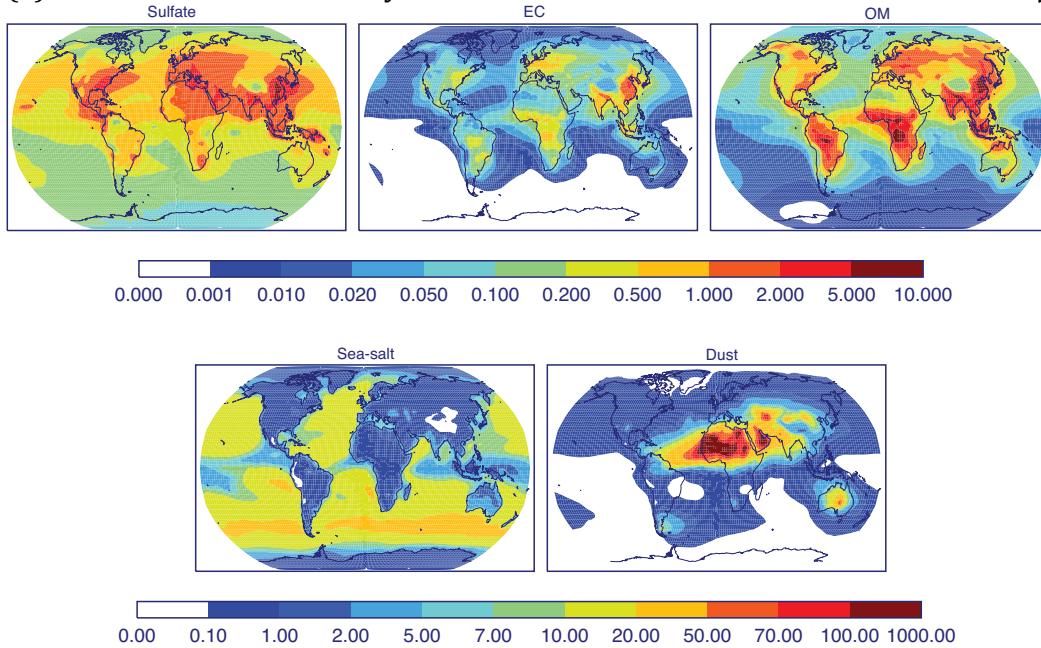
4

5

6

7

8 Figure 6. Annual-mean concentrations of (a) sulfate, (b) EC, (c) OM, (d) sea-salt, and
9 (e) dust in the lowermost layer in the ModelE2-TOMAS BASE run. Units are $\mu\text{g m}^{-3}$.



10

11

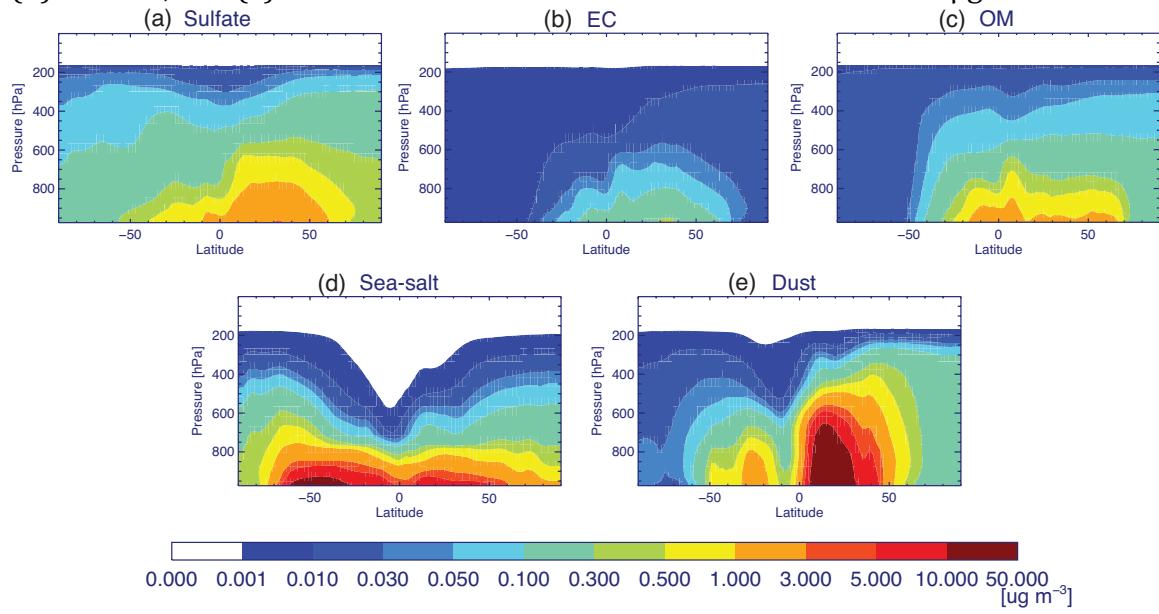
12

13

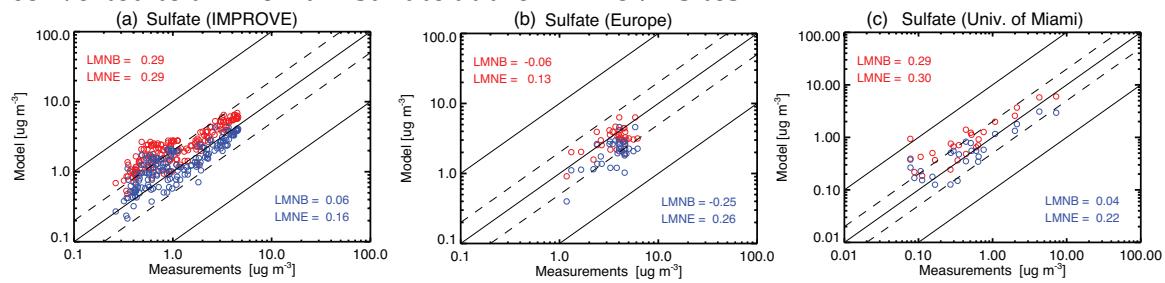
14

15

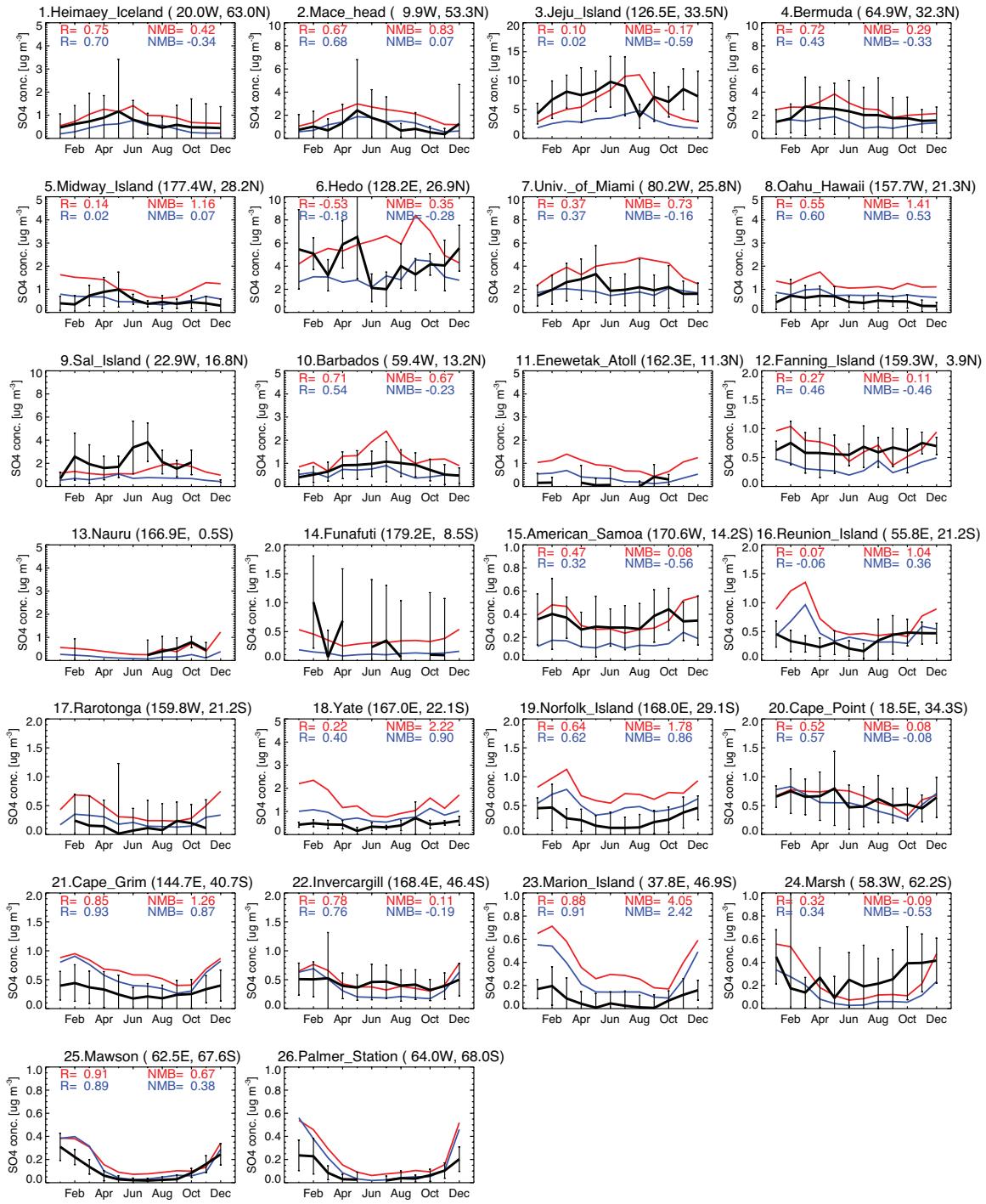
1 Figure 7. Annually and zonally averaged concentrations of (a) sulfate, (b) EC, (c) OM,
 2 (d) sea-salt, and (e) dust in the ModelE2-TOMAS BASE run. Units are $\mu\text{g m}^{-3}$.



