

Simulation of tropospheric chemistry and aerosols with the climate model EC-Earth

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Abstract

We have integrated the atmospheric chemistry and transport model TM5 into the global climate model EC-Earth version 2.4. We present an overview of the TM5 model and the two-way data exchange between TM5 and the IFS model from the European Centre for Medium-Range Weather Forecasts (ECMWF), the atmospheric general circulation model of EC-Earth. In this paper we evaluate the simulation of tropospheric chemistry and aerosols in a one-way coupled configuration. We have carried out a decadal simulation for present-day conditions and calculated chemical budgets and climatologies of tracer concentrations and aerosol optical depth. For comparison we have also performed offline simulations driven by meteorological fields from ECMWF's ERA-Interim reanalysis and output from the EC-Earth model itself. Compared to the offline simulations, the online-coupled system produces more efficient vertical mixing in the troposphere, which reflects an improvement of the treatment of cumulus convection. The chemistry in the EC-Earth simulations is affected by the fact that the current version of EC-Earth produces a cold bias with too dry air in large parts of the troposphere. Compared to the ERA-Interim driven simulation, the oxidizing capacity in EC-Earth is lower

1 in the tropics and higher in the extratropics. The atmospheric lifetime of methane in EC-Earth
2 is 9.4 years, which is 7% longer than the lifetime obtained with ERA-Interim but remains well
3 within the range reported in the literature. We further evaluate the model by comparing the
4 simulated climatologies of surface radon-222 and carbon monoxide, tropospheric and surface
5 ozone, and aerosol optical depth against observational data. The work presented in this study
6 is the first step in the development of EC-Earth into an Earth system model with fully
7 interactive atmospheric chemistry and aerosols.

8 **1 Introduction**

9 Chemically reactive gases and aerosols play important roles in the climate system. They affect
10 the Earth's energy balance by direct interaction with radiation and in various indirect ways.

11 Ozone (O₃) absorbs both solar (shortwave) and terrestrial (longwave) radiation. The
12 absorption of ultraviolet and visible radiation by ozone causes solar heating in the
13 stratosphere, and the absorption of thermal infrared radiation makes ozone an important
14 greenhouse gas.

15 Depletion of stratospheric ozone is the main cause for the observed cooling of the lower
16 stratosphere since the 1980s (Forster et al., 2011). Although the impact on the global radiation
17 balance of the troposphere is thought to be relatively small (Myhre et al., 2013b),
18 stratospheric ozone depletion has been identified as an important driver of tropospheric
19 circulation changes in the Southern Hemisphere (e.g. Gillett et al., 2003; Arblaster and Meehl,
20 2006; Polvani et al., 2011) and of circulation changes in the Southern Ocean (Sigmond et al.,
21 2011). This may also have reduced the uptake of carbon dioxide (CO₂) by the Southern Ocean
22 (Lenton et al., 2009). It is anticipated that the expected recovery of the ozone layer in the
23 coming decades will tend to reverse these trends (Perlwitz et al., 2008; Son et al., 2008;
24 Sigmond et al., 2011).

25 Ozone is not directly emitted into the atmosphere. In the troposphere it is produced by
26 oxidation of carbon monoxide (CO), methane (CH₄) and other volatile organic compounds
27 (VOCs), in the presence of nitrogen oxides (NO_x=NO+NO₂). Increases in tropospheric ozone
28 since the preindustrial era have contributed substantially to global warming by direct radiative
29 effects (Stevenson et al., 2013; Myhre et al., 2013b), especially in the Northern Hemisphere
30 (Mickley et al., 2004; Shindell et al., 2006a). Moreover, increases in ground-level ozone may
31 have contributed to global warming by reducing the CO₂ uptake by vegetation (Felzer et al.,

1 2005; Sitch et al., 2007; Arneth et al., 2013), but the importance of this effect is still uncertain
2 (Myhre et al., 2013b).

3 Methane itself strongly absorbs thermal infrared radiation and is therefore a very potent
4 greenhouse gas. Methane is also important as a precursor of stratospheric water vapour.
5 Increases in methane concentrations have contributed substantially to global warming, and
6 increases in the anthropogenic methane emissions even more so (Shindell et al., 2009; Myhre
7 et al., 2013b).

8 Aerosols affect the Earth's radiation budget by scattering and absorption of sunlight, by
9 absorption of thermal infrared radiation, and by interactions with clouds. Scattering tends to
10 increase the planetary albedo and has a cooling effect. Absorption, on the other hand, causes
11 warming in the atmosphere. The importance of scattering versus absorption depends on the
12 chemical composition, mixing state, size distribution, particle shapes and vertical distribution
13 of the aerosol mixture, as well as on the presence of clouds and the surface albedo.

14 Absorption of infrared radiation mainly takes place by coarse-mode aerosols, and is most
15 relevant for stratospheric aerosols resulting from large volcanic eruptions (Arfeuille et al.,
16 2013), and for tropospheric aerosols containing mineral dust or sea salt (Jacobson, 2001). The
17 shortwave radiative effects of aerosols are generally considered to be more important for the
18 climate (Myhre et al., 2013a; 2013b).

19 Black carbon (Petzold et al., 2013) strongly absorbs sunlight, which makes it an important
20 warming agent (Bond et al., 2013). Sulphate, nitrate and sea salt only weakly absorb sunlight
21 and are mainly scattering. Mineral dust and organic aerosols vary from weakly to strongly
22 absorbing, depending on their composition and the wavelength of the light. Light absorbing
23 aerosols such as black carbon and mineral dust also have a warming effect when deposited on
24 snow or ice (Myhre et al., 2013b).

25 Aerosol-cloud interactions include the effects of aerosols on the albedo and the lifetime of
26 clouds (Boucher et al., 2013). Overall, aerosol-cloud interactions are thought to have a
27 cooling effect (Myhre et al., 2013b).

28 Due to the difficulty of characterizing the concentrations and properties of aerosols and the
29 complexities involved in the processes that determine their effects on the climate, aerosols
30 still are a major source of uncertainty in our understanding of climate change. It is generally
31 believed that increases in anthropogenic aerosols since the preindustrial era have slowed

1 down global warming, but it is highly uncertain by how much (Myhre et al., 2013b).
2 Moreover, the cooling effect of increases in sulphate and other weakly absorbing aerosols,
3 such as organic aerosols and nitrate, has been largely compensated by a substantial warming
4 effect due to increases in black carbon (Bond et al., 2013).

5 Aerosols and chemically reactive gases are coupled in various ways. Many aerosol
6 components are produced from gaseous precursors by chemical reactions and nucleation or
7 condensation processes in the atmosphere. Sulphate (SO₄), nitrate (NO₃) and ammonium
8 (NH₄) result from emissions of sulphur dioxide (SO₂), dimethyl sulphide (DMS), NO_x and
9 ammonia (NH₃), and secondary organic aerosols (SOA) from emissions of VOCs. Nitrate,
10 ammonium and many organic aerosol components are semi-volatile and exist in equilibrium
11 with their gas-phase counterparts. Moreover, aerosols have an influence on photolysis rates by
12 scattering and absorption of ultraviolet light and by their impacts on clouds. They also
13 provide particle surfaces on which heterogeneous chemical reactions can take place.

14 Deposition of chemically reactive gases and aerosols from the atmosphere is a source of
15 nutrients to the terrestrial and marine biosphere. The biogeochemical cycles of nitrogen and
16 carbon are tightly coupled, and it is likely that the availability of reactive nitrogen will be a
17 limiting factor for the land carbon sink in the 21st century (Ciais et al., 2014).

18 Despite the important role aerosols and chemically reactive gases play in the climate system,
19 the description of atmospheric chemistry and aerosols varies strongly among climate models.
20 Most global models that participated in the recent Coupled Model Intercomparison Project
21 Phase 5 (CMIP5) (Taylor et al., 2012) did not include atmospheric chemistry and many did
22 not have fully interactive aerosols (Flato et al., 2013). Even so, atmospheric or atmosphere-
23 ocean general circulation models (GCMs) are increasingly being transformed into chemistry-
24 climate models (CCMs) with interactive representations of chemistry and aerosols (e.g.
25 Zhang, 2008; Dameris and Jöckel, 2013; Lamarque et al., 2012).

26 The work presented in this paper is the first step in the development of an interactive
27 chemistry module in the global climate model EC-Earth (Hazeleger et al., 2010; 2012). EC-
28 Earth is a relatively new climate model that has been developed in recent years by a
29 consortium of partner institutes from currently ten European countries, consisting of the
30 national meteorological services of Denmark, Ireland, the Netherlands, Portugal, Spain and
31 Sweden, universities, high-performance computing centres, and other research institutes. The
32 atmospheric GCM of EC-Earth is based on the Integrated Forecasting System (IFS) model

1 from the European Centre for Medium-Range Weather Forecasts (ECMWF). EC-Earth is
2 used for seasonal to decadal predictions as well as for long-term climate simulations (see
3 Hazeleger et al., 2010).

4 The chemistry module of EC-Earth is based on the chemistry and transport model TM5 (Krol
5 et al., 2005; Huijnen et al., 2010; Aan de Brugh et al., 2011). We have integrated TM5 into
6 EC-Earth by coupling it online to IFS. The model allows for two-way exchange of fields
7 between TM5 and IFS, but in this paper we focus on the impact of the online integration on
8 the performance of TM5, without feedbacks to IFS.

9 To this end, we have carried out a decadal simulation of tropospheric chemistry and aerosols
10 for present-day conditions, and calculated seasonal climatologies of concentration fields and
11 chemical budgets for various tracers. For comparison, we have repeated this simulation with
12 the standalone version of TM5 driven by meteorological fields from the ECMWF ERA-
13 Interim reanalysis (Dee et al., 2011). We have evaluated the results from both simulations
14 against a number of observational datasets.

15 In Sect. 2 we briefly introduce the EC-Earth model and describe the most important aspects of
16 TM5 and the data exchange between TM5 and IFS. In Sect. 3 we describe the setup of the
17 online and offline simulations. An evaluation of the results is presented in Sect. 4. We end
18 with a discussion and conclusions in Sect. 5.

19 **2 Model description**

20 **2.1 EC-Earth version 2.3**

21 The atmosphere-ocean general circulation model (GCM) applied in this study is EC-Earth
22 version 2.3. It consists of an atmospheric GCM based on the IFS model cycle 31r1 with the
23 H-TESSSEL land-surface scheme, and an ocean GCM from the Nucleus for European
24 Modelling of the Ocean (NEMO) version 2 with the Louvain-la-Neuve sea ice model (LIM)
25 version 2. The exchange of two-dimensional fields between IFS/H-TESSSEL and NEMO/LIM
26 takes place through the OASIS3 coupler (Ocean Atmosphere Sea Ice Soil version 3). A
27 description of these components and the coupling interface is given by Hazeleger et al. (2010,
28 2012).

29 A number of improvements in physical parameterizations have been included from more
30 recent cycles of IFS (see Hazeleger et al., 2012). In particular, the convection scheme has

1 been updated to the formulation of cycle 32r3. A detailed description of the changes that are
2 involved in this update is given by Bechtold et al. (2008). It has been shown that the new
3 convection scheme produces higher and more realistic levels of convective activity over land,
4 and leads to improvements in tropical precipitation patterns and extratropical circulation
5 characteristics (Bechtold et al., 2008; Jung et al., 2010).

6 EC-Earth version 2.3 has been used for CMIP5. Compared to the version described by
7 Hazeleger et al. (2012), the aerosol forcings have been improved and made consistent with the
8 CMIP5 recommendations (Taylor et al., 2012). In this study, we applied the same
9 configuration as for the CMIP5 long-term simulations, using the T159 spectral resolution
10 (corresponding to 1.125°) with 62 vertical levels for the atmosphere and the ORCA1 grid
11 (about 1° horizontal resolution and 42 layers) for the ocean.

12 **2.2 TM5**

13 We have extended the atmosphere-ocean GCM version of EC-Earth with a module for
14 simulating atmospheric chemistry and transport, the Tracer Model 5 (TM5). The new model
15 configuration with TM5 has been released as part of EC-Earth version 2.4. TM5 can be used
16 for non-reactive greenhouse gases like CO_2 (Peters et al., 2010) and sulphur hexafluoride
17 (SF_6) (Peters et al., 2004), for diagnostic radioactive tracers like radon-222 (^{222}Rn) and lead-
18 210 (^{210}Pb), as well as for chemically reactive gases and aerosols. The version used in this
19 study is based on the tropospheric chemistry version documented by Huijnen et al. (2010),
20 extended with the aerosol microphysics and optics modules described by Aan de Brugh et al.
21 (2011) and Aan de Brugh (2013). In this section we will give an overview of the main
22 characteristics of this TM5 model version and briefly describe the most important
23 modifications and improvements compared to these earlier publications.

24 **2.2.1 Resolution**

25 The global atmospheric domain of TM5 is discretized on a regular latitude/longitude grid. To
26 limit the computational costs, this grid will typically have a coarser resolution than the grid
27 used in IFS. In this study a horizontal resolution of $3^\circ \times 2^\circ$ (longitude x latitude) is used. Zoom
28 regions with higher horizontal resolutions can be defined and nested into the global domain
29 (Krol et al., 2005), but this option is not used in EC-Earth. To avoid the use of very short time
30 steps near the poles, the number of grid cells in the zonal direction is gradually reduced in the
31 polar regions. Dry deposition velocities and surface emission fluxes that depend on local

1 meteorological conditions and/or other surface variables are calculated on a higher-resolution
2 surface grid and subsequently coarsened to the atmospheric grid. The resolution of the surface
3 grid is currently $1^\circ \times 1^\circ$.

4 In the vertical direction TM5 uses the same hybrid sigma-pressure levels as used in the IFS
5 model version to which it is coupled, or a subset thereof. In the EC-Earth configuration
6 applied in this study a selection of 31 levels is made out of the 62 levels used in IFS. Because
7 of the relatively poor vertical resolution of the 62-level version of IFS in the upper part of the
8 domain (e.g. compared to the 60- or 91-level versions), no merging of levels is applied above
9 ~ 100 hPa. The top of the model is at 5 hPa.

10 2.2.2 Data exchange and transformations

11 As for the exchange between IFS and NEMO, the data exchange between IFS and TM5 takes
12 place through OASIS3 (Valcke, 2013). To prevent the different components having to wait
13 for each other, IFS runs one exchange time interval ahead of the other modules. In the current
14 configuration, the interval for the data exchange between TM5 and IFS is set to 6 hours. A
15 more frequent exchange will be applied in future versions of the model.

16 Since OASIS3 can only deal with 2-dimensional (2-D) fields, 3-dimensional (3-D) fields are
17 transferred layer by layer. The layers that are transferred from IFS to TM5 and vice versa
18 correspond to the full vertical resolution of IFS. The required merging of the layers and
19 interpolation of the data in the vertical direction is performed at the TM5 side.

20 TM5 receives both meteorological data and surface property fields from IFS. The datasets
21 employed in the chemistry version of TM5 used in this study are listed in Table 1. They
22 include instantaneous, time-averaged, and constant fields.

23 Most fields are interpolated by OASIS to TM5's regular latitude/longitude atmospheric or
24 higher-resolution surface grid. However, to avoid unnecessary interpolations of the wind
25 fields, the wind divergence and vorticity fields and the concurrent surface pressure field are
26 received in their native spectral representation and transformed into gridded air mass fluxes
27 following the procedure of Segers et al. (2002). Here the vertical mass fluxes are calculated
28 directly from the spectral fields, and the local mass balance over the exchange interval is
29 closed by slightly adjusting the horizontal mass fluxes. This method has been shown to lead to
30 superior chemistry simulations compared to methods that make use of interpolated wind fields
31 (Bregman et al., 2003).

1 Most instantaneous fields transferred from IFS to TM5, including the spectral fields
2 mentioned above, are valid for the middle of the exchange interval. However, for closing the
3 mass balance the surface pressure is also required at the beginning and at the end of the
4 interval. This is achieved by reading the initial surface pressure field from the TM5 restart
5 file, and including an additional (gridded) surface pressure field valid for the end of the
6 interval in the transfer.

7 The system has also been prepared for data transfer in the other direction. The fields that can
8 currently be transferred from TM5 to IFS are the ozone and methane concentrations, the
9 particle number and component-specific mass concentrations in the different aerosol modes
10 (see below), and aerosol optical property fields (extinction, single-scattering albedo and
11 asymmetry factor) at the wavelengths used in the IFS shortwave radiation scheme. Because
12 IFS runs ahead of TM5, these forcing fields are applied with some delay in IFS. To minimize
13 this delay, they are treated as instantaneous fields calculated at the end of the exchange
14 interval. This reduces the delay to half an exchange time step on average.

15 In this paper we first evaluate the one-way coupled simulation of chemistry and aerosols. In a
16 forthcoming publication two-way coupling will be applied, including feedbacks of the TM5
17 forcing fields to the radiation and cloud scheme of IFS.

18 2.2.3 Transport

19 Tracers in TM5 are moved around by advection, cumulus convection and vertical diffusion.
20 Tracer advection is described using either the first-order moments ('slopes') algorithm
21 developed by Russell and Lerner (1981) or the second-order moments scheme by Prather et
22 al. (1986). Both schemes are conserving the mass of the advected tracers. This is an important
23 requirement especially for chemistry-climate simulations, where the tracer concentrations are
24 not constrained by assimilation. The default option is to use the slopes scheme, which is used
25 in the simulations presented in this study.

26 Convective tracer transport in TM5 is described using a bulk mass flux approach, in which
27 clouds are represented by a single pair of entraining and detraining plumes describing the
28 updraft and downdraft motions. The meteorological fields involved in the calculation are the
29 vertical air mass fluxes and the entrainment and detrainment rates in the updrafts and
30 downdrafts. In EC-Earth the mass fluxes and detrainment rates are taken from IFS. The
31 corresponding entrainment rates follow from mass conservation. Thus, the description of

1 convective tracer transport in TM5 is fully consistent with the representation of convection in
2 IFS.

3 Vertical diffusion of tracers in TM5 is described with a first-order closure scheme, where the
4 diffusion coefficient is assumed to be the same as for heat (Olivié et al., 2004). In the free
5 troposphere it is computed based on wind shear and static stability following Louis (1979). In
6 the boundary layer it is based on the revised Louis-Tiedtke-Geleyn (LTG) scheme of Holtslag
7 and Boville (1993). The boundary layer height is calculated following Vogelesang and
8 Holtslag (1996). Details are given in Olivié et al. (2004a).

9 2.2.4 Chemistry

10 The TM5 version applied in this study is designed to simulate the concentrations of reactive
11 gases and aerosols in the troposphere and their deposition to the Earth's surface. The model's
12 gas-phase, aqueous-phase and heterogeneous chemistry schemes are described by Huijnen et
13 al. (2010). Details on the aqueous-phase chemistry can be found in Roelofs (1992) and
14 Feichter et al. (1996).

15 The gas-phase reaction scheme, representing the oxidation of CO, CH₄ and non-methane
16 volatile organic compounds (NMVOCs) in the presence of NO_x, is based on the Carbon Bond
17 Mechanism 4 (CBM4). It utilizes a structural-lumping technique in which organic species are
18 grouped into one or more surrogate categories according to the carbon bond types present in
19 the molecule. CBM4 was originally developed for simulating urban and regional
20 photochemistry (Gery et al., 1989), and was later extended to the global scale by including
21 reactions important under background conditions (Houweling et al., 1998). Since then,
22 reaction rates and product distributions have been updated (see Huijnen et al., 2010). In
23 addition to CBM4, the gas-phase chemistry scheme in TM5 also includes reactions for the
24 oxidation of SO₂, DMS, and NH₃.

25 Photolysis rates are calculated based on the parameterization of Landgraf and Crutzen
26 (1998), using 7 wavelength bands between 202.0 and 752.5 nm. Variations due to the effects
27 of clouds, overhead ozone, and surface albedo are included following Krol and van Weele
28 (1997).

29 Aqueous-phase chemistry in clouds is included for the oxidation of total dissolved sulphur
30 dioxide, S(IV), by dissolved hydrogen peroxide (H₂O₂) and O₃, depending on the acidity of
31 the droplets. The sulphate production rates due to the oxidation of S(IV) by H₂O₂ and O₃ are

1 calculated following Martin and Damschen (1981), with a temperature dependence from Nair
2 and Peters (1989), and Maahs (1983), respectively. The representation of heterogeneous
3 chemistry is currently limited to the reactive uptake of dinitrogen pentoxide (N_2O_5) at the
4 surface of cloud droplets, cirrus particles and aerosols, which produces nitric acid (HNO_3)
5 (Dentener and Crutzen, 1993).

6 2.2.5 Aerosols

7 Aerosols are represented in the model as described by Aan de Brugh et al. (2011). Sulphate,
8 black carbon (BC), organic carbon (OC), sea salt and mineral dust are described with the size-
9 resolved modal microphysics scheme M7 (Vignati et al., 2004). It uses seven log-normal size
10 distributions or modes with predefined geometric standard deviations. There are four water-
11 soluble modes (nucleation, Aitken, accumulation and coarse) and three insoluble modes
12 (Aitken, accumulation and coarse). The nucleation mode contains only SO_4 particles with dry
13 diameters smaller than 10 nm. The Aitken, accumulation and coarse modes represent particles
14 with dry diameters in the range 10-100 nm, 100 nm-1 μm , and larger than 1 μm , respectively.
15 The insoluble Aitken mode consists of internally mixed particles of BC and OC, while the
16 larger insoluble modes contain only dust particles. The soluble Aitken mode represents
17 internal mixtures of sulphate, BC and OC, while the larger soluble modes also contain sea salt
18 and dust in the mixture. Each mode is characterized by the total particle number and the mass
19 of each component. With this the total number of aerosol tracers in M7 amounts to 25. The
20 microphysical processes included in M7 are the formation of new SO_4 particles by nucleation
21 from gaseous sulphuric acid (H_2SO_4), condensation of H_2SO_4 onto existing particles, water
22 uptake, and intramodal and intermodal coagulation.

