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An isopycnic ocean carbon cycle model

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Abstract

The carbon cycle is a major forcing component in the global climate system. Modelling studies aiming to explain recent and past climatic changes and to project future ones thus increasingly include the interaction between the physical and biogeochemical systems. Their ocean components are generally z -coordinate models that are conceptually easy to use but that employ a vertical coordinate that is alien to the real ocean structure. Here we present first results from a newly developed isopycnic carbon cycle model and demonstrate the viability of using an isopycnic physical component for this purpose. As expected, the model represents interior ocean transport of biogeochemical tracers well and produces realistic tracer distributions. Difficulties in employing a purely isopycnic coordinate lie mainly in the treatment of the surface boundary layer which is often represented by a bulk mixed layer. The most significant adjustments of the biogeochemical code for use with an isopycnic coordinate are in the representation of upper ocean biological production. We present a series of sensitivity studies exploring the effect of changes in biogeochemical and physical processes on export production and nutrient distribution. Apart from giving us pointers for further model development, they highlight the importance of preformed nutrient distributions in the Southern Ocean for global nutrient distributions. Use of a prognostic slab atmosphere allows us to assess the effect of the changes in export production on global ocean carbon uptake and atmospheric CO_2 levels. Sensitivity studies show that iron limitation for biological particle production, the treatment of light penetration for biological production, and the role of diapycnal mixing result in significant changes of modelled air-sea fluxes and nutrient distributions.

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1 Introduction

The human induced increase of atmospheric greenhouse gases, especially that of CO₂, has led to growing concern about the climatic and environmental consequences related to these perturbations. The ocean plays an important role in the regulation of atmospheric gases (e.g., Sillen, 1966). It is known as to represent the major ultimate sink for anthropogenic CO₂ (e.g., Bolin and Eriksson, 1957; Archer, 2005) due to its relatively quick turnover time scale of 1000–2000 years (Matsumoto et al., 2007) and its high buffer capacity for CO₂ additions (Buch et al., 1932; Revelle and Suess, 1957). For correct predictions of the development of the Earth's carbon budget during the coming decades and centuries under given greenhouse gas emission scenarios, the kinetics of marine CO₂ uptake from the atmosphere play a crucial role: how quickly can the ocean neutralise CO₂ additions to the atmosphere through input from fossil fuel burning and cement manufacturing (Boden et al., 2009) as well as land use (Houghton, 1999)? Further, in recent years, carbon cycle climate feedbacks have been identified to provide a major uncertainty in future projections of climate change (Denman et al., 2007). It is likely that these feedbacks will reinforce climate change, but feedback strength varies considerably among different model systems (Friedlingstein et al., 2006).

While quantifying the regulation of the Earth's radiative budget is a key question, we require coupled physical-biogeochemical ocean models to address a number of additional important issues. The rising carbon content of the ocean causes a drop in pH (ocean acidification) with potentially severe impacts for marine biota and related geochemical cycles (Caldeira and Wickett, 2003; Raven et al., 2005). The pH changes have to be predicted appropriately with respect to space and time and impacts such as metal speciation on biogeochemical cycling have to be quantified and upscaled to the Earth system. Global nutrient cycling is undergoing significant changes (Duce et al., 2008) with as yet not well established consequences for ocean biogeochemistry. Trace metal delivery to the ocean may change due to climatic change, especially for iron, a biolimiting micronutrient (Jickells et al., 2005). Further concerns are the growing areas

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of poorly ventilated water masses in the oceans and sinking oxygen water column levels on the large scale (e.g., Whitney et al., 2007; Oschlies et al., 2008).

In addition to these pressing questions, major fundamental oceanographic questions are still not answered conclusively and need model tools for clarification. We are as yet not able to explain the glacial-interglacial variations in atmospheric CO₂ concentration (Archer et al., 2000; Sigman and Boyle, 2000; Toggweiler et al., 2006) for which the ocean has to play a major role. Marine particle fluxes, which provide the major link between the input of matter to the oceans from aeolian deposition as well as river loads and the output of matter through sedimentation and reverse weathering, are not yet well quantified due the difficulties in measuring biological export production rates and particle fluxes through the water column (e.g., Iverson et al., 2000; Buesseler et al., 2007). The marine paleoclimatic record, which is the result of a complex chain of transfer functions which map environmental and climatic change onto the marine sediment, needs explanation and exploitation for the calibration of climate models. In addition, the fate of toxic or otherwise hazardous substances in the oceans has to be identified and quantified including the realistic transport of CO₂ from leaks out of purposefully established sub-sea carbon storage reservoirs (IPCC, 2005).

So far, only a relatively small number of global interactive carbon cycle climate models (“Earth system models”, to be extended also with respect to other processes and cycles) exist. The ocean components in these systems for ocean physics (velocity field, density field, and sea ice) as well as biogeochemistry (inorganic chemistry, biological particle production and degradation, input and output of matter) are conceptually relatively closely related. The ocean biogeochemical models are either inorganic models (without a representation of the marine biosphere, e.g., Maier-Reimer and Hasselmann, 1987), “particles only models” representing only biological export production (based on phytoplankton production, e.g., Bacastow and Maier-Reimer, 1990; Najjar et al., 1992), or differing numbers functional groups and zooplankton (NPZD models and models involving further detailing of functional groups, e.g., Six and Maier-Reimer, 1996; Fasham et al., 1990; Aumont et al., 2003; LeQuéré et al., 2005). A few models

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further include an interactive sediment (Archer and Maier-Reimer, 1994; Heinze et al., 1999; Maier-Reimer et al., 2005; Ridgwell and Hargreaves, 2007; Gehlen et al., 2008). Due to the lack of sufficient knowledge of the “first principles” governing life processes in the ocean, biogeochemical ocean modelling is still in its developmental phase.

5 The various physical ocean models are mostly based on the Primitive Equations (full set of temporally averaged Navier-Stokes equations), usually discretised on a C-grid (Arakawa and Lamb, 1977) within a z-coordinate framework, i.e., the water column is discretised vertically with respect to fixed depth intervals. For physical tracer transport in the ocean, it is intriguing to explore alternative formulations, since different models still differ significantly with respect to reproducing global tracer distributions (Doney et al., 2004; Orr, 2002). While permitting an accurate computation of the horizontal pressure gradient driving geostrophic flow, vertical discretization on z-levels leads to spurious diapycnal mixing and upwelling. Even though the widely used parameterisation of ocean mixing due to large-scale turbulence after Gent and McWilliams (1990) has been shown to be a suitable formulation for oceanic mixing in ocean models, isopycnic ocean models have an advantage over those with geometric vertical layers. Their vertical coordinate mimics the real structure of the water column as stratified layers of constant density, and thus avoids artificial mixing and advection in the ocean interior. Their disadvantages include the problem of massless layers, the necessity of adding a mixed layer model to adequately represent surface processes, and the induction of a horizontal pressure gradient error by the sloping density surfaces. Models with different vertical schemes thus complement each other and can be used as one basis for an uncertainty assessment.

25 We present here a new coupled isopycnic physical-biogeochemical model based on two already well established components: the dynamical isopycnic ocean model MICOM (Bleck and Smith, 1990; Bleck et al., 1992) with a series of modifications (Bentsen et al., 2004) and the biogeochemical model HAMOCC (Maier-Reimer et al., 2005). While MICOM has been used successfully in abiotic tracer transport studies (Orr, 2002; Gao et al., 2005; Orre et al., 2008) and also for carbon uptake studies

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(Matsumoto et al., 2004), integrations including nutrients, oxygen, and carbon had to be limited to short periods since the computations were too demanding for resources at the time (Drange, 1996).

Initially, we will give a description of the model components and the changes made to HAMOCC to make it compatible with an isopycnic ocean model. After this we evaluate the circulation and temperature and salinity distributions in the physical model. We evaluate the main biogeochemical model parameters with respect to the physical results and present results from several sensitivity studies. In addition, we will also consider the uptake of anthropogenic CO₂ in the model and discuss the conclusions that can be drawn from a stand-alone ocean carbon cycle model with a prognostic slab atmosphere.

2 Model Description

2.1 The physical ocean model MICOM

The numerical methods and thermodynamics of MICOM are documented in Bleck and Smith (1990), and Bleck et al. (1992). Several important aspects deviate from the original model in the version of MICOM used in this study. Since the model version employed here includes several new features that have not been published, we will give a fairly detailed model description and evaluate model performance on a global scale, with a particular view on the influence of the physical model on biogeochemistry.

We use incremental remapping as an advection algorithm as proposed by Dukowicz and Baumgardner (2000). This multi-dimensional, second order accurate algorithm is expressed in flux form and assures conservation for tracers. It also guarantees monotonicity for tracers for any velocity field that does not violate the Courant-Friedrichs-Levy condition of the method. Neither of these conditions was assured in the original flux-corrected transport (FCT) scheme (Zalesak, 1979), used for layer thickness and MPDATA (Smolarkiewicz and Margolin, 1998) used for tracers as implemented in

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MICOM. While the incremental remapping is computationally rather expensive for one tracer, the cost of adding additional tracers and age tracers is modest which makes it well-suited for use with a ocean carbon cycle module that contains a large number of passive tracers.

5 The original MICOM uses potential density, ρ_r , with reference pressure at 0 db as the vertical coordinate. This ensures that the very different flow and mixing characteristics in neutral and dia-neutral directions is well represented near the surface since isopycnals and neutral surfaces are similar near the reference pressure. For pressures that differ substantially from the reference pressure, this does not hold. In this study, we
10 choose a reference pressure of 2000 db as the non-neutrality of the isopycnals in the world ocean is then reduced compared to having the reference pressure at the surface (McDougall and Jackett, 2005).

2.1.1 Pressure gradient force

Traditionally, MICOM expresses the pressure gradient force (PGF) as a gradient of a
15 potential on an isopycnic surface,

$$\frac{1}{\rho} \nabla_z \rho \approx \nabla_{\rho_r} M, \quad (1)$$

as such a formulation has favourable numerical properties (Hsu and Arakawa, 1990). This is only accurate if the density can be considered as a function of potential density and pressure alone. This is not the case (de Szoek, 2000) and assuming such a
20 functional form of density causes large PGF errors when the pressure is substantially different from the reference pressure. There have been several attempts to modify the MICOM PGF formulation (1) in order to incorporate the effects of a more accurate representation of density (Sun et al., 1999; Hallberg, 2005). Inspired by recent work of Rainer Bleck (personal communication, 2006), we have based our formulation
25 on Janjić (1977) where the PGF is expressed as a gradient of the geopotential on a

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pressure surface

$$\frac{1}{\rho} \nabla_z \rho = \nabla_p \phi. \quad (2)$$

The geopotential at a certain pressure p is found by integrating the hydrostatic equation from the invariant geopotential at the bottom, ϕ_b :

$$\phi = \phi_b - \int_{p_b}^p \frac{dp}{\rho}. \quad (3)$$

A suitable functional form of the equation of state has been chosen, inspired by Jackett et al. (2005), that ensures an accurate representation of density compared to the Feistel (2004) equation of state and an analytic expression for the integral in Eq. (3).