3
 4 Figure 8. Scatter plot of annually averaged surface sulfate mass concentrations [μg
 5 m^{-3}] in the model (red for ModelE2-TOMAS and blue for ModelE2-OMA) and
 6 observation from (a) the IMPROVE network, (b) the European sites from Putaud et
 7 al. (2010), and (c) University of Miami. Note that the sulfate in the model is simply
 8 converted to ammonium sulfate at the IMPROVE sites.

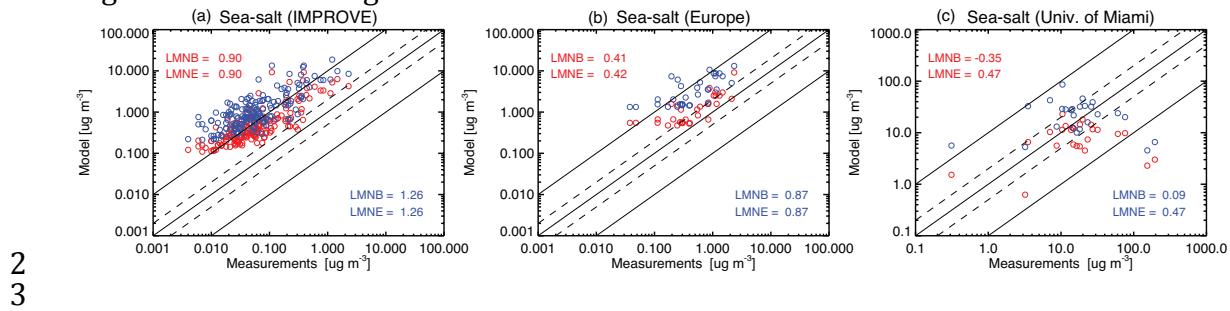


1 Figure 9. Comparisons of monthly averaged surface sulfate mass concentrations [$\mu\text{g m}^{-3}$] simulated (red for ModelE2-TOMAS and blue for ModelE2-OMA) and measured
 2 (black solid line with an error bar showing a standard deviation) by University of
 3 Miami. Correlation (R) and normalized mean bias (NMB) are provided only when
 4 the observation is available for 12 months.
 5