23 Of organic aerosols (OA), also known as particulate organic matter (POM), only the carbon
24 component is included in M7. To account for the other components that may be present, in
25 TM5 the dry mass of organic aerosols is assumed to be 40% higher than the OC mass (e.g.
26 Dentener et al., 2006; Kinne et al., 2006).

27 The current chemistry scheme does not describe the formation of secondary organic aerosols.
28 An additional source of organic aerosols is therefore included near the surface over land,
29 representing SOA formation from biogenic NMVOCs (mainly mono-terpenes) on time scales
30 of a few hours. The total mass of SOA being formed is prescribed using monthly fields from
31 Dentener et al. (2006), which amount to 19.1 Tg POM yr^{-1} . The freshly formed SOA particles

1 are assumed soluble and are added to the soluble Aitken mode, as in Aan de Brugh et al.
2 (2011).

3 The other aerosol components included in the model are nitrate, ammonium and methane
4 sulphonic acid (MSA), which is produced in the oxidation of DMS. These are represented by
5 their total mass, i.e. using a bulk aerosol approach. The gas/aerosol partitioning of the semi-
6 volatile inorganic species (i.e. the ratios between HNO_3 and nitrate aerosol and between NH_3
7 and NH_4) is described with the thermodynamic equilibrium model EQSAM (Metzger et al.,
8 2002).

9 The optical properties of the aerosol mixtures are calculated as a function of wavelength
10 based on Mie theory, using a look-up table (Aan de Brugh et al., 2011; Aan de Brugh, 2013).
11 The optical effects of nitrate aerosol are included by assuming that the ammonium nitrate and
12 the water absorbed by it are present in the soluble accumulation mode (Aan de Brugh et al.,
13 2011). Effective-medium approximations are applied to calculate the refractive indices of the
14 internally mixed modes. Sulphate, nitrate, OC, sea salt and water are treated as homogeneous
15 mixtures described by the Bruggeman mixing rule. When BC and/or dust are present in the
16 mix, these are treated as inclusions in a homogeneous background medium, using the
17 Maxwell-Garnett mixing rule. A more detailed description of the optics module of TM5 is
18 given by Aan de Brugh (2013).

19 2.2.6 Radioactive tracers

20 In addition to reactive gases and aerosols, the model also includes the diagnostic radioactive
21 tracers radon-222 and lead-210. Radon-222 is chemically inert and insoluble in water. It is
22 emitted at a relatively uniform rate from the continental crust and decays with a half-life of
23 3.8 days into lead-210. Because of its short lifetime, radon-222 can be used to study rapid
24 vertical exchange from the continental boundary layer to the free troposphere and further
25 transport to more remote parts of the atmosphere (see reviews in Zahorowski et al., 2004, and
26 Williams et al., 2011). Lead-210 has a much longer half-life (22.3 years). After being formed,
27 it rapidly attaches to submicron aerosol particles. As a consequence, lead-210 is mainly
28 removed from the atmosphere by wet deposition and can be used to diagnose the wet
29 scavenging processes in the model.

1 2.2.7 Dry and wet deposition

2 Dry deposition of gases and aerosols to the Earth's surface in TM5 is described using a
3 standard resistance approach (e.g. Seinfeld and Pandis, 2006). Dry deposition velocities of
4 gaseous species are calculated as the inverse of the sum of an aerodynamic resistance, a quasi-
5 laminar sublayer resistance and a surface resistance. The surface resistance is calculated
6 following the method of Wesely (1989), which distinguishes between deposition to
7 vegetation, bare soils, water or wet surfaces, and snow or ice, and combines the various
8 deposition pathways into a total surface resistance. The gas-phase deposition scheme is
9 described in more detail by Ganzeveld and Lelieveld (1995) and Ganzeveld et al. (1998). An
10 overview is given in Huijnen et al. (2010).

11 Dry deposition velocities of aerosols are determined by the aerodynamic resistance and a
12 quasi-laminar sublayer resistance, enhanced by sedimentation of particles by gravitational
13 settling. The quasi-laminar sublayer resistance depends on particle size and is calculated for
14 land and sea surfaces following Slinn (1976) and Slinn and Slinn (1980), respectively. The
15 implementation for the M7 modes is described by Aan de Brugh et al. (2011).

16 The wet deposition scheme in TM5 describes the removal of gases and aerosols from the
17 atmosphere by raining clouds, and distinguishes between convective and large-scale
18 stratiform precipitation. Scavenging by precipitation formation in convective clouds is
19 included in the convective mass transport operator as part of the mass fluxes entrained in the
20 cumulus updrafts (Balkansi et al., 1993; Guelle et al., 1998). Aerosols and irreversibly soluble
21 gases are assumed to be completely scavenged in vigorous convective updrafts, while the
22 removal efficiencies of other gases are reduced depending on their solubility. Resolution
23 dependencies are reduced by scaling down the convective scavenging rates for all tracers,
24 depending on the grid-cell mean convective precipitation at the surface (see Vignati et al.,
25 2010a).

26 For stratiform precipitation both in-cloud and below-cloud scavenging of gases and aerosols
27 are considered as described by Roelofs and Lelieveld (1995) and Jeuken et al. (2001).
28 Scavenging by precipitation formation inside stratiform clouds is assumed to be five times
29 less effective for ice particles than for liquid droplets. In-cloud scavenging of aerosols is
30 included only for the soluble accumulation and coarse modes, and for bulk aerosols like
31 ammonium, nitrate, MSA, and lead-210. The in-cloud scavenging coefficients for the soluble
32 accumulation and coarse modes are assumed to be equal to the values for irreversibly soluble

1 gases, while for bulk aerosols they are reduced by 30% to account for the presence of
2 interstitial aerosols (Vignati et al., 2010a). Below-cloud scavenging of aerosols is modeled
3 using estimates of bulk washout coefficients for the various modes based on Dana and Hales
4 (1976). Resolution dependencies in wet removal by stratiform precipitation are reduced by the
5 introduction of a mixing time scale, which delays the subgrid-scale mixing between cloudy
6 and cloud-free regions (see Vignati et al., 2010a).

7 2.2.8 Boundary conditions

8 A detailed description of stratospheric chemistry is not included in the model. To simulate
9 stratospheric ozone chemistry a parameterized linear chemistry scheme can be used (Cariolle
10 and Teyssèdre, 2007; McLinden et al., 2000; Van Noije et al., 2004, 2006). Alternatively, the
11 O₃ concentrations in the stratosphere can simply be relaxed towards observational values, as
12 described in Huijnen et al. (2010). In the current relaxation scheme, total O₃ column estimates
13 from a multi-sensor reanalysis (Van der A et al., 2010) are combined with a climatological
14 dataset of vertical profiles constructed from sonde and satellite observations (Fortuin and
15 Kelder, 1998). A similar relaxation procedure is applied to the CH₄ concentrations in the
16 stratosphere, while HNO₃ is constrained by prescribing the concentration ratio of HNO₃ over
17 O₃ at 10 hPa. These stratospheric boundary conditions are primarily based on satellite data
18 (see Huijnen et al., 2010). This is adequate for the present-day decadal simulations presented
19 in this study, but additional datasets based on output from stratospheric chemistry models are
20 needed for representing the longer-term trends and variability in simulations that start in the
21 pre-satellite era or continue into the future.

22 Because of the relatively long lifetime of CH₄, an additional constraint can be imposed on the
23 CH₄ concentrations at the surface. This is common practice in chemistry models in which the
24 CH₄ lifetime, which is mainly determined by the amount and distribution of the hydroxyl
25 radical (OH) in the troposphere, is not prescribed or tuned. It prevents drifts and/or biases in
26 the global CH₄ concentration, which would otherwise result from inconsistencies between the
27 CH₄ sources and sinks. This constraint can be imposed by relaxing the zonal mean surface
28 concentrations of CH₄ to values consistent with observations, while at the same time
29 including the location dependent emissions of CH₄. Alternatively, the CH₄ concentrations at
30 the surface can be prescribed using zonal and monthly mean fields based on observed values.
31 In both cases future concentration scenarios may be imposed by scaling the target
32 concentration fields based on the projected evolution of the global mean concentration.

1 2.2.9 Emissions

2 Emissions from anthropogenic activities and biomass burning are taken from the dataset
3 provided for the Atmospheric Chemistry and Climate Model Intercomparison Project
4 (ACCMIP), which was also used in CMIP5. The historical part of this dataset covers the
5 period 1850 to 2000 (Lamarque et al., 2010). The estimates for the year 2000 are based on a
6 combination of regional and global inventories for the various sectors. The reconstruction for
7 earlier decades is forced to agree with these estimates. For the 21st century emission
8 projections from the representative concentration pathways (RCPs) are used (Van Vuuren et
9 al., 2011). The RCP emissions start from the historical inventory in 2000. The RCP emissions
10 are provided as monthly emissions for the years 2000, 2005, 2010, 2020, etc. A linear
11 interpolation of the seasonal cycle is applied to obtain the emissions in the intermediate years.

12 Oceanic emissions of DMS and NO_x production by lightning are calculated online as in
13 Huijnen et al. (2010). Terrestrial DMS emissions from soils and vegetation are prescribed
14 following Spiro et al. (1992). Sea-salt emissions are calculated online as in Vignati et al.
15 (2010b), based on the parameterization by Gong (2003). The emission rate is assumed to
16 depend on the 10-m wind speed as a power law with exponent 3.41 (Monahan and
17 Muircheartaigh, 1980). Emissions of mineral dust can either be calculated online based on the
18 parameterization by Tegen et al. (2002) or be prescribed using the monthly dataset for the
19 year 2000 from the Aerosol Comparisons between Observations and Models (AeroCom)
20 project, described by Dentener et al. (2006).

21 Natural emissions of CO, NMVOCs, NO_x, NH₃ and SO₂ are prescribed using a monthly
22 varying dataset compiled for the Monitoring Atmospheric Composition and Climate (MACC)
23 project. It includes: (1) biogenic emissions of isoprene and a number of other NMVOC
24 species as well as CO from vegetation based on the Model of Emissions of Gases and
25 Aerosols from Nature (MEGAN) version 2.1 (Guenther et al., 2012) for the year 2000; (2)
26 biogenic emissions of NO_x from soils based on Yienger and Levy (1995); (3) oceanic
27 emissions of CO and NMVOCs from Olivier et al. (2003); (4) biogenic emissions of NH₃
28 from soils under natural vegetation and oceanic emissions of NH₃ from Bouwman et al.
29 (1997); and (5) SO₂ fluxes from continuously emitting volcanoes from Andres and Kasgnoc
30 (1998). The emissions of radon-222 are prescribed as in Dentener et al. (1999). Following the
31 recommendations of Rasch et al. (2000), the emission flux density is set 1.0 atoms cm⁻² s⁻¹ for
32 all land areas between 60°S and 60°N and to 0.5 atoms cm⁻² s⁻¹ for land areas between 60°N

1 and 70°N, except in Greenland. Emissions in Greenland and other parts of the world are
2 assumed to be zero.

3 As in Huijnen et al. (2010), a diurnal cycle is applied to the isoprene emissions from
4 vegetation on top of the monthly estimates provided in the dataset. To account for SO₄
5 formation in sub-grid plumes, 2.5% of the sulphur in the SO₂ emissions provided for the
6 various sources is assumed to be emitted in the form of SO₄ (Huijnen et al., 2010; Aan de
7 Brugh et al., 2011). The size distributions assumed for the different particulate emission
8 sources are listed in Aan de Brugh et al. (2011).

9 The implementation of emission heights has been revised compared to the description in
10 Huijnen et al. (2010), based on estimates from several studies (Dentener et al., 2006; De Meij
11 et al., 2006; Bieser et al., 2011; Simpson et al. 2012). The vertical distributions applied to the
12 different emission sources are given in Table A1.

13 **3 Simulations**

14 In this study we present results from decadal simulations for the years 2000-2009, using 1999
15 as a spin-up year for the chemistry. The various simulations that were carried out are listed in
16 Table 2.

17 In the atmosphere-ocean GCM simulation of EC-Earth the historical part extends up to 2005
18 and is continued by a simulation based on scenario assumptions, in accordance with the
19 CMIP5 experimental design for long-term climate simulations. As a future scenario we adopt
20 the RCP4.5 (Thomson et al., 2011), one of the stabilization scenarios of the representative
21 concentration pathways. Please note that for the period considered the simulated climate will
22 not be sensitive to the chosen scenario. The atmosphere-ocean GCM was initialized on 1
23 January 1999 from one of the CMIP5 20th century simulations performed by the EC-Earth
24 consortium. To be precise, the first ensemble member ('SHC1') provided by the Swedish
25 Meteorological and Hydrological Institute (SMHI) was used.

26 In TM5 the stratospheric O₃ concentrations were relaxed as described in Sect. 2.2.8, using
27 total column estimates for the years 2000-2009. Also, the surface CH₄ concentrations were
28 prescribed according to observations for those years. Emissions of CH₄ were therefore not
29 applied in these simulations.

1 Simulations were carried out both with and without yearly changes in the emissions from
2 anthropogenic activities and biomass burning in TM5. In the reference EC-Earth simulation
3 these emissions were fixed to their 2000 values. This reference simulation is compared with a
4 corresponding TM5 simulation driven by meteorological data from the ECMWF reanalysis
5 ERA-Interim (Dee et al., 2011). Both simulations are also evaluated against observational
6 data. To estimate the impact of possible trends in the emissions, an additional ERA-Interim
7 simulation is used with anthropogenic and biomass burning emissions varying between 2000
8 and 2009. In this simulation the anthropogenic and biomass burning emissions were
9 prescribed according to the RCP4.5 scenario, consistent with the setup of the atmosphere-
10 ocean GCM. Please note, however, that during the RCP development process a harmonization
11 procedure has been applied to ensure that the emissions in the four different RCPs are still
12 nearly identical in 2005. As a consequence the choice of the RCP will only have some albeit
13 small effect on the chemistry during the second half of the simulation. Results from the ERA-
14 Interim simulation with varying emissions have also been provided to the second phase of the
15 AeroCom project. Aerosol concentrations and optical property fields from that simulation
16 have been evaluated within that project (see aerocom.met.no). In all simulations, the
17 emissions of mineral dust were prescribed using the AeroCom dataset for the year 2000 (see
18 Sect. 2.2.9).

19 The ERA-Interim input fields for TM5 have been created from the original ECMWF data
20 during a pre-processing stage (see Krol et al., 2005). In this process the required
21 meteorological and surface property fields are retrieved at a spectral resolution of T255
22 (corresponding to about 0.7°) and converted into TM5 input fields at a $1^\circ \times 1^\circ$ horizontal
23 resolution, keeping the full 60-level vertical resolution of the original data. The ERA-Interim
24 simulation was carried out at the same $3^\circ \times 2^\circ$ horizontal resolution as used in EC-Earth.
25 However, because of the different vertical resolutions of the ERA-Interim dataset and the
26 CMIP5 EC-Earth simulations, the vertical grid is different. The ERA-Interim simulation was
27 carried out using the same selection of 34 levels out of the original 60 levels of ERA-Interim
28 as used in Huijnen et al. (2010). The treatment of the meteorological fields in the temporal
29 dimension is also slightly different. In the ERA-Interim simulation most meteorological fields
30 are updated at a 3-hourly frequency (see Huijnen et al., 2010) and a linear interpolation is
31 applied to the instantaneous fields.

1 Another difference relates to the representation of the tracer transport by cumulus convection.
2 Historically, the required convective air mass fluxes and entrainment and detrainment rates
3 were not archived in the meteorological datasets used to drive TM5. In the standalone version
4 of TM5 these fields are therefore calculated diagnostically. This is done in a pre-processing
5 step according to the parameterization of Tiedtke (1989). This scheme was introduced in
6 ECMWF's operational forecast model in 1989. As the more recent schemes used in later IFS
7 cycles, the original Tiedtke scheme already distinguished between deep, shallow and mid-
8 level convection.

9 To estimate the impact of using diagnostically calculated convective mass fluxes and
10 entrainment and detrainment rates, an additional decadal simulation was performed with the
11 standalone version of TM5, but now driven by meteorological output from EC-Earth. The
12 radon-222 concentrations from this offline simulation are compared with the results from the
13 reference EC-Earth simulation, in which the data transfer from IFS to TM5 is done online
14 through OASIS. For the offline EC-Earth simulation the driving meteorological fields for the
15 years 1999-2005 were taken from the CMIP5 set of long-term historical simulations, which
16 also provided the start fields for the online EC-Earth simulation. From this ensemble the first
17 simulation provided by the Irish National Meteorological Service ('MEI1') was selected. For
18 the years 2006-2009 the corresponding member ('ME41') of the ensemble of RCP4.5
19 simulations was used. As for ERA-Interim, a pre-processing step was required to convert the
20 IFS output into input fields for TM5. The online and offline EC-Earth simulation were
21 performed at the same horizontal and vertical resolutions. Also, the meteorological fields are
22 updated with the same 6-hourly frequency. The remaining minor difference between the two
23 simulations is that in the offline simulation a linear temporal interpolation is applied to the
24 instantaneous meteorological fields, which is not possible in the online simulation (see Table
25 2).

26 The inclusion of atmospheric chemistry and aerosols substantially increases the computational
27 burden of the simulations. In the configuration described above with 47 processors used for
28 IFS and 32 for NEMO, the atmosphere-ocean version of EC-Earth simulates one year within
29 less than two hours on the ECMWF IBM POWER7 high-performance computer facility.
30 Adding 45 processors for TM5 slows down the model by almost a factor 9. We like to point
31 out that the performance of the online coupled IFS-NEMO-TM5 system is similar to that of
32 the standalone version of TM5, when the same number of processors is used for TM5.

1 **4 Evaluation**

2 In this section an evaluation is presented of some important aspects of the atmospheric
3 chemistry simulation with EC-Earth. With this objective monthly, seasonal and/or annual
4 mean 10-year climatologies from the reference EC-Earth simulation are compared with the
5 corresponding climatologies from the ERA-Interim and offline EC-Earth simulations and/or
6 with observational datasets. The variables of interest for which this has been done include
7 temperature, humidity, the concentrations of various tracers, aerosol burdens and optical
8 depth, and some important chemical budget terms. When comparing the climatologies from
9 the different simulations and observational datasets, the interannual variability in the
10 underlying data is used to calculate standard deviations and to determine the statistical
11 significance of the differences. Regions where the differences are not statistically significant
12 at the 5% level are indicated by the stippled areas in the figures.

13 When evaluating the model with ground-based observation from a particular station, we
14 linearly interpolate the grid-cell values from the model to the location of the station. The
15 station height is used to estimate the corresponding pressure level based on a standard
16 atmospheric profile (U.S. Standard Atmosphere, 1976).

17 **4.1 Physical climate**

18 An evaluation of the physical climate of EC-Earth version 2.2 is presented by Hazeleger et al.
19 (2012). In general the large-scale structures of the atmosphere, ocean and sea ice are well
20 simulated and the main patterns of interannual climate variability are well represented. The
21 climate of EC-Earth version 2.3 has qualitatively similar characteristics. In particular, the
22 model has a cold bias in most of the troposphere throughout the year. A warm bias is found
23 over the Southern Ocean, over the stratocumulus regions west of the continents in the
24 subtropics, and over parts of the Northern Hemisphere (NH) extratropics, but only in the
25 lower troposphere or near the surface. The middle and upper troposphere as well as the tropics
26 and most of the Arctic are on average up to 3-4°C too cold (see Fig. 1, top panels).
27 Consequently, the specific humidity is also biased low in most of the troposphere, in
28 particular in the tropics (Fig. 1, bottom panels). According to the Clausius-Clapeyron relation,
29 the saturation vapour pressure in the lower troposphere decreases by about 7% per degree
30 temperature decrease (Held and Soden, 2006). Assuming that the relative humidity is
31 insensitive to the temperature bias, a cold bias of 1°C would result in a local decrease in the

1 specific humidity of about 7%. Such a response of specific humidity to the temperature bias
2 combined with a subsequent redistribution of the resulting local humidity bias due to
3 horizontal and vertical transport can explain a large part of the observed bias in the humidity
4 field in EC-Earth compared to ERA-Interim.

5 Various aspects of the temperature and humidity biases in EC-Earth are also found in other
6 climate models (see e.g. Tian et al., 2013; Flato et al., 2014; Lamarque et al., 2013a). In
7 particular, the CMIP5 model ensemble also produces a cold bias in most of the troposphere
8 and a similar warm and wet bias over the Southern Ocean (Tian et al., 2013). The CMIP5
9 multi-model mean bias in annual mean temperature maximizes at about 2°C in the
10 extratropical upper troposphere. This is somewhat lower than the cold bias produced by EC-
11 Earth in this region, which goes up to about 3°C in the NH. Moreover, EC-Earth produces a
12 stronger cold bias in the tropical troposphere than the CMIP5 model ensemble. As a
13 consequence, the wet bias in the tropical lower troposphere is substantially stronger in EC-
14 Earth. Finally, we note that the spatial distribution of surface temperature biases is
15 qualitatively very similar for EC-Earth and the CMIP5 ensemble, at least in the annual mean
16 (Flato et al., 2014).

17 **4.2 Radon-222**

18 Simulated radon-222 concentrations provide information about transport in the different
19 model configurations, in particular transport from the continental boundary layer to the free
20 troposphere and more remote regions. Differences between the reference, online EC-Earth
21 simulation and the ERA-Interim simulation are caused by a combination of factors; most
22 significantly: (1) biases in the EC-Earth climate, (2) the different treatment of cumulus
23 convection in EC-Earth, and (3) the 6-hourly update of the meteorological fields in EC-Earth
24 (see Table 2). The effect of treating convection differently is evident in the comparison
25 between radon concentrations from the online and offline EC-Earth simulations (see Sect. 3).

