

1 **Supplementary material**

2 **S.1) Filter measurements uncertainty**

3 As mentioned in Sect. 2, undenuded filter measurements, like those performed during the campaign,
4 are subject to artefacts, related to volatilisation of semi-volatile material (SVM, mostly organic
5 compounds and ammonium nitrate) (Pang et al., 2002) and/or adsorption of gaseous species.
6 Additionally, artefacts strongly depend on the type of filter, and notably on its ability to retain
7 gaseous compounds (among which HNO₃ and NH₃) (Yu-Hsiang et al., 1997; and references therein).
8 In southern California, Hering and Cass (1999) have estimated that 28% on average of the particulate
9 nitrate is lost when using Teflon filters as compared to denuded nylon ones, with a large monthly
10 variability. During a filter intercomparison (including Teflon filters) over 10 months at 19 sites with
11 mean temperatures ranging from -20 to 30°C (i.e. roughly comparable to temperatures in Paris over
12 the year), Keck et al. (2005) have shown that negative artefacts on ammonium nitrate (and chloride)
13 are negligible below 0°C for Teflon filters. Above that threshold temperature, losses increase with
14 temperature, to approximately 20-30% at 10°C, around 40% at 20°C, until complete loss above 25°C.
15 Conversely, the low ability to retain gases on Teflon filters (contrary to cellulose acetate-nitrate
16 filters for instance) may lead to lower positive artefacts in ambient conditions with significant
17 gaseous HNO₃ or NH₃ concentrations.

18 For both nitrate and OA, intercomparison exercises have shown reasonable agreement with a
19 particle-into-liquid-sampler (PILS) coupled with ion chromatography (IC) and an OCEC Sunset field
20 instrument respectively (Bressi et al., 2012), but during a wintertime period with potentially low
21 evaporation. These artefacts are likely to increase in spring and summer with changing ambient
22 conditions. Hering and Cass (1999) have shown that the seasonal variation of undenuded Teflon filter
23 losses approximately ranges from 10% in winter to 80% in summer.

24 Thanks to the available TEOM-FDMS measurements at the PAR station, it is possible to investigate
25 the semi-volatile contribution to PM_{2.5}. This instrument operates at 30°C, and provides a reference
26 PM_{2.5} measurement, including both non-volatile and semi-volatile parts, as well as the split between
27 both. Gravimetric and TEOM-FDMS measurements are compared in Fig. S1. The correlation
28 between both measurements is strong (R=0.97). The semi-volatile part (at 30°C) measured by the
29 TEOM-FDMS, ranging from 1 to 14 µg.m⁻³, is stronger in winter and spring, due to higher
30 temperature discrepancies between the instrument and ambient conditions as well as higher
31 particulate concentrations. This semi-volatile matter is likely essentially composed of ammonium
32 nitrate and organic matter (Huffmann et al., 2009). Actually, its correlation with nitrate appears

1 significantly stronger than with OM (0.88 against 0.53), which may indicate a larger contribution of
2 this former compound.

3 Most of the time, gravimetric measurements underestimate the PM_{2.5} concentration, with maximum
4 relative biases reaching around -50%. Strongest negative relative discrepancies occur in early autumn
5 and spring. Additionally, positive biases (probably by adsorption) are observed in August. If they
6 reach +50%, these latter biases are not so problematic as they occur when concentrations are very
7 low, and should thus not influence considerably the budget results. The gravimetric measurement
8 usually gives higher concentrations than the reference non-volatile one, probably because it operates
9 at lower temperature (ambient room temperature compared to 30°C) and is thus expected to
10 evaporate less material.

11 Let us assume that the difference between the corrected TEOM-FDMS PM_{2.5} concentration and the
12 gravimetric measurement is essentially composed of ammonium nitrate and OM. It corresponds to
13 the net artefact. One can compute its relative contribution to the total of ammonium nitrate plus OM
14 corrected by this loss:

$$15 \quad \Delta = \frac{PM_{ref} - PM_{grav}}{OM + NH_4NO_3 + PM_{ref} - PM_{grav}} \quad (15)$$

16 Figure S2 shows the evolution of this relative error over the year. Maximum delta values reach 60%
17 in autumn and summer, while they range between 10 and 40% during winter and spring. Highest
18 (negative) values occur in August when concentrations are low, and are thus not likely to influence
19 considerably budget results. From this figure, one can roughly estimate an OM plus ammonium
20 nitrate loss around -30% in winter and -50% in summer (except August).

21

22 **S.2) PM emissions speciation**

23 The Table S1 gives the PM_{2.5} speciation used in the ESMERALDA simulation platform, for the large
24 and the two other refined domains. Speciation in the Paris region is deduced from the local detailed
25 emission inventory (at a SNAP sector level of 3) and speciation factors from the literature.

26

27 **S.3) Evidence of woodburning local impact at the RNE station**

28 Daily levoglucosan measurements are available at PAR, RNE and RUS stations. Their concentrations
29 are represented in Fig. S3, as well as temperature and OM rural levels. The figure shows strong OM
30 discrepancies between the RNE station on one side, and the two other rural ones on the other side (up

1 to a factor of five on specific days). They are not systematic, but appear occasionally from October to
2 May. Such large discrepancies are also clearly observed for levoglucosan concentrations, which
3 indicate that these differences are probably related to local woodburning pollution. The contribution
4 of road transport is likely to be minor, since no similar trend is observed for EC concentrations
5 during these events. Even during north-easterly wind regimes, these large OM (and levoglucosan)
6 amounts at the RNE station are not advected toward Paris (PAR concentrations below RNE ones),
7 which suggests a local origin. These OM data thus need to be invalidated, as well as the PM one.

