

Supplementary information

Table S1. Set of differential equations used in GEOS-Chem/Fe_d

$\frac{d\text{Fe(II)}}{dt}$	$= j_1\text{Fe(OH)}^{2+} + j_2[\text{Fe(C}_2\text{O}_4)_2^-] + j_3[\text{Fe(C}_2\text{O}_4)_3^{3-}] + k_7[\text{Fe(OH)}^{2+}][\text{O}_2^-]$ $+ k_8[\text{Fe(OH)}^{2+}][\text{HO}_2] + k_{17}[\text{Fe(C}_2\text{O}_4)_n^{3-2n}][\text{O}_2^-]$ $+ k_{18}[\text{Fe(C}_2\text{O}_4)_n^{3-2n}][\text{HO}_2] - k_1[\text{Fe}^{2+}][\text{H}_2\text{O}_2] - k_2[\text{FeC}_2\text{O}_4][\text{H}_2\text{O}_2]$ $- k_3[\text{Fe}^{2+}][\text{O}_2^-][\text{H}^+]^2 - k_4[\text{Fe}^{2+}][\text{HO}_2][\text{H}^+] - k_5[\text{Fe}^{2+}][\text{NO}_3]$ $- k_6[\text{Fe}^{2+}][\text{NO}_2][\text{H}^+] - k_9[\text{Fe}^{2+}][\text{O}_3]$
$\frac{d\text{Fe(III)}}{dt}$	$= -\frac{d\text{Fe(II)}}{dt}$
$\frac{d[\text{FeO}^{2+}]}{dt}$	$= k_9[\text{Fe}^{2+}][\text{O}_3] - k_{10}[\text{FeO}^{2+}][\text{H}_2\text{O}] - k_{11}[\text{FeO}^{2+}][\text{OH}^-][\text{H}^+] - k_{12}[\text{FeO}^{2+}][\text{H}_2\text{O}_2]$ $- k_{13}[\text{FeO}^{2+}][\text{HO}_2]$
$\frac{d[\text{C}_2\text{O}_4^{2-}]}{dt}$	$= j_2[\text{Fe(C}_2\text{O}_4)_2^-] + j_3[\text{Fe(C}_2\text{O}_4)_3^{3-}] - k_{19}[\text{C}_2\text{O}_4^{2-}][\text{OH}^-] - k_{20}[\text{C}_2\text{O}_4^{2-}][\text{NO}_3]$ $- k_{21}[\text{C}_2\text{O}_4^{2-}][\text{O}_2]$
$\frac{d[\text{H}_2\text{O}_2]}{dt}$	$= k_3[\text{Fe}^{2+}][\text{O}_2^-][\text{H}^+]^2 + k_4[\text{Fe}^{2+}][\text{HO}_2][\text{H}^+] + k_{11}[\text{FeO}^{2+}][\text{OH}^-][\text{H}^+] + k_{14}[\text{HO}_2][\text{HO}_2]$ $+ k_{15}[\text{HO}_2][\text{O}_2][\text{H}^+] - k_1[\text{Fe}^{2+}][\text{H}_2\text{O}_2] - k_2[\text{FeC}_2\text{O}_4][\text{H}_2\text{O}_2]$ $- k_{12}[\text{FeO}^{2+}][\text{H}_2\text{O}_2]$
$\frac{d[\text{HO}_2]}{dt}$	$= k_{12}[\text{FeO}^{2+}][\text{H}_2\text{O}_2] - k_4[\text{Fe}^{2+}][\text{HO}_2][\text{H}^+] - k_8[\text{Fe(OH)}^{2+}][\text{HO}_2] - k_{13}[\text{FeO}^{2+}][\text{HO}_2]$ $- k_{14}[\text{HO}_2][\text{HO}_2] - k_{15}[\text{HO}_2][\text{O}_2][\text{H}^+] - k_{18}[\text{Fe(C}_2\text{O}_4)_n^{3-2n}][\text{HO}_2]$
$\frac{d[\text{OH}^-]}{dt}$	$= j_1\text{Fe(OH)}^{2+} + k_1[\text{Fe}^{2+}][\text{H}_2\text{O}_2] + k_2[\text{FeC}_2\text{O}_4][\text{H}_2\text{O}_2] + k_{10}[\text{FeO}^{2+}][\text{H}_2\text{O}]$ $- k_{11}[\text{FeO}^{2+}][\text{OH}^-][\text{H}^+] - k_{19}[\text{C}_2\text{O}_4^{2-}][\text{OH}^-]$
$\frac{d[\text{O}_2^-]}{dt}$	$= k_{16}[\text{CO}_2^-][\text{O}_2] + k_{21}[\text{C}_2\text{O}_4^{2-}][\text{O}_2] - k_3[\text{Fe}^{2+}][\text{O}_2^-][\text{H}^+]^2 - k_7[\text{Fe(OH)}^{2+}][\text{O}_2^-]$ $- k_{17}[\text{Fe(C}_2\text{O}_4)_n^{3-2n}][\text{O}_2^-]$
$\frac{d[\text{C}_2\text{O}_4^-]}{dt}$	$= j_2[\text{Fe(C}_2\text{O}_4)_2^-] + j_3[\text{Fe(C}_2\text{O}_4)_3^{3-}] + k_{19}[\text{C}_2\text{O}_4^{2-}][\text{OH}^-] + k_{20}[\text{C}_2\text{O}_4^{2-}][\text{NO}_3]$ $+ k_{21}[\text{C}_2\text{O}_4^{2-}][\text{O}_2]$