26 Figure 2 shows winter and summer zonal mean radon concentrations from the online EC-
27 Earth simulation. Radon concentrations are highest over the continents, where the emissions
28 take place. The highest zonal mean concentrations overall are simulated in the NH lower
29 troposphere during boreal winter, when the boundary layer is most stable. For both
30 hemispheres the concentrations in the upper troposphere are highest in summer, when the
31 convective activity is strongest.

1 Compared to both offline simulations, the online EC-Earth simulation produces higher radon
2 concentrations in large parts of the upper troposphere, extending from the tropics to mid-
3 latitudes, and, during boreal winter, in the middle troposphere (above about 600 hPa) at
4 northern mid-latitudes. The online simulation generally gives lower concentrations near the
5 surface and in the lower troposphere, especially at northern mid-latitudes during winter. These
6 features are due to the different treatment of convection and are in line with the fact that the
7 updated convection scheme produces more intense continental convection and stronger upper-
8 level convective detrainment (Bechtold et al., 2008).

9 Close to the equator both EC-Earth simulations produce higher zonal mean concentrations
10 than the ERA-Interim simulation in the lower troposphere (up to about 700-600 hPa) and
11 lower concentrations at higher altitudes up to about 200 hPa. The fact that this feature is
12 common to both EC-Earth simulations indicates that it is the result of biases in the EC-Earth
13 climate and/or the 6-hourly update frequency of the meteorological fields.

14 The 6-hour update frequency may not always be sufficient to capture the development of
15 convective boundary layers over continental areas, which may lead to an overestimation of
16 radon concentrations near the surface (Krol et al., 2005). The offline EC-Earth simulation
17 indeed produces higher concentrations than the ERA-Interim simulation in parts of the lower
18 troposphere and at higher latitudes, especially during summer in the NH. However, the online
19 EC-Earth simulations give lower concentrations than the ERA-Interim simulation in these
20 regions due to the opposing effect of stronger vertical transport resulting from the different
21 treatment of convection. The update frequency will be reduced in future versions of the
22 model.

23 We finish this section with a comparison between monthly radon concentrations from the
24 simulations and ground-based observations. We present results for six stations, including two
25 coastal sites (Cape Grim and Richmond, both Australia), three island sites (Mauna Loa,
26 Gosan on Jeju Island, South Korea, and Sado Island, Japan) and one continental site
27 (Hohenpeissenberg, Germany). Data for Hohenpeissenberg have been taken from the World
28 Data Centre for Greenhouse Gases (WDCGG); data for the other stations have been provided
29 by the Australian Nuclear Science and Technology Organisation (ANSTO). All selected
30 stations provide multi-annual datasets of hourly radon measurements (see Table A2). We
31 calculated monthly climatologies based on the full period of available measurements, and
32 compare these with the 10-year climatologies from the simulations.

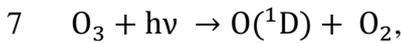
1 The resulting observed and simulated seasonal cycles at the selected stations are shown in
2 Fig. 3. Following earlier studies in which ground-based radon observations were used for
3 evaluating global models (Dentener et al., 1999; Taguchi et al., 2002; Zahorowski et al.,
4 2004), we applied a diurnal sampling window to the observations. The applied window is site
5 specific (see Table A2), and is chosen to minimize contributions from small-scale effects not
6 well resolved by the models. This is most relevant for sites like Richmond, where global
7 models underestimate the trapping of radon in the nocturnal boundary layer, and mountainous
8 and island sites like Mauna Loa, Gosan and Sado, where models underestimate the influence
9 of local emissions for various reasons (Zahorowski et al., 2005; Chambers et al., 2013).
10 Application of a diurnal sampling window vastly improves comparisons with the simulated
11 results for these stations (see Fig. 3). The resulting annual mean biases, root mean square
12 errors, and correlation coefficients between the observed and simulated seasonal cycles are
13 given in Table A2.

14 Many of the discrepancies between the modelled and observed radon seasonal cycles in Fig. 3
15 have also been found in other global modelling studies. Using the same radon emission fluxes
16 as applied in our simulations, Dentener et al. (1999), Tagushi et al. (2002) and Zhang et al.
17 (2011) reported underestimates for Mauna Loa all year round, and overestimates for Cape
18 Grim during the austral summer. Moreover, the simulations reported by Zhang et al. (2008)
19 also failed to reproduce the concentration increase measured at Hohenpeissenberg during
20 boreal summer. Discrepancies may be partially related to inaccuracies in assumed emission
21 fluxes. For example, emissions in continental Asia have been estimated to be higher than
22 previously thought (Williams et al., 2009). Except during the summer monsoon season, higher
23 Asian emissions estimates would substantially increase the simulated concentrations at Gosan
24 and Sado Island (Chambers et al., 2009). Zhang et al. (2011) also obtained better agreements
25 at Mauna Loa and Cape Grim using improved emissions estimates.

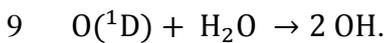
26 It is important to note that discrepancies between simulated and observed radon
27 concentrations are generally larger than the differences amongst the three simulations. In
28 order to use radon observations to evaluate the differences in the simulated transport, in
29 particular the impacts of the different treatment of convection in the online EC-Earth
30 simulation, it will be necessary to improve the setup of the simulations using more accurate
31 estimates of the spatially heterogeneous and seasonally varying radon emission fluxes (Zhang
32 et al., 2011).

1 4.3 Oxidizing capacity and methane lifetime

2 The oxidizing capacity of the atmosphere is determined by the abundance and distribution of
3 the hydroxyl radical (OH) in the troposphere. OH is highly reactive and initiates most of the
4 photochemical reaction chains that oxidize reactive gases in the atmosphere (Levy et al.,
5 1971; Lelieveld et al., 2002; 2004). The production of OH in the troposphere is mainly
6 governed by the photolysis of O₃,



8 followed by the reaction of the excited oxygen atom, O(¹D), with a water molecule:



10 The second step is limited by the availability of water molecules in the gas phase. On average
11 only a few percent of the O(¹D) atoms produced in the first reaction step will encounter a
12 water molecule to react with and produce OH (Lelieveld et al., 2002). As a consequence, OH
13 production rates are highest in the tropical lower and middle troposphere, due to the relatively
14 high amounts of both sunlight and water vapour in those regions.

15 The zonal mean OH concentrations from the reference EC-Earth simulation for winter and
16 summer are presented in the top panels of Fig. 4. Compared to the monthly climatology
17 presented by Spivakovsky et al. (2000), the region of high OH concentrations extends more
18 towards the surface. In other respects, the large-scale features of the spatial distributions are
19 very similar in all seasons. The peak concentrations in the tropics and subtropics are
20 substantially lower in EC-Earth, especially in boreal winter and the transition seasons.
21 However, based on simulations of methyl chloroform it has been concluded that the OH
22 concentrations from Spivakovsky et al. (2000) are likely too high. A better correspondence
23 with the observed decay of methyl chloroform concentrations was obtained by reducing the
24 climatology of Spivakovsky et al. (2000) by 8% (see Huijnen et al., 2010). Compared to the
25 optimized climatology thus obtained (shown in the middle panels of Fig. 4), the peak
26 concentrations from the EC-Earth simulation are quantitatively similar in boreal summer, but
27 at least 20% lower in the other seasons.

28 Compared to the ERA-Interim simulation, EC-Earth produces lower OH concentrations in
29 large parts of the tropical and subtropical troposphere. As shown in the bottom panels of Fig.
30 4, the zonal mean concentrations are lower in a region extending from close to the surface to
31 about 200 hPa. The lower OH concentrations in this region are mainly caused by lower

1 temperatures, resulting in lower specific humidities. Because there is less water vapour
2 available to react with O(¹D), the production of OH via the reaction path described above is
3 lower than in the ERA-Interim simulation (see Table 3).

4 At higher latitudes, the OH concentrations from EC-Earth are generally higher than in the
5 ERA-Interim simulation. As will be shown in Sect. 4.5, EC-Earth produces higher O₃
6 concentrations in most of the lower and middle troposphere, especially in the extratropics.
7 This increases the O(¹D) production rates, especially in the summer hemisphere. In the
8 Southern Hemisphere (SH) extratropical lower troposphere the difference in the OH
9 concentrations between the EC-Earth and ERA-Interim simulations is further enhanced by the
10 higher levels of humidity in EC-Earth associated with the warm bias over the Southern
11 Ocean.

12 In the tropical upper troposphere (above about 200 hPa), the OH concentrations in EC-Earth
13 are also higher than in the ERA-Interim simulation. This is likely related to the fact that some
14 of the tracers involved in the other reactions that produce OH (see Table 3) are more
15 efficiently transported to higher altitudes by deep convection (see Sect. 4.2).

16 Differences in the amount and distribution of the NO_x production by lightning also affect the
17 distribution of OH. The global production of NO_x by lightning is significantly higher in EC-
18 Earth than in the ERA-Interim simulation (see Table A3). The production in EC-Earth is 0.84
19 and 0.51 Tg N/yr higher in the NH and SH extratropics, respectively, while it is 0.74 Tg N/yr
20 lower in the tropics (30°S-30°N).

21 The lower OH concentrations in the tropical and subtropical lower and middle troposphere
22 lead to a slower removal of CH₄ from the atmosphere. The average chemical lifetime of CH₄
23 against reaction with tropospheric OH is 10.9 years in the EC-Earth simulation and 10.1 years
24 in the ERA-Interim simulation. Both values are within the multi-model ranges of 9.7 ± 1.7 ,
25 10.2 ± 1.7 , and 9.7 ± 1.5 years estimated by Shindell et al. (2006b), Fiore et al. (2009), and
26 Naik et al. (2013), respectively, from simulations for present-day conditions (year 2000 or
27 2001). Assuming a lifetime of 120 and 160 years, respectively, for the chemical loss in the
28 stratosphere and the soil sink (Ehhalt et al., 2001), the atmospheric lifetime of CH₄ is 9.4
29 years in the EC-Earth simulations. This is 7% longer than the atmospheric lifetime of 8.8
30 years obtained in the ERA-Interim simulation. Prather et al. (2012) recently estimated from
31 methyl chloroform observations that the present-day atmospheric lifetime of CH₄ is in the

1 range 9.1 ± 0.9 years. The values obtained in the EC-Earth and ERA-Interim simulations are
2 both well within this range.

3 **4.4 Carbon monoxide**

4 Carbon monoxide is emitted into the atmosphere by anthropogenic and natural sources and is
5 chemically produced in the atmosphere by oxidation of CH_4 and NMVOCs and by photolysis
6 of certain NMVOCs. The oxidation of CH_4 and many other hydrocarbons proceeds via the
7 formation of formaldehyde (CH_2O), which is subsequently converted to CO by photolysis or
8 oxidation, mostly by reaction with OH. CO is removed from the atmosphere by reaction with
9 OH and by dry deposition at the surface. The various contributions to the atmospheric budget
10 of CO in the EC-Earth and ERA-Interim simulations are given in Table 4.

11 The average lifetime of CO in the atmosphere (total burden divided by total loss) is 54.6 days
12 in EC-Earth compared to 52.5 days with ERA-Interim. The slightly longer lifetime in EC-
13 Earth is a result of a slower chemical destruction due to the lower OH concentrations in the
14 tropical and subtropical troposphere. In contrast, OH levels in EC-Earth are higher in the
15 extratropics, causing a more efficient removal of CO at higher latitudes, especially in the SH.

16 Also the production of CO in the tropics and the subtropics is lower in EC-Earth. This is a
17 direct consequence of a lower yield from the oxidation of CH_4 , caused by the lower OH
18 concentrations in the tropics and subtropics. In other words, to obtain the same CH_4
19 concentrations lower effective CH_4 emissions are needed in EC-Earth, resulting in a lower
20 production of CO. The total chemical production of CO in both simulations is lower than the
21 range of model estimates reported by Shindell et al. (2006b). This is likely due to an
22 underestimation of the CO production from NMVOCs in the CBM4 chemistry scheme.

23 The global tropospheric burden of CO is similar in both simulations (Table 4). EC-Earth
24 produces a lower burden in the tropics and a higher burden in the NH extratropics, but the
25 differences are only a few percent. Compared to the ERA-Interim simulation, EC-Earth gives
26 higher CO concentrations in the tropical upper troposphere and in the lower stratosphere (Fig.
27 5), mostly due to more efficient transport of CO by deep convection into the tropical upper
28 troposphere. In the tropical lower and middle troposphere both higher and lower
29 concentrations are observed depending on the location and the season. The concentrations in
30 EC-Earth are lower in the NH extratropics, due to the faster chemical destruction. They are

1 also somewhat lower in the SH extratropics in austral summer and, in the middle and upper
2 troposphere, in austral winter.

3 The simulated surface mixing ratios of CO have been evaluated against monthly averages
4 from the network of surface flask sampling of NOAA's Earth System Research Laboratory
5 (ESRL) Global Monitoring Division (GMD). The decadal monthly mean mixing ratios from
6 the simulations are compared with the flask measurements in Fig. 6. Since in the reference
7 EC-Earth simulation and the corresponding ERA-Interim simulation the emissions from
8 anthropogenic activities and biomass burning were fixed to their values for the year 2000, we
9 have also included the results from the ERA-Interim simulation with emissions varying from
10 year to year (see Table 2). The two ERA-Interim simulations give very similar decadal mean
11 CO concentrations at the stations used in the evaluation (see Table A4 for a complete list).
12 Thus, also the simulations with fixed emissions can be directly compared with the
13 measurements.

14 At the measurement locations the concentration differences between the EC-Earth and ERA-
15 Interim simulation are generally small in the SH and in the tropics and become larger in the
16 NH extratropics (see Fig. 6 and Table A2). For the majority of stations the seasonal cycle is
17 very well simulated, as expressed by a high correlation between the simulated and the
18 measured monthly values. The simulated CO concentrations are generally in good
19 quantitative agreement with the measurements in the SH. At the tropical stations in the NH
20 both simulations underestimate the measurements by about equal amounts. At northern mid-
21 latitudes, both simulations underestimate the measurements, but the concentrations in EC-
22 Earth are lower than with ERA-Interim, especially outside of the summer season. In the
23 annual mean the difference can be up to about 11 ppbv. The differences between the two
24 simulations are generally smaller than the amounts by which the measurements are
25 underestimated. We note that other modelling studies have shown similar CO biases in the
26 NH, using identical anthropogenic and biomass burning emissions (Lamarque et al., 2010;
27 Naik et al., 2013).

28 **4.5 Ozone**

29 The sources of ozone in the troposphere are chemical production by the oxidation of CO,
30 CH₄, and NMVOCs in the presence of NO_x, and net transport from the stratosphere. Ozone is
31 removed from the troposphere by chemical destruction and by dry deposition at land surfaces.

1 The chemical destruction of tropospheric O₃ occurs mainly through the photolysis of O₃
2 followed by the reaction of the produced excited oxygen atom with a water molecule (see
3 Sect. 4.3) and through the reaction of O₃ with the peroxy radical (HO₂) and with OH. The
4 main contributions to the tropospheric O₃ budget in the EC-Earth and ERA-Interim
5 simulations are given in Table 5. These numbers can be directly compared with the ACCENT
6 and ACCMIP multi-model results for the present day reported by Stevenson et al. (2006) and
7 Young et al. (2013), who use a similar method for defining the tropopause. Note that
8 Stevenson et al. (2006) give ranges based on the full ensemble of models participating in that
9 study as well as on a subset of models that were selected based on criteria related to the
10 simulation of O₃ and the CH₄ lifetime for the present day.

11 In EC-Earth the average lifetime of tropospheric O₃ is 25.5 days, which is outside the ranges
12 22.3 ± 2.0 days and 22.2 ± 2.2 days estimated by Stevenson et al. (2006) for the full ensemble
13 of ACCENT models and a subset of models, respectively, and corresponds to the highest
14 value out of the six individual model results reported by Young et al. (2013). With ERA-
15 Interim a lifetime of 23.9 days is obtained, which is within the ranges reported by these
16 authors. The longer lifetime in EC-Earth is caused by a slower chemical destruction. The cold
17 bias that exists in most of the troposphere slows down the destruction of O₃ by photolysis
18 because of the lower specific humidity. Lower concentrations of OH and HO₂ in large parts of
19 the tropical and subtropical troposphere (see Sect. 4.3) further slow down the destruction of
20 O₃.

21 EC-Earth produces a tropospheric O₃ burden of 327 Tg, which is well within the ranges $344 \pm$
22 39 Tg and 336 ± 27 Tg reported by Stevenson et al. (2006) and the range 337 ± 23 Tg
23 reported by Young et al. (2013). With ERA-Interim a tropospheric burden of 309 Tg is
24 obtained, which is at the low side of the ranges estimated by Stevenson et al. (2006) and
25 below the range estimated by Young et al. (2013). The higher burden in EC-Earth compared
26 to the ERA-Interim simulation is mainly due to the slower chemical destruction of O₃ in the
27 troposphere and a higher net influx of O₃ from the stratosphere. The influx from the
28 stratosphere is 349 Tg/yr in EC-Earth compared to 306 Tg/yr with ERA-Interim. Both values
29 are below the ranges 552 ± 168 Tg/yr and 556 ± 154 Tg/yr estimated by Stevenson et al.
30 (2006) and the model results reported by Young et al. (2013). Other model studies of
31 stratosphere-troposphere exchange also found the net O₃ flux to be higher than about 400
32 Tg/yr, in line with estimates based on observations (Olsen et al., 2004; Hsu et al., 2005).

1 Thus, compared to the ERA-Interim simulation, the higher net stratosphere-troposphere
2 exchange flux simulated in EC-Earth is likely an improvement.

3 Overall, the total chemical destruction of O₃ in the troposphere is lower in EC-Earth than in
4 the ERA-Interim simulation. The chemical production of O₃ in the troposphere is also
5 somewhat lower, but the net chemical production of O₃ in the troposphere is still higher in
6 EC-Earth. Combined with the higher net influx from the stratosphere this is consistent with a
7 higher deposition of O₃. The total deposition is 978 Tg/yr in EC-Earth, while 851 Tg/yr is
8 obtained in the ERA-Interim simulation. Both results are within the ranges 1003 ± 200 Tg/yr
9 and 953 ± 154 Tg/yr estimated by Stevenson et al. (2006) and the model results reported by
10 Young et al. (2013); however, the higher value obtained with EC-Earth is closer to the central
11 estimates obtained by these authors.

12 EC-Earth produces higher zonal mean O₃ concentrations than the ERA-Interim simulation in
13 large parts of the troposphere, including most of the NH and the lower and middle parts of the
14 SH (Fig. 7). Lower zonal mean concentrations are simulated in the tropical and subtropical
15 upper troposphere and parts of the tropical and subtropical middle troposphere, and in parts of
16 the lower stratosphere of the SH.

17 Differences in the contribution from O₃ originating from the stratosphere explain part of the
18 differences in the simulated O₃ concentrations in the troposphere (compare the lower and
19 middle panels of Fig. 7). This contribution was diagnosed using a stratospheric O₃ tracer, O_{3S}.
20 As in Lelieveld and Dentener (2000), O_{3S} is subject to the same stratospheric boundary
21 conditions and removal processes as regular O₃, but is not produced below a certain pressure
22 level, ~140 hPa in our model setup. Since only small amounts of O₃ are produced in the
23 region between this level and the tropopause, effectively the chemical production is switched
24 off in the troposphere. The O_{3S} tracer therefore provides a robust method for estimating the
25 contribution of O₃ produced in the stratosphere to the tropospheric budget. The total chemical
26 destruction and deposition of O_{3S} in the troposphere is 351 Tg/yr in EC-Earth and 305 Tg/yr,
27 very close to the estimates of the net stratosphere-troposphere exchange flux quoted above,
28 which were obtained by closing the tropospheric budget of O₃.

29 EC-Earth gives higher O₃ concentrations than the ERA-Interim simulation in the lowermost
30 stratosphere at high northern latitudes (Fig. 7). This, combined with the slower chemical
31 destruction in the troposphere, leads to higher O_{3S} concentrations in most of the NH. In the
32 SH, the lower zonal mean O₃ concentrations simulated with EC-Earth in the subtropical upper

1 and middle troposphere are partly due to a lower contribution from O_{3S} , especially in austral
2 winter. At higher latitudes, on the other hand, the contribution from O_{3S} in the troposphere is
3 higher than in the ERA-Interim simulation. Concentration differences in the lower
4 stratosphere between the EC-Earth and ERA-Interim simulations are the combined effect of
5 differences in the large-scale stratospheric circulation, stratosphere-troposphere exchange and
6 vertical resolution.

7 The slower chemical destruction due to the cold bias in EC-Earth increases the lifetime of O_3
8 in the troposphere, and tends to increase the concentration of both O_{3S} and O_3 . The resulting
9 concentration increase is larger for O_3 than for O_{3S} , but the increase in the O_3 concentration is
10 partly compensated by a reduced chemical production in the troposphere.

11 Differences in vertical exchange are also important to explain the differences in the O_3
12 concentrations between the two simulations. The different treatment of convection in EC-
13 Earth leads to more efficient convection to the upper parts of the tropical troposphere (see
14 Sect. 4.2). This tends to decrease the O_3 concentrations in the tropical and subtropical upper
15 troposphere. In the extratropics enhanced vertical mixing in EC-Earth tends to increase the O_3
16 concentrations in the lower parts of the troposphere by bringing down more O_3 from higher
17 altitudes. Enhanced mixing similarly tends to increase the O_{3S} concentrations in the lower
18 extratropical troposphere. The latter effects are unique features of the online EC-Earth
19 simulation, which are not reproduced in the offline EC-Earth simulation (not shown).