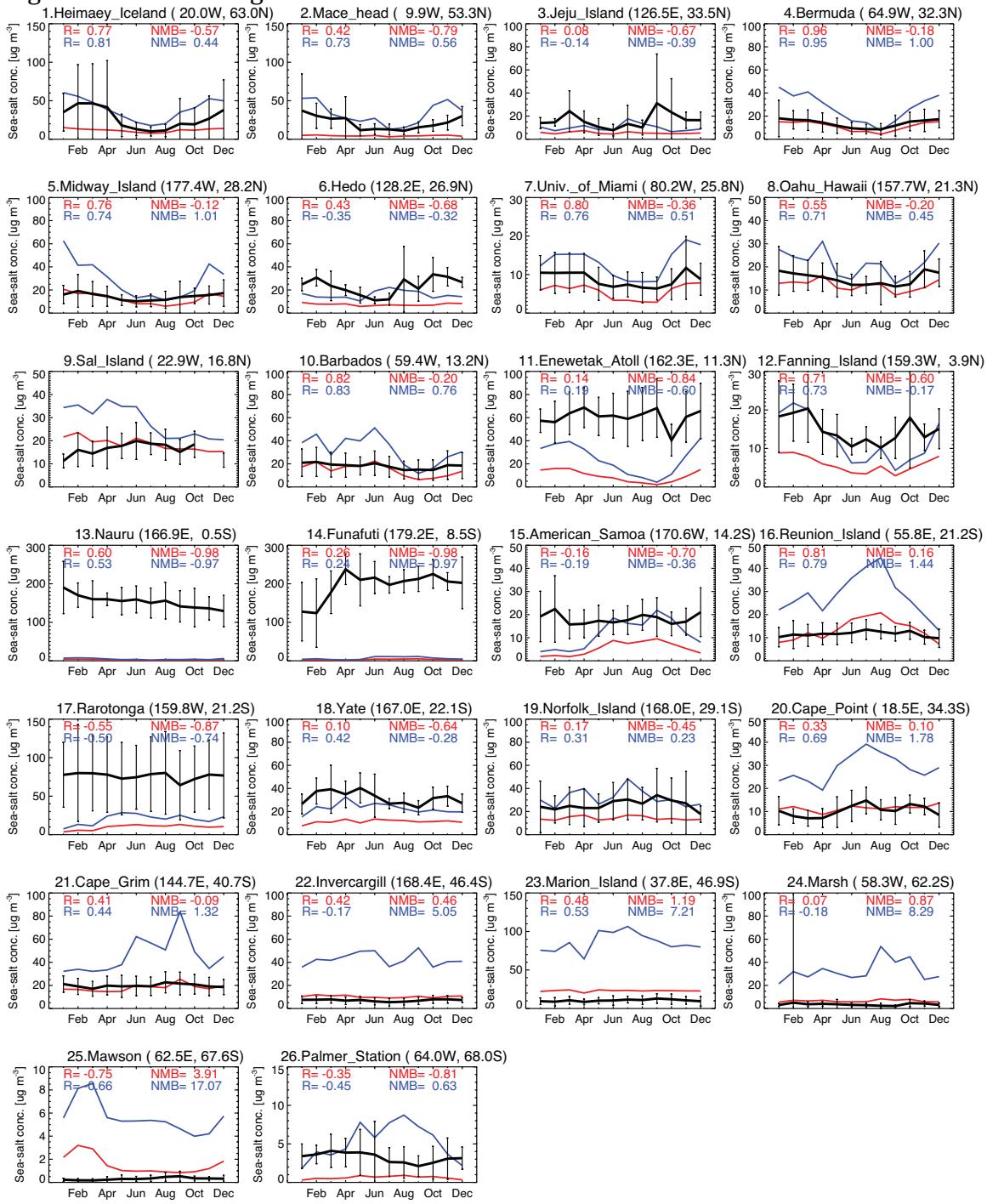


6
7
8
9

1 Figure 10. Same as Figure 8 but for sea-salt mass concentrations

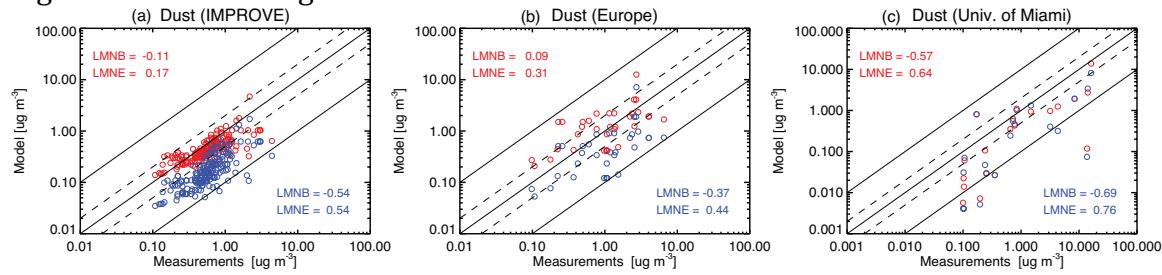


1 Figure 11. Same as Figure 9 but for sea-salt mass concentrations

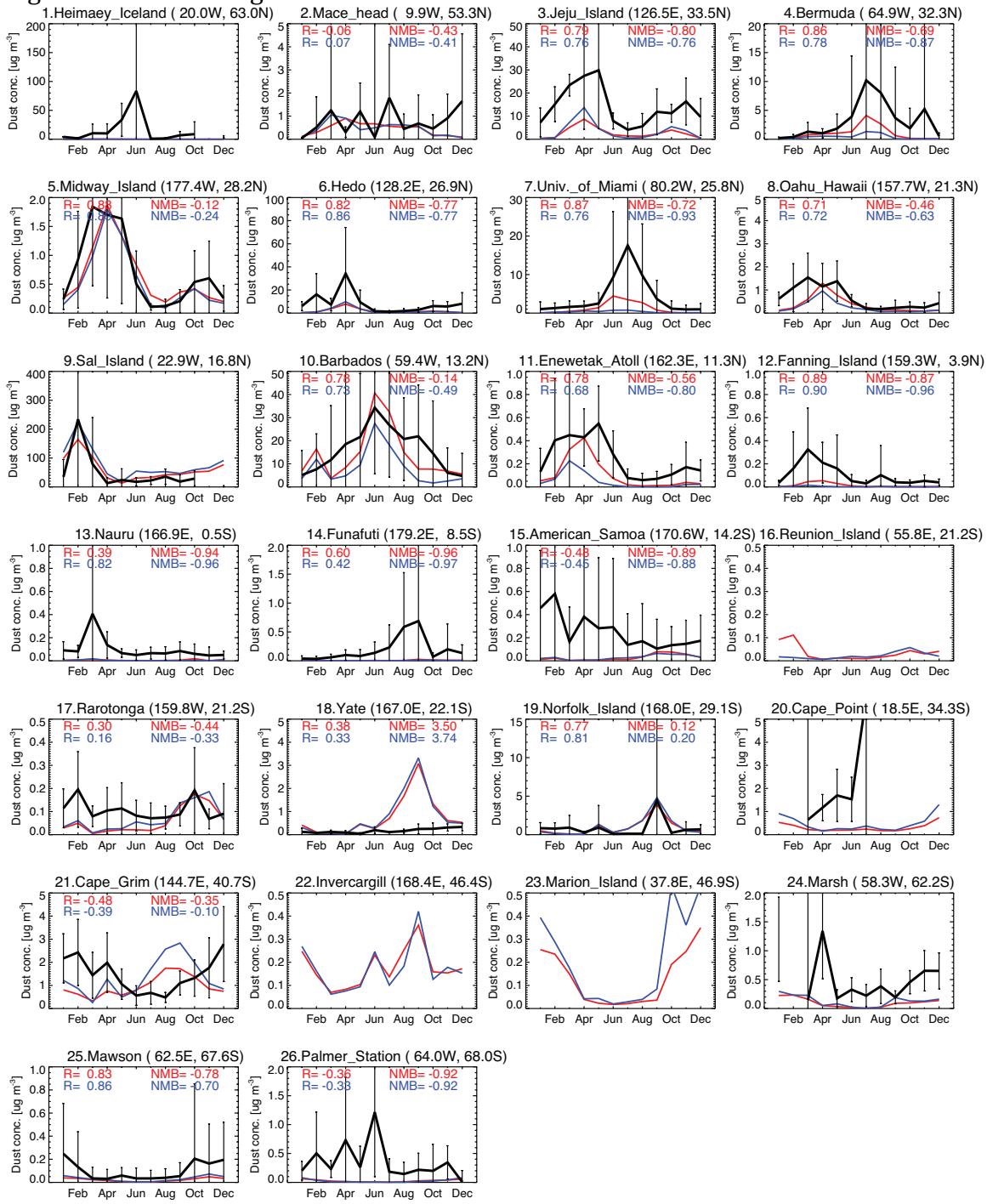


2
3
4

1 Figure 12. Same as Figure 8 but for dust mass concentrations

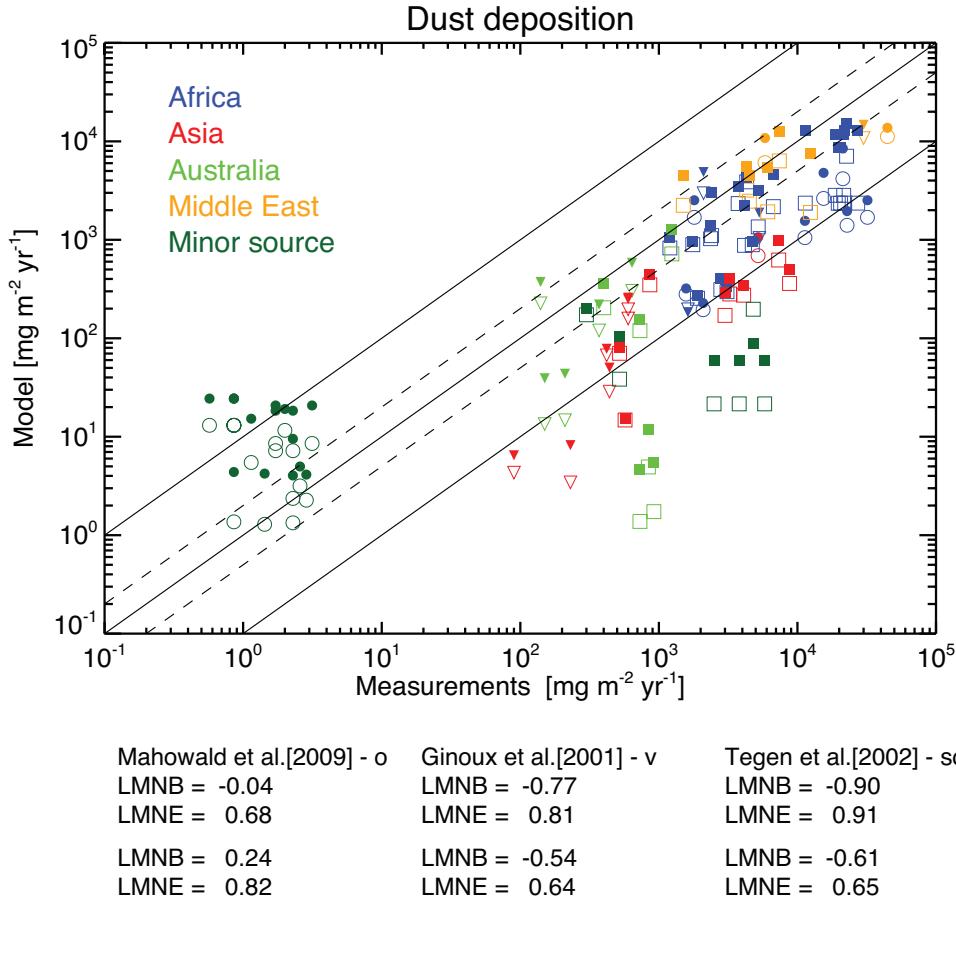


1 Figure 13. Same as Figure 9 but for dust mass concentrations



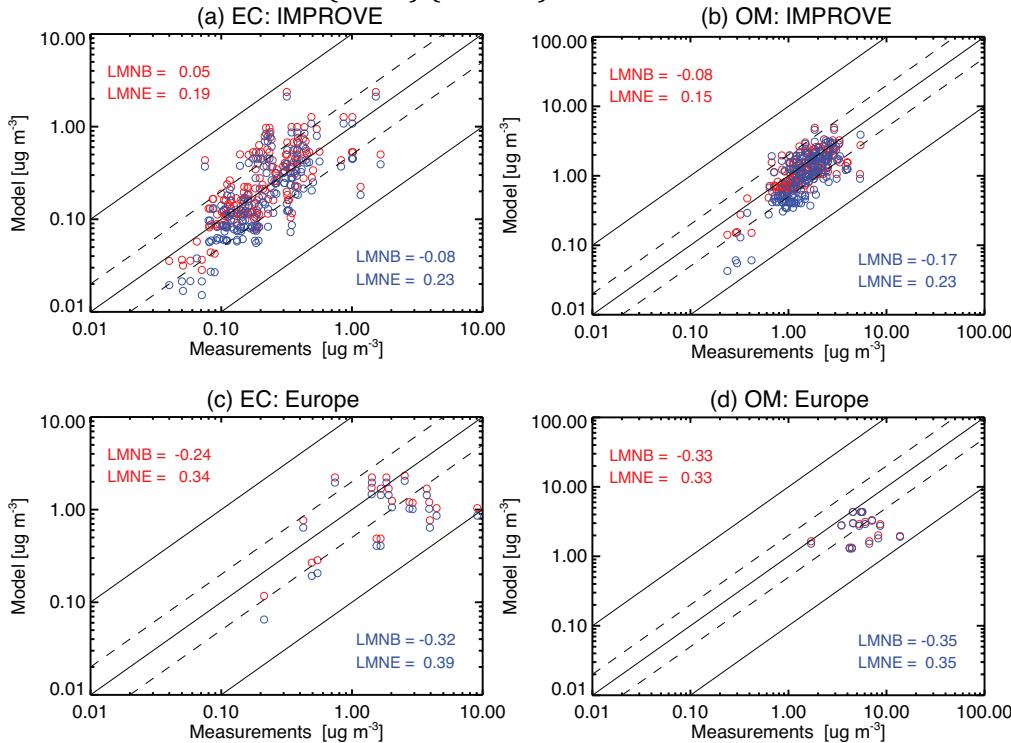
2
3

1 Figure 14. Scatter plot of annual-average dust deposition fluxes [$\text{mg m}^{-2} \text{yr}^{-1}$]
 2 simulated and observations obtained from Ginoux et al. (2001), Tegen et al. (2002)
 3 and Tables S2 in Mahowald et al. (2009). Open symbols are for ModelE2-TOMAS and
