Evaluation of the global aerosol microphysical ModelE2-TOMAS model 1 2 against satellite and ground-based observations

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12 **Abstract**

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14 The TwO-Moment Aerosol Sectional microphysics model (TOMAS) has been integrated 15 into the state-of-the-art general circulation model, GISS ModelE2. This paper provides a 16 detailed description of the ModelE2-TOMAS model and evaluates the model against 17 various observations including aerosol precursor gas concentrations, aerosol mass and 18 number concentrations, and aerosol optical depths. Additionally, global budgets in 19 ModelE2-TOMAS are compared with those of other global aerosol models, and the 20 ModelE2-TOMAS model is compared to the default aerosol model in ModelE2, which is 21 a One-Moment Aerosol (OMA) model (i.e., no aerosol microphysics). Overall, the 22 ModelE2-TOMAS predictions are within the range of other global aerosol model 23 predictions, and the model has a reasonable agreement (mostly within a factor of two) 24 with observations of sulphur species and other aerosol components as well as aerosol 25 optical depth. However, ModelE2-TOMAS (as well as ModelE2-OMA) cannot capture 26 the observed vertical distribution of sulphur dioxide over the Pacific Ocean possibly due 27 to overly strong convective transport and overpredicted precipitation. The ModelE2-28 TOMAS model simulates observed aerosol number concentrations and cloud 29 condensation nuclei concentrations roughly within a factor of two. Anthropogenic aerosol 30 burdens in ModelE2-OMA differ from ModelE2-TOMAS by a few percent to a factor of 31 2 regionally, mainly due to differences in aerosol processes including deposition, cloud 32 processing, and emission parameterizations. We observed larger differences for naturally 33 emitted aerosols such as sea salt and mineral dust, as those emission rates are quite 34 different due to different upper size cutoff assumptions.

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- 36 1. Introduction
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38 Aerosols perturb the energy balance of the Earth-atmosphere system by scattering 39 and absorbing solar and terrestrial radiation, known as the aerosol direct effect, and by 40 modifying cloud properties such as via acting as cloud condensation nuclei (CCN),

1 known as aerosol indirect effects (e.g. Lohmann and Feichter, 2005; Forster and 2 Ramaswamy, 2007). The recently published IPCC AR5 (Intergovernmental Panel on 3 Climate Change Fifth Assessment Report) refers to these as aerosol-radiation interactions 4 and aerosol-cloud interactions, respectively (Boucher et al., 2013). For light-absorbing 5 aerosols such as black carbon and mineral dust, the ambient air can be heated as a result 6 of their direct effect, affecting relative humidity and atmospheric stability, which is 7 known as the semi-direct effect. The largest uncertainty in estimating anthropogenic 8 climate forcing is from the aerosol indirect effects (Myhre et al., 2013b). Since it is not 9 easily estimated from observations due to natural variability in cloud properties and the 10 lack of observations of the pre-industrial atmosphere, estimates of aerosol indirect forcing 11 have been mainly based on general circulation models (GCMs). Thus, there have been 12 growing efforts to develop and improve aerosol microphysics models for a more 13 physically based representation of atmospheric aerosol number and CCN concentrations (e.g. Adams and Seinfeld, 2002; Easter et al., 2004; Vignati et al., 2004; Lauer et al., 14 15 2005; Liu et al., 2005; Spracklen et al., 2005; Stier et al., 2005; Bauer et al., 2008; 16 Trivitayanurak et al., 2008; Yu and Luo, 2009; Mann et al., 2010; Lee and Adams, 2012).

Aerosol microphysics models can be broadly categorized into modal and sectional methods, depending on how they represent the aerosol size distribution. In general, modal-based methods use an analytical function (e.g. a lognormal distribution) to represent a subset of the particle population. Sectional methods represent a size distribution by predicting aerosols in several size sections or "bins". Additionally, sectional and modal methods may differ from each other in numerous ways, including the number of moments of the size distributions that are tracked in each section or mode.

24 Sectional methods can be divided into single-moment sectional methods that 25 typically track either aerosol number or mass in each bin and two-moment sectional methods that explicitly track both aerosol number (i.e. 0th moment) and mass (i.e. 1st 26 mass moment or 3rd radial moment) in each size section. Unlike single-moment sectional 27 28 approaches, two-moment sectional methods can conserve both number and mass very 29 accurately (Tzivion et al., 1987; Feingold et al., 1988; Harrington and Kreidenweis, 1998; 30 Tzivion et al., 2001; Adams and Seinfeld, 2002; Jung et al., 2006) but have a high 31 computational burden. The modal approaches are generally more computationally

efficient but may not represent abrupt transitions in a size distribution well, which can
 occur during cloud processing (Zhang et al., 1999).

3 The TOMAS aerosol microphysics model (Adams and Seinfeld, 2002; Tzivion et 4 al., 1987; Tzivion et al., 1989) has been developed to study tropospheric aerosol 5 microphysics and predict cloud condensation nuclei (CCN) concentrations. The TOMAS 6 model has been previously implemented into the climate model of Goddard Institute for 7 Space Studies General Circulation Model II-prime (GISS GCM II-prime), referred to as 8 "GISS-TOMAS" (Lee and Adams, 2010). It has also been incorporated into GEOS-9 CHEM (Trivitayanurak et al. 2008), the regional model PMCAMx-UF (Jung et al., 10 2010), and the Large-Eddy Simulation model (Stevens et al., 2012; Singh et al., 2014). 11 The GISS GCM II-prime has horizontal grid dimensions of 4° latitude and 5° longitude, 12 with nine vertical sigma layers between the surface to the 10 hPa level (Hansen et al., 13 1983). Modules for each of the major aerosol species have been developed for the GISS 14 GCM II-prime, and the GISS-TOMAS model has been evaluated with ground-level 15 measurements such as number and mass concentrations, deposition fluxes, and remote 16 sensing observations (Adams and Seinfeld, 2002; Pierce and Adams, 2006; Pierce et al., 17 2007; Lee et al., 2009; Lee and Adams, 2010). Despite the accuracy in predicting aerosol 18 microphysical process in TOMAS, the original version of TOMAS has a heavy 19 computational burden. Lee and Adams (2012) developed less computationally expensive 20 configurations of the TOMAS model (Fast TOMAS), which are 2-3 times faster than the 21 original TOMAS model with only a few percent increases in microphysical errors. 22 However, a remaining weakness for the GISS-TOMAS model is the outdated host model, 23 the GISS GCM II-prime.

24 Here, we incorporate the TOMAS model into the new version of GISS GCM (i.e., 25 ModelE2), referred to as "ModelE2-TOMAS". ModelE2 now has three different aerosol 26 models available: TOMAS, the One-Moment Aerosol model (hereafter, referred to as 27 OMA) (e.g. Koch et al., 2006) that has no microphysics, and the modal-based aerosol 28 microphysics model, MATRIX (Multiconfiguration Aerosol TRacker of mIXing state) 29 (Bauer et al., 2008). The combination of several aerosol models into the same host GCM 30 allows ModelE2 to explore the uncertainties in predicting aerosol characteristics and their 31 climate effects that are associated with aerosol modelling (e.g. different numerical approaches) in the same host model. We also note that it was important to implement the
TOMAS aerosol model into the ModelE2 host model because uncertainties in the
estimates of aerosol forcing come not only from aerosol modelling itself but also other
parts of the host GCM (e.g., cloud physics, planetary boundary layer, and advection).

5 ModelE2-TOMAS has been used in several recent studies under the Atmospheric 6 Chemistry and Climate Model Intercomparison Project (ACCMIP) which aims to 7 understand composition changes and the associated radiative forcing between 1850 and 8 2100 (Bowman et al., 2013; Lamarque et al., 2013a; Lamarque et al., 2013b; Lee et al., 9 2013a; Nabat et al., 2013; Naik et al., 2013; Shindell et al., 2013; Stevenson et al., 2013; 10 Young et al., 2013). Here we give a detailed description of ModelE2-TOMAS and 11 evaluate against ModelE2-OMA (e.g., Schmidt et al., 2014) and observations of aerosol 12 mass and number as well as aerosol optical depth. Section 2 provides the description of 13 ModelE2-OMA, and Section 3, the description of the TOMAS aerosol microphysics 14 model. Section 4 explains the emissions and design of the simulations. Section 5 presents 15 global budgets of the simulated aerosols and the evaluation of the ModelE2-TOMAS and 16 ModelE2-OMA against observations of aerosol mass concentrations and aerosol optical 17 depth and the evaluation of the TOMAS number predictions against observations. 18 Conclusions follow in Section 6. We note that aerosol direct and indirect forcings using 19 ModelE2-TOMAS will be discussed in a separate paper.

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2. GISS GCM ModelE2

23 In this section, we briefly describe ModelE2 (Schmidt et al. (2014)), the GISS climate 24 model used to perform Coupled Model Intercomparison Phase 5 (CMIP5; Taylor et al, 25 2012). The model physics are mostly similar to GISS ModelE (CMIP3 version: Schmidt 26 et al., 2006). The model has 2° latitude by 2.5° longitude resolution, with 40 vertical 27 hybrid sigma layers from the surface to 0.1 hPa (80 km). Tracers, heat, and humidity are 28 advected using the highly nondiffusive Quadratic Upstream Scheme (Prather, 1986). The 29 radiation scheme accounts for size-dependent scattering properties of clouds and aerosols 30 based on Mie scattering (Hansen et al., 1983) and non-spherical light scattering of cirrus 31 and dust particles based on T-matrix theory (Mishchenko et al., 1996). It also includes the 32 impact of water uptake by hygroscopic species on their radiative properties. In the model, clouds are distinguished into convective and large-scale stratiform clouds. The clouds
parameterizations are similar to Del Genio (1993) and Del Genio et al. (1996) but have
been improved in several respects (see details in Schmidt et al., 2006; Schmidt et al.,
2014). The physics timestep is 30 minutes, and the radiation is calculated every 2.5 hours.

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2.1 ModelE2-OMA description

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7 ModelE2 includes a default aerosol module, OMA (One-Moment Aerosol), which has 8 no microphysics. ModelE2-OMA has sulphate (Koch et al., 2006; Koch et al., 2007; 9 Koch et al., 2011), carbonaceous aerosols (Koch et al., 2007), secondary organic aerosols 10 (Tsigaridis and Kanakidou, 2007), sea salt (Koch et al., 2006; Tsigaridis et al., 2013), 11 dust (Miller et al., 2006), and nitrate (Bauer et al., 2007). Along with sulphate, the model 12 also predicts sulphur dioxide, dimethyl sulfide (DMS) and methanesulfonic acid (MSA) 13 (Koch et al., 2006). The secondary organic aerosol formation is computed using a two-14 product model with isoprene, monoterpenes, and sesquiterpenes as SOA precursors 15 (described in Tsigaridis and Kanakidou, 2007). Sea salt particles have two size classes 16 with a fine mode (0.1 μ m to 1 μ m in dry radii) and a coarse mode (1 μ m to 4 μ m in dry 17 radii). Dust particles have four size classes with radii between 0.1-1 µm (clay), 1-2 µm 18 (silt1), 2-4 µm (silt2), and 4-8 µm (silt3). The model accounts for heterogeneous 19 chemistry on mineral dust particle surfaces to form nitrate and sulphate (Bauer and Koch, 20 2005).

21 In ModelE2, the surface boundary conditions are defined using dry deposition and 22 interactive surface sources (Koch et al., 2006). The dry deposition scheme is tightly 23 coupled to the model boundary layer process and is based on a resistance-in-series 24 scheme derived from the Harvard GISS-CTM, which is applied between the surface layer 25 (10 m) and the ground (Koch et al., 2006). Wet deposition is determined by several 26 processes including rainout within clouds, washout below precipitating regions, 27 scavenging within and below cloud updrafts, evaporation of falling precipitation, 28 transport along with convective plumes, and detrainment and evaporation from 29 convective plumes (Koch et al., 2006; Shindell et al., 2006). ModelE2 includes a 30 dissolved species budget scheme for stratiform clouds, which has an impact on sulphate formation via aqueous oxidations, since some sulphate formed in clouds undergoes wet
 scavenging instead of being added back to the sulphate in air (Koch et al., 2006).

Tropospheric/stratospheric chemistry in ModelE2 includes 156 chemical reactions among 51 gas species (Shindell et al., 2013). In ModelE2, chemistry and aerosols are fully interactive, so that the oxidation fields used for sulphate formation are from the chemistry model (not prescribed) and the photolysis rates are affected by light attenuation by aerosols (Shindell et al., 2013). Photolysis rates are computed using the Fast-J2 scheme (Bian and Prather, 2002). Aerosol indirect effects are based on an empirical parameterization that compute cloud droplet number concentrations as a function of aerosol mass (Menon et al., 2002; Menon et al., 2008).

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3. ModelE2-TOMAS description

14 The TOMAS aerosol microphysics model uses a sectional approach that represents 15 the aerosol size distribution by predicting the amount of aerosol in several size categories 16 or "bins". TOMAS tracks two moments of the aerosol size distribution in each size bin: total aerosol number (i.e., 0th moment) and mass (i.e., 1st mass moment). Total mass is 17 decomposed into several aerosol species, allowing prediction of the size-resolved aerosol 18 19 composition. In total, ten quantities are tracked for each size bin: sulphate mass, sea-salt 20 mass, mass of pure (hydrophobic) elemental carbon (EC), mass of mixed (aged) EC, 21 mass of hydrophobic organic matter (OM), mass of hydrophilic OM, mass of mineral 22 dust, mass of ammonium, mass of water and the number of aerosol particles in that bin. 23 In TOMAS, all ammonia becomes aerosol ammonium until sulfate is neutralized to form 24 ammonium sulfate; the excess ammonia after neutralization remains as free gas-phase 25 ammonia. The aerosol ammonium is partitioned into each size bin in proportion to the 26 sulfate mass. However, ammonium is not size-resolved (i.e., bulk tracer) for purposes of 27 model processes outside of TOMAS such as advection and deposition. In addition, the 28 model tracks four bulk gas-phase species: sulphur dioxide (SO₂), dimethylsulfide (DMS), 29 sulphuric acid (H₂SO₄), and a lumped gas-phase tracer that represents oxidized organic 30 vapors forming secondary organic aerosol (SOA). The gas-phase H₂SO₄ is assumed to be 31 pseudo-steady state equilibrium between its chemical production in and 32 condensational/nucleation losses (Pierce and Adams, 2009a). Water uptake by sulphate

1 and sea salt is based on a polynomial fit based on ISORROPIA, a thermodynamic 2 equilibrium model for inorganic aerosols (Nenes et al., 1998). For water uptake by 3 hydrophilic OM, it is based on the observations of Dick et al. (2000). The size section 4 boundary is defined by dry particle mass, such that addition or removal of aerosol water 5 mass does not move particles between sections. In general, TOMAS treats all aerosols as 6 internally mixed during microphysics such as calculating condensation and coagulation 7 rates. However, a portion of EC is treated as externally mixed for purpose of wet 8 deposition. Detailed descriptions of the TOMAS microphysics scheme can be found in 9 Adams and Seinfeld (2002), Lee and Adams (2012), and Lee et al. (2013b).

10 Several alternative nucleation schemes are available in TOMAS, including binary 11 nucleation (Vehkamaki et al., 2002), ternary nucleation (Napari et al., 2002), ion-induced nucleation (Modgil et al., 2005), and activation nucleation with an A factor of $2x10^{-6}$ s⁻¹ 12 13 (Sihto et al., 2006) for the boundary layer (~up to 900 mbar). For the simulations used in 14 this paper, only binary nucleation is used. The boundary-layer nucleation is off in all 15 simulations because it tends to overpredict aerosol number concentrations in our model. 16 Also we do not show any run with the ternary nucleation (Napari et al., 2002) because it 17 overpredicts aerosol number concentration severely (not shown).

18 With Fast TOMAS models, the TOMAS microphysics module became more flexible 19 in term of varying particle size resolution, i.e. the number of size bins (Lee and Adams, 20 2012). For the size range of 10 nm to 10 µm, the original TOMAS uses 30 bins (size 21 boundary is defined with mass doubling), and the Fast TOMAS uses 15 bins or 12 bins 22 (size boundary is defined with mass quadrupling). As discussed in Lee and Adams 23 (2012), the Fast TOMAS reduces the computational burden by 2-3 times while generally 24 predicts CCN concentrations within a few percent of the original TOMAS. The lower 25 size cutoff in TOMAS can also vary from 10 nm to 3 nm or from 10 nm to 1 nm (Lee et 26 al., 2013b). Among several possible configurations, ModelE2-TOMAS currently uses 27 either 12 bins covering 10 nm to 10 µm or 15 bins covering 3 nm to 10 µm, which is the 28 most computationally efficient version of TOMAS for the given size range. In this paper, 29 we used TOMAS with 15 bins covering 3 nm to 10 µm (TOMAS15; see Table S1 in the 30 supplementary materials): 3 bins cover from 3nm to 10 nm, 10 bins from 10 nm to 1 µm 31 and the last two bins from 1 µm to 10 µm. The TOMAS15 version becomes a default model configuration for ModelE2-TOMAS, so we will be continuously refereed to as
ModelE2-TOMAS throughout this paper. More configurations will be available in the
near future.

4 The wet deposition scheme in ModelE2-TOMAS is identical to the one used in 5 ModelE2-OMA except for the following. First, ModelE2-TOMAS adds sulphate mass 6 produced in the aqueous phase directly to the bin-resolved sulphate mass in ambient air 7 rather than maintaining a separate tracer for dissolved sulphate. Compared to ModelE2-8 OMA, this is a simplification because the sulphate formed in the aqueous phase will be 9 only released as interstitial aerosol when the cloud water evaporates. It is adopted here 10 for simplicity but will be improved in the future. The other difference is that the wet/dry deposition in ModelE2-TOMAS accounts for particle size dependence. For in-cloud 11 12 scavenging, modified Köhler theory is used to obtain the critical supersaturation for 13 activation of each size section and to determine which particles activate and are subject to in-cloud (nucleation) scavenging (Pierce et al., 2007). To determine activation, we 14 15 assume kappa values of 0.7 for sulfate, 1.3 for sea-salt, and 0.15 for hydrophilic OM. The 16 fraction of activated aerosols removed by wet deposition is proportional to the fraction of 17 cloud water that precipitates, which is computed in each model layer. Wet deposition 18 accounts for re-evaporation of precipitation. For in-cloud scavenging, the large-scale and 19 convective clouds in the model are assumed to have a supersaturation of 0.2%; unlike 20 GISS-TOMAS that used a supersaturation of 1.0% for convective clouds, a 21 supersaturation of 0.2% is assumed in ModelE2-TOMAS in order to capture the observed 22 Hoppel gap (~ 100 nm) in the marine boundary layer. Note that the activation described 23 here to determine in-cloud scavenging is not used for computing cloud droplet number 24 concentrations (see below). For below-cloud scavenging, a first-order removal scheme 25 implemented for bulk aerosols by Koch et al. (1999) is modified for size-resolved 26 aerosols (Adams and Seinfeld, 2002). Dry deposition is identical to the existing 27 resistance-in-series scheme in ModelE2, but ModelE2-TOMAS treats a size-dependent 28 gravitational settling of particles and a size-dependent resistance in the quasi-laminar 29 sublayer (Seinfeld and Pandis, 1998; Adams and Seinfeld, 2002).

To compute the cloud microphysics properties as a function of aerosols (i.e., the aerosol-cloud interactions), ModelE2-TOMAS uses a physical-based activation parameterization from Nenes and Seinfeld (2002). A critical supersaturation is computed
 in the parameterization using a model updraft velocity that is computed based on a large scale vertical velocity and sub-grid velocity.