20 The simulated O_3 concentrations have been evaluated against a vertically resolved, zonal and
21 monthly mean dataset based on the O_3 profile measurements from the Binary DataBase of
22 Profiles (BDBP) of Hassler et al. (2008), which includes both satellite observations and
23 ozonesondes. The dataset used in the evaluation was constructed for the years 1979-2007 by
24 Bodeker Scientific (www.bodeker.com) in a similar way as described in Hassler et al. (2009).

25 In addition, the monthly mean O_3 dataset from Cionni et al. (2011) was included in the
26 evaluation. This dataset provided the O_3 distribution for the CMIP5 climate models that did
27 not calculate O_3 interactively. The historical part of this dataset extends to 2009. In the
28 stratosphere it consists of zonal mean fields derived from a multiple linear regression analysis
29 of satellite observations and polar ozonesonde data. In the troposphere it is based on
30 simulations with the chemistry-climate models CAM3.5 (Lamarque et al., 2010) and GISS-
31 PUCINI (Shindell et al., 2006c) with prescribed sea surface temperatures. The

1 anthropogenic and biomass burning emissions used in these simulations are the same as in the
2 simulations presented here, and are also kept constant from 2000-2009.

3 In Figs. 8 and 9 the monthly mean O₃ mixing ratios from the EC-Earth and ERA-Interim
4 simulations are compared with the observational and CMIP5 datasets in latitude bands of 30°
5 at 750, 500 and 250 hPa. In these figures, the 2000-2009 means from the simulations are
6 compared with the 2000-2007 means from the observational dataset and the 2000-2009 means
7 from the CMIP5 dataset.

8 We first look at the extratropical upper troposphere/lower stratosphere (250 hPa). Here EC-
9 Earth gives higher O₃ concentrations in the NH and lower concentrations in the SH than the
10 ERA-Interim simulation. At high northern latitudes (60-90°N) the higher concentrations
11 obtained with EC-Earth are in better agreement with the observational dataset. However, EC-
12 Earth underestimates the observational data in boreal spring and summer, which indicates that
13 the downward transport in the lower stratosphere may still be too slow. Between 30-60°N the
14 higher values obtained with EC-Earth lead to a somewhat stronger overestimation of the
15 observational data. On the other hand, the CMIP5 dataset shows even higher concentrations in
16 this region during boreal winter and spring. At high southern latitudes (60-90°S) both
17 simulations underestimate the observational data, but the agreement is worse for EC-Earth.
18 EC-Earth also underestimates the observational data between 30-60°S, where the ERA-
19 Interim simulation overestimates the observational data, especially in austral spring and
20 summer. In this region the CMIP5 dataset gives similar values as obtained with EC-Earth.

21 In the extratropical middle and lower troposphere (500 and 750 hPa, respectively), EC-Earth
22 gives significantly higher concentrations than the ERA-Interim simulation. In the NH, EC-
23 Earth agrees well with both the observational data and the CMIP5 dataset in boreal winter. In
24 boreal summer, the concentrations from EC-Earth are significantly higher than in the
25 observational dataset. In this season, the concentrations from the ERA-Interim simulation are
26 either very close to (30-60°N) or slightly lower than the observational estimates (60-90°N). In
27 the SH, EC-Earth shows a much smaller bias relative to the observational dataset. At high
28 southern latitudes (60-90°S), the concentrations from EC-Earth are lower than the
29 observational estimates during austral winter and higher during austral summer. EC-Earth is
30 in excellent agreement with the observational data in the lower troposphere between 30-60°S.

1 In the tropical upper troposphere (30°S-30°N, 250 hPa), EC-Earth produces lower values than
2 the ERA-Interim simulation and is in fairly good agreement with the observational dataset.
3 The CMIP5 dataset gives significantly lower values in this region.

4 In the tropical middle and lower troposphere the differences between the EC-Earth and ERA-
5 Interim simulation are relatively small. In the NH tropics (0-30°N), EC-Earth gives somewhat
6 higher concentrations. In the middle troposphere (500 hPa), EC-Earth reproduces the
7 observational data very well during boreal summer and fall, but gives lower values during
8 winter and spring. In any case, EC-Earth is closer to the observational data than both the
9 ERA-Interim simulation and the CMIP5 dataset. In the lower troposphere (750 hPa), EC-
10 Earth agrees well with the observational dataset in boreal spring and gives slightly higher
11 values in the other seasons. Both simulations are in fairly good agreement with the
12 observational dataset. The CMIP5 dataset gives significantly lower concentrations in this
13 region. In the SH tropics (0-30°S), both simulations underestimate the observation data,
14 especially in the middle troposphere (500 hPa). Here EC-Earth gives somewhat lower values
15 than the ERA-Interim simulation during austral winter and spring. The CMIP5 dataset
16 underestimates the observational data more strongly in the lower troposphere (750 hPa), and
17 also gives somewhat lower values than EC-Earth in the middle troposphere (500 hPa) during
18 austral winter and fall.

19 The results presented in Figs. 8 and 9 can be compared with the evaluation presented by
20 Young et al. (2013), in which present-day tropospheric O₃ concentrations from the ACCENT
21 and ACCMIP model ensembles are compared with monthly climatological datasets based on
22 ozonesonde measurements as well as satellite retrievals from the Tropospheric Emission
23 Spectrometer (TES). In general, our simulations are within the range of concentrations
24 simulated by the ACCMIP models. In particular, most ACCMIP models also underestimate
25 the O₃ concentrations in the middle and lower troposphere between 0-30°S. Likewise they
26 also underestimate the observed seasonal cycle in the upper and middle troposphere between
27 0-30°N.

28 The surface O₃ concentrations simulated with EC-Earth and the differences compared to the
29 ERA-Interim simulation are presented in Fig. 10. EC-Earth gives higher surface
30 concentrations in most of the world, with the exception of some regions located in the tropics
31 and subtropics. In the mid- to high latitudes of the NH, differences up to about 10 ppbv are
32 simulated during the winter season, while even larger differences are found during summer.

1 The simulated surface O₃ concentrations have been evaluated against in-situ surface
2 measurements. The stations used for the evaluation of surface O₃ are listed in Table A5. They
3 include stations from the NOAA GMD network and a selection of stations included in the
4 World Data Centre for Greenhouse Gases (WDCGG). Data for Mace Head were taken from
5 the European Monitoring and Evaluation Programme (EMEP). Monthly averages were
6 calculated from the hourly mixing ratio measurements and then averaged over the available
7 years in the simulation period. Figure 11 shows the resulting monthly mixing values for a
8 subset of stations spanning a broad range of latitudes, together with the decadal mean
9 simulation results obtained at the corresponding locations. Simulation results are included for
10 the reference EC-Earth simulation and the corresponding ERA-Interim simulation, as well as
11 for the ERA-Interim simulation with yearly changes in the emissions from anthropogenic
12 activities and biomass burning (see Table 2). As for CO, the effect of these emission
13 variations on the simulated decadal mean O₃ concentrations at the stations used in the
14 evaluation is very small, and sometimes barely visible. The simulations with fixed emissions
15 can therefore again be directly compared with the measurements (see Fig. 11 and Table A5).

16 EC-Earth produces higher monthly mean surface concentrations than the ERA-Interim
17 simulation at all stations, except at Mauna Loa (Hawaii), Tutuila (American Samoa) and
18 Pyramid on Mount Everest (Nepal).

19 At the Antarctic stations, both simulations underestimate the measurements. Here EC-Earth is
20 closer to the observations than the ERA-Interim simulation, but this is achieved at the expense
21 of the correlation between the measured and simulated monthly concentrations. At Cape Grim
22 (Tasmania), the seasonal cycle in both simulations is weaker than in the observations. EC-
23 Earth is in excellent agreement with the observations during the winter months, but
24 overestimates the observations during summer. The ERA-Interim simulation, on the other
25 hand, is in good agreement during summer, but underestimates the observations during
26 winter.

27 The ERA-Interim simulation also underestimates the measurements at high northern latitudes,
28 especially during winter. Here EC-Earth is on average closer to the observations. At Summit
29 (Greenland), EC-Earth is in excellent agreement with the measurements during summer and
30 underestimates the measurements in the other seasons. At Storhofdi (Iceland), on the other
31 hand, EC-Earth very well reproduces the measured concentrations in winter, but
32 overestimates them in the other seasons.

1 The picture is different at the tropical and subtropical stations, where both simulations show
2 an average positive bias. At Tutuila station, located in the tropics in the SH, the simulations
3 give similar results and systematically overestimate the observations. At Mauna Loa EC-Earth
4 is in reasonable agreement with the observations during spring and summer and overestimates
5 the observations during fall and winter. Here the difference with the ERA-Interim simulation
6 is relatively small compared to the interannual variability in the results. At the other tropical
7 and subtropical stations, EC-Earth shows worse agreement with the observations than the
8 ERA-Interim simulation, depending on the location.

9 EC-Earth also shows an average positive bias at the NH mid-latitude stations. Here the model
10 overestimates the measurements especially during summer. With the exception of the high-
11 altitude station Hohenpeissenberg (Germany), the ERA-Interim simulation is closer to the
12 observations at these stations.

13 **4.6 Aerosols**

14 In this section the model is evaluated with regard to the simulation of aerosols. Maps of the
15 atmospheric loads of the different aerosol components from the EC-Earth and ERA-Interim
16 simulations are presented in Fig. 12. In Table 6 we compare the corresponding global
17 burdens, lifetimes and dry and wet deposition rates as well as the emissions of sea salt and
18 mineral dust with results from recent modelling studies, in particular multi-model estimates
19 from AeroCom and ACCMIP. Results on the global budgets of the sulphate precursors DMS,
20 MSA and SO₂ are given in Table A6.

21 Compared to the ERA-Interim simulation, the sulphate load in EC-Earth is on average lower
22 in the tropics and, especially in boreal winter, higher in the extratropics. The decrease in the
23 tropics is mainly caused by a lower chemical production (see Table A6). The increase in the
24 extratropics is the combined effect of a higher chemical production and a longer lifetime. The
25 loads of black carbon, organic aerosols and nitrate are also higher at high northern latitudes in
26 boreal winter.

27 Furthermore, EC-Earth gives higher sea-salt loads at high latitudes, especially during local
28 winter and spring. This is mainly due to higher emissions from the oceans (see Table 6). Since
29 the emission rate calculated in the model depends strongly on the 10-m wind speed (Sect.
30 2.2.9), the emissions at high and mid-latitudes are highest during the winter season.
31 Moreover, small differences in surface winds over the oceans may introduce substantial

1 differences in the emissions. A comparison of the sea-salt emissions in both simulations
2 shows that EC-Earth produces higher emissions over large parts of the northern Pacific and
3 the Southern Ocean (not shown).

4 In the tropics, the differences in the aerosol distributions between the two simulations also
5 reflect a shift in the location of the inter-tropical convergence zone (ITCZ). In particular,
6 during boreal winter a southward shift in the aerosol loads can be observed over central
7 Africa and the tropical Atlantic for all components except sea salt. As the mineral-dust and
8 biomass burning emissions are the same in both simulations, this shift in the aerosol
9 distributions is likely caused by differences in the location of the ITCZ and in the associated
10 transport.

11 The global burdens of the different aerosol components from both simulations are generally
12 on the low side of the ranges obtained in recent model intercomparison studies (see Table 6).
13 On the other hand, their atmospheric lifetimes and the relevant dry and wet deposition rates
14 are well within the multi-model estimates from these studies. This indicates that the emissions
15 of aerosols and/or aerosol precursors are lower in our simulations. As can be verified in
16 Tables 6 and A6, the emissions of sea salt and DMS in our simulations are indeed on the low
17 side of the ranges used in AeroCom and ACCMIP, respectively. Moreover, the volcanic
18 sulphur emissions applied in our model are substantially lower than in the ACCMIP models.

19 The aerosol optical depth (AOD) field from the EC-Earth simulation is quantitatively similar
20 to the result obtained with ERA-Interim (see Fig. 13). The spatial correlation between the
21 multi-annual mean AOD fields from the two simulations is 0.97 and the global mean AOD
22 values differ by only 3%.

23 EC-Earth gives somewhat higher values at high latitudes, especially during winter and spring.
24 This is primarily due to a higher contribution from sea salt. The higher AOD values simulated
25 by EC-Earth in the Arctic are in somewhat better agreement with ground-based and satellite
26 measurements as well as reanalysis data (see Von Hardenberg et al., 2012).

27 In the tropics and at mid-latitudes the AOD differences between the two simulations can be
28 positive or negative, depending on the location and the season. During boreal winter a
29 southward shift in the AOD pattern can be observed over central Africa and the tropical
30 Atlantic. This is primarily due to a shift in the contributions from mineral dust and biomass
31 burning (see Fig. A1 in the Appendix), and reflects a shift in the location of the ITCZ. During
32 boreal summer, EC-Earth gives smaller AOD values over large parts of northern Africa and

1 the tropical Atlantic. In the NH this can mainly be attributed to a lower contribution from
2 mineral dust, over the southern equatorial Atlantic to a lower contribution from biomass
3 burning. In most of western Africa and the Sahel, EC-Earth produces lower AOD values than
4 the ERA-Interim simulation in all seasons. EC-Earth also produces somewhat lower AOD
5 values over India and the Arabian Sea from summer to winter, especially during the summer
6 season. This is mainly caused by lower contributions from mineral dust and sulphate (see Fig.
7 A1).

8 We have evaluated the AOD fields from both simulations against remote sensing data from
9 the Moderate Resolution Imaging Spectroradiometer (MODIS) instrument aboard the Terra
10 and Aqua satellites, part of NASA's Earth Observing System (EOS). For this analysis, we
11 have used the Level-3 monthly gridded AOD fields from MODIS collection 5.1, which are
12 provided at $1^\circ \times 1^\circ$ resolution. We have included the data for the years 2000-2009 and
13 averaged the results from the Terra and Aqua data products. The resulting mean fields were
14 subsequently coarsened to the $3^\circ \times 2^\circ$ grid of TM5.

15 Both simulations strongly underestimate the AOD values retrieved from MODIS over almost
16 the entire globe (Fig. 14). The mean bias is -0.083 for EC-Earth and -0.086 for the ERA-
17 Interim simulation, which is 53% and 54%, respectively, of the retrieved mean value of 0.158.
18 EC-Earth gives a smaller mean bias than the ERA-Interim simulation from boreal autumn to
19 spring, especially in the winter season (see Table A7).

20 There are some land regions where the simulations produce higher values than observed. This
21 is for instance the case over large parts of Australia, which also include desert areas, in all
22 seasons. As it is difficult to accurately retrieve AOD over deserts because of the high
23 reflectivity of the surface, biases in these regions can be due to model biases as well as errors
24 in the MODIS retrieval. The simulations also give higher AOD values over the southeastern
25 United States during boreal winter and autumn, over southern parts Central America during
26 boreal winter, over eastern parts of South America during austral winter and autumn, and over
27 parts of South-Africa during austral summer and autumn. We have checked that these biases
28 are not caused by the fact that in the reference EC-Earth simulation and the corresponding
29 ERA-Interim simulation the emissions from anthropogenic activities and biomass burning
30 were fixed to their values for the year 2000. Biases of similar magnitude are found in the
31 ERA-Interim simulation with emissions varying from year to year. In other parts of the world,
32 the simulated AOD values are generally much lower than the MODIS values.

1 In all seasons, the AOD fields from the simulations correlate well with the distributions
2 observed by MODIS (Table A7). The spatial correlation between the simulated and observed
3 multi-annual mean AOD fields is 0.80 for EC-Earth and 0.79 for ERA-Interim, which means
4 that 65% and 62%, respectively, of the observed spatial variability is captured by the
5 simulations. The observed spatial distribution is slightly better represented by EC-Earth than
6 by the ERA-Interim simulation during boreal winter and autumn, and slightly worse during
7 boreal summer.

8 We have also calculated the temporal correlation between the simulated and observed
9 monthly mean AOD values as a function of geographical location. As we have only included
10 locations where observations are available for every month of the year, the analysis is
11 restricted to the tropics and mid-latitudes. The resulting seasonal correlation map shows
12 strong spatial variability where regions with high correlations are intermixed with regions
13 with low correlation (Fig. 15, left panel). Moreover, in some regions the observed seasonal
14 cycle is better represented by EC-Earth, in other regions by the ERA-Interim simulation (Fig.
15 15, right panel). For instance, EC-Earth gives higher correlations over the southern equatorial
16 Atlantic and the southern Indian Ocean, but lower correlations over the northern equatorial
17 Atlantic, the Arabian Sea, the Bay of Bengal, and Indonesia.

18 Finally, we compare the contributions from the individual aerosol components to the global
19 mean optical depth in our simulations with estimates from an aerosol reanalysis produced
20 within the MACC project. The MACC reanalysis system for aerosols is based on the IFS
21 meteorological assimilation system with an integrated aerosol module (Morcrette et al., 2009)
22 and uses MODIS retrievals of total AOD at 550 nm to further constrain the aerosol simulation
23 (Benedetti et al., 2009). Because total AOD is assimilated in the reanalysis, it provides a
24 valuable reference dataset for evaluating global aerosol models.

25 Compared to the results from the MACC reanalysis, our simulations underestimate the global
26 mean AOD contributions for all components (see Table 6). Both in the reanalysis and in our
27 simulations, the components that contribute most to the global mean AOD at 550 nm are sea
28 salt, sulphate and mineral dust, followed by organic aerosols. Compared to the reanalysis
29 results, the contributions from these components are underestimated by a factor 2 to 3. It is
30 important to note, however, that the distribution of AOD over the individual aerosol
31 components in the reanalysis is also subject to model uncertainty.

1 **5 Discussion and conclusions**

2 We have integrated the atmospheric chemistry and transport model TM5 into the global
3 climate model EC-Earth. The system allows for two-way exchange of fields between TM5
4 and IFS, the atmospheric GCM of EC-Earth. Here we have tested the system in one-way
5 coupled configuration. We have carried out a decadal simulation of tropospheric chemistry
6 and aerosols for present-day conditions, and calculated chemical budgets and climatologies of
7 tracer concentrations and aerosol optical depth. We have evaluated the results against
8 corresponding TM5 simulations driven offline by meteorological fields from 1) the ERA-
9 Interim reanalysis and 2) the EC-Earth model itself, as well as against various observational
10 datasets.

11 Differences in transport have been diagnosed from the simulated radon-222 concentrations.
12 Compared to the offline simulations, the online-coupled system exhibits more efficient
13 vertical mixing in the troposphere. This is due to the different treatment of cumulus
14 convection in the online-coupled system, in which the relevant convective fields are passed
15 from IFS to TM5. In the offline simulations these fields are calculated diagnostically based on
16 a somewhat outdated parameterization (Tiedtke, 1989). The stronger mixing characteristics
17 seen in the coupled system likely reflect improvements in the convection scheme made in
18 more recent cycles of IFS (Bechtold et al., 2008; Jung et al., 2010).

19 Compared to the ERA-Interim simulation, the oxidizing capacity in EC-Earth is lower in large
20 parts of the tropical and subtropical troposphere, and higher in the extratropics. The lower
21 oxidizing capacity in the tropics and subtropics is primarily driven by the model's cold bias in
22 these regions, which results in a lower specific humidity. As a consequence, the atmospheric
23 lifetime of CH₄ is 7% longer in EC-Earth than in the ERA-Interim simulation: EC-Earth gives
24 a lifetime of 9.4 years, while the ERA-Interim simulation gives 8.8 years. Both values are
25 well within the range 9.1 ± 0.9 years, recently estimated by Prather et al. (2012).

26 Differences in vertical mixing and oxidizing capacity also affect the distribution of other
27 chemically reactive gases, such as CO and O₃. Compared to the ERA-Interim simulation, the
28 total chemical production and destruction of CO are lower in EC-Earth, resulting in very
29 similar total amounts and a 4% longer lifetime. On the other hand, EC-Earth gives lower CO
30 concentrations in the NH extratropics, due to faster chemical destruction in this region. This
31 leads to a somewhat stronger underestimation of the surface concentrations measured by the
32 NOAA GMD flask sampling network. In both configurations, the total chemical production of

1 CO in TM5 is below the range of model estimates reported by Shindell et al. (2006b),
2 suggesting that the CBM4 chemistry scheme underestimates the production of CO from
3 NMVOCs.

4 The influx of O₃ from the stratosphere to the troposphere is 14% higher in EC-Earth than in
5 the ERA-Interim simulation, and is in slightly better agreement with other modelling studies
6 as well as observational estimates. Moreover, the cold bias in EC-Earth tends to slow down
7 the chemical destruction of O₃ in the troposphere, resulting in an increase in the net chemical
8 production of O₃ in the troposphere. Furthermore, enhanced vertical mixing tends to
9 redistribute O₃ from higher to lower parts of the troposphere. Overall, the total amount of O₃
10 in the troposphere is 6% higher in EC-Earth, in better agreement with the ranges reported in
11 the model intercomparison studies by Stevenson et al. (2006) and Young et al. (2013). The
12 average lifetime of tropospheric O₃ increases by 7% from 23.9 days with ERA-Interim to 25.5
13 days in EC-Earth. The latter value is higher than the multi-model range reported by Stevenson
14 et al. (2006) and is on the high side of the model results reported by Young et al. (2013).

15 Overall, EC-Earth produces lower concentrations in the upper parts of the tropical and
16 subtropical troposphere and higher concentrations in most of the lower and middle
17 troposphere, especially in the extratropics. Similarly, EC-Earth gives higher surface
18 concentrations in most of the world, with the exception of some regions located in the tropics
19 and subtropics. This results in a 15% higher total deposition of O₃ compared to the ERA-
20 Interim simulation.