2.1.2 Diapycnal mixing

The diffusivity of diapycnal turbulent mixing, ν_d , in the model is the sum of a background diffusivity and a Richardson number dependent diffusivity:

$$\nu_d = \nu_b + \nu_r, \quad (4)$$

$$\nu_b = \frac{C}{N}, \quad \nu_r = \nu_0 \max \left\{ 0, 1 - \left(\frac{Ri_g}{Ri_0} \right)^2 \right\}^3, \quad (5)$$

$$N^2 = \frac{g}{\rho} \frac{\partial \rho}{\partial z}, \quad Ri_g = \frac{N^2}{(\partial u / \partial z)^2 + (\partial v / \partial z)^2}, \quad (6)$$

where N is the Brunt-Väisälä frequency and Ri_g is the local gradient Richardson number. The parameter determining the background diffusivity is set to $C = 1.8 \times$

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$10^{-7} \text{ m}^2 \text{ s}^{-2}$, while the critical Richardson number is $Ri_0 = 1$. The maximum Ri_g dependent diffusivity ν_0 is set to $500 \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$ in the 300 m closest to the ocean floor to parameterize gravity current mixing, and $50 \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$ elsewhere to parameterize shear instability mixing. The numerical implementation of the diapycnal mixing follows the scheme of McDougall and Jackett (2005) but with additional care in handling the vigorous mixing due to shear instabilities. The original MICOM only used the background diffusivity, and the addition of the Ri_g dependent diffusivity greatly improves the water mass characteristics downstream of overflow regions.

2.1.3 Lateral turbulent mixing

Lateral turbulent mixing of momentum and tracers is parameterized by Laplacian diffusion where diffusive velocities (diffusivities divided by the local grid size Δ) of momentum and tracers are $1.0 \times 10^{-2} \text{ m s}^{-1}$ and $0.35 \times 10^{-2} \text{ m s}^{-1}$, respectively. With a grid size of about 130 km at 60° N and 60° S this gives actual minimum diffusivities of $1300 \text{ m}^2 \text{ s}^{-1}$ and $455 \text{ m}^2 \text{ s}^{-1}$ for momentum and tracers, respectively. Layer interfaces are smoothed with biharmonic diffusion with a diffusive velocity (biharmonic diffusivity divided by Δ^3) of $1.0 \times 10^{-2} \text{ m s}^{-1}$ giving a minimum diffusivity of about $2.2 \times 10^{13} \text{ m}^4 \text{ s}^{-1}$.

2.1.4 Sea-ice

The thermodynamic module incorporates freezing and melting of sea-ice and snow covered sea-ice (Drange and Simonsen, 1996) and is based on the thermodynamics of Semtner (1976). The dynamic part of the sea-ice module follows the viscous-plastic rheology of Hibler (1979). The dynamic ice module has been further modified by Harder (1996) to include description of sea-ice roughness and the age of sea-ice. A third order weighted essential non-oscillatory scheme (Liu et al., 1994) is used for the advection of sea-ice fraction, thickness, and age.

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2.2 The ocean carbon cycle model HAMOCC 5.1

A detailed description of HAMOCC5.1 is given in Maier-Reimer et al. (2005). The model is based on the work by Maier-Reimer (1993) and has been used to analyse contemporary ocean carbon fluxes coupled to the physical ocean model MPIOM (Marsland et al. 2003) in Wetzel et al. (2005). Here we only summarise its main features and those that deviate from this description either because they were introduced later or because they bear on the inclusion of HAMOCC in an isopycnic ocean model. The values of key parameters are summarised in Table 1.

HAMOCC5.1 model uses the formulation of inorganic carbon chemistry in following Maier-Reimer and Hasselmann (1987). Surface $p\text{CO}_2$ is computed as a function of prognostic alkalinity, total DIC, temperature, pressure, and salinity. Dissolution of calcium carbonate at depth is a function of carbonate ion saturation state and a constant dissolution rate. The air-sea gas exchange processes of CO_2 , oxygen and nitrogen are a function of gas solubility, transfer velocity, and the difference between partial pressure tracers in air and water following Wanninkhof (1992). The model uses gas tracer solubilities according to Weiss (1970) and Weiss (1974), whereas the gas transfer velocity depends on the Schmidt number and prognostic wind speed at the surface. HAMOCC5.1 contains a simple diffusive slab atmosphere which allows the prognostic computation of atmospheric CO_2 levels for a stand-alone ocean set-up of the model forced by atmospheric reanalyses.

The model includes an ecosystem model of the NPZD (Nutrient-Phytoplankton-Zooplankton-Detritus) class (Six and Maier-Reimer, 1996) with nutrient co-limitations by phosphate, nitrate and iron (Aumont et al., 2003). Nutrient, carbon, and oxygen uptake and re-dissolution are treated according to Redfield stoichiometry. Growth of the single phytoplankton class is described by Michaelis-Menten kinetics and is in addition to nutrients limited by temperature and light intensity following Eppley (1972). Exported organic matter is remineralised throughout the water column below the euphotic zone, and the non-remineralised particles are collected by a sediment module.

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The organic tissue exported as Particulate Organic Carbon (POC) is associated with two major functional phytoplankton groups: diatoms and coccolithophores. Assuming that diatoms are the faster growing group, the amount of opal shells exported with the POC is computed first as a function of silicate concentration. The remaining POC export is assumed to contain CaCO_3 shells.

In the tropical oligotrophic, nitrate depleted regions, the marine ecosystem module accounts for atmospheric nitrogen fixation essential for cyanobacteria growth. Compared to the model version described in Maier-Reimer et al. (2005) the most recent version of HAMOCC includes sulphate reduction in oxygen poor waters and sediments in addition to nitrate reduction once all nitrate has been utilized. Also, the minimum concentration for phytoplankton is slightly increased, and a quadratic zooplankton mortality has been introduced to eliminate spurious Lotka-Volterra cycles that were present in earlier model versions, in particular in the equatorial Pacific. The dust deposition fields of Mahowald et al. (2005) are now used to provide iron input to the ocean. Remineralisation is now composed from three compartments: remineralisation of particulate organic carbon, dissolved organic carbon as in Maier-Reimer et al. (2005) and newly also dying phytoplankton.

HAMOCC5.1 only allows biological production in the top 90 m of the water column. In a z-level model this simply implies computing biological production in the layers that lie above this depth. In an isopycnal model layer thickness changes over time and space. We thus redefined the original criterium and now test for changes in layer thickness at every time step of the biogeochemical model: in order for biological production to occur at least 10 m of an isopycnal layer must lie within the 90 m euphotic zone. The layer that contains the 90 m boundary is split virtually into a part above and one below the euphotic zone boundary and is assumed to be instantly well-mixed (Fig. 1). Changes in nutrient concentrations etc. due to biological production are calculated with respect to the concentration in the entire layer and then scaled proportionally to the thickness

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of the layer that lies in the euphotic zone.

$$X_{\text{new}} = \frac{X_{\text{old}} \times dz_d + (X_{\text{old}} + dX_{\text{euph}}) \times dz_e}{dz} \quad (7)$$

where X is any biogeochemical tracer, dX_{euph} the change in X due to biological production, dz is the total layer thickness and dz_e and dz_d the parts of the layer that lie in the euphotic zone and below it, respectively. The diagnosed values for primary and export production follow the formulation for biological production and are also scaled to the top 90 m. Similarly, the part of the same layer that lies below the euphotic zone boundary experiences the remineralization processes that occur in the deep ocean resulting in dX_{deep} .

$$X_{\text{new}} = \frac{X_{\text{old}} \times dz_e + (X_{\text{old}} + dX_{\text{deep}}) \times dz_d}{dz} \quad (8)$$

We used HAMOCC's implicit sinking scheme with fixed sinking velocities, but excluded massless layers from the routine. The 12-layer sediment model included in HAMOCC 5.1 was used without amendment. It exchanges fluxes with the lowest ocean layer exceeding 0.5 m in thickness.

HAMOCC's biogeochemical tracers are advected and mixed as passive tracers in MICOM. MICOM uses a leap-frog time stepping scheme. For computational efficiency ran the biogeochemical tracers on only one of its time levels and thus only every second physical time step. Therefore we corrected the global tracer budget for changes in layer thickness caused by time-smoothing between the two physical time levels to ensure tracer conservation.

2.3 Model configuration, initialisation and forcing

The model is configured on a global grid where the North Pole was moved to a position over central Siberia to avoid a singularity in the computational ocean domain (Furevik

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et al., 2003, Fig. 1b). At the equator the grid resolution is 2.4° zonally and 0.8° meridionally. With increasing latitude the grid cells are gradually transformed to have more isotropic metric scale factors in the horizontal directions. The grid spacing ranges from 60 km in the Arctic and Southern Oceans to 180 km in the subtropical gyres. In the vertical, the model has 35 layers of which the uppermost is a mixed layer with temporally and spatially varying density. The potential densities of the isopycnal layers are in the range of 1030.12–1037.80 kg m⁻³.

The physical model is initialised from rest with temperature and salinity distributions based on the January climatologies from Levitus and Boyer (1994) and Levitus et al. (1994), respectively. Nutrients and DIC/Alkalinity in HAMOCC were initialised from global mean vertical profiles based on WOA05 (Boyer et al., 2006) and GLODAP (Sabine et al., 2005), respectively. The model was spun up for 950 years. The first 600 of these were performed with a monthly NCEP-based climatology (Kalnay et al., 1996). This does not include the synoptic and interannual variability necessary to obtain realistic sea ice distributions and surface mixing and mixed layers, and we switched to repeated passes of the NCEP Reanalyses 1950–1999 for another 350 years of spin-up.

The NCEP reanalysis data provide radiative fluxes and turbulent surface fluxes of momentum and heat, along with information on the ocean surface state (temperature and sea ice concentration). The forcing scheme and procedure proposed in Bentsen and Drange (2000) is used here. The scheme reproduces the reanalysis fluxes if the model has the same surface state as in the reanalysis. These states will generally differ, however, and the fluxes are modified accordingly. The turbulent fluxes are modified consistent with the bulk parameterization of Fairall et al. (1996) and long-wave radiative fluxes consistent with the Berliand and Berliand (1952) parameterization. Precipitation is directly provided by the NCEP reanalysis and the evaporation is calculated based on the (modified) latent heat flux. To maintain a stable and realistic Atlantic meridional overturning circulation (AMOC) during the spin-up and years before 1948, a Newtonian relaxation of sea surface salinity (SSS) and sea surface temperature (SST) is applied with a relaxation time scale of 30 days for a 50 m thick mixed layer, linearly decreasing

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with thicker mixed layers (see Bentsen et al., 1999). For experiments forced by NCEP Reanalyses after 1948 this is reduced to 60 days for SSS and 180 days for SST to allow the development of surface SST and SSS anomalies, while still maintaining a realistic AMOC.