8 However, considering the importance of NE sector, both in terms of frequency (31%, see Sect. 5.2)
9 and PM imports, a criteria has been defined to discriminate such local pollution events from real
10 imports at the RNE stations, in order to retain as many data as possible. Despite a large distance
11 between stations (around 100 km), rural OM concentrations appear quite homogeneous. The median
12 values of RNE/RUS, RNE/RNW and RUS/RNW ratios are 1.3, 1.3 and 1.0, respectively. We thus set
13 at 30% the maximum relative discrepancy on OM concentration between RNE station and the two
14 other rural ones, above which RNE data (OM and PM_{2.5}) is invalidated and replaced by mean data
15 from RUS and/or RNW stations, depending on their availability. Results are illustrated in Fig. S4,
16 where the correction of red points leads to green ones.

17 The choice of such a low threshold allows minimizing the risk of taking into account non-
18 representative OM imports in the Paris budget. It implies a large recurrence of RNE data invalidation,
19 210 days out of 361. But actually, the procedure is here applied to all data and not only data
20 associated to north-east origin (i.e. used to compute the advected contribution). The number of
21 corrected points taken into account is thus lower. Sensitivity tests on this criterion show a satisfactory
22 robustness of average (monthly, annual) results (monthly contributions do not change more than $\pm 3\%$
23 with criteria values in the 20-40% range).

24

25 **S.4) Primary organic aerosol volatility**

26 As primary organic aerosol (POA) is considered as non-volatile in the CHIMERE model, it is
27 interesting to evaluate how strong the effect of its unaccounted volatility can be.

28 Let us consider an emission of POA. Given the semi-volatile nature of organic matter, this POA can
29 be considered as semi-volatile OM (SVOM). Following Robinson et al. (2007), we also need to take
30 into account intermediate-volatility OM (IVOM), estimated at $1.5 \cdot \text{SVOM}$. For convenience, this
31 total amount is denominated as SVOM (thus equal to $2.5 \cdot \text{POA}$). As this SVOM includes a large
32 number of species of various volatility, one can group them into 9 bins of volatility, $\{C_i^*\}_{i=(1;9)}$ with

1 respect to their reference saturation concentration at a given temperature of 298 K (Donahue et al.,
2 2006):

$$3 \quad \{C_i^*\} = \{0.01, 0.1, 1, 10, 100, 1000, 10000, 100000\} \mu\text{g}\cdot\text{m}^{-3} \quad (16)$$

4 The lumped species aggregating all the SVOM emitted in the volatility bin i is defined as SVOM_i ,
5 and its concentration as C_i . $\{C_i\}_{i=(1,9)}$ is assumed to follow the distribution illustrated by the bars
6 height in Fig. S4, based on Robinson et al. (2007). At a given temperature, the partitioning theory
7 describes the fraction f_i of the SVOM_i species that would be present in the particulate phase through
8 the following equation (Donahue et al., 2006):

$$9 \quad f_i = \frac{1}{1 + \frac{C_i}{C_{OA}^*}} \text{ with } C_{OA} = \sum_{i=1}^9 f_i C_i \text{ the condensed-phase OA mass concentration.} \quad (17)$$

10 From this, it is now possible to compute the total fraction of emissions that stays in the particulate
11 phase, depending on the temperature and the C_{OA} concentration. A change on C_{OA} and/or temperature
12 (modifying C_i^* through Clausius-Clapeyron law) modifies the partitioning between both phases.
13 Given an OA concentration and a temperature, it is thus possible to derive the amount of initial POA
14 emissions that is likely to volatilize just after emission. The example given in Fig. S5 shows that, at a
15 temperature of 298 K and an OA concentration of $10 \mu\text{g}\cdot\text{m}^{-3}$, only 26% of the traditional POA
16 emissions stay in the particulate phase.

17 A similar exercise is performed for each day during the whole PARTICULES period, on the basis of
18 the OA concentrations observed at the PAR site and the temperature measured in the city
19 (MONTSOURIS station). Results are shown in Fig. S6. The part of volatilized POA emissions turns
20 out to be very large, with values ranging from 40 to 80%. The consideration of POA volatility
21 appears to be crucial in summer where most of emissions are volatilized (up to 4-5 time lower).
22 While values are lower in winter, particularly during episodes, they remain significant.

23

24 **S.5) Meteorology errors and impact on thermodynamic equilibria**

25 As previously mentioned, thermodynamics of secondary inorganic salts (and water) are handled in
26 CHIMERE using a tabulated version of the ISORROPIA module (Bessagnet et al., 2004). The
27 amount of particulate ammonium nitrate depends on the total amounts in both gaseous and condensed
28 phases of each compound of the system (sulfate plus sulfuric acid, nitrate plus nitric acid, ammonium
29 plus ammonia) and on meteorological conditions (temperature, RH). We investigate here the

1 influence of meteorological errors on the thermodynamic equilibria, with a focus on nitrates. As
2 described in Sect. 5.1, over the whole year, simulated temperature and RH in Paris show negative (-
3 1°C) and positive biases (+3.1% RH), respectively. During specific episodes (cold or heat waves),
4 errors can be much larger. This is likely to shift the equilibria towards the particulate phase, and we
5 intend here to quantify the possible overestimation of particulate nitrate, based on the ISORROPIA
6 table.