In ModelE2-TOMAS, Mie theory is used to compute size-resolved AOD. For each grid cell, particle compositions (including aerosol-water) in each individual size bin are used to compute the volume-averaged refractive index and optical properties based on Mie theory. The optical properties are used to compute aerosol optical depth taking into account the aerosol concentration.

9 ModelE2-TOMAS is coupled to the same gas chemistry model (Shindell et al., 2013) 10 as ModelE2-OMA. So the oxidation fields used for sulphate formation are from the 11 chemistry model. However, unlike ModelE2-OMA, the photolysis rates are not affected 12 by aerosols.

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14 4. Description of the simulations

4.1 Emissions

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17 The emissions used in this study are summarized in Table 1. The simulations used year-18 2000 emissions from the anthropogenic emissions inventory created for CMIP5 19 (Lamarque et al., 2012) and climatologically averaged biomass burning emissions from 20 GFED3 for 1997 to 2009 (van der Werf et al., 2010). For SO₂, in addition to the 21 anthropogenic emissions, continuous volcanic emissions from GEIA (Global Emissions 22 InitiAtive; Andres and Kasgnoc, 1998) are used but increased by a factor of 1.5 as in the 23 AEROCOM intercomparison emissions in Dentener et al. (2006). Sea-salt emission is 24 based on Gong (2003), which extends the lower size limit of the Monahan et al. (1986) 25 emission from 0.4 µm to 0.02 µm. Dust emission is based on the source distribution from 26 Ginoux et al. (2001) and is proportional to the third power of the wind speed (at 10 m in 27 height) above a threshold that is a function of soil moisture. Subgrid-variation of the wind 28 speed in a GCM grid box, which is created by boundary-layer turbulence and dry/wet 29 convection, is accounted for the modeled dust emissions (Cakmur et al., 2006; Miller et 30 al., 2006). DMS emission is based on the seawater DMS concentrations of Kettle et al.

(1999). For the sea-to-air transfer function used in the DMS emissions, The ModelE2TOMAS runs are based on Liss and Merlivat (1986), and the ModelE2-OMA model run
is based on Nightingale et al. (2000).

4 Nightingale et al. (2000) provides a revised parameterization based on observations of 5 the sea-to-air transfer rate scatter between two classical parameterizations (i.e., Liss and 6 Merlivat, 1986; Wanninkhof, 1992), and it has been more favoured in many global DMS 7 models than the two classical parameterizations. However, DMS emission is quite uncertain. Estimates of the global DMS emissions range from 16 to 54 Tg S yr⁻¹ (Kettle 8 9 and Andreae, 2000), depending on the choice of DMS sea surface climatology, sea-to-air 10 transfer rate parameterization, and wind speed data. DMS emission rates from ModelE2-11 TOMAS and ModelE2-OMA are within the range. The Liss and Merlivat (1986) 12 parameterization is used in ModelE2-TOMAS, because a ModelE2-TOMAS run based 13 on Nightingale et al. (2000) overpredicts the SO₂ concentrations over remote oceanic 14 regions especially in the southern hemisphere. Koch et al. (2006) showed that the sea-to-15 air transfer function from Nightingale et al. (2000) increased annual DMS emissions by 16 roughly a factor of two compared to the emission based on Liss and Merlivat (1986). This 17 was desirable in ModelE2-OMA because of the underprediction of sulphate in remote 18 oceanic regions in that model, although the model DMS and MSA (oxidized from DMS) 19 tended to be excessive in SH oceanic regions especially near Antarctica. However, 20 despite the higher DMS emissions, it turned out that the sulphate was still underpredicted 21 because sulphate formed by aqueous oxidation was subject to wet scavenging before 22 releasing to the ambient air as a result of the updated dissolved species budget scheme 23 (Koch et al., 2006).

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4.2 ModelE2-TOMAS run setup

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We performed the simulations nudged with winds from the MERRA (Modern Era Retrospective-analysis for Research and Applications; Rienecker et al., 2011) reanalysis meteorological fields from 2000 to 2003 with 3 years spin-up (i.e., 1997-1999). Primary emission of particulate sulphate is assumed to be 1.0 percent of total sulphur emissions. Emissions size distributions assumed for ModelE2-TOMAS are summarized in Table 2. Primary sulphate emission is assumed to have a bi-modal lognormal distribution that

1 assigns 5% of the primary sulphate emissions as a nucleation mode with a geometric 2 number mean diameter (GMD) of 10 nm and a geometric standard deviation (GSD) of 3 1.6 and the rest as an Aitken mode with GMD of 70 nm and GSD of 2. For fossil fuel and 4 biofuel emissions, the size of primary carbonaceous aerosol emissions is assumed to fit a 5 lognormal size distribution with a GMD of 60 nm and a GSD of 1.59 for both EC and 6 OM (Stier et al., 2005). For carbonaceous aerosols of biomass burning emissions, a 7 lognormal size distribution is assumed to have a GMD of 150 nm and a GSD of 1.59. 8 Note that although the emission size distribution for biofuel emissions is generally 9 assumed to be the same as that for biomass burning emissions (e.g. Dentener et al., 2006), 10 in ModelE2-TOMAS run we assumes the biofuel emission size distributions follow the 11 finer fossil fuel size settings because the CMIP5 emissions does not provide a separate 12 category for biofuel emissions (e.g. biofuel used for cooking and heating are assigned as 13 the residential sector, which also includes fossil fuel usage). The OC (organic carbon): 14 OM (organic matter) ratio is assumed to be 1:1.4.

15 ModelE2-TOMAS assumes larger particles for primary sulphate and carbonaceous 16 aerosols than GISS-TOMAS (e.g., Lee et al., 2013b) to capture the observed aerosol 17 number concentrations better. This is very likely due to the following: 1) GISS-TOMAS 18 attributed the primary sulphate emissions only to anthropogenic sulphur emissions 19 (excluding biomass burning emissions), while ModelE2-TOMAS attributes these to all 20 sulphur emissions; 2) GISS-TOMAS applied the biomass-burning emission size 21 distributions of carbonaceous aerosols to the biofuel emissions, which is coarser than the 22 fossil fuel emission size distribution. Note also that the emission size distributions used 23 for biomass burning and volcanic emissions are finer than the AEROCOM 24 recommendations in Dentener et al. (2006). However, the model number concentrations 25 and size distributions are changed little when applying the AEROCOM recommended 26 emission distributions (not shown). Note that the biomass burning and volcanic emissions for sulphur are 1.4 Tg S yr⁻¹ and 12.5 Tg S yr⁻¹, respectively. 27

Following the soil size assumptions used in GISS-TOMAS (Lee et al., 2009), the clay distribution is assumed to have a GMD of 0.14 μ m and a GSD of 2, and the silt distribution, a GMD of 1.15 μ m and a GSD of 2. Using this distribution, fifteen percent

of the silt emissions flux falls out of the upper size cutoff (i.e., 10 µm), and is therefore 1 2 not received by any of the TOMAS size bins.

3 Compared to the run setup described above (hereafter, referred to as the "BASE" 4 run), we additionally ran two other sensitivity runs with the ModelE2-TOMAS model by 5 perturbing the nucleation process to evaluate changes in number concentrations (Table 1). 6 The first sensitivity run is called "NoNUC", in which we turned off nucleation to 7 estimate the contribution of primary emissions to aerosol number concentrations. The 8 other run is called "LowNUC", in which we reduced the nucleation rate by using 5 times 9 lower sulphuric acid concentrations to compute nucleation rates. Note that sulphuric acid 10 concentrations are not perturbed in other processes, and the model sulphuric acid budget 11 is little influenced by this treatment.

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4.3 ModelE2-OMA run setup

14 To compare to the ModelE2-TOMAS run, we also ran the ModelE2-OMA model 15 nudged to the same MERRA reanalysis meteorology with 3 years spin-up. However, the 16 natural emissions and associated settings are not always the same between the two 17 models because we chose to maintain the natural emissions/setup used in ModelE2-18 OMA, which has been chosen carefully in previous studies. To assist the interpretation 19 of the results, we briefly summarize the differences between ModelE2-OMA and 20 ModelE2-TOMAS. First, as mentioned in Section 4.1, the ModelE2-OMA model uses 21 the sea-to-air transfer function of Nightingale et al. (2000) instead of Liss and Merlivat 22 (1986), because Koch et al. (2006) prefers the newer DMS emissions to improve a 23 sulphate prediction in the remote marine locations. Second, the same sea salt and dust 24 emission schemes are applied in both aerosol models, but different assumptions for the 25 upper limit of particle size are used: 8 µm in diameter for sea salt and 16 µm in diameter 26 for dust in the ModelE2-OMA model; 10 µm in ModelE2-TOMAS. Third, the ModelE2-27 OMA model assumes 2.5% of the total sulphur as primary sulphate as followed by the 28 AEROCOM study (Dentener et al., 2006) whereas ModelE2-TOMAS assumes only 1%. 29 Aerosol number predictions are sensitive to the primary sulphate assumption, but 30 sulphate mass concentrations are not. When using the 2.5% assumption in ModelE2-31 TOMAS, we found that the simulated aerosol number concentrations were biased high,

and the model size distribution predictions wer also poor. Note that Pierce and Adams
 (2009b) shows that GISS-TOMAS also overpredicts aerosol number concentration with
 the 2.5% assumptions.

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5. Model results and evaluations

7 In this section, we present global-annual budgets, spatial distributions, and 8 evaluations of the model aerosol precursor gases (in Sect. 5.1), aerosol mass (in Sect. 5.2 9 and 5.3), aerosol optical depths (AODs; in Sect. 5.4) and aerosol number (in Sect. 5.5 and 10 5.6). The observations used for model evaluations are from surface-based, aircraft-based 11 and remote-sensing measurements. More details of the observations are provided in each 12 subsection. To compare with the ModelE2-TOMAS results, we included the ModelE2-13 OMA results in global-annual budgets and model evaluations. Only the BASE run results 14 are used in Section 5.1 to 5.4 because the predicted aerosol precursor gases 15 concentrations, aerosol mass concentrations, and AODs from the nucleation sensitivity 16 runs are quite similar to the BASE run.

Model skill is quantified in terms of log-mean normalized bias (LMNB) and logmean normalized error (LMNE) when evaluating with annual-mean concentrations measurements and normalized mean bias (NMB) and correlation coefficient (R) when evaluating with an observed annual cycle and aerosol optical depth (both monthly and annually averaged AODs).

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5.1 Aerosol precursor gases

24 Global budgets of DMS and SO₂ in ModelE2-TOMAS are presented in Table 3 with 25 a range obtained from several global models including Wang et al. (2011), Liu et al. 26 (2005), and those listed in Liu et al. (2005). The DMS and SO₂ budgets in ModelE2-27 TOMAS are within the ranges of the other global models. In case of ModelE2-OMA (in 28 Table 3), the global burden of DMS is about a factor of two higher than the ModelE2-29 TOMAS model because the DMS emission rate is \sim 78% higher by using the sea-to-air 30 transfer functions by Nightingale et al. (2000). Despite the different DMS emissions and 31 SO_2 emissions (due to the primary sulphate emission assumption, 1% versus 2.5%), the global burden of SO₂ is quite similar to that in the ModelE2-TOMAS model. The dominant SO₂ removal processes are aqueous oxidation and dry deposition in both simulations. Boucher et al. (2003) simulates atmospheric DMS in the LMD-ZT model using the same DMS emission schemes as ModelE2-OMA (i.e., EXP1 in their study) and ModelE2-TOMAS (i.e., EXP4 in their study). The global DMS budgets from ModelE2-OMA and ModelE2-TOMAS agree quite well with the EXP1 (within 25 %) and EXP4 (within 15%).

8 When using the same DMS emissions in ModelE2-TOMAS as in ModelE2-OMA, the 9 DMS global budgets are almost identical, but the SO₂ budgets vary substantially due to 10 the differences in SO_2 modelling, i.e., heterogeneous SO_2 oxidation and photolysis (see 11 Section 2). The heterogeneous sulphur dioxide oxidation on dust aerosol surfaces, which 12 is only included in ModelE2-OMA, accounts for 25% of the total gas-phase oxidation 13 loss. Based on Bauer and Koch (2005), including the heterogeneous chemistry, global SO₂ burden can decrease 32%, and global sulphate burden can increase 3%. The 14 15 simulated photolysis rates in ModelE2-OMA are affected by aerosol optical depth, 16 affecting hydroxide (OH) and other gas tracer concentrations - Naik et al. (2013) show a 17 higher OH concentration in ModelE2-TOMAS than ModelE2-OMA. Overall, using the 18 same DMS emissions in ModelE2-TOMAS results in a higher SO₂ burden and worse 19 agreement for SO₂ and sulphate concentrations over remote oceanic regions (not shown).

20 The global budgets of H₂SO₄ and SOA precursor gas in the ModelE2-TOMAS model 21 are not included in Table 3 but are summarized here. The simulated H₂SO₄ has a total production rate of 12.3 Tg S yr⁻¹, matching the SO₂ gas-phase oxidation, and is used in 22 aerosol microphysics (i.e. 12 Tg yr⁻¹ for condensation and 0.3 Tg yr⁻¹ for nucleation). 23 The model SOA precursor gas has a total production rate of 17.1 Tg yr⁻¹, assumed to be 24 25 10% of the terpene emission, and is condensed as hydrophilic OM. For ModelE2-OMA, the total production rate of SOA is 14.6 Tg yr⁻¹. This is quite comparable to ModelE2-26 TOMAS, which treats SOA much more simply and has a production rate of 17.1 Tg yr⁻¹. 27 Global burden of SOA in ModelE2-OMA is 0.6 Tg yr⁻¹. 28

Figure 1 shows global maps of annual-mean DMS and SO₂ column mass concentrations. The spatial distribution of DMS concentrations shown in Fig. 1a is driven by its emission and interactive OH and NO₃ concentrations, which oxidize DMS to form 1 MSA and SO₂. The model DMS concentrations are most pronounced in the Southern 2 Ocean and the Northern Atlantic oceans due to high seawater DMS concentrations during 3 summer. The simulated SO₂ concentration shown in Fig. 1b is very high over industrial 4 regions due to the anthropogenic emissions and is also high over the Southern Ocean due 5 to DMS oxidation. Several local hotspots of SO₂ shown in Fig. 1b are due to volcanic 6 emissions.

7 Annually averaged surface-layer SO₂ concentrations from both ModelE2-TOMAS 8 and ModelE2-OMA are evaluated against observations from the EMEP (European 9 Monitoring and Evaluation Programme, http://www.emep.int) and CASTNET (Clean Air 10 Status and Trends Network, http://epa.gov/castnet/javaweb/index.html) networks (see 11 Fig. 2). We used 2000-2004 mean SO₂ measurements for the EMEP network and 1995-12 2005 mean SO₂ data for the CASTNET network. Performance of ModelE2-TOMAS and ModelE2-OMA for predicted SO₂ concentrations in these locations is almost the same 13 14 (i.e. LMNB=0.25-0.26 and LMNE=0.34 for the EMEP network; LMNB=0.09 and 15 LMNE=0.29 for the CASTNET network), and the continental SO₂ predictions agree with 16 the observation on average roughly within a factor of two. The two aerosol models are 17 almost the same because the anthropogenic emissions, which are identical in both 18 models, are dominant at these locations.

19 Figure 3 compares surface-layer SO₂ and DMS mixing ratios from the two aerosol 20 models against observations at three Southern Hemisphere remote sites: Amsterdam Island (DMS from Sciare et al., 2000; SO₂ from Nguyen et al., 1992), Cape Grim (Ayers 21 22 et al., 1995), and Dumont (Jourdain and Legrand, 2001). Note that Dumont has only 23 DMS measurements. For DMS, both models capture the observed seasonal cycle (i.e. 24 R>0.8) generally well but are less successful over Dumont. The DMS concentrations 25 seem to agree well against the observations when using the sea-air transfer function of 26 Liss and Merlivat (1986), i.e., the case for ModelE2-TOMAS, but this run underpredicts 27 during winter season at the Amsterdam Island site (in Fig. 3a) and during all season at 28 Dumont site (in Fig. 3c). Earlier, we mentioned that the global DMS budgets from 29 Boucher et al. (2003) agree well with those from ModelE2 when using the same DMS 30 emission parameterization. However, Boucher et al. (2003) shows the better agreement to 31 the same DMS measurements when using the sea-air transfer function of Nightingale et

al. (2000) at Amsterdam Island and Cape Grim (i.e., the case for ModeleE2-OMA),
because their DMS mixing ratios from that simulations are actually more close to
ModelE2-TOMAS. This may suggest that, at least over SH high latitude regions, the
surface wind speed in ModelE2 is much stronger than that in LMD-ZT, resulting in
higher DMS emissions and burden. We need to investigate further to find out a source for
the difference though.

7 The ModelE2-TOMAS SO₂ shows very good agreement at Amsterdam Island but a 8 high bias at Cape Grim (see Figs. 3d and e). We considered whether the overpredicted 9 SO_2 in the model might be influenced by the emissions in the adjacent grids, which is 10 mentioned in Mann et al. (2010), because the measured SO₂ at Cape Grim is filtered to 11 include the marine sector only (Ayers et al., 1991). Sampling the model SO₂ from 12 adjacent grids toward marine areas, the overprediction is reduced significantly (LMB is 13 reduced from 10 to 3) but is still severe. The most plausible reason for the overprediction 14 of SO₂ at Cape Grim might be the lack of SO₂ oxidation by ozone on sea salt particles, 15 which is missing in our model. Korhonen et al. (2008) shows a reduction of SO₂ 16 concentrations by a factor of 5 in January and a factor of 20 in July at Cape Grim when 17 including SO₂ oxidation on sea spray particles, although their treatment of the reactions 18 might overestimate the SO₂ oxidation rates.

19 Simulated DMS and SO₂ vertical profiles over the Pacific Ocean are compared 20 against two sets of aircraft observations in Figures 4 and 5: PEM-Tropics-A performed 21 during August-October, 1996 (Hoell et al., 1999) and PEM-Tropics-B during March-22 April, 1999 (Raper et al., 2001). Note that PEM-Tropics-A DC8 aircraft data is used and 23 most of them were during September 1996, but model evaluation is little changed by 24 comparing with model outputs in September or August-October average. The observed 25 vertical profile data used here are binned into altitude ranges (Emmons et al., 2000). 26 Model outputs are averaged over the observational time period and domain. Simulated 27 DMS vertical profiles are very similar between the two aerosol models, although the 28 surface DMS is different as their emissions are not the same. Both models show good agreement with the observations (mostly within 25th and 75th percentile of observed 29 30 values), especially capturing a strong concentration decrease from the surface to the free 31 troposphere.