5 The model is initialised with a preindustrial atmospheric CO₂ level of 278 ppm and is then allowed to evolve freely during the spin-up. It stabilises at a value of 286 ppm. We start our emission scenario runs in 1860 using values for anthropogenic CO₂ emissions from Boden et al. (2009). Since no gridded physical atmospheric forcing data exist prior to 1948, we use perpetual forcing with the NCEP year 1959 for the period 1860–1947. 1959 is both relatively neutral for both NAO (see e.g. <http://www.cru.uea.ac.uk/cru/climon/data/nao/>) and ENSO (see e.g. <http://www.cdc.noaa.gov/people/klaus.wolter/MEI/>), as well as being early in the NCEP period. This should ensure that air temperatures and Southern Ocean winds are still relatively close to preindustrial conditions. Prior to starting the emission simulations the model was run with year 1959 forcing, but no emissions for 20 years to avoid a drastic change at the start of the emission simulation. Initially, four simulations were performed:

- CLIM: The reference simulation that continues the spin-up. Forced by NCEP year 1959 without CO₂ emissions for the entire 1860–2007 period.
- 20 – CLIM+EMS: Forced by NCEP year 1959 for the entire 1860–2007 period, but with anthropogenic CO₂ emissions from 1860 to 2007.
- SYN: Forced by NCEP Reanalyses 1948–2007 without CO₂ emissions. Initialised from CLIM at the end of year 1947.
- 25 – SYN+EMS: Forced by NCEP Reanalyses 1948–2007 and including anthropogenic CO₂ emissions for this period. Initialised from CLIM+EMS at the end of 1947.

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3 Model results and evaluation

Since most observational data available was obtained over the last three decades we use an average of the years 1978–2007 of the SYN+EMS simulation for comparison to global climatologies. We summarise model performance in a Taylor diagramme (Fig. 2), and provide a more detailed discussion in the following sections.

3.1 Physical model

Amended versions of MICOM have been evaluated in several studies throughout its development particularly in the North Atlantic (e.g. Nilsen et al., 2003; Bentsen et al., 2004; Gao et al., 2004; Drange et al., 2005; Hatun et al., 2005; Lohmann et al., 2008). Here we present an evaluation of the main features of the physical model mainly to provide a context for the biogeochemical tracer distributions.

Overall, the simulated circulation as well as the potential temperature and salinity fields are realistically described by the model. The model reproduces both global salinity S and potential temperature Θ distributions well, with a slightly better pattern representation for temperature (Fig. 2). Simulated distributions have larger normalised standard deviations (STD), i.e. stronger gradients, than observed. This is likely due to MICOM not being very diffusive and to low exchange between the isopycnal layers and their outcropping region in the surface mixed layer across their diapycnal interface. Surface distributions for both salinity and temperature are close to the observed ones (Figs. 3 and 4). Since salinity is relaxed quite strongly to observations, SSS deviations are generally below 0.2 psu. Notable exceptions from this are the estuaries/plumes of big rivers, especially in the Arctic, in whose vicinity the model overestimates salinity. Rather than explicitly prescribing river fresh water fluxes, MICOM uses a run off scheme to route NCEP precipitation on land into the ocean. While this ensures fresh water conservation, it means that fresh water inputs from continents in the model do not completely agree with the real ones, especially in the vicinity of big rivers. Deviations in both SSS and SST in the North Atlantic can be explained by deviations in the

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path of the North Atlantic Current that leaves the North American coast too far North as is common in coarse resolution models. The resulting large heat loss may lead to low SSTs in the sub-polar gyre and northwestern sub-tropical gyre. The North Pacific and Kuroshio show a similar pattern. SSTs are generally too cold. A notable exception are the upwelling regions where overestimated temperatures indicate that the upwelling strength is too weak.

Meridional sections in the Atlantic show that the model produces deep and bottom water masses that are too fresh and cold (Figs. 3 and 4). Simulated sub-surface and deep waters in the Greenland Sea are warmer and slightly fresher than observed, possibly due to underestimated convection intensity in the model. The overly cold and fresh waters in the Labrador Sea in our simulation stem from two origins: a Denmark Strait overflow where water appears to experience too little entrainment during its descent and local convection of overly fresh and cold surface waters in the Labrador Sea caused by misrepresentations in the course of the North Atlantic Current.

Overestimated Θ and S above 1000 m in the tropics and subtropics indicate weak replenishment of Sub-antarctic Mode Water (SAMW) and Antarctic Intermediate Water (AAIW). Water with the characteristically low salinity of AAIW is subducted to depths of up to 3000 m in the AAIW formation regions, because its temperatures are too low by 2.5°C and thus its density allows it to sink to these depths. The source of this cold water can be found further South. While deep convection and formation of cold and saline waters is only observed on Antarctic shelves, the water column of the deep Weddell Sea is only weakly stratified when using the surface as a reference level for potential density. In fact, the choice of 2000 m as a reference level for our model density layers means that the observed stratification in this area is unstable (Fig. 5). This initially leads to overestimated vertical mixing in our model which erodes the Warm Deep Water core in the Weddell Sea, an area whose representation in ocean models is particularly sensitive to the parametrisation of vertical mixing (Timmermann and Beckmann, 2005). Due to the long spin-up of the model this leads to a relatively homogeneous cold water column throughout the Southern Ocean, and both Antarctic Bottom Water (AABW) and

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Central Intermediate Water (CIW) exported are colder than observed. Consequently, water upwelled in the southern limb of the Deacon Cell (Fig. 6a) is too cold and leads to the excessive subduction of AAIW described above.

This process also leads to the relatively strong AABW cell of 14 Sv in the global meridional overturning circulation (Fig. 6a) since the deep AABW outflow is joined by water which would normally exit the Southern Ocean as AAIW at much shallower depths. The model does, however, show a realistic North Atlantic overturning of 14 Sv. Major features of the horizontal circulation (Fig. 6b) are in general well-reproduced with an ACC strength of 150–160 Sv, a Weddell Gyre of 50 Sv (Beckmann et al., 1999), a Ross Gyre of 20–30 Sv (Assmann and Timmermann, 2005) and a North Atlantic Sub-tropical and Sub-polar Gyres of 30–40 Sv.

3.2 Biogeochemistry

Correlations for all biogeochemical 3-D fields that we compared lie above 0.8 (Fig. 2). As for Θ and S simulated standard deviations are higher than observed by 10–35% again indicating that the model overestimates gradients in the distributions.

Phosphate is useful for a baseline assessment of the biogeochemical tracers since it is not affected by air-sea gas exchange. Its surface concentrations are generally low due to uptake during biological production. Higher deep ocean concentrations arise from the remineralisation of this biological matter after sinking out of the euphotic zone. Areas of higher surface phosphate are associated with upwelling, e.g., the Southern Ocean and the Equatorial Pacific, or enhanced mixing, e.g. the North Pacific. Lower deep concentrations indicate areas of deep water formation like the North Atlantic.

Simulated surface phosphate concentrations are generally underestimated (Fig. 7). Exceptions are the Siberian shelves, where dissolution from the sediment increases concentrations, and the Southern Ocean below 60° S. The weak upwelling in the model discussed in the previous section leads to low phosphate concentrations in the Equatorial Pacific. Sarmiento et al. (2004) argue that strong vertical tidal mixing is an important process for resupplying the surface waters in the North Pacific. This process

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is not included in our model and likely contributes to low phosphate concentrations in the model. The North Pacific and the Southern Ocean between 40° and 60° are the formation regions for intermediate waters that supply much of the low-latitude ocean with nutrients (Sarmiento et al., 2004). In our simulation surface phosphate concentrations in these regions are too low and the most obvious deviation from observations. Meridional sections in the Atlantic (Fig. 7) illustrate how the low simulated preformed surface concentrations are communicated into the intermediate water masses. Analysis of the physical fields showed that the model's Antarctic Intermediate Water (AAIW) tongues are less pronounced than in observations. The subduction of low-salinity water that would normally contribute to AAIW to much greater depth aids the trapping of nutrients in the Antarctic Marginal Seas and Antarctic Bottom Water (AABW) where the model overestimates nutrient concentrations. We will explore the relative roles of physics and biology for the preformed concentrations in the AAIW formation regions through sensitivity studies in Sect. 5.

To first order, the global distribution of oxygen is a mirror image of the distribution of nutrients. In addition, surface oxygen concentrations bear the imprint of temperature dependent dissolution with higher concentrations in cold high-latitude waters. According to the Taylor diagram (Fig. 2) oxygen, along with potential temperature, is the tracer that the model simulates best. A comparison of modelled and observed surface oxygen distributions (Fig. 8) confirms this, as deviations are small except in the Arctic and Southern Oceans. The modelled sea ice extent in the Southern Ocean is too large in summer and thus prohibits air-sea gas exchange. Also, our model – as most other biogeochemical ocean models – treats sea ice as an impregnable lid to air-sea fluxes, which it may not be. In addition, observations are generally summer-biased in polar regions. This may lead to higher observed oxygen concentrations since they are dominated by oxygen release during biological production. Oxygen content of the deep and bottom water is too high indicating that vertical mixing in deep water formation regions is overestimated by the model (Fig. 8). The simulated sub-equatorial oxygen minima are not as pronounced in the model. Reasons for this lie in underestimated upwelling

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and nutrient supply which lead to low biological production and remineralisation of sinking organic matter. However, it also indicates that our model is not prone to the nutrient trapping at the Equator commonly seen in biogeochemical ocean models (Aumont et al., 1999). Low AAIW oxygen concentrations confirm the weak formation rates of this water mass in the model.

Dissolved Inorganic Carbon (DIC, Fig. 9) shows in principle the same deviations from the observations as phosphate, but performs worse in the Taylor diagramme since it is also affected by air sea gas exchange and opal/CaCO₃ production. Again, the main feature is the trapping of carbon in the far Southern Ocean and low carbon in the NADW.

Global mean values for biological production are within observation-based estimates. Mean global primary production is at $48.5 \pm 2.0 \text{ Gt C yr}^{-1}$, POC export production at $12.4 \pm 0.53 \text{ Gt C yr}^{-1}$, calcium carbonate export at $0.91 \pm 0.04 \text{ Gt C yr}^{-1}$ and silicate export at $157.6 \pm 7.2 \text{ Tmol Si yr}^{-1}$ (Treguer et al., 1995). Our value for global primary production agrees well with a satellite-based estimate of 48 Gt C yr^{-1} by Behrenfeld et al. (2006) and is within the range of 35 to 70 Gt C yr^{-1} given by Carr et al. (2006). Export production is well within the observational estimates which range from 11 to 22 Gt C yr^{-1} (Eppley and Peterson, 1979; Laws et al., 2000; Schlitzer, 2000), and markedly higher than that in other contemporary coupled climate carbon cycle models (Schneider et al. 2008).