21 The simulated O₃ concentrations have been evaluated against a vertically resolved, zonal and
22 monthly mean dataset produced from satellite and ozonesonde observations (Hassler et al.,
23 2009; www.bodeker.com) and against surface measurements from various networks. Both
24 simulations show reasonable agreement with the observational datasets and both have their
25 relative strengths and weaknesses depending on the location and the season. EC-Earth tends
26 to overestimate the O₃ concentrations in the lower troposphere and at the surface in the NH
27 extratropics during boreal summer and fall. Note that this is partly a resolution effect. It is
28 well known that the relatively coarse horizontal resolutions applied in the current generation
29 of global chemistry models tends to overestimate the production of O₃ in the boundary layer
30 (Wild and Prather, 2006). During boreal winter and spring, on the other hand, as well as in the
31 SH extratropics, the higher surface and lower-tropospheric concentrations produced by EC-

1 Earth are generally in better agreement with the measurements than the results from the ERA-
2 Interim simulation.

3 The aerosol climatologies from both simulations are quantitatively very similar. EC-Earth
4 gives somewhat higher AOD values at high latitudes especially during winter and spring,
5 mainly due to a higher contribution from sea salt. The global burdens of the various aerosol
6 components are generally on the low side of the ranges obtained in recent model
7 intercomparison studies like AeroCom and ACCMIP. This is partly caused by lower natural
8 emissions, in particular of sea salt, DMS and volcanic sulphur. However, previous studies
9 indicate that the wet removal of aerosols in TM5 may also be too fast (see Aan de Brugh et
10 al., 2011). A comparison with AOD fields retrieved from MODIS shows that the simulations
11 capture a large part of the spatial variability, but strongly underestimate the observed values
12 over almost the entire globe.

13 In the system we have developed, the description of atmospheric chemistry and aerosols is not
14 integrated into IFS. Instead, it is taken care of by a separate module, TM5, which is coupled
15 to IFS through OASIS. Previously, a coupled TM5-IFS system using OASIS was developed
16 within the GEMS project (Flemming et al., 2009). However, the data exchange in that system
17 was designed specifically for short-term forecasts and reanalysis purposes, in which chemical
18 data assimilation plays a central role. We also needed to completely redo the technical
19 implementation of the coupling, because the GEMS system made use of an OASIS version
20 that is incompatible with the version used in EC-Earth.

21 A more recent activity is the development of the Composition-IFS (C-IFS) model within the
22 MACC project. In C-IFS the description of atmospheric chemistry is integrated into the IFS
23 model (Flemming et al., 2012). As part of this development, the aerosol scheme of IFS
24 (Morcrette et al., 2009) is also being upgraded and coupled to the gas-phase chemistry. The
25 main advantages of C-IFS compared to the system we have developed are (1) that the
26 description of chemistry and aerosols can be more tightly coupled to the relevant dynamical
27 and physical processes described in IFS, and (2) that there is no external exchange of data.
28 Conversely, the main advantages of our system compared to C-IFS are (1) that the tracer
29 transport in TM5 is locally mass conserving, (2) that TM5 can be run at a lower resolution
30 than IFS. The latter points are crucial to enable long-term climate integrations, but are less
31 relevant for the short-term forecast and reanalysis simulations which C-IFS has been

1 developed for, and (3) that the TM5 module can be kept more easily up to date with the
2 offline version.

3 The work presented in this study is the first step in the development of EC-Earth into an Earth
4 system model with fully interactive chemistry and aerosols. A number of developments are
5 planned for the near future. First, to improve the simulation of aerosol burdens and optical
6 depths, we intend to improve the representation of natural emissions of aerosols and aerosol
7 precursors, and revisit the description of the wet removal processes in TM5. At the same time,
8 the calculation of photolysis rates will be updated following Williams et al. (2006; 2012) and
9 coupled to the simulated aerosols. Moreover, the representation of heterogeneous chemistry
10 will be improved following Huijnen et al. (2014). Another line of work focuses on improving
11 the representation of stratospheric chemistry through simplified schemes, but this is more a
12 long-term project.

13 Meanwhile, the performance and scalability of TM5 has recently been strongly improved, and
14 will be further improved in the near future. The new, massively parallel model (named TM5-
15 mp) is currently being implemented into the latest version of EC-Earth (version 3.0). In this
16 system the couplings with the radiation and cloud schemes of IFS will be made, and the
17 exchange period will be reduced from 6 to 3 hours or less.

18 A parallel development is the introduction of a carbon cycle in EC-Earth, based on the
19 dynamic vegetation model LPJ-GUESS (Smith et al., 2001; Weiss et al., submitted) and the
20 biogeochemical component of NEMO. As part of these developments, various couplings will
21 be made between TM5 and the terrestrial and marine biosphere.

1 **Appendix**

2

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- 13

1 Table 1. Meteorological data and surface property fields transferred from IFS to TM5.

2

Field	Type
<u>Three-dimensional</u>	
Wind divergence/vorticity	Instantaneous
Temperature	Instantaneous
Specific humidity	Instantaneous
Cloud liquid/ice water content	Instantaneous
Cloud fraction	Instantaneous
Overhead/underfeet cloud fraction	Instantaneous
Updraught/downdraught convective air mass flux	Average
Updraught/downdraught convective air mass detrainment rate	Average
<u>Two-dimensional</u>	
Surface pressure	Instantaneous
10-m wind (west-east/south-north components)	Instantaneous
2-m temperature	Instantaneous
2-m dewpoint temperature	Instantaneous
Surface east-west/north-south momentum stress	Average
Surface sensible/latent heat flux	Average
Surface solar radiation	Average
Stratiform precipitation as rain	Average
Convective precipitation as rain	Average
Skin reservoir water content	Instantaneous
Snow depth	Instantaneous
Soil wetness in top soil layer	Instantaneous
Low/high vegetation cover fractions	Instantaneous
Vegetation type fractions	Instantaneous
Surface roughness	Instantaneous
Surface orography	Constant
Land/sea fraction	Constant
Sea-ice fraction	Instantaneous

3

1 Table 2. Overview of the decadal simulations used in this study.

2

Simulation	Focus	Emissions from anthropogenic activities and biomass burning	Number of vertical levels in TM5/IFS	Highest update frequency of meteorological fields in TM5	Temporal interpolation of instantaneous meteorological fields in TM5	Convective fields
EC-Earth	Chemistry and aerosols, radon-222	Year 2000	31/62	6 hourly	No	Received from IFS
ERA-Interim	Chemistry and aerosols	Year 2000	34/60	3 hourly	Linear	Diagnosed offline
ERA-Interim, varying emissions	Chemistry and aerosols	2000-2009	34/60	3 hourly	Linear	Diagnosed offline
Offline EC-Earth	Radon-222	-	31/62	6 hourly	Linear	Diagnosed offline

3

1 Table 3. Contributions to the chemical production of OH in the troposphere (Tg OH yr^{-1}) for
 2 the reference EC-Earth simulation and the corresponding ERA-Interim simulation. The
 3 contributions from 90-30°S, 30°S-30°N, and 30-90°N are given by the numbers between
 4 parentheses. Results have been obtained from a monthly analysis with a fixed tropopause
 5 level, set to the uppermost model layer for which the monthly mean O_3 mixing ratio is below
 6 150 ppbv. Standard deviations are based on the simulated interannual variability.

7

Reaction	EC-Earth	ERA-Interim
$\text{O}(^1\text{D}) + \text{H}_2\text{O}$	1432 (112 / 1120 / 199)	1554 (102 / 1243 / 208)
$\text{NO} + \text{HO}_2$	994 (81 / 649 / 264)	1002 (65 / 682 / 255)
$\text{O}_3 + \text{HO}_2$	393 (42 / 249 / 102)	393 (37 / 263 / 93)
$\text{H}_2\text{O}_2 + \text{h}\nu$	197 (18 / 149 / 29)	230 (20 / 175 / 34)
Other	168 (12 / 139 / 17)	176 (13 / 144 / 18)
Total	3184 ± 20 (266 / 2307 / 611)	3355 ± 30 (238 / 2508 / 608)

8

1 Table 4. Contributions to the budget of CO in the atmosphere (Tg CO yr⁻¹), together with the
 2 tropospheric and atmospheric burdens of CO (Tg CO), and the atmospheric lifetime of CO
 3 (days) for the reference EC-Earth simulation and the corresponding ERA-Interim simulation.
 4 The contributions from 90-30°S, 30°S-30°N, and 30-90°N are given by the numbers between
 5 parentheses. Results have been obtained from a monthly analysis with a fixed tropopause
 6 level, set to the uppermost model layer for which the monthly mean O₃ mixing ratio is below
 7 150 ppbv. Standard deviations are based on the simulated interannual variability.

8

	EC-Earth	ERA-Interim
Emissions	1166 (23 / 762 / 381)	1166 (23 / 762 / 381)
Tropospheric chemical production	1105 (81 / 838 / 186)	1170 (76 / 905 / 189)
Total gain*	2284 (107 / 1606 / 572)	2351 (102 / 1674 / 574)
Dry deposition	173 ± 1.0 (6 / 107 / 60)	180 ± 1.5 (6 / 105 / 69)
Tropospheric chemical destruction	2065 (191 / 1447 / 427)	2129 (166 / 1541 / 422)
Total loss*	2284 (208 / 1570 / 506)	2352 (184 / 1662 / 506)
Tropospheric burden	316 ± 2.0 (53 / 179 / 84)	317 ± 3.3 (53 / 175 / 88)
Atmospheric burden	341 ± 2.5 (61 / 185 / 95)	338 ± 3.7 (59 / 181 / 97)
Atmospheric lifetime	54.6	52.5

9

10 * The total gain and loss also include small contributions from, respectively, chemical
 11 production and destruction in the stratosphere.

1 Table 5. Contributions to the budget of O₃ in the troposphere (Tg O₃ yr⁻¹), together with the
 2 tropospheric O₃ burden (Tg O₃), and the tropospheric O₃ lifetime (days) for the reference EC-
 3 Earth simulation and the corresponding ERA-Interim simulation. The contributions from 90-
 4 30°S, 30°S-30°N, and 30-90°N are given by the numbers between parentheses. Results have
 5 been obtained from a monthly analysis with a fixed tropopause level, set to the uppermost
 6 model layer for which the monthly mean O₃ mixing ratio is below 150 ppbv. Standard
 7 deviations are based on the simulated interannual variability.

8

	EC-Earth	ERA-Interim
Chemical production	4328 ± 17 (339 / 2890 / 1099)	4419 ± 38 (278 / 3070 / 1071)
Chemical destruction	3698 ± 24 (327 / 2656 / 714)	3873 ± 39 (291 / 2896 / 687)
Dry deposition	978 ± 3.2 (115 / 471 / 392)	851 ± 4.1 (84 / 425 / 341)
Stratosphere-troposphere exchange	349 ± 10	306 ± 15
Burden	327 ± 1.3 (66 / 161 / 100)	309 ± 1.7 (59 / 162 / 88)
Lifetime	25.5	23.9

9

1 Table 6. Global budgets, burdens, and lifetimes of the different aerosol components for the
 2 reference EC-Earth simulation and the corresponding ERA-Interim simulation. Estimates of
 3 their contributions to the optical depth at 550 nm are also included. The results are compared
 4 with estimates from other studies.

	EC-Earth	ERA-Interim	Other studies
Sulphate			
Emissions (Tg S yr ⁻¹)		1.47 ^a	
Chemical production (Tg S yr ⁻¹)			
SO ₂ + OH	7.80	8.24	
S(IV) + H ₂ O ₂	23.9	24.2	
S(IV) + O ₃	5.42	4.58	
Burden (Tg S)	0.522	0.498	0.67 ± 0.17 ^d
Lifetime (days)	4.93	4.73	5.0 ± 2.0 ^d 4.1 ± 0.7 ^e
Dry deposition rate (day ⁻¹)	4.68 10 ⁻³	4.57 10 ⁻³	0.03 ± 0.02 ^e
Wet deposition rate (day ⁻¹)	0.198	0.207	0.22 ± 0.05 ^e
Optical depth	2.13 10 ⁻²	2.08 10 ⁻²	0.044 ^f
Black carbon			
Emissions (Tg yr ⁻¹)		7.77	
Burden (Tg)	0.145	0.149	0.16 ± 0.07 ^d
Lifetime (days)	6.81	6.99	7.4 ± 3.4 ^d 7.1 ± 2.3 ^e
Dry deposition rate (day ⁻¹)	6.17 10 ⁻³	5.65 10 ⁻³	0.03 ± 0.02 ^e
Wet deposition rate (day ⁻¹)	0.141	0.137	0.12 ± 0.04 ^e
Optical depth	1.11 10 ⁻³	1.15 10 ⁻³	0.0085 ^f
Organic aerosols			
Emissions (Tg yr ⁻¹)		69.5 ^b	
Burden (Tg)	1.18	1.16	1.6 ± 0.8 ^g
Lifetime (days)	6.18	6.08	5.7 ± 1.6 ^g
Dry deposition rate (day ⁻¹)	5.23 10 ⁻³	4.69 10 ⁻³	0.029 ± 0.046 ^g
Wet deposition rate (day ⁻¹)	0.157	0.160	0.16 ± 0.04 ^g
Optical depth	9.28 10 ⁻³	9.29 10 ⁻³	0.024 ^f
Nitrate			
Burden (Tg N)	2.29 10 ⁻²	1.27 10 ⁻²	0.1 ± 0.0 ^h
Optical depth	6.82 10 ⁻⁴	3.99 10 ⁻⁴	0.007 ± 0.001 ^h
Sea salt			
Emissions (Pg yr ⁻¹)	7.35 ± 0.11 ^c	6.83 ± 0.09 ^c	8.2 ± 8.2 ⁱ 16.6 ± 33.0 ^e
Burden (Tg)	6.81	6.17	7.9 ± 5.5 ⁱ 7.5 ± 4.1 ^e
Lifetime (days)	0.338	0.330	0.48 ± 0.28 ^e
Dry deposition rate (day ⁻¹)	2.42	2.40	4.3 ± 9.4 ^e
Wet deposition rate (day ⁻¹)	0.538	0.630	0.79 ± 0.61 ^e
Optical depth	2.66 10 ⁻²	2.35 10 ⁻²	0.055 ± 0.016 ^j
Mineral dust			
Emissions (Pg yr ⁻¹)		1.78	1.84 ± 0.90 ^e
Burden (Tg)	12.1	13.4	19.2 ± 7.7 ^e
Lifetime (days)	2.48	2.75	4.1 ± 1.8 ^e
Dry deposition rate (day ⁻¹)	0.311	0.287	0.23 ± 0.19 ^e
Wet deposition rate (day ⁻¹)	9.20 10 ⁻²	7.60 10 ⁻²	0.08 ± 0.03 ^e
Optical depth	1.55 10 ⁻²	1.71 10 ⁻²	0.043 ± 0.014 ^j

- 1 ^a Includes 0.12 Tg S yr⁻¹ from volcanoes.
- 2 ^b Includes 19.1 Tg yr⁻¹ representing SOA (see Sect. 2.2.5).
- 3 ^c Standard deviations calculated from the simulated interannual variability.
- 4 ^d ACCMIP multi-model means and standard deviations for the year 2000 from Shindell et al.
5 (2013).
- 6 ^e AeroCom phase-I multi-model means and standard deviations from Textor et al. (2006).
- 7 ^f MACC reanalysis (Benedetti et al., 2009) results for the year 2003 as provided on the
8 AeroCom phase-II web interface (http://aerocom.met.no/cgi-bin/aerocom/surfobs_annualrs.pl,
9 simulation labelled 'ECMWF_FBOV').
- 10 ^g AeroCom phase-II multi-model means and standard deviations from Tsigaridis et al. (2014).
- 11 ^h Results for 1998-2002 from a CMIP5 simulation with the Hadley Centre climate model
12 HadGEM2-ES by Bellouin et al. (2011).
- 13 ⁱ AeroCom phase-I multi-model means and standard deviations from Textor et al. (2007),
14 based on a selection of seven models from Textor et al. (2006).
- 15 ^j MACC reanalysis results with uncertainty estimates from Bellouin et al. (2013).

1 Table A1. Vertical distributions applied to the different emission sources. The emissions from
 2 aviation are distributed as provided in the CMIP5 emission dataset.

Vertical distribution type	Emission sector/source	Fraction per height range (%)					
		0-30 m	30-100 m	100-300 m	300-600 m	600-1000 m	1000-2000 m
Energy	Energy production and distribution	0	10	70	20	0	0
Industrial	Industrial processes and combustion	10	20	60	10	0	0
Residential	Residential and commercial combustion	40	40	20	0	0	0
Waste	Waste treatment and disposal	10	20	40	30	0	0
Near- surface ¹		80	20	0	0	0	0
Surface ²		100	0	0	0	0	0
Volcanic	Volcanic SO ₂	0	0	10	30	40	20
		0-100 m	100-500 m	500-1000 m	1000-2000 m	2000-3000 m	3000-6000 m
Forest fires							
Tropical (30° S-30° N)		20	20	20	40	0	0
Temperate (30-60° S/N)		20	20	20	40	0	0
High-latitude (60-90° S/N)		10	10	20	20	40	0

3
 4 ¹ Includes solvent production and use, maritime transport, agricultural waste burning,
 5 grassland fires, SOA, and mineral dust.
 6 ² Includes land transport, agriculture, biogenic emissions from soils and vegetation, oceanic
 7 emissions, and radon-222.

1 Table A2. List of stations used for the evaluation of the simulated surface radon-222
 2 concentrations. The mean bias and root mean square errors (RMSE) in the simulated monthly
 3 mean concentrations are indicated for the online and offline EC-Earth simulations and the
 4 ERA-Interim simulation. Also given are the linear correlation coefficients between the
 5 simulated and measured monthly mean concentrations. A diurnal window has been applied to
 6 the measurements, except for Cape Grim. The results for the online EC-Earth simulation and
 7 the ERA-Interim simulation are separated by a slash; the results for the offline EC-Earth
 8 simulation are shown between parentheses.

Station	Lat. (deg)	Lon. (deg)	Height (m)	Period (years)	Window (hours LT, inclusive)	Bias (Bq/m ³)	RMSE (Bq/m ³)	Correlation Coefficient
Cape Grim, Tasmania	-40.68	144.69	90	1991-2012	Not Applied	0.11 / 0.30 (0.19)	0.27 / 0.33 (0.28)	0.47 / 0.93 (0.79)
Richmond, Australia	-33.62	150.75	24	2007-2012	14-18	-0.10 / 0.33 (0.07)	0.50 / 0.53 (0.59)	0.81 / 0.84 (0.62)
Mauna Loa, Hawaii	19.54	-155.58	4170	2004-2012	8-10	-0.03 / -0.04 (-0.03)	0.04 / 0.04 (0.04)	0.78 / 0.78 (0.78)
Gosan, Jeju Island, Korea	33.29	126.16	70	2001-2010	13-17	-0.97 / -0.82 (-0.96)	1.07 / 0.95 (1.10)	0.66 / 0.59 (0.45)
Sado Island, Japan	38.25	138.40	130	2002-2005	11-16	-0.77 / -0.43 (-0.62)	0.94 / 0.73 (0.84)	-0.70 / -0.71 (-0.64)
Hohenpeissenberg, Germany	47.80	11.02	985	1999-2005	14-18	-1.30 / 0.12 (-0.56)	1.45 / 0.83 (0.97)	0.26 / 0.52 (0.13)

9

1 Table A3. Global budgets of oxidized and reduced reactive nitrogen (Tg N yr^{-1}) for the
 2 reference EC-Earth simulation and the corresponding ERA-Interim simulation, compared
 3 with the ACCMIP multi-model means and standard deviations for the year 2000 from
 4 Lamarque et al. (2013b). The standard deviations indicated for the NO_x production by
 5 lightning in our simulations are calculated from the interannual variability.

	EC-Earth	ERA-Interim	ACCMIP
Oxidized nitrogen (NO_y)			
Total NO_x emissions	49.6	49.0	49 ± 3
NO_x emissions by soils		4.95	
NO_x production by lightning	6.45 ± 0.06	5.83 ± 0.21	6 ± 2
NO_y dry deposition	21.1	20.9	21 ± 7
NO_y wet deposition	29.6	28.8	29 ± 5
NO_y net chemical production	1.02	0.76	~ 1
Reduced nitrogen (NH_x)			
NH_3 emissions		48.5	49 ± 2
NH_3 dry deposition	21.9	22.3	15 ± 6
NH_4 dry deposition	0.0	0.0	5 ± 1
NH_3 wet deposition	8.23	8.14	7 ± 4
NH_4 wet deposition	18.1	17.8	23 ± 6
NH_x net chemical destruction	0.24	0.27	

6

1 Table A4. List of stations from the NOAA GMD network of flask measurements used for the
 2 evaluation of the simulated surface CO mixing ratios. The mean bias and root mean square
 3 errors (RMSE) in the simulated monthly mean mixing ratios are indicated for the reference
 4 EC-Earth simulation and the corresponding ERA-Interim simulation. Also given are the linear
 5 correlation coefficients between the simulated and measured monthly mean concentrations.
 6 The results for the EC-Earth and ERA-Interim simulations are separated by a slash.