 4 the filled symbol for ModelE2-OMA. LMNB and LMNE are presented below the plot:
 5 the first values are for ModelE2-TOMAS.

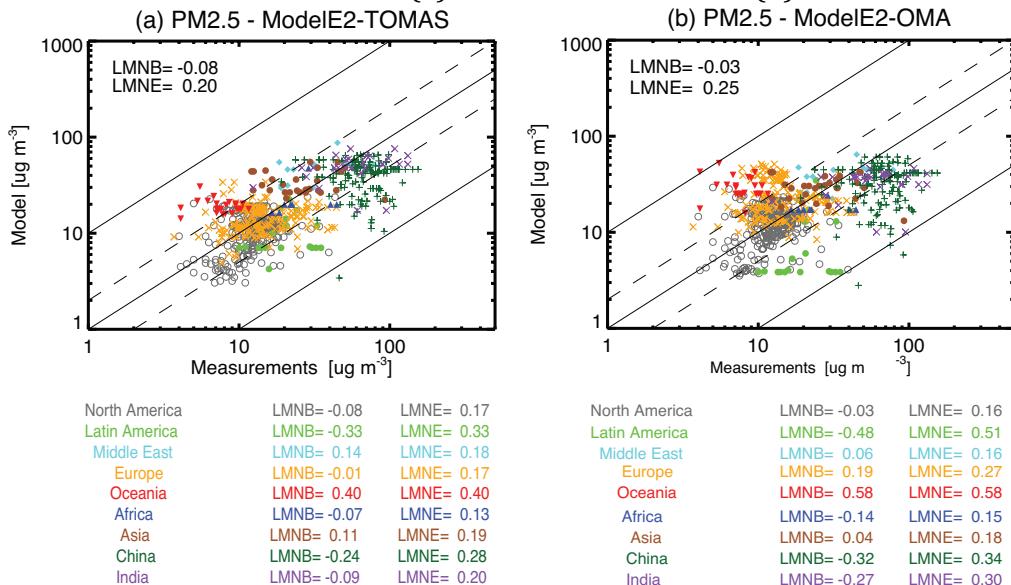


6
 7
 8

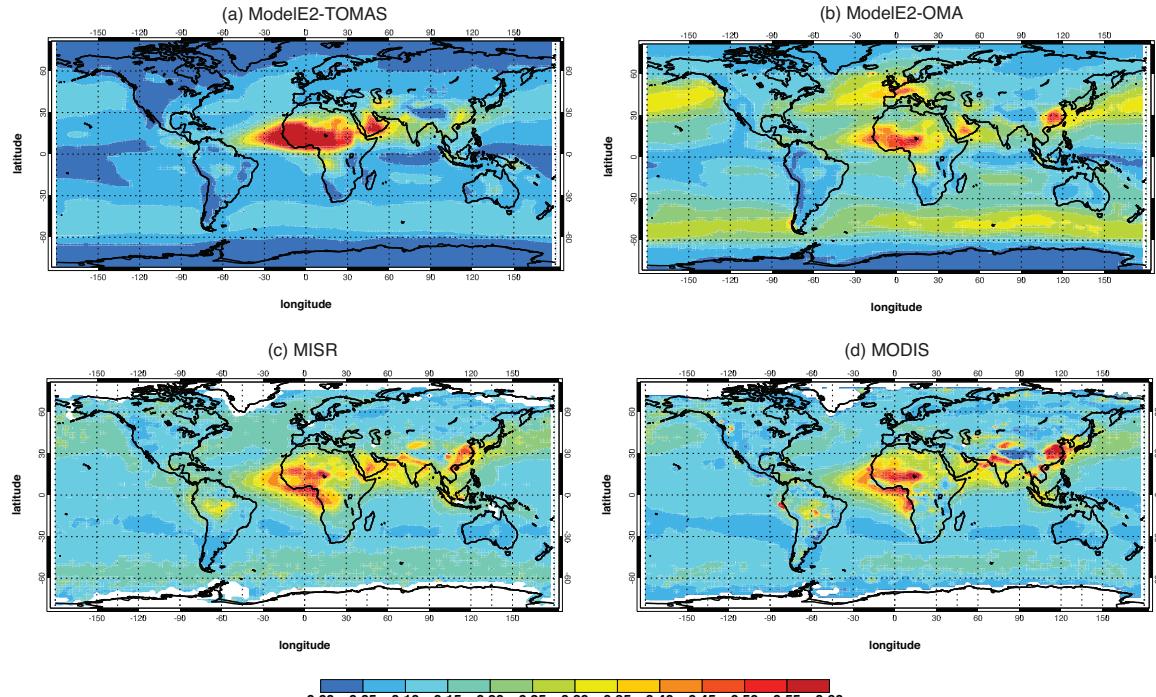
1 Figure 15. Scatter plots of annually averaged surface EC and OM mass
 2 concentrations [$\mu\text{g m}^{-3}$] simulated (red for ModelE2-TOMAS and blue for ModelE2-
 3 OMA) and observations from the IMPROVE network (a and b) and the European
 4 sites from Putaud et al. (2010) (c and d).



5
 6
 7 Figure 16. Scatter plots of annually averaged surface PM2.5 concentration [$\mu\text{g m}^{-3}$]
 8 simulated and observations compiled by the Global Burden of Disease (GBD) Study.
 9 ModelE2-TOMAS is shown in (a) and ModelE2-OMA in (b).
 10