The global distribution of primary and export production (Fig. 10) shows maxima in the North Atlantic and Pacific and in the coastal and equatorial upwelling regions, as expected (e.g. Schneider et al., 2008, Fig. 6). Biological production in the shallow shelf areas is underestimated as in most global models due to the lack of resolution and nutrient inputs by rivers and run-off. Due to the low nutrient supply to the North Pacific and Atlantic simulated biological production in these areas is also low. The most striking deviation from the observation-based distribution of primary and export production is the broad band of high production in the Southern Ocean reaching south

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to the simulated summer sea ice edge. We will explore possible reasons for this in a series of sensitivity studies in Sect. 5.

4 Air-sea fluxes and the uptake of anthropogenic CO₂

There is a steady ocean carbon uptake of $0.05 \pm 0.006 \text{ Gt C yr}^{-1}$ in the reference simulation, CLIM, which indicates that this simulation has reached equilibrium after the spin-up (Fig. 11a). In the simulation forced by NCEP Reanalyses, SYN, annual ocean carbon fluxes vary between -0.2 and 0.3 Gt C yr^{-1} . Notable is a positive trend in the carbon fluxes in SYN after the late 1990s indicating that changes in the physical system reduce the ocean's ability to take up carbon. In the equivalent simulation including emissions, SYN+EMS, the increase in ocean carbon uptake slows after this point, while uptake in the CLIM+EMS simulation continuously forced by NCEP year 1959 increases more sharply due to an acceleration in anthropogenic emissions. Thus our model results agree with recent findings of a weakening of the ocean carbon sink due to a climate-carbon cycle feedback (Schuster and Watson, 2007; Schuster et al., 2009; LeQuéré et al., 2007; Metzl, 2009).

Air-sea carbon flux climatologies based on observations estimate an ocean carbon uptake of $2.2 \pm 0.5 \text{ Gt C yr}^{-1}$ for 1995 (Takahashi et al., 2002) and in an updated version $1.4 \pm 0.7 \text{ Gt C yr}^{-1}$ ($2.0 \pm 0.7 \text{ Gt C yr}^{-1}$ anthropogenic uptake) for 2000 (Takahashi et al., 2008). Simulated ocean uptake for these years is 2.8 Gt C yr^{-1} and 2.7 Gt C yr^{-1} , respectively, and thus just outside upper bounds for the observational values. It is, however, worth noting that Takahashi et al. (2008) assume a preindustrial ocean carbon source of 0.4 Gt C yr^{-1} which brings their estimate of anthropogenic carbon uptake closer to ours since the simulated 2000 flux without emissions in SYN is only $0.06 \text{ Gt C yr}^{-1}$.

A comparison of the observed year 2000 climatology and the simulated fluxes for the same year shows that the model underestimates outgassing in the upwelling regions like the equatorial Pacific (Fig. 12f). This is due to the underestimation of upwelling

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rates discussed in Sect. 3.1. In addition, uptake in the high latitude sink regions of the North Pacific, North Atlantic and Southern Ocean is generally overestimated. Reasons for this are likely associated with the artificially large vertical mixing in the model which allows fast subduction of carbon taken up particularly in winter and overestimated export production in the Southern Ocean. The fact that MICOM has a bulk mixed layer probably also contributes to this phenomenon: carbon taken up at the ocean surface is instantaneously mixed with a volume of water that particularly in winter is large. The resulting changes to surface $p\text{CO}_2$ are consequently small and further uptake is easily possible.

The Southern Ocean is the region with the largest disagreement between model and climatology. When the area south of 56°S is excluded, the correlation between simulated and observation-based air-sea fluxes and $\Delta p\text{CO}_2$ patterns improves from 0.38 to 0.73 for $\Delta p\text{CO}_2$ and from 0.55 to 0.68 for the air sea fluxes. This is particularly striking for $\Delta p\text{CO}_2$ since a thick perennial sea ice cover in the model prohibits outgassing in the central Weddell Sea and leads to excessively high surface ΔCO_2 values. However, the observed climatology is likely summer-biased in this region which would lead to underestimated $\Delta p\text{CO}_2$. There also is as yet no parametrisation of carbon fluxes through sea ice available for large-scale models. Observations indicate that these fluxes are large (Semiletov et al., 2004), but most ocean carbon cycle models represent sea ice as an impregnable lid making the interpretation of their results in sea-ice covered areas difficult.

Using NCEP Reanalyses after 1948 leads to an atmospheric CO_2 about 2 ppm higher than in the experiment that was continuously forced with NCEP year 1959 (Fig. 11b). All models of the Coupled Climate Carbon Cycle Model Intercomparison Project (C^4MIP) have a carbon cycle-climate feedback of less than 15 ppm in 2006 (Friedlingstein et al., 2006) and this small effect is thus not unreasonable for a stand-alone ocean model. Start and end values of the simulated time series are close to the observed values (Fig. 11b). However, the simulated atmospheric CO_2 increases more slowly than that observed up to 1950, likely because our model misses a land

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carbon model and a representation of the “pioneer effect” which made the land a significant carbon source during this period (Houghton, 1999). Towards the end of the 20th century the simulated increase exceeds that observed, indicating a weakening of the ocean sink and a change of the land from source to sink. Of the 242.7 Gt C emitted between 1860 and 1994, the ocean in our model takes up 90 Gt C, leaving 152.7 Gt C in the atmosphere. Both of these numbers are smaller than those deduced from observations by Sabine et al. (2004) due to the lack of the net land source in the model.

The column inventory of anthropogenic CO₂ (Fig. 13) agrees well in pattern and magnitude with that deduced from observations by Sabine et al. (2004). Column inventories that exceed those of the observations may both be due to overestimated ventilation of the deep basins in the model and inaccuracies of the C* method used to deduce anthropogenic CO₂ from the available observations (Matsumoto and Gruber, 2005). Underestimated column loads of anthropogenic CO₂ in the tropical Atlantic can be explained both by the weak renewal of AAIW in the model and by the fact that our simulation misses emissions due to the “pioneer” effect early in the emission scenario which would have had time to penetrate further into the ocean interior.

5 Sensitivity studies

One of the most striking deviations from observations in the model is the large biological production and the excessively high nutrients in the Southern Ocean, an area known as a HNLC-region (High Nutrient Low Chlorophyll). Biological production in the Southern Ocean is primarily limited by light and iron (e.g. Martin, 1990). We thus focus our sensitivity studies exploring the origin of the artificially large Southern Ocean primary production on these factors. We ran each of the sensitivity simulations for the NCEP period from 1948–2007 with anthropogenic CO₂ emissions starting from year 1947 of the CLIM+EMS simulation, much like the setup of SYN+EMS simulation. We compare annual means for 2007 for each of the sensitivity runs to the same year in

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SYN+EMS. Since the model has not been run to equilibrium for each of these 60 year long simulations, the differences shown are indicative only of the sign of the changes and the relative magnitude, but not of the final magnitude of the induced changes and total feedback on the carbon cycle. The following sensitivity experiments were conducted:

- IRON: Biological production in the model is phosphate, nitrate and iron co-limited by the expression

$$\text{nutrient} = \min \left(PO_4, \frac{NO_3}{R_{N:P}}, \frac{Fe}{R_{Fe:P}} \right)$$

where $R_{Fe:P} = 3. \times 10^{-6} \times R_{C:P}$, $R_{C:P} = 122$ and $R_{N:P} = 16$, according to Redfieldian stoichiometry. To enhance iron limitation we used $R_{Fe:P} = 5. \times 10^{-6} \times R_{C:P}$ globally in this sensitivity study. Sarthou et al. (1997) derived an iron to carbon uptake ratio of $R_{Fe:C} = 3.47 \times 10^{-6}$ from observations in the Indian Sector of the Southern Ocean.

- ABS: In our standard model version the entire mixed layer (or the part that is within the 90 m euphotic zone) receives the full incoming short wave radiation. To reduce biological production in regions with deep mixed layers in spring, the mixed layer is split virtually into a 20 m thick surface layer and the rest below this up to a maximum euphotic zone depth of 90 m. Only the top 20 m receive the full incoming short wave radiation, while the short wave radiation incident on the lower part of the layer has been attenuated by absorption in the top 20 m.
- DIAPYC: One issue in isopycnal models are layers that outcrop at the surface. In order to represent surface processes a mixed layer with varying density is added in MICOM which contains these outcrops. There is thus an artificial diapycnal boundary between the isopycnal layers and their surface expression. Exchange across this boundary happens by changes in mixed layer thickness due to surface

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buoyancy fluxes and diapycnal mixing as used between the standard isopycnal layers. To account for the fact that fluxes between the mixed layer and the isopycnal layer are isopycnal rather than diapycnal we increased the parameter C in the expression for the background diffusivity ν_b in Eq. (5) for the layer directly below the mixed layer by a factor of 10.

In both IRON and ABS, export production decreases in a broad band between 40° S and the sea ice edge (Fig. 14). In both cases this coincides with an increase in phosphate concentrations in this area (Fig. 15) suggesting that the underestimation of phosphate concentration seen in Fig. 7 is due to excessive export production. However, the globally enhanced iron limitation leads to a significant and unrealistic increase in surface phosphate throughout the Pacific, South Atlantic and Indian Ocean. No change in phosphate is seen in IRON in the North Atlantic and Indic where large dust input from the Sahara leads to nitrate rather than iron limitation. Consequently, there is no change in export production in the North Atlantic in IRON, while there is a reduction in both the North and Equatorial Pacific.

In ABS increased surface phosphate (Fig. 15) in the North Pacific and Atlantic coincides with reduced export production (Fig. 14). While the former is a desirable improvement of the model results (Fig. 7), the latter is already underestimated in the standard model version. This indicates that we are possibly missing additional nutrient sources in these areas. The model only receives fresh water, but not the accompanying nutrient fluxes from the large Siberian rivers. These are suggested to be a significant source of nutrients in the North Atlantic (Jones et al., 2003), in addition to nutrient rich waters from the North Pacific that enter the Arctic and North Atlantic through Bering Strait. In the North Pacific, tidal mixing has been suggested as an important factor for transporting nutrient-rich intermediate waters into the surface layer to sustain vigorous biological production (Sarmiento et al., 2004). Our model, like most global OGCMs, does not include tides. However, the enhanced mixing between the surface and that layer directly below in DIAPYC lead to both enhanced export production and surface phosphate in this area (Figs. 14 and 15). This confirms that enhanced mixing is indeed

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needed to make the highly nutrient-rich deep and intermediate waters of the North Pacific accessible for biological production in the surface layer. In contrast, there is no significant effect in DIAPYC in the North Atlantic where waters directly underlying the surface are low in nutrients. Similarly to the North Pacific, enhanced diapycnal mixing between the mixed layer and sub-surface enhances both surface phosphate and export production in the Southern Ocean, extending both to just north of 40° S. This northward extension brings our model distributions closer to observed ones and highlights the importance of exchange between the surface and intermediate as well as mode waters in this region. The strong increase of export production further south in DIAPYC may in part be due to the overestimated deep nutrient concentrations south of the ACC.