1 In the case of SO_2 , even though the agreement is not as good as that seen for DMS, 2 both aerosol models seem to capture the observed magnitude approximately within a 3 factor of two (see Fig. 5). The overall vertical patterns shown in the models are 4 frequently not in agreement with the observations. Except at Hawaii, our model do not 5 capture the enhanced SO₂ concentrations in the boundary layer shown in the observation, 6 even though the model DMS is quite well captured. The poor prediction of SO₂ vertical 7 profile might be due to 1) too much precipitation near tropics in ModelE2 (see Fig. 9 in 8 Schmidt et al., 2014) and 2) too strong vertical transport (e.g., via deep convection over 9 the tropical Pacific Ocean) in the model. The latter can be supported by that the small 10 DMS peak at 8 km and the elevated SO_2 in the upper/free troposphere in the models (see Figs. 4 and 5),. Although the elevated SO₂ mixing ratios might be due to too weak wet 11 12 scavenging (including aqueous chemistry) of SO₂, we did not see any noticeable improvement when increasing SO₂ Henry's law constant by a factor of two in the model 13 14 (not shown). A large peak in the mid-troposphere at Hawaii in the models results from 15 volcanic SO₂ emissions, while the observations show a similar peak only during March-16 April 1999, which is heavily influenced by volcanic emissions (Thornton et al., 1999). 17 During August-October 1996, the observations at Tahiti and Easter Island show transport 18 of volcanic SO₂ emissions in the middle and upper troposphere (Thorton et al., 1999), 19 which does not capture in the model. Since our model includes only continuous volcanic 20 emission with a yearly resolution, our model fails to simulate variability in volcanic SO₂ 21 emissions at higher time resolution.

22 23

5.2 Aerosol mass budgets and distributions

24 Globally and annually averaged budgets of aerosols in the ModelE2-TOMAS model 25 are shown in Table 4. For the sulphate and EC budgets, we use the ACCMIP multi-model 26 mean from Shindell et al. (2013) and Lee et al. (2013a), which is based on 8 ACCMIP 27 models using the same AR5 emission scenario, Note that the biomass burning emission in 28 this study is GFEDv3 inventory averaged from 1997 to 2009, while the ACCMIP models 29 use GFEDv2 inventory for 2000. We do not compare with the AEROCOM phase 2 30 multi-model mean presented in Myhre et al. (2013a) because the aerosol budgets in 31 Myhre et al. (2013) are for anthropogenic aerosols, which is defined as the difference between the present-day run and pre-industrial run. For the lifetime and deposition rate
 coefficient budgets, we compare with the AEROCOM Phase 1 multi-model mean
 presented in Textor et al. (2006) –hereafter, referred to as AEROCOM Phase 1.

4 For sulphate, the ModelE2-TOMAS total source rate is lower than the ACCMIP mean (43.7 Tg yr⁻¹ vs. 51.7 Tg S yr⁻¹), and the global burden is the same as the ACCMIP 5 mean burden (0.67 Tg S) due to the slightly longer lifetime in the ModelE2-TOMAS 6 7 model (5.7 days vs. 5.0 days). Note that the GISS-E2-R-TOMAS model used for 8 ACCMIP is almost identical to ModelE2-TOMAS evaluated here except for the sulfate 9 modeling. The sulphate and DMS emissions used in GISS-E2-R-TOMAS are identical to 10 those used in ModelE2-OMA in this paper. For dry deposition coefficient (the inverse of 11 the lifetime), ModelE2-TOMAS has a particularly small value. However, the longer 12 overall sulphate lifetime is contributed by both dry and wet deposition, rather than dry 13 deposition. When increasing the dry deposition coefficient to the AEROCOM Phase 1 14 mean alone, the overall lifetime is decreased from 5.6 days to 4.8 days. Doing the same 15 for wet deposition with no change in the dry deposition, the overall lifetime decreases 16 from 5.6 days to 4.5 days.. Wet deposition accounts for 98% of the total deposition in 17 ModelE2-TOMAS, which is much higher than AEROCOM Phase 1, and convective 18 clouds contribute 27% of the wet deposition.

19 Global-annual EC in ModelE2-TOMAS is 0.19 Tg, which is very comparable to the 20 ACCMIP mean (0.16 Tg). Similar to sulphate, wet deposition contributes >95% of total 21 deposition of EC, which is higher than other ACCMIP models (see Table 3 in Lee et al., 22 2013a), and 24% of wet deposition is by convective clouds. For EC and OM, their 23 lifetimes are longer than the ACCMIP mean as well as AEROCOM Phase 1 means but 24 still within the standard deviation. Despite smaller dry deposition coefficients for EC and 25 OM compared to the AEROCOM mean, their wet deposition coefficients are quite 26 comparable to the AEROCOM mean.

The global annual burden of sea salt and dust in ModelE2-TOMAS are 3.6 Tg and 9.1 Tg, respectively. For sea salt and dust, dry deposition is as important as wet deposition due to their large particle sizes, accounting for 68% and 52% of total deposition, respectively. Since the size coverage of sea salt and dust in our model do not necessarily match with those in the AEROCOM Phase 1 models, we do not compare the lifetime and removal rate coefficients, which is strongly influenced by the upper size cutoff used in
 their emissions.

3 Despite the same host model and the same anthropogenic emission scenarios as the 4 ModelE2-TOMAS model, the ModelE2-OMA model shows significantly different 5 aerosol mass budgets (in Table 5), which must arise from using different deposition 6 assumptions and other aerosol modelling treatments (see Section 2 for the details). 7 Sulphate burden and lifetime in ModelE2-OMA is roughly half of that in ModelE2-8 TOMAS. The total source rate of SO₄ is about 20% higher than ModelE2-TOMAS and is 9 close to the ACCMIP mean value. The ModelE2-OMA model has a shorter lifetime for 10 EC, leading to ~40% lower burden compared to ModelE2-TOMAS. The OM burden is quite similar between two models. SOA formation rate is slightly different (14.6 Tg yr⁻¹ 11 for ModelE2-OMA and 17.1 Tg yr⁻¹ for ModelE2-TOMAS), but the difference is only a 12 13 small portion (about 3-4%) of the total OM source rate. The sea-salt emission rate is 14 lower than that in ModelE2-TOMAS due to the maximum size cutoff of 8 µm assumed in 15 sea-salt emission, but its burden is more than a factor of two higher. For dust particles, 16 the emission rate is higher than that in ModelE2-TOMAS due to the coarser size cutoff in 17 their emissions (i.e. up to 16 µm), and the burden is higher. For sea salt, the contribution 18 of wet deposition to total deposition is more than 2 times higher than that in ModelE2-19 TOMAS. Unlike ModelE2-TOMAS, ModelE2-OMA has nitrate aerosol, which has a 20 global burden of 1.6 Tg with a lifetime of 6.4 days.

21 The removal rate coefficient of dry deposition is about 50-60 times higher for 22 sulphate, EC and OM in ModelE2-OMA, making it more comparable to the AEROCOM 23 mean values. The lower dry deposition rates with the ModelE2-TOMAS model is likely 24 due to the size-dependent dry deposition parameterization. The dry deposition velocity is 25 not saved in the ModelE2 output currently, so alternatively we refer to the Figure 1 in 26 Adams and Seinfeld (2001) that presents the global and annual-average of size-resolved 27 dry deposition velocities in GISS-TOMAS (sulphate alone) compared to the size-28 independent one. Although the dry deposition velocities from ModelE2-TOMAS might 29 not be exactly the same as the ones in the model version used in Adams and Seinfeld 30 (2001) due to the updates made in ModelE2 (e.g. the boundary layer module), this point 31 should be valid because the dry deposition parameterizations in both models have been little changed. Despite the large differences in dry deposition rates for accumulation
 mode particles, dry deposition is a fairly minor removal pathway in both models.

3 Figure 6 shows simulated global distributions of annual-mean concentrations of 4 ModelE2-TOMAS sulphate, EC, OM, sea salt and dust in the lowermost layer. The 5 sulphate concentrations are high over industrial regions, driven by the SO₂ emissions and OH/H₂O₂ oxidant concentrations. Simulated EC and OM concentrations are high over the 6 7 biomass burning regions and the industrial regions, especially East Asia and South Asia, 8 but the OM concentrations are particularly pronounced over biomass burning regions due 9 to their higher emissions. Due to the SOA formation, the OM concentrations over 10 Midwest US and Central Siberia are also noticeably high. The sea salt concentrations are 11 distributed fairly uniformly over the oceans but are higher over the Southern Ocean and 12 lower over the oceans near the tropics as expected due to wind speed variations. The dust 13 concentrations are pronounced over the source regions such as Northern Africa, Arabia, 14 Northern/Northwestern China, and Australia.

15 Figure 7 shows zonal distributions of annual-mean aerosol concentrations in 16 ModelE2-TOMAS. The sulphate concentrations are highest between 0° N and 50° N due 17 to the high anthropogenic emissions in the NH. Over the SH, the sulphate concentrations 18 are mostly a result of DMS oxidation. The zonal-mean EC and OM concentrations are 19 high from the tropics to $\sim 50^{\circ}$ N. Similar to Fig. 6, the high EC and OM concentrations are around 30° N to 50° N due to fossil fuel emissions, but the OM concentrations are 20 21 also large around the tropics due to biomass burning emissions. The two small spikes shown the EC and OM concentrations between 10° S and 10° N are a result of the 22 23 injection height used in the biomass burning emissions. Since a significant amount of 24 sulfate and OM are also formed through chemical reactions in the atmosphere, their 25 vertical gradients are relatively small. The sea salt concentrations are high from 60° S to 26 50° N with a peak around 30° S to 60° S due to the large open ocean in the SH. A strong 27 dust plume is shown at around 0°-30° N due to the large Northern African and Middle 28 East dust emissions, and a small dust plume at around 30° S due to Australian emissions. 29 In the model, the PM10 concentrations in upper troposphere are dominated by sulfate and 30 dust particles.

1

5.3 Aerosol mass concentrations evaluation

2

3 The simulated surface-layer mass concentrations of aerosols are evaluated against 4 various observations: 1) sulphate/sea salt/dust concentrations at 23 long-term observation 5 sites operated by University of Miami (e.g. Prospero and Bonatti, 1969; Savoie and Prospero, 1989; Arimoto et al., 1990); 2) speciated PM2.5 concentrations from the 6 7 Interagency Monitoring of Protected Visual Environment (IMPROVE) sites in the United 8 States that are annual-averages from 2000 to 2008 (Debell et al., 2006; Hand et al., 9 2011); 3) speciated PM2.5 concentrations from various European observations (Putaud et 10 al., 2010) (hereafter, referred to as European sites); 4) a large set of PM2.5 observations 11 assembled in support of the Global Burden of Disease Study (GBD Study 2010, 12 http://www.globalburden.org); 5) deposition flux measurements obtained from Ginoux et 13 al. (2001), Tegen et al. (2002), and Mahowald et al. (2009). The PM2.5 dataset from the 14 GBD Study consists of a worldwide set of annual-average PM2.5 largely drawn from 15 official monitoring networks for 2005 (in some cases from 2004-2006). The GBD PM2.5 16 dataset includes the IMPROVE network and the European site measurements, so we only 17 present the PM2.5 evaluation with the GBD dataset. For details of the GBD PM2.5 18 dataset, the reader is referred to the description in the supplementary material in Shindell 19 et al. (2011).

20 Figure 8 compares the model annual-mean surface-layer sulphate mass concentrations 21 to the observations from the IMPROVE network, the European sites, and the University 22 of Miami network. Simulated sulphate agrees well with observations, mostly within a 23 factor of two. Compared to ModelE2-OMA, the ModelE2-TOMAS model shows better 24 agreement in the European sites (i.e. LMNB=-0.06 and LMNE=0.13 in Fig. 8b) but 25 worse in the IMPROVE sites (i.e. LMNB=0.06 and LMNE=0.16 in Fig. 8a) and the 26 remote oceanic sites (i.e. LMNB=0.04 and LMNE=0.22 in Fig. 8c). Over the US, both models overpredict systematically at lower observed concentrations (i.e. below 1 μ g m⁻³ 27 28 of measured SO₄ concentrations), which are mostly located over the western US.

Monthly mean surface-layer sulphate concentrations are evaluated using observations from the University of Miami in Figure 9. The simulated sulphate mass concentrations from both models usually falls within the standard deviation of the observed values. The

1 sulphate predictions in both ModelE2-OMA and ModelE2-TOMAS become quite similar 2 over the SH because about a factor of two lower DMS emissions are used in ModelE2-3 TOMAS. Using the same DMS emissions as ModelE2-OMA, ModelE2-TOMAS tends to 4 overpredict sulphate concentrations noticeably over the SH (not shown), and the higher 5 SO_4 concentration with ModelE2-TOMAS could be explained by 1) a longer lifetime due 6 to different deposition parameterizations, 2) letting all SO₄ formed from aqueous 7 oxidation to evaporate without accounting for cloud evaporation, 3) a stronger oxidation 8 state resulted from un-degraded photolysis rates by aerosol optical depth (more SO₄ can 9 be formed from DMS oxidation). Despite the fact that ModelE2-OMA accounts for the 10 heterogeneous SO₄ formation on dust particles, SO₄ concentrations in near dust source 11 regions are still higher in the ModelE2-TOMAS model due to the shorter lifetime in the 12 ModelE2-OMA model.

13 Simulated annual-mean surface-layer sea salt concentrations are evaluated with the 14 IMPROVE network, the European sites, and the University of Miami network (Fig. 10). 15 Both aerosol models are biased strongly high over the US and Europe but biased low near 16 the tropics. Unlike ModelE2-OMA, ModelE2-TOMAS underpredicts sea salt 17 concentrations at several remote sites (see Fig. 10c). Compared to other aerosol 18 components, the agreement between model sea salt and the observations is worse over the 19 United States and Europe. For the evaluation of monthly mean surface-layer sea salt 20 concentrations against the University of Miami dataset in Figure 11, the model 21 predictions fall within the observed standard deviation at about a half of the 26 sites, but 22 these sites are not necessarily the same between the two aerosol models. The 23 overprediction of sea salt in continents may suggest that ModelE2 has a stronger 24 transport from ocean to in-land, as there is no obvious overprediction over adjacent 25 oceanic sites. The ModelE2-OMA model exhibits a particularly large overprediction over 26 most SH sites. Both models tend to be biased significantly low at some of the sites near 27 the tropics where the observed sea salt concentrations are high. Similar underprediction is 28 also shown in mineral dust (see Fig. 13). This might be due to fast wet scavenging due to 29 overpredicted precipitation in that area (see Fig. 9 in Schmidt et al., 2014).

30 For ModelE2-TOMAS, simulated annual-mean surface-layer dust concentrations are 31 mostly within a factor of two of the measurements (in Fig. 12): 6 sites are excluded in

1 Fig. 12 due to an incomplete annual cycle. ModelE2-TOMAS shows good agreement at 2 the IMPROVE and European sites, while ModelE2-OMA tends to underpredict. This is 3 due to the emission size assumptions (resulting in more clay emissions than ModelE2-4 OMA) and the differences in the deposition parameterizations (resulting in slightly longer 5 lifetime for clay particles; ~ 9 days in ModelE2-TOMAS and 6.5 days in ModelE2-6 OMA). Both models are biased low compared to the University of Miami dataset, and 7 only 7-8 sites among 20 sites fall within a factor of two agreement. A few sites show a 8 severe underprediction for both models, mostly located near the tropics and in SH high 9 latitudes.

10 Monthly mean surface-layer dust concentrations are evaluated using the observations 11 from the University of Miami in Figure 13: no measurement data is available at 3 sites (12 Reunion Island, Invercargill, and Marion Island), but we still include them to compare the 13 two aerosol model predictions. The model captures the observed annual cycle of dust 14 very well at most NH sites but not in the SH sites except Norfolk and Mawson. The 15 model captures the observed magnitude well at sites located relatively near the source 16 regions (e.g., Sal Island, Barbardos, Bermuda for African dust; Jeju Island and Hedo for 17 Asian dust; Cape Grim and Norfolk Island for Australian dust). Both models underpredict 18 dust concentrations in the NH/SH high latitude (45°) regions except Mace Head, but the 19 simulated dust seems to be within the observed standard deviation as the observations 20 have a large standard deviation for dust. The observed peak concentration at Heimaey 21 Iceland is the second highest after Sal island. Our models underpredict this site severely 22 probably because our dust emission parameterization is not designed to simulate a dust 23 event in humid areas such as Iceland. Prospero et al. (2012) points out that dust emissions 24 at high latitudes (e.g., Alaska and Iceland) are mostly due to individual dust events or 25 single seasons and link large dust events at Heimaey Iceland during 1997 to 2002 with 26 glacial outburst floods.

Figure 14 compares simulated annual-mean dust deposition fluxes against observations obtained from Ginoux et al. (2001), Tegen et al. (2002), and Mahowald et al. (2009). Data is classified by the influencing source region, presented in different colours in Fig. 14. Except for the minor source category, the model dust deposition fluxes tend to be underestimated at most locations and agree with observations only within a factor of 5-8 on average. This may indicate that dust emissions are too low in ModelE2, but deposition fluxes measurements could contain large particles especially close to source regions (e.g., Duce, 1995) or local emissions (Uematsu et al., 1985) that are not simulated in the model. For ModelE2-TOMAS, the upper size limit being 10 μm, which is too small for the dust particles near the sources, can explain some of the severe underprediction, and Lee et al. (2009) also shows similar disagreement using the GISS-TOMAS model.

8 Figure 15 shows simulated annual-mean surface-layer EC and OM concentrations 9 compared against the observations from the IMPROVE network and the European sites 10 from Putaud et al. (2010). The simulated EC and OM in both models agree very well at 11 the IMPROVE sites (for EC, LMNB=-0.08 to 0.05; for OM, LMNB=-0.17 to -0.08). 12 Note that we applied an OM/OC ratio of 1.4 to the IMPROVE network to make it 13 consistent with our model assumption, but the IMPROVE OM data provided to us was 14 based on the ratio of 1.8. Over Europe, the model predictions are still reasonable (within 15 a factor of 2-3), but the agreement is slightly worse than the IMPROVE sites. Since these 16 sites are mostly adjacent to the source/emissions, the good agreement suggests that the 17 emission inventory (used in this study) is well represented for these regions.

18 Figure 16 compares simulated annual-mean PM2.5 concentrations against the GBD 19 dataset. Note that ModelE2-OMA includes nitrate mass into its PM2.5, and the nitrate 20 contributes to PM2.5 rather significantly. The GBD data is classified/presented by a 21 region listed in Figs. 16a and b. The aerosol models capture the observation quite well in 22 most locations (overall LMNB is -0.08 to -0.03 and LMNE is 0.2 to 0.25), but both 23 models show the worst agreement for Oceania regions (LMNB/LMNE=0.4-0.58) and 24 Latin America (LMNB/LMNE =-0.33 to -0.48). The PM2.5 overprediction in Oceania is 25 mainly due to too much fine mode sea-salt particles (the overall agreement in Oceana is 26 little influenced by sulfate or dust particles). Note that the sea salt comparison to the 27 Miami dataset (in Figures 11) shows severe underpredictions in several sites in Oceania 28 because their concentrations are likely dominantly by coarse mode sea salt particles. The 29 underprediction of PM2.5 in Latin America might be related to the biomass burning 30 emissions. It is consistent with the model AOD being biased low over biomass burning 31 source regions that is shown in Section 5.4. Model evaluation with the observed PM10

1 concentrations using the IMPROVE and European sites was also performed and is similar 2 to the PM2.5 evaluation results (not shown): LMNB and LMNE are -0.01 and 0.17 for 3 ModelE2-TOMAS and 0.0 and 0.29 for ModelE2-OMA, respectively.