Station	Lat. (deg)	Lon. (deg)	Height (m)	Period (years)	Bias (ppbv)	RMSE (ppbv)	Correlation Coefficient
South Pole	-89.98	-24.80	2810	2000-2009	-0.2 / 0.1	2.9 / 3.7	0.94 / 0.91
Halley Station, Antarctica	-75.58	-26.5	30	2000-2009	-1.2 / -0.6	2.4 / 2.7	0.97 / 0.94
Cape Grim, Tasmania	-40.68	144.69	94	2000-2009	6.5 / 6.4	8.8 / 8.7	0.70 / 0.67
Tutuila, American Samoa	-14.25	-170.56	42	2000-2009	-3.9 / -4.3	4.6 / 4.9	0.90 / 0.86
Ascension Island	-7.97	-14.40	85	2000-2009	-5.2 / -3.1	9.3 / 7.9	0.69 / 0.83
Ragged Point, Barbados	13.16	-59.43	15	2000-2009	-18.3 / -18.2	18.6 / 18.7	0.97 / 0.96
Mariana Islands, Guam	13.39	144.66	0	2000-2009	-19.5 / -15.9	19.6 / 16.9	0.99 / 0.96
Mauna Loa, Hawaii	19.54	-155.58	3397	2000-2009	-18.5 / -18.4	18.8 / 18.8	0.98 / 0.98
Izaña, Tenerife	28.31	-16.50	2373	2000-2009	-25.3 / -23.9	25.9 / 24.6	0.96 / 0.94
Tudor Hill, Bermuda	32.27	-64.88	30	2002-2009	-29.3 / -23.2	30.9 / 24.0	0.97 / 0.99
Terceira Island, Azores	38.77	-27.38	19	2000-2009	-33.9 / -27.6	34.5 / 28.1	0.95 / 0.96
Mace Head, Ireland	53.33	-9.90	5	2000-2009	-30.2 / -19.9	31.2 / 20.7	0.97 / 0.97
Cold Bay, Alaska	55.21	-162.72	21	2000-2009	-36.7 / -29.2	39.0 / 31.5	0.92 / 0.90
Storhofdi, Iceland	63.40	-20.29	118	2000-2009	-34.0 / -23.9	35.3 / 24.9	0.96 / 0.97
Barrow, Alaska	71.32	-156.61	11	2000-2009	-38.1 / -28.7	41.0 / 31.3	0.94 / 0.96
Ny-Ålesund, Spitsbergen	78.91	11.89	474	2000-2009	-39.0 / -28.0	41.0 / 29.5	0.97 / 0.98
Alert, Canada	82.45	-62.51	200	2000-2009	-38.9 / -29.4	41.4 / 31.5	0.95 / 0.97

7

1 Table A5. List of stations used for the evaluation of the simulated surface O₃ mixing ratios.
 2 The mean bias and root mean square errors (RMSE) in the simulated monthly mean mixing
 3 ratios are indicated for the reference EC-Earth simulation and the corresponding ERA-Interim
 4 simulation. Also given are the linear correlation coefficients between the simulated and
 5 measured monthly mean concentrations. The results for the EC-Earth and ERA-Interim
 6 simulations are separated by a slash.

7

Station	Lat. (deg)	Lon. (deg)	Height (m)	Period (years)	Bias (ppbv)	RMSE (ppbv)	Correlation Coefficient
South Pole	-89.98	-24.80	2810	2000-2009	-11.2 / -17.1	11.6 / 17.3	0.70 / 0.94
Arrival Heights, Antarctica	-77.83	166.20	250	2000-2008	-6.1 / -13.3	8.0 / 14.1	0.86 / 0.89
Neumayer, Antarctica	-70.65	-8.25	42	2000-2009	-3.3 / -10.5	5.6 / 11.1	0.86 / 0.97
Syowa, Antarctica	-69.00	39.58	16	2000-2009	-2.8 / -11.1	4.7 / 11.4	0.91 / 0.99
Cape Grim, Tasmania	-40.68	144.68	94	2000-2009	3.5 / -3.3	5.0 / 4.9	0.88 / 0.89
Tutuila, American Samoa	-14.25	-170.56	42	2000-2009	6.6 / 5.8	6.7 / 5.8	0.99 / 0.99
Ragged Point, Barbados	13.16	-59.43	15	2006-2009	12.8 / 8.0	13.1 / 8.3	0.79 / 0.91
Cape Verde	16.85	-24.87	10	2006-2009	11.1 / 5.9	12.5 / 7.3	0.31 / 0.68
Mauna Loa, Hawaii	19.54	-155.58	3397	2000-2009	2.6 / 2.7	3.0 / 3.1	0.95 / 0.96
Assekrem, Algeria	23.27	5.63	2710	2000-2001, 2003-2009	9.0 / 5.1	9.5 / 5.6	0.66 / 0.78
Mount Everest, Nepal	27.96	86.82	5079	2006-2009	10.3 / 8.9	12.1 / 10.5	0.78 / 0.81
Izaña, Tenerife	28.30	-16.50	2367	2000-2009	10.6 / 2.7	11.6 / 3.9	0.76 / 0.83
Tudor Hill, Bermuda	32.27	-64.88	30	2003-2009	6.2 / 1.7	7.9 / 5.8	0.87 / 0.85
Trinidad Head, California	41.05	-124.15	107	2002-2009	11.0 / 2.9	11.9 / 5.9	0.44 / 0.38
Jungfrauoch, Switzerland	46.55	7.99	3580	2000-2009	6.9 / -1.5	7.7 / 2.8	0.93 / 0.94
Payerne, Switzerland	46.82	6.95	490	2000-2009	9.4 / 1.8	10.1 / 4.3	0.93 / 0.93
Zugspitze, Germany	47.42	10.98	2960	2000-2002	7.9 / 0.1	8.7 / 3.5	0.90 / 0.90
Hohenpeissenberg, Germany	47.80	11.02	985	2000-2007	2.8 / -10.2	5.5 / 11.6	0.88 / 0.91
Mace Head, Ireland	53.33	-9.90	25	2000-2009	9.3 / 0.0	10.2 / 4.1	0.39 / 0.49
Storhofdi, Iceland	63.40	-20.29	118	2003-2009	5.5 / -6.2	7.2 / 7.1	0.40 / 0.69
Barrow, Alaska	71.32	-156.61	11	2000-2009	2.5 / -4.5	8.3 / 8.0	-0.26 / -0.16
Summit, Greenland	72.58	-38.48	3216	2000-2009	-4.0 / -13.1	5.2 / 13.5	0.63 / 0.68

8

1 Table A6. Global budgets of DMS, MSA, SO₂ and total reactive sulphur together with the
2 global burden and lifetime of SO₂ for the reference EC-Earth simulation and the
3 corresponding ERA-Interim simulation. For comparison we have also included ACCMIP
4 multi-model means and standard deviations of emissions and depositions for the year 2000
5 from Lamarque et al. (2013b). For the gas-phase SO₂ and aqueous-phase S(IV) oxidation
6 reactions, which result in the production of SO₄, the numbers between parentheses give the
7 separate contributions from 90-30°S, 30°S-30°N, and 30-90°N.

	EC-Earth	ERA-Interim	ACCMIP
DMS			
Emissions (Tg S yr ⁻¹)	19.4 ± 0.2 ^a	19.1 ± 0.3 ^a	23 ± 5
Chemical destruction (Tg S yr ⁻¹)			
DMS + OH	14.4	14.5	
DMS + NO ₃	4.93	4.60	
MSA			
Chemical production (Tg S yr ⁻¹)			
DMS + OH	1.87	1.79	
Wet deposition (Tg S yr ⁻¹)	1.87	1.79	~ 2
SO₂			
Total emissions (Tg S yr ⁻¹)		57.2	65 ± 2
Volcanic emissions (Tg S yr ⁻¹)		4.67	~ 12 ± 2
Chemical production (Tg S yr ⁻¹)			
DMS + OH	12.6	12.7	
DMS + NO ₃	4.93	4.60	
Chemical destruction (Tg S yr ⁻¹)			
SO ₂ + OH	7.80 (0.43/ 3.84/ 3.53)	8.24 (0.34 / 4.39/ 3.52)	
S(IV) + H ₂ O ₂	23.9 (2.75 / 12.7 / 8.46)	24.2 (2.63 / 13.4/ 8.11)	
S(IV) + O ₃	5.42 (1.01 / 2.58 / 1.83)	4.58 (0.91 / 2.02/ 1.64)	
Dry deposition (Tg S yr ⁻¹)	26.1	26.3	
Wet deposition (Tg S yr ⁻¹)	11.5	11.2	
Burden (Tg S)	0.317	0.261	
Lifetime (days)	1.55	1.28	
Total reactive sulphur			
Emissions (Tg S yr ⁻¹)	78.1	77.8	89 ± 6
Dry deposition (Tg S yr ⁻¹)	27.0	27.2	37 ± 10
Wet deposition (Tg S yr ⁻¹)	51.1	50.7	52 ± 8

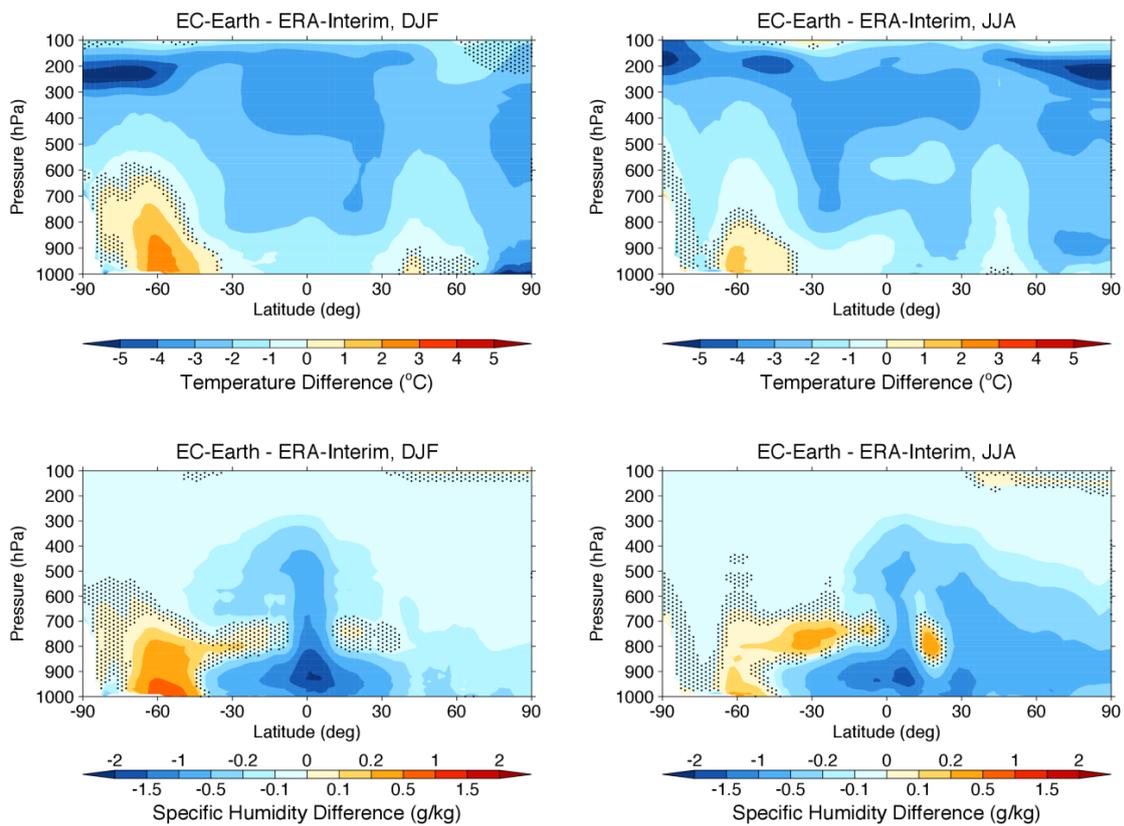
8 ^a Standard deviations calculated from the simulated interannual variability.

1 Table A7. Mean biases and linear correlation coefficients between the time averaged AOD
 2 fields from the simulations and the MODIS Level-3 product coarsened to the same resolution
 3 and averaged over the years 2000-2009. Grid areas where, for all consecutive years, the
 4 MODIS data are missing for one or more months during the year or season of interest have
 5 not been included in the calculations. The results for the EC-Earth and ERA-Interim
 6 simulations are separated by a slash.

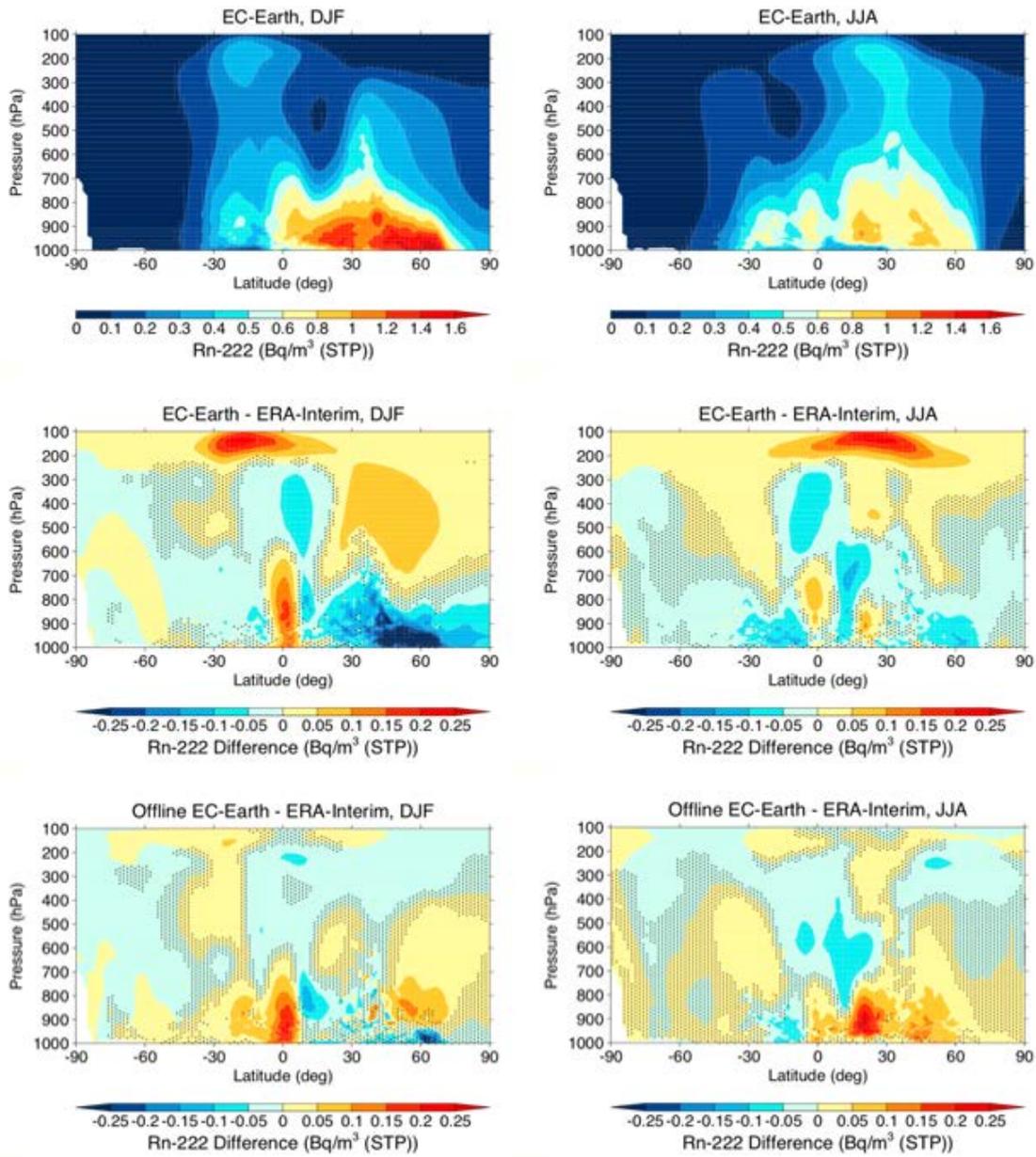
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Season	Mean bias	Correlation Coefficient	Observed Area (%)
Annual	-0.083 / -0.086	0.80 / 0.79	71.8
DJF	-0.073 / -0.080	0.76 / 0.72	82.2
MAM	-0.093 / -0.095	0.78 / 0.78	82.1
JJA	-0.087 / -0.085	0.72 / 0.78	84.1
SON	-0.078 / -0.080	0.62 / 0.60	84.8

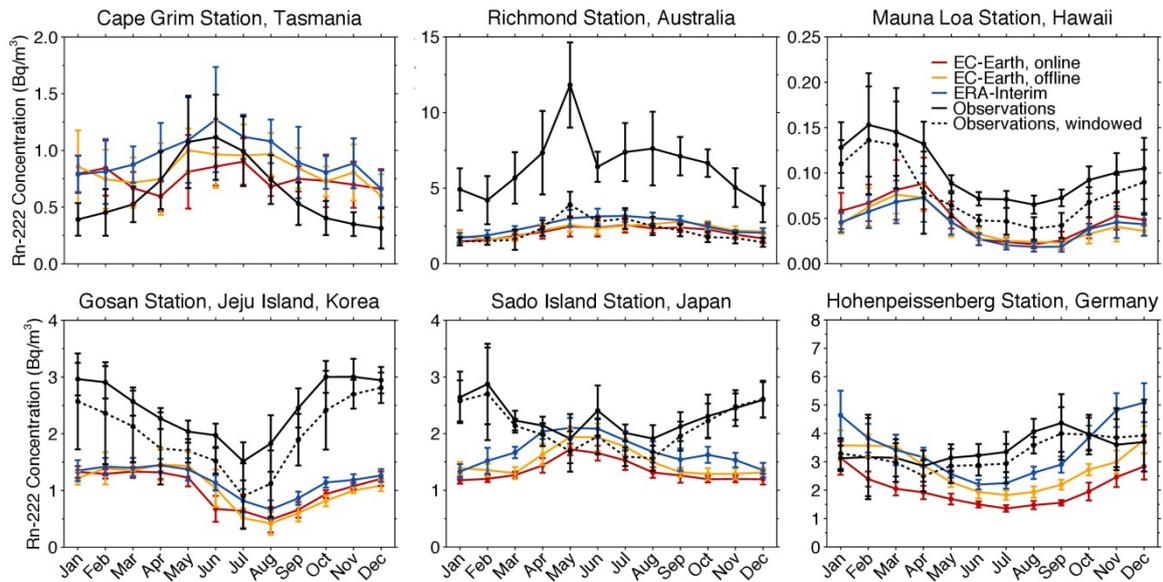
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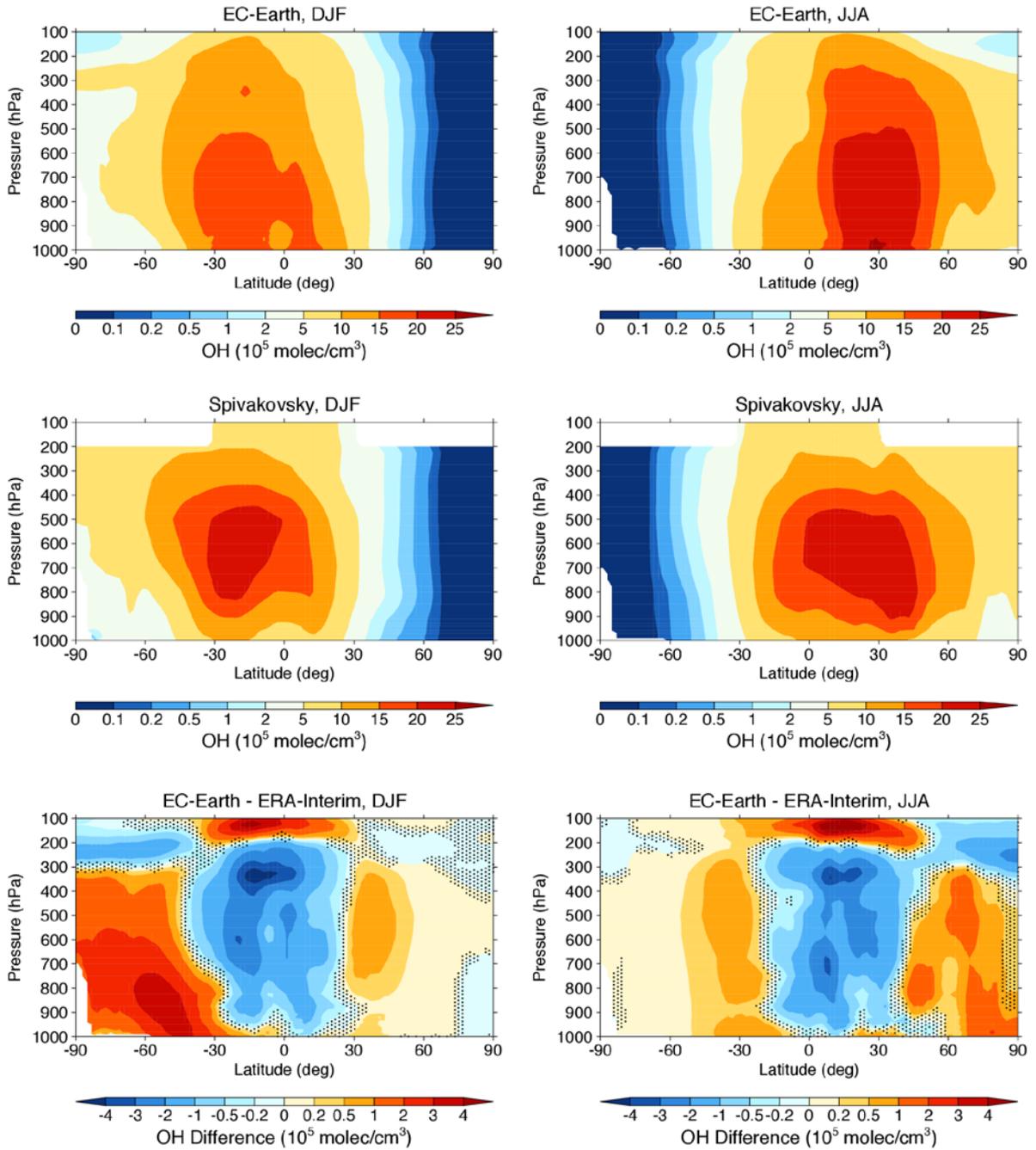
1
 2 Figure 1. Zonal mean bias in temperature (top) and specific humidity (bottom) in EC-Earth
 3 compared to ERA-Interim for boreal winter (December-February, left) and boreal summer
 4 (June-August, right) for the period 2000-2009. Regions where the differences are not
 5 significant at the 5% level are indicated by the stippled areas.