Evaluating the vertical phosphate distribution along a meridional section in the Atlantic, we found that while simulated concentrations south of the ACC are overestimated, concentrations in the AAIW and NADW are too low (Fig. 7). All sensitivity studies remedy the distribution in the Southern Atlantic by lowering nutrient concentrations in both CDW and AABW and increasing them in the AAIW and SAMW (Fig. 16). Decreasing export production in the IRON and ABS experiments leads to an increase of pre-formed surface phosphate in those regions where AAIW and SAMW are subducted. The decrease of fast-sinking detritus that remineralises below the intermediate waters lowers phosphate concentrations south of the ACC and finds a surface expression in a decrease of phosphate concentrations where these deep waters upwell underneath sea ice in the Weddell Sea. The existence of the same pattern in DIAPYC albeit with a lesser reduction in the south indicates that the main factor for the enhanced intermediate water concentrations is indeed the increased northward extension of both pre-formed nutrients and biological production.

While the enhanced mixing in DIAPYC does lead to a 2.3 Pg C yr⁻¹ increase in POC export production (Fig. 17a and Table 2), the coincident exposure of carbon-rich surface waters to the atmosphere leads to a slight decrease in oceanic carbon uptake (Fig. 17b) and higher atmospheric CO₂ (Fig. 17c). Export production in the IRON experiment is reduced by 36% to 8.3 Pg C yr⁻¹ which results in a 26.8 ppm increase

in atmospheric CO₂ over the 60yr time period of the sensitivity experiment. The 2.5 Pg C yr⁻¹ reduction in export production in the ABS experiment leads to an atmospheric CO₂ increase of 14.4 ppm. Even though changes in export production from the reference run are almost constant over time (Fig. 17a), it takes around 20 years for the initial reaction of global ocean CO₂ uptake (Fig. 17b) to subside. Especially in ABS and IRON the reduction of summer pCO₂ drawdown due to the abrupt reduction in biological production initially leads to carbon outgassing. As the ocean surface adjusts to the new annual cycle some of this carbon is taken up again aided by the simulated increase in export production after 1980.

6 Discussion

We have presented a new ocean carbon cycle model consisting of the isopycnic ocean model MICOM and the biogeochemistry model HAMOCC. After a thorough evaluation we conclude that it is well-suited for global carbon cycle studies. Previous attempts to use isopycnic ocean models as the physical part of ocean carbon cycle models are scarce and had their problems (Drange, 1997; Haigh et al., 2001). However, continuing development of layered ocean models now has improved performance so much that our attempt has been more successful. The strength of an isopycnic vertical coordinate lies in it being a close approximation to the surfaces on which tracer transport in the interior ocean actually takes place. We find that our model reproduces, e.g., the invasion of anthropogenic carbon into the interior ocean extremely well.

A disadvantage of the isopycnic coordinates lies in the need to use a mixed layer of varying density to handle layer outcropping at the ocean surface and air-sea exchange processes. MICOM uses a bulk mixed-layer which implies that vertical resolution at the ocean surface is limited. It is therefore not particularly suited to detailed surface-ocean ecosystem studies that require a highly resolved euphotic zone. A relatively simple ecosystem model like the one in HAMOCC is therefore a good match to MICOM. Most of the adaptation needed to make HAMOCC compatible with an isopycnic vertical

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coordinate was indeed in representing biological processes in the euphotic zone, e.g., phytoplankton growth and light absorption. A hybrid-coordinate model like HYCOM (Bleck, 2002) where the surface layer is split into closely spaced z-levels is the better tool where dedicated ecosystem studies are concerned.

5 A series of sensitivity studies were performed, initially with the purpose of exploring the origin of several model shortcomings like the vigorous biological production in the Southern Ocean. A result common to all of these is the importance of the distribution of preformed nutrients in the Southern Ocean on the global nutrient distribution as highlighted by Sarmiento et al. (2004) and Marinov et al. (2006). The overesti-
10 mated biological production in our reference model experiment depletes surface nutrients leading to low nutrient concentrations in the AAIW. Instead, these nutrients sink into the upwelling CDW where they are remineralised and trapped within the unproductive southern loop of the MOC (Marinov et al., 2006). Reducing this export by both enhanced iron limitation and a more sophisticated light absorption scheme leads to a
15 reduction in phosphate in this southern loop and an increase in nutrients in mode and intermediate waters. While enhanced diapycnal mixing between the mixed layer and the isopycnal layer immediately below leads to an increase in export production in the Southern Ocean, it produces the same pattern as the biogeochemical changes in the phosphate distribution, since it is also a mechanism that increases surface nutrients in
20 the mode and intermediate water formation regions.

From the large-scale surface increase in phosphate in IRON we conclude that this approach is either too simplistic or that iron is not at the root of the problem. Our standard carbon to iron ratio appears to be realistic (Sarhou et al., 1997) and modelled iron concentrations agree with available observations (Moore and Braucher, 2008).
25 If iron were to be the solution, uptake ratios would have to vary with either species or location, which they probably do. However, observations of iron and its uptake in plankton are sparse and therefore a more sophisticated parametrisation is hard to come by. In any case it would probably go beyond the level of complexity of the rest of our biogeochemical model.

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As discussed above most of the challenges in adapting an ocean biogeochemistry module for use with an isopycnic coordinate lay in making it suitable for use with a bulk mixed layer. An important point here was light absorption. While we explored various parametrisations during model development, we settled on a fairly simple formulation for our reference run. The ABS sensitivity run shows that this was maybe too simple, but also that the improvement of this process made a large impact on our results and will be used in further studies.

In the evaluation of our model results the effect of sea ice on biogeochemical processes and its representation as an impregnable lid was touched upon several times. There is mounting evidence that sea ice is not simply a lid to air-sea exchange and biology (Semiletov et al., 2004), but more sophisticated parametrisations are necessary to include this in ocean carbon cycle and Earth System Models. Motivated by excessive DIC values under sea ice particularly in the Southern Ocean, we allowed some exchange through sea ice by limiting the maximum sea ice concentration in HAMOCC experiences to 0.5. More permeable sea ice would allow outgassing of some of this carbon and some biological production, attempting to counteract the overestimated summer sea ice extent in the Southern Ocean. After a 60 year simulation the effects are still largely confined to sea ice covered areas. Changes to atmospheric CO₂, ocean carbon uptake and export production are not significant. There is a substantial reduction in phosphate in Arctic surface waters that extend into the Labrador Sea where it is subducted into the deep ocean. This implies that Arctic waters play an important role in North Atlantic nutrient supply. Missing nutrient inputs from Arctic rivers (e.g. McClelland et al., 2006) and an underestimated Pacific water layer (Falck et al., 2005; Jones et al., 2003) may explain the low nutrient concentrations and biological production in these regions in our model.

Based on this evaluation we conclude that the new isopycnic ocean carbon cycle model we have presented here is a useful tool to investigate ocean biogeochemistry processes and their interaction with the physical system. It has been included in the Bergen Climate Model and first results prove that its capabilities in projecting

the climate system's response to anthropogenic CO₂ emissions lie well within the bounds of other coupled carbon cycle-climate models (Tjiputra et al., 2009). Work on a new model version currently in progress includes some of the parametrisations explored in our sensitivity studies as well as a parametrization of vertical mixing that circumvents the issue of unstable Southern Ocean sub-surface stratification with a 2000 m reference pressure.

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Table 1. HAMOCC parameters, their symbols and values in the reference experiment.

Euphotic zone		
Phytoplankton		
initial slope of the P-I-curve	α	$0.02 \text{ d}^{-1} (\text{W m}^{-2})^{-1}$
maximum growth rate	V_P	0.4 d^{-1}
light attenuation coeff. of water	k_w	0.04 m^{-1}
light attenuation coeff. of chlorophyll	k_c	$0.006 \times 10^{-6} \times 122. \text{ m}^{-1} (\text{kmol P m}^{-3})^{-1}$
half-saturation constant for PO_4	$R_{\text{PO}_4}^{\text{PHY}}$	$10^{-7} \text{ kmol P m}^{-3}$
half-saturation constant for Si(OH)_4	$R_{\text{Si(OH)}_4}^{\text{PHY}}$	$10^{-6} \text{ kmol P m}^{-3}$
Fe:P uptake ratio	$R_{\text{Fe:P}}$	$3 \times 10^{-6} \times 122. \text{ mol Fe mol P}^{-1}$
Opal:P uptake ratio	$R_{\text{Si:P}}$	$25 \text{ mol Si mol P}^{-1}$
CaCO_3 uptake ratio	$R_{\text{CaCO}_3:\text{P}}$	$35 \text{ mol C mol P}^{-1}$
remineralsation rate	$\lambda_{\text{DOM,PO}_4}^{\text{surf}}$	0.03 d^{-1}
mortality rate	$\lambda_{\text{PHY,DET}}^{\text{surf}}$	0.008 d^{-1}
exudation rate	$\lambda_{\text{PHY,DOM}}$	0.03 d^{-1}
Zooplankton		
max grazing rate	μ_{ZOO}	$1. \text{ d}^{-1}$
half-saturation constant for grazing	K_{ZOO}	$4 \times 10^{-8} \text{ kmol P m}^{-3}$
mortality rate	$\lambda_{\text{ZOO}}^{\text{surf}}$	$5. \text{ m}^3 \text{ kmol}^{-1} \text{ d}^{-1}$
excretion rate	$\lambda_{\text{ZOO,DOM}}$	0.03 d^{-1}
Deep Ocean		
Detritus remineralisation rate	$\lambda_{\text{DET,PO}_4}^{\text{deep}}$	0.003 d^{-1}
DOC remineralisation rate	$\lambda_{\text{DOM,PO}_4}^{\text{deep}}$	0.004 d^{-1}
Opal dissolution rate	$\lambda_{\text{OPAL,Si(OH)}_4}$	0.005 d^{-1}
CaCO_3 dissolution rate	$\lambda_{\text{CaCO}_3,\text{DIC}}$	0.075 d^{-1}
Phytoplankton mortality rate	$\lambda_{\text{PHY,DET}}^{\text{deep}}$	0.1 d^{-1}
Zooplankton mortality rate	$\lambda_{\text{ZOO,DET}}^{\text{deep}}$	0.2 d^{-1}
Detritus sinking speed	W_{POC}	5 m d^{-1}
Opal sinking speed	$W_{\text{Si(OH)}_4}$	80 m d^{-1}
CaCO_3 sinking speed	W_{CaCO_3}	30 m d^{-1}

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Table 2. Atmospheric CO₂ level for 2007 and mean 5-year (2003–2007) global air-sea CO₂ flux and POC export production for the reference run SYN+EMS and the sensitivity studies IRON, ABS, DIAPYC.