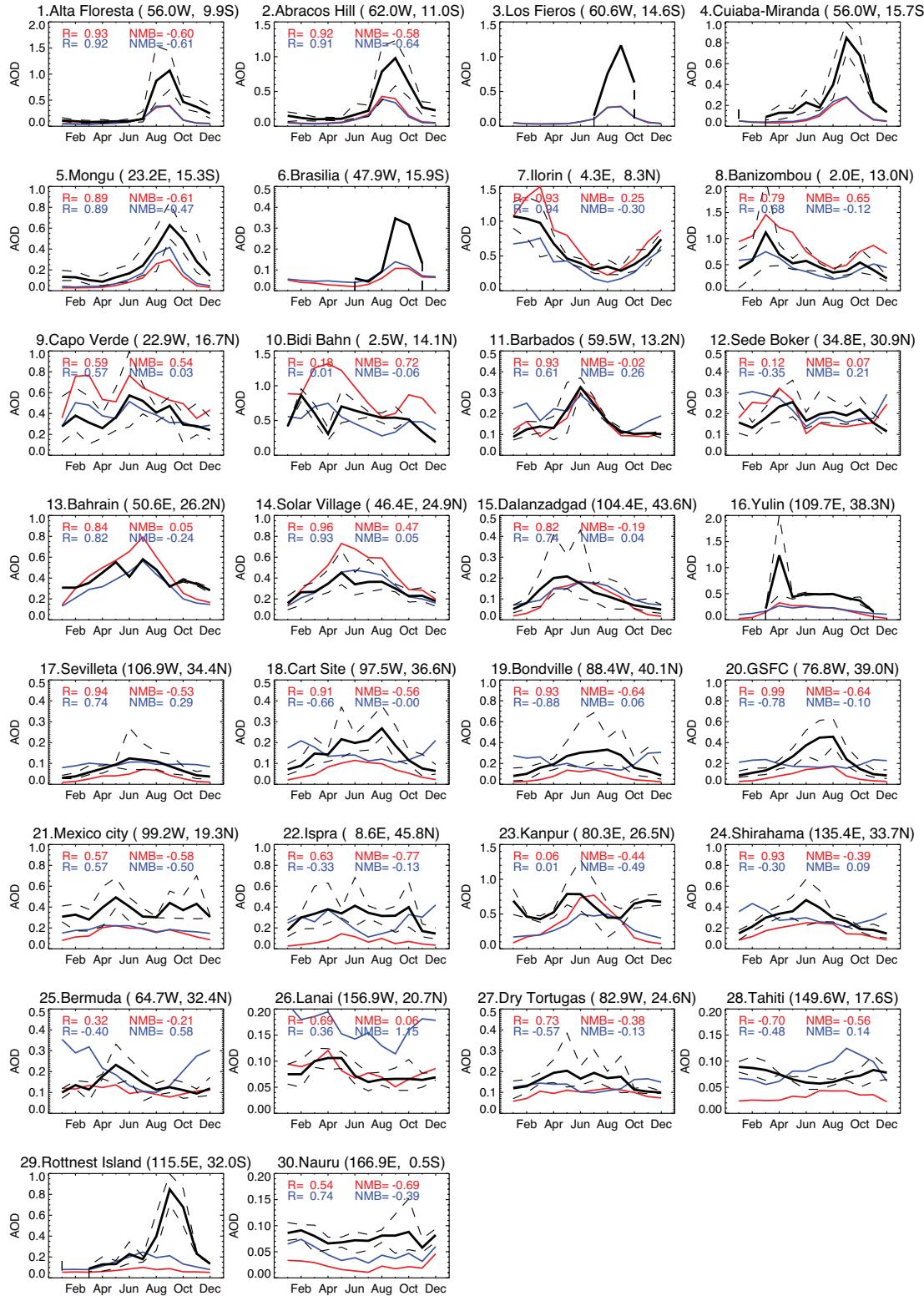


1 Figure 17. Global distributions of the annual-mean AOD from (a) ModelE2-TOMAS,
 2 (b) ModelE2-OMA, (c) MISR, and (d) MODIS. See Section 5.4 for the details of the
 3 MISR and MODIS AOD information.



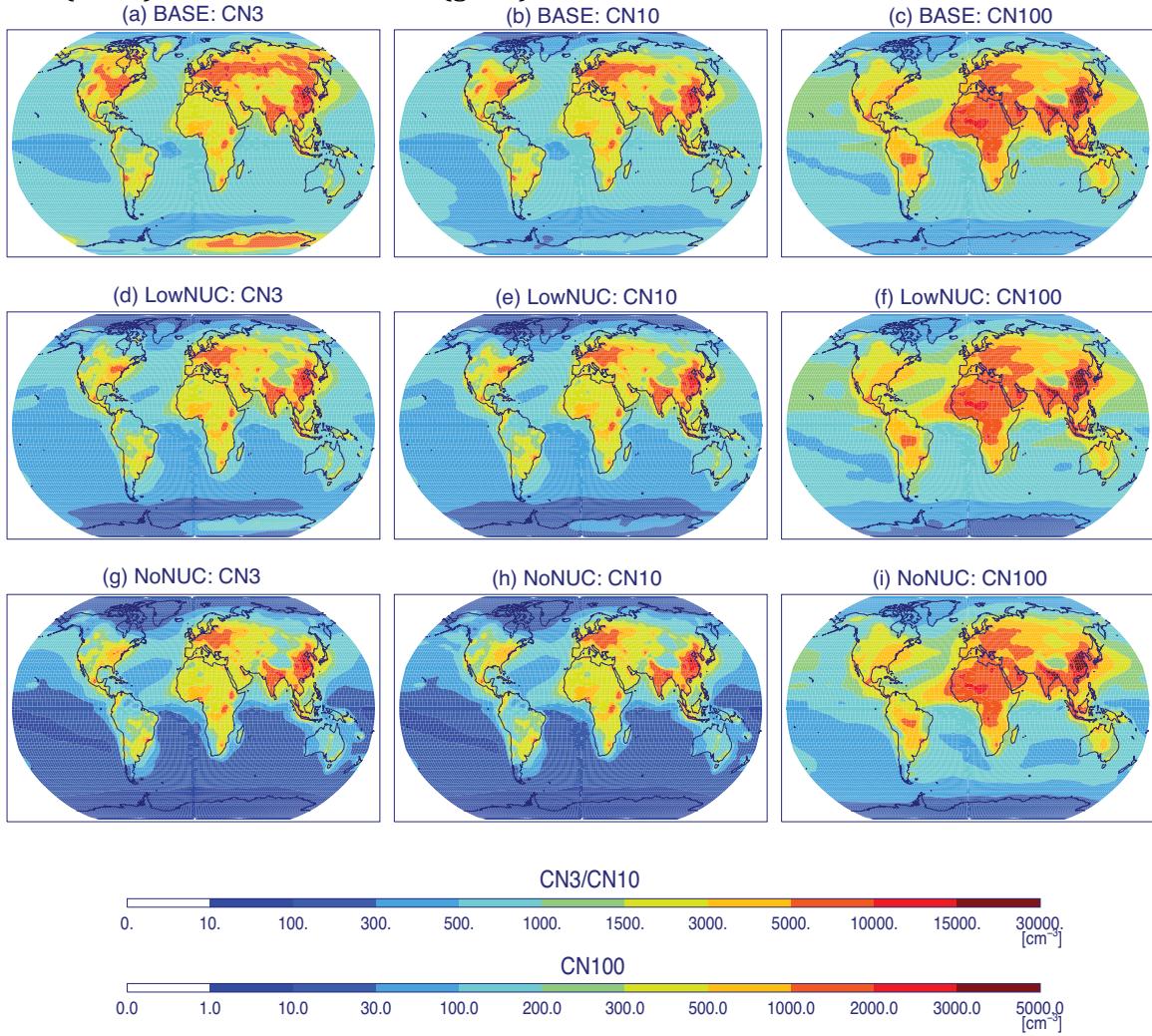
4
 5
 6
 7

1 Figure 18. Comparisons of monthly averaged model AOD (red for ModelE2-TOMAS
 2 and blue for ModelE2-OMA) and AERONET AOD (black solid line). Correlation (R)
 3 and normalized mean bias (nmb) are provided only when the observation is
 4 available for 12 months.