7 Let us consider over the Paris hourly temperature, RH and concentrations of each compound
8 simulated by the model. By assuming for simplification that equilibrium is instantaneously achieved,
9 one can compute with the ISORROPIA table the nitrate partition between gaseous and condensed
10 phases. The same can be done with the temperature and the RH observed at MONTsouris station
11 in Paris. Results are shown in Fig. S7.

12 As expected, the correction of temperature and RH reduces the amount of nitrate in the condensed
13 phase. On average, the nitrate overestimation induced by these meteorological errors remains quite
14 low, with absolute and relative bias around $0.3 \mu\text{g}\cdot\text{m}^{-3}$ and 10%, respectively. However, during
15 summer, when most of nitrate is volatilized due to higher temperatures, biases can reach a factor of
16 two. More precisely, discrepancies significantly depend on the distance between the simulated RH
17 and the deliquescence RH (DRH, around 62% for ammonium nitrate). Therefore, meteorological
18 errors in Paris are not likely to affect so much the overall nitrate budget simulated by CHIMERE
19 since they are far lower than measurement uncertainties on average.

20 However, they can be more critical during specific meteorological events, during which errors on
21 temperature and RH are likely to increase. Such an event occurs in the second half of June, where the
22 heat wave in most of Europe is simulated with a cold bias (between -1 to more than -3°C), as shown
23 over the continental domain on Fig. S8 (meteorological observations at all worldwide synoptic
24 stations available from the British Atmospheric Data Center (BADC) (UK meteorological office)).

25 The previous analysis is applied to meteorological errors in Paris. Similar errors at the regional scale
26 are not likely to influence directly the amount of particulate nitrate advected towards the city, since
27 equilibrium is expected to be fast enough (compared to transportation time), depending mainly on the
28 local meteorological conditions. However, an overestimation of the condensed phase far from Paris
29 decreases gaseous nitric acid (HNO_3) concentrations and dry deposition, faster than particulate
30 nitrate, along the transport and thus increases the available reservoir (nitric acid plus nitrate), which
31 may finally be a factor of nitrate overestimation (in the case of negative temperature biases).

32

1 Table S1: PM_{2.5} speciation (in %) for different SNAP sectors.

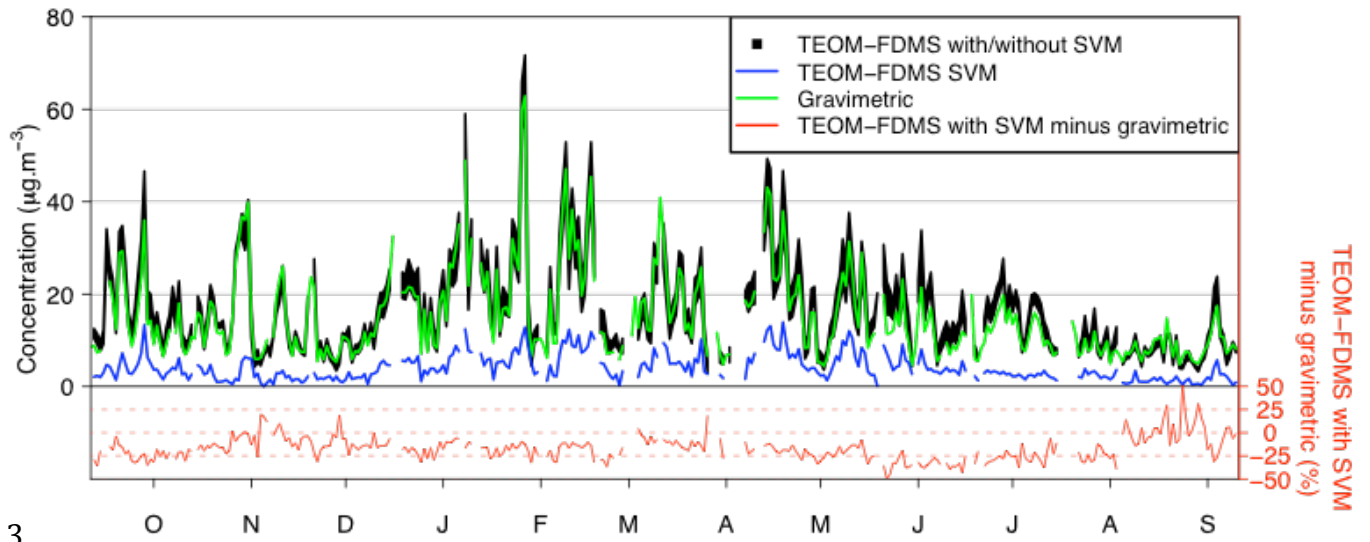
SNAP sector	Continental			Paris region		
	EC	OC	MPM*	EC	OC	MPM*
1 Public electricity and other energy transformation	10	10	80	45.1	38.3	16.6
2 Small combustion plants	15.8	54.7	29.5	15	54.2	30.8
3 Industrial combustion and processes with contact	25	30	45	32.9	32.4	34.6
4 Industrial process emission	0	10	90	0	14.1	85.9
5 Fossil fuel production	24.3	40	35.7	**	**	**
6 Solvent and product use	24	40	36	0	0	100
7 Road transport	54.3	45.7	0	61.3	37	1.7
8 Other (non-road) transport and mobile machinery	54.7	20	25.3	59.4	32.8	7.8
9 Waste disposal	24.7	40	35.3	1.7	7.7	90.6
10 Agriculture	24	40	36	4.8	3.3	91.9

2 * MPM: Mineral particulate matter.