	Atm. CO ₂ (ppm)	Air-Sea Flux (Pg C yr)	Export Production (Pg C yr)
SYN+EMS	382.1	−2.91	13.0
IRON	408.9	−2.22	8.3
ABS	396.5	−2.53	10.5
DIAPYC	387.4	−2.76	15.3

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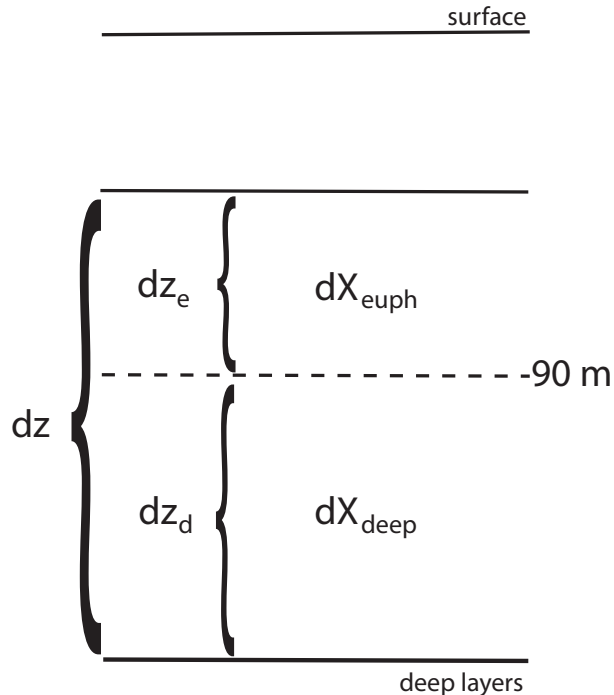


Fig. 1. Schematic diagramme illustrating the virtual split of the model layer containing the 90 m deep lower boundary of the euphotic zone. Please note that this 90 m boundary may lie either within the mixed layer or below the mixed layer and a number of thin or massless layers, not necessarily in the second layer below the surface.

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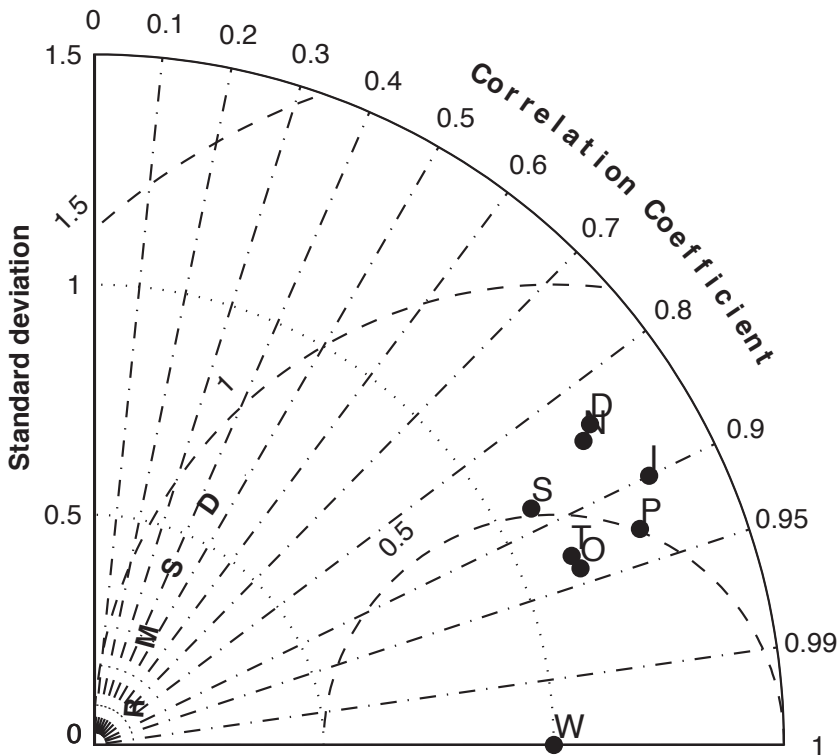


Fig. 2. Taylor diagramme (Taylor, 2001) comparing simulated mean (1978–2007) and observed 3-D fields weighted by volume. *T* potential temperature, *S* salinity, *P* phosphate, *N* nitrate, *I* silicate, *O* oxygen, *D* DIC, *W* denotes the observations. Climatologies for temperature, salinity, phosphate, nitrate, silicate, and oxygen are from WOA05 (Boyer et al., 2006), those for DIC from GLODAP (Sabine et al., 2005). Standard deviations were normalised to combine the different variables in one diagramme.

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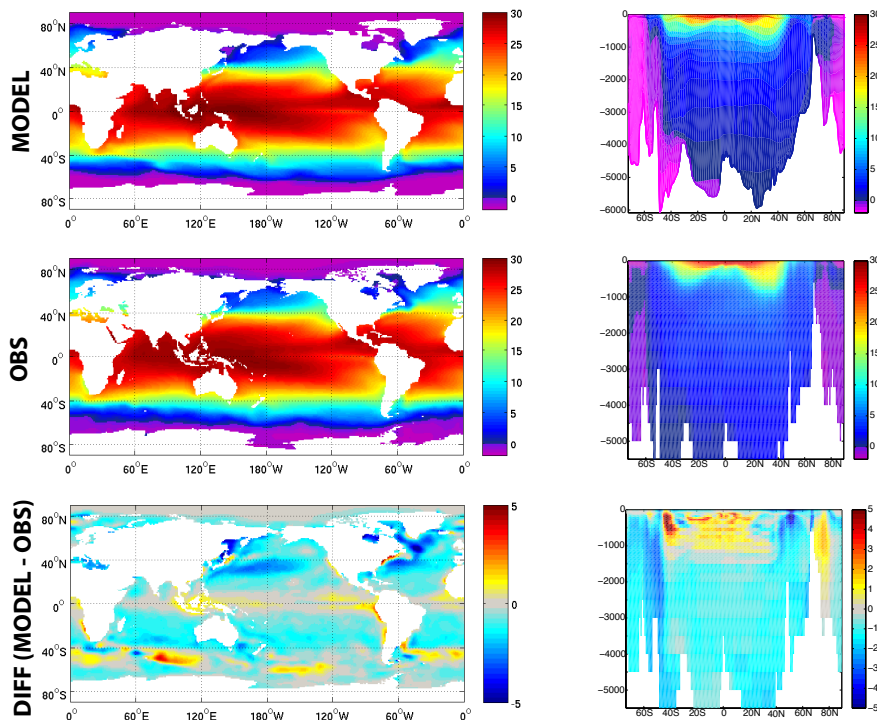



Fig. 3. Surface distributions (left) and meridional sections through the western Atlantic (right) of simulated (top), and observed (middle) potential temperature ($^{\circ}\text{C}$) and their difference (bottom). Model results are a mean 1978–2007 and observations from the same sources as in Fig. 2. Model results were interpolated onto the same $1^{\circ}\times 1^{\circ}$ grid as the observations for the comparison.

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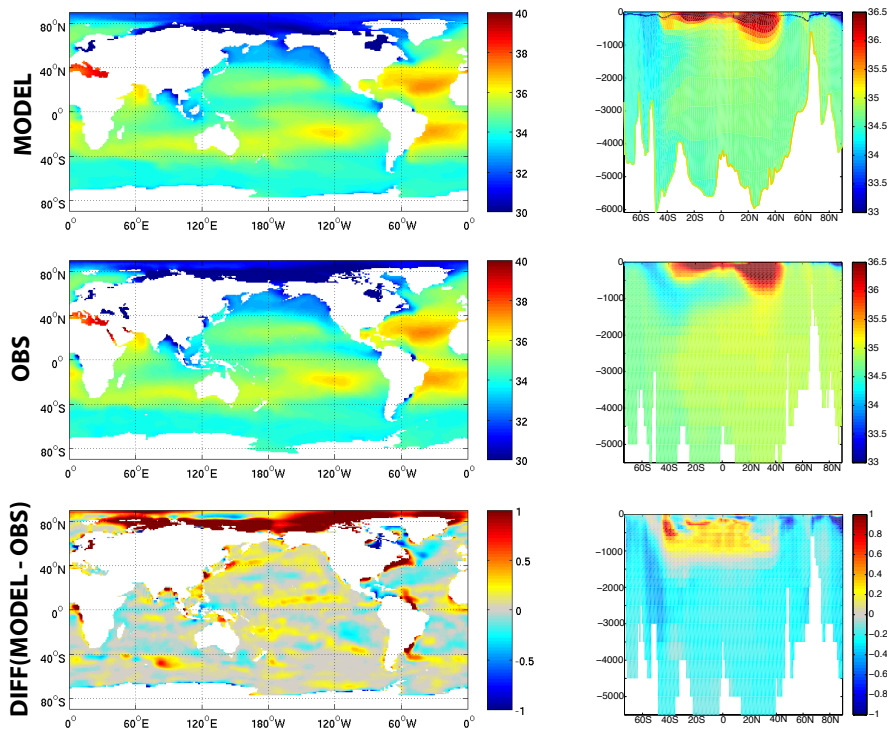


Fig. 4. As Fig. 3, but for salinity (psu).

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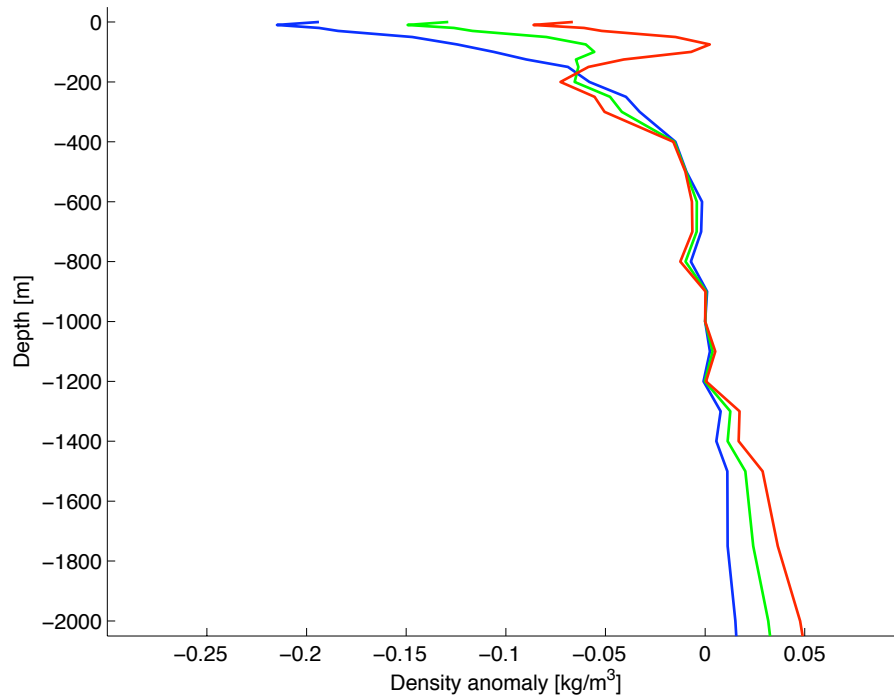


Fig. 5. Potential density profiles with reference pressure of 0 db (blue line), 1000 db (green line), and 2000 db (red line) in the Weddell Sea (70° S–60° S, 40° W–0° W). The profiles are computed using August data from Levitus and Boyer (1994) and Levitus et al. (1994) and the density anomaly is relative to density at 1000 m.