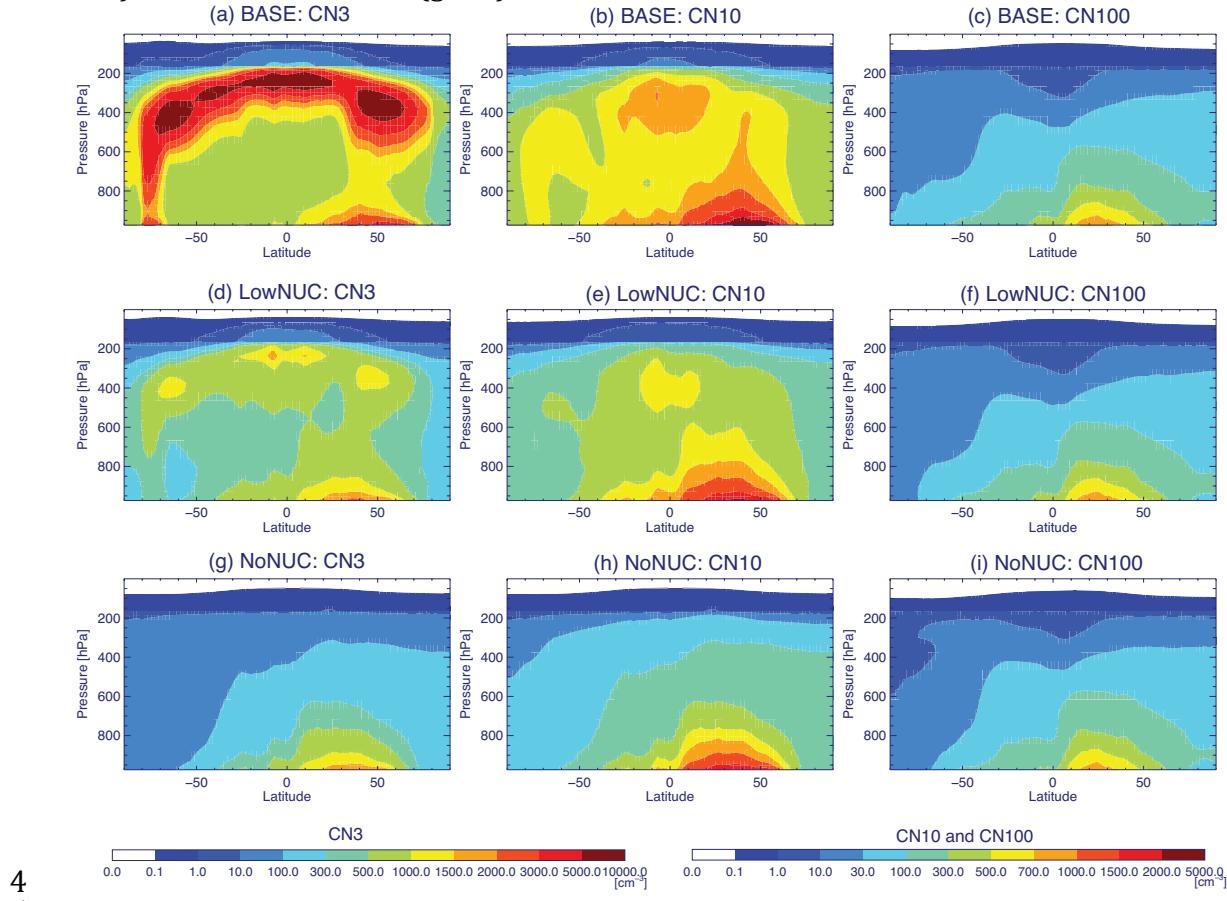
5

1 Figure 19. Annually averaged CN3 ($D_p \geq 3$ nm), CN10 ($D_p \geq 10$ nm), CN100 ($D_p \geq 100$ nm) concentrations in the lowermost layer for the BASE run (a to c), the LowNUC run (d to f), and the NoNUC run (g to i) of ModelE2-TOMAS. Units are cm^{-3} .



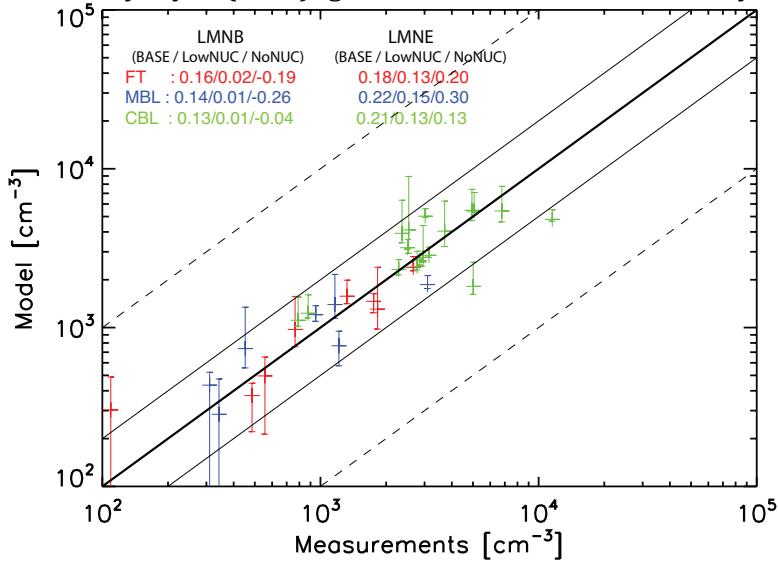
4
5
6
7

1 Figure 20. Annually and zonally averaged CN3 ($D_p \geq 3$ nm), CN10($D_p \geq 10$ nm), and
 2 CN100($D_p \geq 100$ nm) concentrations for the BASE run (a to c), the LowNUC run (d
 3 to f), and the NoNUC run (g to i) of ModelE2-TOMAS. Units are cm^{-3} .



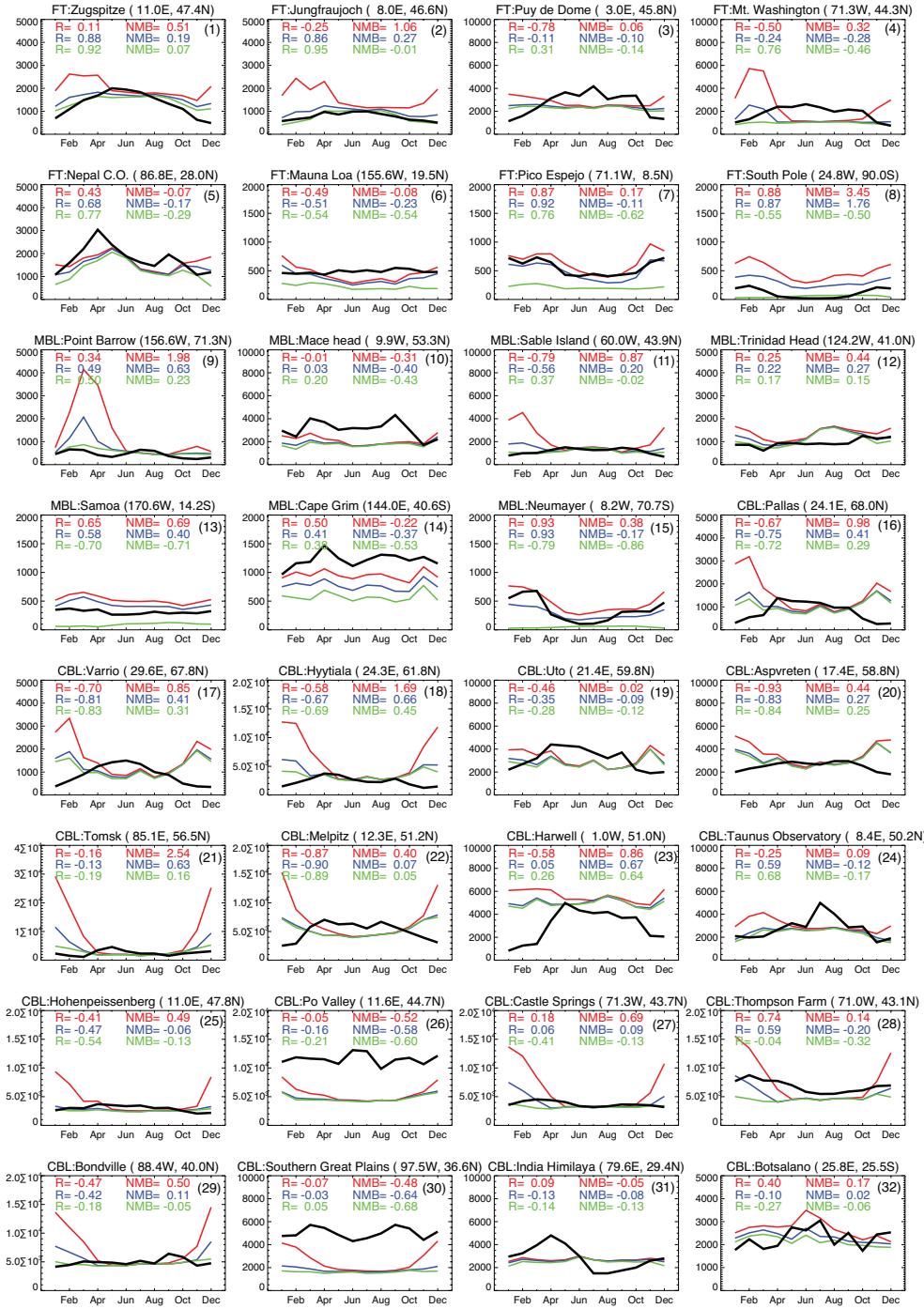
4
 5
 6
 7

1 Figure 21. Scatter plot of simulated annual-mean aerosol number concentrations in
 2 comparison with a dataset of surface number concentrations measurements at 36
 3 sites around the world compiled by Spracklen et al. (2010). The top horizontal bar
 4 represents the BASE results, and the middle bar the LowNUC results and the lower
 5 bar the NoNUC results. Red color is for free troposphere (FT); blue for marine
 6 boundary layer (MBL); green for continental boundary layer (CBL).