- 4

5.4 Aerosol optical depth evaluation

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6 Simulated annual-mean clear-sky aerosol optical depths (AODs) at 550 nm are 7 compared with observations from the Terra MODIS (MODerate resolution Image 8 Spectroradiometer; e.g. Abdou et al., 2005; Remer et al., 2008) and MISR (Multiangle 9 Image SpectroRadiometer; e.g. Diner et al., 1998; Kahn et al., 2005) satellite instruments 10 averaged over 2004-2006 (Fig. 17). Specifically, we use Terra MODIS Level 3 11 (MOD08 M3.051), Aqua MODIS Level 3 (MYD08 M3.051), and Terra MISR Level 3 12 (MIL3MAE4), which are monthly products with 1x1 degree resolution. We also use the 13 Deep Blue AOD (e.g., Hsu et al., 2006) from Terra MODIS to increase its spatial satellite 14 all coverage, and the data was obtained from 15 http://disc.sci.gsfc.nasa.gov/giovanni. When the Deep Blue AOD and the "regular" AOD 16 from Terra MODIS are both available, the former is used. However, we notice that the 17 Deep Blue AOD shows some local hotspots (e.g. over South America) that are not shown 18 in the "regular" MODIS AOD as well as the MISR AOD. Our model calculates clear-sky 19 AOD by including only AOD values calculated in model locations where clouds are not 20 present (i.e., cloud-free grid-box only). The ModelE2-TOMAS and ModelE2-OMA 21 models capture the broad spatial features seen in the satellite measurements: 1) very high 22 AODs over desert regions in and near Northern Africa and the Arabian Peninsula 23 associated with mineral dust, 2) the band of locally enhanced AOD over the Southern 24 Ocean associated with sea salt, 3) high AODs over East Asia and India due to high 25 anthropogenic emissions. However, the models underestimate the AOD over East Asia 26 and the Indo-Gangetic plain, especially compared to MODIS. Including nitrate in 27 ModelE2-OMA, the AOD predictions increase particularly over Europe and East Asia. 28 Without nitrate, they become quite similar to the ModelE2-TOMAS AOD values in the 29 regions where the model nitrate predictions are significant. The models show an 30 enhanced AOD over the biomass burning regions such as tropical South America, Africa 31 and Indonesia but it is clearly underestimated. The simulated AOD in North America and

high latitude regions appears also to be lower than the satellite observations. Comparing
with ModelE2-OMA AOD, ModelE2-TOMAS shows a stronger AOD over Africa due to
its higher dust burden and a lower AOD over the marine areas especially the Southern
Ocean associated with sea salt.

5 We present spatial correlations and biases between the models and the satellite data 6 (Table 6). For these analyses, the annual-mean satellite AOD fields are regridded to 7 $2^{\circ}x2.5^{\circ}$ horizontal resolution, and the models are sampled only where the satellite AOD 8 is available. Correlation coefficients between the model and satellite AODs are around 9 0.6-0.7 for ModelE2-TOMAS and around 0.4-0.5 for ModelE2-OMA. Given that the 10 correlation coefficient between MODIS and MISR is 0.79, the ModelE2-TOMAS model 11 shows a good correlation with these satellite data. Both models show better correlations 12 with MISR AOD. Compared to ModelE2-OMA, ModelE2-TOMAS shows a strong 13 negative bias (about -29 to -34%) for both satellite datasets because of noticeably low 14 AODs over the oceanic regions (where sea salt is dominant) shown in Fig. 16 and 15 possibly because of missing component such as nitrate aerosols, which contributes to 16 AOD significantly over Europe and China in the ModelE2-OMA model. Over Europe, 17 the ModelE2-OMA model overpredicts AOD due to nitrate though. Without nitrate in 18 ModelE2-OMA, the normalized mean bias (NMB) falls from 8-16% to -16 to -21%. Both 19 models show lower AOD over China, India, and biomass burning regions and a similar 20 underprediction is shown in by the ACCMIP models (Shindell et al., 2013), indicating a 21 possibility of aerosol emissions being underestimated in these regions.

22 Figure 18 presents simulated monthly mean AODs compared against AERONET 23 (AErosol RObotic NETwork; Holben et al., 1998; Holben et al., 2001) measurements at 24 28 sites that represent the following characteristic regions: polluted continental, marine, 25 biomass-burning, and dusty regions (see Table 7 for individual site information). Both 26 ModelE2-TOMAS and ModelE2-OMAunderestimate the maximum AOD during 27 summer by a factor of 2-3 in the biomass burning sites (1 to 6) but capture the observed 28 annual cycle quite well (R > 0.9). Unlike other biomass burning sites, simulated AODs 29 at Ilorin and Banizoumbou are comparable to the observations possibly due to the 30 influence of mineral dust. The agreement between the models and the AERONET AOD 31 is generally good in the dusty regions (9 to 16). The ModelE2-TOMAS model shows a slight overprediction of AOD during spring at Capo Verde and Bidi Bahn which are
 located near the African dust sources, although it shows a good agreement at Barbados
 where is also influenced by African dust.

4 For ModelE2-TOMAS, all polluted continental sites (17 to 24) show large 5 underpredictions, while the model tends to capture the observed annual cycle well. 6 However, the model surface aerosol mass concentrations agree well with the observations 7 from IMPROVE network and several European sites (see Figs. 8, 10, 12, 15, and 16). 8 This might be due to the fact that column AOD depends on many additional factors (e.g., 9 optical properties and vertical distribution of aerosols) and provides a measure of total 10 radiatively active aerosols in the atmosphere. Misrepresentation of these factors or 11 missing a chemical component could introduce a bias in the model AODs. For ModelE2-12 OMA, the model AOD without nitrate is quite similar to the ModelE2-TOMAS AOD. 13 Including nitrate in ModelE2-OMA, it simulates the annual-mean AERONET AOD 14 relatively well but the observed annual cycle worse. Although the inclusion of nitrate is 15 helpful for the underpredicted AOD in polluted regions, the overprediction of wintertime 16 AOD suggests that the model nitrate is too large.

17 With the ModelE2-TOMAS model, the oceanic sites (25 to 30) are generally 18 underpredicted roughly by a factor of two except for Bermuda (25) and Lanai (26). By 19 contrast, the ModelE2-OMA model captures the observed magnitude relatively well but 20 overpredicts at Bermuda and Lanai. Note that Bermuda (23) and Rottnest Island (27) are 21 influenced by long-range transported mineral dust. The ModelE2-OMA model predicts 22 AOD that is more comparable to observations in these oceanic sites than ModelE2-23 TOMAS. Compared to ModelE2-OMA, the underprediction of sea salt concentrations at 24 the remote sites (shown in Figs. 10 and 11 in Sect. 5.3) and the underpredictions of 25 AODs in the remote oceanic sites in ModelE2-TOMAS may be due to a faster sea salt 26 removal rate (see global budgets in Table 4 for the details). It is worth mentioning that 27 the inclusion of marine organic aerosols, which are not included in this paper, may not 28 increase AOD noticeably in remote oceanic sites. Using the same host model, Tsigaridis 29 et al. (2013) shows that Southern Ocean AOD is quite insensitive to the inclusion of 30 marine organic particles but is strongly sensitive to the sea-salt emissions 31 parameterization (see Figure 9 of Tsigaridis et al. (2013).

1 2

5.5 Aerosol number budgets and its distributions

3 This section includes only ModelE2-TOMAS results, as ModelE2-OMA does not 4 predict aerosol number concentrations. Global mean number budgets for all three 5 simulations, including two sensitivity runs for nucleation rates, are presented in Table 8 6 (see Section 4.2 and Table 1 for the details of run descriptions). Compared to the BASE 7 run, global mean CN3 (particles with dry diameters larger than 3 nm), CN10 (particles 8 with dry diameters larger than 10 nm), and CN100 (particles with dry diameters larger 9 than 100 nm) concentrations in the LowNUC run are decreased by 74%, 33%, and 7% in 10 the troposphere and 29%, 17%, and 3% in the lowermost layer, respectively. The smaller 11 impact on CN in the lowermost layer is due to the presence of the primary emissions near 12 the surface. It is also because the binary nucleation parameterization used in this model 13 produces few particles in the boundary layer; however, nucleated particles in the 14 upper/free troposphere may be carried down to the surface and influence CN there. Since 15 the LowNUC run produces fewer nucleated particles than the BASE run, each nucleated 16 particle is more likely to grow more efficiently to form CCN-sized particles, as there is 17 less competition for sulphuric acid and condensable organics. We can see this using the number budgets in Table 8. The increase in CN70 with the BASE case nucleation (i.e., 18 BASE CN70 – NONUC CN70) is 51 cm⁻³ for the nucleation rate (J3) of 0.131 cm⁻³ s⁻¹ 19 and that with the LowNUC case nucleation, 37 cm⁻³ for J3 of 0.013 cm⁻³ s⁻¹. In the BASE 20 21 run, J3 is 10 times higher but the CN70 increase by nucleation is only ~1.4 times higher 22 than those in the LowNUC run.

Aerosol number burdens normalized by tropospheric volume are dramatically reduced when nucleation is turned off in the NoNUC run. For example, global mean CN3 and CN10 are decreased by 95% and 76% in the troposphere and by 42% and 31% in the surface layer. We found that 24% of tropospheric CCN-sized particles (i.e., CN100) and ~10% of surface-layer CCN-sized particles result from binary nucleation in our model; the contribution of the nucleated particles to the CCN concentrations is larger as the cutoff size (e.g. 100 nm in CN100) gets lower.

Figure 19 shows global distributions of annually averaged CN3, CN10, and CN100 in the lowermost layer for the BASE, LowNUC, and NoNUC runs, and Fig. 20 presents their zonal distributions. For the CN3 distributions, the BASE run shows a high

concentration (over 5000 cm⁻³) in the upper/free troposphere and the entire Antarctica 1 2 troposphere due to nucleation and over polluted continental regions due to the primary 3 emissions (see Figs. 19a and 20a). Note that the CN3 near the surface in Antarctica 4 comes from nucleated particles from the upper/free troposphere, as the binary 5 parameterization does not predict nucleation in the boundary layer. CN3 in the LowNUC 6 run (Figs. 19d and 20d) is decreased significantly in the free troposphere and in the 7 surface layer over Eurasia and western Antarctica (fewer nucleated particles formed 8 above subside to the surface). When nucleation is switched off, CN3 is very close to 9 CN10 near to the surface (Figs. 19g and h) because nucleation contributes most CN 10 between 3 nm to 10nm. For CN10, its spatial pattern over the continents is quite similar 11 among the runs as it is mostly driven by primary emissions except for some locations 12 heavily influenced by the nucleated particles formed in the upper/free troposphere (Figs. 13 19b, e and h and Figs. 20b, e, h). A pronounced difference in CN10 is shown over 14 oceans, indicating a larger contribution of nucleated particles to CN10 in these regions. 15 This is consistent with Merikanto et al. (2009), which shows higher contribution of 16 nucleation to CN over oceanic regions. CN100 differs little among runs except for SH 17 high latitudes where binary nucleation plays an important role (Figs. 19c, f, and i). This is 18 consistent with Pierce and Adams (2009) showing that the exhibited high SH latitude 19 region for the most positive changes in CCN(0.2%) by turning on binary nucleation and, 20 again, with Merikanto et al. (2009) showing 65% of CCN(0.2%) in Antarctica resulted from upper tropospheric nucleation. CN100 shows a maximum at 10° N to 40° N because 21 22 most anthropogenic primary emissions are located in mid-NH latitudes. Rather 23 surprisingly, dust particles in our model contribute to CN100 quite significantly over the 24 source regions. This is opposite to the results from GISS-TOMAS (Lee et al., 2009), 25 which shows a 10-20% reduction in CCN(0.2%) when introducing mineral dust 26 emissions. Despite the direct source of CCN-sized particles from dust emissions, CCN 27 and ultrafine particles that grow to become CCN are scavenged via coagulation with 28 coarse dust particles, and dust particles compete for condensable sulfuric acid. leading to 29 a slower growth rate of ultrafine particles.

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5.6 Aerosol number evaluation

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1 The CN measurement dataset compiled by Spracklen et al. (2010) is used to evaluate 2 simulated annual-mean CN concentrations (in Fig. 21) and monthly mean CN 3 concentrations (in Fig. 22). Details of the measurement procedures for each site, 4 including the instrument type and minimum cutoff diameter (varying from 3nm to 14 5 nm), can be found in Table 2 of Spracklen et al. (2010). Due to an incomplete annual 6 cycle, we excluded Mt. Waliguan, Finokalia, Listvyanka, and Weybourne. The 7 measurement sites are classified into three categories: FT (free troposphere; 1 to 8 in Fig. 8 22), MBL (marine boundary layer; 9 to 15 in in Fig. 22), and CBL (continental boundary 9 layer; 16 to 32 in Fig. 22). We sampled the model values to match the altitude of each 10 measurement site except for the free troposphere sites that use an altitude 30% lower 11 because it improved the annual cycles prediction significantly. Previous studies pointed 12 out that free tropospheric sites can be influenced by upslope winds that carry the 13 planetary boundary layer air, so it cannot be assumed to be in the free troposphere all the time (e.g., Baltensperger, 1997; Collaud Coen et al., 2011). Our model does not seem to 14 15 simulate this well, so sampling the model predictions at lower altitude (i.e., 30% lower) 16 helps to increase the influence of PBL air. In Fig. 21, a whisker plot is used to present the 17 three run results; the maximum of the whisker line for the BASE run; the circle symbol in 18 the middle of the whisker line for the LowNUC run; the minimum for the NoNUC run.

19 On average, the annual-mean CN concentrations in the model agree with the 20 observations well for the all three categories (LMNB= -0.26 to 0.16; LMNE=0.13 to 21 0.22; see Fig. 21), although the LowNUC simulation shows the best agreement to 22 observation. Simulated annual cycles at individual sites also show that the model captures 23 the measured magnitude reasonably but overpredicts the CN during winter (November to 24 March) that is worse at the CBL sites and results in poor seasonality (especially in the 25 BASE run). The poor seasonality in ModelE2-TOMAS in all three runs suggests that 26 other factors may play a role such as missing seasonal variation in primary emissions or 27 scavenging that causes the poor seasonality rather than nucleation. The evaluation of CN 28 at South Pole shows that the model predicts too strong nucleation throughout the year.

Figure 23 compares the observed size distributions at six European sites during winter (DJF: December to February; Figs. 23a to f) and summer (JJA: June to August; Figs. 23d to l) to the model. The observed size distributions are averaged during morning,

1 afternoon, and night, while the model results are not broken into the three periods. This 2 data is obtained from Putaud et al. (2003) and, for the same sites shown in Figure 21, the 3 temporal coverage used in Putaud et al. (2003) is not necessarily matched with them. 4 Most sites are reasonably close to the dataset used in Figure 21. However, for 5 Jungfraujoch, the total CN concentrations summed from the size distribution data, which 6 covers from June 1997 to May 1998, is about a factor of two lower than the CN shown in 7 Figure 21, which covers from 1995 to 1999 and 2003-2007. As expected, the simulated 8 size distributions of the model simulations differ according to what nucleation scheme is 9 used. The impact of nucleation is more notable for particles smaller than ~50 nm, as 10 expected, and also during winter. In general, the higher nucleation rates tend to 11 overpredict Aitken model particles at most sites. For Harwell (Figs. 23c and i), particles 12 below 50 nm are overpredicted even without nucleation. For the summer season, the 13 model shows less sensitivity to nucleation rates and has better skill at capturing the 14 observed size distributions.

15 The observed CN5, CN15, and CN120 concentrations from the LACE campaign 16 (Petzold et al., 2002) are compared with three model runs (Figs. 24a to c). Below 700-17 800 mbar, all three model runs predict concentrations roughly within the observed CN5, 18 CN15, and CN120 ranges. Above 700-800 mbar, the BASE simulation overpredicts CN5 19 and CN15 by approximately an order of magnitude and 2-3 times, respectively. For the 20 NoNUC simulation, it captures the lower side of the observed CN15 but fails to capture 21 the increasing CN5 concentrations with height above 600 mbar. All model runs basically 22 show almost identical CN120 concentrations and fall on the lower edge of the observed 23 range.

We compare CN3 vertical profile measurements averaged into the 3 latitude bands over the Pacific Ocean (Clarke and Kapustin, 2002) with the model (Figs. 24 d to f). The simulated CN3 profiles in LowNUC agree well with the observation, capturing the increasing CN3 with height. Although the BASE run shows the increasing pattern correctly, it overpredicts CN3 severely above approximately 6 km. The NoNUC run disagrees with the observations for all latitudes and altitutes and clearly fails to reproduce the high number concentrations in the upper troposphere that result from nucleation.

1 The observed Aitken mode and accumulation mode concentrations and size 2 distributions in the marine boundary layer (Heintzenberg et al., 2000) are compared with 3 the model in Figs. 25 and 26. The measurements are aggregated into 15-degree latitude 4 ranges. For the Aitken mode (Fig. 25a), the LowNUC run shows the closest agreement to 5 observations. However, whereas the observations show higher concentrations in the SH 6 than in the NH, all model simulations show the opposite tendency. Similarly, other 7 global models with binary nucleation show underpredicted CN concentrations in the SH 8 and either well-simulated or overpredicted CN in the NH (e.g. Easter et al., 2004; 9 Spracklen et al., 2005; Pierce and Adams, 2006; Pierce et al., 2007; Trivitayanurak et al., 10 2008; Mann et al., 2010; Wang et al., 2011). For the accumulation mode (Fig. 25b), the 11 three model runs are quite similar to each other and are within the observed range.

12 In Fig. 26, ModelE2-TOMAS captures the bimodal size distribution shown in the 13 observations reasonably, which is mainly determined by the activation diameter assumed 14 in wet deposition (and cloud processing): ~80 nm as an activation diameter 15 (supersaturation of 0.2%) for both large-scale clouds and convective clouds in the model. 16 Note that ModelE2-TOMAS cannot capture the observed bimodal distribution when a 17 supersaturation of 1.0% is assumed for convective clouds, unlike the other TOMAS 18 models (e.g. Pierce et al., 2007; Trivitayanurak et al., 2008): a peak at around 20-30 nm 19 appears when the supersaturation of 1.0% is assumed for convective clouds. This 20 suggests that, compared to GISS GCM II', convective clouds are more frequent in 21 ModelE2. Although the model captures the observations successfully using fixed 22 supersaturation assumptions, future work is needed to link the in-cloud supersaturation to 23 cloud and aerosol properties.

24 Simulated CCN concentrations are compared against a dataset of CCN measurements 25 compiled by Spracklen et al. (2011) in Figs. 27 and 28: see Table 1 in Spracklen et al. 26 (2011) for the details regarding each site. The CCN dataset includes a total of 277 27 measurements at 80 locations using various instruments from 1971 to 2009. 28 Approximately 70% of the observations were taken after 1990. Most have sampling 29 periods of days to weeks except the observations at Cape Grim and Mace Head. For Cape 30 Grim and Mace Head, an annual cycle is available, so we present them separately in Fig. 31 28. In Fig. 27, the CCN data is divided into two groups: CCN in the MBL (marine boundary layer) and CCN in the CBL (continental boundary layer). Note that all CCN measurements used here are in the boundary layer. For CCN in the CBL, the model CCN shows good agreement with the observation in all three simulations (LMNB=0.11-0.2 and LMNE=0.31-0.34). For CCN in the MBL, the model predictions are, on average, within a factor of 1.5-2 of the observations for all three runs but, relative to several measurements of CCN concentration between ~100 and ~300 cm⁻³, are biased high, by roughly a factor of two.

8 For the annual cycle of CCN concentrations at Cape Grim and Mace Head (Fig. 28), 9 the model overpredicts in all months even without nucleation (NMB>1.0). The CCN 10 overpredictions at Cape Grim might be influenced by overpredicted SO_2 (shown in Fig. 11 3), which could lead to overpredicted condensational growth of Aitken mode particles. 12 As with the SO₂ evaluation in Fig. 3, the CCN overprediction decreases by sampling 13 adjacent grids toward the ocean (not shown), but the model CCN is still higher than the 14 measurements. Given the fact that most CCN observations have very short duration (days 15 to weeks) in a single year and, according to Spracklen et al. (2011), the relative 16 uncertainties in the measurement data range from about 5-40%, mostly in 10-20% very 17 roughly, the overall model-to-observation agreement is satisfying.