3 ** No emissions in Paris region for this SNAP 5.

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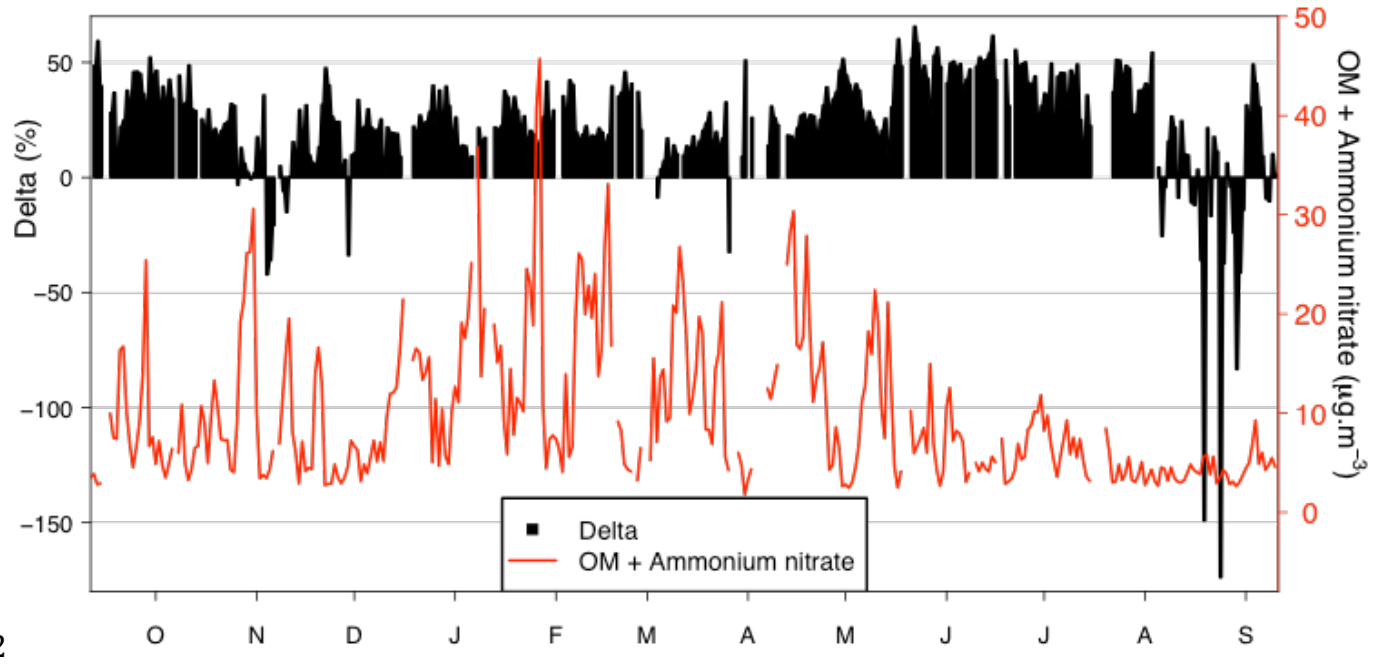


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4 Figure S1: PM_{2.5} concentrations (µg.m⁻³) at the PAR site. PM_{2.5} semi-volatile material (SVM)
5 concentration is represented within the PM_{2.5} total concentration in black (i.e. bottom and top of
6 black surface correspond to TEOM-FDMS without and with SVM, respectively), and alone in blue.
7 The gravimetric PM_{2.5} at 20% RH is plotted in green. The relative discrepancy between reference
8 (TEOM-FDMS with SVM) and gravimetric PM_{2.5} (%) is plotted in red.