1
 2 Figure 2. Zonal mean radon-222 concentrations in the online EC-Earth simulation (top) and
 3 the differences compared to the ERA-Interim (middle) and offline EC-Earth (bottom)
 4 simulations for boreal winter (left) and boreal summer (right).

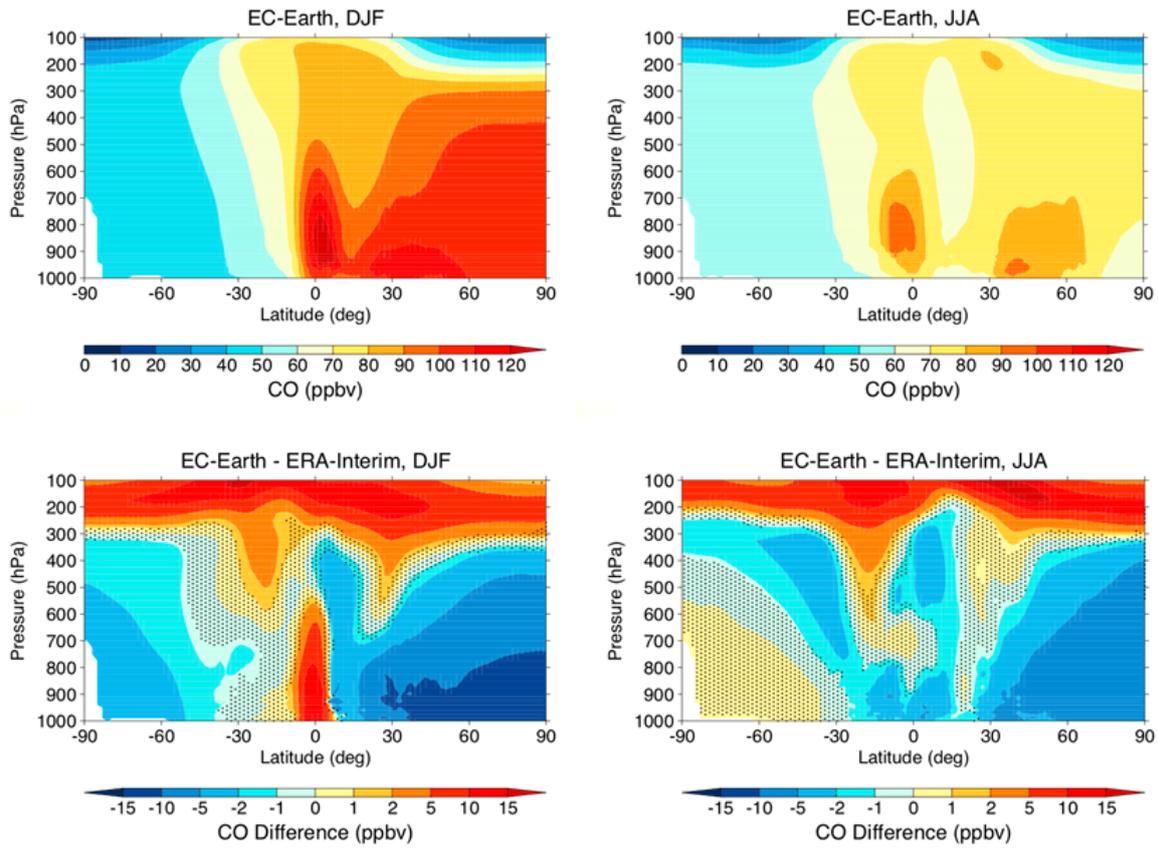


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 2 Figure 3. Comparison of monthly mean surface radon-222 concentrations from the online and
 3 offline EC-Earth simulations (solid red and orange lines, respectively) and the ERA-Interim
 4 simulation (solid blue lines) against surface measurements (black lines). The observational
 5 means shown by the dashed lines only include the measurements made within a certain station
 6 dependent diurnal window (see Table A2).

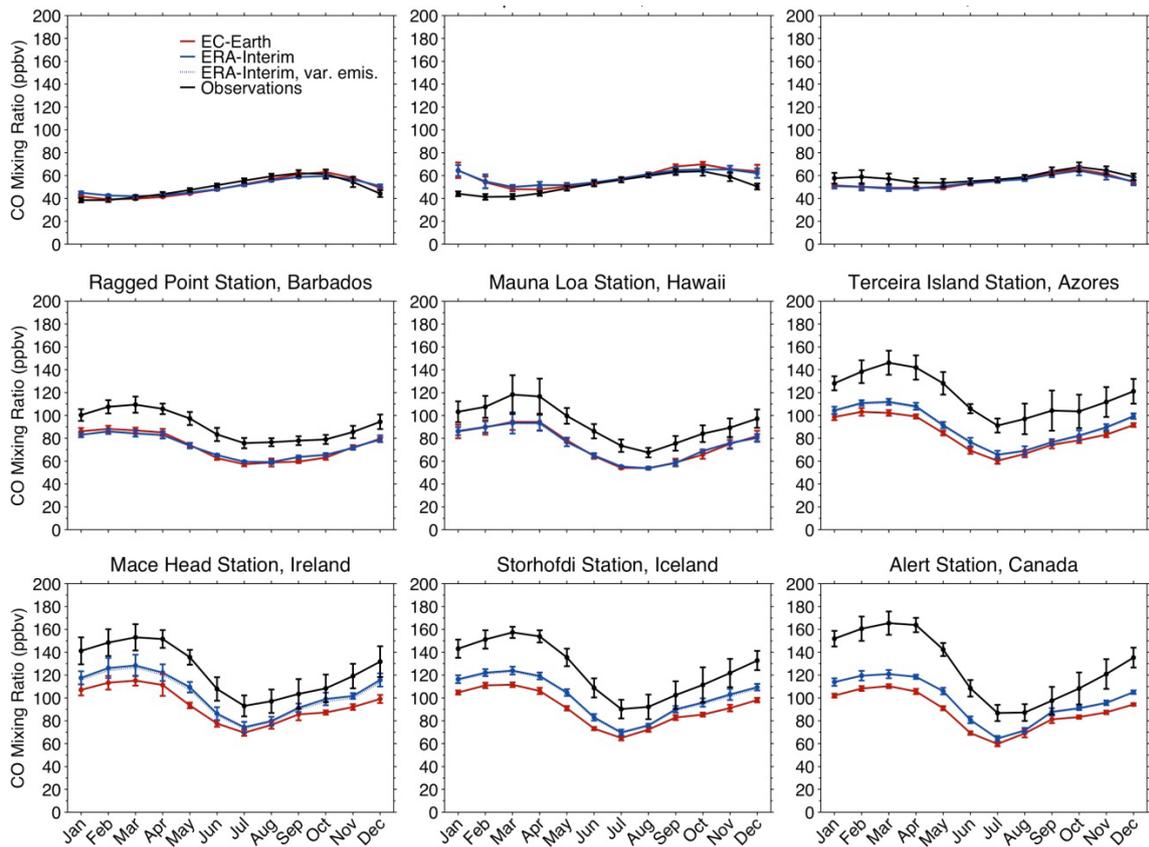


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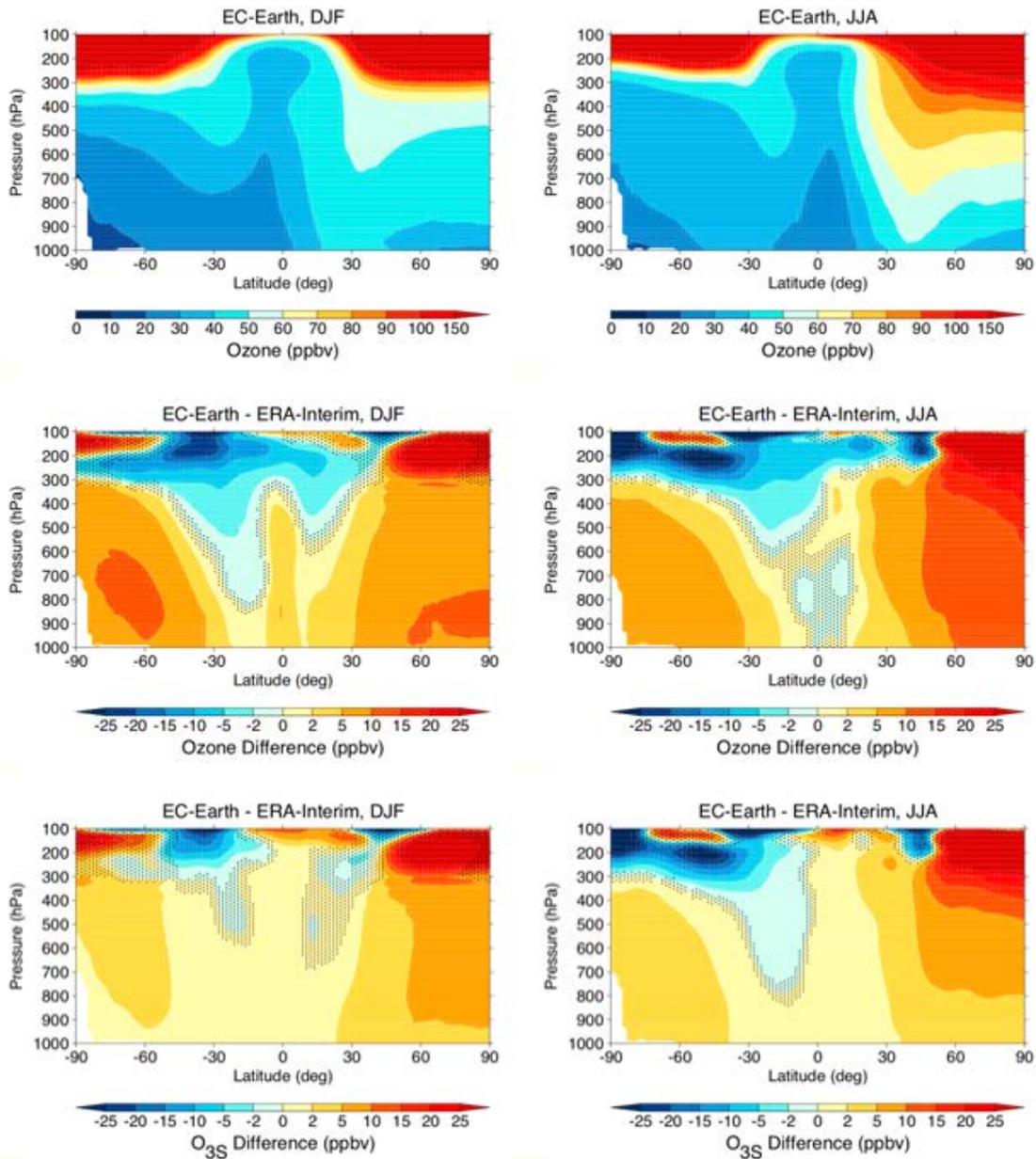
2 Figure 4. Zonal mean OH concentrations in the reference EC-Earth simulation (top), the
 3 climatology of Spivakovsky et al. (2000) reduced by 8% (middle), and the differences
 4 between the reference EC-Earth simulation and the corresponding ERA-Interim simulation
 5 (bottom) for boreal winter (left) and boreal summer (right).



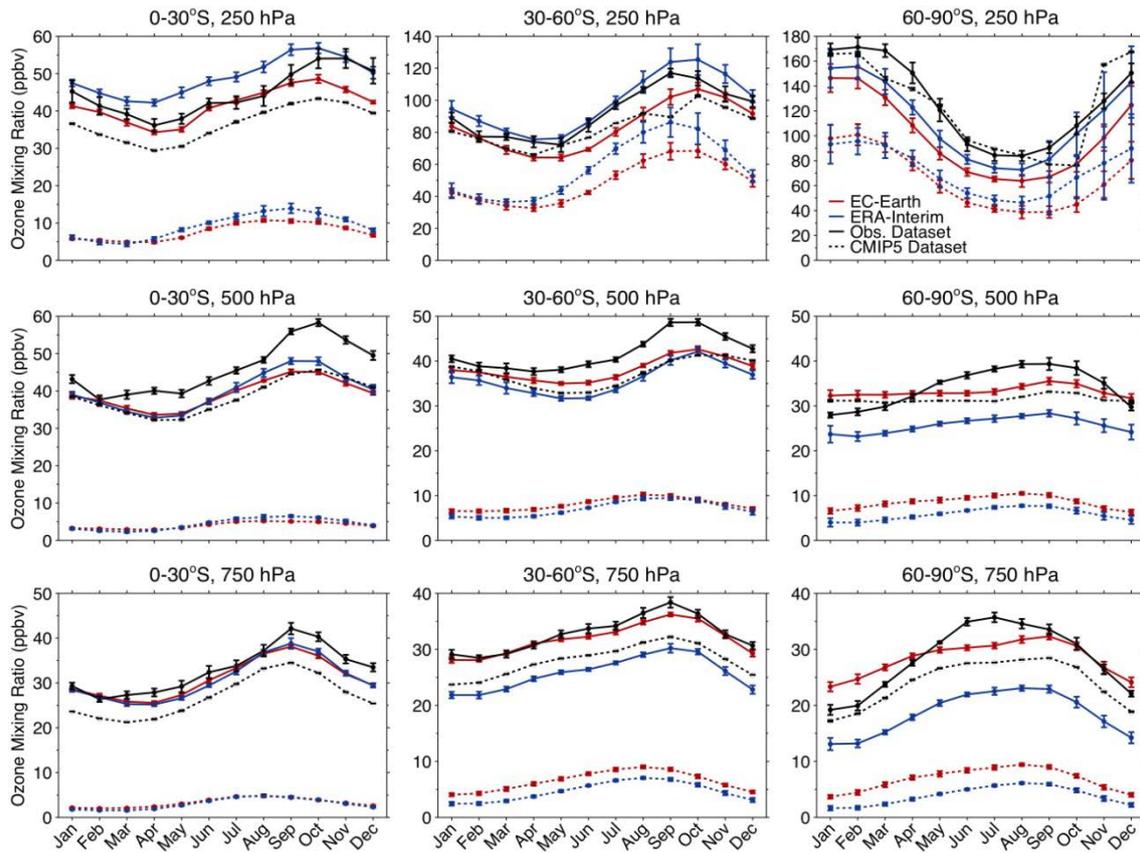
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 2 Figure 5. Zonal mean CO mixing ratios in the reference EC-Earth simulation (top) and the
 3 differences compared to the corresponding ERA-Interim simulation (bottom) for boreal
 4 winter (left) and boreal summer (right).



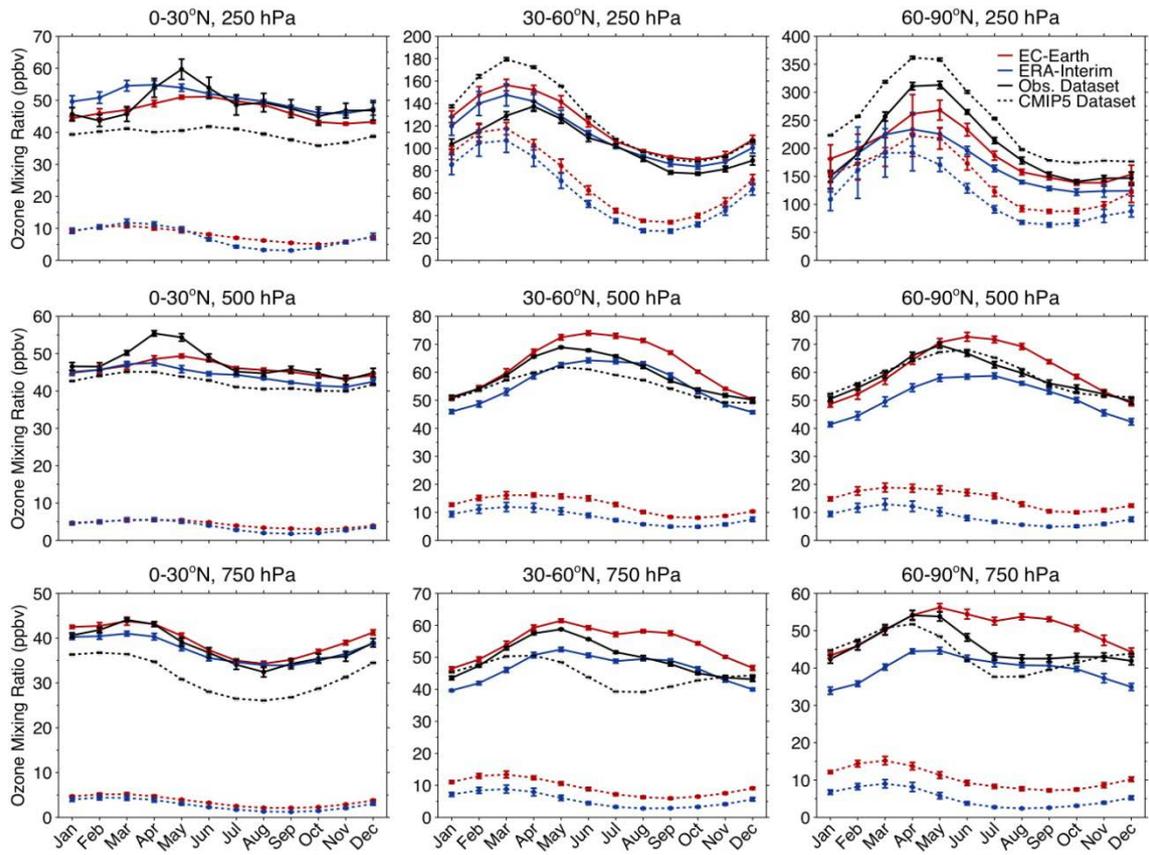
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 2 Figure 6. Comparison of monthly mean surface CO mixing ratios from the reference EC-
 3 Earth simulation (solid red lines) and the two ERA-Interim simulations (solid and dotted blue
 4 lines) against flask measurements at a number of stations selected from the NOAA GMD
 5 network. The results from the ERA-Interim simulation with interannual variations in the
 6 emissions from anthropogenic activities and biomass burning (dotted blue lines) nearly
 7 coincide with those from the ERA-Interim simulation where these emissions are fixed to their
 8 values for the year 2000 (solid blue lines).



1
 2 Figure 7. Zonal mean O₃ mixing ratios in the reference EC-Earth simulation (top), the
 3 differences compared to the corresponding ERA-Interim simulation (middle), and the
 4 contribution from the stratospheric O₃ tracer (O_{3S}) to these differences (bottom) for boreal
 5 winter (left) and boreal summer (right).

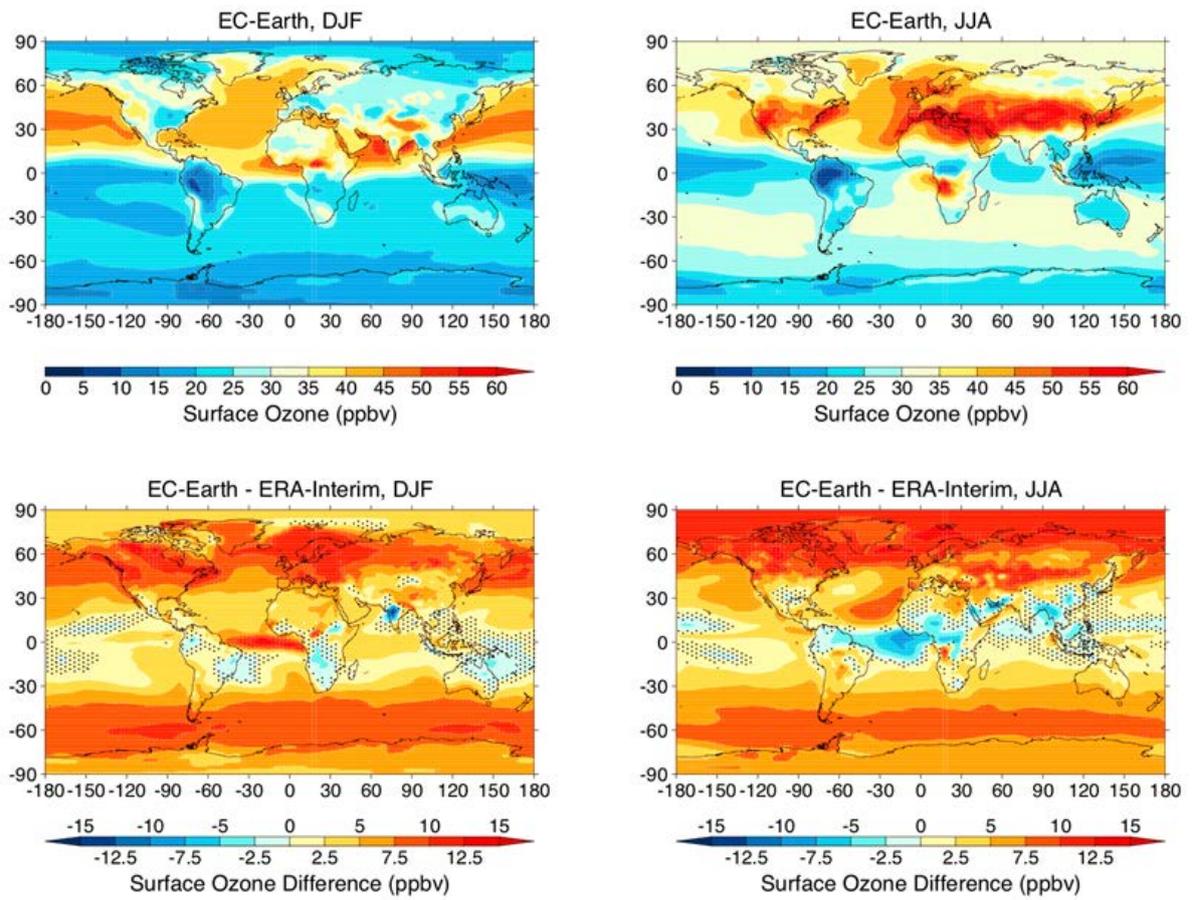


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 2 Figure 8. Monthly mean O₃ mixing ratios at 750, 500 and 250 hPa averaged over different
 3 latitude bands in the SH. The results from the reference EC-Earth simulation (solid red lines)
 4 and the corresponding ERA-Interim simulation (solid blue lines) are compared against the
 5 observational dataset (solid black lines) constructed from the Binary Database of Profiles
 6 (BDBP). The CMIP5 dataset from Cionni et al. (2011) is indicated by the dashed black lines.
 7 The contributions from O_{3S} for the EC-Earth and ERA-Interim simulations are shown by the
 8 dashed red and blue lines, respectively.