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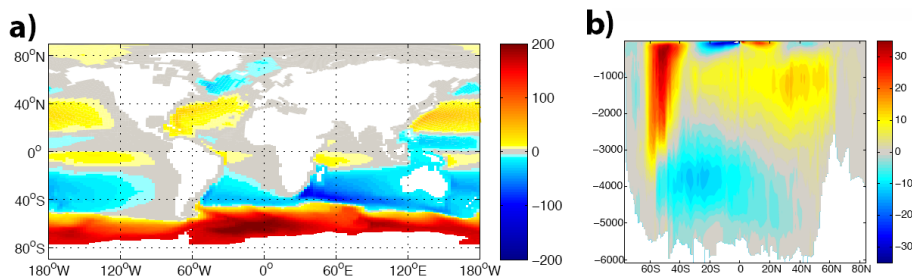


Fig. 6. Simulated mean (1978–2007) global circulation pattern. **(a)** Barotropic stream function in Sv ($10^6 \text{ m}^3 \text{ s}^{-1}$). **(b)** Global meridional overturning circulation in Sv.

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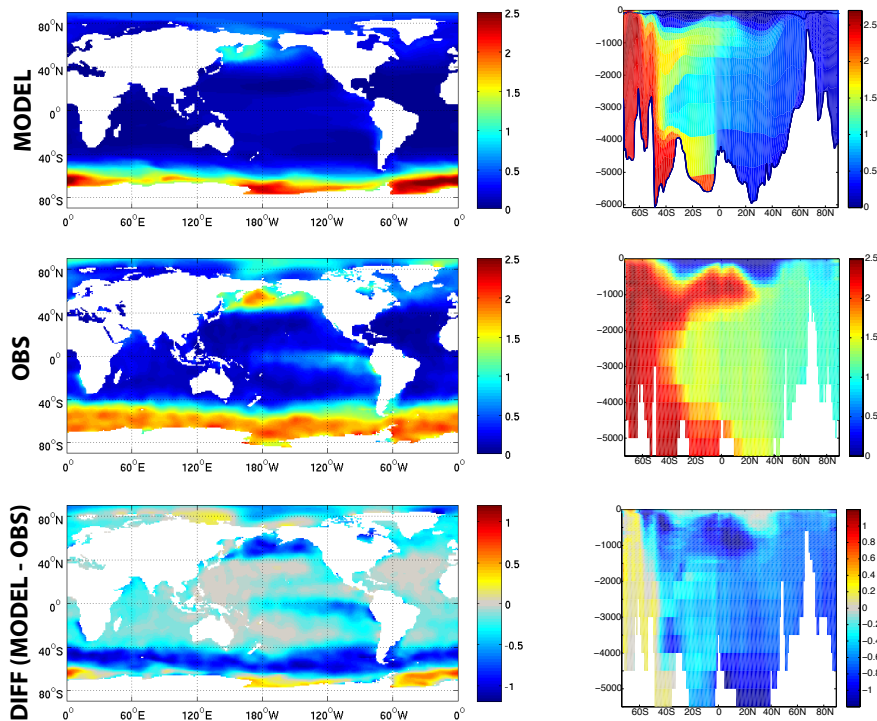


Fig. 7. As Fig. 3, but for phosphate ($\mu\text{mol l}^{-1}$).

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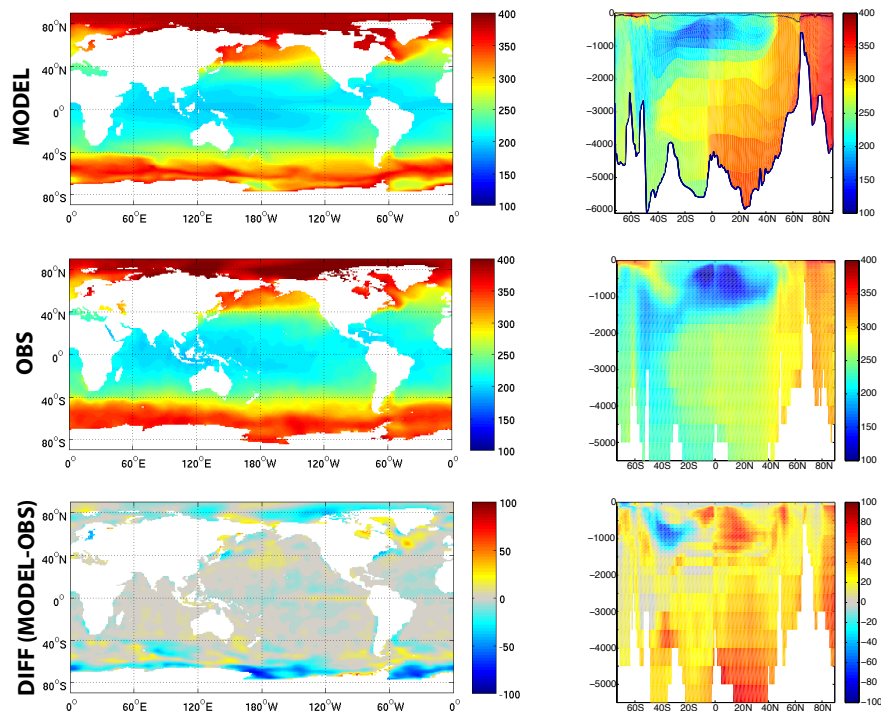


Fig. 8. As Fig. 3, but for oxygen ($\mu\text{mol l}^{-1}$).

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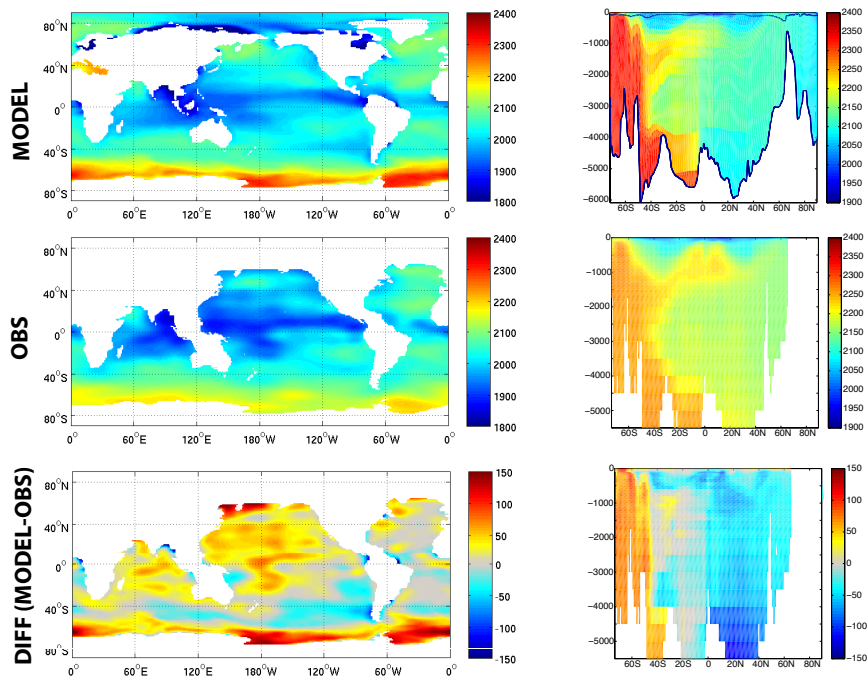


Fig. 9. As Fig. 3, but for DIC ($\mu\text{mol l}^{-1}$).

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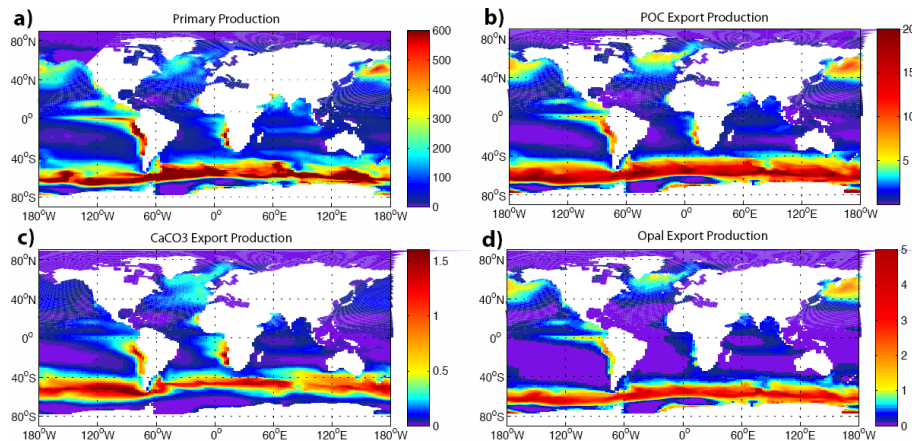


Fig. 10. Mean (1978–2007) simulated (a) primary production ($\text{mol C m}^{-2} \text{yr}^{-1}$), (b) POC export production ($\text{mol C m}^{-2} \text{yr}^{-1}$), (c) CaCO_3 export ($\text{mol C m}^{-2} \text{yr}^{-1}$), and (d) opal export ($\text{mol Si m}^{-2} \text{yr}^{-1}$).

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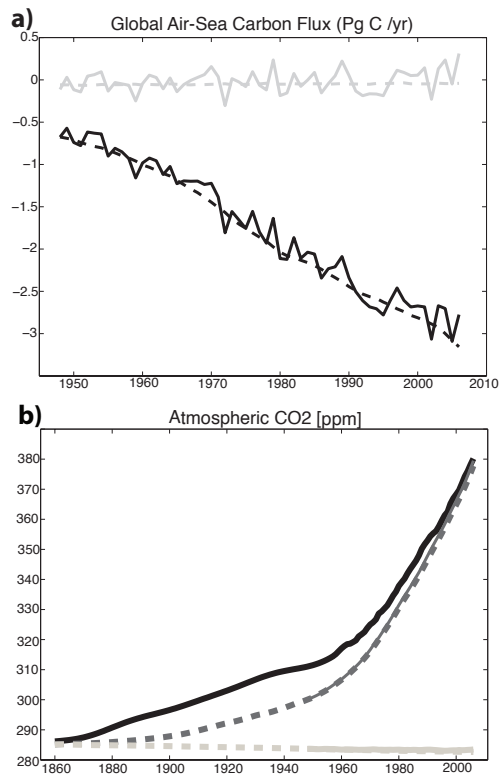


Fig. 11. (a) Time series of annual mean globally integrated air-sea CO_2 flux (Pg C yr^{-1} , negative downward). Black dashed CLIM+EMS, black solid SYN+EMS, light grey dashed CLIM, light grey solid SYN. (b) Time series of annual mean atmospheric CO_2 concentrations 1860–2007 (ppm). Thick black line observed, dark grey dashed CLIM+EMS, dark grey solid SYN+EMS, light grey dashed CLIM, light grey solid SYN.