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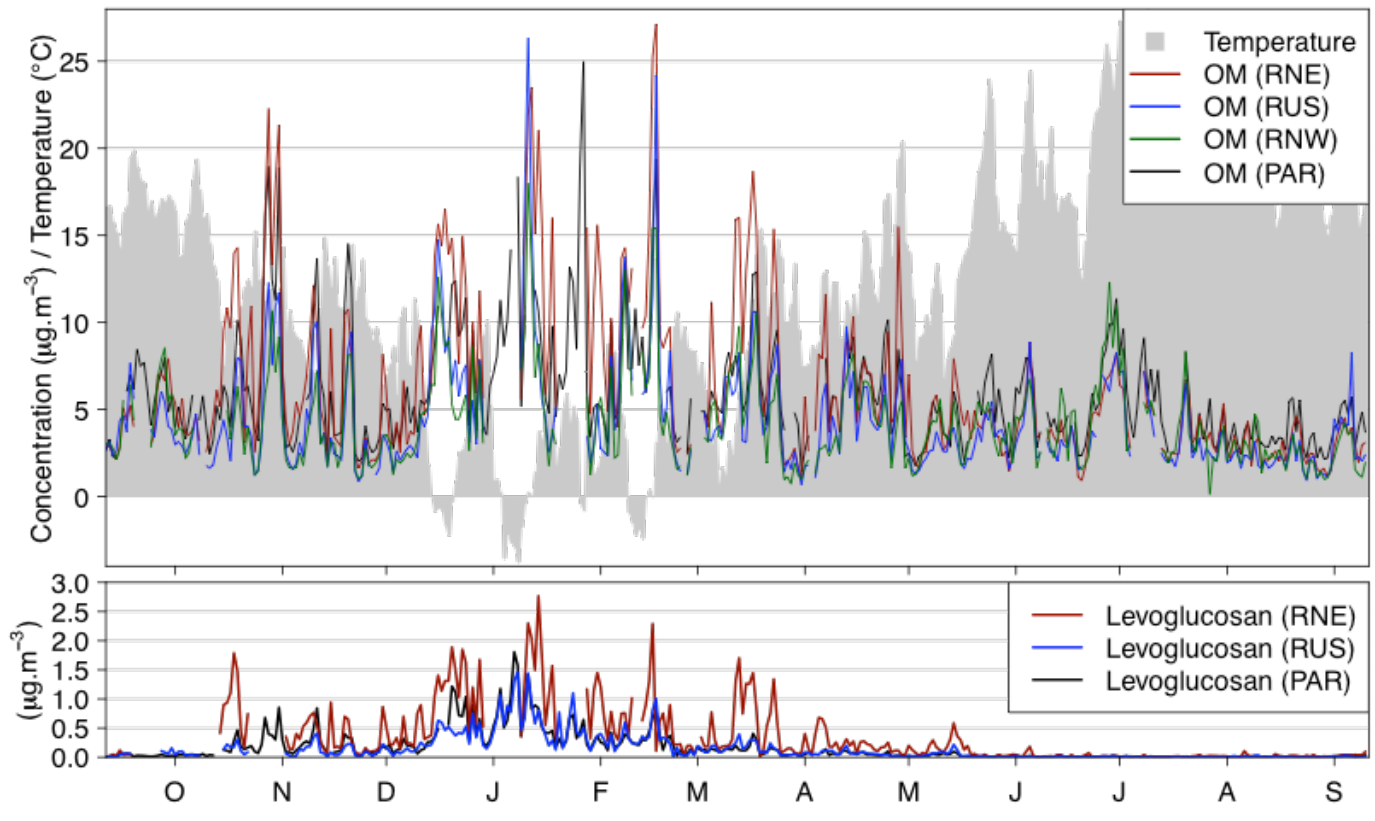


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3 Figure S2: Delta time series (in black) (see text for explanations) and OM plus ammonium nitrate
4 concentration (in red).

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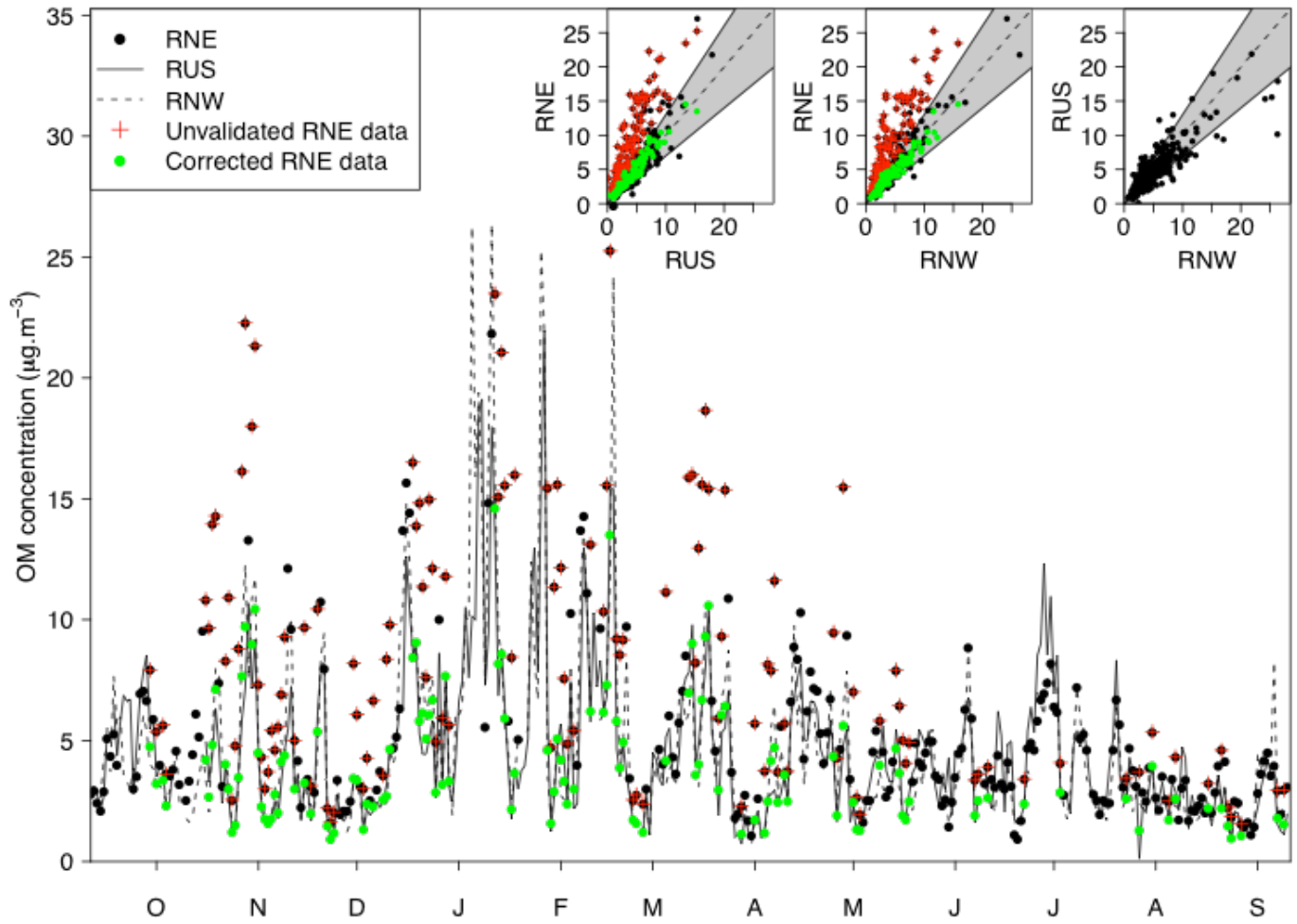