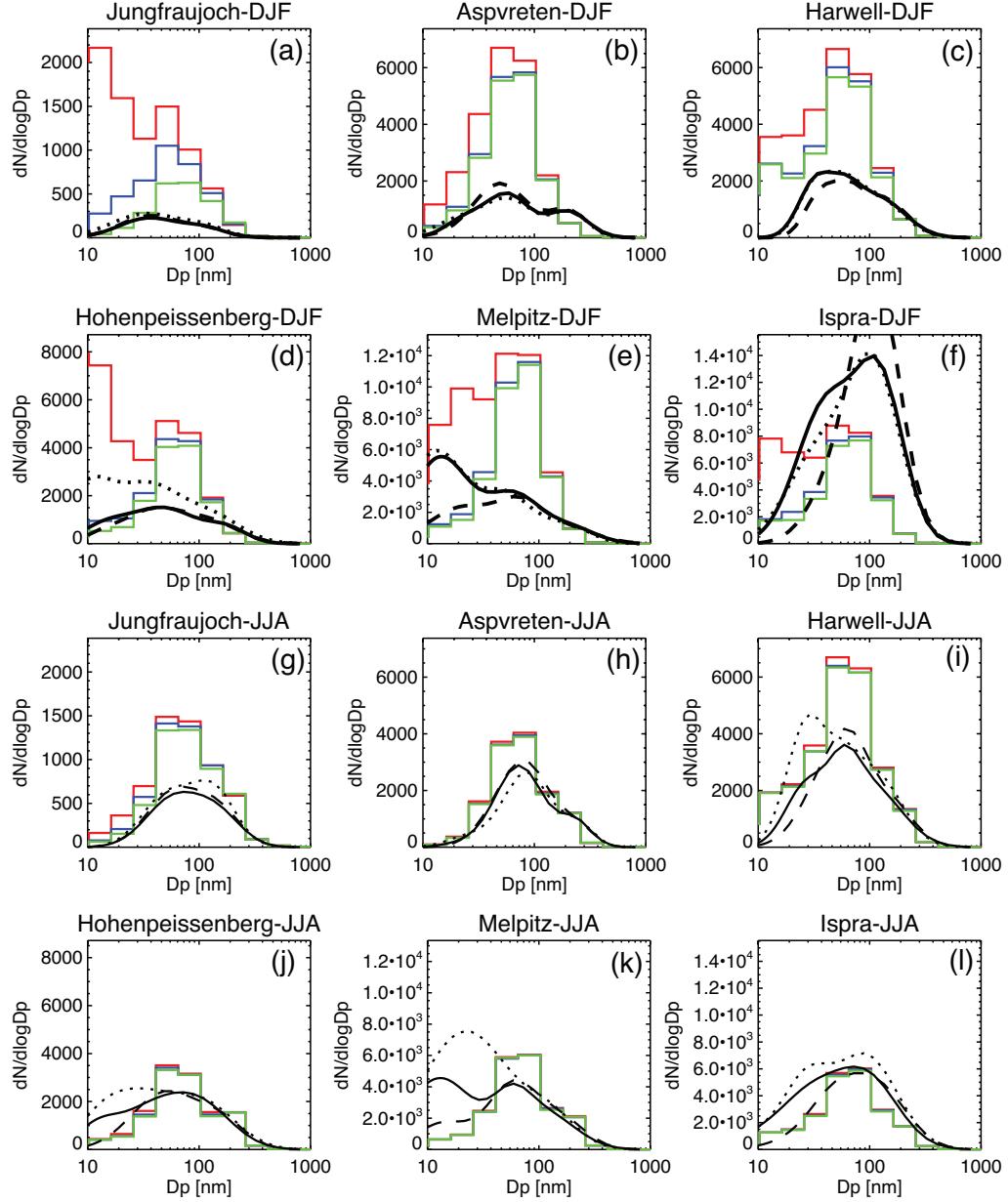


7
 8
 9
 10
 11

1 Figure 22. Annual cycle of aerosol number concentrations [cm^{-3} at ambient
 2 conditions] at 32 sites. The observations are shown in black, and three model
 3 results are presented: red for BASE, blue for LowNUC, and green for NoNUC. The
 4 free tropospheric (FT) sites are from 1 to 8; the marine boundary layer (MBL) sites,
 5 9 to 15; the continental boundary layer (CBL) sites, 16 to 32. Correlation (R) and
 6 normalized mean bias (NMB) are provided only when the observation is available
 7 for 12 months.

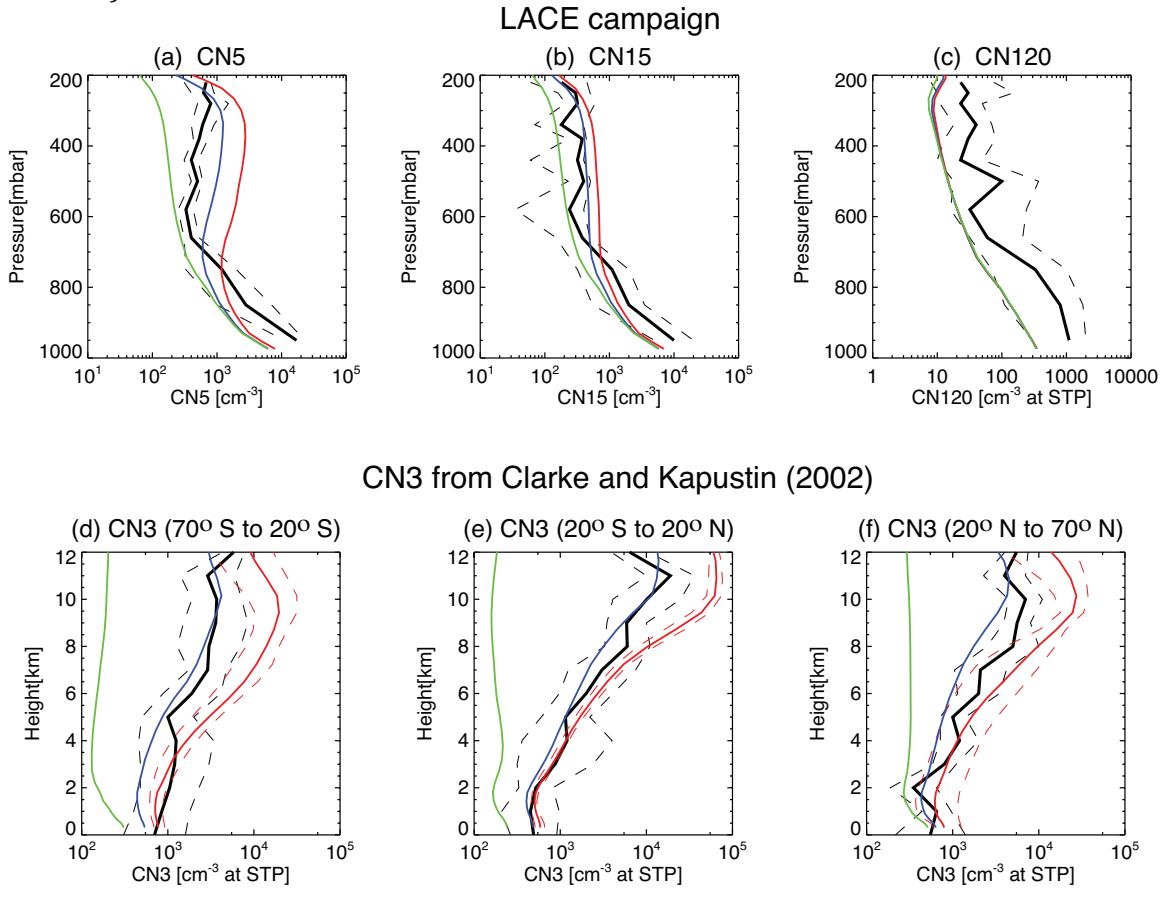


1 Figure 23. Number size distributions from European sites during winter (DJF; a to f)
 2 and summer (JJA; g to l) that are obtained from log-normal 3-mode fits during
 3 morning (black solid), afternoon (black dotted), and night (black dashed). The
 4 model results are the seasonal mean, shown in red lines for the BASE run, blue lines
 5 for the LowNUC run, and green lines for the NoNUC run.