18

19 **6.** Conclusions

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We have implemented the TwO-Moment Aerosol Sectional (TOMAS) microphysics model into the new version of the GISS GCM (i.e. ModelE2), called "ModelE2-TOMAS". This paper has compared the global budgets of ModelE2-TOMAS to other global aerosol models and evaluates the model with various observations such as aerosol precursor gas concentrations, aerosol mass and number concentrations, and aerosol optical depth.

Global budgets of aerosols and aerosol precursor gases in ModelE2-TOMAS are similar to those in other global aerosol models, and the ModelE2-TOMAS model agrees reasonably (mostly within a factor of two) with long-term observed aerosol precursor gas and aerosol mass concentrations. The model captures the broad spatial features shown in the MODIS and MISR annual-mean AOD distributions as well as the observed seasonal

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trends of AOD at several AERONET sites. The model predicts the observed annual-mean CN (the minimum cutoff varying from 3nm to 14 nm) concentrations very well and the observed vertical profiles of aerosol number over Germany (i.e., LACE campaign) and in the marine boundary layer. For CCN, the model shows good skill in capturing the observations. We conclude that the model is realistic enough to be useful for many types of scientific study.

7 However, the evaluations have also highlighted some weaknesses in ModelE2-8 TOMAS to be revisited in the future. First, ModelE2-TOMAS predicts too much SO₂ 9 lifted into the upper/free troposphere over the Pacific Ocean possibly due to overly strong 10 convective transport. This is also seen in ModelE2-OMA (and in gaseous tracers), suggesting this might be a host model problem. Second, the model AOD is 11 12 underpredicted over polluted continents, even though mass concentrations of each aerosol 13 component at the surface are well simulated (or at least not underpredicted) in the model. 14 Missing nitrate in ModelE2-TOMAS may not be the main contributor, as the inclusion of 15 nitrate in ModelE2-OMA decreases its ability to capture the observed seasonality in 16 AERONET polluted continents sites. Third, the model tends to underpredict aerosol 17 loading (and AOD) over biomass burning emission regions. This is a common issue in 18 global aerosol models, and this might be due to underestimation of biomass burning 19 emissions. Fourth, the ModelE2-TOMAS AOD prediction is biased low over the SH 20 high-latitude oceans, which suggests underpredicted sea salt burden in this area. Fifth, the 21 simulated CN seasonality is poor at some CBL sites due to overpredicted CN during the 22 wintertime. Similarly, ModelE2-TOMAS predicts the observed number size distributions 23 over European sites during the summer season reasonably but not for the winter season. 24 The model overpredicts Aitken mode particles during the winter season, which happens 25 even without nucleation, possibly due to a problem in primary emissions representation 26 or a bias in the model scavenging that causes the poor seasonality. Finally, ModelE2-27 TOMAS seems to predict faster nucleation rates using binary nucleation (Vehkamaki et 28 al., 2002) than other global microphysics models using the same binary nucleation 29 (including GISS-TOMAS). For instance, the observed CN vertical profiles are captured 30 the best when nucleation rates are reduced (our LowNUC run), while Lauer et al. (2005) 31 presents significantly underpredicted CN profiles with the same binary nucleation scheme

1 using ECHAM/MADE. To investigate this issue further, we need to perform a model 2 evaluation against observation-derived nucleation-relevant metrics, which has been done 3 in Westervelt et al. (2013). Because of the fine size assumptions for the primary 4 emissions (see Section 4.2 for the details) and/or the fast nucleation rates using binary 5 nucleation in ModelE2-TOMAS, we do not include boundary nucleation as it 6 overpredicted CN near the surface. We conclude that further work is necessary to 7 improve the realism of some aspects of ModelE2-TOMAS and to better understand the 8 size-resolved physical processes (e.g. microphysics, emissions, and depositions).

9 Comparing the ModelE2-TOMAS model with the ModelE2-OMA model in 10 ModelE2, some aerosol species burdens and concentrations differ significantly, even with 11 using the same host model and identical anthropogenic emissions, mainly because the 12 deposition parameterizations and some of the emission-related assumptions are different. 13 In the case of sea salt and dust, the size ranges assumed by each model are different, 14 resulting in different emission rates and burdens.

Analysis of multiple aerosol model results help to identify where a model bias might originate from: aerosol modelling or the host GCM or elsewhere such as emissions. We found that some of the large differences in aerosol predictions between the two aerosol models are due to aerosol modelling. This is valuable information, as this is not easy to constrain using observations. Having more than one aerosol physics representation in the NASA GISS ModelE2 will serve as a useful tool to study the uncertainty in aerosol modelling and to guide our efforts to improve the models.

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- 10
- 11

12 **References**

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- 14 Abdou, W. A., Diner, D. J., Martonchik, J. V., Bruegge, C. J., Kahn, R. A., Gaitley, B. J.,
- 15 Crean, K. A., Remer, L. A., and Holben, B.: Comparison of coincident Multiangle
- 16 Imaging Spectroradiometer and Moderate Resolution Imaging Spectroradiometer aerosol
- 17 optical depths over land and ocean scenes containing Aerosol Robotic Network sites, J.
- 18 Geophys. Res.-Atmos., 110, D10S07, doi: 10.1029/2004jd004693, 2005.
- 19 Adams, P. J., and Seinfeld, J. H.: Predicting global aerosol size distributions in general
- 20 circulation models, J. Geophys. Res.-Atmos., 107, 4370, doi:10.1029/2001JD001010,
- 21 2002.
- 22 Andres, R. J., and Kasgnoc, A. D.: A time-averaged inventory of subaerial volcanic
- sulfur emissions, Journal of Geophysical Research: Atmospheres, 103, 25251-25261,
- 24 10.1029/98JD02091, 1998.
- Arimoto, R., Ray, B. J., Duce, R. A., Hewitt, A. D., Boldi, R., and Hudson, A.:
- 26 CONCENTRATIONS, SOURCES, AND FLUXES OF TRACE-ELEMENTS IN THE
- 27 REMOTE MARINE ATMOSPHERE OF NEW-ZEALAND, J. Geophys. Res.-Atmos.,
- 28 95, 22389-22405, 10.1029/JD095iD13p22389, 1990.
- 29 Ayers, G. P., Ivey, J. P., and Gillett, R. W.: COHERENCE BETWEEN SEASONAL
- 30 CYCLES OF DIMETHYL SULFIDE, METHANESULFONATE AND SULFATE IN
- 31 MARINE AIR, Nature, 349, 404-406, 10.1038/349404a0, 1991.
- 32 Baltensperger, U.: Aerosol climatology at the high, ÄêAlpine site Jungffaujoch,
- 33 Switzerland, Journal of Geophysical Research, 102, 19707-19715, 1997.
- 34 Bauer, S. E., and Koch, D.: Impact of heterogeneous sulfate formation at mineral dust
- 35 surfaces on aerosol loads and radiative forcing in the Goddard Institute for Space Studies
- 36 general circulation model, J. Geophys. Res.-Atmos., 110, D17, D17202, doi:
- 37 10.1029/2005jd005870, 2005.
- 38 Bauer, S. E., Koch, D., Unger, N., Metzger, S. M., Shindell, D. T., and Streets, D. G.:
- 39 Nitrate aerosols today and in 2030: a global simulation including aerosols and
- 40 tropospheric ozone, Atmospheric Chemistry and Physics, 7, 5043-5059, 2007.
- 41 Bauer, S. E., Wright, D. L., Koch, D., Lewis, E. R., McGraw, R., Chang, L. S., Schwartz,
- 42 S. E., and Ruedy, R.: MATRIX (Multiconfiguration Aerosol TRacker of mIXing state):
- 43 an aerosol microphysical module for global atmospheric models, Atmospheric Chemistry
- 44 and Physics, 8, 6003-6035, 2008.
- 1 Bian, H. S., and Prather, M. J.: Fast-J2: Accurate simulation of stratospheric photolysis in
- 2 global chemical models, Journal of Atmospheric Chemistry, 41, 281-296,
- 3 10.1023/a:1014980619462, 2002.
- 4 Boucher, O., Moulin, C., Belviso, S., Aumont, O., Bopp, L., Cosme, E., von Kuhlmann,
- 5 R., Lawrence, M. G., Pham, M., Reddy, M. S., Sciare, J., and Venkataraman, C.: DMS
- 6 atmospheric concentrations and sulphate aerosol indirect radiative forcing: a sensitivity
- 7 study to the DMS source representation and oxidation, Atmos. Chem. Phys., 3, 49-65,
- 8 10.5194/acp-3-49-2003, 2003.
- 9 Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen,
- 10 V.-M., Kondo, Y., Liao, H., Lohmann, U., Rasch, P., Satheesh, S. K., Sherwood, S.,
- 11 Stevens, B., and Zhang, X. Y.: Clouds and Aerosols. In: Climate Change 2013: The
- 12 Physical Science Basis. , Contribution of Working Group I to the Fifth Assessment
- 13 Report of the Intergovernmental Panel on Climate Change, Cambridge University Press.
- 14 Cambridge, United Kingdom and New York, NY, USA, 571-657, 2013.
- 15 Bowman, K. W., Shindell, D. T., Worden, H. M., Lamarque, J. F., Young, P. J.,
- 16 Stevenson, D. S., Qu, Z., de la Torre, M., Bergmann, D., Cameron-Smith, P. J., Collins,
- 17 W. J., Doherty, R., Dalsoren, S. B., Faluvegi, G., Folberth, G., Horowitz, L. W., Josse, B.
- 18 M., Lee, Y. H., MacKenzie, I. A., Myhre, G., Nagashima, T., Naik, V., Plummer, D. A.,
- 19 Rumbold, S. T., Skeie, R. B., Strode, S. A., Sudo, K., Szopa, S., Voulgarakis, A., Zeng,
- G., Kulawik, S. S., Aghedo, A. M., and Worden, J. R.: Evaluation of ACCMIP outgoing
 longwave radiation from tropospheric ozone using TES satellite observations.
- 22 Atmospheric Chemistry and Physics, 13, 4057-4072, 10.5194/acp-13-4057-2013, 2013.
- 23 Cakmur, R. V., Miller, R. L., Perlwitz, J., Geogdzhayev, I. V., Ginoux, P., Koch, D.,
- 24 Kohfeld, K. E., Tegen, I., and Zender, C. S.: Constraining the magnitude of the global
- 25 dust cycle by minimizing the difference between a model and observations, J. Geophys.
- 26 Res.-Atmos., 111, D06207, doi:10.1029/2005jd005791, 2006.
- 27 Clarke, A. D., and Kapustin, V. N.: A pacific aerosol survey. Part I: A decade of data on
- 28 particle production, transport, evolution, and mixing in the troposphere, Journal of the
- 29 Atmospheric Sciences, 59, 363-382, 10.1175/1520-
- 30 0469(2002)059<0363:apaspi>2.0.co;2, 2002.
- 31 Collaud Coen, M., Weingartner, E., Furger, M., Nyeki, S., Prévôt, A. S. H., Steinbacher,
- 32 M., and Baltensperger, U.: Aerosol climatology and planetary boundary influence at the
- 33 Jungfraujoch analyzed by synoptic weather types, Atmos. Chem. Phys., 11, 5931-5944,
- 34 10.5194/acp-11-5931-2011, 2011.
- 35 Debell, L. J., Gebhart, K. A., Malm, W. C., Pitchford, M. L., Schichtel, B. A., and White,
- 36 W. H.: Spatial and seasonal patterns and temporal variability of haze
- and its constituents in the United States: report IV, 2006.
- 38 Del Genio, A. D., Yao, M. S., Kovari, W., and Lo, K. K. W.: A prognostic cloud water
- 39 parameterization for global climate models, J Climate, 9, 270-304, 1996.
- 40 Del Genio, A. D., and Yao, M-S: Efficient cumulus parameterization for long-term
- 41 climate studies: The GISS scheme, The Representation of Cumulus Convection in
- 42 Numerical Models, American Meteorological Society, Boston, Mass., 181-184 pp.,
- 43 1993.
- 44 Dentener, F., Kinne, S., Bond, T., Boucher, O., Cofala, J., Generoso, S., Ginoux, P.,
- 45 Gong, S., Hoelzemann, J. J., Ito, A., Marelli, L., Penner, J. E., Putaud, J. P., Textor, C.,
- 46 Schulz, M., van der Werf, G. R., and Wilson, J.: Emissions of primary aerosol and

- 1 precursor gases in the years 2000 and 1750 prescribed data-sets for AeroCom,
- 2 Atmospheric Chemistry and Physics, 6, 4321-4344, 2006.
- 3 Dick, W. D., Saxena, P., and McMurry, P. H.: Estimation of water uptake by organic
- 4 compounds in submicron aerosols measured during the Southeastern Aerosol and
- 5 Visibility Study, J. Geophys. Res.-Atmos., 105, 1471-1479, 10.1029/1999jd901001,
- 6 2000.
- 7 Diner, D. J., Beckert, J. C., Reilly, T. H., Bruegge, C. J., Conel, J. E., Kahn, R. A.,
- 8 Martonchik, J. V., Ackerman, T. P., Davies, R., Gerstl, S. A. W., Gordon, H. R., Muller,
- 9 J. P., Myneni, R. B., Sellers, P. J., Pinty, B., and Verstraete, M. M.: Multi-angle Imaging
- 10 SpectroRadiometer (MISR) Instrument description and experiment overview, Ieee
- Transactions on Geoscience and Remote Sensing, 36, 1072-1087, 10.1109/36.700992,
 1998.
- 13 Duce, R. A.: Sources, distributions, and fluxes of mineral aerosols and their relationship
- to climate, Aerosol Forcing of Climate, edited by: Charlson, R. J., and Heintzenberg, J.,
 43-72 pp., 1995.
- 16 Easter, R. C., Ghan, S. J., Zhang, Y., Saylor, R. D., Chapman, E. G., Laulainen, N. S.,
- 17 Abdul-Razzak, H., Leung, L. R., Bian, X. D., and Zaveri, R. A.: MIRAGE: Model
- 18 description and evaluation of aerosols and trace gases, J. Geophys. Res.-Atmos., 109,
- 19 D20210, doi:10.1029/2004JD004571, 2004.
- 20 Emmons, L. K., Hauglustaine, D. A., Muller, J. F., Carroll, M. A., Brasseur, G. P.,
- Brunner, D., Staehelin, J., Thouret, V., and Marenco, A.: Data composites of airborne
 observations of tropospheric ozone and its precursors, J. Geophys. Res.-Atmos., 105,
- 23 20497-20538, 10.1029/2000jd900232, 2000.
- Feingold, G., Tzivion, S., and Levin, Z.: EVOLUTION OF RAINDROP SPECTRA .1.
- 25 SOLUTION TO THE STOCHASTIC COLLECTION BREAKUP EQUATION USING
- 26 THE METHOD OF MOMENTS, Journal of the Atmospheric Sciences, 45, 3387-3399,
- 27 10.1175/1520-0469(1988)045<3387:eorspi>2.0.co;2, 1988.
- 28 Forster, P., and Ramaswamy, V.: Changes in Atmospheric Constituents and in Radiative
- 29 Forcing, Climate Change 2007: The Physical Science Basis, edited by: Solomon, S., Qin,
- 30 D., Manning, M., Marquis, M., Averyt, K., Tignor, M. M. B., Miller, H. L., and Chen, Z.
 31 L., 129-234 pp., 2007.
- 32 Ginoux, P., Chin, M., Tegen, I., Prospero, J. M., Holben, B., Dubovik, O., and Lin, S. J.:
- 33 Sources and distributions of dust aerosols simulated with the GOCART model, J.
- 34 Geophys. Res.-Atmos., 106, 20255-20273, 2001.
- 35 Gong, S. L.: A parameterization of sea-salt aerosol source function for sub- and super-
- micron particles, Global Biogeochemical Cycles, 17, 1097, doi:10.1029/2003gb002079,
 2003.
- 38 Hand, J. L., Copeland, S. A., Day, D. E., Dillner, A. M., Indresand, H., Malm, W. C.,
- 39 McDade, C. E., Moore, C. T., Pitchford, M. L., Schichtel, B. A., and Watson, J. G.:
- 40 Spatial and Seasonal Patterns and Temporal Variability of Haze and its Constituents in
- 41 the United States: Report V, Colo. State Univ., Fort Collins, 2011.
- 42 Hansen, J., Russell, G., Rind, D., Stone, P., Lacis, A., Lebedeff, S., Ruedy, R., and
- 43 Travis, L.: Efficient 3-Dimensional Global-Models for Climate Studies Model-I and
- 44 Model-Ii, Mon Weather Rev, 111, 609-662, 1983.