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3 Figure S3: Temperature at MONTSOURIS station and OM concentration at rural stations (top panel),
4 and levoglucosan concentrations (bottom panel).

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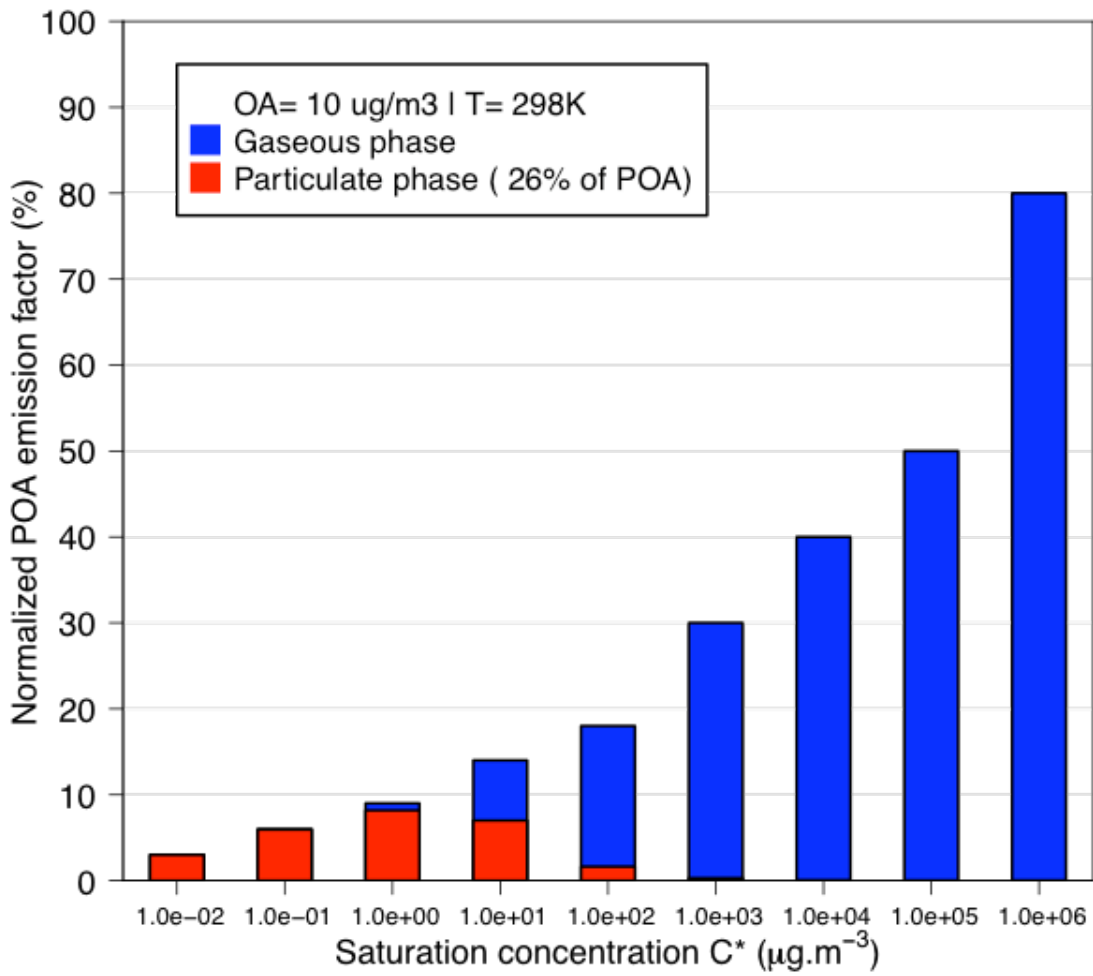


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3 Figure S4 : OM concentration at all rural background stations (RNE, RUS, RNW) before correction
4 (in black), unvalidated and corrected RNE points (in red and green, respectively). At the top of the
5 figure, associated OM concentrations scatter plots of rural stations pairs ($y=x$ in dotted line, and
6 $y=x\pm 30\%x$ in grey).