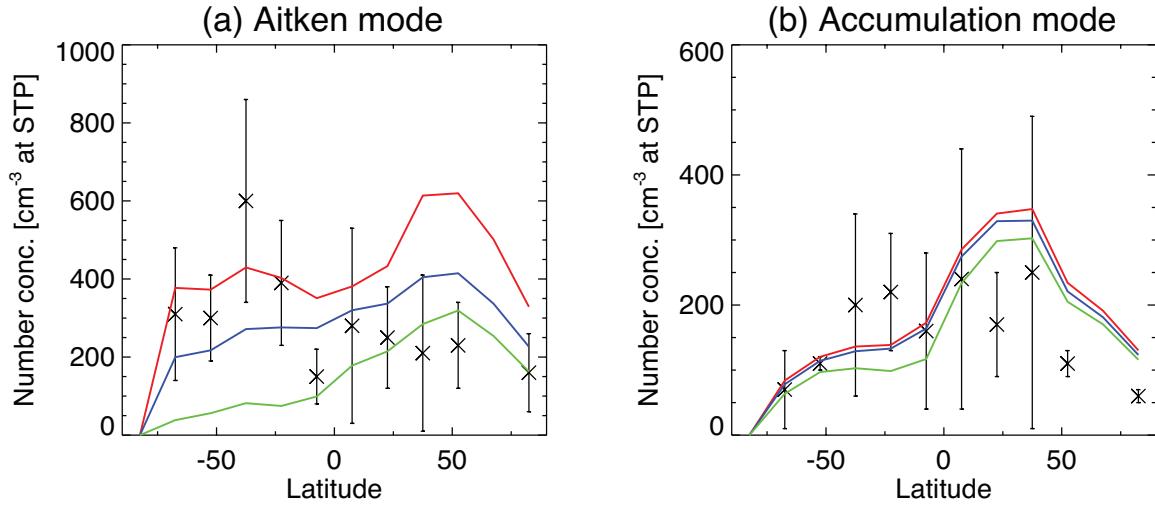
6
 7
 8
 9

1 Figure 24. Vertical profiles of aerosol number concentrations from the observations
 2 (black lines) and ModelE2-TOMAS (red lines for the BASE run, blue lines for the
 3 LowNUC run, and green lines for the NoNUC run). The CN5, CN15, CN120
 4 concentrations (ambient conditions) from LACE campaign (Petzold et al., 2002, in
 5 north-east Germany) are shown in a to c, respectively. The observed CN3
 6 concentrations (STP conditions: 1atm, 273 K) over the Pacific Ocean, which are
 7 averaged into the 3 latitude bands (70 S to 20S, 20S to 20N, and 20N to 70N; Clarke
 8 and Kapustin, 2002), are shown in d to f. The dashed lines show the standard
 9 deviation for the observations and the min/max monthly mean for the model (only
 10 in d to f).



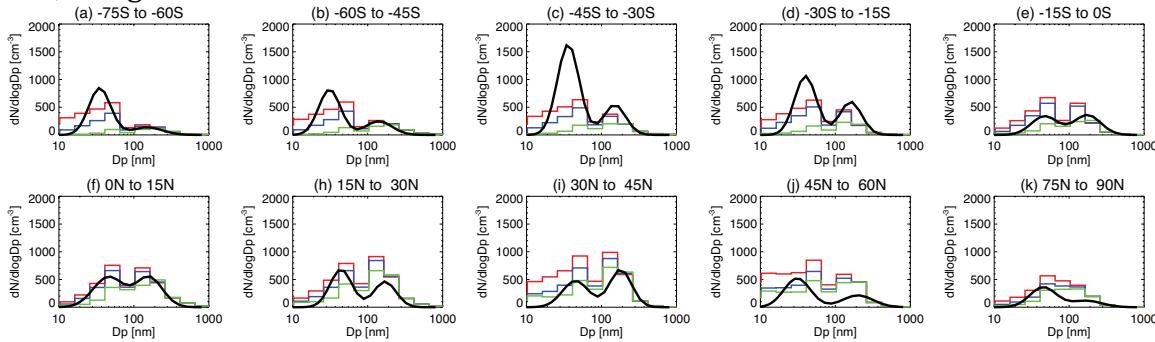
11
 12
 13
 14
 15

1 Figure 25. Simulated number concentrations in (a) Aitken mode ($10 \text{ nm} \leq D_p < 100 \text{ nm}$) and (b) accumulation mode ($100 \text{ nm} \leq D_p < 1 \mu\text{m}$) compared to the
 2 observations ("x" symbol with error bar representing minimum and maximum
 3 observed concentrations) that were compiled and aggregated into 15-degree
 4 latitude ranges (Heintzenberg et al. 2002). The model is also averaged to the 15-
 5 degree grid and is shown in red lines for the BASE run, blue lines for the LowNUC
 6 run, and green lines for the NoNUC run.
 7



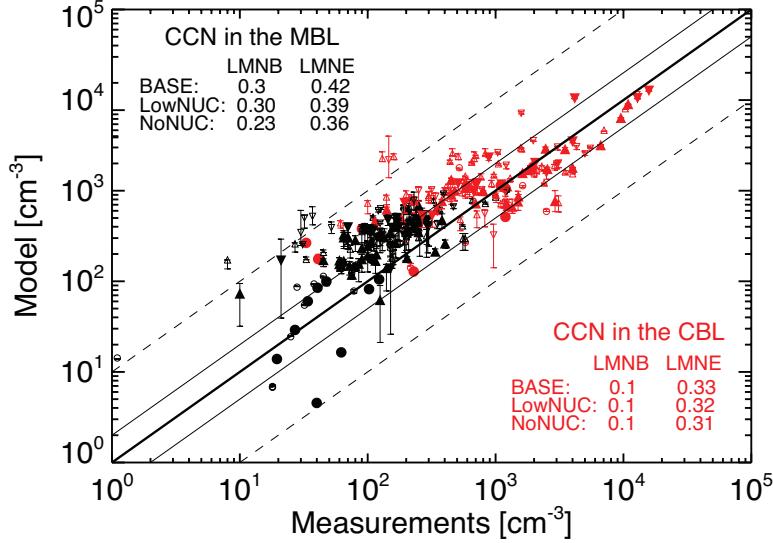
8
 9
 10
 11
 12
 13
 14
 15
 16
 17
 18
 19

Figure 26. Annually and zonally averaged aerosol size distributions in the marine boundary layer. Observations are from Heintzenberg et al. (2000) and were compiled and aggregated into a $15^\circ \times 15^\circ$ grid. The model is also averaged to the 15-degree grid and is shown as red lines for the BASE run, blue lines for the LowNUC run, and green lines for the NoNUC run.



20
 21
 22
 23
 24

1 Figure 27. Scatter plot of simulated CCN concentrations in comparison with a
 2 dataset of CCN measurements compiled by Spracklen et al. (2011). Data is classified
 3 into two categories: CCN in the MBL (in black) and CCN in the CBL (in red). The
 4 upper error bar is for the BASE run, the middle symbol for the LowNUC run, and the
 5 lower error bar for the NoNUC run. Large and filled symbols are for measurement
 6 duration longer than 10 days, and small and open symbols for less than 10 days.
 7 Circle symbols are for supersaturations (s) less than 0.2%; upward triangles for s
 8 greater than 0.2% and less than 0.8%; downward triangle for s greater than 0.8%.



9
 10 Figure 28. Annual cycle of CCN concentrations at Cape Grim (supersaturations (s) of
 11 1.2% in a, 0.75% in b, and 0.23% in c) and Mace Head (s of 0.5% in d). The measured
 12 CCN concentrations are shown in black, and the simulated CCN in red for the BASE
 13 run, blue for the LowNUC run, and green for the NoNUC run. Correlation (R) and
 14 normalized mean bias (NMB) are given.
 15