- 1 Harrington, D. Y., and Kreidenweis, S. M.: Simulations of sulfate aerosol dynamics Part
- II. Model intercomparison, Atmospheric Environment, 32, 1701-1709, 10.1016/S1352 2310(97)00453-6, 1998.
- 4 Heintzenberg, J., Covert, D. C., and Van Dingenen, R.: Size distribution and chemical
- 5 composition of marine aerosols: a compilation and review, Tellus Series B-Chemical and
- 6 Physical Meteorology, 52, 1104-1122, 10.1034/j.1600-0889.2000.00136.x, 2000.
- 7 Hoell, J. M., Davis, D. D., Jacob, D. J., Rodgers, M. O., Newell, R. E., Fuelberg, H. E.,
- 8 McNeal, R. J., Raper, J. L., and Bendura, R. J.: Pacific Exploratory Mission in the
- 9 tropical Pacific: PEM-Tropics A, August-September 1996, J. Geophys. Res.-Atmos., 104,
- 10 5567-5583, 10.1029/1998jd100074, 1999.
- 11 Holben, B. N., Eck, T. F., Slutsker, I., Tanre, D., Buis, J. P., Setzer, A., Vermote, E.,
- 12 Reagan, J. A., Kaufman, Y. J., Nakajima, T., Lavenu, F., Jankowiak, I., and Smirnov, A.:
- 13 AERONET A federated instrument network and data archive for aerosol
- 14 characterization, Remote Sensing of Environment, 66, 1-16, 10.1016/s0034-
- 15 4257(98)00031-5, 1998.
- 16 Holben, B. N., Tanre, D., Smirnov, A., Eck, T. F., Slutsker, I., Abuhassan, N., Newcomb,
- 17 W. W., Schafer, J. S., Chatenet, B., Lavenu, F., Kaufman, Y. J., Castle, J. V., Setzer, A.,
- 18 Markham, B., Clark, D., Frouin, R., Halthore, R., Karneli, A., O'Neill, N. T., Pietras, C.,
- 19 Pinker, R. T., Voss, K., and Zibordi, G.: An emerging ground-based aerosol climatology:
- Aerosol optical depth from AERONET, J. Geophys. Res.-Atmos., 106, 12067-12097,
- 21 10.1029/2001jd900014, 2001.
- 22 Hsu, N. C., Tsay, S. C., King, M. D., and Herman, J. R.: Deep blue retrievals of Asian
- aerosol properties during ACE-Asia, Ieee Transactions on Geoscience and Remote
 Sensing, 44, 3180-3195, 10.1109/tgrs.2006.879540, 2006.
- 25 Jourdain, B., and Legrand, M.: Seasonal variations of atmospheric dimethylsulfide,
- 26 dimethylsulfoxide, sulfur dioxide, methanesulfonate, and non-sea-salt sulfate aerosols at
- 27 Dumont d'Urville (coastal Antarctica) (December 1998 to July 1999), J. Geophys. Res.-
- 28 Atmos., 106, 14391-14408, 10.1029/2000jd900841, 2001.
- Jung, J. G., Adams, P. J., and Pandis, S. N.: Simulating the size distribution and chemical
- composition of ultrafine particles during nucleation events, Atmospheric Environment,
 40, 2248-2259, 2006.
- 32 Jung, J. G., Fountoukis, C., Adams, P. J., and Pandis, S. N.: Simulation of in situ ultrafine
- 33 particle formation in the eastern United States using PMCAMx-UF, J. Geophys. Res.-
- 34 Atmos., 115, D03203, doi: 10.1029/2009jd012313, 2010.
- 35 Kahn, R. A., Gaitley, B. J., Martonchik, J. V., Diner, D. J., Crean, K. A., and Holben, B.:
- 36 Multiangle Imaging Spectroradiometer (MISR) global aerosol optical depth validation
- 37 based on 2 years of coincident Aerosol Robotic Network (AERONET) observations, J.
- 38 Geophys. Res.-Atmos., 110, D10s04
- 39 10.1029/2004jd004706, 2005.
- 40 Kettle, A. J., Andreae, M. O., Amouroux, D., Andreae, T. W., Bates, T. S., Berresheim,
- 41 H., Bingemer, H., Boniforti, R., Curran, M. A. J., DiTullio, G. R., Helas, G., Jones, G. B.,
- 42 Keller, M. D., Kiene, R. P., Leck, C., Levasseur, M., Malin, G., Maspero, M., Matrai, P.,
- 43 McTaggart, A. R., Mihalopoulos, N., Nguyen, B. C., Novo, A., Putaud, J. P.,
- 44 Rapsomanikis, S., Roberts, G., Schebeske, G., Sharma, S., Simo, R., Staubes, R., Turner,
- 45 S., and Uher, G.: A global database of sea surface dimethylsulfide (DMS) measurements

- 1 and a procedure to predict sea surface DMS as a function of latitude, longitude, and
- 2 month, Global Biogeochemical Cycles, 13, 399-444, 10.1029/1999gb900004, 1999.
- 3 Kettle, A. J., and Andreae, M. O.: Flux of dimethylsulfide from the oceans: A
- 4 comparison of updated data sets and flux models, Journal of Geophysical Research:
- 5 Atmospheres, 105, 26793-26808, 10.1029/2000JD900252, 2000.
- 6 Koch, D., Jacob, D., Tegen, I., Rind, D., and Chin, M.: Tropospheric sulfur simulation
- 7 and sulfate direct radiative forcing in the Goddard Institute for Space Studies general
- 8 circulation model, J. Geophys. Res.-Atmos., 104, 23799-23822, 10.1029/1999jd900248,
 9 1999.
- 10 Koch, D., Schmidt, G. A., and Field, C. V.: Sulfur, sea salt, and radionuclide aerosols in
- 11 GISS ModelE, J. Geophys. Res.-Atmos., 111, D06206, doi:10.1029/2004jd005550, 2006.
- 12 Koch, D., Bond, T. C., Streets, D., Unger, N., and van der Werf, G. R.: Global impacts of
- 13 aerosols from particular source regions and sectors, J. Geophys. Res.-Atmos., 112,
- 14 D02205, doi: 10.1029/2005jd007024, 2007.
- 15 Koch, D., Bauer, S. E., Del Genio, A., Faluvegi, G., McConnell, J. R., Menon, S., Miller,
- 16 R. L., Rind, D., Ruedy, R., Schmidt, G. A., and Shindell, D.: Coupled Aerosol-
- 17 Chemistry-Climate Twentieth-Century Transient Model Investigation: Trends in Short-
- 18 Lived Species and Climate Responses, J Climate, 24, 2693-2714,
- 19 10.1175/2011jcli3582.1, 2011.
- 20 Korhonen, H., Carslaw, K. S., Spracklen, D. V., Mann, G. W., and Woodhouse, M. T.:
- 21 Influence of oceanic dimethyl sulfide emissions on cloud condensation nuclei
- 22 concentrations and seasonality over the remote Southern Hemisphere oceans: A global
- 23 model study, J. Geophys. Res.-Atmos., 113, D15204, doi: 10.1029/2007jd009718, 2008.
- 24 Lamarque, J.-F., Shindell, D. T., Josse, B., Eyring, V., Young, P. J., Cionni, I.,
- 25 Bergmann, D., Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi,
- 26 G., Folberth, G., Ghan, S. J., Horowitz, L. W., Lee, Y. H., McKenzie, I., Nagashima, T.,
- 27 Naik, V., Plummer, D., Rumbold, S., Skeie, R., Stevenson, D. S., Strode, S., Sudo, K.,
- 28 Szopa, S., Voulgarakis, A., and Zeng, G.: The Atmospheric Chemistry and Climate
- 29 Model Intercomparison Project (ACCMIP): Overview and description of models,
- 30 simulations and climate diagnostics, Geoscientific Model Development, accepted, 2012.
- 31 Lamarque, J. F., Dentener, F., McConnell, J., Ro, C. U., Shaw, M., Vet, R., Bergmann,
- 32 D., Cameron-Smith, P., Dalsoren, S., Doherty, R., Faluvegi, G., Ghan, S. J., Josse, B.,
- 33 Lee, Y. H., MacKenzie, I. A., Plummer, D., Shindell, D. T., Skeie, R. B., Stevenson, D.
- 34 S., Strode, S., Zeng, G., Curran, M., Dahl-Jensen, D., Das, S., Fritzsche, D., and Nolan,
- 35 M.: Multi-model mean nitrogen and sulfur deposition from the Atmospheric Chemistry
- 36 and Climate Model Intercomparison Project (ACCMIP): evaluation of historical and
- 37 projected future changes, Atmospheric Chemistry and Physics, 13, 7997-8018, 2013a.
- 38 Lamarque, J. F., Shindell, D. T., Josse, B., Young, P. J., Cionni, I., Eyring, V.,
- 39 Bergmann, D., Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi,
- 40 G., Folberth, G., Ghan, S. J., Horowitz, L. W., Lee, Y. H., MacKenzie, I. A., Nagashima,
- 41 T., Naik, V., Plummer, D., Righi, M., Rumbold, S. T., Schulz, M., Skeie, R. B.,
- 42 Stevenson, D. S., Strode, S., Sudo, K., Szopa, S., Voulgarakis, A., and Zeng, G.: The
- 43 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP):
- 44 overview and description of models, simulations and climate diagnostics, Geoscientific
- 45 Model Development, 6, 179-206, 10.5194/gmd-6-179-2013, 2013b.

- 1 Lauer, A., Hendricks, J., Ackermann, I., Schell, B., Hass, H., and Metzger, S.: Simulating
- 2 aerosol microphysics with the ECHAM/MADE GCM Part I: Model description and
- 3 comparison with observations, Atmospheric Chemistry and Physics, 5, 3251-3276, 2005.
- Lee, Y. H., Chen, K., and Adams, P. J.: Development of a global model of mineral dust
 aerosol microphysics, Atmospheric Chemistry and Physics, 9, 2441-2458, 2009.
- 5 aerosol microphysics, Atmospheric Chemistry and Physics, 9, 2441-2458, 2009.
- 6 Lee, Y. H., and Adams, P. J.: Evaluation of aerosol distributions in the GISS-TOMAS
- 7 global aerosol microphysics model with remote sensing observations, Atmospheric
- 8 Chemistry and Physics, 10, 2129-2144, 2010.
- 9 Lee, Y. H., and Adams, P. J.: A Fast and Efficient Version of the TwO-Moment Aerosol
- 10 Sectional (TOMAS) Global Aerosol Microphysics Model, Aerosol Science and
- 11 Technology, 46, 678-689, 10.1080/02786826.2011.643259, 2012.
- 12 Lee, Y. H., Lamarque, J. F., Flanner, M. G., Jiao, C., Shindell, D. T., Berntsen, T.,
- 13 Bisiaux, M. M., Cao, J., Collins, W. J., Curran, M., Edwards, R., Faluvegi, G., Ghan, S.,
- 14 Horowitz, L. W., McConnell, J. R., Ming, J., Myhre, G., Nagashima, T., Naik, V.,
- 15 Rumbold, S. T., Skeie, R. B., Sudo, K., Takemura, T., Thevenon, F., Xu, B., and Yoon, J.
- 16 H.: Evaluation of preindustrial to present-day black carbon and its albedo forcing from
- 17 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP),
- 18 Atmospheric Chemistry and Physics, 13, 2607-2634, 10.5194/acp-13-2607-2013, 2013a.
- 19 Lee, Y. H., Pierce, J. R., and Adams, P. J.: Representation of nucleation mode
- 20 microphysics in a global aerosol model with sectional microphysics, Geoscientific Model
- 21 Development, 6, 1221-1232, 2013b.
- 22 Liss, P. S., and Merlivat, L.: Air-sea gas exchange rates: Introduction and synthesis, The
- 23 Role of Air-Sea Exchange in Geochemical Cycling, Springer, New York, 1986.
- 24 Liu, X. H., Penner, J. E., and Herzog, M.: Global modeling of aerosol dynamics: Model
- 25 description, evaluation, and interactions between sulfate and nonsulfate aerosols, J.
- 26 Geophys. Res.-Atmos., 110, D18206, doi: 10.1029/2004jd005674, 2005.
- 27 Lohmann, U., and Feichter, J.: Global indirect aerosol effects: a review, Atmospheric
- 28 Chemistry and Physics, 5, 715-737, 2005.
- 29 Mahowald, N. M., Engelstaedter, S., Luo, C., Sealy, A., Artaxo, P., Benitez-Nelson, C.,
- 30 Bonnet, S., Chen, Y., Chuang, P. Y., Cohen, D. D., Dulac, F., Herut, B., Johansen, A. M.,
- 31 Kubilay, N., Losno, R., Maenhaut, W., Paytan, A., Prospero, J. A., Shank, L. M., and
- 32 Siefert, R. L.: Atmospheric Iron Deposition: Global Distribution, Variability, and Human
- 33 Perturbations, in: Annual Review of Marine Science, Annual Review of Marine Science,
- 34 245-278, 2009.
- 35 Mann, G. W., Carslaw, K. S., Spracklen, D. V., Ridley, D. A., Manktelow, P. T.,
- 36 Chipperfield, M. P., Pickering, S. J., and Johnson, C. E.: Description and evaluation of
- 37 GLOMAP-mode: a modal global aerosol microphysics model for the UKCA
- 38 composition-climate model, Geoscientific Model Development, 3, 519-551,
- 39 10.5194/gmd-3-519-2010, 2010.
- 40 Menon, S., Del Genio, A. D., Koch, D., and Tselioudis, G.: GCM Simulations of the
- 41 aerosol indirect effect: Sensitivity to cloud parameterization and aerosol burden, Journal
- 42 of the Atmospheric Sciences, 59, 692-713, 10.1175/1520-
- 43 0469(2002)059<0692:gsotai>2.0.co;2, 2002.
- 44 Menon, S., Del Genio, A. D., Kaufman, Y., Bennartz, R., Koch, D., Loeb, N., and
- 45 Orlikowski, D.: Analyzing signatures of aerosol-cloud interactions from satellite

- 1 retrievals and the GISS GCM to constrain the aerosol indirect effect, J. Geophys. Res.-
- 2 Atmos., 113, D14s22
- 3 10.1029/2007jd009442, 2008.
- 4 Merikanto, J., Spracklen, D. V., Mann, G. W., Pickering, S. J., and Carslaw, K. S.:
- Impact of nucleation on global CCN, Atmospheric Chemistry and Physics, 9, 8601-8616,
 2009.
- 7 Miller, R. L., Cakmur, R. V., Perlwitz, J., Geogdzhayev, I. V., Ginoux, P., Koch, D.,
- 8 Kohfeld, K. E., Prigent, C., Ruedy, R., Schmidt, G. A., and Tegen, I.: Mineral dust
- 9 aerosols in the NASA goddard institute for Space Sciences ModelE atmospheric general
- 10 circulation model, J. Geophys. Res.-Atmos., 111, D06208, doi: 10.1029/2005jd005796,
- 11 2006.
- 12 Mishchenko, M. I., Travis, L. D., and Mackowski, D. W.: T-matrix computations of light
- 13 scattering by nonspherical particles: A review, Journal of Quantitative Spectroscopy &
- 14 Radiative Transfer, 55, 535-575, 10.1016/0022-4073(96)00002-7, 1996.
- 15 Monahan, E. C., Spiel, D. E., and David, K. L.: A model of marine aerosol generation via
- 16 whitecaps and wave disruption, in Oceanic Whitecaps, edited by: Monahan, E. C., and
- 17 Mac Niocaill, G., Oceanographic Sciences Library, 167–174 pp., 1986.
- 18 Myhre, G., Samset, B. H., Schulz, M., Balkanski, Y., Bauer, S., Berntsen, T. K., Bian, H.,
- 19 Bellouin, N., Chin, M., Diehl, T., Easter, R. C., Feichter, J., Ghan, S. J., Hauglustaine, D.,
- 20 Iversen, T., Kinne, S., Kirkev, eg, A., Lamarque, J. F., Lin, G., Liu, X., Lund, M. T., Luo,
- 21 G., Ma, X., van Noije, T., Penner, J. E., Rasch, P. J., Ruiz, A., Seland, o., Skeie, R. B.,
- 22 Stier, P., Takemura, T., Tsigaridis, K., Wang, P., Wang, Z., Xu, L., Yu, H., Yu, F., Yoon,
- 23 J. H., Zhang, K., Zhang, H., and Zhou, C.: Radiative forcing of the direct aerosol effect
- 24 from AeroCom Phase II simulations, Atmos. Chem. Phys., 13, 1853-1877, 10.5194/acp-
- 25 13-1853-2013, 2013a.
- 26 Myhre, G., Shindell, D., Brèon, F.-M., Collins, W., Fuglestvedt, J., Huang, J., Koch, D.,
- 27 Lamarque, J. F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G.,
- 28 Takemura, T., and Zhang, H.: Anthropogenic and Natural Radiative Forcing. In: Climate
- 29 Change 2013: The Physical Science Basis. , Contribution of Working Group I to the Fifth
- 30 Assessment
- 31 Report of the Intergovernmental Panel on Climate Change, Cambridge University Press.
- 32 Cambridge, United Kingdom and New York, NY, USA, 2013b.
- 33 Nabat, P., Somot, S., Mallet, M., Chiapello, I., Morcrette, J. J., Solmon, F., Szopa, S.,
- 34 Dulac, F., Collins, W., Ghan, S., Horowitz, L. W., Lamarque, J. F., Lee, Y. H., Naik, V.,
- 35 Nagashima, T., Shindell, D., and Skeie, R.: A 4-D climatology (1979-2009) of the
- 36 monthly tropospheric aerosol optical depth distribution over the Mediterranean region
- 37 from a comparative evaluation and blending of remote sensing and model products,
- 38 Atmospheric Measurement Techniques, 6, 1287-1314, 10.5194/amt-6-1287-2013, 2013.
- 39 Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L. W., Lamarque, J. F., Lin, M.,
- 40 Prather, M. J., Young, P. J., Bergmann, D., Cameron-Smith, P. J., Cionni, I., Collins, W.
- 41 J., Dalsoren, S. B., Doherty, R., Eyring, V., Faluvegi, G., Folberth, G. A., Josse, B., Lee,
- 42 Y. H., MacKenzie, I. A., Nagashima, T., van Noije, T. P. C., Plummer, D. A., Righi, M.,
- 43 Rumbold, S. T., Skeie, R., Shindell, D. T., Stevenson, D. S., Strode, S., Sudo, K., Szopa,
- 44 S., and Zeng, G.: Preindustrial to present-day changes in tropospheric hydroxyl radical
- 45 and methane lifetime from the Atmospheric Chemistry and Climate Model

- 1 Intercomparison Project (ACCMIP), Atmospheric Chemistry and Physics, 13, 5277-
- 2 5298, 10.5194/acp-13-5277-2013, 2013.
- 3 Napari, I., Noppel, M., Vehkamaki, H., and Kulmala, M.: Parametrization of ternary
- 4 nucleation rates for H2SO4-NH3-H2O vapors, J. Geophys. Res.-Atmos., 107, 4381, doi:
 5 10.1029/2002JD002132, 2002.
- 6 Nenes, A., Pandis, S. N., and Pilinis, C.: ISORROPIA: A New Thermodynamic
- 7 Equilibrium Model for Multiphase Multicomponent Inorganic Aerosols, Aquatic
- 8 Geochemistry, 4, 123-152, 10.1023/A:1009604003981, 1998.
- 9 Nguyen, B. C., Mihalopoulos, N., Putaud, J. P., Gaudry, A., Gallet, L., Keene, W. C., and
- 10 Galloway, J. N.: COVARIATIONS IN OCEANIC DIMETHYL SULFIDE, ITS
- 11 OXIDATION-PRODUCTS AND RAIN ACIDITY AT AMSTERDAM ISLAND IN
- 12 THE SOUTHERN INDIAN-OCEAN, Journal of Atmospheric Chemistry, 15, 39-53,
- 13 10.1007/bf00053608, 1992.
- 14 Nightingale, P. D., Malin, G., Law, C. S., Watson, A. J., Liss, P. S., Liddicoat, M. I.,
- 15 Boutin, J., and Upstill-Goddard, R. C.: In situ evaluation of air-sea gas exchange
- 16 parameterizations using novel conservative and volatile tracers, Global Biogeochemical
- 17 Cycles, 14, 373-387, 10.1029/1999gb900091, 2000.
- 18 Petzold, A., Fiebig, M., Flentje, H., Keil, A., Leiterer, U., Schroder, F., Stifter, A.,
- 19 Wendisch, M., and Wendling, P.: Vertical variability of aerosol properties observed at a
- 20 continental site during the Lindenberg Aerosol Characterization Experiment (LACE 98),
- 21 J. Geophys. Res.-Atmos., 107, 8128, 8128, doi: 10.1029/2001jd001043, 2002.
- 22 Pierce, J. R., and Adams, P. J.: Global evaluation of CCN formation by direct emission of
- sea salt and growth of ultrafine sea salt, J. Geophys. Res.-Atmos., 111, D06203,
- 24 doi:10.1029/2005JD006186, 2006.
- 25 Pierce, J. R., Chen, K., and Adams, P. J.: Contribution of primary carbonaceous aerosol
- 26 to cloud condensation nuclei: processes and uncertainties evaluated with a global aerosol
- 27 microphysics model, Atmospheric Chemistry and Physics, 7, 5447-5466, 2007.
- 28 Pierce, J. R., and Adams, P. J.: A Computationally Efficient Aerosol
- 29 Nucleation/Condensation Method: Pseudo-Steady-State Sulfuric Acid, Aerosol Science
- 30 and Technology, 43, 216-226, 2009a.
- 31 Pierce, J. R., and Adams, P. J.: Uncertainty in global CCN concentrations from uncertain
- aerosol nucleation and primary emission rates, Atmospheric Chemistry and Physics, 9,
 1339-1356, 2009b.
- 34 Prather, M. J.: Numerical Advection by Conservation of 2nd-Order Moments, J.
- 35 Geophys. Res.-Atmos., 91, 6671-6681, 1986.
- 36 Prospero, J. M., and Bonatti, E.: CONTINENTAL DUST IN ATMOSPHERE OF
- 37 EASTERN EQUATORIAL PACIFIC, Journal of Geophysical Research, 74, 3362-3371,
- 38 doi: 10.1029/JC074i013p03362, 1969.
- 39 Prospero, J. M., Bullard, J. E., and Hodgkins, R.: High-Latitude Dust Over the North
- 40 Atlantic: Inputs from Icelandic Proglacial Dust Storms, Science, 335, 1078-1082,
- 41 10.1126/science.1217447, 2012.
- 42 Putaud, J. P.: A European aerosol phenomenology; physical and chemical characteristics
- of particulate matter at kerbside, urban, rural and background sites in Europe, European
 Commission, EUR 20411 EN, 2003.
- 45 Putaud, J. P., Van Dingenen, R., Alastuey, A., Bauer, H., Birmili, W., Cyrys, J., Flentje,
- 46 H., Fuzzi, S., Gehrig, R., Hansson, H. C., Harrison, R. M., Herrmann, H., Hitzenberger,