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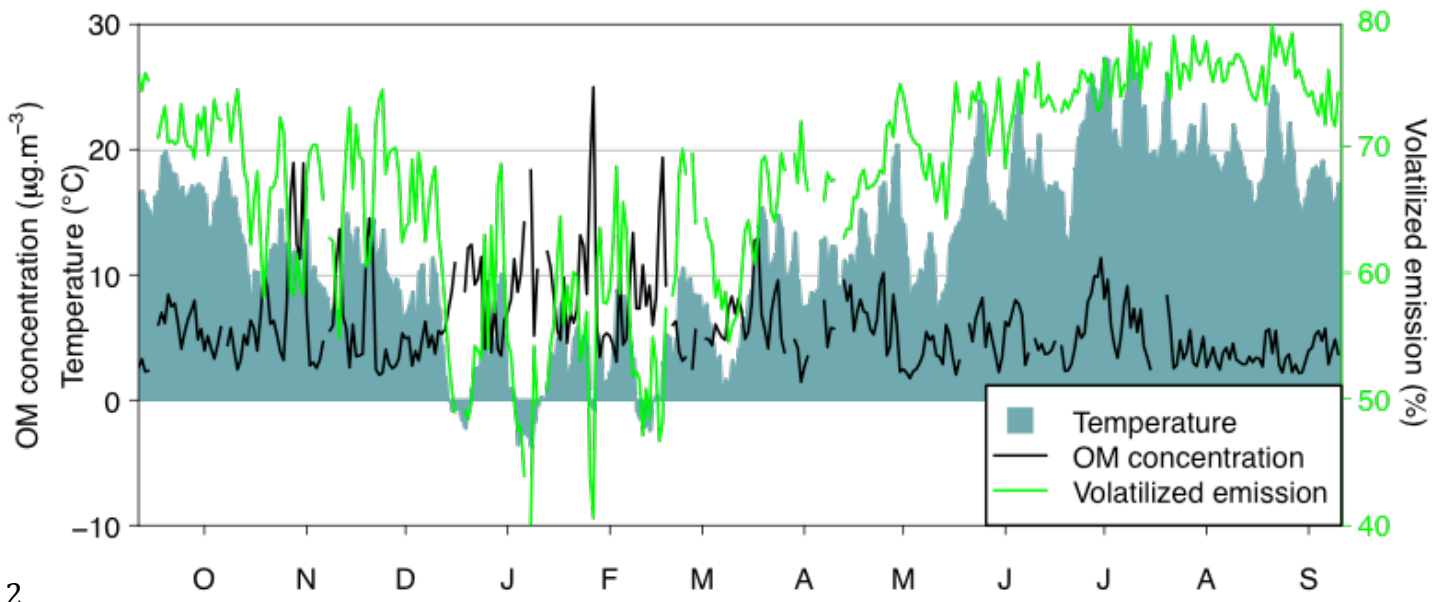


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3 Figure S5: Distribution of SVOM emissions per volatility class normalized by POA emissions (i.e.
4 total area equal to 250% of POA), with partitioning between particulate (in red, equal to 26% of
5 POA) and gaseous (in blue) phases at a temperature of 298K and an OA concentration of $10 \mu\text{g}\cdot\text{m}^{-3}$.

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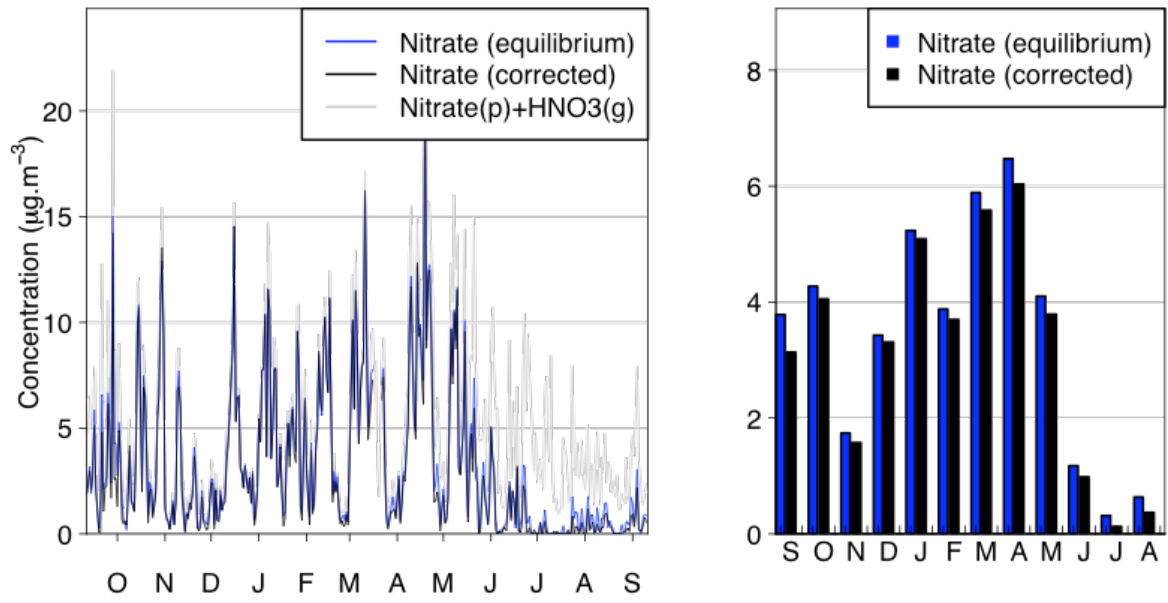
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3 Figure S6: Volatilized POA emission fraction (green). Temperature at the MONTSOURIS station (in
4 grey) and OM concentration at the PAR station (black) are also represented.