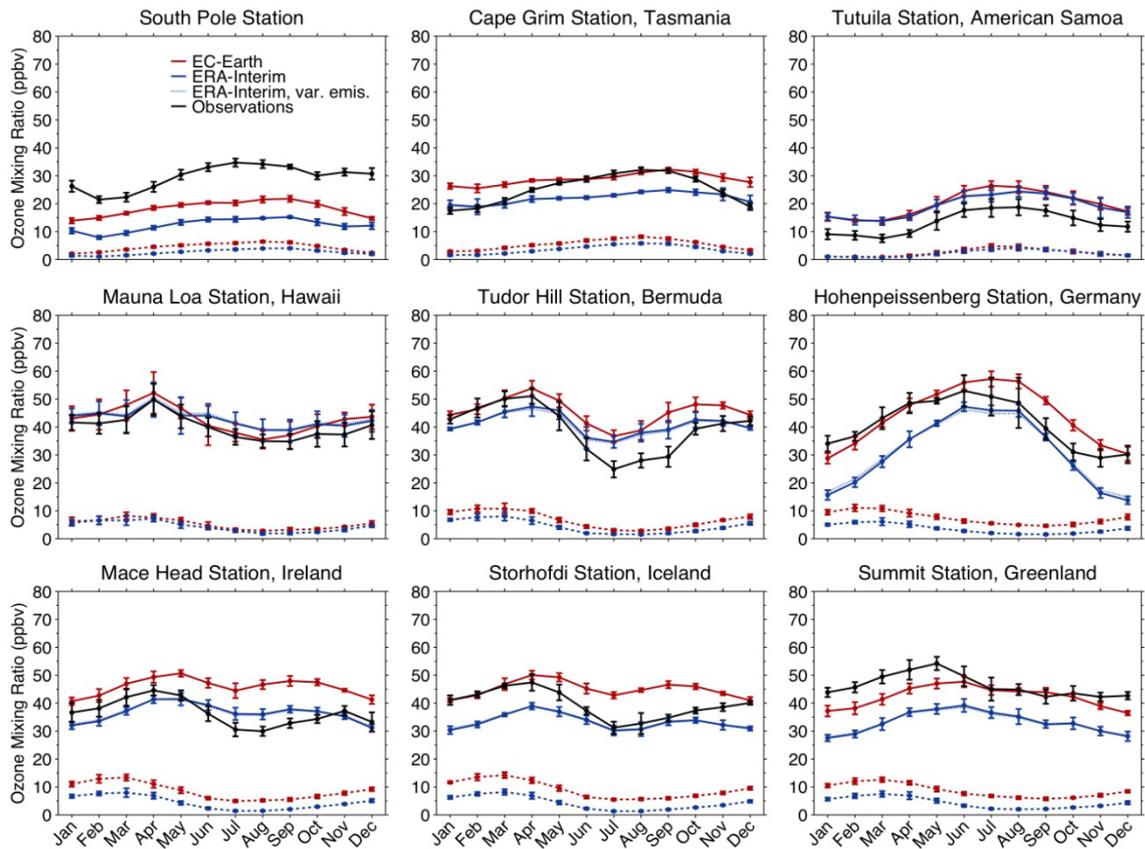


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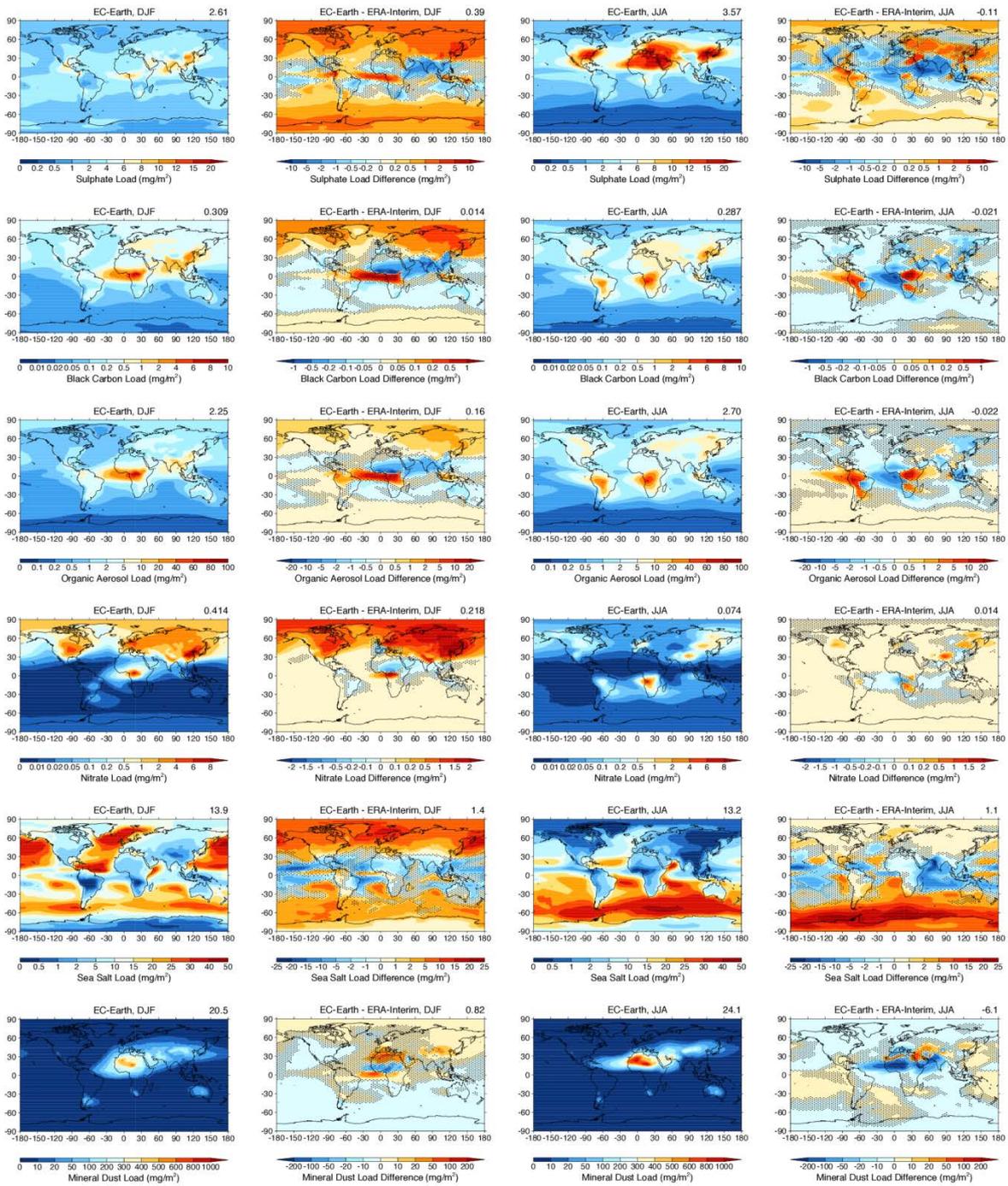
2 Figure 9. Same as Fig. 8, but for the NH.



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 2 Figure 10. Surface O₃ mixing ratios in the reference EC-Earth simulation (top) and the
 3 differences compared to the corresponding ERA-Interim simulation (bottom) for boreal
 4 winter (left) and boreal summer (right).

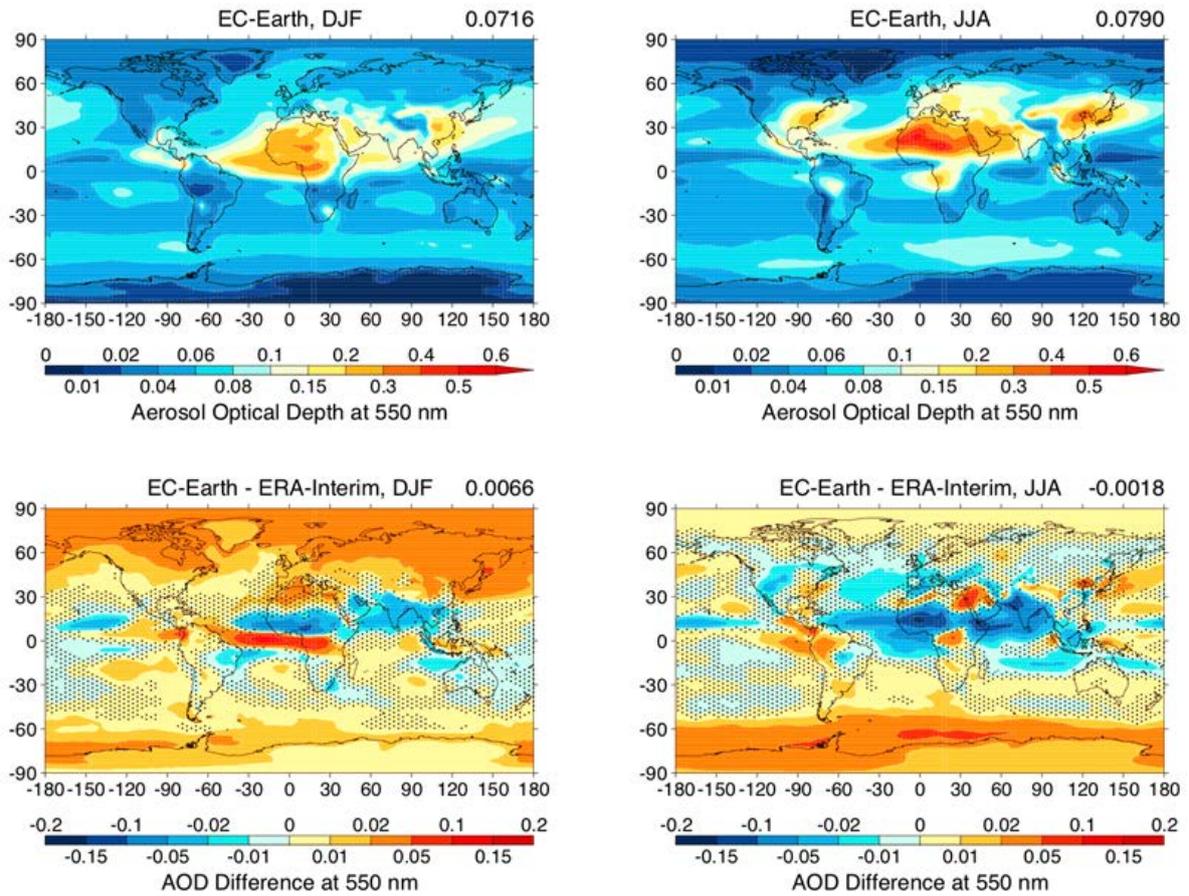


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2 Figure 11. Comparison of monthly mean surface O₃ mixing ratios from the reference EC-
3 Earth simulation (solid red lines) and the two ERA-Interim simulations (solid and dotted blue
4 lines) against in-situ measurements at a number of stations included in the NOAA GMD,
5 WDCGG and/or EMEP databases (solid black lines). The results from the ERA-Interim
6 simulation with interannual variations in the emissions from anthropogenic activities and
7 biomass burning (dotted blue lines) nearly coincide with those from the ERA-Interim
8 simulation where these emissions are fixed to their values for the year 2000 (solid blue lines).
9 The contributions from O_{3S} are shown by the dashed lines at the bottom of each panel (red for
10 EC-Earth, blue for ERA-Interim).



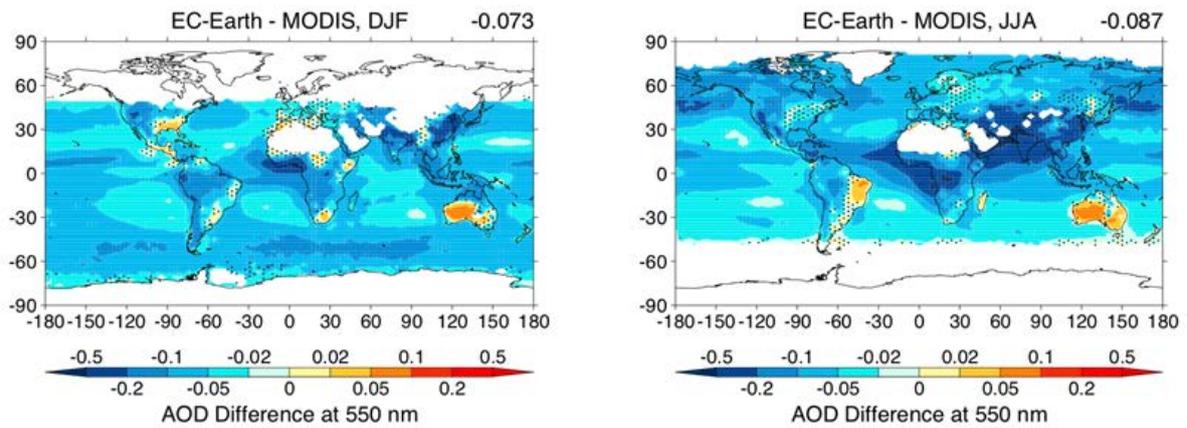
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2 Figure 12. Total loads of the different aerosol components in the reference EC-Earth
 3 simulation and the differences compared to the corresponding ERA-Interim simulation for
 4 boreal winter (left two columns) and boreal summer (right two columns). The global mean
 5 values of the displayed fields are indicated at the top of each panel.

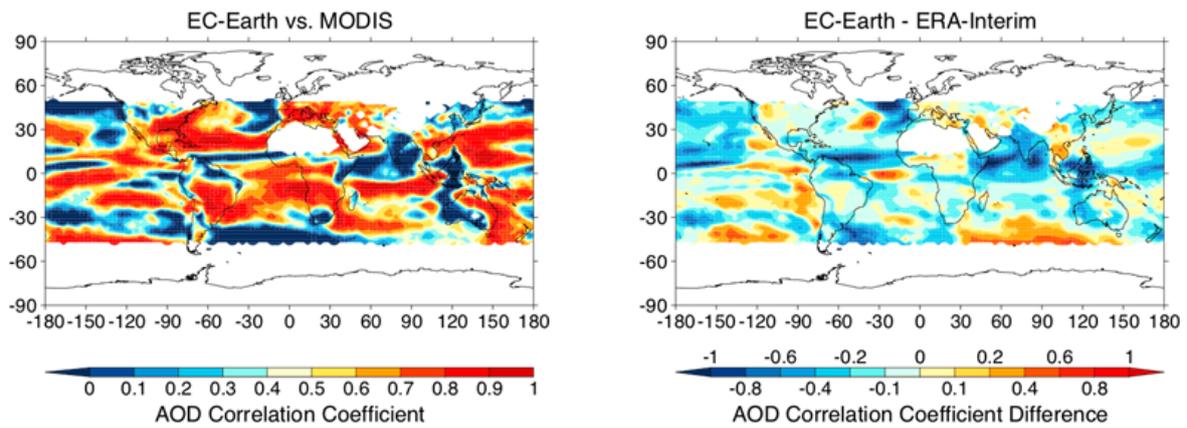


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2 Figure 13. Aerosol optical depth at 550 nm in the reference EC-Earth simulation (top) and the
 3 differences compared to the corresponding ERA-Interim simulation (bottom) for boreal
 4 winter (left) and boreal summer (right). The global mean values of the displayed fields are
 5 indicated at the top of each panel.

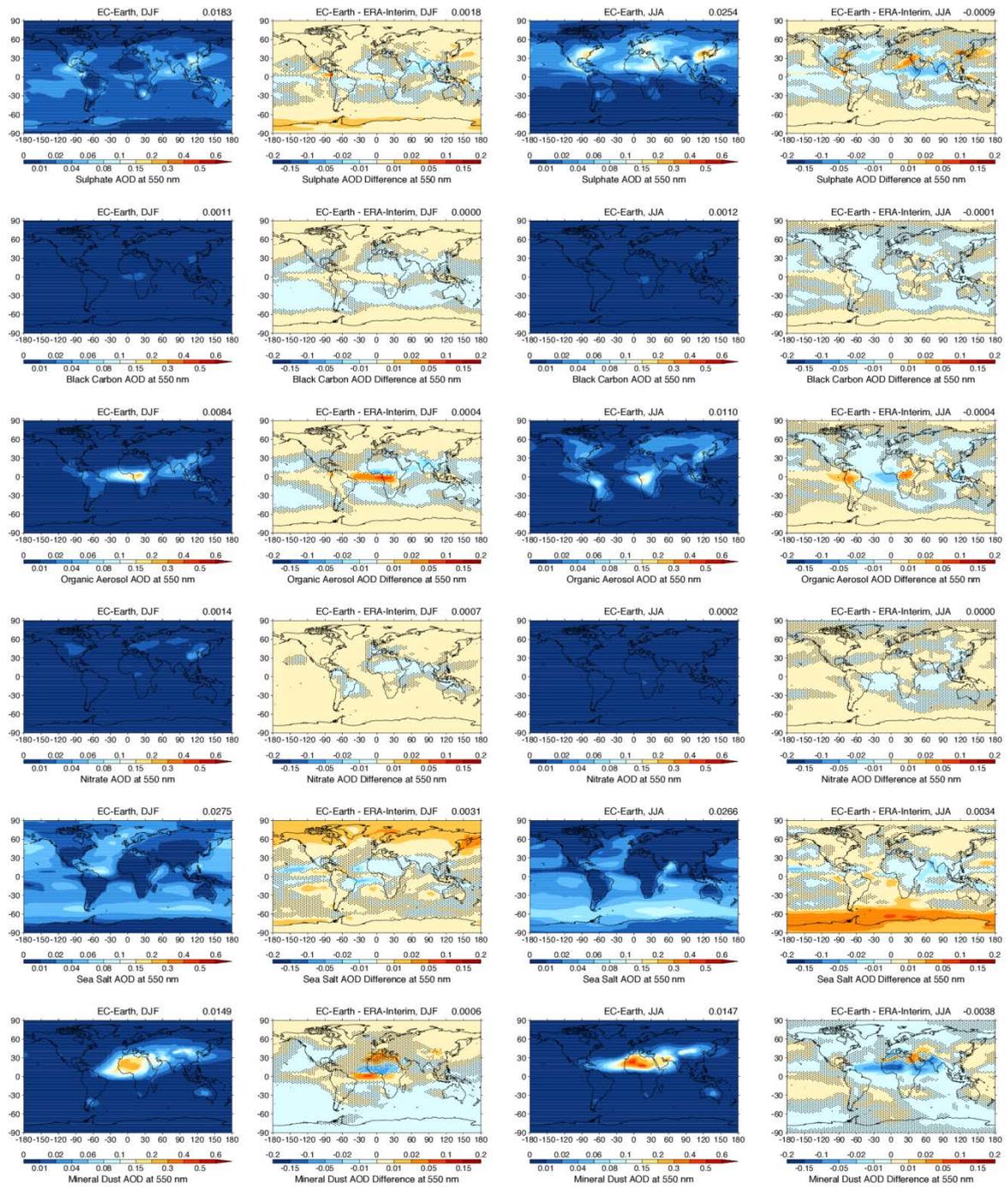


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 2 Figure 14. Difference in aerosol optical depth at 550 nm from the reference EC-Earth
 3 simulation compared to the MODIS Level-3 product coarsened to the same resolution and
 4 averaged over the years 2000-2009 for boreal winter (left) and boreal summer (right). Grid
 5 areas where, for all consecutive years, the MODIS data are missing for one or more months
 6 during the presented seasons are not included. The mean values of the displayed fields are
 7 indicated at the top of each panel.



1

2 Figure 15. Linear correlation coefficient between the decadal monthly mean aerosol optical
 3 depths at 550 nm from the reference EC-Earth simulation and the MODIS Level-3 product
 4 coarsened to the same resolution (left), and the difference compared to the corresponding
 5 correlation coefficient for the ERA-Interim simulation (right). Grid areas where, for all
 6 consecutive years, the MODIS data are missing for one or more months are not included.



1

2 Figure A1. Contributions from the different aerosol components to the aerosol optical depth at
 3 550 nm in the reference EC-Earth simulation and the differences compared to the
 4 corresponding ERA-Interim simulation for boreal winter (left two columns) and boreal
 5 summer (right two columns). The global mean values of the displayed fields are indicated at
 6 the top of each panel.