- 1 R., Hueglin, C., Jones, A. M., Kasper-Giebl, A., Kiss, G., Kousa, A., Kuhlbusch, T. A. J.,
- 2 Loeschau, G., Maenhaut, W., Molnar, A., Moreno, T., Pekkanen, J., Perrino, C., Pitz, M.,
- 3 Puxbaum, H., Querol, X., Rodriguez, S., Salma, I., Schwarz, J., Smolik, J., Schneider, J.,
- 4 Spindler, G., ten Brink, H., Tursic, J., Viana, M., Wiedensohler, A., and Raes, F.: A
- 5 European aerosol phenomenology-3: Physical and chemical characteristics of particulate
- 6 matter from 60 rural, urban, and kerbside sites across Europe, Atmospheric Environment,
- 7 44, 1308-1320, 10.1016/j.atmosenv.2009.12.011, 2010.
- 8 Raper, J. L., Kleb, M. M., Jacob, D. J., Davis, D. D., Newell, R. E., Fuelberg, H. E.,
- 9 Bendura, R. J., Hoell, J. M., and McNeal, R. J.: Pacific Exploratory Mission in the
- 10 tropical Pacific: PEM-Tropics B, March-April 1999, J. Geophys. Res.-Atmos., 106,
- 11 32401-32425, 10.1029/2000jd900833, 2001.
- 12 Remer, L. A., Kleidman, R. G., Levy, R. C., Kaufman, Y. J., Tanre, D., Mattoo, S.,
- 13 Martins, J. V., Ichoku, C., Koren, I., Yu, H., and Holben, B. N.: Global aerosol
- 14 climatology from the MODIS satellite sensors, J. Geophys. Res.-Atmos., 113, D14s07
- 15 10.1029/2007jd009661, 2008.
- 16 Rienecker, M. M., Suarez, M. J., Gelaro, R., Todling, R., Bacmeister, J., Liu, E.,
- 17 Bosilovich, M. G., Schubert, S. D., Takacs, L., Kim, G.-K., Bloom, S., Chen, J., Collins,
- 18 D., Conaty, A., da Silva, A., Gu, W., Joiner, J., Koster, R. D., Lucchesi, R., Molod, A.,
- 19 Owens, T., Pawson, S., Pegion, P., Redder, C. R., Reichle, R., Robertson, F. R., Ruddick,
- 20 A. G., Sienkiewicz, M., and Woollen, J.: MERRA: NASA's Modern-Era Retrospective
- Analysis for Research and Applications, J Climate, 24, 3624-3648, 10.1175/JCLI-D-11 00015.1, 2011.
- 23 Savoie, D. L., and Prospero, J. M.: COMPARISON OF OCEANIC AND
- 24 CONTINENTAL SOURCES OF NON-SEA-SALT SULFATE OVER THE PACIFIC-
- 25 OCEAN, Nature, 339, 685-687, 10.1038/339685a0, 1989.
- 26 Schmidt, G. A., Ruedy, R., Hansen, J. E., Aleinov, I., Bell, N., Bauer, M., Bauer, S.,
- 27 Cairns, B., Canuto, V., Cheng, Y., Del Genio, A., Faluvegi, G., Friend, A. D., Hall, T.
- 28 M., Hu, Y. Y., Kelley, M., Kiang, N. Y., Koch, D., Lacis, A. A., Lerner, J., Lo, K. K.,
- 29 Miller, R. L., Nazarenko, L., Oinas, V., Perlwitz, J., Perlwitz, J., Rind, D., Romanou, A.,
- 30 Russell, G. L., Sato, M., Shindell, D. T., Stone, P. H., Sun, S., Tausnev, N., Thresher, D.,
- and Yao, M. S.: Present-day atmospheric simulations using GISS ModelE: Comparison
- 32 to in situ, satellite, and reanalysis data, J Climate, 19, 153-192, 10.1175/jcli3612.1, 2006.
- 33 Schmidt, G. A., Kelley, M., Nazarenko, L., Ruedy, R., Russell, G. L., Aleinov, I., Bauer,
- 34 M., Bauer, S. E., Bhat, M. K., Bleck, R., Canuto, V., Chen, Y.-H., Cheng, Y., Clune, T.
- 35 L., Del Genio, A., de Fainchtein, R., Faluvegi, G., Hansen, J. E., Healy, R. J., Kiang, N.
- 36 Y., Koch, D., Lacis, A. A., LeGrande, A. N., Lerner, J., Lo, K. K., Matthews, E. E.,
- 37 Menon, S., Miller, R. L., Oinas, V., Oloso, A. O., Perlwitz, J. P., Puma, M. J., Putman,
- 38 W. M., Rind, D., Romanou, A., Sato, M., Shindell, D. T., Sun, S., Syed, R. A., Tausnev,
- 39 N., Tsigaridis, K., Unger, N., Voulgarakis, A., Yao, M.-S., and Zhang, J.: Configuration
- 40 and assessment of the GISS ModelE2 contributions to the CMIP5 archive, Journal of
- 41 Advances in Modeling Earth Systems, 6, 141-184, 10.1002/2013ms000265, 2014.
- 42 Seinfeld, J. H., and Pandis, S. N.: Atmospheric Chemistry and Physics, John Wiley and
- 43 Sons, New York, 1998.
- 44 Shindell, D., Faluvegi, G., Walsh, M., Anenberg, S. C., Van Dingenen, R., Muller, N. Z.,
- 45 Austin, J., Koch, D., and Milly, G.: Climate, health, agricultural and economic impacts of

- 1 tighter vehicle-emission standards, Nature Climate Change, 1, 59-66,
- 2 10.1038/nclimate1066, 2011.
- 3 Shindell, D. T., Faluvegi, G., Unger, N., Aguilar, E., Schmidt, G. A., Koch, D. M., Bauer,
- 4 S. E., and Miller, R. L.: Simulations of preindustrial, present-day, and 2100 conditions in
- 5 the NASA GISS composition and climate model G-PUCCINI, Atmospheric Chemistry
- 6 and Physics, 6, 4427-4459, 2006.
- 7 Shindell, D. T., Lamarque, J. F., Schulz, M., Flanner, M., Jiao, C., Chin, M., Young, P. J.,
- 8 Lee, Y. H., Rotstayn, L., Mahowald, N., Milly, G., Faluvegi, G., Balkanski, Y., Collins,
- 9 W. J., Conley, A. J., Dalsoren, S., Easter, R., Ghan, S., Horowitz, L., Liu, X., Myhre, G.,
- 10 Nagashima, T., Naik, V., Rumbold, S. T., Skeie, R., Sudo, K., Szopa, S., Takemura, T.,
- 11 Voulgarakis, A., Yoon, J. H., and Lo, F.: Radiative forcing in the ACCMIP historical and
- 12 future climate simulations, Atmospheric Chemistry and Physics, 13, 2939-2974,
- 13 10.5194/acp-13-2939-2013, 2013.
- 14 Sihto, S. L., Kulmala, M., Kerminen, V. M., Dal Maso, M., Petaja, T., Riipinen, I.,
- 15 Korhonen, H., Arnold, F., Janson, R., Boy, M., Laaksonen, A., and Lehtinen, K. E. J.:
- 16 Atmospheric sulphuric acid and aerosol formation: implications from atmospheric
- 17 measurements for nucleation and early growth mechanisms, Atmospheric Chemistry and 18
- Physics, 6, 4079-4091, 2006.
- 19 Singh, S., Adams, P. J., Misquitta, A., Lee, K. J., Lipsky, E. M., and Robinson, A. L.:
- 20 Computational Analysis of Particle Nucleation in Dilution Tunnels: Effects of Flow
- 21 Configuration and Tunnel Geometry, Aerosol Science and Technology, 48, 638-648, 22 10.1080/02786826.2014.910291, 2014.
- 23 Spracklen, D. V., Pringle, K. J., Carslaw, K. S., Chipperfield, M. P., and Mann, G. W.: A
- 24 global off-line model of size-resolved aerosol microphysics: I. Model development and
- 25 prediction of aerosol properties, Atmospheric Chemistry and Physics, 5, 2227-2252, 26 2005.
- 27 Spracklen, D. V., Carslaw, K. S., Merikanto, J., Mann, G. W., Reddington, C. L.,
- 28 Pickering, S., Ogren, J. A., Andrews, E., Baltensperger, U., Weingartner, E., Boy, M.,
- 29 Kulmala, M., Laakso, L., Lihavainen, H., Kivekas, N., Komppula, M., Mihalopoulos, N.,
- 30 Kouvarakis, G., Jennings, S. G., O'Dowd, C., Birmili, W., Wiedensohler, A., Weller, R.,
- 31 Gras, J., Laj, P., Sellegri, K., Bonn, B., Krejci, R., Laaksonen, A., Hamed, A., Minikin,
- 32 A., Harrison, R. M., Talbot, R., and Sun, J.: Explaining global surface aerosol number
- 33 concentrations in terms of primary emissions and particle formation, Atmospheric
- 34 Chemistry and Physics, 10, 4775-4793, 10.5194/acp-10-4775-2010, 2010.
- 35 Spracklen, D. V., Carslaw, K. S., Poeschl, U., Rap, A., and Forster, P. M.: Global cloud
- 36 condensation nuclei influenced by carbonaceous combustion aerosol, Atmospheric
- 37 Chemistry and Physics, 11, 9067-9087, 10.5194/acp-11-9067-2011, 2011.
- 38 Stevens, R. G., Pierce, J. R., Brock, C. A., Reed, M. K., Crawford, J. H., Holloway, J. S.,
- 39 Ryerson, T. B., Huey, L. G., and Nowak, J. B.: Nucleation and growth of sulfate aerosol
- 40 in coal-fired power plant plumes: sensitivity to background aerosol and meteorology,
- 41 Atmos. Chem. Phys., 12, 189-206, 10.5194/acp-12-189-2012, 2012.
- 42 Stevenson, D. S., Young, P. J., Naik, V., Lamarque, J. F., Shindell, D. T., Voulgarakis,
- 43 A., Skeie, R. B., Dalsoren, S. B., Myhre, G., Berntsen, T. K., Folberth, G. A., Rumbold,
- 44 S. T., Collins, W. J., MacKenzie, I. A., Doherty, R. M., Zeng, G., van Noije, T. P. C.,
- 45 Strunk, A., Bergmann, D., Cameron-Smith, P., Plummer, D. A., Strode, S. A., Horowitz,
- L., Lee, Y. H., Szopa, S., Sudo, K., Nagashima, T., Josse, B., Cionni, I., Righi, M., 46

- 1 Eyring, V., Conley, A., Bowman, K. W., Wild, O., and Archibald, A.: Tropospheric
- 2 ozone changes, radiative forcing and attribution to emissions in the Atmospheric
- 3 Chemistry and Climate Model Intercomparison Project (ACCMIP), Atmospheric
- 4 Chemistry and Physics, 13, 3063-3085, 10.5194/acp-13-3063-2013, 2013.
- 5 Stier, P., Feichter, J., Kinne, S., Kloster, S., Vignati, E., Wilson, J., Ganzeveld, L., Tegen,
- 6 I., Werner, M., Balkanski, Y., Schulz, M., Boucher, O., Minikin, A., and Petzold, A.: The
- 7 aerosol-climate model ECHAM5-HAM, Atmospheric Chemistry and Physics, 5, 1125-
- 8 1156, 2005.
- 9 Tegen, I., Harrison, S. P., Kohfeld, K., Prentice, I. C., Coe, M., and Heimann, M.: Impact
- 10 of vegetation and preferential source areas on global dust aerosol: Results from a model
- 11 study, J. Geophys. Res.-Atmos., 107, 4576, 4576, doi: 10.1029/2001jd000963, 2002.
- 12 Textor, C., Schulz, M., Guibert, S., Kinne, S., Balkanski, Y., Bauer, S., Berntsen, T.,
- 13 Berglen, T., Boucher, O., Chin, M., Dentener, F., Diehl, T., Easter, R., Feichter, H.,
- 14 Fillmore, D., Ghan, S., Ginoux, P., Gong, S., Kristjansson, J. E., Krol, M., Lauer, A.,
- 15 Lamarque, J. F., Liu, X., Montanaro, V., Myhre, G., Penner, J., Pitari, G., Reddy, S.,
- 16 Seland, O., Stier, P., Takemura, T., and Tie, X.: Analysis and quantification of the
- 17 diversities of aerosol life cycles within AeroCom, Atmospheric Chemistry and Physics, 6,
- 18 1777-1813, 2006.
- 19 Thornton, D. C., Bandy, A. R., Blomquist, B. W., Driedger, A. R., and Wade, T. P.:
- Sulfur dioxide distribution over the Pacific Ocean 1991–1996, Journal of Geophysical
 Research: Atmospheres, 104, 5845-5854, 10.1029/1998JD100048, 1999.
- Trivitayanurak, W., Adams, P. J., Spracklen, D. V., and Carslaw, K. S.: Tropospheric
- 22 aerosol microphysics simulation with assimilated meteorology: model description and
- intermodel comparison, Atmospheric Chemistry and Physics, 8, 3149-3168, 2008.
- 25 Tsigaridis, K., and Kanakidou, M.: Secondary organic aerosol importance in the future
- atmosphere, Atmospheric Environment, 41, 4682-4692, 10.1016/j.atmosenv.2007.03.045,
 2007.
- 28 Tsigaridis, K., Koch, D., and Menon, S.: Uncertainties and importance of sea spray
- 29 composition on aerosol direct and indirect effects, J. Geophys. Res.-Atmos., 118, 220-
- 30 235, 10.1029/2012jd018165, 2013.
- 31 Tzivion, S., Feingold, G., and Levin, Z.: An Efficient Numerical Solution to the
- Stochastic Collection Equation, Journal of the Atmospheric Sciences, 44, 3139-3149,1987.
- 34 Tzivion, S., Feingold, G., and Levin, Z.: THE EVOLUTION OF RAINDROP SPECTRA
- 35 .2. COLLISIONAL COLLECTION BREAKUP AND EVAPORATION IN A
- 36 RAINSHAFT, Journal of the Atmospheric Sciences, 46, 3312-3327, 10.1175/1520-
- 37 0469(1989)046<3312:teorsp>2.0.co;2, 1989.
- 38 Tzivion, S., Reisin, T. G., and Levin, Z.: A new formulation of the spectral multi-moment
- 39 method for calculating the kinetic collection equation: More accuracy with fewer bins,
- 40 Journal of Computational Physics, 171, 418-422, 10.1006/jcph.2001.6776, 2001.
- 41 Uematsu, M., Duce, R. A., and Prospero, J. M.: DEPOSITION OF ATMOSPHERIC
- 42 MINERAL PARTICLES IN THE NORTH PACIFIC-OCEAN, Journal of Atmospheric
- 43 Chemistry, 3, 123-138, 10.1007/bf00049372, 1985.
- 44 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P.
- 45 S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire emissions
- 46 and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997-

- 1 2009), Atmospheric Chemistry and Physics, 10, 11707-11735, 10.5194/acp-10-11707-
- 2 2010, 2010.
- 3 Vehkamaki, H., Kulmala, M., Napari, I., Lehtinen, K. E. J., Timmreck, C., Noppel, M.,
- 4 and Laaksonen, A.: An improved parameterization for sulfuric acid-water nucleation
- 5 rates for tropospheric and stratospheric conditions, J. Geophys. Res.-Atmos., 107, 4622,
- 6 DOI: 10.1029/2002JD002184, 2002.
- 7 Vignati, E., Wilson, J., and Stier, P.: M7: An efficient size-resolved aerosol microphysics
- 8 module for large-scale aerosol transport models, J. Geophys. Res.-Atmos., 109, D22202,
- 9 doi:10.1029/2003jd004485, 2004.
- 10 Wang, M., Ghan, S., Ovchinnikov, M., Liu, X., Easter, R., Kassianov, E., Qian, Y., and
- 11 Morrison, H.: Aerosol indirect effects in a multi-scale aerosol-climate model PNNL-
- 12 MMF, Atmospheric Chemistry and Physics, 11, 5431-5455, 10.5194/acp-11-5431-2011,
- 13 2011.
- 14 Wanninkhof, R.: Relationship between wind speed and gas exchange over the ocean,
- 15 Journal of Geophysical Research: Oceans, 97, 7373-7382, 10.1029/92jc00188, 1992.
- 16 Westervelt, D. M., Pierce, J. R., Riipinen, I., Trivitayanurak, W., Hamed, A., Kulmala,
- 17 M., Laaksonen, A., Decesari, S., and Adams, P. J.: Formation and growth of nucleated
- 18 particles into cloud condensation nuclei: model-measurement comparison, Atmos. Chem.
- 19 Phys., 13, 7645-7663, 10.5194/acp-13-7645-2013, 2013.
- 20 Young, P. J., Archibald, A. T., Bowman, K. W., Lamarque, J. F., Naik, V., Stevenson, D.
- 21 S., Tilmes, S., Voulgarakis, A., Wild, O., Bergmann, D., Cameron-Smith, P., Cionni, I.,
- 22 Collins, W. J., Dalsoren, S. B., Doherty, R. M., Eyring, V., Faluvegi, G., Horowitz, L.
- 23 W., Josse, B., Lee, Y. H., MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M.,
- 24 Rumbold, S. T., Skeie, R. B., Shindell, D. T., Strode, S. A., Sudo, K., Szopa, S., and
- 25 Zeng, G.: Pre-industrial to end 21st century projections of tropospheric ozone from the
- 26 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP),
- 27 Atmospheric Chemistry and Physics, 13, 2063-2090, 10.5194/acp-13-2063-2013, 2013.
- 28 Yu, F., and Luo, G.: Simulation of particle size distribution with a global aerosol model:
- 29 contribution of nucleation to aerosol and CCN number concentrations, Atmospheric
- 30 Chemistry and Physics, 9, 7691-7710, 2009.
- 31 Zhang, Y., Seigneur, C., Seinfeld, J. H., Jacobson, M. Z., and Binkowski, F. S.:
- 32 Simulation of aerosol dynamics: A comparative review of algorithms used in air quality
- 33 models, Aerosol Science and Technology, 31, 487-514, 1999.
- 34 35

1 Table 1. Aerosol and precursor gas emissions used in ModelE2-TOMAS and

2 ModelE2-OMA and the nucleation scheme used in the ModelE2-TOMAS simulations

3

Emission/Process	TOMAS model	Bulk model
Anthropogenic	CMIP5 2000 emissions	CMIP5 2000 emissions (Lamarque
emissions	(Lamarque et al., 2012)	et al., 2012)
Biomass burning	Climatological-average GFED3	Climatological-average GFED3
emissions	emissions from 1997 to 2009	emissions from 1997 to 2009 (van
	(van der Werf et al., 2010)	der Werf et al., 2010)
Primary sulfate	1.0% of total sulfur emissions	2.5% of total sulfur emissions
emission assumption		
DMS emission	Seawater DMS concentrations	Seawater DMS concentrations from
	from Kettle et al. (1999)	Kettle et al. (1999)
	Sea-to-air transfer function	Sea-to-air transfer function from
	from Liss and Merlivat (1986)	Nightingale et al. (2000)
Sea-salt emission	Gong et al. (2002); the upper	Gong et al. (2002); the upper
	diameter limit of 10 um	diameter limit of 8 um
Dust emission	See text for the details; the	See text for the details; the upper
	upper size diameter of 10 um	size diameter of 16 um
Nucleation	Three nucleation cases	N/A
	1. BASE - Binary nucleation	
	2. LowNUC - Binary	
	nucleation with 5 times lower	
	sulfuric acid concentrations	
	3. NoNUC – no nucleation	

4

5

6 Table 2. Emission size distributions assumed in ModelE2-TOMAS. *This is the soil

7 size assumption used in ModelE2-TOMAS, and the dust emission size distribution is

8 additionally influenced by meteorological variables. GMD stands for geometric mean

9 diameter, and GSD for geometric standard deviation.