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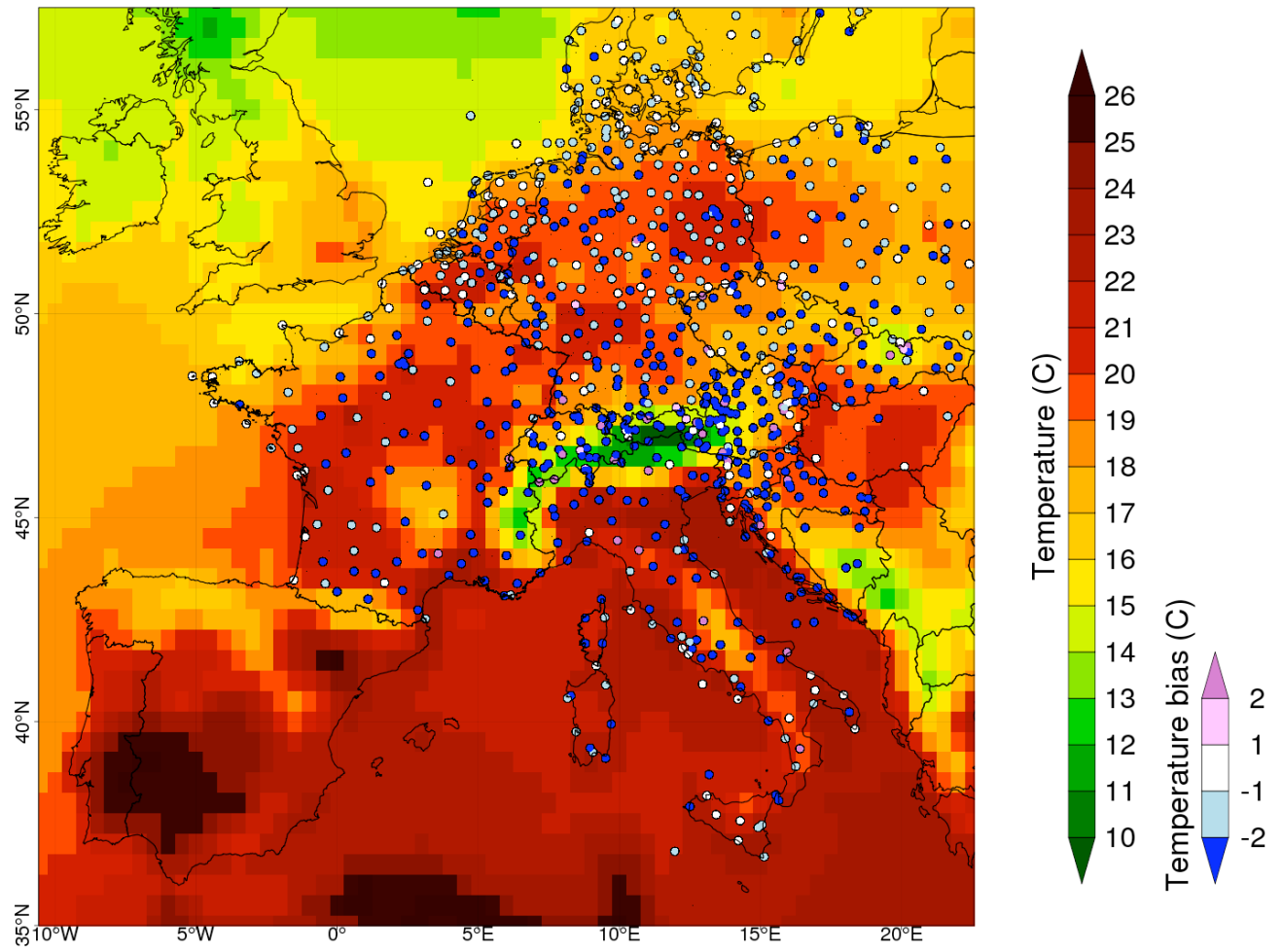
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3 Figure S7: Daily (left panel) and monthly (right panel) particulate nitrate concentrations in Paris,
4 taking into account the simulated (so-called *equilibrium*, blue line) and the observed (so-called
5 *corrected*, black line) meteorology. The reservoir (particulate nitrate plus gaseous nitric acid) is
6 represented in grey line.

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3 Figure S8: Simulated average temperature field over the period of 25 June – 5 July on the continental
4 LAR domain, and mean temperature biases at BADC synoptic stations in a selection of countries.

5