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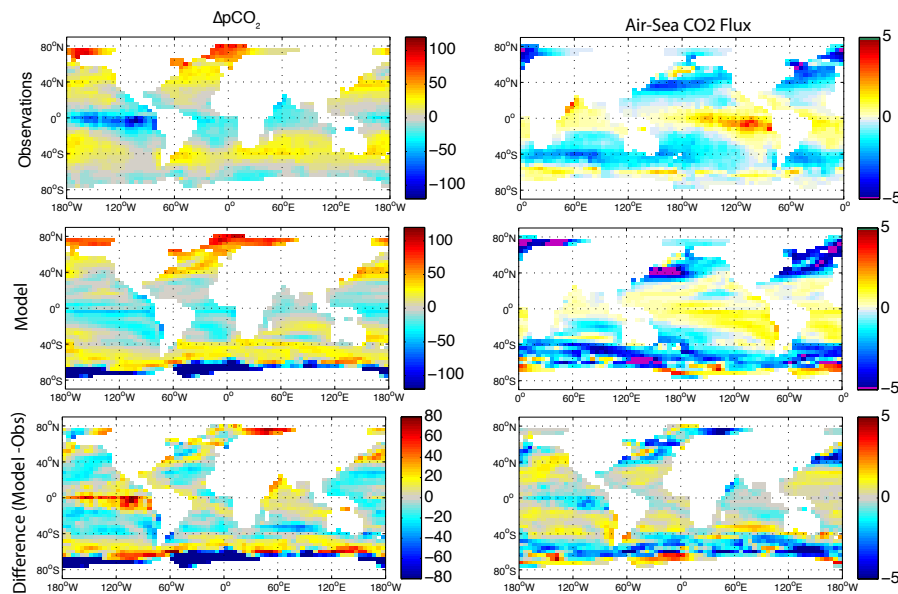


Fig. 12. Air-sea disequilibrium ($\Delta p\text{CO}_2$, ppm, left column) and CO_2 fluxes (positive upward, $\text{mol C m}^{-2} \text{yr}^{-1}$, right column). Observations from Takahashi et al. (2008), referenced to 2000 (top), simulated mean for year 2000 (middle) and their difference (bottom). Model results were interpolated to the $4^\circ \times 5^\circ$ grid used by Takahashi et al. (2008).

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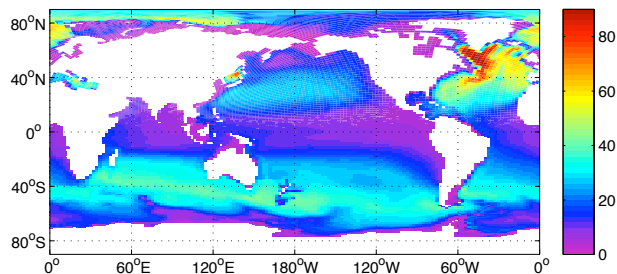


Fig. 13. Simulated column inventory of anthropogenic CO₂ in the ocean (mol m⁻²) for 1994 in the SYN+EMS experiment.

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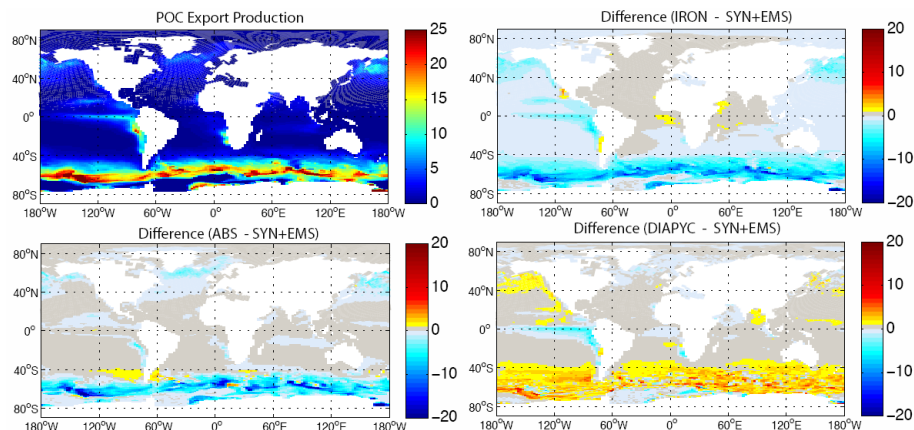


Fig. 14. POC export ($\text{mol C m}^{-2}\text{yr}^{-1}$, left) for 2007 for SYN+EMS as a reference and differences from this for IRON, ABS, and DIAPYC sensitivity experiments.

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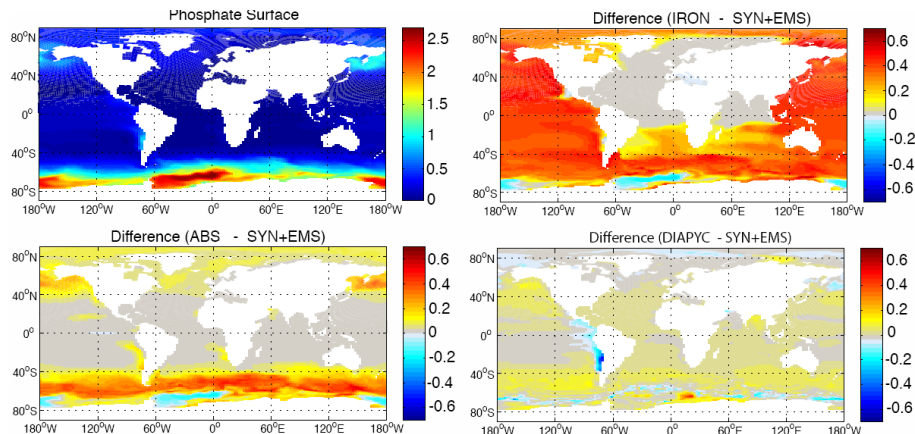


Fig. 15. Annual mean phosphate surface concentration ($\mu\text{mol l}^{-1}$) for 2007 for SYN+EMS as a reference and differences from this for IRON, ABS, and DIAPYC sensitivity experiments.

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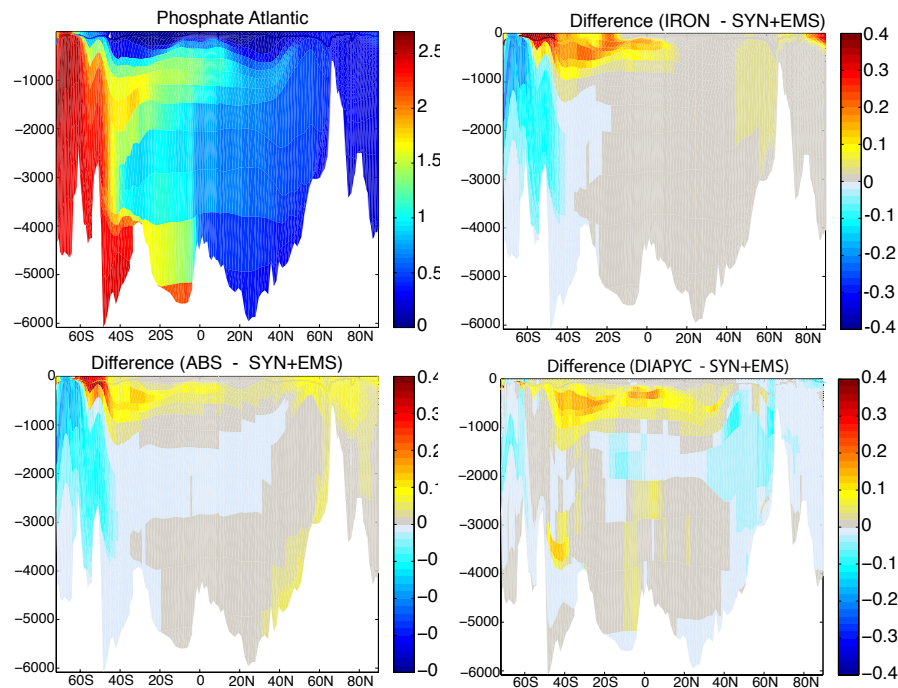


Fig. 16. Annual mean phosphate concentration along a meridional section in the western Atlantic ($\mu\text{mol l}^{-1}$) for 2007 for SYN+EMS as a reference and differences from this for IRON, ABS, and DIAPYC sensitivity experiments.

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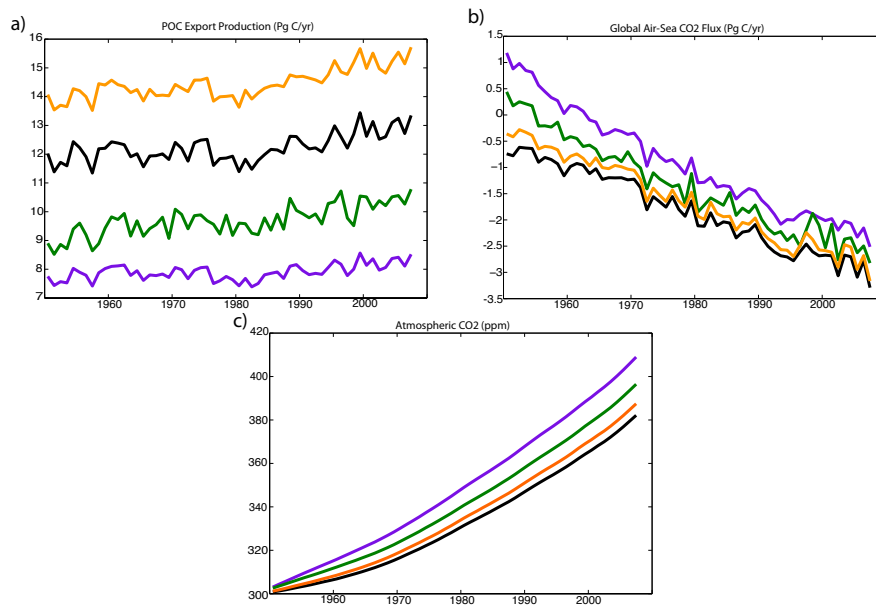


Fig. 17. Annual mean time series of **(a)** global export production (Pg C yr^{-1}), **(b)** global air-sea CO_2 flux (Pg C yr^{-1}) and **(c)** atmospheric CO_2 concentration (ppm) for SYN+EMS (black), IRON (purple), ABS (green) and DIAPYC (orange).

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