Species	Emissions	Size assumptions
		Bimodal distribution
Sulfate	All emissions	GMD=10nm, GSD=1.6 (5% of total mass)
		GMD=70 nm, GSD=2.0 (95% of total mass)
EC and OC	Fossil fuel and Biofuel	GMD=60 nm, GSD=1.59
EC allu OC	Biomass burning	GMD=150 nm, GSD=1.59
Du at*	Clay	GMD=140 nm, GSD=2.0
Dust	Silt	GMD=1.15 μm, GSD=2.0
	Species Sulfate EC and OC Dust*	SpeciesEmissionsSulfateAll emissionsEC and OCFossil fuel and Biofuel Biomass burningDust*Clay Silt

1 Table 3. Global budgets for DMS and SO₂ from the BASE run in ModelE2-TOMAS and

2 ModelE2-OMA. The ModelE2-TOMAS values are presented before slash and the

3 ModelE2-OMA values are after slash. Values in parentheses are ranges from other

4 global models including Wang et al. (2011), Liu et al. (2005), and those listed in Liu

- 5 et al. (2005).
- 6

	DMS	SO2
	0.05 / 0.11	0.36 / 0.38
Burden [Tg S]	(0.02-0.15)	(0.2-0.69)
Total source [Tg S yr⁻¹]	16.1 / 28.7	80 / 90
Emission	16.1/ 28.7	65.6 / 64.7
EIIIISSIOII	(10.7-23.7)	(61.2-92.0)
Chemistry	-	14.4 / 25.3
Sink [Tg S yr ⁻¹]	16.1/ 28.7	80 / 89
Gas phase evidation		12.3 / 14.6
Gas-phase oxidation	16.1/ 28.7	(6.1-22.0)
Aquaque phase evidation		30.8 / 35.8
Aqueous-phase oxidation	-	(24.5-57.8)
Wat deposition		0.36 / 0.4
wei deposition	-	(0-19.9)
Dry denosition		37 / 38.8
Dry deposition	-	(15.78-55)
	1.2 / 1.5	1.9 / 1.5
Lifetime [days]	(0.5-3.0)	(0.6-2.6)

1 Table 4. Global aerosol budgets of the BASE run in ModelE2-TOMAS. Values in the

2 parentheses are the mean and normalized standard deviations obtained from Table

3 10 in Textor et al. (2006). The mass budgets for sulfate are presented as Tg S.

4

	Sulfate	Elemental carbon (EC)	Organic matter (OM)	Sea-salt	Dust
Burden [Tg]	0.67	0.19	1.2	3.6	9.1
Total source [Tg yr ⁻¹]	43.7	7.4	60.8	3231.9	705.8
Emission	0.66	7.4	43.7	3231.9	705.8
Wet deposition [Tg yr ⁻¹]	42.9	7.1	59.1	1046.9	336.8
By convective clouds [%]	27	24	24	54	29
Dry deposition [Tg yr ⁻¹]	0.8	0.3	1.6	2184.9	369.9
Lifatima [days]	5.6	9.6	7.2	0.4	4.7
	(4.1, 18%)	(7.1, 33%)	(6.5, 27%)	(0.5 <i>,</i> 58%)	(4.1, 43%)
Removal rate coefficient [day ⁻¹]	0.18	0.1	0.14	2.4	0.21
	(0.25, 18%)	(0.15, 21%)	(0.16, 24%)	(5.1, 188%)	(0.31, 62%)
Wet deposition	0.18	0.1	0.14	0.79	0.1
werdeposition	(0.22 <i>,</i> 22%)	(0.12, 31%)	(0.14, 32%)	(0.79, 77%)	(0.08, 42%)
Dry deposition	0.0032	0.004	0.0037	1.6	0.11
	(0.03 <i>,</i> 55%)	(0.03 <i>,</i> 55%)	(0.03 <i>,</i> 49%)	(4.3 <i>,</i> 219%)	(0.23 <i>,</i> 84%)
	98	96	97	32	48
wet/(wet+Dry)[%]	(89, 8%)	(79, 17%)	(80, 16%)	(31, 65%)	(33, 54%)

5 6

Table 5. Global aerosol budgets in ModelE2-OMA. Note that the sulfate and nitrate

7 budgets are presented as Tg S and Tg N, respectively.

	Sulfate	EC	ОМ	Sea-salt	Dust	Nitrate
Burden [Tg]	0.38	0.12	1.1	9.3	11.4	0.37
Total source [Tg yr-1]	52.0	7.4	58.5	2866.7	1071.8	21.0
Emission	1.7	7.4	43.7	2866.7	1071.8	
Wet deposition [Tg yr-1]	46.6	5.4	44.9	2059.1	407.7	17.7
By convective clouds [%]	21	37	32	39	52	29
Dry deposition [Tg yr-1]	5.4	2.0	13.7	806.9	664.1	3.3
Lifetime [days]	2.6	5.8	7.1	1.2	3.9	6.4
Removal rate coefficient [day-1]	0.39	0.17	0.14	0.84	0.26	0.16
Wet deposition	0.35	0.13	0.11	0.61	0.10	0.13
Dry deposition	0.04	0.046	0.033	0.24	0.16	0.025
Wet/(Wet+Dry) [%]	90	73	77	72	38	84

- Table 6. Statistical measures of model predictions compared to satellites. For ModelE2-OMA, the model prediction without nitrate aerosols is also presented in
- parentheses.

		Model vs. MODIS	Model vs. MISR	MODIS vs. MISR
ModelE2-	Correlation	0.63	0.73	0.79
TOMAS	NMB [%]	-29%	-34%	8%
ModelE2-	Correlation	0.45 (0.45)	0.52 (0.55)	0.79
OMA	NMB [%]	16% (-16%)	8% (-21%)	8%

	Sites	Longitude	Latitude	Years
1	Alta Floresta	56.0° W	9.9° S	1999-2005
2	Abracos Hill	62.0° W	11° S	1999-2005
3	Cuiaba-Miranda	56.0° W	15.7° S	2001-2005
4	Mongu	23.2° E	15.2° S	1995-2005
5	Ilorin	4.3° E	8.3° N	1998-2005
6	Banizombou	2.0° E	13.0° N	1995-2005
7	Capo Verde	22.9° W	16.7° N	1994-2004
8	Bidi Bahn	2.5° W	14.1° N	1996-1997
9	Barbados	59.5° W	13.2° N	1996-2000
10	Sede Boker	34.8° E	30.9° N	1998-2005
11	Bahrain	50.6° E	26.2° N	2004-2005
12	Solar Village	46.4° E	24.9° N	1999-2005
13	Dalanzadgad	104.4° E	43.6° N	1997-2005
14	Yulin	109.7° E	38.3° N	2001-2002
15	Sevilleta	106.9° W	34.4° N	1994-2005
16	Cart site	97.5° W	36.6° N	1996-2005
17	Bondville	88.4° W	40.1° N	1996-2005
18	GSFC	76.8° W	39.0° N	1995-2005
19	Mexico city	99.2° W	19.3° N	1999-2005
20	Ispra	8.6° E	45.8° N	2001-2005
21	Kanpur	80.3° E	26.5° N	2001-2005
22	Shirahama	135.4° E	33.7° N	2000-2005
23	Bermuda	64.7° W	32.4° N	1996-2005
24	Lanai	156.9° W	20.7° N	1996-2004
25	Dry Tortugas	82.9° W	24.6° N	1996-2003
26	Tahiti	149.6° W	17.6° S	1999-2005
27	Rottnest Island	115.5° E	32.0° N	2001-2004
28	Nauru	166.9° E	0.5° S	1999-2005

Table 7. Locations of AERONET sites and corresponding measurement time periods.

1 Table 8. Summary of global annual-average of tropospheric and surface-layer

2 aerosol number budgets including J3 (new particle formation rates at 3 nm), CN3

3 (number concentration of particles with $Dp \ge 3$ nm), $CN10(Dp \ge 10$ nm), CN100(Dp

 $4 \ge 100 \text{ nm}$) in ModelE2-TOMAS. Values normalized by tropospheric volume at 273 K

5 and 1 atm.

	Emission	Troposp	Tropos	Tropos	Tropos	Surface-	Surface-	Surface-
	rate	heric J3	pheric	pheric	pheric	layer	layer	layer
	$[cm^{-3} s^{-1}]$	$[cm^{-3} s^{-1}]$	CN3	CN10	CN100	CN3	CN10	CN100
			[cm-3]	[cm ⁻³]	[cm-3]	[cm-3]	[cm ⁻³]	[cm-3]
Base	5.47x10 ⁻⁴	0.131	4852	939	211	1622	1331	416
LowNUC	5.47 x10 ⁻⁴	0.013	1277	628	197	1152	1111	405
NoNUC	5.47 x10 ⁻⁴	0.000	222	221	159	935	919	374

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1 Figure 1. Annual-average column mass concentrations of (a) DMS and (b) SO₂ in the



4 5

6 Figure 2. Scatter plot of annual-mean surface SO_2 concentrations [µg m⁻³] for the

7 model (red for ModelE2-TOMAS and blue for ModelE2-OMA) compared to the

8 observations in the EMEP (European Monitoring and Evaluation Programme,

9 Loevblad et al., 2004; a) and CASTNET (Clean Air Status and Trends Network, Malm

10 et al., 2002; b) networks. Log-mean normalized bias (LMNB) and log-mean



- 1 Figure 3. Comparisons of monthly averaged surface DMS (a to c) and SO₂ (d to e)
- 2 mixing ratios [pptv] simulated (red for ModelE2-TOMAS and blue for ModelE2-
- 3 OMA) and measured (black) at Amsterdam Island [Nguyen et al., 1992] and Cape
- 4 Grim [Ayers et al., 1991]. Only DMS at Dumont D'Urville [Jourdain and Legrand,
- 5 2001]. Correlation (R) and normalized mean bias (NMB) are given.



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10 Figure 4. Comparison of DMS vertical profiles over the Pacific Ocean simulated (red

- 11 for ModelE2-TOMAS and blue for ModelE2-OMA) and observed (black solid line).
- 12 Observations are from PEM-Tropic-A (August-October 1996 in the tropical Pacific;
- 13 Christmas Island in a; Hawaii in c; Tahiti in e; Easter Island in g; Fiji in h) and PEM-
- 14 Tropic-B (March-April 1999 in the tropical Pacific; Christmas Island in b; Hawaii in
- d; Tahiti in f). The dashed lines represent 25th and 75th percentiles of the observed
 values.





Figure 6. Annual-mean concentrations of (a) sulfate, (b) EC, (c) OM, (d) sea-salt, and



Figure 7. Annually and zonally averaged concentrations of (a) sulfate, (b) EC, (c) OM,
 (d) sea-salt, and (e) dust in the ModelE2-TOMAS BASE run. Units are µg m⁻³.







9 converted to ammonium sulfate at the IMPROVE sites.



- 1 Figure 9. Comparisons of monthly averaged surface sulfate mass concentrations [μg
- 2 m⁻³] simulated (red for ModelE2-TOMAS and blue for ModelE2-OMA) and measured
- 3 (black solid line with an error bar showing a standard deviation) by University of
- 4 Miami. Correlation (R) and normalized mean bias (NMB) are provided only when











Figure 11. Same as Figure 9 but for sea-salt mass concentrations







- 1 Figure 14. Scatter plot of annual-average dust deposition fluxes [mg m⁻² yr⁻¹]
- 2 simulated and observations obtained from Ginoux et al. (2001), Tegen et al. (2002)
- and Tables S2 in Mahowald et al. (2009). Open symbols are for ModelE2-TOMAS and
- 4 the filled symbol for ModelE2-OMA. LMNB and LMNE are presented below the plot:
- 5 the first values are for ModelE2-TOMAS.



- Figure 15. Scatter plots of annually averaged surface EC and OM mass
- concentrations [µg m⁻³] simulated (red for ModelE2-TOMAS and blue for ModelE2-
- OMA) and observations from the IMPROVE network (a and b) and the European





ModelE2-TOMAS is shown in (a) and ModelE2-OMA in (b).



- 1 Figure 17. Global distributions of the annual-mean AOD from (a) ModelE2-TOMAS,
- 2 (b) ModelE2-OMA, (c) MISR, and (d) MODIS. See Section 5.4 for the details of the
- 3 MISR and MODIS AOD information. (a) ModelE2-TOMAS



0.00 0.05 0.10 0.15 0.20 0.25 0.30 0.35 0.40 0.45 0.50 0.55 0.60

- 1 Figure 18. Comparisons of monthly averaged model AOD (red for ModelE2-TOMAS
- 2 and blue for ModelE2-OMA) and AERONET AOD (black solid line). Correlation (R)

3 and normalized mean bias (nmb) are provided only when the observation is



Feb Apr Jun Aug Oct Dec

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Feb Apr Jun Aug Oct Dec

- 1 Figure 19. Annually averaged CN3 ($D_p \ge 3 \text{ nm}$), CN10 ($D_p \ge 10 \text{ nm}$), CN100 ($D_p \ge 100 \text{ nm}$)
- 2 nm) concentrations in the lowermost layer for the BASE run (a to c), the LowNUC
- 3 run (d to f), and the NoNUC run (g to i) of ModelE2-TOMAS. Units are cm⁻³. (a) BASE: CN3 (b) BASE: CN10 (c) BASE: CN100



CN3/CN10

0.	10.	100.	300.	500.	1000.	1500.	3000.	5000.	10000.	15000.	30000. [cm ⁻³]
CN100											
0.0	1.0	10.0	30.0	100.0	200.0	300.0	500.0	1000.0	2000.0	3000.0	5000,0 [cm ⁻³]

- 1 Figure 20. Annually and zonally averaged CN3 ($Dp \ge 3 \text{ nm}$), $CN10(Dp \ge 10 \text{ nm})$, and
- $CN100(Dp \ge 100 \text{ nm})$ concentrations for the BASE run (a to c), the LowNUC run (d
- 3 to f), and the NoNUC run (g to i) of ModelE2-TOMAS. Units are cm⁻³. (a) BASE: CN3 (b) BASE: CN10 (c) BASE: CN100



- 1 Figure 21. Scatter plot of simulated annual-mean aerosol number concentrations in
- 2 comparison with a dataset of surface number concentrations measurements at 36
- 3 sites around the world compiled by Spracklen et al. (2010). The top horizontal bar
- 4 represents the BASE results, and the middle bar the LowNUC results and the lower
- 5 bar the NoNUC results. Red color is for free troposphere (FT); blue for marine
- 6 boundary layer (MBL); green for continental boundary layer (CBL).





- 1 Figure 22. Annual cycle of aerosol number concentrations [cm⁻³ at ambient
- 2 conditions] at 32 sites. The observations are shown in black, and three model
- 3 results are presented: red for BASE, blue for LowNUC, and green for NoNUC. The
- 4 free tropospheric (FT) sites are from 1 to 8; the marine boundary layer (MBL) sites,
- 5 9 to 15; the continental boundary layer (CBL) sites,16 to 32. Correlation (R) and
- 6 normalized mean bias (NMB) are provided only when the observation is available
- 7 for 12 months.



- 1 Figure 23. Number size distributions from European sites during winter (DJF; a to f)
- 2 and summer (JJA; g to l) that are obtained from log-normal 3-mode fits during
- 3 morning (black solid), afternoon (black dotted), and night (black dashed). The
- 4 model results are the seasonal mean, shown in red lines for the BASE run, blue lines



5 for the LowNUC run, and green lines for the NoNUC run.

- 1 Figure 24. Vertical profiles of aerosol number concentrations from the observations
- 2 (black lines) and ModelE2-TOMAS (red lines for the BASE run, blue lines for the
- 3 LowNUC run, and green lines for the NoNUC run). The CN5, CN15, CN120
- concentrations (ambient conditions) from LACE campaign (Petzold et al., 2002, in 4
- 5 north-east Germany) are shown in a to c, respectively. The observed CN3
- concentrations (STP conditions: 1atm, 273 K) over the Pacific Ocean, which are 6
- averaged into the 3 latitude bands (70 S to 20S, 20S to 20N, and 20N to 70N; Clarke 7
- 8 and Kapustin, 2002), are shown in d to f. The dashed lines show the standard
- 9 deviation for the observations and the min/max monthly mean for the model (only
- 10 in d to f).


- 1 Figure 25. Simulated number concentrations in (a) Aitken mode (10 nm ≤ Dp < 100
- 2 nm) and (b) accumulation mode (100 nm \leq Dp < 1 μ m) compared to the
- 3 observations ("x" symbol with error bar representing minimum and maximum
- 4 observed concentrations) that were compiled and aggregated into 15-degree
- 5 latitude ranges (Heintzenberg et al. 2002). The model is also averaged to the 15-
- degree grid and is shown in red lines for the BASE run, blue lines for the LowNUC
 run, and green lines for the NoNUC run.



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15 Figure 26. Annually and zonally averaged aerosol size distributions in the marine

16 boundary layer. Observations are from Heintzenberg et al. (2000) and were

- compiled and aggregated into a 15° x 15° grid. The model is also averaged to the 15degree grid and is shown as red lines for the BASE run, blue lines for the LowNUC
- 19 run, and green lines for the NoNUC run.



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- Figure 27. Scatter plot of simulated CCN concentrations in comparison with a
- 25 dataset of CCN measurements compiled by Spracklen et al. (2011). Data is classified

- 1 into two categories: CCN in the MBL (in black) and CCN in the CBL (in red). The
- 2 upper error bar is for the BASE run, the middle symbol for the LowNUC run, and the
- 3 lower error bar for the NoNUC run. Large and filled symbols are for measurement
- 4 duration longer than 10 days, and small and open symbols for less than 10 days.
- 5 Circle symbols are for supersaturations (s) less than 0.2%; upward triangles for s
- 6 greater than 0.2% and less than 0.8%; downward triangle for *s* greater than 0.8%.



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9 Figure 28. Annual cycle of CCN concentrations at Cape Grim (supersaturations (*s*) of

- 10 1.2% in a, 0.75% in b, and 0.23% in c) and Mace Head (*s* of 0.5% in d). The measured
- 11 CCN concentrations are shown in black, and the simulated CCN in red for the BASE
- 12 run, blue for the LowNUC run, and green for the NoNUC run. Correlation (R) and
- 13 normalized mean bias (NMB) are given.

